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Proceedings

Design of Novel Electrochemical Sensors for the Selective Detection of Glyphosate[†]

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Abstract: This study concerns the development of two molecularly imprinted polymers (MIP) based electrochemical sensors for glyphosate detection. In both cases polypyrrole (PPy) was chosen as matrix and glyphosate molecules were the templates. The main difference between the two strategies is related to the investigated electrode: gold surface in the first case, and ZnO nanorods vertically grown on ITO diazonium modified substrates in the second case. The limits of detection (LOD) of these sensors were about 10^{-13} M and 10^{-10} M respectively.

Keywords: glyphosate; omethoate; molecularly imprinted polymer (MIP); electrochemical sensor; atomic force microscopy (AFM)

1. Introduction

Glyphosate is the most frequently used herbicide worldwide. Its hazard potential is unclear, as it possesses amino acid-like structures which might interfere with the formation of amino acids and other chemicals in plants. Among the major concerns, glyphosate was found to affect the human hormonal system, chromosomal changes, tumors in the pancreas, and increased risk of skin cancer [1]. These health concerns have motivated the development of numerous analytical methods to track glyphosate in the environment as well as in body fluids. Despite their highly sensitive and accurate detection, many of the developed methods are costly, complicated, time-consuming or require extensive instrumentation.

An interesting alternative approach consists in the design of molecularly imprinted polymer (MIP) based electrochemical sensors [2,3] for the selective detection of glyphosate. The rationale for making MIPs, is that they are low-cost, sensitive and powerful analytical tools capable of determining low concentrations of this herbicide, inferior to the limits required by the European standards (EU limit in drinking water is $0.1 \mu\text{gL}^{-1}$ (5.9×10^{-10} M) [4]. To achieve high sensitivities and low limits of detection, a promising route consists in increasing the specific surface area of the sensors by incorporating ZnO nanorods, which have ultra-high surface/volume ratio, and coating them with the PPy-MIP [5,6].

2. Materials and Methods

2.1. Chemicals

Pyrrole, lithium perchlorate, glyphosate, omethoate, methanol (MeOH), acetic acid (AcCOOH) H₂SO₄ (95%) and H₂O₂ (30%), zinc acetate dihydrate (Zn(CH₃COO)₂·2H₂O, 99.99% of purity), sodium hydroxide (NaOH, purity ≈ 97%), hexamethylenetetramine (C₆H₁₂N₄, 99% purity) and ITO electrodes were purchased from Sigma Aldrich. Absolute ethanol was supplied by VWR Prolabo; Zinc nitrate hexahydrate (Zn(NO₃)₂, Purity ≈ 99%) was from Merck. Pyrrole was purified before its use by filtering through basic alumina column and stored in dark under argon at −4 °C. Lithium perchlorate was used as a supporting electrolyte for electrochemical measurements. Glyphosate and omethoate were used as received.

2.2. Electrochemical Measurements

Electrochemical measurements were performed at room temperature with a CHI 650 electrochemical workstation. A conventional three-electrodes system was used with a steel grid, a saturated calomel, and gold (or ITO) substrates as auxiliary, reference and working electrodes respectively.

2.3. Atomic Force Microscopy

Atomic force microscopy measurements were carried out at room temperature, with a Nanosurf (Liestal, Switzerland) easyScan 2 Flex AFM system in the dynamic force mode, with cantilever's resonance frequency of about 165 kHz. Commercially available tips with typical curvature radius of 6 nm were used (ACLA silicon AFM probes, from AppNANO, Mountain View, CA, USA).

2.4. Design of MIP Based Sensors

Two routes were investigated to design the MIP based sensors (Figure 1). Prior to any electrochemical process, the working electrodes were cleaned and activated by a drop of a piranha solution (98% H₂SO₄/30% H₂O₂ 1:1 V/V), during 20 min.

For the first route, the films were electrodeposited by chronoamperometry (CA) to control their growth. A thin polypyrrole blocking layer was deposited by CA before MIP electrosynthesis to prevent the formation of glyphosate complexes with gold electrodes (Figure 1a).

In the second strategy, MIP coated ZnO nanorods were considered to attempt enhancement of sensitivity to glyphosate. ZnO nanorods were prepared as follow [7]: first, 75mM of zinc acetate were mixed homogenously and stirred during 6h at 75°C with 100mM of NaOH in 75mL of ethanol to obtain ZnO nanoparticles which were deposited by spin coating on Gold (ITO) surfaces modified by diazonium salts. The further growth step was carried out by immersing the electrodes in a beaker which contains 500mL of water, 30mM of zinc nitrate hexahydrate and 30mM of hexamethylenetetramine (HMT). The mixture was heated up to 95 °C and stirred for 5 h. Finally, the samples were rinsed with deionized water and dried at 120 °C.

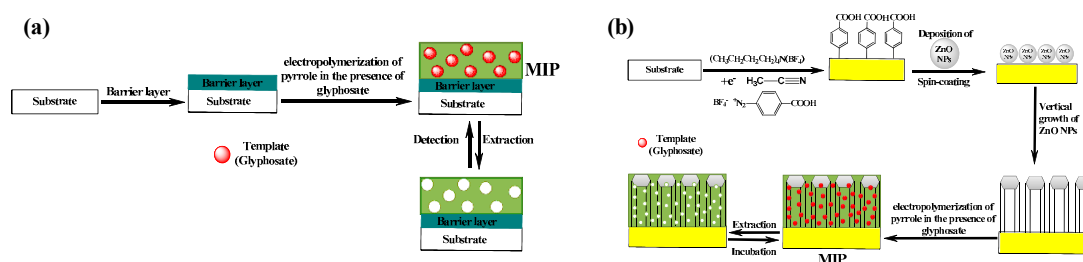


Figure 1. Schematic representation of MIPs preparation (a) PPy electropolymerized MIP and (b) ZnO-MIP based sensors.

3. Results and Discussion

3.1. Electrochemical Measurements

Square wave voltammetry (SWV) was used to record the sensors response to glyphosate at different concentrations varying from 10^{-13} M to 10^{-4} M. Results presented in Figure 2 indicate that the current peaks values increase with glyphosate concentration and that the peak corresponding to the lowest concentration remains clearly distinguishable, with respect to the response to the blank. This value is three orders of magnitude lower than the limit value set by the EU for drinking water.

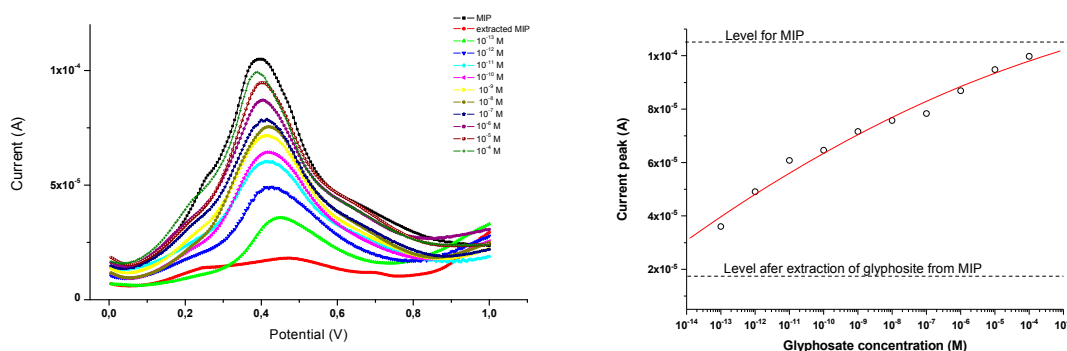


Figure 2. Follow up of glyphosate concentration variation, by square wave voltammetry (left), and current variation versus glyphosate concentration (right).

Selectivity tests were performed with omethoate, an organophosphorous pesticide widely used in the EU. Figure 3 shows that the sensors’ response for omethoate at concentration as high as 10^{-4} M is comparable to that for glyphosate at concentration of 10^{-13} M. These results are encouraging and other pesticides are under test.

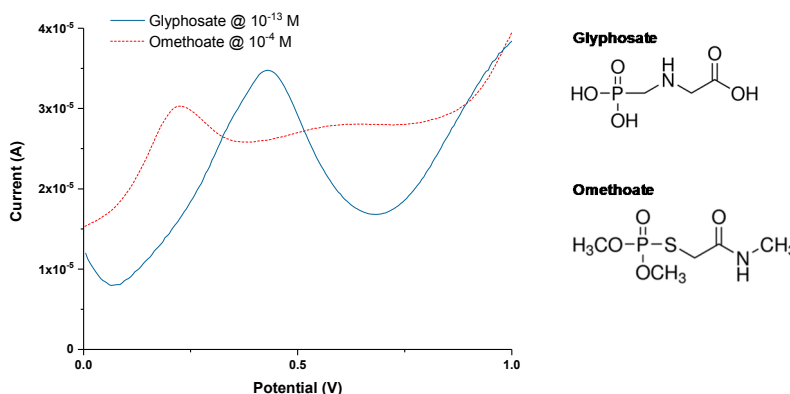


Figure 3. Current variation versus potential corresponding to glyphosate and omethoate detection by MIP based sensor (left); chemical structure of the two pesticides (right).

3.2. New Route and Perspectives

To reach high sensitivities and low LOD values, we chose to increase the specific surface area of the MIP by incorporating ZnO nanorods, due to their ultra-high surface/volume ratio. The preliminary experiments show that is possible to create nucleation sites on nonflexible electrodes (ITO or gold) for vertical growth of ZnO nanorods (Figure 4c). Nevertheless, by coating ZnO nanorods electrochemically with PPy-glyphosate-MIP, we do not reach the expected enhancement of sensors’ metrological performances. Indeed, the LOD obtained in this case, of about 10^{-10} M, is in the same order of magnitude as the EU recommendations but largely superior to that obtained with the first strategy (10^{-13} M). The main reasons lie in the distribution density and inclination of the ZnO

nanorods, as shown in AFM images (Figure 4c). Optimization of these crucial parameters and nanorod spacing, essential for significant improvement of the sensors performances.

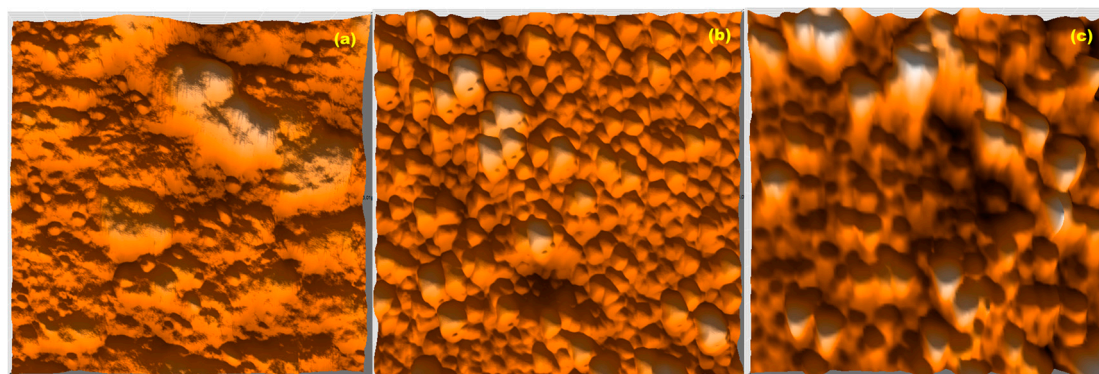


Figure 4. AFM images ($3 \times 3 \mu\text{m}^2$) of (a) bare ITO electrode; (b) After deposition of ZnO nanoparticles; (c) after further ZnO nanorods growth.

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Conflicts of Interest: The authors declare no conflict of interest. The founding sponsors had no role in the design of the study; in the collection, analyses, or interpretation of data; in the writing of the manuscript, and in the decision to publish the results.

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