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# Technology of medieval mortars: an investigation into the use of organic additives

L. RAMPAZZI<sup>1\*</sup>, M. P. COLOMBINI<sup>2</sup>, C. CONTI<sup>3</sup>, C. CORTI<sup>1</sup>, A. LLUVERAS-TENORIO<sup>2</sup>,  
A. SANSONETTI<sup>3</sup> AND M. ZANABONI<sup>2</sup>

<sup>1</sup>Dipartimento di Scienza e Alta Tecnologia, Università dell'Insubria, via Valleggio 11, 22100 Como, Italy

<sup>2</sup>Dipartimento di Chimica e Chimica Industriale, Università di Pisa, Via Risorgimento 35, 56126 Pisa, Italy

<sup>3</sup>Istituto per la Conservazione e la Valorizzazione dei Beni Culturali, Consiglio Nazionale delle Ricerche, Unità di Milano 'Gino Bozza', Area della Ricerca Milano 3 Bicocca, via Cozzi 53, 20125 Milano, Italy

\*Corresponding author

Laura Rampazzi, email address: laura.rampazzi@uninsubria.it; phone number: +390312386475; fax number: +390312386449

## Abstract

*This work proposes a multi-analytical approach to determine the additives in historical mortars, the use of which is widely described in bibliographical sources, but has rarely been reported in the literature. A protocol to thoroughly analyse mortars was created (optical microscopy, X-ray diffraction, infrared spectroscopy, thermal analyses and gas chromatography – mass spectrometry). These techniques had already been carried out on samples from various sites from the Roman to the modern era, finding that additives had only been used in the mortars from the internal masonry at our sampling site: the medieval military shipyard of Amalfi (Italy). The investigations yielded information on the production technology, and FT–IR and GC–MS revealed a saccharide material-based additive in the mortars, of plant origin. The FT–IR spectra suggested the presence of a natural gum, which has been used since ancient times to strengthen the cohesion properties of mortars and their resistance to tensile stress.*

Keywords: Amalfi, medieval mortars, organic additives, infrared spectroscopy, thermal analysis, microscopic analyses, gas chromatography – mass spectrometry, mortar technology

## INTRODUCTION

The most common components of historical mortars are known through the bibliographical sources on ancient technologies. A wide number of scientific papers have confirmed calcium carbonate as the most common binder component, together with a hydraulic binder, and usually with sand being the main aggregate phase. The use of additives in mortars has rarely been reported in papers on conservation science or the history of technology. Most of the literature focuses on the addition of organic compounds to the mortar paste for various purposes (Sickels 1981; Arcolao 1998; Carbonara 2007). In the past few years, we have analysed a large number of historical mortars (architectural mortars and stuccoes) from various sites, dating from Roman times to the modern era (Bugini et al. 2006; Rampazzi and Bugini 2006; Corti et al. 2013; Bugini et al. 2014). The aim of our multi-analytical approach (Rampazzi et al. 2010) is to shed light on the main features of mortars; that is, the composition of the binder and aggregate fractions, the morphology and structure, the binder/aggregate ratio and the mineralogical phases. Using this approach, we have previously studied the organic phases, and in the case of Baroque stuccoes, an egg-based additive was successfully

characterized by combining spectroscopic and chromatographic techniques (Rampazzi *et al.* 2008, 2012). In the current study, we found organic additives in mortars from the brick bedding in the masonry at our archaeological site in Amalfi (Italy)—a medieval shipyard/ building, known in Italian as *Arsenale* (which is not to be confused with the English word ‘arsenal’).

The Amalfi Arsenale is the only surviving example of a medieval shipyard in southern Italy (Gargano 1994). The ships and galleys were constructed in a building dating back to the ninth century, the ruins of which still remain. The grandiose Gothic construction was located on the waterfront, to the left of the ‘Porta della Marina’, and extended as far as the quay. It was renovated in 1240 and 1272, and was partially destroyed during a devastating sea-storm in 1343. The need to build using a mortar that was resistant to high humidity levels and sea salt spray had to be considered by the ancient builders. The need to use mortars with organic additives may have arisen due to this issue. Ten pillars divide the Arsenale into two stone brick passages (Fig. 1), with only 12 pointed arches still standing out of the 22 that had originally opened out towards the sea (Gargano 1994). The building is currently 44.60 m long and 6.65 m wide. The blocks of stone of the interior walls are joined together by a bedding mortar, which was sampled and analysed. The bedding mortars are whitish in colour and irregular in shape and thickness, the latter varying between 2 and 4 cm; the texture of the mortar is rough and the presence of organic additives cannot be discerned from its appearance.



Figure 1 *The Arsenale of Amalfi: details of the stone brick passages.*

This was an opportunity to shed light on the building techniques and medieval engineering, and to verify the recipe for the mortar mix of the historical masonry. Also, attempts to highlight the presence of organic substances were planned in the analytical survey.

The literature (Sickels 1981; Carbonara 2007) reports the use of additives with varying aims:

- improving the durability and hardness;
- reducing the drying shrinkage;
- increasing the resistance to traction;
- delaying/accelerating the setting time;
- facilitating adhesion on to the surface;
- stabilizing any emulsion, if present;
- increasing resistance to decay caused by freeze/thaw cycles; and
- increasing the plasticity and workability.

The use of organic compounds dates back to ancient times and became common in Europe in the medieval epoch under Byzantine influence. Pliny the Elder in his *Naturalis historia* and Vitruvius in his *De architectura* mention that the

Romans commonly used additives in order to enhance the resistance of mortars (Pliny 1968; Vitruvius 1999). The most common additives were animal glue, casein and other dairy products, beer, animal fat, linseed oil, albumen, blood and natural resins (Sickels 1981; Pecchioni *et al.* 2008). Pliny mentions a special mortar called '*maltha*', composed of lime slaked in wine, hogs' lard and fig juice, and designed for wet rooms, such as thermal baths (Arcolao 1998). Cennino Cennini, in his *Il libro dell'arte*, describes a mortar produced with lime putty and cheese, which adhered extremely well to stone blocks (Cennini 2009). In the 18th century, Francesco Milizia reports a similar mixture, with the addition of oil in order to enhance the impermeability of mortars used in aqueducts and cisterns (Arcolao 1998). The organic products mentioned above were used without any specific knowledge of their properties in the recipe for the mix and the performance they were able to induce. Most probably, they were simply added, making use of local resources, after which the resulting performance was observed empirically. Carbohydrate-based materials such as malt, sugar, molasses and gum arabic were also often used. Lime and sugar derivatives react quickly to form calcium saccharate, which strengthens the cohesion properties of the mortars and the traction resistance, especially in a warm and dry climate and in the case of large masonry, where the lime sets slowly (Carbonara 2007). Cardenas and co-workers investigated the role of polysaccharides in lime mortars also mixed with nopal (Cardenas *et al.* 1998). Cactus juice was commonly added to mortars in historical Mexican buildings, to improve the performance against water penetration and cracking. Cardenas demonstrated that the mechanical properties increased due to the formation of an interpenetrated network of cactus polysaccharide mucilage and  $\text{Ca}(\text{OH})_2$ . In the 17th century, fir pine cones were added to mortars: the tannin and sugar content of the cones ensured water resistance and the plasticity of the paste (Arcolao 1998). As far as modern mortars are concerned, additives are again used both to slow down and accelerate the setting time, and to increase the workability (Pecchioni *et al.* 2008). When added to repair mortars, cactus gum increased the mechanical performance (Alonso *et al.* 2002) and the carbonation front (Ventola *et al.* 2011). The plasticity and resistance of Portland cement are also increased when it is mixed with cactus extract, due to the formation of complexes with  $\text{Ca}(\text{OH})_2$  (Chandra *et al.* 1998).

Despite the above-mentioned investigations, to the best of our knowledge the determination of additives in historical mortars has rarely been documented. By extraction treatment with chloroform–ethanol and infrared analysis, Luxan *et al.* (1995) identified the presence of wax, resin and drying oil in ancient gypsum mortars from the St Engracia Basilica (16th century, Zaragoza, Spain). Elsewhere, HPLC analyses have pointed to the use of egg white in Byzantine lime mortars (Kurugol and Gulec 2012). Kuckova identified proteinaceous compounds in Romanesque mortars by peptide mass mapping and mass spectrometry analyses (Kuckova *et al.* 2009). The same authors also set up two efficient spectrometric methods to analyse protein additives in mortars (Krizkova *et al.* 2014). Sticky rice has been determined in 18th-century mortars (Zeng *et al.* 2008), while Duran *et al.* (2008) have highlighted the presence of cellulosic materials from fibres in mortars dating back to the 16th–17th centuries. Plant gums have been identified by ion chromatography in Macedonian tombs (Greece) wall paintings of the fourth to third centuries BC (Colombini *et al.* 2002).

The composite nature of bedding mortars is normally investigated by an integrated protocol of analyses, since a complementary approach is strongly recommended for reliable and complete results on the binder and aggregate fractions. X-ray diffraction (XRD) analysis and optical microscopy (OM) highlight the mineralogical phases. Fourier transform infrared spectroscopy (FT–IR) spectra determine the nature of the inorganic compounds and of the organic additive. Thermogravimetric analysis and differential scanning calorimetry (TG/DSC) determine organic compounds and quantify the binder used in the preparation of the mortar, indirectly determining the binder/aggregate ratio, while gas chromatography – mass spectrometry (GC–MS) identifies the organic phases. The analyses also help to verify the state of conservation of the mortars, determining decay markers such as efflorescences and gypsum. In this survey, the

abovementioned techniques constituted a multi-analytical protocol for the characterization of the mortars from the Arsenale in Amalfi.

## MATERIALS AND METHODS

The samples were collected by using a scalpel, according to the indications of UNI-Normal 3/80 (CNR–ICR 1980). Figure 2 shows the location of the samples. Small fragments of each one were analysed as polished cross-sections and thin sections.

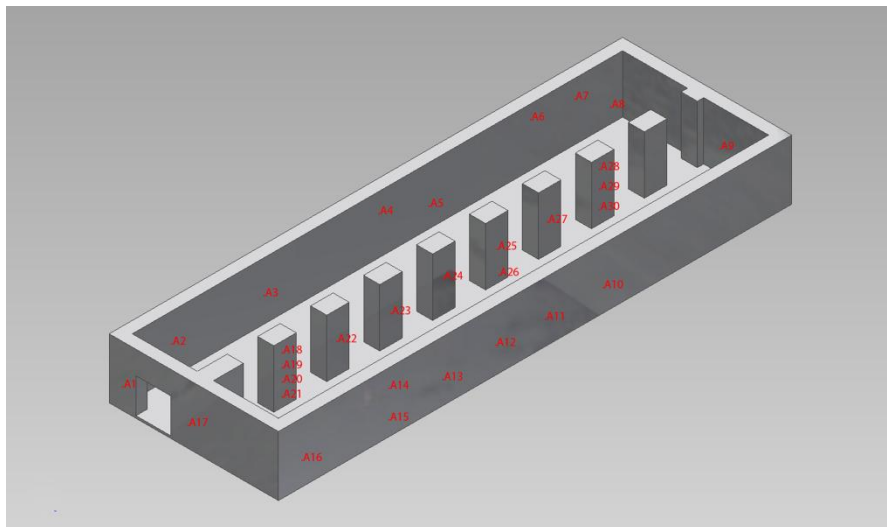


Figure 2 The location of the mortar samples

### *Optical microscopy*

Preliminary morphological observations were carried out on micro-fragments using a LeitzWild M420 stereomicroscope. Thin cross-sections (30  $\mu\text{m}$  thick) of the samples were observed in reflected light using a Leitz Ortholux microscope with an Ultropack illuminator, and in polarized light using a Nikon Eclipse E400Pol microscope. All microscopes were equipped with a digital image capture system.

### *X-ray diffraction*

Fine powders of the samples were analysed using a Bruker AXS D8 ADVANCE diffractometer with the following conditions: generator settings 40 kV and 40 mA; fenditure divergence slit 1.0; RS 0.2 mm;  $R = 217.5$  mm; radiation Cu– $K\alpha$ ,  $\lambda = 1.5418$  Å; pyrolytic graphite monochromator with diffracted ray; geometry  $\theta:\theta$ ; soller slits 2.3°; continuous scan 1°/min;  $\Delta 2\theta = 0.02^\circ$ ; scan range 5–55°; and deep quarter sampler 200  $\mu\text{m}$ .

### *Fourier transform infrared spectroscopy*

The samples were analysed as KBr (Sigma-Aldrich FT–IR Grade) pellets using a BioRad Excalibur Series FTS 3000 FT–IR spectrophotometer and a DTGS detector, in the transmission mode (400–4000  $\text{cm}^{-1}$ , 4  $\text{cm}^{-1}$  resolution, 16 scans). Regarding the extraction protocol, around 50 mg of the fine powders of the samples were put in a glass test tube, completely dipped into organic solvents (around 0.5 mL) and sonicated for 15 min. The extractions were conducted using hexane (Sigma-Aldrich 95% anhydrous), toluene (Sigma-Aldrich 99.8% anhydrous) and ethyl acetate (Sigma-

Aldrich 99% anhydrous), in order to extract non-polar and polar components, respectively (Doménech Carbó *et al.* 1996; Derrick *et al.* 1999; Colombini *et al.* 2003). The test tubes were centrifuged for 3 min. The soluble fraction was placed on a NaCl plate and analysed in the transmission mode after gentle evaporation of the solvent. Solvent blanks were prepared to check for contaminations. Extractions with deionized water (MilliQ Millipore grade) were also carried out. The soluble fraction was placed on a watch glass and left to evaporate. The residue was analysed as KBr pellets.

#### *Thermal analysis*

Determinations were performed using a NETZSCH STA 409 PC. Samples were placed in alumina crucibles, and the temperature programme ranged from room temperature to 900°C, at a heating rate of 10°C min<sup>-1</sup> under a pure nitrogen atmosphere. Raw data from thermogravimetric analysis (TG) and differential scanning calorimetry (DSC) were corrected on the basis of a background curve.

#### *Gas chromatography – mass spectrometry analysis*

Various selected samples were analysed according to a complex analytical procedure that characterizes all the organic compounds from the same microsample, avoiding interferences from inorganic media (Lliveras *et al.* 2010). The system used was a 6890N GC System Gas Chromatograph from Agilent Technologies, coupled with a 5975 Mass Selective Detector (also Agilent Technologies) with a single quadrupole mass spectrometer equipped with a PTV injector. The mass spectrometer operated in the electron impact (EI) positive mode (70 eV). The MS transfer line temperature was 280°C, the MS ion source temperature was kept at 230°C and the MS quadrupole temperature was 180°C. The samples were digested in a Milestone MLS-1200 MEGA microwave oven. The acidic hydrolysis of proteins was performed using powers of 250 W for 10 min and 500 W for 30 minutes, with 30 mL of 6N HCl at 160°C for 40 min. The acidic hydrolysis of polysaccharides was performed using a power of 500 W for 20 min, at 120°C, with 500 mL of TFA 2M. Saponification was performed using a power of 200W for 60 min at 80°C, with 300 mL of KOH in EtOH 10 wt%. The detailed analytical procedure and operating conditions are described elsewhere (Lliveras *et al.* 2010). The detection (LOD) and the quantification (LOQ) limits of amino acids, aldoses, uronic acids and fatty and dicarboxylic acids were calculated at a statistical significance level of 0.05 and used for the quantitative evaluation of each fraction. The LODs of the proteinaceous, lipid and saccharide materials were 0.35, 1.22 and 0.64 µg, respectively. The LOQs of the proteinaceous, lipid and saccharide materials were 0.75, 1.47 and 1.48 µg, respectively.

## **RESULTS AND DISCUSSION**

#### *Mortar investigation*

Although our focus was on organic additives, we also made a complete characterization of the mortar. In fact, combining the analysis of the inorganic fraction with the organic added substances enabled us to better understand the design criteria for the mortar mix. The mineralogical characterization of the mortars was performed by comparing the results achieved with X-ray diffractometry (Table 1) and polarized microscopy observations (Fig. 3). On the basis of the XRD results, samples were selected for the preparation of thin sections. Each mortar showed a high binder/aggregate ratio, with the aerial binder mainly consisting of calcite [CaCO<sub>3</sub>], even though there was frequently a small quantity of magnesium carbonates as nesquehonite [Mg(HCO<sub>3</sub>)(OH)·2(H<sub>2</sub>O)], magnesite [MgCO<sub>3</sub>] and dolomite [MgCa(CO<sub>3</sub>)<sub>2</sub>]. The wide variety of carbonates could be ascribed to the geological features of the area, which are characterized by

Liassic limestones and dolomites. The aggregate was exclusively silicatic, consisting mainly of feldspathoids as analcime [NaAlSi<sub>2</sub>O<sub>6</sub>·H<sub>2</sub>O] and leucite [K[AlSi<sub>2</sub>O<sub>6</sub>]]. These minerals are commonly found in volcanic rocks enriched in alkaline elements; in fact, they occur in the most abundant Monte Somma–Vesuvio lavas, together with augite [(Ca,Mg,Fe<sub>2+</sub>,Fe<sub>3+</sub>,Ti,Al)<sub>2</sub>(Si,Al)<sub>2</sub>O<sub>6</sub>] and olivine [(Mg,Fe)<sub>2</sub>SiO<sub>4</sub>].

Sample label	Calcite	Mg-Carbonates	Quartz	Feldspathoids	K-feldspars	Muscovite	Gypsum	Halite
A1	+++	+	-	++	-	+	-	+++
A2	+++	-	-	+	++	++	-	+++
A3	+++	++	-	+	-	-	-	++
A4	+++	-	-	+	+	-	+	-
A5	+++	+	-	+	+	-	-	+
A7	+++	+	-	+	-	+	-	+
A8	+++	+/-	+	+	-	+	-	+
A9	+++	+	-	++	-	-	-	-
A10	+++	+	-	+	++	-	-	-
A11	+++	+/-	-	-	-	-	++	+
A12	+++	++	-	+	-	-	-	+
A13	++	+++	-	+	-	-	-	+
A14	+++	-	-	+++	-	-	+	+
A15	+++	+	-	++	-	-	-	+
A16	+++	-	-	-	-	-	-	+/-
A17	+++	++	-	+++	+	-	-	+++
A18	+++	++	-	+	-	-	-	+++
A19	+++	+	-	++	-	+++	-	-
A20	+++	++	-	-	-	+	-	+++
A22	+++	++	+	+	-	+	-	++
A23	+++	+/-	+	++	-	+	-	++
A24	+++	+	-	++	+	-	-	++
A25	+++	+	-	++	-	-	-	++
A26	+++	+	-	++	-	+	-	++
A27	+++	++	-	+	+	-	-	++
A28	+++	+	-	++	-	-	-	-
A29	+++	-	+	+	-	++	-	-
A30	+++	+	-	++	-	-	-	-

Table 1 X-ray diffraction semiquantitative data. [Legend: +++ very abundant (>40%); ++ abundant (15-40%); + present (3-15%); +/- scarce (<3%)].

The occurrence of forsterite [ $Mg_2SiO_4$ ], the magnesium-rich end-member of the olivine solid solution series, verified by examining thin sections, confirmed the use of a volcanic aggregate belonging to the Monte Somma–Vesuvio lavas for the preparation of the mortars. Although to a lesser extent, the aggregate was also characterized by more common minerals such as K-feldspars [ $KAlSi_3O_8$ ], muscovite [ $KAl_2(AlSi_3O_{10})(F,OH)_2$ ] and quartz [ $SiO_2$ ].

Two granulometric fractions of aggregate were measured: quartz and feldspars have a finer granulometry, ranging from 0.1 to 0.3 mm; feldspathoids and forsterites show a coarser size, from 1 to 4 mm. Neither natural pozzolan nor other hydraulic additives have been detected. Some decay products were determined, such as gypsum [ $CaSO_4 \cdot 2H_2O$ ] and halite [ $NaCl$ ]. The first is probably caused by the sulphation process of the lime binder fraction; the latter is clearly due to the close proximity to the seashore. The pattern of XRD and the observations of the thin sections did not provide any evidence of organic substances in the mix.

The detection of quartz in the pillar mortars proves that in these samples, another kind of aggregate (sand) was used together with the volcanic fragments. In fact, it is well known that the co-presence in rocks of quartz and feldspathoids is not possible. The use of a sand aggregate was probably due to the need to enhance the mechanical resistance of the mortar.

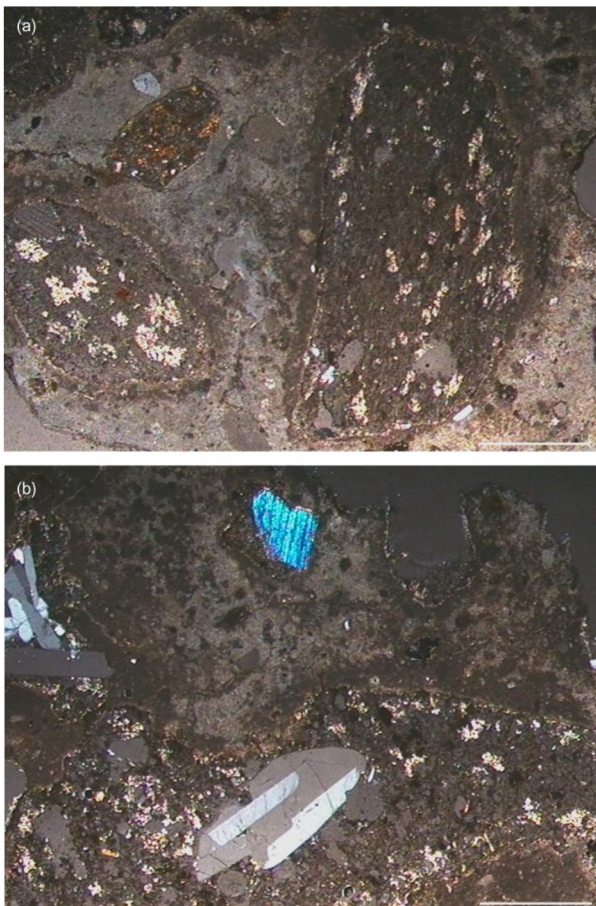


Figure 3 A thin section micrograph of the mortars observed in crossed polarised light: (a) sample A29; (b) sample A14. The main components of the mortar aggregates are feldspathoids, the coarse grains clearly visible in (a); olivine, the crystal with high interference colours in the upper part of (b); quartz and feldspars, with their low order interference colours, visible both in (a) and (b) (bar = 1 mm).

Infrared spectra confirmed the XRD results as far as calcite and silicates are concerned. Analyses by medium infrared spectroscopy can rarely determine the silicate family; thus it is not possible to discriminate between the different phases highlighted with the mineralogical investigations. The presence of calcite was suggested by the asymmetric C–O stretching band between 1430 and 1440  $cm^{-1}$  and by the absorbances at 874 (out-of-plane bending vibration), 713 (in-



plane bending vibration), 1798 and 2514  $\text{cm}^{-1}$  (combination modes). Silicates present bands in the regions of 1200–900 and 500–400  $\text{cm}^{-1}$ , due to their  $\text{SiO}_4$  tetrahedra. Gypsum was determined in low amounts only in a few samples, which showed characteristic peaks at 671 and 600  $\text{cm}^{-1}$  (S–O bending bands). Some organic matter was found in all the samples, in most cases just as traces, with peaks in the regions around 2800–2900  $\text{cm}^{-1}$ —that is, C–H signals—and around 1650  $\text{cm}^{-1}$ . This suggested the presence of the carbonyl function, which is normally in the range 1680–1820  $\text{cm}^{-1}$ , but which could have been lowered by conjugation effects or by the presence of a calcium salt, probably calcium saccharate. We thus made further analyses in order to identify the organic compound. A preliminary FT–IR survey on organic solvent-soluble portions of the most meaningful samples was designed, so as to focus on the organic family for GC–MS analyses.

### *Investigation of organic fraction*

*Infrared spectroscopy (FT–IR)* All the samples were extracted using various organic solvents, as described in the ‘Materials and Methods’ section, and the residues were analysed using a FT–IR spectrophotometer. Analyses of residues after extraction with hexane and toluene presented no absorbance peaks, except for not yet evaporated solvent. In contrast, spectra recorded after ethyl acetate extraction showed peaks typical of organic compounds. Generally, the C–H bond is suggested by stretching and bending absorbance signals, respectively, at 2921 and 2854  $\text{cm}^{-1}$ . The stretching absorbances of C=O and C–O bonds were observed around 1635  $\text{cm}^{-1}$  and 1230  $\text{cm}^{-1}$ , respectively. Figure 4 shows the spectrum of sample A27 after extraction with ethyl acetate. Some peaks are due to residues of the inorganic fraction; that is, calcite and silicates. Most samples left no residues after extraction with hot water, except for samples A7, A9, A15 and A23, the absorbance peaks of which are summarized in Table 2. The FT–IR spectra confirmed aliphatic C–H stretches around 3000–2800  $\text{cm}^{-1}$  and group C=O stretching absorption around 1650  $\text{cm}^{-1}$ . Figure 5 presents the spectra recorded after extraction of sample A23 with water and ethyl acetate.

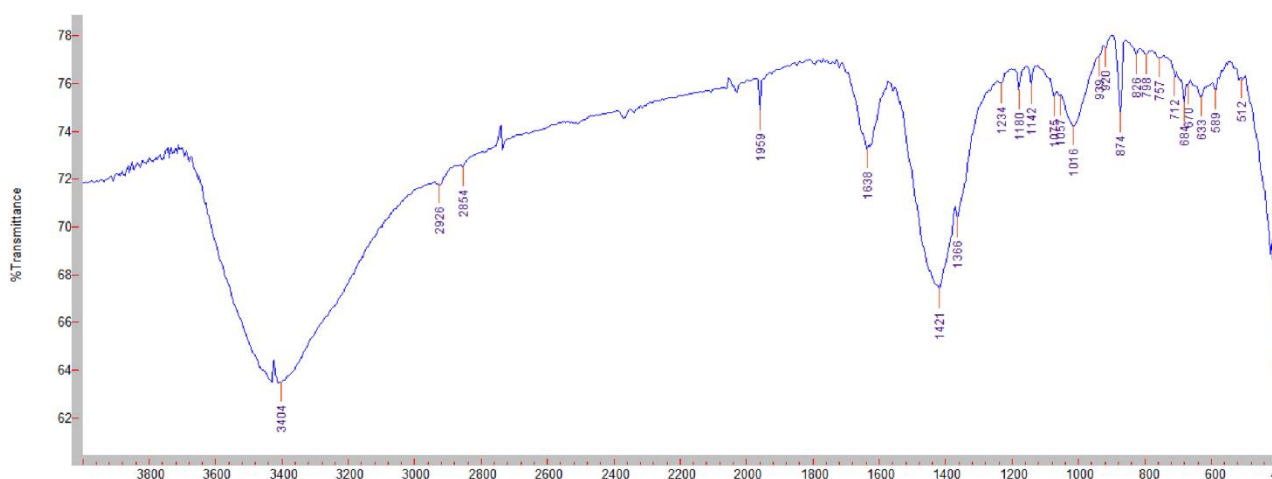


Figure 4 An FT–IR (infrared spectroscopy) transmission spectrum of sample A27 after extraction with ethyl acetate, showing the signals of calcite (1421, 874 and 712  $\text{cm}^{-1}$  and organic compounds (2926, 2854, 1638, 1366, 1234, 1075 and 920  $\text{cm}^{-1}$ ).

<i>Sample label</i>	<i>3600-3200cm<sup>-1</sup> stretching O-H</i>	<i>3000-2800cm<sup>-1</sup> stretching C-H</i>	<i>1650cm<sup>-1</sup> stretching C=O</i>	<i>1480-1300cm<sup>-1</sup> bending C-H</i>	<i>1300-900cm<sup>-1</sup> stretching C-O</i>
<b>A7</b>	3415		1633	1384	920
<b>A9</b>	3418	2855	1633	1384	
<b>A15</b>	3423	2924, 2855	1640	1383	
<b>A23</b>	3420, 3240	2926, 2855	1632	1384	

Table 2 Infrared absorbance peaks of residue after extraction with water

The comparison with reference standards of aged organic binders and the spectra found in the literature (Derrick 1989; Cui *et al.* 2007) suggested that the samples obtained after extraction with water and ethyl acetate might contain polysaccharides. The strong bands around 3300 cm<sup>-1</sup>, ascribed to O–H group stretching, and around 1080 cm<sup>-1</sup>, due to the C–O fingerprint, are particular features of polysaccharides and usually have the same intensity. Other clues are the weak signals around 2800–3000 cm<sup>-1</sup> (methyl and methylene symmetric and asymmetric stretching modes) and the moderately strong carbonyl signal at 1620 cm<sup>-1</sup>, assigned to the carboxylic group, and in particular the asymmetric stretching mode. Although some bands might be hidden by the calcite bands, most of the typical signals are clearly evident. A study of the literature enabled us to recognize a plant gum, possibly gum arabic. The peaks around 1070 cm<sup>-1</sup>, often present in the samples, could be assigned to β-(1→6) and β-(1→3)-linked-galactan; that is, polymerized galactose, the basic unit of gum arabic (Renard *et al.* 2006). Also, the peaks around 1420–1450 cm<sup>-1</sup>, present in some cases, probably indicate COO<sup>-</sup> asymmetric stretching of the compound. The aging of gum arabic initiates a new, weak absorbance at 1730 cm<sup>-1</sup>, due to oxidation reactions, which was sometimes observed in the samples (Caruso 2006).

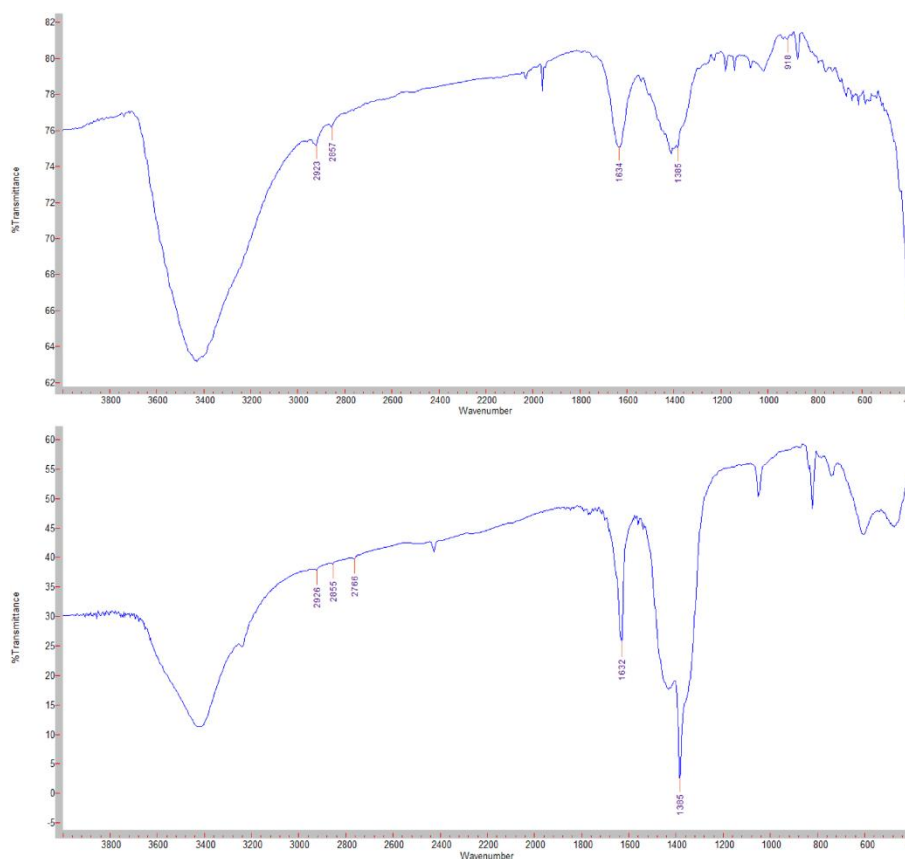


Figure 5 An FT-IR (infrared spectroscopy) transmission spectrum of sample A23 after extraction with ethyl acetate (above) and water (below), showing the signals of organic compounds around 2920, 2852, 1634, 1385 and 918 cm<sup>-1</sup>.

Due to natural and anthropic decay, only very low amounts of organic matter are generally recovered in ancient works of art. Thus, only the samples (A7, A8, A11, A15, A23 and A27) that after extraction with ethyl acetate showed the strongest diagnostic peaks belonging to organic compounds—for example, signals around 2900, 2800, 1730, 1630, 1230 and 1070  $\text{cm}^{-1}$ —were selected for GC–MS analyses, in order to have the best chance of determining their nature.

*Gas chromatography – mass spectrometry analysis (GC–MS)* There was no proteinaceous or lipid material in any of the samples, which showed a content below the limit of detection (LOD) of the procedure; however, the saccharide content was above the LOD of the procedure. The chromatogram of the saccharide fraction of sample A15 is presented in Figure 6, as an example of the chromatographic profiles obtained from the samples. Table 3 shows the relative percentage sugar composition, as well as the saccharide content, calculated as the sum of the nine sugars quantified, for all samples. The glycoside profiles reveal several sugars, suggesting that a polysaccharide material could be present in the samples. The absence of glucose and of peaks corresponding to ketose sugars enables materials such as starch to be ruled out as the additive added to the mortars. However, neither the quantitative nor the qualitative profiles are in agreement with those in the literature for reference materials (Lliveras-Tenorio *et al.* 2012a). Given the high content of inorganic material, responsible for the modification of sugar profiles (Lliveras-Tenorio *et al.* 2012b), the low percentage of organic material in each of the samples in terms of the total amount of sample analysed (around 0.01%) and the absence of other organic materials (e.g., proteinaceous) that could contribute to the saccharide profile (Lliveras-Tenorio *et al.* 2012b), it is not possible to identify the source of the saccharide material present in the samples. However, as both aldoses and uronic acids were present, the glycoside profiles seem to point to the presence of a polysaccharide material of plant origin.

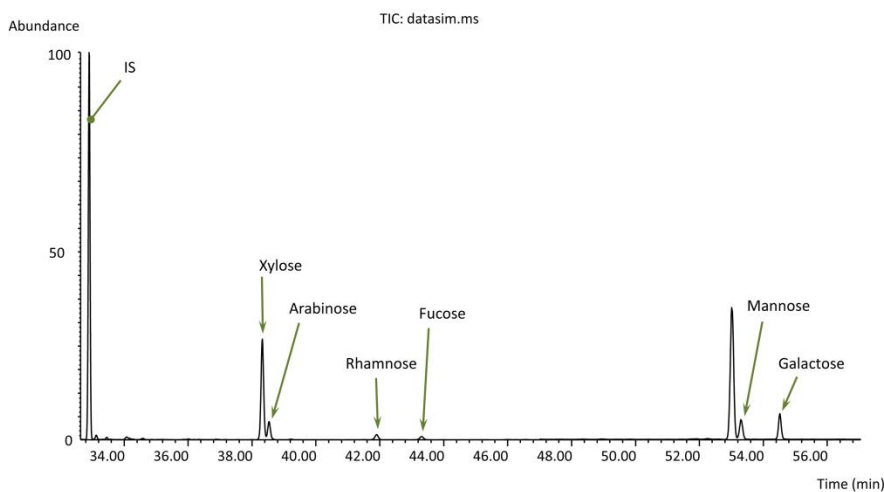


Figure 6 A chromatogram in single ion monitoring (SIM) of the saccharide fraction of sample A15. IS, internal standard mannitol.

Sample label	Xylose	Arabinose	Rhamnose	Fucose	Galacturonic acid	Glucuronic acid	Glucose	Mannose	Galactose	Saccharide content (µg)
A23	33.1	15.9	4.9	2.7	-	-	-	30.5	20.6	1.3
A27	23.4	20.7	2.6	2.6	-	3.7	-	20.2	26.8	1.1
A7	16.4	24.6	10.1	4.3	-	3.0	-	21.1	20.4	0.6
A8	35.1	34.3	3.9	2.3	-	-	-	17.6	13.0	1.2
A11	19.8	14.1	8.8	6.0	-	-	-	20.0	19.7	1.9
A15	14.6	6.1	3.9	2.3	-	-	-	28.7	21.0	1.1

Table 3 Relative percentage sugar composition of samples quantified.

*Thermogravimetric analysis (TGA)* Thermal analysis may also reveal the presence of natural gums, by the weight losses around 90°C and 320°C, respectively due to water loss and polysaccharide depolymerization (Zohuriaan and Shokrolahi 2004). The investigation was carried out on the same samples as analysed by GC–MS, confirming the presence of gum. In particular, a weight loss was observed around 340°C. The signal was weak, as the organic fraction was expected to be a minor component of the mortar sample. Figure 7 shows the TG/DSC curve of sample A23. The differential scanning calorimetry (DSC) thermogram shows a peak around 300°C, which may be associated with the exothermic transition typical of the depolymerization of natural gums (Zohuriaan and Shokrolahi 2004).

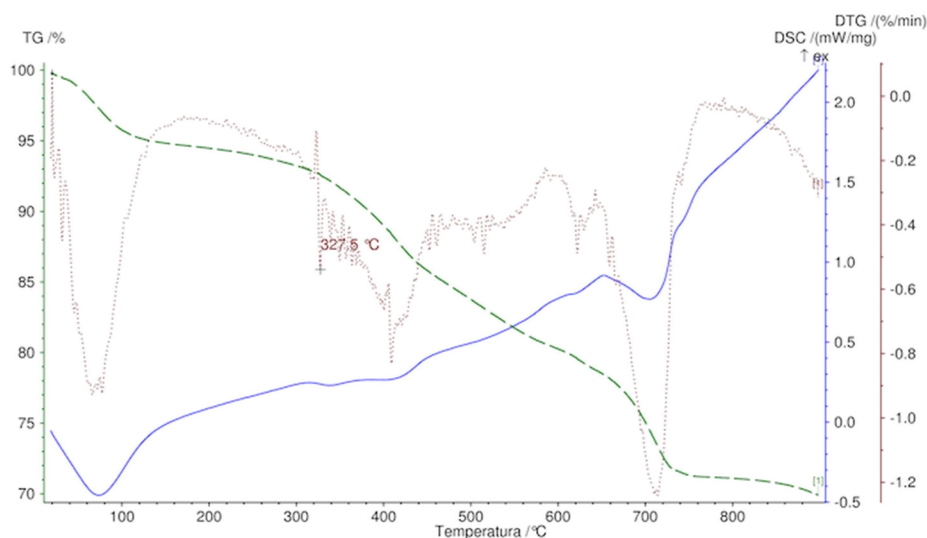


Figure 7 A thermogravimetric analysis and differential scanning calorimetry (TG/DSC) thermogram of sample A23, showing the DSC peak of depolymerization of natural gums at 340°C. Weight losses at 67, 410 and 712°C are due to water loss, organic compounds and CaCO<sub>3</sub> decomposition, respectively. Continuous curve, DSC; long-dashed curve, TG; dotted line, first derivative of TG curve.

## CONCLUSIONS

The mortars taken from the Arsenale were investigated using both mineralogical and chemical techniques. The mineralogical techniques revealed the minerals were present both in the aggregate and in the binder fractions. The results identified an aerial mortar, where the aggregate phase was made up of volcanic rocks. The presence of halite was due to the proximity to the seashore. The FT–IR analyses were important in determining the organic components and suggesting the presence of a saccharide material, possibly a natural gum. FT–IR confirmed its role as a rapid determination method for the preliminary screening, since the results influenced the choice of confirmatory sensitive analytical techniques, such as GC–MS and TGA. The GC–MS results enabled us to rule out the presence of lipid and proteinaceous materials and to confirm the presence of a saccharide material in all the analysed samples: the glycoside profiles were in agreement with the presence of a polysaccharide material of plant origin. The TGA investigations also revealed the presence of a natural gum, thus indicating the use of organic additives to strengthen the mixture and confirming the mortar recipes for the historical masonry. This study demonstrates that the presence of organic additives can be detected even if the mortar is rough and no clues are perceivable in its texture, due to its rough appearance. In fact, very often there are organic additives when a smooth and partially glossy surface is visible. This is often the case for stuccowork or mortars applied as a final render. Thus, evidence of organic additives in a mortar bedding is considered quite unique.

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