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# Low effect of phenanthrene bioaccessibility on its biodegradation in diffusely contaminated soil

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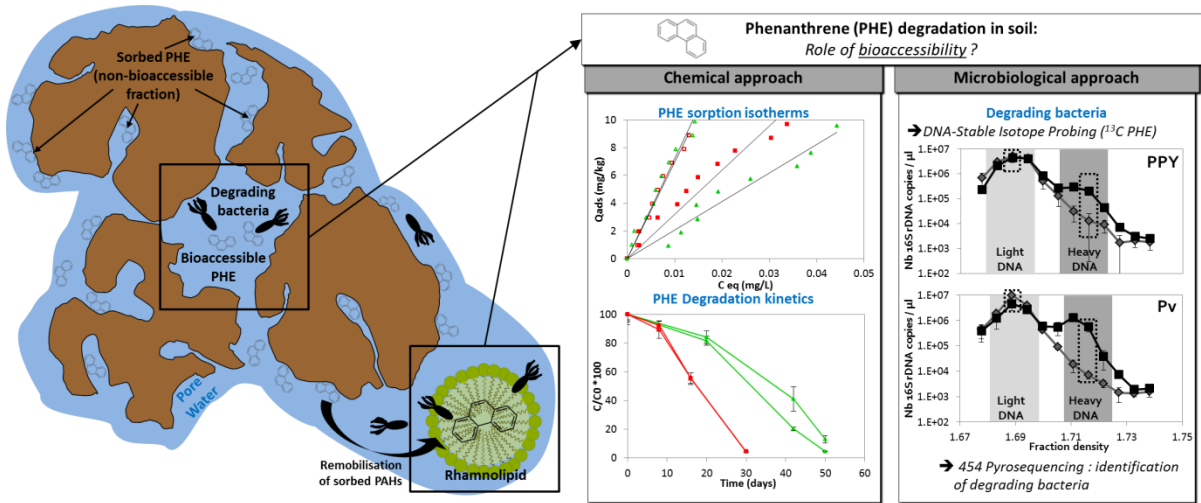
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**Short title:** Low effect of bioaccessibility on phenanthrene biodegradation in soil

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*Rhodocyclaceae*

18 **Graphical Abstract**



20 **Abstract**

21 This study focused on the role of bioaccessibility in the phenanthrene (PHE) biodegradation in  
22 diffusely contaminated soil, by combining chemical and microbiological approaches. First, we  
23 determined PHE dissipation rates and PHE sorption/desorption kinetics for two soils (PPY and  
24 Pv) presenting similar chronic PAH contamination, but different physico-chemical properties.  
25 Our results revealed that the PHE dissipation rate was significantly higher in the Pv soil  
26 compared to the PPY soil, while PHE sorption/desorption kinetics were similar. Interestingly,  
27 increases of PHE desorption and potentially of PHE bioaccessibility were observed for both  
28 soils when adding rhamnolipids. Second, using <sup>13</sup>C-PHE incubated in the same soils, we  
29 analysed the PHE degrading bacterial communities. The combination of stable isotope probing  
30 (DNA-SIP) and 16S rRNA gene pyrosequencing revealed that *Betaproteobacteria* were the  
31 main PHE degraders in the Pv soil, while a higher bacterial diversity (*Alpha*-, *Beta*-,  
32 *Gammaproteobacteria* and *Actinobacteria*) was involved in PHE degradation in the PPY soil.  
33 The amendment of biosurfactants commonly used in biostimulation methods (*i.e.* rhamnolipids)  
34 to the two soils clearly modified the PHE sorption/desorption kinetics, but had no significant  
35 impact on PHE degradation rates and PHE-degraders identity. These results demonstrated that

36 increasing the bioaccessibility of PHE has a low impact on its degradation and on the functional  
37 populations involved in this degradation.

38

39

#### 40 **Capsule**

41 The increase of the bioaccessibility level of phenanthrene has a low impact on its degradation  
42 in diffusely contaminated soil.

## Introduction

Polycyclic aromatic hydrocarbons (PAHs) are ubiquitous carcinogenic organic contaminants, originating from multiple sources (*i.e.* biogenic, petrogenic and pyrogenic). Among them, human activities are responsible for the major releases into the environment, particularly through atmospheric deposition (Tobiszewski and Namieśnik, 2012). PAHs physico-chemical properties (high chemical stability, low vapor pressure, low aqueous solubility, high water-organic carbon  $K_{oc}$  partition coefficient) make them persistent in soils. However, this persistence is also strongly dependent on soil physico-chemical properties and soil microbial activities (Doyle et al., 2008).

In the case of diffuse contamination, soils are continuously exposed to low and repeated deposits of PAHs. In these soils, PAHs can be found in different soil compartments, depending on the age and history of the contamination. A small fraction of these PAHs is usually present in the soil aqueous phase, and is considered as directly accessible to microorganisms for degradation. The bioaccessible fraction is available to cross cellular membranes if the organism is present and has access to it, even after a period of time or through spatial rearrangements, whereas the bioavailable fraction is that is freely available to cross the organism's cellular membrane at a given time (Riding et al., 2013). The PAHs recently adsorbed on soil constituents are also generally directly accessible to microorganisms, while PAHs in contact with soil constituents for a longer time become poorly bioaccessible. A last fraction of PAHs can be strongly bound to soils, by slow diffusion into soil micropores. The strong sorption/sequestration mechanisms of PAHs to organo-mineral complexes eventually creates non-extractable bound-residues (Ahangar, 2010), considered as not bioaccessible (Semple et al., 2004).

After adaptation and selection processes, the soil bacterial communities acquire the potential to biodegrade the bioaccessible PAH fraction (Haritash and Kaushik, 2009). Indeed, different bacterial strains isolated from various contaminated soils have been described as capable of degrading PAHs and of growing with PAHs as sole carbon source. These bacteria belong to different phyla or classes such as  $\alpha$ -,  $\beta$ - and  $\gamma$ -*Proteobacteria* (Haritash and Kaushik, 2009; Jeon et al., 2006; Kim and Jeon, 2009; Niepceon et al., 2013), *Bacteroidetes*, *Actinobacteria* (Timmis et al., 2010) or *Firmicutes* (Wu et al., 2008).

The bioaccessibility factor depends on the physico-chemical properties of PAH compounds and on the soil properties (clays and organic matter contents and nature) (Brandli et al., 2008; Jonker et al., 2005; Mechlińska et al., 2009; Xia et al., 2010). Besides that, the bioavailability factor depends on the physiological and catabolic potential of soil microorganisms (Crampon et al., 2014). The combination of these complex processes determines the fate of PAHs in soils (Semple et al., 2004; Semple et al., 2003). Enhancement of PAH bioaccessibility in contaminated soils through addition of surfactants has already been reported (Ahn et al., 2008; Paria, 2008; Zhou et al., 2007; Zhou and Zhu, 2008). Biosurfactants (*e.g.* surfactants naturally produced by (micro)organisms) are more environmental friendly and interesting due to their low critical micellar concentration (CMC) and their lower toxicity for soil microorganisms than synthetic ones (Bustamante et al., 2012). Rhamnolipids, produced by *Pseudomonas aeruginosa* (Bordas et al., 2005), have been described for their ability to remobilize sorbed PAHs and to potentially enhance their biodegradation in slurry batches or column experiments (Bordas et al., 2005; Congiu and Ortega-Calvo, 2014). However, the real impact of these rhamnolipids on PAHs accessibility after direct treatment of contaminated soils remains to be determined.

In this context, the aim of this study was to determine the parameters mostly governing phenanthrene (PHE) degradation rates in diffusely contaminated soils. PHE was chosen as a

model PAH representative of low molecular weight PAHs (2-3 fused aromatic rings). To address such an objective, two soil types (PPY and Pv) characterized by similar chronic PAH contamination, but different physico-chemical characteristics, were analyzed and compared. In each soil type, chemical and molecular biological methods were combined to investigate the fate of phenanthrene (PHE) by focusing on: (i) PHE accessibility, (ii) PHE degradation kinetics, and (iii) PHE-degrading bacterial diversity in relation to the amendment or not of rhamnolipids. PHE degradation was monitored using a microcosm approach during two months. In microcosm, PHE sorption isotherms with or without rhamnolipids were determined to estimate PHE accessibility. PHE degradation kinetics were determined using  $^{13}\text{C}$ -PHE and PHE-degrading bacterial populations were identified through DNA-Stable Isotope Probing (DNA-SIP) and 16S rRNA gene amplicon pyrosequencing.

## **Materials and Methods**

### **Soil sampling and soil analyses**

Soil samples were collected in the Seine river basin (France), which is a highly urbanized and industrialized location accounting for 40% of the French economic activity and including 33% of oil refining and 50% of river traffic. In spring 2014, approximately 15 kg of soil (0–15 cm depth) of two diffusely contaminated soil types (PPY and Pv) were sampled. PPY soil is an agricultural soil characterized by a permanent grassland and located near Yvetot (Seine-Maritime department, 49°36'40.6"N 0°44'10.7"E), at about 15 km from the Seine River. The PPY soil is classified as Luvisol according to the World Reference Base for soil resources (IUSS Working Group WRB 2006). The second soil type sampled (Pv) is located near Petiville (Seine Maritime department, 49°25'54.0"N 0°36'25.4"E) in the Seine river estuary, and is

partially covered by young willow. This soil, periodically flooded, is classified as fluvic Gleysol according to the World Reference Base for soil resources.

After sampling, field-moist soils were sieved through a 5 mm mesh and stored at room temperature for 8 days for acclimation. Aliquots were stored at -20°C for molecular characterization of the initial microbial communities in the two soil types before PAH spiking. The natural background concentrations of PAHs in both soils were lower than 1 mg.kg<sup>-1</sup> for the sum of the 16 PAHs classified as priority by the US-Environmental Protection Agency (USEPA) (Table 1).

Pv and PPY soil organic matter (OM) was analyzed on dried crushed soils by RockEval 6 pyrolysis (Vinci Technologies, France) and S2 peak deconvolution (Sebag et al., 2006) was obtained using peakfit software (SigmaPlot, Systat) and the “Bulk Rock” method, as described previously (Crampon et al., 2014) (Table 1). Nature of clays in soils was analyzed by X-Ray diffraction (XRD) (Philips PW3040 diffractometer, France), as described previously (Crampon et al., 2014).

Soil pH was determined according to the French NF ISO 10-390 standard. Granulometry was analyzed according to the NF X31-107 standard. CEC (Cationic Exchange Capacity) was obtained following the French NF ISO 31-130 standard by the cobaltihexamine method. The analyses of pH, texture and CEC were carried out at the LAS-INRA laboratory (Arras, France).

#### **Phenanthrene sorption isotherm determination**

Sorption isotherm experiments were performed for both soils as previously described (Groboillot et al., 2011), in the presence or not of rhamnolipids. Briefly, sorption reactors consisted in 4 g of soil sieved at 355 µm and mixed with 40 mL deionized water in glass tubes under magnetic agitation. Reactors were incubated at 25°C along the experiment and HgCl<sub>2</sub> (500mg.L<sup>-1</sup>) was added as a biocide at the beginning of the experiment to avoid bacterial



degradation, which could decrease PHE concentration in water after many hours. PHE was then introduced in the reactors with a known volume of PHE stock solution (100 mg.L<sup>-1</sup> in acetonitrile), to obtain a range of concentrations (from 0.1 to 1 mg.L<sup>-1</sup>), below its limit of solubility in water (*i.e.* 1.6 mg.L<sup>-1</sup>). The volume of acetonitrile injected (<1% of the total volume) was low enough to have no impact on PHE sorption, as previously studied (Groboillot et al., 2011). To evaluate the remobilization capacity of rhamnolipids, the biosurfactant (AGAE Technologies (USA), 90% purity) was added as a powder in the reactors 5 h after the introduction of PHE, to a final concentration of 0.55 g.L<sup>-1</sup> (concentration ten times higher than its CMC of 0.05 g.L<sup>-1</sup>) to ensure the formation of micelles in the aqueous phase. Five hours was chosen as the time to add the biosurfactant after PHE spiking because it was observed that a “quasi” plateau was reached for PHE sorption after this time (data not shown). Mixtures treated or not with rhamnolipids were stirred during 24 h at 25°C to be sure to reach thermodynamic equilibrium (time evaluated during preliminary tests) (Barnier et al., 2014). Thereafter, reactors were centrifuged at 7000 rpm during 20 min to recover the aqueous phase. The concentration of PHE at equilibrium in the aqueous phase after 24 h ( $C_{eq}$ ) was then measured after solid-phase extraction (SPE) and analysis using high-performance liquid chromatography coupled with fluorescence detector (HPLC-FLD) as developed, in the section below. As our systems were closed, PHE spiked at  $t_0$  was fully distributed between the water and the soil phases after 24h. So, the sorbed concentration ( $C_{ads}$ ) was deduced knowing the initial PHE concentration  $C_0$ , following the equation  $C_{ads} = C_0 - C_{eq}$ . The  $K_d$  partition coefficients were then obtained from the slope of the isotherms.

### **Phenanthrene dissipation kinetics**

Microcosms consisted of 4.5 g of dry equivalent fresh soil (90%) and 0.5 g of dry sterilized (autoclaved at 120°C for 20 min) soil (10%) spiked with <sup>12</sup>C-PHE (purity > 98%) or <sup>13</sup>C-PHE

166 (uniformly labeled  $^{13}\text{C}_{14}$ , 99 atom %; Sigma-Aldrich), mixed in a hermetically-closed 100 mL  
167 sterile glass flask. A PHE solution was prepared in acetone and spiked to the dry sterilized soils  
168 in order to obtain a final PHE concentration of  $3,000 \text{ mg.kg}^{-1}$  of soil. The dry sterilized soil was  
169 mixed under an extractor hood for 24 h to homogenize the PHE distribution and to evaporate  
170 the remaining acetone. After one week of stabilization at ambient temperature, the dry PHE  
171 contaminated and sterilized soil was mixed with the wet native soil (humidity of 24.1 and 31.7%  
172 wt of fresh soil for PPY and Pv soils, respectively) and gently homogenized to obtain final soil  
173 microcosms contaminated with PHE ( $300 \text{ mg.kg}^{-1}$  dw soil). This spiked PHE final  
174 concentration was 337-940 times higher than the sum of the 16 USEPA background PAHs  
175 (Table 1) and 2000-2857 times higher than the concentration of the native 2-3 aromatic ring  
176 PAHs. So we could reasonably consider that the main effects of PAHs on our studied systems  
177 were mainly due to the addition of PHE at  $300 \text{ mg.kg}^{-1}$ , and so the impact of co-metabolism  
178 was negligible in our study. It was decided to add the initial spiked dried sterilized soil (10%)  
179 to the native wet soil (90%) after contact time of only 7 days, in order to avoid PHE dissipation  
180 in the initial spiked soil. Soil water content of each microcosm was adjusted with distilled water  
181 to 70% of its water holding capacity. It corresponded to a total of 43.5 g and 49.1 g of water  
182 per 100 g soil for PPY and Pv soils, respectively. Rhamnolipids were added to distilled water  
183 to a final concentration of  $300 \text{ mg.kg}^{-1}$  dw soil into half of the microcosms. Rhamnolipids  
184 correspond to a mixture of mono and di-rhamnolipids (R90 product). So the concentration of  
185 the added biosurfactant was around  $0.61\text{-}0.69 \text{ g.L}^{-1}$  in the water fraction, which was near the  
186 concentration tested in sorption studies. The flasks were opened once a week in sterile  
187 conditions for 15 min to renew the atmosphere, and were incubated at  $20^{\circ}\text{C}$  in static mode in a  
188 dark room. Analyses of microcosms were carried out after 0, 8, 16 and 30 days of incubation  
189 for Pv soil and after 0, 20, 42 and 50 days for PPY soil, based on the different PHE degradation  
190 rates evaluated through a preliminary experiment (data not shown). These different sampling

times between the two soils were chosen in order to be able to compare the labeled DNA between the two soils (in terms of percentage of dissipated PHE), to potentially have the same quantity of  $^{13}\text{C}$  possibly incorporated into the degrading bacteria DNA. All the microcosms were performed in triplicates giving a total of 96 microcosms (2 soils, with and without rhamnolipids, 4 incubation times, 3 replicates and with  $^{12}\text{C}$  and  $^{13}\text{C}$  PHE). The experimental protocol is available in supplementary materials (Fig. S1).

### **Phenanthrene extraction and measurements**

All organic solvents were purchased from Fisher Scientific (Illkirch, France) and were of HPLC grade. PHE extractions from soils were performed using the microwave accelerated extraction (MAE; Mars X, CEM Corporation, USA) as described previously (Portet-Koltalo et al., 2008). For PHE degradation kinetics, two aliquots of 1.5–2 g of wet soil were taken from each of the three microcosms (in triplicates) at each time point. Then at each studied time, 6 sub-samples of a soil without biosurfactant and 6 sub-samples of the same soil amended with biosurfactant were dried 24 h at 35°C (to avoid the semi volatile PAH loss) and crushed separately. One g of each of the 6 sub-samples was extracted simultaneously using MAE technique in 20 mL of an acetone/toluene 50/50 (v/v) mixture at 130°C for 30 min, at 1,200 W. So for each studied time, extraction yields were obtained from a mean of 6 values. To determine the PAHs concentration in native soils, 10 g of crushed dried soil were processed using MAE technique, with 40 mL of the acetone/toluene 50/50 mixture, in the same conditions as described above. After filtration (0.45  $\mu\text{m}$  Teflon filters; Fisher Scientific, France), extracts were evaporated to reach a volume of 1.5 mL. An internal standard (10  $\mu\text{L}$  of perdeuterated PHE at 100  $\text{mg}\cdot\text{L}^{-1}$ ) was added to 990  $\mu\text{L}$  of each extract before GC-MS analyses. Quantification analyses were performed using a 6850 gas chromatograph from Agilent (USA) fitted with a mass spectrometer (MS) detector (5975C, Agilent) and using a DB5-MS capillary column (60 m $\times$ 0.25 mm inner diameter, film

thickness 0.25  $\mu\text{m}$ ; J&W Scientific, USA) according to Crampon *et al.* (Crampon et al., 2014). The mass detector (electron impact ionization 70 eV) operated in the selected ion monitoring (SIM) mode for better sensitivity. The detection and quantification thresholds, calculated respectively as 3 and 10 times the standard deviation of the blank sample noise, were 1.5 and 5  $\mu\text{g.L}^{-1}$ , respectively. The mean recovery of PHE after MAE-GC-MS extraction and analysis was  $90.0 \pm 7.0\%$  ( $n=5$ ).

PHE extractions from aqueous media were done by SPE using Strata-X cartridges (Phenomenex, Le Pecq, France). After cartridge conditioning and sample percolation (Portet-Koltalo et al., 2013), PHE was eluted with 5 mL methylene chloride, and was evaporated under nitrogen flow and recovered in 1 mL acetonitrile before HPLC analysis. PHE quantification was performed using a HPLC (Beckman Coulter, USA) coupled to a FLD (Prostar 363, Varian, USA) and using a C<sub>18</sub> Envirosep PP column (150 $\times$ 4.6 mm, dp = 5  $\mu\text{m}$ ) (Phenomenex, Le Pecq, France) according to Portet-Koltalo *et al.* (Portet-Koltalo et al., 2013). Detection and quantification limits, calculated respectively as 3 and 10 times the standard deviation of the sample noise, were evaluated as 0.4 and 1.3  $\mu\text{g.L}^{-1}$ , respectively. The mean recovery of PHE after SPE-HPLC-FLD extraction and analysis was  $94.2 \pm 1.2\%$  ( $n=5$ ). Taking into account the concentration factor using SPE, the limit of quantification of PHE in the aqueous medium was 33  $\text{ng.L}^{-1}$ . So PHE was preferentially measured in the aqueous medium after sorption experiments because SPE-HPLC-FLD allowed us to reach lowest concentrations than using GC-MS and because of the highest accuracy of the method.

### **<sup>13</sup>C labeled DNA isolation**

For DNA-based diversity analyses, samples were selected for further analysis according to the level of <sup>13</sup>C PHE degradation. Initial soils collected before spiking (triplicates) and soil samples collected during the experiment at 20% (DT20) and 90% of PHE degradation (DT90) were used

for soil genomic DNA extractions using the PowerSoil DNA isolation kit (Mo-Bio Laboratories) according to the manufacturer recommendations. Extracted DNA was quantified using NanoDrop 2000 (Thermo Scientific) and samples were stored at -20°C until analysis. Heavy ( $^{13}\text{C}$  labeled) and light DNA ( $^{12}\text{C}$ ) were separated according to the protocol described by Neufeld et al. (Neufeld et al., 2007). Gradients were prepared using a CsCl solution (purity  $\geq 99\%$ ; Sigma-Aldrich) with an initial density of ca 1.85. Gradient buffer and DNA sample were then added (for a final DNA quantity in the centrifuge tube of 5 to 10  $\mu\text{g}$ ) to obtain a final density of 1.725. Ultracentrifugations were carried out on a Beckman centrifuge (Optima L-80 XP) with a vertical rotor Vti 65.2 (Beckman), at 42,400 rpm for 40 h at 20°C, with maximal acceleration and no deceleration to avoid gradient disturbance. After centrifugation, tubes were fractionated in 12 density fractions that were weighed to check the gradient formation. DNA contained in each fraction was then precipitated using 2 volumes of polyethylene glycol (~0.8-1 mL), 1  $\mu\text{L}$  glycogen (20  $\mu\text{g}$ ), centrifuged (13000g for 30 min) and washed with 70% purified ethanol (500  $\mu\text{L}$ ), and finally recovered in 30  $\mu\text{L}$  TE buffer (10mM Tris-HCl and 1mM EDTA in deionized water) for further use.

On DNA recovered from each DNA fraction, the abundance of 16S rRNA genes was quantified using real-time quantitative PCR to determine the distribution of heavy and light DNA. Real-time PCR assays were performed in a 20  $\mu\text{L}$  final volume, as previously described (Cebon et al., 2008) using: 2X iQ SYBR Green Supermix (Bio-Rad), 0.4  $\mu\text{mol.L}^{-1}$  of each 16S rDNA universal primers 968F and 1401R (Nubel et al., 1996), BSA (0.06%), 0.2  $\mu\text{L}$  DMSO (dimethylsulfoxide), 0.08  $\mu\text{L}$  t4gp32 protein (MP-Biomedicals) and 1  $\mu\text{L}$  of DNA. The amplification was carried out with 40 cycles of four steps: 20 sec at 95°C, 20 sec at 56°C, and 30 sec at 72°C and 5 sec at 82°C (Cebon et al., 2008). On similar DNA fractions, fungal 18S rRNA genes were also quantified using real-time quantitative PCR (Thion et al., 2013) with the previously described primer pairs Fung5f-FF390r (Lueders et al., 2004), but no enrichment of

fungi was detected in heavy DNA fractions (Fig. S3) meaning that fungi were not significantly involved into PHE degradation because they did not incorporate  $^{13}\text{C}$  in their DNA.

### **Identification of phenanthrene-degrading bacteria**

Altogether, DNA from the 16 microcosm treatments (2 soil types, with and without rhamnolipids, at DT90, light and heavy DNA fractions n°3 and 7, from  $^{12}\text{C}$  and  $^{13}\text{C}$  incubations) and the two initial soils, in triplicate, were chosen for sequencing, resulting in 54 samples. Libraries of 16S rRNA gene amplicons were prepared for 454 pyrosequencing after amplification using universal primers 27F (5' – AGA GTT TGA TCC TGG CTC AG - 3') and 533R (5' – TTA CCG CGG CTG CTG GCA C - 3'), extended at the 5' end by 10 bp multiplex identifiers (MIDs, Roche). Amplifications were done with GoTaq G2 Green Master Mix (Promega). The amplification step was carried out with 25 and 30 cycles (30 sec at 95°C, 30 sec at 60°C, 30 sec at 72°C) for microcosm and initial soil samples, respectively. After quality checking on agarose gel electrophoresis and quantification using NanoDrop, three independent PCR products per sample were pooled together to minimize PCR bias on sequencing, and an equimolar mixture was then prepared. Pyrosequencing was carried out using a GS FLX 454 (Beckman Coulter Genomics, Danvers, MA, USA) and produced 316.5 megabases corresponding to a total of 716,250 partial 16S rRNA gene sequences having a median size of 490 bp. Sequence analyses, i.e. reducing sequencing error, removing chimeras, grouping into OTUs, subsampling to 1000 sequences, SILVA database alignment, and classifying using the RDP reference files were performed using MOTHUR (version 1.36), as recommended by its author (Schloss et al., 2011). Following the trimming, we obtained 250 bp sequences from the forward primer (corresponding to the V1 and V2 16S rDNA regions). The alpha diversity of the samples was estimated using the Chao1 richness estimator and the nonparametric Shannon diversity index considering a same number of sequences for each sample (i.e. 1000 sequences).

291

## 292 **Statistical analyses**

293 Statistical analyses (ANOVA, Student's t tests, Kruskal Wallis tests, Kolmogorov Smirnov  
294 tests), data analysis and Factorial Correspondence Analysis (FCA) on OTUs table, by using  
295 Bray-Curtis dissimilarity distance, were performed using XLStat software (version 2.01.16906,  
296 Addinsoft, Paris). Chao-1 and nonparametric Shannon diversity indexes were calculated on  
297 OTUs table, using MOTHUR (version 1.36). ADONIS tests were performed by using Bray-  
298 Curtis dissimilarity distance (R software, version 3.1.2, VEGAN package).

299

## 300 **Nucleotide sequence accession numbers**

301 All 16S rRNA gene sequences have been submitted to the NCBI Short Read Archive under the  
302 Bioproject accession number PRJNA312913.

303

304

## 305 **Results**

306

### 307 **Physico-chemical properties of soils**

308 Soil analyses show that the two soils were very different (Table 1), with the PPY soil  
309 characterized by acidic and moderate nutrient conditions and the Pv soil characterized by  
310 alkaline and higher nutrient availability. For both soils, the major grain size fraction was coarse  
311 silt, representing more than 40% dry weight of the mineral fraction (Table 1). Although both  
312 particle size distributions were close, by comparison with the Shepard diagram (Shepard, 1954),  
313 the PPY and the Pv soils correspond to sandy silt clayey and silt soils, respectively. Concerning  
314 the clayey fraction, total clay content was significantly higher ( $P < 0.05$ ; Student's t test) in the  
315 Pv soil (20.2%) compared to the PPY (15.5%), but the nature of clays was different (Table 1).

For PPY soil, chlorites, illites and kaolinites (clays with the lower CEC, 10-40, 20-30 and 3-15 meq.g<sup>-1</sup>, respectively; Morel, 1996) were dominant whereas smectites (80 meq.g<sup>-1</sup> < CEC < 150 meq.g<sup>-1</sup>) were dominant in the Pv soil. Consequently, the CEC of the Pv soil was higher than that of the PPY soil, especially due to clay types.

Concerning total organic carbon content (TOC), the PPY was the richest soil (Table 1). Residual carbon (RC) was dominant in both soils compared to pyrolysable carbon (PC), showing that OM tended to be mature and resistant to pyrolysis, and could potentially strongly sorb PHE. The main OM fraction in PPY soil corresponded to biopolymers whereas mature geomacromolecules, that are higher PHE sorbents, were dominant in the Pv soil.

#### **Phenanthrene sorption and bioaccessibility**

PHE sorption isotherms (Fig. 1) were similar for the PPY and the Pv soils ( $P > 0.05$ , Kolmogorov-Smirnov Test). In the low PHE concentration range, they can both be described by a linear partitioning model ( $R^2=0.965$  and  $0.981$  for PPY and Pv soils, respectively) (Crampon et al., 2016). So  $K_d$  values were calculated from the slopes of the linear part of the sorption isotherms. The  $K_d$  partition coefficients were  $718.7 \text{ L.kg}^{-1}$  and  $695.5 \text{ L.kg}^{-1}$  for the PPY and the Pv soils, respectively. Water/organic carbon partition coefficients  $K_{oc}$  could be calculated from the  $K_d$  values and the total organic carbon fraction ( $f_{oc}$ ), using the equation:  $K_{oc}=K_d/f_{oc}$ . Log  $K_{oc}$  values were  $4.58$  and  $4.73$  for PHE in the PPY and the Pv soils, respectively. Increases of PHE desorption and potentially of PHE bioaccessibility were observed for both soils when adding rhamnolipids in the aqueous medium (Fig. 1). Indeed,  $K_d$  values obtained in the presence of rhamnolipids were significantly lower (Student's  $t$  test,  $P < 0.05$ ) than those obtained without biosurfactant:  $206.2 \text{ L.kg}^{-1}$  and  $318.7 \text{ L.kg}^{-1}$  for the PPY and the Pv soils, respectively ( $R^2=0.959$  and  $0.929$  for PPY and Pv soils, respectively).



## **Phenanthrene dissipation kinetics**

Although no marked lag phase was observed in our experimental conditions, PHE dissipation was slow at the beginning of the experiments. Only 20% of PHE was degraded during the first 8 and 20 days for the Pv and the PPY soils, respectively (Fig. 2). Then, PHE dissipation was faster in Pv soil compared to the PPY soil, as ca 90% of PHE was dissipated in 16 days in the Pv soil, while 42 days were necessary in the PPY soil. The same trends were observed in a previous experiment (Crampon et al., 2014), aiming, inter alia, to follow the degradation of 7 selected PAHs in different soils with the same sampling terms.

Kinetics of PHE dissipation in the presence of rhamnolipids did not show any significant impact of surfactant addition compared to the controls ( $P > 0.05$ , Student's t test; Fig. 2).

## **SIP experiment and $^{13}\text{C}$ labeled DNA isolation**

Total community ( $^{12}\text{C}$ ) and active  $^{13}\text{C}$ -labeled PHE-degrading bacterial populations were studied at two incubation times corresponding to DT20 (time where 20% PHE is dissipated) and DT90 (time where 90% PHE is dissipated) for both soils: it corresponded to 8 and 16 days of incubation for the Pv soil and 20 and 42 days of incubation for the PPY soil (Fig. 2). For controls, DNA was analyzed from both  $^{12}\text{C}$  and  $^{13}\text{C}$  microcosms.

Quantitative PCR revealed a higher number of 16S rRNA gene copies in the lower densities ( $<1.70$ ) of the CsCl gradient, corresponding to light DNA (that should contain only  $^{12}\text{C}$  DNA) (Fig. 3). For further analyses the fraction n°3 ( $d = 1.689$ ) was selected. Heavy DNA (that should contain all  $^{13}\text{C}$  DNA) was recovered in the fractions with density ranging from 1.705 to 1.722 (Fig. 3). Higher numbers of 16S rRNA gene copies in these heavy fractions collected from the  $^{13}\text{C}$  microcosms corresponded to the active  $^{13}\text{C}$ -labeled PHE-degrading bacteria. For further analyses the fraction n°7 ( $d = 1.711$ ) was selected. At DT20, the difference between the number of 16S rRNA gene copies in the fraction n°7 from  $^{12}\text{C}$  and  $^{13}\text{C}$  microcosms was only significant

for Pv soil microcosms ( $P < 0.05$ , ANOVA test; Fig. S2). In PPY soil microcosms, no  $^{13}\text{C}$  labeled DNA was detected at DT20. At DT90, significant differences ( $P < 0.05$ , ANOVA test) were observed between  $^{12}\text{C}$  and  $^{13}\text{C}$  microcosms, for both soils (Fig. 3). The amounts of 16S rRNA gene copies in the heavy fraction of DNA were about 10 and 50 times higher for the  $^{13}\text{C}$  microcosms than for the  $^{12}\text{C}$  microcosms for the PPY and the Pv soils, respectively. Therefore, samples at DT90 were chosen for sequencing and analyzing the bacterial diversity. No significant difference was observed in the level of  $^{13}\text{C}$  incorporation between microcosms with and without rhamnolipids (Fig. 3).

No significant difference in the number of fungal 18S rRNA gene copies was observed between  $^{12}\text{C}$  and  $^{13}\text{C}$  microcosms (Fig. S3), meaning that  $^{13}\text{C}$  from PHE was not used as a growth carbon source by fungi.

#### **Characterization of bacterial communities and identification of PHE-degrading bacteria**

454-pyrosequencing was carried out on 54 soil samples, corresponding to the soils before spiking and the fractions from the DNA-SIP experiment. After de-noising, a total of 211,519 16S rRNA gene sequences were obtained, with an average number of 3,917 sequences per sample. Even if the rarefaction curves based on operational taxonomic units (OTUs; at 97% identity) did not reach a plateau for the soils before spiking, the diversity corresponding to the labeled  $^{13}\text{C}$  DNA (*i.e.* PHE-degrading populations) seemed to be well sampled (Fig. S4). Although species richness given by Chao1 index was slightly higher for the Pv soil comparatively to the PPY soil, no significant differences ( $P > 0.05$ , Kruskal-Wallis test) of alpha diversity were observed between the two native soils (Table 2). Concerning the heavy fractions from the  $^{13}\text{C}$  microcosms (fractions PPY/Pv\_13C\_Rh+/-\_H in Table 2), the observed richness were lower than in the initial soil types or in controls, meaning that PHE-degrading bacteria represent only a small fraction of the total community in both soils. This finding is more

pronounced for the Pv soil, meaning that a significantly smaller number of OTUs ( $P < 0.05$ , Kruskal-Wallis test) was involved in PHE degradation.

Factorial correspondence analysis (Fig. 4) was performed using the 100 most represented OTUs (in each soil, separately) to highlight similarities and dissimilarities between replicates, sample conditions (soil before spiking,  $^{12}\text{C}$  and  $^{13}\text{C}$  microcosms, or with and without rhamnolipids) and DNA fractions (light and heavy). Whatever the soil type, the bacterial diversity retrieved from the heavy DNA fractions of  $^{13}\text{C}$  microcosms was significantly different from the diversity found in the heavy DNA fractions of  $^{12}\text{C}$  control microcosms ( $P < 0.01$ ; ADONIS test). A specific analysis of the effect of rhamnolipids addition revealed no differences in each soil type between treatments with and without rhamnolipids, in both light or heavy fractions ( $P > 0.05$ ; ADONIS test). Notably, no differences appeared between the diversity in light DNA fractions from both  $^{12}\text{C}$  and  $^{13}\text{C}$  microcosms and the diversity of soils before spiking.

The analysis of the sequence distribution in the OTUs represented in the heavy DNA fractions from the  $^{13}\text{C}$  labeled microcosms and having a significantly higher abundance than in the  $^{12}\text{C}$  control microcosms allowed us to identify the main  $^{13}\text{C}$ -labeled PHE degrading bacteria (Table 3). Notably, this analysis revealed that the main PHE degraders identified were poorly represented in the initial soils (from  $<0.01\%$  to  $0.29\%$  of the total sequences, Table 3), as in the light and heavy fractions of the  $^{12}\text{C}$  microcosms (from  $<0.01\%$  to  $3.0\%$  of the total sequences, data not shown).

The six OTUs, representing the active PHE degraders in the Pv soil, were identified as members of the *Betaproteobacteria* class, with a high proportion of the *Rhodocyclaceae* family and some *Burkholderiales* (including the *Polaromonas* genus). Concerning the PPY soil, members of a higher number of OTUs were involved in PHE degradation. The 13 OTUs representing the active PHE degraders belonged to *Alpha*-, *Beta*-, *Gamma-Proteobacteria* and *Actinobacteria*. Notably the OTUs Pv\_PHE6 and PPY\_PHE6 were both related to the *Polaromonas* genus

(Table 3), and appeared phylogenetically close (Fig. S5). Moreover, OTUs Pv\_PHE4 and PPY\_PHE12 corresponded to the same OTU affiliated to the *Rhodoferrax* genus (Table 3 and Fig. S5).

The relative abundance of some PHE degraders increased or decreased depending on the presence or absence of rhamnolipids, but only in the case of some of the low abundance PHE degraders (Table 3). Notably, the proportion of the two OTUs related to the *Polaromonas* genus was significantly higher in the presence of rhamnolipids in both soils. By contrast, when rhamnolipids were added, the relative proportion of sequences related to the *Rhodoferrax* genus was significantly higher for the PPY soil ( $P < 0.05$ ), whereas it tended to be lower for the Pv soil ( $P > 0.05$ ). Altogether, our results revealed that direct addition of rhamnolipids to soil has a limited effect on the active PHE degrading bacteria.

## Discussion

PHE accessibility was assessed by studying different physico-chemical properties of soil, and by characterizing PHE sorption isotherms with or without rhamnolipids. Overall, the two soils considered were characterized by a similar OM content, but a high variation in their clay contents and natures. While total organic carbon content was slightly higher in the PPY soil, OM was dominated by mature geomacromolecules in the Pv soil (Table 1). This fraction of OM combines constituents naturally stable or stabilized by physico-chemical processes, residues of fires and some anthropogenic pollutants like soot, *i.e.* black carbon (Sebag et al, 2006). The higher content of this more mature, carbonaceous and crystallized fraction in Pv soil make it probably more prone to a stronger PHE sorption than the PPY soil. The characterization of OM maturation state in soils is important because binding energies and sorption properties towards PAHs increase over time (Ran et al., 2007). Indeed, PAHs sorption is generally lower and more reversible on fresh OM compared to aged OM because with ageing, the polar mixtures of

441 biomolecules transforms into a more hydrophobic form with higher sorptive properties for  
442 lipophilic organic compounds (Mechlińska et al., 2009). Moreover, both total clay contents and  
443 natures varied significantly between the two soil types considered. Chlorites, illites and  
444 kaolinites were dominant in the PPY soil, whereas smectite was dominant in the Pv soil.  
445 Although clay is generally described as a parameter affecting less PAH sorption to soil particles  
446 compared to organic matter, it has to be considered here because of the potential impact on  
447 rhamnolipids (Delle Site, 2001; Hwang and Cutright, 2002). Indeed, it has been reported that  
448 sorption of rhamnolipids to soil is more strongly influenced by clay contents and nature than  
449 by OM : at low biosurfactant concentrations, that are relevant for enhancing organic  
450 contaminant biodegradation, clays and particularly kaolinite are the strongest sorbents for  
451 rhamnolipids (Ochoa-Loza et al., 2007). So clays could influence observed results on PHE  
452 remobilization.

453 Interestingly, despite these different physico-chemical properties and biodegradation kinetics,  
454 PHE sorption was found to be quite similar in the two soil types. These results are consistent  
455 with a previous study, where we showed that the bioaccessible fraction of PHE, estimated using  
456 a mild extraction method (extraction with hydroxypropyl- $\beta$ -cyclodextrin) was close between  
457 these two soil types (Crampon et al., 2014). Furthermore, rhamnolipids addition increased PHE  
458 desorption in both soil types, but with a greater effect in the PPY soil. Although no consensus  
459 exists on the real positive or negative effect of rhamnolipids on PAH removal, several studies  
460 reported a strong impact of soil organic matter content and contamination ageing (Congiu and  
461 Ortega-Calvo, 2014; Szulc et al., 2014). Thus, the lowest PHE desorption from the Pv soil by  
462 rhamnolipids addition could be explained by higher contents of mature OM, smectites and  
463 kaolinite in the Pv soil. These soil constituents may enhance the formation of clay-humic  
464 complexes and thus contribute to a strong adsorption of PHE on these complexes. Moreover,  
465 these clay minerals increase the surface available for PAH as well as rhamnolipids sorption.

The presence of higher contents of kaolinite in Pv soil (Table 1) can explain the strongest sorption of the biosurfactant on Pv soil, compared to PPY soil. Consequently, there may be less micelles in the soil solution to remobilize PHE and more additional sorption sites through rhamnolipids sorbed on soil aggregates (Ochoa-Loza et al., 2007; Zhou and Zhu, 2007).

Although rhamnolipids significantly increased PHE desorption in the two soil types tested, addition of a biosurfactant had a limited impact on the PAH biodegradation efficiency (Fig. 2). While some studies report that surfactants may increase PAH biodegradation rates, others have shown that the effect of their application may be negligible (Laha and Luthy, 1991; Rouse et al., 1994; Tiehm, 1994; Tsomides et al., 1995). Indeed, Leonardi *et al.* (Leonardi et al., 2007) showed that when a soil is characterized by a low non-bioaccessible PAH fraction, addition of surfactant has a limited impact on the PAH biodegradation efficiency. Moreover, the effect of the surfactant could also depend on the identity of PAH-degrading bacteria involved (Allen et al., 1999). At last, we can also suppose that if rhamnolipids can partially desorb PAHs from soils, they can also desorb them from the surface of bacteria, which can lead to the decrease of their bioavailability while they increase their bioaccessibility.

More generally, other factors might explain the difference in degradation rate observed between the two soils. Indeed, the optimum pH range can be very variable depending on the type of soil and the PAH-degrading bacteria (Simarro et al., 2011). It could not be excluded that the basic Pv soil allowed a greater activity of the PAH-degrading bacteria than the acidic PPY soil. Moreover, difference in initial concentrations of nitrogen and carbon between the two soils may also affect the activity of the PAH-degrading bacteria (Teng et al., 2010). Temperature and toxicity of PAHs could also impact the biodegradation rate (Coover and Sims, 1987; Martin et al., 2012). Indeed, native high molecular weight PAHs (4-6 ring PAHs), which can be considered as the most toxic PAHs, were 2.5 times more concentrated in Pv soil than in PPY soil. However, it must be noted that they represent only 0.09-0.22% of the total amount of PAHs

491 after PHE spiking, which is quite negligible. Interestingly, all the factors having a potential  
492 impact on the PHE degradation kinetics have in common to influence the physiology of the  
493 PAH-degrading bacteria.

494 In a second phase, we therefore studied the diversity of the PAH-degrading bacteria in order (i)  
495 to compare the PAH-degrading populations between these two soils and (ii) to evaluate the  
496 impact of a bioavailability variation on these degrading populations. The combination of <sup>13</sup>C  
497 labeled PHE and 16S rRNA gene amplicon pyrosequencing enabled us to identify the PHE-  
498 degrading bacterial communities in these two contrasted soil types.

499 Most of the PHE degraders identified in the two soil types belonged to the *Proteobacteria*  
500 phylum. Some members of the *Alphaproteobacteria* (Demaneche et al., 2004),  
501 *Betaproteobacteria* (Goyal and Zylstra, 1997; Laurie and Lloyd-Jones, 1999; Seo et al., 2009)  
502 and *Gammaproteobacteria* (Balashova et al., 2001; Ensley et al., 1982; Kiyohara et al., 1994;  
503 Resnick and Gibson, 1996) classes possess specific PAH dioxygenases involved in the catalysis  
504 of the initial aromatic ring oxidation (Cebron et al., 2008; Habe and Omori, 2003). Some OTUs  
505 belonging to these classes were found among PHE degraders in the PPY soil. In addition, one  
506 OTU of the PHE degraders belongs to the *Mycobacterium* genera (*Actinobacteria*) that has  
507 already been described as a good PAH degrader in soils (Timmis et al., 2010). Altogether, these  
508 PHE degraders were poorly represented in both initial soil bacterial communities, indicating  
509 that when soil PAH are diffusely concentrated, the PHE-degraders belong to the rare biosphere,  
510 but could rapidly become active when available PAHs enter the system.

511 The Pv soil presented a lower diversity of PHE degraders than the PPY soil, all belonging to  
512 the *Betaproteobacteria*. It was further shown that the abundance of *Betaproteobacteria*  
513 increased after three months of incubation in microcosms spiked with PHE, pyrene or a mix of  
514 seven PAHs, members of this bacterial class being largely described as capable of degrading  
515 PAHs (Jeon et al., 2003; Martin et al., 2012; Niepceron et al., 2013; Núñez et al., 2012;

Singleton et al., 2006). Among the *Betaproteobacteria* class, sequences related to members of the *Rhodocyclaceae* family were the most represented in the  $^{13}\text{C}$  fraction (more than 80% of the sequences), suggesting a major role in PHE degradation in the Pv soil. Interestingly, representatives of these taxa have also been described as able to degrade PAHs and to possess specific PAH RHD enzymes (Chemerys et al., 2014; Martin et al., 2012; Regonne et al., 2013; Singleton et al., 2006; Uhlik et al., 2012), and a cultivable member of this *Rhodocyclaceae* family that degrades PAHs has been recently sequenced (Singleton et al., 2015). The exceptional PAH degradation capacity of *Rhodocyclaceae* in various soils suggests classifying them as very efficient PHE degraders. Since previous representatives of this taxon have already been shown to efficiently degrade PAHs (Singleton et al., 2015), the *Rhodocyclaceae* family may be used as a bioindicator of soils with a high PHE degradation capacity. Moreover, cultivation-dependent analysis combined to genome sequencing should permit to further understand the functional role of the *Rhodocyclaceae* family in PAH degradation. Interestingly, the diversity and the structure of the main PHE degraders seem only slightly impacted by rhamnolipids addition and no impact on the total bacterial community was detected. Such results suggest that the rhamnolipid amendment used in biostimulation processes is probably not used as carbon source, and may be a good alternative in term of toxicity for the soil bacterial communities compared to synthetic surfactants. However, its use for soils where bioaccessibility is not a limiting degradation parameter does not seem to be an advantage even if it improved PHE desorption in both soils.

## Conclusion

By using chemical and microbiological methods, we demonstrated that the PHE bioaccessibility has a low impact on its degradation. Indeed, the difference in biodegradation rates we found between the two soils considered cannot be explained by a difference in



bioaccessibility. Moreover, although the amendment of biosurfactants (*i.e.* rhamnolipids) did change the PHE sorption/desorption kinetics in the two soils tested, no significant impact on PHE degradation rates was detected. In addition, we showed that variation in biodegradation rate between these diffusely contaminated soils could be more explained by the differences in PHE-degrading populations, which presence and activities might be driven by physico-chemical properties of the soils. Finally, we showed, for the first time to our knowledge, that rhamnolipids present a low toxicity for soil microorganisms as they do not modify the global bacterial diversity as well as the activity of the phenanthrene-degraders. In the future, it could be interesting to investigate the degradation kinetics of a mixture of PAHs (e.g. the 16 priority PAHs) with or without rhamnolipids and to identify the bacteria more particularly involved in the degradation of heavy high molecular weight PAHs.

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## **Supporting information**

Five Figures (Figure S1, S2, S3, S4 and S5) are included in the supporting information.

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771 TABLE 1: Physico-chemical properties of the two studied soils (Pv and PPY).

Method	Data	PPY	Pv
<b>Granulometry without decarbonatation (% dry weight)</b>	<i>Clays (&lt;2<math>\mu</math>m)</i>	15.5	20.2
	<i>Fine silts (2 &lt; D &lt; 20 <math>\mu</math>m)</i>	23.4	26.0
	<i>Coarse silts (20 &lt; D &lt; 50 <math>\mu</math>m)</i>	42.3	44.2
	<i>Fine sands (50 &lt; D &lt; 200 <math>\mu</math>m)</i>	17.8	9.5
	<i>Coarse sands (200 &lt; D &lt; 2000 <math>\mu</math>m)</i>	1.0	0.1
<b>XRD (g.kg<sup>-1</sup>)</b>	<i>Chlorite</i>	36.5	0.0
	<i>Illite / Chlorite</i>	11.2	0.0
	<i>Illite</i>	53.7	44.8
	<i>Kaolinite</i>	53.6	66.8
	<i>Smectite</i>	0.0	90.4
<b>RockEval6 Pyrolysis &amp; S2 peak deconvolution <sup>a</sup> (g.kg<sup>-1</sup>)</b>	<i>Total Organic Carbon (TOC)</i>	18.7	13.1
	<i>Pyrolysable carbon (PC)</i>	4.2	2.6
	<i>Residual carbon (RC)</i>	14.5	10.6
	<i>Biomacromolecules</i>	8.2	4.5
	<i>Immature geomacromolecules</i>	5.4	2.4
	<i>Mature geomacromolecules</i>	5.1	6.2
<i>pH (H<sub>2</sub>O)</i>		5.7	8.2
<i>Cationic Exchange Capacity CEC (cmol.kg<sup>-1</sup>)</i>		8.7	14.2
□ 16 PAHs (US-EPA)		0.320 ± 0.040	0.890 ± 0.290
<b>PAHs (mg.kg<sup>-1</sup>) (n=6)</b>	<i>BaP</i>	0.019 ± 0.006	0.046 ± 0.022
	<i>PHE</i>	0.035 ± 0.012	0.013 ± 0.001

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773 <sup>a</sup> The deconvolution of S2 peak, which can be related to soil OM (cracking between 200 to

774 650 °C), allowed a precise determination of the maturation of OM in soils.

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TABLE 2: Alpha diversity indexes of the two studied soils (Pv and PPY) based on OTUs formed at 97% similarity.

<i>Samples<sup>b</sup></i>	0.03 <sup>c</sup>								
	OTU			Chao1			H'		
Pv <sup>d, x</sup>	446	±	105	1194	±	139	6.19	±	0.33
Pv_12C_Rh+_L <sup>x</sup>	417	±	64	1031	±	124	5.75	±	0.42
Pv_12C_Rh+_H <sup>x</sup>	376	±	59	840	±	114	5.34	±	0.59
Pv_12C_Rh-_L <sup>x</sup>	411	±	135	965	±	356	5.75	±	0.67
Pv_12C_Rh-_H <sup>x</sup>	423	±	148	1000	±	348	5.76	±	0.82
Pv_13C_Rh+_L <sup>x</sup>	417	±	12	1028	±	48	5.78	±	0.04
Pv_13C_Rh+_H <sup>y</sup>	54	±	1	195	±	16	1.72	±	0.06
Pv_13C_Rh-_L <sup>x</sup>	452	±	6	1143	±	59	5.94	±	0.04
Pv_13C_Rh-_H <sup>y</sup>	40	±	3	137	±	28	1.62	±	0.19
PPY <sup>d, x</sup>	413	±	20	880	±	61	5.82	±	0.09
PPY_12C_Rh+_L <sup>x</sup>	310	±	61	632	±	123	5.17	±	0.57
PPY_12C_Rh+_H <sup>x</sup>	334	±	107	668	±	220	5.25	±	0.70
PPY_12C_Rh-_L <sup>x</sup>	347	±	31	741	±	82	5.45	±	0.19
PPY_12C_Rh-_H	319	±	NA <sup>e</sup>	597	±	NA <sup>e</sup>	5.04	±	NA <sup>e</sup>
PPY_13C_Rh+_L <sup>x</sup>	347	±	4	738	±	23	5.47	±	0.02
PPY_13C_Rh+_H <sup>z</sup>	171	±	59	406	±	136	3.65	±	0.92
PPY_13C_Rh-_L <sup>x</sup>	335	±	61	710	±	122	5.34	±	0.47
PPY_13C_Rh-_H <sup>z</sup>	241	±	67	609	±	189	4.22	±	0.56

<sup>a</sup> x, y and z represent the different groups characterized statistically (p<0.05, Kruskal Wallis test)

<sup>b</sup> Rh<sup>±</sup>: with or without rhamnolipid addition; L/H: light or heavy DNA fractions.

<sup>c</sup> 0.03 is the OTU cutoff in distance units, OTU means the number of OTUs observed, Chao1 means the Chao1 estimated minimum number of OTUs and H' means the nonparametric Shannon diversity index.

<sup>d</sup> Soil before spiking

<sup>e</sup> Not Applicable, for samples with not enough reads for calculation of standard deviation with triplicates.

785 TABLE 3: Active <sup>13</sup>C-labeled PHE degraders in the two studied soils (Pv and PPY). OTUs  
786 were considered as PHE-degraders when a significantly higher proportion was recovered in  
787 heavy DNA fractions from <sup>13</sup>C microcosms than from <sup>12</sup>C microcosms.

Soil	OTU	Taxonomic identification		Relative abundance of this OTU found (%)		
		Phylum / Class	Best resolution <sup>a</sup>	With Rhamnolipids	Without Rhamnolipids	Native soil
Pv	<i>Pv_PHE1</i>	Betaproteobacteria	Rhodocyclaceae family	52.48 ± 2.91	50.06 ± 11.22	0.01 ± 0.01
	<i>Pv_PHE2</i>	Betaproteobacteria	Rhodocyclaceae family	19.77 ± 1.44	17.95 ± 3.9	<0.01
	<i>Pv_PHE3</i>	Betaproteobacteria	Rhodocyclaceae family	15.69 ± 2.02	13.69 ± 2.42	<0.01
	<i>Pv_PHE4<sup>b</sup></i>	Betaproteobacteria	<i>Rhodoferax</i>	3.16 ± 2.9	12.02 ± 8.28	0.12 ± 0.05
	<i>Pv_PHE5</i>	Betaproteobacteria	<i>Hydrogenophaga</i>	0.76 ± 0.08	1.11 ± 1.3	0.03 ± 0.04
	<i>Pv_PHE6<sup>c</sup></i>	Betaproteobacteria	<i>Polaromonas</i>	0.89 ± 0.44	0.15 ± 0.13	0.02 ± 0.03
PPY	<i>PPY_PHE1</i>	Gammaproteobacteria	<i>Nevskia</i>	23.90 ± 18.53	23.92 ± 11.07	<0.01
	<i>PPY_PHE2</i>	Betaproteobacteria	Burkholderiales order	11.61 ± 10.5	6.31 ± 2.57	0.01 ± 0.02
	<i>PPY_PHE3</i>	Betaproteobacteria	<i>Collimonas</i>	2.13 ± 0.82	7.89 ± 6.53	0.29 ± 0.2
	<i>PPY_PHE4</i>	Alphaproteobacteria	Hyphomicrobiaceae family	6.22 ± 3.91	2.55 ± 1.09	0.25 ± 0.12
	<i>PPY_PHE5</i>	Gammaproteobacteria	<i>Dyella</i>	4.86 ± 2.53	3.53 ± 2.57	0.04 ± 0.03
	<i>PPY_PHE6<sup>c</sup></i>	Betaproteobacteria	<i>Polaromonas</i>	6.59 ± 4.59	1.11 ± 0.22	<0.01
	<i>PPY_PHE7</i>	Alphaproteobacteria	Hyphomicrobiaceae family	2.30 ± 0.9	3.12 ± 3.83	0.03 ± 0.06
	<i>PPY_PHE8</i>	Gammaproteobacteria	<i>Nevskia</i>	0.91 ± 0.55	1.60 ± 2.24	<0.01
	<i>PPY_PHE9</i>	Alphaproteobacteria	Acetobacteraceae family	1.51 ± 0.47	1.49 ± 1.19	<0.01
	<i>PPY_PHE10</i>	Actinobacteria	<i>Mycobacterium</i>	2.04 ± 1.96	0.73 ± 0.46	<0.01
	<i>PPY_PHE11</i>	Betaproteobacteria	<i>Herbaspirillum</i>	0.93 ± 1.20	1.17 ± 0.76	<0.01
	<i>PPY_PHE12<sup>b, c</sup></i>	Betaproteobacteria	<i>Rhodoferax</i>	1.40 ± 0.15	0.37 ± 0.64	0.10 ± 0.09
	<i>PPY_PHE13</i>	Gammaproteobacteria	<i>Nevskia</i>	0.46 ± 0.46	1.21 ± 1.52	<0.01

788 <sup>a</sup> These taxonomic identifications were confirmed phylogenetically (FIG S5)

789   <sup>b</sup> The two OTUs Pv\_PHE4 and PPY\_PHE12 are similar  
790   <sup>c</sup> These OTUs showed significant difference between microcosms with and without rhamnolipids ( $P < 0.05$ ,  
791   Student's t test)  
792

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## FIGURE LEGENDS

FIG 1: Phenanthrene sorption isotherms (expressed as the quantity of PHE adsorbed on the solid phase  $Q_{ads}$ , relative to the concentration of PHE in solution at the equilibrium  $C_{eq}$ ) at 25°C with and without rhamnolipids (0.55 g.L<sup>-1</sup>) for (a) PPY soil and (b) Pv soil.

FIG 2: PHE degradation on PPY and Pv soils in the presence (at 300 mg.kg<sup>-1</sup>) and absence of rhamnolipids in microcosms spiked with <sup>13</sup>C PHE (at 300 mg.kg<sup>-1</sup>). DT20 and DT90 corresponding respectively to 20% and 90% of PHE dissipation are represented for both soils.

FIG 3: Quantification of bacterial 16S rRNA gene copies in the 12 fractions separated by CsCl gradients using real-time quantitative PCR. Data are presented for the two studied soils (PPY and Pv) with and without rhamnolipids and at DT90. Fractions n°3 and n°7 were selected for light DNA and heavy DNA sequencing and analyses, respectively.

FIG 4: Factorial correspondence analysis (FCA) of the relative proportion of the main OTUs (top 100) present in the (a) PPY and (b) Pv samples. Full-tone graphic patterns correspond to microcosms with rhamnolipids. For each soil type the principal component axes F1 and F2 explain most of the variance in the data cumulatively.

Figure 1

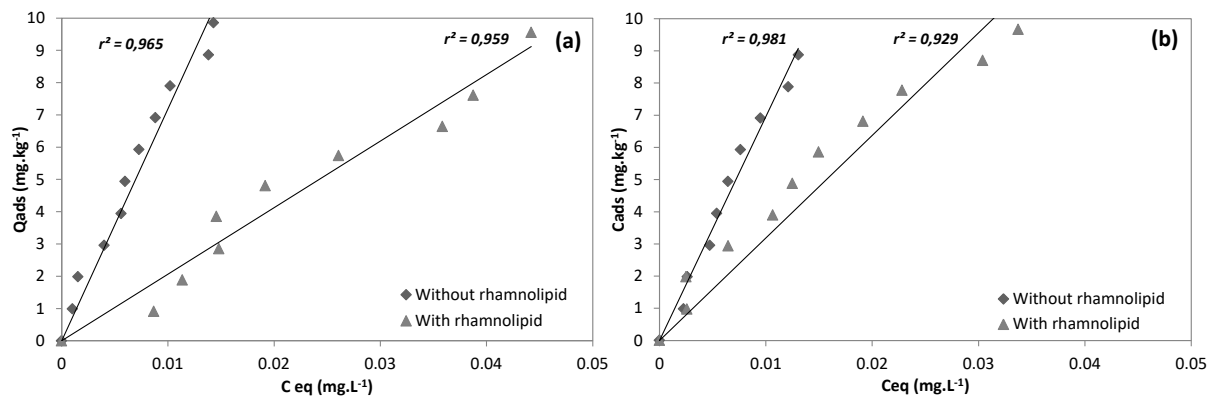




Figure 2

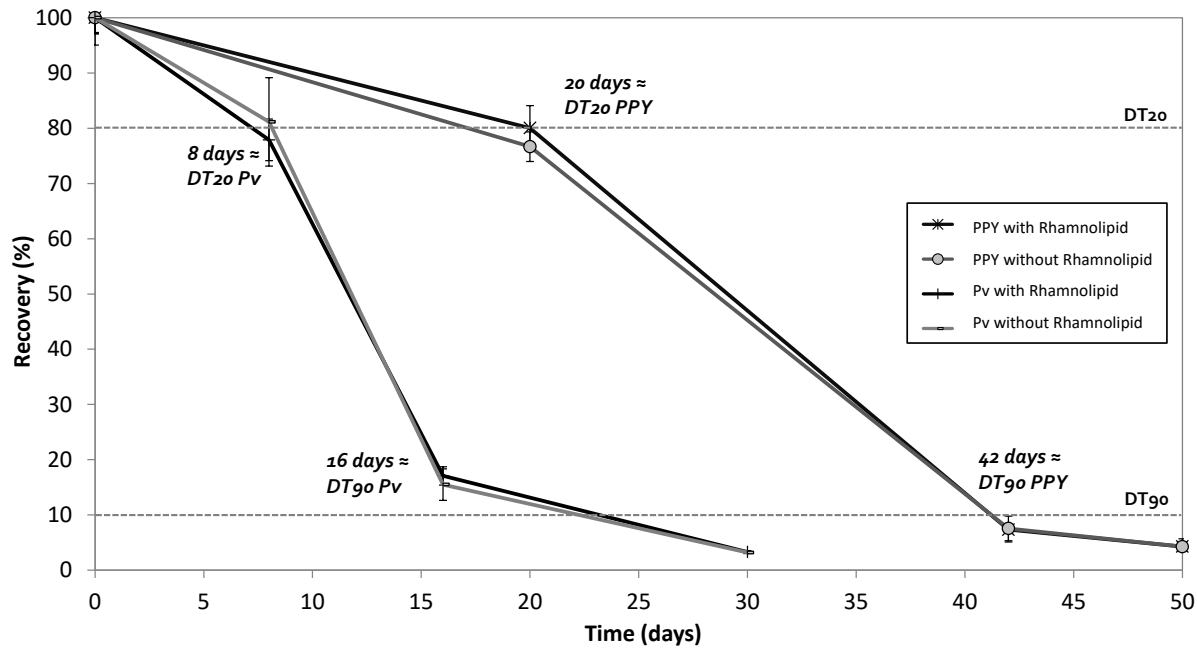


Figure 3

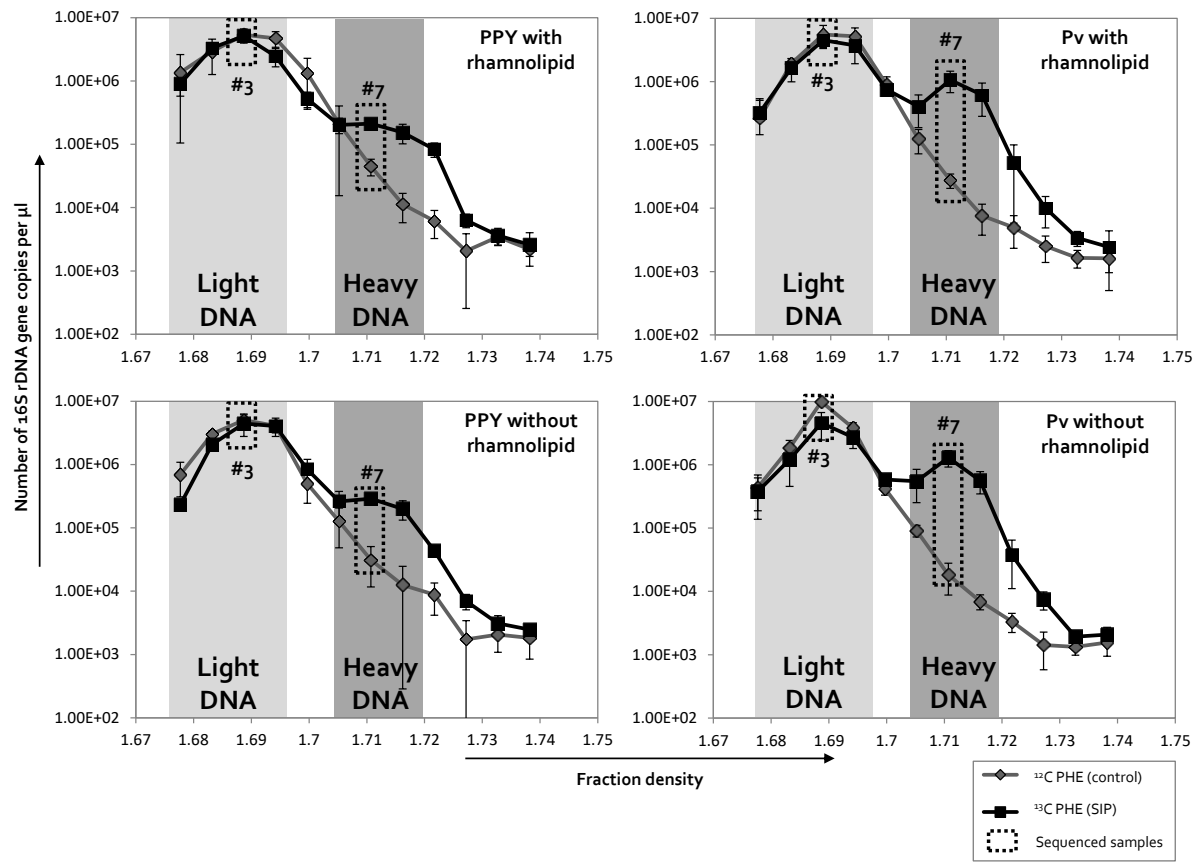
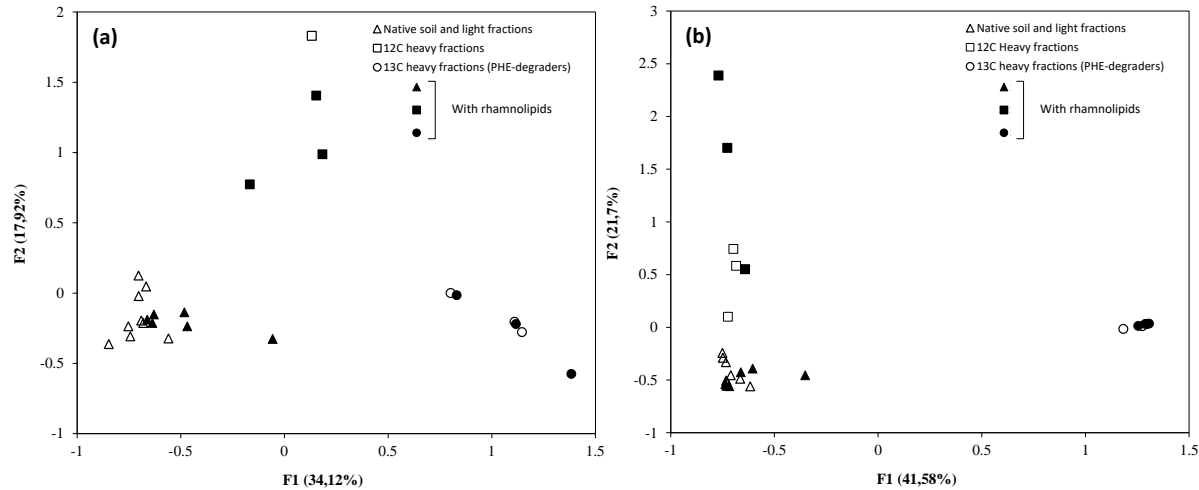
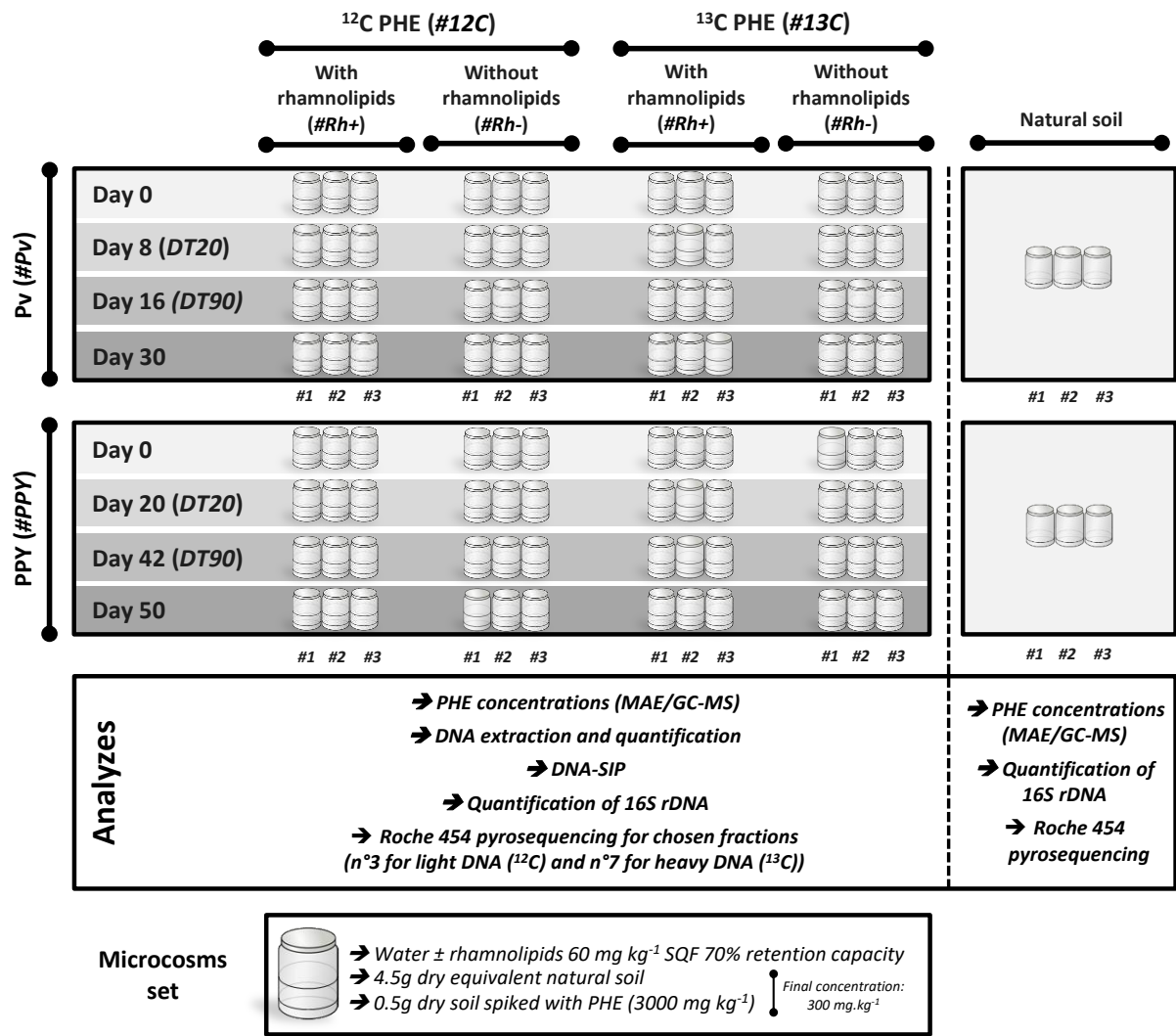


Figure 4



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Supplemental materials



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834 FIG S1: Experimental set up corresponding to the phenanthrene-contaminated microcosms.

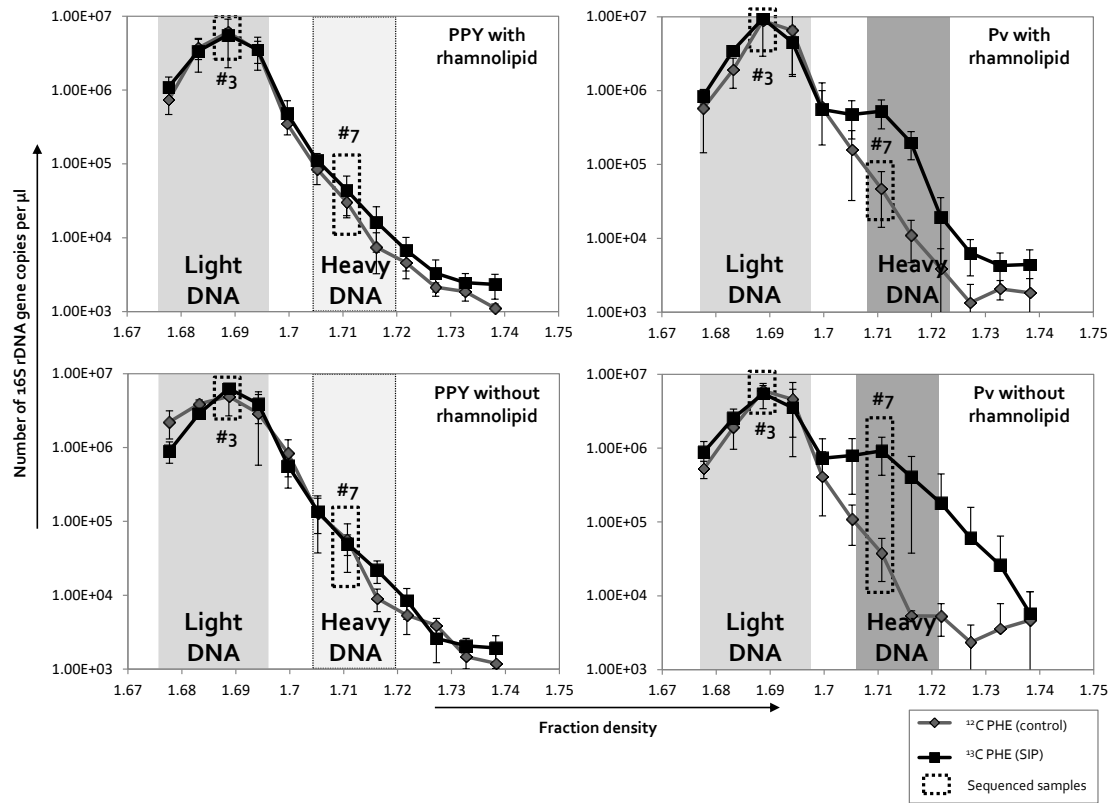
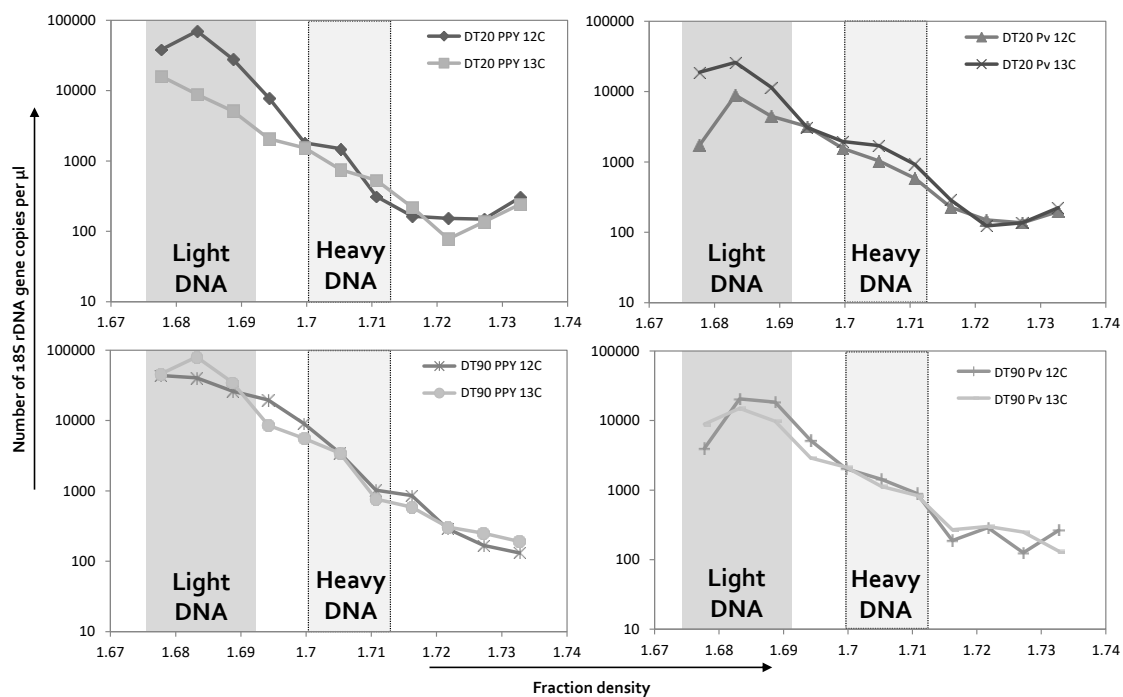


FIG S2: Quantification of bacterial 16S rRNA gene copies in the 12 fractions separated by CsCl gradients using real-time quantitative PCR. Data are presented for the two studied soils (PPY and Pv) with and without rhamnolipids and at DT20. Fractions n°3 and n°7 were selected as models for light DNA and heavy DNA containing fractions, respectively.

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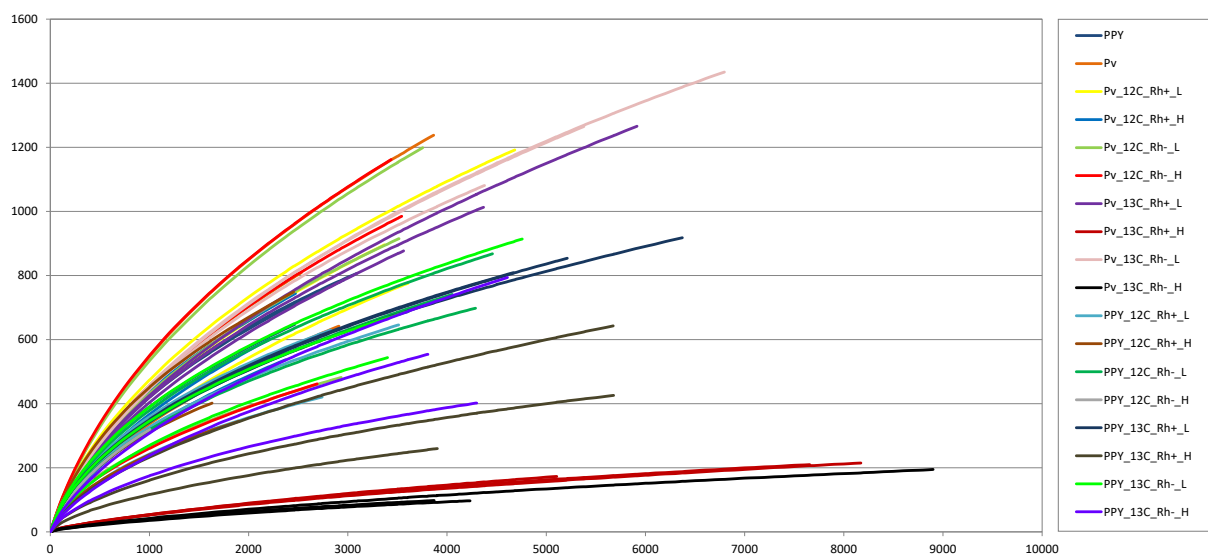


846

847 FIG S3: Quantification of fungal 18S rRNA gene copies in the 12 fractions separated by CsCl  
 848 gradients using real-time quantitative PCR. Data are presented for the two studied soils (PPY  
 849 and Pv) with rhamnolipids, at DT20 and DT90, only for one replicate.

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851

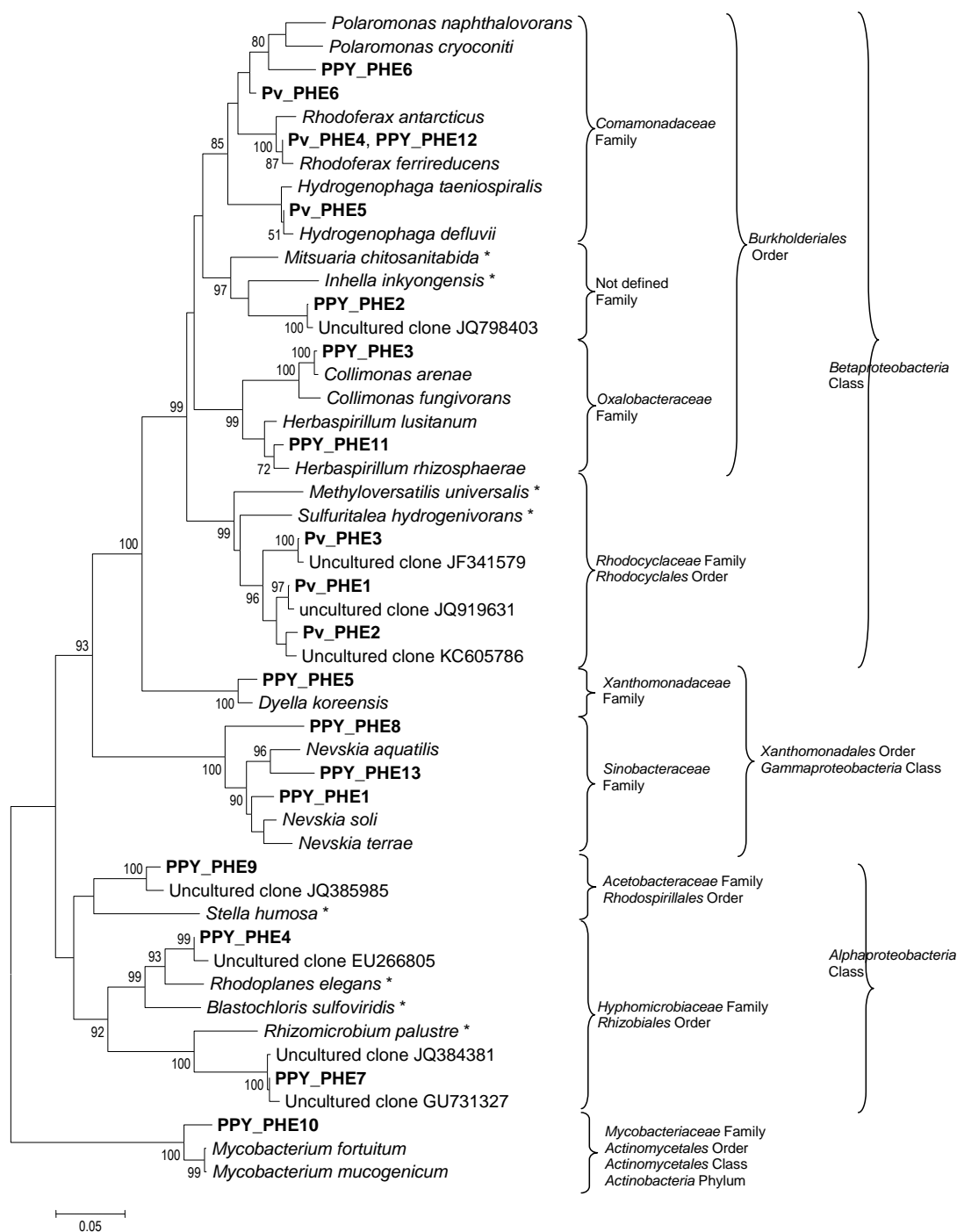


852

853 FIG S4: Rarefaction curves based on OTU at 97% similarity. Sequencing of 16S rDNA from  
 854 light (L) and heavy (H) DNA fractions ( $n^{\circ}3$  and  $n^{\circ}7$ , respectively) isolated from soils (Pv and  
 855 PPY) incubated with  $^{12}\text{C}$  or  $^{13}\text{C}$ -PHE, and with and without rhamnolipids (Rh+ and Rh-). For  
 856 each sample, triplicates are represented.

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860 FIG S5: 16S rRNA gene phylogeny of the dominant PHE-degrading bacteria from soil  
 861 microcosms enriched with PHE. Sequences of closest relative bacteria are included in the  
 862 analysis. The unrooted dendrogram was generated using neighbor joining analysis. The scale  
 863 bar corresponds to 0.05 substitutions per nucleotide position. Numbers on tree branches indicate  
 864 bootstrap for those branches having >80% support. Sequences provided in this study are in  
 865 bold.



