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# High-spin → low-spin relaxation kinetics and metal dilution effects in 1D spin-crossover chain compounds

Chérif Baldé<sup>a,\*</sup>, Mame Seyni Sylla<sup>a</sup>, Cédric Desplanches<sup>b</sup>, Guillaume Chastanet<sup>b</sup>

<sup>a</sup>Laboratoire de Chimie et Physique des Matériaux (LCPM), Université Assane Seck de Ziguinchor, Senegal

<sup>b</sup>CNRS, Univ. Bordeaux, ICMCB UPR9048, 87 Avenue du Dr. Schweitzer, F-33600 Pessac, France

## A B S T R A C T

Light Induced Excited Spin State Trapping (LIESST) is a well known phenomenon in spin crossover (SCO) materials, studied since 1982 in solution and 1984 in the solid state, as it offers some interesting prospects in data storage. In this paper we discuss the influence of metal dilution on the lifetime of the photo induced state in two 1D chain solid solution series:  $[\text{Fe}_x\text{Zn}_{1-x}(\text{btzp})_3](\text{BF}_4)_2$  and  $[\text{Fe}_x\text{Zn}_{1-x}(\text{endi})_3](\text{BF}_4)_2$  (**btzp** 1,2 bis(tetrazol 1 yl)propane and **endi** 1,2 bis(tetrazol 1 yl)ethane). For each composition, we have determined the thermal spin transition temperature,  $T_{1/2}$ , and the relaxation temperature of the photoinduced high spin state,  $T(\text{LIESST})$ . A detailed analysis of the relaxation kinetics allowed a discussion of the evolution of this lifetime as a function of metal dilution.

### Keywords:

Spin Crossover (SCO)

Dilution effects

Light-induced spin switching

Metastable state lifetime

Molecular switches

## 1. Introduction

The recent developments in technology in which molecules or assemblies of molecules are used for information processing has enabled scientists to investigate their potential in several areas, namely for the development of display and memory devices [1]. Among them, the materials that undergo a spin crossover (SCO) phenomena upon application of external stimuli, such as temperature, pressure, light and electric field [2–6], have received a high interest within the construction of molecular devices with optical switch or/and magneto optical storage properties [7–17]. In this context, SCO coordination compounds of a  $3d^4$  to  $3d^7$  electronic configuration, displaying a central metallic ion in an octahedral environment, have been the focus of research for nearly 80 years [1–6]. Several aspects of the SCO phenomenon have been reviewed in the past [2, 4, 16–20]. Of these, by far the majority are pseudo octahedral  $3d^6$  iron(II) complexes, thus, as a function of the ligand field strength, the six electrons of the iron(II) ion can occupy the 3d orbitals in two different ways, giving rise to two different stable states: a diamagnetic low spin (LS) electronic configuration state  $t_{2g}^6e_g^0$  ( $^1A_{1g}$ ,  $S = 0$ ) in the case of a strong ligand field and a paramagnetic high spin (HS) electronic state  $t_{2g}^4e_g^2$  ( $^5T_{2g}$ ,  $S = 2$ ) when the ligand field is weak. In the case of an intermediate ligand field,

the two spin states may become so close in energy that a small external perturbation can switch one state into the other.

LS ↔ HS conversions are accompanied by profound changes in all the properties that depend on the distribution of the 3d valence electrons. Besides the most obvious change of magnetic properties [21], including switchable diamagnetism ↔ paramagnetism [10], the reversible change of color [10], refractive index [22], electrical conductivity, [23] luminescence, [24] non linear optical [25] and mechanical properties [26] may accompany the SCO. The bistability of SCO materials can be addressed with the application of various external stimuli, like temperature [5,27], pressure [28–30], magnetic field [31,32] or light irradiation [5,27,33,34]. The optical switching is particularly regarded since ultrafast switches (femtoseconds) can be achieved [15]. Light can reversibly write and erase the LS and HS states according to the direct or reverse light induced excited spin state trapping (LIESST) effects [34(d),35]. In 1998, Létard et al. introduced the  $T(\text{LIESST})$  measurement, which aims to estimate the limiting temperature above which the photoinduced HS metastable state is erased [19,36]. From the determination of the  $T(\text{LIESST})$  value of several compounds, a comparison has been made in order to draw trends for chemists to increase the lifetime of the photo induced state [37]. In particular, the correlation between  $T(\text{LIESST})$  and  $T_{1/2}$  was evidenced based on a  $[T_{1/2}; T(\text{LIESST})]$  database. From this  $[T_{1/2}; T(\text{LIESST})]$  database [36,37], Létard et al. observed a linear dependence between  $T(\text{LIESST})$  and  $T_{1/2}$  and extracted the empirical relationship  $T(\text{LIESST}) = T_0 - 0.3 T_{1/2}$  where  $T_0$  is the initial value of the linear function. This phenomenological relationship appears to

\* Corresponding author.

E-mail address: cbalde@univ-zig.sn (C. Baldé).

reflect rearrangement processes in the coordination sphere accompanying the SCO at the metal center. It is well documented that the crystal packing obviously plays a key role both on the thermal spin crossover feature, as often and early evidenced [19,20,38–41], and on the lifetime of the metastable state.

Dilution has proven to be an elegant way to learn more about cooperativity effects [42–46]. In fact, the first effect of dilution is to weaken the propagation of the elastic interactions between the Fe(II) SCO entities and, therefore, the cooperativity. Dilution also induces an internal pressure, positive or negative, depending on the ionic radius of the guest ion with respect to that of the HS M(II) ion. Dilution can be defined as the incorporation of isostructural M(II) complexes which do not exhibit SCO behavior towards the Fe(II) SCO crystal lattice. To date dilution has appeared very useful in understanding the thermal spin crossover and the relaxation behavior, but it is also a powerful tool to stabilize the photo induced metastable HS state [22,47]. This finding stimulates a true challenge towards applications in photonic devices, i.e. to identify the parameters affecting the lifetime of the photo induced excited state, with the ultimate goal to operate at room temperature [36].

In previous works we have studied extensively the effect of metal dilution on the LIESST temperature in mixed  $[\text{Fe}_x\text{M}_{1-x}]$  zero dimensional mononuclear complexes with various metal ions [45,46]. Whilst at a first glance the substitution of Fe(II) by Zn(II) ions appeared to not influence the  $T(\text{LIESST})$ , a comparison of several other ions, such as Mn(II), Cd(II) and Ni(II), revealed a different picture. The use of anions bigger than the Fe(II) ion in its HS state induced an increase of the  $T(\text{LIESST})$  of several degrees. On the contrary, insertion of Ni(II) ions, that is exactly inbetween Fe(II) LS and Fe(II) HS, induced a slight decrease of  $T(\text{LIESST})$  [46g]. The effect of internal pressure, that has been widely documented on the thermal SCO, is also effective on the relaxation temperature  $T(\text{LIESST})$  and the lifetime of the photo induced state.

The goal of the present study is to investigate the influence of isomorphous metal dilution on the photo induced state's lifetime in SCO compounds of higher dimensionality than molecular complexes, where the iron centers are linked together by covalent bonds, as a complement to our previous study on the relaxation temperature [46b]. We chose the two 1D compounds,  $[\text{Fe}_x\text{Zn}_{1-x}(\text{btzp})_3](\text{BF}_4)_2$  and  $[\text{Fe}_x\text{Zn}_{1-x}(\text{endi})_3](\text{BF}_4)_2$  ( $\text{btzp} = 1,2$  bis(tetrazol-1-yl)propane and  $\text{endi} = 1,2$  bis(tetrazol-1-yl)ethane), and diluted them gradually with zinc, which is appropriate for the formation of isomorphous mixed crystal series with iron(II) ions. The results are discussed on the basis of a kinetic study governing the photo induced back conversion. The two ligands  $\text{endi}$  and  $\text{btzp}$  are depicted in Scheme 1.

## 2. Experimental

### 2.1. Chemicals

*General:* All syntheses were carried out under a nitrogen atmosphere. The syntheses of the ligands  $\text{btzp}$  [1,2 bis(tetrazol-1-yl)propane] and  $\text{endi}$  [1,2 bis(tetrazol-1-yl)ethane] are described in the literature [48,49].

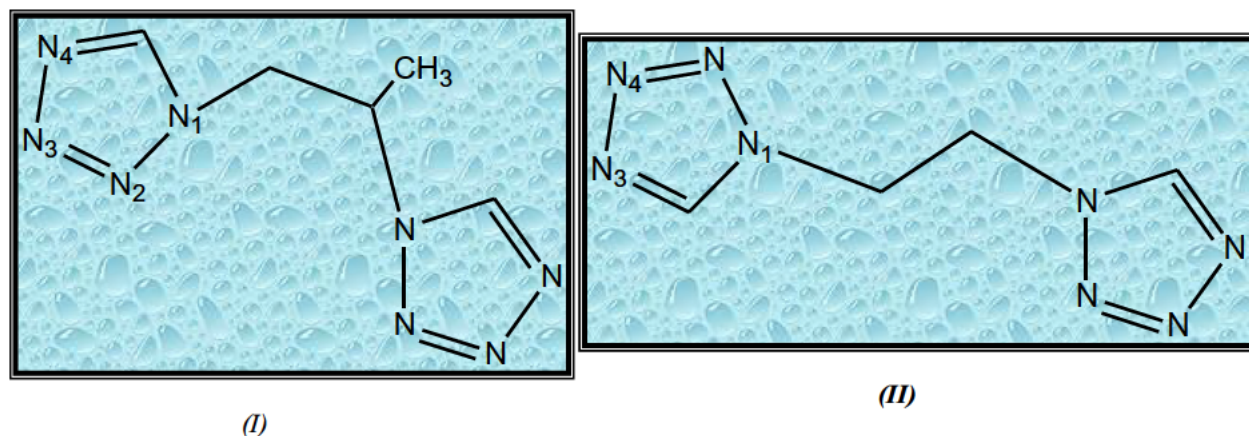
*Materials:* The ligand 1,2 bis(tetrazol-1-yl)propane [ $\text{btzp}$ ] was prepared from sodium azide, triethyl orthoformate and 1,2 diaminopropane according to the general method described by Kamiya and Saito [50]. The ligand 1,2 (tetrazol-1-yl)ethane [ $\text{endi}$ ] was prepared in a similar method to that used by Satoh et al. [51]. The preparation of the pure compounds  $[\text{Fe}(\text{btzp})_2](\text{BF}_4)_2$  and  $[\text{Fe}(\text{endi})_2](\text{BF}_4)_2$  has been described earlier by Gutlich et al. [48] and Linert et al. [49]. The diluted compounds  $[\text{Fe}_x\text{Zn}_{1-x}]$  were synthesised according to the same procedures, replacing iron(II) tetrafluoroborate by mixtures of iron(II) and zinc(II) in a given ratio. The iron fraction ( $x$ ) values were calculated from the iron and zinc atomic percentage determined by quantitative analysis. Elemental analyses of C, H, N and F were performed and the results agreed well with the calculated data (See Tables 1 and 2). The slight difference in values (about 1 for some doped compounds) may be due to minor errors made during the estimation of the iron and zinc atomic percentage, as determined by a microprobe spectrometer.

### 2.2. X ray powder diffraction

Powder X ray diffraction measurements were conducted at room temperature in the range  $2\theta = 10$ – $35^\circ$  for all compounds. Above  $35^\circ$ , the peak intensity was too weak to be relevant compared to the background. The complete diffraction patterns of all the compounds are shown in Fig. 1. The presence of well defined peaks proves that these samples consisted of highly crystallized phases. Moreover, the positions of all of the (hkl) reflections match for all compounds of the  $[\text{Fe}_x\text{Zn}_{1-x}(\text{btzp})]$  and  $[\text{Fe}_x\text{Zn}_{1-x}(\text{endi})]$  series. This observation shows that all these compounds are isostructural, without miscibility issues.

### 2.3. Magnetic and photomagnetic studies

Magnetic susceptibility data were collected using a Quantum Design MPMS 5 SQUID magnetometer under an applied field of 1 T. All measurements were performed on homogenous polycrystalline samples and in the temperature range 10 to 290 K. Diamag



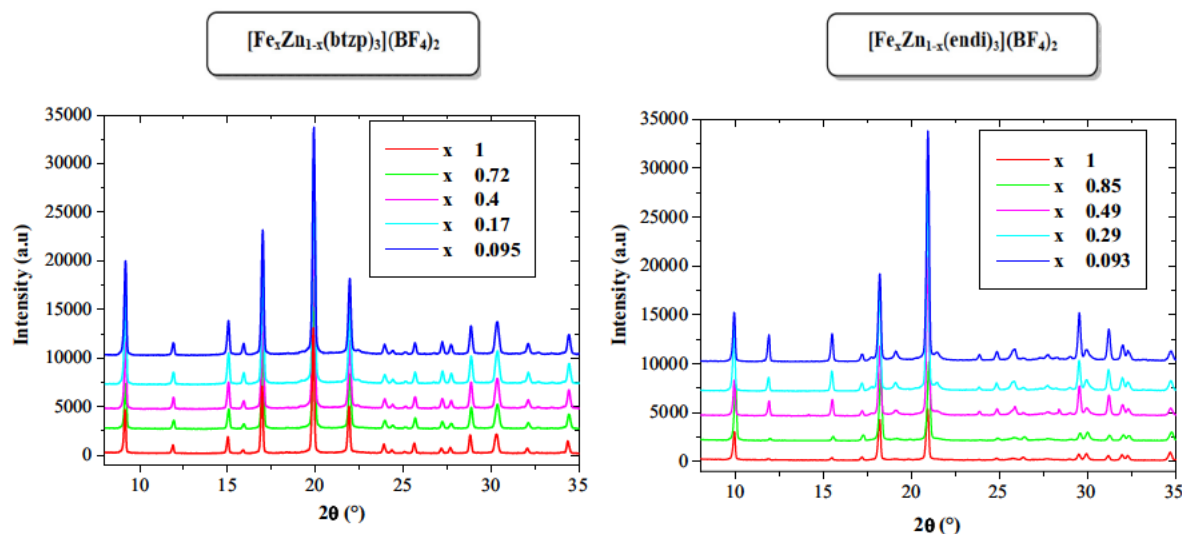
Scheme 1. Derivatives of the bis(tetrazol) ligand used in the present study. (I) 1,2-bis(tetrazol-1-yl)propane ( $\text{btzp}$ ) and (II) 1,2-bis(tetrazol-1-yl)ethane ( $\text{endi}$ ).

**Table 1**  
Elemental analysis for the mixed-crystal complexes:  $[\text{Fe}_x\text{Zn}_{1-x}(\text{btzp})_3](\text{BF}_4)_2$ .

Compound	Experimental (% weight)			Calculated (% weight)		
	C	H	N	C	H	N
1: $[\text{Fe}(\text{btzp})_3](\text{BF}_4)_2$	23.17	3.10	43.40	23.40	3.14	43.66
2: $[\text{Fe}_{0.72}\text{Zn}_{0.28}(\text{btzp})_3](\text{BF}_4)_2$	23.10	3.08	43.33	23.32	3.13	43.51
3: $[\text{Fe}_{0.4}\text{Zn}_{0.6}(\text{btzp})_3](\text{BF}_4)_2$	22.50	3.02	42.11	23.23	3.12	43.34
4: $[\text{Fe}_{0.17}\text{Zn}_{0.83}(\text{btzp})_3](\text{BF}_4)_2$	22.21	3.01	42.08	23.16	3.11	43.22
5: $[\text{Fe}_{0.095}\text{Zn}_{0.905}(\text{btzp})_3](\text{BF}_4)_2$	22.20	3.00	42.01	23.14	3.11	43.18

**Table 2**  
Elemental analysis for the mixed-crystal complexes:  $[\text{Fe}_x\text{Zn}_{1-x}(\text{endi})_3](\text{BF}_4)_2$ .

Compound	Experimental (% weight)			Calculated (% weight)		
	C	H	N	C	H	N
1: $[\text{Fe}(\text{endi})_3](\text{BF}_4)_2$	19.70	2.49	45.53	19.80	2.49	46.18
2: $[\text{Fe}_{0.85}\text{Zn}_{0.15}(\text{endi})_3](\text{BF}_4)_2$	19.75	2.48	45.68	19.76	2.49	46.09
3: $[\text{Fe}_{0.49}\text{Zn}_{0.51}(\text{endi})_3](\text{BF}_4)_2$	19.58	2.46	45.38	19.67	2.47	45.87
4: $[\text{Fe}_{0.29}\text{Zn}_{0.71}(\text{endi})_3](\text{BF}_4)_2$	19.52	2.45	45.29	19.62	2.47	45.75
5: $[\text{Fe}_{0.093}\text{Zn}_{0.907}(\text{endi})_3](\text{BF}_4)_2$	19.53	2.47	45.10	19.57	2.46	45.64



**Fig. 1.** Room temperature X-ray powder diffraction patterns of  $[\text{Fe}_x\text{Zn}_{1-x}(\text{btzp})_3](\text{BF}_4)_2$  (a) and  $[\text{Fe}_x\text{Zn}_{1-x}(\text{endi})_3](\text{BF}_4)_2$  (b) series.

netic corrections for the sample holder and the material (using Pascal constants) were applied.

Photomagnetic measurements were performed using a  $\text{Kr}^+$  laser coupled via an optical fiber to the cavity of a MPMS 55 Quantum Design SQUID magnetometer operating at 2 T. The crystalline sample was prepared in a thin layer ( $\sim 0.1$  mg) to promote full penetration of the irradiated light. The sample mass was obtained by comparison with the thermal spin transition curve measured on a larger accurately weighed polycrystalline sample. The calculated mass is of course approximate, however the adequacy between the bulk and thin layer magnetic measurements is reasonable, taking into account the fact that the data correction of such a low amount of sample is challenging. Our previously published standardized method for obtaining LIESST data was followed [36a]. The sample was first slowly cooled to 10 K, ensuring that potential trapping of HS species at low temperatures did not occur. Irradiation was carried out at a set wavelength and the power of the sample surface was adjusted to  $5 \text{ mW cm}^{-2}$ . Once photo saturation was reached, irradiation was ceased, the temperature increased at a rate of  $0.3 \text{ K min}^{-1}$  to  $\sim 100 \text{ K}$  and the magnetization measured every 1 K to determine the  $T(\text{LIESST})$  value [36,37], given by the

extreme of the  $\delta \chi_{MT}/\delta T$  versus  $T$  curve for the relaxation. The  $T(\text{LIESST})$  value describes the limiting temperature above which the light induced magnetic high spin information is erased in a SQUID cavity. In the absence of irradiation, the magnetism was also measured over the temperature range 10–290 K to follow the thermal spin transition and to obtain a low temperature baseline. In addition to the  $T(\text{LIESST})$  measurement, kinetic studies were performed by irradiating the sample at 10 K until photo saturation, then, under constant irradiation, the sample was warmed to a desired temperature around the  $T(\text{LIESST})$  region. At the desired temperature, irradiation was ceased and the decay of the magnetization signal was followed for several hours, or until complete relaxation back to the low spin baseline.

### 3. Results and discussion

#### 3.1. Magnetic characterization

The magnetic measurements were performed on polycrystalline samples and the thermal evolution of the molar magnetic suscep

tibility,  $\chi_M$ , time and the temperature were followed. The measured  $\chi_M T$  product is, in fact, the sum of the contributions of the iron(II) ions and zinc(II) ions (Eq. (1)). Based on these data and considering that the zinc(II) metal ion has a diamagnetic response in regard to the filled  $3d^{10}$  level, the iron(II) HS molar fraction,  $\gamma_{HS}$ , may be directly deduced from the  $(\chi_M T)_{Fe}$  product through Eq. (3), where  $(\chi_M T)_{Fe}^{HT}$  refers to the neat iron compound ( $x = 1$ ) at high temperatures, typically at ambient temperature.

$$\chi_M T = x(\chi_M T)_{Fe} + (1-x)(\chi_M T)_{Zn} \quad (1)$$

$$(\chi_M T)_{Fe} = \frac{\chi_M T - (1-x)(\chi_M T)_{Zn}}{x} \quad (2)$$

$$\gamma_{HS} = \frac{(\chi_M T)_{Fe}}{(\chi_M T)_{Fe}^{HT}} \quad (3)$$

In a previous paper, we have investigated the mixed  $[Fe_xZn_{1-x}(btzp)]$  and  $[Fe_xZn_{1-x}(endi)]$  compounds with the objective to study the influence of metal dilution on the  $T(LIESST)$  properties [46b]. Fig. 2 recalls the temperature dependence of the high spin fraction,  $\gamma_{HS}$ , deduced from the magnetic susceptibility measurements for different metal dilution factors,  $x$ , for the two mixed crystal series  $[Fe_xZn_{1-x}(btzp)_3](BF_4)_2$  and  $[Fe_xZn_{1-x}(endi)_3](BF_4)_2$ . As originally mentioned [46b], when  $x$  decreases, (i) the spin crossover regime becomes more gradual, (ii) the thermal spin transition is lowered, and (iii) the residual HS fraction at low temperature increases. The information given by Fig. 2 is in fact close to the original studies of Gütllich et al. [42]. All that was so far explained in the following way:

- (1) The increase of the gradual character with the metal dilution reflects the progressive loss of cooperativity, as expected from the increasing distance between active iron(II) metal centers in the diluted [Zn] lattice [5].
- (2) The shift of the thermal spin transition towards lower temperature can be understood through the change of internal pressure. The radius of the zinc(II) ion ( $r = 74$  pm) is close to the ionic radius of the iron(II) HS ion ( $r = 78$  pm) and higher than the ionic radius of the iron(II) LS ion (61 pm) [52]. Thus, if iron(II) LS ions are highly diluted in a [Zn] lattice, the zinc lattice induces a 'negative' pressure on the Fe(II) site, leading to an increase in the Fe–N bond lengths (decrease of the crystal field potential) which favors the HS state [5,53]. On the contrary, the effect of zinc lattice is

negligible when the iron(II) HS ions are highly diluted. Consequently, the thermal spin transition,  $T_{1/2}$ , is shifted towards lower temperature with increasing metal dilution. This temperature is defined as the temperature at which the high spin fraction  $\gamma_{HS}$  is equal to the mean value of  $\gamma_{HS}$  at high and low temperature.

- (3) The increase of the residual HS fraction at low temperature is also a consequence of the negative pressure occurring on the iron(II) LS ions in the highly diluted [Zn] lattice. The HS state is in fact stabilized by the [Zn] lattice.

### 3.2. Photomagnetic properties

The low  $\rightarrow$  high spin photoconversion was investigated in bulk conditions using a SQUID magnetometer coupled to a continuous wave optical source. At 10 K, the samples were irradiated with the most efficient wavelength to induce a LIESST effect, that is 532 nm. In fact, it has been shown in our previous paper [46b] that the absorbance spectra recorded exhibit a broad band at 850 nm and a sharper band at 575 nm. The band at 575 nm is attributed to both d d and MLCT transitions of the LS spin complex, namely  ${}^1A_1 \rightarrow {}^1T_1$ . On the other hand, the broad peak at 850 nm is attributed to a d d band of the HS state of the compound, namely the  ${}^5T_2 \rightarrow {}^5E$  transition. Thus, 532 nm corresponds to a shoulder in the MLCT band of the diffuse absorption spectra. After light excitation at 532 nm at 10 K, the temperature was increased in the dark to record the  $T(LIESST)$  curve. The  $\chi_M T$  product first increases upon warming from 10 to 30 K due to zero field splitting of the high spin iron(II) ion [37]. Then after a plateau, the  $\chi_M T$  value decreases sharply. The value of  $\chi_M T$  at this plateau gives information about the efficiency of the photoswitching.  $T(LIESST)$  is determined by the minimum of the derivative  $\partial(\chi_M T)/\partial T$  versus  $T$  (Inset, Figs. 3 and 4). Figs. 3 and 4 represent the temperature dependence of  $\chi_M T$  for the different mixed crystal systems  $[Fe_xZn_{1-x}(btzp)_3](BF_4)_2$  and  $[Fe_xZn_{1-x}(endi)_3](BF_4)_2$ , respectively, before and after irradiation. For all the complexes, a drastic increase of the magnetic signal under light irradiation was observed at 10 K. It should, however, be noticed that the amount of iron in the photo excited high spin state is not the same for all the compounds, and for none of the complexes it is quantitative. The level of the percentage of photo conversion at 10 K has been calculated dividing the  $\chi_M T$  value after photo irradiation by the  $\chi_M T$  value at room temperature. Depending on the compound and the metal dilution, this percentage varies from 61

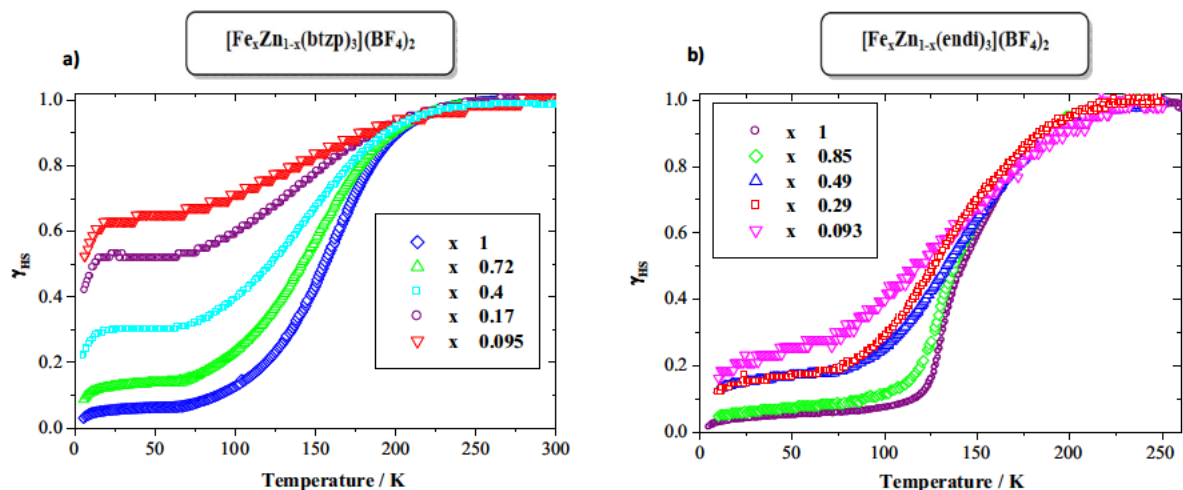


Fig. 2. Evolution as a function of  $x$  of the  $\gamma_{HS}$  vs.  $T$  curves obtained from magnetic susceptibility measurements for the mixed-crystal series (a)  $[Fe_xZn_{1-x}(btzp)_3](BF_4)_2$  and (b)  $[Fe_xZn_{1-x}(endi)_3](BF_4)_2$ .

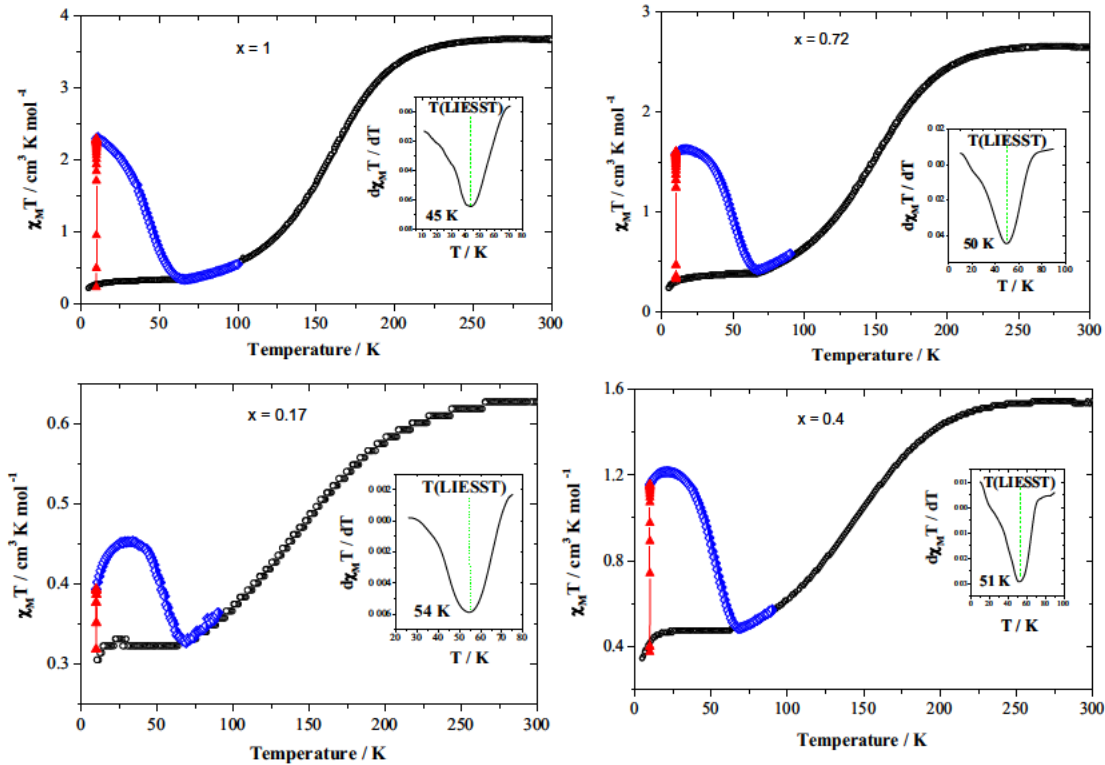
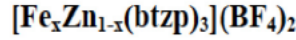


Fig. 3. Temperature dependence of  $\chi_M T$  for  $[\text{Fe}_x\text{Zn}_{1-x}(\text{btzp})_3](\text{BF}_4)_2$  ( $\circ$ ) data recorded without irradiation; ( $\blacktriangle$ ) data recorded with irradiation at 10 K; ( $\diamond$ )  $T(\text{LIESST})$  measurement, data recorded in the warming mode with the laser turned off after one hour of irradiation.

to 86%. Values of  $T(\text{LIESST})$  are reported in Table 2. For the  $[\text{Fe}_x\text{Zn}_{1-x}(\text{btzp})_3](\text{BF}_4)_2$  compounds, the change in  $T(\text{LIESST})$  is somewhat more pronounced. It increases from 46 K ( $x = 1$ ) to 50 K ( $x = 0.72$ ) and 51 K ( $x = 0.4$ ), reaching a value of 54 K ( $x = 0.17$ ). Attempts to determine a  $T(\text{LIESST})$  value for  $x = 0.095$  were unsuccessful due to the weak  $\chi_M T$  signal. For the samples of  $[\text{Fe}_x\text{Zn}_{1-x}(\text{endi})_3](\text{BF}_4)_2$ ,  $T(\text{LIESST})$  has been recorded between 61 and 63 K for  $x$  values between 1 and 0.49, and at 67 K for  $x = 0.29$ . Determining a  $T(\text{LIESST})$  value for  $x = 0.093$  was unsuccessful due to the weak  $\chi_M T$  signal.

### 3.3. Relaxation kinetics

To learn more about the influence of the kinetic parameters on the lifetime of the photo induced state in comparison to previous results, we systematically investigated the relaxation kinetics in the 40–70 K temperature range, where the HS  $\rightarrow$  LS state relaxation is thermally activated for both systems and accessible with our SQUID apparatus. Figs. 5 and 6 display the relaxation processes for  $[\text{Fe}_x\text{Zn}_{1-x}(\text{btzp})_3](\text{BF}_4)_2$  (with  $x = 1, 0.72, 0.4$ ) and  $[\text{Fe}_x\text{Zn}_{1-x}(\text{endi})_3](\text{BF}_4)_2$  (with  $x = 1, 0.85, 0.49$ ) respectively, recorded at different temperatures. Both materials show stretched exponential behavior such that incomplete relaxations are observed, even after 8 hours. Such stretched exponential shapes are typical for a broad distribution of relaxation times, which are expected to be associated with structural disorders, defects and inhomogeneities of photoexcitation. This behavior is also coherent for non cooperative systems. The relaxation curves for all the compounds under investigation have been satisfactorily fitted (solid

lines in Figs. 5 and 6) using the distribution of the activation energy involved in Eq. (4). The activation energy is supposed to have a normal (Gaussian) distribution with a standard deviation  $\sigma$  ( $E_a$ ).

$$k_{HL}(T) = k_0 + k_\infty \cdot \exp[-E_a/k_B T] \quad (4)$$

All the kinetic parameters, such as the apparent activation energy  $E_a$ , the apparent pre exponential factor,  $k_\infty$ , of the activated region, as well as the rate constant,  $k_0$ , of the quantum tunneling region, are calculated from the straight line fits given by the Arrhenius plot  $\ln k_{HL}(T)$  versus  $1/T$  (Figs. 5 and 6) and listed in Table 3. The rate constant  $k_0$  characterizes the relaxation in the quantum tunneling region and is estimated as an upper limit from the last complete kinetic recorded at the lowest temperature and was estimated to be around  $10^{-4} \text{ s}^{-1}$ . We observed that for the (btzp) series,  $E_a$  increases with the dilution, whereas  $k_\infty$  hardly changes. For the (endi) series,  $E_a$  and  $k_\infty$  both increase with dilution.

### 3.4. Discussion

In the present study, we have fully investigated the kinetics of  $[\text{Fe}_x\text{Zn}_{1-x}(\text{btzp})_3](\text{BF}_4)_2$  and  $[\text{Fe}_x\text{Zn}_{1-x}(\text{endi})_3](\text{BF}_4)_2$  metal diluted materials to determine the evolution of the different kinetic parameters governing the metastable HS  $\rightarrow$  LS relaxation process (activation energy, frequency factor). According to Eq. (4), an increase of  $k_\infty$  will lead to a decrease of  $T(\text{LIESST})$ , whereas an increase of  $E_a$  will lead to an increase of  $T(\text{LIESST})$ . For the  $[\text{Fe}_x\text{Zn}_{1-x}(\text{btzp})_3](\text{BF}_4)_2$  series, we observed that  $E_a$  increases while  $k_\infty$  remains almost constant on increasing the doping level, leading to an increase of the  $T(\text{LIESST})$  values. In the case of the  $[\text{Fe}_x\text{Zn}_{1-x}$

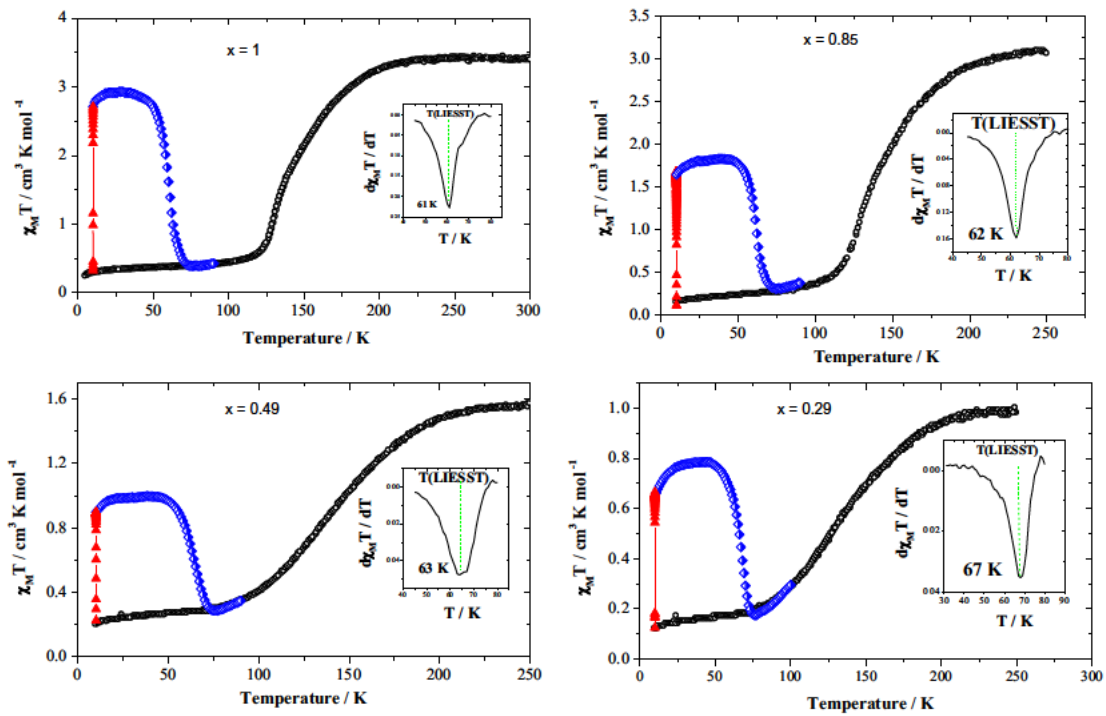
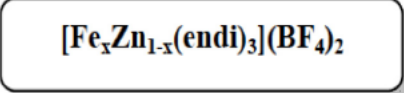


Fig. 4. Temperature dependence of  $\chi_M T$  for  $[\text{Fe}_x\text{Zn}_{1-x}(\text{endi})_3](\text{BF}_4)_2$  ( $\circ$ ) data recorded without irradiation; ( $\blacktriangle$ ) data recorded with irradiation at 10 K; ( $\diamond$ )  $T(\text{LIESST})$  measurement, data recorded in the warming mode with the laser turned off after one hour of irradiation. The insets show the derivate of the  $d\chi_M T/dT$  vs.  $T$  curves, whose minimum corresponds to the  $T(\text{LIESST})$  value.

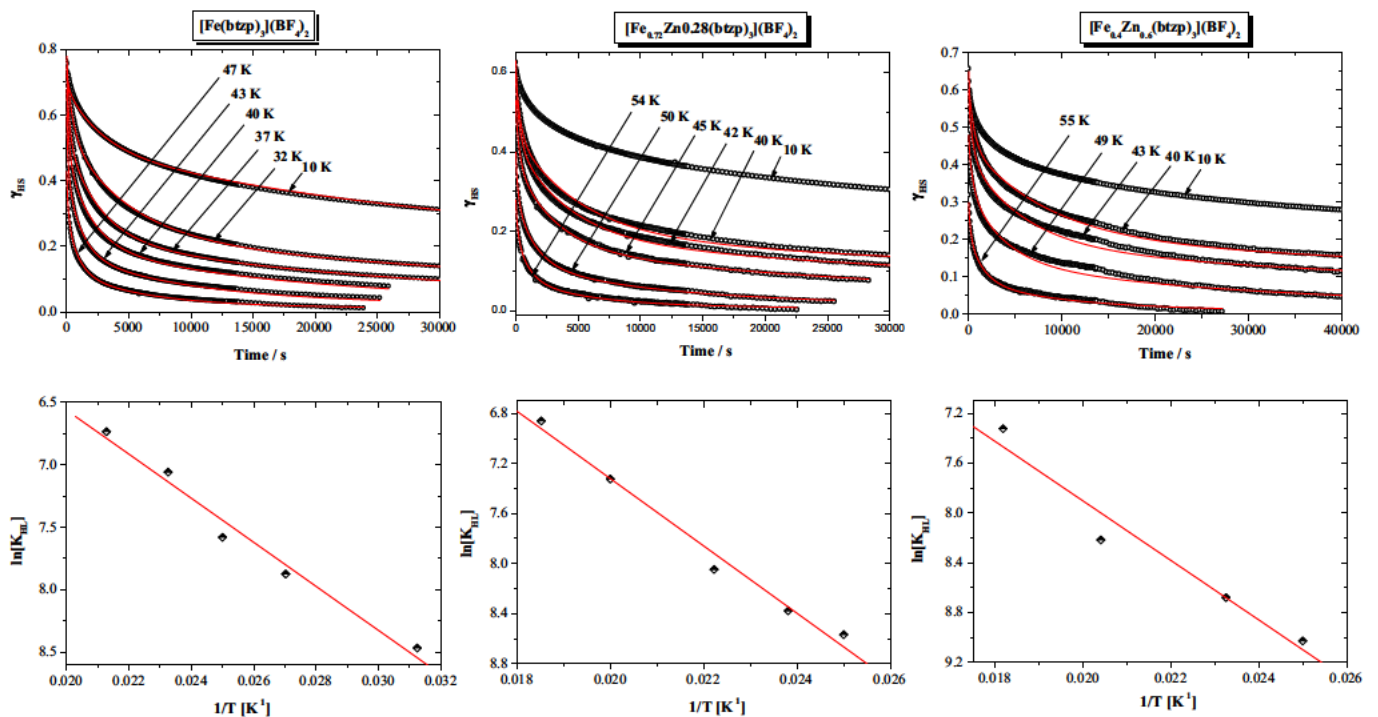


Fig. 5. HS  $\rightarrow$  LS relaxation at various temperatures for the mixed  $[\text{Fe}_x\text{Zn}_{1-x}(\text{btzp})_3](\text{BF}_4)_2$  systems ( $x = 1, 0.72$  and  $0.4$ ). The solid lines are the fits obtained with the models described in the text. The corresponding Arrhenius plot of  $\ln(k_{\text{HS}})$  vs.  $1/T$  are given in front. The solid lines represent the best linear simulations.

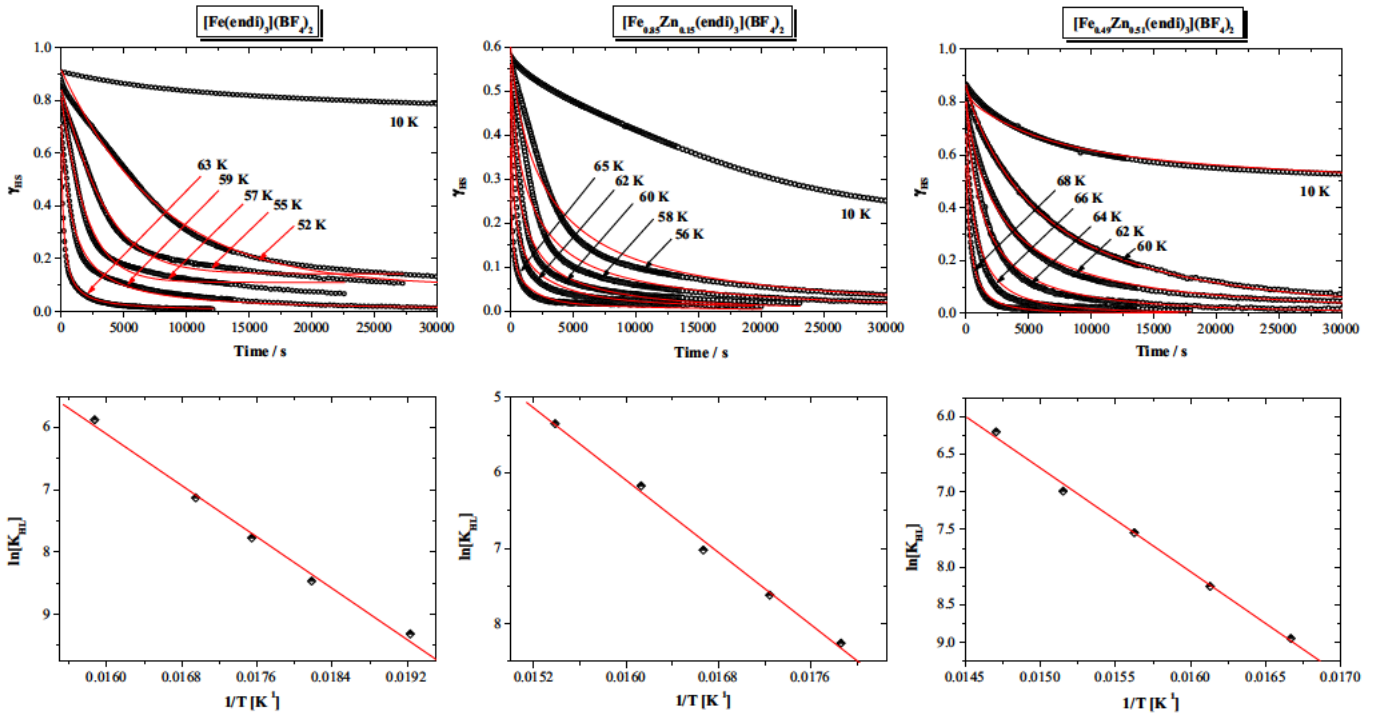


Fig. 6. HS  $\rightarrow$  LS relaxation at various temperatures for the mixed  $[\text{Fe}_x\text{Zn}_{1-x}(\text{endi})_3](\text{BF}_4)_2$  systems ( $x = 1, 0.85$  and  $0.49$ ). The solid lines are the fits obtained with the models described in the text. The corresponding Arrhenius plot of  $\ln(k_{\text{HS}})$  vs.  $1/T$  are given in front. The solid lines represent the best linear simulations.

**Table 3**  
Magnetic and photomagnetic properties of the mixed  $\text{Fe}_x\text{Zn}_{1-x}(\text{btzp})_3(\text{BF}_4)_2$  and  $\text{Fe}_x\text{Zn}_{1-x}(\text{endi})_3(\text{BF}_4)_2$  series.  $T_{1/2}$  is the temperature at which the sample contains 50% of LS and HS molecules.  $T(\text{LIESST})$  is the temperature at which the light-induced HS information was erased during warming from 10 K at a rate of  $0.3 \text{ K min}^{-1}$ . Kinetic parameters as defined in Eqs. (5) and (6) are reported.

SCO systems	$T_{1/2}$ (K)	$T(\text{LIESST})$ (K)	$k_{\infty}$ ( $\text{s}^{-1}$ )	$E_a$ ( $\text{cm}^{-1}$ )	$\sigma(E_a)$ ( $\text{cm}^{-1}$ )
$[\text{Fe}(\text{btzp})_3](\text{BF}_4)_2$	158	46	$5 \times 10^{-2}$	122	10
$[\text{Fe}_{0.72}\text{Zn}_{0.28}(\text{btzp})_3](\text{BF}_4)_2$	150	50	$7 \times 10^{-2}$	175	13
$[\text{Fe}_{0.4}\text{Zn}_{0.6}(\text{btzp})_3](\text{BF}_4)_2$	142	51	$6 \times 10^{-2}$	180	10
$[\text{Fe}_{0.17}\text{Zn}_{0.83}(\text{btzp})_3](\text{BF}_4)_2$	135	54	/	/	/
$[\text{Fe}(\text{endi})_3](\text{BF}_4)_2$	143	61	$3.2 \times 10^4$	717	33
$[\text{Fe}_{0.85}\text{Zn}_{0.15}(\text{endi})_3](\text{BF}_4)_2$	142	62	$4.6 \times 10^5$	831	35
$[\text{Fe}_{0.49}\text{Zn}_{0.51}(\text{endi})_3](\text{BF}_4)_2$	138	63	$1 \times 10^6$	956	33
$[\text{Fe}_{0.29}\text{Zn}_{0.71}(\text{endi})_3](\text{BF}_4)_2$	136	67	/	/	/

$x(\text{endi})_3](\text{BF}_4)_2$  series, the situation is different. Both  $E_a$  and  $k_{\infty}$  parameters increase with dilution. It seems that the metal dilution through the increase of  $k_{\infty}$  and  $E_a$  parameters causes a kind of compensation, leading to a smaller variation of  $T(\text{LIESST})$  value as compared to the  $[\text{Fe}_x\text{Zn}_{1-x}(\text{btzp})_3](\text{BF}_4)_2$  series.

At this stage, it is interesting to compare the evolution of  $T(\text{LIESST})$  and  $T_{1/2}$  as a function of dilution and discuss the observed behaviors on the basis of the relaxation kinetic parameters. Table 3 collects all the data obtained during this study and the results are reported in Fig. 7. The experimental data of  $[\text{Fe}_x\text{Zn}_{1-x}(\text{btzp})_3](\text{BF}_4)_2$  and  $[\text{Fe}_x\text{Zn}_{1-x}(\text{endi})_3](\text{BF}_4)_2$  are close to the line  $T_0 = 100$ , in accord with the prediction of the Létard database for bidentate complexes. In other words, the values of  $T(\text{LIESST})$  and  $T_{1/2}$  obey the relation  $T(\text{LIESST}) = T_0 - 0.3 T_{1/2}$ . According to this equation, in the  $[\text{Fe}_x\text{Zn}_{1-x}(\text{btzp})_3](\text{BF}_4)_2$  series, the variation of  $T_{1/2}$  from 158 to 135 K, should lead to a variation of 7 degrees of  $T(\text{LIESST})$ . This agrees well with the current observation of an 8 K variation. In  $[\text{Fe}_x\text{Zn}_{1-x}(\text{endi})_3](\text{BF}_4)_2$ , the  $T_{1/2}$  value decreases from 143 to 136 K, leading to an expected decrease of  $T(\text{LIESST})$  of only 2 K. In this series, the agreement between experiment and the linear relation  $T(\text{LIESST}) = T_0 - 0.3 T_{1/2}$  is less good as  $T(\text{LIESST})$  varies by 6 K.

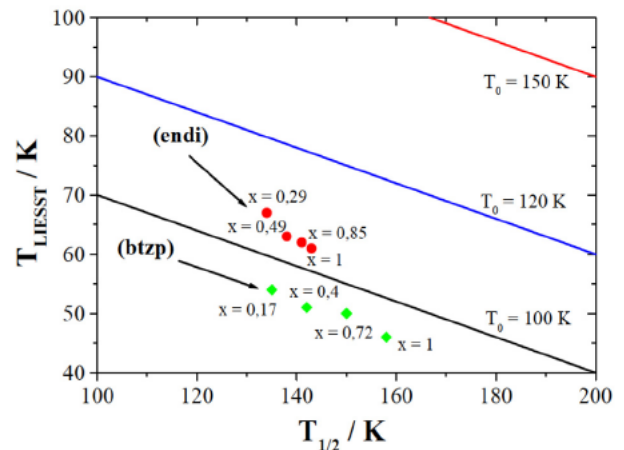


Fig. 7. HS  $\rightarrow$  LS Evolution of the  $T(\text{LIESST})$  vs.  $T_{1/2}$  for the mixed  $\text{Fe}_x\text{Zn}_{1-x}(\text{btzp})_3(\text{BF}_4)_2$  and  $\text{Fe}_x\text{Zn}_{1-x}(\text{endi})_3(\text{BF}_4)_2$  series.



This result contrasts with that obtained for the zero dimensional mononuclear compounds previously studied. Indeed, it was shown that Zn(II) dilution does not lead to any significant change of  $T(\text{LIESST})$  [46]. In the polymeric systems under study, the metal dilution does not follow this trend and moreover it does not exactly follow the  $T(\text{LIESST})$  versus  $T_{1/2}$  empirical relationship. Even if the variations look minor, they are still not understood.

One clue to explain such behavior should lie in the relaxation rate,  $k_\infty$ . Indeed, despite the great structural similarity between the two series of compounds *endi* and *btzp*,  $k_\infty$  has values that vary in magnitude. Hauser, on the basis of the SCC approach (Single Configurational Coordinate), predicted values of  $k_\infty$  of the order of  $10^6$ – $10^9$  s<sup>-1</sup> [5,27,33,54]. In SCC theory, the Fe ligand distance is taken as a reaction coordinate with a single vibration mode. Regarding the values of  $k_\infty$ , especially those relating to the *endi* and *btzp* series, it is reasonable to assume that a single reaction coordinate may not be sufficient to correctly describe the HS → LS transition. Therefore, the SCC approach may be insufficient, especially to obtain values of  $k_\infty$  of the order of  $10^2$  s<sup>-1</sup>. It becomes essential to consider a system that considers several modes of vibration to describe correctly the HS → LS relaxation. In 2006, Hauser has elsewhere proposed a model with two modes of vibration [53]. Compared to a single mode system, a system involving a low distortion mode vibration energy allows to the anticipation of low values of  $k_\infty$ . Interestingly, from a crystallographic point of view, the multimode theory, which takes into account the concept of distortion, is the most suitable for describing the spin crossover phenomenon. Indeed, a direct relationship has been established between the variation of the octahedron [FeN<sub>6</sub>] distortion,  $T_{1/2}$  and  $T(\text{LIESST})$  [20,55].

#### 4. Conclusions

In the present work, we have investigated the effects of metal dilution on thermal properties and LIESST relaxation kinetics time/temperature dependence for [Fe<sub>x</sub>Zn<sub>1-x</sub>(btzp)<sub>3</sub>](BF<sub>4</sub>)<sub>2</sub> and [Fe<sub>x</sub>Zn<sub>1-x</sub>(endi)<sub>3</sub>](BF<sub>4</sub>)<sub>2</sub> SCO materials. As a first approximation, the results related to the compounds of the two series obey roughly a linear relationship between  $T(\text{LIESST})$  and  $T_{1/2}$ . Contrary to the dilution series of zero dimensional mononuclear compounds, the iron atoms in the compounds under study are not located in isolated molecules, linked only by van der Waals forces or  $\pi$ – $\pi$  stacking, but are linked by covalent bonds. This can justify the somewhat different behaviors between the systems studied in this article and the previously studied dilution studies on zero dimensional mononuclear compounds.

As far as kinetic parameters governing the metastable HS → LS relaxation process (activation energy, frequency factor) parameters are concerned, it appears that dilution increases the energy barrier, whereas the derived frequency factor is increased in the case of the [Fe<sub>x</sub>Zn<sub>1-x</sub>(endi)<sub>3</sub>](BF<sub>4</sub>)<sub>2</sub> series and remains essentially constant in the case of the [Fe<sub>x</sub>Zn<sub>1-x</sub>(btzp)<sub>3</sub>](BF<sub>4</sub>)<sub>2</sub> series. It seems that, in both cases, the increased activation energy is not completely compensated by the higher frequency factor, so that globally  $T(\text{LIESST})$  is slightly increased with dilution.

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#### Appendix A. Supplementary data

Supplementary data to this article can be found online at <https://doi.org/10.1016/j.poly.2018.11.046>.

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