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The Mg-rich phase NdNiMg₁₅: structural and magnetic properties

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Abstract

The intermetallic NdNiMg₁₅ is the Mg-richest phase (more than 88 at% of Mg) discovered in the Mg-Nd-Ni system. Its structure was determined by X-ray diffraction on single-crystal with the following crystal data: tetragonal system, P4/*nmm*, Z = 2, *a* = 10.0602(1) Å, *c* = 7.7612(2) Å, d_{calc} = 2.40 g.cm⁻³. Its structure is made of a 3-Dimensional framework of magnesium atoms showing channels filled by 1D chain consisting of alternating Nd and Ni atoms along the *c*-axis. Antiferromagnetic ordering was observed with $T_N = 9K$, which is remarkably high considering the long distances between magnetic atoms, *i.e* Nd atoms. The effective magnetic moment μ_{eff} is equal to 3.58 μ_B which is consistent with magnetic Nd³⁺ ions and weakly or non-magnetic Ni-atoms. Below T_N , the M(H) curves show field induced metamagnetic transitions at critical fields increasing with decreasing temperatures. The magnetic structure of NdNiMg₁₅ was determined from neutron powder diffraction data by considering the propagation vector $\mathbf{k} = (\frac{1}{2} \frac{1}{2} 0)$. This magnetic structure consists in ferromagnetic chains along the *c*-axis of Nd atoms carrying moments, only separated by Ni atoms. The chains are ferromagnetically coupled within planes perpendicular to the [110] direction and these planes are antiferromagnetically coupled to neighboring planes forming a checkerboard-like magnetic structure.

Introduction

The interest in magnesium alloys has increased over the past thirty years, especially because magnesium is abundant in nature, cheap and lightweight and has very diverse applications (automobile, aerospace, hydrogen storage...). Numerous studies have been devoted to the search of new Mg-based light structural materials, mainly for automobile and aeronautic industry components^{1,2} and for biomedical applications^{3,4}. Besides these structural applications, magnesium and magnesium alloys are also good candidates for solid storage of hydrogen through the formation of metal hydrides^{5,6}. Magnesium exhibits very high hydrogen storage capacity (*i.e.* 7.6 wt.%, 110 gH/L by formation of MgH₂), but its use remains limited due to the high operating temperature (around 300°C) and the heat released by the highly exothermic absorption reaction⁵. In addition, a new interest in the hydrolysis reaction between Mg or Mg alloys and water for hydrogen generation has recently emerged. This would be an interesting way to valorize Mg-based alloys wastes^{7,8}. For all these reasons, a reinvestigation of the Mg-rich zone of many ternary (and quaternary) systems has been carried out in the last years.

In the case of structural materials, extensive studies of the RE-Zn-Mg systems close to the Mg corner have led to the discovery of a new series of so-called Long Period Stacking Ordered (LPSO) phases⁹. The strength and ductility of Mg–Y–Zn alloys are strongly enhanced when they are produced by rapid solidification processing¹⁰. After this process, intermetallic particles, so-called LPSO, precipitate and are distributed along the Mg grains boundaries. The LPSO phases are long period stacking variants of the structure of hcp-Mg and the most frequently polytypes observed crystallize with a 18R, 14H or 10H structure^{11,12}. The RE and Zn atoms are located in some of the layers and they form Zn₆RE₈ clusters. The Mg-rich corner of many RE-TM-Mg ternary systems (RE = Rare-Earth, TM = Transition Metal) have been also studied for solid hydrogen storage, in order to combine the low weight of magnesium and the catalytic effect of rare-earth and transition metal^{13,14,15}. Among these systems one can cite the La-Ni-Mg^{15,16} and Nd-Ni-Mg¹⁷⁻²¹ systems. The different studies show a decomposition of the intermetallic compounds into MgH₂, binary rareearth hydrides (e.g. LaH₃) and ternary magnesium transition metal hydrides (e.g. Mg₂NiH₄). An increase of the kinetics of the hydrogen sorption properties of Mg is observed but no real breakthrough concerning the solid storage of hydrogen was achieved. The stabilization of these ternary phases upon hydrogenation is still an interesting challenge to overcome to find new candidates for solid hydrogen storage.

The reinvestigation of many ternary RE-TM-Mg systems has led to the discovery of numerous new very-rich Mg phases and has extended our knowledge of these ternary systems. Some of these ternary phases are disordered^{22,11} or derive from binary phases^{23,24}. For instance, Gd₁₃Ni₉Mg₇₈²² crystallizes with a modulated structure which derives from an average cubic structure and

LaCuMgs²³ crystallizes in the La₂Mg₁₇–type structure with Cu atoms located in interstitial sites or partially substituted to Mg atoms on one atomic position. Among the ordered ones one can cite NdNiMg₅¹⁸, Mg₃₀Co₂Y9²⁵, Ce₂Ru₄Mg₁₇²⁶, Nd₁₆Mg9₆Ni₁₂²¹, Nd₄Ni₈Mg₈₀²⁰ and the title compound NdNiMg₁₅²⁷ with increased molar ratio of magnesium (from 71.4% to 88.2%). Except of NdNiMgs which crystallizes with an ordered variant of the Nd₃Co₃Ga-type structure, all these compounds crystallize with new structural types. Only the magnetic properties of Ce₂Ru₄Mg₁₇ and NdNiMg5 have been studied. No magnetic ordering was observed down to 3 K for the former compound and this has been correlated to an almost tetravalent state of Ce atoms. An antiferromagnetic ordering is observed below $T_N = 12$ K for NdNiMg5, and the value of the effective magnetic moment confirms that the prevalent contribution to magnetism comes from the Nd atoms. Among the LPSO phases discovered, recently Kishida *et al.* have determined the crystal structure of three phases with composition Y₄Zn₃Mg₂₃, Y₄Zn₃Mg₂₉ and Y₄Zn₃Mg₃₅¹². Up to now, NdNiMg₁₅ is the richest Mg ordered ternary crystalline phase ever discovered. In a recent paper we have presented the interest of this compound for hydrogen generation²⁷. The crystal structure and the magnetic properties and structure of this peculiar compound are reported herein.

Experimental section

The starting materials for the preparation of the sample were neodymium pieces (ZLX Tech, >99.9%), a nickel rod (Strem Chemicals, >99.9%), and a magnesium rod (Alpha Aesar, >99.8%). To avoid oxides impurities, the surfaces of the magnesium and neodymium pieces were carefully cleaned with abrasive tools. The Nd, Ni and Mg elements were then weighed in the 1:1:15 atomic ratio and sealed in small tantalum ampoules under an argon pressure of ca. 800 mbar. The argon was purified before with magnesium sponge (673 K). Then, the ampoules were placed in a high frequency furnace under argon and heated at about 1773 K and kept at that temperature for 2 min. The tantalum ampoule was then enclosed in an evacuated quartz tube and annealed at 973 K for 10 days. The temperature was decreased to room temperature at 6 K/h.

A crystal suitable for single-crystal X-ray diffraction was selected on the basis of the size and sharpness of the diffraction spots. Data collection was carried out using a Bruker Kappa Apex II diffractometer (Mo-radiation) at room temperature. Numerical absorption correction using the shape of the crystal (face indexed with the help of the video microscope) was made with SADABS- $2014/5^{28}$ software. The unit cell is tetragonal with lattice parameters a = 10.0602(1) and c = 7.7612(2) Å. The space group and structural determinations were performed with the Jana2006 program package²⁹. The extinction conditions observed agree with the *P4/nmm* space group (the origin choice 2 has been chosen for comparison with the isotype structures). All the atomic positions were located using the SUPERFLIP program³⁰. Refinement of the occupancy parameters

showed full occupancy within two standard deviations. At the end of the refinement, with anisotropic atomic displacement parameters for all atomic positions, the reliability factors were $R[F2 > 2\sigma(F2)] = 0.012$ and wR(F2) = 0.030. R/Rw, S = 1.00 for 31 parameters, 805 independent reflections and with residual electron density in the range [-0.33, +0.51 e.Å⁻³]. Details of data collections and structure refinement can be found in the crystallographic information file (CIF) deposited in the Fachinformationszentrum Karlsruhe (FIZ, 76344 Eggenstein-Leopoldshafen Germany); with reference CSD 434727. The atomic positions are given in table 1.

Magnetization measurements were performed using a SQUID magnetometer from Quantum Design for temperatures from 1.8 to 360 K and magnetic fields up to 7 T.

Neutron diffraction experiments were carried out at the Laue Langevin Institute (ILL) on the D1B diffractometer with the use of a wavelength of 2.524 Å. A cylindrical vanadium sample holder of 6mm inner diameter was used to collect neutron data. The FullProf Suite program was used for nuclear and magnetic structure refinements using the Rietveld method³¹.

Results and Discussion

Crystal structure

The structure of NdNiMg15 is made of a three-dimensional network of magnesium atoms giving rise to channels along the c axis where chains made of nickel and neodymium atoms are located (figure 1). The channels consist in infinite face-sharing chains of [Mg8] square antiprism filled alternatively by Ni and Nd atoms and a connection between the chains of Mg8 square prisms through Mg6-Mg6 bonds can be considered (figure 2a). The Ni-Nd distances observed in the chain, equal to 3.858 and 3.903 Å (table 2), are very long compared to the sum of the metallic radii, 3.07 Å ($r_{Nd} = 1.82$ and $r_{Ni} = 1.25$ Å³² and of the covalent radii, 2.79 Å ($r_{Nd} = 1.64$ and $r_{Ni} = 1.15$ Å³³). Then, the bonding between Nd1 and Ni1 is expected to be rather weak. The Ni-Mg distances in the [Mg₈] square antiprism, equal to 2.641 and 2.669 Å, are very close to the sum of the covalent radius of Ni and atomic of Mg, 2.75 Å ($r_{Mg} = 1.60$ Å³³). For the discussion of distances with Mgatoms, the covalent radius ($r_{Mg} = 1.36 \text{ Å}^{33}$) has not been taken into account because of its too low value which is not realistic in the case of intermetallic chemistry. Then the atomic radius of 1.60 Å has been considered and moreover it corresponds also to the metallic radius³⁴. By considering the two shortest Nd-Mg distances of 3.394 and 3.474 Å (table 2), the Nd atom has eight Mgneighbors which form a square antiprism. These distances are very close to the sum of the metallic radii (1.82 + 1.60 = 3.42 Å). By slightly increasing the Nd-Mg distances up to 3.6 Å the coordination polyhedra becomes more complex (figure 2b) with 16 Mg-neighbors, 24 triangular faces and 2 square faces. In NdNiMg5¹⁸ different coordination polyhedra around Ni and Nd atoms are observed but the Ni-Mg and Nd-Mg distances are very close to the ones observed for NdNiMg15. They range from 2.668 to 2.804 Å for Ni-Mg and from 3.292 to 3.416 Å for Nd-Mg (only the distances below 3.60 Å are considered). To describe the 3D-network of magnesium atoms one can consider three roughly regular Mg4 tetrahedra sharing faces or edges (figure 3a and 3b). The most regular tetrahedron is [Mg4]₄ with Mg4-Mg4 distances equal to 3.055 and 3.182 Å and the angles between the center of the tetrahedra and the vertices equal to 105.2 and 111.7°. The two other types of tetrahedra are more distorted with in each case a long edge with a length equal to 3.265 Å for Mg3-Mg7 and 3.451 Å for Mg4-Mg7. A view of the structure with a drawing of these tetrahedra is shown in figure 3c. The average distance between the center of these three tetrahedra and their vertices is equal to 1.94 Å which is exactly the same value as the one observed in Mg-metal²⁷. Moreover the Mg-Mg distances are ranging from 2.944 to 3.451 (table 2) and are consistent with the value of 3.2 Å observed in Mg-metal. All these structural features suggest the metallic character of the Mg-network. The structure of NdNiMg15 can be considered as an ordered variant of the $V_{15}Sb_{18}^{35}$ structure type. Only few compounds crystallize with this structural type, one can cite Ti₈Bi₉³⁶, Zr₂V₆Sb₉³⁷, Ti_{5,42}Mo_{2.58}Sb₉³⁸ or Ce₂Pd₁₄Si³⁹. It may be noticed that the ICSD database⁴⁰ has chosen the Zr₂V₆Sb₉ structure as structure-type even if the V₁₅Sb₁₈ structure (or Bi₈Ti₉) has been published earlier. To our knowledge NdNiMg15 is the first Mg-based phase crystallizing with this structural-type.

The structures of the four very Mg-rich phases recently discovered in the Nd-Ni-Mg system can be described as made of a three-dimensional metallic framework of magnesium atoms and of a covalent network of Nd-Ni atoms with lower dimensionality. A 2D-network is observed in NdNiMg5¹⁸, the Nd and Ni atoms forming ordered graphitic-type layer perpendicular to the *c*-axis and a 1D-network in NdNiMg15 with chains alternatively made of Nd- and Ni-atoms running along the *c* axis. In the case of Nd4Ni8Mg80²⁰ only Nd and Ni isolated atoms are observed whereas in Nd₁₆Ni₁₂Mg96²¹ Nd4Ni₂ clusters and isolated Ni-atoms are coexisting. As briefly discussed in the introduction, the coexistence of a metallic Mg-matrix and clusters of rare-earth and transition metal was previously observed in Mg-rich LPSO phases (see Y-Zn-Mg¹² system for instance) and led to drastic changes in mechanical properties¹⁰.

Magnetic properties

The plot of the magnetization divided by the applied magnetic field, M/H, of NdNiMg₁₅ is reported on figure 4. On this curve, we clearly distinguish a peak at 9 K indicating the establishment of an antiferromagnetic ordering. As shown in the inset of figure 4, the inverse of the magnetic susceptibility measured at 3 T exhibits a small curvature in the paramagnetic domain. Thus,

the data above T_N, were fitted with a modified Curie-Weiss law: $\chi = \chi_0 + C/(T-\theta_p)$, where χ_0 is a temperature-independent term, C the Curie constant and θ_p the paramagnetic Curie temperature. From this fit, we obtain the following values: $\chi_0 = 4.8 \ 10^{-4} \ \text{emu/mol}$, C = 1.60 emu.K.mol⁻¹ and θ_p = -4 K. The effective moment corresponding to the Curie constant is equal to 3.58 μ_B which is very close to the value expected for the trivalent state of the Nd³⁺ ion ($\mu(Nd^{3+})= 3.62 \ \mu_B$). This also indicates that Ni is likely not or very weakly magnetic in this compound. The slightly negative paramagnetic Curie temperature (*i.e.* -4K), close to the Néel temperature, is coherent with local antiferromagnetic interactions between Nd atoms.

Figure 5 displays the isothermal magnetization curves at different temperatures between 1.8 and 15 K. Above T_N , the M(H) curve is linear as expected for a paramagnetic state. Below the Néel temperature, the curves exhibit a linear behavior at low fields which is typical of an antiferromagnet. Besides, the curves show field induced metamagnetic transitions at critical fields increasing with decreasing temperatures. For example, the critical fields H_c at 1.8 and 7 K are around 4.8 and 2.75 T, respectively (see figure 5). This metamagnetic transition most likely results in a transition from an antiferromagnetic to a ferromagnetic state. The maximum value of the magnetization reaches around 2 $\mu_B/f.u.$ at 7 T but as the curve is clearly not completely satrated, one can expect a slightly higher saturated moment for Nd atoms.

Magnetic structure

In order to know the type of antiferromagnetic order present in NdNiMg15 we performed neutron diffraction measurements above and below T_N, namely between 20 and 1.8 K (see Figure. 6). Above the Néel temperature, the Bragg peaks, indexed with the P4/nmm space group, only correspond to the nuclear contribution. At 20 K, the cell parameters are equal to a = 10.009(1) Å and c = 7.731(1) Å and the refined crystal structure is in agreement with the results obtained from singlecrystal measurements at room temperature. Some small extra peaks not indexed with the P4/nmm space group, are accounted for the presence of the residual secondary phases Mg, Mg₂Ni and NdMg₁₂. Below 10 K, magnetic contributions start to grow at 2θ -positions different from those of the nuclear contribution, as it can be seen on Figure 6. These magnetic peaks are indexed with the propagation vector $\mathbf{k} = (1/2 \ 1/2 \ 0)$, which is coherent with the antiferromagnetic state observed at $T_N = 9$ K by magnetic measurements. This implies that the magnetic cell corresponds to the nuclear cell multiplied by two along both the *a*-axis and the *b*-axis $(2a \times 2b \times c)$. At 1.8 K, the best refinement of the difference diagram 1.8K-20K is unambiguously found for an antiferromagnetic model where the magnetic moments are pointed along the c-axis. The refinement at 1.5 K and the corresponding magnetic structure are shown on figures 7 and 8. This magnetic structure consists in ferromagnetic chains along the *c*-axis of Nd atoms carrying moments, only separated by Ni atoms. These chains are coupled to each other ferromagnetically within planes perpendicular to the [110]

direction and antiferromagnetically between these planes leading to a checkerboard pattern in the (ab) plane (figure 8b). The Nd-atoms form a puckered square lattice in the (a, b) plane with Nd-Nd distances of 7.19 and 10.06 Å. This latter distance corresponds to the diagonal of the square (see figure 8). The "square lattices" of Nd-atoms are stacked along the c-axis with Nd-Nd distances of 7.76 Å. In the (a,b) plane, antiferromagnetic coupling is observed between Nd-atoms along the [100] or [0-10] directions (Nd-Nd distance of 10.06 Å). For the Nd-Nd distance of 7.19 Å, antiferromagnetic and ferromagnetic couplings that alternate are observed along the [110] directions. Ferromagnetic couplings between Nd atoms are observed along [-110] directions (distance of 7.19 Å) and along the [001] direction (Nd-Nd distance of 7.76 Å). From the collinear magnetic structure, one can interpret the metamagnetic transition observed on the M(H) curve as a spin-flip transition. At the critical field value, half of the magnetic moments turn over along the direction of the field, which results in a ferromagnetic state. The magnetic moment value of Nd reaches μ_{Nd} = 2.70(5) µ_B at 1.5 K in consistency with the magnetization measurements M(H) at low temperatures. From the various neutron patterns recorded below T_N, we also refined µ_{Nd} in function of the temperature as depicted in figure 9. This thermal evolution confirms that the Néel temperature is around 10(1) K. No significant magnetic moment is found on the Ni site as expected from previous susceptibility measurement. However, the Néel temperature is remarkably high considering only the Nd sublattice with its high Nd-Nd distances within the crystal structure. It is known that the magnetic interactions between rare-earth atoms of RKKY (Ruderman-Kittel-Kasuya-Yosida) type are indirect and mediated by the spin polarization of conduction electrons induced by the localized 4f-moments. This polarization which has an oscillatory character allows the interaction between localized 4f-moments and its absolute value decreases with increasing interatomic distance. The magnetic ordering temperature is also dependent on the density of states at the Fermi level. Among the other effects that can modulate this temperature one can cite the crystalline electric field⁴¹ and the hybridization between the 5d orbitals of rare-earth and 3d orbitals of transition metal⁴². In NdNiMg₁₅, with a Néel temperature $T_N = 9$ K, we observe very long Nd-Nd distances, equal to 7.76 Å and 7.19 Å between Nd atoms inside and between the chains, respectively. In the literature, only few compounds having very long RE-RE distances (higher than 7 Å to achieve a 3D network of RE-atoms) exhibit magnetic ordering temperature higher than 10 K. In the series RERhIn5 (RE = Rare-Earth) with tetragonal symmetry, NdRhIn₅^{43,44} is antiferromagnetic below $T_N = 11$ K. The anisotropic structure is made of NdIn3 and RhIn2 layers with Nd-Nd distances equal to 4.63 and 6.55 Å in the NdIn₃ layers and to 7.50 Å between these layers. In the RE₂Pt₆Ga₁₆ series, a magnetic transition $T_M \sim 13K$ is observed for RE = Nd⁴⁵. These compounds crystallize with an hexagonal symmetry and are made of Pt₂Ga₁₄ double layers and RE-Ga disordered layers. The RE-RE distances are nearly equal to 4.3 Å within the RE-Ga layers and to 8.7 Å between them. The last example is NdNiMg5¹⁸

which exhibits antiferromagnetic ordering below $T_N = 12$ K and its structure is made of Nd-Ni graphite-type and Mg layers. The Nd-Nd distances are equal to 4.48 and 5.47 Å inside the Nd-Ni layers and to 7.23 and 7.77 Å between these layers. It can be noticed that in all the examples, a strong structural anisotropy is observed. To show that the Nd-Nd distance is not the only criterion to take into account to estimate the magnetic transition temperature, the following examples are given. Nd₂Zn₁₇⁴⁶ has Néel temperature close to 1.1 K whereas the Nd-Nd distances are ranging from 4.35 Å to 6.84. Å. Moreover, in NdNi₄Mg, Nd atoms form a tridimensional FCC network with Nd-Nd distances of 5.02 Å, and no magnetic order is observed⁴⁷. Band structure calculation are planned to better characterize the magnetic interactions in NdNiMg₁₅ and to see if the Ni-atoms play a role in the interactions.

Conclusion

The search of new phases in the Mg-rich corner of the Nd-Ni-Mg was fruitful with the discovery of four phases; NdNiMg5¹⁸, Nd4Ni8Mg80²⁰, Nd16Ni12Mg96²¹ and the title compound NdNiMg15. Its structure is an ordered variant of the V15Sb18 structure where a 3-Dimensional network of Mg-atoms and 1D-chains of alternating Ni and Nd atoms are coexisting. Such arrangement with a 3-Dimensional metallic network of Mg atoms and a covalent network of Ni and Nd atoms with lower dimensionality was previously observed in the other Mg-rich phases in the Nd-Ni-Mg system. Despite the very long Nd-Nd distances, higher than 7 Å, an antiferromagnetic ordering has been observed with $T_N = 9K$. The magnetic structure determined by neutron diffraction shows ferromagnetic coupling within the chains of Nd atoms which are running along the *c*-axis. The chains are ferromagnetically coupled within planes perpendicular to the [110] direction and these planes are antiferromagnetic transition observed on the M(H) curve can be interpreted as a spin-flip transition leading to an overall ferromagnetic structure.

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Position	Wyck.	x	У	Ζ	U_{eq} (Å ²)
Nd1	2c	1/4	1/4	0.93340(2)	0.00968(4)
Ni2	2c	1/4	1/4	0.43049(5)	0.01473(8)
Mg3	4d	0	0	0	0.0166(2)
Mg4	8i	3/4	0.40182(5)	0.15049(7)	0.0155(1)
Mg5	2b	3/4	1/4	1/2	0.0317(3)
Mg6	8j	0.91021(4)	0.91021(4)	0.39055(7)	0.0172(1)
Mg7	8i	3/4	0.98025(5)	0.73297(7)	0.0162(1)

Table 1. Atomic positions and equivalent displacement parameters of NdNiMg₁₅. *P*4/*nmm* space group (origin choice 2) a = 10.0602(1) and c = 7.7612(2) Å.

 Table 2. Interatomic distances in Å for NdNiMg15.

Nd1	Mg6 ×4	3.3936(5)	Mg5	Mg4 ×4	3.1130(5)
	Mg7 ×4	3.4743(5)		Mg7 $\times 4$	3.2610(5)
	Mg4 ×4	3.5627(5)			
	Mg3 ×4	3.5942(1)	Mg6	Ni2 ×1	2.6692(5)
	Ni2 ×1	3.8580(4)		Mg6 ×1	3.0682(6)
	Ni2 ×1	3.9032(5)		Mg4 ×2	3.1056(6)
				Mg7 ×2	3.1870(7)
Ni2	Mg7 ×4	2.6410(6)		Mg6 ×2	3.2235(6)
	Mg6 ×4	2.6692(5)		Mg3 ×1	3.2893(5)
	Nd1 ×1	3.8580(4)		Nd1 ×1	3.3936(5)
	Nd1 ×1	3.9032(5)			
			Mg7	Ni2 ×1	2.6410(6)
Mg3	Mg4 ×4	2.9437(3)		Mg6 ×2	3.1870(7)
	Mg7 ×4	3.2650(3)		Mg4 $\times 2$	3.2427(5)
	Mg6 ×2	3.2893(5)		Mg5 ×1	3.2610(5)
				Mg3 ×2	3.2650(3)
Mg4	Mg3 ×2	2.9437(3)		Mg7 ×2	3.2758(5)
	Mg4 ×1	3.0547(8)		Mg4 ×1	3.4509(7)
	Mg6 ×2	3.1056(6)		Nd1 ×1	3.4743(5)
	Mg5 ×1	3.1130(5)			
	Mg4 ×2	3.1816(7)			
	Mg7 ×2	3.2427(5)			
	Mg7 ×1	3.4509(7)			



Figure 1. Perspective view of the structure of NdNiMg₁₅. The Mg–Mg bonds are drawn (Mg-Mg distance lower than 3.3 Å).



Figure 2. (a) Nd-Ni chains of atoms with Nd-Mg, Ni-Mg and Mg6-Mg6 bonds are drawn. (b) Environment of Nd1 position by considering distances Nd-Mg lower than 3.6 Å.



Figure 3. (a) Surrounding of a [Mg4]4 tetrahedra drawn in green. (b) Same view as (a) with all [Mg4] drawn in green. (c) View of the structure of NdNiMg15 with all Mg4 tetrahedra drawn in green.



Figure 4. Temperature dependence of the magnetization divided by the magnetic field of NdNiMg₁₅. Inset: Inverse of the magnetic susceptibility versus temperature with the modified Curie-Weiss fit represented in dashed line.



Figure 5. Field dependence of the magnetization curves at various temperatures between 1.8 and 15 K.



Figure 6. Neutron diffraction patterns of NdNiMg₁₅ recorded at 1.5 and 20 K with the difference pattern between both temperatures showing the magnetic contribution.



Figure 7. Rietveld refinement of the magnetic structure of NdNiMg₁₅ from the neutron diffraction pattern at 1.5 K. The two first lines of Bragg position ticks correspond to the nuclear and magnetic phases of NdNiMg₁₅ respectively and the third one to residual Mg phase. The symbol * corresponds to the secondary phase Mg₂Ni.



Figure 8. Magnetic structure of NdNiMg₁₅. (left) Four unit cells are represented in blue and only Nd and Ni atoms are shown for clarity (right) Representation along the *c*-axis. The + and - signs represent the magnetic moments pointing forward and backward along the c-axis, respectively



Figure 9. Temperature dependence of the magnetic moment of Nd. The line is a guide for the eyes.