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Graphical Abstract:



Discovery of new hit-molecules targeting *Plasmodium falciparum* through a global SAR study of the 4-substituted-2-trichloromethylquinazoline antiplasmodial scaffold

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Abstract:

From 4 antiplasmodial hit-molecules identified in 2-trichloromethylquinazoline series, we conducted a global Structure-Activity relationship (SAR) study involving 26 compounds and covering 5 molecular regions (I – V), aiming at defining the corresponding pharmacophore and identifying new bioactive derivatives. Thus, after studying the aniline moiety in detail, thienopyrimidine, quinoline and quinoxaline bio-isosters were synthesized and tested on the K1 multi-resistant *P. falciparum* strain, along with a cytotoxicity evaluation on the human HepG2 cell line, to define selectivity indecies. SARs first showed that thienopyrimidines and quinolines were globally more cytotoxic, while quinoxaline analogs appeared as active as-and less cytotoxic than their quinazoline counterparts. Such pharmacomodulation in quinoxaline series not only provided a new antiplasmodial reference hit-molecule (IC₅₀ = 0.4 μ M) and quite selective (SI = 265) synthesis intermediate.

Highlights:

► Antiplasmodial pharmacomodulation of CCl₃-substituted-nitrogen containing heterocycles was made. ► Thienopyrimidine derivatives appeared more cytotoxic. ► Original 3-substituted-2-trichloromethylquinoxaline analogs were prepared. ► Two quinoxaline derivatives displayed *in vitro* IC₅₀ values of 0.4 and 0.5 µM on the K1 multi-resistant *P. falciparum* strain. ► Cytotoxicity was assessed on the human HepG2 cell line showing low cytotoxicity (CC₅₀ ~ 40 µM) and improved selectivity indecies (77-100).

Keywords: Quinazoline; Quinoline; Quinoxaline; Thienopyrimidine; Trichloromethyl goup; *Plasmodium falciparum*; *In vitro* antiplasmodial activity; *In vitro* HepG2 cytotoxicity, Structure-Activity Relationships.

1. Introduction

Cerebral malaria, caused by *Plasmodium falciparum*, is the leading cause of death among parasitic infections worldwide. According to the 2015 World Malaria Report **[1]**, 214 million people were infected by *Plasmodium* in 2014, leading to 438.000 deaths out of which about 88% occurred in Africa, mainly in children under five. It is to note that a significant improvement of the situation has been noted since the beginning of the 21st century, thanks to the action of the WHO and the financial involvement of several non-governmental organizations.

However, the control of the infection is facing worldwide the emergence of drug-resistant strains of the parasite which turns into a major concern for the medical and scientific community. Indeed, the WHO recommended the treatment of *P. falciparum* malaria is based on an artemisinin-combination therapy (ACT), but resistances to artemisinin derivatives are emerging in Asia [2], and it was demonstrated that they were responsible for therapeutic failures in several infected patients [3,4]. Moreover, it was also highlighted that the African *Anopheles gambiae* mosquito could transmit such Asian resistant parasites [5], indicating a major worldwide spreading risk. Thus, research efforts have to be developed, in order to discover new chemical entities presenting novel mechanisms of action, to complete and guaranty the durable efficiency of ACTs.

Several research teams working in the field of antimalarial agents previously reported novel quinazoline derivatives displaying significant antiplasmodial activities, in particular when bearing an amino- [6], alkylamino- [7,8] or an aniline- substituent [9] at position 4 of the quinazoline ring.

The research activity of our group is focused on the synthesis and anti-infective evaluation of new nitrogen-containing heterocycles [10]. We have intensively studied a large series of antiplasmodial derivatives based on the original 2-trichloromethylquinazoline scaffold. Thus, the synthesis and *in vitro* biological evaluation of a large quinazoline chemical library bearing an arylamino- [11,12], aryl- [13], phenoxy- [14], phenylthio- [15], sulfonamido- [16], alkynyl- [17], heteroarylamino- [18], benzyloxy- or alkoxy- [19] substituent at position 4 of the quinazoline ring, revealed several hit-molecules (A-D) which are presented in Figure 1.





In the light of these hit-molecules, within the aim of defining the antiplasmodial Structure Activity Relationships (SARs) of the corresponding series, 5 key-regions (I-V) were chosen and modulated, as summarized in Figure 2. Thus, for region I, we first investigated the role of the -CCl₃ group by replacing it by various substituents including closely related halogenated groups. For region II, we studied the influence of the combination of the halogen atoms on the aniline moiety of hit A. The effect of the methylation of the secondary amine nitrogen of hit A was then investigated in region III. Further modulations in region IV were made by replacing the quinazoline ring of hits **A-D** by a bio-isosteric thienopyrimidine. Indeed, starting from the natural quinazoline-based antimalarial molecule Febrifugine, such ring replacement was successfully carried out [20]. Moreover, some recent thienopyrimidine derivatives demonstrated promising antiplasmodial activities [21,22]. Finally, in region V, we looked for novel analogs, centered either on the quinoline nucleus, as this heterocycle is encountered in numerous antimalarial drugs such as quinine, chloroquine, mefloquine, amodiaquine, piperaguine and primaguine, or on the guinoxaline nucleus, taking advantage of previously reported results obtained in our team in the pyrrolo[1,2-a] quinoxaline series [23]. The synthesis of all new derivatives, their *in vitro* biological evaluation and the resulting SARs is presented and discussed herein.



Figure 2. Antiplasmodial SAR study involving 5 key-regions (I-V) from the quinazoline hits

2. Results and discussion

2.1. Synthesis of hit A analogues

2.1.1. Region I: Variation of the nature of the substituent at position 2 of the quinazoline ring

The nature of the substituent at position 2 of hit A was first studied. Thus, the synthesis of the 2-methyl- 1, 2-chloromethyl- 2, 2-dichloromethyl- 3, 2-chloro-2-difluoro-methyl- 7, 2trifluoromethyl-8 and the 2-unsubstituted-9 quinazoline analogs were operated, starting from appropriate 4-chloroquinazolines (Scheme 1). Compound 1 [11] was already synthesized by our team via an Aromatic Nucleophilic Substitution (S_NAr) reaction, starting from 4-chloro-2trichloromethylquinazoline E [24]. Derivatives 2 and 3 were prepared by S_NAr reactions [25] 4-chloro-2-chloromethylquinazoline and starting from F 4-chloro-2dichloromethylquinazoline G [26]. The synthesis of the parent 2-unsubstituted compound 9 was already reported in the literature, but to the best of our knowledge, it was never studied for its antiplasmodial potential [27]. The biological evaluation of the above prepared quinazolines is summed up in Table 1.



Scheme 1. SAR modulation of region I: modification of the substituent at position 2 of the quinazoline ring.

Reagents and conditions : (i) 2,4-dichloroaniline 0.8 equiv, conc. HCl cat., iPrOH, 70 °C, 2 h.

For the synthesis of derivative **7**, the corresponding 4-chloroquinazoline intermediate was not reported in the literature. Thus, starting from 2-aminobenzonitrile, acylation of the amine was made by reacting with 2-chloro-2-difluoroacetic acid in the presence of the coupling agent *N*-dimethylaminopropyl-*N*'-ethylcarbodiimide (EDCI) and dimethylaminopyridine

(DMAP), affording **4** in low yield (15%). Then, a smooth oxidative hydration of nitrile **4** was operated with H_2O_2 , followed by a cyclization in alkaline medium [**17,28**], leading to quinazolinone **5**. This last was further chlorodehydroxylated using POCl₃, affording the expected intermediate **6** (Scheme 2).



Scheme 2. Variation of the region I: Synthesis of intermediate 6.

Reagents and conditions : (i) 2-chloro-2-difluoroacetic acid 1 equiv, EDCI.HCl 1 equiv, DMAP 0.8 equiv, CH₂Cl₂, 0 °C then rt, 24 h; (ii) H₂O₂ 30%, NaOH, EtOH/H₂O, 0 °C then rt, 90 min; (iii) POCl₃, 140 °C, MW, 10 min, 800 W.

2.1.2. Region II: Variation of the halogen atoms at position 2' and 4' of the 4anilino-substituent

The synthesis of the anilino derivatives **10-13** was easily performed under classical S_NAr reaction conditions by reacting the appropriate 2,4-dihaloaniline with 4-chloro-2-trichloromethylquinazoline **E** in refluxing isopropanol, in 55-70 % range yield (**Scheme 3**). The results of their *in vitro* biological evaluation are presented in **Table 1**.



Scheme 3. Chemical modulations at regions II and III of the pharmacophore *Reagents and conditions:* (i) 2,4-dihaloaniline (2 equiv) or 2,4-dichloro-*N*-methylaniline (1.1 equiv), *i*PrOH, reflux, 2 h.

2.1.3. Region III: N-methylation of the 4-(2',4'-dichlorophenyl)amine substituent

The influence of the hydrogen atom on the secondary amine, in region III, was also studied. For this purpose, a S_NAr reaction was performed between **E** and *N*-methyl-2,4-dichloroaniline, affording the expected methylated compound **14** in low yield (**Scheme 3**). The biological evaluation of compound **14** is summarized in **Table 1**.

2.2. Biological evaluation of the series belonging to regions I, II and III

These new derivatives were evaluated *in vitro* toward both their antiplasmodial activity against the K1 multi-resistant *P. falciparum* strain (determination of the IC_{50} = inhibitory concentration 50%) and their cytotoxicity (determination of the CC_{50} = cytotoxic concentration 50%) on the HepG2 human cell line. The results were compared with three commercial antimalarial reference-drugs (atovaquone, chloroquine and doxycycline) and a cytotoxic reference drug (doxorubicine). For all tested compounds, the corresponding selectivity indecies (SI) were calculated (SI = CC_{50}/IC_{50}). The results are presented in **Table 1**.

The aim of the modulation of region I was to confirm that the trichloromethyl group, at position 2 of the quinazoline ring, was the only substituent providing the antiplasmodial activity. Indeed, the antiplasmodial activity was totally lost when replacing it with other groups, including closely related ones such as CF₃ or CHCl₂. Looking at the combination of the halogen substituents on the aniline ring at position 4 (region II), the antiplasmodial activity was maintained for brominated compounds **10-12**, in comparison with hit **A** (respectively 0.3, 0.5 and 0.6 *vs* 0.4 μ M), but the cytotoxicity was significantly increased, leading to less selective molecules. Contrary to the results obtained with brominated analogues, fluorinated analogue **13** presented the same cytotoxicity profile as hit **A** but was less active (IC₅₀ = 3.1 μ M). Finally in region III, the methylation of the aniline moiety (compound **14**) significantly impaired both antiplasmodial activity and cytotoxicity.

N N R ₂						
Malaanla	- R ₂	D	HepG2 CC ₅₀	K1 P. falciparum	Selectivity	
Molecule		- K 4	(µM)	$IC_{50}(\mu M)$	Index ^c	
1	-CH ₃	CI H CI	15	85	0.18	
2	-CH ₂ Cl	CI ₩————————————————————————————————————	0.5	>12.5 ^d	< 0.04	
3	-CHCl ₂	H H CI	4.7	>50	< 0.09	
7	-CF ₂ Cl	H N CI CI CI	7	48.4	0.14	
8	-CF ₃	CI H CI	>50	>50	-	
9	-H	cı H————————————————————————————————————	35	>50	<0.7	
10	-CCl ₃	H N Br Br	2	0.3	6.7	
11	-CCl ₃	CI Br	5	0.5	10	
12	-CCl ₃	H-CI	2.9	0.6	4.8	
13	-CCl ₃	K F F	19.6	3.1	6.3	
14	-CCl ₃		10	3.4	2.9	
Hit A	-CCl ₃	CI H N-CI	16	0.4	40	
Doxorubicine ^a		0.2	-	-		
Atovaqı	lone ^b		> 15.6 ^d	0.001	15600	
Chloroquine ^b			30	0.6	50	
Doxycycline ^b			20	6.0	3.3	

Table 1. In vitro antiplasmodial and cytotoxicity evaluation of compounds 1-3 and 7-14:Modulations at regions I, II and III.

 R_4

^aDoxorubicine was used as a cytotoxic reference-drug; ^bAtovaquone, Chloroquinine and doxycycline were used as antimalarial reference-drugs, ^cSelectivity indecies were calculated according to the formula : SI = HepG2 $CC_{50} / K1 IC_{50}$. ^dHighest concentration tested due to a lack of solubility.

2.3. Region IV: Synthesis of thieno[3,2-*d*]pyrimidine series.

The replacement of the phenyl ring of the quinazoline nucleus, by a thiophene nucleus as a bio-isostere, was next studied. We first tried to synthesize 4-substituted-2-(trichloromethyl)thieno[3,2-*d*]pyrimidine derivatives by a S_NAr reaction between the key chlorinated intermediate **20** and various nucleophilic species, corresponding to the substituents borne by hits **A-D**, respectively 2,4-dichloroaniline **a**, 3-trifluoromethylaniline **b**, 4-chlorophenol **c** and 4-chlorothiophenol **d** (**Figure 3**).



Figure 3. Thieno[3,2-*d*]pyrimidines initial retrosynthesis pathway.

Starting from the commercially available 3-aminothiophene-2-carboxylate, we first acetylated the amino group in presence of Ac₂O, leading to **15** in 78% yield **[29]**. Then, the cyclization was made by heating **15** in presence of 25% ammonia in a sealed vial, affording thienopyrimidinone **16 [30]** in 63% yield. Unfortunately, in our hands, the chlorination reaction of **16**, to access **20**, did not succeed. Thus, we decided to realize the chlorination reaction in 2 steps. We first synthesized 4-chloro-2-methylthieno[3,2-*d*]pyrimidine **17 [30]**. Then, **17** was engaged in a second chlorination reaction with PCl₅/POCl₃, under classical heating or microwave activation to convert the 2-methyl group into a 2-trichloromethyl substituent. Unfortunately, here again, no trace of the key intermediate **20** was formed, and only inseparable mixtures of polychlorinated compounds were observed by LC-MS analysis (**Scheme 4**).



Scheme 4. Synthesis of key intermediate 20.

Reagents and conditions: (i) Ac₂O, AcOH cat., 120 °C, MW, 10 min; (ii) NH₄OH 25%, 105 °C, 3 h, sealed vial; (iii) PCl₅ 6 equiv, POCl₃ as solvent, 100 °C, MW, 20 min, 800 W; (iv) POCl₃, pyridine cat., reflux 24 h (molecule **17**) or 1 h (molecule **20**); (v) Et₃N, trichloroacetyl chloride, 5 °C, 30 min; (vi) AcOH saturated with gaseous HCl, trichloroacetonitrile, 100 °C, 18 h, evaporated then refluxed in *i*PrOH, 5 min.

To introduce the $-\text{CCl}_3$ group, we chose to start again from 3-aminothiophene-2carboxylate and carried out the acylation reaction, using trichloroacetyl chloride in presence of Et₃N. The corresponding product **18** was obtained in quantitative yield. Unfortunately, the following cyclization step to lactam **19** was unsuccessful under the reaction conditions used (either NH₄OH at 105 °C in sealed vial or NH₃ in MeOH at 120 °C under MW). Finally, we succeeded in preparing lactam **19** by reacting 3-aminothiophene-2-carboxylate with trichloroacetonitrile in AcOH, saturated with gaseous HCl **[31]**. This was a similar approach to the one of Ried **[32]** who used a more expensive reagent: methyl 2,2,2-trichloroacetimidate. Then, the chlorodehydroxylation of lactam **19** in refluxing POCl₃ in the presence of a catalytic amount of pyridine, furnished the key intermediate **20** in almost quantitative yield. S_NAr reactions between **20** and some N, O or S centered nucleophiles were then performed in a similar way to those operated in quinazoline series. The target 4-substituted-2-(trichloromethyl)thieno[3,2-*d*]pyrimidine derivatives **21a-d** were obtained, however, in low to moderate yields (7-55%) (**Scheme 5**). The results of the biological evaluation of this series are presented in **Table 2**.



Scheme 5. Synthesis of 4-substituted-2-(trichloromethyl)thieno[3,2-*d*]pyrimidine. *Reagents and conditions:* (i) aniline derivative **a** or **b** 1 equiv, conc. HCl cat., EtOH, 100 °C, 18 h, sealed vial; (ii) NaH 2 equiv, 4-chlorophenol **c** or 4-chlorothiophenol **d** 1 equiv, DMSO, rt or 50 °C, 24 h.

2.4. Region IV: Synthesis of the thieno[2,3-d]pyrimidine series.

To complete this SAR study with more thiophene-containing bio-isosteres, we decided to synthesize the position isomer analogues in thieno[2,3-*d*]pyrimidine series. This could be performed by a S_NAr reaction between the desired nucleophiles **a-d** and a key chlorinated intermediate **25**, obtained from 2-aminothiophene-3-carbonitrile (**Figure 4**).



Figure 4. Thieno[2,3-d]pyrimidines initial retrosynthesis pathway.

Thus, 2-aminothiophene-3-carbonitrile was acetylated with Ac₂O to give 22 [33]. Then, the cyclization was made in the presence of H_2O_2 30% in alkaline medium (NaOH) [17,28] leading to thienopyrimidinone 23 [34] which was then chlorodehydroxylated into 24. Unfortunately, as previously observed in the thieno[3,2-*d*]pyrimidine series, we did not succeed in synthesizing 25, neither by the *gem*-trichloromethylation of 24 [35], nor by the tetrachlorination of 23 with in a PCl₅/POCl₃ mixture. (Scheme 6). Acylation of 2-aminothiophene-3-carbonitrile was then realized with trichloroacetyl chloride, to give 26. Cyclization into lactam 27 was next conducted with a mixture of polyphosphoric acid (PPA)

and phosphoric acid. Nevertheless, once again, it remained impossible to isolate key intermediate 25 *via* a chlorodehydroxylation reaction. It seemed that 25 was too unstable to be isolated, indeed some traces were detected by LC-MS, but direct engagement of crude 25 in S_NAr reaction did not afforded target compounds.



Scheme 6. Various synthesis routes investigated for the synthesis of key intermediate 25. *Reagents and conditions:* (i) Ac₂O, AcOH cat., 120 °C, MW, 10 min; (ii) H₂O₂ 35%, NaOH, EtOH/H₂O, 0 to 55 °C, 15 min; (iii) PCl₅ 6 equiv, POCl₃ as solvent, 100 °C, MW, 20 min, 800 W; (iv) POCl₃, pyridine cat., reflux 24 h; (v) Et₃N, trichloroacetyl chloride, rt, 36 h; (vi) H₃PO₄, polyphosphoric acid, 70 °C, 3 h; (vii) POCl₃ or PCl₅ or oxalyl chloride or P₂O₅/tetrabutylammonium chloride (TBACl) or SOCl₂.

Facing the difficulty to prepare the intermediate **25** and in order to access to the target thieno[2,3-*d*]pyrimidine derivatives, we tried another synthesis pathway based on the cyclization of thiophene derivatives which could be cyclized into thieno[2,3-*d*]pyrimidine rings bearing already both the CCl₃ group and an aniline (or phenol or thiophenol) moiety (**Scheme 7**). Thus the synthesis of amidines **28a** and **28b** was conducted by reacting nitrile **26** with appropriate anilines in presence of AlCl₃ at room temperature in CH₂Cl₂ [**36**]. The crudes residues obtained were a mixture of unreacted starting material **26** (even after 72 h at rt), the expected amidines **28a-b** and cyclized target derivatives **29a** and **29b**. When the temperature was increased to improve the conversion, undesirable side products appeared. Refluxing the obtained amidines **28a-b** in ethanol led to the target compounds **29a** and **29b**, respectively in 36 and 15% yields [**37**]. For phenol **c** and thiophenol **d**, the same conditions using AlCl₃ did not afford the expected derivatives. We then considered the synthesis work of Baati and al.,

describing the addition of thiols onto nitrile groups in presence of gaseous HBr in ether at 0 $^{\circ}$ C [38]. Indeed, to our delight, by using these reaction conditions, thioimidate 28d was produced, and further cyclization in refluxing ethanol afforded compound 29d in low yield (Scheme 7). Unfortunately, the same reaction conditions were inefficient to obtain compounds 28c and 29c. Other conditions were attempted in order to produce the imidate 28c *via* the addition of 4-chlorophenol into the nitrile group, in the presence of NaH [39], Na [40], K₂CO₃ [41] or gaseous HCl [42], but all of them conducted to the same disappointing result. The results of the biological evaluation are presented in Table 2.



Scheme 7. Synthesis of 4-substituted-2-(trichloromethyl)thieno[2,3-*d*]pyrimidine 29a-d. *Reagents and conditions:* (i) anilines **a** or **b**, AlCl₃, CH₂Cl₂, rt, 72 h; (ii) 4-chlorothiophenol, gaseous HBr, Et₂O, - 20 °C, 30 min then 18 h at rt; (iii) *i*PrOH, reflux, 5 min.

2.5. Modulation of region IV: Biological evaluation

In region IV, the replacement of the quinazoline ring by thienopyrimidine scaffolds led to molecules displaying similar antiplasmodial activity as hit-molecules **A-D**, with IC₅₀ values in the submicromolar range (0.4 to 0.9 μ M) (**Table 2**). However, all the thienopyrimidine derivatives, belonging either to the [3,2-*d*] or [2,3-*d*] series, appeared more cytotoxic (CC₅₀ = 0.7 to 13.3 μ M) than the corresponding hit-molecules, leading to lower selectivity indecies. Globally, SARs indicate that there is no significant difference between the two thienopyrimidine series, regarding both activity and cytotoxicity.

Malaanla	Ctore atoma	HepG2 CC ₅₀	K1 P.falciparum	Selectivity
Molecule	Structure	(µM)	$IC_{50}(\mu M)$	Index ^c
	CI			
21a	HN S	3.2	0.6	5.3
	N CCl3			
	HNCCE			
21b	S N N	0.7	0.9	0.8
	N CCl ₃)
21c	S N	4.3	0.6	7.2
	N CCl₃		\sim	
	s			17.0
21d	S N	6.9	0.4	17.2
	CICI			
20.5	Г. Ц ŅН	6.2	0.5	12 /
29a	N N	0.2	0.5	12.4
	s `N´ `CCŀ			
29b		4.0	0.6	6.7
_> 0	S N CCh		0.0	0.7
	CI			
29d	s s	13.3	0.8	16.6
	S N CCI3			
H	Hit A	16	0.4	40
H	Hit B	150	1.8	83
Hit C		50	1.1	45
Hit D		>25 ^d	0.9	>28
Doxorubicine ^a		0.2	-	-
Atovaquone^b		$> 15.6^{d}$	0.001	15600
Chloroquine^b		30	0.6	50
Doxycvcline ^b		20	6.0	3.3

Table 2. In vitro antiplasmodial and cytotoxicity evaluation of compounds 21a-d and 29a-d:modulations at region IV

^aDoxorubicine was used as a cytotoxic reference-drug; ^bAtovaquone, Chloroquinine and doxycycline were used as antimalarial reference-drugs, ^cSelectivity indecies were calculated according to the formula : SI = HepG2 $CC_{50} / K1 IC_{50}$. ^dHighest concentration tested due to a lack of solubility.

2.6. Region V: Synthesis of quinoline series.

In order to evaluate the influence of the pyrimidine moiety of the quinazoline ring onto the biological profile, we chose to synthesize some analogs in 4-substituted-2-trichloromethylquinoline series bearing, at position 4, the same substituents **a-d** as the hit-molecules. Thus, starting from 4-chloro-2-trichloromethyquinoline **H** obtained by chlorination of 2-methylquinolin-4(1H)-one with PCl₅ and POCl₃ [42], we first tried solvent-free S_NAr reaction conditions [**11**] with anilines but without any conversion. Compounds **30a** and **30b** were finally obtained by using a catalytic amount of conc. HCl in refluxing EtOH to obtain the desired compounds in 45 and 55% yields respectively. For the phenol **c** and thiophenol **d**, we generated the corresponding anions using NaH before reacting **H**. Thus, phenol derivative **30c** and thiophenol derivative **30d** were obtained, respectively, in 72 and 32% yields (**Scheme 8**). The biological evaluation of these molecules is summarized in **Table 3**.



Scheme 8. Synthesis of 4-substituted-2-trichloromethylquinoline series. *Reagents and conditions :* (i) aniline derivative 1 equiv, conc. HCl cat., EtOH, 100 °C, 18 h, sealed vial; (ii) NaH 2 equiv, 4-chlorophenol or 4-chlorothiophenol reagent 1 equiv, DMSO, rt or 50 °C, 24 h.

2.7. Region V: Synthesis of quinoxaline derivatives.

Previously, we reported that 2-trichloromethylquinoxaline **I** displayed an *in vitro* antiplasmodial activity ($IC_{50} = 1.5 \mu M$), hindered by a significant *in vitro* cytotoxicity ($CC_{50} = 3.1 \mu M$) **[23]**. Moreover, the introduction of a phenyl substituent at position 3 (compound **J**) gave a more selective derivative **J** (SI = 17.5 *versus* 2.1, respectively), thanks to a better

antiplasmodial IC₅₀ value (0.2 μ M) [23]. Such interesting results noted with compound J prompted us to carry on the investigation of the antiplasmodial potential of the 3-substituted-2-trichloromethylquinoxaline scaffold, in continuation of the pharmacomodulation of the quinazoline moiety in region V (Figure 5).



Figure 5. Rational for the synthesis of 2-substituted-3-trichloromethylquinoxaline derivatives.

The first attempted synthetic route involved the reaction of nucleophiles **a-d** with 2-chloro-3-trichloromethylquinoxaline **K** [44] under S_NAr conditions, as previously reported in quinazoline series. Unfortunately, all tested conditions failed to afford the expected derivatives **32a-d**, where no conversion or degradation of starting material could occur.

To solve this problem, we next investigated another pathway by reacting 2-chloro-3methylquinoxaline **L** [45] with the nucleophiles **a-d** *via* a S_NAr reaction, before operating the *gem*-trichloromethylation reaction (Scheme 9). The best reaction conditions found were to carry out the S_NAr reaction in DMF, under moderate heating, in the presence of Cs₂CO₃ leading to **31a**, **31b**, **31c** and **31d** in, respectively, 19%, 5%, 85% and 83% respectively. Another strategy to improve these low reaction yields was successful only with 3trifluoromethylaniline **b**, by reacting **L** in aniline **b** (without any solvent) under microwave heating in a sealed vial (69%). Then, from intermediates **31a-d**, a final chlorination step was needed to transform the methyl group into a $-CCl_3$ one. This step was achieved using the classical route elaborated by our team using a $PCl_5/POCl_3$ mixture, under microwave heating. Unfortunately, only the ether and thioether derivatives **32c** and **32d** were obtained in good yields (70-72%) (Scheme 9). It seemed that the presence of the amine function of **31a** and **31b** inhibited the chlorination reaction. The results of the biological assays of this series are presented in **Table 3**.



Scheme 9. Synthesis of 2-substituted-3-trichloromethylquinoxaline series.

Reagents and conditions: (i) 2,4-dichloroaniline, 4-chlorophenol or 4-chlorothiophenol 1 equiv, Cs₂CO₃ 1 equiv, anh. DMF, 70 °C, 12 h, sealed vial, N₂; (ii) 3-trifluoromethylaniline 7 equiv, 140 °C, MW, 45 min, sealed vial; (iii) PCl₅ 6 equiv, POCl₃ as solvent, 100 °C, MW, 20 min, 800 W.

2.8. Modulation of region V: biological evaluation.

Thus, the ring variation of region V afforded two series of compounds: a quinoline and a quinoxaline series.

Among the four tested compounds in quinoline series, it appeared that all exerted an antiplasmodial activity globally similar to the ones of **hits A-D**. However, the cytotoxicity toward the HepG2 human cell line was increased for all of the quinoline series, in comparison to the Hits **A-D** and particularly for aniline **b** (6.5 μ M vs 150 μ M for **hit B**). It must be pointed out that for quinoline 3d, a lack of solubility in the biological media was observed, as previously reported for **hit D**. Thus, the substitution of the quinazoline ring by a quinoline ring was globally detrimental.

Contrary to the quinoline series, the quinoxaline series appeared very promising. Thus, derivative **32c** showed both a better antiplasmodial activity than its analog **hit C** (0.4 μ M vs 1.1 μ M) and a great improvement in the cytotoxic profile, its SI reaching 100, compared to 45 for **hit C**. Similarly, the same profile was observed for compound **32d** with a 2-fold activity increase compared to **hit D** (0.45 μ M vs 0.9 μ M) and a SI of 77 without any solubility impairment, as observed for **hit D**.

To conclude, we noted that for the region V, the suppression of nitrogen N-3 of the quinazoline led to generally more cytotoxic compounds while the replacement of the quinazoline moiety by a quinoxaline one was a significant improvement of the biological profile, compound **32c** reaching a SI of 100.

Molecule	Structure	HepG2 CC ₅₀ (μM)	K1 P.falciparum IC ₅₀ (μM)	Selectivity Index ^c
30a		11.7	1.3	9.0
30b		6.5	1.3	5.0
30c		31	1.5	20.7
30d	S C C I	>15.6 ^d	1.1	>14.2
32c		40.2	0.4	100.5
32d		38.6	0.5	77.2
	Hit A	16	0.4	40
	Hit B	150	1.8	83
	Hit C	50	1.1	45
	Hit D	>25 ^d	0.9	>28
	Doxorubicine ^a	0.2	-	-
	Atovaquone ^b	> 15.6 ^d	0.001	15600
	Chloroquine ^b	30	0.6	50
	Doxycycline ^b	20	6.0	3.3

Table 3. In vitro antiplasmodial and cytotoxicity evaluation of compounds 30a-d, and 32c-d:modulations at region V.

^aDoxorubicine was used as a cytotoxic reference-drug; ^bAtovaquone, Chloroquinine and doxycycline were used as antimalarial reference-drugs, ^cSelectivity indecies were calculated according to the formula : SI = HepG2 $CC_{50} / K1 IC_{50}$. ^dHighest concentration tested due to a lack of solubility.

Finally, when comparing the biological activities of the synthetic intermediates prepared for studying region V, it was interesting to note that the intermediate **E** was not active (IC₅₀ = 54.5 μ M) while intermediates **H** and **K** displayed a significant IC₅₀ (0.8 and 0.4 μ M respectively). Noticeably, the intermediate **K** possessed a very good SI (265) thanks to a moderate value of its CC₅₀. Moreover, the presence of a substituent at position 3 appeared to be mandatory for displaying a good antiplasmodial profile by comparison of compound **I** which displayed a SI of only 2 (**Figure 6**). It is to note that both derivatives **32c** (IC₅₀ = 0.15 μ g/mL) and **K** (IC₅₀ = 0.11 μ g/mL) meet the hit to lead *in vitro* TDR criteria [46] for "selectively active antiplasmodial agents": IC₅₀<0.2 μ g/mL, SI>100.



Figure 6. Biological assessment of synthetic intermediates.

3. Conclusion

From previously identified antiplasmodial hits A-D, in 2-trichloromethyl-4-substitutedquinazoline series, new derivatives were synthesized in order to study the SARs in 5 precisely defined regions (I-V). The modulation of region I showed that the -CCl₃ group was mandatory for providing antiplasmodial activity in quinazoline series. In the same series, the study of regions II and III demonstrated that the optimal substitution of the aniline moiety by halogens atoms was 2,4-dichloro and that the N-methylation was detrimental toward activity. Then, by modulating regions IV and V, it respectively appeared that the replacement of the quinazoline ring by a quinoline ring led to activity levels similar to the ones of Hits A-D but also to increased cytotoxicities values, while the replacement of the quinazoline ring by a quinoxaline ring highlighted derivatives **32c-d**, displaying both a preserved activity and a reduced cytotoxicity, in comparison with hits A-D, as summarized in Table 4. These molecules now appear as new reference antiplasmodial hits. Finally, the great antiplasmodial potential of intermediate K was revealed. Both derivatives 32c (IC₅₀ = 0.15 µg/mL) and K $(IC_{50} = 0.11 \ \mu g/mL)$ meet the hit to lead *in vitro* TDR criteria [46] for "selectively active" antiplasmodial agents": IC₅₀<0.2 µg/mL, SI>100, opening the way to the synthesis of novel lead-compounds.

R		R N N CCl ₃ Previous hits A-D	S N CCl ₃ 21a-d	R N S N CCL 29a,b,d	R N CC 30a-d	$ \begin{array}{c} $
CI	IC ₅₀	0,4	0.6	0.5	1.3	-
HN	CC ₅₀	16	3.2	6.2	11.7	-
mandana	SI	40	5.3	12.4	9.0	-
CF ₃	IC ₅₀	1,8	0.9	0.6	1.3	-
HN	CC ₅₀	150	0.7	4.0	6.5	-
reaching	SI	83	0.8	6.7	5.0	-
	IC ₅₀	1.1	0.6	-	1.5	0.4
	CC ₅₀	50	4.3	-	31	40.2
*****	SI	45	7.2	-	20.7	100.5
	IC ₅₀	0.9	0.4	0.8	1.1	0.5
s-CI	CC ₅₀	>25	6.9	13.3	>15.6*	38.6
~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	SI	>28	17.2	16.6	>14.2	77.2

**Table 4.** Summary of SAR data at regions IV and V.

*Highest concentration tested due to a lack of solubility.

In bold, newly identified hit-molecules

#### 4. Experimental

#### 4.1. Chemistry

Commercial reagents were used as received without additional purification. Melting points were determined on a Kofler bench and are uncorrected. Elemental analysis and HRMS were carried out at the Spectropole, Faculté des Sciences et Techniques de Saint-Jérôme, Marseille, France. NMR spectra were recorded on a Bruker ARX 200 spectrometer or a Bruker AV 250 spectrometer at the Faculté de Pharmacie de Marseille or a BRUKER Avance III nanobay 400 at the the Spectropole, Faculté des Sciences et Techniques de Saint-Jérôme, Marseille (¹H-NMR: 200, 250 or 400 MHz, ¹³C-NMR: 50, 63 or 100 MHz). NMR references were the following: ¹H: CHCl₃  $\delta$  = 7.26, DMSO-*d*₆  $\delta$  = 2.50 and ¹³C: CHCl₃  $\delta$  = 76.9, DMSO-*d*₆  $\delta$  = 39.5. Solvents were dried by conventional methods. The following adsorbent was used for column chromatography: silica gel 60 (Merck, particle size 0.063–0.200 mm, 70–230 mesh ASTM). TLC was performed on 5 cm × 10 cm aluminium plates coated with silica gel 60F-

254 (Merck) in an appropriate eluent. Visualization was made with ultraviolet light (234 nm). HRMS spectra were recorded on QStar Elite (Applied Biosystems SCIEX) spectrometer. PEG was the matrix for HRMS. The experimental exact mass was given for the ion which has the maximum isotopic abundance. Purity of synthetized compounds was checked with LC-MS analyses which were realized at the Faculté de Pharmacie de Marseille with a Thermo Scientific Accela High Speed LC System[®] coupled with a single quadrupole mass spectrometer Thermo MSQ Plus[®]. The RP-HPLC column used is a Thermo Hypersil Gold[®]  $50 \times 2.1$  mm (C18 bounded), with particles of 1.9 µm diameter. The volume of sample injected on the column was 1 µL. The chromatographic analysis, total duration of 8 min, is made with the gradient of following solvents: t = 0 min, water/methanol 50/50; 0 < t < 4 min, linear increase in the proportion of water to a ratio water/methanol 95/5; 4 < t < 6 min, water/methanol 95/5; 6 < t < 7 min, linear decrease in the proportion of water to return to a ratio 50/50 water/methanol; 6 < t < 7 min, water/methanol 50/50. The water used was buffered with 5 mM ammonium acetate. The retention times  $(t_{\rm R})$  of the molecules analyzed are indicated in min. The preparation of 4-chloro-2-trichloromethylquinazoline E [24], 2methyl-N-(2,4-dichlorophenyl)quinazolin-4-amine 1 [10], N-(2,4-dichlorophenyl)quinazolin-4-amine 9 [27], 4-chloro-2-trichloromethyquinoline H [43], methyl 3-acetamidothiophene-2carboxylate 15 [28], 4-chloro-2-chloromethylquinazoline F [24], 4-chloro-2dichloromethylquinazoline G [26], 4-chloro-2-trifluoromethylquinazoline [47], 2-chloro-3trichloromethylquinoxaline K [44] and 2-chloro-3-methylquinoxaline L [45] was achieved as described in the literature and characterization were consistent as reported in literature.

#### 4.1.1. General procedure for the preparation of *N*-aryl-2trichloromethylquinazolin-4-amine (2, 3, 7, 8 and 14)

To a mixture of appropriate 4-chloroquinazoline (1 equiv) in isopropanol (10 mL) and appropriate aniline (0.8 equiv) were added a few drops of concentrated HCl. The vial was then sealed and heated at 70 °C for 2 h. The mixture was allowed to cool to room temperature and was poured into 20 mL of iced water. The mixture was then extracted twice with  $CH_2Cl_2$ . The combined organic phases were washed twice with water, then dried ( $Na_2SO_4$ ) and the solvent was evaporated under reduced pressure. The residue was then purified by silica gel column chromatography (Petroleum Ether/CH₂Cl₂ 1/1).

#### 4.1.1.1. 2-Chloromethyl-*N*-(2,4-dichlorophenyl)quinazolin-4-amine (2)

Starting from 4-chloro-2-chloromethylquinazoline [25] (500 mg, 2.35 mmol) and 2,4-dichloroaniline (304 mg, 1.88 mmol).

Yield 71%. White powder. mp 171 °C. ¹H NMR (200 MHz, CDCl₃)  $\delta$  = 8.97 (d, *J* = 8.9 Hz, 1H), 8.21 (bs, 1H), 7.99-7.82 (m, 3H), 7.68-7.60 (m, 1H), 7.47-7.35 (m, 2H), 4.75 (s, 2H). ¹³C NMR (50 MHz, CDCl₃)  $\delta$  =160.9, 157.1, 149.9, 133.6, 128.9, 128.7, 128.0, 127.6, 127.5, 123.9, 122.9, 120.1, 113.9, 47.6. A quaternary C was not observed under these experimental conditions. Anal. Calcd for C₁₅H₁₀Cl₃N₃: C, 53.20; H, 2.98; N, 12.41. Found: C, 53.60; H, 3.12; N, 11.61. LC-MS (ESI, 35 eV): *t*_R = 4.21 min, *m/z* 338 [M+H]⁺.

**4.1.1.2. 2-Dichloromethyl-***N***-(2,4-dichlorophenyl)quinazolin-4-amine (3)** Starting from 4-chloro-2-dichloromethylquinazoline G [26] (500 mg, 2.02 mmol) and 2,4dichloroaniline (259 mg, 1.62 mmol).

Yield 42%. White powder. mp 203 °C. ¹H NMR (200 MHz, CDCl₃)  $\delta$  = 9.11 (d, *J* = 8.9 Hz, 1H), 8.30 (bs, 1H), 8.02-7.85 (m, 3H), 7.73-7.66 (m, 1H), 7.49-7.39 (m, 2H), 6.78 (s, 1H). ¹³C NMR (50 MHz, CDCl₃)  $\delta$  =160.6, 157.6, 149.6, 133.9, 133.6, 129.5, 128.7, 128.3, 128.2, 123.7, 122.9, 120.1, 114.5, 71.8. A quaternary C was not observed under these experimental conditions. Anal. Calcd for C₁₅H₉Cl₄N₃: C, 48.29; H, 2.43; N, 11.26. Found: C, 48.41; H, 2.29; N, 11.03. LC-MS (ESI, 35 eV): *t*_R = 4.71 min, *m/z* 372 [M+H]⁺.

#### 4.1.1.3. 2-(chlorodifluoromethyl)-*N*-(2,4-dichlorophenyl)quinazolin-4amine (7)

Starting from 4-chloro-2-chlorodifluoromethylquinazoline **6** (500 mg, 2.01 mmol) and 2,4-dichloroaniline (293 mg, 1.8 mmol).

Yield 30%. White powder. mp 195 °C. ¹H NMR (400 MHz, CDCl₃)  $\delta$  = 8.93 (d, *J* = 9.0 Hz, 1H), 8.38 (bs, 1H), 8.11 (d, *J* = 8.3 Hz, 1H), 8.00-7.92 (m, 2H), 7.79-7.74 (m, 1H), 7.48 (d, *J* = 2.5 Hz, 1H), 7.42-7.35 (m, 1H). ¹³C NMR (100 MHz, CDCl₃)  $\delta$  =157.3, 155.6 (t, *J* = 29 Hz), 149.4, 134.2, 133.2, 130.0, 129.2, 129.0, 128.8, 128.2, 123.9, 122.9 (t, *J* = 292 Hz), 122.8, 120.1, 114.9. HRMS (ESI): m/z calcd. for C₁₅H₈Cl₃F₂N₃ [M+H]⁺: 374.9908. Found: 374.9909. LC-MS (ESI, 35 eV): *t*_R = 4.58 min, *m/z* 376 [M+H]⁺.

**4.1.1.4.** *N*-(**2,4-dichlorophenyl**)-**2**-(trifluoromethyl)quinazolin-4-amine (**8**) Starting from 4-chloro-2-trifluoromethylquinazoline [45] (500 mg, 2.15 mmol) and 2,4dichloroaniline (279 mg, 1.72 mmol).

Yield 37%. White powder. mp 198 °C. ¹H NMR (200 MHz, CDCl₃)  $\delta$  = 8.88 (d, *J* = 9.0 Hz, 1H), 8.29 (bs, 1H), 8.11-8.07 (m, 1H), 7.97-7.89 (m, 2H), 7.78-7.70 (m, 1H), 7.47-7.35 (m, 2H). ¹³C NMR (50 MHz, CDCl₃)  $\delta$  =157.3, 152.1 (q, *J* = 36 Hz), 149.5, 134.2, 133.2, 130.1, 129.2, 129.1, 128.8, 128.2, 123.9, 122.8, 120.1, 119.8 (q, *J* = 275 Hz), 115.2. Anal. Calcd for C₁₅H₈Cl₂F₃N₃: C, 50.30; H, 2.25; N, 11.73. Found: C, 50.39; H, 2.21; N, 11.80.

#### 4.1.1.5. *N*-(2,4-dichlorophenyl)-*N*-methyl-2-trichloromethylquinazolin-4amine (14)

Starting from 4-chloro-2-(trichloromethyl)quinazoline **E** [24] (500 mg, 1.77 mmol) and *N*-methyl-2,4-dichloroaniline (188  $\mu$ L, 1.95 mmol).

Yield 15%. Colorless oil. ¹H NMR (200 MHz, CDCl₃)  $\delta$  = 8.12-8.08 (m, 1H), 7.61-7.60 (m, 2H), 7.35-7.17 (m, 3H), 6.93-6.89 (m, 1H), 3.63 (s, 3H). ¹³C NMR (50 MHz, CDCl₃)  $\delta$  =161.8, 160.1, 150.9, 143.1, 134.3, 133.1, 133.0, 131.3, 129.9, 129.5, 129.1, 127.1, 124.4, 114.9, 97.4, 41.3. HRMS (ESI): m/z calcd. for C₁₆H₁₀Cl₅N₃ [M+H]⁺: 418.9317. Found: 418.9319. LC-MS (ESI, 35 eV):  $t_{\rm R}$  = 5.18 min, m/z 419.7 [M+H]⁺.

#### 4.1.2. Synthesis of 4-chloro-2-chlorodifluoromethylquinazoline (6) 4.1.2.1. 2-Chloro-*N*-(2-cyanophenyl)-2,2-difluoroacetamide (4)

In a 250 mL round-bottom flask was successively introduced 2-aminobenzonitrile (2.0 g, 16.9 mmol), 4-(*N*,*N*-dimethylamino)pyridine (DMAP) (1.7 g, 13.3 mmol) and *N*-(3-dimethylaminopropyl)-*N*'-ethylcarbodiimide hydrochloride (EDCI.HCl) (3.4 g, 17.6 mmol) in CH₂Cl₂ (50 mL). Then, 2-chloro-2,2-difluoroacetic acid (1.4 mL, 16.9 mmol) was added into the mixture and it was stirred for 24 h at rt. The solvent was evaporated under reduced pressure and water (200 mL) was added to the residue. The mixture was gentle heated at 40 °C until a homogeneous solution was obtained. After cooloing down to rt, conc. HCl was added to adjust the pH = 1. The resulting precipitate was filtered, thoroughly washed with water and dried in an oven leading 2-chloro-*N*-(2-cyanophenyl)-2,2-difluoroacetamide.

Yield 15%. White powder. mp 84 °C. ¹H NMR (400 MHz, CDCl₃)  $\delta$  = 8.37 (d, *J* = 8.5 Hz, 2H), 7.71 (t, *J* = 7.6 Hz, 2H), 7.36 (t, *J* = 7.6 Hz, 1H). ¹³C NMR (100 MHz, CDCl₃)  $\delta$  =157.1 (t, *J* = 30 Hz), 137.6, 134.5, 132.7, 126.3, 121.6, 121.5, (t, *J* = 300 Hz), 115.3, 103.7. Anal. Calcd for C₉H₅ClF₂N₂O: C, 46.88; H, 2.19; N, 12.15. Found: C, 46.98; H, 2.13; N, 12.35. LC-MS (ESI, 35 eV): *t*_R = 0.85 min, *m/z* 229 [M-H]⁻.

#### 4.1.2.2. 2-Chlorodifluoromethylquinazolin-4(3*H*)-one (5)

In a 100 mL round-bottom flask was successively introduced 2-chloro-*N*-(2-cyanophenyl)-2,2-difluoroacetamide **4** (500 mg, 2.17 mmol), 30%  $H_2O_2$  (2 mL) and a mixture of EtOH (6 mL) and water (8 mL). Then, NaOH (140 mg, 3.47 mmol) was added at 0 °C. The mixture was stirred at rt for 90 min and the volatiles were removed under vacuum. Water (100 mL) was added to the residue and the resulting mixture was acidified to pH = 4 with conc. HCl.

The precipitate obtained was filtered, thoroughly washed with water and dried in an oven gave the 2-chlorodifluoromethylquinazolin-4(3H)-one (5).

Yield 80%. Beige powder. mp 246 °C. ¹H NMR (400 MHz, CDCl₃)  $\delta$  = 10.44 (bs, 1H), 8.37 (d, *J* = 8.5 Hz, 1H), 7.89 (d, *J* = 3.7 Hz, 2H), 7.67-7.61 (m, 1H). ¹³C NMR (100 MHz, CDCl₃)  $\delta$  =135.6, 129.2, 128.7, 126.9. Five quaternary C were not observed under these experimental conditions. LC-MS (ESI, 35 eV): *t*_R = 0.66 min, *m/z* 229 [M-H]⁻.

#### 4.1.2.3. 4-Chloro-2-chlorodifluoromethylquinazoline (6)

In a 20 mL vial was successively introduced 2-chlorodifluoromethylquinazolin-4(3*H*)-one **5** (500 mg, 2.17 mmol) and POCl₃ (1 mL, 10.8 mmol). The mixture was stirred at 140 °C for 10 min under microwave irradiation (800 W). After cooling at rt, the crude was purified by silica gel column chromatography (CH₂Cl₂) to afford 4-chloro-2-chlorodifluoromethylquinazoline.

Yield 75%. Beige powder. mp 90 °C. ¹H NMR (400 MHz, CDCl₃)  $\delta$  = 8.38 (d, *J* = 8.3 Hz, 1H), 8.23 (d, *J* = 8.3 Hz, 1H), 8.12-8.08 (dt, *J* = 8.2, 1.2 Hz, 1H), 7.89 (dt, *J* = 8.2, 1.2 Hz, 1H). ¹³C NMR (100 MHz, CDCl₃)  $\delta$  =164.2, 155.0 (t, *J* = 30 Hz), 150.4, 136.2, 130.9, 129.6, 126.1, 123.8, 122.0 (t, *J* = 292 Hz). Anal. Calcd for C₉H₄Cl₂F₂N₂: C, 43.40; H, 1.62; N, 11.25. Found: C, 43.29; H, 1.56; N, 11.42. LC-MS (ESI, 35 eV): *t*_R = 3.42 min, *m/z* 249 [M+H]⁺.

#### 4.1.3. Representative procedure for the preparation of *N*-(2,4dihalogenophenyl)-2-trichloromethylquinazolin-4-amine (10-13)

A 100 mL round-bottomed single-neck flask equipped with a condenser and a magnetic stir bar was charged with 4-chloro-2-trichloromethylquinazoline **E** [24] (500 mg, 1.77 mmol), 2 equiv. of the appropriated dihalogeno-aniline (3.54 mmol) and 30 mL of isopropanol. The mixture was refluxed for 48 h. After cooling, water was added (50 mL), and the resulting solution was extracted three times with  $CH_2Cl_2$ . Then, the combined organic layers were washed with water and dried (Na₂SO₄). Finally, the solvent was evaporated under reduced pressure. The residue was purified by trituration with  $CH_2Cl_2$  and filtration yielding the *N*-(2,4-dihalogenophenyl)-2-trichloromethylquinazolin-4-amines (**10-13**).

#### 4.1.3.1. *N*-(2,4-dibromophenyl)-2-trichloromethylquinazolin-4-amine (10)

Yield 55%. White powder. mp 243 °C. ¹H NMR (200 MHz, DMSO- $d_6$ )  $\delta$  = 10.32 (bs, 1H), 8.55 (d, J = 8.2 Hz, 1H), 8.02- 7.91 (m, 3H), 7.82-7.59 (m, 3H). ¹³C NMR (50 MHz, DMSO- $d_6$ )  $\delta$  =160.0, 159.5, 148.9, 136.3, 134.7, 134.4, 131.1, 130.8, 128.5, 128.3, 123.1, 122.5, 119.5, 113.5, 97.7. Anal. Calcd for C₁₅H₈Br₂Cl₃N₃: C, 36.59; H, 1.62; N, 8.46. Found: C, 36.16; H, 1.53; N, 8.50. LC-MS (ESI, 35 eV):  $t_R$  = 5.41 min, m/z 494 [M+H]⁺.

# 4.1.3.2. *N*-(4-bromo-2-chlorophenyl)-2-trichloromethylquinazolin-4-amine (11)

Yield 63%. White powder. mp 222 °C. ¹H NMR (200 MHz, DMSO- $d_6$ )  $\delta$  = 10.38 (bs, 1H), 8.62 (d, J = 8.3 Hz, 1H), 8.07-7.97 (m, 2H), 7.93 (s, 1H), 7.87-7.80 (m, 1H), 7.75 - 7.65 (m, 2H). ¹³C NMR (50 MHz, DMSO- $d_6$ )  $\delta$  =160.2, 159.7, 149.1, 135.0, 134.6, 132.1, 131.8, 130.8, 130.6, 128.7, 128.5, 123.4, 119.3, 113.7, 97.9. Anal. Calcd for C₁₅H₈BrCl₄N₃: C, 39.86; H, 1.78; N, 9.30. Found: C, 39.83; H, 1.70; N, 9.45. LC-MS (ESI, 35 eV):  $t_R$  = 5.41 min, m/z 450 [M+H]⁺.

# 4.1.3.3. N-(2-bromo-4-chlorophenyl)-2-trichloromethylquinazolin-4-amine (12)

Yield 70%. White powder. mp 223 °C. ¹H NMR (200 MHz, DMSO-*d*₆)  $\delta$  = 9.09 (d, *J* = 9.0 Hz, 1H), 8.38 (s, 1H), 8.11 (d, *J* = 8.3 Hz, 1H), 8.00-7.89 (m, 2H), 7.77-7.70 (m, 1H), 7.63 (d, *J* = 2.3 Hz, 1H), 7.44 (dd, *J* = 9.0, 2.3 Hz, 1H). ¹³C NMR (50 MHz, DMSO-*d*₆)  $\delta$  =160.6, 157.2, 149.6, 134.5, 134.0, 131.7, 130.3, 129.2, 128.8, 128.7, 122.8, 120.1, 120.0, 114.1, 97.5. Anal. Calcd for C₁₅H₈BrCl₄N₃: C, 39.86; H, 1.78; N, 9.30. Found: C, 40.01; H, 1.69; N, 9.26. LC-MS (ESI, 35 eV): *t*_R = 4.82 min, *m/z* 450 [M+H]⁺.

4.1.3.4. *N*-(2,4-difluorophenyl)-2-trichloromethylquinazolin-4-amine (13)

Yield 70%. White powder. mp 165 °C. ¹H NMR (200 MHz, DMSO-*d*₆)  $\delta$  = 9.00-8.87 (m, 1H), 8.12-8.08 (m, 1H), 7.95-7.66 (m, 4H), 7.07-6.92 (m, 2H). ¹³C NMR (50 MHz, DMSO-*d*₆)  $\delta$  =160.7, 157.4, 153.2 (d, *J* = 260 Hz), 152.9 (d, *J* = 260 Hz), 149.6, 133.9, 130.2, 128.5, 123.5 (d, *J* = 8.8 Hz), 123.1 (dd, *J* = 9.5, 4.1 Hz), 120.1, 113.8, 111.4 (dd, *J* = 21.6, 3.6 Hz), 104.3-103 (m, 1CH), 97.6. Anal. Calcd for C₁₅H₈Cl₃F₂N₃: C, 48.09; H, 2.15; N, 11.22. Found: C, 48.45; H, 1.99; N, 11.25. LC-MS (ESI, 35 eV): *t*_R = 4.58 min, *m/z* 374 [M+H]⁺.

#### 4.1.4. 2-Methylthieno[3,2-*d*]pyrimidin-4(3*H*)-one (16)

A solution of methyl 3-acetamidothiophene-2-carboxylate **15** [29] (760 mg, 3.81 mmol) in NH₄OH 25% (9.5 mL) was heated, in a sealed vial, at 105 °C for 3 h. After cooling to room temperature, concentrated HCl was added until pH 8. The precipitate thus formed was filtered, washed with water and dried in a vacuum oven at 40 °C to afford 2-methylthieno[3,2-d]pyrimidin-4(3*H*)-one. No further purification necessary.

Yield 63%. White solid. mp 91°C. ¹H NMR (200 MHz, DMSO- $d_6$ )  $\delta$  = 12.31 (bs, 1 H, NH), 8.12 (d, J = 5.2 Hz, 1 H), 7.29 (d, J = 5.2 Hz, 1 H), 2.36 (s, 3 H). ¹³C NMR (63 MHz, DMSO- $d_6$ )  $\delta$  = 181.0, 158.8, 158.0, 156.3, 133.9, 124.5, 21.1. LC-MS (ESI, 35eV):  $t_{\rm R}$  = 0.64 min, m/z 167 [M+H]⁺.

#### 4.1.5. 4-Chloro-2-methylthieno[3,2-d]pyrimidine (17) [30]

To a solution of compound **16** (200 mg, 1.20 mmol) in POCl₃ (5.0 mL) were added a few drops of pyridine. The mixture was then refluxed for 24 h. After cooling down to room temperature, the solution was poured on ice and a saturated Na₂CO₃ solution was added until the excess of POCl₃ was neutralized. The aqueous phase was then extracted with EtOAc, the combined organic phases were washed with brine, dried (Na₂SO₄) and the solvent was evaporated under reduced pressure. The residue was then purified by silica gel column chromatography (eluent: Petroleum Ether/EtOAc 8/2) to afford 4-chloro-2-methylthieno[3,2-*d*]pyrimidine.

Yield 97%. White solid. mp 84 °C. ¹H NMR (250 MHz, CDCl₃)  $\delta$  = 7.96 (d, *J* = 5.5 Hz, 1H), 7.45 (d, *J* = 5.5 Hz, 1H), 2.78 (s, 3H). ¹³C NMR (63 MHz, CDCl₃)  $\delta$  = 164.7, 162.6, 154.7, 136.7, 127.9, 124.8, 25.5. LC-MS (ESI, 35 eV): *t*_R = 1.74 min, *m/z* 185 [M+H]⁺.

#### 4.1.6. Methyl 3-(2,2,2-trichloroacetamido)thiophene-2-carboxylate (18)

To a solution of methyl 3-aminothiophene-2-carboxylate (500 mg, 3.18 mmol) and triethylamine (440  $\mu$ L, 3.18 mmol) in tetrahydrofuran (4.0 mL) at 5 °C was added trichloroacetyl chloride (350  $\mu$ L, 3.18 mmol). The reaction was stirred at 5 °C for 30 min. Water (5.0 mL) was then added and the aqueous phase was extracted with EtOAc. The organic phase was then washed, dried (Na₂SO₄) and the solvent was evaporated to afford methyl 3-(2,2,2-trichloroacetamido)thiophene-2-carboxylate. No further purification necessary.

Yield 99%. White solid. mp 244 °C. ¹H NMR (200 MHz, DMSO-*d*₆)  $\delta$  = 11.42 (bs, 1 H), 8.06 (d, *J* = 5.4 Hz, 1 H), 7.89 (d, *J* = 5.4 Hz, 1 H), 3.88 (s, 3 H). ¹³C NMR (63 MHz, DMSO-*d*₆)  $\delta$  = 163.4, 158.1, 140.9, 133.5, 121.0, 113.6, 91.8, 52.2. LC-MS (ESI, 35 eV): *t*_R = 4.11 min, *m*/*z* 300 [M+H]⁺.

#### 4.1.7. 2-(Trichloromethyl)thieno[3,2-d]pyrimidin-4(3H)-one (19)

 $HCl_{(g)}$  was bubbled into acetic acid (1.23 mL) for 30 minutes. Then, nitrogen was swept across the solution. This solution was added to a mixture of methyl 3-aminothiophene-2-carboxylate (3.0 g, 19.08 mmol) and trichloroacetonitrile (0.85 mL, 8.48 mmol) in acetic acid (1.86 mL). The reaction was then heated at 100 °C for 18 h. After cooling down to rt, the acetic acid was evaporated under reduced pressure and isopropanol (4.5 mL) was added. This solution was refluxed for 5 min, then refrigerated. The precipitate thus formed was filtered, and then purified by silica gel column chromatography (eluent: Petroleum Ether 100% + 10

mL of EtOAc every 100 mL of eluent) to afford 2-(trichloromethyl)thieno[3,2-*d*]pyrimidin-4(3*H*)-one.

Yield 63%. White solid. mp 237 °C. ¹H NMR (200 MHz, CDCl₃)  $\delta$  = 10.50 (bs, 1 H), 7.93 (d, *J* = 5.3 Hz, 1 H), 7.51 (d, *J* = 5.3 Hz, 1 H). ¹³C NMR (50 MHz, CDCl₃)  $\delta$  = 157.7, 155.7, 151.8, 135.7, 126.1, 123.7, 92.0. LC-MS (ESI, 35 eV):  $t_{\rm R}$  = 0.78 min, m/z 267 [M-H]⁻.

#### 4.1.8. 4-Chloro-2-(trichloromethyl)thieno[3,2-d]pyrimidine (20)

To a solution of 2-(trichloromethyl)thieno[3,2-d]pyrimidin-4(3*H*)-one **19** (200 mg, 0.74 mmol) in POCl₃ (3.0 mL) were added a few drops of pyridine. The reaction was refluxed for 1 h. After cooling to room temperature, the solution was poured on ice and the aqueous phase was extracted with EtOAc. The combined organic phases were washed with brine, dried (Na₂SO₄) and the solvent was evaporated under reduced pressure. The residue was then purified by silica gel column chromatography (eluent: Petroleum Ether/EtOAc 8/2) to yield 4-chloro-2-(trichloromethyl)thieno[3,2-d]pyrimidine.

Yield 99%. White solid. mp 99 °C. ¹H NMR (200 MHZ, CDCl₃)  $\delta$  = 8.21 (d, *J* = 5.5 Hz, 1 H), 7.74 (d, *J* = 5.5 Hz, 1 H). ¹³C NMR (50 MHZ, CDCl₃)  $\delta$  = 161.9, 161.7, 155.6, 139.0, 130.7, 125.6, 96.0. LC-MS (ESI, 35 eV): *t*_R = 4.05 min (no ionization).

#### 4.1.9. *N*-(2,4-Dichlorophenyl)-2-(trichloromethyl)thieno[3,2-*d*]pyrimidin-4amine (21a)

To a solution of 4-chloro-2-(trichloromethyl)thieno[3,2-*d*]pyrimidine **20** (190 mg, 0.66 mmol) and 2,4-dichloroaniline (106 mg, 0.66 mmol) in ethanol (3.0 mL) were added a few drops of concentrated HCl. The vial was sealed and the reaction was heated at 100 °C for 24 h. The solvent was evaporated under reduced pressure and the residue was purified by silica gel column chromatography (eluent: Petroleum Ether/CH₂Cl₂ 8/2 + 10 mL of CH₂Cl₂ every 100 mL of eluent) to afford *N*-(2,4-dichlorophenyl)-2-(trichloromethyl)thieno[3,2-d]pyrimidin-4-amine.

Yield 45%. White solid. mp 197 °C. ¹H NMR (200 MHz, CDCl₃)  $\delta$  = 8.72 (d, *J* = 8.9 Hz, 1 H), 7.96 (d, *J* = 5.4 Hz, 1 H), 7.66 (d, *J* = 5.4 Hz, 1 H), 7.61 (s, 1 H), 7.48 (d, *J* = 2.3 Hz, 1 H), 7.37 (dd, *J* = 8.9, 2.3 Hz, 1 H). ¹³C NMR (200 MHz, CDCl₃)  $\delta$  = 161.6, 161.0, 154.6, 134.1, 133.2, 130.0, 129.1, 128.3, 126.2, 125.1, 124.0, 116.2, 97.1. LC-MS (ESI, 35 eV):  $t_{\rm R}$  = 4.90 min, *m*/*z* 410 [M-H]⁻. Anal. Calcd for C₁₃H₆Cl₅N₃S: C, 37.76; H, 1.46; N, 10.16. Found: C, 38.03; H, 1.26; N, 10.30.

#### 4.1.10. 2-(Trichloromethyl)-N-[3-(trifluoromethyl)phenyl]thieno[3,2-d]pyrimidin-4-amine (21b)

To a solution of 4-chloro-2-(trichloromethyl)thieno[3,2-*d*]pyrimidine **20** (150 mg, 0.52 mmol) and 3-(trifluoromethyl)aniline (87  $\mu$ L, 0.52 mmol) in ethanol (2.5 mL) were added a few drops of concentrated HCl. The vial was sealed and the reaction was heated at 100°C for 24 h. The solvent was evaporated under reduced pressure and the residue was purified by silica gel column chromatography (eluent: Petroleum Ether/EtOAc 8/2) to yield 2-(trichloromethyl)-*N*-[3-(trifluoromethyl)phenyl]thieno[3,2-*d*]pyrimidin-4-amine.

Yield 55%. White solid. mp 141 °C.¹H NMR (200MHz, CDCl₃)  $\delta$  = 8.42 (s, 1 H), 7.88-7.83 (m, 2 H), 7.58-7.39 (m, 4 H). ¹³C NMR (50 MHZ, CDCl₃)  $\delta$  = 161.9, 161.0, 155.0, 138.3, 133.9, 131.9, 131.3, 129.7, 125.7, 124.3, 121.3, 118.6, 115.7, 97.4. Anal. Calcd for C₁₄H₇Cl₃F₃N₃S: C, 40.75; H, 1.71; N, 10.18. Found: C, 40.99; H, 1.80; N, 10.39. LC-MS (ESI, 35 eV):  $t_{\rm R}$  = 4.87 min, m/z 412 [M+H]⁺

**4.1.11. 4-(4-Chlorophenoxy)-2-(trichloromethyl)thieno[3,2-d]pyrimidine (21c)** To a solution of 4-chloro-2-(trichloromethyl)thieno[3,2-*d*]pyrimidine **20** (210 mg, 0.73 mmol) and 4-chlorophenol (281 mg, 2.19 mmol) in acetonitrile (5.0 mL) was added K₂CO₃ (302 mg, 2.19 mmol). The reaction was heated at 160 °C for 1h10 under microwave irradiation. The mixture was poured onto a NaOH 10% solution. The aqueous phase was extracted with EtOAc. The organic phases were combined, washed with brine, dried (Na₂SO₄) and the solvent was evaporated under reduced pressure. The residue was then purified by silica gel column chromatography (eluent: Petroleum Ether/CH₂Cl₂ 8/2 + 10 mL of CH₂Cl₂ every 100 mL of eluent) to yield 4-(4-chlorophenoxy)-2-(trichloromethyl)thieno[3,2-*d*]pyrimidine.

Yield 39%. White solid. mp 161 °C. ¹H NMR (200 MHz, CDCl₃)  $\delta$  = 8.09 (d, *J* = 5.4 Hz, 1 H), 7.70 (d, *J* = 5.4 Hz, 1 H), 7.45-7.31 (m, 4 H). ¹³C NMR (50 MHz, CDCl₃)  $\delta$  = 163.5, 163.1, 161.5, 150.3, 136.9, 131.4, 129.9, 129.5, 125.2, 122.9, 117.7, 116.4, 96.4. Anal. Calcd for C₁₃H₆Cl₄N₂OS: C, 41.08; H, 1.59; N, 7.37. Found: C, 41.28; H, 1.72; N, 7.45. LC-MS (ESI, 35 eV):  $t_{\rm R}$  = 5.08 min, m/z 379 [M+H]⁺.

**4.1.12. 4-[(4-Chlorophenyl)thio]-2-(trichloromethyl)thieno[3,2-d]pyrimidine (21d)** To a solution of 4-chlorothiophenol (100 mg, 0.69 mmol) in DMSO (0.5mL), NaH (33 mg, 1.39 mmol) was added portionwise under inert atmosphere. The mixture was stirred at room temperature for 20 mi. 4-Chloro-2-(trichloromethyl)thieno[3,2-d]pyrimidine **20** (200 mg, 0.69 mmol) was then added portionwise and the reaction was stirred for 24 h at rt. Water (1.0 mL) was added and the aqueous phase was then extracted with  $CH_2Cl_2$ . The organic phase was washed with water, brine and then dried (Na₂SO₄). The solvent was evaporated and the residue was purified by silica gel column chromatography (eluent: Petroleum Ether 100% +

10 mL of  $CH_2Cl_2$  every 100 mL of eluent) to afford 4-[(4-chlorophenyl)thio]-2-(trichloromethyl)thieno[3,2-*d*]pyrimidine.

Yield 7%. White solid. mp not determined. ¹H NMR (200 MHz, CDCl₃)  $\delta$  = 8.02 (d, *J* = 5.5 Hz, 1 H), 7.67-6.64 (m, 3 H), 7.45 (d, *J* = 8.8 Hz, 2 H). ¹³C NMR (50 MHz, CDCl₃)  $\delta$  = 164.5, 161.1, 159.9, 137.0, 136.6, 136.4, 129.5, 127.5, 125.0, 124.7, 96.8. Anal. Calcd for C₁₃H₆Cl₄N₂S₂: C, 39.41; H, 1.53; N, 7.07. Found: C, 39.75; H, 1.72; N, 7.20. LC-MS (ESI, 35 eV): *t*_R = 5.36 min, *m*/*z* 395 [M+H]⁺.

#### 4.1.13. *N*-(3-Cyanothiophen-2-yl)acetamide (22)

To a solution of 2-aminothiophene-3-carbonitrile (1.020 g, 8.21 mmol) in acetic anhydride (3.0 mL) were added a few drops of acetic acid. The reaction was heated at 120 °C for 10 min under microwave irradiation. The brown solid thus obtained was filtered, washed with water and dried in a vacuum oven at 40 °C to afford *N*-(3-cyanothiophen-2-yl)acetamide. No further purification necessary.

Yield 90%. Brown solid. mp 210 °C (litt. 210-211 °C). ¹H NMR (200 MHz, DMSO- $d_6$ )  $\delta = 11.70$  (bs, 1 H), 7.13-7.12 (m, 2 H), 2.20 (s, 3 H). ¹³C NMR (50 MHz, DMSO- $d_6$ )  $\delta = 168.4$ , 149.4, 124.7, 118.7, 114.9, 91.8, 22.5. LC-MS (ESI, 35 eV):  $t_{\rm R} = 0.81$  min, m/z 165 [M-H]⁻.

#### 4.1.14. 2-Methylthieno[2,3-*d*]pyrimidin-4(3*H*)-one (23)

To a solution of *N*-(3-cyanothiophen-2-yl)acetamide **22** (1.232 g, 7.41 mmol) and  $H_2O_2$  30% (6.5mL) in ethanol (22 mL) and water (4 mL), at 0 °C, was added NaOH in pellets (0.495 g, 12.4 mmol). The reaction was stirred at room temperature for 15 min and then heated at 55 °C until a homogeneous solution was obtained. After cooling to rt, the solvent was evaporated under reduced pressure. Water (10 mL) was added and the mixture was heated to 50 °C. After cooling to rt, a HCl 1N solution was added until pH 4. The precipitate was filtered, washed with water and dried in a vacuum oven at 40°C to afford 2-methylthieno[2,3-*d*]pyrimidin-4(3*H*)-one. No further purification necessary.

Yield 72%. White solid. mp 210 °C.¹H NMR (200 MHz, DMSO- $d_6$ )  $\delta$  = 7.90 (s, 1 H), 7.39 (d, J = 5.8 Hz, 1 H), 6.93 (d, J = 5.8 Hz, 1 H), 2.19 (s, 3 H). ¹³C NMR (50 MHz, DMSO- $d_6$ )  $\delta$  = 166.9, 166.8, 145.8, 123.0, 115.8, 114.8, 23.3. LC-MS (ESI, 35 eV):  $t_{\rm R}$  = 0.65 min, m/z 167 [M+H]⁺.

#### 4.1.15. 4-Chloro-2-methylthieno[2,3-d]pyrimidine (24)

To a solution of 2-methylthieno[2,3-*d*]pyrimidin-4(3*H*)-one **23** (300 mg, 1.80 mmol) in POCl₃ (5.0 mL) were added a few drops of pyridine. The mixture was refluxed for 24 h and, after

cooling to rt, was poured onto ice. A saturated  $Na_2CO_3$  solution was added until complete neutralization of the excess of POCl₃. The aqueous phase was extracted with EtOAc. The combined organic phases were washed with brine, dried ( $Na_2SO_4$ ) and the solvent was evaporated under reduced pressure. The residue was purified by silica gel column chromatography (eluent: Petroleum Ether/EtOAc 8/2) to yield 4-chloro-2-methylthieno[2,3*d*]pyrimidine.

Yield 61%. Yellow solid mp 93 °C. ¹H NMR (200 MHz, CDCl₃)  $\delta$  = 7.48 (d, *J* = 6.0 Hz, 1 H), 7.35 (d, *J* = 6.0 Hz, 1 H), 2.79 (s, 3 H). ¹³C NMR (50 MHz, CDCl₃)  $\delta$  = 169.2, 163.0, 154.4, 126.8, 126.6, 119.5, 25.5. LC-MS (ESI, 35 eV): *t*_R = 1.84 min, *m/z* 185 [M+H]⁺.

#### 4.1.16. 2,2,2-Trichloro-N-(3-cyanothiophen-2-yl)acetamide (26)

To a solution of 2-aminothiophene-3-carbonitrile (1.08 g, 8.0 mmol) and trichloroacetyl chloride (0.89 mL, 8.0 mmol) in dioxane (15 mL), triethylamine (1.67 mL, 12.0 mmol) was added dropwise. The reaction was stirred at room temperature for 36 h. The aqueous phase was extracted with diethyl ether and the combined organic phases were washed with brine and dried (Na₂SO₄). The solvent was evaporated under reduced pressure and the residue was purified by silica gel column chromatography (eluent: Petroleum Ether/EtOAc 9/1) to afford 2,2,2-trichloro-N-(3-cyanothiophen-2-yl)acetamide.

Yield 76%. White solid. mp 127 °C. ¹H NMR (200 MHz, CDCl₃)  $\delta$  = 9.48 (s, 1 H), 7.08 (s, 2 H). ¹³C NMR (50 MHz, CDCl₃)  $\delta$  = 158.7, 147.0, 124.8, 120.5, 113.4, 96.8, 90.8. LC-MS (ESI, 35 eV):  $t_{\rm R}$  = 0.80 min, m/z 267 (M⁻). HRMS (ESI): m/z calcd. for C₇H₃Cl₃N₂OS [M+H]⁺: 268.9104. Found: 268.9110.

#### 4.1.17. 2-(Trichloromethyl)thieno[2,3-d]pyrimidin-4(3H)-one (27)

Compound 2,2,2-trichloro-*N*-(3-cyanothiophen-2-yl)acetamide **26** (809 mg, 3.0 mmol) was added to a mixture of phosphoric acid and polyphosphoric acid (60 mL, 1:1). The solution was stirred at rt for 1 h and then heated at 70 °C for 3 h. After cooling to room temperature, the mixture was poured onto ice. The precipitate was filtered, washed with ice-cold water and dried in a vacuum oven at 40 °C to yield 2-(trichloromethyl)thieno[2,3-*d*]pyrimidin-4(3*H*)-one. No further purification necessary.

Yield 90%. White solid. mp 195 °C. ¹H NMR (200 MHz, DMSO- $d_6$ )  $\delta$  = 8.23 (s, 1 H), 7.56 (d, J = 5.8 Hz, 1 H), 7.25 (d, J = 5.8 Hz, 1 H). ¹³C NMR (50 MHz, DMSO- $d_6$ )  $\delta$  = 167.4, 158.8, 144.6, 124.1, 119.2, 118.5, 91.6. LC-MS (ESI, 35 eV):  $t_{\rm R}$  = 0.69 min, m/z 267 [M-H]⁻.

#### 4.1.18. *N*-(2,4-Dichlorophenyl)-2-(trichloromethyl)thieno[2,3-*d*]pyrimidin-4amine (29a)

To a solution of 2,2,2-trichloro-*N*-(3-cyanothiophen-2-yl)acetamide **27** (200 mg, 0.74 mmol) and 2,4-dichloroaniline (118 mg, 0.74 mmol) in CH₂Cl₂ (0.3 mL), AlCl₃ (100 mg, 0.74 mmol) was added portionwise. The mixture was stirred at rt for 72 h. The solvent was evaporated under reduced pressure and the residue was purified by silica gel column chromatography (eluent: Petroleum Ether/EtOAc 7/3) to afford a white solid **28a** (LC-MS (ESI, 35 eV):  $t_{\rm R} = 4.00$  min, m/z 430 (M+H)⁺) which was dissolved in ethanol (2 mL) and the resulting solution was refluxed for 4 h. The solvent was evaporated under reduced pressure and the residue solution chromatography (eluent: Petroleum Ether/CH₂Cl₂ 8/2 to 5/5) to afford *N*-(2,4-dichlorophenyl)-2-(trichloromethyl)thieno[2,3-*d*]pyrimidin-4-amine.

Yield 36%. White solid. mp 181 °C. ¹H NMR (200 MHz, CDCl₃)  $\delta$  = 8.94 (d, *J* = 9.0 Hz, 1 H), 7.77 (bs, 1 H), 7.64 (d, *J* = 6.0 Hz, 1 H), 7.47 (d, *J* = 2.3 Hz, 1 H), 7.37 (dd, *J* = 5.4, 2.3 Hz, 2 H). ¹³C NMR (50 MHz, CDCl₃)  $\delta$  = 167.3, 160.2, 153.8, 133.5, 128.9, 128.8, 128.2, 127.9, 123.4, 122.6, 116.8, 116.2, 96.3. Anal. Calcd for C₁₃H₆Cl₅N₃S: C, 37.76; H, 1.46; N, 10.16. Found: C, 37.95; H, 1.52; N, 10.39. LC-MS (ESI, 35 eV): *t*_R = 5.26 min, *m*/*z* 412 [M+H]⁺.

#### 4.1.19. 2-(Trichloromethyl)-*N*-(3-(trifluoromethyl)phenyl)thieno[2,3-*d*]pyrimidin-4-amine (29b)

To a solution of 2,2,2-trichloro-*N*-(3-cyanothiophen-2-yl)acetamide **26** (320 mg, 1.18 mmol) and 3-(trifluoromethyl)aniline (200  $\mu$ L, 1.18 mmol) in CH₂Cl₂ (0.3 mL), AlCl₃ (160 mg, 1.18 mmol) was added portionwise. The mixture was stirred at rt for 72 h. The solvent was evaporated under reduced pressure and the residue **28b** (LC-MS (ESI, 35 eV):  $t_{\rm R}$  = 4.99 min, m/z 430 [M+H]⁺) was dissolved in ethanol (2.0 mL) and the solution was refluxed for 18 h. The solvent was evaporated under reduced pressure and the residue was purified by silica gel column chromatography (eluent: Petroleum Ether/CH₂Cl₂ 8/2 to 5/5) to afford 2-(Trichloromethyl)-*N*-(3-(trifluoromethyl)phenyl)thieno[2,3-*d*]pyrimidin-4-amine.

Yield 15%. Pale blue solid. mp 124 °C. ¹H NMR (200 MHz, CDCl₃)  $\delta$  = 8.42 (s, 1 H), 7.92 (d, *J* = 8.1 Hz, 1 H), 7.60 (d, *J* = 5.9 Hz, 1 H), 7.52 (t, *J* = 7.9 Hz, 1 H), 7.48-7.40 (m, 2 H), 7.35 (d, *J* = 5.9 Hz, 1 H). ¹³C NMR (50 MHz, CDCl₃)  $\delta$  = 167.4, 160.4, 154.3, 138.7, 132.0, 131.3, 129.7, 127.6, 123.4, 120.9, 117.7, 116.6, 116.4, 97.2. Anal. Calcd for C₁₄H₇Cl₃F₃N₃S: C, 40.75; H, 1.71; N, 10.18. Found: C, 41.03; H, 1.86; N, 10.24. LC-MS (ESI, 35 eV): *t*_R = 5.22 min, m/z 412 [M+H]⁺.

#### 4.1.20. 4-((4-Chlorophenyl)thio)-2-(trichloromethyl)thieno[2,3-d]pyrimidine (29d)

HBr_(g) was bubbled into a solution of 2,2,2-trichloro-*N*-(3-cyanothiophen-2yl)acetamide **26** (270 mg, 1.0 mmol) and 4-chlorothiophenol (145 mg, 1.0 mmol) in diethyl ether (2.5 mL), at -20 °C, for 30 min. The bubbling was stopped when the mixture became homogeneous (bright yellow solution). The reaction was then stirred at rt for 18 h. Water (2.0 mL) was then added and the aqueous phase was extracted with EtOAc. The organic phase was dried (Na₂SO₄) and the solvent was evaporated under reduced pressure. The residue was purified by silica gel column chromatography (eluent: Petroleum Ether/CH₂Cl₂ 8/2 + 5 mL of CH₂Cl₂ every 100 mL of eluent) to afford a white solid **28d** (86 mg) (LC-MS (ESI, 35 eV):  $t_R$ = 4.46 min, m/z 413 [M+H]⁺) which was dissolved in ethanol (1.0 mL) and the resulting solution was refluxed for 5 minutes. The solvent was evaporated under reduced pressure to yield 4-((4-chlorophenyl)thio)-2-(trichloromethyl)thieno[2,3-*d*]pyrimidine. No further purification necessary.

Yield 22%. White solid. mp 150 °C (dec.). ¹H NMR (200 MHz, CDCl₃)  $\delta$  = 7.70 (d, *J* = 6.0 Hz, 1 H), 7.61 (d, *J* = 8.6 Hz, 2 H), 7.46-7.42 (m, 3 H). ¹³C NMR (50, MHz, CDCl₃)  $\delta$  = 166.5, 164.6, 159.5, 136.8, 136.3, 129.8, 129.5, 126.7, 125.1, 118.8, 96.7. Anal. Calcd for C₁₃H₆Cl₄N₂S₂: C, 39.41; H, 1.53; N, 7.07. Found: C, 39.61; H, 1.61; N, 7.16. LC-MS (ESI, 35 eV):  $t_{\rm R}$  = 5.45 min, m/z 395 [M+H]⁺.

#### 4.1.21. *N*-(2,4-Dichlorophenyl)-2-trichloromethylquinoline-4-amine (30a)

To a solution of 4-chloro-2-(trichloromethyl)quinoline **H** [43] (281 mg, 1.0 mmol) and 2,4dichloroaniline (160 mg, 1.0 mmol) in ethanol (5.0 mL) were added a few drops of concentrated HCl. The vial was then sealed and heated at 100 °C overnight. The mixture was allowed to cool to rt and the organic phase was then washed with a solution of NaOH 2N. The aqueous phase was extracted with CH₂Cl₂. The combined organic phases were dried (Na₂SO₄) and the solvent was evaporated under reduced pressure. The residue was then purified by silica gel column chromatography (gradient: Petroleum Ether 100% + 10 mL of EtOAc every 100 mL of eluent) to afford 158 mg of the expected *N*-(2,4-dichlorophenyl)-2trichloromethylquinoline-4-amine.

Yield 39%. Orange powder. mp 121 °C. ¹H NMR (200 MHz, CDCl₃)  $\delta$  = 8.22 (d, *J* = 8.3 Hz, 1 H), 8.02 (d, *J* = 8.3 Hz, 1 H), 7.84-7.76 (m, 1 H), 7.68-7.64 (m, 2 H), 7.54-7.51 (m, 1 H), 7.47 (s, 1 H), 7.35-7.29 (m, 1 H), 7.11 (s, 1 H). ¹³C NMR (50 MHz, CDCl₃)  $\delta$  = 158.6, 147.5, 147.0, 135.4, 131.0, 130.3, 129.7, 128.3, 127.8, 126.9, 122.5, 119.8, 99.7, 98.1. Anal. Calcd for C₁₆H₉Cl₅N₂: C, 47.27; H, 2.23; N, 6.89. Found: C, 47.01; H, 2.28; N, 6.99. LC-MS (ESI, 35 eV): *t*_R = 5.10 min, m/z 403 [M-H]⁻.

4.1.22. 2-(Trichloromethyl)-*N*-[3-(trifluoromethyl)phenyl]quinoline-4-amine (30b) To a solution of 4-chloro-2-(trichloromethyl)quinoline H [43] (281 mg, 1.0 mmol) and 3-(trifluoromethyl)aniline (170 µL, 1.0 mmol) in ethanol (5.0 mL) were added a few drops of concentrated HCl. The vial was then sealed and heated at 100 °C overnight. The mixture was allowed to cool to rt and the organic phase was then washed with a solution of NaOH 2N. The aqueous phase was extracted with  $CH_2Cl_2$ . The combined organic phases were dried ( $Na_2SO_4$ ) and the solvent was evaporated under reduced pressure. The residue was then purified by silica gel column chromatography (gradient: Petroleum Ether/CH₂Cl₂ 8/2 + 5 mL of CH₂Cl₂ every 100 mL of eluent) 2-(trichloromethyl)-N-[3to afford (trifluoromethyl)phenyl]quinoline-4-amine.

Yield 73%. Yellow powder. mp 159 °C. ¹H NMR (200 MHz, CDCl₃)  $\delta$  = 8.22 (d, *J* = 8.3 Hz, 1 H), 8.12 (d, *J* = 8.3 Hz, 1 H), 7.75 (t, *J* = 7.7 Hz, 1 H), 7.69 (s, 1 H), 7.57-7.42 (m, 6 H). ¹³C NMR (50 MHz, CDCl₃)  $\delta$  = 157.8, 149.0, 146.4, 140.0, 132.8, 131.3, 130.6, 130.1, 127.6, 125.3, 121.7, 120.3, 119.3, 119.1, 119.0, 98.8, 97.5. Anal. Calcd for C₁₇H₁₀Cl₃F₃N₂: C, 50.34; H, 2.48; N, 6.91. Found: C, 50.73; H, 2.69; N, 6.66. LC-MS (ESI, 35 eV) *t*_R = 5.09 min, *m*/*z* 403 [M-H]⁻.

#### 4.1.23. 4-(4-Chlorophenoxy)-2-(trichloromethyl)quinoline (30c)

To a solution of 4-chlorophenol (141 mg, 1.1 mmol) in DMSO (1.0 mL), NaH (48 mg, 2.0 mmol) was added portionwise under inert atmosphere. The mixture was stirred at rt for 20 min. 4-chloro-2-(trichloromethyl)quinoline **H** [43] (281 mg, 1.0 mmol) was then added portionwise and the reaction was stirred for 24 h at 50 °C. The aqueous phase was then extracted with  $CH_2Cl_2$  and the organic phase was washed with water, dried (Na₂SO₄) and the solvent was evaporated. The residue was then purified by silica gel column chromatography (eluent Petroleum Ether/CH₂Cl₂ 6/4) to afford 4-(4-chlorophenoxy)-2-(trichloromethyl)quinolone.

Yield 72%. White solid. mp 166 °C. ¹H NMR (200 MHz, CDCl₃)  $\delta$  = 8.34 (d, *J* = 8.3Hz, 1 H), 8.20 (d, *J* = 8.6 Hz, 1 H), 7.89-7.81 (m, 1 H), 7.71-7.64 (m, 1 H), 7.48 (d, *J* = 8.8 Hz, 2 H), 7.22 (s, 1 H), 7.18 (d, *J* = 8.8 Hz, 2 H). ¹³C NMR (50 MHz, CDCl₃)  $\delta$  = 162.6, 158.4, 152.6, 147.4, 131.3, 130.6, 130.0, 128.0, 122.2, 121.6, 121.0, 100.6, 97.7. Anal. Calcd for C₁₆H₉Cl₄NO: C, 51.51; H, 2.43; N, 3.75. Found: C, 51.69; H, 2.53; N, 3.81. LC-MS (ESI, 35 eV): *t*_R = 5.52 min, *m*/*z* 372 [M+H]⁺.

#### 4.1.24. 4-((4-Chlorophenyl)thio)-2-(trichloromethyl)quinoline (30d)

To a solution of 4-chlorothiophenol (159 mg, 1.1 mmol) in DMSO (1.0 mL), NaH (48 mg, 2.0 mmol) was added portionwise under inert atmosphere. The mixture was stirred at rt for 20

min. 4-Chloro-2-(trichloromethyl)quinoline **H** [43] (281 mg, 1.0 mmol) was then added portionwise and the reaction was stirred for 24 h at rt. The aqueous phase was then extracted with  $CH_2Cl_2$  and the organic phase was washed with water, dried (Na₂SO₄) and the solvent was evaporated. The residue was then purified by silica gel column chromatography (eluent: Petroleum Ether/CH₂Cl₂ 8/2) to afford 4-((4-chlorophenyl)thio)-2-(trichloromethyl)quinolone.

Yield 32%. White powder. mp 151 °C. ¹H NMR (200 MHz, CDCl₃)  $\delta$  = 8.24-8.18 (m, 2 H), 7.87-7.78 (m, 1 H), 7.72-7.64 (m, 1 H), 7.57-7.46 (m, 5 H). ¹³C NMR (50 MHz, CDCl₃)  $\delta$  = 156.8, 150.2, 145.4, 136.4, 136.1, 131.0, 130.8, 130.5, 129.3, 128.4, 127.7, 125.6, 123.2, 114.1, 97.7. Anal. Calcd for C₁₆H₉Cl₄NS: C, 49.39; H, 2.33; N, 3.60. Found: C, 49.49; H, 2.23; N, 3.53. LC-MS (ESI, 35 eV):  $t_{\rm R}$  = 5.76 min, m/z 388 [M+H]⁺.

#### 4.1.25. N-(2,4-Dichlorophenyl)-3-methylquinoxalin-2-amine (31a)

To a solution of 2-chloro-3-methylquinoxaline **L** [45] (1.0 g, 5.6 mmol) and 2,4dichloroaniline (0.91 g, 5.6 mmol) in anhydrous DMF (10 mL),  $Cs_2CO_3$  (1.82 g, 5.6 mmol) was added under inert atmosphere. The mixture was stirred at 70 °C for 24 h. After cooling, water then CH₂Cl₂ were added. The organic layer was then washed five times with water and dried with Na₂SO₄. After filtration and evaporation, the resulting solid was purified by silica gel column chromatography (eluent: Petroleum Ether/CH₂Cl₂ 7/3 then 1/1) to afford *N*-(2,4dichlorophenyl)-3-methylquinoxalin-2-amine.

Yield 19%. White powder. mp 152 °C. ¹H NMR (250 MHz, CDCl₃)  $\delta$  = 9.06 (d, *J* = 9.0 Hz, 1H), 7.90 (d, *J* = 8.2 Hz, 1H), 7.83 (d, *J* = 8.2 Hz, 1H), 7.67- 7.57 (m, 1H), 7.57- 7.30 (m, 4H), 2.78 (s, 3H). ¹³C NMR (63 MHz, CDCl₃)  $\delta$  = 147.5, 145.2, 140.4, 138.0, 135.3, 129.6, 128.8, 128.4, 128.0, 127.6, 127.0, 126.4, 123.2, 121.4, 21.1. LC-MS (ESI, 35 eV): *t*_R = 5.49 min, m/z 304 [M+H]⁺.

#### 4.1.26. 3-Methyl-*N*-(3-(trifluoromethyl)phenyl)quinoxalin-2-amine (31b)

A solution of 2-chloro-3-methylquinoxaline L [45] (500 mg, 2.8 mmol) and 3trifluoromethylaniline (2,5 mL, 20 mmol) in anhydrous DMF (10 mL) was heated at 140 °C in a sealed vial under microwave irradiation. After completion of the reaction (45 min),  $CH_2Cl_2$  was added, and the organic phase was washed successively with 1M HCl and brine. The organic layer was then dried with Na₂SO₄, filtered and evaporated. The resulting solid was purified by silica gel column chromatography (eluent:  $CH_2Cl_2$ ) to afford 3-Methyl-N-(3-(trifluoromethyl)phenyl)quinoxalin-2-amine.

Yield 69%. Amber powder. mp 98 °C. ¹H NMR (250 MHz, CDCl₃)  $\delta$  = 8.24 (s, 1H), 7.93 (d, *J* = 8.1 Hz, 1H), 7.87 (dd, *J* = 8.1, 1.1 Hz, 1H), 7.78 (dd, *J* = 8.2, 1.0 Hz, 1H), 7.64 –

7.53 (m, 1H), 7.52-7.39 (m, 2H), 7.37- 7.27 (m, 1H), 6.77 (bs, 1H, NH), 2.65 (s, 3H). ¹³C NMR (63 MHz, CDCl₃)  $\delta$  = 147.7, 144.6, 140.3 (d, *J* = 3.2 Hz), 137.9, 131.5 (q, *J* = 33.1 Hz), 129.4 (d, *J* = 6.0 Hz), 128.2, 126.9, 126.1, 124.3 (q, *J* = 272.9 Hz), 122.8, 119.6 (d, *J* = 3.8 Hz), 116.6 (d, *J* = 3.4 Hz), 21.03. LC-MS (ESI, 35 eV):  $t_{\rm R}$  = 4.75 min, m/z 304 [M+H]⁺.

#### 4.1.27. 2-(4-Chlorophenoxy)-3-methylquinoxaline (31c)

To a solution of 2-chloro-3-methylquinoxaline **L** [45] (650 mg, 3.6 mmol) and 4chlorophenol (0.37 mL, 3.6 mmol) in anhydrous DMF (15 mL),  $Cs_2CO_3$  (1.19 g, 3.6 mmol) was added under inert atmosphere. The mixture was stirred at 70 °C overnight. After completion of the reaction, water was added, leading to a precipitate which was separated by filtration. The resulting precipitate was then thoroughly washed with water. The precipitate was dissolved in  $CH_2Cl_2$  and dried with  $Na_2SO_4$ . After filtration and evaporation, the resulting solid was purified by silica gel column chromatography (eluent: Petroleum Ether/CH₂Cl₂ 1/1) to afford 2-(4-chlorophenoxy)-3-methylquinoxaline.

Yield 85%. Off-white powder. mp 108 °C. ¹H NMR (250 MHz, CDCl₃)  $\delta$  = 7.69 (dd, *J* = 6.1, 3.7 Hz, 1H), 7.57 (dd, *J* = 6.3, 3.5 Hz, 2H), 7.46-7.38 (m, 2H), 7.28 – 7.19 (m, 2H), 2.81 (s, 3H). ¹³C NMR (63 MHz, CDCl₃)  $\delta$  = 155.8, 151.5, 147.9, 139.6, 139.4, 130.6, 129.7, 129.4, 128.1, 127.6, 127.4, 123.3, 20.7. LC-MS (ESI, 35 eV): *t*_R = 4.35 min, m/z 271 [M+H]⁺.

#### 4.1.28. 2-((4-Chlorophenyl)thio)-3-methylquinoxaline (31d)

To a solution of 2-chloro-3-methylquinoxaline **L** [45] (500 mg, 2.8 mmol) and 4chlorothiophenol (405 mg, 2.8 mmol) in anhydrous DMF (10 mL),  $Cs_2CO_3$  (912 mg, 2.8 mmol) was added under inert atmosphere. The mixture was stirred at 70 °C overnight. After completion of the reaction, water was added, leading to a precipitate which was separated by filtration. The resulting precipitate was then thoroughly washed with water. The precipitate was dissolved in  $CH_2Cl_2$  and dried with  $Na_2SO_4$ . After filtration and evaporation, the resulting solid was purified by silica gel column chromatography (eluent: Petroleum Ether/CH₂Cl₂ 1/1) to afford 2-((4-chlorophenyl)thio)-3-methylquinoxaline.

Yield 83%. Beige powder. mp 118 °C. ¹H NMR (250 MHz, CDCl₃)  $\delta$  = 7.96-7.92 (m, 1H), 7.72-7.68 (m, 1H), 7.60-7.54 (m, 4H), 7.44 (d, *J* = 6.6 Hz, 2H), 2.77 (s, 3H). ¹³C NMR (101 MHz, CDCl₃)  $\delta$  = 155.3, 151.4, 141.4, 140.0, 136.7, 135.6, 129.5, 129.2, 128.6, 128.3, 128.1, 127.2, 22.4. LC-MS (ESI, 35 eV): *t*_R = 5.51 min, m/z 287 [M+H]⁺.

#### 4.1.29. 2-(4-Chlorophenoxy)-3-(trichloromethyl)quinoxaline (32c)

To a solution of 2-(4-chlorophenoxy)-3-methylquinoxaline **31c** (624 mg, 2.8 mmol) and PCl₅ (2.88 g, 16.8 mmol), POCl₃ was added to make a slurry (ca 5 mL). The mixture was then

heated in a multimode microwave oven at 100 °C, 800 W for 20 min. After completion of the reaction, the mixture was poured into ice and was then neutralized with  $Na_2CO_3$ . The resulting solution was extracted with  $CH_2Cl_2$  and dried with  $Na_2SO_4$ . After filtration and evaporation, the resulting solid was purified by silica gel column chromatography (eluent: Petroleum Ether/CH₂Cl₂ 7/3) to afford 2-(4-chlorophenoxy)-3-(trichloromethyl)quinoxaline.

Yield 72%. White powder. mp 108 °C. ¹H NMR (250 MHz, CDCl₃)  $\delta$  = 8.11 (d, *J* = 8.0 Hz, 1H), 7.72-7.63 (m, 3H), 7.38 (dd, *J* = 6.6, 3.5 Hz, 2 H), 7.22 (dd, *J* = 6.6, 3.5 Hz, 2 H). ¹³C NMR (101 MHz, CDCl₃)  $\delta$  = 152.5, 151.2, 142.4, 141.2, 137.0, 132.3, 131.1, 129.9, 129.8, 128.8, 127.1, 123.2, 94.9. LC-MS (ESI, 35 eV): *t*_R = 5.63 min, (no ionization). HRMS (ESI): m/z calcd. for C₁₅H₈Cl₄N₂O [M+H]⁺: 372.9464. Found: 372.9461. Anal. Calcd for C₁₅H₈Cl₄N₂O: C, 48.16; H, 2.16; N, 7.49. Found: C, 48.37; H, 2.32; N, 7.42.

#### 4.1.30. 2-((4-Chlorophenyl)thio)-3-(trichloromethyl)quinoxaline (32d)

To a solution of 2-((4-chlorophenyl)thio)-3-methylquinoxaline **31d** (623 mg, 2.2 mmol) and PCl₅ (2.7 g, 13.0 mmol), POCl₃ was added to make a slurry (ca 6 mL). The mixture was then heated in a multimode microwave oven at 100 °C, 800 W for 20 min. After completion of the reaction, the mixture was poured into ice and was then neutralized with Na₂CO₃. The resulting solution was extracted with CH₂Cl₂ and dried with Na₂SO₄. After filtration and evaporation, the resulting solid was purified by silica gel column chromatography (eluent: Petroleum Ether/CH₂Cl₂ 9/1) to afford 2-((4-chlorophenyl)thio)-3-(trichloromethyl)quinoxaline.

Yield 72%. Yellow powder. mp 92 °C. ¹H NMR (250 MHz, CDCl₃)  $\delta$  = 8.15- 8.04 (m, 1H), 7.76-7.64 (m, 3H), 7.56 (d, *J* = 8.6 Hz, 2H), 7.43 (d, *J* = 8.6 Hz, 2H). ¹³C NMR (63 MHz, CDCl₃)  $\delta$  = 152.9, 147.7, 142.0, 137.2, 137.0, 136.0, 132.1, 129.9, 129.8, 129.5, 128.7, 127.8, 96.7. LC-MS (ESI, 35 eV):  $t_{\rm R}$  = 6.59 min, m/z 389 [M+H]⁺. Anal. Calcd for C₁₅H₈Cl₄N₂S: C, 46.18; H, 2.07; N, 7.18. Found: C, 46.33; H, 2.26; N, 6.97.

#### 4.2. Biology

#### 4.2.1. In vitro Antiplasmodial evaluation

In this study, a K1 culture-adapted *P. falciparum* strain resistant to chloroquine, pyrimethamine and proguanil was used in an *in vitro* culture. Maintenance in continuous culture was done as described previously by Trager and Jensen [48]. Cultures were maintained in fresh A+ human erythrocytes at 2.5% hematocrit in complete medium (RPMI 1640 with 25 mM HEPES, 25 mM NaHCO₃, 10% of A+ human serum) at 37 °C under reduced O₂ atmosphere (gas mixture 10% O₂, 6% CO₂, and 84% N₂). Parasitaemia was maintained daily between 1% and 6%. The *P. falciparum* drug susceptibility test was carried

out by comparing quantities of DNA in treated and control cultures of parasite in human erythrocytes according to a SYBR Green I fluorescence-based method [49] using a 96-well fluorescence plate reader. Compounds, previously dissolved in DMSO (final concentration less than 0.5% v/v) were incubated in a total assay volume of 200 µL of synchronized culture suspension (2% hematocrit and 0.4% parasitaemia) for 72 h in a humidified atmosphere (84%  $N_2$ , 10%  $O_2$  and 6%  $CO_2$ ) at 37 °C, in 96-well flat bottom plates. Duplicate assays were performed for each sample. After incubation, 125 µL supernatant was discarded and cells were washed twice with 125 µL 1X PBS. 15 µL re-suspended cells were transferred to 96well flat bottom nonsterile black plates (Greiner Bio-one) already containing 15 µL of the SYBR Green I lysis buffer (2X SYBR Green I, 20 mM Tris base pH 7.5, 20 mM EDTA, 0.008% w/v saponin, 0.08% w/v Triton X-100). Negative control, treated by solvents (DMSO or H₂O) and positive controls (atovaquone, chloroquine and doxycycline) were added to each set of experiments. Plates were incubated for 15 min at 37 °C and then read on a TECAN Infinite F-200 spectrophotometer with excitation and emission wavelengths at 485 and 535 nm, respectively. The concentrations of compounds required to induce a 50% decrease of parasite growth ( $IC_{50}$  K1) were calculated from three independent experiments.

#### 4.2.2. In vitro Cytotoxicity evaluation

HepG2 cell line was maintained at 37 °C, 6% CO2 with 90% humidity in RPMI supplemented with 10% feetal bovine serum, 1% L-glutamine (200 mM) and penicillin (100 U/mL) / streptomycin (100 µg/mL) (complete RPMI medium). The evaluation of the tested molecules cytotoxicity on the HepG2 (hepatocarcinoma cell line purchased from ATCC, ref HB-8065) cell line was performed according to the method of Mosmann [50] with slight modifications. Briefly,  $5.10^3$  cells in 100 µL of complete medium were inoculated into each well of 96-well plates and incubated at 37 °C in a humidified 6% CO₂. After 24 h incubation, 100 µL of medium with various product concentrations dissolved in DMSO (final concentration less than 0.5% v/v) were added and the plates were incubated for 72 h at 37 °C. Triplicate assays were performed for each sample. Each plate-well was then microscope-examined for detecting possible precipitate formation before the medium was aspirated from the wells. 100 µL of MTT (3-(4,5-dimethyl-2-thiazolyl)-2,5-diphenyl-2H-tetrazolium bromide) solution (0.5 mg/mL in medium without FCS) were then added to each well. Cells were incubated for 2 h at 37 °C. After this time, the MTT solution was removed and DMSO (100 µL) was added to dissolve the resulting blue formazan crystals. Plates were shaken vigorously (700 rpm) for 10 min. The absorbance was measured at 570 nm with 630 nm as reference wavelength using a

BIO-TEK ELx808 Absorbance Microplate Reader. DMSO was used as blank and doxorubicin (purchased from Sigma Aldrich) as positive control. Cell viability was calculated as percentage of control (cells incubated without compound). The 50% cytotoxic concentration ( $CC_{50}$ ) was determined from the dose–response curve by using the TableCurve software 2D v.5.0.  $CC_{50}$  values represent the mean value calculated from three independent experiments.

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#### **Highlights:**

► Antiplasmodial pharmacomodulation of CCl₃-substituted-nitrogen containing heterocycles was made. ► Thienopyrimidine derivatives appeared more cytotoxic. ► Original 3-substituted-2-trichloromethylquinoxaline analogs were prepared. ► Two quinoxaline derivatives displayed *in vitro* IC₅₀ values of 0.4 and 0.5 µM on the K1 multi-resistant *P. falciparum* strain. ► Cytotoxicity was assessed on the human HepG2 cell line showing low cytotoxicity (CC₅₀ ~ 40 µM) and improved selectivity indecies (77-100).