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Carbon nanotube/alumina and graphite/alumina composite coatings on stainless steel for tribological applications

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ABSTRACT

Carbon/alumina coatings on stainless steel are prepared by a sol-gel route, using either carbon nanotubes (8 walls on average) or graphite flakes. The friction coefficient against a steel ball is decreased by a factor of 4–5 compared to pure alumina and wear is reduced by a factor of 2 with graphite flakes. A Raman spectroscopy study of selected specimens outside and inside the worn surface shows that the carbon nanotubes are not dramatically damaged whereas the graphite flakes are broken into graphene layers. The reasons why graphite is more effective than the carbon nanotubes, for the same carbon content, to improve the tribological behavior are discussed.

Keywords: Carbon nanotubes Graphite Alumina Sol gel coatings Friction Wear

1. Introduction

Austenitic stainless steels are widely used in the aerospace, energy, medical and food industries because of their resistance to corrosion but they are highly susceptible to adhesive wear (seizure), therefore leading to a significant loss of profitability and possibly to some environmental impact. Carbon-containing composites are of particular interest as self-lubricating materials showing a high resistance to friction and wear, preventing the need for liquid lubricants. Reports on the tribological behavior of metal-[1–11] and ceramic-matrix [12–17] bulk composites containing carbon nanotubes (CNTs) are increasingly abundant. However, the comparison of the results reported by different groups is hampered notably because different CNTs are used, the preparation routes markedly differ and the tribological testing conditions (counterface, load, sliding distance, relative humidity, temperature) vary widely. Moreover, it is either undesirable or impossible to use bulk composites for applications such as the protection of austenitic steels. Therefore, composite coatings are to be preferred, such as CNT-metal electroless coatings [3] and plasma-sprayed CNT/Al₂O₃ coatings [18,19]. Balani et al. [18] reported that the sliding wear volume loss of a 8 wt.% CNT/Al₂O₃ coating against a ZrO₂ pin (dry conditions, normal load 48 N) was 49 times lower than for an Al₂O₃ coating. Keshri et al. [19] reported a 72% increase in wear resistance against a WC ball (298 K, ball-on-disk tribometer) for a 8 wt.% CNT/Al₂O₃ coating compared to Al₂O₃. Note that these results are partly attributed to indirect effects of the presence of CNTs, such as a locally enhanced densification of the wear surface and a higher toughness of the coating through CNTs bridging between the splats and/or Al₂O₃ grains. Graphite/Al₂O₃ bulk composites were shown [20] to have a friction coefficient half of that of pure Al₂O₃. A low shear strength in the sliding direction and high compression strength in the direction of the load (i.e. perpendicular to the sliding direction) are beneficial to lower the friction coefficient. Laminated graphite/Al₂O₃ composites were reported [21] to show significantly better friction and wear behaviors than monolithic graphite/Al₂O₃ composites (against an Al₂O₃ ball, dry conditions room temperature) because graphite particles can be easily dragged on to the friction surface from the graphite layers. Changes in the layer spacing and in the graphite volume fraction affect the formation of lubricating and transferring films, loadbearing capacities and wear mechanisms of the materials. The aims of the present work are to shape CNT/Al₂O₃ coatings onto an austenitic 304-L stainless steel substrate, to compare their tribological behavior to that of pure Al₂O₃ and graphite/Al₂O₃ coatings

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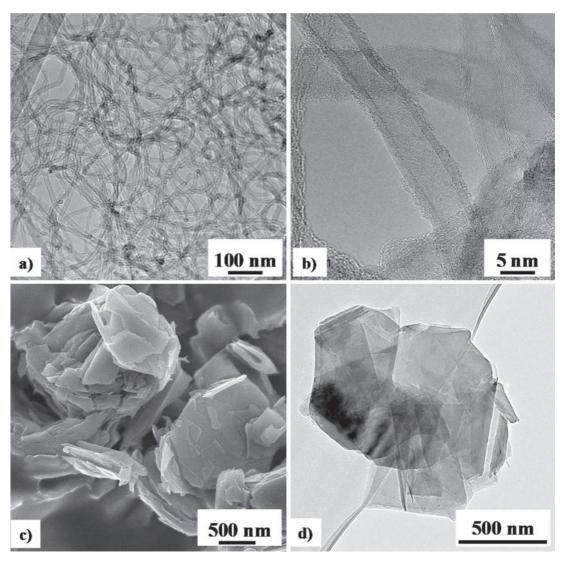


Fig. 1. TEM images of the CNTs (a and b) and FESEM image (c) and TEM image (d) of graphite flakes.

and to explain why graphite is more effective than CNTs to improve the tribological behavior.

2. Experimental methods

2.1. Raw materials

A CNT sample described in detail elsewhere [10] was purchased from Nanocyl (Belgium). TEM images (Fig. 1a and b) show that the CNTs are mostly not bundled and contain a fair number of defects along their lengths. CNTs with 3–22 walls are observed. CNTs with 8 walls are dominant (30%) with CNTs with 7 and 9 walls (both 16%) the second most abundant populations.

The average number of walls is equal to 8.5 (rounded to 8). The average CNT outer diameter is 10.2 nm and length is below 1.5 μm . The specific surface area of the sample is $242\,m^2\,g^{-1}$, in agreement with calculations from geometrical data [22]. The CNTs were carboxyl-functionalized with a nitric acid solution (3 mol L $^{-1}$) as described elsewhere [23]. Graphite platelets (Fig. 1c and d) were purchased from Abcr (Germany). Flakes about 15 nm thick are agglomerated into platelets about 1.5 μm in size. The specific surface area of the sample is $20\,m^2\,g^{-1}$.

An aqueous aluminum chloride hexahydrate (AlCl $_3\cdot 6H_2O$) solution (0.13 mol L^{-1}) was poured into an excess of ammonia solution

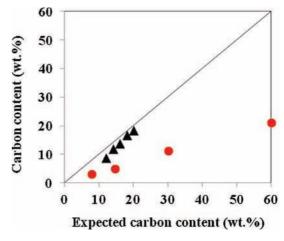


Fig. 2. Measured carbon content versus the expected carbon content for the CNT/Al_2O_3 (\blacktriangle) and graphite/ Al_2O_3 coatings (\bullet). The solid line shows the expected carbon content (i.e. 1 for 1).

(5 mol L⁻¹) under magnetic stirring at room temperature. The soobtained boehmite (AlOOH) precipitate was filtered, washed with deionized water and oven-dried overnight, producing a boehmite powder. A known amount of this powder was dispersed into deionized water and the solution was peptized with a 3.1 vol.% solution of acetic acid (100%) and stirred for 24 h, producing a boehmite colloidal sol. Carbon was added in different proportions (6, 8, 10, 12 and $14\,\mathrm{g\,L^{-1}}$ for CNTs and 5, 10, 25 and $50\,\mathrm{g\,L^{-1}}$ for graphite), as required for the study. The carbon-boehmite sols were sonicated (Vibra Cell 75042 sonotrode), 1 h for CNTs and 0.5 h for graphite.

Stainless steel disks (AISI 304-L, diameter 30 mm, thickness 5 mm, arithmetic average roughness Ra 0.6 µm) were used as substrates. The disks were first pre-treated using a routine involving alkaline degreasing, acid pickling and nitric acid passivation [24].

2.2. Preparation of the coatings

Al $_2$ O $_3$ and carbon-Al $_2$ O $_3$ coatings were prepared by dip-coating followed by thermal treatments [24]. The stainless steel substrates were dipped into the boehmite or carbon-boehmite colloidal sols and withdrawn at a controlled speed (300 mm min $^{-1}$), resulting in the deposition of a sol layer onto the substrate. After removing of the excess liquid, drying in air (80 °C, 2 h) provoked solvent evaporation and transformation of the sol in a xerogel. A heat-treatment in N $_2$ (500 °C, 25 min, heating rate 100 °C h $^{-1}$, natural cooling) then resulted in the formation of the Al $_2$ O $_3$ and carbon-Al $_2$ O $_3$ coatings. The coatings will be referred to hereafter by a letter (T for CNTs or G

for graphite) and a number denoting its measured carbon content (weight%), i.e. T18 or G21.

2.3. Characterization

The viscosity of the sols was measured using a viscometer (LAMY RM100) in a Taylor-Couette configuration at a shear range of 966–2900 s⁻¹. Grazing incidence X-ray diffraction (XRD) was carried out on the substrate and on the coatings using a Siemens D5000 diffractometer with Cu K α X-ray source. XRD patterns were collected at room temperature by 0.02° (20) scanning steps over the 10-100° range. The carbon content in the carbon-Al₂O₃ coatings was obtained using an electron microprobe (EPMA, Cameca SXFive). The surface of the coatings was observed by field-emissiongun scanning electron microscopy (FESEM, JEOL JSM 6700F) and interferential rugosimetry (Zygo NewView 100). The thickness of the coatings was determined on cross-sectional SEM images (SEM JEOL 35CF). Cross sections were prepared by cryo-fracturation on notched samples. The wear tracks were observed by 3D optical profilometry (SENSOFAR S neox) and optical microscopy (Keyence VHX-1000E). Selected samples were studied by Raman spectroscopy (Horiba 800 spectrometer using 633 nm laser excitation). The spectra represent results obtained from three different areas in a given sample.

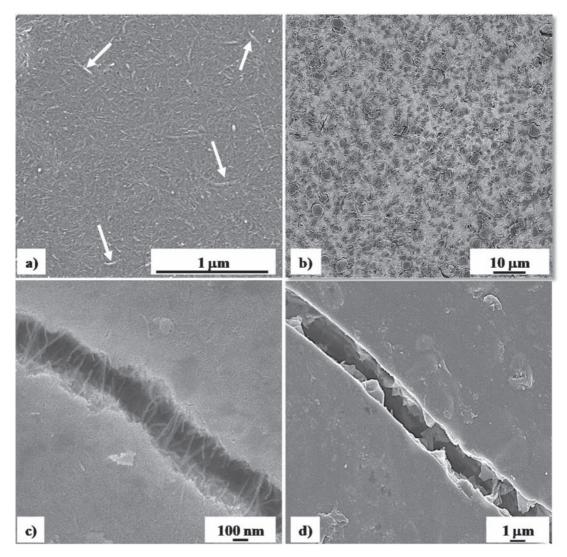


Fig. 3. FESEM images showing the surface of the CNT/Al_2O_3 (18 wt.%) (a) and graphite/ Al_2O_3 (21 wt.%) (b) coatings and higher magnification images showing cracks made on purpose, with bridging CNTs for the CNT/Al_2O_3 (18 wt.%) coating (c) and non-bridging graphite platelets for the graphite/ Al_2O_3 (21 wt.%) coating (d).

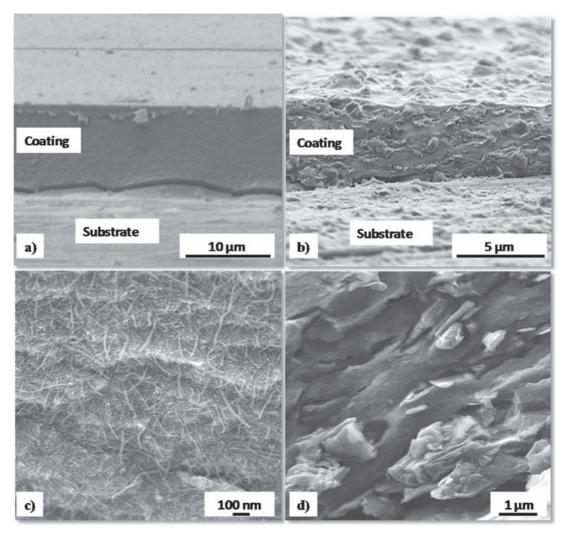


Fig. 4. FESEM images showing cross-sections of the CNT/Al₂O₃ (18 wt.%) coating (a) (white arrows indicate CNTs) and graphite/Al₂O₃ (21 wt.%) (b) coating; higher magnification images show CNTs (c) and graphite platelets (d), respectively.

2.4. Nano-indentation and tribological testing

Instrumented indentation tests were conducted (in air at room temperature, maximum normal load $600~\mu N$, load/unloading rate $800~\mu N$ min $^{-1}$, dwell 60~s, maximum penetration depth 100--300~nm) with a Berkovich nano-indenter (CSM Instrument Ultra Nanoindenter) in order to record the load versus penetration depth and to evaluate the Young's modulus and hardness of the raw (i.e. un-polished) coatings. Friction tests were performed using a ball-on-disk geometry in rotary mode (CSM Tribometer) in compliance to the ASTM G99 international standard. Tests (normal load 2 N, rotating speed $10~cm~s^{-1}$ and total sliding distance 250~m) were performed at room temperature in ambient air with a 40--60% relative humidity. The un-polished surfaces were rubbed against a 316--L steel ball 10~mm in diameter. The frictional force transferred to a load cell was recorded throughout the test. All friction tests were repeated three times, showing identical results.

3. Results and discussion

3.1. Composition and microstructure of the coatings

The viscosity of the sol increases markedly upon the increase in CNT content, from 11 mPa s (pure boehmite sol) to 28 mPa s for the sol with a carbon proportion of $14\,\mathrm{g\,L^{-1}}$. By contrast, the viscosity

of the graphite-boehmite sols remains constant for all the studied carbon concentration range $(5-50\,\mathrm{g\,L^{-1}})$ and is only marginally higher than for pure boehmite $(13\,\mathrm{vs}\,11\,\mathrm{mPa\,s})$. This could reflect the higher aspect ratio (length/diameter) of the 8CNTs (150) compared to that of graphite (1). The different carbon proportions $(6,8,10,12\,\mathrm{and}\,14\,\mathrm{g\,L^{-1}}$ for 8CNTs and $5,10,25\,\mathrm{and}\,50\,\mathrm{g\,L^{-1}}$ for graphite) in the starting sols should yield carbon contents in the coatings equal to 12, 14, 16, 18, 20 wt.% for 8CNTs and 7.9, 14.6, 30, 60 wt.% for graphite. The actual measured carbon content is 7–27% lower than expected for the CNT/Al₂O₃ coatings (Fig. 2), but the discrepancy tends to diminish for higher contents, and is 62–68% lower than expected for the graphite/Al₂O₃ coatings (Fig. 2), indicating that carbon species, in particular the graphite particles, tend to be drained by gravity during the withdrawing step of dip-coating if the viscosity of the sol is too low.

Analysis of the XRD pattern (not shown) for the un-coated 304-L steel substrate reveals the austenite (111) and (200) peaks (major phase) and a weak ferritic or martensitic peak (minor phase). For the Al_2O_3 coating, the steel peaks are less intense and no peaks corresponding to crystallized Al_2O_3 are detected. For the carbon-Al $_2\text{O}_3$ coatings, very weak steel peaks are detected in addition to a strong CNT- or graphite (002) peak. Again, no peaks corresponding to Al $_2\text{O}_3$ are detected, showing that Al $_2\text{O}_3$ here is either amorphous or made up of extremely small crystallites.

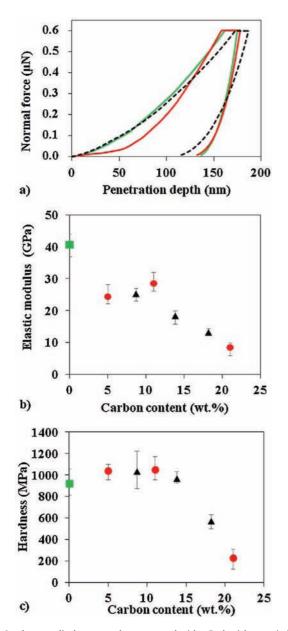


Fig. 5. Load versus displacement plots measured with a Berkovich nano-indenter (a), Young's modulus (b) and hardness (c) versus carbon content.

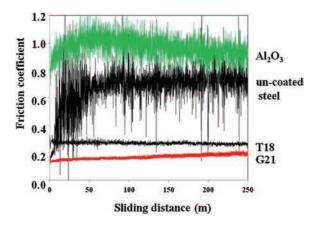


Fig. 6. Friction coefficient against a steel ball versus the sliding distance for the uncoated substrate and the Al_2O_3 , CNT/Al_2O_3 (18 wt.%) and graphite/ Al_2O_3 (21 wt.%) coatings. The test load is equal to 2 N.

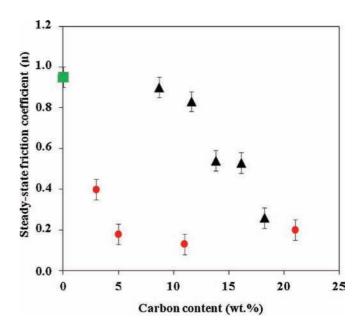


Fig. 7. Steady-state friction coefficient against a steel ball versus carbon content for the Al_2O_3 (\blacksquare), CNT/Al_2O_3 (\blacktriangle) and graphite/ Al_2O_3 (\blacksquare) coatings.

FESEM observations of the surface of the coatings reveal a homogeneous distribution of the CNTs (Fig. 3a) and graphite platelets (Fig. 3b, back-scattered electron image in chemical composition mode). Cracks (Fig. 3c and d), were generated on purpose by performing high-load indentations. CNTs are observed bridging the crack (Fig. 3c), as could be expected from previous results [23,25], whereas the shorter platelets do not (Fig. 3d).

Cross-sections images such as those shown for T18 and G21 suggest that the T coatings (Fig. 4a) are less rough than the G coatings (Fig. 4b), as confirmed by white-light interferometry. Note that the crack between substrate and coating (Fig. 4a) was formed because the cross section was prepared by cryo-fracturation. The CNTs are homogenously dispersed within the coating (Fig. 4c). The graphite platelets mostly form micrometer-sized agglomerates (Fig. 4b) but individual platelets not thicker than 50 nm are observed in a higher-magnification image (Fig. 4d). The thickness of the coatings was evaluated on similar images. It increases upon the increase in carbon content from 1 μ m for Al $_2$ O $_3$ to about 7 μ m for T18 and about 5 μ m for G21.

Above a carbon content of 10 wt.%, the higher thickness of the T coatings compared to that of the G coatings could be attributed to the higher viscosity of the corresponding sols. The arithmetic average roughness (Ra) calculated from white-light interferential rugosimetry images (not shown) is equal to about 0.4 μm for both the steel substrate and the Al2O3 coating, 0.8 μm for the G coatings and only 0.2 μm for the T coatings, in agreement with the FESEM observations where the T18 coating (Fig. 4a) seems to be more leveling towards the roughness of the steel substrate

3.2. Nano-indentation

Typical load-penetration depth plots measured with a Berkovich nano-indenter are shown in Fig. 5a. The maximum penetration depth was kept lower than the tenth of the coating thickness for all samples, in order to neglect the influence of the substrate on the calculated mechanical properties. Analysis of such curves by the Oliver and Pharr method [26] shows that the Young's modulus of the coatings (Fig. 5b) steadily decreases upon the increase in carbon content, from 41 GPa for Al₂O₃ to 9 GPa for G21. The hardness (Fig. 5c) is constant (about 1000 MPa) from 0 to 14 wt.% of carbon

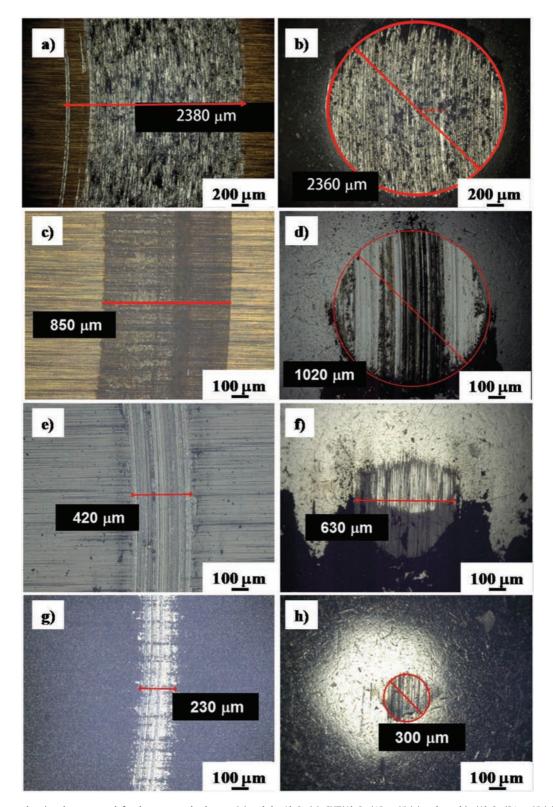


Fig. 8. Optical images showing the wear track for the un-coated substrate (a) and the Al_2O_3 (c), CNT/Al_2O_3 (18 wt.%) (e) and graphite/ Al_2O_3 (21 wt.%) (g) coatings and the corresponding steel balls (b, d, f and h, respectively).

and then decreases sharply, reaching only 200 MPa for G21. FESEM observation of the surfaces do not indicate a correspondingly lower densification, thus this could reflect a lower crystallization state of Al_2O_3 , less cohesive grain boundaries or a weak carbon matrix interface.

3.3. Tribological testing

Typical curves showing the friction coefficient against a steel ball versus the sliding distance are shown in Fig. 6. The curves for the un-coated steel and $\rm Al_2O_3$ coating are very unstable, which could reflect that wear is severe and a lot of debris is visibly generated

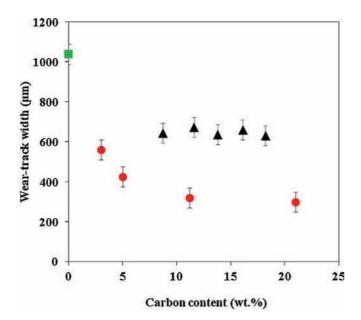


Fig. 9. Width of the wear tracks versus carbon content for the Al_2O_3 (\blacksquare), CNT/Al_2O_3 (\blacktriangle) and graphite/ Al_2O_3 (\blacksquare) coatings.

(it was neither quantified nor imaged). For the T18 coating, there is a running-in period of about 20 m, with a marked increase of the friction coefficient, and after which it is stabilized. This could reveal that the coating is separated from the substrate (reflecting total wear) very early in the test, possibly reflecting a weak coating/substrate interface. The so-produced debris would stay in the track and seemingly show a lubricating role that pure Al_2O_3 does not, which could be in agreement with reports by other authors [18,19]. For the G21 coating, there is a small but regular increase up to a sliding distance of about 200 m.

Therefore, for all specimens, the average steady-state friction coefficient (μ) was calculated for the last 50 m of the tests. The Al₂O₃ coating (μ = 0.95) does not provide any improvement on friction behavior over un-coated steel (μ = 0.80). By contrast, μ decreases upon the increase in carbon content (Fig. 7), very progressively for the T coatings (reaching 0.26 for T18) and more abruptly for the G coatings for which a plateau at about μ = 0.17 is reached for G5.

Post-test optical observations of the wear tracks of selected coatings and the corresponding steel balls are presented in Fig. 8. The wear tracks of the composite coatings appear to be less wide and partially covered with a black film-like layer. This tribologically-formed film was transferred to the steel ball, suggesting that the observed lubricating effect during sliding may be at least partially related to the smearing of a tribofilm over the contact area. The width of the wear tracks (Fig. 9), measured on similar optical images, is lower for the composite coatings than for the Al₂O₃ coating (>1000 μ m). The value is constant (about 650 μ m for the T samples) and decreasing down to 300 μ m for the G specimens.

Wear tracks of selected samples (T18 and G21) were observed by non-contact optical profiler imaging (Fig. 10a and c) and the corresponding profiles perpendicular to the track (Fig. 10b and d) were used to measure their half-width and depth. Note that for T18 (Fig. 10b), the depth of the profile is close to the coating thickness (7 μ m, Fig. 7) which could confirm total wear as noted above. The wear volume (V_w) was calculated for a half-ellipse track according to Eq. (1):

$$V_W \left(mm^3 \right) = 1/2.\pi rd.2\pi L \tag{1}$$

with r=track half-width (mm), d=track depth (mm), L=outer radius of the circular track (10 mm).

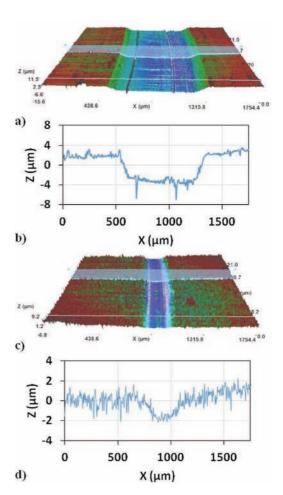


Fig. 10. White-light interferometry images of the wear tracks and the corresponding profiles for the CNT/ Al_2O_3 (18 wt.%) coating (a and b) and the graphite/ Al_2O_3 (21 wt.%) coating (c and d).

The calculated wear volume for T18 and G21 is equal to 0.20 and 0.06 mm^3 , respectively. G21 is the most wear-resistant composite coating compared to un-coated steel (0.30 mm^3) or Al_2O_3 coating (0.12 mm^3).

4. Discussion

Although a detailed investigation of the wear mechanisms as performed by other authors [18,19] is well beyond the scope of this work and warrants further studies, it was found of particular interest to study the wear tracks by Raman spectroscopy. Typical Raman spectra for T18 and G21, inside and outside the worn surface, are shown in Fig. 11. The ratio between the intensities of the D band and the G band (I_D/I_G) was calculated from the spectra. A higher ratio is generally attributed to the presence of more structural defects (more sp³ carbon). Interestingly, the spectra for T18 (Fig. 11a) are remarkably similar to each other and the I_D/I_G ratio are similar too (1.83 outside and 1.90 inside). Moreover, it is also similar to the I_D/I_G ratio (1.86 and 1.88) reported [23] for the raw and functionalized CNTs (same provider, same batch), respectively. These results indicate that the CNTs do not suffer drastic deterioration. It is probable that the more defective CNTs, those with a high number of walls, are destroyed because of the strong wear, very early in the friction test, resulting in the formation of a carbon/CNT/Al₂O₃ tribofilm, the subsequent behavior of which does not favor CNT destruction and delamination. By contrast, the I_D/I_G ratio for G21 (Fig. 11b) increases significantly from 0.42 (outside, and 0.40 for raw graphite) to 1.22 (inside), which could indicate some destruction of the graphite par-

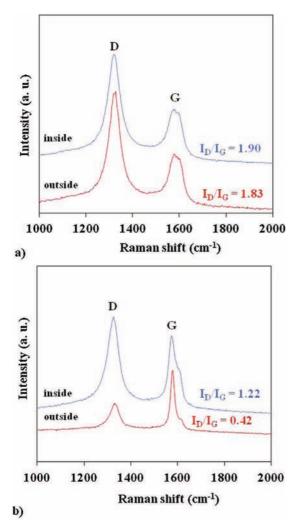


Fig. 11. Typical Raman spectra outside and inside the wear track for the CNT/ Al_2O_3 (18 wt.%) (a) and graphite/ Al_2O_3 (21 wt.%) (b) coatings.

ticles during the friction test. The so-generated debris would form a lubricating film, lowering the shearing resistance.

As mentioned above, the CNTs do not appear to be strongly damaged during the friction test and therefore the total carbon content in the samples is higher than the contributing, or "useful", carbon. The relative weight of each wall was calculated [27] and a plot (similar to the one in Fig. 7) of the steady-state friction coefficient versus the corrected carbon content was drawn (Fig. 12a), considering that the innermost wall is inactive (i.e. one counts only 7 walls), then that the two innermost walls are inactive (i.e. one counts only 6 walls), and so on until one counts only the outer wall, which contributes only 16.5% to the total weight of the CNT. The corresponding curves gradually shift to the left and interestingly the data when considering only the outer wall (labelled 1 W in Fig. 12a) fall reasonably in line with those of the graphite curve, forming what we consider to be a "master-curve". This suggests that instead of using weight or volumic carbon contents, the appropriate unit for tribological behavior could be the surface area developed by the involved carbon species. For the CNT/Al₂O₃ specimens, the contributing surface area is thus estimated, admittedly roughly, as that calculated for the outer wall only, considering the geometric characteristics of the CNTs described in Section 2.1 and no bundling. Assuming that the contributing surface area is constant during the friction test, for a given sample, the corresponding plot of friction coefficient versus surface area forms the basis of

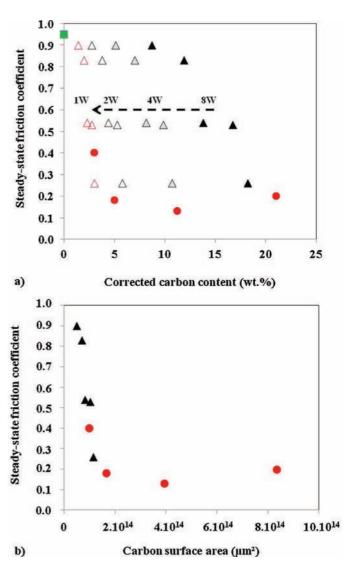


Fig. 12. (a) Steady-state friction coefficient versus corrected carbon content, considering only the number of walls considered active for the process (see text for details). Curves for 8, 4, 2 and 1 walls are shown. The arrow shows the gradual shift to the left; (b) steady-state friction coefficient versus the surface area of the carbon species involved in the friction phenomena (see text for details) for the CNT/Al₂O₃ (triangles) and graphite/Al₂O₃ (coatings. The error bars have been omitted for clarity

another "master-curve" (Fig. 12b). For the graphite/Al₂O₃ specimens, the carbon flakes were first described as cylinders 1.5 µm in diameter and 1.5 µm thick (which obviously is far too thick) to get a the starting data for the calculation: the initial number of carbon flakes for a given weight content and the corresponding surface area (the vertical lateral area of the cylinder was neglected). Then, the cylinders were horizontally "sliced" in half, thus each time doubling the surface area developed by the carbon, until a reasonable fit with the "master-curve" (Fig. 12b) was obtained. This occurred after slicing 11 times and corresponds to a carbon layer about 730 pm thick, i.e. close to two layers of graphene, which reasonably supports the finding that it is the graphene (graphite) surface available that is important for the reduction of friction and wear and that its initial form (either CNTs or graphite flakes) is un-important. Therefore, our results suggest that selecting graphite flakes is more convenient because, first it is easier to disperse into the boehmite sol and second it is more readily delaminated and broken into thinner flakes (also called few-layered-graphene) during the test, therefore providing more desirable lubricating surface area.

5. Conclusions

Carbon/Al₂O₃ coatings on 304-L stainless steel prepared using either CNTs (average number of walls = 8) or graphite flakes present a lower friction coefficient (reduced by a factor of 4-5) and lower wear (reduced by a factor of 2 for graphite flakes) against steel compared to a pure Al₂O₃ coating. The CNTs and micrometersized graphite agglomerates are homogenously dispersed within the coatings, the thickness of which is in the 1–7 μ m range. There is no dramatic deterioration of the structure of the CNTs during the friction test whereas the graphite flakes are partially destroyed. The CNT/Al₂O₃ coating is separated from the substrate very early in the test, reflecting total wear, and the so-produced debris would show a lubricating role that pure Al₂O₃ does not, but however less efficiently than for graphite/Al₂O₃ coatings. The so-generated debris form a lubricating film, which is transferred to the steel ball, suggesting that the observed lubricating effect during sliding may be at least partially related to the smearing of a transferred film over the contact area. Some simple modeling shows that only the outer wall of the CNTs contributes to the sliding, in agreement with the observed absence of major deterioration, and that the total available carbon surface area in the specimen is the relevant parameter. Therefore, it is shown that graphite flakes are more efficient than CNTs, at least in these experimental tribological conditions, because they are readily delaminated into thinner flakes (few-layered-graphene) during the test, providing more desirable lubricating surface area. These results provide important guidelines for the design of self-lubricating carbon/Al₂O₃ coatings and future work directions include: the preparation of graphite/Al₂O₃ coatings with thinner graphite flakes, and an in-depth study of the influence of the change in the coatings thickness, Young's modulus and hardness evolutions that were shown to occur upon the increase in carbon content.

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