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Meng Xia, Diego Zabrodiec, Philippe Scoufflaire, Benoît Fiorina, Nasser Darabiha. Experimental and numerical studies of pulverized coal devolatilization and oxidation in strained methane/air flames. Proceedings of the Combustion Institute, 2017, 36 (2), pp.2123 - 2130. 10.1016/j.proci.2016.07.080 . hal-01542060

HAL Id: hal-01542060

<https://hal.science/hal-01542060>

Submitted on 6 Apr 2020

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Experimental and numerical studies of pulverized coal devolatilization and oxidation in strained methane/air flames

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Experimental and numerical investigations of the combustion of a methane/air mixture fed with coal particles are performed. The retained configuration is a laminar 1-D strained flame impinging a wall. The wall surface promotes the formation of a methane/air flame front parallel to the stagnation plane. Coal particles oxidation takes place in the hot region between the flame front and the wall. Laser induced fluorescence PLIF measurements of OH radical and imaging of CH*, C₂* spontaneous emission are performed to identify the chemical flame structure. The inlet fresh gas boundary conditions are well characterized in the experiments in terms of gas velocity but also number of coal particles and mean particle diameter. These conditions are used to perform numerical simulations with the 1-D REGATH code developed at EM2C laboratory. The governing equations are fully coupled between gas and particle phase. Detailed chemical kinetics based on USC-Mech II reaction scheme, NO_x chemistry and OH*/CH*/C₂* sub-mechanisms are considered. The numerical predictions showed good agreement with the experimental data. The analysis of obtained results confirmed that the composition of volatile species of coal has a significant influence on the CH* and C₂* concentration in the post-flame region. Finally, the effects of volatile matter composition, strain rate and particle size on the NO_x, C₂H₂ and C₆H₆ formation are studied.

1. Introduction

Coal has been and will continue to be one of the most important resources in the long term due to its abundant worldwide reserves and competitively low prices, especially used in the power generation [1]. However, with the growing demand of more reliable and cleaner energy supply, pollutants such

as NO_x , unburned hydrocarbons (HC) and soot, emitted during coal combustion are one of the key issues in global air pollution. Hence, effectively reducing pollution from coal combustion, especially from those with low rank coal, is significant to limit greenhouse gases and respect air pollutant emission standards.

Coal combustion is a complex phenomenon that involves devolatilization, gasification and both volumetric and surface chemical reactions. In order to characterize the behavior of pulverized coal flames such as the flame structure, stability, burnout and pollutant formation, both experimental techniques and numerical simulations are essential to get a better understanding of these physical and chemical processes. In particular, the gas phase reaction of coal volatile matter is an extremely important process that controls the behaviors of pulverized coal combustion [2]. Xu and Tomita [3,4] studied experimentally the pyrolysis of different ranks of coals in an inert atmosphere. The pyrolysis products were classified into inorganic gases (H_2 , CO_2 , CO , and H_2O), hydrocarbon gases (C1–C3), light hydrocarbon liquids (benzene, toluene, xylene, phenol and cresol) and tar. The yields of different species depend significantly on the coal type as well as pyrolysis temperature.

Entrained flow reactors, such as flat flame burners, were used in several studies [5–7] to investigate the effect of coal devolatilization and gasification in a preheated unstrained laminar flow under air and oxy-combustion conditions. Optical pyrometry and particle image velocimetry (PIV) as well as char and burnt gases composition measurements were carried out. Coal devolatilization and char oxidation sub-models able to predict the product composition in this configuration have been validated [7,8]. Experimental studies [9–11] have been also conducted on pulverized coal jet flame structure using OH planar laser-induced fluorescence (OH PLIF). Balusamy et al. [11] investigated the effect of oxidizer/diluent concentrations and coal loading rate.

The LES simulation performed by Muto et al. [12] successfully predicted the characteristics of Balusamy configuration [11]. Another well-studied setup is the CRIEPI jet flame burner [10] and several validations of LES modeling were conducted and showed good results [13]. In these simulations, the volatile matter is modeled as mono-species, i.e. $\text{C}_x\text{H}_y\text{O}_z$ [12,13] or only CH_4 [14] using global reaction schemes for reducing computational cost. Although the results show a good accordance with velocity statistics and temperature profiles, the predictions of species concentration are qualitative.

Whilst devolatilization and gasification sub-models are involved in LES/RANS simulations of pulverized coal turbulent flames, their impact on the flame structure and prediction of pollutants still remains unknown. According to Peters [15],

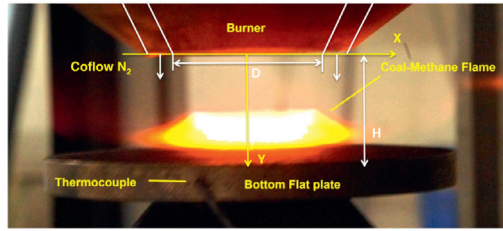


Fig. 1. Direct view of coal/methane/air flame.

instantaneous flame element embedded in a turbulent flow has the structure of a 1-D strained laminar flame. This elemental configuration, retained in many studies on gaseous [16,17] or two-phase flow flames [18,19] is attractive to understand such fundamental coal combustion properties.

The objective of the present work is to study experimentally and numerically the combustion of pulverized coal in a mixture of methane/air reactive flow and to investigate the impact of coal pyrolysis sub-models on flame structure and pollutant formation. A laboratory-scale laminar strained configuration is designed. The configuration is a premixed methane/air flow that impacts on a wall surface. The pulverized coal conveyed by the flow reacts in the hot region confined between the methane-air flame front and the wall. Spontaneous emission of CH^* and C_2^* as well as LIF of OH spectrum are experimentally measured across the jet axis. Numerical analyses are performed by implementing kinetic models of pulverized coal combustion in the 1-D REGATH code developed at EM2C laboratory. The coal particles are simulated by devolatilization and gasification reactions, coupling with the gas phase calculated with a detailed chemical reaction scheme for the methane-air combustion [20]. The validity of modeling is assessed by comparing the experimental and numerical results of OH, CH^* and C_2^* radicals. The effects of gas-phase volatile reactions on the flame structure and pollutant formation, including unburned hydrocarbons, NO_x and soot precursor such as C_2H_2 , C_6H_6 are examined in detail.

2. Experimental configurations

2.1. Experimental setup

The experiments were carried out using a strained flow burner shown in Fig. 1. Additional details are found in [16,19]. The convergent nozzle is axisymmetric and has an inner diameter of 10 mm. Pulverized coal particles are carried by premixed methane/air flow at ambient pressure and temperature, which then impacts a horizontal metallic brass wall. In this paper, the distance between the nozzle and the wall is kept constant to 10 mm. A coaxial nozzle fed by nitrogen

creates an inert coflow, which protects the reaction zone from ambient perturbations that could disturb the measurements. The flow impacting the wall forms a strained flow. A laminar flame front stabilizes at a distance from the wall, which depends on the laminar flame speed and the injection velocity. The hot gases from the methane/air combustion are located between the flame front and the wall. Pulverized coal particles are heated in this hot region. Figure 1 shows a direct view of the coal/methane/air flame. The high luminosity is only observed in coal-seeded flames.

2.2. Diagnostic techniques

PLIF measurements of OH radical are performed. The laser used for the PLIF is a tunable Dye laser (using Rhodamine 590 as a lasing medium) pumped by a Nd:Yag pulsed laser at 532 nm. A frequency doubling unit is mounted at the exit of the Dye Laser in order to reach the wavelength of 283.09 nm which corresponds to the Q1(6) rotational branch from the OH transition system. The duration of the laser pulse is 10 ns with the power of 18 mJ per pulse. An ICCD camera of 1024 × 1024 pixel detector is used with a Nikon UV-Nikor 105 mm. The planar laser sheet is aligned with the jet axis and includes the flame reaction zone. The OH Narrow-band filter is centered at 313 nm, which has 10 nm bandpass and 68% transmission in the maximum. The OH profiles are corrected with the laser sheet energy profile on a single shot basis. An average of over 200 OH PLIF images is done. From the averaged images, the relative concentration profiles are obtained from the pixels values at the jet axis. The scale in the resulting measurements represents pure fluorescence intensity, and does not directly correspond to OH mole fraction because the LIF signals are not corrected for quenching, the thermal distribution and saturation effects. However it has been observed that quenching effects on the OH PLIF signals are constant. The OH LIF signals can therefore be considered linearly proportional to the OH radical population within a calibration constant.

Direct view images of CH* and C₂* spontaneous emission have also been recorded using the same camera and optics used for the PLIF. Emission profiles are obtained by averaging 200 images. The uncertainty of measurements corresponds to the RMS of the signals captured by the camera and is less than 5% in all experiments. The images are corrected by means of dark background subtraction. Narrow-band interference filters are interposed along the optical path for capturing the CH* emission. The filter used has 60% transmission and a 10 nm wide bandpass centered around 430 nm. The filter for C₂* bandpass is centered around 516.5 nm and has 10 nm bandwidth. The spatial resolution obtained by the optical arrangements is 0.1 mm.

2.3. Experimental cases studied

Two experimental flame configurations, named case METH and case METH-COAL, are performed with and without the injection of coal particles, respectively. For both flames, the equivalence ratio of the carrying methane/air stream is characterized by an injection temperature of 310 K and an equivalence ratio of $\phi = 0.82$. All gaseous mass flow rates are controlled by the mass flow meters with an accuracy of $\pm 1.5\%$. The minimum and maximum coal particle diameters found in case METH-COAL are, respectively, 3.4 μm and 98 μm . The measured Sauter Mean Diameter (D32) is about 15 μm . The distribution of particle size is available as Supplemental material.

3. Numerical simulations

3.1. Mathematical modeling

The methane-air mixture is injected conveying mono-disperse pulverized coal particles in the one-dimensional axisymmetric formulation. The similarity approach is employed by searching for similar solutions of both gaseous and solid phases balance equations in the vicinity of the central axis. The similarity analysis leads to solutions of the form: $u_g = x U_g(y)$, $v_g = v_g(y)$, $T_g = T_g(y)$, $\rho_g = \rho_g(y)$, $Y_k = Y_k(y)$, $k = 1, \dots, K$, $u_s = x U_s(y)$, $v_s = v_s(y)$, $T_s = T_s(y)$, $\rho_s = \rho_s(y)$, $\alpha_s = \alpha_s(y)$, and $Y_{sdry} = Y_{sdry}(y)$. The superscript g represents gaseous phase and the superscript s the solid phase. ρ is density, u and v the radial and axial velocities, respectively, T temperature. Y_k is k th species mass fraction, α_s particles volume fraction, and Y_{sdry} dry coal mass fraction. U_g and U_s , respectively, describe the y -dependence of the transverse velocities u_g and u_s . Assuming a constant radial pressure-gradient, J describes $J = -\frac{1}{x} \frac{\partial p}{\partial x}$, and is considered constant along the axial coordinate: $\partial J / \partial y = 0$. f_x and f_y are the drag forces in x and y directions. The parameter j corresponds to 0 and 1 for two-dimensional and axisymmetric configurations, respectively. The governing equations, which describe the flow conservation of gas and solid phases, can be written in the following form:

$$\frac{\partial \rho_g}{\partial t} + (1 + j) \rho_g U_g + \frac{\partial \rho_g v_g}{\partial y} = n \dot{m}_s \quad (1)$$

$$\rho_g \frac{\partial Y_k}{\partial t} + \rho_g v_g \frac{\partial Y_k}{\partial y} = - \frac{\partial}{\partial y} (\rho_g Y_k V_{k,y}) + W_k \dot{\omega}_k + n \dot{m}_s (\gamma_{sk} - Y_k), \quad k = 1, \dots, K \quad (2)$$

$$\rho_g \frac{\partial u_g}{\partial t} + \rho_g U_g^2 + \rho_g v_g \frac{\partial U_g}{\partial y} = J + \frac{\partial}{\partial y} \left(\mu_g \frac{\partial U_g}{\partial y} \right) + n \dot{m}_s (U_s - U_g) - n f_x \quad (3)$$

$$\begin{aligned} \rho_g \frac{\partial h_g}{\partial t} + \rho_g v_g \frac{\partial h_g}{\partial y} &= \frac{\partial}{\partial y} \left(\lambda_g \frac{\partial T_g}{\partial y} \right) \\ &- \frac{\partial}{\partial y} \left(\sum_{k=1}^K (\rho_g Y_k V_{ky} h_k) \right) + n \dot{m}_s \sum_{k=1}^K (\gamma_{sk} h_k(T_s)) \\ &- Y_k h_k(T_g) - n \dot{q} - \dot{q}_R \end{aligned} \quad (4)$$

$$\alpha_s \frac{\partial \rho_s}{\partial t} + \alpha_s v_s \frac{\partial \rho_s}{\partial y} = -n \dot{m}_s \quad (5)$$

$$\frac{\partial \alpha_s}{\partial t} + (1+j) \alpha_s U_s + \frac{\partial (\alpha_s v_s)}{\partial y} = 0 \quad (6)$$

$$\alpha_s \rho_s \frac{\partial Y_{s, dry}}{\partial t} + \alpha_s \rho_s v_s \frac{\partial Y_{s, dry}}{\partial y} = -\dot{\omega}_{s, pyr} + Y_{s, dry} n \dot{m}_s \quad (7)$$

$$\alpha_s \rho_s \frac{\partial U_s}{\partial t} + \alpha_s \rho_s U_s^2 + \alpha_s \rho_s v_s \frac{\partial U_s}{\partial y} = n f_x \quad (8)$$

$$\alpha_s \rho_s \frac{\partial v_s}{\partial t} + \alpha_s \rho_s v_s \frac{\partial v_s}{\partial y} = \alpha_s \rho_s g + n f_y \quad (9)$$

$$\begin{aligned} \alpha_s \rho_s C_{\rho_s} \frac{\partial T_s}{\partial t} + \alpha_s \rho_s v_s C_{\rho_s} \frac{\partial T_s}{\partial y} &= -\dot{\omega}_{s, pyr} \Delta h_{pyr} \\ &- \dot{\omega}_{s, char} \Delta h_{char} - \sum_{k=1}^K \gamma_{sk} h_k(T_s) n \dot{m}_s + n \dot{q} \end{aligned} \quad (10)$$

In these equations n represents the particle number density given by $n = \alpha_s / v_p$, with v_p the volume of a single particle. The mass of a particle is $m_s = \rho_s v_p$, and $n \dot{m}_s$ is the total mass devolatilization rate of particles. V_{ky} is the k th species diffusion velocity modeled by using mixture averaged multicomponent transport [21]. γ_{sk} is the k th species percentage of the total volatilization rate. $\dot{\omega}_k$ and W_k are the k th species molar production rate and molar mass, respectively. $\dot{\omega}_{s, pyr}$ and $\dot{\omega}_{s, char}$ represent production rate of coal and oxidation of char, respectively. \dot{q} is the heat transfer rate between the gaseous and solid phases involving convective and radiative heat transfer. The Stefan–Boltzmann model is considered for the solid phase radiation. \dot{q}_R is radiative heat transfer from the gaseous phase to the exterior using model as in [22] including reabsorption by species such as CO₂, H₂O and CO.

3.2. Numerical setup

The simulation are performed using REGATH code [18] for the strained flow flame with detailed chemistry and mixture averaged multicomponent transport. To model the gaseous combustion kinetics, we have employed the H₂/CO/C₁–C₄ mechanism USC-Mech II [20]. NO_x chemistry from GRI-Mech 3.0 [23] is considered for predictions of NO_x formation. To enable comparison with CH* measurements, the CH* sub-mechanisms used by Alviso et al. [19] have been also added. The total

chemical mechanism contains 131 species and 916 reactions.

According to experiments, an inlet stream of methane and air characterized by an equivalence ratio $\phi = 0.82$ is injected from the nozzle at a gas flow velocity equal to 1.6 m/s. The fresh gases stream is fed with coal particles. The particle velocity has been measured by PIV. PIV results show that the particle average velocity matches the gas velocity at the injector exit. A typical profile is available in Supplemental material. The diameter of injected particles corresponds to the Sauter Mean Diameter measured by microscope: $D_s^0 = 15 \mu\text{m}$ and the particle number density (per unit volume) is estimated $n_s = 4.54 \times 10^{10}$ particles per m³ from coal mass flow rate measurements, which leads to $\alpha_s = 8.0 \times 10^{-5}$. The brass plate is modeled as an isothermal wall of 600 K, which corresponds to the temperature measured during the experiment. The initial properties of coal particles used for the simulations are: $\rho_s = 909 \text{ kg/m}^3$, and the constant pressure heat capacity $c_{ps} = 1.5 \text{ kJ/(kg K)}$. The pressure is equal to one atmosphere.

The coal particle is considered dry and ash-free. We consider only the transformation of solid phase to gaseous phase, i.e. the process of pyrolysis: $Coal \implies (1 - v_{char}) Volatile + v_{char} Char$ and the surface oxidation: $Char \implies (1 + v_{O_2}) CO_2$. The total mass loss rate of particle phase is given by: $\dot{m}_s = (1 - v_{char}) \dot{\omega}_{s, pyr} + \dot{\omega}_{s, char}$. Coal devolatilization source term $\dot{\omega}_{s, pyr}$ will be either modeled by first-order single reaction model [24] or by a 2-step model [25]. The kinetic-diffusion model [26] is used for char oxidation source term $\dot{\omega}_{s, char}$. The parameter for single reaction model and char oxidation model are modeled as in [27], and the 2-step model parameters are taken from [7]. The heat transfer between gas and particle phase decomposed as: $n \dot{q} = \dot{Q}_{conv} + \dot{Q}_{rad}$. The convective heat transfer is given by $\dot{Q}_{conv} = \alpha_s \sigma_s k_{conv} (T_g - T_s)$ where σ_s is the particle surface/volume ratio, k_{conv} is the heat transfer coefficient of coal particles calculated assuming a Nusselt number of 2. The radiative heat transfer is given by $\dot{Q}_{rad} = \epsilon \alpha_s \sigma_s \sigma (T_w^4 - T_s^4)$ where the radiation temperature T_w is set as the boundary temperature in this configuration, the emissivity of coal particles $\epsilon = 0.85$ is kept constant in all simulations, and σ is the Stefan–Boltzmann constant.

3.3. Determination of volatile compositions

The composition of volatile matter can be obtained from pyrolysis models such as CPD (Chemical Percolation Devolatilization) model [28], FG-DVC model [29] and FLASHCHAIN model [30]. Heizprofil (HP) lignite briquettes are grinded to get pulverized coal particles employed in the experiment. In this study, its total volatile matter (TVM) composition is considered as similar to the composition of lignite coals South Beulah (SB) and Morwell (MW) identified in [2]. We use the data

Table 1
HP, SB and MW coal properties considered for numerical simulations.

Coal name	Proximate analysis (wt %)				Ultimate analysis (wt %)				
	Moisture	Ash	Volatile matter	Fixed carbon	C	H	O	N	S
HP	19.0	4.3	50.6	45.1	69.0	5.0	24.7	0.8	0.5
SB	18.1	13.7	38.6	47.7	71.8	4.7	19.2	1.4	2.9
MW	19.6	2.0	51.5	46.5	67.4	5.0	26.8	0.5	0.3

Table 2
Mass percentage of species in TVM from pyrolysis of SB and MW.

Coal name	CH ₄	CO	CO ₂	H ₂	H ₂ O	C ₂ H ₄	C ₂ H ₆	C ₃ H ₆	C ₃ H ₈	tar(C ₆ H ₆)
SB	3.4	17	15	0.66	12	1.04	1.04	1.04	0.52	48.3
MW	4.7	19	24	0.64	15	0.96	0.96	0.96	0.48	33.3

Table 3
Different cases studied with estimated laminar flame speed S_f (m/s).

Case	Description	TVM	$1 - v_{char}$	Devolatilization	S_f
A	$\dot{m}_s = 0$ in Eqs. (1)–(4)	SB	0.6	1-step	0.388
B	TVM HC considered as CH ₄ only	SB	0.6	1-step	0.435
C	Reference case	SB	0.6	1-step	0.425
D	MW composition	MW	0.65	1-step	0.42
E	2-step devolatilization	SB	0.6	2-step	0.424
F	$V = 1.3$ m/s	SB	0.6	1-step	0.41
G	$D_s^0 = 30$ μ m	SB	0.6	1-step	0.412

published in [2] obtained from the CPD model. It predicts the formation of CH₄, CO, CO₂, H₂, H₂O, C₂H₄, C₂H₆, C₃H₆, C₃H₈, and tar. The properties of these lignite coals: HP, SB and MW coal, are listed in Table 1. Table 2 shows the mass percentage of different volatile species of SB and MW coal.

3.4. Numerical cases studied

Variations of TVM composition and coal devolatilization model are explored first. Table 3 shows the different cases studied. The case A corresponds to a simulation where coal oxidation and devolatilization do not impact the gaseous phase (i.e. $\dot{m}_s = 0$ in Eqs. (1)–(4)). Flame temperature is then only impacted by the combustion of methane injected through the nozzle. Cases B, C, E, F and G represent the TVM composition of SB coal (Table 2). In order to analyze the influence of unburned hydrocarbons present in the volatile matter on the flame structure, they are considered in case B as only CH₄ (55.34 %). The heating rate of coal particles is estimated $\approx 5.10^5$ K/s from the simulation of temperature and local velocity, leading to a TVM yield ($1 - v_{char}$) of 0.6 according to [2]. Case D corresponds to the TVM composition of MW coal (Table 2) with $1 - v_{char} = 0.65$. Cases A to D use the 1-step devolatilization model [24], whereas the 2-step devolatilization model [25] is used in case E. Case C is considered as a reference case, all other cases differing by only one parameter from it. Cases F and G are parametric studies further explained in 4.5.

4. Results and discussion

4.1. Flame speed

The laminar flame speeds of the stretched flames are estimated by using the minimum of the velocity in the reaction zone [31] and are indicated in Table 3. A variation of 9.5% is observed when the impact of coal oxidation and devolatilization on the gaseous phase is accounted for. This increase of laminar flame speed is caused by the local augmentation of the fuel/air equivalence ratio induced by the release of volatile gases.

4.2. Temperature profiles and flame structure

Figure 2 plots the simulated temperature profiles of gas and coal particles along the jet axis. A shift of 0.6 mm towards the burner exit due to a difference of 0.037 m/s in the laminar flame speed combined with a difference of 140 K in the maximum temperature are also observed between cases A and C. Comparing cases B and C, the influence of volatile gases composition is less significant as only differences of maximum temperature and laminar flame speed of 40 K and 0.01 m/s, respectively. The TVM composition variation illustrated by case D shows slight differences in temperature maximum and flame front position in comparison with the reference case C. Also the complexity of the devolatilization model (case E: two-step) does not impact significantly the maximum temperature and position predicted by case C (one-step).

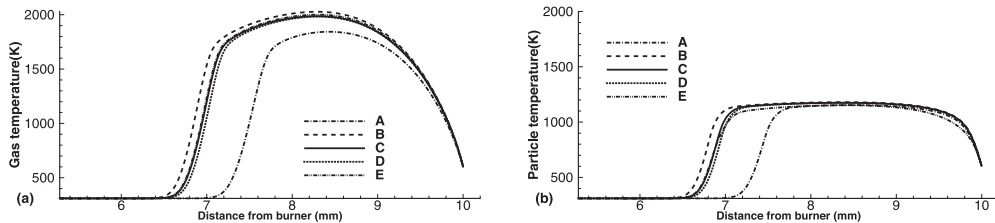


Fig. 2. Comparison of temperature profiles of (a) gas, and (b) coal particles, in cases A–E .

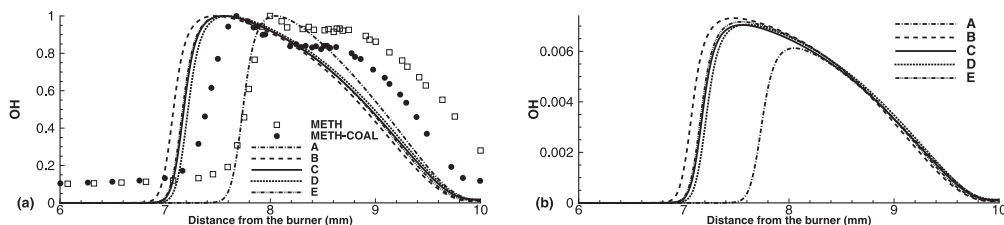


Fig. 3. Comparison of (a) normalized OH mole fraction profiles between experimental and numerical results, (b) absolute numerical OH profiles, in cases A–E .

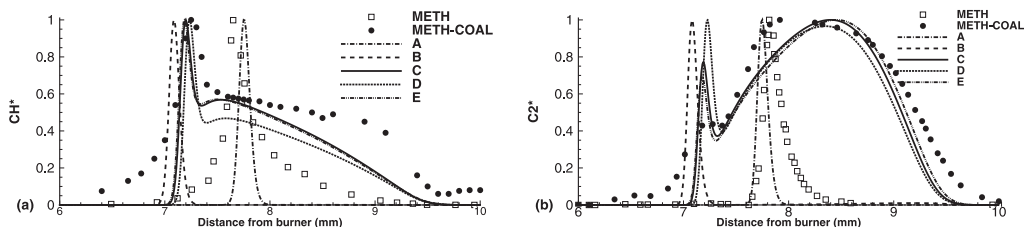


Fig. 4. Comparison of normalized (a) CH^* , and (b) C_2^* mole fraction profiles between experimental and numerical results in cases A–E.

4.3. Experimental and numerical profiles of OH radical

The camera intensity used to measure OH radical varies between METH and METH-COAL cases. To avoid misleading interpretations of OH predictions, each numerical and experimental data is normalized between 0 and 1 for comparisons in Fig. 3a. Absolute non-normalized profiles of OH mole fraction are shown in Fig. 3b. The flame front position predicted by case A (where the impact of coal particles on the flow is neglected) matches well that of pure methane experiment (METH case). This result validates the numerical simulation of the gaseous phase. When coal particles are injected, the OH measured profiles are shifted toward the burner inlet because of the impact on the flame speed as discussed previously. This trend is quantitatively retrieved by the simulations B–E, which include coal devolatilization and oxidation. The variation of TVM composition and coal devolatilization model show slight influence in the predicted OH concentration.

4.4. CH^* and C_2^* spontaneous emission

Numerical and experimental profiles of CH^* and C_2^* emission are normalized between 0 and 1 for comparisons in Fig. 4a and b, respectively. The peak location of CH^* measured in case METH and METH-COAL are well retrieved by simulation A and B, respectively. Prediction the reactive layer position is however significantly improved when HC are included in the TVM composition (see cases C to E). Measurement in case METH-COAL also highlights a “prolongation” of the coal reaction zone in the burnt gases which is not observed in the methane flame. This effect is not observed in case B where the TVM unburned hydrocarbons are simplified as CH_4 only. Figure 4a shows that in cases C and D, the results of CH^* emission in the hot region are well predicted. Figure 4b compares normalized C_2^* emission intensity with numerical results. In the experiment, the intensity of C_2^* emission in the hot region between the coal flame front and the wall is significantly more important than the methane/air flame. This effect is predicted in cases C–E. The case

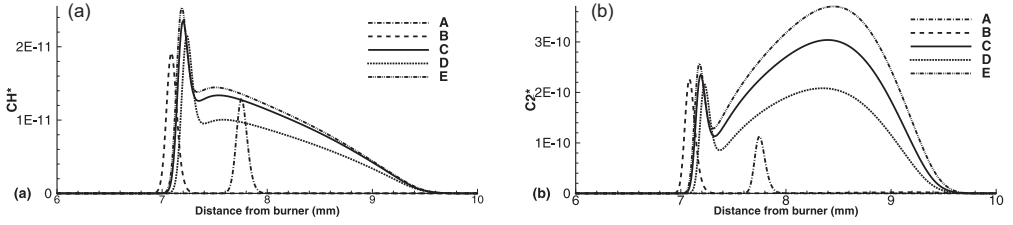


Fig. 5. Comparison of absolute numerical (a) CH^* , and (b) C_2^* mole fraction profiles in cases A–E.

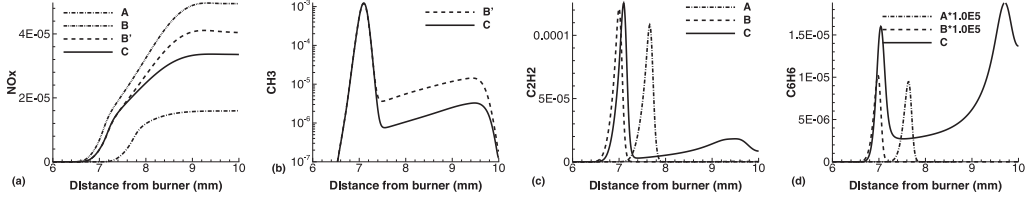


Fig. 6. Comparison of (a) NO_x mole fraction in cases A–C, (b) CH_3 mole fraction in cases B' and C, (c) C_2H_2 , and (d) C_6H_6 mole fraction profiles in cases A–C.

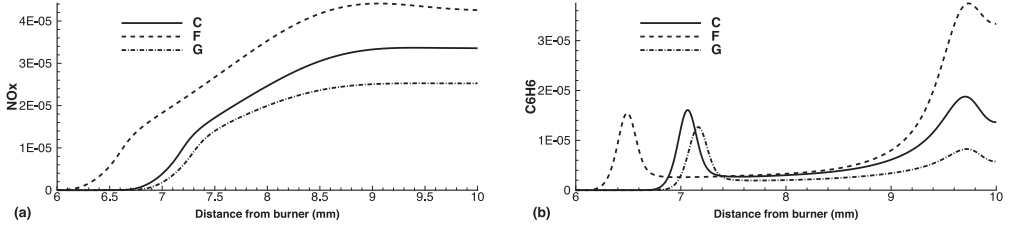


Fig. 7. Comparison of (a) NO_x , (b) C_6H_6 mole fraction profiles in cases C, F and G.

B where organic species are considered to be only CH_4 does not show similar trend in the post-flame region. Absolute non-normalized profiles of CH^* and C_2^* mole fraction predicted by the computation are shown in Fig. 5a and b, respectively. The maximum CH^* visible in Fig. 5a varies with the coal type. The complexity of the devolatilization model used in case E has a little influence on both CH^* and C_2^* emission.

4.5. Pollutant formation

Figure 6 shows the formations of NO_x , CH_3 , C_2H_2 and C_6H_6 . Due to the simplification of the devolatilization model, only NO_x produced by the gas-phase is discussed in this study. The NO_x production from coal-nitrogen is neglected. Comparison between cases A and C shows that the presence of volatile gases, by increasing the temperature, promotes the NO_x production. Differences observed between cases B and C predictions highlight a significant influence of the TVM composition. Case C differs from case B by the presence of heavier burnt species in the TVM (this explains the lower burnt gases temperature observed in Fig. 2). To identify the origin of these differences, we compute an additional case B' where

methane/air equivalence ratio is lowered to 0.8 to mimic the burnt gases temperature and flame speed (and position) of case C (see Fig. 2). Figure 6b shows that in case B', higher CH_4 concentration in the TVM leads to higher concentration of CH_3 radicals, which is directly involved in the prompt- NO reaction pathway. This explains the higher NO_x in the coal reaction zone compared to case B.

Formation of C_6H_6 and C_2H_2 , involved in the soot formation process, are influenced by the coal devolatilization, as shown in Fig. 6c and d. In particular, the presence of heavy hydrocarbons in the TVM promotes the formation of soot precursors in the burnt gases, localized between the flame front and the wall.

To study the influence of strain rate, the injection gas velocity V in case F (Table 3) is set to 1.3 m/s ($V = 1.6$ m/s in case C). Numerical profiles of NO_x and C_6H_6 mole fraction are presented in Fig. 7a and b, respectively. The flame front position is displaced according to the strain rate. The augmentation of residence time of coal particles induced by the decrease of strain rate explains the higher production of NO_x and C_6H_6 .

In case G (Table 3), the particle number density n is divided by 8, keeping α_s constant as well as the mass flow rate of coal particles. As all the coal

particles are represented by spheres, the particle diameter D_p^0 is doubled. The particles with smaller size have higher heating rate and the mixing of the air and volatile reactions at the particle surfaces are more active, promoting NO_x and C_6H_6 formation, presented in Fig. 7a and b, respectively. This effect of particle size has been observed experimentally by Sung et al. [32].

5. Conclusions

Both experimental and numerical studies of coal/methane/air mixture combustion in a strained flow configuration have been presented. The retained configuration is a reactive flow impinging a wall surface. The coal particles burn in the hot region confined between the flame front and the stagnation plane. OH PLIF, CH^* and C_2^* chemiluminescence signals are measured in the experiments. Reacting flow equations of gaseous and solid phases have been developed and implemented in the REGATH 1-D detailed chemistry flame solver. The analysis highlights the influence of coal devolatilization on the temperature and on the OH, CH^* and C_2^* profiles. The good agreement between experimental and numerical prediction of the reactive layer position shows that the flame speed is fairly predicted when unburned hydrocarbons presents in TVM are considered as only CH_4 . A significant improvement is however observed when heavy hydrocarbons are included the devolatilization gases composition. The influence of TVM composition on the pollutant formation is more pronounced, as unburned hydrocarbons significantly promotes the production of gas-phase NO_x and soot precursors. Finally parametric studies highlight the influence of strain rate and particle size on the pollutant formation.

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