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How to optimize metabolites concentrations and fluxes in a metabolic pathway?



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Introduction

The modelling of metabolic pathways is developed here for application to sugar metabolism of tomatoes during fruit expansion. In the now classic calculation of network fluxes, we add concentrations. Under assumptions of steady state and of "minimal effort" as proposed in [1], fluxes and concentrations, respecting constraints of experimental measurements and of thermodynamics, are searched. In order to obtain a well-posed problem, a preliminary mathematical study is in progress.

Backgrounds and Basics

The development time of tomato fruit is about 50 days. This period is subdivided in 50 daily phases numbered by l. At the scale of a phase (day), we suppose that the flux vector is at steady state and follows some fixed constraints. But at the scale of the development, these constraints will change. So we formulate a general problem at a phase scale l and solve it for $l = 1 \cdots 50$.

Notations

Previous approach and results: problem nº1

- 1. $X = (X_1, \dots, X_m)$ the vector of internal metabolites,
- 2. $V = (v_1, \dots, v_n)$ the flux vector composed by the rate v_j of each reaction j in the network.
- 3. $S = (s_{ij})_{i=1\cdots m, j=1\cdots n}$ the stoichiometry matrix (only refers to the internal metabolites) where s_{ij} is the stoichiometric coefficient of metabolite X_i in reaction j.
- 4. $\tilde{S} = (\tilde{s}_{ij})_{i=1\cdots m, j=1\cdots n_i}$ a submatrix extracted from the stoichiometry matrix containing only the stoichiometric coefficients of internal reactions (reactions between internal metabolites).

Thermodynamics Basics

Let us recall that the **Gibbs free energy change** for a reaction r_i

$$\mathbf{n}_1\mathbf{X}_1 + \mathbf{n}_2\mathbf{X}_2 \longleftrightarrow \mathbf{n}_3\mathbf{X}_3 + \mathbf{n}_4\mathbf{X}_2$$

is defined by

$$\Delta G'_{r_i} = \Delta G^{0'}_{r_i} + RT \ln \frac{X_3^{n_3} X_4^{n_4}}{X_1^{n_1} X_2^{n_2}}$$

where $\Delta G_{r_i}^{0'} = n_3 \Delta G_{f_{X_3}}^{0'} + n_4 \Delta G_{f_{X_4}}^{0'} - n_1 \Delta G_{f_{X_1}}^{0'} - n_2 \Delta G_{f_{X_2}}^{0'}$ is a constant, *R* the gas constant, and *T* the absolute temperature in Kelvin. **Generalisation:** With *m* metabolites et n_i internal reactions, let us set

 $\left(\Delta G_{r_1}(X) \right) \qquad \left(\Delta G_{f_{X_1}}^{0'} + RT \ln(X_1) \right)$

Constraints Steady-state (for all internal metabolites) assumption

dX/dt = SV = 0

Bounds on the fluxes (the values $v_{l,j,min}$ and $v_{l,j,max}$ are constants)

 $v_{l,j,min} \le v_j \le v_{l,j,max}$ for $j = 1 \cdots n$

Proposition. $\min_{V \in A_l} ||V||^2$ admits a unique solution where A_l is the non empty following feasible set

 $A_l = \{ V \in \mathbb{R}^n, SV = 0, v_{l,j,min} \le v_j \le v_{l,j,max} \quad \text{for } j = 1 \cdots n \}$

Simp	le	Ex	am	ple	e: b	ro	ble	m r	nº1		
$S = \begin{pmatrix} \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\$	1 0 0 0 0 , <i>mit</i> <i>nin</i>	0 1 0 0 0 $n = 0$ $m = v_{l_{s}}$ nique	0 0 0 0 -1 $0 \forall j = -1$ j, max $solut$	$0 \\ 0 \\ 0 \\ -1 \\ 0 \\ \neq 1, 2 \\ c = 1 \\ ion is$	-1 1 0 0 0 0 2, 8 an for <i>j</i>	$-1 \\ 0 \\ 1 \\ 0 \\ 0 \\ 0 \\ d \\ = 1,$	${\scriptstyle -1 \atop 0 \atop 0 \atop 1 \atop 0 \atop 0}$	$\begin{array}{c} 0 \\ -1 \\ 1 \\ 0 \\ 0 \\ 0 \end{array}$	$\begin{array}{c} 0 \\ -1 \\ 0 \\ 0 \\ 0 \\ 1 \end{array}$	$\begin{array}{c} 0 \\ 0 \\ -1 \\ -1 \\ 1 \\ 1 \end{array}$	
$V \approx$	(1,]	1, 1.3	51, 0.6	9, 0, 0	0.31,	0.69,	0.38,	0.61,	(0.69)		
			Ŝ		$ \begin{array}{c} -1 \\ 1 \\ 0 \\ 0 \\ 0 \\ 0 \\ 0 \end{array} $	$ \begin{array}{c} -1 \\ 0 \\ 1 \\ 0 \\ 0 \\ 0 \\ 0 \end{array} $	$ \begin{array}{c} -1 \\ 0 \\ 0 \\ 1 \\ 0 \\ 0 \\ 0 \end{array} $	$ \begin{array}{c} 0 \\ -1 \\ 1 \\ 0 \\ 0 \\ 0 \\ 0 \end{array} $	$ \begin{array}{c} 0 \\ -1 \\ 0 \\ 0 \\ 0 \\ 1 \end{array} $	$\begin{array}{c} 0 \\ 0 \\ -1 \\ -1 \\ 1 \\ 1 \end{array}$	

$$\Delta G_r(X) = \begin{pmatrix} & & & \\ &$$

Proposition.

 $\Delta G_r(X) = {}^t \tilde{S} Y(X)$

$A \xrightarrow{r_{5}} B \xrightarrow{r_{9}} B \xrightarrow{r_{9}} B \xrightarrow{r_{10}} B \xrightarrow{r_{$

New Approach

In this new modelling, in addition to the fluxes, we introduce metabolite concentrations (as in [2]), which are also supposed to be at steady state. In order to ensure thermodynamic reliability, we add the Gibbs free energy change of each reaction. Still assuming some "minimal effort", we simultaneously minimize $||V||^2$ and $||\Delta G_r(X)||^2$. We add an other set of constraints B_l for the metabolites to take into account some experimental measurements.

Math	ematical Formulation	Study of problem nº2			
	$\min_{(V,X)\in K_l} \left(V ^2, \Delta G_r(X) ^2 \right)$	Proposition. $\min_{X \in \mathbb{R}^m_+} \Delta G_r(X) ^2$ admits a space of			
with		solutions of dimension equals to $dim(ker({}^t \tilde{S}))$.			
and	$K_l = \{ (V, X) \in A_l \times B_l, \Delta G_{r_i}(X) \ v_i < 0 \text{ for } i = 1 \cdots n_i \}$	The problem without constraints is solved			
	$B_l = \{ X \in \mathbb{R}^m_+, M_l X = T_l \}.$	Simple example: problem n°2			

second law of thermodynamics $\Delta G_{r_i}(X) v_i < 0$ that may be conflicting. So a rigorous answer to this problem comes from understanding the separated problems.

This problem is a multi-objective optimization with uncoupled functions but with constraints (respect the

References

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Conclusion

Why a mathematical study ? With only numerical computations, the set of optimal solutions may not be exhaustively described. To find all the solutions (if they exist) of the multiobjective problem, understand each separated problem is a pre-requisite. Moreover this formal outcome will fit to any other network.