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# Structure and Biosynthesis of Xenoamicins from Entomopathogenic *Xenorhabdus*

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**Abstract:** During the search for novel natural products from entomopathogenic *Xenorhabdus doucetiae* DSM17909 and *X. mauleonii* DSM17908 novel peptides named xenoamicins were identified in addition to the already known antibiotics xenocoumacin and xenorhabdin. Xenoamicins are acylated tri-decadepeptides consisting of mainly hydrophobic amino acids. The main derivative xenoamicin A (**1**) was isolated from *X. mauleonii* DSM17908, and its structure elucidated by detailed 1D

and 2D NMR experiments. Detailed MS experiments, also in combination with labeling experiments, confirmed the determined structure and allowed structure elucidation of additional derivatives. Moreover, the xenoamicin biosynthesis gene cluster was identified and analyzed in *X. doucetiae*

DSM17909, and its participation in xenoamicin biosynthesis was confirmed by mutagenesis. Advanced Marfey's analysis of **1** showed that the absolute configuration of the amino acids is in agreement with the predicted stereochemistry deduced from the nonribosomal peptide synthetase XabABCD. Biological testing revealed activity of **1** against *Plasmodium falciparum* and other neglected tropical diseases but no antibacterial activity.

**Keywords:** biosynthesis • natural products • peptides • structure elucidation

## Introduction

Natural products have always been a great source of novel bioactive compounds, in particular in the field of anti-infectives. As the number of multiresistant pathogens is increasing, it is necessary to continue research in the field of natural products,<sup>[1]</sup> and thus novel approaches for the identification of natural products from well-established or novel natural-product producers are required. Additionally, novel molecular approaches also allow the activation of previously silent gene clusters or heterologous expression.<sup>[2,3]</sup> Among novel natural-product producers, bacteria of the genus *Xen-*

*orhabdus* have shown promise.<sup>[4-6]</sup> The bacteria live in symbiosis with nematodes of the genus *Steinernema* and are released from the nematode gut when the nematode infects an insect. Besides insecticidal protein toxins *Xenorhabdus* also produces small molecules that are involved in insect virulence or act as antibiotics. Examples are the benzylideneacetone,<sup>[7]</sup> rhabduscin,<sup>[8]</sup> xenocoumacins,<sup>[9]</sup> xenorhabdins,<sup>[10]</sup> xenofuranones,<sup>[11]</sup> nematophin,<sup>[12]</sup> PAX-Peptides,<sup>[13]</sup> GameXPep-tides<sup>[14]</sup> and even depsipeptides, such as xenematides,<sup>[15]</sup> szentiamides,<sup>[15]</sup> and xentrival peptides,<sup>[16]</sup> which show antibiotic, antifungal, or cytotoxic activity. As several of the compounds may be released to protect the insect cadaver from food competitors living in the soil, they may also be potential anti-infectives. Here we describe the structure and biosynthesis of a novel class of large hydrophobic depsipeptides named xenoamicins from *Xenorhabdus* showing activity against *Plasmodium* and *Trypanosoma*.

## Results and Discussion

During our screening for novel natural products by UPLC-ESI-MS of XAD-16 extracts<sup>[17]</sup> from *Xenorhabdus* cultures in standard LB medium we identified a new family of natural products. According to characteristic neutral losses for amino acids such as valine, leucine, and isoleucine in their MS<sup>2</sup> fragmentation patterns, these compounds were identified as peptides.<sup>[18]</sup> Their molecular formulas were determined on the basis of MALDI-Orbitrap-MS data (Table S1 of the Supporting Information) revealing C<sub>64</sub>H<sub>109</sub>N<sub>15</sub>O<sub>15</sub> for xenoamicin A (**1**; *m/z* 1300.822 [*M*+H]<sup>+</sup>, Δppm 0.5) as the

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main product in *X. mauleonii* DSM17908. The determination of the molecular formula was facilitated and confirmed by the identification of the correct number of carbon and nitrogen atoms from labeling experiments with strain DSM 17908 in standard growth medium,  $^{13}\text{C}$ -labeled medium, and  $^{15}\text{N}$ -labeled medium (Figure S1 of the Supporting Information), as described previously.<sup>[14,16]</sup> Compound **1** (29.2 mg) was subsequently isolated from strain DSM17908 by preparative chromatography with HPLC-UV/ELSD in a three step protocol (Figures S2–S4 of the Supporting Information). The first two chromatographic steps were performed on  $\text{C}_{18}$  columns with different particle sizes under acidic conditions. In the third step the selectivity was changed through a higher pH value of the eluent. Due to the lack of a chromophore, **1** shows only weak absorption in the UV spectrum even at 215 nm. Despite several peaks with a strong absorption at 215 nm in the extract of *X. mauleonii* DSM17908, **1** could easily be identified with the universal signal of the evaporative light scattering detector (ELSD). Due to the relatively large amount of **1** in the extract of DSM17908 a large signal in the ELSD but not in the UV spectra could be observed, whereas other compounds in this extract with a chromophore gave large signals in both spectra (Figure S2 of the Supporting Information). Thus, the ELSD signal gives a very good indication of the actual amount of the individual compounds in the extract, independent of the physicochemical properties, and therefore ELSD is especially valuable for the detection of peptides without any chromophore.<sup>[19]</sup>

The structure of **1** was determined by comprehensive NMR studies. In most standard solvents, such as  $[\text{D}_6]\text{DMSO}$ ,  $\text{CD}_3\text{OD}$ , or  $\text{CDCl}_3$ , NMR spectra of poor quality were obtained. The resonances of the amide protons in particular appeared as very broad signals (Figures S5, S6 and Annex 1 of the Supporting Information). However, in  $[\text{D}_6]\text{benzene}$  at 293 K well-resolved NMR spectra with reasonable dispersion of the amide signals were obtained (Figure S6 of the Supporting Information). The assignment of all proton and carbon resonances was carried out with various 2D NMR techniques, including DQF-COSY, TOCSY, ROESY, multiplicity-edited HSQC, HSQC-TOCSY, and HMBC spectra (Table 1, Annexes 1–20 in the Supporting Information). The assignment was hampered by the presence of two sets of signals in a ratio of 1:1 resulting from proline *cis/trans* conformers (see below). Primary analysis of the spin systems in the TOCSY and HSQC-TOCSY spectra revealed the presence of 13 amino acid residues: five valines, one leucine, one isoleucine, two alanines, two prolines, one threonine, and one  $\beta$ -alanine residue. In addition a butyric acid moiety was observed. The two sets of signals are caused by different orientations of the amide bond between the  $\beta$ -alanine residue and one of the proline residues. In the *cis* conformer an intensive ROE correlation between proline  $\text{H}\alpha$  and  $\beta$ -alanine  $\text{H}\beta$  is observed. In the *trans* conformer such an intensive ROE should appear between proline  $\text{H}\delta$  and  $\beta$ -alanine  $\text{H}\beta$ . Unfortunately, this ROE cannot be assigned, because of peak overlaps between proline  $\text{H}\delta$  and

the  $\text{H}\beta$  of  $\beta$ -alanine. The presence of a *cis* and a *trans* conformer is also confirmed by the difference in  $^{13}\text{C}$  chemical shifts of  $\text{C}\beta$  and  $\text{C}\gamma$  of proline  $[\Delta(\beta\gamma) = \delta(^{13}\text{C}\beta - ^{13}\text{C}\gamma)]$ .<sup>[20]</sup> The differences for the *cis* and *trans* conformers are  $\Delta(\beta\gamma) = 8.83$  ppm and  $\Delta(\beta\gamma) = 5.17$  ppm, respectively, which are in agreement with literature data.<sup>[20]</sup> Due to overlap of the carbon resonances in the carbonyl region and line broadening of some amide protons, the HMBC spectrum was not sufficient for the sequential assignment (see Annex 3 of the Supporting Information). Therefore, the peptide sequence was determined by analysis of ROE correlations which resulted in a depsipeptide structure with ring formation between the C-terminal valine residue and the threonine side chain at position 7 (Figure 1). The ring closure could be confirmed by an ROE between threonine  $\text{H}\beta$  and valine 14  $\text{H}\alpha$  in both conformers, whereby the ROE of the *cis* conformer is only weak (Annex 13 of the Supporting Information). All other sequential NH/ $\text{H}\alpha$  ROE correlations of both conformers were observed with the exception of the correlation between valine 10 and alanine 11 in both conformers. Nevertheless, the connection of valine 10 and alanine 11 was in accordance with the molecular formula and was shown by MS fragmentation data (see below). The assignment of the single spin systems to the corresponding conformer was confirmed by additional long-range ROE correlations between valine 10 NH and isoleucine 6  $\text{H}\alpha$ . Moreover, a long-range ROE correlation between threonine  $\text{H}\beta$  and alanine 11  $\text{H}\alpha$  confirmed the assignment of the *trans* conformer. The N-terminal proline residue is acylated with butyric acid, as could be concluded from an ROE correlation between  $\text{H}\alpha$  of the acid and  $\text{H}\alpha$  of proline in position 2. Unfortunately, the alanine NH in position 11 of the *trans* conformer could not be detected due to very broad signals.

The configuration of the amino acids in **1** was elucidated by using the advanced Marfey's method<sup>[16,21,22]</sup> leading to the identification of the L configuration for isoleucine and both proline residues. The D configuration was revealed for leucine, *allo*-threonine, and both alanine residues. The ratio of L to D valine was found to be 1:4. Therefore, four valine residues in **1** may have the D configuration and one valine residue has the L configuration (Figure 2, Table S2 of the Supporting Information). However, the stereochemical assignment of the valine residues was not possible at this stage.

To verify and support the structure elucidation by NMR spectroscopy and to enable the structural elucidation of further derivatives in *X. doucetiae* DSM17909, MS-based structure elucidation was additionally performed for **1** in DSM1908. To confirm the building blocks, standard  $^{12}\text{C}$ -L-amino acids were added to bacterial cultures growing in a reversed labeling experiment in fully  $^{13}\text{C}$  labeled medium (Figure S7 of the Supporting Information), as described previously.<sup>[14,16]</sup> Incorporation could be detected by HPLC-ESI-MS as shifts to lower masses of  $m/z$  1.5 for  $\beta$ -alanine, 2 for threonine, 2.5 for proline, 3 for isoleucine, 3 for leucine, 1.5 for alanine, and 2.5 for valine, related to half of the numbers of carbon atoms incorporated due to the presence of doubly charged ions.

Table 1. NMR spectroscopic data of conformer mixture (600 MHz ( $^1\text{H}$ ), 150 MHz ( $^{13}\text{C}$ ) in  $[\text{D}_6]\text{benzene}$ ) of xenoamicin A (**1**),  $\delta$  in ppm. Spin system numbering from acylated N- to C-terminus. Only key sequential ROESY correlations are listed. Pos. = Position.

Amino acid	Pos.	<i>trans</i> Conformer			<i>cis</i> Conformer			
		$\delta_{\text{C}}$	$\delta_{\text{H}}$	ROESY	$\delta_{\text{C}}$	$\delta_{\text{H}}$	ROESY	
1-BA	$\alpha$	36.46	2.34	2- $\alpha$	36.46	2.34	2- $\alpha$	
	$\alpha$	36.46	1.95		36.46	1.95		
	$\beta$	18.63	1.90		18.63	1.90		
	$\beta$		1.85			1.85		
	$\gamma$	14.23	0.98		14.23	0.98		
2-Pro	$\alpha$	60.30	4.58	1- $\alpha$ , 3-NH	60.30	4.58	1- $\alpha$ , 3-NH	
	$\beta$	29.54	1.82	3-NH	29.54	1.82	3-NH	
	$\beta$		1.47	3-NH		1.47	3-NH	
	$\gamma$	24.96	2.08		24.96	2.08		
	$\gamma$		1.29			1.29		
3-Ala <sup>[a]</sup>	NH		8.71	2- $\alpha$ , 2- $\beta$		8.71	2- $\alpha$ , 2- $\beta$	
	$\alpha$	48.06 <sup>[b]</sup>	5.01	4-NH	48.01 <sup>[b]</sup>	5.03	4-NH	
	$\beta$	16.56	1.69		16.56	1.67		
	4-Val <sup>[a]</sup>	NH		7.82	3- $\alpha$		7.82	3- $\alpha$
		$\alpha$	60.04	4.84	5-NH	60.08	4.78	5-NH
$\beta$		30.70	2.25		30.70	2.25		
$\gamma$		19.63	1.10		19.64	1.06		
$\gamma$		19.45	1.04		19.38	1.00		
5-Leu	NH		9.22	4- $\alpha$		9.23	4- $\alpha$	
	$\alpha$	50.76	5.14	6-NH	50.93	5.13	6-NH	
	$\beta$	40.34	2.03		40.34	2.00		
	$\gamma$	25.56 <sup>[b]</sup>	1.97		25.50 <sup>[b]</sup>	1.97		
	$\delta$	22.59	1.09		22.59	1.09		
6-Ile	$\delta$	23.45	1.10		23.45	1.10		
	NH		8.70	5- $\alpha$		8.74	5- $\alpha$	
	$\alpha$	58.05	5.00	7-NH, 7- $\beta$ , 10-NH	58.01	4.92	7-NH, 7- $\beta$ , 10-NH	
	$\beta$	34.91	2.34		34.83	2.34		
	$\gamma_1$	9.75	0.96		9.75	0.94		
7-Thr	$\gamma_2$	25.44	1.72		25.39	1.72		
	$\gamma_2$		1.52			1.52		
	$\delta$	15.91	1.14		15.87	1.12		
	NH		9.60	6- $\alpha$		9.54	6- $\alpha$	
	$\alpha$	56.71	5.18	8-NH	56.14	5.15	8-NH	
8-Val	$\beta$	70.03	6.05	6- $\alpha$ , 11- $\alpha$ , 14- $\alpha$	71.82	5.82	6- $\alpha$ , 14- $\alpha$	
	$\gamma$	17.66	1.47		17.24	1.49		
	NH		9.03	7- $\alpha$ , 7- $\beta$		8.84	7- $\alpha$ , 7- $\beta$	
	$\alpha$	61.88	4.62	9-NH	61.54	4.56	9-NH	
	$\beta$	29.36	2.57		29.54	2.56		
9-Val	$\gamma$	19.72	1.37		19.71	1.33		
	$\gamma$	19.55	1.27		19.93	1.28		
	NH		8.97	8- $\alpha$		8.85	8- $\alpha$	
	$\alpha$	60.82	4.76	10-NH	60.57	4.79	10-NH	
	$\beta$	30.45 <sup>[b]</sup>	2.55		30.49 <sup>[b]</sup>	2.56		
10-Val	$\gamma$	19.99	1.24		19.91	1.25		
	$\gamma$	19.99	1.24		19.91	1.25		
	NH		9.15	6- $\alpha$ , 9- $\alpha$		9.01	6- $\alpha$ , 9- $\alpha$	
	$\alpha$	60.30	4.90		58.64	5.49		
	$\beta$	29.68	2.56		31.07	2.38		
11-Ala	$\gamma$	19.44 <sup>[b]</sup>	1.23		19.50 <sup>[b]</sup>	1.23		
	$\gamma$	19.36	1.18		19.09	1.17		
	NH					9.35		
	$\alpha$	49.93	5.45	7- $\beta$ , 12-NH	50.31	5.16	12-NH	
	$\beta$	17.24	1.75		19.29	1.58		
12- $\beta$ -Ala	NH		7.97	11- $\alpha$		6.70	11- $\alpha$	
	$\alpha$	35.66	4.20		35.79	4.42	13- $\alpha$	
	$\alpha$		3.38			4.26		
	$\beta$	34.80	2.84		35.13	3.00		
	$\beta$		2.18			2.35	13- $\alpha$	

Table 1. (Continued)

Amino acid	Pos.	<i>trans</i> Conformer			<i>cis</i> Conformer		
		$\delta_{\text{C}}$	$\delta_{\text{H}}$	ROESY	$\delta_{\text{C}}$	$\delta_{\text{H}}$	ROESY
13-Pro	$\alpha$	63.19	4.47		61.47	4.26	12- $\alpha$ , 12- $\beta$
	$\beta$	30.21	2.17		31.39	1.76	
	$\beta$		1.73			1.66	
	$\gamma$	25.04	1.40		22.56	1.26	
	$\gamma$		1.68			1.11	
14-Val	$\delta$	46.57	3.20	14-NH	46.48	3.54	14-NH
	$\delta$		2.81	14-NH		3.22	14-NH
	NH		6.60	7- $\beta$ , 13- $\delta$		6.65	13- $\delta$
	$\alpha$	57.70	4.69	7- $\beta$	56.14	4.51	7- $\beta$
	$\beta$	32.77	2.27		32.05	2.17	
	$\gamma$	18.83	0.97		18.81	0.66	
	$\gamma$	18.33	0.96		16.98	0.63	

[a] Interchangeable spin systems in the different conformers at the same position. [b] Interchangeable positions in the spin systems of the different conformers at the same position.

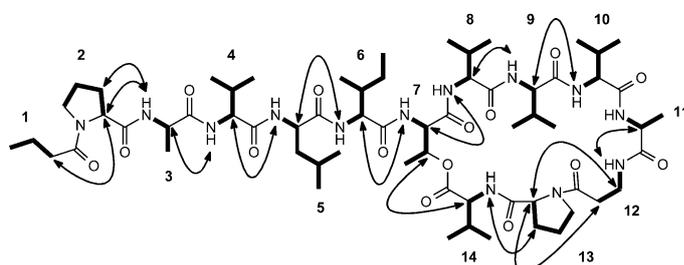


Figure 1. Connectivities in xenoamicin A (**1**), as determined by COSY and TOCSY (bold lines). Arrows indicate ROESY correlations used for the sequential assignment. Additionally the numeration of the spin systems is shown.

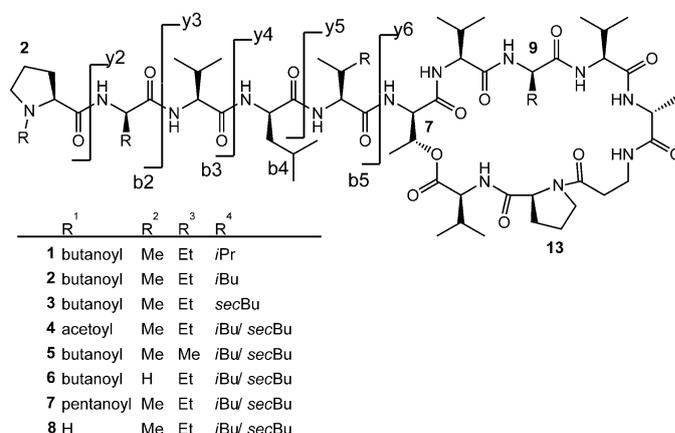


Figure 2. Structures of xenoamicins A–H (**1–8**) indicating b2–b5 and y2–y6 fragmentation ions of the MS<sup>2</sup> spectra. No differentiation between leucine or isoleucine at position 9 (R<sup>4</sup>) could be obtained for **4–8**.

The more intense doubly charged ions were used for fragmentation and showed mainly b2–b5 and y2–y6 ions (Figure 2, Figure S8 and Table S3 of the Supporting Information), as xenoamicins exhibited a characteristic collision-induced dissociation (CID) MS<sup>2</sup> fragmentation pattern. However, as the ring stayed intact in the CID experiment, the sequence of the ring could not be determined. There-

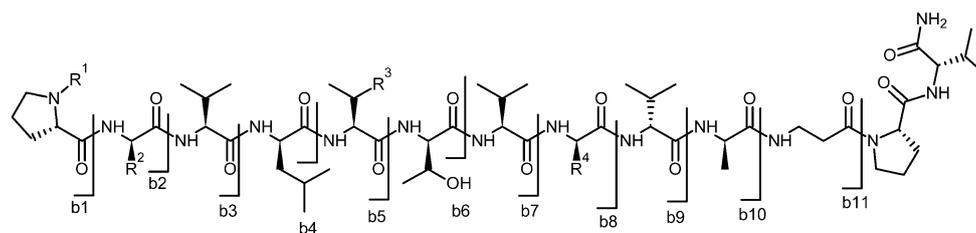


Figure 3. Structures of linear xenoamicins A–C (**1–3**) after reaction with  $\text{NH}_3$ . Described b1–b11 ions could be observed in  $\text{MS}^n$  fragmentations. For  $\text{R}^1$ – $\text{R}^4$  see Figure 2.

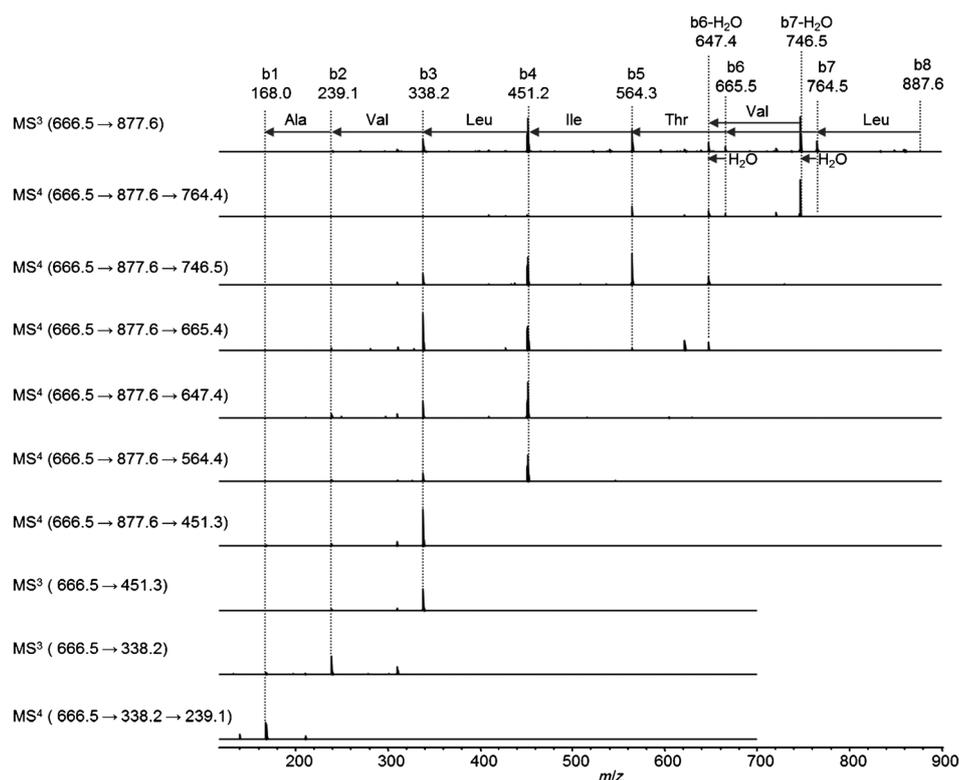


Figure 4.  $\text{MS}^3$  and  $\text{MS}^4$  spectra of linear xenoamicin B (**2**) after reaction with  $\text{NH}_3$ . According to  $\text{MS}^3$  and  $\text{MS}^4$  spectra, b2–b5 and b8 ions could be assigned in  $\text{MS}^2$  spectra. Assignment of b9–b11 ions were proposed due to complex  $\text{MS}^3$  and  $\text{MS}^4$  fragmentation.

fore, xenoamicin-containing extracts were hydrolyzed with 28% aqueous  $\text{NH}_3$  solution, which resulted in positive mass shifts of 8.5 Da and 9 Da in doubly charged species for the addition of  $\text{NH}_3$  (17 Da; resulting in the linear amide) or for the addition of water (18 Da; resulting in the linear acid), respectively (Figure 3, Table S4 of the Supporting Information).<sup>[23]</sup> Based on CID  $\text{MS}^2$  of the linearized peptide (Figure S9 of the Supporting Information) and  $\text{MS}^n$  experiments (Figure S10 of the Supporting Information), b1–b8 ions could be identified and also used for determination of the amino acid sequence of the ring, which confirmed the sequence of **1** as determined by NMR spectroscopy. The same analyses were performed for compounds **2** and **3**, which appear in *X. mauleonii* DSM17908 and *X. doucetiae* DSM17909 (Figures S8 and S9, Tables S3 and S4 of the Sup-

porting Information). Both show  $m/z$  1314.8  $[\text{M}+\text{H}]^+$  indicating the presence of an additional methyl group in both compounds. The location of this additional methyl group in **2** and **3** could be identified as position 9 due to the difference between b7 and b8 ions (Figures 3 and 4; Figure S11 of the Supporting Information). In **1** the difference between the b7 and b8 ions of 99 Da represents the neutral loss of a valine building block, whereas the difference in **2** and **3** of 113 Da indicates leucine or isoleucine, respectively. Isobaric leucine and isoleucine could be distinguished by analyzing the results from inverse feeding experiments (Figure S7 of the Supporting Information). By feeding  $^{12}\text{C}$ -leucine to a culture in  $^{13}\text{C}$ -enriched media two shifts of 3 Da from the double charged species could be seen for **2** and show the incorporation of two leucine residues instead of one leucine residue in **1** and **3**. For **3**

two shifts for isoleucine could be observed instead of one shift in **1** and **2** after the feeding of  $^{12}\text{C}$ -isoleucine. The extracts of the inverse feeding experiments were also hydrolyzed with 28% aqueous  $\text{NH}_3$  solution. However, the intensity of the b6 and b7 ions of the  $^{13}\text{C}$ -peptides with incorporated  $^{12}\text{C}$ -leucine or  $^{12}\text{C}$ -isoleucine was too low for detection (Table S4 of the Supporting Information). In summary,  $\text{MS}^4$  fragmentation and CID  $\text{MS}^2$  fragmentation of the linearized peptides together with inverse feeding experiments were sufficient for the complete sequential structural elucidation of **1–3**.

Additionally, xenoamicins D–H (**4–8**) could be detected in DSM17909, but not in DSM17908, as doubly charged species  $[\text{M}+2\text{H}]^{2+}$ ,  $m/z$  644.0, 651.0, 651.0, 665.0, although these were only produced in trace amounts (Table S1 of the

Supporting Information). Their structure could only partially be determined from their b<sub>2</sub>–b<sub>5</sub> and y<sub>2</sub>–y<sub>6</sub> ions in MS<sup>2</sup> spectra (Figure S12 and Table S5 of the Supporting Information). Determination of the ring amino acids of **4–8** was not possible. However, as variability in the ring was only observed at position 9 in compounds **1–3**, this may also be the variable position for compounds **4–8**, although no differentiation between leucine and isoleucine is possible. As described above, the differentiation of leucine and isoleucine can only be achieved by reversed feeding experiments and subsequent linearization if the structure elucidation is performed by MS. However, the amounts of **4–8** were not sufficient for further characterization.

Nevertheless, **5** and **6** show variability in positions 6 and 3, respectively. Furthermore, two additional N-terminal acyl moieties, acetate and pentanoate, were observed in **4** and **8**. Experimentally proven characterization of the ring amino acids is not essential for elucidation of these variabilities.

The b<sub>2</sub> ion of **4** with a mass of 211 Da revealed that acetate is acylated with the N-terminal proline residue. The neutral loss of proline is 97 Da and that of alanine 71 Da. Therefore the remaining component of the b<sub>2</sub> ions had a mass of 43 Da indicating acetate. By comparison of the b<sub>2</sub> ions with 211 Da and the y<sub>2</sub> ions with 1147 Da, alanine in position 3 is confirmed. Valine in position 4, leucine in position 5, and isoleucine in position 6 could be predicted by the mass shifts of 99, 113, and 113 Da between the y<sub>3</sub>, y<sub>4</sub>, y<sub>5</sub>, and y<sub>6</sub> ions, respectively. Leucine and isoleucine cannot be distinguished by their mass, but it can be assumed that leucine is incorporated at position 5 and isoleucine at position 6, as in **1–3**. However, CID MS<sup>2</sup> fragmentation was not sufficient for the analysis of the ring. Therefore, it had to be assumed that the ring consists of the same or very similar building blocks as in **1–3**. However, because the y<sub>6</sub> ions of **4** showed a mass of 751 Da, like for the y<sub>6</sub> ions of **2** and **3**, we propose that leucine or isoleucine was incorporated at position 9, as in **2** and **3**, and that all other amino acids in the ring were as in **1–3**.

The CID MS<sup>2</sup> fragmentation revealed that compound **5** has alanine in position 3, valine in position 4, leucine in position 5, and valine in position 6 due to the shifts of 71, 99, 113, and 99 Da between the y<sub>2</sub>, y<sub>3</sub>, y<sub>4</sub>, y<sub>5</sub>, and y<sub>6</sub> ions, respectively. Therefore this compound shows variability in position 6 due to the incorporation of valine instead of isoleucine. The b<sub>2</sub> ions with a mass of 239 Da revealed a butanoyl acid unit in position 1. The obtained y<sub>6</sub> ions for compound **5** exhibited the same size as for compound **4**, **6**, **7**, and **8**, thus the structure shown in Figure 2 resulted.

In compound **6** glycine instead of alanine is incorporated in position 3, as revealed by the mass shift of 57 Da between the y<sub>2</sub> and y<sub>3</sub> ions. According to the shifts between y<sub>3</sub>, y<sub>4</sub>, y<sub>5</sub>, and y<sub>6</sub> ions, valine, leucine, and isoleucine are incorporated in positions 4, 5, and 6, respectively. The loss of 14 Da of the b<sub>2</sub> ions in comparison with the b<sub>2</sub> ions of **1**, **2**, **3**, and **5** resulted from incorporation of glycine instead of alanine.

Compound **7** showed the same mass shifts as **2–4**. Therefore, the same amino acid sequence can be assumed. How-

ever, the b<sub>2</sub> ions with a mass of 253 Da showed that **7** contained a pentanoyl moiety as the difference of the y<sub>2</sub> and the y<sub>3</sub> ions indicated an alanine in position 3. When the mass of alanine and proline were subtracted from the mass of the b<sub>2</sub> ions the mass of a pentanoyl moiety is left.

The MS<sup>2</sup> spectrum of **8** is more complex. Therefore the MS<sup>2</sup> spectrum of the sodium adducts of **8** is additionally presented (Figure S12 of the Supporting Information). The y<sub>2</sub>–y<sub>6</sub> ions showed that **8** exhibited the same amino acid sequence and ring size as **2–4** and **7**. Taking the molecular mass of 1244 Da into account, it can be assumed that the N-terminus of **8** is not acylated.

To fully assign the absolute configuration of xenoamicin, which was not completely possible from the NMR data and amino acid analysis, the biosynthesis gene cluster was identified and analyzed. As the genome of *X. mauleonii* DSM17908 is not available yet, the already finished genome sequence of strain DSM17909 (originally strain FRM16) (<http://www.genoscope.cns.fr/agc/microscope/home/index.php>) was searched for a biosynthesis gene cluster encoding nonribosomal peptide synthetases (NRPS) involved in xenoamicin biosynthesis (*xab*) by using the antismash software tool.<sup>[24–26]</sup> The only candidate that fits the predicted biosynthesis gene cluster encodes five nonribosomal peptidases XabABCD and an aspartic acid decarboxylase XabE putatively involved in the formation of β-alanine.<sup>[14,27,28]</sup>

To prove that this gene cluster is indeed involved in xenoamicin biosynthesis, the gene *xabB* encoding the second NRPS was disrupted by plasmid insertion. Comparison between the *xabB:cat* mutant and the wild type showed complete loss of **1–8** in the mutant, whereas all other natural products such as xenorhabdins and xenocoumacins were still produced (Figures S13 and S14 of the Supporting Information).

Overall 13 modules and all domains for the biosynthesis of the xenoamcins could be identified (Figure 5, Table S6 of the Supporting Information). The presence of an adenylation (A) domain specific for β-alanine was readily detectable due to the differences in the specificity conferring code (Table S7 of the Supporting Information).<sup>[29–31]</sup> Analysis of the condensation (C) domains<sup>[32]</sup> revealed the presence of dual condensation/epimerization (C/E) domains<sup>[33]</sup> in modules 3, 5, 7, 9, and 11. Thus, it can be assumed that the amino acids incorporated in the previous modules 2, 4, 6, 8, and 10 are epimerized, which is in accordance with the results from the advanced Marfey's analysis (Table S2 of the Supporting Information). For example, exclusively L-proline and D-alanine were detected in the Marfey's analysis and only C and C/E domains were identified following the modules responsible for the incorporation of proline (modules 1 and 12) and alanine (modules 2 and 10), respectively. Furthermore, **1–3** were identical in both strains according to the MS analysis of the linearized peptides as well as the retention times of cyclic or linear peptides. Thus, the absolute configuration of all amino acids in xenoamcins can be predicted accordingly and this also allows the stereochemical assignment for the valine moieties to be made (Figure 2).

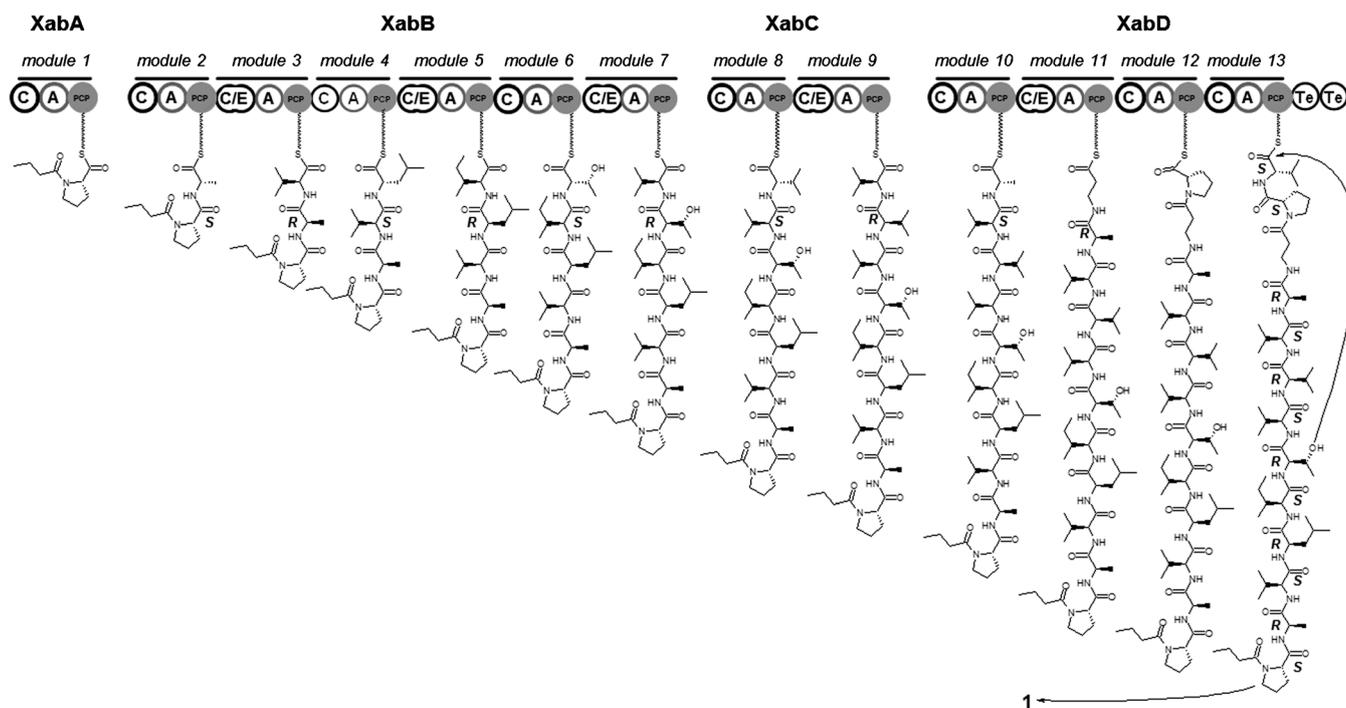


Figure 5. Proposed biosynthesis of xenoaamicin A (**1**) in *X. doucetiae* DSM 17909, showing the nonribosomal peptide synthetases XabABCD involved in its biosynthesis. Domains and modules are labeled: adenylation domain (A), peptidyl carrier protein domain (PCP), condensation domain (C), condensation/epimerization domain (C/E), thioesterase domain (Te).

With alternating L and D configuration in the peptide chain, xenoaamicins belong to the class of peptides which can form  $\beta$  helices in membranes and hence ion channels. The best-studied example is the antibiotic gramicidin D,<sup>[34]</sup> which is also formed nonribosomally and requires two molecules to span a membrane. Recently, a novel class of ribosomally made alternating D/L peptides, namely, proteusins, which has strong cytotoxic activity and polytheonamide as most prominent member, has been identified and its biosynthesis elucidated.<sup>[35]</sup>

In contrast to these D/L peptides, **1** most likely cannot form such a  $\beta$ -helix due to the depsipeptide structure. Nevertheless, the presence of such a large number of hydrophobic amino acids and a N-terminal acyl moiety in **1** may indicate an interaction with the membrane. Hence, the bioactivity of xenoaamicin A (**1**) was investigated. Whereas no antibacterial or antifungal activity could be observed in an agar diffusion assay with *E. coli* DH10B, *B. subtilis*, *M. luteus*, *P. aeruginosa*, *S. cerevisiae*, and *C. albicans*,<sup>[36]</sup> good activity against *Plasmodium falciparum* NF 54 (IC<sub>50</sub> of 2.35  $\mu\text{g mL}^{-1}$ ) and *Trypanosoma brucei* rhodesiense STIB900 (IC<sub>50</sub> of 6.41  $\mu\text{g mL}^{-1}$ ) was observed. Much weaker activity against *Trypanosoma cruci* Tulahuen C4 (IC<sub>50</sub> of 30.5  $\mu\text{g mL}^{-1}$ ), *L. donnovanni* MHOM-ET-67L82 (IC<sub>50</sub> of 50.1  $\mu\text{g mL}^{-1}$ ), and mammalian L6 cells (IC<sub>50</sub> of 68.5  $\mu\text{g mL}^{-1}$ ) indicated a specific target.<sup>[37]</sup> However, further analysis is needed to identify this target and also to elucidate the natural function of this novel class of compounds in the complex life cycle of *Xenorhabdus* with its nematode host and insect prey.

## Experimental Section

**Xenoaamicin A (1):** White powder; <sup>1</sup>H NMR (600 MHz, [D<sub>6</sub>]benzene, 25°C): see Table 1; <sup>13</sup>C NMR (150 MHz, [D<sub>6</sub>]benzene, 25°C, see Table 1; MS/MS (70 eV) and HRMS (MALDI): see Table S1 of the Supporting Information.

**Cultivation and extraction:** *X. doucetiae* DSM17909 and *X. mauleonii* DSM17908 were always cultivated at 30°C, 200 rpm. For inoculation, 1% overnight preculture and 2% XAD-16 in LB medium were used. For analytical and preparative HPLC, LB medium was used (20 mL and 12 L, respectively). XAD-16 resin beads were collected with a sieve after 72 h, washed with a small amount of water, and extracted with methanol to yield the crude extract (9.5 g) after evaporation from the large-scale cultivation.

**Feeding experiments:** The strains DSM17909 and DSM17908 were cultivated at 30°C overnight. The cell pellet of the overnight culture was washed once with ISOGRO-<sup>13</sup>C medium (1 mL) and dissolved again in <sup>13</sup>C medium (1 mL). ISOGRO-<sup>13</sup>C medium was prepared with ISOGRO-<sup>13</sup>C powder (1 g), K<sub>2</sub>HPO<sub>4</sub> (1.8 g L<sup>-1</sup>), KH<sub>2</sub>PO<sub>4</sub> (1.4 g L<sup>-1</sup>), and CaCl<sub>2</sub>·xH<sub>2</sub>O (11.1 mg L<sup>-1</sup>) dissolved in H<sub>2</sub>O (100 mL). The feeding culture in <sup>13</sup>C medium (5 mL) was inoculated with washed overnight culture (50  $\mu\text{L}$ ). After incubation for 6 h at 30°C, a stock solution (50  $\mu\text{L}$ , 100 mM) of the respective amino acid ( $\beta$ -alanine, L- $\alpha$ -threonine, L- $\alpha$ -proline, L- $\alpha$ -isoleucine, L- $\alpha$ -leucine, L- $\alpha$ -alanine, or L- $\alpha$ -valine) was added. Another two additions were carried out at 24 and 48 h. The cultures were extracted with ethyl acetate (5 mL) after incubation for 72 h. The extract was then prepared for HPLC-MS analyses.<sup>[14]</sup>

**Hydrolysis of the extract:** The extract was treated with ammonium hydroxide solution (Sigma-Aldrich, 28%) to open the ring in the depsipeptides.<sup>[23]</sup> Hydrolysis was started with methanol extract solution (50  $\mu\text{L}$ ) and ammonium hydroxide solution (500  $\mu\text{L}$ ). After incubation for 2 h at 40°C, the reaction mixture was neutralized with HCl (3M) solution. After evaporation of the solvent, the sample was then prepared for HPLC-MS analyses.

**HPLC and mass spectrometric analysis of the extract:** Analysis of the extracts was carried out by using an Ultimate 3000 LC system from Dionex, coupled to an amaZon X electrospray ionization mass spectrometer from Bruker Daltonics. Peptides were separated on a C18 Column (ACQITY UPLC BEH, 1.7  $\mu\text{m}$ , 2.1  $\times$  50 mm, flow rate 0.6 mL min<sup>-1</sup>, Waters). Acetonitrile/water containing 0.1% HCOOH was used as mobile phase under a linear gradient from 40–50% acetonitrile over 17.5 min for the separation. Collision-induced dissociation (CID) was performed on the ion trap in the amaZon X in positive mode with a scan range of  $m/z$  100–1500.<sup>[17]</sup> HR-MALDI-MS data were obtained with a MALDI LTQ Orbitrap XL from Thermo Fisher Scientific equipped with a 337 nm nitrogen laser. The extract was diluted in acetonitrile for MALDI-MS analysis. 0.2  $\mu\text{L}$  of the extract solution and 0.5  $\mu\text{L}$  of a 20 mM 4-chloro- $\alpha$ -cyanocinnamic acid (CICCA) solution (70% acetonitrile) were spotted on the MALDI target and air-dried. The following instrument parameters were used: laser energy, 25  $\mu\text{J}$ ; automatic gain control, on; auto spectrum filter, off; resolution, 100000; plate motion, survey CPS. Mass spectra were internally calibrated with calibration mixture 2 (Applied Biosystems, Sequazyme peptide mass standard kits). The molecular formulas were calculated by using the Qual Browser software V. 2.0.7 (Thermo Fisher Scientific).<sup>[18]</sup>

**Advanced Marfey's method:** Compound **1** (ca. 0.5 mg) was hydrolyzed with HCl (6M, 0.8 mL) in a high-pressure Ace tube at 110°C for 16 h. The hydrolysate was evaporated to dryness and resuspended in H<sub>2</sub>O (100  $\mu\text{L}$ ). To 50  $\mu\text{L}$  of this solution NaHCO<sub>3</sub> (10  $\mu\text{L}$ , 1M) and *N*<sub>α</sub>-(5-fluoro-2,4-dinitrophenyl)-L-leucinamide or -D-leucinamide (L-FDLA or D-FDLA, 100  $\mu\text{L}$  of a 1% solution in acetone) were added. The reaction vials were covered with aluminum foil and placed in a water bath at 40°C for 1 h. Then the reaction mixtures were cooled to room temperature, quenched with HCl (10  $\mu\text{L}$ , 1M), and evaporated to dryness.<sup>[21,22]</sup> The residue was dissolved in methanol (400  $\mu\text{L}$ ). The analysis of L- and D-FDLA-modified amino acids was carried out by LC-MS, as mentioned above for the analysis of the extract. Acetonitrile/water containing 0.1% HCOOH was used as mobile phase under a linear gradient from 20–60% acetonitrile over 34 min for the separation. At same time, the commercial standard amino acids were prepared as reference.

**Isolation:** For isolation of **1**, the XAD extract (9.5 g) of *X. doucetiae* DSM 17909 was dissolved in 50 mL of DMSO/MeOH/iPrOH (7:2:1). The resulting suspension was centrifuged at 13.3  $\times$  10<sup>3</sup> rpm for 4 min and the pellet was discarded after further analysis. The following preparative HPLC setup was used: 4 mL per injection by using a Gilson 231 XL sample injector with a sandwich injection method, a Gilson 402 syringe pump, a Gilson 215 liquid handler and two Varian Prep Star SD-1 pumps with a flow rate of 95 mL min<sup>-1</sup> and an acetonitrile/water gradient from 0–1 min 5%, 1–16 min 5–95%, 16–19 min 95%, a Gilson 306 pump for the addition of 10% HCOOH buffer with a flow rate of 2.5 mL min<sup>-1</sup>, a Dionex P580 pump for the makeup flow of 1 mL min<sup>-1</sup> acetonitrile, a Jasco UV 975 recording at 215 nm, an MRA Active Splitter, a Varian 380LC ELSD, and a Luna C18 10  $\mu\text{m}$  50  $\times$  50 column from Phenomenex (for the chromatogram, see Figure S2 of the Supporting Information). The obtained fractions were freeze-dried. The fractions containing **1** were combined and purified in a second step on the same instrument setup, but with a flow rate of 75 mL min<sup>-1</sup> and an acetonitrile/water gradient from 0–1 min 45%, 1–16 min 45–65%, 16–19 min 65–95% and a Luna C18 5  $\mu\text{m}$  30  $\times$  75 column. The buffer for the second step was changed to 3.33% trifluoroacetic acid buffer, which was added at a flow rate of 2.5 mL min<sup>-1</sup> (Figure S3 of the Supporting Information). In a third purification step the selectivity of the chromatographic system was changed through the pH value. Therefore, an aqueous solution of ammonium acetate (50 g L<sup>-1</sup>) was used as buffer. A flow rate of 75 mL min<sup>-1</sup> with an acetonitrile/water gradient from 0–1 min 5%, 1–16 min 5–95%, 16–19 min 95% and a Luna C18 5  $\mu\text{m}$  30  $\times$  75 column (Figure S4 of the Supporting Information) were used.

**NMR spectroscopy:** NMR spectra were recorded on a Bruker AVANCE 600 spectrometer operating at a <sup>1</sup>H frequency of 600.2 MHz and a <sup>13</sup>C frequency of 150.9 MHz. The instrument was equipped with a 5 mm TCI probe head. The final experiments were carried out with a sample of **1** (10 mg) dissolved in [D<sub>6</sub>]benzene (600  $\mu\text{L}$ ) at 293 K. For structure eluci-

ation and complete assignment of proton and carbon resonances, 1D <sup>1</sup>H, 1D <sup>13</sup>C, DQF-COSY, ROESY (mixing time 150 ms, spinlock field 2 kHz), multiplicity-edited HSQC, and HMBC spectra were acquired. <sup>1</sup>H chemical shifts were referenced to the solvent signals (<sup>1</sup>H: 7.15 ppm, <sup>13</sup>C: 128.0 ppm).

Two-dimensional homonuclear experiments (DQF-COSY, TOCSY, and ROESY) were performed with a spectral width of 11 ppm. Spectra were recorded with 1024 increments in  $t_1$  and 4096 complex data points in  $t_2$ . For each  $t_1$  value 4 (DQF-COSY), 8 (TOCSY), or 16 (ROESY) transients were averaged.

For the HSQC spectrum 1024 increments with 3072 complex data points in  $t_2$  were collected with sweep widths of 8 ppm in the proton and 90 ppm in the carbon dimension. For each  $t_1$  value four transients were averaged. For the HSQC-TOCSY spectrum 1024 increments with 3072 complex data points in  $t_2$  were collected with a sweep width of 11 ppm in the proton and 80 ppm in the carbon dimension. For each  $t_1$  value 32 transients were averaged, and a mixing time of 80 ms was used for the TOCSY transfer. The HMBC spectrum was acquired with a sweep width of 11 ppm in the proton and 200 ppm in the carbon dimension by using a defocusing delay of 62 ms (optimized for coupling constants of 8 Hz). A total of 32 transients were averaged for each of 1024 increments in  $t_1$ , and 4096 complex points in  $t_2$ .

**Identification and verification of the gene cluster:** The proposed gene cluster responsible for the biosynthesis of xenoamcins from *Xenorhabdus doucetiae* DSM17909 (originally strain FRM16) was analyzed by using PKS/NRPS Analysis Web-site (<http://nrps.igs.umaryland.edu/nrps/>).<sup>[24]</sup> To confirm the gene cluster, gene *xabB* was disrupted by insertion of a plasmid. *Xenorhabdus doucetiae* DSM17909 was naturally resistant to ampicillin, which was used to select *Xenorhabdus* after conjugation. Primer pair 5'-GAACTGGCATGCGGAAATTGAGGCGCAAC and 5'-ACAAGAGAGCTCCCTTCTGAAGCGGTG were used to amplify parts of gene *xabB*, which were then digested and cloned into pDS132 vector and transformed into *E. coli* S17 ( $\lambda$ pir). After conjugation, primer pair 5'-GCCCCGATATCCTGTCCATTG and 5'-ACATGTG-GAATTGTGAGCGG was used to verify the insertion of the plasmid in the mutants. The detailed method was used by us before.<sup>[38]</sup> The extract from the mutant was prepared as for the wild type, except for chloramphenicol in the LB medium for the mutant.

## Acknowledgements

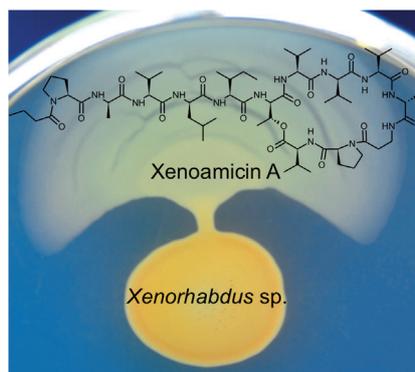
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**Large and alternating:** Xenoamicins, novel tridecadesipeptides with alternating D- and L-amino acid residues, were isolated from *Xenorhabdus* sp (see scheme). Their structures were elucidated by detailed NMR experiments, the advanced Marfey's method, and feeding experiments followed by MS<sup>n</sup> studies. Additionally, identification of the the biosynthesis gene cluster indicated promiscuity of specific domains leading to the various derivatives identified.



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**Natural Products**

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*Q. Zhou, F. Grundmann, M. Kaiser, M. Schiell, S. Gaudriault, A. Batzer, M. Kurz, H. B. Bode\** ..... ■■■■-■■■■

**Structure and Biosynthesis of Xenoamicins from Entomopathogenic *Xenorhabdus*** 