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2	Silver potentiates aminoglycoside toxicity by enhancing their uptake
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27	
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29 E. coli, Clostridium difficile

30 Summary

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The predicted shortage in new antibiotics has prompted research for chemicals that could 32 act as adjuvant and enhance efficacy of available antibiotics. In this study, we tested the 33 effects of combining metals with aminoglycosides on Escherichia coli survival. The best 34 synergising combination resulted from mixing aminoglycosides with silver. Using genetic 35 and aminoglycoside uptake assays, we showed that silver potentiates aminoglycoside 36 action in by-passing the PMF-dependent step, but depended upon protein translation. 37 We showed that oxidative stress or Fe-S cluster destabilization were not mandatory 38 39 factors for silver potentiating action. Last, we showed that silver allows aminoglycosides 40 to kill an E. coli gentamicin resistant mutant as well as the highly recalcitrant anaerobic pathogen Clostridium difficile. Overall this study delineates the molecular basis of 41 42 silver's potentiating action on aminoglycoside toxicity and shows that use of metals might offer solutions for battling against increased bacterial resistance to antibiotics. 43 44 45

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- 48 Introduction
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Increasing bacterial resistance to antibiotics is a major health issue, particularly in the 50 context of the low discovery rate for new antibiotics (Nordmann et al., 2007). A solution to 51 this problem is to use antibiotic-potentiating or adjuvant agents to improve the efficacy of 52 existing antibiotics. Augmentin that combines a β-lactam and a β-lactamase inhibitor is a 53 well-known example of an antibiotic-adjuvant association, which has been used extensively to 54 55 treat infections caused by β -lactam resistant bacteria (Leigh *et al.*, 1981; Worthington and Melander, 2013). Other types of potentiating effects can be obtained by facilitating the entry 56 57 of antibiotics into target bacterial cells (Mislin and Schalk, 2014).

Aminoglycosides are bactericidal antibiotics, which target ribosomes. They comprise 58 kanamycin, tobramycin, gentamicin (Gm), and streptomycin, and are commonly used 59 worldwide, thanks to their high efficacy and low cost (Davis, 1987). One limitation of these 60 61 antibiotics is their ineffectiveness against anaerobic bacteria. Recently, we showed that the efficacy of aminoglycosides could be lowered by reduced iron bioavailability, which reduced 62 Gm uptake (Ezraty et al., 2013; Ezraty and Barras, 2016). Uptake of aminoglycosides in 63 Gram-negative bacteria is a three-step process (Taber et al., 1987). Aminoglycosides, which 64 are strongly cationic drugs at physiological pH, first bind electrostatically to the negatively 65 charged outer membrane and then reach the periplasm through porin channels. Uptake across 66 the inner membrane is an energy-dependent process that can be described in two sub-steps 67 forming a feed forward loop. Energy dependent phase I (EDP-I) requires the proton motive 68 force (PMF), which allows aminoglycoside to reach the cytosol via an electrophoresis-like 69 process. Eventually aminoglycosides act on translating ribosomes. Energy dependent phase 70 II (EDP-II) is when inhibited ribosomes release misfolded aborted hydrophobic proteins, 71 which incorporate into, and disturb membrane organization. The consequence is a massive 72 uptake of aminoglycosides (Hurwitz et al., 1981). 73

Respiratory complexes I and II, which are the major sources of PMF, contain numerous 74 Fe-S clusters, which are essential for their activity. Biogenesis of Fe-S cluster is carried out 75 either by the ISC or by the SUF system. ISC is used under normal growth conditions, 76 whereas SUF is used under stress conditions, such as iron limitation. For reasons that 77 remain unclear, the SUF system is much less efficient in targeting Fe-S clusters into complexes 78 I and II. The consequence is that complexes I and II are less active. Thus, under iron 79 80 limitation, low efficiency of maturation of complexes I and II by SUF leads to lowered PMF, reduced EDP-I controlled aminoglycoside uptake, and enhanced aminoglycoside 81 resistance. Interestingly, a low level of PMF could also account for the high resistance of 82

anaerobic bacteria to aminoglycosides as they use fermentative metabolism. Another connection between aminoglycosides and metals was highlighted by a previous study that attributed an aminoglycoside-potentiating effect to silver. It was proposed that silver enhanced reactive oxygen species (ROS) production, which contributed to the silver potentiating effect on Gm toxicity (Morones-Ramirez *et al.*, 2013).

In this study, we screened a series of metals for their influence on the susceptibility of 88 E. coli to aminoglycosides. Silver, and to some extent copper, stood out as potentiating 89 aminoglycoside efficacy. We reinvestigated the molecular basis of the silver potentiating effect and 90 concluded that it is (i) independent of PMF and allows by-passing EDP-I, (ii) dependent on 91 protein translation, and (iii) independent of ROS production. Finally we showed that silver 92 could be used to combat highly natural aminoglycoside resistant bacteria, such as the Gram-93 positive anaerobic pathogen, Clostridium difficile. 94

- 95 **Results**
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Assessing the potentiating effects of metals on Gm toxicity

In LB and under aerobic conditions, E. coli exhibited a minimal inhibitory concentration 99 (MIC) value to Gm (Gm) of 2.8 _{+/- 0.2} µg/ml (Fig. 1A). This value was not modified upon 100 addition of sub-lethal concentrations of iron, cobalt, zinc or nickel-containing solutions (100 101 μM, see Table S1) (Fig. 1A). In contrast, addition of silver (8 μM) led to a 10 fold reduction 102 of the MIC value (0.17 _{+/- 0.1} µg/ml) (Fig. 1A). A dose-response effect of silver was 103 observed (Fig. S1A). Addition of copper (100 µM) produced also a small but significant 104 decrease of the MIC value, i.e. 2 +/- 0.4 µg/ml (Fig. 1A). In M9 minimal medium, the MIC value 105 to Gm decreased to 0.25 $_{\pm 0.02}$ µg/ml and was influenced neither by the addition of sub-lethal 106 concentration of silver (4 nM see Table S1) nor by that of copper (0.25 µM see Table S1) (Fig. 107 S1B). In M9 minimal medium supplemented with 0.2% Casamino acids, E. coli exhibited a 108 MIC value to Gm of 0.5 +/- 0.06 μ g/ml, addition of silver (50 or 100 nM) produced a small 109 but significant 2 fold decrease of the MIC value. However, survival rate of E. coli to Gm 110 treatment in M9 minimal medium supplemented with 0.2% Casamino acids was not modified 111 by the presence of silver (50 or 250 nM) (Fig. S1D). In Neidhardt's low phosphate minimal 112 medium, E. coli exhibited a MIC value to Gm of 0.2 +/- 0.02 μ g/ml, which was not modified 113 by the addition of sub-lethal concentration of silver (0.01 µM see Table S1) (Fig. S1E). 114 Consistently, survival rate of *E. coli* to Gm treatment in Neidhardt's low phosphate minimal 115 medium was not modified by the presence of silver (Fig. S1F). Altogether these studies 116 highlighted the importance of the growth conditions, which can modify both the MIC values 117 of Gm and the capacity of metals to enhance the toxicity of Gm. The reason why silver seems 118 to potentiate Gm toxicity only in LB grown E. coli is unknown and this issue is discussed later 119 in the manuscript. A thorough comparative study between media was out of the scope of the 120 present study and we used LB throughout all next experiments. 121

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124 Silver potentiates the activity of all aminoglycosides in *E. coli*

We tested the potentiating activity of silver on an extended set of antibiotics from diverse families. Silver was found to lower more than 10 fold the MIC values for all three bactericidal aminoglycosides tested, i.e. kanamycin, tobramycin and streptomycin. A reduction of only 2 fold was observed with spectinomycin, an aminoglycoside that is bacteriostatic (Fig. 1B). Similarly, a slight potentiating effect (less than 20% decrease of the MIC value) was observed when used in conjunction with nalidixic acid and norfloxacin but not ciprofloxacin, although all three belong to the quinolone family (Fig. 1B). Modest reduction of the MIC values was also noted when silver was used in conjunction with ampicillin, tetracycline or chloramphenicol (Fig. 1B). Collectively, these results showed that silver exerts a potentiating effect mostly on aminoglycosides.

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Silver enhances Gm toxicity by acting independently of PMF

Gm penetrates across the cytoplasmic membrane via a PMF-dependent process. We therefore 140 wondered whether PMF was required for the silver potentiating effect. To test this, we used 141 pharmacological and genetic approaches. Carbonyl cyanide-m-chlorophenylhydrazone 142 (CCCP), an uncoupler H⁺ ionophore, was used to dissipate the PMF. Then, we followed the 143 survival rate of E. coli after 1.5 hours of Gm treatment, in absence or in presence of CCCP 144 and with or without silver. As previously reported by us and others (Ezraty et al., 2013; 145 146 Allison et al., 2011), the presence of CCCP enhanced resistance of E. coli to Gm-mediated killing, because it abolishes PMF. In contrast, adding silver reversed the effect and re-147 148 sensitized E. coli even in the presence of CCCP (Fig. 2A). We also tested mutants known to have low levels of PMF (Ezraty et al., 2013). Specifically, we tested either a strain deleted of 149 150 nuo and sdh operons, which lacks complexes I and II, or a strain deleted of iscUA genes, 151 which poorly matures complexes I and II. Survival tests showed that $\Delta nuo \Delta sdh$ and $\Delta iscUA$ strains exhibited enhanced resistance to Gm (Fig. 2A). Consistently, MIC values of both 152 $\Delta nuo \Delta sdh$ (3.6 +/- 0.3) and $\Delta iscUA$ (6.1 +/- 0.5) strains were higher than wt (Fig. 2B). Adding 153 silver cancelled differences between mutant and wt strains. All three strains, wt, $\Delta nuo \Delta sdh$ 154 and $\Delta iscUA$, showed the same survival loss of about 5 log in 1.5 hours (Fig. 2A) and all 155 three exhibited the same MIC value, i.e. 0.17 +/- 0.1 µg/ml (Fig. 2B). Taken together, both 156 the pharmacological and genetic approaches demonstrated that silver enhanced Gm toxicity in 157 a PMF-independent manner. 158

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160 Silver restores Gm uptake in a PMF-deficient strain

We then tested whether the potentiating effect of silver was due to an increased uptake of Gm. The intracellular accumulation of Gm was monitored over time by using tritiated Gm (³H-Gm). As previously reported, both the $\Delta nuo \Delta sdh$ and $\Delta iscUA$ strains were drastically altered in Gm uptake (Fig. 2C) (Ezraty *et al.*, 2013). In contrast, addition of silver restored Gm uptake in both mutants (Fig. 2C). Uptake in the $\Delta nuo \Delta sdh$ mutant was more efficient as 167 compared with both the wt and the $\Delta iscUA$ mutant; we have no explanation for this 168 phenomenon. Moreover, in wt strain, silver allowed uptake of Gm applied at sub-lethal 169 concentration (Fig. 2D). These results showed the potentiating effect of silver to be due to its 170 capacity to enhance uptake of Gm. Taken together with the results above, these data 171 demonstrated that silver by-passes the EDP-I PMF-dependent step of the Gm entry process.

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173 Silver potentiating effect is EPD-II dependent

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The EDP-II is protein translation-dependent. This is shown by the fact that chloramphenicol (Cm) a bacteriostatic antibiotic inhibiting translation, has a protective effect against Gm (Fig. 3). Therefore we asked whether Cm could interfere with the silver potentiating effect. Survival rate of *E. coli* to Gm treatment was the same whether silver was added or not to Cm-treated cells (Fig. 3). This result showed that silver needs the EDP-II, i.e. protein translation, to potentiate Gm toxicity.

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182 ROS production does not explain the positive combination between silver and 183 aminoglycosides

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Silver is known to destabilize Fe-S clusters (Liu and Imlay, 2013). Morones-Ramirez et al. 185 attributed potentiating effect of silver on aminoglycoside activity to enhanced production of 186 187 harmful ROS following Fe-S destabilization by silver, release of free iron and Fenton chemistry (Morones-Ramirez et al., 2013). To further investigate this issue, we tested the 188 silver-potentiating effect of Gm on anaerobically grown cultures. An E. coli strain grown in 189 anaerobic fermentative conditions exhibited enhanced resistance to Gm (Fig. 4A). However, 190 adding silver to anaerobically growing E. coli allowed Gm to cause a loss of 4 logs in 191 survival (Fig. 4A). This did not support the hypothesis of a contribution of ROS. We tested 192 the role of ROS in silver potentiating effect by using *E. coli* strains altered in their redox 193 homeostasis control. ROS-hyper-sensitive strains lacking superoxide dismutases ($\Delta sodA$ 194 $\Delta sodB$) or the H₂O₂-stress responding master regulator ($\Delta oxvR$) exhibited similar 195 sensitivity to silver potentiating effect as compared with the wild type parental strain (Fig. 196 197 4B, S2). Similarly, a strain expressing constitutively the H₂O₂-stress responding regulon (oxyRc) exhibited a wild type-like sensitivity to the silver-potentiating effect of Gm (Fig. 198 199 4B, S2). Altogether, these observations indicate that ROS do not play a major role in the Gm-potentiating activity of silver. 200

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202 Fe/S clusters destabilization is not mandatory for silver potentiating action.

An iscS mutant exhibits enhanced resistance to silver (Fig S3A) (Morones-Ramirez et al., 203 2013). A possibility put forward by others is that the *iscS* enhanced resistance was due to a 204 reduced level of Fe-S cluster biogenesis, hence a reduced possibility of fueling Fenton 205 chemistry upon Fe-S cluster destabilization (Morones-Ramirez et al., 2013). However, iscS 206 207 mutation is pleiotropic and it is unsafe to interpret its phenotype as being the sole consequence of an alteration of Fe-S cluster biogenesis. Therefore, we tested an iscUA mutant, which 208 is specifically altered in Fe-S cluster biogenesis. The *iscUA* mutant was hyper-sensitive to 209 silver (Fig. S3B). The reason for this is probably because silver targets Fe-S clusters in key 210 enzymes, whose decreased activities are detrimental for fitness, as this has already been 211 proposed (Xu and Imlay, 2012). In any case, the phenotype of *iscUA* mutant shows that 212 decrease in Fe-S cluster capacity sensitizes cells to silver toxicity, rather than protects it. 213

Then the question arose of why an *iscS* mutant still exhibits enhanced resistance to silver. 214 215 IscS degrades L-cysteine and L-cysteine is known like histidine to help E. coli to resist to silver by sequestering it (Fig. S3C) (Matsumura et al., 2003). Hence a possibility was that the 216 iscS mutant showed enhanced resistance to silver because it accumulated a higher 217 concentration of cysteine. To test this hypothesis, we over-produced CsdAE, a highly efficient 218 hetero-dimeric cysteine desulphurase, in an iscS mutant and observed that this did 219 suppress silver resistance of the *iscS* strain. As a control, we showed that a strain 220 synthesizing a mutated inactive version of CsdA was unable to act as a suppressor of the *iscS* 221 resistance (Fig. S3D). The data above suggest that Cys accumulation is the reason for the 222 hyper-resistance to silver of *iscS* mutant. 223

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225 Silver sensitizes *E. coli* Gm^R strain and aminoglycoside resistant 226 pathogens bacteria

The *aaC1* gene encodes a N-acetyltransferase that modifies and inactivates Gm. Accordingly, 228 an *E. coli aac1*+ strain (Gm^R) exhibited a MIC value of 121+/-6 µg/ml, i.e. ca. 50 fold higher 229 than wt and remained partially viable in the presence of 100 µg/ml Gm (Fig. 5A and B). 230 Uptake experiments revealed that E. coli aacl+ strain failed to accumulate intracellular Gm 231 (Fig. 5C). Remarkably, adding silver reduced the MIC value to $4.3_{+/-1}$, and provoked 232 bacteriostatic and bactericidal effects with 5 and 100 µg/ml Gm treatment, respectively (Fig. 233 5A and B). Moreover, Gm uptake tests showed that silver allowed both Gm incorporation 234 and accumulation to resume in the Gm^R strain (Fig. 5C). Taken together, these results 235 showed that silver is able to sensitize a Gm^R strain to Gm treatment. 236

Clostridium difficile, an anaerobic enteropathogen causing infectious diarrhea, is well known 237 for its aminoglycoside resistance (Båverud et al., 2004). Indeed, MIC values of C. difficile for 238 Gm, tobramycin and kanamycin were found to be very high (Fig. 5D). The question of the 239 silver potentiating capacity was therefore worth exploring. Spectacularly, in the presence of 240 silver, the MIC values of C. difficile dropped down from 43+/-4 to 13+/-3 µg/ml for Gm, from 241 115_{+/-8} to 15_{+/-4} µg/ml for tobramycin and from 313_{+/-26} to 13_{+/-4} µg/ml for kanamycin (Fig. 242 5D). Thus, overall treatment with silver decreased the MIC values for aminoglycosides from 3-243 to 24-fold. Taken together, these results expanded the spectrum of silver potentiating effect to a 244 recalcitrant Gram-positive 245 pathogen.

246 **Discussion**

Screening of different metals allowed us to identify silver as an enhancer of aminoglycoside 247 treatment efficacy against E. coli. Moreover, we showed that silver allowed sensitizing the 248 249 Gram-positive aminoglycoside resistant enteropathogen C. difficile to Gm, tobramycin or 250 kanamycin treatments. The potentiating effect of silver has been reported before by Morones-Ramirez et al., (2013), but we offer a different explanation regarding the underlying 251 mechanism. In particular we rule out a role for ROS in this process as the enhancing effect 252 was also observed to arise under anaerobiosis and in mutants altered in anti-ROS activities. 253 Moreover, we show that silver enhances Gm uptake in a PMF-independent, protein 254 translation-dependent way. 255

Silver has been used as an antiseptic for centuries. Here we investigated the molecular basis 256 of both its toxicity and its adjuvant activity. We found an *iscUA* mutant, specifically altered 257 258 in Fe-S biogenesis, to be highly sensitive to silver. The simplest interpretation is that Fe-S clusters are targeted by silver ions, due to their thiophilicity (Xu and Imlay, 2012). One can 259 wonder why the back-up stress responding Fe-S cluster biogenesis system, SUF, does not 260 take over the absence of ISC in the iscUA mutant. Presumably, at the silver concentration 261 used, the SUF system is not sufficient either to keep up with the overall cellular demand in 262 the presence of silver or to target clusters efficiently enough to essential enzymes such as 263 IspGH. In both cases, bacterial viability is compromised. 264

Silver potentiating effect was observed when E. coli was grown in LB but was not observed 265 when E. coli was grown in minimal medium, M9 or Neidhardt's low Pi medium. The 266 influence of medium is best illustrated by the fact that the MIC values for Gm are 10 fold 267 lower in minimal medium than in LB. One possibility was that silver damaged Fe-S bound 268 dehydratase enzymes required for branched-chain amino acid synthesis. If so translation could 269 have been arrested, hence preventing silver to potentiate Gm toxicity as we showed its effect 270 271 to be translation-dependent. This was investigated by asking whether the addition of Casamino acids could allow silver potentiating effect to resume even in M9 medium. Results were 272 ambiguous as some potentiating effect was observed using MIC values but it was much less 273 than in LB (i.e. 2 fold in minimal medium supplemented with Casamino acids vs. 10 fold in 274 LB). Moreover, survival tests failed to reveal a silver potentiating effect. Therefore, for now 275 we have no clear-cut explanation to offer for the medium dependency. It could be due to 276 277 metabolic changes in different media, differences in pH and possible consequences on ΔpH trans-membrane component of the PMF, sequestering of Gm by LB contained components, 278 occurrence of phosphate silver complexes, or a combination of these different effects. A 279 dedicated systematic study will need to be conducted to solve this issue. 280

Silver potentiating effect of Gm was observed under anaerobiosis. Moreover, we failed to observe a correlation between anti-ROS capacity of the cell and its sensitivity to a mix of silver and Gm. Altogether, this strongly suggested that ROS has no major role in in silver potentiating effect.

285 Our characterization of the silver-potentiating effect demonstrated that it was independent of PMF, as it restored Gm sensitivity of strains with reduced PMF levels such as $\Delta nuo \Delta sdh$ 286 or $\Delta iscUA$ mutants. A possibility is that silver destabilizes the membrane allowing Gm to 287 get access to the cytosol more easily. In support of this view, we might cite electron 288 microscopy studies, which showed that silver perturbs cell wall perturbation after silver 289 treatment (Feng et al., 2000) and a recent system-based analysis of silver stress pointed out 290 291 potential dysregulation of fatty acid homeostasis (Saulou-Bérion et al., 2015). However, we also showed that silver-potentiating effect of Gm toxicity remained dependent on 292 293 translation. This implies that the membrane disturbance induced by silver is not sufficient for massive Gm uptake and needs additional contribution from mis-294 localised aborted polypeptides, predicted to occur during EDP-II. Alternatively, silver 295 might act directly on ribosomes, and might allow released of mis-folded aborted 296 polypeptides that would eventually go to the membrane and allow massive Gm 297 uptake. In this hypothesis, silver would cause an EDP-II like step. Further studies 298 will be required for establishing the actual mechanism of action of silver. Under some 299 aspects, it seems similar to toluene and polymixin B (Davis, 1960). In contrast, it 300 differs from the mode of action of others chemicals recently reported to potentiate Gm 301 toxicity. For instance, mannitol and fructose led to the generation of PMF via increased 302 NADH production, hence facilitating the EDP-I step (Allison et al., 2011). Another study 303 304 showed that non-buffered L-arginine, by impacting the ΔpH transmembrane component of the PMF, increased susceptibility to Gm (Lebeaux et al., 2014). 305

Strict anaerobes are unable to efficiently take up aminoglycosides and therefore are resistant 306 to this family of antibiotics (Schlessinger, 1988) as observed for C. difficile (Båverud et al., 307 2004). However, it is important to stress that ribosomes of clostridia can be bound and 308 inhibited by aminoglycosides (Bryan et al., 1979). Our study points to a new approach to try 309 to fight vegetative cells of this bacterium, as silver was found to overcome the probable 310 natural lack of permeability of their membrane for aminoglycoside. This is an important 311 312 result as C. difficile infection is the most common cause of identifiable diarrhea in hospitalized adult patients and the incidence and severity of C. difficile infections increased 313 (Lessa et al., 2015; Rupnik et al., 2009). In conclusion, our study depicts a new molecular 314 mechanism for the positive potentiating effect of silver on aminoglycoside activity, and shows 315

that use of metals might offer solutions for battling against increased bacterial resistance to antibiotics. It is noteworthy to remind that copper was also proposed for boosting various drug treatments (Speer *et al.*, 2013). Moreover, our study opens the way for fighting natural aminoglycoside resistant anaerobic pathogens.

322 General microbiology and molecular biology 323 In this study, we used E. coli K-12 derivatives strains and C. difficile 630 Δerm (Hussain et 324 al., 2005) (for details see Table). The LB medium used in this study comes from Difco. The 325 phosphate limiting minimal medium was the MOPS (pK_a 7.2) medium described by Neidhardt 326 contained 0.1 mM K₂HPO₄ plus 9.8 mM KCl and 0.2 % glucose (Neidhardt et al., 1974). 327 P1vir transduction was used to move alleles into a different background. pCsdAE^{C358A} was 328 constructed by site-directed mutagenesis using pCsdAE plasmid as template (Trotter et al., 329 2009), the oligonucleotides csdAE C358A F 330 (CGGGCCGGGCAGCATGCCGCTCAGCCGCTAC) and csdAE C358A R 331 (GTAGCGGCTGAGCGGCATGCTGCCCGGCCCG), Pfu Turbo 332 DNA polymerase 333 (Invitrogen) and *DpnI* from NEB. All plasmids were sequenced (Beckman Coulter Genomics) for verification. 334 335 Strains Source or Reference **Relevant Genotype** 336 337 *E. coli* K-12 strains 338 339 Lab collection MG1655 Parental strain 340 MG1655 $\Delta iscUA$::Cm^r 341 DV597 (Vinella *et al.*, 2009) MG1655 Δ*iscS*::Cm^r LL111 (Trotter *et al.*, 2009) 342 343 BEFB20 MG1655 $\Delta sdhB \Delta nuo::nptI Kan^{r}$ (Ezraty *et al.*, 2013) MG1655 *trp*::Hfr::Gm^r BEFB10 (Ezraty *et al.*, 2013) 344 BEFB01 MG1655 $\Delta sodA$::Cm^r $\Delta sodB$::Kan^r (Ezraty *et al.*, 2013) 345 **BEFB02** MG1655 $\Delta oxvR$::Cm^r (Ezraty *et al.*, 2013) 346 QC2433 MG1655 oxyRc Lab collection 347 348 *C. difficile* 349 350 630∆*erm* (Hussain *et al.*, 2005) 351 352 Plasmids 353 354 pCsdAE Lab collection (Trotter *et al.*, 2009) 355 pCsdAE^{C358A} This study 356

Experimental procedures

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357 MIC determination

The drugs tested were dissolved in LB medium for E. coli tests and in water for C. difficile 358 tests. The range of concentrations of each drug was twice the MIC. For E. coli, one hundred 359 microliters (100µL) of each 2X concentration tested were added in a 96-well microplate. 360 Each well was inoculated with 100 μL of a fresh LB bacterial inoculum of 2 x $10^5~\text{CFU/mL}$ 361 and the plates were incubated at 37°C for 18 hours under aerobic conditions and agitation 362 150 rpm. C. difficile strain 630∆erm was grown anaerobically (5 % H₂, 5 % CO₂, and 90 % 363 N₂) in microplate. Each well was inoculated with 1 mL of a fresh BHI bacterial inoculum at 364 an OD₆₀₀ of 0.01 and the drug was added at different concentrations in the presence or 365 absence of AgNO₃ at 20 µM. The plates were incubated at 37°C for 20 h under anaerobic 366 conditions. MIC was defined as the lowest drug concentration that exhibited complete 367 inhibition of microbial growth. 368

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370 Lethality studies on LB agar plates

Strains were grown aerobically (shaking 150 rpm) in minimal media M9 at 37°C. When the OD₆₀₀ reached 0.2, serial dilutions of cell suspensions in phosphate buffer (0.05 M, pH 7.4) were spread onto M9 media agar plates with different concentrations of silver nitrate and cysteine (specified in each figures). C.f.u. were counted after incubation for 16 h at 37°C.

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376 *Metals*

For synergetic effect with antibiotics $AgNO_3$ was used at 8 μ M for *E. coli*, unless noted otherwise, and 20 μ M for *C. difficile*, FeSO₄, CoCl₂, Cu₂(OAc)₄, Zn(OOCCH₃)₂ and NiCl₂ were used at 0.1 mM.

380

381 Survival rate

Overnight cultures (16 h) were diluted 100 times and grown aerobically in LB at 37°C to an 382 OD₆₀₀ of 0.2. At this point, antibiotics and/or silver were added to the cells and after 1.5 383 h cells were taken, diluted in PBS buffer, spotted on LB agar and incubated at 37°C for 384 16 h. Cell survival was determined by counting colony-forming units per ml (c.f.u./ml). The 385 absolute c.f.u at time-point 0 (used as the 100 %) was $\sim 5 \times 10^7$ cells/ml in all 386 experiments. When used, the PMF uncoupler, carbonyl cyanide-m-chlorophenylhydrazone 387 (CCCP) was added at a final concentration of 1 μ g/ml. When used, chloramphenicol (Cm) 388 was added 30 min before Gm treatment at a final concentration of 200 µg/ml. The Cm stock 389 solution was extemporaneously prepared at 20 mg/ml with 1/3 ethanol and 2/3 H₂O. 390 391 Synergic effect in anaerobic conditions was performed as previously described but in

- anaerobic chamber (Coy Chamber) with anaerobic media (previously equilibrated in theanaerobic chamber for 24 h).
- For silver sensitivity experiments M9 minimal media was used instead LB in order to minimized the variability of the results due to the chelation of the metal by medium components present in the LB (Feng *et al.*, 2000). Overnight cultures (16 h) were diluted 100 times and grown aerobically in minimal media M9-0.2 % glycerol at 37°C to an OD₆₀₀ of 0.2. At this point, AgNO₃ was added to the cells and after different incubation times (specified on each figure) cells were taken, diluted in phosphate buffer (0.05 M, pH 7.4), spotted on LB agar and incubated at 37°C for 16 h.

402 *Gm uptake*

- ⁴⁰³ [³H]-Gm (20 μCi/mg; Hartmann Analytic Corp.) was added to a final concentration of
- 5 or 0.25 μ g/ml and cultures were incubated at 37°C on a rotary shaker. At given times, 500 μ l aliquots were removed and collected on a 0.45 μ m-pore-size HAWP membrane filter (Millipore) pretreated with 1 ml of unlabeled Gm (250 μ g/ml). Filters were subsequently washed with 10 ml of 3 % NaCl, placed into counting vials, dried for 30 min at
- 408 52°C whereafter 8 ml of scintillation liquid were added and incubated overnight at room 409 temperature. Vials were counted for 5 min. Gm uptake efficiency is expressed as total 410 accumulation of Gm (nanograms) per 10^8 cells.
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412 Statistical analysis

- 413 Mann-Whitney U tests were performed using the QI-Macros software (KnowWare
- 414 International, Inc., Denver, CO)

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521 Figure 1. Silver potentiates aminoglycosides in *E. coli*

523 (A) Minimal Inhibitory Concentration (MIC) of Gm (in μ g/ml) in the presence of different 524 metals was calculated. *E. coli* strain MG1655 was grown in LB under aerobic conditions in 525 presence of 8 μ M AgNO₃, or 0.1 mM FeSO₄, CoCl₂, Cu₂(OAc)₄, Zn(OOCCH₃)₂ and NiCl₂. 526 MIC values were calculated as described in Materials and Methods. (B) MIC values of *E. coli* 527 MG1655 for different antibiotics as indicated on top of the graphs. AgNO₃ concentration used 528 was 8 μ M. *P ≤ 0.05; **P ≤ 0.01; and ***P ≤ 0.001 (Mann-Whitney U test).

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531 Figure 2. Silver potentiates Gm activity against *E. coli* by a PMF-independent uptake

(A) Percentage survival of E. coli MG1655 after 1.5 h treatment with Gm (5 µg/ml) and/or 532 AgNO₃ (8 μ M). Genotypes of strains studied (wt, $\Delta nuo \Delta sdh$ and $\Delta iscUA$) are indicated on 533 top of the graph. Wt strain was also studied after growth in the presence of CCCP (1 534 µg/ml) as indicated. Black, dark grey, light grey and red bars are for untreated, AgNO₃, Gm 535 and Gm and AgNO₃ together. (B) MIC values of wt, $\Delta iscUA$ and $\Delta nuo \Delta sdh$ strains grown in 536 the presence of Gm (dark bar) or Gm and AgNO₃ (8 µM) (red bar). (C) Presence of silver 537 restore tritiated Gm (³H-Gm) uptake in $\Delta iscUA$ and $\Delta nuo \Delta sdh$ mutants. Uptake was 538 measured by incubating early exponential-phase cultures ($OD_{600} \sim 0.2$ with 5 µg/ml ³H-Gm in 539 presence or absence of AgNO₃ (8 µM) at 37°C. (D) Gm uptake test with sub-lethal Gm 540 concentration. The wt strain was grown up to $OD_{600} \sim 0.2$ with 0.25 µg/ml ³H-Gm in the 541 presence or absence of AgNO₃ (8 μ M) at 37°C. *P \leq 0.05; **P \leq 0.01; and ***P < 0.001 542 (Mann-Whitney U test). 543

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Figure 3. Silver potentiates Gm activity against *E. coli* by a translation-dependent process

Percentage survival of *E. co*li MG1655 after 1.5 h treatment with Gm (5 μ g/ml) and/or AgNO₃ (8 μ M) with or without chloramphenicol (Cm). Strain was grown in the presence of saturating concentration of Cm (200 μ g/ml) for 30 min before adding Gm and/or AgNO₃. Black, dark grey, light grey and red bars represent untreated, AgNO₃-treated, Gm-treated and Gm + AgNO₃-treated cultures.

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Figure 4. ROS production does not explain the positive combination between silver and aminoglycoside

(A) Percentage survival of an *E. coli* MG1655 after 1.5 h treatment with AgNO₃ (dark grey 556 bar) Gm (light grey bar), $Gm + AgNO_3$ (red bar) in anaerobic condition. Gm was used at 5 557 µg/ml and silver at 8 µM. Anaerobic condition was realized in anaerobic chamber (Cov 558 559 Chamber; 5 % H₂, and 90 % N₂) with anaerobic LB media previously equilibrated in the anaerobic chamber for 24 h. (B) Percentage survival of E. coli MG1655 after 1.5 h 560 treatment with AgNO₃ (dark grey bar) Gm (5 μ g/ml) (light grey bar), Gm + AgNO₃ (2 μ M) 561 (blue bar), $Gm + AgNO_3$ (4 μ M) (yellow bar), $Gm + AgNO_3$ (6 μ M) (green bar), Gm +562 AgNO₃ (8 μ M) (red bar) in aerobic condition. Genotypes of strains studied (wt, $\Delta sodA$ 563 $\Delta sodB$, $\Delta oxyR$ and $oxyR^{C}$) are indicated on top of the graphs. *P ≤ 0.05 and **P ≤ 0.01 564 (Mann- Whitney U test). 565

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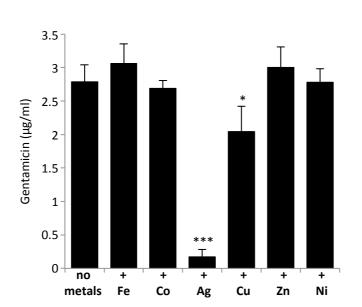
567 Figure 5. Silver sensitizes Gm resistant strain of *E. coli* and *C. difficile*

(A) MIC values (μ g/ml) of an *E. coli* Gm^R mutant (*aacl*+ strain) grown in presence of 569 AgNO₃ (8 µM). (B) Percentage survival of an *E. coli* Gm^R mutant after 1.5 h treatment with 570 571 AgNO₃ (dark grey bar) Gm (light grey bar), or Gm and silver together (red bar). Gm was used at 5 or 100 µg/ml, as indicated, and AgNO₃ 8 µM. (C) Gm uptake test. The Gm^R mutant 572 was grown up to $OD_{600} \sim 0.2$ with 5 µg/ml ³H-Gm in the presence or absence of AgNO₃ (8) 573 μ M) at 37°C. (D) MIC in μ g/ml of C. difficile strain 630 Δ erm was calculated for different 574 antibiotics. Strain $630 \Delta erm$ inoculated at an OD₆₀₀ of 0.01 was grown for 20 hours in the 575 presence or absence of 20 µM AgNO₃ and in the presence of increasing concentrations of Gm, 576 tobramycin and kanamycin. After 20 h of incubation at 37°C, the OD₆₀₀ of the strain 577 $630\Delta erm$ grown in the absence of antibiotics was similar in the presence or absence of 20 578 μ M. *P \leq 0.05; **P \leq 0.01; and ***P \leq 0.001 (Mann-Whitney U test). 579

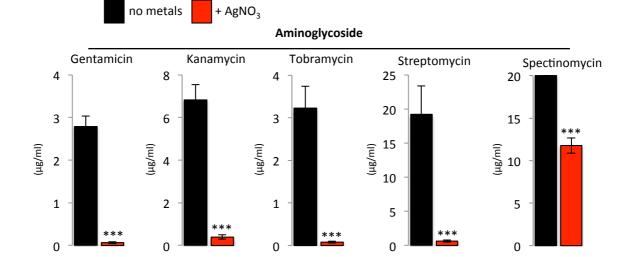
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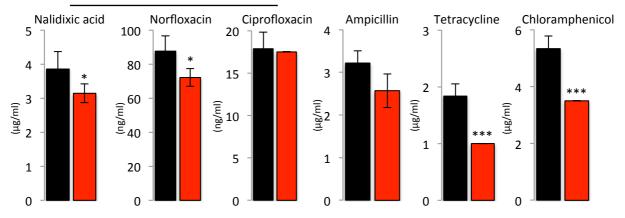
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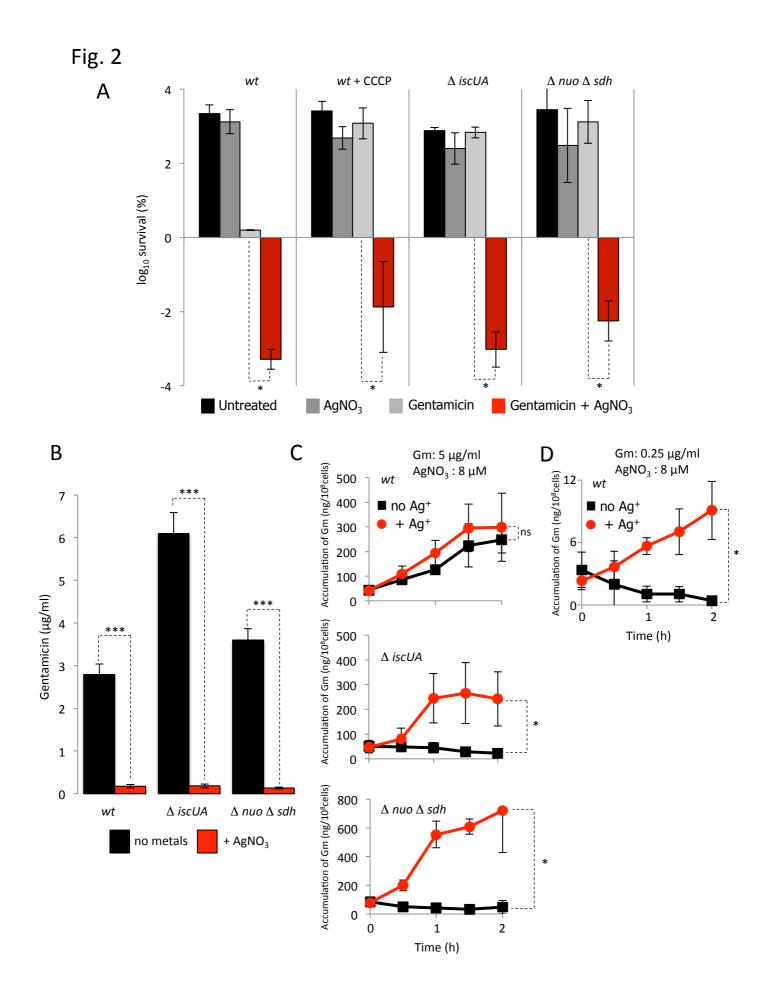


Fig. 3

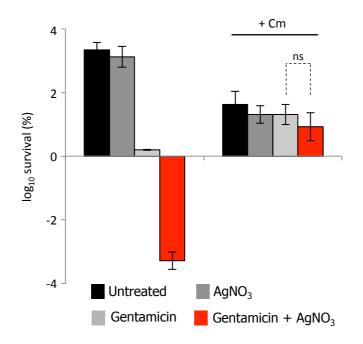
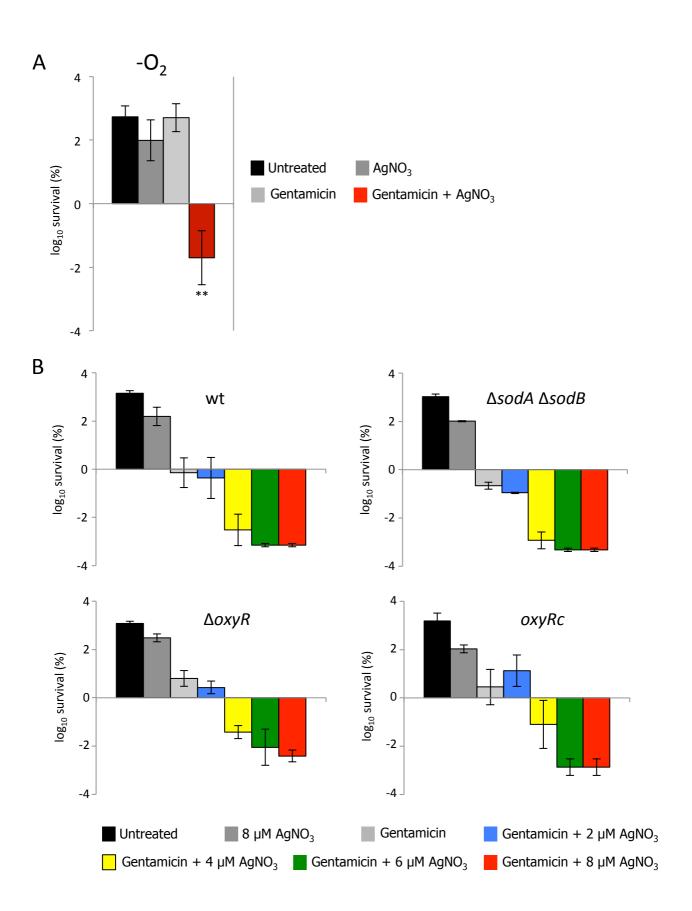
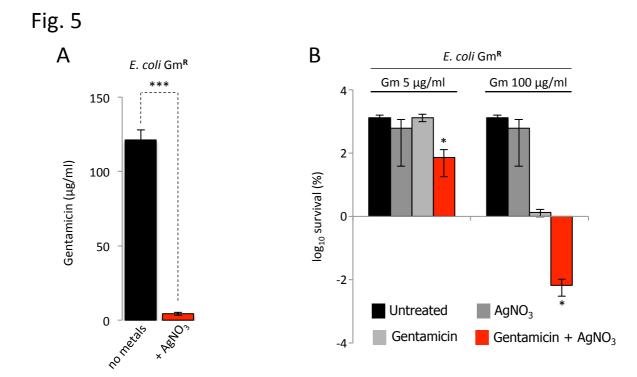
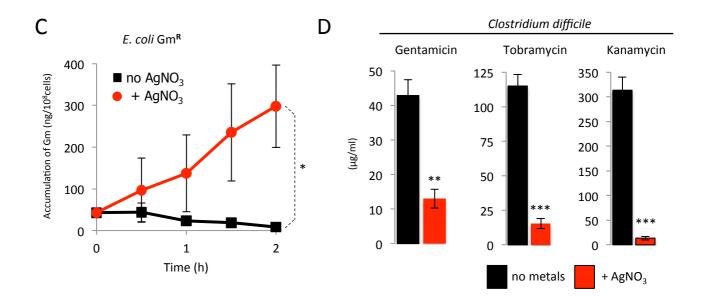


Fig. 4







Silver potentiates aminoglycoside toxicity by enhancing their uptake

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Supplementary Information

Figure Legends

Figure S1. The medium influences the silver potentiating effect

(A) MIC values (µg/ml) of Gm for an E. coli wt strain grown in LB under aerobic conditions in presence of different concentrations of AgNO₃. Black, blue, yellow, green and red bars represent conditions with no metals, 2, 4, 6 and 8 µM AgNO₃, respectively. (B) MIC values of Gm (µg/ml) for an E. coli wt strain growing in M9 minimal medium in presence of AgNO₃ (4 nM) or Cu₂(OAc)₄ (0.25 μ M). (C) MIC values of Gm (μ g/ml) for an E. coli wt strain growing in M9 minimal medium supplemented with 0.2% Casamino acids in presence of AgNO₃ (50 or 100 nM). (D) Percentage survival of E. coli MG1655 grown in M9 minimal medium supplemented with 0.2% Casamino acids after 1.5 h treatment with Gm (2, 3 and 5 µg/ml) and/or AgNO₃ (50 and 250 nM). Black, yellow and red bars represent cultures untreated, AgNO₃ (50 nM) treated and AgNO₃ (250 nM) treated, respectively. Gm concentrations are indicated on bottom of the graph. (E) MIC values of Gm (µg/ml) for an *E. coli* wt strain growing in Neidhardt's low Pi medium in presence of AgNO₃(10 nM). (F) Percentage survival of *E. coli* MG1655 grown in Neidhardt's low Pi medium after 1.5 h treatment with Gm (0.3 µg/ml) and/or AgNO₃ (10 nM). Black, dark grey, light grey and red bars represent cultures untreated, AgNO₃ treated, Gm treated and Gm + AgNO₃ treated, respectively. *P ≤ 0.05 ; **P ≤ 0.01 ; and ***P ≤ 0.001 (Mann-Whitney U test).

Figure S2. Silver potentiating effect is independent of the level of ROS production.

MIC values of Gm (μ g/ml) for *E. coli* wt and anti-ROS mutant strains grown in LB under aerobic conditions in presence of AgNO₃ (8 μ M). Genotypes of strains studied (wt, $\Delta sodA$ $\Delta sodB$ and $oxyR^{C}$) are indicated on top of the graph.

Figure S3. Fe/S clusters destabilization does not explain the positive combination between silver and aminoglycoside

(A) Survival of *E. coli* MG1655 and $\Delta iscS$ exposed to AgNO₃ (13 µM). Genotypes of strains studied (wt and $\Delta iscS$) are indicated on the graph. Black and red lines are for untreated and AgNO₃-treated cells. (B) The *iscUA* mutant is more sensitive to silver.

Dilutions of wild type and *iscUA* strains cultures were spotted on M9 medium added with, or not, silver as indicated above the figure. Growth was recorded after 20 h incubation at 37° C. (C) Cysteine protects *E. coli* against silver stress. Dilutions of wild type strain culture were spotted on M9 medium added with increasing concentrations of silver and cysteine as indicated on the figure. Growth was recorded after 20 h incubation at 37° C. (D) Percentage survival of $\Delta iscS$ mutant carrying an empty vector (pBad24) or a vector containing *csdE* gene (encoding a hetero-dimeric cysteine desulfurase) and *csdA* gene (wild type or mutated allele) after 2 h treatment with AgNO₃ (2 µM) in M9 media.

Table S1. MIC values for MG1655 E. coli strain

Metals	Products	Minimal inhibitory concentration (MIC)	Concentrations used in Fig. 1A
Iron (Fe)	FeSO ₄	> 5 mM	0.1 mM
Cobalt (Co)	CoCl ₂	2.5 mM	0.1 mM
Silver (Ag)	AgNO ₃	70 μΜ	8 μΜ
Copper (Cu)	Cu(CO ₂ CH ₃) ₂	> 5 mM	0.1 mM
Zinc (Zn)	$Zn(CO_2CH_3)_2$	1.5 mM	0.1 mM
Nickel (Ni)	NiCl ₂	2.5 mM	0.1 mM

In LB medium

In M9 minimal medium

Metals	Products	Minimal inhibitory concentration (MIC)	Concentrations used in Fig. S1
Silver (Ag)	AgNO ₃	40 nM	4 nM
Copper (Cu)	$Cu(CO_2CH_3)_2$	3 μM	0.25 μΜ

In M9 minimal medium + CASA (0.2 %)

Metals	Products	Minimal inhibitory concentration (MIC)	Concentrations used in Fig. S1
Silver (Ag)	AgNO ₃	500 nM	50; 100; 250 nM

In low phosphate minimal medium

Metals	Products	Minimal inhibitory concentration (MIC)	Concentrations used in Fig. S1
Silver (Ag)	AgNO ₃	0.1 μM	10 nM

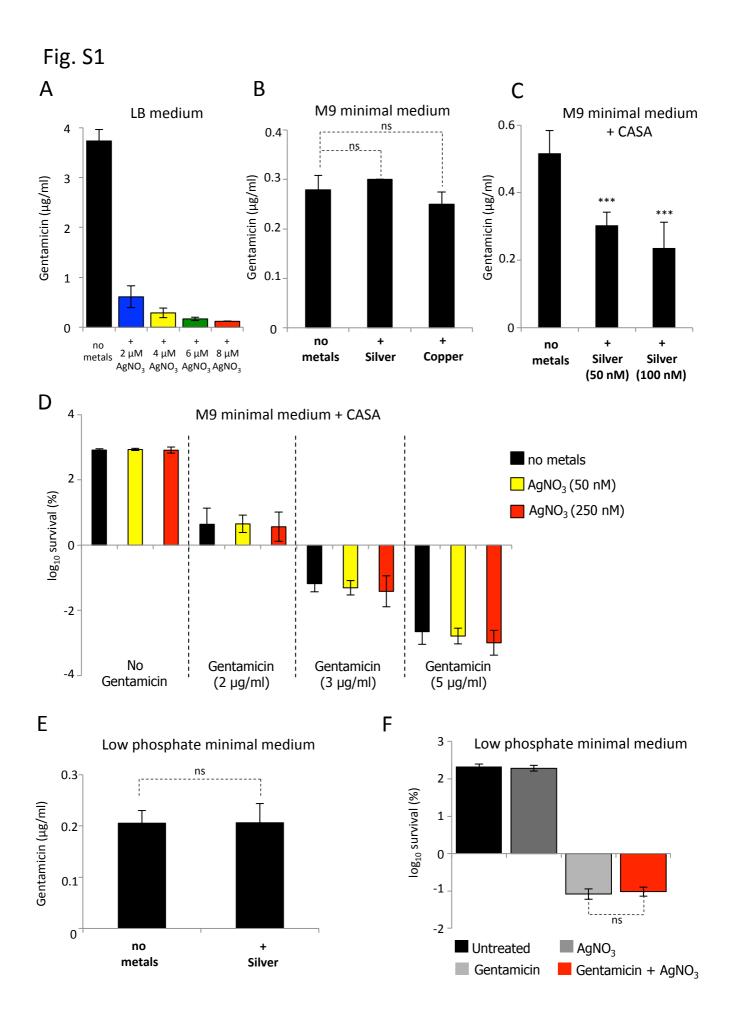


Fig. S2

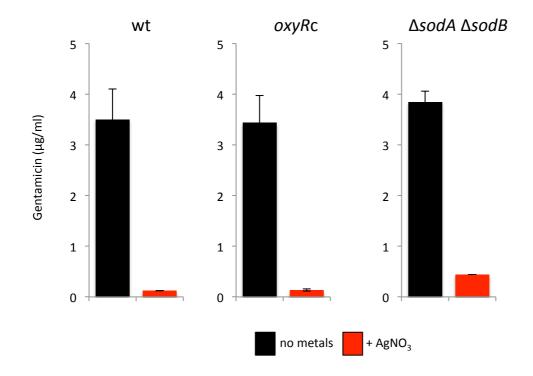


Fig. S3

