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# Mechanical Resistance due to Shearing of Partially Wet Granular Media<sup>†</sup>

Driss Oulahna\*, Romain Collet and Alain de Ryck

<sup>1</sup> Université de Toulouse, Mines Albi, CNRS, Centre RAPSODEE

## Abstract

Mixing granular media with a given amount of liquid is an operation conducted and used intensively for the preparation of concrete. The power and time required to obtain good homogeneity of the granular paste are known to have a complex relationship with the physical properties of the particles, with the liquid and with the mixer design. In this paper some of these issues are addressed by evaluating the shear resistance of a granular paste. Using model spherical materials (glass beads) and ground and sieved minerals (calcite  $\text{CaCO}_3$ ), we investigate experimentally the impact of the particle size, liquid amount and morphology of the particles.

From quasi-static experiments in shear cells, different regimes of shear resistance are revealed. In dry conditions, van der Waals' forces dominate. In wet conditions, a capillary or consolidation regime where shear resistance is dominated by capillary forces is strongly impacted by the morphology of the particles and by the formation of texture. These regimes are qualitatively observed in a bowl mixer for which the variation of the current intensity correlates with the shear resistance observed in quasi-static experiments.

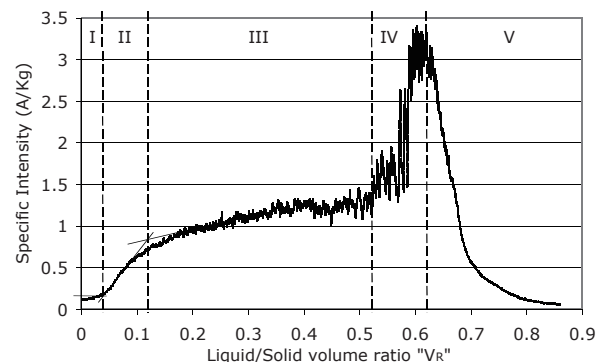
**Keywords:** wet granular mixing, shearing, quasi-static system, powder mechanics

## 1. Introduction

In order to understand why the mixing of a composition requires more or less energy than another composition, we need to classify the different factors (powders and liquid properties, process parameters) and the different forces (capillary, van der Waals, gravitational, frictional and viscous forces) that have an effect on the shear resistance<sup>1-3</sup>. Usually, power or intensity consumption measurements are taken during mixing to evaluate the different wetting states<sup>4-8</sup>. The use of progressive addition, with a controlled flow, makes it possible to differentiate the various wetting regimes. The consumption intensity profile, represented as a function of mixing time or of the amount of liquid added, may be related to different states of liquid saturation (Fig. 1a, 1b): *dry state*, *pendular state* when particles are held together by a

liquid bridge at their contact points, *capillary state*, occurring when all the voids are filled with the liquid and the surface liquid is drawn back into the pores under capillary action, and *funicular state* as a transition state between the pendular and capillary state where the voids are not fully saturated with liquid.

In a recent work<sup>9-11</sup>, we showed that the behaviour



**Fig. 1a** Intensity consumption of the calcite 1 mixed with a progressive water addition (40 ml/min; 120 rpm) and schematic wetting regime states (*pendular II, funicular III, capillary IV*)

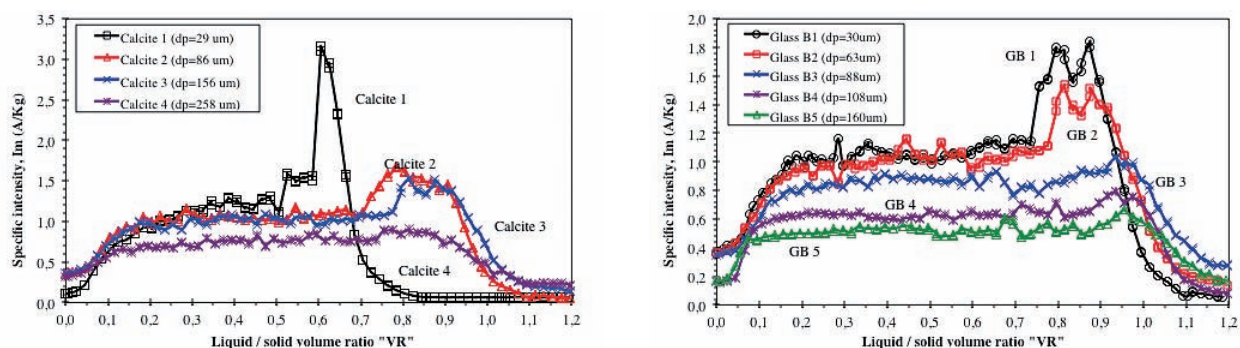
<sup>†</sup> Accepted: September 7, 2012

<sup>1</sup> Campus Jarlard, 81013 Albi, France

\* Corresponding author:

E-mail: oulahna@mines-albi.fr

TEL: +33-5-63-49-30-00 FAX: +33-5-63-49-30-25



**Fig. 1b** Intensity consumption of the calcite (left) or glass beads (right) mixed with a progressive water addition (40 ml/min; 120 rpm) in a planetary mixer CAD.

of the granular medium during mixing (planetary mixer) is greatly affected by the granular properties, by the mixer process parameters and by the liquid addition method. We also studied the interparticulate porosity of granular samples removed during mixing, and the effects of particle size. Measurements of calcite granular porosities during mixing (with water) showed that the porosity is not constant from the dry to the capillary state<sup>9-11)</sup>, the capillary state being denser than the funicular state.

The mixing energy consumption depends on the particle interaction forces, their density and how the shear deformation occurs in the mixer. To distinguish these contributions, an alternative method is to study particulate interactions in a quasi-static shear tester, for which the shear deformation is well controlled.

Generally, quasi-static tests are used with dry powders to study cohesive flow behaviour. The flow data obtained can then be used to design silos and hoppers that provide gravitational flow<sup>12, 18)</sup>. We used such

a device to obtain information about cohesive (capillary or van der Waals' forces) and frictional forces acting in wet granular media. The low velocity used leads to negligible viscous forces when compared to capillary forces. The second advantage of using a shear cell is that it simplifies definition of the shearing zone, which is reduced to a known surface plane. The simplicity of the shear area allows the shear stress to be calculated.

Section II of this paper presents the shear test, a short description of the Schulze annular shear cell and the materials used. Section III summarises the results obtained for the quasi-static shear cell experiments, taking into account the influence of the consolidated normal load, the amount of liquid, the mean diameter, and the shape of the particles. Finally, the last section presents a comparative evaluation between quasi-static and shear mixing experiments.

**Table 1** Solids properties

	$d_{v10}$ ( $\mu\text{m}$ )	$d_{v50}$ ( $\mu\text{m}$ )	$d_{v90}$ ( $\mu\text{m}$ )	$\frac{d_{v90} - d_{v10}}{d_{v50}}$	True density $\rho_s$ ( $\text{g}/\text{cm}^3$ )	Bulk density $\rho$ ( $\text{g}/\text{cm}^3$ )	Porosity $\varepsilon$ (%) (dry state)	Contact angle (water) $\theta$	$i_{v10}$	Sphericity index $i_{v50}$	$i_{v90}$
Glass beads 1	18	30	48	1.00	2.48	1.44	42	0	0.45	0.91	0.95
Glass beads 2	45	63	88	0.68	2.51	1.43	43	0	-	-	-
Glass beads 3	64	88	122	0.66	2.51	1.49	41	0	0.47	0.96	0.98
Glass beads 4	78	108	149	0.66	-	-	-	0	-	-	-
Glass beads 5	117	160	219	0.64	2.52	1.49	41	0	0.68	0.96	0.99
Glass beads 6	164	218	295	0.60	-	-	-	0	-	-	-
Glass beads 7	194	261	353	0.61	2.52	1.46	42	0	0.60	0.94	0.99
Glass beads 8	244	332	455	0.64	2.49	1.49	40	0	-	-	-
Calcites 1	3	29	68	2.24	2.73	1.32	51	51	0.35	0.71	0.90
Calcites 2	55	86	142	1.01	0.73	1.31	52	51	0.49	0.77	0.92
Calcites 3	102	156	225	0.79	0.73	1.36	50	51	0.51	0.73	0.89
Calcites 4	181	258	354	0.67	0.73	1.38	49	51	0.45	0.71	0.88

## 2. Materials, Experimental Set-up and Methods

The materials used (**Table 1**) were spherical glass beads supplied by SOVITEC<sup>®</sup> and calcites supplied by OMYA<sup>®</sup>. The volume size distributions were obtained by a MALVERN<sup>®</sup> Mastersizer 2000. The wetting angles with water  $\theta$  were obtained from capillary rise experiments, using the methodology of Galet<sup>13)</sup>. The average volume roundness index  $i_{v,50}$  (ratio of the smaller and higher diameters of the particle) came from a volumetric roundness distribution obtained with the MALVERN<sup>®</sup> PharmaVision 830 (PVS) automated optical device (2D). The true density of the solids  $\rho_s$  was obtained using a helium pycnometer (Accupyc 1330-MICROMERITIC<sup>®</sup>), and the dry porosities  $\varepsilon$  (or the compacity C) by using the values of the true  $\rho_s$  and bulk  $\rho$  densities:

$$\varepsilon = 1 - \frac{\rho}{\rho_s} = 1 - C \quad (1)$$

The liquid used for all the tests was water.

Every quasi-static shear test was carried out with a dry or wet granular medium previously mixed in a planetary mixer from the CAD<sup>®</sup> Company (Mlx60)<sup>9-11)</sup>. This mixer can be used to make a paste, and it allows the power consumption to be recorded at a frequency of 10 points per second. The available rotation velocity is between 0 and 300 rpm and the volume of the bowl is 4 litres. A preliminary study allowed the process parameters to be set at the following values: an impeller speed of 120 rpm, a droplet-by-droplet liquid flow rate of 40 ml/min, and an initial bulk volume of 1250 cm<sup>3</sup>, corresponding to a 32% filling rate of the bowl mixer.

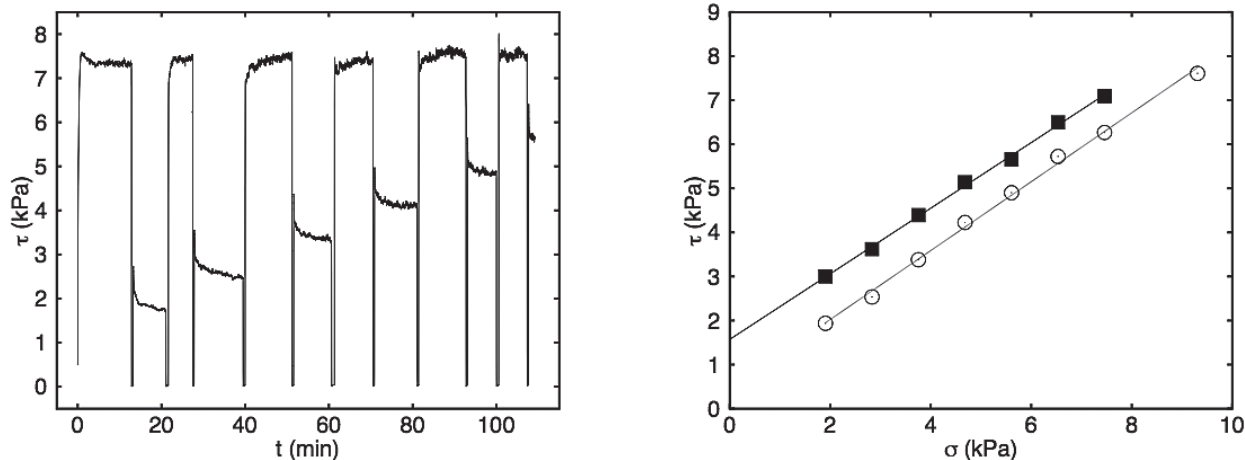
To study the interaction forces that take place dur-

ing the shearing, we used an annular shear cell allowing a sheared interface of known constant area. The shear velocity was given by the rotational speed ( $\omega = 3.5 \cdot 10^3$  rad/s, ranges from 100 to 200  $\mu\text{m/s}$  from the inner (6 cm) to the outer (12 cm) diameter of the cell). These velocities give very low capillary numbers (of the order  $10^{-5}$ ), allowing the viscous forces to be neglected in these quasi-static experiments.

An experimental standard method developed by Jenike<sup>14, 18)</sup> was used. This method consists of measuring a yield locus, i.e. the tangential stress necessary to shear a granular medium for a given normal stress and a powder sample previously prepared by a consolidation process. It is a two-step measurement. During the first step, the shear under a normal stress  $\sigma_c$  consolidates the granular sample to a critical state where there are no further variations in volume. The resulting consolidated shearing stress  $\tau_c$  is constant. The sample is then sheared under a normal load  $\sigma$ , which is lower than  $\sigma_c$ . This second step allows the yield shear stress  $\tau_y$  to be obtained at the peak, and the steady-state shear stress  $\tau_D$  to be measured when a plateau is reached again in the shear stress versus deformation curve or time. These two stages are repeated several times with a constant consolidated normal stress  $\sigma_c$  and an increasing normal stress<sup>15-16)</sup> (see **Fig. 2**).

The curves representing the yield and the steady-state shear stress ( $\tau_s$  or  $\tau_D$ ) versus the normal stress  $\sigma$  are called the yield<sup>17-19)</sup> and the dynamic locus, respectively. In the present study, they are both well fitted by a linear Coulombic law:

$$\tau_x = \mu_x \sigma + c_x \quad (2)$$



**Fig. 2** Shear test (left) and corresponding yield (squares) and dynamic (circles) locii for the 50-125  $\mu\text{m}$  sieved calcite with a liquid to solid ratio of 0.5. The consolidation normal stress here is  $\sigma_c = 9.3$  kPa.

It was observed experimentally that the dynamic cohesion is close to zero and the dynamic behaviour is only characterised by the dynamic friction coefficient ( $\mu_D$ ) linked to the frictional forces between particles. The cohesion  $c$  ( $= c_s$ ), coming from the yield locus built with peak stresses is directly linked to the cohesive forces (capillary and van der Waals' forces).

Since the cohesion appears to increase with the consolidation, this procedure is repeated for at least three normal consolidation stresses. Finally, following the Jenike methodology, rather than using the pair ( $c$ ,  $\sigma_c$ ), we compute the unconfined yield stress  $f_c$  and the maximal principal stress  $\sigma_I$ . These quantities are obtained from the yield locus using the Mohr circles representative of the stress state of an unconfined material at failure, and the maximal stress state for this consolidation (the left and right circles in **Fig. 3**, respectively).  $f_c$  is the minimal stress necessary to shear an unconfined consolidated material and  $\sigma_I$  the maximal stress that this consolidated material has experienced previously. It is observed that the greater the consolidation, the greater the unconfined yield stress. The one-point Flow Function as defined by Schultze<sup>18)</sup> is the ratio ( $\sigma_I/f_c$ ). In this paper, we investigate the evolution of the ratio ( $f_c/\sigma_I$ ): inverse ratio of the definition of Schultze. Changes in the ratio is an increasing function whose slope is limited. Indeed, the yield stress  $\tau_y$  remains lower than the

shear stress at consolidation ( $\tau_c = \mu_D \sigma_c$ ), which gives the yield locus in **Fig. 3** a positive slope. As a consequence, the diameter of the left Mohr circle is, at most, equal to the diameter of the right Mohr circle, giving the following:

$$\frac{f_c}{\sigma_I} \leq \frac{2\mu_D}{1 + \mu_D} \quad (3)$$

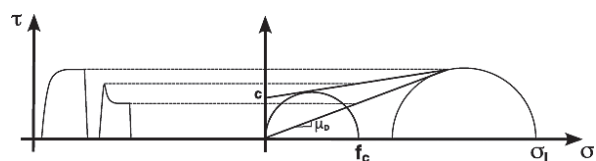
### 3. Quasi-Static Results and Discussion

#### 3.1 Consolidation and liquid amount effect

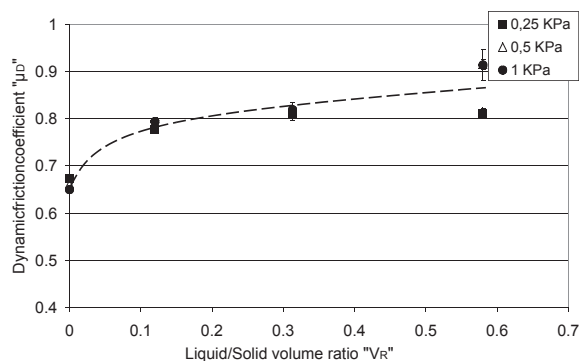
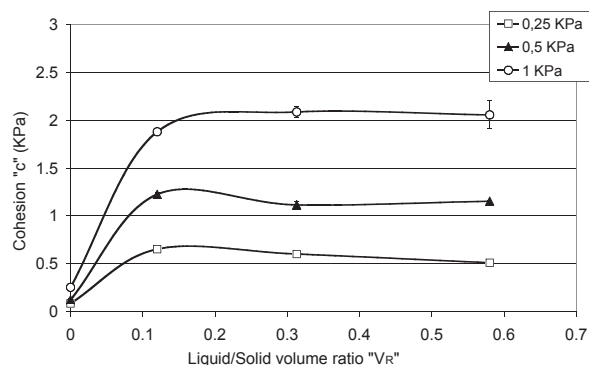
The effects of consolidation and liquid amount on cohesive and frictional interactions were studied with both glass beads and calcites at different particle size distributions (**Table 1**). They were sheared in dry conditions and wet states with a volume-liquid ratio ( $V_R$ : ratio of liquid volume by solid volume) up to 0.6. From each yield locus, the cohesion is linearly extrapolated and **Fig. 4-left** gives the results obtained for the calcite sieved between 50 and 125  $\mu\text{m}$ . The cohesion, which has a non-zero value at the dry state due to the van der Waals' forces, increases when adding the liquid to reach a plateau value when the liquid amount attains 10% in volume. This plateau is maintained until 60% in volume.

Three consolidation normal stresses were used (0.25, 0.5 and 1 kPa). It was observed that the cohesion increases proportionally to the consolidation normal stress. As a consequence, for reasons of clarity, the results will be presented using the ratio  $f_c/\sigma_I$ . This ratio is independent of the consolidation, in the range of liquid content between 10 to 60%. The Flow Function (the relationship between the unconfined yield stress  $f_c$  and the maximal principal stress  $\sigma_I$ ) is therefore fully characterised by their ratio.

To conclude this section, the cohesion for wetted



**Fig. 3** Yield locus construction from shear experiment data. Mohr circles for the determination of  $f_c$  and  $\sigma_I$ .



**Fig. 4** Cohesion (left) and dynamic friction coefficient (right) of the calcite 50-125  $\mu\text{m}$  as a function of liquid amount, for 3 consolidation normal stresses (0.25, 0.5 and 1 kPa).

granular media is linked both to the liquid and to the compacity (caused by the consolidation procedure in a shearing test). Strengthening by the addition of liquid is considerable from the dry state to a liquid/solid volume ratio equal to 0.1. For higher liquid amounts, the cohesion remains constant, and the strength is only dependent on the consolidation. These results are in accordance with the fact that the capillary force does not vary very much with the volume of the liquid bridge<sup>20,21</sup>. As a consequence, in the following section, we shall restrict discussion to only two cases: the dry state and the funicular state. The latter is chosen at a mean value for a liquid ratio between 30 to 40%, for which the quasi-static results and energy consumption for mixing presented in section IV are mainly independent of the liquid ratio. Finally, the presence of liquid also modifies the dynamic friction behaviour of the granular medium.

### 3. 2 Particle size effect

For both glass beads and calcites, the evolutions of the dynamic coefficient of friction  $\mu_D$  and the ratio  $(f_c/\sigma_f)$  were studied while varying the mean size of the quasi-monodisperse granular materials.

Glass beads and calcites with different sizes (between 30 and 330  $\mu\text{m}$ ) were used to perform shear experiments in the dry and funicular wet state. The ratio  $(f_c/\sigma_f)$  and dynamic friction coefficient data from yield and dynamic loci are displayed as a function of the inverse of particle diameter in **Fig. 5**.

For the unconfined yield stress normalised by the principal maximal consolidation stress, **Fig. 5**-left displays the behaviour at dry and funicular conditions.

**Dry case, calcites and glass beads:** The evolution of  $(f_c/\sigma_f)$  for the dry case is close to being

inversely proportional to the particle size, as may be inferred by the Rumpf relation<sup>22</sup>. Rumpf writes the tensional resistance  $T$  of a cohesive granular material as:

$$T = \frac{2}{\pi} \frac{f}{d^2} (1 - \varepsilon) \kappa, \quad (4)$$

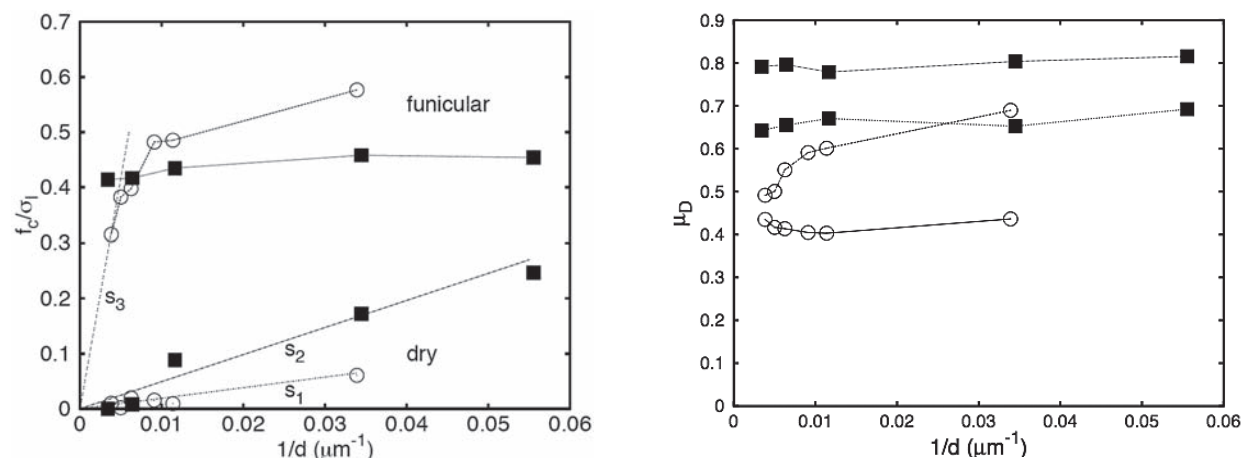
where  $f$  is the interparticulate force,  $d$  the particle diameter,  $\varepsilon$  the porosity and  $\kappa$  the coordination number. For dry particles, the interparticulate forces are van der Waals' forces, which for spheres are given by:

$$f_{vdW} \approx \frac{Ad}{h^2}, \quad (5)$$

where  $A$  is the Hamaker constant and  $h$  the interdistance between particles. They are proportional to the particle size and therefore the tensional or shear resistance of the dry powder is expected and observed in **Fig. 5**-left to be proportional to the inverse of the particle size (Eq. 5 in Eq. 4).

It is possible to obtain the constant of proportionality between the Flow Function and the particle size for the calcites and the glass beads by measuring the slopes  $s_1$  and  $s_2$  in **Fig. 5**-left. Their difference when changing the material is due to several factors: the Hamaker constant  $A$  is different, the roughness and sphericity of both materials are different and finally, the texture of the particle assembly, i.e. the porosity and coordination number may be different. For  $\varepsilon$  and  $\kappa$ , which are geometric numbers, we expect equivalent values for both materials since they are close to monodispersity.

We find that the ratio of the slopes of the linear fits of  $(f_c/\sigma_f)$  versus  $(d^{-1})$  in the dry conditions for the



**Fig. 5** Inverse of the one-point Flow Function of calcites and glass beads (left) or coefficient of dynamic friction (right) as a function of the inverse of the mean diameter for dry and funicular states. Squares: calcites, circles: glass beads.

calcite and glass beads gives a factor of 2.4, which is similar to the ratio of the Hamaker constants of the two materials ( $A_{\text{glass}} = 6.5 \cdot 10^{20}$  J,  $A_{\text{calcite}} = 10.1 \cdot 10^{20}$  J). This result is quite satisfying and indicates that the particle roughness and the morphology of the calcite, compared to the glass beads, have a low impact on the magnitude of the van der Waals' forces.

**Wet case, glass beads:** In the funicular state, the dominant interparticulate force is the capillary force, which can be written as:

$$f_{\text{capillary}} \approx \gamma d, \quad (6)$$

in the case of wetted spheres (Fig. 6-left) where  $\gamma$  is the surface tension of the liquid employed. Here again, the interparticulate force is proportional to the particle size, and we therefore expected shear resistance to increase proportionally with  $d^{-1}$ , with a constant of proportionality greater than in the dry case. However, this is not what was observed. Although we noticed an increase in the shear resistance when decreasing the size (Fig. 5-left, upper data points), the proportionality with  $d^{-1}$  was only perceptible for the glass beads at particle sizes greater than 100  $\mu\text{m}$ . The linear fit between  $(f_c/\sigma_f)$  and  $(d^{-1})$  in this range gives a slope s3 (Fig. 5-left) 500 times greater than the slope in the dry case (slope s1 in Fig. 5-left). The ratio of capillary and van der Waals' forces is then of the order  $(f_{\text{capillary}}/f_{\text{vdW}}) \sim \frac{\gamma h^2}{A} \sim 500$ , leading to a van der Waals' interdistance (roughness)  $h_{\text{glass-beads}} \approx 20$  nm, a reasonable value giving some confidence in the interpretation of the quasi-static shear resistance for glass beads of a size greater than 100  $\mu\text{m}$ .

For the smaller sizes of glass beads, the shear strength becomes lower than that extrapolated from the behaviour of  $(d^{-1})$ . This is due to the fact that the ratio  $f_c/\sigma_f$  is bound to a value around 0.8 for the glass beads (Eq. 3). We then observe the transition between a capillary regime for large particles (larger than 100  $\mu\text{m}$ ) where the shear resistance decreases as the inverse of the particle size when increasing the size, and a consolidation regime for the small sizes (smaller than 100  $\mu\text{m}$ ) where the shear resistance cannot increase proportionally with the inverse of the particle size, but is limited by the consolidation.

In this small size regime, the Rumpf relation (Eq.



**Fig. 6** Liquid bridge of wetted spheres and (left) radius  $r$  of curvature of the meniscii for particle with some flatness (right).

4) suggests that a formation of texture takes place and that the porosity and/or coordination number decreases when the particle size decreases. This observation is corroborated by the observation of the dynamic coefficient of friction (Fig. 5-right).

For glass beads at high particle sizes, the difference in behaviour for the funicular and dry states tends towards zero, which is expected if the capillary interparticle forces become negligible compared to the particle weight, together with a similar texturation of the powder bed.

**Wet case, calcites:** For the wet calcites, in the range of sizes used (30 to 250  $\mu\text{m}$ ), the shear strength is quasi-constant ( $f_c/\sigma_f \approx 0.42 \pm 0.2$ ). In particular, we do not observe the regime at high particle size approaching the dry case, as for the glass beads. This difference of behaviour between the glass beads and the calcites is interpreted as follows: the tensional or shear resistance is scaled by  $f_{\text{capillary}}/d^2$ , where the capillary force scales like the product of the Laplace pressure multiplied by the area of the meniscii. For wetted particles, the area of the meniscii scales like  $d^2$  and therefore we have:

$$f_{\text{capillary}} \approx \frac{\gamma}{r} d^2, \quad (7)$$

where  $r$  is the radius of curvature of the meniscii. If the particles are smooth spheres like the glass beads,  $r \sim d$  and  $T \sim \gamma/d$ . If the particles are rough with some flatness,  $r$  is close to the roughness, and then  $T \sim \gamma/r$  is independent of the particle size (see Fig. 6-right). The independency of the strength and the particle size for the calcites suggests that this morphology effect takes place. For grains of irregular shapes with flattened surface areas, the curvature of the meniscii is then controlled by the local roughness of these areas, leading to a constant shear resistance. Nevertheless, this argument supposes that the local roughness is independent of the particle size for the range of particle sizes studied here. This point merits further investigation.

This size independency is also observed in dynamic conditions for the corresponding  $\mu_D$  in dry and funicular conditions (Fig. 5-right). In particular, there is no convergence of the wet dynamic coefficient of friction towards the dry one when the particle size is increased, indicating that the capillary forces remain greater than the weight of the particles for all the sizes of calcites used.

**Conclusion on the particle size effect in quasi-static experiments:** The use of a shear cell allowed us to measure quantitatively the Flow Function of spherical and non-spherical particles in dry

and wet conditions. From the results obtained, we extracted different shear resistance mechanisms. When the particles are dry, the shear resistance is induced by van der Waals' interparticle forces (Eq. 4), which leads to a shear resistance inversely proportional to the mean size of the particles. This shear resistance remains weak enough to remain below the saturation due to consolidation, and we observe only this van der Waals-dominated regime for all the range of particle sizes for both calcites and glass beads.

When they are wetted by 10 to 60% v/v by water, the shear resistance is induced by the capillary forces, which are of higher magnitude than the van der Waals' forces. Depending on the morphology of the particles, this capillary force is given by Eqs. (5) or (6), leading to a shear resistance inversely proportional to the mean particle size for the glass beads and a constant shear resistance for the non-spherical calcites. This capillary regime may be dominated by a consolidation regime where the shear resistance is limited by the consolidation (Eq. 3).

These quantitative measurements allow us to discriminate between different regimes of shear resistance. We shall now investigate whether these different regimes can be qualitatively observed in a planetary mixer, for which neither the surface of the sheared area nor the stresses acting upon it can be obtained, and only the torque on the impeller is measurable.

#### 4. Comparison Between Quasi-Static and Shear-Mixing Experiments

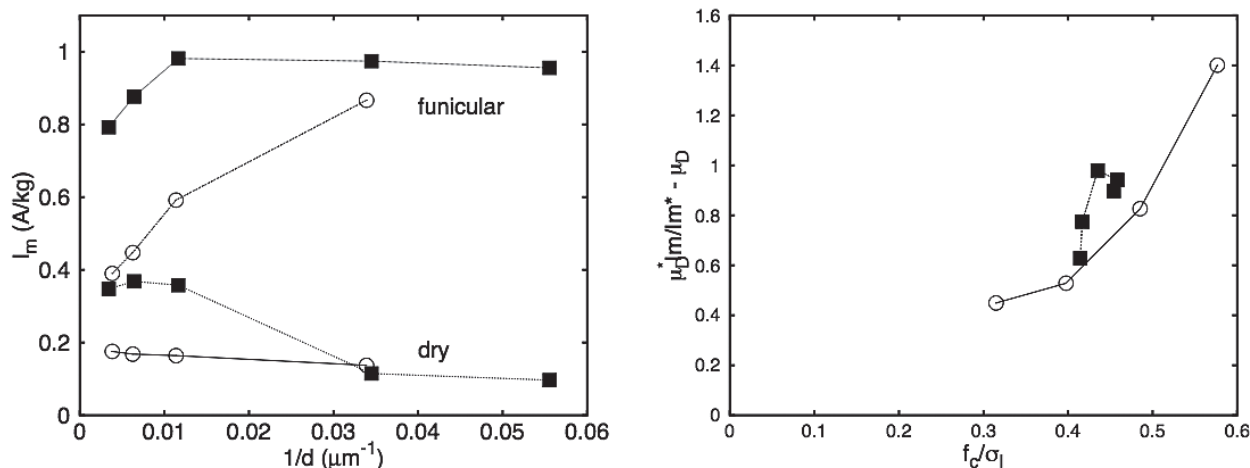
In order to compare the quasi-static shear experi-

ments with the shear-mixing experiments, we recorded the current intensity required to mix at constant velocity in dry and funicular conditions obtained by reading the actual current intensity minus the current intensity when the bowl is empty. The rotational velocity of the impeller was also recorded in order to check that it remained constant and thus provided a proportionality between this current intensity and the torque on the impeller or the power consumed. For these experiments, the mass was adjusted to obtain the same bulk volume in the mixer (leading to  $M_{\text{glass}} = 1830 \pm 30$  g and  $M_{\text{calcite}} = 1650 \pm 150$  g). We should then expect that the surface sheared remained approximately the same for all the experiments. **Fig. 7** reports the ratio  $I_m = I/M$  obtained for the calcites and glass beads.

This mixing intensity was expected to be proportional to a mean force acting on the bowl, which may be heuristically written as:

$$I \propto S(c + \mu_D \sigma), \quad (8)$$

where  $S$  is the mean sheared surface in the mixer, which is not properly defined, as in the quasi-static experiments of section III.c, and  $\sigma$  the mean cohesion and normal stress exerted by the powder on this shear plane. The normal stress was again not measured, but for a half-filled bowl, we expected that the main contribution would be the weight of the particles themselves.  $\sigma$  is then expected to be proportional to the mass  $M$ , introduced as: ( $\sigma = \beta M$ ), where  $\beta$  is a geometrical proportionality constant. The powder in the mixer to be consolidated by this normal stress, and therefore the cohesion, writes



**Fig. 7** Left: specific intensity at dry or funicular state versus the inverse of the mean diameter. Right: scaled specific intensity at funicular state (Eq. 6) versus the inverse of the one-point Flow Function. Squares: calcites, circles: glass beads.



as:  $c = \lambda (f_c/\sigma) \sigma$ , where  $\lambda$  is a coefficient of order unity. We then have:

$$I_m = S\beta(\lambda \frac{f_c}{\sigma_I} + \mu_D) \quad (9)$$

In the latter expression,  $(S\beta)$  is unknown. In order to eliminate this unknown parameter, we can only compare two experiments with the same geometry. In dry conditions and for particle sizes greater than 100  $\mu\text{m}$ , the mechanical behaviour of the powders used are those of a non-cohesive powder. As a consequence, we have:

$$I_m^* = S\beta\mu_D^*, \quad (10)$$

where the asterisk indicates the dry non-cohesive conditions. From Eqs. (9) and (10), we can now relate the measured intensities and quasi-static characterizations of the powders with the following expression:

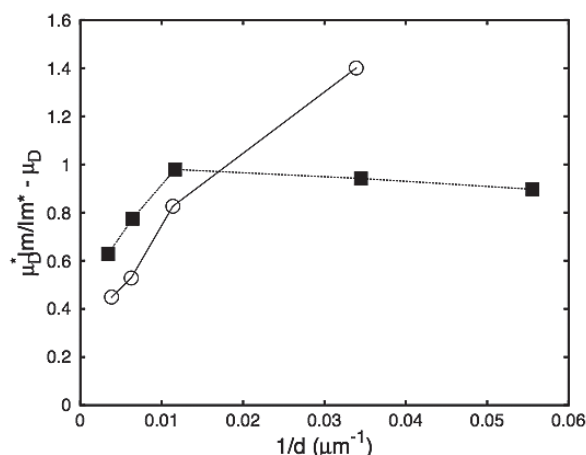
$$\mu_D^* \frac{I_m}{I_m^*} - \mu_D \propto \frac{f_c}{\sigma_I}, \quad (11)$$

where the coefficients of order unity are taken to unity. This relation was checked by plotting the left member of Eq. (11) versus the right member in **Fig. 7-right**.

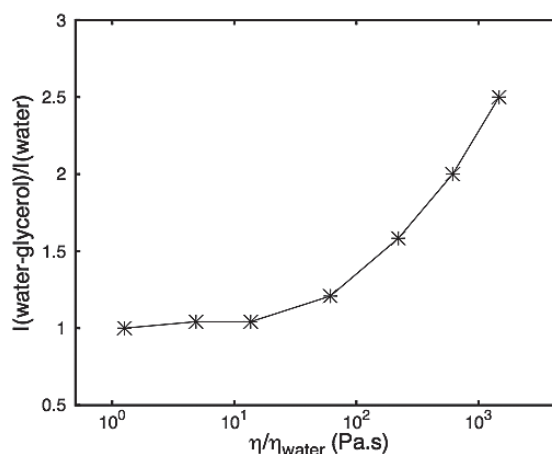
We observe in this figure that the calcites present a weak variation in both coordinates: in quasi-static and dynamic experiments, the mechanical resistance is mainly constant. The data points representing the calcite collapse, along with those obtained for the glass beads. For the glass beads, the variations for both coordinates are in correct linear correlation. These correlations remain qualitative and although it would be of interest to add more experimental data for different materials and equipments, they nevertheless give some credence to the assertion that the shear resistance in the mixer at funicular state may be described by the quasi-static resistance of a powder consolidated and submitted to its own weight.

The correlations indicate that the variation of the shear resistance with the mean size of the particles is similar in both quasi-static shear cell and mixer for dry and funicular conditions as directly observed in **Fig. 8**, where the results of **Fig. 7** are plotted versus  $d^{-1}$  for direct comparison with **Fig. 5-left** (two upper curves).

Finally, for the glass beads of 100  $\mu\text{m}$  mean diameter, we checked that the viscous forces remain negligible in the mixer under the operating conditions or the rotational velocity used. To do this we performed a series of experiments with water-glycerol mixtures of different viscosities. The results obtained are dis-



**Fig. 8** Results of **Fig. 7-right** versus the inverse of the mean diameter. Squares: calcites, circles: glass beads.



**Fig. 9** Intensity consumption at funicular state for water-glycerol, scaled by the intensity using water, versus the corresponding scaled viscosity for 100  $\mu\text{m}$  glass beads.

played in **Fig. 9**, which shows the intensity versus the viscosity, both scaled with the values obtained for pure water. It is particularly noticeable that the experiments with water are on a plateau:  $I$  does not depend on the viscosity, indicating that the viscous forces are negligible compared to friction and capillary forces in our experiments with water.

## 5. Concluding Remarks

A quasi-static system was used to obtain the cohesion (due to capillary or van der Waals' forces) and frictional forces acting in sheared granular media. They present a rich phenomenology, depending on the dominant forces in action and on the texturation

taken by the paste.

This quasi-static behaviour is shown to explain qualitatively the behaviour of the same granular media sheared in a planetary mixer at the funicular state. The agreement between the two types of experiments suggests that the consolidation process in the bowl mixer is identical for all the samples explored and close to a consolidation due to the weight of the particles themselves.

This correlation between quasi-static experiments and planetary mixer experiments indicates that the strengthening of the granular material is similar in the funicular state in both cases. The shear resistance regimes investigated quantitatively in the quasi-static shear cell (van der Waals, capillary particle morphology and texturation regimes) all apply.

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### List of symbols

$A$	Hamaker constant
$C$	Compacity (%)
$c$	Cohesion (Pa)
$d=dv_{50}$	Mean particle diameter ( $\mu\text{m}$ )= $dp$
$f_c$	Unconfined yield stress (Pa), (another common designation is $\sigma_c$ )
$f_{capillary}$	Capillary force
$f_{vdW}$	van der Waals force
$h$	Interdistance between solid particles
$iv_{50}$	Mean volume roundness index (-), sphericity index
$I$	Intensity (A)
$Im$	Specific intensity (A/Kg), = $I/M_s$
$k$	Coordination number
$m_s$	Solid mass (g)
$M_s$	Solid mass (kg)
$r$	Radius of curvature of meniscii.
$S$	Mean sheared surface
$T$	Tensional resistance of a cohesive granular materials
$t$	Time (s)
$(\sigma_I/f_c)$	One-point Flow Function (-)
$V_p$	Porous volume ( $\text{cm}^3$ )
$V_s$	Volume of dry solid ( $\text{cm}^3$ )
$V_L$	Volume of liquid ( $\text{cm}^3$ )
$V_R$	Ratio of liquid volume by solid volume (-)

$V_T$  Total volume ( $\text{cm}^3$ ) = ( $V_p + V_s$ )

### Greek symbols

$\varepsilon$	Interparticulate porosity of dry granular medium
$\gamma = \gamma_L$	Liquid surface tension ( $\text{mN.m}^{-1}$ )
$\theta$	Liquid-solid contact angle ( $^\circ$ ), by capillary rise experiments
$\rho_s$	Solid true density ( $\text{g.cm}^{-3}$ )
$\rho$	Solid bulk density ( $\text{g.cm}^{-3}$ )
$\omega$	Angular shear velocity ( $\text{rad.s}^{-1}$ )
$\sigma$	Normal stress (Pa)
$\sigma_c$	Consolidation normal stress (Pa)
$\tau_s$	Yield shear stress (Pa)
$\tau_D$	Steady-state shear stress (Pa)
$\tau_C$	Consolidation shear stress (Pa)
$\mu_D$	Dynamic friction coefficient (-)
$\sigma_I$	Major principal stress (Pa),
$\eta_{water}$	Water viscosity (Pa.s)
$\eta_{W-G}$	Water-glycerol viscosity (Pa.s)
$\beta$	Coefficient: geometrical proportionality constant (Eq.9)
$\lambda$	Coefficient of order unity (Eq.9)

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## Author's short biography



### Driss Oulahna

Driss Oulahna received his chemical engineering degree in 1988 and his PhD in chemical and process engineering in 1992 from École des Mines de Saint-Etienne, France. After a post-doctoral position, he joined the Process Engineering Laboratory of Rapsodee Center-CNRS 5302, École des Mines d'Albi (France) as assistant professor in 1996. His research is in the field of powder technology (agglomeration, granulation, compaction) and particle characterisation methods such as solids-liquids interactions and their role in chemical engineering processes.



### Romain Collet

Romain Collet obtained a master's degree in "Applied physico-chemistry of Materials" in 2006 from Montpellier II university. After a PhD at École des Mines d'Albi where he studied "Criteria of blending wet granular mixtures", he joined Parex-Group (R&D center, Saint Quentin Fallavier, France). He is currently an R&D engineer and works on the formulation of ready-to-use mortars for floors (tile adhesive, grout and floor covering).



### Alain De Ryck

Alain de Ryck is a graduate of the École polytechnique (1990), has a MSc in solid physics (Université Paris-Sud) and a PhD in physics (UPMC, Paris 1994). After a post-doctoral position in fluid mechanics at the Instituto Pluridisciplinar-Universidad Complutense, Madrid, he joined the Rapsodee Center-CNRS 5302, Ecole des Mines d'Albi in 1996. Professor and former director, he currently works on granular friction and flows of cohesive bulk materials, gas-solid interactions.