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Composite Nature of Layered Hybrid Perovskites : Assessment on Quantum and Dielectric Confinements and Band Alignment

Boubacar Traore,^{*,†} Laurent Pedesseau,[‡] Linda Assam,^{‡,¶} Xiaoyang Che,^{†,‡}
Jean-Christophe Blancon,[§] Hsinhan Tsai,[§] Wanyi Nie,[§] Constantinos C.
Stoumpos,^{||} Mercuri G. Kanatzidis,^{||} Sergei Tretiak,[§] Aditya D. Mohite,[§] Jacky
Even,[‡] Mikaël Kepenekian,^{*,†} and Claudine Katan^{*,†}

[†]*Univ. Rennes, ENSCR, INSA Rennes, CNRS, ISCR (Institut des Sciences Chimiques de
Rennes) - UMR 6226, F-35000 Rennes, France*

[‡]*Univ. Rennes, INSA Rennes, CNRS, FOTON (Fonctions Optiques pour les Technologies
de l'Information) UMR 6082, F-35000 Rennes, France*

[¶]*TOTAL SA, Tour Coupole, 2 place Jean Miller - La Défense 6 - Courbevoie, France*

[§]*Los Alamos National Laboratory, Los Alamos, NM 87545, USA*

^{||}*Department of Chemistry, Northwestern University, Evanston, IL 60208, USA*

E-mail: boubacar.traore@univ-rennes1.fr; mikael.kepenekian@univ-rennes1.fr;

claudine.katan@univ-rennes1.fr

Abstract

Layered hybrid organic-inorganic perovskites (HOPs) have re-emerged as potential technological solutions for next generation photovoltaic and optoelectronic applications. Their two dimensional (2D) nature confers them a significant flexibility and results in the appearance of quantum and dielectric confinements. Such confinements are at the origin of their fascinating properties and understanding them from a fundamental level is of paramount importance for optimization. Here, we provide an in-depth investigation of band alignments of 2D HOP allowing access to carriers' confinement potentials. 2D HOPs are conceptualized as composite materials in which pseudo inorganic and organic components are defined. In this way, computational modeling of band alignments becomes affordable using first-principles methods. First, we show that the composite approach is suitable to study the position dependent dielectric profiles and enables clear differentiation of the respective contributions of inorganic and organic components. Then we apply the composite approach to a variety of 2D HOP, assessing the impact on the confinement potentials of well and barrier thickness, of nature of the inorganic well and of structural transitions. Using the deduced potentials, we further discuss the limitations of the effective mass approximation, scrutinizing the electronic properties of this family of composite materials. Our simulations demonstrate type-I dominant band alignment in 2D HOP. Finally, we outline design principles on band alignment towards achieving specific optoelectronic properties. Thus, we present alternative theoretical methods to inspect the properties of 2D hybrid perovskites and expect that the composite approach will be applicable to other classes of layered materials.

Keywords

Band alignment, type-I, quantum confinement, dielectric confinement, composite, photovoltaics, light emitters

The rapid rise of hybrid organic-inorganic perovskites' (HOP) power conversion efficiencies¹⁻⁶ over the past few years has placed them as feasible technological solutions for

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3 future photovoltaic and optoelectronic applications. The unique combination of organic
4 and inorganic constituents of the material as exemplified in the so-called methylammonium
5 (MA=CH₃NH₃) lead iodide (MAPbI₃), underscores superior photo-physical properties of
6 this family of materials.^{7,8} Thin films of these materials can be prepared using low-cost
7 solution-based growth methods. However, to bridge the gap from a laboratory characterized
8 material to a technological viable solution compliant with industrial standards, HOP must be
9 resistant to lighting, moisture and humidity stress conditions. The latter proved to be detri-
10 mental to 3D perovskites.⁹⁻¹² Recently, layered perovskites have re-emerged as promising
11 alternatives to their 3D counterparts due to their ability to overcome the above-mentioned
12 mechanisms.^{5,13,14} Besides, the strong vertical orientation of the layers with respect to the
13 substrate and the self-assembly of organic molecules separating the inorganic sheets in lay-
14 ered HOP offers many opportunities for material tuning targeting specific properties *via* for
15 instance chemical substitution of organic moieties,^{15,16} increasing the length of the inorganic
16 layer^{17,18} or metal substitution in the inorganic framework.¹⁹ Furthermore, they present
17 attractive optical properties related to large exciton binding energies,^{20,21} promising light
18 emission characteristics²²⁻²⁴ and multiexcitonic resonances.^{25,26} Achieving such optimization
19 towards efficient and performing optoelectronic devices requires a thorough understanding
20 of their fundamental photophysical and electronic properties.²⁷

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39 An important criterion for the design of efficient solar or light emitting devices is the
40 band alignment of their constituent materials^{28,29} allowing access to carriers' confinement
41 potentials. However, accessing band lineups is not straightforward both from experimental
42 and theoretical perspectives.³⁰ Electron affinity models have traditionally been used to align
43 band profiles of heterostructures.³¹ It relies on the affinity of the respective bulk components
44 with the main inconvenient that interface related effects on band offsets are ignored. To
45 calculate band profiles in semiconductors, the transitivity rule has also been used. Here, the
46 band lineup of a heterostructure A/C (consisting of materials A and C) is deduced from
47 those of A/B and B/C separately.^{32,33} First principles based approaches with self-consistent
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3 charge densities and potentials have also been suggested for that purpose.^{34,35} The direct
4 alignment of band edges cannot be deduced from the first-principles studies of the separate
5 bulk materials due the lack of an intrinsic energy scale,^{36,37} but interface related effects can
6 be accounted for in heterostructures with short periods.
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11 The self-assembled alternating organic and inorganic sheets of layered HOP is reminiscent
12 of an heterostructure. Mitzi and coworkers proposed a type-I quantum-well (QW) picture
13 (Figure 1a), with organic barriers separating inorganic wells.³⁸ This idea was verified later
14 for $(\text{C}_{10}\text{H}_{21}\text{NH}_3)_2\text{PbI}_4$ material³⁹ where band alignment and confinement potentials for holes
15 and electrons were deduced based on first-principles calculations. In the field of conventional
16 semiconductors, further modeling of heterostructures is performed using such confinement
17 potentials in empirical Hamiltonians. For instance, in a quantum well model, confinement
18 potentials can be used to calculate carriers' densities⁴⁰ and subsequently their mobilities,
19 quantities of significant importance in device operation. The applicability of this method to
20 layered HOP still requires detailed investigation, especially on how materials' structure and
21 composition affect band alignments.
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33 In this work, we introduce the foundation of the composite approach and subsequently
34 demonstrate, by means of first-principles calculations, that it applies to a wide range of lay-
35 ered hybrid organic-inorganic perovskites. It provides fruitful insights for inspecting quantum
36 and dielectric confinement effects. First, we show that high-frequency dielectric constant pro-
37 files, related to dielectric confinement, exhibit a composite behavior. Next, we perform a
38 detailed investigation of band alignment and confinement potentials within the framework
39 of our composite approach considering sequentially the effects of: *(i)* the composition of the
40 inorganic well and organic barrier *(ii)* the thickness of the molecular barriers and phase tran-
41 sitions, and *(iii)* the thickness of the perovskite layers. The later case is then used to inspect
42 the validity of the effective mass approximation in layered HOP. We demonstrate that Type-I
43 confinement is the dominant band alignment in 2D HOP. Before concluding, material design
44 principles are outlined for achieving specific optoelectronic properties. We believe that our
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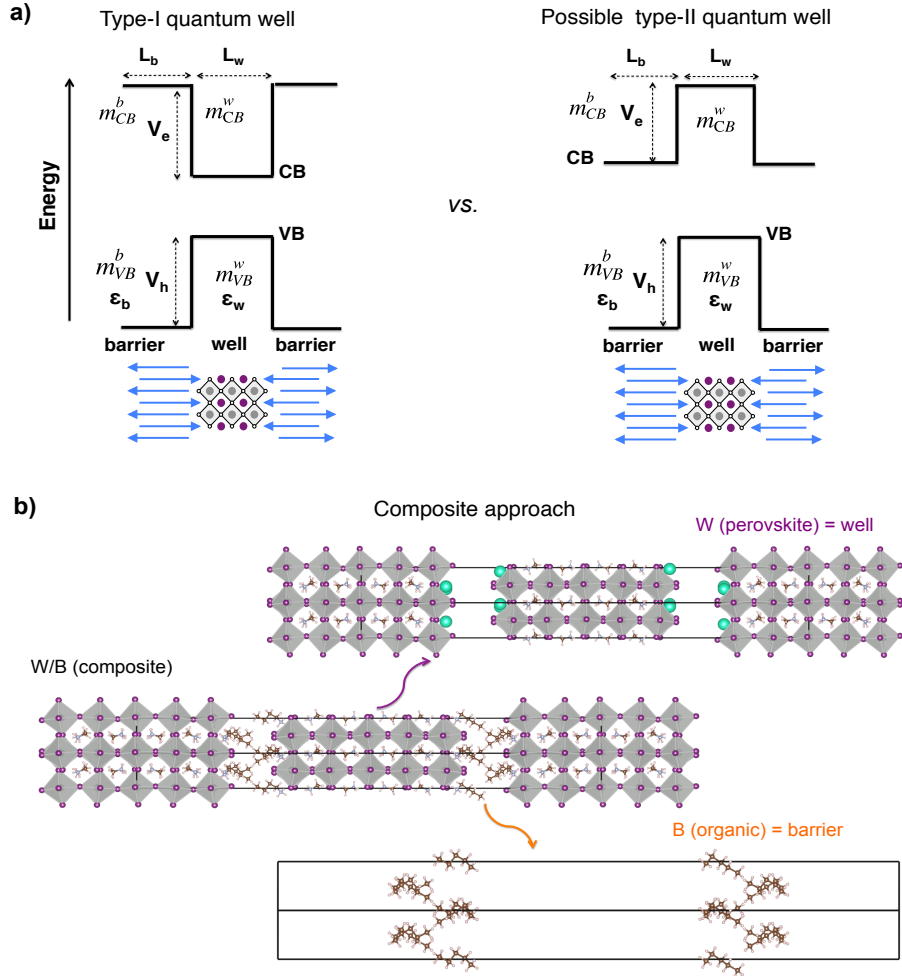


Figure 1: Composite approach to layered hybrid perovskites. (a) schematics of type-I³⁸ and possible type-II quantum wells. VB and CB correspond to valence and conduction bands, respectively. Note that, for a type-II quantum well, VB and CB are not spatially localized in the same layer. L , m , ϵ and V stand for thickness, effective mass, dielectric constant and confinement potential, respectively. Subscripts and superscripts b , w , e and h denote barrier, well, electron and hole, respectively. The schematics structure below shows the definition of the well and the barrier regions in a pictorial perovskite structure (arrows correspond to the barrier and corner shared octahedra to the well). (b) General presentation of the composite approach: the real composite material W/B is decomposed into two parts: the perovskite well section W where the organic molecules of the barrier are replaced by Cs^+ atoms and the organic barrier section B, in which NH_3^+ are replaced by CH_3 groups.

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3 composite approach will be applicable to other classes of 2D hybrid materials, where organic
4 and inorganic components coexist.
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8 9 **Results/Discussion**

10 11 12 **Concept of the Composite approach for dielectric and quantum con-** 13 14 15 **finements in layered HOP** 16 17

18 We start by defining the composite approach for layered HOP. In a layered HOP, the organic
19 molecules exhibit a wider band gap as compared to the inorganic perovskite layer. Such
20 structure can be viewed as a type-I quantum well³⁸ (Figure 1a). In this configuration, the
21 organic molecules and inorganic (perovskite) layers correspond to the barrier and the well of
22 the potential, respectively. Hence, in the composite approach, the quantum well like struc-
23 ture of layered HOP is a heterostructure³⁹ where we define three model systems (Figure 1b).
24 For the proof of concept and without loss of generality, we use $(\text{BA})_2(\text{MA})_4\text{Pb}_5\text{I}_{16}$ ($\text{BA} =$
25 $\text{CH}_3(\text{CH}_2)_3\text{NH}_3$) member of Ruddelsden-Popper (RP) phases type of layered perovskites.¹⁸
26 In the figure, W/B corresponds to the complete crystallographic structure.¹⁸ Although, we
27 rely on the experimental structures, the method is applicable to DFT based optimized ge-
28 ometries as well. Structure W ($\text{Cs}_2(\text{MA})_4\text{Pb}_5\text{I}_{16}$) is the five perovskite layers of W/B with
29 the inter-layer molecules BA replaced by Cs. The validity of such Cs substitution is discussed
30 in the supplementary information (Figure S1). Structure B consists of the inter-layer organic
31 molecules $(\text{CH}_3(\text{CH}_2)_3\text{CH}_3)_2$ where the NH_3^+ group of the BA cation is substituted by the
32 CH_3 moiety. In both W and B structures, the electroneutrality of the system is enforced.
33 Next, we inspect the dielectric profiles and band alignments within the composite method
34 using $(\text{BA})_2(\text{MA})_4\text{Pb}_5\text{I}_{16}$ as a model system.
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52 Layered perovskites have large exciton binding energies due to the dielectric and quantum
53 confinement effects.^{41,42} The barrier dielectric constant is usually much smaller than that of
54 the inorganic well. Consequently, the effective electron-hole Coulomb interaction is enhanced
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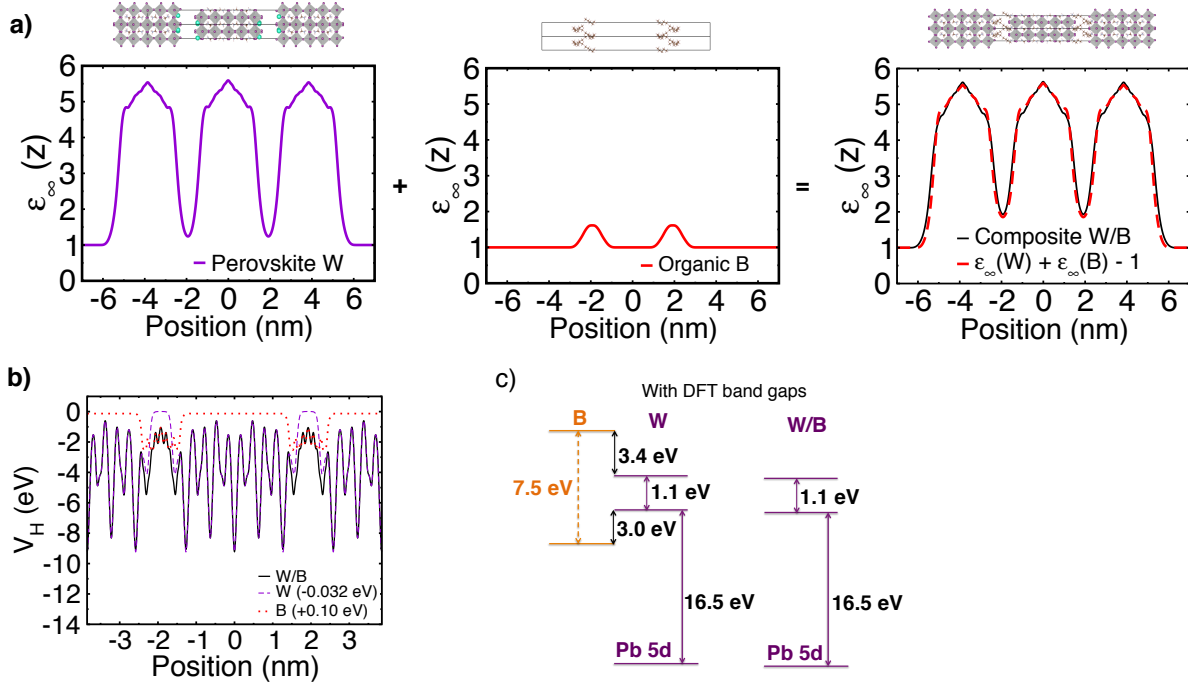


Figure 2: Composite approach to dielectric profiles and band alignment. (a) Dielectric profiles computed for each of the sections W and B of the layered hybrid perovskite $(\text{BA})_2(\text{MA})_4\text{Pb}_5\text{I}_{16}$ ¹⁸ from Figure 1b. The dielectric profile of the real composite structure W/B computed for $(\text{BA})_2(\text{MA})_4\text{Pb}_5\text{I}_{16}$ is also compared to the sum of the individual components W and B. It reveals the composite nature of the dielectric profiles for the 5-octahedron thick layered perovskite. b) Hartree potential profiles (V_H) obtained for the structures of Figure 1b. This affords potential shifts needed to align V_H of the individual components W and B to that of the real composite W/B. c) Band alignment between W and B for $(\text{BA})_2(\text{MA})_4\text{Pb}_5\text{I}_{16}$ obtained considering the V_H shifts deduced in panel b. Noteworthy, for the perovskite layer, the shift in V_H between W/B and W is equal to that of Pb 5d states between W/B and W.

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3 in the well. The image-charge method assuming abrupt interface between the barrier and
4 the well has often been used to account for electron-hole Coulomb interaction in the two
5 particle empirical Hamiltonian.⁴² First principles methods including interface related effects
6 in dielectric profiles have also been developed.^{43,44} Subsequently, we apply these methods
7 to hybrid organic-inorganic perovskites to account for dielectric confinement effects.^{45,46} To
8 show the composite behavior of dielectric profiles in these layered perovskites, we calculate
9 the position dependent high frequency dielectric constant along the stacking axis in the
10 composite W/B, the perovskite well W and the organic barrier B (Figure 2a). ϵ_∞ varies
11 from 1.9 in the organic barrier B to 5.5 in the perovskite well W, clearly highlighting the
12 dielectric confinement. In the perovskite layer W, ϵ_∞ reaches that of bulk 3D MAPbI₃,¹⁸
13 and is in a reasonable agreement with the experimental 6.5 value of the latter.⁴⁷ Figure
14 2a compares the dielectric profile of W/B and the addition of those of W and B. The two
15 dielectric profiles are superimposed using:
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$$\epsilon_\infty^{W/B}(z) = \epsilon_\infty^W(z) + \epsilon_\infty^B(z) - 1, \quad (1)$$

31 where -1 is used to avoid the double counting of the contribution of the vacuum to the
32 dielectric constant. Equation 1 can be interpreted as the sum of the induced polarizations
33 due to structures W and B and from which the equivalent electric susceptibility χ ($\chi = \epsilon -$
34 1) for the composite gives:
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$$\chi_{W/B}(z) = \chi_W(z) + \chi_B(z). \quad (2)$$

42 Hence, using the composite approach, the dielectric profile accounting for the sole elec-
43 tronic contributions (optical dielectric constant profile) of the complete structure can be
44 viewed as the sum of the contributions of the constituting components of the whole struc-
45 ture. The computation of the dielectric profiles presented in this work is performed only for
46 the dielectric component along the stacking axis. It allows identifying clearly the contribu-
47 tion of the respective layers to the total dielectric profile in the composite approach. We
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3 further checked that the composite approach towards the high-frequency dielectric constant
4 also applies in the plane of the quantum well and that the dielectric constant is anisotropic in
5 these structures. Additionally, for thin inorganic layers especially $n=1$, the dielectric profiles
6 of the composite and the addition of the constituting parts do not perfectly match and the
7 error is about 7% (Figure S2). This error becomes negligible with thicker perovskite layers
8 (Figure S2). This indicates that $n=1$, consisting of only one inorganic octahedron, is not a
9 true composite and the relative large error suggests that in this limiting case the properties
10 are influenced by interface effects.

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13 In hybrid perovskites, electron-phonon coupling becomes significant in the THz region
14 which also gets manifest in the dielectric response of the structures.⁴⁸⁻⁵⁰ Such an effect of
15 polar phonons on the dielectric constant can, in principle, be accounted for computationally
16 through the relaxation of both the electronic density and the atomic positions under an
17 external applied electric field.⁴⁶ However, the relaxation of the perovskite W layer alone with
18 Cs^+ substitution or the organic barrier B alone is not relevant due to the strong influence of
19 barrier-perovskite interactions on the relaxation of the composite W/B. The same conclusion
20 can be drawn for the contribution of the stochastic motions of molecular cations to the low-
21 frequency dielectric constant.⁷ Besides, the perovskite and barrier layers in the composite
22 method are mere conceptual structures and are not suitable for structural optimization.

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25 The computed high frequency dielectric constant profiles involve the relaxation of the sole
26 electronic density under an applied external field along the stacking axis. This electronic
27 density involves the valence band (VB) states (occupied states) as the conduction bands
28 are completely unoccupied in these undoped semiconductor materials. We expect that the
29 composite method will permit the evaluation of carrier's confinement potentials, particularly
30 relevant to the VB. We show next its applicability to compute band alignment with the
31 subsequent deduction of relevant confinement potentials.

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34 The evaluation of band alignment in a heterostructure using either electron affinity rule
35 or first principles based DFT methods requires information on its bulk materials. In the

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3 former, alignment is achieved through the affinities of the respective bulk materials; in the
4 latter, band edge energies are referenced to the Hartree potentials obtained from the bulk
5 components. However, bulk structures for the perovskite and organic parts of layered HOP
6 cannot be defined as in classical semiconductor heterostructures.³⁹ This is possible, however,
7 in the present composite approach where the band alignment is carried out through the
8 Hartree potential profiles. The latter are calculated from DFT simulations of three different
9 structures (W/B, W, and B). This is illustrated for $(\text{BA})_2(\text{MA})_4\text{Pb}_5\text{I}_{16}$ in Figure 2b. Hartree
10 potentials are then obtained by averaging the potentials in the plane perpendicular to the
11 stacking axis of the layered structure. Shifting the potential profile of the perovskite layer
12 W with Cs^+ substitution (-0.032 eV for our example) results in a perfect match to that of
13 the perovskite part of the whole W/B structure. Thus the electronic states of the inorganic
14 layer and close to the band gap edges are fairly well described by the W model structure,
15 otherwise such a match would not be possible. Similarly, a shift (+0.10 eV for our example)
16 of the potential profile of the organic structure B results in a perfect match with that of the
17 organic part of W/B. This procedure demonstrates that, with regards to band alignment,
18 layered perovskites can be considered as composite materials with weak interactions between
19 the organic and inorganic layers.³⁹

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21 From the Hartree potentials alignment, we further deduce an appropriate band lineup
22 between the perovskite and organic layers of the 2D material. The procedure consists of
23 applying the potential shifts obtained from V_H alignment to corresponding VB of W and B
24 structures. For our example, a downward shift of 0.032 eV (as obtained from Figure 2b) is
25 applied to VB of the perovskite part W to match it to that of the structure W/B as shown
26 in Figure 2c. Notably, the very same shift (-0.032 eV) is derived to align the Pb 5d states of
27 W to those of W/B. Indeed Pb 5d states are low-lying orbitals unaffected by the chemical
28 substitution for states close to the electronic gap and could be used as electronic markers in
29 Pb based perovskites.⁵¹ To obtain the level of Pb-5d states, we use orbital resolved projected
30 density of states. The fluctuations observed between the different 5d orbital entries (*vide*
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3 *infra*) are related to the chemical environment around Pb. As the slight difference with Ref.
4 51, it may essentially be due to the choice of the basis-set and pseudo-potentials adopted for
5 those calculations. A similar procedure can be applied to systems with a different chemical
6 composition in which case low-lying orbital entries could be used as markers (for example,
7 4d states for Sn based 2D HOP, *vide infra*). The alignment of structure B is deduced by
8 applying an upward shift of 0.10 eV to its VB as obtained from Hartree potentials alignment
9 of Figure 2b. Hence, using the composite approach of 2D HOP, we afford a complete band
10 alignment of the system and shows the type-I nature of confinement for this structure.

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19 Although conduction band (CB) lineup is also shown in Figure 2c, we stress that this
20 approach is more relevant to VB alignment as the polarized electron gas is related to the
21 occupied states only. Furthermore, due to the underestimation of the electronic band gap
22 by DFT, it does not allow a precise alignment of CBs. Such treatment would require the use
23 of more sophisticated and computationally demanding methods including both many-body
24 corrections and spin-orbit coupling interactions.^{51,52} Thus, whenever possible we use available
25 experimental band gaps for CB alignment. We remind that these band gaps are usually
26 optical ones that include excitonic contributions (Table S6). Such procedure assumes that
27 band gap corrections mostly affect the conduction band.³⁶ As such, confinement potentials
28 at CB may present larger errors. In the following, we apply the composite approach to a
29 wide range of layered HOP and assess various effects on the confinement potentials.
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43 **Chemical tuning of the inorganic well**

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45 Because band edge states (valence band maximum (VBM) and conduction band minimum
46 (CBM)) in HOP are mainly determined by the hybridization between metal and halogen
47 orbitals, substituting halogen atoms as well as mixing them have been used to tailor the
48 band gap and photoluminescence (PL) properties of these materials.^{53,54} For instance, in
49 (EA)₄Pb₃X₁₀ (EA=CH₃CH₂NH₃) layered perovskites, an increase of the band gap was ob-
50 served in the absorption or emission spectra when X was varied from Br to Cl.²⁴ We inspect
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the effect of such change of the halogen atom on band alignment with the phenyl methylammonium lead halide $(\text{PMA})_2\text{PbX}_4$ ($\text{PMA}=\text{C}_6\text{H}_5\text{CH}_2\text{NH}_3$) series of layered perovskites (Figure 3a). Here, DFT band gaps are used for CB alignment since experimental data are not available for the entire series. Figure 3b presents their deduced type-I band alignments considering the experimental structures (Figure 3(a)) for $\text{X} = \text{I}$,⁵⁵ Br ¹⁵ and Cl .⁵⁶ The DFT band gap increases with lighter halogen atoms whereas the in-plane lattice constants decrease in the same series. This stabilizes (destabilizes) Pb-halogen hybridization at VBM (CBM) and results in the band gap increase for lighter halogens. Incidentally, significant structural

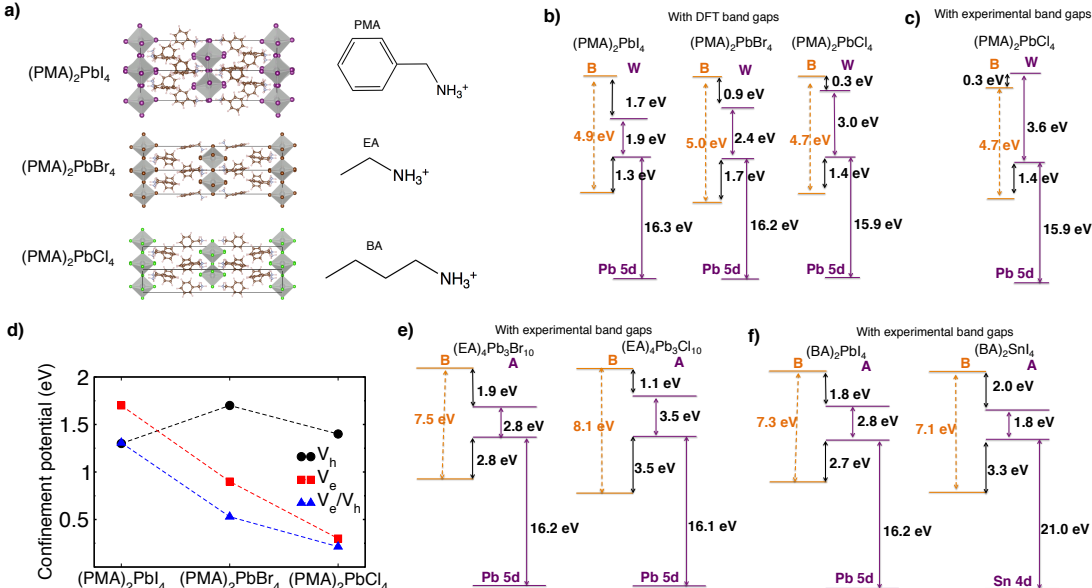


Figure 3: Effect of the chemical composition of the perovskite well on band alignments. (a) Experimental structures used for $(\text{PMA})_2\text{PbX}_4$ series to investigate the effect of halogen atom on band alignment. X refers to the halogen (I, Br or Cl); and schematics of the corresponding molecular barriers considered in this panel. PMA for phenyl methylammonium, EA for ethylammonium and BA for butylammonium. (b) Computed band alignments of $(\text{PMA})_2\text{PbX}_4$ and corresponding band offsets. Since, we do not have the experimental band gaps for all the compositions, DFT band gaps are used for conduction band lineup. For $\text{X}=\text{Cl}$, the band alignment tends to deviate from type-I. (c) Same as panel b considering the experimental optical band gap for $(\text{PMA})_2\text{PbCl}_4$, showing the possibility of a type-II band alignment. (d) Variation of hole (circle) and electron (square) confinement potentials extracted from panel b and corresponding ratios. (e) Band alignment computed for $(\text{EA})_4\text{Pb}_3\text{X}_{10}$ using experimental optical band gaps²⁴ to align CBM (for alignments with DFT band gaps see Figure S4). (f) Band alignment showing the effect of metal substitution $(\text{BA})_2\text{MI}_4$ ($\text{M} = \text{Pb}, \text{Sn}$) on confinement potentials. Experimental optical band gaps are used for CB alignment (see Figure S5 for DFT band gaps).

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3 distortions occur, consistent with noticeable octahedral tiltings (β , δ) induced in the inor-
4 ganic layer when the halogen is varied from I to Cl. The tiltings (β , δ) are defined in Figure S3
5 and their values are reported in Table S1. Such distortions affect more the VB than the CB
6 since the former consists of an antibonding hybridization between halogen X (p states) and
7 lead Pb (s states); and the latter composes a bonding hybridization between Pb (p states).
8 One would then expect that a pronounced effect on VB states would significantly affect hole
9 confinements from I to Cl. However, we notice that the hole confinement potentials (V_h)
10 have similar values for the different $(\text{PMA})_2\text{PbX}_4$ wells (Figure 3d). The 0.3 eV difference of
11 V_h between Br based well as compared to that of Cl falls within the uncertainties in the DFT
12 band gaps of the organic and inorganic sheets. At first, the electron confinement potential
13 (V_e) for $(\text{PMA})_2\text{PbI}_4$ appears larger than V_h considering DFT band gaps. This conclusion
14 is however expected to be revised using experimental band gaps or under accurate band gap
15 correction *via* sophisticated many body correction schemes. Noteworthy, despite systematic
16 errors in the band gap values, $(\text{PMA})_2\text{PbBr}_4$ and $(\text{PMA})_2\text{PbCl}_4$ present much lower V_e as
17 compared to V_h . This trend suggests that the confinement type for $(\text{PMA})_2\text{PbX}_4$ compounds
18 may undergo a crossover from type-I to type-II under a proper band gap correction. For in-
19 stance, considering the experimental 3.65 eV optical band gap of $(\text{PMA})_2\text{PbCl}_4$,⁵⁶ the band
20 alignment of $(\text{PMA})_2\text{PbCl}_4$ may change from type-I to a type-II confinement (Figure 3c).
21 Although such a type-II alignment has not yet been reported in layered hybrid perovskites,
22 an experimental measurement aiming at probing the lifetime of excitons could be decisive
23 about its possible existence. From such a spectroscopic measurement, longer lived indirect
24 excitons could point to a type-II band heterostructure.

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47 We further inspect the effect of halogen substitution considering $(\text{EA})_4\text{Pb}_3\text{X}_{10}$ type of
48 perovskites²⁴ where X represents Br and Cl ($\text{EA} = \text{CH}_3\text{CH}_2\text{NH}_3$). The corresponding band
49 alignments showing the relevant confinement potentials are shown in Figure 3e. We notice a
50 similar trend of higher V_h as compared to V_e . Notably, the decrease of V_e in $(\text{EA})_4\text{Pb}_3\text{Cl}_{10}$
51 also points to the less pronounced type-I carrier confinement when Cl is used as the halogen.
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We finally investigate the effect of metal substitution on band alignment considering $(\text{BA})_2\text{MI}_4$ ($\text{M}=\text{Pb}, \text{Sn}$).^{57–59} Notice that Sn 4d, unaffected low-lying orbitals, were used as electronic markers for $(\text{BA})_2\text{SnI}_4$ as shown in Figure 3f. $(\text{BA})_2\text{SnI}_4$ presents both larger hole and electronic confinement potentials as compared to $(\text{BA})_2\text{PbI}_4$ probably due to the narrower band gap of $(\text{BA})_2\text{SnI}_4$.

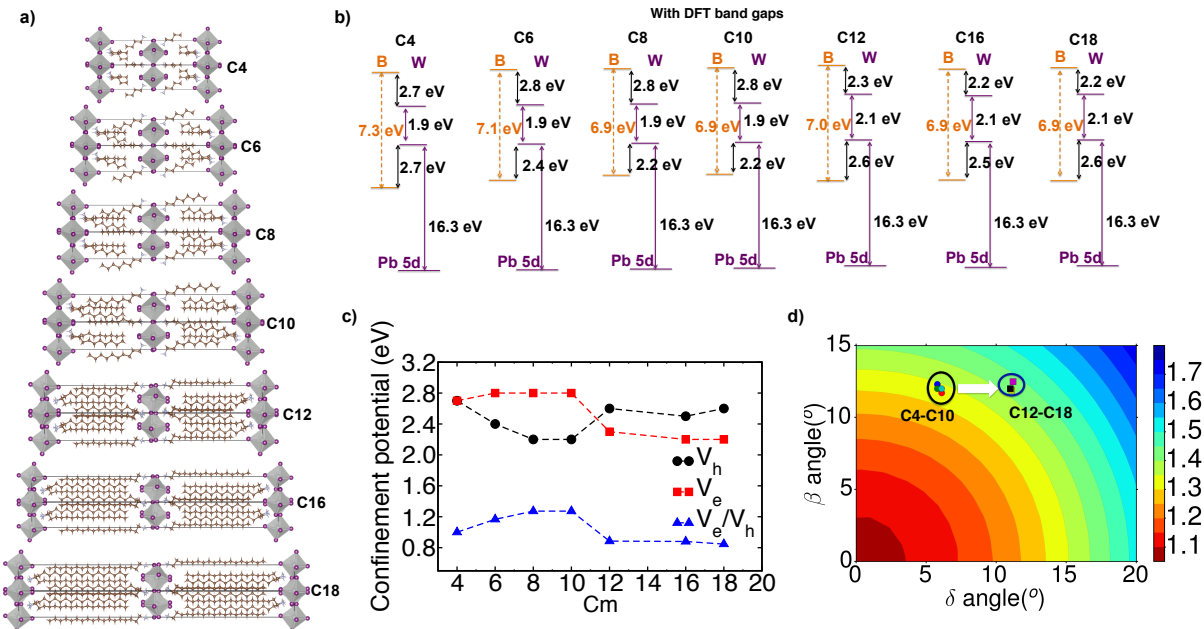


Figure 4: Band alignment for various barrier thicknesses. (a) Structures of $(\text{C}_m\text{H}_{2m+1}\text{NH}_3)_2\text{PbI}_4$. C_m next to the structures refers to the number m of carbon atoms in the alkyl chain. (b) Computed band alignments for the alkyl-based structures shown in panel a, taking crystallographic data recorded at room temperature. Since, all the experimental band gaps are not available for the series, we use DFT band gaps for conduction band lineup. (c) Variation of hole (circle) and electron (square) confinement potentials extracted from panel b and corresponding ratios. (d) Computed band gaps plotted on the 2D color map illustrating the variation of the band gap with octahedral in-plane (β) and out-of-plane (δ) tilt angles for the Cs_2PbI_4 model structure (spin-orbit coupling is included).⁶⁰ Tilt angles are summarized in Table S2.

Effect of barrier length

Yet another attractive feature of 2D HOP is the flexibility of the organic barrier and the possibility of tuning its length and chemical composition. To assess the effect of the barrier length on the band lineup, we investigate some members of the alkyl ammonium lead iodide

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($C_mH_{2m+1}NH_3$)₂PbI₄ series of layered HOP (C_m with m carbons in the barrier molecule), in their orthorhombic phases (Figure 4a, Table S2).^{57,61,62} Figures 4b,c show their band alignments and the variation of the deduced confinement potentials with increasing barrier length, respectively. All plots reveal a type-I alignment. From C4 to C10, V_e is larger than V_h with almost no variation as a function of the barrier length. This behaviour is reversed for the C12-C18 series where V_h is larger than V_e with little dependence on the length of the barrier. Additionally, a band gap increase of about 0.2 eV is also deduced for C12-C18 compounds as compared to C4-C10.

This band gap increase and the associated reverse trend of the confinement potentials can be related to distortions in the structures of C12-C18, as compared to C4-C10. In fact, structural distortions have a significant effect on the band gap of hybrid perovskites with a direct impact on their optical properties.^{8,27,52,63} To rationalize the interplay between octahedral distortions and band gap variation with the associated reversed confinement potentials, we use a Cs₂PbI₄ model system from which two characteristic angles β (in-plane tilt) and δ (out-of-plane tilt) can be defined.⁶⁰ The angles deduced from the different experimental structures (Table S2) can then be plotted on the computed 2D map of band gap variations with octahedral tilting, Figure 4d. The entire alkyl series (C4-C18) exhibits similar in-plane distortion β . However, the out-of-plane tilt angle δ is larger for C12-C18 as compared to C4-C10 which is accompanied by an increase of the band gap in the 2D map (Figure 4d).⁶⁰ Hence, the band gap increase and the reversed confinement potentials are related to the octahedral distortions that build up when going from C12 to C18. A noticeable change in the in-plane lattice constants also occurs in C12-C18 compared to C4-C10 (Table S2). Thus, increasing the barrier length can result in a change of hole and electron confinement potentials, which in turn can influence the efficiency of carrier transport or confinement in the structure. As such, selecting an appropriate cation length is, among others, a practical way to fine tune the optoelectronic properties of layered perovskites

Effect of lattice distortion related to structural phase transitions

For a given composition, structural distortions can also be induced in the inorganic layer due to temperature or pressure driven structural phase transitions. It may have a noticeable influence on the hybridization of the metal-halogen states especially at the valence band, which in turn influences hole confinement. The effect of such phase transitions is inspected with C10, which adapts two orthorhombic structures with space group $Pbca$, recorded at 268 K and 293 K.⁶² Their corresponding confinement potentials are shown in Figure 5a and the relevant band alignment is shown in Figure S6. Here, DFT band gaps are used. The band gap of C10 increases from 1.9 eV to 2.1 eV by lowering the temperature (Figure S6), in agreement with the structure's phase color change from orange to yellow observed experimentally.⁶² Again, using the 2D map shows that the band gap increase results from larger octahedral tilt angles in the low temperature phase (Table S3, Figure 5b) which leads to the inversion of the respective amplitudes of V_h and V_e .

Effect of the perovskite layer thickness

The self-organized nature of 2D HOP confers them a high degree of flexibility for tuning the size and chemical composition of the constituent layers. An interesting example of such tuning with potential technological promises is the RP series of layered perovskites of general formula $(BA)_2(MA)_{n-1}Pb_nI_{3n+1}$. n corresponds to the number of octahedra in each perovskite layer and defines its thickness (Table S4).^{5,17,21,57,64} We consider $n=1-5$ (Figure 6a) members of the series and summarize their crystallographic information in Table S4. Figures 6b and c show the band alignments and the variation of the confinement potentials with increasing n of the perovskite layer, respectively. Here, the experimental continuum band gaps²¹ excluding excitonic contributions are used to align CBM (band alignments with DFT band gaps are provided in Figure S7). For all n , a type-I band lineup is obtained. A larger hole confinement potential V_h is deduced for all n with $n=3$ showing the largest

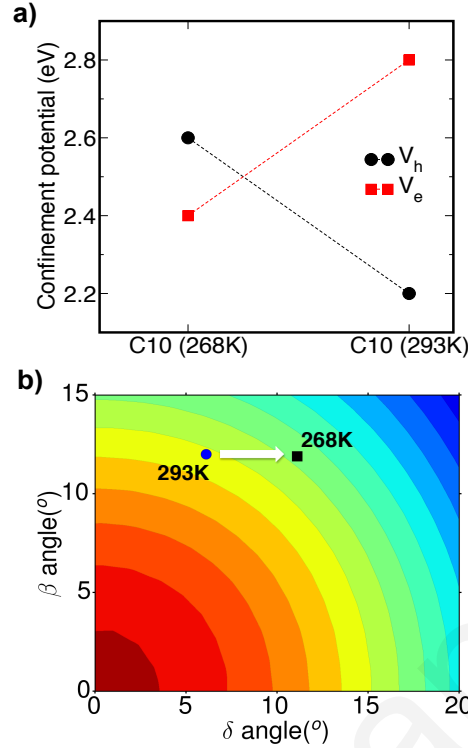


Figure 5: Assessing the impact of phase transition on band alignment. (a) Variation of the confinement potentials with phase transition. C10 (Figure 4a) is used for this example. (b) 2D color map of the electronic band gap variation of C10 with respect to (β, δ) as a result of phase transition.

$V_h - V_e$ difference. Noteworthy, the band gap decreases with increasing n which is a signature of diminishing quantum confinement effect in thicker perovskite layers.^{18,21} Hence, confinement potentials and quantum confinement effects are inter-related in these structures. In the section applying with the composite approach to band alignment, we showed that the shift of Pb 5d states between the composite and the inorganic layer is the same as their Hartree potentials difference. This does not exactly hold for $n=1$ where the difference between the Hartree potentials and Pb 5d states amounts to 0.14 eV. In general, we obtain 0.1 eV to 0.15 eV differences for other structures with one inorganic layer (see section S8 of the supplementary information). A similar conclusion was reached from the study of dielectric profiles, suggesting that the $n=1$ system is influenced by interface effects. Thus, the composite approach is best suited for $n > 1$. The effect of well thickness can be further analyzed

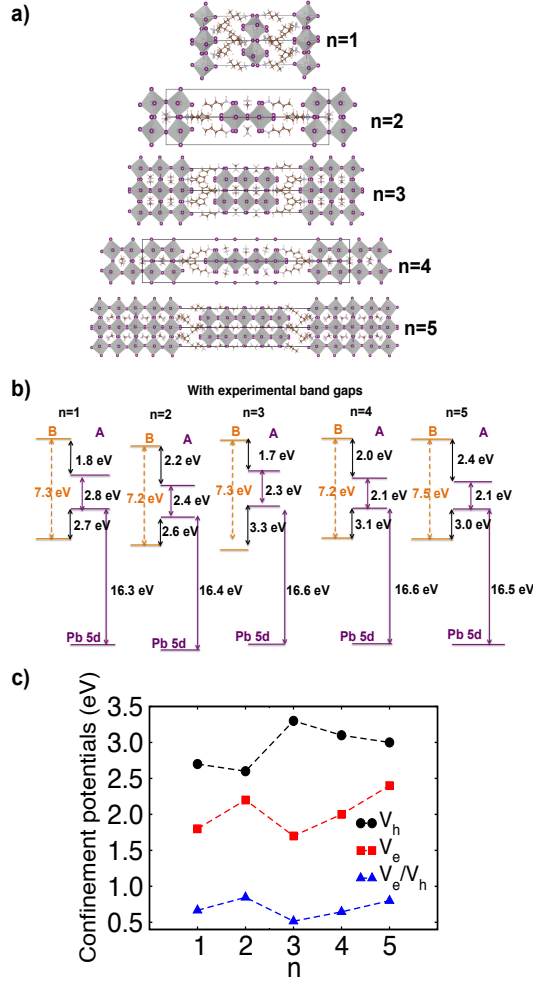


Figure 6: Band alignment for various well thicknesses. (a) Schematics of the HOP structures $(\text{BA})_2(\text{MA})_{n-1}\text{Pb}_n\text{I}_{3n+1}$ where n indicates the number of octahedron in the perovskite layer. These structures belong to the so-called Ruddlesden-Popper (RP) phases.¹⁷ (b) Computed band alignment for the RP HOP with $n= 1$ to 5 . Experimental band gaps, extracted from continuum absorption thresholds,²¹ are used for conduction band lineup. (c) Hole (circle) and electron (square) confinement potentials extracted from panel b with corresponding ratios.

in the framework of the effective mass model, whose limitations will be assessed.

Validity of the effective mass approximation

To investigate the validity of the effective mass approximation (EMA) in predicting electronic properties, such as the band gap, we further consider the RP series (Figure 6a). For a 2D system modeled as a quantum well with finite confinement potentials, the band gap can be

calculated using:

$$E_g = E_g^{3D} + E_{VB} + E_{CB}, \quad (3)$$

where E_{VB} and E_{CB} are ground state VB and CB energies respectively. Using the EMA with finite potential barriers, one obtains the following for the valence band:⁴⁰

$$\alpha = \frac{m_{VB}^b}{m_{VB}^w} k \tan\left(k \frac{L}{2}\right), \quad (4)$$

where

$$k = \sqrt{\frac{2m_{VB}^w E_{VB}}{\hbar^2}} \quad \text{and} \quad \alpha = \sqrt{\frac{2m_{VB}^b (V_h - E_{VB})}{\hbar^2}} \quad (5)$$

Here \hbar is Planck's constant and materials parameters entering these equations are defined in Figure 1a. For conduction band ground state energy E_{CB} , a similar Equation 4 can be written, where V_h and subscript VB are replaced with V_e and CB, respectively.

Subsequently, Equations 4 and 5 can be solved numerically for E_{VB} and E_{CB} entering Equation 3. This requires knowledge of effective masses of holes (m_h) and electrons (m_e). These masses can be computed either in the plane parallel to the perovskite (well) layer (m_{\parallel}) or along the stacking axis (m_{\perp}). From the composite approach, the pseudo-structures W and B are defined along the stacking axis (Figure 1b) and hence the effective masses in equation 4 are relevant to m_{\perp} for both holes and electrons. Nevertheless, the computation of effective masses presents some challenges. Indeed, it has been shown in HOP that accurate prediction of effective masses can only be achieved using self-consistent many-body and relativistic effects⁶⁵ that are currently out-of-reach for the layered structures of our study. Hence, we use effective masses calibrated to a reduced mass of about $0.200m_0$ as measured from magneto-absorption experiments⁶⁶ as well as those used to fit the experimental band gaps (Table S5).

Next, for different n values we employ the computed V_h and V_e (Figure 6b). Figure 7a

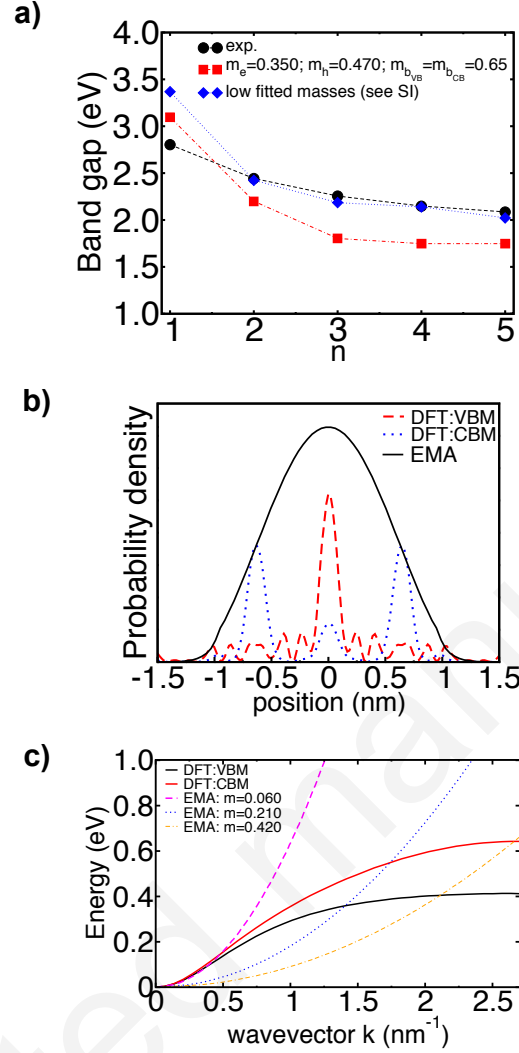


Figure 7: Limitations of the effective mass approximation when applied to layered HOP. (a) Experimental continuum absorption thresholds, band gaps,²¹ (black circles) compared to those calculated using the effective mass approximation (Eq. 2 and 3), considering either effective masses fitted so as to reproduce the experimentally determined reduced masses (red square) or significantly lower effective masses (Table S5) that fit reasonably experimental band gaps except for $n=1$ (blue diamond). (b) Probability densities of VBM and CBM obtained within DFT compared to that computed for the CB within the effective mass approximation (For DFT, $n=3$ from Figure 6a was used). (c) Energy dispersion as derived within the effective mass approximation (dashed-dotted lines) and DFT (straight-lines). DFT dispersion for both the CBM and the VBM evidences the non-parabolicity that cannot be captured in the effective mass approximation.

compares the experimental continuum band gaps²¹ with those deduced within the effective mass model. Using the effective masses calibrated to the experimental reduced ones ($m_e = 0.350m_0$, $m_h = 0.470m_0$) fails to reproduce the experimental band gaps; however, lower

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3 values of effective masses ($m_e \leq 0.105m_0$, $m_h \leq 0.220m_0$) are able to recover them (see
4 Table S5 for fitted masses). Nevertheless, these reduced masses are more than 3 times lower
5 than the previous ones. In all cases, the effective mass approximation fails for $n = 1$.^{39,67}
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7 This is consistent with the above mentioned difference between the shift of Pb 5d states and
8 the Hartree potentials difference, showing that for $n=1$ the system is dominated by interface
9 effects. Hence, the composite approach towards $n=1$ presents a larger error compared to the
10 other n members of 2D RP series ($n=2-5$).

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12 In contrast to earlier work,⁶⁷ the effective mass approximation cannot reproduce experi-
13 mental band gaps for realistic reduced effective mass values even for larger wells ($n=2$ to 5).
14 Thus, such approximation fails for this family of materials.

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16 This failure can be understood through the absence of common Bloch functions between
17 the inorganic well and the organic barrier, along the stacking axis, that is a basic assumption
18 in quantum well models through the continuity equation across the boundaries.³⁹ Besides,
19 the probability density profiles computed using VBM and CBM wavefunctions from DFT are
20 clearly different from the square of cosine probability density expected with the EMA (Figure
21 7b). As such, the envelope function approximation (another basic approach in semiconductor
22 heterostructures within the EMA) is not valid in these 2D HOP structures. Finally, Figure
23 7c compares the energy dispersions of VBM and CBM obtained with DFT to the variation
24 of the energy with respect to k -wavevector within the EMA. Only for k values lower than
25 0.5 nm^{-1} , DFT based VBM and CBM vary reminiscent to the effective mass model when
26 $0.06m_0$ value is used for the latter. With realistic effective masses, DFT based dispersions
27 cannot be reproduced by the effective mass model for any k values. This is a consequence
28 of the non-parabolicity of the DFT dispersions along the stacking axis, which differs from
29 the parabolicity assumption in the EMA. Therefore, first-principles methods are necessary to
30 accurately describe opto-electronic properties of this family of materials. We stress, however,
31 that there are systems (for instance, GaAs/AlAs heterostructure) for which the effective mass
32 approximation performs reasonably well.⁶⁸

Design guidelines

Having analyzed both the roles of material composition and structural properties on band alignment in 2D HOP, here we bring together a holistic view of the different effects aiming to formulate guidelines for device optimization. The major difference between type-I and type-II alignments is the spatial localization of holes and electrons. As such the type of 2D HOP alignment has a major influence on their excitonic properties. Resorting to 2D HOP containing only one inorganic layer with Cl as halogen can be good candidates for long-living excitons with enhanced emission properties due to its trending towards a possible type-II alignment. Such type-II alignment could be achieved by adopting organic barriers with narrower band gaps (by using, for example, conjugated moieties). The combination of 2D HOP with I as halogen and organic barriers with narrower band gaps may lead to lower confinement potentials in a type-I alignment, which is desirable for efficient carrier extraction from the perovskite layer. Moreover, self-assembled 2D HOPs with wider band gap molecular barriers potentially result in larger confinement potentials. Such systems with larger confinements could be interesting for optical confinement and may have potential application in guided optical emission devices as in edge light emitting diodes (ELED).⁶⁹ The thickness of the perovskite slab along with the length of the organic barriers are also playground for materials engineering of these devices. Finally, tuning distorted structures through thermal processing or doping is a possible optimization route to optimize carrier confinement potentials and subsequently, charge transport or emission efficiency.

Conclusions

We presented a systematic approach that treats two dimensional hybrid organic-inorganic as composite materials. The approach evaluates band alignments of the layered HOP allowing access to carriers' confinement potentials. The latter depend on the composition and structural properties of the layered perovskites. In particular, monolayered ($n=1$) HOPs

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3 with Cl as the halogen trend toward a type-II band alignment. Thus, adopting similar
4 confinement potentials for different 2D HOP structures within a modeling framework may
5 not be valid. The composite approach is shown to be useful for studying position depen-
6 dent dielectric profiles and enables clear separation of contributions of inorganic and organic
7 components. However, the composite approach is best suited for $n > 1$ since in monolayered
8 HOP ($n=1$) interface related effects are dominant. Effective mass approximation, a powerful
9 method frequently used in semiconductor quantum wells, has limited applicability to 2D
10 HOP. Its delinquency is related to the non-parabolicity of 2D HOP's band dispersions and
11 the absence of common Bloch functions along the stacking axis between the organic barrier
12 and the inorganic well. Finally, the composite approach allows to draft a set of practical
13 structure-electronic property relationships helpful to guide future materials design efforts.
14 Overall, this method is an attractive modeling route applicable to low-dimensional hybrid
15 organic-inorganic perovskites and other 2D materials.
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31 Methods

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34 The calculations were performed within the Density Functional Theory (DFT)^{70,71} as imple-
35 mented in SIESTA package⁷² with a basis set of finite-range of numerical atomic orbitals.
36 We used a van der Waals corrected functional with C09 exchange (VDW-C09)⁷³ to describe
37 the exchange-correlation term. Norm-conserving Troullier-Martins pseudopotentials were
38 used for each atomic species to account for the core electrons.⁷⁴ $1s^1$, $2s^22p^2$, $2s^22p^3$, $3s^23p^5$,
39 $4s^24p^5$, $5s^25p^5$ and $5d^{10}6s^26p^2$ were used as valence electrons for H, C, N, Cl, Br, I and
40 Pb respectively. Polarized Double-Zeta (DZP) basis set with an energy shift of 200 meV
41 and a real space mesh grid energy cutoff of 200 Rydberg were used for the calculations.
42 The Brillouin zone was sampled with a $4 \times 4 \times 1$ Monkhorst-Pack grid. The electronic and
43 dielectric properties were calculated using the experimental lattice parameters, and atomic
44 coordinates were transformed to their primitive cells whenever applicable for their band
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3 structure computations. The used experimental structures were recorded at room tempera-
4 ture otherwise specified. Since the positions of H are not well resolved by X-Ray diffraction
5 techniques, they were optimized within the primitive cells while keeping the other atomic
6 positions fixed. Although its effect is significant in Pb based perovskites, spin-orbit coupling
7 (SOC) was not taken into account in the band alignment procedures. This is because SOC
8 effect is important at the conduction band maximum⁵¹ and band gaps are already under-
9 estimated within DFT. Hence, we focused our discussion on the valence band alignment.
10 For the high-frequency dielectric constant profiles, slabs based on the respective systems
11 were constructed, and an electric field of 0.01 eV/Å was applied along the [001] direction to
12 compute the relaxed electron densities, as described elsewhere.^{45,46}
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Supporting Information Available

The supplementary information file is available free of charge.

- The supplementary information contains: validity of substituting inter-layer molecules by Cs for states close to the band gap; Band alignments with DFT band gaps when the experimental band gaps are used in the main text; Definition of the octahedral tiltings; Tables containing octahedral titl angles for different structures (C_m , n=1-5 2D RP, $(PMA)_2PbX_4$, C10 for phase transition effect); Note on band alignment for monolayered (n=1) structures; Effective masses used to fit band gaps.

This material is available free of charge via the Internet at <http://pubs.acs.org/>.

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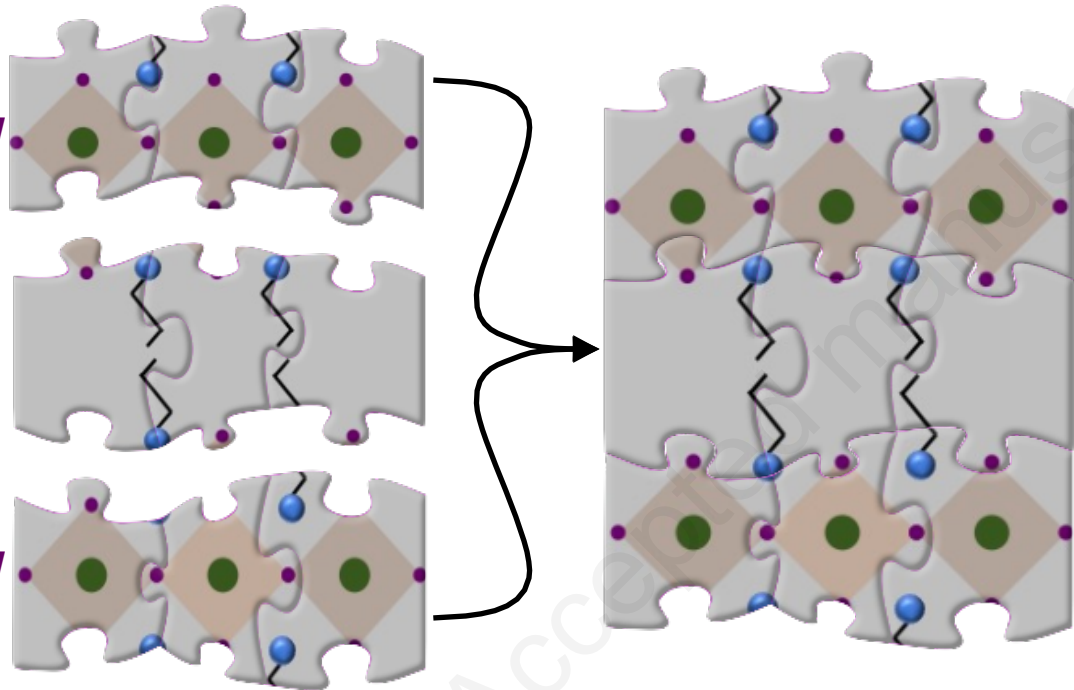
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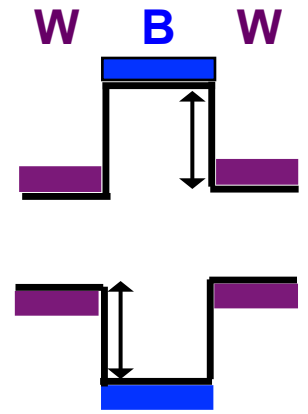
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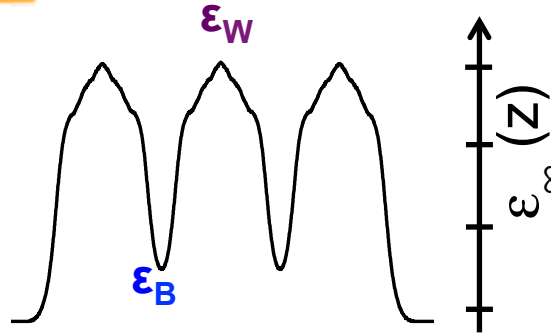
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Composite (W/B)



Band alignment



Dielectric contrast