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Benigni P., J. Rogez. High temperature thermal diffusivity measurement by the periodic cylindrical method: The problem of contact thermocouple thermometry. Review of Scientific Instruments, 1997, 68 (7), pp.2767 - 2773. 10.1063/1.1148193 . hal-01788840

HAL Id: hal-01788840 https://hal.science/hal-01788840v1

Submitted on 9 May 2018

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Citation: Review of Scientific Instruments **68**, 2767 (1997); doi: 10.1063/1.1148193 View online: http://dx.doi.org/10.1063/1.1148193 View Table of Contents: http://scitation.aip.org/content/aip/journal/rsi/68/7?ver=pdfcov Published by the AIP Publishing

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High temperature thermal diffusivity measurement by the periodic cylindrical method: The problem of contact thermocouple thermometry

P. Benigni and J. Rogez^{a)} Centre de Thermodynamique et Microcalorimétrie du C.N.R.S., 26 Rue du 141 ème R.I.A., 13003 Marseille, France

(Received 25 November 1996; accepted for publication 3 March 1997)

The conditions required for the accurate measurement of the thermal diffusivity of solids at high temperature (800-1800 K) by the Angström method with cylindrical geometry are studied. It is shown that the main error in the measurement arises from the difficulty in measuring the temperature accurately with contact thermocouples. On the basis of new experimental results on Cecorite and polycrystalline alumina, the effect of the finite size of the sensors and the consequence of the imperfect thermal contact between the sample and the sensors are investigated. Two models which allow a greater insight into the simultaneous influences of both effects are developed. © 1997 American Institute of Physics. [S0034-6748(97)03006-2]

I. INTRODUCTION

An apparatus for thermal diffusivity measurements of refractory solids at high temperature (800-1800 K) has been recently built. The method and the description of the experimental setup are published elsewhere¹ with the first measurements on a cordierite based ceramic, the Cecorite 130 P.

The principle of the apparatus is the classical Angström method adapted to cylindrical geometry. In this method, a thermal wave of angular velocity ω , is generated over the lateral side of the cylindrical sample, the amplitude and the phase of which vary continuously all along the radius. The two measurement points are located at the center of the cylinder and at a distance *d* from it. The solution of the heat transfer equation relates the thermal diffusivity *K*, through the dimensionless group $(u = \sqrt{\omega/Kd})$, to the phase change Φ and the amplitude ratio Θ :

$$\Phi(u) = \tan^{-1} \left(\frac{\operatorname{bei} (u)}{\operatorname{ber} (u)} \right), \tag{1}$$

$$\Theta(u) = \frac{1}{\sqrt{\operatorname{ber}^2(u) + \operatorname{bei}^2(u)}},$$
(2)

where ber and bei are the Kelvin functions.

Our primary results on the Cecorite 130 P (Ref. 1) approached but were not in sufficiently close agreement with those of the CODATA program;² moreover, some difficulties remained:

- (1) the measurements have been found to be dependent on the frequency of the thermal signal; and
- (2) a significant difference between the K_{Θ} and K_{Φ} values has been observed.

It is well known that the accuracy of temperature measurements using contact thermocouples is limited by the finite size of the sensors and the imperfect thermal contact between the sensors and the sample. In the field of the periodic method, these effects can be the most significant ones on the measurement uncertainties.³ Within the frame of our method, the aim of this article is to quantify these sources of error and to check if they can account for the abovementioned phenomena. The study is based on new experimental results and refinement in the mathematical model of the measurement process.

II. EFFECT OF THE FINITE SIZE OF THE SENSORS

The distance between the axes of the thermocouples' holes can be accurately estimated within ± 0.1 mm, inducing a systematic but moderate error on the diffusivity measurement (e.g., $\approx 1.3\%$ if d=15 mm). This kind of error analysis is the most commonly used in the description of experimental devices from the literature.⁴ In the model of data processing, *d* represents the distance between the points where the temperature is effectively measured.

The thermocouples are constructed by inserting Pt and Pt-Rh 10% wires (\emptyset =0.3 mm) in alumina twin-bore tubes (outer \emptyset =1.7 mm). The wires are welded with an oxyacetylene torch, to form a small bead. The use of exposed hot junctions allows the sensors to have very low thermal capacity. The thermocouples are inserted in holes of radius (r_h = 1 mm) drilled in the sample. As a consequence, the location of the contact point between the bead of a thermocouple and the sample is not exactly known. Let r be the radial position of a thermocouple hole, this thermocouple will measure any value of temperature between ($r-r_h$) and (r+ r_h).⁵ Moreover, the size and shape of the bead can vary from one thermocouple to another and the conditions of contact are not controlled. This last point will be the purpose of the next paragraph.

Hence, a knowledge of the distance *d* is limited by the finite size of the sensors. This geometric effect can be represented in a simple model by a displacement parameter ϵ of the thermocouples around their expected locations. The magnitude of ϵ is obviously restricted to the radius of the holes in which the thermocouples are inserted. At the center of the sample, due to cylindrical symmetry, the inner thermocouple lies between 0 and r_h , yielding a decrease of *d*. The location of the outer thermocouple varies within the range $(d - r_h) - (d + r_h)$. For the usual experimental setups the radius

Rev. Sci. Instrum. 68 (7), July 1997

0034-6748/97/68(7)/2767/7/\$10.00

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a)Electronic mail: jrz@ctm.cnrs-mrs.fr

TABLE I. Diffusivity of Cecorite 130 P obtained with d=9.3 mm. Each value is an average over at least four measurements.

	Period/s										
T/K	14	145		252		333		490		646	
	а	b	а	b	а	b	а	b	а	b	
861 ^c	6.85	6.62	6.38	6.55	6.01	6.52	5.39	6.45	4.92	6.35	
1231 ^d	6.73	6.28	6.66	6.32	6.54	6.34	6.03	6.27	5.85	6.24	

 ${}^{a}K_{\Theta} \times 10^{3}/\text{cm}^{2} \text{ s}^{-1}$.

 ${}^{b}K_{\Phi} \times 10^{3} / \text{cm}^{2} \text{ s}^{-1}$.

 ${}^{c}K_{CODATA} \times 10^{3} = 5.46 \text{ cm}^{2} \text{ s}^{-1}.$

 ${}^{d}K_{CODATA} \times 10^{3} = 5.28 \text{ cm}^{2} \text{ s}^{-1}.$

 $(r_h = 1 \text{ mm})$, is nonnegligible with respect to the distance d and thus ϵ may have a strong effect on the results. The resulting absolute error on d is limited to the range -2 to +1 mm corresponding to a relative error on K of -13% to +26% for d=15 mm.

Some new measurements were performed on the same Cecorite sample by decreasing the distance between the thermocouples to d=9.3 mm. Such a procedure allows us to check the coherence of the results in another way. These new measurements are presented in Table I together with the recommended values of the CODATA program² at the corresponding temperatures. These new results lie outside the range of the main scattering of $\pm 10\%$ around the fit. A new approach was intended to increase our understanding of these experimental results.

The $\Theta(u)$ and $\Phi(u)$ charts are convenient for the discussion as they allow in a single representation, a comparison between the Θ and Φ measurements obtained under various experimental conditions (different temperatures and



FIG. 1. Cecorite sample: +: ideal curves [Eqs. (1) and (2)], \bigcirc : d = 13.2 mm, large circles: experimental results, small circles: calculated values with Model I where $\epsilon_1 = \epsilon_2 = 0.85$ mm, \triangle : d = 9.3 mm, large triangles: experimental results, small triangles: calculated values with Model I where $\epsilon_1 = \epsilon_2 = 1$ mm.

hence different diffusivity values, distances and frequencies).

Considering that the diffusivity value from the CODATA fit is the true value, each Θ and Φ measurement can be plotted versus its true *u* quantity which is calculated from the diffusivity data issued from the CODATA fit at this temperature, and the ω and *d* values. Such experimental plots are compared to the ideal ones [Eqs. (1) and (2)] in Fig. 1. A systematic shift between experimental and ideal curves, the magnitude of which depends on *d*, is observed, showing a systematic error.

In Fig. 1 are also plotted the results of the following model, which can simply account for this effect. This model gives the phase change and the amplitude ratio between the points $(r = \epsilon_1)$ and $(r = d - \epsilon_2)$ in the infinite cylinder.¹

$$\Theta_{\epsilon_1/d-\epsilon_2} = \frac{\sqrt{\operatorname{ber}^2(e\,\epsilon_1) + \operatorname{bei}^2(e\,\epsilon_1)}}{\sqrt{\operatorname{ber}^2[e(d-\epsilon_2)] + \operatorname{bei}^2[e(d-\epsilon_2)]}},\qquad(4)$$

where *e* is the reciprocal thermal diffusion length ($e = \sqrt{\omega/K}$). Equations (3) and (4) reduce to Eqs. (1) and (2) as $\epsilon_1 = \epsilon_2 = 0$.

Figure 1 shows that the experimental results on Cecorite 130 P for two different values of d, at various frequencies and around two temperatures, are quite well fitted by the model. The Φ results are better fitted than Θ ones. It should be kept in mind that the model is very simple and that such an agreement significantly evidences the finite size effect.

Figure 2 shows the limiting curves which can be calculated by the former model, corresponding to the phase change and the amplitude ratio between the points r=0 and $r=d+r_h$ or $r=r_h$ and $r=d-r_h$ for each value of d. The

comparison between Figs. 1 and 2, which are identically scaled, shows that all the experimental points remain in the range of the limiting curves.

III. IMPERFECT THERMAL CONDUCTANCE BETWEEN THE SENSORS AND THE SAMPLE

The finite thermal conductance at the interface between the sensors and the sample tends to decrease the amplitude ratio Θ and to increase the phase change Φ toward the values expected with a perfect contact. This effect has been experimentally tested, on a dense AL 23 polycrystalline alumina provided by the Degussa-France company. Alumina was chosen because it has been extensively studied, even if experimental data present a great deal of scatter. According to the manufacturer's data, the chemical analysis of our sample is 99.7 wt % Al₂O₃ nominal and 99.5 wt % Al₂O₃ minimal.

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FIG. 2. Cecorite sample: +: ideal curves [Eqs. (1) and (2)], the other curves are calculated with Model I and $r_h = 1 \text{ mm}$: d = 13.2 mm: $\bigcirc: \epsilon_1 = \epsilon_2 = r_h$, •: $\epsilon_1 = 0$ and $\epsilon_2 = -r_h$, d = 9.3 mm: $\triangle: \epsilon_1 = \epsilon_2 = r_h$, $\blacktriangle: \epsilon_1 = 0$ and $\epsilon_2 = -r_h$.

Its density lies between 3.7 and 3.95 $g \text{ cm}^{-3}$, with an average grain size of 20 μ m. The sample is a cylinder of 35 mm diameter and 70 mm height, made up of a stack of disks of various thicknesses. This configuration minimizes the axial heat flow and allows easier drilling of the thermocouples' holes. The location of the outer thermocouple is d = 14.5 mm from the center.

The diffusivity was measured at four temperatures (Table II) under argon and helium atmospheres. The pressure of the inert gas is about 300×10^2 Pa before heating. The approximate pressures at the measurement temperature are indicated in Table II.

Several diffusivity measurements reported in the literature for polycrystalline alumina, the density of which is close to that of our sample, were fitted versus temperature for comparison (Fig. 3). The details of the references are given in Table III. The density of the various samples lies in the range $3.7-4 \text{ g cm}^{-3}$, except for Plummer⁶ (ρ =3.04 g cm⁻³) and Paladino⁷ (ρ =3.6 g cm⁻³). Bonnerot⁸ and Fétiveau⁹ indicate for their samples, densities lower than 3.5 g cm⁻³. However, it is believed that the true densities of their samples are higher because they used the same dense AL 23 alumina as in the present study.

K/cm² s⁻¹ = 0.0103 + 0.189 exp
$$\left(-\frac{T/K}{288}\right)$$
. (5)



FIG. 3. The thermal diffusivity of dense polycrystalline alumina vs temperature. At each temperature, an average of the K_{Θ} and K_{Φ} values, over all the periods, is plotted.

Using the values calculated by Eq. (5) (Table II), and similarly as for the Cecorite results, experimental charts $\Theta(u)$ and $\Phi(u)$ are plotted in Fig. 4. The measurements performed under argon atmosphere can be fitted with a single curve independent of the temperature. The measurements carried out under helium atmosphere can be represented by two distinct curves depending on the temperature. This difference is mainly a temperature effect. Actually, the two measurements under helium are 631 K apart, resulting in very different heat transfer conditions (gas conductivity multiplied by 1.5, radiation heat flux multiplied by 6) whereas the measurements under argon are only 186 K apart.

For all the temperatures, Fig. 4 shows that helium, the thermal conductivity of which is greater than that of argon (i.e., ten times at room temperature and atmospheric pressure¹⁰), gives a lower Φ and higher Θ . Moreover, helium at 1413 K gives a lower Φ and higher Θ than helium at 782 K. All these observations are consistent with an improvement of the interfacial heat transfer when helium is used or when the temperature is raised. However, this simple reasoning does not take into account the fact that the contact conductance is the sum of three contributions acting in parallel, the magnitudes of which need to be estimated. Moreover, the conductivity of the sample plays a role in the contact phenomena. A quantitative analysis of the contact influence is presented in the following using a model which combines the

TABLE II. Diffusivity of AL 23 alumina obtained with d = 14.5 mm. Each value is an average over at least three measurements.

	Period/s															
T/K	67 80		30	113 184		84	4 252		333		491		647			
	а	b	а	b	а	b	а	b	а	b	а	b	а	b	а	b
1095 ^c					1.18	1.24			1.40	1.34	1.44	1.35	1.47	1.34	1.60	1.33
1281 ^d			1.11	1.03			1.13	1.12			1.09	1.12	1.06	1.08	1.11	1.10
782 ^e	2.20	2.27							1.93	2.21			1.47	2.22	1.27	2.20
1413 ^f	1.30	1.26							1.17	1.28			1.03	1.26	.955	1.25
$ \begin{array}{c} \hline & \\ {}^{a}K_{\Theta} \times 10^{2}/\mathrm{cm}^{2} \mathrm{s}^{-1}. \\ {}^{b}K_{\Phi} \times 10^{2}/\mathrm{cm}^{2} \mathrm{s}^{-1}. \\ {}^{c}P_{\mathrm{Ar}} = 480 \times 10^{2} \mathrm{Pa}. \end{array} $								${}^{d}P_{Ar}$ ${}^{e}P_{He}$ ${}^{f}P_{He}$	=510> =400> =530>	<10 ² Pa <10 ² Pa <10 ² Pa	a. a. 1.					

TABLE III. Thermal diffusivity measurements of dense polycrystalline alumina: overview of the samples, the methods, and the experimental conditions used by various authors. In some original papers, numerical values are not explicitly given hence they are deduced from the curves. List of the symbols used in the table; FF-front face, IR-infrared, K-chromel-alumel thermocouple, P-pressure, RF-rear face, S-Pt-Pt-Rh 10% thermocouple, and t-time.

Reference and year	Sample	Method, procedure, and heater	Т, Р	Sensors and range of error
11 (1977)	Sintered ALCOA A-16SG alumina, ρ_{geom} =3.93 g cm ⁻³ , Al ₂ O ₃ 99.5%	Flash, axial heat flow, laser	298–1073 K	K thermocouple attached to the RF
8 (1986)	Degussit AL 23, $\rho_{geom} = 3350 \text{ kg m}^{-3}$, disk: $\emptyset = 10 \text{ mm}$, $e = 1.676 \text{ mm}$	Flash, axial heat flow, Degiovanni method	1183–1833 K	or the sample
12 (1987)	Degussit AL 23+Pt coating, $\rho = 3.916 \text{ g cm}^{-3} - \emptyset_{\text{grains}} = 13.5 \ \mu\text{m},$ Al ₂ O ₃ 99.5% - SiO ₂ < 0.03% - Fe ₂ O ₃ < 0.03%, -CaO < 0.05% - MgO < 0.3%, disk: $e = 2.73 \text{ mm} \cdot e_{\text{D},\text{cm}} = 10 \text{ to } 15 \ \mu\text{m}$	Flash, axial heat flow, $t_{1/2}$ +correction for finite pulse width and heat loss, neodyme glass laser	300-900 K, under vacuum, 10^{-5} -10^{-6} mbar	IR In–Sb detector RF, 3%–5%
13 (1993)	Sintered ALCOA A-16SG alumina ($T_{\text{sintering}}$ =1873 K)+graphite coating, ρ 1873=3.75 g cm ⁻³ , volumetric porosity =0.058, Al ₂ O ₃ 99.79%, disk: \emptyset =12 mm- e =1.2 mm	Flash, axial heat flow, neodyme glass laser 5–95 J, λ =1.067 μ m, \emptyset _{pulse} =16 mm	373-1473 K, under vacuum, Ar, N ₂ , He	IR In–Sb detector RF, λ =5.5 μ m
8 (1961)	Degussit AL 23, $\rho = 3.43 \text{ g cm}^{-3}$, Al ₂ O ₃ >99.5%, slab: $40 \times 40 \times 50 \text{ mm}^{3}$	Sine wave (20 s), axial heat flow, amplitude ratio measurement by a short period galvanometer, lamp heating or contact heating by Joule effect in Pt thin foil ($e = 0.02$ mm)	340–1273 K, under N ₂	K thermocouples (\emptyset =0.2 mm) inserted in the grooved sample, 3%-10%
14 (1963)	Alumina + carbone coating on the FF, $a = 4 \text{ g cm}^{-3}$	Sine wave (20–50 s), axial heat flow, phase change measurement by photographing the face of an oscilloscope, electron gun	1290-1400 K, under vacuum 10^{-5} mm Hg	FF and RF radiometers calibrated against a blackbody, 28%
15 (1972)	$\rho = 3.95 \text{ g cm}^{-3},$ $\emptyset_{\text{grains}} = 20 \ \mu\text{m},$ $\emptyset_{\text{pores}} = 1 \ \mu\text{m},$ $Al_2O_3 99\% - \text{SiO}_2 1\%,$ $e = 6.45 \text{ mm}$	Square wave (10–30 s), axial heat flow, phase change measurement+parameter identification with confidence interval, CO_2 gas laser	530–1294 K, N ₂ +5% H ₂ flow under 1–5 Torr	PbS IR detector
6 (1962)	Alumina, $\rho = 3.04 \text{ g cm}^{-3}$, slab section=76×127 mm ²	Constant heat flux on a sample having initially a uniform <i>T</i> , axial heat flow, recording of the <i>T</i> on the FF and RF of the sample vs <i>t</i> , Joule effect in a chromel sheet $(5 \times 76 \times 180 \text{ mm}^3)$	298–1273 К	2 K thermocouples (\emptyset =3.2 mm) spot welded to the 2 chromel sheet (heater and heat sink) which compose the sandwich assembly with the sample, precision 10%, accuracy 15%
17 (1967)	Lucalox alumina, $\rho = 3.89 \text{ g cm}^{-3}$, $Al_2O_3 > 99.8\% - SiO_2 0.03\% -$ $Fe_2O_30.01\% - CaO 0.01\% -$ MgO 0.1%, 2 cylinders: $\emptyset = 25.4 \text{ mm} - h = 152.4 \text{ mm}$ $\emptyset = 50.8 \text{ mm} - h = 457.2 \text{ mm}$	Scanning temperature with constant rate (1.5 K min ⁻¹), radial heat flow, recording of the <i>T</i> at 2 points of the sample vs <i>t</i> , Pt–Rh 40% wire wound tubular furnace or W mesh heater	808–1479 or 368–1692 K under purified He	S thermocouples inserted in 2 $(\emptyset = 1.6 \text{ mm})$ holes 11.4 mm distant, or, 4 $(\emptyset = 2.3 \text{ mm})$ holes, 7.1–14.3–23.5 mm distant, 10%
4 (1986)	Mac Danel alumina, calculated $\rho = 3.87 \text{ g cm}^{-3}$, Al ₂ O ₃ 99.8%, cylinder: $\emptyset = 54.1 \text{ mm-h} = 139.7 \text{ mm}$	Scanning temperature with constant rate (1.5 K min ⁻¹), radial heat flow, recording of the <i>T</i> at 2 points of the sample vs <i>t</i> , horizontal resistance furnace	573-1413 K, under vacuum 10^{-2} Torr	2 sheathed S thermocouples $(\emptyset = 2.03 \text{ mm})$ inserted in $(\emptyset = 2.39 \text{ mm})$ holes 23.71 mm distant, 2.4%-5.4%
16 (1950)	Alumina, $\rho = 3.8 \text{ g cm}^{-3}$, porosity 0.05, cylinder: $\emptyset = 25.4 \text{ mm-h} = 228.6 \text{ mm}$	-50 K T step on a sample having initially a uniform T , radial heat flow (infinite cylinder) recording of the T change vs t , quenching from a Kanthal resistance furnace to a mixed liquid Pb bath	673–973 K, air or liquid Pb	S thermocouple in an axial hole (\emptyset =3 mm), 2.5%
7 (1962)	Alumina, $\rho = 3.6 \text{ g cm}^{-3}$, porosity<0.1, 2 cylinders: $\emptyset = 15.2 \text{ mm-h} = 53 \text{ mm}$ $\emptyset = 24.4 \text{ mm-h} = 53.8 \text{ mm}$	250 K T step on a sample having initially a uniform T, axial and radial heat flow (finite cylinder), measurement of the slope of the curve ln (T) vs t for 2 samples with different \emptyset , identification of the diffusivity and the Biot modulus, displacement of the sample from a Glo-bar element furnace (T_{min}) , to a graphite resistance furnace (T_{max})	1773–2073 K, constant He flow	W-Mo thermocouple in an axial hole (\emptyset =3 mm), 6%.



FIG. 4. Effect of the finite thermal conductance of the thermocouples/ sample contact under outer steady temperature. H=0: perfect contact, H > 0: imperfect contact.

effect of the sensors/sample conductance with the displacement effect.

Let the one-dimensional heat transfer equation in cylindrical coordinates, where [T(r,t)] is the temperature at the radial position r and the time t, be

$$\frac{\partial^2 T}{\partial r^2} + \frac{1}{r} \frac{\partial T}{\partial r} = \frac{1}{K} \frac{\partial T}{\partial t}.$$

The heat transfer boundary conditions are described at the interfaces by the H main coefficient (Fig. 5)

$$r = \epsilon_1 > 0: \quad \left(\frac{\partial T}{\partial r}\right)_{\epsilon_1} = \frac{1}{H} \left[T(\epsilon_1, t) - T_{\text{IT}}\right],$$
$$r = d - \epsilon_2 \quad \left(\frac{\partial T}{\partial r}\right)_{d - \epsilon_2} = \frac{1}{H} \left[T_{\text{OT}} - T(d - \epsilon_2, t)\right].$$

 $T_{\rm IT}$ and $T_{\rm OT}$ are the temperatures of the inner and outer thermocouples, respectively. The outer thermocouple is taken as the phase origin and unit amplitude yielding



FIG. 5. Alumina sample: +: ideal curves [Eqs. (1) and (2)]. Experimental points: \bigcirc : under argon (1095 and 1281 K), \triangle : under helium (782 K), \blacktriangle : under helium (1413 K). ---: fit of the argon results. ——: Model II, $H = 1 \text{ mm}, \epsilon_1 = \epsilon_2 = 0.5 \text{ mm}.$

$$T_{\rm OT} = \cos(\omega t),$$
$$T_{\rm TT} = \Theta \cos(\omega t - \Phi).$$

Knowing the measured quantities, Φ and Θ , the problem is to find the phase change Φ' and the amplitude ratio Θ' between the points $r = \epsilon_1$ and $r = d - \epsilon_2$. Using complex temperature T^* , the solution is quite straightforward and omitting the mathematical developments, the solution can be written as follows.

Model II

$$\Theta' = \frac{|T^*(\epsilon_1, t)|}{|T^*(d - \epsilon_2, t)|} = \frac{|\text{NUM}|}{|\text{DEN}|},\tag{6}$$

$$\Phi' = \operatorname{Arg} \left[T^*(\epsilon_1, t) \right] - \operatorname{Arg} \left[T^*(d - \epsilon_2, t) \right]$$

= Arg (NUM) - Arg (DEN), (7)

where

NUM=A
$$I_0(e\epsilon_1\sqrt{i})$$
 – B $K_0(e\epsilon_1\sqrt{i})$, (8)

DEN=A
$$I_0[e(d-\epsilon_2)\sqrt{i}]$$
-B $K_0[e(d-\epsilon_2)\sqrt{i}]$, (9)

$$A = C \Theta \exp(-i\Phi) + D, \qquad (10)$$

$$B = E \Theta \exp(-i\Phi) + F, \qquad (11)$$

and

$$\mathbf{C} = e H \sqrt{i} K_0' [e(d - \boldsymbol{\epsilon}_2) \sqrt{i}] + K_0 [e(d - \boldsymbol{\epsilon}_2) \sqrt{i}], \qquad (12)$$

$$\mathbf{D} = e H \sqrt{i} K_0'(e \epsilon_1 \sqrt{i}) - K_0(e \epsilon_1 \sqrt{i}), \qquad (13)$$

$$\mathbf{E} = e H \sqrt{i} I_0'[e(d - \boldsymbol{\epsilon}_2) \sqrt{i}] + I_0[e(d - \boldsymbol{\epsilon}_2) \sqrt{i}], \qquad (14)$$

$$\mathbf{F} = e H \sqrt{i} I_0'(e \epsilon_1 \sqrt{i}) - I_0(e \epsilon_1 \sqrt{i}).$$
(15)

In the above equations, *i* is the complex number, I_0 and K_0 are the modified Bessel functions of zero order and I'_0 and K'_0 their first derivatives. *H* is the ratio between the conductivity of the sample λ and the conductance of the sensor/sample contacts *h*. It has the dimension of a length

$$H=\frac{\lambda}{h}.$$

H always appears in the dimensionless form *e H* [Eqs. (12) to (15)], which means that its effect in the calculation depends on the frequency of the thermal signal. Note *h* is in fact a global heat transfer coefficient which takes into account conduction through the contact points or through the gaseous atmosphere and linearized radiating heat transfer. H=0 represents a perfect thermal contact; in this case the model yields obviously $\Theta = \Theta'$ and $\Phi = \Phi'$.

Assuming that the sensors/sample contact is nearly perfect under helium at 1413 K, the experimental results obtained under these conditions are fitted using Model I [Eqs. (3) and (4)]. This procedure allows a rough estimation of the displacement parameters yielding $\epsilon_1 = \epsilon_2 = 0.5$ mm. Then, the results of the measurements under argon atmosphere are fitted with Eqs. (3) and (4) and give a set of (Θ, Φ) initial

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TABLE IV. Estimation of the h parameter.

	$H \approx$	$\lambda \ Al_2O_3$	$h \approx$	λ gas	$h_{ m gas} \approx$	$h_{ m rad} \approx$	$h_{ m sol} \approx$
T/K	cm	$\mathrm{Wcm^{-1}K^{-1}}$	$\mathrm{Wcm^{-2}K^{-1}}$	$\mathrm{W}\mathrm{cm}^{-1}\mathrm{K}^{-1}$	$\mathrm{Wcm^{-2}K^{-1}}$	$\mathrm{Wcm^{-2}K^{-1}}$	$W cm^{-2} K^{-1}$
782	0.1	0.11	1	0.0030 ^a	0.060^{a}	0.0055	0.9
1095	0.3	0.072	0.2	$0.000 \ 60^{b}$	0.012 ^b	0.015	0.2
1291	0.3	0.063	0.2	0.000 67 ^b	0.013 ^b	0.024	0.2
1413	0	0.060		0.0047 ^a	0.094	0.032	

^aHe. ^bAr.

data for the calculations by Model II [Eqs. (6)–(15)]. Finally, lists of (Θ', Φ') values are calculated, using the estimated ϵ_1 and ϵ_2 , for different *H* values (Fig. 5).

It can be seen in Fig. 5 that the displacement effect, $\epsilon_1 = \epsilon_2 = 0.5$ mm, is approximately counterbalanced when H = 1 mm. To account for the shift between the results obtained under argon (1095 and 1281 K) and helium atmosphere at 1413 K, a value of H=3 mm is necessary. This value cannot be considered as a measurement of H at the sensor/thermocouple contact under argon around 1100 K, only the order of magnitude is significant. For comparison, Paladino⁷ (1957) reported H values lying in the range 10–4 mm in the case of heat transfer through the surface of an alumina sample submitted to constant helium flow between 1773 and 2073 K. This author also notices that this coefficient is "extremely sensitive to slight experimental variations" and that only a range of variation of H can be reported.

In the above-mentioned case, the displacement of the thermocouples from their ideal position results in the decrease of the distance between them. The *H* parameter may partially or even totally cancel this effect, depending on the relative values of H, ϵ_1 , and ϵ_2 . However, another case can be imagined, where the displacement increases the distance between the thermocouples; the two effects become now additive and an increase in *H* will increase the measurement error.

It was mentioned before that the observed differences on the $(\Theta(u), \Phi(u))$ charts between the four temperatures were due to the changes in the contact conditions: nature of the gas and also increasing radiation heat transfer as the temperature raises. However, (Θ, Φ) do not only depend on the contact conditions, i.e., on *h*, they also depend on the conductivity of the sample through the parameter *H*. This effect is non-negligible in the present case as the thermal conductivity of alumina is approximately divided by a factor of 2 between 800 and 1400 K. The error on (Θ, Φ) decreases as the sample conductivity decreases.

A priori estimation of the h coefficient is difficult, this coefficient is the sum of three contributions; solid conduction through the contact points, gas conduction, and radiation

$$h = h_{\rm sol} + h_{\rm gas} + h_{\rm rad}$$

Among these three terms, h_{sol} is the hardest to estimate. However, it is possible to try to use the *H* values deduced from the experiments (Table IV) to check the order of magnitude of *h*. From thermal capacity values¹⁸ and Eq. (5), with a theoretical density of 4 g cm⁻³, the conductivity is calculated as a function of temperature (Table IV). Dividing the *H* experimental values by the λ values, *h* values are obtained (Table IV).

For the h_{gas} term, relative information is given by the conductivities of the gas. The conductivities values reported in Table IV are extrapolated from the data of references^{19,20} at atmospheric pressure, to the temperatures of the measurements. This is an overestimation of the true values because the pressure in the present case lies in the range 400 $\times 10^2 - 530 \times 10^2$ Pa. The gas conductance under argon is, at least, 5 times lower than that of helium. The temperature changes do not strongly modify conduction heat transfer by the gas compared to changing the nature of the gas. To know the magnitude of h_{gas} , it is necessary to know the average gap distance g between the sample and the thermocouple bead. Knowing that the diameter of the bead is about 1 mm and the diameter of the hole is 2 mm, this distance can be very roughly estimated as g = 0.5 mm. Then, $h_{\text{gas}} = \lambda/g$ is 20 times the conductivity value (Table IV).

On the hypothesis that the emissivity *E* of the participating surfaces is close to 0.5,²¹ the $h_{\rm rad}$ term can be estimated by $h_{\rm rad} = 4 E \sigma T^3$, where σ is the Stefan–Boltzmann constant. Note $h_{\rm rad}$ increases from 0.0055 to 0.032 W cm⁻² K⁻¹ between 782 and 1413 K (Table IV).

Finally, subtraction of the gas and radiation terms from the total conductance gives the solid conduction term h_{sol} (Table IV). Note h_{sol} is estimated in the range $0.2-0.9 \text{ W cm}^{-2} \text{ K}^{-1}$. The h_{sol} values remain in the range $0.1-1 \text{ W cm}^{-2} \text{ K}^{-1}$, which is a classical order of magnitude for wavy metallic or nonmetallic surfaces in contact.

The models which have been developed show that the influence of displacement and conductance effects are indistinguishable, both contributing to the overall results. Obviously, the setup is improved as the diameter of the holes is decreased and the thermal conductance of the interface is increased. This is the main problem to be solved in order to increase the accuracy of the apparatus. At the present, the use of intrinsic split thermocouples directly welded on metallic samples, is under investigation. It is expected that this feature will greatly enhance the accuracy of our apparatus as it diminishes the influence of the contact conductance and divides the diameter of the thermocouples holes by two.

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