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Structural study and ferroelectricity of epitaxial BaTiO₃ films on silicon grown by molecular beam epitaxy

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Integration of epitaxial complex ferroelectric oxides such as BaTiO₃ on semiconductor substrates depends on the ability to finely control their structure and properties, which are strongly correlated. The epitaxial growth of thin BaTiO₃ films with high interfacial quality still remains scarcely investigated on semiconductors; a systematic investigation of processing conditions is missing although they determine the cationic composition, the oxygen content, and the microstructure, which, in turn, play a major role on the ferroelectric properties. We report here the study of various relevant deposition parameters in molecular beam epitaxy for the growth of epitaxial tetragonal BaTiO₃ thin films on silicon substrates. The films were grown using a 4 nm-thick epitaxial SrTiO₃ buffer layer. We show that the tetragonality of the BaTiO₃ films, the crystalline domain orientations, and SiO₂ interfacial layer regrowth strongly depend on the oxygen partial pressure and temperature during the growth and on the post-deposition anneal. The ferroelectricity of the films, probed using piezoresponse force microscopy, is obtained in controlled temperature and oxygen pressure conditions with a polarization perpendicular to the surface. © 2014 AIP Publishing LLC.

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I. INTRODUCTION

Integration of epitaxial complex ferroelectric oxides such as BaTiO₃ on semiconductor substrates has been proposed for the fabrication of new silicon-based electronic and photonic devices.^{1–4} Since the first epitaxial growth of the perovskite SrTiO₃ on silicon by molecular beam epitaxy (MBE),⁵ various studies were carried out to integrate functional oxides on semiconductor platforms.^{5–17} Among the functionalities, ferroelectricity is one of the most appealing, in particular, for applications such as ferroelectric field-effect transistors (FETs) for steep subthreshold slope transistors.^{1,2} Such devices would enable low-voltage operation of logic devices and therefore reduction of power consumption. BaTiO₃ is an attractive candidate for applications such as negative-capacitance field-effect devices since it has a moderate Curie temperature.^{1,18,19} It is a well-known perovskite largely studied for its dielectric, piezoelectric, and ferroelectric properties. For the targeted applications, it is desirable that BaTiO₃ be ferroelectric with a polarization pointing out-of-plane. However, the growth of tetragonal *c*-axis oriented BaTiO₃ (the *c*-axis is perpendicular to the substrate plane) tends to be impeded by the large lattice mismatch (~4.4%) and thermal mismatch between BaTiO₃ and silicon. Due to the difference in the thermal expansion coefficients, the deposited film experiences a tensile bi-axial stress during the cooling down period, which favors the in-plane orientation of the long axis of the cell. Moreover, the out-of-plane polarization is not favored because of the depolarization field that

arises due to incomplete charge screening at the interface with the semiconductor or insulating buffer. Tetragonal *c*-axis films have been obtained using Ba(Sr)TiO₃ (Ref. 7) or SrTiO₃ (Refs. 3, 4, 8, 11, 13, and 14) as a buffer layer. The SrTiO₃/Si system has been widely studied^{20–23} and substrate's like quality can be obtained. Reversible switching of the ferroelectric polarization of BaTiO₃ thin films on SrTiO₃-buffered silicon with no bottom electrode has been reported recently by few groups.^{4,13,14} However, a systematic study of relevant molecular beam epitaxy parameters on the growth of BaTiO₃ on silicon is missing, although the resulting crystalline orientation and microstructure strongly depend on the processing conditions and strongly impact the ferroelectric properties.

Among the growth parameters, oxygen pressure has been shown to have a significant influence on the structure and properties of BaTiO₃ films. Zhao *et al.* observed a dependence of the crystallographic orientation on the oxygen pressure in the range of 1.5×10^{-6} to 10^{-1} Torr for films grown on SrTiO₃ by laser MBE.²⁴ Zhu *et al.*²⁵ observed a change in the type of predominant defects in laser-MBE grown BaTiO₃ on SrTiO₃ by varying the oxygen pressure in the range of 1.5×10^{-7} to 1.5×10^{-4} Torr. Rutkowski *et al.* observed a dependence with P(O₂) of (Ba,Sr)TiO₃ film composition grown by MBE on SrTiO₃.²⁶ There are also numerous studies on the oxygen dependence of BaTiO₃ microstructure for films grown by sputtering or conventional pulsed laser deposition (PLD), either on SrTiO₃ or MgO substrates.^{27–36} In these works, it has been reported that by

increasing the oxygen pressure, the out-of-plane lattice parameter decreases and that the films are in-plane oriented. This phenomenon was reported for $\text{Ba}_x\text{Sr}_{1-x}\text{TiO}_3$ as well.^{37,38} The increase of the surface roughness with oxygen pressure was also observed.^{27,28,35,36} Temperature is another crucial growth parameter. For the MBE of BaTiO_3 on silicon, the optimal window has to be determined; indeed, a compromise has to be found between crystallization of the perovskite phase and interdiffusion at the silicon interface with the SrTiO_3 template. As already pointed out, a systematic study of relevant molecular beam epitaxy parameters on the growth of BaTiO_3 on silicon is missing.

Here, we report the investigation of molecular beam epitaxy processing conditions for the epitaxial growth of BaTiO_3 on SrTiO_3 -buffered silicon. The evolution of the crystalline structure of the films with critical deposition parameters such as temperature, oxygen pressure, or cooling conditions is studied. We discuss the conditions that favor flat surfaces and mainly *c*-axis oriented films. Ferroelectricity is evidenced by piezoresponse force microscopy (PFM) in specific growth conditions, with a polar axis pointing perpendicular to the substrate's plane.

II. EXPERIMENTAL

The epitaxial stacks were grown by MBE using Sr, Ba effusion cells. The flux of each cell was carefully measured and adjusted prior to the growth. For Ti, we used either an e-beam gun or an effusion cell. The 4-nm-thick epitaxial SrTiO_3 buffer layer was grown on HF-cleaned p-type Si (001) substrate. After the HF clean, the substrate was exposed to ultraviolet/ozone for 3 min to remove organic contaminants. Once in the MBE chamber, the substrate was heated to 500 °C and exposed to Sr flux (2 min) and then heated up to 780 °C in order to remove the SiO_2 native oxide and form a 2×1 surface reconstruction corresponding to $1/2$ monolayer (ML) of Sr. Such a passivation allows the direct epitaxial growth of SrTiO_3 on Si. The SrTiO_3 growth was performed at a temperature of 400 °C under a reduced oxygen partial pressure of $P(\text{O}_2) = 5 \times 10^{-8}$ Torr in order to limit the oxidation of the Si surface. The film was then fully crystallized at 460 °C for typically 20 min under ultra high vacuum. The subsequent growth of BaTiO_3 was realized by co-evaporating Ba and Ti at a temperature ranging typically from 440 °C to 525 °C and an oxygen partial pressure of typically 1 to 5×10^{-7} Torr. The temperature, oxygen pressure, and cooling down conditions were varied to optimize the growth conditions. An oxygen plasma source could be used for post-deposition anneal of the films.

A 30 kV reflection high-energy electron diffraction (RHEED) system was used to *in situ* monitor the film surface during the growth. X-ray diffraction was performed using a high-resolution four-circle smartlab diffractometer from Rigaku. θ - 2θ and grazing incident scans were measured. The out-of-plane and in-plane lattice parameters were determined from the 002 and 200 diffraction peaks, respectively (Si 004 and 220 peak positions were used respectively as internal references). X-ray reflectometry (XRR) was performed for thickness determination. High-resolution transmission

electron microscopy (HRTEM) was carried out on an aberration-corrected Hitachi HF3300S microscope (I2TEM-Toulouse) on cross-section samples prepared by mechanical tripod polishing. Atomic force microscope (AFM) images were recorded in tapping mode using Veeco Dimension 3100 microscope with a Nanoscope V controller. Vertical PFM was performed on an NT-MDT Ntegra AFM using single-frequency. A DC voltage of ± 7 V was applied to the sample while scanning over the surface to write domains. No top electrodes were used. The doped p-type Si substrate ($\sim 8 \times 10^{15} \text{ cm}^{-3}$) served as a back electrode. To read the written patterns, a small AC voltage was applied. The operating frequency was chosen to be close to the cantilever resonance in order to enhance the signal to noise ratio.

III. RESULTS

A. Effect of oxygen pressure

To study the impact of oxygen pressure on the crystalline quality, BaTiO_3 films (~ 16 – 18 nm) were prepared at 450 °C under an oxygen pressure ranging from 1×10^{-7} to 3×10^{-6} Torr. This set of films was slowly cooled down (5 °C/min) under a high oxygen partial pressure $P(\text{O}_2)$ of 10^{-5} Torr (with a total cooling time of about 55 min). As can be inferred from RHEED and AFM measurements (Fig. 1),

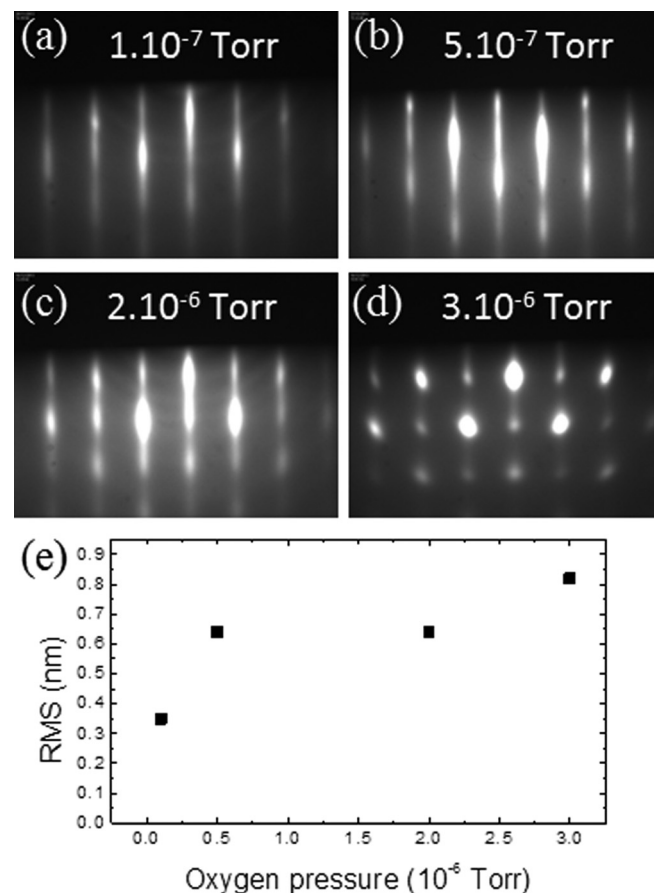


FIG. 1. (a)–(d) RHEED patterns recorded along the [100] azimuth during BaTiO_3 growth at 450 °C under an oxygen pressure of (a) 1×10^{-7} Torr, (b) 5×10^{-7} Torr, (c) 2×10^{-6} Torr, (d) 3×10^{-6} Torr and (e) root mean square roughness (RMS) of the film surfaces measured by AFM as a function of oxygen pressure during BaTiO_3 growth.

the sample surface becomes rougher with increasing oxygen pressure. Until 2×10^{-6} Torr of oxygen, the RHEED patterns present well-contrasted streaky lines indicating a flat and single-crystalline surface. Then, when the oxygen pressure gets higher, the streaky lines tend to become spotty which characterizes a rougher surface. The AFM images confirm this behavior as the root mean square roughness (RMS) increases with oxygen pressure (Fig. 1(e): RMS = 0.35 nm at 1×10^{-7} Torr, RMS = 0.82 nm at 3×10^{-6} Torr). The thickness of the films was determined from XRR measurements. A three-layer model was used to describe the stacks: an interfacial SiO₂ layer, SrTiO₃ buffer layer, and BaTiO₃ layer. The density, rugosity, and thickness of each layer were left as free fitting parameters. A typical XRR scan is shown in Fig. 2(a). In this example, the density of both SrTiO₃ and BaTiO₃ films is found to be close to their bulk values and their roughness is 0.5 nm. A relative density of 1.5 is found for the SiO₂ layer. The thicknesses are 16.4 nm for BaTiO₃, 4.0 nm for SrTiO₃, and 2.7 nm for SiO₂.

X-ray diffraction shows that the films are single crystalline with no parasitic phase, as shown in Fig. 2(b); the epitaxial relationship between BaTiO₃ and Si is as follows: $[110]_{\text{BaTiO}_3} // [100]_{\text{Si}}$ and $(001)_{\text{BaTiO}_3} // (001)_{\text{Si}}$ (as determined from ϕ -scans). The intensities of the in-plane and out-of-plane diffraction peaks are weaker for the samples grown at pressures larger than 5×10^{-7} Torr (Figs. 3(a) and 3(b)), indicating a degraded epitaxial crystalline quality. In the following, the average in-plane and out-of-plane lattice

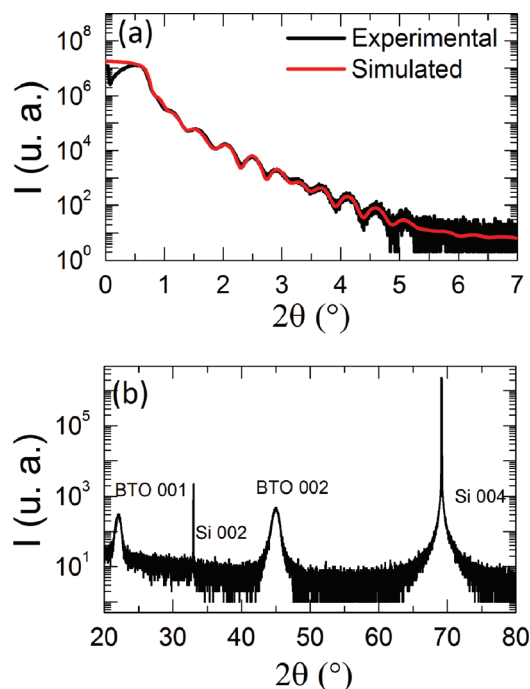


FIG. 2. (a) X-ray reflectometry scan on a BaTiO₃/SrTiO₃ stack deposited on Si at 450 °C under an oxygen pressure of 1×10^{-7} Torr. A SiO₂ interfacial layer is formed between Si and SrTiO₃ during the heterostructure growth. A three-layer model was used with the interfacial SiO₂ layer, SrTiO₃ buffer layer, and BaTiO₃ layer. The density, rugosity, and thickness of each layer were left as free fitting parameters. The density of both SrTiO₃ and BaTiO₃ films is found to be close to their bulk values and their roughness is 0.5 nm. A relative density of 1.5 is found for the SiO₂ layer. The thicknesses are 16.4 nm for BaTiO₃, 4.0 nm for SrTiO₃, and 2.7 nm for SiO₂. (b) $\theta/2\theta$ X-ray diffraction on a BaTiO₃/SrTiO₃ stack deposited on Si.

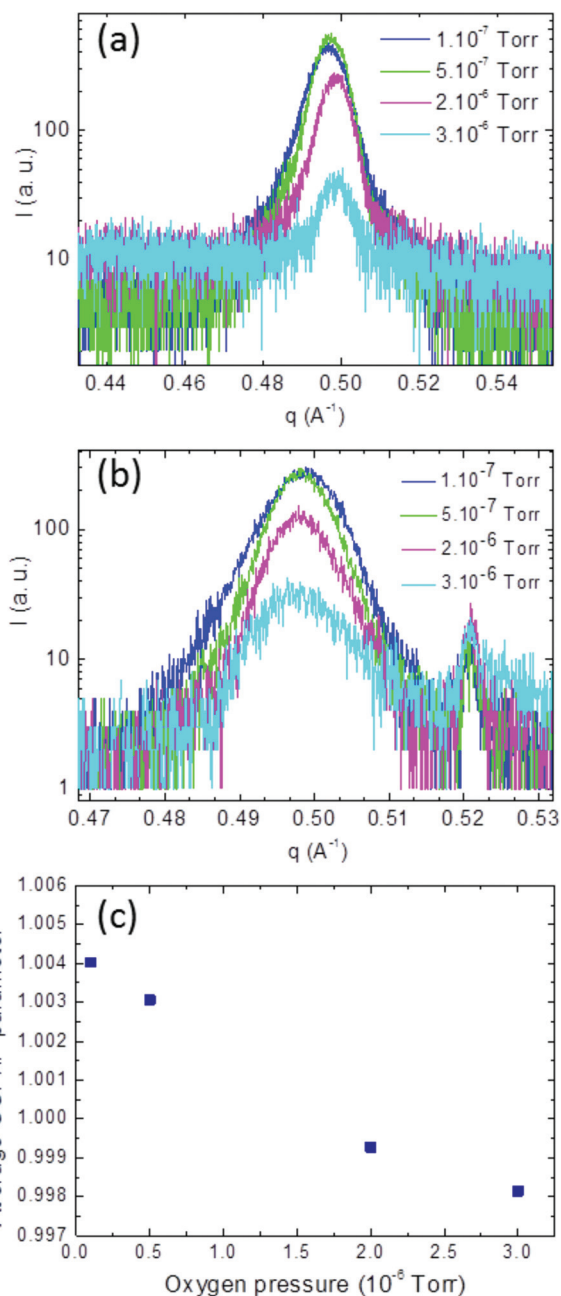


FIG. 3. X-ray diffraction scans of BaTiO₃/SrTiO₃ stack deposited on Si (a) BaTiO₃ 002 peak and (b) BaTiO₃ 200 peak, (c) ratio of average out-of-plane parameter/average in-plane parameter as a function of oxygen pressure during BaTiO₃ growth.

parameters were determined from the maximum of the 200 and 002 diffraction peaks, respectively. Although some of the 16–18 nm-thick films consist in a mixture of *c*- and *a*-domains, we did not determine the individual contributions due to the limited resolution. We observe that the out-of-plane lattice parameter progressively decreases when the oxygen pressure is raised, while the in-plane parameter increases. At 5×10^{-7} Torr, the in-plane average parameter is 4.015 Å and the out-of-plane average parameter is 4.024 Å. These values reflect the fact that there is a mixture of *c*- and *a*-domains. For thinner films of typically 7 nm grown in the same conditions, which are fully *c*-axis oriented, the lattice parameters are: $a = 3.996$ Å and

$c = 4.027 \text{ \AA}$. These values are close to the bulk parameters, indicating that the films are relaxed.

The ratio of the average parameters is plotted as a function of oxygen pressure in Fig. 3(c). Starting from $2 \times 10^{-6} \text{ Torr}$ and above, the out-of-plane/in-plane ratio becomes lower than 1, meaning preferentially a -oriented films for samples grown at high oxygen pressure, whereas below $2 \times 10^{-6} \text{ Torr}$, BaTiO_3 films are mainly c -oriented (ratio > 1). There is thus a strong effect of the oxygen pressure on the crystalline orientation of the films, which will be discussed in Sec. IV.

B. Effect of temperature

BaTiO_3 films ($\sim 18\text{--}20 \text{ nm}$) were grown on SrTiO_3/Si under $1 \times 10^{-7} \text{ Torr}$ of oxygen at different temperatures ranging from 410°C to 580°C . This set of films was cooled down rapidly ($25^\circ\text{C}/\text{min}$) under vacuum ($5 \times 10^{-8} \text{ Torr}$) until 180°C (the total cooling time was about 10 min); the

temperature was then set at 180°C and the oxygen plasma (400 W) was switched on for 40 min ($P(\text{O}_2) = 1 \times 10^{-5} \text{ Torr}$). The RHEED patterns displayed in Figs. 4(a)–4(f) exhibit rings when the temperature exceeds 480°C . This could denote a polycrystalline surface but all the samples, except the one grown at 580°C , are found to be single crystalline by X-ray diffraction. Particulates are observed on the surface by AFM and additional spots are clearly seen on the RHEED patterns for films thicker than 20 nm. Off-stoichiometry might be induced by the temperature, leading to the segregation of Ba on the surface. A Ba-rich-surface oxide layer has been reported for MBE-grown BaTiO_3 films on SrTiO_3 substrates.³⁹ No crystalline growth happens at 580°C . At this temperature, interface reactions occur and impede the epitaxy. At lower temperature (down to 440°C), streak lines are observed and their intensity increases with temperature. At 410°C , the lines are weak which is related to the fact that the temperature is not high enough to achieve full crystallization of BaTiO_3 . AFM images displayed in Figs. 4(g)–4(i) show

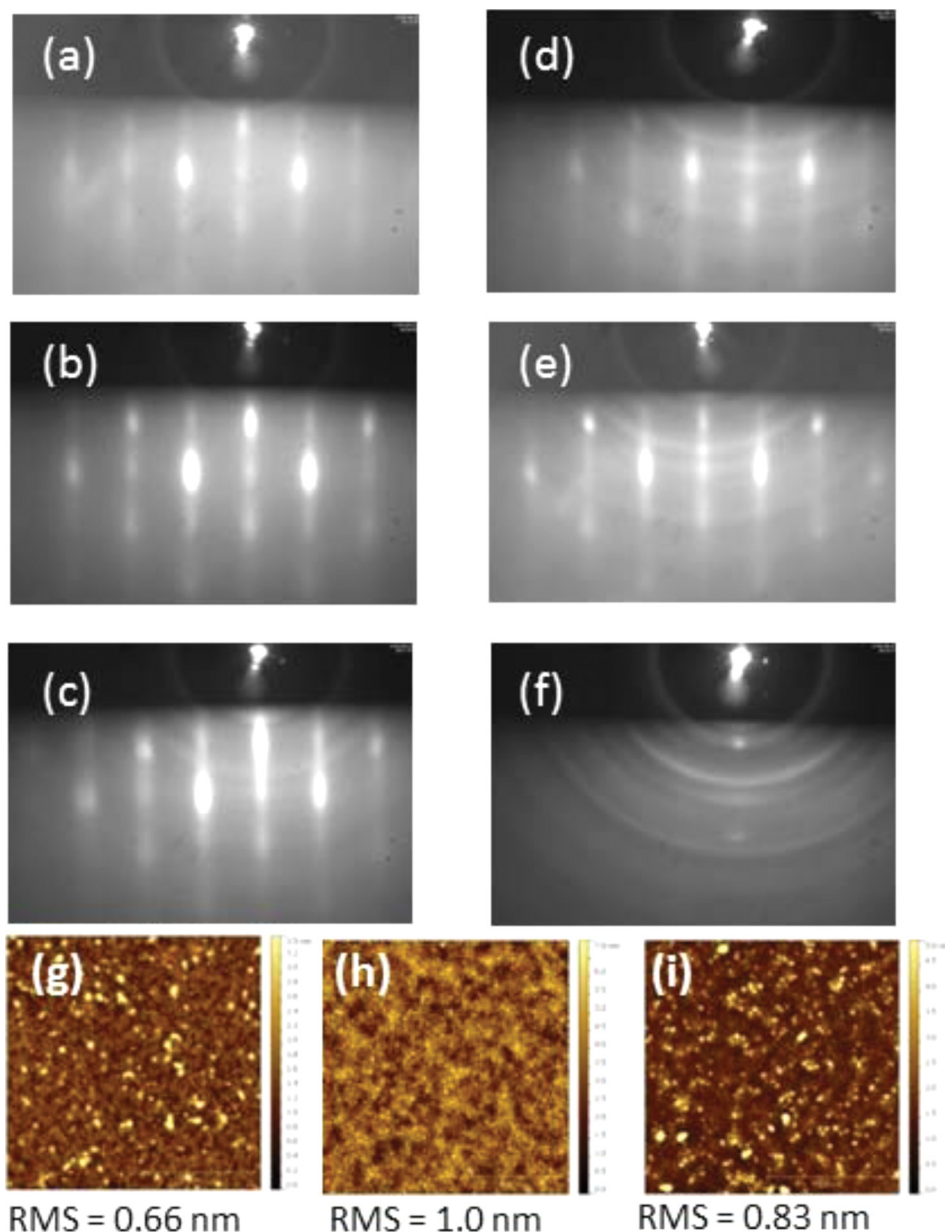


FIG. 4. (a)–(f) RHEED patterns recorded along the $[100]$ azimuth of BaTiO_3 grown under an oxygen pressure of $1 \times 10^{-7} \text{ Torr}$ at (a) 410°C , (b) 440°C , (c) 480°C , (d) 500°C , (e) 525°C , (f) 580°C . (g)–(i) AFM images ($1 \mu\text{m} \times 1 \mu\text{m}$) of the films elaborated at (g) 410°C , (h) 480°C , and (i) 525°C .

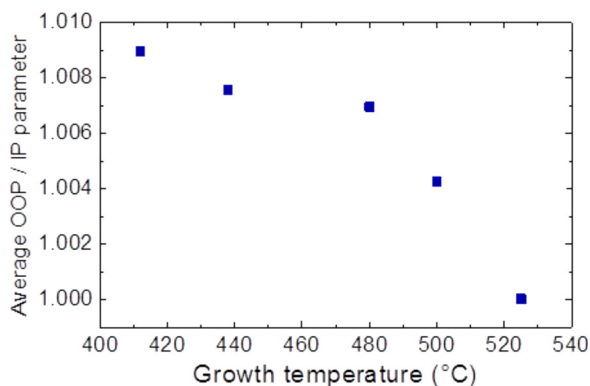


FIG. 5. Ratio of average out-of-plane parameter/average in-plane parameter as a function of growth temperature for BaTiO₃/SrTiO₃ deposited under P(O₂) = 1 × 10⁻⁷ Torr.

homogeneous surfaces with RMS ranging from 0.7 to 1.0 nm for the different growth temperatures. The average out-of-plane and in-plane parameters were determined by X-ray diffraction and their ratio is plotted in Fig. 5. The ~18–20 nm-thick films are composed of a mixture of *c*- and *a*-axis crystallites. At 525 °C, films have similar in-plane and out-of-plane lattice parameters of 4.01–4.02 Å. These values are comparable to those determined locally by geometrical phase analyses in 16 nm films with mixed *c*- and *a*-domains.¹³ As the temperature increases, we observe a decrease of the tetragonal ratio, which may indicate an increase of the *a*-axis crystallite content and/or a change in the crystalline structure parameters.

C. Impact of growth conditions on the interface

During the SrTiO₃ and BaTiO₃ growth and anneals, oxygen diffusion occurs through the film leading to the formation of an interfacial amorphous SiO₂ layer between the Si substrate and the SrTiO₃ buffer layer, but preserving the original epitaxial relationship. In view of integrating such materials in negative-capacitance FETs, the low permittivity interfacial layer should be suppressed. It is therefore desirable to find optimal conditions to minimize it. After the

growth of 18–20 nm-thick BaTiO₃ films at 440–450 °C under P(O₂) = 1 × 10⁻⁷ Torr, two different post-deposition processes were tested. The first one (process 1) is a slow cooling down (10 °C/min) under high oxygen pressure (1 × 10⁻⁵ Torr) and the second one (process 2) is a quick cooling down (25 °C/min) under vacuum (5 × 10⁻⁸ Torr) followed by a plasma oxidation under 1 × 10⁻⁵ Torr of atomic oxygen. HRTEM images shown in Figs. 6(a) and 6(b) indicate that a slow cooling down favors the formation of an amorphous SiO₂ interfacial layer with an average thickness of ~2.5–3.0 nm, while a ~1.0 nm SiO₂ thickness is observed for the other procedure (process 2). The effect of growth temperature was also investigated with a BaTiO₃ film grown at 525 °C under P(O₂) = 1 × 10⁻⁷ Torr and cooled according to process 2. As can be seen in Fig. 6(c), a thicker interfacial layer is obtained at 525 °C (SiO₂ ~1.7 nm versus ~1.0 nm at 440 °C), which is expected as a higher temperature favors oxygen diffusion. In all investigated films, an abrupt interface between SrTiO₃ and BaTiO₃ was observed.

D. Ferroelectricity characterization

Ferroelectricity was investigated by vertical PFM. Three square patterns were successively written: an outer square of 7 × 7 μm² was poled using a bias of -7 V, then a 4.5 × 4.5 μm² square was poled by reversing the bias to +7 V, and finally an inner 2 × 2 μm² pattern was poled back with -7 V. Hysteresis loops were recorded in spectroscopic mode (Vac = 0.5 V). We report in Fig. 7 the images and hysteresis loops obtained for a sample grown at 525 °C and with an oxygen pressure of 1 × 10⁻⁷ Torr. Domains are written and switched; the amplitude between opposite polarization is similar and a 180° phase difference is observed. We did not observe topographic changes on the film surface. Similar images were obtained for samples grown at 450 °C with an oxygen pressure of 5 × 10⁻⁷ Torr with mainly *c*-axis orientation. Our results are similar to previous results reported for MBE-grown BaTiO₃ on SrTiO₃-templated Si substrates.^{13,14,40} The domains are stable over 24 h for

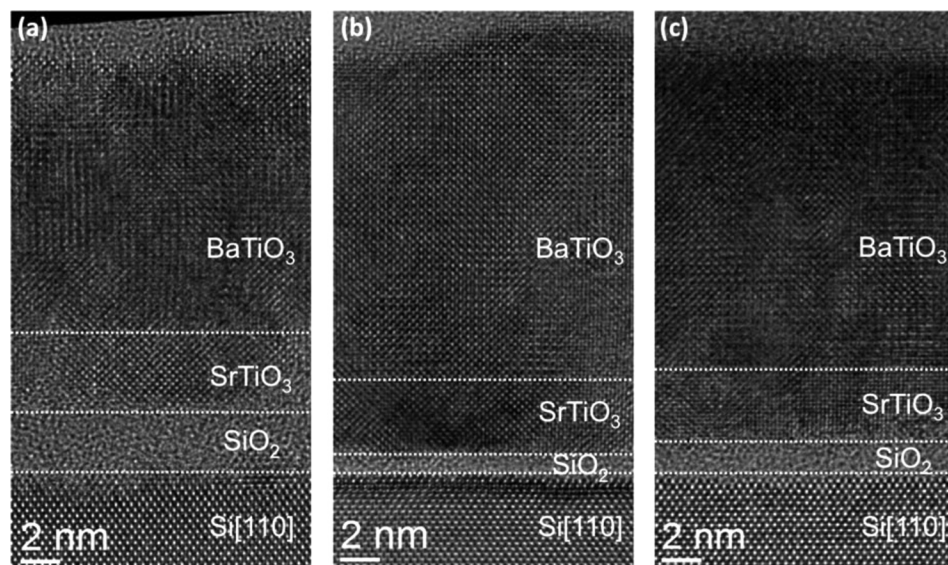


FIG. 6. High resolution transmission electron microscopy images of BaTiO₃/SrTiO₃ stacks grown under P(O₂) = 1 × 10⁻⁷ Torr for different temperatures and post-deposition process. (a) 450 °C—slow cooling down procedure at P(O₂) = 1 × 10⁻⁵ Torr, (b) 440 °C—rapid cooling down under UHV followed by annealing under an oxygen plasma (1 × 10⁻⁵ Torr) for 40 min, (c) 525 °C—rapid cooling down under UHV followed by annealing under an oxygen plasma (1 × 10⁻⁵ Torr) for 40 min. A SiO₂ interfacial layer between Si and SrTiO₃ is formed upon SrTiO₃ annealing and BaTiO₃ growth and its thickness depends on the cooling down conditions. Horizontal dotted lines are only to guide the eyes.

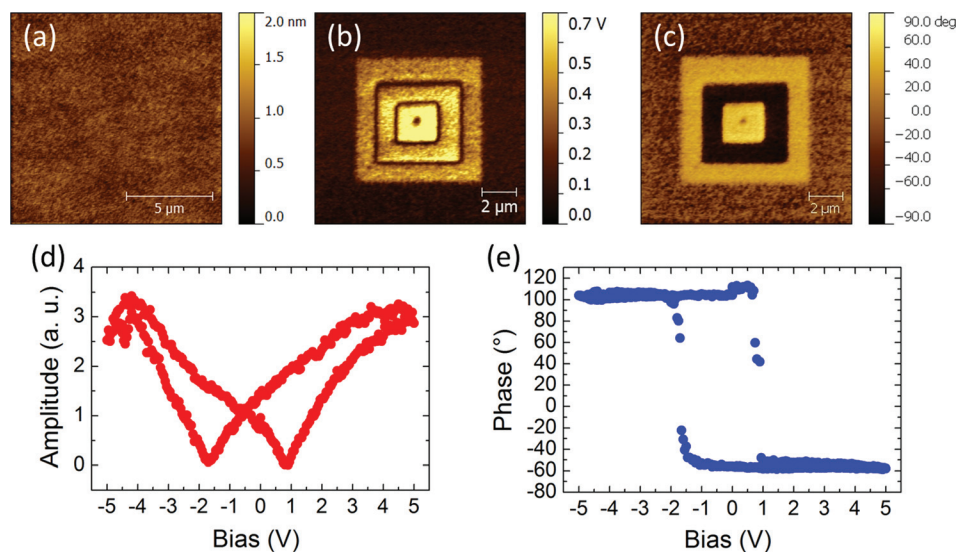


FIG. 7. (a) Topography, (b) PFM amplitude, and (c) PFM phase for a 18 nm-thick BaTiO₃ film grown at $P(\text{O}_2) = 1 \times 10^{-7}$ Torr poled with $-7 \text{ V}/+7 \text{ V}/-7 \text{ V}$ —(d) amplitude and (e) phase of PFM spectroscopy on the same stack ($V_{\text{ac}} = 0.5 \text{ V}$).

optimized films. In some cases, the contrast decreases after several hours, which can be attributed to relaxation phenomena. Domain instability might also originate from the depolarization field that is present in thin ferroelectric films on oxide or semiconductor substrates and from the fact that the domains poled by the tip might not extend throughout the film thickness. Hysteresis loops were also performed locally on the sample surface and show an abrupt hysteresis for the phase and a typical butterfly shape for the amplitude (Figs. 7(d) and 7(e)). No stable domain could be written in the mainly *a*-axis oriented films grown at oxygen pressures of 2×10^{-6} Torr or above.

IV. DISCUSSION

A. Morphology and crystalline structure

Both oxygen pressure and temperature are shown to strongly impact the morphology, crystalline orientation, and lattice parameters of the films. One origin may be the change in the diffusion of species at the sample surface with these two parameters, which may affect the morphology, local cationic composition, the oxygen content, and/or strain state. This could, in turn, influence the formation of the crystalline domain structure upon cooling down the sample.

We report a progressive decrease of the out-of plane lattice parameter when the oxygen pressure is raised while the in-plane parameter increases. The ratio of the out-of-plane parameter/in-plane parameter becomes lower than 1 at 2×10^{-6} Torr, indicating that *a*-axis growth is predominant. We also observe an increase of the surface roughness with increasing oxygen pressure as shown by AFM and RHEED. The same effects have been reported for BaTiO₃ films grown by laser MBE,^{24,25} PLD,^{28,29,34–36} or sputtering.²⁷ Regarding surface morphology, Zhang *et al.* noted a change of the topography from flat surface at 5×10^{-3} Torr to hilly surface at 0.5 Torr for BaTiO₃ films grown by PLD on SrTiO₃ substrate.²⁹ This phenomenon was also observed by Chen *et al.* for films grown on SrRuO₃/SrTiO₃: the microstructure varies from a dense large grained structure with a smooth surface to a small columnar grain structure with rough surface as the

deposition pressure increases from 5 to 200 mTorr.³⁶ To our knowledge, there is no similar study reported for BaTiO₃ films grown by MBE. We explain the morphology trend with the fact that the energy of evaporated species is lower with increasing oxygen pressure because of scattering and multiple collisions between the evaporated species and oxygen molecules. Thus, the mobility of the adatoms at the surface of the substrate becomes too low to move down to lower level at the edge of islands and surface flatness is deteriorated.

The oxygen pressure also affects strongly the crystalline orientation of the films. For example, Zhao *et al.* observed a dependence of the crystallographic orientation on the oxygen pressure in the range of 1.5×10^{-6} to 10^{-1} Torr for films grown on SrTiO₃ by laser MBE.²⁴ It is interesting to note that completely different deposition processes involving different mechanisms and energy range for the impinging species at the substrate surface show a similar trend with oxygen pressure: an increasing oxygen pressure leads to a change from *c*-axis to *a*-axis growth (although the pressure ranges are necessarily different depending on the techniques). Oxygen pressure is expected to impact the occurrence of oxygen vacancies, the cationic composition, the nature of the defects, and the resulting strain state.

Decreasing the oxygen pressure during the growth should favor the formation of oxygen vacancies V_{O}^{2+} , which cause an increase of the *c* lattice parameter.²⁴ Since our films are *in situ* annealed at high $P(\text{O}_2)$ during cooling down, the oxygen pressure during growth should not affect the final oxygen content of the films. This point should be further investigated in our heterostructures; in particular, the impact of the oxygen pressure during BaTiO₃ growth on the SrTiO₃ underlying template should be explored. Indeed, it has been shown that the BaTiO₃ perovskite phase could be formed by MBE from the coevaporation of metallic Ba and Sr (from effusion cells) in ultrahigh vacuum conditions, without introducing oxygen.⁴¹ Shimoyama *et al.* showed that oxygen was fed from the SrTiO₃ substrates to the growing film surface and could evidence the incorporation of oxygen vacancies deep into the SrTiO₃ substrate.⁴¹ Moreover,

the interface between BaTiO₃ film and SrTiO₃ substrate remained sharp despite the amount of oxygen vacancies having moved through the interface. Recently, Rutkowski *et al.* reported an optical study of oxygen vacancy formation at SrTiO₃/(Ba,Sr)TiO₃ heterostructures grown by MBE (oxygen plasma pressure of 5×10^{-7} Torr).²⁶ They showed that defects accumulate at the interface and that this accumulation depends on the growth conditions, particularly P(O₂). It is therefore possible that oxygen from the SrTiO₃ template moves towards the BaTiO₃ film surface during the growth of our heterostructures. Defects in the BaTiO₃ film have also been shown to depend on the oxygen pressure. Zhu *et al.*²⁵ observed a change in the type of predominant defects in laser-MBE grown BaTiO₃ on SrTiO₃ by varying the oxygen pressure in the range of 1.5×10^{-7} to 1.5×10^{-4} Torr: with the lowering of oxygen pressure, the major defects in the films change from threading dislocations into (111) nanotwins. Advanced TEM studies will be performed to determine whether the underlying SrTiO₃ template as well as the BaTiO₃ film exhibit a different defect structure depending on the P(O₂) conditions.

Another strong impact of the oxygen pressure during BaTiO₃ growth is a possible change in the cationic ratio Ti/Ba, which has been reported for PLD-grown films.^{30,32,36} Chen *et al.* reported an increasing Ti/Ba ratio with increasing oxygen pressure for films on SrTiO₃ substrates.³⁶ The films change from Ti-deficient to Ti-rich with increasing the oxygen pressure. We can suspect that compositional changes occur in our films with P(O₂) (this point will be investigated in the future by EELS/EDX in a transmission electron microscope). In addition to changing the diffusion of the species on the substrate's surface, an additional effect may arise from oxygen pressure in MBE. Although MBE should enable a straightforward control of the composition due to the independent control of each metal source, the presence of oxygen in the chamber makes such a control challenging. Indeed, oxygen induces surface oxidation of the hot metal charges in the effusion cells as well as oxidation of the evaporated species. As a result, the oxygen pressure used during the film growth may modify the stoichiometry.

We also think that the crystalline changes we observed with increasing the growth temperature are related to cationic modifications.

Change in the cationic ratio is expected to modify the lattice parameter of the films. Off stoichiometries will indeed lead to structural buckling and defects. In a detailed study of the MBE growth of LaCrO₃, it was shown that the crystalline quality, film surface morphology, and defect density are sensitive to the La-to-Cr cationic ratio.⁴² Many perovskite oxides can accommodate cationic composition deviation while keeping a perovskite structure; the physical properties are, however, strongly impacted.^{39,42–45} We did not observe change in the RHEED patterns for the different films. However, % level change in the cationic ratio cannot be precluded. In Ref. 39, it is reported that changes in composition of at least $\pm 20\%$ are needed to observe noticeable changes in the BaTiO₃ RHEED patterns for films deposited on bulk SrTiO₃ substrates. The control of the cationic composition during the growth process remains one of the major

challenges for the molecular beam epitaxy of complex oxides.

Finally, regarding oxygen pressure, one should note that the working conditions reported in the literature for the epitaxial growth of BaTiO₃ on SrTiO₃-buffered Si by MBE span an extremely large range, of about three orders of magnitude: $4\text{--}5 \times 10^{-8}$ Torr of molecular oxygen in Ref. 7, 1 to 5×10^{-7} Torr of molecular oxygen in this work, 5×10^{-6} Torr of molecular oxygen in Ref. 13, and 3×10^{-5} Torr under rf oxygen plasma in Ref. 4. It is not yet understood why different equipments/processes result in such a difference in the optimal oxygen pressure.

B. Ferroelectricity

Ferroelectricity in epitaxial BaTiO₃ deposited on silicon without a conducting bottom electrode cannot be evidenced by conventional C-V electrical measurements. Indeed, in such capacitive structures, the voltage applied across the heterostructure is mainly dropped in the low permittivity interfacial SiO₂ layer (of typically 20–30 Å). Moreover, the silicon in depletion also contributes to the total capacitance. Hence, it is not possible to reach an effective electrical field that can switch the ferroelectric layer. Probes in which the electrical field applied locally is large enough to switch the ferroelectric are required to assert ferroelectricity.

For our optimized films with mainly *c*-axis orientation or grown at higher temperature (525 °C), we observe stable domain written by PFM and hysteresis loops, that are typical for ferroelectric materials. Since few years, it has been pointed out that ionic and electrochemical phenomena may play a major role in scanning probe microscopy.⁴⁶ It has been shown that domain writing/switching as well as hysteresis loops can be obtained in non-ferroelectric compounds such as crystalline LaAlO₃/SrTiO₃ heterostructures,⁴⁷ amorphous LaAlO₃ (Ref. 48), or transition metal oxides involved in memristive devices like TiO₂ or SrTiO₃.⁴⁹ Oxygen vacancies may play a major role in the electromechanical response. In epitaxial LaAlO₃/SrTiO₃ heterostructures, it was suggested that the oxygen vacancy distribution throughout the LaAlO₃ thickness is reversibly modified, leading to the switchable electromechanical response. In this system, the PFM signal of the written domain is found to decay in tens of minutes. Stable domains were written in amorphous 3 nm LaAlO₃ films deposited on Si by MBE.⁴⁸ However, in this study, the LaAlO₃ film was deposited from the evaporation of a LaAlO₃ pellet without additional oxygen pressure in the first stage of the growth (~ 1 nm) in order to limit interface reactions with Si, which promotes a high density of oxygen vacancies in the film. In our study, oxygen is provided throughout the whole growth process of the SrTiO₃/BaTiO₃ heterostructure; moreover, the films were annealed *in situ* after deposition under oxygen. We thus exclude that the main contribution to the electromechanical response is related to oxygen vacancies in our films. The PFM results, together with the crystalline characterization (*c*-axis oriented tetragonal BaTiO₃ phase), show that the BaTiO₃ films are ferroelectric with a polarization that can be switched between up/down states.

V. CONCLUSION

In this work, different deposition conditions were investigated to optimize the epitaxial *c*-axis growth of 16–20 nm-thick tetragonal BaTiO₃ films on silicon by MBE. Low oxygen pressures (in the range of 1×10^{-7} to 5×10^{-7} Torr) are favorable for an out-of-plane *c*-axis growth together with smooth films and high epitaxial crystalline quality. A limited window is available for the growth temperature (440–525 °C) of BaTiO₃ in these MBE conditions. Although a high crystalline quality of this refractory material requires a high temperature, a compromise has to be found with interfacial reactivity. Lower temperatures (~450 °C) and quick cooling down with a post-deposition oxygen plasma anneal are also prone to limit the regrowth of the unfavorable low permittivity SiO₂ layer at the Si/SrTiO₃ interface. A challenging aspect of the epitaxial growth of BaTiO₃ by MBE is the control of the cationic composition in the first stages of the growth, which may be affected by the processing conditions (such as oxygen pressure and temperature) and the control of its stability with time. Finally, the ferroelectricity of 16–20 nm thick films was evidenced by PFM measurements coupled to the structural characterization showing *c*-axis oriented tetragonal crystalline structure. The critical thickness allowing ferroelectricity to be stabilized on such SrTiO₃-Si substrates will be investigated, which is relevant for integrating such ferroelectric films in field-effect devices.

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