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# Advanced source apportionment of carbonaceous aerosols by coupling offline AMS and radiocarbon size-segregated measurements over a nearly 2-year period

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**Abstract.** Carbonaceous aerosols are related to adverse human health effects. Therefore, identification of their sources and analysis of their chemical composition is important. The offline AMS (aerosol mass spectrometer) technique offers quantitative separation of organic aerosol (OA) factors which can be related to major OA sources, either primary or secondary. While primary OA can be more clearly separated into sources, secondary (SOA) source apportionment is more challenging because different sources – anthropogenic or natural, fossil or non-fossil – can yield similar highly oxygenated mass spectra. Radiocarbon measurements provide unequivocal separation between fossil and non-fossil sources of carbon. Here we coupled these two offline methods and analysed the OA and organic carbon (OC) of different size fractions (particulate matter below 10 and 2.5  $\mu\text{m}$  –  $\text{PM}_{10}$  and  $\text{PM}_{2.5}$ , respectively) from the Alpine valley of Magadino (Switzerland) during the years 2013 and 2014 (219 samples). The combination of the techniques gave further insight into the characteristics of secondary OC (SOC) which was rather based on the type of SOC precursor and not on the volatility or the oxidation state of OC, as typically considered. Out of the primary sources separated in this study, biomass burning OC was the dominant one in winter, with average concentrations of  $5.36 \pm 2.64 \mu\text{g m}^{-3}$  for  $\text{PM}_{10}$  and  $3.83 \pm 1.81 \mu\text{g m}^{-3}$  for  $\text{PM}_{2.5}$ , indicating that wood combustion particles were predominantly generated in the fine

mode. The additional information from the size-segregated measurements revealed a primary sulfur-containing factor, mainly fossil, detected in the coarse size fraction and related to non-exhaust traffic emissions with a yearly average  $\text{PM}_{10}$  ( $\text{PM}_{2.5}$ ) concentration of  $0.20 \pm 0.24 \mu\text{g m}^{-3}$  ( $0.05 \pm 0.04 \mu\text{g m}^{-3}$ ). A primary biological OC (PBOC) was also detected in the coarse mode peaking in spring and summer with a yearly average  $\text{PM}_{10}$  ( $\text{PM}_{2.5}$ ) concentration of  $0.79 \pm 0.31 \mu\text{g m}^{-3}$  ( $0.24 \pm 0.20 \mu\text{g m}^{-3}$ ). The secondary OC was separated into two oxygenated, non-fossil OC factors which were identified based on their seasonal variability (i.e. summer and winter oxygenated organic carbon, OOC) and a third anthropogenic OOC factor which correlated with fossil OC mainly peaking in winter and spring, contributing on average  $13\% \pm 7\%$  ( $10\% \pm 9\%$ ) to the total OC in  $\text{PM}_{10}$  ( $\text{PM}_{2.5}$ ). The winter OOC was also connected to anthropogenic sources, contributing on average  $13\% \pm 13\%$  ( $6\% \pm 6\%$ ) to the total OC in  $\text{PM}_{10}$  ( $\text{PM}_{2.5}$ ). The summer OOC (SOOC), stemming from oxidation of biogenic emissions, was more pronounced in the fine mode, contributing on average  $43\% \pm 12\%$  ( $75\% \pm 44\%$ ) to the total OC in  $\text{PM}_{10}$  ( $\text{PM}_{2.5}$ ). In total the non-fossil OC significantly dominated the fossil OC throughout all seasons, by contributing on average  $75\% \pm 24\%$  to the total OC. The results also suggested that during the cold period the prevailing source was residential biomass burning while during the warm period primary

biological sources and secondary organic aerosol from the oxidation of biogenic emissions became important. However, SOC was also formed by aged fossil fuel combustion emissions not only in summer but also during the rest of the year.

## 1 Introduction

The field deployment of the time-of-flight aerosol mass spectrometer (HR-ToF-AMS, Canagaratna et al., 2007) has advanced our understanding of aerosol chemistry and dynamics. The HR-ToF-AMS provides quantitative mass spectra of the non-refractory particle component, including, but not limited to, organic aerosol (OA), ammonium sulfate and nitrate, by combining the flash vaporization of particle species and the electron ionization of the resulting gases. The application of positive matrix factorization (PMF, Paatero, 1997) techniques has demonstrated that the collected OA mass spectra contain sufficient information to quantitatively distinguish aerosol sources. However, the cost and intensive maintenance requirements of this instrument significantly hinder its systematic, long-term deployment as part of a dense network and most applications are limited to few weeks of measurements (Jimenez et al., 2009; El Haddad et al., 2013; Crippa et al., 2013). This information is critical for model validation and policy directives. The Aerodyne aerosol chemical speciation monitors (ACSM, Ng et al., 2011; Fröhlich et al., 2013) were developed as a low-cost, low-maintenance alternative to the AMS; however, their reduced chemical resolution can limit the factor separation achievable by source apportionment.

The recent utilization of the AMS for the offline analysis of ambient filter samples (Daellenbach et al., 2016) has significantly broadened the spatial and temporal scales accessible to high-resolution AMS measurements (Daellenbach et al., 2017; Bozzetti et al., 2017a, b). In addition, the technique enables measurement of aerosol composition outside the normal size transmission window of the AMS; the standard AMS can measure up to only 1  $\mu\text{m}$ , or  $\sim 2.5 \mu\text{m}$  with a newly developed aerodynamic lens (Williams et al., 2013; Elser et al., 2016). This capability has been used to quantify the contributions of primary biological organic aerosol to OA in  $\text{PM}_{10}$  filters (Bozzetti et al., 2016). Finally, the offline AMS technique allows a retrospective reaction to critical air quality events. For example, one of the applications of this approach had been to examine a severe haze event in China which affected a total area of  $\sim 1.3$  million  $\text{km}^2$  and  $\sim 800$  million people (Huang et al., 2014).

A major limitation of the technique is the resolution of low water solubility fractions, as the recoveries of some of them are not accessible. Despite this, source apportionment results obtained using this technique are in good agreement with online AMS or ACSM measurements. PMF analysis of offline AMS data has yielded factors related with primary

emissions from traffic, biomass burning and coal burning, and secondary organic aerosols (SOA) differentiated according to their different seasonal contributions. Nevertheless, the identification of SOA precursors using the AMS has proven challenging, due to the evolution of different precursors towards chemically similar species and the extensive fragmentation by the electron ionization used in the AMS.

The radiocarbon ( $^{14}\text{C}$ ) analysis of particulate matter has proven to be a powerful technique providing an unequivocal distinction between non-fossil (e.g. biomass burning and biogenic emissions) and fossil (e.g. traffic exhaust emissions and coal burning) sources (Lemire et al., 2002; Szidat et al., 2004, 2009). The measurement of the  $^{14}\text{C}$  content of total carbon (TC), which comprises the elemental carbon (EC) originating from combustion sources and the organic carbon (OC), had been the subject of many studies (Schichtel et al., 2008; Glasius et al., 2011; Genberg et al., 2011; Zhang et al., 2012, 2016; Zotter et al., 2014b; Bonvalot et al., 2016). Results have shown that in European sites, especially in Alpine valleys, the non-fossil sources play an important role during winter due to biomass burning and in summer due to biogenic sources (Gelencsér et al., 2007; Zotter et al., 2014b). Moreover, at regional background sites close to urbanized areas in Europe (Dusek et al., 2017) as well as in megacities such as Los Angeles and Beijing, fossil OA may also exhibit significant contributions to the total OA (Zotter et al., 2014a; Zhang et al., 2017). However, the determination of the  $^{14}\text{C}$  content in EC and OC separately is challenging and therefore not often attempted for extended datasets.

The coupling of the offline AMS/PMF with radiocarbon analysis provides further insight into the sources of organic aerosols and in particular those related to SOA precursors. Such combination has been already attempted (Minguillón et al., 2011; Zotter et al., 2014a; Huang et al., 2014; Beekmann et al., 2015; Ulevicius et al., 2016); however, the focus has rather been on high OA concentration episodes, while little is known about the yearly cycle of the most important SOA precursors and the size resolution of the different fossil and non-fossil OA fractions.

Here, we present offline AMS measurements of a total of 219 samples, 154 of which are  $\text{PM}_{10}$  samples representative of the years 2013 and 2014 and 65  $\text{PM}_{2.5}$  concurrent with  $\text{PM}_{10}$  samples for the year 2014 (January to September).  $^{14}\text{C}$  analysis was also performed on a subset of 33  $\text{PM}_{10}$  samples, covering the year 2014. The size-segregated samples offered better insights into the mechanism by which the different fractions enter the atmosphere, while the coupling of offline AMS/PMF and  $^{14}\text{C}$  analysis provided a more profound understanding of the SOA fossil and non-fossil precursors on a yearly basis.

## 2 Methods

### 2.1 Site and sampling collection

Magadino is located in an Alpine valley in the Southern part of Switzerland, south of the Alps (Fig. S1 in the Supplement). The station (46°9′37″ N, 8°56′2″ E, 204 m a.s.l.) belongs to the Swiss National Air Pollution Monitoring Network (NABEL) and is classified as a rural background site. It is located relatively far from busy roads or residential areas and surrounded by agricultural fields and forests. It is ca. 1.4 km away from Cadenazzo train station, ca. 8 km from Lake Maggiore (Lago Maggiore) and ca. 7 km from the small Locarno Airport.

The filter samples under examination are 24 h integrated PM<sub>10</sub> (from 4 January 2013 to 28 September 2014, with a 4-day interval) and PM<sub>2.5</sub> (from 3 January to 28 September 2014, with a 4-day interval). PM was sampled and collected on 14 cm (exposed diameter) quartz fibre filters, using a high volume sampler (500 L min<sup>-1</sup>). After the sampling, filter samples and field blanks were wrapped in lint-free paper and stored at -20 °C.

### 2.2 Offline AMS method

The offline AMS method is thoroughly described by Daelenbach et al. (2016). Briefly, four punches of 16 mm diameter from each filter sample are extracted in 15 mL of ultrapure water (18.2 MΩ cm at 25 °C with total organic carbon, TOC, < 3 ppb), followed by insertion in an ultra-sonic bath for 20 min at 30 °C. The water-extracted samples are then filtered through a 0.45 μm nylon membrane syringe and inserted to an Apex Q nebulizer (Elemental Scientific Inc., Omaha, NE, USA) operating at 60 °C. The resulting aerosols generated in Ar (≥ 99.998 % vol., Carbagas, 3073, Gümliigen, Switzerland) were dried by a Nafion dryer and subsequently injected and analysed by the HR-ToF-AMS.

To correct for the interference of NH<sub>4</sub>NO<sub>3</sub> on the CO<sub>2</sub><sup>+</sup> signal as described in Pieber et al. (2016), several dilutions of NH<sub>4</sub>NO<sub>3</sub> in ultrapure water were measured regularly as well. The CO<sub>2</sub><sup>+</sup> signal was then calculated as

$$\text{CO}_{2,\text{real}} = \text{CO}_{2,\text{meas}} - \left( \frac{\text{CO}_{2,\text{meas}}}{\text{NO}_{3,\text{meas}}} \right)_{\text{NH}_4\text{NO}_3,\text{pure}} \cdot \text{NO}_{3,\text{meas}}, \quad (1)$$

where CO<sub>2,real</sub> represents the corrected CO<sub>2</sub><sup>+</sup> signal, CO<sub>2,meas</sub> and NO<sub>3,meas</sub> are signals from the samples measured, and the correction factor  $\left( \frac{\text{CO}_{2,\text{meas}}}{\text{NO}_{3,\text{meas}}} \right)_{\text{NH}_4\text{NO}_3,\text{pure}}$  was determined during the campaign by measuring aqueous NH<sub>4</sub>NO<sub>3</sub>.

### 2.3 <sup>14</sup>C analysis

Based on the instrumentation setup described in Agrios et al. (2015) and on the method described in Zotter et

al. (2014b), radiocarbon analysis of TC and EC was conducted on a set of 33 filters. The <sup>14</sup>C content of blank filters was measured for TC only, as there was no EC found on these filters. All the <sup>14</sup>C results are given in fractions of modern carbon (*f<sub>M</sub>*) representing the <sup>14</sup>C / <sup>12</sup>C ratios of each sample relative to the respective <sup>14</sup>C / <sup>12</sup>C ratio of the reference year 1950 (Stuiver and Polach, 1977).

#### 2.3.1 <sup>14</sup>C measurements of TC

For the determination of the <sup>14</sup>C content of TC, a Sunset OC / EC analyser (Model 4L, Sunset Laboratory, USA) equipped with a non-dispersive infrared (NDIR) detector was first used in order to combust each filter punch (1.5 cm<sup>2</sup>) under pure O<sub>2</sub> (99.9995 %) at 760 °C for 400 s. The generated CO<sub>2</sub> was then captured online by a zeolite trap within a gas inlet system (GIS) and then injected in the accelerator mass spectrometer (AMS\*) mini carbon dating system (MICADAS) at the Laboratory for the Analysis of Radiocarbon with AMS\* (LARA), University of Bern, Switzerland (Szidat et al., 2014) for <sup>14</sup>C measurement. (Note that we used AMS\* and AMS as abbreviations for the accelerator mass spectrometer and the aerosol mass spectrometer, respectively, to avoid confusion.)

The *f<sub>M</sub>* of TC underwent a blank correction following an isotopic mass balance approach:

$$f_{M_b,\text{cor}} = \frac{mC_{\text{sample}} \cdot f_{M,\text{sample}} - mC_b \cdot f_{M,b}}{mC_{\text{sample}} - mC_b}, \quad (2)$$

where *f<sub>M<sub>b,cor</sub></sub>* is the blank corrected *f<sub>M</sub>*; *mC<sub>sample</sub>* and *mC<sub>b</sub>* are the carbon mass in sample and blank, respectively; and *f<sub>M,sample</sub>* and *f<sub>M,b</sub>* are the *f<sub>M</sub>* measured for sample and blank, respectively. Error propagation was applied for the determination of the *f<sub>M<sub>b,cor</sub></sub>* uncertainty. The *f<sub>M,b</sub>* was 0.61 ± 0.10 and the concentration of the blank 1.1 ± 0.2 μg C m<sup>-3</sup>.

#### 2.3.2 <sup>14</sup>C measurements of EC

For the EC isolation of the samples, each filter punch (1.5 cm<sup>2</sup>) was analysed by the Sunset EC / OC analyser with the use of the Swiss\_4S protocol developed by Zhang et al. (2012). According to the protocol, the heating is conducted in four different steps under different gas conditions: step one under pure O<sub>2</sub> at 375 °C for 150 s, step two under pure O<sub>2</sub> at 475 °C for 180 s, step three under He (> 99.999 %) at 450 °C for 180 s followed by an increase in the temperature up to 650 °C for another 180 s, and step four under pure O<sub>2</sub> at 760 °C for 150 s. Each filter sample was previously water extracted and dried, in order to minimize the positive artefact induced by the OC by removing the water-soluble OC (WSOC), which is known to produce charring (Piazzalunga et al., 2011; Zhang et al., 2012). By this method, the water-insoluble OC (WINSOC) was removed during the first three steps of the Swiss\_4S protocol. In the fourth step, EC was

combusted and then trapped in the GIS and measured by the AMS\* MICADAS, as described above.

This protocol was preferred over the protocols commonly used in thermo-optical methods (EUSAAR 2 or NIOSH) because it optimises the separation of the two fractions OC and EC by minimizing (i) the positive artefact of charring produced by WSOC during the first three steps and (ii) the premature losses, during the removal of the WINSOC in the third step, of the less refractory part of EC which may preferentially originate from non-fossil sources such as biomass burning.

Following a similar principle to Zotter et al. (2014b), both charring and EC yield, which is the part of EC that remained on the filter after step three and before step four in the Swiss\_4S protocol, were quantified and corrected for with the help of the laser mounted on the Sunset analyser. The laser transmittance is monitored continuously during the heating process. Charring in step three was quantified as

$$\text{Charring}_{S_3} = \frac{\text{maxATN}_{S_3} - \text{initial ATN}_{S_2}}{\text{initial ATN}_{S_1}}, \quad (3)$$

where ATN refers to the laser attenuation,  $\text{maxATN}_{S_3}$  is the maximum attenuation in step three, and  $\text{initial ATN}_{S_2}$  and  $\text{initial ATN}_{S_1}$  are the initial attenuations in step two and one, respectively.

The EC yield in step three was quantified as

$$\text{ECyield}_{S_3} = \frac{\text{initial ATN}_{S_3}}{\text{maxATN}_{S_3}} \cdot \frac{\text{initial ATN}_{S_2}}{\text{maxATN}_{S_1}}, \quad (4)$$

The average charred OC was found to be  $4 \pm 2\%$  and the recovered EC for all samples was on average  $71 \pm 7\%$ .

As there is a linear relationship between the fraction of modern carbon for EC ( $f_{M_{EC}}$ ) and the EC yield (Zhang et al., 2012), the slope can be used to extrapolate  $f_{M_{EC}}$  to 100 % EC yield. According to Zotter et al. (2014), a slope of  $0.35 \pm 0.11$  was considered to correct all  $f_{M_{EC}}$  to 100 % of EC yield, such that

$$f_{M_{EC, \text{total}}} = \text{slope} \cdot (1 - \text{ECyield}_{S_3}) + f_{M_{EC}}. \quad (5)$$

### 2.3.3 Calculation of $^{14}\text{C}$ content of OC

The fraction of modern carbon of OC ( $f_{M_{OC}}$ ) was calculated following a mass balance approach:

$$f_{M_{OC}} = \frac{\text{TC} \cdot f_{M_{TC}} - \text{EC} \cdot f_{M_{EC}}}{\text{TC} - \text{EC}}, \quad (6)$$

where TC and EC are the concentrations of total and elemental carbon, respectively, and  $f_{M_{TC}}$  and  $f_{M_{EC}}$  are the fractions of modern carbon of TC and EC, respectively. The uncertainty of  $f_{M_{OC}}$  was calculated by propagating the error of each component of Eq. (6).

### 2.3.4 Nuclear bomb peak correction

The expected  $f_M$  coming from fossil samples should be equal to zero due to the complete decay of  $^{14}\text{C}$  until now, whereas the  $f_M$  from non-fossil samples is expected to be unity. However, due to the extensive nuclear bomb testing during the late 1950s and early 1960s, the radiocarbon amount in the atmosphere increased dramatically because of the high neutron flux during the explosions. Therefore the measured  $f_M$  of non-fossil samples may exhibit values greater than one (Levin et al., 2010a). To correct for this effect, the  $f_M$  is normalized to a reference non-fossil fraction ( $f_{M_{NF, \text{ref}}}$ ) which represents the amount of  $^{14}\text{C}$  currently in the atmosphere compared to 1950, before the nuclear bomb tests. As EC comes from either biomass burning or fossil sources, the non-fossil fraction of EC ( $f_{M_{NF, EC}}$ ) equals the  $f_M$  coming from biomass burning ( $f_{M, \text{bb}}$ ). The latter was estimated by a tree growth model (Mohn et al., 2008) and was equal to 1.101. The non-fossil fraction of OC ( $f_{M_{NF, OC}}$ ) is calculated as

$$f_{M_{NF, OC}} = p_{\text{bio}} \cdot f_{M, \text{bio}} + p_{\text{bb}} \cdot f_{M, \text{bb}}, \quad (7)$$

where  $f_{M, \text{bio}}$  ( $= 1.023$ ) is the fraction of modern carbon of biogenic sources and was estimated from  $^{14}\text{CO}_2$  measurements in Schauinsland (Levin et al., 2010a). The fractions of biogenic sources ( $p_{\text{bio}}$ ) and biomass burning ( $p_{\text{bb}}$ ) to the total non-fossil sources were set to 0.5 since both sources are important in Magadino during the year (biomass burning in winter, biogenic sources in summer).

## 2.4 Additional measurements

Organic and elemental carbon fractions were determined by a Sunset EC / OC analyser with the use of the EUSAAR-2 thermal-optical transmittance protocol (Cavalli et al., 2010). Water-soluble organic carbon was measured by a total organic carbon analyser (Jaffrezo et al., 2005) with the use of catalytic oxidation of water-extracted filter samples and detection of the resulting  $\text{CO}_2$  with an NDIR. The concentrations of major ionic species ( $\text{K}^+$ ,  $\text{Na}^+$ ,  $\text{Mg}^{2+}$ ,  $\text{Ca}^{2+}$ ,  $\text{NH}_4^+$ ,  $\text{Cl}^-$ ,  $\text{NO}_3^-$  and  $\text{SO}_4^{2-}$ ) as well as methane sulfonic acid (MSA) were determined by ion chromatography (Jaffrezo et al., 1998). Anhydrous sugars (levoglucosan, mannan, galactosan) were analysed by an ion chromatograph (Dionex ICS3000) using high-performance anion exchange chromatography (HPAEC) with pulsed amperometric detection. Cellulose was analysed by performing enzymatic conversion of cellulose to D-glucose (Kunit and Puxbaum, 1996) and D-glucose was determined by HPAEC.

## 3 Source apportionment

### 3.1 Method

The obtained organic mass spectra from the offline AMS measurements were analysed by positive matrix factoriza-

tion (Paatero and Tapper, 1994; Ulbrich et al., 2009). PMF attempts to solve the bilinear matrix equation,

$$\mathbf{X}_{ij} = \sum_k \mathbf{G}_{i,k} \mathbf{F}_{k,j} + \mathbf{E}_{i,j}, \quad (8)$$

by following the weighted least-squares approach. In the case of aerosol mass spectrometry,  $i$  represents the time index,  $j$  the fragment and  $k$  the factor number. If  $\mathbf{X}_{ij}$  is the matrix of the organic mass spectral data and  $\mathbf{s}_{i,j}$  the corresponding error matrix,  $\mathbf{G}_{i,k}$  the matrix of the factor time series,  $\mathbf{F}_{k,j}$  the matrix of the factor profiles and  $\mathbf{E}_{i,j}$  the model residual matrix, then PMF determines  $\mathbf{G}_{i,k}$  and  $\mathbf{F}_{k,j}$  such that the ratio of the Frobenius norm of  $\mathbf{E}_{i,j}$  over  $\mathbf{s}_{i,j}$  is minimized. The allowed  $\mathbf{G}_{i,k}$  and  $\mathbf{F}_{k,j}$  are always non-negative. The input error matrix  $\mathbf{s}_{i,j}$  includes the measurement uncertainty (ion-counting statistics and ion-to-ion signal variability at the detector) (Allan et al., 2003) as well as the blank variability. Fragments with a signal-to-noise ratio (SNR) below 0.2 were removed and the ones with SNR lower than 2 were down-weighted by a factor of 3, as recommended by Paatero and Hopke (2003). Both input data and error matrices were scaled to the calculated water-soluble organic matter (WSOM<sub>*i*</sub>) concentration:

$$\text{WSOM}_i = \frac{\text{OM}}{\text{OC}} \cdot \text{WSOC}_i, \quad (9)$$

where  $\frac{\text{OM}}{\text{OC}}$  is determined from the AMS measurements and  $\text{WSOC}_i$  is the water-soluble OC measured by the TOC analyser.

The Source Finder toolkit (SoFi v.4.9, Canonaco et al., 2013) for IGOR Pro software package (Wavemetrics, Inc., Portland, OR, USA) was used to run the PMF algorithm. The PMF was solved by the multilinear engine 2 (ME-2, Paatero, 1999), which allows the constraining of the  $\mathbf{F}_{k,j}$  elements to vary within a certain range defined by the scalar  $\alpha$  ( $0 \leq \alpha \leq 1$ ), such that the modelled  $\mathbf{F}'_{k,j}$  equals

$$\mathbf{F}'_{k,j} = \mathbf{F}_{k,j} \pm \alpha \cdot \mathbf{F}_{k,j}. \quad (10)$$

Here we constrained only the hydrocarbon-like factor (HOA) from high-resolution mass spectra analysed by Crippa et al. (2013).

### 3.2 Sensitivity analysis

To understand the variability of our dataset we explored 4–10 factor solutions and retained the 7-factor solution as the best representation of the data. The exploration of the PMF solutions is thoroughly described in Sect. S.1 in the Supplement.

We assessed the accuracy of PMF results by bootstrapping the input data (Davison and Hinkley, 1997). New input data and error matrices were created by randomly resampling the time series from the original input matrix (223 samples in total: 219 + 4 remeasurements from the PM<sub>10</sub> samples), with replacement; i.e. any sample from the whole population

can be resampled more than once. Each sample measurement included on average blocks of 12 mass spectral repetitions; therefore, resampling was performed on the blocks. Out of the 223 original samples, some of them were represented several times, while some others not at all. Overall, the resampled data made up on average  $64 \pm 2\%$  of the total original data per bootstrap run. We performed 180 bootstrap runs, with each of the generated matrices being perturbed by varying the  $\mathbf{X}_{ij}$  element within twice the corresponding error matrix  $\mathbf{s}_{i,j}$ . Within the resampling operation, the  $\alpha$  value used to set the HOA constraining strength was varied between 0 and 1 with an increment of 0.1 to assess the sensitivity of the results on the  $\alpha$  value.

To select the physically plausible solutions we applied two criteria:

1. We accepted solutions where the average absolute concentrations of all factors in PM<sub>2.5</sub> did not statistically significantly exceed their concentrations in PM<sub>10</sub>. For this we performed a paired  $t$  test with a significance level of 0.01 (Fig. S2 and Table S1 in the Supplement).
2. We excluded outlier solutions identified by examining the correlation of factor time series from bootstrap runs with their respective factor time series from the average of all bootstrap runs. The rejected solutions included factors that did not correlate with the corresponding average factor time series, meaning that one of the factors was not separated (Fig. S3 in the case of water-soluble primary biological organic carbon, PBOC).

In total 24 bootstrap runs were retained after the application of the aforementioned criteria.

### 3.3 Recoveries

In order to rescale the WSOC concentration of a factor  $k$  to its total concentration  $\text{OC}_k$ , we used factor recoveries ( $R_k$ ) as proposed by Daellenbach et al. (2016). First, the  $\text{WSOM}_k$  was calculated as

$$\text{WSOM}_k = f_{k,\text{WSOM}} \cdot \text{WSOC}_{\text{measured}} \cdot \left( \frac{\text{OM}}{\text{OC}} \right)_{\text{bulk}}, \quad (11)$$

where

$$f_{k,\text{WSOM}} = \frac{\text{WSOM}_{k,\text{measured}}}{\sum_k \text{WSOM}_{k,\text{measured}}} \quad (12)$$

and

$$\left( \frac{\text{OM}}{\text{OC}} \right)_{\text{bulk}} \quad (13)$$

is estimated from the input data matrix for the PMF.

The  $\text{WSOM}_k$  was converted to  $\text{WSOC}_k$  to fit the measured OC concentrations (determined by the Sunset EC / OC analyser). The  $\text{WSOC}_k$  was determined as

$$\text{WSOC}_k = \frac{f_{k,\text{WSOM}} \cdot \text{WSOC}_{\text{measured}} \cdot \left( \frac{\text{OM}}{\text{OC}} \right)_{\text{bulk}}}{\left( \frac{\text{OM}}{\text{OC}} \right)_k}, \quad (14)$$

where  $(\frac{OM}{OC})_k$  is calculated from each factor profile.

Finally, the recoveries were applied following Eq. (15):

$$OC_{i,k} = \frac{WSOC_{i,k}}{R_k}. \quad (15)$$

To assess the recoveries and their uncertainties, we evaluated the sum of  $OC_{i,k}$  against the measured OC ( $OC_{i,measured}$ ) by fitting Eq. (16). The starting values for the  $R_k$  fitting were based on Bozzetti et al. (2016) (for  $R_{PBOA}$ ) and Daellenbach et al. (2016) except  $R_{SCOAA}$ , which was randomly varied between 0 and 1 (increment:  $10^{-4}$ ). While  $R_{HOA}$  and  $R_{SCOAA}$  were constrained,  $R_{PBOA}$ ,  $R_{BBOA}$ ,  $R_{WOOA}$ ,  $R_{AOOA}$  and  $R_{SOOA}$  were determined by a non-negative multilinear fit (see below in Sect. 4.3 for a description of these PMF factors from offline AMS results). The multilinear fit was chosen to be non-negative because a negative  $R_k$  would mean a negative concentration of  $WSOC_k$  or  $OC_k$ . The fit was performed 100 times for each of the retained bootstrap solutions.

$$OC_{i,measured} = \sum_k \frac{WSOC_{i,k}}{R_k} \quad (16)$$

Each fit was initiated by perturbing the  $OC_{i,k}$  and the  $WSOC_{i,k}$  concentrations within their uncertainties, assuming a normal distribution of errors, to assess the influence of measurement precision on  $R_k$ . Additionally, we introduced a constant 5 % accuracy bias corresponding to the OC and WSOC measurement accuracy.

To select the environmentally meaningful solutions we applied the following criteria:

1. To retain the recoveries that achieved the OC mass closure, we estimated the OC residuals and discarded solutions where OC residuals were statistically different from 0 within 1 standard deviation for each size fraction individually and for winter and summer individually.
2. We also examined the dependence between the WSOC residuals and each factor  $WSOC_{i,k}$  ( $t$  test,  $\alpha = 0.001$ ). Overall, 55 % of the solutions were retained.
3. The physically plausible range of the recoveries is [0, 1]. However, the mathematically possible range can exceed the upper limit.  $R_k$  larger than 1 would mean that  $WSOC_k$  is larger than  $OC_k$  and is, therefore, non-physical. For this reason, out of the accepted solutions that survived the previous two criteria, the retained  $R_k$  combinations were weighted according to their physical interpretability. More specifically, fitting results with  $R_k$  larger than 1 were down-weighted according to the measurement uncertainties of WSOC and OC (see Sect. S.2, Fig. S4).

## 4 Results and discussion

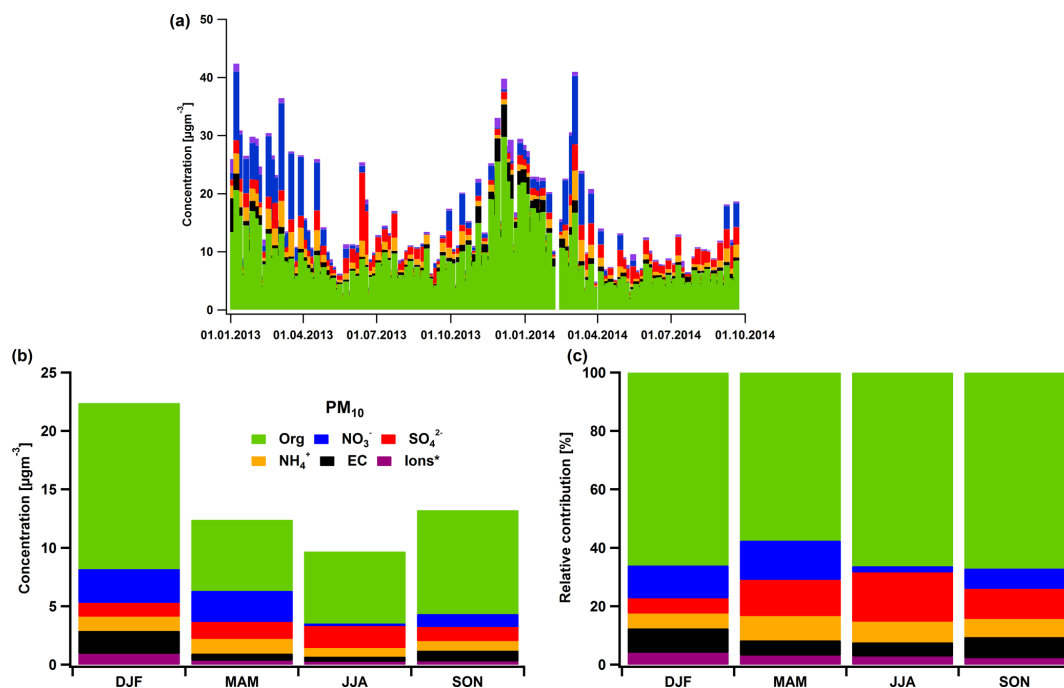
### 4.1 PM<sub>10</sub> composition

PM<sub>10</sub> in Magadino has been characterized by high carbonaceous concentrations during winter (Gianini et al., 2012a; Zotter et al., 2014b). This is clearly illustrated in Fig. 1 where an overview of the PM<sub>10</sub> composition is presented in Fig. 1a with Fig. 1b and c summarizing the concentrations and relative contributions of each component to the total PM<sub>10</sub> averaged per season. The peaks of OM and EC during winter (daily averages up to 26 and 5.9  $\mu\text{g m}^{-3}$ , respectively) are indications of the increased wood-burning activity. Other Alpine sites close to Magadino, such as Roveredo and San Vittore in Switzerland, have also exhibited high OM concentrations due to residential wood burning (Szidat et al., 2007, for PM<sub>10</sub> in Roveredo, Lanz et al., 2010, for PM<sub>1</sub> in Roveredo and Zotter et al., 2014b, for PM<sub>10</sub> in San Vittore and Roveredo). The organic contribution dominated the inorganic fraction not only in winter, but also throughout both years (Fig. 1c). Note that the EC concentrations are much lower in spring compared to winter (Fig. 1b). The main inorganic aerosols contributing to the total PM are  $\text{NO}_3^-$ ,  $\text{SO}_4^{2-}$  and  $\text{NH}_4^+$ .  $\text{NO}_3^-$  represented the second major component of PM<sub>10</sub>, exhibiting a seasonal cycle with higher concentrations during winter (2.9  $\mu\text{g m}^{-3}$ ). The notable discrepancy of  $\text{NO}_3^-$  concentrations between the first (2013) and second (2014) winter could be explained by the lower temperatures in January–February 2013 compared to 2014. Conversely,  $\text{SO}_4^{2-}$  showed a rather stable yearly cycle with slightly higher concentrations in summer (1.9  $\mu\text{g m}^{-3}$ ) compared to winter (1.3  $\mu\text{g m}^{-3}$ ), despite a shallower boundary layer height in winter.

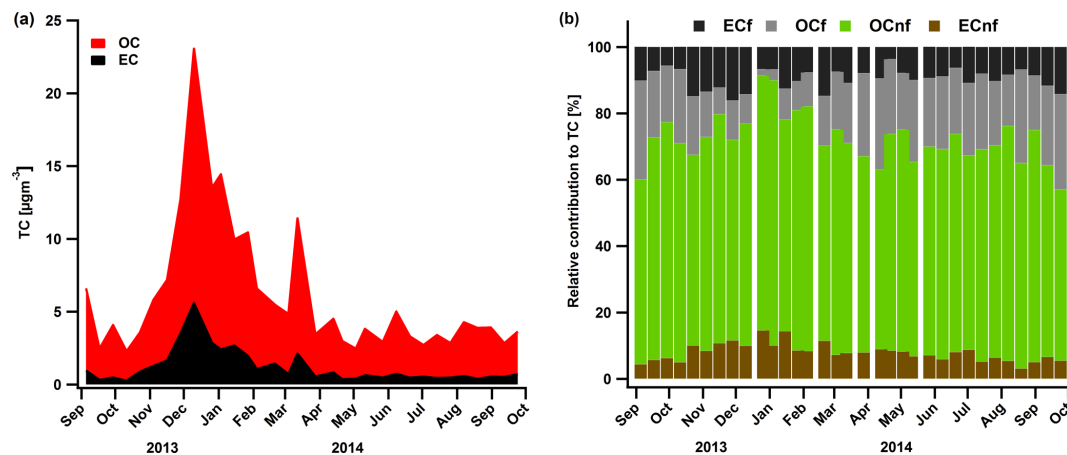
### 4.2 <sup>14</sup>C analysis results

So far radiocarbon results have been reported mostly for relatively short periods of time (Bonvalot et al., 2016), mainly describing high concentration events, and only a few studies report measurements on a yearly basis (Genberg et al., 2011; Gilardoni et al., 2011; Zotter et al., 2014b; Zhang et al., 2016, 2017; Dusek et al., 2017). Here, for a subset of 33 PM<sub>10</sub> filters from the year 2014, we present yearly contributions of  $OC_{nf}$ ,  $OC_f$ ,  $EC_{nf}$  and  $EC_f$ .

Overall the total carbon concentrations followed a yearly pattern mainly caused by the shallow planetary boundary layer and the enhanced biomass burning activity during winter, with OC reaching on average ( $\pm 1$  standard deviation)  $9.4 \pm 4.5$  and EC  $2.6 \pm 1.5 \mu\text{g m}^{-3}$  (Fig. 2a). During the rest of the year, TC remained rather stable with much lower concentrations ( $OC_{avg} = 3.7 \pm 1.9$  and  $EC_{avg} = 0.8 \pm 0.7 \mu\text{g m}^{-3}$ ). <sup>14</sup>C results indicate that non-fossil sources prevail over the fossil ones in Magadino. More specifically, we found that in winter on average  $f_{NF,OC} = 0.9 \pm 0.1$  and  $f_{NF,EC} = 0.5 \pm 0.1$ , which is in



**Figure 1.** Concentrations of OM, EC and major ionic species for the years 2013 and 2014 (a), their seasonal concentrations (b) and relative contributions to the total measured mass within the particulate matter ( $\text{PM}_{10}$ ) (c). The sum of the ions  $\text{Na}^+$ ,  $\text{K}^+$ ,  $\text{Mg}^{2+}$ ,  $\text{Ca}^{2+}$  and  $\text{Cl}^-$  are included in the indication “Ions\*”.



**Figure 2.** Time series of OC and EC (a) concentrations in  $\text{PM}_{10}$ .  $^{14}\text{C}$  analysis results with the relative contributions of EC fossil, OC fossil, OC non-fossil and EC non-fossil to the TC (b).

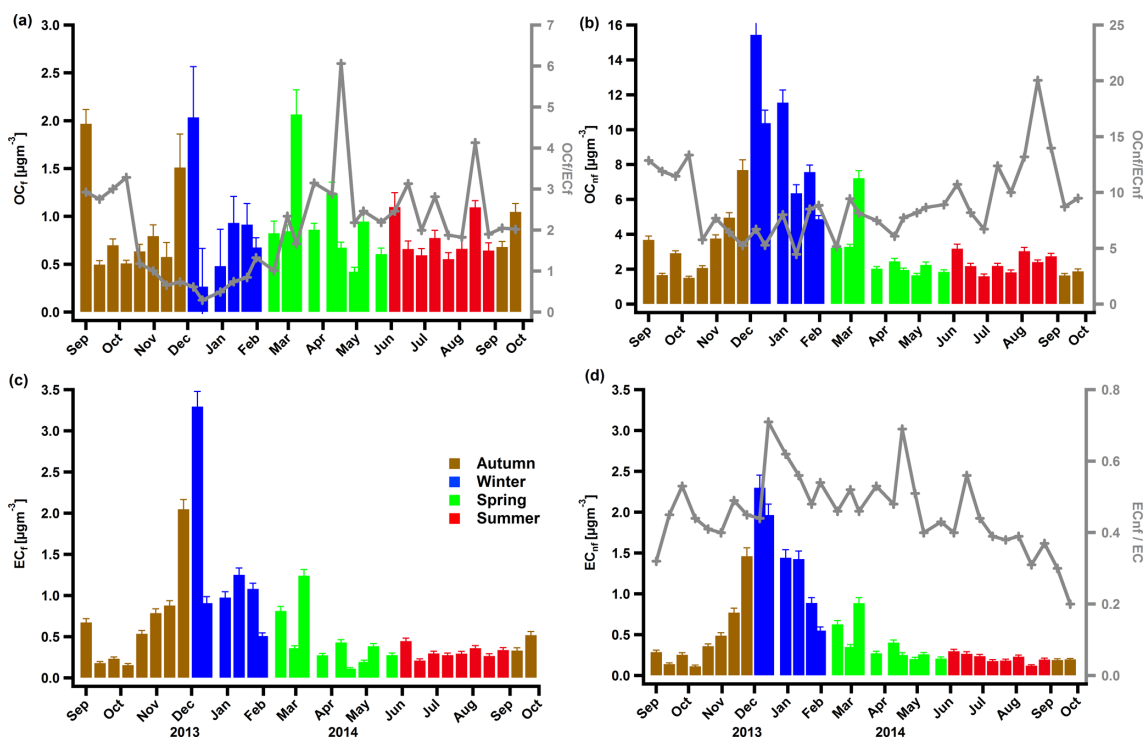
agreement with the reported fractions by Zotter et al., 2014b ( $f_{\text{NF,OC}} = 0.8 \pm 0.1$  and  $f_{\text{NF,EC}} = 0.5 \pm 0.2$ ). Table 1 summarizes the  $f_{\text{NF}}$  per fraction season wise.

$\text{OC}_{\text{nf}}$  was the dominant part of TC throughout the year with contributions of up to 80% in winter and 71% in summer (Fig. 2b) and average concentrations of  $8.5 \pm 4.2$  and  $2.4 \pm 0.6 \mu\text{g m}^{-3}$  in winter and summer, respectively (Fig. 3b). Such high contributions in winter strongly indicate that biomass burning (BB) from residential heating is the main source of carbonaceous aerosols in this

region, similar to previous reports (Jaffrezo et al., 2005; Puxbaum et al., 2007; Sandradewi et al., 2008; Favez et al., 2010; Zotter et al., 2014b). The coefficient of determination  $R^2$  between  $\text{OC}_{\text{nf}}$  and levoglucosan, a characteristic marker for BB, was 0.92 (Fig. S7a), and the slope ( $\text{OC}_{\text{nf}} / \text{levoglucosan} = 4.8 \pm 0.3$ ) lies within the reported range by Zotter et al. (2014b) for Magadino (which was  $6.9 \pm 2.6$ ).

The concentration of  $\text{EC}_{\text{nf}}$  was significantly higher in winter (average  $1.3 \pm 0.7 \mu\text{g m}^{-3}$ ) compared to the rest of





**Figure 3.** Concentrations in PM<sub>10</sub> of OC<sub>f</sub> (a), OC<sub>nf</sub> (b), EC<sub>f</sub> (c) and EC<sub>nf</sub> (d) colour-coded by seasons. The ratios OC<sub>f</sub>/EC<sub>f</sub>, OC<sub>nf</sub>/EC<sub>nf</sub>, and EC<sub>nf</sub>/EC are also displayed in (a), (b) and (d), respectively.

**Table 1.** Median OC and EC non-fossil fractions per season in PM<sub>10</sub> with interquartile range.

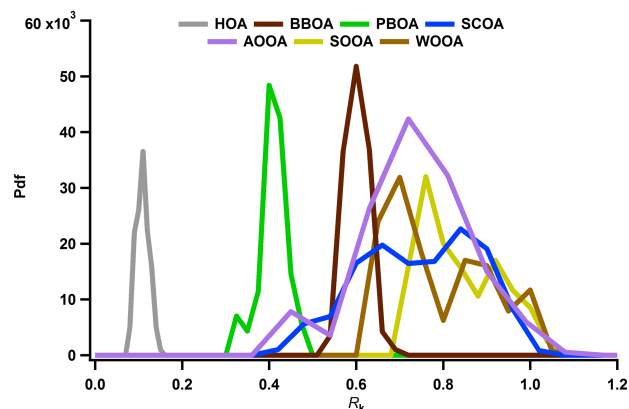
	Autumn			Winter			Spring			Summer		
	Q <sub>25</sub>	Q <sub>50</sub>	Q <sub>75</sub>	Q <sub>25</sub>	Q <sub>50</sub>	Q <sub>75</sub>	Q <sub>25</sub>	Q <sub>50</sub>	Q <sub>75</sub>	Q <sub>25</sub>	Q <sub>50</sub>	Q <sub>75</sub>
$f_{\text{NF,OC}}$	0.71	0.77	0.83	0.87	0.88	0.93	0.70	0.75	0.79	0.73	0.76	0.79
$f_{\text{NF,EC}}$	0.36	0.41	0.44	0.44	0.52	0.56	0.42	0.49	0.51	0.38	0.39	0.42

the year (spring average:  $0.4 \pm 0.2 \mu\text{g m}^{-3}$ , summer average:  $0.21 \pm 0.06 \mu\text{g m}^{-3}$ , autumn average:  $0.43 \pm 0.41 \mu\text{g m}^{-3}$ ) (Fig. 3d). EC<sub>nf</sub> is considered to originate solely from BB, for instance from residential wood burning in winter. This assumption is supported by the very high correlation ( $R^2 = 0.95$ ) with levoglucosan (Fig. S7b) and the slope ( $\text{EC}_{\text{nf}} / \text{levoglucosan} = 0.82 \pm 0.03$ ) which is also in agreement with the literature (Zotter et al., 2014b; Herich et al., 2014).

The strong correlation between OC<sub>nf</sub> and EC<sub>nf</sub>, driven mainly by the winter data points, supports the fact that OC<sub>nf</sub> is mostly from biomass burning in winter (Fig. S6a). In late spring, summer and early autumn, the contribution of EC<sub>nf</sub> decreased significantly (on average to  $0.23 \pm 0.07 \mu\text{g m}^{-3}$ ). The low correlation of OC<sub>nf</sub> and EC<sub>nf</sub> during this period (Fig. S6a), in combination with the increase in the OC<sub>nf</sub>/EC<sub>nf</sub> ratio in summer (Fig. 3b), suggests that a part of the secondary OC<sub>nf</sub> originates from non-combustion sources, e.g. biogenic/natural sources.

In total, the relative contribution of the fossil fraction to the TC was 27%. Excluding winter, EC<sub>f</sub> exhibited slightly higher concentrations than EC<sub>nf</sub> (Fig. 3c and d). The average concentrations of EC<sub>f</sub> were  $1.26 \pm 0.93$ ,  $0.41 \pm 0.35$ ,  $0.31 \pm 0.07$  and  $0.63 \pm 0.56 \mu\text{g m}^{-3}$  for winter, spring, summer and autumn, respectively (Fig. 3c). The increase in EC<sub>f</sub> witnessed in winter could be mainly attributed to the shallower planetary boundary layer (PBL) rather than to an increase in the emissions (Fig. S8a). The sources of EC<sub>f</sub> in the coarse (PM<sub>10</sub>–PM<sub>2.5</sub>) size fraction are typically related to re-suspension of abrasion products of vehicle tires or brake wear (Bukowiecki et al., 2010; Zhang et al., 2013). The fine part of EC<sub>f</sub> is due to fossil fuel burning, here mostly due to traffic exhaust emissions. It is significantly correlated with NO<sub>x</sub> (Fig. S8b) and the EC<sub>f</sub>/NO<sub>x</sub> = 0.020 ratio lies within the reported slopes (Zotter et al., 2014b, and references therein).

The contribution of OC<sub>f</sub> to TC decreased during winter (8%) but remained roughly stable throughout the rest of the year (22% in spring, 21% in summer and 19% in



**Figure 4.** Probability density functions of factor recoveries: hydrocarbon-like OA (HOA) in grey, biomass burning OA (BBOA) in dark brown, sulfur-containing OA (SCOA) in blue, primary biological OA (PBOA) in green, anthropogenic oxygenated OA (AOOA) in purple, summer oxygenated OA (SOOA) in yellow and winter oxygenated OA (WOOA) in light brown.

**Table 2.** Variability of OM / OC and factor recoveries.

	OM / OC			$R_k$		
	$Q_{25}$	$Q_{50}$	$Q_{75}$	$Q_{25}$	$Q_{50}$	$Q_{75}$
HOA	1.32	1.33	1.36	0.10	0.11	0.13
BBOA	1.76	1.77	1.78	0.60	0.61	0.63
SCOA	2.03	2.16	2.20	0.68	0.81	0.89
PBOA	1.74	1.76	1.82	0.41	0.42	0.44
AOOA	2.12	2.14	2.16	0.72	0.79	0.87
SOOA	1.66	1.67	1.68	0.78	0.84	0.94
WOOA	1.76	1.79	1.83	0.72	0.78	0.92

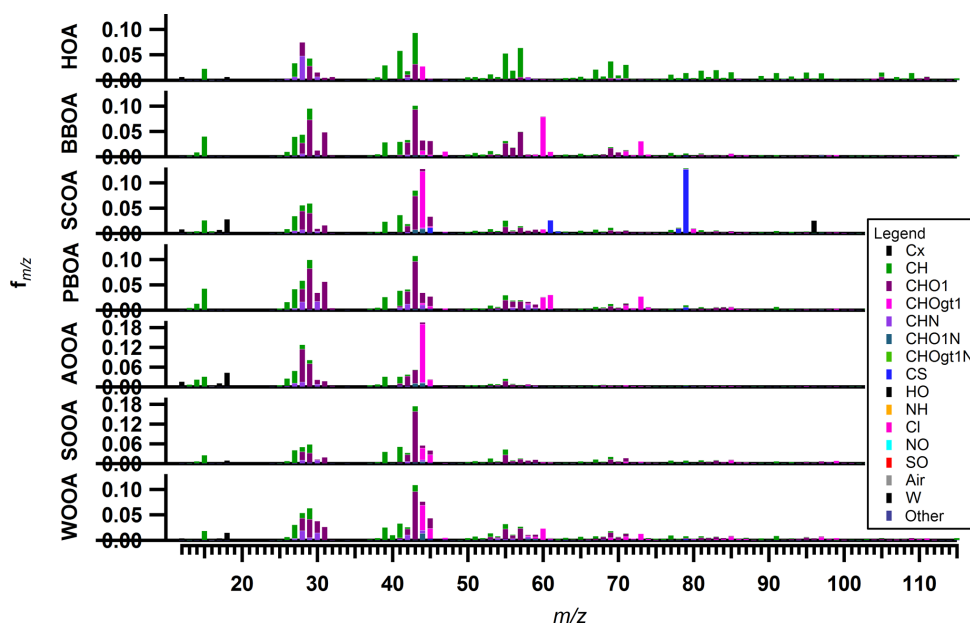
autumn, Fig. 2b) with average concentrations  $0.87 \pm 0.30$ ,  $0.96 \pm 0.12$ ,  $0.89 \pm 0.14$  and  $0.76 \pm 0.10 \mu\text{g m}^{-3}$  for winter, spring, summer and autumn, respectively (Fig. 3a). The low correlation overall observed between  $\text{OC}_f$  and  $\text{EC}_f$  (Fig. S6b) may indicate that a fraction of  $\text{OC}_f$  is not directly emitted but formed as secondary OC (SOC) from fossil-fuel-related emissions (e.g. traffic). This is supported by low  $\text{OC}_f/\text{EC}_f$  ratios in winter (on average  $0.7 \pm 0.3$ ) and much higher values in spring and summer (on average  $2.7 \pm 1.1$ ) (Fig. 3a). The low ratios are consistent with tunnel measurement studies (Li et al., 2016; Chirico et al., 2011; El Haddad et al., 2009) and the increase in  $\text{OC}_f/\text{EC}_f$  in spring and summer above these values is an indication of anthropogenic SOA formation. We also note that fossil SOA may be formed by other sources besides traffic. A recent study revealed that fossil SOA is produced by the oxidation of volatile chemical products coming from petrochemical sources (McDonald et al., 2018).

### 4.3 Offline AMS analysis results: factor interpretation

In this section, we will interpret the PMF outputs. The factor recoveries for all factors,  $R_k$ , determined as described in Sect. 3.3, are shown in Fig. 4. Factor mass spectra are displayed in Fig. 5. The contribution of the different factors to OA is presented in Fig. 6. In addition, for some cases we will discuss the factor contribution to OC to check the consistency of our results with previous literature reports. Recovery values determined and used in this study will also be compared for each factor to previous values. Median values of the recoveries as well as the OM / OC ratios with their interquartile range are compiled in Table 2. The  $R_k$  values were in general consistent with previous reports (Daellenbach et al., 2016, 2017; Bozzetti et al., 2016). Here we report for the first time the recoveries of each SOA factor individually which were in agreement with the ones reported by Daellenbach et al. (2016). The consistency of the recovery results with not only previous offline AMS/PMF studies but also with on-line AMS measurements (Xu et al., 2017) points out that this method is rather robust and universal for different datasets.

Hydrocarbon-like OA (HOA), typically associated with traffic emissions, was constrained using the reference HOA high-resolution profile from Crippa et al. (2013). The resulting factor profile (Fig. 5) exhibited a low OM / OC (Table 2) and the time series followed the one from  $\text{NO}_x$  (Fig. 6). As the offline AMS technique requires water-extracted samples, it is expected that HOA, which mostly contains water-insoluble material, will be poorly represented. This is also shown by the low recovery  $R_{\text{HOA,median}}$  which was estimated to be 0.11 ( $Q_{25} = 0.10$  and  $Q_{75} = 0.13$ ) as reported in Daellenbach et al. (2016) (Fig. 4). Therefore, the correlation between HOA and  $\text{NO}_x$  was weak (Fig. S9). However, the HOA/ $\text{NO}_x$  ratio was 0.017 for  $\text{PM}_{10}$  and 0.008 for  $\text{PM}_{2.5}$  and these values are consistent with already reported ones in the literature (Daellenbach et al., 2017; Lanz et al., 2007). In addition, the HOC time series followed a similar yearly cycle as  $\text{EC}_f$  (Fig. S10a) and the HOC /  $\text{OC}_f$  ratio was  $0.37 \pm 0.12$  (Fig. S10b), in agreement with Zotter et al. (2014a).

Biomass burning OA (BBOA) was identified by its significant contributions of the oxygenated fragments  $\text{C}_2\text{H}_4\text{O}_2^+$  (at  $m/z$  60) and  $\text{C}_3\text{H}_5\text{O}_2^+$  (at  $m/z$  73), common markers for wood burning formed by fragmentation of anhydrous sugars (Alfarra et al., 2007) (Fig. 5). It was also identified by its distinct seasonal variation which exhibited exclusively high concentrations in winter, reaching up to  $20.0 \pm 0.7 \mu\text{g m}^{-3}$  for  $\text{PM}_{10}$  in December 2013 and  $12.3 \pm 0.5 \mu\text{g m}^{-3}$  for  $\text{PM}_{2.5}$  in January 2014 (Fig. 6). The median value for the OM / OC ratio was 1.8 and the  $R_{\text{BBOA}}$  was consistent with the low end of the reported one by Daellenbach et al. (2016) (Table 2). The identification of this factor as BBOA was further confirmed by its remarkable correlation with levoglucosan. Similar to levoglucosan, this factor did not exhibit a significant difference between  $\text{PM}_{2.5}$  and  $\text{PM}_{10}$  concentrations (Fig. S5a), suggesting that most of these particles are



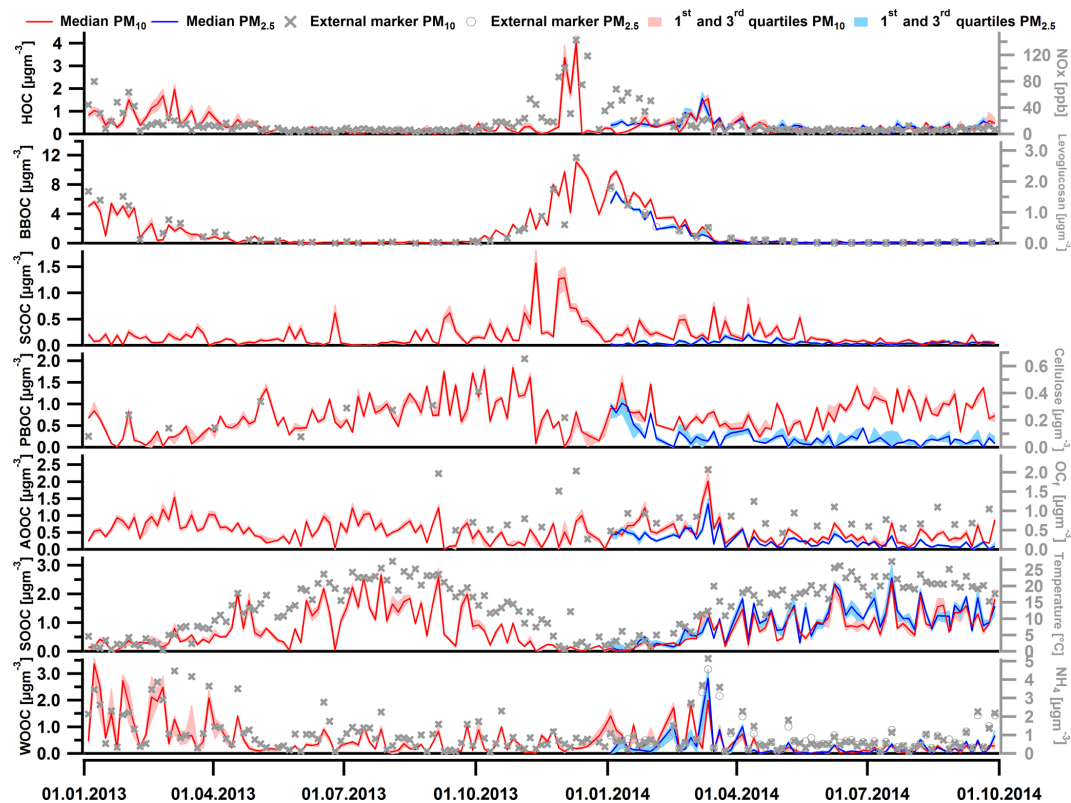
**Figure 5.** Offline AMS/PMF (ME-2) factor profiles: hydrocarbon-like OA (HOA), biomass burning OA (BBOA), sulfur-containing OA (SCOA), primary biological OA (PBOA), anthropogenic oxygenated OA (AOOA), summer oxygenated OA (SOOA) and winter oxygenated OA (WOOA).

present in the fine mode, consistent with previous observations (Levin et al., 2010b). The BBOA/levoglucosan ratio was 7.1 for  $PM_{10}$  and 5.8 for  $PM_{2.5}$ , which falls into the range reported by Daellenbach et al. (2017) and was also consistent with the ratio reported by Bozzetti et al. (2016). The difference of BBOA/levoglucosan for the two size fractions is due to four samples in BBOA  $PM_{10}$  with high concentrations. Lastly, BBOC showed a strong correlation with  $EC_{nf}$ , with a slope of 4.9 (Fig. 7b) which fell within the range of the compiled  $EC_{nf}$ /BBOC ratios in Ulevicius et al. (2016).

Sulfur-containing OA (SCOA) was identified by its spectral fingerprint which is described by a high contribution of the fragment  $CH_3SO_2^+$  (at  $m/z$  79) (Fig. 5) and high OM/OC ratio (Table 2). The  $R_{SCOA}$  (Fig. 4, Table 2) showed a much broader distribution than the rest of the primary OC recoveries yet more limited towards the strongly water-soluble fractions compared to Daellenbach et al. (2017). SCOA concentrations were higher in the coarse fraction compared to  $PM_{2.5}$  (Figs. 6 and 7c, S5) and exhibited higher concentrations during autumn and winter compared to summer (Table 3). A similar profile had previously been linked to a marine origin by Crippa et al. (2013) in Paris; however, Daellenbach et al. (2017) found that SCOA in Switzerland was rather a primary locally emitted source with no marine origin due to its anti-correlation with methanesulfonic acid (MSA). Here we confirm that SCOA did not follow the MSA time series (Fig. S11) but rather the time series of  $NO_x$ . These observations suggest that this factor is connected to a primary coarse particle episodic source related to traffic.

Primary biological OA (PBOA) exhibited significant contributions from the fragment  $C_2H_5O_2^+$  (part of  $m/z$  61) (Fig. 5) and was more enhanced in summer and spring (Fig. 6). The  $R_{PBOA}$  (Fig. 4, Table 2) met the high end of  $R_{PBOA}$  in Bozzetti et al. (2016). PBOA appeared mostly in the coarse mode (Table 3, Fig. S5). The mass spectral features, the seasonality and coarse contribution suggested the biological nature of this factor possibly including plant debris. Additional support of this interpretation is provided by the correlation of PBOA with cellulose (Fig. 7d), a polymer mostly found in the cell wall of plants. The correlation improved if only data from summer and spring were considered. The outliers here were the late autumn and winter points when BBOA was more important and PBOA could not as easily be separated by the PMF technique.

One out of the three oxygenated OAs (OOA) was identified as a highly oxidized factor, due to the significant contribution of the fragment  $CO_2^+$  (Fig. 5) and the high OM/OC ratio (Table 2) which was consistent with the reported OM/OC ratio by Turpin et al. (2001) for non-urban aerosols. This factor peaked mainly in winter and spring and the  $PM_{2.5}$  size fraction exhibited higher concentrations during this period compared to the coarse size fraction (Table 3, Fig. 6). The water solubility of this oxygenated factor was high (Fig. 4, Table 2), which is consistent with the literature values (Daellenbach et al., 2016, 2017) that refer to the sum of all oxygenated factors, as well as with reported water-soluble fractions for highly oxidized compounds (Xu et al., 2017). The yearly median concentration for  $PM_{10}$  was  $0.97 \mu g m^{-3}$  ( $Q_{25} = 0.86$  and  $Q_{75} =$



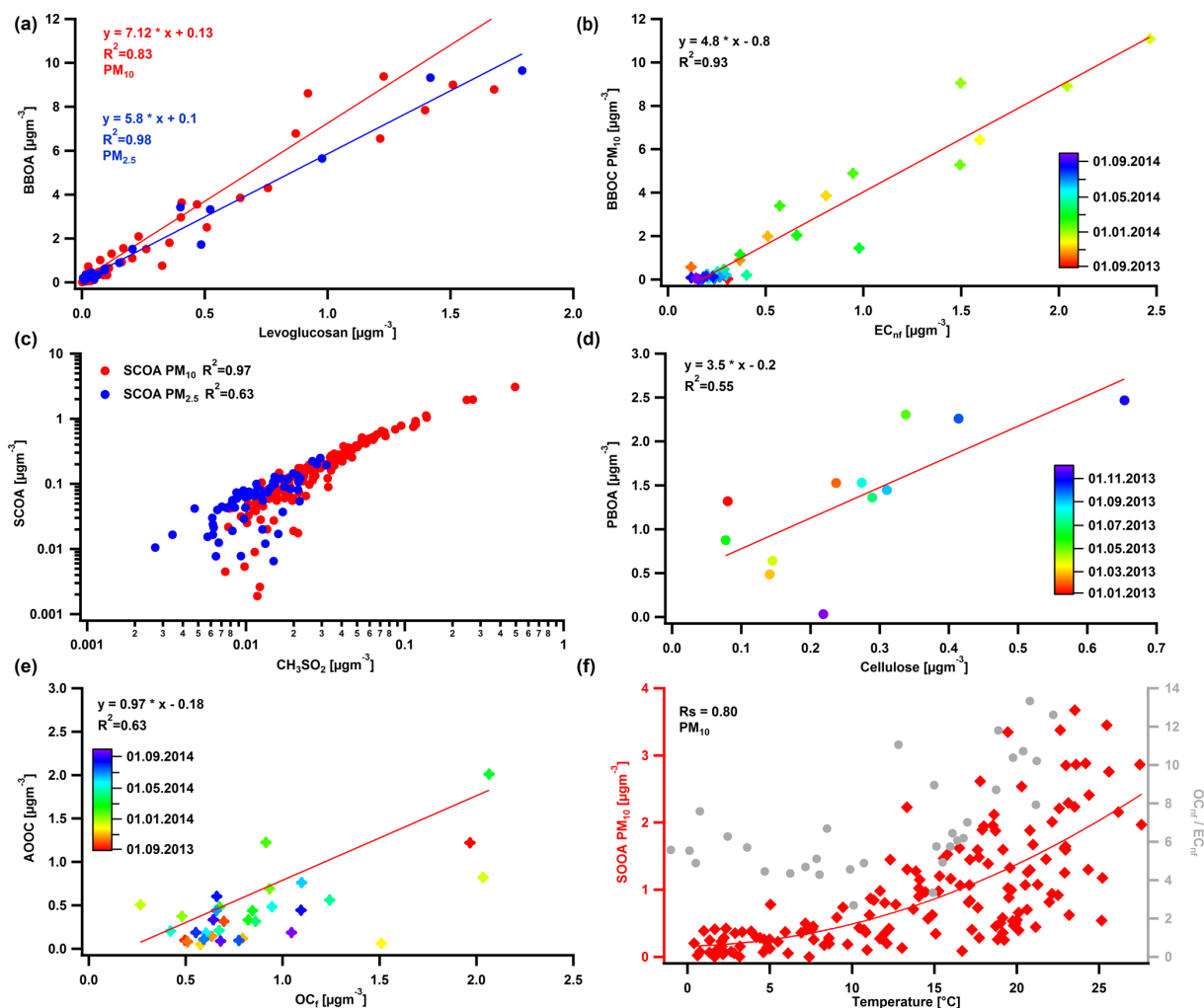
**Figure 6.** Factor (in red for  $\text{PM}_{10}$  and blue for  $\text{PM}_{2.5}$ ) and external marker (in grey markers) time series for the two size fractions: HOC and  $\text{NO}_x$ , BBOC and levoglucosan, SCOC, PBOC and cellulose, AOOC and  $\text{OC}_f$ , SOOC and temperature, and WOOC and  $\text{NH}_4^+$ . Note that here, different from Fig. 5, the factors are quantified according to their carbon mass concentration, with HOC, BBOC, SCOC, PBOC, AOOC, SOOC, and WOOC referring to hydrocarbon-like organic carbon (OC), biomass burning OC, sulfur-containing OC, primary biological OC, anthropogenic oxygenated OC, summer oxygenated OC, and winter oxygenated OC, respectively.

$1.09 \mu\text{g m}^{-3}$ ), which accounts for approximately 13 % of the total OA. Out of all the possible correlations with external markers, this factor correlated best with  $\text{OC}_f$  (Fig. 7e); therefore, we chose to name it anthropogenic OOA (AOOA) (see also discussion in Sect. 4.4.2). Both AOOC and  $\text{OC}_f$  followed very similar annual cycles (Fig. S12) with average  $\text{AOOC} / \text{OC}_f = 0.97 \pm 2.49$ . This observation along with the increase in  $\text{OC}_f / \text{EC}_f$  as already discussed in Sect. 4.2 could indicate that this factor is linked to secondary organic aerosol from traffic emissions or to transported air masses from industrialized areas. It may also be connected to the oxidation of volatile chemical products such as pesticides, coatings, printing inks or cleaning agents (McDonald et al., 2018). Further discussion about AOOC can be found in Sect. 4.4.

Summer oxygenated OA (SOOA) was mainly identified by the high contribution of the fragment  $\text{C}_2\text{H}_3\text{O}^+$  ( $m/z$  43) (Fig. 5) ( $f_{\text{C}_2\text{H}_3\text{O}^+} = 0.15$ ) as well as its seasonal behaviour (Fig. 6). Like all the oxygenated OA factors, it was highly water soluble (Fig. 4, Table 2). The highest concentrations were witnessed in July with values of  $4.4 \mu\text{g m}^{-3}$  for  $\text{PM}_{10}$  in 2013 and  $4.3 \mu\text{g m}^{-3}$  for  $\text{PM}_{2.5}$  in 2014. The bulk contribution of this factor was present in the  $\text{PM}_{2.5}$  fraction (Table 3,

Fig. S5). The seasonal variability of SOOA followed the daily temperature average (Fig. 6). In fact, SOOA exponentially increased with temperature (Fig. 7f). Such behaviour was also observed in Daellenbach et al. (2017), where they connected this factor to the oxidation of terpene emissions and therefore to biogenic SOA formation. The exponential dependence of SOOA with temperature was also similar to the temperature dependence of the biogenic SOA concentrations from a Canadian terpene-rich forest, reported by Leitch et al. (2011). A similar factor was identified with an online instrument in Zurich during summer 2011, where the semi-volatile OOA was mainly formed by biogenic sources as the high temperatures favour the biogenic emissions compared to the rest (Canonaco et al., 2015). Finally, the O/C ratio (0.37) fell into the range of the reported O/C ratios measured by chamber-generated SOA (Aiken et al., 2008), which was similar to biogenic SOA produced in flow tubes (Heaton et al., 2007).

Named after its seasonal behaviour (Daellenbach et al., 2017), the third oxygenated factor, winter oxygenated OA (WOOA), exhibited the highest concentrations during winter. WOOA mass spectrum exhibited elevated contributions of



**Figure 7.** Correlations between BBOA and levoglucosan for the two size fractions (a), BBOC and EC<sub>nf</sub> for PM<sub>10</sub> (b), SCOA and CH<sub>3</sub>SO<sub>2</sub><sup>+</sup> for the two size fractions (c) (the regression lines show a linear relationship), PBOA and cellulose for PM<sub>10</sub> (d), AOOC and OC<sub>f</sub> (the regression fit was weighted by the standard deviation of AOOC) (e), and SOOA and daily averaged temperature as well as OC<sub>nf</sub>/EC<sub>nf</sub> ratio and temperature for PM<sub>10</sub> (f).

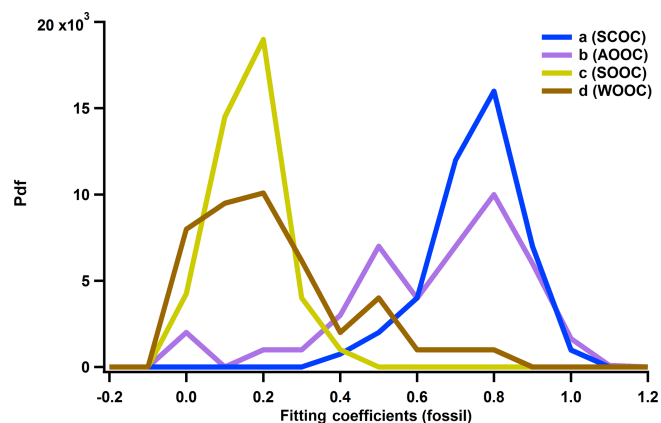
the fragment C<sub>2</sub>H<sub>3</sub>O<sup>+</sup> (Fig. 5), but lower compared to SOOA (for WOOA  $f_{\text{C}_2\text{H}_3\text{O}^+} = 0.11$ ). It also exhibited a slightly enhanced contribution of the fragment C<sub>2</sub>H<sub>4</sub>O<sub>2</sub><sup>+</sup> which can be an indication that this factor originated from aged biomass burning emissions. Moreover, a similar mass spectral pattern (peaks of fragments C<sub>3</sub>H<sub>3</sub>O<sup>+</sup>, C<sub>3</sub>H<sub>5</sub>O<sub>2</sub><sup>+</sup>, C<sub>4</sub>H<sub>5</sub>O<sub>2</sub><sup>+</sup> and C<sub>5</sub>H<sub>7</sub>O<sub>2</sub><sup>+</sup> at  $m/z$  55, 73, 85 and 99, respectively) to the one coming from oxygenated products from a wood-burning experiment (Bruns et al., 2015) was found. The recovery of this factor manifested high values (Table 2) and consisted mainly of fine-mode particles (Fig. S5). WOOA also correlated with NH<sub>4</sub><sup>+</sup> (Fig. S13), which is directly connected to the inorganic secondary ions NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup>.

#### 4.4 Coupling of offline AMS and <sup>14</sup>C analyses

In this section of the paper we will show the combined results of AMS/PMF and radiocarbon analyses. The first part will elaborate on the technical aspect of the analysis by presenting the calculation of the contribution of each factor to the fossil OC. In the second part, a thorough description of each fossil and non-fossil major source will be given. The time series of each fossil and non-fossil fraction for the whole AMS dataset is illustrated in Fig. 10. Contributions of the primary and secondary OC to the total OC will be also discussed and shown in Fig. 11.

**Table 3.** Season-wise average ( $\pm 1$  standard deviation) concentrations (in  $\mu\text{g m}^{-3}$ ) of different OA factors per size fraction. Note that for the 2 different years the months per season can vary.

$\mu\text{g m}^{-3}$	Autumn				Winter				Spring				Summer			
	2013 (Sep, Oct, Nov)		2014 (Sep)		2013 (Jan, Feb, Dec)		2014 (Jan, Feb)		2013 (Mar, Apr, May)		2014 (Mar, Apr, May)		2013 (Jun, Jul, Aug)		2014 (Jun, Jul, Aug)	
	PM <sub>10</sub>	PM <sub>2.5</sub>	PM <sub>10</sub>	PM <sub>2.5</sub>	PM <sub>10</sub>	PM <sub>2.5</sub>	PM <sub>10</sub>	PM <sub>2.5</sub>	PM <sub>10</sub>	PM <sub>2.5</sub>	PM <sub>10</sub>	PM <sub>2.5</sub>	PM <sub>10</sub>	PM <sub>2.5</sub>	PM <sub>10</sub>	PM <sub>2.5</sub>
HOA	0.23 ± 0.20	0.44 ± 0.19	0.46 ± 0.20	0.44 ± 0.19	1.38 ± 1.35	0.45 ± 0.36	0.66 ± 0.26	0.67 ± 0.55	0.51 ± 0.52	0.54 ± 0.44	0.51 ± 0.52	0.12 ± 0.13	0.27 ± 0.14	0.26 ± 0.19	0.27 ± 0.14	0.26 ± 0.19
BBOA	3.09 ± 3.78	0.21 ± 0.18	0.21 ± 0.22	0.21 ± 0.18	8.32 ± 5.58	9.46 ± 4.65	6.76 ± 3.19	1.35 ± 1.20	0.66 ± 0.95	0.50 ± 0.54	0.66 ± 0.95	0.14 ± 0.11	0.21 ± 0.11	0.19 ± 0.08	0.21 ± 0.11	0.19 ± 0.08
SCOA	0.61 ± 0.68	0.08 ± 0.03	0.13 ± 0.11	0.08 ± 0.03	0.48 ± 0.59	0.47 ± 0.26	0.06 ± 0.06	0.23 ± 0.18	0.46 ± 0.37	0.14 ± 0.09	0.46 ± 0.37	0.17 ± 0.27	0.12 ± 0.06	0.06 ± 0.04	0.12 ± 0.06	0.06 ± 0.04
PBOA	2.04 ± 0.96	0.24 ± 0.14	1.82 ± 0.75	0.24 ± 0.14	0.66 ± 0.57	1.60 ± 0.68	1.01 ± 0.72	1.02 ± 0.58	1.12 ± 0.45	0.38 ± 0.25	1.12 ± 0.45	1.63 ± 0.64	1.99 ± 0.51	0.31 ± 0.22	1.99 ± 0.51	0.31 ± 0.22
AOOA	0.79 ± 0.68	0.22 ± 0.19	0.89 ± 0.63	0.22 ± 0.19	1.35 ± 0.69	1.37 ± 0.53	0.98 ± 0.27	1.38 ± 0.70	1.06 ± 1.01	0.66 ± 0.66	1.06 ± 1.01	0.70 ± 0.65	0.75 ± 0.44	0.36 ± 0.27	0.75 ± 0.44	0.36 ± 0.27
SOOA	1.16 ± 1.13	2.22 ± 0.78	2.05 ± 0.67	2.22 ± 0.78	0.29 ± 0.21	0.31 ± 0.25	0.43 ± 0.38	1.05 ± 0.77	1.44 ± 0.70	1.59 ± 0.76	1.44 ± 0.70	2.41 ± 1.08	1.97 ± 0.90	2.37 ± 0.82	1.97 ± 0.90	2.37 ± 0.82
WOOA	0.64 ± 0.62	0.49 ± 0.47	0.33 ± 0.30	0.49 ± 0.47	2.02 ± 1.75	1.27 ± 1.03	0.49 ± 0.49	1.98 ± 0.88	0.49 ± 0.79	0.59 ± 1.17	0.49 ± 0.79	0.59 ± 0.51	0.21 ± 0.19	0.23 ± 0.22	0.21 ± 0.19	0.23 ± 0.22



**Figure 8.** Probability density functions of the fitting coefficients of the relative fossil contributions: SCOC in blue, AOOC in purple, SOOC in yellow and WOOC in light brown.

#### 4.4.1 Calculation of fossil and non-fossil fraction per factor

To combine the AMS/PMF with the  $^{14}\text{C}$  results, the identified sources from AMS/PMF were divided into fossil and non-fossil fractions. HOC was fully assigned to fossil sources assuming that the percentage of biofuel content is negligible. BBOC and PBOC were considered totally non-fossil. To explore the fossil and non-fossil nature of the rest of the factors, we performed multilinear regression using Eq. (17):

$$\text{OC}_{f,i} - \text{HOC}_i = a \cdot \text{SCOC}_i + b \cdot \text{AOOC}_i + c \cdot \text{SOOC}_i + d \cdot \text{WOOC}_i, \quad (17)$$

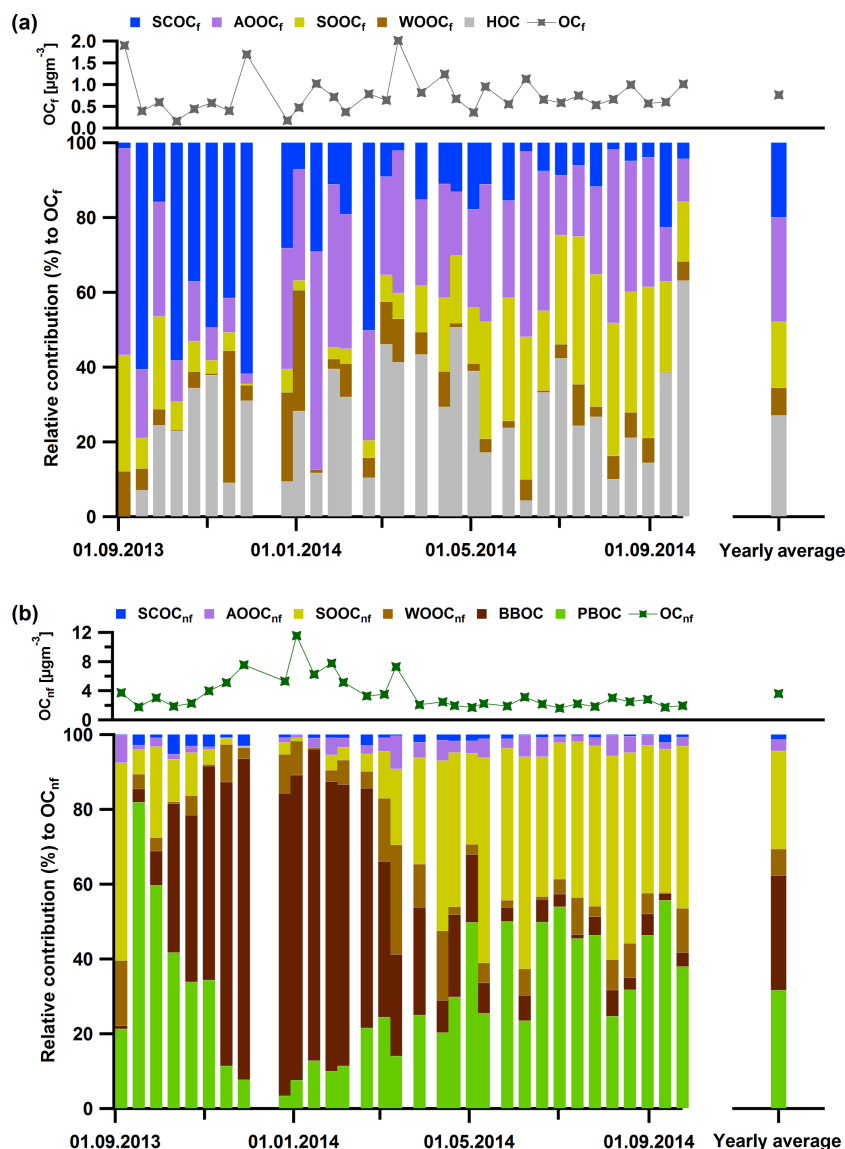
where  $a$ ,  $b$ ,  $c$  and  $d$  are the fitting coefficients, weighted by the relative uncertainty of  $\text{OC}_{f,i} - \text{HOC}_i$ . To investigate the stability of the solution, we obtained distributions of the fitting coefficients by performing 100 bootstrap runs where input data were randomly selected (Fig. 8). The median values (and first and third quartiles) were as follows:  $a = 0.81$  ( $Q_{25} = 0.73$ ,  $Q_{75} = 0.88$ ),  $b = 0.77$  ( $Q_{25} = 0.54$ ,  $Q_{75} = 0.85$ ),  $c = 0.21$  ( $Q_{25} = 0.15$ ,  $Q_{75} = 0.26$ ) and  $d = 0.23$  ( $Q_{25} = 0.13$ ,  $Q_{75} = 0.39$ ).

We chose to apply the multilinear regression to the fossil fraction because for the non-fossil part, the errors related to fitting coefficients were very high and the dependences of the  $\text{OC}_{\text{nf}}$  on the input factors were not statistically significant ( $p$  values  $> 0.1$ ).

To calculate the non-fossil part of each factor  $k$  ( $k\text{OC}_{\text{nf}}$ ), we used the following equation:

$$k\text{OC}_{\text{nf},i} = k\text{OC}_i - k\text{OC}_{f,i}. \quad (18)$$

This analysis suggests that the major fossil primary sources were HOC and SCOC ( $81\% \pm 11\%$  fossil), while AOOC ( $77\% \pm 23\%$  fossil) was the only major fossil secondary source. In terms of the non-fossil sources, the



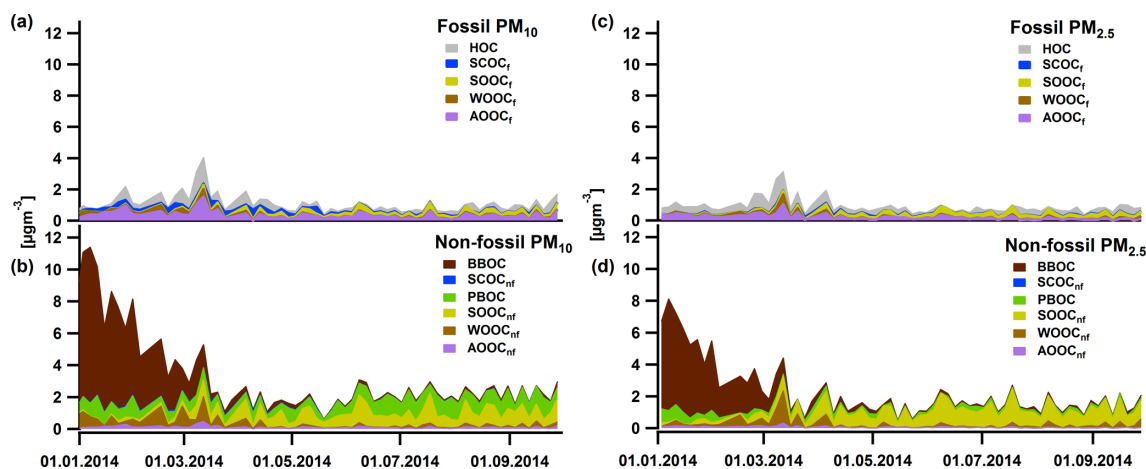
**Figure 9.** Relative contributions to the fossil OC per factor ( $\text{PM}_{10}$ ) (a) and to the non-fossil OC per factor ( $\text{PM}_{10}$ ) (b): BBOC in dark brown,  $\text{SCOC}_f$  and  $\text{SCOC}_{nf}$  in blue, PBOC in green,  $\text{AOOC}_f$  and  $\text{AOOC}_{nf}$  in purple,  $\text{SOOC}_f$  and  $\text{SOOC}_{nf}$  in yellow, and  $\text{WOOC}_f$  and  $\text{WOOC}_{nf}$  in light brown. Note that the total non-fossil concentrations (dark green markers) are on average 6 times higher compared to the fossil ones (dark grey markers).

dominating primary sources included BBOC and PBOC, whereas the most important secondary sources were SOOC ( $79\% \pm 11\%$  non-fossil) and WOOC ( $77\% \pm 23\%$  non-fossil).

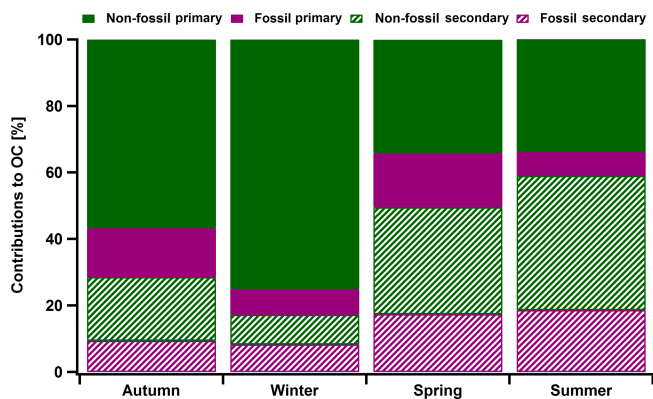
#### 4.4.2 Contribution of fossil and non-fossil, primary and secondary OC to the total OC

The results point out that  $81\% \pm 11\%$  (average and 1 standard deviation) of SCOC was fossil ( $\text{SCOC}_f$ ). Taking into account the enhanced contribution of SCOC in the coarse size fraction, its sulfur content and its fossil nature, we assume that this factor is linked to primary anthropogenic sources re-

lated to traffic, such as tire wear, resuspension of road dust (Bukowiecki et al., 2010), resuspension from asphalt concrete (Gehrig et al., 2010) or asphalt mixture abrasion (in bituminous binder, Fullova et al., 2017). The contribution of  $\text{SCOC}_f$  to the  $\text{OC}_f$  was more important during autumn and winter (up to 62%, Fig. 9a) in contrast to spring and summer (on average  $9\% \pm 5\%$ ), while on average the contribution to the  $\text{OC}_f$  was  $20\% \pm 19\%$ . The concentrations in winter and autumn were similar and on average for  $\text{PM}_{10}$  ( $\text{PM}_{2.5}$ )  $0.22 \pm 0.21 \mu\text{g m}^{-3}$  ( $0.03 \pm 0.03 \mu\text{g m}^{-3}$ ) (Fig. 10, Table S2), which accounted for 73% of the total SCOC for this period.



**Figure 10.** Yearly cycles of fossil PM<sub>10</sub> (a), non-fossil PM<sub>10</sub> (b), fossil PM<sub>2.5</sub> (c), and non-fossil PM<sub>2.5</sub> (d) OC factors: BBOC in dark brown, SCOC<sub>f</sub> and SCOC<sub>nf</sub> in blue, PBOC in green, AOOC<sub>f</sub> and AOOC<sub>nf</sub> in purple, SOOC<sub>f</sub> and SOOC<sub>nf</sub> in yellow, and WOOC<sub>f</sub> and WOOC<sub>nf</sub> in light brown. Note that the covered time periods in (a), (b) and (c), (d) are different.



**Figure 11.** Averaged contributions of the fossil and non-fossil primary and secondary OC to the total OC season wise for PM<sub>10</sub>.

However, the contribution of SCOC<sub>f</sub> to the total OC for the coarse size fraction was not high ( $5\% \pm 8\%$  on average).

The combined <sup>14</sup>C / AMS analysis supported the initial hypothesis that AOOC was mainly related to the oxidation of fossil fuel combustion emissions (e.g. traffic), as AOOC was  $77\% \pm 23\%$  fossil (AOOC<sub>f</sub>) on average. The average contribution of AOOC<sub>f</sub> to the OC<sub>f</sub> was  $28\% \pm 14\%$  (Fig. 9a), larger than SCOC<sub>f</sub>, while its contribution to the total OC was  $10\% \pm 5\%$  for the coarse OC and  $7\% \pm 7\%$  of the fine OC. The yearly cycle exhibited elevated contributions in winter and spring compared to summer and autumn with average values for PM<sub>10</sub>:  $0.47 \pm 0.22$ ,  $0.43 \pm 0.30$ ,  $0.39 \pm 0.23$  and  $0.29 \pm 0.23 \mu\text{g m}^{-3}$ , respectively (Fig. 10, Table S2). In winter and spring most of the mass concentration came from the PM<sub>2.5</sub> size range in contrast to the other two seasons.

The fossil fractions of SOOC (SOOC<sub>f</sub>) and WOOC (WOOC<sub>f</sub>) were low (21 and 23 %, respectively) and could also be attributed to traffic emissions or less likely (due to

low emissions) to aged aerosols from residential fossil fuel heating. SOOC<sub>f</sub> was important during summer with contributions up to 40 % to the OC<sub>f</sub> and WOOC<sub>f</sub> was more distinctively present during a few days in autumn and winter (up to 35 % to the OC<sub>f</sub>) in contrast to the rest of the year (Fig. 9a).

From the non-fossil sources, apart from non-fossil SCOC (SCOC<sub>nf</sub>) and non-fossil AOOC (AOOC<sub>nf</sub>), the rest of the factors exhibited a very distinct yearly cycle with BBOC contributing up to 86 % to the OC<sub>nf</sub> in late autumn and winter (Fig. 9b, yearly average  $28\% \pm 30\%$ ) and with PBOC and SOOC<sub>nf</sub> becoming more important in late spring, summer and early autumn with contributions up to 82 and 57 %, respectively (Fig. 9b).

SOOC was 79 % non-fossil which supported the AMS/PMF results: the significance of non-fossil SOOC (SOOC<sub>nf</sub>) during summer can be attributed to SOA formation from biogenic emissions. The average contribution of SOOC<sub>nf</sub> to OC<sub>nf</sub> was  $25\% \pm 19\%$  (Fig. 9b). SOOC<sub>nf</sub> was more pronounced in PM<sub>2.5</sub> (on average  $1.12 \pm 0.40 \mu\text{g m}^{-3}$  in summer and  $0.75 \pm 0.35 \mu\text{g m}^{-3}$  in spring, Fig. 10, Table S2). This factor along with PBOC was the main and almost equally important source of OC during spring and summer, with PBOC contributing to OC in the coarse mode (on average  $35\% \pm 16\%$  from April to August 2014) and SOOC<sub>nf</sub> in the fine mode ( $46\% \pm 15\%$  from April to August 2014). PBOC made up  $30\% \pm 18\%$  of the OC<sub>nf</sub> and the average concentrations of PBOC<sub>coarse</sub> for 2014 were  $1.00 \pm 0.23 \mu\text{g m}^{-3}$  in summer and  $0.56 \pm 0.21 \mu\text{g m}^{-3}$  in spring.

Non-fossil WOOC (WOOC<sub>nf</sub>) dominated over WOOC<sub>f</sub> (77 % over 23 %). The average yearly contribution to OC<sub>nf</sub> was low ( $6\% \pm 6\%$ , Fig. 9b); however, WOOC<sub>nf,coarse</sub> was apparent during the cold period especially in 2013 with concentrations of  $0.88 \pm 0.74 \mu\text{g m}^{-3}$  on average for winter ( $0.28 \pm 0.28 \mu\text{g m}^{-3}$  for autumn) (Fig. 10). In



2014 the concentrations dropped for winter (autumn) with  $0.53 \pm 0.43 \mu\text{g m}^{-3}$  ( $0.15 \pm 0.13 \mu\text{g m}^{-3}$ ) for  $\text{PM}_{10}$  and  $0.22 \pm 0.19 \mu\text{g m}^{-3}$  ( $0.21 \pm 0.21 \mu\text{g m}^{-3}$ ) for  $\text{PM}_{2.5}$ . Based on its yearly cycle (Fig. 10b and d)  $\text{WOOC}_{\text{nf}}$  could be linked to aged OA influenced by wintertime and early spring biomass burning emissions. Therefore, not only  $\text{AOOC}_{\text{f}}$  but also  $\text{WOOC}_{\text{nf}}$  can be related to anthropogenic activities. In other studies (Daellenbach et al., 2017; Bozzetti et al., 2016) this factor was more pronounced; however, in our case in winter most of the  $\text{OC}_{\text{nf}}$  was related to primary biomass burning.

Overall for  $\text{PM}_{10}$  the non-fossil primary OC contributions were more important during autumn (57 %) and winter (75 %), whereas in spring and summer the non-fossil secondary OC contributions became more pronounced (32 and 40 %, respectively) (Fig. 11). The dominance of the SOC during the warm period is likely related to the stronger solar radiation which favours the photo-oxidation of biogenic volatile organic compounds and to the elevated biogenic volatile organic compounds emissions.

## 5 Conclusions

The coupling of offline AMS and  $^{14}\text{C}$  analyses allowed a detailed characterization of the carbonaceous aerosol in the Alpine valley of Magadino for the years 2013–2014. The seasonal variation along with the two size-segregated measurements ( $\text{PM}_{10}$  and  $\text{PM}_{2.5}$ ) gave insights into the source apportionment, by for example quantifying the resuspension of road dust or asphalt concrete and estimating its contribution to the OC or by identifying SOC based on SOC precursors. More specifically, seven sources including four primary and three secondary ones were identified. The non-fossil primary sources were dominating during autumn and winter, with BBOC exhibiting by far the highest concentrations. During spring and summer again two non-fossil sources, PBOC in the coarse fraction and  $\text{SOOC}_{\text{nf}}$  in the fine mode, prevailed over the fossil ones. The size-segregated measurements and  $^{14}\text{C}$  analysis enabled a better understanding of the primary SCOC factor, which was enhanced in the coarse fraction and was mainly fossil, suggesting that it may originate from resuspension of road dust or tire – asphalt abrasion. The results also showed that SOC was formed mainly by biogenic sources during summer and anthropogenic sources during winter. However, SOC formed possibly by oxidation of traffic emissions or volatile chemical products was also apparent during summer ( $\text{AOOC}_{\text{f}}$ ).  $\text{AOOC}_{\text{f}}$  was also important during winter along with SOC linked to transported non-fossil carbonaceous aerosols coming from anthropogenic activities such as biomass burning ( $\text{WOOC}_{\text{nf}}$ ).

*Data availability.* The data are available upon request from the corresponding author.

**The Supplement related to this article is available online at <https://doi.org/10.5194/acp-18-6187-2018-supplement>.**

*Competing interests.* The authors declare that they have no conflict of interest.

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