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Advances in 3D imagingand volumetric reconstruction of fluid and melt inclusions by high resolution X-ray computed tomography

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Abstract

Fluid and melt inclusions are tiny pockets of fluid and melt trapped in natural and synthetic minerals. Characterizing the 3D distribution of fluid and melt inclusions within minerals, their shape and the volume fractionof their different phases is crucial for determining the conditions for crystal growth and paleostress analysis. However, their relatively small size (typically 5 to 100 μm), complex shape, heterogeneous content, the opaque nature of some host minerals and projection bias frequently hamper accurate imaging and volumetric reconstruction using conventional microscopic techniques. High resolution X-ray computed tomography (HRXCT) is a non-destructive method which uses contrasts of X-ray attenuation in a series of contiguous radiographs with different view angles to reconstruct the 3D distribution of areas of different densities within a large variety of materials. In this work, we show the capabilities HRXCT for: (i) imaging the 3D distribution of aqueous and hydrocarbon-bearingfluid inclusionsand silicate melt inclusionsin a crystal; (ii) characterizing the shape of fluid and melt inclusions and (iii)reconstructing the total volume and the volume of the different phases (liquid, glass, crystal, vapor) of fluid inclusions and melt inclusions. We have used a varietyof hand specimens and chips of transparent and opaque minerals(olivine, quartz, feldspar, garnet, emerald, wolframite), that we analyzed using three different HRXCT setups. When a resolution of~ 1 μm^3/voxel is achieved, HRXCT allows identifying > 5 μm fluid inclusions, and the identification and volumetric reconstruction of the different phases can be carried out with reasonable confidence for relatively large (> 25 μm) inclusions. Density contrasts are high enough to properly identify: (i) a silicate melt inclusion,and its different phases (glass, vaporand crystalssuch as clinopyroxene and spinel) in an olivine crystal; (ii) aqueous monophase (liquid) and two-phase (liquid + vapor) fluid inclusions in transparent and opaque minerals (quartz, garnet, emerald, wolframite). In the

case of hydrocarbon-bearingfluid inclusions containing twoliquid phases (oil andaqueous solution) and a vapor phase, the two liquid phases could not be distinguished from each other. Volumetric reconstruction of liquid and vapor phases of aqueous and hydrocarbon-bearingfluid inclusionsshow compatible results with independent calculations using known pressure, temperature, molar volume and composition (*P*-*T*-*V*-*x*)conditions of trapping or imaging using confocal laser scanning microscopy respectively. Collectively, our results show that HRXCT is a promising tool for non-destructive characterization of fluidand melt inclusions.

Keywords

Fluid inclusions; melt inclusions; high resolution X-ray computed tomography; volume;shape; phase

1. Introduction

Tiny droplets of fluid and melt are trapped within natural and synthetic minerals as fluid and melt inclusions which represent invaluable toolsto understand crystal growth and mass transfer in the Earth's mantle and crust. The 3D distribution of fluid and melt inclusionswithin crystals, their shapeand the volumetric proportions of their different phases are key information for the determination of the relationship between crystal growth, deformation and the nature of the fluids and melts involved (e.g. Roedder, 1984; Frezzotti, 2001; Bodnar and Student, 2006; Kesler et al., 2013). However, such data aredifficultto acquire with good confidence because optical observation is frequently hampered by perspective and projection issues. For instance, the estimation of the volumetric fraction of the vapor phase to the entire inclusion at room temperature (φ_{vap}) may be incorrect when deduced from area fractions (Bakker and Diamond, 2006). This led to the development of a series of tools and methods allowing significant improvement in textural, microstructural, morphological and volumetric reconstruction of fluid and melt inclusions like for instancethe four-axis universal stage (Tuttle, 1949); relationships between vapor phasediameter, the temperatures of freezing and liquid-vapor homogenization and the volumetric properties of the fluids in the H_2O-NaCl system (Bodnar, 1983), the spindle stage (Anderson and Bodnar, 1993), confocal scanning laser microscopy (Petford et al., 1995),transmission electron microscopy (Viti and Frezzotti, 2001); AnIma (Lespinasse et al., 2005); second harmonic generation microscopy (Stoller et al., 2007)and synchrotron-based X-ray fluorescence computed tomography (Cauzid et al., 2007)and nano X-ray tomography (Yao et al., 2015).

High resolution X-ray computed tomography(HRXCT) is a non-destructive method which uses contrasts of X-ray attenuation in a series of contiguous radiographs with different view angles to reconstruct the 3D distribution of areas of different densities within a large

variety of materials. HRXCT has been increasingly used for characterizing Earth materials over the past two decades (Ketcham and Carlson, 2001;Cnuddle and Boone, 2013). Among various examples, HRXCT has been successfully applied to unravel the shape, orientation and 3D distribution of minerals,porosity and cracksin various rocks showing complex textures like sedimentary rocks (Noiriel, 2015), metamorphic rocks (Sayab et al., 2015), magmatic rocks (Baker et al., 2012), mantle rocks (Créon et al., 2017), ores (Kyle and Ketcham,2015) and extraterrestrial materials (Hanna and Ketcham, 2017).

Since the pioneering study of Nakashima et al. (1997),fluid and melt inclusions have been the subject of only a few studies by HRXCT even with more recent and powerful equipment(Kyle et al., 2004;Kyle and Ketcham, 2007a,b, 2008, 2015; Nowecki, 2014; Pamukcu et al., 2013, 2015; Gaetani et al., 2015; Farley and Horton, 2017).These studies have shown that HRXCT allows imaging isolated fluid and melt inclusions,fluid inclusionclusters and trails, as well as discriminatingand quantifying different phases within fluid inclusionsand silicate melt inclusions in transparent (e.g. quartz, olivine) and opaque (e.g. pyrite, pyrrhotite, sphalerite) minerals.Here, in an attempt to provide an overview of the current capabilities and future perspectives of HRXCT for the study of fluid and melt inclusions,we show the results of a series of experiments using three HRXCT setups at Université de Lorraine (France), University of Strathclyde (UK) and Universidad Nacional Autónoma de México (Mexico). Nine samples containing eitheraqueous fluid inclusions, hydrocarbon-bearingfluid inclusions or silicate melt inclusionsin transparent(natural and synthetic quartz, “trapiche” emerald, gem “demantoid” garnet, olivine) and opaque (wolframite) minerals were analyzed. Both 3D imaging of fluid inclusions distribution within single crystals and quantitative volumetric reconstruction of the different phases within single fluid and melt inclusions were carried out.

2. Sample description

The samples were selected to cover some common types of fluid and melt inclusion compositions found in nature orused as synthetic inclusions, as well as various transparent and opaque host minerals. The targeted inclusions have all in common to be relatively large (>25 µm) to event “giant” (~ 5 mm), and display variable shape and infilling. Their characteristics are provided below and in Table 1.

2.1.Silicate melt inclusions(Sample A)

The volcanic material (Sample A)wascollected at Los Humeros volcanic complex located in the easternpart of the Trans-Mexican Volcanic Belt in Mexico (Table 1). Both the Trans-Mexican Volcanic Beltand Los Humeros volcanic complex exhibit a large diversity of alkaline to calk-alkaline volcanism (Gómez-Tuena et al., 2016; Créon et al., 2018). Sample A was collected in porphyritic lavas from the 8.91 ka post-caldera event (Carrasco-Nuñez et al., 2018; Dávila-Harris and Carrasco-Núñez, 2014). Sample A contains plagioclase, olivine, orthopyroxene and clinopyroxene phenocrysts together with large vesicles and a glass matrix.Olivine phenocrysts are 0.5-2.0 mm large and contain isolated, round silicate melt inclusionsup to 230 µm in diameter, with a vapor bubble (Fig.1). Euhedral crystals of spinel and clinoyroxene (as identified by secondary electron microscopy) are observed inside the silicate melt inclusion (Fig.1). These crystals nucleated after the trapping of the silicate melt inclusion.The analyzed material consists of a single olivine phenocryst (1.0 mm large) separated from Sample A.

2.2. Aqueous fluid inclusions (SamplesB, C, D, E, F)

Sample B consists of water + H_2fluid inclusionstrapped in synthetic quartz grown by the hydrothermal temperature gradient method using refractory autoclaves, produced at theInstitute of Experimental Mineralogy, Russian Academy of Sciences (Chernogolovka, Moscow oblast, Russia)(Table 1). The full experimental procedures for quartz and fluid inclusion synthesis and characterization are described in Balitsky et al. (2011, 2014, 2015). Sample B is a 3×5×2 cm monocrystal showing arrays of elongated and oriented two-phase (liquid + vapor) aqueous fluid inclusions(Fig.2). No sectioning was carried out prior to HRXCT scanning and Sample B was scanned both entirely and on a 1.4×1.4×1.4 mm region of interest.

Sample C consists of synthetic two-phase liquid + vapor aqueous fluid inclusions trapped in a natural quartz, produced at the GeoRessources laboratory, Université de Lorraine (Vandoeuvre-lès-Nancy, France) (Table 1). Fluid inclusion synthesis was conducted following the procedure exposed in Jacquemet et al. (2014), inspired from the one of Sterner and Bodnar (1984). A pre-fractured quartz lamella (5×5×1 mm)wasused to trap the fluid inclusions at 600°C-2.0 kbars. The experimental H_2O-NaCl solution hada salinity of 7.7 wt.% NaCl. Considering these pressure, temperature, molar volume and composition *(P-T-V-x)* conditions for trapping, the expectedφ_{vap} valuewas42% (Bakker, 2003). After fluid inclusion synthesis, the quartz lamella was doubly polished to reach a 0.4 mm thickness for optical microscopy and HRXCT analyses. Examination under optical microscope showedthe presence of numerous two-phase (liquid + vapor) fluid inclusions (1-100 μm) with apparent φ_{vap} value compatible with the theoretical value of 42% (Fig.3).

Sample D was collected from the Antetezambato gem-quality “demantoid” garnet deposit, near Ambanja, in the Ambato peninsula of northern Madagascar (Table 1). This

deposit is classified as a skarn-type deposit related to Upper Mesozoic to Cenozoic magmatism affecting the limestones of the Mesozoic Isalo Group (Giuliani et al., 2015 and references therein). The greenish gem-quality“demantoid” andradite crystal measures 1.5×0.6 cm and hosts a gianttwo-phase (liquid + vapor) aqueous fluid inclusion visible to the naked eye (Fig.4; Giuliani et al., 2015; Morlot et al., 2016). No sectioning was carried out prior to HRXCT scanning and Sample D was scanned entirely.

Sample E was collected from the Muzo depositon the western side of the Eastern Cordillera Basin, Colombia (sample T12 in Pignatelli et al., 2015) (Table 1). The emerald deposits are hosted by Lower Cretaceous sedimentary rocks forming two mineralized zones located on the eastern and western border of the basin, respectively. The emeralds are formed by interaction between hydrothermal H_2O-NaCl-CO_2fluids and black shales. Trapiche emeralds are very rare and are only found in a few depositsfrom the western emerald zone. Their occurrence is related to a complex context of fluid pressure variations along faults and thrusts, affecting the growth of the emerald. On a section perpendicular to the c-axis of the emerald, the trapiche texture is typified by a central core, six arms, which are crystallographically equivalent growth sectors, and dendritesbetween the arms and around the core (Fig.5). The trapiche emeralds host primary multi-phase (liquid + vapor + halite + carbonates ± sylvite) aqueous fluid inclusionsand large secondary monophase (liquid) aqueous fluid inclusionsstudied here, developed parallel to the dislocations in the arms (Fig.5; Pignatelli et al., 2015). No sectioning was carried out prior to HRXCT scanning and Sample E was scanned entirely.

Sample F was collected from the Piaotang W-Sn deposit located in the southern Jiangxi province (SE China) (Table 1).At Piaotang, a buried Jurassic granite shows alteration by hydrothermal fluids in its upper part. The W-Sn mineralsare located within vertical hydrothermal veins that are rooted in the granite and are mostly hosted in the

metasedimentary country rocks (Legros et al., 2018).Sample F is a three centimeter-longopaque wolframite crystal with an average formula of $Fe_{0.5}Mn_{0.5}WO_4$. The wolframite hosts primary (5-40 µm) two-phase liquid + vapor aqueous fluid inclusionseither isolated or aligned and elongated along the c-axis as well as secondary inclusions with similar infilling to primary inclusions (Legros et al., submitted, this volume; Fig.6).Sulfide veinlets cut the wolframite crystal parallel and perpendicular to the c-axis. The analyzed material is a chip (3×4×0.2 mm) of a doubly polished thick section prepared from Sample F.

2.3. Hydrocarbon-bearingfluid inclusions (Samples G, H, I)

Samples G and H consist of hydrocarbon-bearingfluid inclusionstrapped in synthetic quartz, produced at theInstitute of Experimental Mineralogy, Russian Academy of Sciences (Chernogolovka, Moscow oblast, Russia)(Table 1). Quartz crystals with fluid inclusions were grown by the hydrothermal temperature gradient method using refractory autoclaves. The full experimental procedures for quartz and fluid inclusion synthesis and characterization weredescribed in Balitsky et al. (2011, 2014, 2015). Sample G was synthetized using 7.5 wt. % Na_2CO_3 aqueous solution and oil. Sample H was synthetized using 7.0 wt. % Na_2CO_3aqueous solution and boghead. Temperature gradients between the top and bottom of the autoclave were 290/300 °C and 291/347 °C respectively. Sample Gshows an elongated (200 µm) fluid inclusion containing a vaporphase, an aqueous solution and oil as seen under optical microscope using ultraviolet (UV) illumination (Fig.7). Sample H shows an elongated (500 µm) fluid inclusion containing a vapor phase, an aqueous solution, oil and solid bitumen as observed under optical microscope using visible light and UV illumination (Fig.8). The analyzed materialswerechips (5×5×0.5 mm) of doubly polished thick sections prepared from Samples G and H.

Sample I was collected from a petroleum field whose location and geological settings are kept confidential (Table 1). Sample I displays authigenic feldspar cements in which an elongated, 25 µm oil-bearing primary fluid inclusion was found (Fig.9). The inclusion was analyzed by optical microscope using UV illumination showing that the liquid phase contains fluorescentoil. Then the inclusion was analyzed by confocal laser scanning microscopy (Pironon et al., 1998) which allows 3D reconstruction of the inclusion and visualization of the vapor and liquid oil phases through ImageJ 3D Viewer (Fig. 9), with a spatial resolution lower than 0.5µm. By using this method, the calculated φ_{vap} value of the fluid inclusion is 10 ± 2%. The analyzed material wasa chip (3×4×0.15 mm) of a doubly polished thick section prepared from Sample I.

3. Analytical methods

The principles of the method as well as analytical conditions of the three setups used are described below as well as in Table 1.

3.1. Principles of HRXCT analysis and data processing

HRXCT is a non-destructive and non-invasive method which uses contrasts of X-ray attenuation (a function of density and atomic number) to reconstruct the 3D distribution of areas of different densities within a large variety of materials with a resolution down to ~ 1 µm/pixel(px) or 1 μm^3/voxel(vx) (see Ketcham and Carlson, 2001; Cnuddle and Boone, 2013 for more details about the HRXCT data acquisition and processing and application to Earth materials). The 3D reconstruction is based on a series of contiguous 2D radiographs taken with different view angles, by rotating a sample around a single axis in small steps. In the

following experiments, a correction of the ring artifact was done during analysis using specific filters and beam hardening corrections were carried out with the softwares mentioned below.Each phase (mineral, glass, liquid, vapor) wasseparatedmanually using simple greyscale threshold tool, by segmentation of the volumes corresponding to the relative density range of each individual phase. Separation is considered as optimal when the selected range of gray values selects all the voxels belonging to a phase while it does not select voxels that belong to a different phase. Such optimal separation is achieved thanks to a strong contrast between phases (as visible on 2D slicesin Fig. 2,4,7,8) and a reduced noise. Then, the volume of the different phases were calculated by the count of voxelsin the different voxelclusters.

3.2. HRXCT at Universidad Nacional Autónoma de México (Sample A)

Sample A was analyzed through X-ray micro-tomography at the Laboratorio Universitario de Microtomografía de Rayos X (Centro de Geociencias, Universidad Nacional Autónoma de México, Querétaro, Mexico) on the ZEISS Xradia 510 Versa instrument (Table 1). The detector was a CCD ANDOR camera, with 1080×1080 px and 32 bits resolution. Polychromatic tomography with 30 kV radiation was performed on Sample A. The sample-detector source-sample distances were optimized for a spatial resolution of2.06µm/px (8.7 μm^3/vx) with magnifications of 4.0×. The 3D images obtained were composed of 885×885×999 vx (elementary image volume) with 32-bit grey-scale resolution. Acquisition time was 280min for 1601 projections (8 s per projection) together with around 200 reference images (blanks). Tomographic volume reconstructions were performed using the ZEISS XRM software. Data processing was performed using ImageJ® and Avizo 9.2® softwares.

3.3. HRXCT at University of Strathclyde (Sample B)

The HRXCT scans on Sample B were performed at the University of Strathclyde, Glasgow (UK) using an industrial Nikon XTH 320/225 system, equipped with a 225 kV X-ray tube, a microfocus multimetal target, and a 2000×2000 px flat panel photodetector (cell size 0.2×0.2 mm), leading to volumes of 2000×2000×2000 vx (Table 1). Two different scan conditions were tested on Sample B.One scan of the complete sample (Fig. 2) was performed using a silver target under an accelerating voltage of 160 kV, 71 µA current corresponding to a power of 11.4W, for a spatial resolution of 25µm/px(15625 μm^3/vx). The exposure time for each of the 3141 projections was 0.708 s, leading to a scan time of 37 min (1 frame per projection). The second scan focusing on a 1.4×1.4×1.4 mm region of interest of the sample was performed using a silver target under an accelerating voltage of 160 kV, 46 µA current, corresponding to a power of 7.4W, for a spatial resolution of7.7 µm/px(456 μm^3/vx). The exposure time for each of the 3141 projections was 1.415 s, leading to a scan time of 75 min (1 frame per projection). X-ray tube conditions would not saturate photodetector, consequently no metallic filter was required during the scans. Volume reconstructions were carried out with CT Pro 3D software. Both 3D dataset were treated using Avizo® 9.2.0, noise was corrected by using a median filter that uses morphological operators to set the voxelsvalue to the median for the defined neighborhood (3 px 2D square). After separation of phases, the remaining selected voxelssmaller than 100 connected voxels in 3D were removed from the selection to ensure no noise was left.

3.4. HRXCT at Université de Lorraine (Samples C-I)

HRXCT scans of Samples C to I were acquired at the Université de Lorraine, Vandoeuvre-lès-Nancy (France) with a Phoenix Nanotom S scanner (Table 1).Data processing was carried out using Avizo® 9.2.0. The detector used was a CCD Hamamatsu, with 2300×2300 px.For Samples C, D, E, F, G, H and I, the source-sample distance was 12.5, 55, 28, 6, 6, 10 and 10 mm respectively. The accelerating voltage was 100, 115, 100, 90, 90, 110 and 110 kV respectively. The current was 100, 100, 65, 100, 115, 115 and 115 µA respectively. The following number of projections were carried out: 1500 (as averages of 4 projections of 1000 ms each), 2000 (as averages of 3 projections of 1000 ms each), 2000 (as averages of 5 projections of 750 ms each), 2000 (as averages of 3 projections of 1250 ms each), 1500 (as averages of 6 projections of 750 ms each), 1200 (as averages of 6 projections of 750 ms each) and 1200 (as averages of 6 projections of 750 ms each) respectively. This led to a scan time of 100, 100, 125, 125, 125, 105and 105min. respectively. The spatial resolution was 1.25 µm/px (1.95 μm^3/vx), 1.96 µm/px (7.5μm^3/vx), 3.5 µm/px (42 μm^3/vx), 0.77 µm/px (0.45 μm^3/vx), 0.77 µm/px (0.45 μm^3/vx), 1 µm/px (1 μm^3/vx) and 1 µm/px (1 μm^3/vx) respectively.

4. Results and discussion

4.1. Identification and 3D distribution of fluid and melt inclusions

For Sample A, the density of the various solid components of the silicate melt inclusion(glass: d ~ 2.7 $g.cm^{-3}$; clinopyroxene: d = 3.3 $g.cm^{-3}$; spinel: d = 3.6 $g.cm^{-3}$) contrasts sufficiently with that of the host olivine to allow to proper segmentation of the whole silicate

melt inclusion (Fig. 1).Similarly, for Samples B to I, density differences between the liquid phase of aqueous and/or hydrocarbon-bearingfluid inclusions(d ≈ 1g.cm^{-3} and 0.7-1g.cm^{-3} respectively), and their host minerals (quartz: d = 2.6 g.cm^{-3}; garnet: d = 3.5-4.3 g.cm^{-3}; emerald: d = 2.7-2.9 g.cm^{-3}; wolframite: d = 7.1-7.5 g.cm^{-3} and feldspar = 2.6 g.cm^{-3}) are high enough to induce detectable contrasts of X-ray attenuation, as visible on 2D slices (Fig. 2-9).HRXCT 3D reconstructions allow to identify isolated inclusions (Fig. 1,4,5,7-9, Samples A,D,E,G-I) and 2D and 3D fluid inclusionclusters (Fig. 2,3,6, Samples B,C,F). Based on the analysis of relatively small fluid inclusions and when a resolution of ~ 1 µm^3/vx is achieved (Fig. 3,6, Samples C,F) the minimum size of the fluid inclusions that can be clearly identified using HRXCT is about 5 µm in diameter (assuming spherical shape). In Sample E,HRXCT 3D reconstruction shows that monophase liquid aqueous fluid inclusionsare developed parallel to the dislocations in one of the arms of the trapiche emerald (Fig. 5). In Sample F, HRXCT 2D and 3D reconstructions show thatmost fluid inclusions are elongated perpendicular to the main sulfide veinlet and parallel to the c-axis.

Here, fluid inclusions and silicate melt inclusionswere successfully imaged by HRXCT in both transparent (olivine, natural and synthetic quartz, demantoid garnet, “trapiche” emerald, feldspar) and opaque (wolframite) minerals.The present results complement previous studies in which fluidand melt inclusionswere imaged by HRXCT in quartz, olivine, sphalerite, pyrite and pyrrhotite (Nakashima et al., 1997; Kyle et al., 2004; Kyle and Ketcham, 2007a,b, 2008, 2015; Nowecki, 2014; Pamukcu et al., 2013, 2015; Gaetani et al., 2015; Farley and Horton, 2017). Common geologic fluids differ in linear X-ray attenuation coefficient from common minerals by several orders of magnitudes, indicating that HRXCT is theoretically suitable for identification of fluid inclusionsin most natural and synthetic crystals (Kyle and Ketcham, 2015).HRXCT is therefore a powerful technique for non-destructive reconnaissance study of fluid and melt inclusion study in many types of

minerals, including: (i) opaque minerals for which observation of inclusions needs access to an infrared camera and sometimes are not even for suited for infrared cameras (Lüders, 2017); (ii) fragile minerals (e.g. carbonate, gypsum) subjected to fluid inclusionstretching during sample preparation (Bourdet and Pironon, 2008) and (iii) gemstones for which the presence of fluid inclusions impacts their commercial value (Giuliani et al., 2014). HRXCT reconstruction could be a prerequisite before any sectioning of the sample or destructive analysis, for example before laser ablation - inductively coupled plasma - mass spectrometry analysis (to locate fluid inclusions to be ablated in opaque minerals;Kouzmanov and Pettke, 2010; Colin et al., 2013), or crushing (to evaluate the volume of fluid that can be extracted for further elemental of isotopic analysis; Nowecki, 2014). HRXCT reconstruction could help defining the relationships between the inclusions and their host crystals and their primary or secondary originif the orientation of the crystal is known (Roedder, 1984). Finally,3D mapping of fluid inclusionplanes by HRXCT could be of great help for microtectonic characterization and paleostress analysis (Nakashima et al., 1997; Lespinasse et al., 2005).

4.2. Identification of the different phases within fluid and melt inclusions

For Sample A,the respective densities of the different phases within the silicate melt inclusion(vapor bubble, glass, clinopyroxene and spinel) are sufficiently contrasted to allow proper thresholding and labelling, and to reconstruct the complex distribution of the vaporbubble and the solidswithin the glass matrix (Fig. 1). The multiple clinopyroxene and spinel crystals are mostly located close to the walls of the inclusion or close to the vaporbubble, as also seen by secondary electron microscopy. Mineral crystallization at the vaporbubble rim implies a fluid saturation event and discards, by the same, a possible free volume vaporbubble (known to be due to the melt-glass transition).Within two-phase aqueous

and hydrocarbon-bearingfluid inclusions, the density contrast between the vapor phase ($d \approx 10^{-2}$-10^{-1} g.cm^{-3}) and the liquid phase(s) ($d \approx 1$ g.cm^{-3}) is high enough to allow their proper identification and reconstruction (Fig. 2-4,6-9, Samples B-D,F-I). For Samples G and H, the two liquid phases (oil and aqueous solution) and the solid bitumen are indistinguishable, probably due to their similar density(Fig. 7-8). For Samples C and F, no vapor phase could be identified in relatively small inclusions (<10 µm in diameter assuming spherical shape, Fig. 3,6).The minimum diameter of the vapor phase allowing to firmly discriminate the liquid phase from the vapor phase is about 2 µm.

Here, multiphase silicate melt inclusionsand one phase (liquid) and two-phase (liquid + vapor) aqueous and hydrocarbon-bearingfluid inclusionswere successfully imaged.Therefore 3D HRXCT imaging can not only identify the presence and location of fluidand melt inclusionsbut also can provide information about the number, the composition, the shape andthe spatial distribution of their different phases. This has direct implications for the nature of the parental melt and trapping conditions of crystallized silicate melt inclusions(e.g., homogeneous or heterogeneous) and on the origin of the vaporbubble (free-volume or fluid saturation) as well as for the trapping conditions and the nature of the fluids trapped in fluid inclusions.As a complement to previous studies (Nakashima et al., 1997; Kyle et al., 2004; Kyle and Ketcham, 2007a,b, 2008, 2015; Nowecki, 2014; Pamukcu et al., 2013, 2015; Gaetani et al., 2015; Farley and Horton, 2017), this is the first time that liquid hydrocarbon is identified in fluid inclusionsand that different solids are characterized within a melt inclusionby HRXCT. Further investigation could include the identification of daughter minerals within aqueous fluid inclusionsand the characterization of gas-bearing inclusions like monophase vapor CH_4-N_2 inclusions and three-phase (aqueous solution, liquid CO_2 and vapor CO_2) aquocarbonic inclusions. Further improvements in phase discrimination and

identificationcould include more systematic use of propagation phase-contrast tomography (Pamukcu et al., 2013; Sayab et al., 2016).

4.3. Shape and volumetric reconstruction of fluid and melt inclusions

HRXCT reconstruction of the analyzed inclusions revealsa broad range in shape (nearly spherical, elongated, tubular or complex; Fig. 1-9, Samples A-I). For Sample D, the morphology of the fluid inclusioncavity is sufficiently well reconstructed that it is possible to identify a negative crystal shape, with polygonal surfaces parallel to the host crystal faces(Fig. 4). Besides polygonal faces, the wall of the fluid inclusion shows a rough surface.In some cases, the vapor phase of both aqueous and hydrocarbon-bearingfluid inclusions (Samples B,D,G,I, Fig. 2,4,7,9) and the vapor bubble of the silicate melt inclusion(Sample A, Fig. 1) are located in such places that they are slightly flattened at the contact of the inclusion wall. This configuration usually hampers accurate optical estimation of the volume of the vapor phase. As for secondary electron microscopy imaging, HRXCT 3D reconstruction of the silicate melt inclusionshows that the multiple clinopyroxene and spinel crystals are euhedral (Fig. 1, Sample A).

The 3D reconstruction also allows the calculation of the relative volume of the different phases. For Sample A (Fig. 1) the calculations return the following volume percentages for the different phases within the silicate melt inclusion: glass (67.5 vol.%), clinopyroxene (20.4 vol. %, as the sum of all clinopyroxene crystals), vaporbubble (7.2 vol.%), spinel (4.9 vol.%, as the sum of all spinel crystals). For Sample D (Fig. 4), the demantoid garnet has a calculated total volume of 896.1 mm^3. The fluid inclusionrepresents 6.2 vol.% of the total volume of the garnet. The liquid phase has a calculated volume of 52.16 mm^3 and the vapor phase has a calculated volume of 3.52 mm^3. This corresponds to a φ_{vap}

value of 6.3%.For Sample G (Fig. 7), the calculated φ_{vap} value is 9.2%.For Sample B (Fig. 2) preliminary tests of the sensitivity of thecalculated phase volume to threshold values and to scanning resolution were carried out (Table 2). The volumetric calculations using a scanning resolution of 25 µm/px (15625 μm^3/vx) and optimal threshold values return a φ_{vap} value ranging between 13% and 24% (average 20.1%) for the population of imaged inclusions (Table 2). When scanned with a resolution of 7.7 µm/px (456 μm^3/vx) and using optimal threshold values the calculated φ_{vap} valueis of 24% for fluid inclusion#3 while the 25 µm/px (15625 μm^3/vx) resolution scan returns a φ_{vap} value of 22.5%. At a resolution of 7.7 µm/px (456 μm^3/vx), when the threshold value is changed by 5%, the calculations return a φ_{vap} value of 76.2% (Table 2). Therefore, it appears that the volumetric reconstruction is sensitive to the scanning resolution (the higher the better) and most importantly to the choice of the threshold values during the post-processing stage. For Sample C, the calculated φ_{vap} value of relatively large inclusions (>25 µm in diameter assuming spherical shape)is between 40 and 45%, which is compatible with the expectedvalue of 42% (considering known *P-T-V-x* conditions of trapping).For Sample I, 3D reconstruction of the hydrocarbon-bearingfluid inclusionusing confocal laser scanning microscopy and visualization through ImageJ 3D Viewer returns aφ_{vap} value of 10 ± 2%, while HRXCT 3D reconstruction of the sameinclusion returns a compatibleφ_{vap} valueof 9.8%.For samples B, C, F, H and I, the reconstructed surfaces ofvapor phases are slightly more complex than spherical or flattened shapes observed under optical microscope (Fig. 2,3,6,8,9). Those differences arise from the thresholding and segmentation procedures and would induce more discrepancies between actual and reconstructed shapes and volumes when the resolution and density contrasts are the lowest. However, based on independent analysis of Samples C and I, this does not seem to introduce significant bias on volume estimation.

The present results show that HRXCT reconstruction allows 3D visualization of the fluid and melt inclusions shape which may help correcting for projection bias under the optical microscope (Bakker and Diamond, 2006) and better identify post-trapping deformation (i.e. stretching, necking down). 3D reconstruction may also overcome the frequent limitation in estimation of vapor phase volume when the vaporbubble is flattened or when the shape of the inclusion is complex.Finally, 3D volumetric reconstruction of melt inclusions by HRXCT could be used as a promising alternative approach to post-entrapment modification correction and calculation of the parental melt composition without inducing leakage of volatiles by heating and related diffusion (Créon et al., 2018). Further improvement would consist of more in-depth sensitivity tests on various parameters andquantifying the uncertainties in volume calculation (Lin et al. 2015). Also, one could test the volume changes in the fluid inclusion cavities induced by freezing of the liquid phase in the case of inclusions hosted in fragile minerals like carbonates. This tests would require adaptation of a microthermometric stage in the HRXCT setup. It could be also interesting to test if sample heating due to X-ray absorption could induce measurable change in the volume of the vapor phase. Here, the present observations show no visible variation of the vapor phase diameter between radiographs taken before and during the experiments.Similarly, the vapor phase has to remain immobile during the experiments. Consequently and considering that the vapor phase tends to move in the liquid phase in many types of fluid inclusions, the applicability of the HRXCT is inherently limited to fluid inclusions with relatively large vapor phase, or trapped morphologically.

5. Conclusions

Collectively, the present results complement significantly the sparse previous studies of HRXCT characterization of fluid and melt inclusions. HRXCT can definitely be a useful and versatile tool for non-destructive characterization of the 3D distribution of fluid and melt inclusions, their shape and the volume fractions of their different phases, within both transparent and opaque minerals. Among other applications, HRXCT is a promising powerful method for fluid and melt inclusion petrography, paleostress analysis and *P-T-V-x* reconstruction.Future developments should focus on: (i) achieving higher spatial resolution and more efficient phase discrimination with more sophisticated data processing and experimental setups and (ii) more quantitative approaches to fluid and melt inclusion distribution and volumetric reconstruction including better estimation of accuracy and precision of the calculated volumes.

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Table captions

Table. 1: Summary of key parameters for the experiments. “Phases” refer to the different phases identified at room temperature within the inclusions. MI: melt inclusion; FI: fluid inclusion, IEM Institute of Experimental Mineralogy, Cpx: clinopyroxene; Spl: spinel; Vap.: vapor; Aq. sln.: solution; sb.: solid bitumen; px: pixel; vx: voxel. * corresponds to the high resolution scanning in the region of interest shown in Fig. 2 (fluid inclusion#3).

Table 2: Preliminary tests of the sensitivity of thecalculated phase volume to threshold values and to scanning resolution, carried out on fluid inclusions (FI) from Sample B.The fluid inclusions are numbered according to Fig. 2. * corresponds to calculated volumes with optimal thresholding value. **corresponds to calculatedvolumes by varying the threshold values by 5% during post-processing stage. The scanning resolution for all fluid inclusions except *** is 25µm/px(15625 μm^3/vx). *** corresponds to calculatedvolumesusing a scan with a resolution of7.7 µm/px (456 μm^3/vx). FI_{tot}: total fluid inclusion volume; Vap.: vapor phase; Liq.: liquid phase; φ_{vap}: volumetric fraction of the vapor phase to the entire inclusion; Δ: relative difference between measurements with different threshold values; px: pixel; vx: voxel.

Figure captions

Fig.1:Silicate melt inclusion (SMI)in olivine (Sample A). (A) An olivine phenocryst (1.0 mm large) mounted with epoxy (carried out after HRXCT analysis for further microprobe analyses). The olivine phenocryst hosts a SMI outlined in a red dashed circle. (B)

Secondary electron microscopyimagingof the SMI showing a glass phase, a vapor bubble and several euhedral crystals of spinel (Spl) and clinopyroxene (Cpx) identified by secondary electron microscopy analysis. (C) HRXCT 3D reconstruction of the SMI with four different view angles showing the 3D distribution of the different phases within the SMI.

Fig.2:Aqueousfluid inclusions in synthetic quartz (Sample B). (A) Photograph of Sample B showing an array of elongated and oriented two-phase (liquid + vapor) aqueousfluid inclusions. Inclusions being a bit out of focus are framed by rectangles.Fluid inclusions are numbered for correspondence with (B). Note that inclusion#5 is completely out of focus and is therefore not indicated. The red asterisk indicates the same inclusion(#3) as in (B) to (F).(B) HRXCT 3D reconstruction of an array oftwo-phasefluid inclusions.Surfaces of the liquidphase (blue) and vaporphase (green) are represented in a 3D view with perspective effect (vx size is 15625 μm^3). Surfaces were produced for each object as triangular mesh (smoothing extent of 3 px for vapor phases and 1 for FI walls). (C) 2D slice-view of the fluid inclusion indicated by the red asterisk in (A) and (B) showing the liquid phase (Liq.) and the vapor phase (Vap.)after image filter application. Grey values refer to relative density, brighter being denser. From (C) to (F) the spatial resolution is7.8 µm/px(456 μm^3/vx).(D) Equivalent 2D-view after use of threshold and labelling.The darker pixelsbelonging to the vaporphase are gathered as green material, brighter black pixels belonging to liquidphase are gathered as blue material, while brighter grey pixels belonging to quartz host are removed from the dataset. (E) Equivalent surface view in 3D with no perspective enabled.(F) Surface view of the vapor phase.

Fig.3:Synthetic aqueous fluid inclusionstrapped in a natural quartz (Qtz) (Sample C). (A)A typical two-phase liquid (Liq.) + vapor (Vap.) fluid inclusion with apparent φ_{vap} value compatible with the expectedvalue of 42% (see section 2.2. for details). (B) HRXCT 3D reconstruction of a cluster of aqueous fluid inclusions. Pixelsbelonging to the vapor phase are gathered as green material and pixelsbelonging to liquid phase are gathered as blue material.

Fig.4: Aqueous fluid inclusion in “demantoid” garnet (Sample D). (A) A demantoid garnet (Grt) crystal with a giant two-phase liquid (Liq.) + vapor (Vap.) aqueous fluid inclusion (photo by Michel Cathelineau). (B) HRXCT 2D slice-view of the aqueous fluid inclusion and its host crystal after image filter application.Grey values refer to relative density, brighter being denser. (C) 3D reconstruction of the aqueous fluid inclusion and its host crystal showing the morphology of the inclusioncavity. The inner wall of the inclusion is shown in light grey, the outer wall of the inclusion is shown in light blue and the vapor phase is shown in dark blue.

Fig.5: Aqueous fluid inclusions in “trapiche” emerald (Sample E). (A) Hand specimen. The trapiche texture in this sample appears as formed by a central, small hexagonal core, six arms of intense green color, and dark dendritesbetween the arms and around the core. (B) Flat monophase liquid (Liq.)aqueous fluid inclusion. (C, D, E) HRXCT 2D and crossed 2D slice-views of the monophase fluid inclusions and their host crystal.

Fig.6:Aqueous fluid inclusions in wolframite (Wf)(Sample F). (A) Typical two-phase liquid (Liq.) + vapor (Vap.) aqueous fluid inclusionsobserved under optical microscope using an infrared camera. (B) HRXCT 2D slice showing the distribution of fluid inclusions

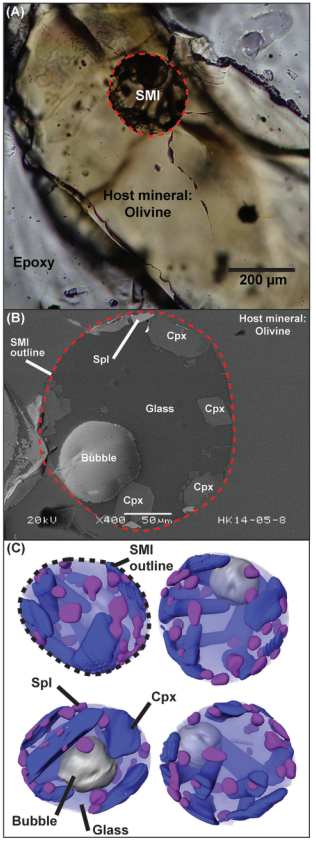
(in blue and yellow) in a selected area of Sample F. (C) 3D reconstruction of fluid inclusions from the area selected in (B). After use of threshold and labelling, the darker pixelsbelonging to the vaporphase are gathered as yellow material, brighter black pixels belonging to liquid phase are gathered as blue material.

Fig.7:Hydrocarbon-bearingfluid inclusion in synthetic quartz (Qtz) (Sample G). (A) The hydrocarbon-bearingfluid inclusion as seen under optical microscope using UV illumination showing that the inclusion contains a vaporphase (Vap.), an aqueous solution (Liq. 1) and oil showing UV fluorescence(Liq. 2). (B) HRXCT 2D slice-view of the hydrocarbon-bearingfluid inclusion after image filter application.Grey values refer to relative density, brighter being denser. (C) Equivalent 2D-view after use of threshold and labelling.The darker pixelsbelonging to the vaporphase are gathered as green material, brighter black pixels belonging to the aqueous solution + oil phases are gathered as blue material. (D) 3D reconstruction of the inclusion with three different view angles. Pixelsbelonging to quartz host are removed from the dataset.

Fig.8:Hydrocarbon-bearingfluid inclusionin synthetic quartz (Qtz) (Sample H). (A) The hydrocarbon-bearingfluid inclusion as seen under optical microscope (natural light) showing that the inclusion contains a vaporphase (Vap.), an aqueous solution (Liq. 1), oil (Liq. 2) and solid bitumen (Sb.). (B) The hydrocarbon-bearingfluid inclusion as seen under optical microscope using UV illumination, where hydrocarbon-bearing phases (Liq. 2 and Sb.) show UV fluorescence. (C) HRXCT 2D slice-view of the hydrocarbon-bearingfluid inclusion after image filter application.Grey values refer to relative density, brighter being denser. (D) Equivalent 2D-view after use of threshold and labelling.The darker pixelsbelonging to the vaporphase are gathered as

greenmaterial, brighter black pixels belonging to aqueous solution + oil phase are gathered as blue material.

Fig.9: Hydrocarbon-bearingfluid inclusionin feldspar (Fsp) (Sample I). (A) The hydrocarbon-bearingfluid inclusion as seen under optical microscope (natural light). (B) The hydrocarbon-bearingfluid inclusion as seen under optical microscope using UV illumination, where the oil phase (Liq.) shows UV fluorescence. (C) 3D reconstruction of the hydrocarbon-bearingfluid inclusion using confocal laser scanning microscopy and visualization through ImageJ 3D Viewer. (D) HRXCT 3D reconstruction of the hydrocarbon-bearingfluid inclusion.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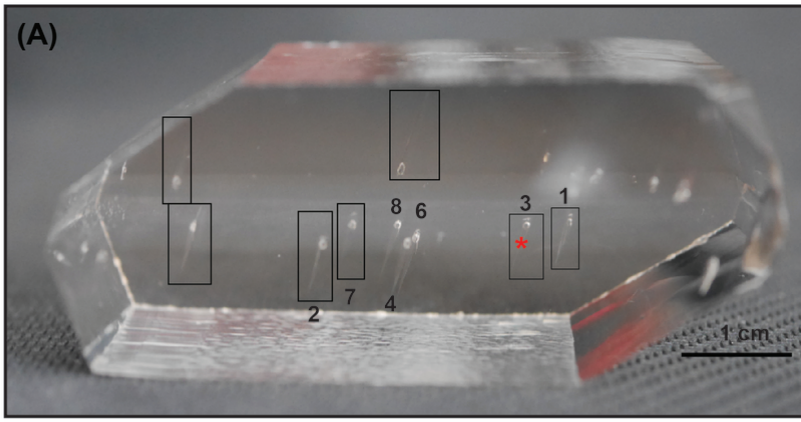
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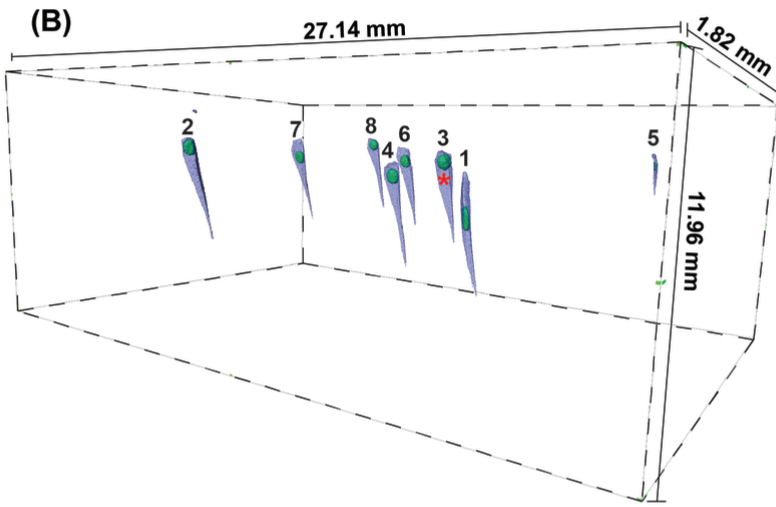
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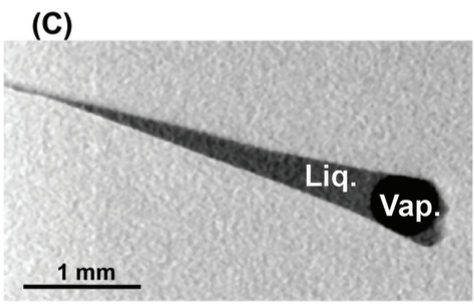
(C)
Liq.
Vap.
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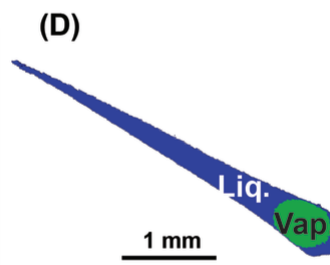
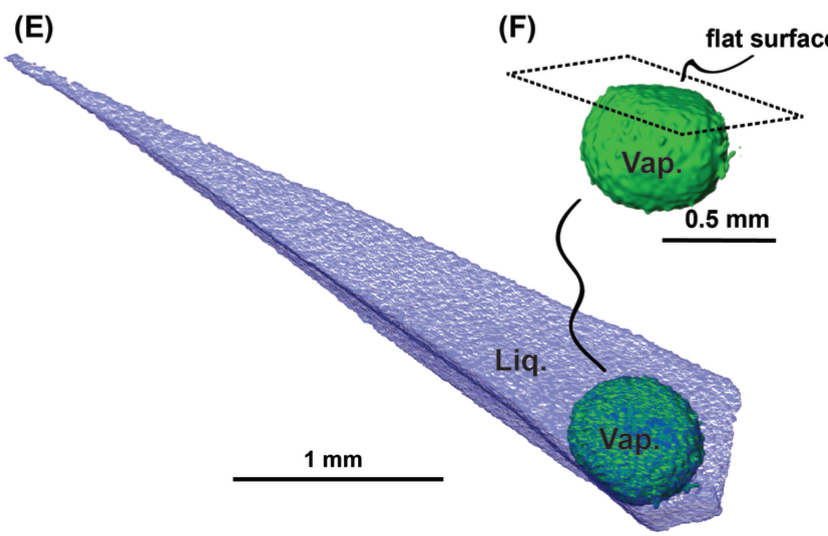
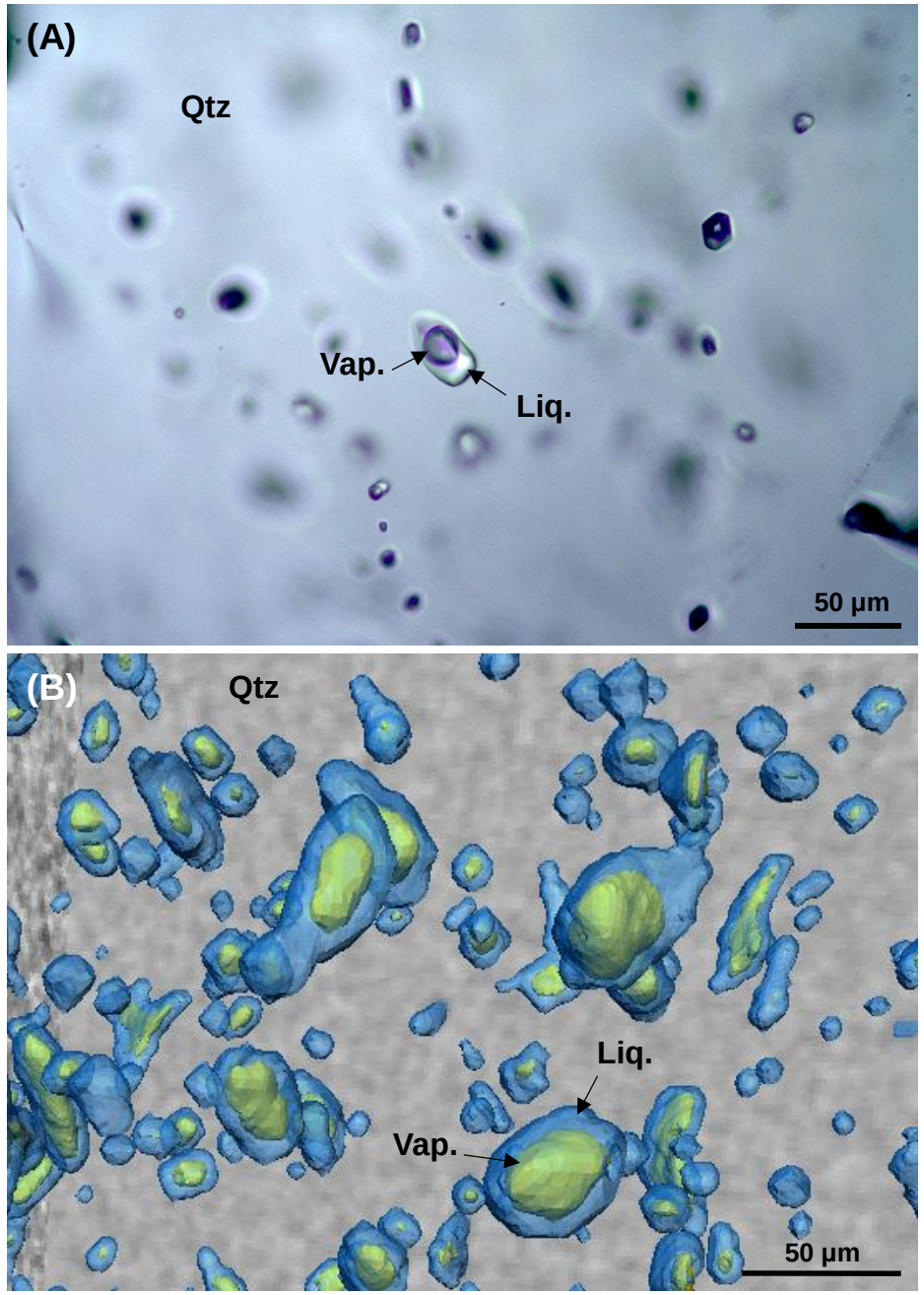
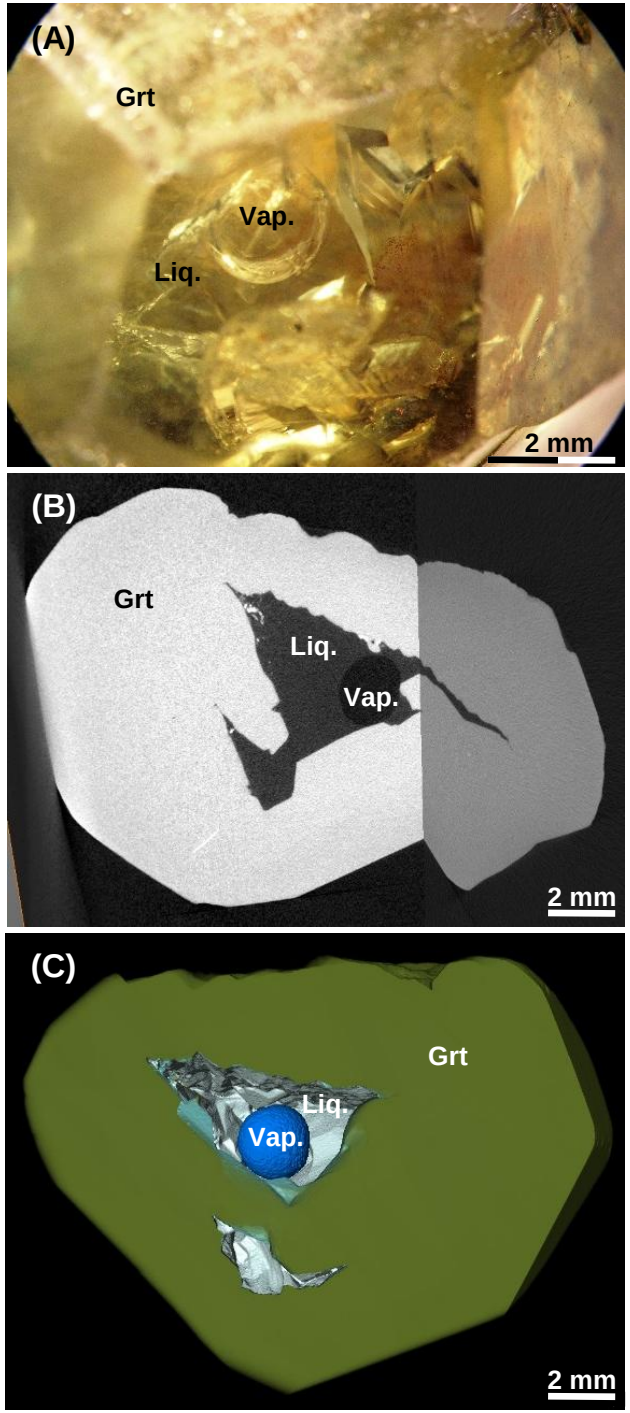

(D)
Liq.
Vap.
1 mm

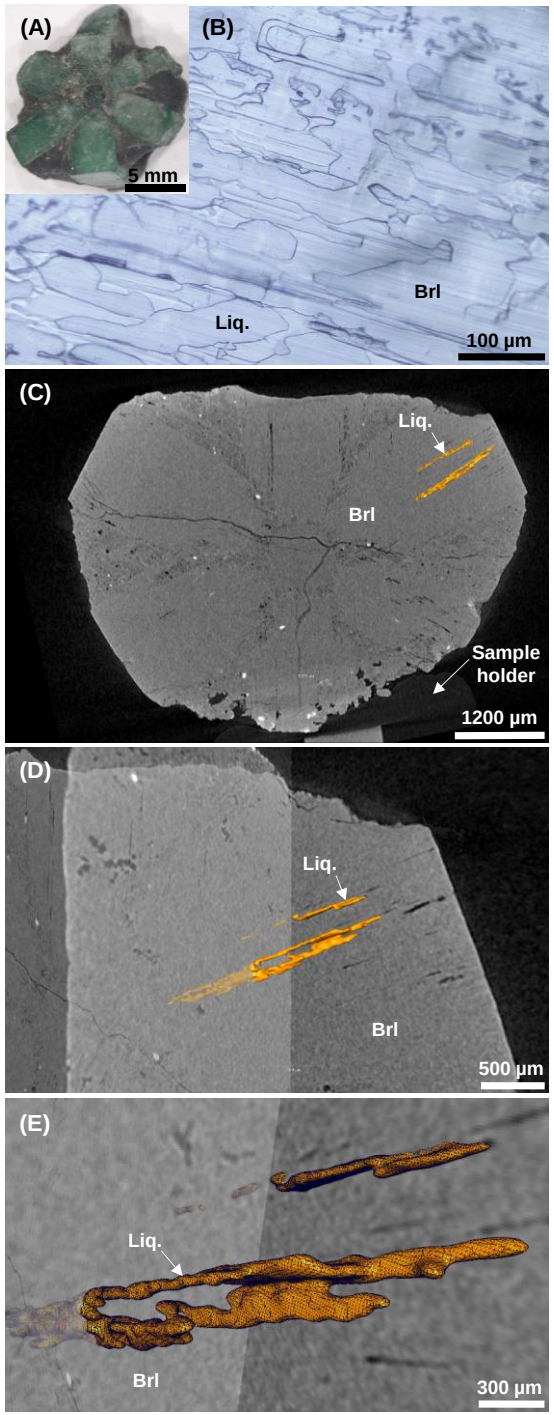

(E)
Liq.
Vap.
1 mm
(F)
flat surface
Vap.
0.5 mm



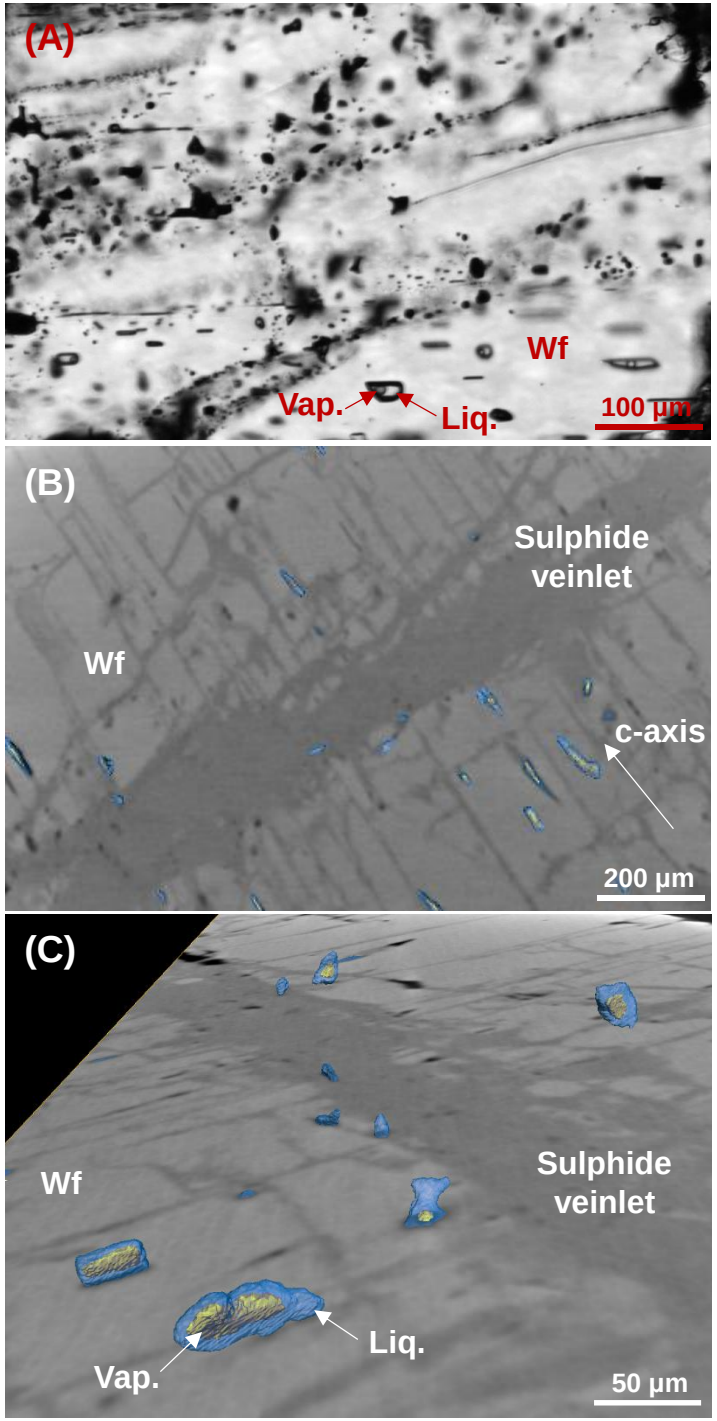
Richard et al. Fig. 3 one column



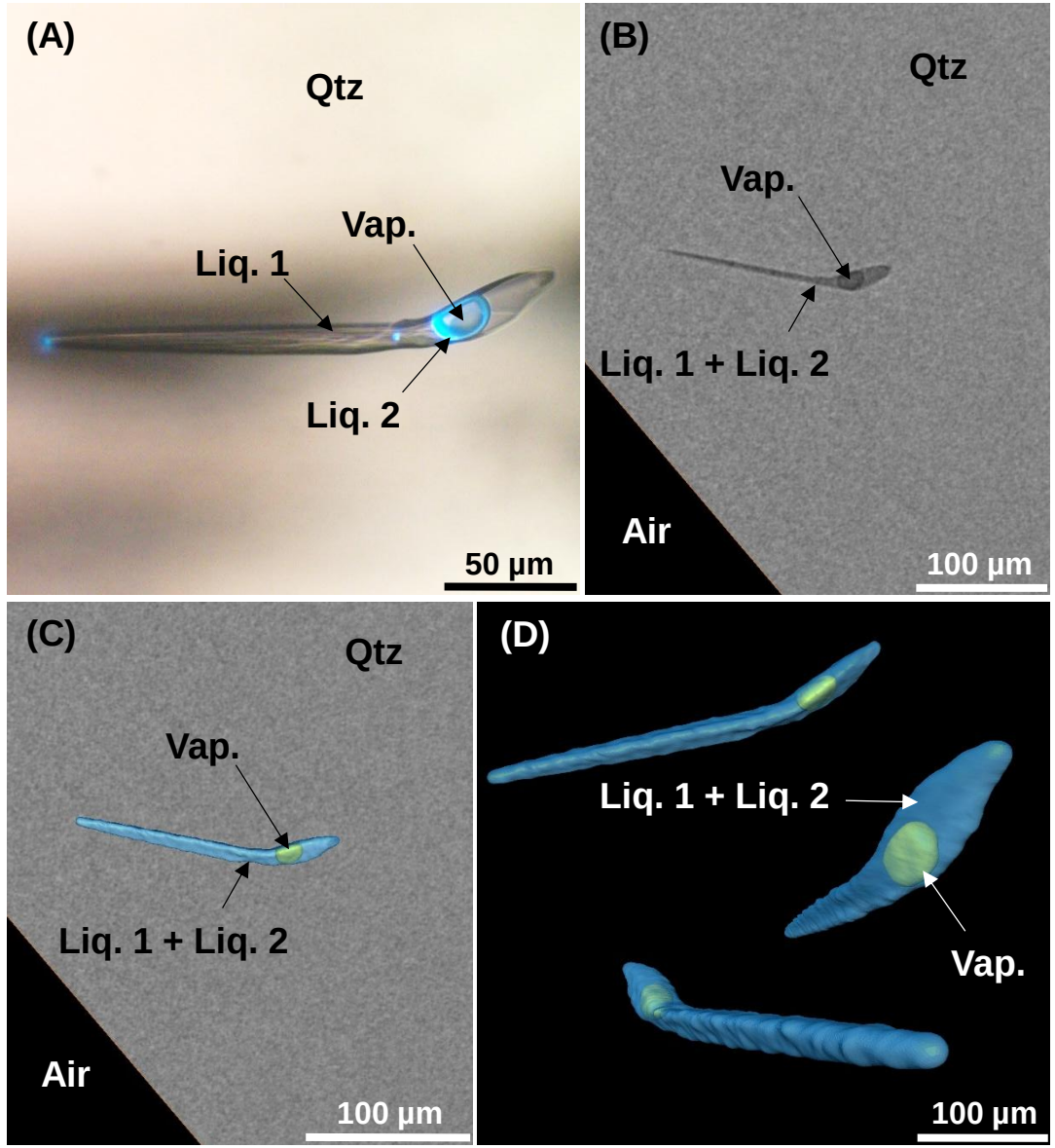
Richard et al. Fig. 4 one column



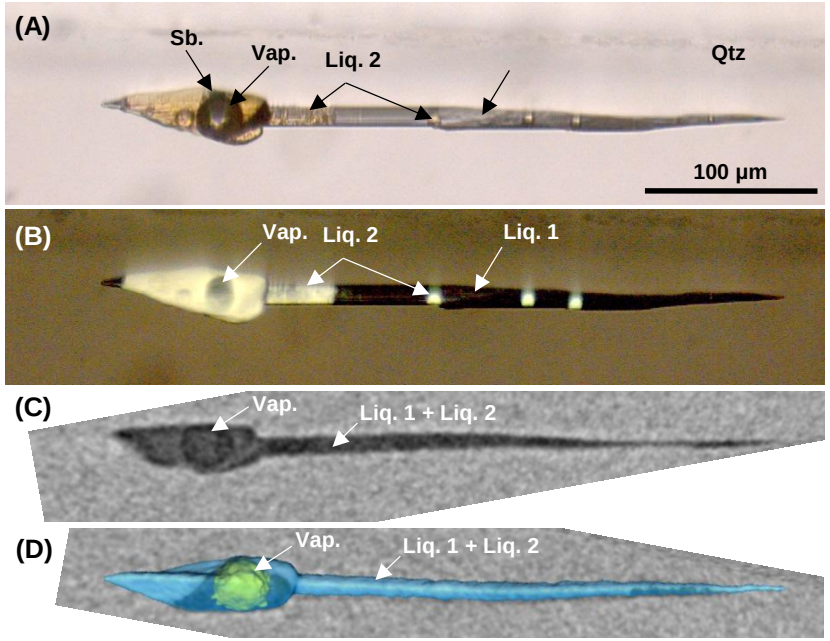
Richard et al. Fig. 5 one column



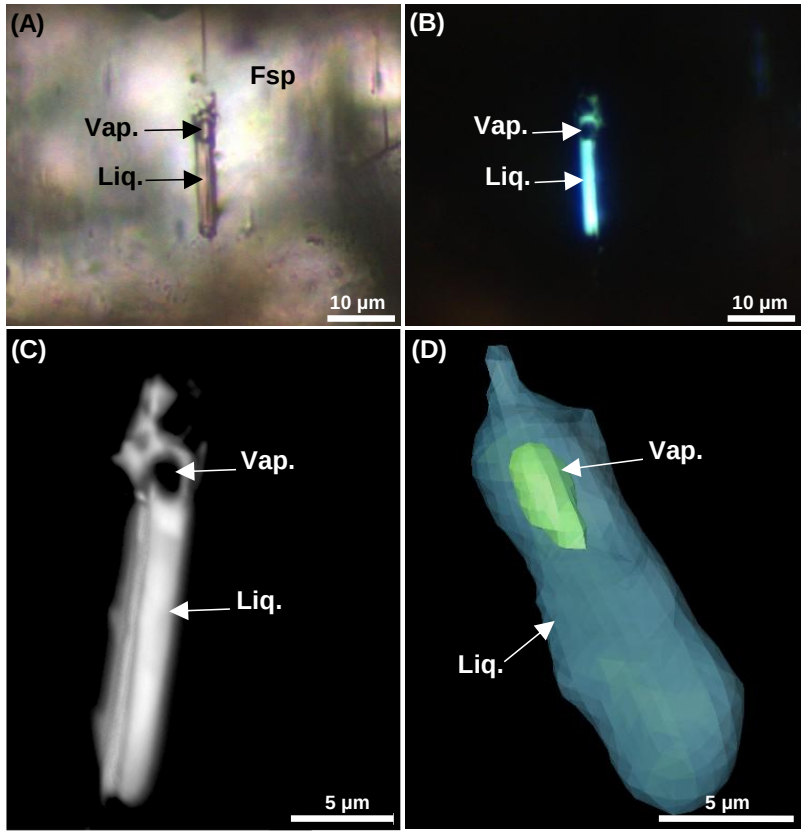
Richard et al. Fig. 6 one column



Richard et al. Fig. 7 one column



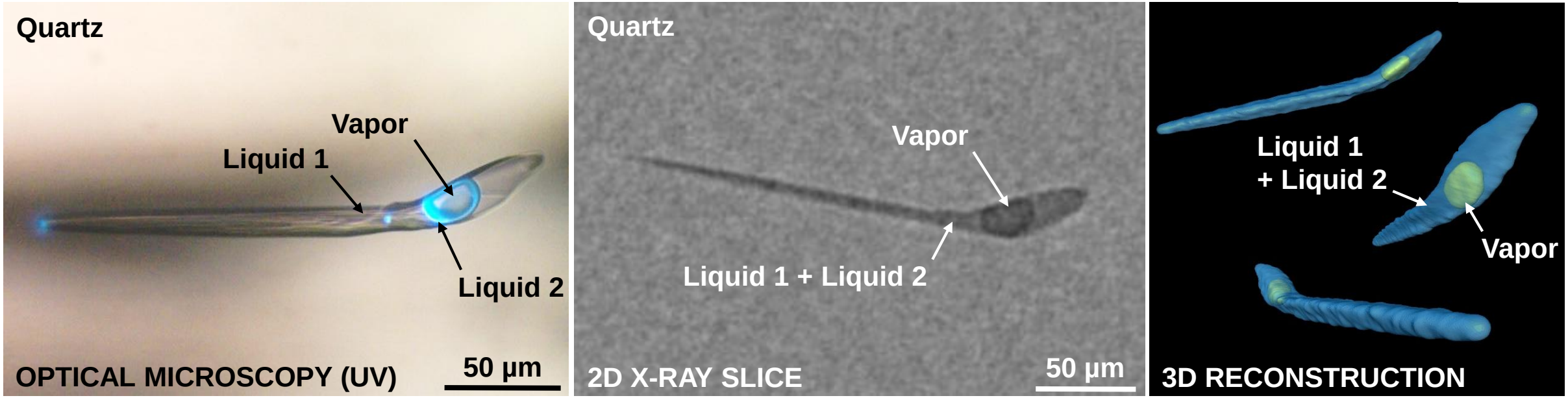
Richard et al. Fig. 8 one column



Richard et al. Fig. 9 one column

Setup	Univ. Nacional Autónoma de México (Mexico) ZEISS Xradia 510 Versa	Univ. of Strathclyde (UK) Nikon XTH 320/225		Univ. de Lorraine (France) Phoenix Nanotom S scanner						
Sample #	A	B		C	D	E	F	G	H	I
Provenance	Los Humeros volcanic complex (Mexico)	Synthetized at IEM (Russia)		Synthetized at Univ. de Lorraine (France)	Antetezambato garnet deposit (Madasascar)	Muzo emerald deposit (Colombia)	Piaotang W-Sn deposit (China)	Synthetized at IEM (Russia)	Synthetized at IEM (Russia)	Kept confidential
Mineral composition	Olivine	Synthetic quartz		Quartz	Gem "demantoid" garnet	Trapiche emerald	Wolframite	Synthetic quartz	Synthetic quartz	Feldspar
Sample size	~1 mm	3×5×2 cm		5×5×0.4 mm	1.5×0.6 cm	~11×11×7 mm	3×4×0.2 mm	5×5×0.5 mm	5×5×0.5 mm	3×4×0.15 mm
Inclusion type	Silicate MI	Aqueous FI		Aqueous FI	Aqueous FI	Aqueous FI	Aqueous FI	Hydrocarbon-bearing FI	Hydrocarbon-bearing FI	Hydrocarbon-bearing FI
Inclusion size	~225 µm	~1-3.5 mm		~10-50 µm	~5 mm	~100-200 µm	5-40 µm	2-4 mm	200 µm	25 µm
Phases	Glass + Cpx + Spl + vap.	Aq. sln. + vap.		Aq. sln. + vap.	Aq. sln. + vap.	Aq. sln.	Aq. sln. + vap.	Aq. sln. + oil + vap.	Aq. sln. + oil + vap. + sb.	Oil + vap.
Voltage	30 kV	160 kV	160 kV*	100 kV	115 kV	100 kV	90 kV	90 kV	110 kV	100 kV
Scan time per projection	8 s	0.708 s	1.415 s*	4 s	3 s	3.75 s	3.75 s	4.5 s	4.5 s	4.5 s
Scan time (total)	280 min	37 min	75 min*	100 min	100 min	125 min	125 min	125 min	105 min	105 min
Projections	1601	3141	3141*	1500	2000	2000	2000	1500	1200	1200
Spatial resolution	2.06 µm/px	25 µm/px	7.7 µm/px *	1.25 µm/px	1.96 µm/px	3.5 µm/px	0.77 µm/px	0.77 µm/px	1 µm/px	1 µm/px
	8.7 μm^3/vx	15625 μm^3/vx	456 μm^3/vx*	1.95 μm^3/vx	7.5 μm^3/vx	42 μm^3/vx	0.45 μm^3/vx	0.45 μm^3/vx	1 μm^3/vx	1 μm^3/vx

FI #	FI_{tot} (Vap. + Liq.)* (mm^3)	Vap.* (mm^3)	Liq.* (mm^3)	φ_{vap}* (%)	FI_{tot} (Vap. + Liq.)** (mm^3)	Vap.** (mm^3)	Liq.** (mm^3)	φ_{vap}** (%)	Δ Vap. (%)	Δ Liq. (%)	Δ φ_{vap} (%)
1	0.28568	0.05581	0.22987	19.5	0.23726	0.05397	0.18146	22.7	3.3	21.1	16
2	0.51785	0.11246	0.40539	21.7	0.45623	0.11926	0.34377	26.1	6.1	15.2	20
3	0.48057	0.10791	0.37266	22.5	0.42364	0.10994	0.31573	26.0	1.9	15.3	16
4	0.40321	0.08927	0.31394	22.1	0.35030	0.08670	0.26103	24.8	2.9	16.9	12
5	0.02901	0.00378	0.02523	13.0	0.01911	0.00350	0.01532	18.3	7.4	39.3	41
6	0.37026	0.08004	0.29022	21.6	0.32465	0.07781	0.24461	24.0	2.8	15.7	11
7	0.45170	0.10092	0.35078	22.3	0.39701	0.09746	0.29609	24.5	3.4	15.6	10
8	0.33186	0.07337	0.25849	22.1	0.28893	0.07102	0.21556	24.6	3.2	16.6	11
3***	0.46119	0.11086	0.35033	24.0	0.14485	0.11042	0.03399	76.2	0.4	90.3	217



Richard et al. Graphical abstract