

Chromium bioavailability in aquatic systems impacted by tannery wastewaters. Part 1: Understanding chromium accumulation by indigenous chironomids

Davide Anselmo Luigi Vignati, Benoît J.D. Ferrari, Jean-Louis Roulier, Marina Coquery, Ewa Szalinska, Andrzej Bobrowski, Anna Czaplicka, Andrzej Kownacki, Janusz Dominik

▶ To cite this version:

Davide Anselmo Luigi Vignati, Benoît J.D. Ferrari, Jean-Louis Roulier, Marina Coquery, Ewa Szalinska, et al.. Chromium bioavailability in aquatic systems impacted by tannery wastewaters. Part 1: Understanding chromium accumulation by indigenous chironomids. Science of the Total Environment, 2019, 653 (653), pp.401-408. 10.1016/j.scitotenv.2018.10.259. hal-01907047

HAL Id: hal-01907047 https://hal.science/hal-01907047v1

Submitted on 28 Oct 2018

HAL is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers. L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.

1	This is the final peer-reviewed	version of the article:
2	Chromium bioavailability in aquatic system	ns impacted by tannery wastewaters. Part 1:
3	understanding chromium accumulation by	indigenous chironomids
4		
5 6 7	Published in Science of the Total Environmen https://doi.org/10.1016/j.scitotenv.2018.10.25	nt, doi: <u>9</u>
8	Vignati D.AL. ^{a,b} *, Ferrari B.J.D. ^{a,c} , Roulier J.	-L. ^d , Coquery M. ^d , Szalinska E. ^e , Bobrowski A. ^f ,
9	Czaplicka A. ^g , Kownacki A. ^h , Dominik, J. ^{a,i}	
10		
11	a) Department FA. Forel for Environmenta	l and Aquatic Sciences, University of Geneva,
12	Uni Carl Vogt, 66 boulevard Carl-Vogt C	H-1211 Geneva, Switzerland
13		
14	b) Université de Lorraine, CNRS, LIEC, F-5	7000 Metz, France
15		
16	c) Swiss Centre for Applied Ecotoxicology	Eawag-EPFL (Centre Ecotox), EPFL-ENAC-
17	IIE-GE, Station 2, 1015 Lausanne, Switze	rland
18		
19	d) Irstea, UR RiverLy, centre de Lyon-Viller	urbanne, F-69625 Villeurbanne, France
20		
21	e) Department of Environment Protection, F	aculty of Geology, Geophysics and
22	Environmental Protection, AGH Universi	ty of Science and Technology, 30 A.
23	Mickiewicza Av., 30-059 Krakow, Polan	d.
24		

25	f)	Department of Building Materials Technology Faculty of Materials Science and Ceramics,
26		AGH University of Science and Technology, 30 A. Mickiewicza Av., 30-059 Krakow,
27		Poland
28		
29	g)	Department of Water Supply, Sewerage and Environmental Monitoring, Cracow
30		University of Technology, 24 Warszawska ul., 31-155 Krakow, Poland
31		
32	h)	Karol Starmach Department of Freshwater Biology, Institute of Nature Conservation,
33		Polish Academy of Sciences, 33 A. Mickiewicza Av., 31-120 Krakow, Poland
34		
35	i)	Institute of Marine Science - National Research Council (ISMAR-CNR) Arsenale - Tesa
36		104, Castello 2737/F, 30122 Venice, Italy
37		
38	*C	orresponding Author:
39	dav	vid-anselmo.vignati@univ-lorraine.fr
40	Po	stal address: LIEC UMR7360, Campus Bridoux, UFR SciFa, Bât Ibise, 8 rue Delestraint,
41	570	070 Metz, France
42		
43		
44		
45		
46		
47		
48		
49		

50 Abstract

51 The tanning industry uses large quantities of Cr whose contribution to the contaminant burden 52 of aquatic organisms is not yet fully understood. The present study investigated Cr 53 bioaccumulation by indigenous chironomids in a freshwater ecosystem impacted by tannery 54 effluents. Total Cr content in sediments and in chironomids was determined on several 55 occasions. Chromium distribution among sediments and pore waters, and Cr speciation in 56 overlying and pore waters were studied in detail to understand possible factors controlling Cr 57 bioavailability to chironomids. Total chromium concentration ranged from 69 to over 3000 µg g^{-1} dry weight in sediments and from negligible to over 300 μg^{-1} dry weight in chironomids 58 59 (values corrected for sediment gut content). Filterable ($<0.45 \mu m$) Cr concentration in 60 overlying waters and pore waters from the surface sediment layers (upper 2 cm) ranged from 3 to 120 μ g L⁻¹, with Cr(VI) representing 0.5–28% of the total filterable Cr. Chromium 61 62 profiles in pore waters as determined by diffusive equilibration in thin films (DET) and 63 diffusive gradient in thin films (DGT) were comparable. DGT-labile Cr accounted for less 64 than 2% of the total Cr measured by DET. Although Cr concentrations in sedimentary and 65 aqueous matrices were not directly proportional to Cr levels measured in chironomids, the available findings suggested that Cr inputs from tanneries were bioavailable to resident 66 chironomids. These observations are of particular importance considering that Cr(III), 67 68 putatively of limited bioavailability and ecotoxicological concern, is the predominant redox 69 form of Cr in bed sediments impacted by tannery discharges. The companion paper provides 70 further insight into Cr bioavailability and effects in tannery impacted ecosystems using a 71 combination of in situ and laboratory approaches.

72

73 Keywords. Chromium speciation; bioaccumulation; DGT; DET; voltammery; sediment cores

74 Introduction

75 Chromium has a complex biogeochemical cycle which is essentially controlled by the 76 simultaneous environmental occurrence of two redox forms, Cr(III) and Cr(VI), having contrasting (geo)chemical and (eco)toxicological properties (2016). Hexavalent Cr is 77 78 considered of most concern due to its ecotoxicity and proven carcinogenicity in humans, 79 while trivalent Cr is usually regarded as relatively harmless (WHO, 2009; 2013). Hexavalent Cr also has high environmental mobility, while Cr(III) tends to bind to suspended particulate 80 81 matter and colloids and eventually accumulates in bed sediments (de Sousa et al., 2016; 82 Dominik et al., 2007; Gunkel-Grillon et al., 2014; Hayzoun et al., 2015). Remobilization of 83 Cr(III) from bed sediments is however possible following re-oxidation to Cr(VI) (Gorny et al., 84 2016; Gunkel-Grillon et al., 2014). While the concern about Cr(VI) is justified, increasing 85 evidence points to a possible underestimation of the ecotoxicological risks posed by Cr(III) 86 for unicellular organisms (Bencheikh-Latmani et al., 2007; Jasso-Chávez et al., 2010; Lira-87 Silva et al., 2011; Vignati et al., 2010), crustaceans (Ponti et al., 2014), aquatic plants (Dazy 88 et al., 2008) and terrestrial plants and organisms (Bigorgne et al., 2010; Speranza et al., 2007). 89 The implications of a possible underestimation of Cr(III) ecotoxicity would be particularly 90 relevant in aquatic ecosystems directly impacted by tannery wastewaters.

91 Leather tanning uses large quantities of Cr(III), generally as the basic chromium sulphate 92 $Cr_2(OH)_2(SO_4)_2$. Chromium inputs from tanneries can markedly affect Cr concentrations and 93 fluxes in aquatic environments at local scales (Alves and Rietzler, 2015; Dominik et al., 2007; 94 Hayzoun et al., 2015; Sanyal et al., 2015) and the toxicological impact of tannery wastewaters 95 containing numerous potentially harmful compounds was recently reviewed by Saxena et al. 96 (2017). However, understanding the contribution of Cr(III) to the overall biological response 97 is far from straightforward because tannery effluents contain several toxicologically active 98 substances. The difficulties in apportioning the observed effects to the presence of Cr(III) are especially big in the case of bed sediments where Cr preferentially accumulates and Cr(III)
tends to be the predominant redox form associated with the solid phase (Pereira et al., 2012;
Rosales et al., 2017).

102 Research on the bioavailability and effects of Cr(III) to benthic organism under realistic 103 exposure conditions or field scenarios is actually very limited. Authoritative risk assessment 104 documents (WHO, 2009; European Union, 2005) focus on ecotoxicity data from water-only 105 exposure; neglecting that sediment ingestion can be a major uptake route of Cr(III) for benthic 106 deposit feeders (Baumann and Fisher, 2011). Furthermore, studying the ecotoxicity of Cr(III) 107 to benthic organisms by spiking sediment-water systems with soluble Cr(III) salts is difficult, 108 because the spiked soluble Cr(III) undergoes hydrolysis within seconds and forms scarcely 109 bioavailable Cr(III) oxy-hydroxides that, in turn, undergo polymerization reaction on the time 110 scale of minutes (Pettine et al., 2008). The formation of such precipitates was actually 111 reported in laboratory studies on the relative ecotoxicity of Cr(VI) and Cr(III) on various 112 amphipod species (Berry et al., 2004; Besser et al., 2004); studies that did not highlight 113 measurable effects of Cr(III) on the test organisms. Assessing the possible ecotoxicity of 114 Cr(III) in sediments containing chromite ore processing residues, Becker et al. (2006) 115 concluded that Cr(III) did not impact survival and growth of two amphipods for concentrations as high as 1,310 µg Cr g⁻¹. However, both Cr oxy-hydroxides formed during 116 117 spiking with Cr(III) salts and chromite ore processing residues poorly mimic the organic-118 Cr(III) forms usually discharged from tannery effluents (Outridge and Scheuhammer, 1993; 119 Walsh and O'Halloran, 1996; Wang et al., 2016). Furthermore, Bervoets et al. (2016) determined that Cr body burdens as low as 10 $\mu g \ g^{\text{-1}}$ d.w. (range 10–69 $\mu g \ g^{\text{-1}}$ d.w.) in 120 121 specimen of *Chironomus* sp. collected at several field locations in the Flanders (Belgium) can 122 be associated with changes in the macroinvertebrate community composition. Studies 123 simultaneously assessing Cr geochemistry including redox speciation and the corresponding 124 Cr bioavailability to benthic organisms in tannery impacted environments are therefore 125 required to understand the bioavailability and possible biological effects associated with Cr in 126 such specific environmental settings.

127 The present work studies Cr bioaccumulation by indigenous chironomids' assemblages in 128 a river/reservoir system impacted by tannery wastewater discharges (Pawlikowski et al., 2006; 129 Szalinska et al., 2010). Investigations of increasing complexity were carried out during four 130 different sampling periods over four years. A detailed characterization of Cr geochemistry and 131 speciation allows examination of the possible importance of different routes of exposure in 132 contributing to the observed Cr body burden under field conditions. Part two of the present 133 study (Ferrari et al., this issue) compares results from laboratory-based bioaccumulation tests 134 with results obtained using an in situ a customized Sediment-Water Interface Study System 135 (Ferrari et al., 2014) to provide further insight in the bioavailability and effects of Cr in 136 aquatic ecosystems receiving tannery discharges.

- 137
- 138

139 **2. Methods**

140 2.1. Study area and sampling

141 The Dunajec river/Czorsztyn reservoir system (Southern Poland, Fig. 1) is impacted by 142 wastewater discharges from about 300 small tanneries in the Nowy Targ region. Due to 143 inefficient tannery wastewater management, pre-treated or raw wastewaters from these 144 tanneries are discharged into local sewers or directly into nearby watercourses. From late 145 October through February, river sediments in the proximity of tannery effluents are impacted 146 by chromium, while spring and summer floods tend to flush the contaminated sediments into 147 the Czorsztyn reservoir (Pawlikowski et al., 2006; Szalinska et al., 2010). Samples of 148 sediments and indigenous chironomids were collected in June and November 2004,

November 2006 and June 2007 at two sites. Site 1 (hereinafter Waksmund) was located in a depositional zone of the Dunajec river and directly impacted by untreated tannery discharges.
Site 2 (hereinafter Maniowy) was located in the Czorsztyn reservoir in a small bay, locally fed by a small creek which receives the effluent from the local wastewater treatment plant before flowing into the bay (Fig. 1). Tanneries are also present in the region of Maniowy, however only a few of them discharge their waste to the wastewater treatment plant.

155 For all sampling periods, sediment samples (2-10 kg) were collected using a plastic scoop 156 (Waksmund, wading against the river current along the right bank of the river) or a Van Veen 157 grab (Maniowy, the grab was operated from a small boat) and directly homogenised in the 158 field. Larvae of indigenous chironomids were recovered in the field by sieving additional 159 aliquots through 2 and 1 mm sieves. Fourth instar larvae were recovered from the sieves with 160 plastic tweezers, washed in deionised water (5 min), 1 mM EDTA solution (10 min) and again 161 deionised water (5–10 minutes). After blotting with paper towels, 14 to 20 larvae were pooled 162 and placed in cryogenic vials, snap frozen in liquid nitrogen and freeze-dried upon return to 163 the laboratory. The focus being on Cr accumulation in indigenous chironomids' assemblages, 164 care was taken to select large larvae showing a marked red colour, but no taxonomical 165 identification was performed at this stage. In 2007, samples for a more detailed 166 characterization of the chironomids' communities were collected at the two sites using a 20x20 cm (400 cm²) Surber Sampler (Waksmund) and a 15x15 cm (225 cm²) Ekman Grab 167 168 Sampler (Maniowy). Chironomids were then recovered for identification by further sieving these samples through a 0.3 mm net. 169

In both 2004 and 2007, independent aliquots of homogenised sediments for analytical determinations were sieved (< 1 mm) in the field and collected into preconditioned polypropylene jars that were kept at 4 °C in the dark and freeze-dried upon return to the laboratory. In November 2006 and June 2007, sediment cores were also collected by short (20 174 cm) hand corers either by wading against the river current at 4–5 m from the right bank of the 175 Dunajec River (Waksmund) or by scuba divers in the Czorsztyn reservoir (Maniowy). The 176 corers allowed recovery of sediments (15-25 cm depending on texture and compactness) and 177 associated overlying water (10-15 cm). Further details on core handling are given in section 178 2.2. In 2006, nine cores were collected at each site and analyzed for total Cr content in the 179 uppermost 0.5 cm (see section 2.2). In 2007, one core was collected at each site and sliced for 180 the determination of the following concentration profiles: total Cr in sediments, total Cr in 181 pore waters and Cr(VI) in pore waters (see section 2.2).

Diffusive Equilibration in Thin Films (DET) and Diffusive Gradients in Thin Films (DGT) probes were deployed at undisturbed locations close to the sampling sites to obtain information on Cr speciation and lability. The DGT and DET probes (arranged back to back) were recovered after exposures of 46h 25 min (Waksmund) or 45 h (Maniowy). A time-series of Cr-DGT profiles (Maniowy only) was also obtained by exposing DGT probes for 22, 45, 68 and 89 h at the same location.

188

189 2.2. Core slicing and pore waters extraction

190 Sediments cores were placed under nitrogen atmosphere in a portable glove box. Overlying 191 waters (OW; height of water column = 10-15 cm) were removed using plastic syringes and 192 taking care to avoid sediment resuspension. Aliquots of OW were filtered (1.2 µm syringe 193 filters, Whatman, Puradisc 30, cellulose acetate), collected into acid washed polypropylene 194 ICP tubes and acidified to 1% (v/v) with concentrated HNO₃ (VWR, Suprapur). Prior to use, 195 filters were prewashed with 10 mL of diluted (10% v/v) HCl followed by three rinses with 196 MilliQ water and then kept individually in plastic zip bags. After removing OW, sediment 197 sections from 0.5 to 1 cm thick were transferred to acid washed 100 mL polypropylene 198 centrifuge tubes. Tubes were capped and sealed with Parafilm[™] inside the glove box and 199 centrifuged at 4000 rpm for 15 + 15 minutes. Tubes showing visible sediment compaction 200 after the second centrifugation cycle were further centrifuged for an additional 30 minutes. 201 Tubes were then re-introduced into the glove box and the pore waters filtered and handled in 202 the same way as OW. Centrifuge tubes, syringes and filters were purged with N₂ immediately 203 before core slicing.

204

205 2.3. DET and DGT methodologies

Total dissolved metal profiles were assessed by the DET technique (Davison et al., 1994), which is based on the equilibration between solutes and a hydrogel casted in a sampling device. The DET probes consisted of a base plate with a window of $1.8 \text{ cm} \times 15 \text{ cm}$ and a set of 1 mm slits with 1 mm interval between the slits. The slits were filled with 1.5 % agarose gel (Leermakers et al., 2005).

Free and labile metal species were determined using the DGT technique (Davison et al., 1994; Zhang and Davison, 1995). The DGT sediment probes consisted of a plastic base (15×5 cm) loaded with a "sandwich" of 3 layers: a resin gel (Chelex 100), covered by a polyacrylamide gel (0.8 mm thickness) and a protective 0.45 µm cellulose nitrate filter (0.13mm thickness, Millipore). Materials for the DGT and DET probes were purchased from DGT Research Ltd. (Lancaster, UK).

Before deployment, all the probes were deoxygenated, transported to the field in deoxygenated water and rapidly inserted into the sediment to avoid the co-precipitation of iron (Fe) and other metal ions from pore waters within the gel. After retrieving, the DET and DGT probes were processed in a laminar flow hood. Both the DET gel and Chelex resins from DGT were removed from selected slits, transferred to preweighed plastic tubes and eluted with 2.5 mL of 1 M HNO₃ (VWR, Suprapur). Average DGT-labile metal concentrations (C_{DGT} , in $\mu g L^{-1}$) were calculated according to the formula:

225
$$C_{DGT} = \frac{M \times \Delta g}{D \times t \times A}$$
(1)

226

where: M is the accumulated mass of metal (ng), D is the diffusive coefficient of the metal (cm² s⁻¹) in the diffusive gel of thickness Δg (cm) for a resin surface A (cm²) and t is the deployment time (s). Chelex 100-loaded DGT cannot complex anionic Cr(VI) species and therefore allow the selective measurement of labile Cr(III) species, while generally excluding organic complexes and colloidal species (Ernstberger et al., 2002; Giusti and Barakat, 2005). The diffusive coefficient of Cr was obtained from the temperature-dependent values tabulated by DGT Research Ltd (http://www.dgtresearch.com/diffusion-coefficients/).

234

235 2.4. Chemical analyses

236 Chromium was assayed by ICP-MS (HP4500, Agilent) using external calibration and internal 237 standard correction in overlying waters, pore waters, DET and DGT eluates, sediment and 238 chironomids. Prior to analysis, sediment aliquots (about 60-300 mg d.w.) and chironomids 239 (about 15-40 mg d.w.) were mineralized with a 5:1 (v/v) mixture of concentrated HNO₃ 240 (65%) and H₂O₂ (30%), both from VWR, Suprapur grade. Overall accuracy of the digestion 241 procedures was evaluated using reference materials LKSD-1 (lake sediments) and STSD-2 242 (stream sediments) (Lynch, 1990) for sediments and TORT-2 (lobster hepatopancreas, NRC 243 Canada) for chironomids. Reference waters SLRS-3 from NRC and 1643e from NIST were 244 used to verify instrumental accuracy, with material 1643e being analyzed after a 10-fold 245 dilution.

The potentially bioavailable fraction of sediment-bound Cr liable to be released during transit of ingested sediments through chironomids' gut was estimated using the biomimetic approach (Chen and Mayer, 1999), consisting in a cold extraction of sediments aliquots with synthetic Bovine Serum Albumin (BSA). Because BSA was supplied in a 3% NaCl matrix, additional aliquots were extracted with a 3% solution of NaCl to account for salt-related matrix effects. The results of 'biomimetic Cr' concentrations are corrected for Cr remobilization due to NaCl. Sediment Loss on Ignition (LOI) at 550°C was performed to estimate the content of total organic matter (OM).

254 In overlying and pore waters (June 2007), Cr(VI) was quantified by Catalytic Adsorptive 255 Stripping Voltammetry (CAdSV) using the differential pulse mode (DPV) and standard 256 addition procedure (Bobrowski et al., 2004). In some cases, pore waters from different 257 sediment layers were combined to obtain enough volume for voltammetric measurements that 258 were performed in triplicate for all samples. To reduce possible interferences due to the 259 presence of large quantities of dissolved organic carbon (DOC) in pore waters, Amberlite 260 XAD-7 resin was added to the (stirred) voltammetric cell five minutes before Cr(VI) 261 determination (Bobrowski et al., 2012).

262

263

264 **3. Results**

265 *3.1. Quality control procedures*

For analysis of waters and DGT/DET eluates, typical detection and quantification limits for total chromium (calculated as 3 and 10 times the standard deviation of analytical blanks) were 0.015 μ g L⁻¹ and 0.049 μ g L⁻¹, respectively. Four analytical replicates were measured for each sample and the typical precision (for concentrations above the quantification limit) was 5% or better. Measured values for certified reference waters (arithmetic means±1 standard deviation) were 0.29±0.02 μ g L⁻¹ for SLRS-3 (n=7) and 21.3±0.1 μ g L⁻¹ (n=7) for 1643e; comparing favourably with the corresponding certified values of 0.30±0.04 μ g L⁻¹ (uncertainty expressed as 95% confidence interval) and 20.4 ± 0.24 µg L⁻¹ (uncertainty expressed as expanded uncertainty). Triplicate measurements for Cr(VI) agreed to 25% or better. Recovery of Cr(VI) from spiked samples was 93% or higher (Bobrowski et al., 2012).

Percentage Cr recovery from certified sediments was about 130% for both materials with 276 measured absolute values of 16±2.5 μ g g⁻¹ (arithmetic mean±[Abs(range/ $\sqrt{2}$)]; n=2) for 277 LKSD-1 and $64\pm3.7 \ \mu g \ g^{-1}$ (arithmetic mean ± 1 s.d.; n=3) for STSD-2. Our extraction 278 procedure consisted of concentrated HNO₃ + H_2O_2 while certified values (12±2 µg g⁻¹ for 279 LKSD-1 and 50±9 μ g g⁻¹ for STSD-2, uncertainties expressed as±1 s.d.) are given for an 280 extraction with concentrated HNO3 and concentrated HCl. Sediments LKSD-1 and STSD-2 281 282 have a total certified Cr content (based on extraction with HF or analysis by XRF or analogous techniques) of 31 ± 3 and 116 ± 13 µg g⁻¹, respectively. For biological materials, 283 percentage recoveries from TORT-2 were between 77% and 98% of the certified value 284 $(0.77\pm0.15 \ \mu g \ g^{-1};$ uncertainty given as 95% confidence limits) and the averaged measured 285 concentration was $0.69\pm0.1 \ \mu g \ g^{-1}$ (arithmetic mean $\pm 1 \text{ s.d.}$; n=5). 286

287

288 3.2. Chromium content in bulk sediments and indigenous chironomids

289 In the 2004 and 2007 sampling periods, Cr concentrations in chironomids tended to be 290 inversely related to those in sediments (Table 1). Biomimetic extractable Cr levels (2004 291 samples only) were about 0.1-0.2 % of total Cr in sediments at both sites and were comparable between sampling periods and sampling sites (Table 1). In November 2006, 292 surface sediment samples at Waksmund were more contaminated (mean ± 1 s.d. = 1802 ± 744 293 $\mu g g^{-1}$, n=9) than at Maniowy (118±39 $\mu g g^{-1}$; n=9). In this sampling period, concentration in 294 295 chironomids increased with increasing concentration in sediments; organisms from Maniowy 296 having lower Cr concentrations than those from Waksmund. However, based on average values, Cr concentrations differed by 17-fold in sediments, but less than 3-fold inchironomids.

299

300 3.3. Chromium concentration, distribution and speciation in sediment cores

301 Total Cr concentration in sediments from the 2007 cores was higher at Waksmund than at 302 Maniowy (Table 2). The upper 2 cm of the Waksmund core had Cr concentrations in excess of 2500 µg g⁻¹ and overlaid three layers (2–4 cm, 4–9 cm, and 9–11 cm) with fluctuating 303 304 levels of Cr contamination. In the core layer between 5 and 8 cm, we visually noticed the 305 presence of fibers and hairs, which is indicative of material originating from tanneries 306 (Pawlikowski et al., 2006). The alternation of sediment layers with different levels of Cr 307 contamination reflects the seasonality of tanning activities (which are concentrated during 308 winter time — the cold season in the boreal hemisphere) coupled with the deposition of spring 309 and summer flood sediments linked to river hydrology (Szalinska et al., 2010). At Maniowy, Cr concentration was highest (452 μ g g⁻¹) in the subsurface layer (0.5–1 cm) and then 310 311 decreased progressively towards the bottom of the core. Considering the upper 2 cm of 312 sediment, which are the most relevant for many chironomids (Stief and de Beer, 2006), Cr 313 concentrations at Waksmund were about 10-fold higher than those at Maniowy.

314 A similar situation was observed in overlying and pore waters. At Waksmund, total Cr concentrations in pore waters were very stable in the first 2 cm (114 \pm 7 µg L⁻¹, n=3) and 315 fluctuated markedly (30-300 µg L⁻¹) in the deeper layers. In the Maniowy core, Cr 316 concentrations in pore waters from the uppermost 2 cm were more variable ($43\pm15 \ \mu g \ L^{-1}$, 317 318 n=3), but decreased below this depth (Table 2). The proportion of Cr(VI) relative to total Cr 319 could not be measured for all the pore water samples collected from the various layers due to 320 the water volume required for voltammetric measurements (Bobrowski et al., 2012). Based on the available 'combined values' (Table 2), the proportion of Cr(VI) relative to the total Cr 321

322 content of pore waters varied between 0.5% and 15% at Waksmund and between 5% and 15%
323 at Maniowy. At both sites, the proportion of Cr(VI) was higher in overlying waters (20% and
324 28% at Waksmund and Maniowy, respectively) than in interstitial waters. In absolute terms,
325 Cr(VI) concentrations were always higher at Waksmund than at Maniowy (Table 2).

326

327 *3.4. Chromium concentration by DET and DGT*

328 In June 2007, the mean dissolved Cr concentration estimated by DET in interstitial waters at Waksmund was 284±85 μ g L⁻¹ (n=11; range 178–437 μ g L⁻¹) and the corresponding mean 329 DGT-labile concentration was $4.9\pm1.7 \ \mu g \ L^{-1}$ (n=11; range 2.9–9.0) (Fig. 2). Based on these 330 331 data, DGTs trapped about 1.7% of the total dissolved species (as measured by DETs) present 332 in pore waters. At Maniowy, average Cr concentrations measured by DET and DGT were 74±21 µg L⁻¹ (n=12; range 39–113 µg L⁻¹) and 1.0±0.54 µg L⁻¹ (n=13; range 0.33–2.02 µg L⁻¹ 333 ¹), respectively; with DGT trapping 1.3% of the total dissolved concentrations. Chromium 334 335 concentrations measured by DET were higher than the corresponding Cr levels determined in pore waters after centrifugation (Table 2): $130\pm86 \ \mu g \ L^{-1}$ (n=10) and $25\pm16 \ \mu g \ L^{-1}$ (n=9) for 336 337 Waksmund and Maniowy, respectively. These difference are linked to the predominance of 338 colloidal Cr(III) (Bobrowski et al., 2004; Dominik et al., 2007) that can be partly removed 339 during centrifugation but is caught by the DET probe. Despite all precautions, precipitation of 340 Fe and Mn oxides may have occurred during core slicing and sediment centrifugation. 341 Regardless of the methodological aspects, Cr concentrations at Maniowy were lower than 342 those at Waksmund for the same period, but the average proportion of dissolved concentration 343 which was labile (DGT with respect to DET) was small (1 to 2 %) and roughly similar for the 344 2 sites.

Finally, Cr concentrations in overlying waters were comparable with those in pore waters from the uppermost 2 cm at Waksmund, while they were lower than the values in pore waters at Maniowy. A flux of chromium from sediments to the water column is therefore possible at
Maniowy, while the situation at Waksmund is more variable and no definite conclusions can
be drawn (Fig. 2).

- 350
- 351

352 **4. Discussion**

353 4.1. Chromium levels in environmental matrices and chironomids

354 At Waksmund, the bioaccumulation factor (BAF - calculated as the ratio between total Cr 355 content in sediments vs. total Cr content in indigenous chironomids) ranged from less than 0.1 356 in 2004 and 2006 to 0.17 in June 2007. The same calculation for Maniowy gives values one 357 order of magnitude higher with a maximum of 1.3 for June 2007 (Table 1). Assuming that 358 ingested sediments represent 6% of the chironomids' body weight (Ferrari et al. (2014), 359 correction for the sediment contained in the organisms' gut (Hare et al., 1989) decreases Cr 360 concentration in chironomids by 60-100% at Waksmund and by, at most, 30% at Maniowy 361 (Table 1). Calculation of BAFs using Cr gut-corrected concentrations (data not shown) 362 therefore confirms that Cr bioavailability is higher at Maniowy than at Waksmund. 363 Pawlikowski et al. (2006) reported the presence of microscopic pieces of tanned leather in 364 sediments from Waksmund; a situation which increases the total Cr content of sediments, but 365 not its bioavailable fraction. However, the biomimetic analysis performed in 2004 suggests 366 that the quantity of bioavailable Cr (sensu Chen and Mayer, 1999) is comparable for the two 367 sites (Table 1). Because no major changes in river/reservoir hydrodynamics or tanning 368 operations occurred between 2004 and 2007, the amount of bioavailable, sediment-bound Cr 369 at the two sites should have remained similar also for the 2006 and 2007 sampling periods. 370 Another possible explanation for the observed discrepancies between sediment

371 contamination and Cr accumulation by chironomids is that pore waters and/or overlying

372 waters contribute most of the bioavailable Cr for uptake by living organisms. However, 373 concentrations of filterable Cr in pore waters (Table 2) and concentrations obtained by 374 DET/DGT (Fig. 2) were systematically higher at Waksmund than at Maniowy; contrary to 375 what was observed in chironomids (Table 1). Even the bioavailable Cr(VI) form was more 376 abundant in pore- and overlying waters from Waksmund than in those from Maniowy. 377 Although these analyses (Table 2) and the results of DET/DGT (Fig. 2) represent a snapshot 378 of the variable aqueous concentrations experienced by chironomids at the sediment-water 379 interface, Cr accumulation by DGT probes exposed for different times at Maniowy was 380 similar (Fig. S1). These additional findings indicate that the sediments' particulate phase was 381 able to constantly resupply Cr to the pore waters (Roulier et al., 2008), thus ensuring a certain 382 stability of DGT-labile Cr concentrations (assumed as a proxy of bioavailable Cr) is this 383 compartment.

384 On the other hand, direct comparisons between Cr concentrations in aqueous/sedimentary 385 matrices vs. indigenous chironomids do not capture the actual dynamic nature of contaminant 386 accumulation by living organisms. Contaminant accumulation is the net result of uptake from 387 different exposure routes, excretion and growth dilution, as formalized by e.g. the biodynamic 388 modelling approach (Baumann and Fisher, 2011; Luoma and Rainbow, 2005; Wang et al., 389 1997). In the present study, the use of biodynamic modelling to estimate the relative 390 contribution of different routes of exposure to the Cr burden of indigenous chironomids is not 391 straightforward because the model's parameters vary among species, food types and 392 contaminants. To the best of our knowledge, only a handful of studies have examined Cr 393 uptake by aquatic organisms from aqueous and dietary exposure routes (Baumann and Fisher, 394 2011; Roditi and Fisher, 1999; Roditi et al., 2000; Wang et al., 1997). With the exception of 395 Roditi et al. (2000), who estimated that only 9-45% of Cr accumulation by zebra mussel 396 originated from particle ingestion, these studies identified uptake from food ingestion (sediments, suspended particles or algae) as the predominant contributor to the total Cr body burden. None of these studies used chironomids as model organisms, but the predominance of contaminant uptake via food ingestion has been confirmed for chironomids in the case of other trace elements (De Jonge et al., 2010; Martin et al., 2008). It is therefore important to examine more in detail the possible factors that can influence the bioavailability of sedimentbound Cr at the study sites.

403

404 4.2. Factors affecting the bioavailability of sedimentary Cr

405 Given the similar concentrations of biomimetic extractable Cr measured at the two sampling 406 sites (Table 1), a similar Cr burden in indigenous chironomids would be expected under the 407 assumption that uptake from sediments determines most of the Cr body burden and that 408 contaminant desorption from sediments is the limiting factor for their subsequent absorption 409 (Wang and Fisher, 1999). These expectations not being confirmed by experimental data, other 410 aspects of Cr geochemistry (not captured by the biomimetic approach) may make Cr more 411 bioavailable at Maniowy than at Waksmund or limit Cr bioavailability in the highly 412 contaminated Waksmund sediments.

413 In the cores collected in June 2007 and in November 2006 (Table 2 and Tables S1a, S1b), 414 a relationship between sedimentary Cr and Fe existed at Maniowy (r=0.88, n=10, p<0.01 in 415 2007 and r=0.69, n=11, p<0.05 in 2006; Spearman Rank Correlation test), but not at 416 Waksmund. The Fe-Cr correlation is indicative of an association between Cr and Fe-417 oxyhydroxides that is often highlighted using sequential extraction procedures of sedimentary 418 Cr in aquatic systems impacted by tannery wastewaters (Hursthouse et al., 2001; Khwaja et 419 al., 2001; Rodrigues and Formoso, 2006). However, in such procedures, the Fe-Mn oxides 420 extraction step usually employs hydroxylamine hydrochloride at pH values around 2; much 421 lower than those of 6-7 typical of the chironomids' gut (Crozier, 1924). Nonetheless, anoxic

conditions in the chironomids' gut could favour reductive dissolution of Fe oxides with 422 423 subsequent liberation of Cr directly in the gut lumen (Stief and Eller, 2006). The redox 424 potential (E_h) in the gut lumen of chironomids is about -50 mV (Stief and Eller, 2006); low enough to allow for the dissolution of amorphous Fe(OH)₃ given the E_h value of 14 mV at pH 425 426 7 for the redox couple Fe(OH)₃/Fe(II) (Weber et al., 2006). The kinetics of Fe oxy-hydroxides 427 dissolution can also be fast enough for the whole process to become important during feeding. 428 Larval gut-passage time (GPT) for chironomids ingesting natural sediments varies from less 429 than 1 hour for assemblages of indigenous chironomids (Romito et al., 2010) to about 2 to 4 430 hours for Chironomus riparius (Wästlund, 1999); the variability in the latter case being linked to sediment characteristics. Under some conditions (pH=7 and partial H_2S pressure of 10^{-4} 431 432 atm), dissolution of Fe oxy-hydroxides is fast enough to occur during GPT (Dos Santos 433 Afonso and Stumm, 1992). All these physiological factors and kinetics aspects of sediment 434 digestion by chironomids would not be captured by the in vitro biomimetic approach. A more 435 effective chromium liberation in the gut lumen following dissolution of Fe oxy-hydroxides 436 could then be responsible for the higher level of Cr bioaccumulation at Maniowy.

437 The presence of heterogeneous levels of Cr contamination in sediments and the possible 438 preference of chironomids for relatively clean sediment patches could also contribute to the 439 observed discrepancies between Cr concentrations in water and sediment vs. accumulation by 440 chironomids. Chromium concentrations in sediments did vary markedly in time and space. 441 Results from November 2006 show that Cr concentrations from surface sediment layers (0-442 0.5 cm) can vary 3-4-fold inside both sampling zones (Table 1). Surface sediment samples 443 collected at Waksmund in September 2005 from 9 sub-sites arranged in a regular grid covering an area of 1 m² had a Cr content ranging between 132 and 1497 μ g g⁻¹ (arithmetic 444 mean ± 1 standard deviation = 431 \pm 379 µg g⁻¹; Table S2). Other samples from Waksmund 445 446 collected in March, May, August 2005 showed a 38-fold variability of total Cr concentrations (Table S2). However, only a 4-fold change in Cr levels was measured in indigenous chironomids recovered from these sediments at the same time (Table S2) and organisms from the cleanest sediments were those with the highest Cr burden after correction for the sediment gut content (Table S2). Whether this may be indicative of a decreased Cr availability in highly contaminated sediments or avoidance of such sediments by benthic organisms remains to be ascertained.

453 In any case, it is worth noting that the gut-corrected Cr concentrations measured in the 454 pools of indigenous chironomids at Waksmund and Maniowy (Table 1 and Table S2) often exceeded the range of 10–69 μ g g⁻¹ d.w. associated with a decrease in the presence of other 455 456 less tolerant taxonomic groups (Bervoets et al., 2016). In June 2004, chironomids were also 457 sampled at two sites not directly impacted by tannery discharges: the Bialka river before its 458 entrance in the Czorsztyn reservoir (Figure 1) and a small sandy beach located east of the mouth of the same river. Chromium levels in sediments from these sites were 44 $\mu g~g^{\text{-1}}$ and 32 459 $\mu g g^{-1}$, respectively, and the corresponding gut-corrected Cr concentrations in chironomids 460 were 2.8 μ g g⁻¹ and 11.3 μ g g⁻¹. The latter values compare favourably with Cr levels that 461 462 should not negatively affect the structure of benthic communities according to Bervoets et al. 463 (2016). Under this assumption, Cr accumulation at Waksmund and Maniowy (Table 1 and S2) 464 could therefore be high enough to cause adverse effects on the benthic community structure.

Finally, the actual species composition of the organisms we described under the generic name of 'indigenous chironomids' may also contribute to the observed discrepancies between ambient Cr concentrations and Cr body burdens. Because different species of chironomids accumulate contaminants to different extents (Martin et al., 2008), we compared our simplified sampling method with a more inclusive procedure of organism collection to identify the taxonomic groups dwelling at each site in June 2007 (see section 2.1). The chironomids' community at Waksmund turned out to be more abundant and more 472 heterogeneous than at Maniowy, where the community included only specimen of the genus 473 Chironomus (Fig. 3 and Table S3). This situation reflects typical differences between 474 piedmont streams (i.e., Waksmund) where biodiversity is typically high (Kownacki, 1995) 475 and reservoirs (i.e., Rożnów reservoir on the Dunajec River) where larvae of the genus 476 Chironomus usually dominate (Dumicka et al., 1986). Practically speaking, our simplified 477 sampling procedure did not alter the community pattern at Maniowy, but selectively sampled 478 individuals of the genera Chironomus and Stictochironomus, representing only 20 % of the 479 total species, at Waksmund (Fig. 3 and Table S3). Stictochironomus can actively migrate 480 along the sediment column (Kiyashko et al., 2004) and feed on less contaminated sediments 481 (see the sediment profile in Table 2). This scenario may then explain why observed Cr 482 concentrations at Waksmund were lower than those expected from the concentration in 483 sediments. The differences in species composition also indicate that chironomids at Maniowy 484 may accumulate higher body burdens of Cr because of their feeding strategy (i.e., surface 485 sediment feeders); a situation similar to that reported by Martin et al. (2008) for two 486 sympatric species of Chironomus.

Given their ecological traits (McLachlan, 1977; Stief and de Beer, 2006), chironomids may indeed accumulate part of their Cr burden from freshly deposited suspended particulate matter which, in the Dunajec river, is highly variable in both concentration and Cr content (Szalinska et al., 2010; Szalinska et al., 2013). The companion paper (Ferrari et al., this issue) will present evidence supporting the hypothesis that ingestion of freshly deposited suspended matter at the sediment water interface can constitute a prominent route of exposure for surface deposit feeders such as *Chironomus* spp. (i.e. the predominant genus occurring at Maniowy).

494

495

497 **5.** Conclusions

498 High Cr concentrations in sedimentary and aqueous matrices impacted by tannery discharges 499 were not systematically accompanied by high Cr body burdens in surface deposit feeders 500 dwelling at the two sampling locations. However, Cr accumulation by indigenous 501 chironomids did occur despite the low proportion of sediment-bound Cr liable to desorption 502 during gut passage (about 0.1% as estimated using a biomimetic approach). Concentrations of 503 DGT-labile Cr, considered as a proxy for bioavailable Cr(III), and Cr(VI) in pore waters were 504 less than 1% of the total Cr levels. On the other hand, Cr(VI) accounted for 20 to 30% of the 505 total in the overlying water confirming that oxidation of Cr(III) used in the tanning process to 506 Cr(VI) occurred and suggesting that sediments can act as a source of Cr(VI) to the overlying 507 water column. While there is a consensus that deposit feeders get most of their contaminant 508 body burden from sediment ingestion, the possible contribution of water born Cr species to 509 the element body burden in chironomids requires further study.

510 Although Cr accumulation by assemblages of indigenous chironomids is potentially 511 affected by numerous factors (from spatial patchiness in sediment contamination to specific 512 traits of various chironomids genera and species), the present work confirms that Cr released 513 by tannery discharges can accumulate in benthic organisms. These findings are particularly 514 important considering that Cr(III), usually regarded as of little environmental concern, is 515 expected to be the predominant redox form of Cr in sediments impacted by tannery 516 discharges. Metal-tolerant organisms such as chironomids were apparently able to cope with total Cr content exceeding 1000 μ g g⁻¹ in the bulk sediment phase. However, such levels of 517 518 contamination may be high enough to exclude more sensitive taxa, resulting in a simplified 519 community composition and, eventually, non compliance with good ecological quality as 520 required by the Water Framework Directive.

522 Acknowledgments

This research was supported by the Swiss National Fund (grants 200020-101844 and 200020-117942). The authors express their sincerest thanks to Michal Krzyszkowski and the direction of the Niedzica Hydro-Electric Power Plant (Poland) for their precious cooperation. We also thank our colleagues Vincent Chanudet, Régis Kottelat and Stephanie Jüstrich from the Institute F.-A. Forel for their help in field work. Ewa Szalinska thanks the Université de Lorraine (Bourse Chercheurs Invités) for supporting a scientific visit to LIEC.

529

530

531 **References**

532 Alves RH, Rietzler AC. 2015. Ecotoxicological evaluation of sediments applied to

533 environmental forensic investigation. Brazilian Journal of Biology. 75:886-893.

- 534 Baumann Z, Fisher NS. 2011. Modeling metal bioaccumulation in a deposit-feeding
- polychaete from labile sediment fractions and from pore water. Science of The Total
 Environment. 409(13):2607-2615.
- 537 Becker DS, Long ER, Proctor DM, Ginn TC. 2006. Evaluation of potential toxicity and
- 538 bioavailability of chromium in sediments associated with chromite ore processing
- 539 residue. Environ Toxicol Chem. 25(10):2576-2583.
- 540 Bencheikh-Latmani R, Obraztsova A, Mackey MR, Ellisman MH, Tebo BM. 2007. Toxicity
- of Cr(III) to *Shewanella* sp. strain mr-4 during Cr(VI) reduction. Environ Sci Technol.
 41:214-220.
- 543 Berry WJ, Boothman WS, Serbst JR, Edwards PA. 2004. Predicting the toxicity of chromium
 544 in sediments. Environ Toxicol Chem. 23:2981-2992.

Bervoets L, De Jonge M, Blust R. 2016. Identification of threshold body burdens of metals for
the protection of the aquatic ecological status using two benthic invertebrates.

547 Environmental Pollution. 210(Supplement C):76-84.

- 548 Besser JM, Brumbaugh WG, Kemble NE, May TW, Ingersoll CG. 2004. Effects of sediment
- 549 characteristics on the toxicity of chromium(III) and chromium(VI) to the amphipod,

550 *Hyalella azteca*. Environ Sci Technol. 38:6210-6216.

- Bigorgne E, Cossu-Leguille C, Bonnard M, Nahmani J. 2010. Genotoxic effects of nickel,
 trivalent and hexavalent chromium on the *Eisenia fetida* earthworm. Chemosphere.
 80(9):1109-1112.
- Bobrowski A, Bas B, Dominik J, Niewiara E, Szalinska E, Vignati D, Zarebski J. 2004.
- 555 Chromium speciation study in polluted waters using catalytic adsorptive stripping 556 voltammetry and tangential flow filtration. Talanta. 63:1003-1012.
- 557 Bobrowski A, Kapturski P, Zarębski J, Dominik J, Vignati DAL. 2012. Catalytic adsorptive

558 stripping voltammetric determination of chromium(VI) in overlying and interstitial

559 waters isolated from sediments contaminated by tannery waste. Analytical Letters.

560 45(5-6):495-507.

- 561 Chen Z, Mayer MM. 1999. Assessment of sedimentary Cu availability: A comparison of
 562 biomimetic and AVS approaches. Environ Sci Technol. 33(4):650-652.
- 563 Croizier WJ. 1924. Hydrogen ion concentration within the alimentary tract of insects. The
 564 Journal of General Physiology. 6(3):289-293.
- Davison W, Zhang H, Grime GW. 1994. Performance characteristics of gel probes used for
 measuring the chemistry of pore waters. Environ Sci Technol. 28(9):1623-1632.
- 567 Dazy M, Béraud E, Cotelle S, Meux E, Masfaraud J-F, Férard J-F. 2008. Antioxidant enzyme
- 568 activities as affected by trivalent and hexavalent chromium species in *Fontinalis*

569 *antipyretica* hedw. Chemosphere. 73:281-290.

571 and metal accumulation in aquatic invertebrates: Implications of feeding behavior and 572 ecology. Environmental Pollution. 158(5):1381-1391. de Sousa EA, Luz CC, de Carvalho DP, Dorea CC, de Holanda IBB, Manzatto ÂG, Bastos 573 574 WR. 2016. Chromium distribution in an amazonian river exposed to tannery effluent. 575 Environmental Science and Pollution Research. 23(21):22019-22026. 576 Dominik J, Vignati DAL, Koukal B, de Abreu MHP, Kottelat R, Szalinska E, Bas B, 577 Bobrowski A. 2007. Speciation and environmental fate of chromium in rivers 578 contaminated with tannery effluents. Engineering in Life Sciences. 7(2):155-169. 579 Dos Santos Afonso M, Stumm W. 1992. Reductive dissolution of iron(III) (hydr)oxides by 580 hydrogen sulfide. Langmuir. 8(6):1671-1675. 581 Dumicka E, Zieba J, Zurek R. 1986. Characteristics of zooplankton and macrobenthos in the 582 Rożnów dam reservoir (southern poland). Acta Hydrobiologica. 28(3/4):393-413. 583 Ernstberger H, Zhang H, Davison W. 2002. Determination of chromium speciation in natural 584 systems using DGT. Anal Biochem Chem. 373:873-879. 585 European Union 2005. European Risk Assessment Report on chromium trioxide, sodium 586 chromate, sodium dichromate, ammonium dichromate and potassium dichromate. 587 Volume 53. 588 Ferrari B, Vignati D, Roulier J-L, Coquery M, Szalinska E, Bobrowski A, Czaplicka A, 589 Dominik J. submitted. Chromium bioavailability in aquatic systems impacted by 590 tannery wastewaters. Part 2: New insights from laboratory and in situ testing with 591 Chironomus riparius. Science of The Total Environment. 592 Ferrari BJD, Vignati DAL, Dominik J. 2014. Bioaccumulation kinetics and effects of 593 sediment-bound contaminants on chironomids in deep waters: New insights using a 594 low-disturbance in situ system. Environmental Technology. 35(4):456-469.

De Jonge M, Blust R, Bervoets L. 2010. The relation between acid volatile sulfides (AVS)

- Giusti L, Barakat S. 2005. The monitoring of Cr(III) and Cr(VI) in natural water and
 synthetic solutions: An assessment of the performance of the dgt and dpc methods.
 Wat Air Soil Poll. 161:313-334.
- Gorny J, Billon G, Noiriel C, Dumoulin D, Lesven L, Madé B. 2016. Chromium behavior in
 aquatic environments: A review. Environmental Reviews. 24(4):503-516.
- Gunkel-Grillon P, Laporte-Magoni C, Lemestre M, Bazire N. 2014. Toxic chromium release
 from nickel mining sediments in surface waters, New Caledonia. Environ Chem Lett.
 12(4):511-516.
- Hare L, Campbell PGC, Tessier A, Belzile N. 1989. Gut sediment in a burrowing mayfly

604 (ephemeroptera, *Hexagenia limbata*): Their contribution to animal trace element

- burderns, their removal, and the efficacy of a correction for their presence. Can J Fish
 Aquat Sci. 46:451-456.
- 607 Hayzoun H, Garnier C, Durrieu G, Lenoble V, Le Poupon C, Angeletti B, Ouammou A,

Mounier S. 2015. Organic carbon, and major and trace element dynamic and fate in a

609 large river subjected to poorly-regulated urban and industrial pressures (Sebou river,

610 Morocco). Science of The Total Environment. 502(Supplement C):296-308.

611 Hursthouse AS, Matthews J, Figures J, Iqbal-Zahid P, Davies IM, Vaughan DH. 2001.

612 Chromium behaviour in intertidal sediments and pore waters, R. Clyde, UK.

Environmental Geochemistry and Health. 23(3):253-259.

614 Jasso-Chávez R, Pacheco-Rosales A, Lira-Silva E, Gallardo-Pérez JC, García N, Moreno-

- 615 Sánchez R. 2010. Toxic effects of Cr(VI) and Cr(III) on energy metabolism of
- 616 heterotrophic *Euglena gracilis*. Aquatic Toxicology. 100(4):329-338.
- Khwaja AR, Singh R, Tandon SN. 2001. Monitoring of Ganga water and sediments vis-à-vis
 tannery pollution at Kanpur (India): A case study. Environmental Monitoring and
- 619 Assessment. 68(1):19-35.

620	Kiyashko SI, Imbs AB, Narita T, Svetashev VI, Wada E. 2004. Fatty acid composition of
621	aquatic insect larvae Stictochironomus pictulus (Diptera: Chironomidae): evidence of
622	feeding upon methonotrophic bacteria. Comp Biochem Physiol Part B. 139:705-711.
623	Kownacki A. 1995. The use of chironomid pupal exuviae for ecological characterization of
624	the Upper Vistula (Southern Poland). Acta Hydrobiologica. 37(1).
625	Leermakers M, Gao Y, Gabelle C, Lojen S, Ouddane B, Wartel M, Baeyens W. 2005.
626	Determination of high resolution pore water profiles of trace metals in sediments of
627	the Rupel River (Belgium) using DET (Diffusive Equilibrium in Thin Films) and DGT
628	(Diffusive Gradients in Thin Films) techniques. Water Air Soil Pollut. 166(1-4):265-
629	286.
630	Lira-Silva E, Ramírez-Lima IS, Olín-Sandoval V, García-García JD, García-Contreras R,
631	Moreno-Sánchez R, Jasso-Chávez R. 2011. Removal, accumulation and resistance to
632	chromium in heterotrophic Euglena gracilis. Journal of Hazardous Materials. 193:216-
633	224.
634	Luoma SN, Rainbow PS. 2005. Why is metal bioaccumulation so variable? Biodynamics as
635	unifying concept. Environ Sci Technol. 39:1921-1931.
636	Lynch J. 1990. Provisional elemental composition values for eight new geochemical lake
637	sediment and stream sediment reference materials LKSD-1, LKSD -2, LKSD -3,
638	LKSD -4, STSD-1, STSD -2, STSD -3, STSD -4. Geostandards Newsletter.
639	14(1):153-167.
640	Martin S, Proulx I, Hare L. 2008. Explaining metal concentrations in sympatric Chironomus
641	species. Limnology and Oceanography. 53(2):411-419.
642	McLachlan AJ. 1977. Some effects of tube shape on the feeding of Chironomus plumosus L.
643	(Diptera: Chironomidae). J Anim Ecol. 46:139-146.

644	Outridge PM, Scheuhammer AM. 1993. Bioaccumulation and toxicology of chromium:
645	Implications for wildlife. Rev Environ Contam Toxicol. 130:31-77.
646	Pawlikowski M, Szalinska E, Wardas M, Dominik J. 2006. Chromium originating from
647	tanneries in river sediments: A preliminary investigation from the upper Dunajec
648	River (Poland). Polish Journal of Environmental Studies. 15(6):885-894.
649	Pereira CD, Techy JG, Ganzarolli EM, Quinaia SP. 2012. Chromium fractionation and
650	speciation in natural waters. Journal of Environmental Monitoring. 14(6):1559-1564.
651	Pettine M, Genneri F, Campanella L, Millero FJ. 2008. The effect of organic compounds in
652	the oxidation kinetics of Cr(III) by H_2O_2 . Geochim Cosmochim Acta. 72:5692-5707.
653	Ponti B, Bettinetti R, Dossi C, Vignati DAL. 2014. How reliable are data for the ecotoxicity
654	of trivalent chromium to Daphnia magna? Environmental Toxicology and Chemistry.
655	33(10):2280-2287.
656	Roditi HA, Fisher NS. 1999. Rates and routes of trace elements uptake in zebra mussel.
657	Limnol Oceanogr. 44(7):1730-1749.
658	Roditi HA, Fisher NS, Sañudo-Wilhelmy SA. 2000. Field testing a metal bioaccumulation
659	model for zebra mussels. Environ Sci Technol. 34(13):2817-2825.
660	Rodrigues M, Formoso M. 2006. Geochemical distribution of selected heavy metals in stream
661	sediments affected by tannery activities. Water, Air, & Soil Pollution. 169(1):167-184.
662	Romito AM, Eggert SL, Diez JM, Wallace JB. 2010. Effects of seasonality and resource
663	limitation on organic matter turnover by Chironomidae (Diptera) in southern
664	Appalachian headwater streams. Limnology and Oceanography. 55(3):1083-1092.
665	Rosales RM, Faz A, Gómez-Garrido M, Muñoz MA, Murcia FJ, González V, Acosta JA.
666	2016. Geochemical speciation of chromium related to sediments properties in the
667	riverbed contaminated by tannery effluents. J Soils Sediments.1-12.

669	dynamic mobilization of trace metals in sediments using DGT and comparison with
670	bioaccumulation in Chironomus riparius: first results of an experimental study.
671	Chemosphere. 70(5):925-932.
672	Sanyal T, Kaviraj A, Saha S. 2015. Deposition of chromium in aquatic ecosystem from
673	effluents of handloom textile industries in Ranaghat–Fulia region of West Bengal,
674	India. Journal of Advanced Research. 6(6):995-1002.
675	Saxena G, Chandra R, Bharagava RN. 2017. Environmental pollution, toxicity profile and
676	treatment approaches for tannery wastewater and its chemical pollutants. In: de Voogt
677	P, editor. Reviews of environmental contamination and toxicology. 240 :31-69.
678	Speranza A, Ferri P, Battistelli M, Falcieri E, Crinelli R, Scoccianti V. 2007. Both trivalent
679	and hexavalent chromium strongly alter in vitro germination and ultrastructure of
680	kiwifruit pollen. Chemosphere. 66:1165-1174.
681	Stief P, de Beer D. 2006. Probing the microenvironment of freshwater sediment macrofauna:
682	Implications of deposit-feeding and bioirrigation for nitrogen cycling. Limnology and
683	Oceanography. 51(6):2538-2548.
684	Stief P, Eller G. 2006. The gut microenvironment of sediment-dwelling Chironomus
685	plumosus larvae as characterised with O2, pH, and redox microsensors. J Comp
686	Physiol B. 176(7):673-683.
687	Szalinska E, Dominik J, Vignati DAL, Bobrowski A, Bas B. 2010. Seasonal transport pattern
688	of chromium(III and VI) in a stream receiving wastewater from tanneries. Applied
689	Geochemistry. 25(1):116-122.
690	Szalinska E, Smolicka A, Dominik J. 2013. Discrete and time-integrated sampling for
691	chromium load calculations in a watershed with an impoundment reservoir at an

Roulier JL, Tusseau-Vuillemin MH, Coquery M, Geffard O, Garric J. 2008. Measurement of

- 692 exceptionally low water level. Environmental Science and Pollution Research.
 693 20(6):4059-4066.
- Vignati DAL, Dominik J, Beye ML, Pettine M, Ferrari BJD. 2010. Chromium(VI) is more
 toxic than chromium(III) to freshwater algae: A paradigm to revise? Ecotoxicology
 and Environmental Safety. 73(5):743-749.
- Walsh AR, O'Halloran J. 1996. Chromium speciation in tannery effluent II. Speciation in the
 effluent and receiving estuary. Water Research. 30(10):2401-2412.
- Wang D, He S, Shan C, Ye Y, Ma H, Zhang X, Zhang W, Pan B. 2016. Chromium speciation
- in tannery effluent after alkaline precipitation: Isolation and characterization. Journalof Hazardous Materials. 316:169-177.
- Wang W-X, Fisher NS. 1999. Assimilation efficiencies of chemical contaminants in aquatic
 invertebrates: a synthesis. Environ Toxicol Chem. 18(9):2034-2045.
- Wang W-X, Griscom SB, Fisher NS. 1997. Bioavailability of Cr(III) and Cr(VI) to marine
- mussels from solute and particulate pathways. Environ Sci Technol. 31:603-611.
- Wästlund D. 1999. The role of sediment characteristics and food regimen in a toxicity test

707 with *Chironomus riparius*. Swedish University of Agricultural Sciences.

- 708 Weber KA, Achenbach LA, Coates JD. 2006. Microorganisms pumping iron: anaerobic
- microbial iron oxidation and reduction. Nature Reviews Microbiology. 4:752.
- 710 World Health Organization. 2009. Inorganic Chromium(III) Compounds. Geneva,
- 711 Switzerland.
- 712 World Health Organization. 2013. Inorganic Chromium(VI) Compounds. Geneva,

713 Switzerland.

714 Zhang H, Davison W. 1995. Performance characteristics of diffusion gradients in thin films

- for the in situ measurement of trace metals in aqueous solution. Anal Chem. 67:3391-
- 716 3400.

Figure captions

Figure 1. Study area (Dunajec River and Czorsztyn Reservoir; southern Poland) and location of the two sampling sites of Waksmund and Maniowy. The region of Nowy Targ, characterized by the presence of numerous small tanneries, is also indicated. Figure 2. Vertical profiles of dissolved and labile Cr concentrations (in µg L⁻¹) as estimated by DET and DGT probes exposed in the field at Waksmund (top panel) and Maniowy (bottom panel) for 45 hours in June 2007. Dashed lines represent the sediment-water interface. Note that DET and DGT depth do not exactly match. Figure 3. Relative percentage abundance of different taxomic groups in samples at Waksmund (Wak) and Maniowy (Man) following a full procedure (Wak and Man) or a simplified procedure (WakA and ManA). Grey bars, Chironomus spp.; White bars, Stictochironomus spp.; Black bars, other (see Table S3 for details).





...







787 Table 1. Chromium concentrations (µg g⁻¹) in field sediment and indigenous chironomids collected at 788 Waksmund and Maniowy during different periods. For samples collected in 2004 and 2007, concentrations in 789 sediments are arithmetic means (± 1 standard deviation; s.d.) of triplicate measures of homogenised sediment 790 aliquots (unless otherwise indicated in the footnotes). For samples collected in 2006, the range and the 791 arithmetic mean ± 1 s.d. of Cr concentration in sediments were obtained from analysis of surface sediment layers 792 (0-0.5 cm) obtained from sediment cores (see footnotes for details). Chromium concentrations extracted from 793 sediments using the biomimetic approach (see section 2.4 for more details) are also reported for the 2004 794 campaigns. Total and gut corrected Cr concentrations are for pooled chironomids samples (n=1, except for June 795 2007 where n=2). Correction for sediment gut content was done according to Hare et al. (1989) assuming a 796 sediment gut content equal to 6% of the organisms' weight (Ferrari et al. 2014). BAF, Bioaccumulation 797 Factor=total Cr in Chironomids/total Cr in sediments; Wak, Waksmund; Man, Maniowy; OM, Organic Matter; 798 n.d., not determined.

Period	Site	Sediments					
		Total Cr	Biomimetic Cr	OM	Total Cr	Gut corrected	BAF
		$(\mu g g^{-1})$	$(\mu g g^{-1})$	(%)	$(\mu g g^{-1})$	$Cr (\mu g g^{-1})$	
Jun 04	Wak	695 ^a	0.99±0.33	13.5	46	4.6	0.07
	Man	427±17 ^a	0.54 ± 0.06	4.47	202	188	0.47
Nov 04	Wak	761±27	0.57±0.051	9.73	43	0^{c}	0.06
	Man	367±10	0.51±0.08	6.52	120	104	0.33
Nov 06	Wak	884–3415 ^b	1	10.2	0.1	30–0 ^c	0.10-0.02
		(1802±744)	n.d. ±744)		81	0^{c}	
	Man 69–174 ^b		n.d.	4.20	30	27–20	0.43–0.29
		(118±39)				24	
Jun 07	Wak	1300±21	n.d.	17.6	222	153	0.17
	Man	241±11	n.d.	5.17	311±37	315	1.3

a) Waksmund, n=1; Maniowy, n=2. Error calculated as $Abs[(range)/\sqrt{2}]$

800 b) Range and, in parenthesis, arithmetic mean \pm one standard deviation of Cr concentration measured in surface

801 sediment layers (0–0.5 cm) from 9 independent cores.

802 c) Correction for sediment content in the gut yielded a negative value that was reported as zero in the table.

803

Table 2. Concentrations of total chromium in sediments, overlying water (OW) and porewaters from cores collected at Waksmund and Maniowy in June 2007. Cr(VI) concentrations in OW and porewaters are also reported. For Cr(VI) determinations, porewaters recovered from different sediment layers were sometimes combined due to analytical constraints (see Material and Methods and Bobrowski et al., 2012). n.a., porewater could not be recovered from these layers; ---, Maniowy core was sliced down to 9 cm.

	Waksmund			1	Maniowy	
Depth	Sediment	Porewater		Sediment	Porew	ater
(cm)	$(\mu g g^{-1})$	(µg	L ⁻¹)	$(\mu g g^{-1})$	(µg L	⁻¹)
	Total Cr	Total Cr	Cr (VI)	Total Cr	Total Cr	Cr
						(VI)
OW	n.a.	116	22.8	n.a.	3.16	0.90
0-0.5	5123	115		339	26.8	
0.5-1	2596	120	3.53	452	44.8	1.48
1–2	2488	106		291	56.2	
2–3	356	34.7	5.47	201	20.1	
3–4	161	18.0	0.73	214	26.2	
4–5	2155	74.6	n.a.	213	24.6	
5–6	6198	198	1.00	141	13.7	1.03
6–7	4789	187	1.04	69.9	7.16	
7–8	3642	310	1.16	80.0	n.a	
8–9	982	134	1.10	52.6	8.58	
9–10	470	n.a	n.a			
10–11	591	n.a	n.a			

Supplementary information for:

Chromium bioavailability in aquatic systems impacted by tannery wastewaters. Part 1:





851	Table S1a. Vertical profiles of total concentrations of Cr, Fe, and Mn (all in $\mu g g^{-1}$ d.w.) in
852	two sediment cores collected in June 2007 at Waksmund and Maniowy. Raw data were
853	centred and reduced before performing the correlation statistics presented in section 4.3 of the
854	main text.

		Waksmund			Maniowy	
Depth (cm)	Cr	Fe	Mn	Cr	Fe	Mn
0–0.5	5123	37,333	860	339	18,870	219
0.5–1	2596	40,268	943	452	22,209	238
1–2	2488	36,503	974	291	17,111	197
2–3	356	42,301	1371	201	15,487	206
3–4	161	42,321	1513	214	19,146	238
4–5	2155	32,965	1106	213	17,980	214
5–6	6198	29,898	774	141	13,868	186
6–7	4789	22,590	392	69.9	14,847	181
7–8	3642	27,691	405	80.0	15,917	212
8–9	982	15,254	199	52.6	12,879	167
9–10	470	10,247	140	n.a.	n.a.	n.a.
10-11	591	11,983	143	n.a.	n.a.	n.a.

Table S1b. Total concentrations of Cr, Fe, and Mn (all in $\mu g g^{-1}$ d.w.) in 12 (Waksmund) or 11 (Maniowy) sediment cores collected in November 2006 at Waksmund and Maniowy. Concentrations are for the uppermost sediment layer (0–0.5 cm). Cores from 1 to 9 are those considered in compiling Table 1 of the main text; cores from 10 to 12 are additional measurements. Raw data were centred and reduced before performing the correlation statistics presented in section 4.3 of the main text.

	Waksmund				Maniowy	
Core	Cr	Fe	Mn	Cr	Fe	Mn
number						
1	2131	28,740	631	94.5	9,740	115
2	1371	28,273	496	101	10,072	133
3	1959	26,545	470	122	11,776	161
4	2073	29,761	669	144	10,199	144
5	3415	28,287	545	95.8	10,324	117
6	1329	31,776	672	174	16,063	235
7	2017	29,101	575	169	12,193	150
8	884	31,643	640	69.2	10,842	155
9	1853	28,056	536	68.7	8,171	123
10	1325	29,776	528	114	10,210	176
11	2096	26,456	561	103	11,653	113
12	1231	26,617	500		Not available	

893	Table S2. Chromium concentrations ($\mu g g^{-1}$) in field sediment and indigenous chironomids
894	collected at Waksmund during different periods. Total and gut corrected Cr concentrations are
895	for pooled chironomids samples (n=1). Correction for sediment gut content was done
896	according to Hare et al. (1989) assuming a sediment gut content equal to 6% of the

897 organisms' weight (Ferrari et al. 2014).

Sampling date	Sediments	Chironomids		
	Total Cr	Total Cr	Gut-corrected Cr	
29/03/2005	4471±449 ^a	217	0 ^b	
27/05/2005	391±21	101	82.6	
15/08/2005	116±1.2 ^a	104	103	
Sept. 2005 ^d	431±379 ^c	52.9	28.8	

898 a) Analytical variability ; n=2 . Error calculated as Abs[(range)/ $\sqrt{2}$]

b) Correction for sediment content in the gut yielded a negative value that was reported aszero in the table.

901 c) Arithmetic mean±1 standard deviation of 9 different samples collected over an area of 1

 m^2 . Analytical variability on individual samples was less than 5%.

903 d) Exact day of sampling not in the record

Table S3. Number of individuals belonging to different taxa in samples collected at
Waksmund and Maniowy in June 2007 using different sampling strategies: Wak, collection
with a Surber Sampler (400 cm²); Man, collection with an Ekman grab sampler (225 cm²);
WakA and ManA, simplified collection procedure for determination of Cr content in
assemblages of indigenous chironomids described in section 2.1 of the main text.

	Waksmund		Maniowy	
	Wak	WakA	Man	ManA
TANYPODINAE				
Procladius spp.	2	0	0	0
Macropelopia sp.	4			
Thienemannimyia - complex	2	0	0	0
Tanypodinae (juv.)	4			
DIAMESINAE				
Pothastia longimanus KIEFFER				
1922	12.5	0	0	0
PRODIAMESINAE				
Prodiamesa olivacea (MEIGEN				
1818)	12	0	0	0
ORTHOCLADIINAE				
Eukiefferiella sp	8	0	0	0
Paracladius conversus (WALKER				
1856)	66	0	0	0
Orthocladiinae (juv.)	4			
CHIRONOMINAE				
Chironomini				
Chironomus spp. ¹	224	42	53	116
Cryptotendipes sp.	4	0	0	0
Microtendipes pedellus group	394	1	0	0
Stictochironomus sp.	40	51	0	0
Polypedilum (Polypedilum) sp.(cf.				
pedestre MEIGEN 1830) (p)				
Polypedilum (Polypedilum) spp.	12	2	1	0
Polypodilum (Tripodura) sp	2	0	0	0
Table S4 (continued)		I		

	Waksmund		Maniowy	
	Wak	WakA	Man	ManA
Paratendipes sp.	4	0	0	0
Chironomini (juv)	152			
Tanytarsini				
Micropsectra atrofasciata				
(KIEFFER 1911) (p)	1			
Micropsectra sp.	5	0	0	0
Tanytarsus brundini LINDEBERG				
1963	98	0	0	0
Tanytarsus pallidicorris				
(WALKER 1856) (p)				
Tanytarsus pallidicorris group)	66	0	0	0
Tanytarsus spp.	140	0	0	0
Cladotanytarsus sp.				
Tanytarsini (juv.)	88			
			54	
	1 345	96	(ind/225	116
TOTAL DENSITY	$(ind/400 \text{ cm}^2)$		cm ²)	

- 924 1,two species of *Chironomus* were found in the examined material on the basis of genetic
- study: *Chironomus plumosus* and *Chironomus bernensis* (Michailova et al., 2009).
- 926
- 927 References in SI:
- Ferrari B, Vignati D, Roulier J-L, Coquery M, Szalinska E, Bobrowski A, Czaplicka A,
 Dominik J. this issue. Chromium bioavailability in aquatic systems impacted by
 tannery wastewaters. Part 2: New insights from laboratory and in situ testing with
 chironomus riparius. Science of The Total Environment.
- Hare L, Campbell PGC, Tessier A, Belzile N. 1989. Gut sediment in a burrowing mayfly
 (ephemeroptera, *hexagenia limbata*): Their contribution to animal trace element
 burderns, their removal, and the efficacy of a correction for their presence. Can J Fish
 Aquat Sci. 46:451-456.
- Michailova P, Szarek-Gwiazda E, Kownacki A. 2009. Effect of contaminants on the genome
 of some species of genus chironomus (chironomidae, diptera) live in sediments of
 dunajec river and czorsztyn reservoir. Water Air Soil Pollut. 202(1):245-258.