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Photomodulation of DNA-templated supramolecular assemblies

Jenifer Rubio-Magnieto,^[a] Tuan-Anh Phan,^[b,c] Mathieu Fossépré,^[a] Valérie Matot,^[a] Jérémie Knoops,^[a] Thibaut Jarrosson,^[b] Pascal Dumy,^[c] Françoise Serein-Spirau,^[b] Claude Niebel,^{*,[b]} Sébastien Ulrich,^{*,[c]} and Mathieu Surin^{*,[a]}

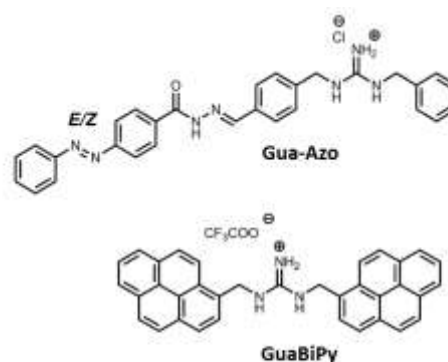
Abstract: A new type of DNA ligand that contains a phosphate-binding group and a photoresponsive azobenzene moiety is reported. When the azobenzene is in *trans* configuration, the ligand binds to the minor groove of a double-stranded DNA, whereas it partially desorbs upon *trans-cis* isomerization with light. The ability to photoswitch the ligand upon interaction with DNA is evidenced by (chir)optical signatures, and deciphered by the differences of binding geometry, stability, and dynamics of the DNA/ligand complexes for the two isomers. We exploit these properties to photomodulate DNA-templated self-assembly, through the incorporation of another π -stacking DNA ligand, which together with the photoresponsive ligand form mixed supramolecular complexes along DNA. Our study demonstrates that well-designed photo-responsive DNA binders can be used to modulate multicomponent supramolecular DNA assemblies.

Introduction

The fine control over the formation of supramolecular assemblies and polymers via DNA templating has recently gained great interest to achieve functional nanostructures, with possible applications in sequence-controlled polymers, (bio)sensing, and delivery.^[1-11] In this regard, achieving DNA-templated multicomponent assemblies for which the organization and function can be modulated upon external stimuli is a fascinating challenge.^[6,12-18] Among the different stimuli, light is a practical trigger for enabling responsive materials,^[13,17] as exemplified by the incorporation of photoswitchable ligands into biomolecular systems in view of

controlling their conformation and activity in a reversible manner.^[12] Several groups have already developed photoresponsive DNA molecules by covalently linking photoswitching molecules within the nucleic acid structure (e.g. azobenzenes in between base pairs).^[15,19-27] This covalent approach permits the formation and dissociation of DNA duplex by illuminating the azo-modified oligonucleotide with UV light, which is valuable for the photoregulation of bioreactions.^[19,22,28-29] However, the main drawback of this incorporation is the difficulty in synthesis/preparation of these covalently modified-DNA, which restricts its use. In this context, the use of photoswitchable molecules that can interact with the DNA *via* supramolecular recognition offers an alternative approach.^[24,30] For instance, spiroopyran or azobenzene derivatives were used as photosensitive DNA intercalators, and the use of an adequate light source permitted to switch between association/dissociation of the intercalator with DNA or between assembly/disassembly of DNA nanostructures.^[31-32]

In this article, we report on the design and characterization of a photoswitch (**Gua-Azo**, Chart 1) containing a guanidinium group, which is purposeful for DNA binding and has been shown to be valuable for applications in DNA sensing and gene delivery.^[25,33-42] The photoswitch is brought by an azobenzene moiety, which can undergo a reversible photoisomerization with UV light (365 nm, *trans* \rightarrow *cis*, *E/Z*) and visible light (450 nm, *cis* \rightarrow *trans*, *Z/E*).



ssDNA d(T)₄₀: 5'-TTT-TTT-TTT-TTT-TTT-TTT-TTT-TTT-TTT-TTT-TT-3', dT₄₀

dsDNA d(R)₂₀d(R)_{rev 20}: 5'-CGT CAC GTA AAT CGG TTA AC-3'
3'-GCA GTG CAT TTA GCC AAT TG-5', dsR₂₀

stDNA : salmon DNA , ~2000 base pairs

Chart 1. Top: Chemical structures of **Gua-Azo** photoswitch and **GuaBiPy** ligand. Bottom: DNA templates used in this study.

[a] Dr. J. Rubio-Magnieto, Dr. M. Fossépré, V. Matot, J. Knoops, Prof. M. Surin

Laboratory for Chemistry of Novel Materials
Center of Innovation and Research in Materials and Polymers
University of Mons – UMONS,
20 Place du Parc, 7000 Mons, Belgium
E-mail: mathieu.surin@umons.ac.be

[b] T.-A. Phan, Dr. T. Jarrosson, Prof. F. Serein-Spirau, Dr. C. Niebel
Institut Charles Gerhardt, UMR 5253, CNRS, Université de Montpellier

ENSCM, Ecole Nationale Supérieure de Chimie de Montpellier,
240 Avenue du Professeur Emile Jeanbrau, 34296 Montpellier
cedex 5, France. E-mail: claudie.niebel@enscm.fr

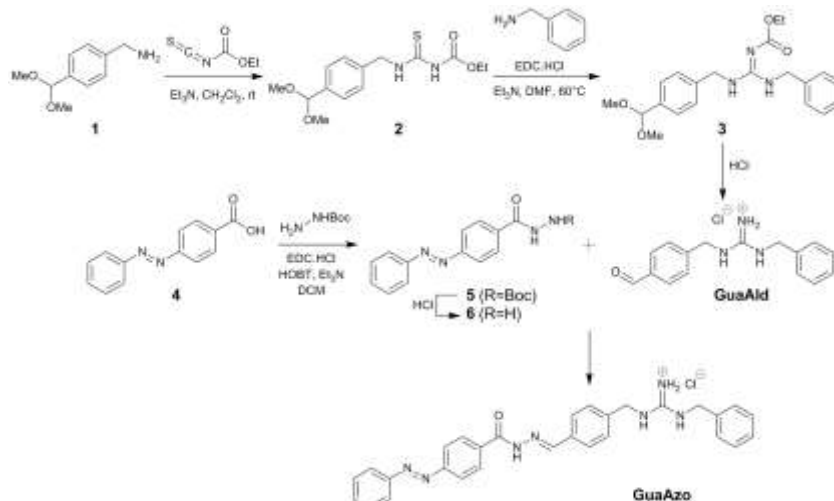
[c] T.-A. Phan, Prof. P. Dumy, Dr. S. Ulrich

Institut des Biomolécules Max Mousseron (IBMM)
UMR 5247 CNRS, Université de Montpellier, ENSCM,
Ecole Nationale Supérieure de Chimie de Montpellier,
240 Avenue du Professeur Emile Jeanbrau, 34296 Montpellier
cedex 5, France.

E-mail: sebastien.ulrich@enscm.fr

Supporting information for this article is given via a link at the end of the document.

We demonstrate that the binding of this photoswitch molecule along the DNA can be modulated by light: when the azobenzene is in *trans* conformation, **Gua-Azo** binds DNA and show induced chiroptical signals, whereas the *cis* conformation has a lower affinity and show no induced chiroptical signal. This is rationalized by molecular modelling simulations, showing that the *trans* isomer perfectly accommodate into the DNA minor-groove of double-stranded DNA (dsDNA), whereas the geometry of the *cis* leads to partial desorption of the azobenzene moiety from DNA minor-groove, a lower stability and different dynamics of the complex. Moreover, the photoresponsive DNA binder **Gua-Azo** can form mixed assemblies with another guanidinium-based compound such as **GuaBiPy** (Chart 1), which permits the photomodulation of the supramolecular assembly along the DNA strands and therefore represents a step forward toward the photo-control of multicomponent DNA-templated self-assemblies.



Scheme 1. Synthetic route for the preparation of compound **Gua-Azo**.

Results and Discussion

Design and synthesis of **Gua-Azo**

We have previously reported the supramolecular self-assembly of **GuaBiPy** (Chart 1) along single-stranded and double-stranded DNA templates.^[11,43] This initial study revealed that, in addition to salt-bridge interactions between the guanidinium groups and the DNA phosphodiester, π -stacking interactions are essential for stabilizing these DNA-templated complexes. We reasoned that the insertion of an azobenzene photoswitch would enable the photocontrol of this type of supramolecular assemblies given that the *E* isomer (*trans*) is planar and should therefore stabilize the assemblies, whereas the *Z* isomer (*cis*) is forced to be non-planar, which should destabilize the assembly.

Based upon our previous experience in using acylhydrazone coupling reactions for synthesizing bioconjugates,^[43,44] we selected a strategy that involves preparing **Gua-Azo** compound through the reaction between aldehyde **GuaAld** and the azohydrazone **5** (Scheme 1 and Figs. S1-S13 in the Supporting Information). **GuaAld** was prepared in three steps involving: i) the formation of thiourea **2** by addition of 4-(dimethoxymethyl)benzylamine **1** onto ethoxycarbonyl isothiocyanate; ii) the subsequent EDC-mediated coupling with benzylamine to afford compound **3**, and iii) the hydrolysis of the ketal and carbamate protecting groups. 4-(phenylazo)benzohydrazide **6** was obtained in two steps: peptide coupling between 4-(phenylazo)benzoic acid **4** and *tert*-butyl carbazate affords compound **5** which was deprotected in acidic conditions to furnish the desired hydrazide **6**. Finally, compounds **GuaAld** and **6** are combined to yield after acylhydrazone ligation the desired **Gua-Azo** in a 40% isolated yield.

Photoisomerization of **Gua-Azo**

The photoisomerization experiments were carried out using a LED source by illuminating a solution of **Gua-Azo** compound ($\sim 120 \mu\text{M}$) at 365 nm during 5 minutes, yielding the *trans* \rightarrow *cis* (*E/Z*) isomerization of the azobenzene moiety. The UV-Vis spectra (Figure 1) showed a decrease of the initial π - π^* absorption band at 330 nm with a large blue shift (~ 20 nm) and a weak increase of the n - π^* absorption band at around 430 nm, narrowed compared to the absorption of the *trans* isomer in this wavelength range (the n - π^* electronic transition is only allowed for the *cis* isomer).^[20,25] Under illumination of the solution at 450 nm, we observed the recovery of signals of the *trans* form, both on the intensity of the π - π^* transition and the decrease of the n - π^* absorption band, thereby demonstrating the reversible and photocontrolled *cis* \rightarrow *trans* (*Z/E*) isomerization.

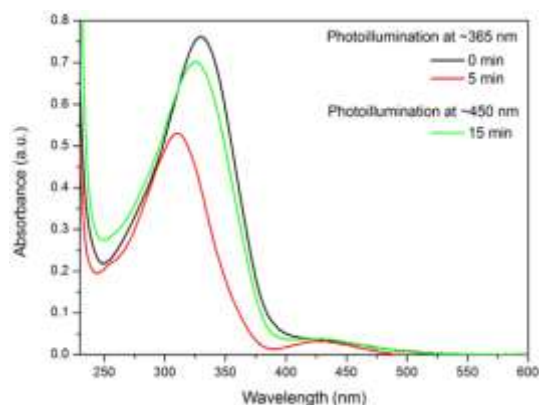


Figure 1. UV-Vis absorption spectra of **Gua-Azo** ($\sim 120 \mu\text{M}$) in Milli-Q water upon illumination. [**Gua-Azo**]_{stock} = 10 mM in DMSO.

^1H NMR studies in d_6 -DMSO also confirmed the reversible photoswitch of **Gua-Azo** compound. Upon irradiation at 365 nm complete conversion of the *trans* isomer into the *cis* isomer takes place within 15 min and the subsequent *cis* \rightarrow *trans* back-isomerization was achieved by excitation of the $n\text{-}\pi^*$ absorption band (Fig. S14).

Photoisomerization of Gua-Azo in presence of DNA

DNA binding of *trans* isomer of Gua-Azo. The interactions between **Gua-Azo** and DNA were studied by UV-Vis and Circular Dichroism (CD) at different molar ratios at 20 °C. For that, **Gua-Azo** in its stable *trans* (*E*) configuration was mixed with selected DNA, either with short oligonucleotides (ODN, dT₄₀ or dsR₂₀) or long genomic DNA (salmon sperm DNA, stDNA), see Chart 1. With a homopyrimidine single-stranded DNA dT₄₀, the CD spectra show the characteristic CD bands of DNA, centred at 276 nm and 250 nm, which are less intense upon the addition of **Gua-Azo** (see Figure 2). Interestingly, an induced circular dichroism (ICD) appears in the 300-400 nm range (see red line in Figure 2b), *i.e.* where the **Gua-Azo** absorbs albeit the pure compound is CD-silent (Fig. S15-S16).

This positive ICD signal shows a maximum intensity at a molar ratio around 1:50 in dT₄₀:**Gua-Azo** (Fig. S15), showing that the **Gua-Azo** adopts a chiral organization upon DNA binding. Note that ICD signals of ligands upon DNA binding were previously reported for other guanidinium derivatives,^[11,32,36] including **GuaBiPy** for which the maximum ICD is around 1:20 in dT₄₀:**GuaBiPy** (single-stranded DNA template).^[11] The same experiments were repeated on double-stranded DNAs (dsDNA), a duplex oligonucleotide (dsR₂₀) and a long dsDNA (stDNA). Upon addition of **Gua-Azo** with each of the dsDNA, important positive ICD signals also appear in the 300-400 nm region (Figure 2d,f), similar to that observed with the ssDNA. However, for the long DNA, we observed a bisignate (-/+) ICD signal in the **Gua-Azo** region with a zero crossing at around 403 nm.

DNA binding of *cis* isomer of Gua-Azo. We added a solution of DNA right after having UV illuminated a **Gua-Azo** solution under a LED (working wavelength at around 365 nm), *i.e.* yielding a large proportion of **Gua-Azo** in *cis* form. The UV-Vis absorption spectra of the [**Gua-Azo**_{cis} · DNA] mixture (see green lines in Figure 2) show a decreased intensity of the $\pi\text{-}\pi^*$ band and a change in the absorption maxima towards lower wavelengths ($\lambda_{max} \sim 313$ nm), which corresponds to the λ_{max} of the **Gua-Azo** in *cis* configuration (~ 310 nm). Importantly, we observe the appearance of a weaker $n\text{-}\pi^*$ absorption band in the visible region (~ 429 nm), as shown with pure **Gua-Azo**_{cis}. The CD spectra of complexes with oligonucleotides (either dT₄₀ or dsR₂₀) show no ICD signal in the range where the **Gua-Azo** absorbs (see Figure 2b,d), which indicates that the *cis* conformer does not adopt a preferential chiral organization upon binding with the short DNA. In contrast, with the long dsDNA (stDNA), an ICD bisignate signal is observed (Figure 2f) but weaker than that for the *trans* form.

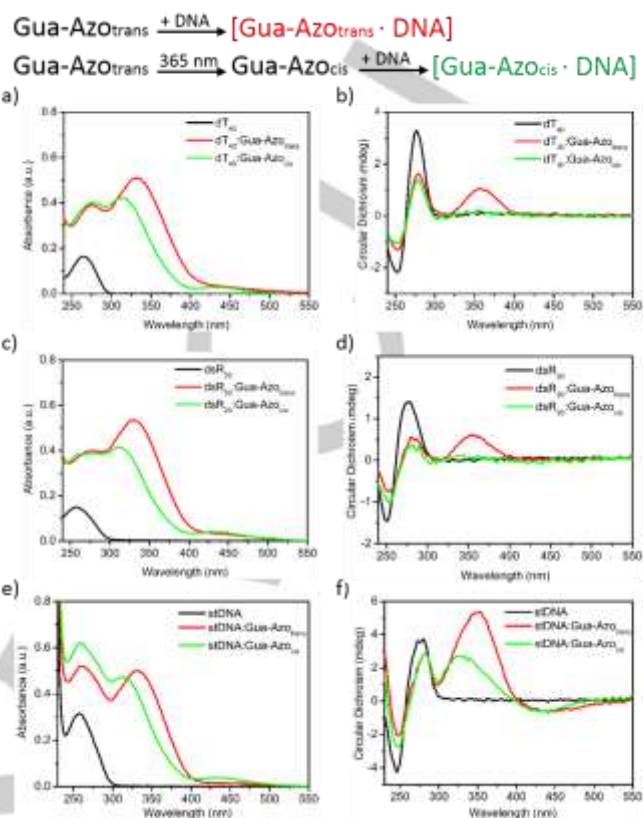


Figure 2. Changes in UV-Vis (left) and CD (right) spectra of several DNA/**Gua-Azo** mixtures at 20 °C in MQ-Water before (red lines) and after illumination at 365 nm (green lines): (a-b) dT₄₀/**Gua-Azo** at 1:50 molar ratio; (c-d) dsR₂₀/**Gua-Azo** at 1:50 molar ratio; and (e-f) stDNA_{bp}/**Gua-Azo** at 1:1 molar ratio.

In situ *cis* \rightarrow *trans* photoisomerization. In order to use **Gua-Azo** as photo-controllable DNA-binding switch, we considered a *cis-trans* cycle. A solution of the complex [**Gua-Azo**_{cis} · DNA] was illuminated at 450 nm (*cis-to-trans*) for 15 minutes. We observed a red-shifted absorption that coincides with the characteristic band of **Gua-Azo**_{trans} and with the vanishing of the $n\text{-}\pi^*$ band (see Figure 3), which is consistent with a *trans* form of **Gua-Azo** within the mixture. Interestingly, the ICD from [**Gua-Azo**_{trans} · DNA] complex is recovered for the short oligonucleotides upon exposure at 450 nm (Figure 3b,d), which indicates that the photoisomerization process could be reversed to give the *trans* isomer. Upon *cis-to-trans* photoisomerization with stDNA, the bisignate ICD from [**Gua-Azo**_{cis} · DNA] turns into a positive monosignate signal, red-shifted ($\Delta\lambda_{max} = 27$ nm) and with a λ_{max} similar to the one observed for [**Gua-Azo**_{trans} · DNA] (Figure 2f, 3f), which indicates that the photoisomerization occurs when the ligand is bound to DNA. In terms of comparison, Baigl *et al.* pointed out that other guanidinium-azobenzene derivatives possessing either one or two terminal guanidinium groups (with a flexible spacer between the guanidinium group and the azobenzene) also bind via non-covalent interactions with different DNAs.^[28] In this case, only the bis-terminated compound exhibited ICD bands in its *trans* conformation,

indicating the importance of the geometrical arrangement of the ligands in the assembly with the DNA (see below).

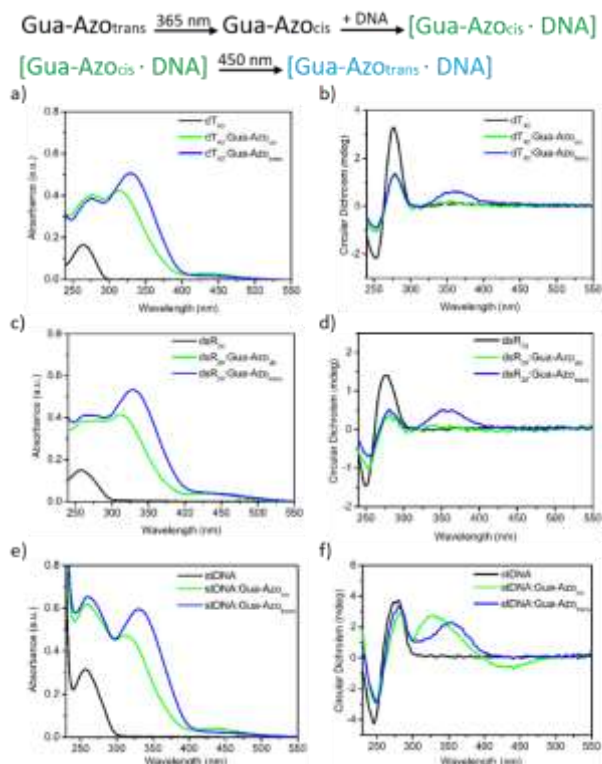


Figure 3. Changes in UV-Vis (left) and CD (right) spectra of several DNA/Gua-Azo mixtures at 20 °C in MQ-Water upon illumination at 365 nm and at 450 nm: (a-b) dT₄₀/Gua-Azo at 1:50 molar ratio; (c-d) dsR₂₀/Gua-Azo at 1:50 molar ratio; and (e-f) stDNA_{bp}/Gua-Azo at 1:1 molar ratio (in base pair/ligand ratio).

In situ trans → cis → trans photoisomerization. A solution of the complex [Gua-Azo_{trans} · DNA] was illuminated at 365 nm for 5 minutes. The CD spectra showed the disappearance of the ICD bands, typical of the [Gua-Azo_{cis} · DNA] complex (Fig. S18), which indicates that the photoisomerization from *trans* → *cis* occurred in interaction with DNA (*in situ*). We then tested a subsequent *cis* → *trans* photoisomerization *in situ*, by illuminating the same solution for 15 minutes at 450 nm. We noticed that the ICD signal typical of [Gua-Azo_{trans} · DNA] was not completely recovered, even after longer illumination times (up to 90 min, Fig. S19). Regarding the UV-Vis absorption spectra, we observed that there is lesser extent of red-shift of the maximum absorption band when both photoisomerization processes (*trans* → *cis* and then *cis* → *trans*) are realized in interaction with DNA, which may indicate structural changes of the complex after a complete photoisomerisation cycle (see below).

DNA binding modes: docking studies

We performed docking studies to compare the affinity and the binding modes of Gua-Azo with double-stranded DNA (using a genetic algorithm in the AutoDock4 package on the dsR₂₀

sequence, see protocol in the Supporting Information), for both *cis* and *trans* stereoisomers. The most energetically favorable docking solution for each Gua-Azo/DNA complex is depicted in Figure 4. The calculated binding energy is more stable for the [Gua-Azo_{trans} · DNA] complex ($E_b = -9.99$ kcal/mol) than for the [Gua-Azo_{cis} · DNA] complex ($E_b = -8.82$ kcal/mol). Although Gua-Azo isomers are both adsorbed in the DNA minor grooves, their binding modes and interactions at play are different, which lead to these differences in stability (see next section), and also on the DNA sequence over which the isomers are adsorbed, as described below for each isomer in these energetically-favorable docking solutions.

Binding modes of trans isomer of Gua-Azo. The nearest DNA bases of dsR₂₀ for Gua-Azo_{trans} (*i.e.* located at a maximum distance of 4.0 Å), are 5'-...7GTAAATCG¹⁴...-3' (3'-...³⁵GCATTTA²⁹...-5'), see Figure 4. Remarkably, the shape of Gua-Azo in *trans* configuration perfectly accommodates the DNA minor-groove, see Figure 4 center in which Gua-Azo is shown in van der Waals spheres. We noticed two main interaction sites between Gua-Azo_{trans} and DNA (Figure 4 left): 1) one strong H-bond between the polar hydrogen of the acylhydrazone group of Gua-Azo (HN1 atom, see numbering Figure 4 top) and the nucleobase oxygen of T32 residue (Figure 4a), with optimal H-bond parameters (distance of 2.2 Å and angle of 158° between donor-hydrogen-acceptor atoms); 2) a pattern of hydrogen-bonds and electrostatic interactions involving the guanidinium moiety with DNA. The HN2 atom is H-bonded to a phosphate oxygen of the C13 residue (distance 3.5 Å and angle 150°, see Figure 4b), and HN3 is H-bonded to the sugar oxygen of the same residue, with optimal geometrical parameters of 3.0 Å and 175° (Figure 4c). Concerning the nitrogen bearing two hydrogen atoms (HN4), one charge-assisted H-bond was detected with the phosphate oxygen of the residue T32 (geometrical parameters: 2.4 Å and 151°, see Figure 4d).

Binding modes of cis isomer of Gua-Azo. The nearest bases of dsR₂₀ for Gua-Azo_{cis} are 5'-...¹⁰AATCGGT¹⁶...-3' (3'-...³³ATTTAG²⁸...-5'), *i.e.* a different preferential sequence site than Gua-Azo_{trans}, see Figure 4. In contrast to the *trans* stereoisomer, the *cis* conformation yield to a partial desorption of the azobenzene moiety, as the phenyl at the extremity of Gua-Azo is outside the DNA minor-groove (Figure 4 right). The interactions between Gua-Azo_{cis} and DNA (see Fig. S20) are: 1) a H-bond between the polar hydrogen of the acylhydrazone group of Gua-Azo_{cis} (HN1) and the sugar oxygen of the residue T12; 2) H-bond between the HN3 hydrogen atom of the guanidinium moiety, and a sugar oxygen of the A29 residue, with non-optimal geometrical parameters (distance 3.1 Å, angle 132°, Figure S20). The same hydrogen participates also in H-bonding with a phosphate oxygen of the residue T30, again with non-optimal geometrical parameters (3.3 Å and 145°). Finally, the HN4a hydrogen atom is at a distance of 2.5 Å from an oxygen phosphate of the residue G15. The binding mode of the Gua-Azo_{cis} isomer can be therefore summarized in two anchor points: one weak H-bond that implicates the HN1 hydrogen atom, and another anchor point that involves several weak H-bonds of the guanidinium moiety, reinforced by electrostatic interactions.

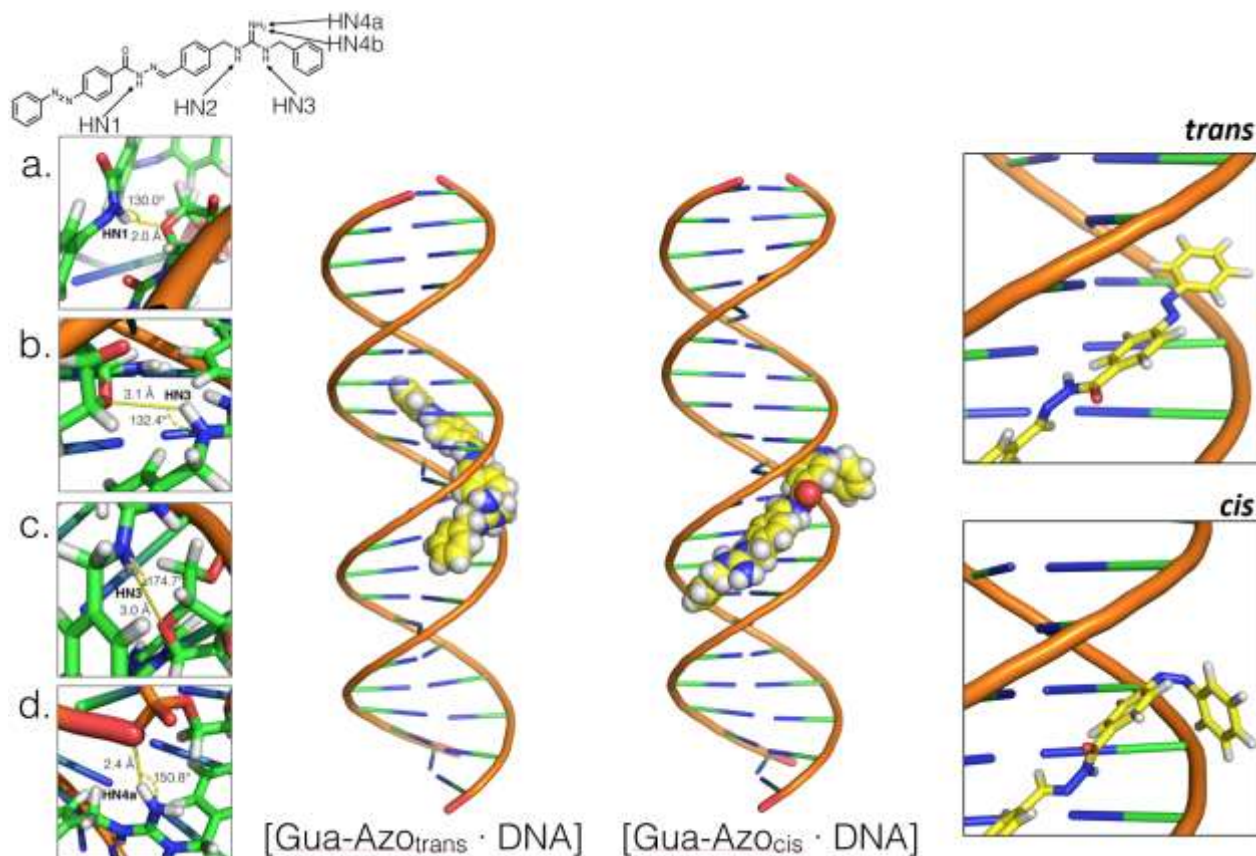


Figure 4. Center: most energetically favorable binding modes for the two **Gua-Azo** isomers with dsR₂₀ DNA. Center: **Gua-Azo** is depicted in van der Waals spheres and DNA as ribbons with bases as sticks. Left: close-up views for the main intermolecular interactions for **Gua-Azo**_{trans}, in stick representation. Right: close-up views at the location of the azobenzene moiety for the two isomers. Water molecules are omitted for clarity.

Stability and dynamics of the complex DNA/photoswitch

The comparison of the interactions involved in docking solutions of [Gua-Azo_{trans} · DNA] versus [Gua-Azo_{cis} · DNA] indicates that H-bonds are more optimal in terms of directionality in the [Gua-Azo_{trans} · DNA] structure, thereby explaining why the **Gua-Azo**_{trans} isomer has a higher affinity with DNA. When considering the ten energetically most favorable complexes obtained from our docking studies for both **Gua-Azo** isomers, our results show that the *trans* azobenzene stereoisomer of **Gua-Azo** is always found to be more stable: the calculated binding energy of the [Gua-Azo_{trans} · DNA] goes from -9.99 to -9.51 kcal/mol for the 1st and 10th most stable complex, and from -8.82 to -7.15 kcal/mol for the corresponding [Gua-Azo_{cis} · DNA] complexes (see Table S2 in the Supporting Information). Let us note that, for **Gua-Azo**_{trans}, the docking scores for different sequence sites were estimated very close in energy, suggesting many ligands interacting along the same DNA structure, in agreement with the titration experiments. The 10th most stable [Gua-Azo_{trans} · DNA] complex is indeed more stable than the 1st most stable [Gua-Azo_{cis} · DNA] complex, reinforcing clearly the preferential binding of the **Gua-Azo**_{trans} isomer to DNA.

The difference of stability between the *trans* and *cis* isomers of **Gua-Azo** when bound to DNA is further shown by DNA melting temperature (T_m) experiments with stDNA, where we monitored the UV-Vis absorbance at 260 nm of the samples as a function of temperature (Fig. S25). The value of T_m is defined as the temperature for which the normalized absorbance equals 0.5; this value is ~ 50 °C for the pure stDNA in MilliQ water. When a solution of **Gua-Azo** in *trans*, this value goes up to 5°C (T_m ~ 55 °C), whereas it goes up to 2°C when the **Gua-Azo** was primarily from *cis* (T_m ~ 53 °C). Besides, we noticed that, at temperatures higher than T_m , the plot for **Gua-Azo**_{trans} does not show a purely sigmoidal curve, and the ΔT_m between pure stDNA and [Gua-Azo_{trans} · stDNA] further increases. When the same experiment was carried out after an *in situ* photoisomerisation of the complex, the curves show the same trend than for a non-irradiated mixture of [Gua-Azo_{trans} · stDNA], with an increase of the melting temperature (T_m ~ 59 °C). Altogether, this shows the higher stability of the [Gua-Azo_{trans} · DNA] compared to [Gua-Azo_{cis} · DNA].

To further assess the differences in dynamics between these two states when bound in the DNA minor-grooves, we performed Molecular Dynamics (MD) simulations for the two complexes, starting from coordinates of the best docking poses for 100 ns MD production (using the AMBER16, see MD methodology in the Supporting Information). For both complexes, a so-called “*induced fit*” in the DNA minor groove was observed in the beginning of the MD simulations, as shown by the important variation of the RMSD of ligands, followed by a plateau until the end of MD simulations (Fig. S21). In the last part of the MD simulations (90-100 ns), *i.e.* when the complexes are stabilized, the variation of the RMSD is much higher for **Gua-Azo_{cis}** than for **Gua-Azo_{trans}** (Fig. S22). The average RMSD and standard deviations over the last 10 ns are 2.4 / 0.4 Å and 2.0 / 0.2 Å for **Gua-Azo_{cis}** and **Gua-Azo_{trans}**, respectively. In addition, the dynamics of [**Gua-Azo_{cis}** · DNA] and [**Gua-Azo_{trans}** · DNA] complexes are different. Concerning the acylhydrazone part of the molecule (around HN1), the distances between the HN1 atom and DNA show a normal distribution profile in a narrow range of values, meaning that the central part of ligands is rather static for both complexes (Fig. S23). In contrast, the dynamics of the guanidinium moiety of **Gua-Azo** are completely different. For the [**Gua-Azo_{trans}** · DNA] complex, the guanidinium moiety is highly mobile, with an oscillating position between two phosphate groups. The distance between the HN4a atom and DNA presents a two-steps profile and a distribution with two peaks, due to the electrostatic competition between the phosphates of C13 and T32 DNA residues (Fig. S24). This is not the case for the [**Gua-Azo_{cis}** · DNA] complex, for which the guanidinium group remains static, as shown by the distance profile between HN4a atom and DNA and its one-peak distribution (Fig. S24). Therefore, the docked **Gua-Azo** stereoisomers feature different dynamics: **Gua-Azo_{trans}** has a static azobenzene anchored to the DNA minor groove, and a mobile guanidinium moiety, whereas the azobenzene in **Gua-Azo_{cis}** is mobile, flipped outside of the DNA minor groove, with a rather static guanidinium moiety. The static behavior of the **Gua-Azo_{trans}** stereoisomer when bound to DNA, well accommodated with its crescent shape in the DNA minor-groove in a helical manner, is in agreement with the observed induced CD signal in the region where the **Gua-Azo** absorbs. The numerous possible sites of adsorption along DNA could further explain the intense ICD signals observed. However, a complete identification on the origin of this ICD signal for the *trans* isomer would need an in-depth theoretical analysis (as for other well-known DNA groove binder),^[45] which is beyond the scope of this study. In contrast, **Gua-Azo_{cis}** stereoisomer is more mobile and less anchored to the DNA, with the azobenzene moiety partially desorbed from the minor-groove, outside the DNA double helix, which could explain the absence of CD signals in the same range.

Photoswitching in DNA-templated heteromolecular complexes

The photo-control of the supramolecular association of **Gua-Azo** molecules onto DNA templates is attractive to modulate the assembly of other DNA binders. Recently, we have shown that a guanidinium-bispyrene derivative, **GuaBiPy** (Chart 1), undergoes DNA-templated self-assembly through a combination of interaction with the phosphodiester DNA backbone and π -stacking between pyrene moieties, with the pyrene moieties adopting a chiral organization within the complex.^[11] In order to evaluate the possibility of hetero-molecular complexes of guanidinium derivatives with DNA, mixtures of **GuaBiPy** and **Gua-Azo** were prepared in varying proportions but maintaining constant the total concentration of guanidinium compounds ($\text{Gua} = \text{GuaBiPy} + \text{Gua-Azo}$, the latter either in *trans* or *cis*). Each type of DNA was added to these mixtures using the same molar ratio than in the previous experiments, *i.e.* 1:50 molar ratio for both dT₄₀:**Gua** and dsR₂₀:**Gua**, and a 1:1 ratio in base pairs for stDNA:**Gua** (*i.e.* one guanidinium derivative per base pair). For mixtures with the oligonucleotides (dT₄₀ or dsR₂₀), the CD measurements show that, while both pure **GuaBiPy** and pure **Gua-Azo_{trans}** interact with dT₄₀ and dsR₂₀, yielding ICD signals in the 320-400 nm range, the ICD signals become less intense than for pure **GuaBiPy** for the three different component mixtures (**GuaBiPy**:**Gua-Azo** 20:80, 50:50, 80:20) (Figs. S26-29). This indicates that **Gua-Azo_{trans}** competes with **GuaBiPy** in terms of DNA binding. The shape of the ICD signals shows mixed signals between ICD of pure guanidinium derivatives, suggesting the formation of heteromolecular complexes with DNA. A weak ICD signal is maintained for the mixtures with **Gua-Azo_{cis}** only for large proportion of **GuaBiPy** within the mixture (*i.e.* for 50:50 and 80:20 in **GuaBiPy**:**Gua-Azo**). When the [ODN · **GuaBiPy** · **Gua-Azo_{cis}**] solutions are illuminated *in situ* at 450 nm (*cis* → *trans*), the ICD signals remained weak, and are difficult to interpret.

In contrast, with the long DNA template (stDNA), we observed intense ICD signals for all the mixtures (Figure 5 and Figs. S30-31), with more intense signals for the 20:80 and 50:50 **GuaBiPy**:**Gua-Azo_{trans}** complexes with stDNA than for the pure stDNA:**GuaBiPy** or pure stDNA:**Gua-Azo_{trans}**. This may indicate a cooperative effect in the chiral organization of the ligands when these are interacting with long templates.^[46] Upon illuminating a **Gua-Azo** solution at 365 nm (*trans* → *cis*) and then adding **GuaBiPy** and DNA, the ICD signal of the [stDNA · **GuaBiPy** · **Gua-Azo_{cis}**] mixture became less intense (except for the 80:20 in **GuaBiPy**:**Gua-Azo**, for which it remained constant, Fig. S31h), as the signature of the isomerization of **Gua-Azo** *trans* → *cis*. The subsequent *in situ* illumination of this [stDNA · **GuaBiPy** · **Gua-Azo_{cis}**] solution at 450 nm yield ICD signals typical of **Gua-Azo_{trans}** with stDNA (Fig. S30f), indicative of the *cis* → *trans* isomerization even when the complex is composed of a mixed assembly along the DNA scaffold.^[47]

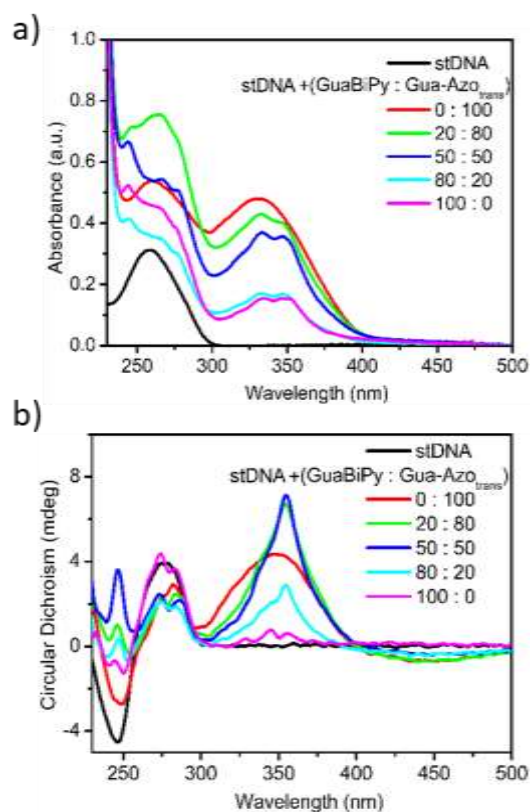


Figure 5. Changes in UV-Vis (a) and CD (b) spectra of several stDNA/GuaBiPy/Gua-Azo_{trans} mixtures (GuaBiPy:Gua-Azo_{trans} 0:100, 20:80, 50:50, 80:20, 100:0) at 20 °C in Milli-Q Water (Gua-Azo in *trans* configuration).

To further assess the effect of photoisomerization over the self-assembly of **Gua-Azo/GuaBiPy** along the DNA template, fluorescence spectroscopy was carried out, as the pyrene moieties are efficient fluorescent probes to assess DNA-templating effects. It was previously observed in the emission spectra that excimer emission **GuaBiPy** (maximum at around 470 nm) is favoured at the expense of monomer emission (peaking at 396 nm) upon interaction with single-stranded DNA, indicative of pure **GuaBiPy** templated self-assembly, stabilized by π -stacking of pyrenes.^[11,36]

In a three-component mixture [ODN · **GuaBiPy** · **Gua-Azo**], when **Gua-Azo** is in *trans*, the excimer emission is quenched compared to excimer emission of a solution of [ODN · **GuaBiPy**], see Figs.S33-34. This is further illustrated by plots of excimer/monomer ratio as a function of the composition of **GuaBiPy:Gua-Azo** in the mixture (Figs. S35). This is also the case for mixtures with the long template [stDNA · **GuaBiPy** · **Gua-Azo**], showing a significant decrease in excimer emission with respect to a DNA-templated assembly of pure **GuaBiPy** (i.e. [stDNA · **GuaBiPy**]), as observed in E/M ratio of Figure 6b (black squares vs. blue triangle). The weaker excimer emission of pyrenes observed in these three-component mixtures indicates that **Gua-Azo**_{trans} competes with **GuaBiPy** in the assembly along the DNA template.

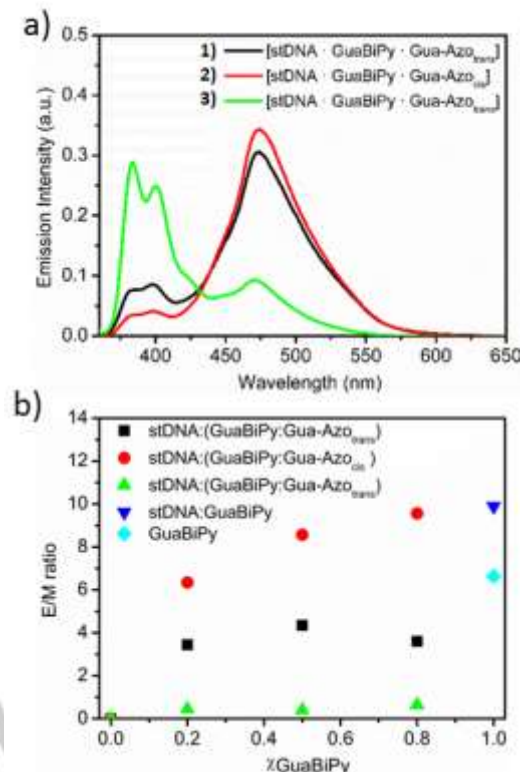


Figure 6. (a) Emission spectra (at $\lambda_{exc} = 348$ nm) of a mixture of stDNA:GuaBiPy:Gua-Azo (stDNA:ligand: 1:1 ratio in base pair, [GuaBiPy]=[Gua-Azo]). [stDNA]_{bp} = 122 μ M in base pair; [GuaBiPy] + [Gua-Azo] = 120 μ M. (b) Plot of the intensity ratio of excimer emission (taken at 472 nm) to monomer emission (taken at 396 nm) as a function of the molar ratio in [GuaBiPy]:[Gua-Azo] in the mixtures.

The evolution of emission and spectra excimer/monomer ratio (E/M) for the different mixtures for a complete photoisomerization cycle has been monitored as a function of the composition **GuaBiPy:Gua-Azo** in the mixture, see Figure 6 (and Figs. S33-36 in the Supporting Information). While the excimer emission of **GuaBiPy** is lowered for short dT₄₀ templates in presence of **Gua-Azo** (whatever the isomer of **Gua-Azo**), the situation is different for mixtures with long dsDNA templates (Figure 6a). When going from *trans* \rightarrow *cis*, we observed a recovery of the excimer emission of **GuaBiPy**: the E/M ratio of [stDNA · **GuaBiPy** · **Gua-Azo**_{cis}] are relatively close to that of pure **GuaBiPy** with this template, see Figure 6b (red dots and blue triangle). This indicates that **Gua-Azo**_{cis} is expelled – at least partially – upon *trans* \rightarrow *cis* photoisomerization because of lower affinity with dsDNA, and replaced by **GuaBiPy**, as sketched in Figure 7. When this mixture is then illuminated at 450 nm (*cis* \rightarrow *trans*), the excimer emission is quenched compared to monomer emission (E/M ratio <1), which indicates that **Gua-Azo** in *trans* replaces **GuaBiPy** in the templated assembly. Given the highly dynamic behaviour of these DNA-templated self-assemblies, the fact that this process is not fully reversible in terms of E/M ratio for a

complete photoisomerization cycle (as for ICD signals of [DNA · **Gua-Azo**] complexes, see above) may indicate kinetic processes, with structural changes within the complex from the initial *trans* to the final *trans* **Gua-Azo** configuration after photoisomerization. These structural changes could also originate from DNA damages upon UVA irradiation,^[48-49] albeit we used a limited irradiation time (5 min with a LED at 365 nm) to perform photoisomerization of **Gua-Azo** with DNA. This is comparable to UVA irradiation times used in other DNA-binding studies with azonbenzene-based peptidomimetic ligand, for which the nativity of DNA was stated to be unchanged.^[30]

Let us highlight that the E/M ratio of [stDNA · **GuaBiPy** · **Gua-Azo**_{trans}] complexes are higher than E/M ratio obtained for complexes with dT₄₀ templates (Figure 6 and Fig. S34), which suggests a better π -stacked pyrenes arrangement along long dsDNA templates.

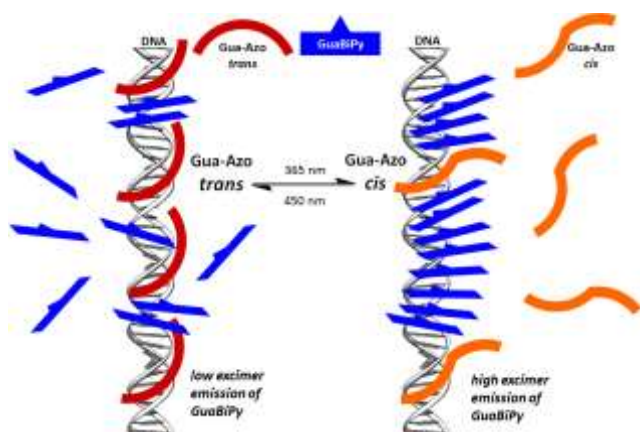


Figure 7. Sketch of the DNA-templated self-assembly of **GuaBiPy/GuaAzo** depending on the isomerization of **Gua-Azo**.

Conclusions

We have reported the design of a photoresponsive DNA ligand, **Gua-Azo**, which photo-isomerizes between two configurations – *cis* and *trans* – that have different binding modes and affinities to DNA. We can pinpoint that:

- **Gua-Azo** binds to the DNA minor groove when the azobenzene moiety is in *trans* configuration, which is detectable by characteristic chiroptical signals. **Gua-Azo** is particular in view of recently reported guanidinium-based DNA ligands,^[11,32] as it binds to the DNA minor-groove from two sites of interactions (acylhydrazone and guanidinium groups), with a crescent shape that accommodates helically to the DNA minor groove, as shown by docking studies;
- in contrast, the *cis* configuration is partially desorbed from DNA minor groove and has a lesser affinity towards DNA;
- when **Gua-Azo** is mixed with another guanidinium-based DNA ligand (**GuaBiPy**) in presence of DNA, the ligands form heteromolecular supramolecular complexes in a chiral organization along the template;

- the stacking propensity of **GuaBiPy** along the DNA, reflected by excimer emission of pyrenes, can be indirectly controlled through **Gua-Azo** photoisomerisation. This amounts to a primitive supramolecular network,^[50-53] where the structural photoswitching of one compound has a remote control over the DNA-templated assembly of another compound present in the chemical system, which is detected by unique fluorescence signal output changes.

Altogether, these results show that well-designed photo-responsive DNA binders can be exploited to *photo-modulate the supramolecular DNA assembly* in templated hetero-molecular complexes. Besides the ability of light-induced ligand redistribution,^[54] our approach could be exploited to tune the delivery of competitive small molecules DNA binders and address the dynamics of DNA-templated multicomponent assemblies. This could also serve to photo-manipulate DNA nanostructures (e.g. DNA origami) through a non-covalent, dynamic approach.^[55]

Experimental Section

Details of synthesis, characterization, preparation of samples, additional spectra, molecular modelling methodology and analysis, see Supplementary Information at...

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