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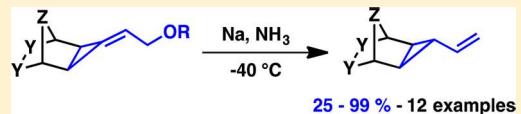
Preparation of Vinylcyclopropanes by Sodium Mediated Reductive Isomerization of Methylenecyclopropanes

Karel Le Jeune, Sabine Chevallier-Michaud, David Gatineau, Laurent Giordano, Alphonse Tenaglia, and Hervé Clavier*

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Supporting Information

ABSTRACT: We disclosed therein a new reaction of reductive isomerization of methylenecyclopropanes (MCPs) to vinylcyclopropanes (VCPs). On treatment with sodium metal in liquid ammonia, MCPs bearing a C—O bond at allylic position undergo both a reductive cleavage of the C—O bond and an isomerization of the C—C double bond giving rise to VCPs. The scope of the reductive isomerization was investigated and showed a broad applicability since various functional groups are tolerated. MCP substrates were straightforwardly prepared by a palladium-promoted [2 + 1] cycloaddition between norbornene derivatives with alkynes.



25 - 99 % - 12 examples

Vinylcyclopropanes (VCPs) are versatile synthons in various organic transformations,¹ mainly cycloadditions, as well as important structural motifs in natural product² such as chrysanthemic acid, dictyopterenes, ambruticin S or lindenane-type sesquiterpenoids. Therefore, their synthesis has attracted much attention and has been translated into the development of several methodologies involving the formation of cyclopropane ring: cyclopropanations between either a diazo compound and a 1,4 diene or an allylic diazo compound and an alkene,³ ring rearrangement,⁴ Michael addition or substitution.⁵ Vinylcyclopropanes can be also prepared by functional group transformations from cyclopropyl precursors.⁶ In 1982, Catellani and Chiusoli reported a palladium-promoted vinylcyclopropane formation starting from norbornene and 1-bromo-1-octene (**Scheme 1**).⁷ The reaction occurring by two consecutive insertions of palladium species and subsequent β -H elimination was unfortunately not exemplified. Alternatively, Murakami⁸ and Lautens⁹ reported independently an interesting rhodium-promoted vinylcyclopropanation of strained olefins using vinyl boronate derivatives. Herein, we report a new approach for the preparation of the VCP substructure, which consists in a two steps procedure: a [2 + 1] cycloaddition followed by a reductive isomerization (**Scheme 1**).

In 2011, we disclosed an unprecedented intermolecular tandem [2 + 1]/[3 + 2] cycloaddition sequence catalyzed by platinum-phosphinito-phosphinous acid complexes **3** between norbornadiene and alkynes giving rise to tricyclic compounds **2**.¹⁰ Further investigations showed that the reaction occurred in two steps (**Scheme 2**). First, a [2 + 1] cycloaddition promoted by catalyst **3**,¹¹ or palladium analogues *in situ* generated from Pd(OAc)₂ and secondary phosphine oxides, afforded methylenecyclopropanes (MCPs) **1**. Then, platinum-based complex **3** was able to achieve the diastereoselective [3 + 2] cycloaddition between **1a** and another alkyne partner with the proviso that MCPs **1** bear an oxygen substituent on

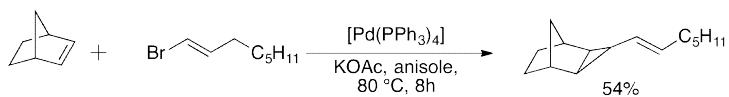
the propargylic carbon atom. Whereas these [2 + 1] cycloadditions have been well studied in our laboratories¹² and by others,¹³ the platinum-promoted [3 + 2] cycloaddition has not been further investigated so far.¹⁴

To broaden the scope of this reaction, we wondered whether we could achieve the [3 + 2] cycloaddition in absence of the bicyclo[2.2.1]heptane moiety linked to the MCP. The ring strain of the norbornene pattern might enhance the reactivity of MCP derivatives. To that end, we planned to prepare suitable substrates by a [2 + 1] cycloaddition of diazabicyclic alkene and propargyl acetate to afford quantitatively the adduct **1b** and attempted the reductive cleavage of the N—N bond. Following the reported conditions for the reductive N—N cleavage of hydrazines,^{15,16} cyclic hydrazine **1b** was treated with 2.5 equiv of sodium in liquid ammonia and no traces of the expected product **4b** could be detected (**Scheme 3**). A new compound was isolated and identified as **5b**. Its formation resulted from a reductive isomerization of the MCP with release of the acetoxy group by C—O bond cleavage.¹⁷ Of note, **5b** was obtained as a single diastereomer, with the vinyl substituent *trans* to other substituents of the cyclopropane. Increasing the quantity of sodium to 5 equiv led effectively to the cleavage of the N—N bond, which is subsequent to the formation of the VCP since compound **6b** was isolated in 55%. Single-crystal X-ray analysis of **6b** unambiguously confirmed the atom connectivity and the relative stereochemistry of all substituents (**Figure 1**).

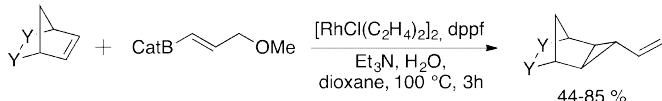
This methodology represents an interesting alternative to rhodium-promoted vinylcyclopropanation of olefins developed by Murakami⁸ and Lautens;⁹ therefore, we decided to pursue the investigation of this original reactivity of MCPs.¹⁸ We started by the preparation of a wide range of [2 + 1] cycloadducts through the reaction of norbornene derivatives

Scheme 1. Preparation of Vinylcyclopropanes for Norbornene Derivatives

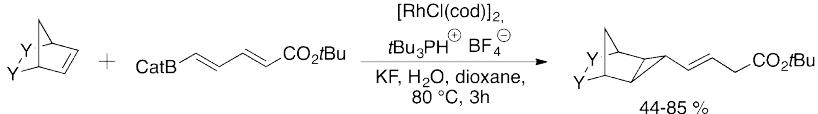
Catellani & Chiusoli



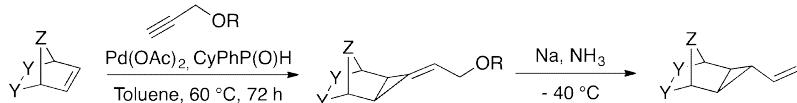
Murakami



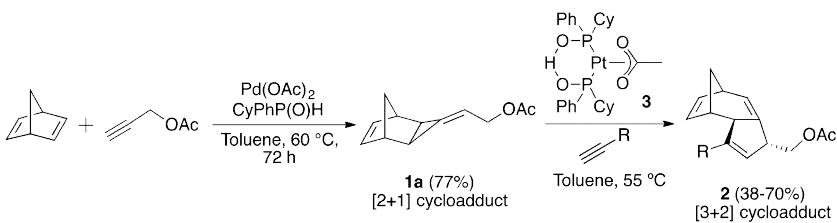
Lautens



This work



Scheme 2. Platinum-Catalyzed [3 + 2] Cycloadditions



Scheme 3. Reductive N–N Cleavage of Cyclic Hydrazine 1b

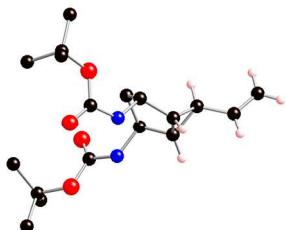
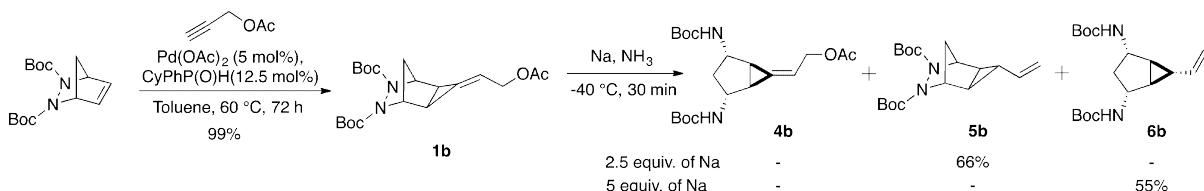


Figure 1. Ball-and-stick representation of vinylcyclopropane **6b** obtained by X-ray analysis (most of the hydrogen atoms have been omitted for clarity).

with alkynes catalyzed by $\text{Pd}(\text{OAc})_2$ in association with secondary phosphine oxides ($\text{PhCyP}(\text{O})\text{H}$) (Table 1). Various functional groups are well tolerated and, in a general manner, good yields were obtained with the exception of cycloadduct **1f** (entry 4). Of note, due to the *E/Z* isomerism of the C–C double bond and the presence of a stereogenic center, **1h** was isolated as an inseparable 1:1 mixture of two diastereomers (entry 6).

We then examined several reducing agents and reaction conditions using the benchmark substrate **1d** (Table 2). With the use of sodium in liquid ammonia at -40°C , the expected reduced product **5d** was isolated in an excellent yield and a good diastereoselectivity since small amounts of the cyclopropane with a *cis* arrangement were detected (entry 1). When the reaction was performed under the reaction conditions of the Bouveault-Blanc reduction,¹⁹ with absolute ethanol as proton source, only a low yield in **5d** was obtained (entry 2). The major product isolated is the alcohol **1n** resulting from the reduction of the ester function. The use of sodium naphthalenide²⁰ in THF at room temperature led to the selective formation of alcohol **1n** in quantitative yield (entry 3). Samarium(II) iodide was also tested but in absence of a protic solvent no reaction occurred, and with methanol a low yield in **1n** was obtained (entries 4 and 5).

Having established the optimal reaction conditions, we further investigated the scope of reaction with a range of ethylenecyclopropanes bearing an acetate group (Table 3). Bicyclic hydrazines **1b** and **1c** gave the corresponding VCP derivatives with moderate yields but an excellent diastereoselectivity (entries 1 and 2). With the substrate **1e**, a

Table 1. Preparation of Alkylidenecyclopropanes by [2 + 1] Cycloaddition^a

The general reaction scheme shows a norbornene derivative (with substituents Y and Z) reacting with an alkyne (R-CH=CH₂) in the presence of Pd(OAc)₂ (5 mol %) and CyPhP(O)H (12.5 mol %) in Toluene at 60 °C for 72 h. The product is an alkylidenecyclopropane where the norbornene double bond has been replaced by a C=C double bond, and the alkyne has added across it.

Entry	[2+1] Cycloadduct	Yield (%)
1		71
2		69
3		54
4		33
5		56
6		99
7		80
8		62
9		63
10		99
11		86

^aReaction conditions: norbornene derivative/alkyne 1:3, Pd(OAc)₂ (5 mol %), CyPhP(O)H (12.5 mol %), toluene, 60 °C, 72 h.

complex mixture was obtained, probably due to the competitive reduction of the norbornene double bond (entry 4).²¹ To our delight, VCP **1o**, prepared through chemoselective epoxidation of **1a**,²² was quantitatively

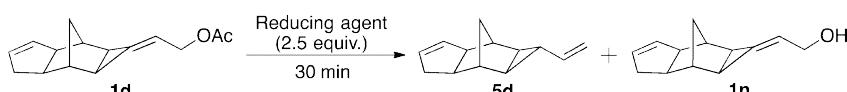
converted to VCP **5o** with high diastereoselectivity (entry 5). The reductive isomerization was found also tolerant to other ether functions such as the 7-oxa bicyclo[2.2.1]heptane skeleton (entry 6). However, the presence of cyano groups decreased significantly both yield and diastereoselectivity (entry 7). This might be the result of competitive cyano groups reduction. Despite a good selectivity, with the phenyl-substituted compound **1h**, only a low yield could be reached due to the formation of the over-reduced product **7h** in 62% yield (entry 8). Unfortunately, it was not possible to separate **5h** from **7h** by silica gel chromatography.

We then studied the influence of the nature of the allylic substituent on the transformation outcome (Table 4). The replacement of the acetate group with a benzoate led to diminish slightly both yield and selectivity and required a 2-fold excess of sodium metal (entries 1 and 2). In addition to carboxylates, the phenoxy group was found to be effective (entry 3). With the benzyl ether substrate **1k**, only the deprotection delivering alcohol **1n** was observed (entry 4). Several alcohols have been also tested and satisfactorily corresponding VCPs have been isolated in moderate yields and modest to good diastereoselectivities (entries 5–7). This includes the gem-dimethyl substituted alcohol **1l** that, with 6 equiv of sodium, gave rise to the 2-methyl-propen-1-yl-cyclopropane **5l**. The modest yields obtained with **1n** and **1l** (entries 5 and 6) can be explained by the fact that, even with 6 equiv of sodium, reactions did not reach completion and starting material alcohols were recovered. This suggests that the alkoxide formation is competing pathway to the reductive isomerization. In absence of the allylic substituent, such as for the benzylidene cyclopropane **1m**, only the fully diastereoselective reduction the C–C double bond occurred leaving the cyclopropane ring intact (entry 8).

As depicted in Scheme 4, we assumed that the reaction mechanism involves a first single-electron transfer to generate the radical anion intermediate **B**. An elimination of the acetate group leads to the formation of vinyl species **C** and a second electron transfer gives **D**, which upon protonolysis releases the VCP **E**.

In summary, we developed a new transformation allowing the formation of vinylcyclopropanes from MCPs. The treatment of the ethylidene cyclopropanes bearing an allylic group with sodium metal in liquid ammonia led to a reductive isomerization and generated VCPs with moderate to quantitative both yields and diastereoselectivities. This reaction was found tolerant to various function groups to the exception of those highly reducible. We are now exploring the reactivity of the VCP derivatives.

Table 2. Investigation of the Reducing Agent

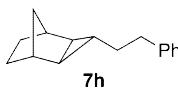


entry	reducing agent	conditions	yield (%)	
			5d (d.r.)	1n
1	Na	NH ₃ , -40 °C	86 (8.5:1)	-
2	Na	NH ₃ , EtOH, 25 °C	30 (1:0)	67
3	Na ⁺ C ₁₀ H ₈ ^{-•}	THF, 25 °C	-	98
4	Sml ₂	THF, 25 °C	-	-
5	Sml ₂	THF, MeOH, 25 °C	-	22

Table 3. Reductive Isomerization with a Variety of MCPs^a

Entry	Substrate	Product	Yield (d.r.)
1			66% (>20:1)
2			55% (>20:1)
3			86% (8.5:1)
4			Complex mixture
5			99% (>20:1)
6			64% (10:1)
7			25% (3.3:1)
8 ^b			12% (>20:1)

^aReaction conditions: substrate (0.5 mmol), Na (29 mg, 1.25 mmol, 2.5 equiv), NH₃ (20 mL). ^b62% of over-reduced compound **7h** was also isolated as unseparable mixture with **5h**.



EXPERIMENTAL SECTION

General Considerations. All reagents were obtained from commercial sources and used as received. Solvents (THF, DCM, toluene, and Et₂O) were purified and dried over solvent purification system or dried by standard procedures prior to use.²³ Petroleum ether (PE) (fraction between 40 and 60 °C) and ethyl acetate used for chromatographic separation were used as technical grade. Unless otherwise stated, all reactions were carried out in an atmosphere of dry nitrogen or argon using oven-dried (120 °C) glasswares. Analytical TLCs were performed on ready-made plates coated with silica gel on aluminum. Products were visualized by ultraviolet light and treatment either with *p*-anisaldehyde or phosphomolybdic acid stain followed by gentle heating or with iodine stain. Flash chromatography was performed using silica gel 60 (230–400 mesh). ¹H and ¹³C NMR spectra were recorded in CDCl₃ at ambient temperature on spectrometers operating at 300 or 400 MHz for ¹H. ¹³C nucleus was observed with ¹H decoupling. Solvent residual signals were used as internal standard.²⁴ Chemical shifts (δ) and coupling constants (J) are given in ppm and Hz, respectively. The peaks patterns are indicated as the following format multiplicity (s, singlet; d, doublet; t, triplet; q, quartet; sept, septuplet; m, multiplet; dd, doublet of doublet; dt, doublet of triplet; dm, doublet of multiplet, etc.). The prefix br. indicates a broadened signal. HRMS were recorded on spectrometers equipped with an Atmospheric

Pressure Ionization (API) source. Mass spectra were obtained a Time Of Flight (TOF) analyzer. X-ray Diffraction: intensity data were collected on a diffractometer using Mo K α radiation (0.71073 Å) at 293(2) K. The CIF file of compound **6b** has been deposited with the CCDC number 1048899.

General Procedure A: Palladium-Catalyzed [2 + 1] Cycloaddition. In a flame-dried Schlenk, Pd(OAc)₂ (5 mol %) and PhCyP(O)H (12.5 mol %) were introduced under argon and dissolved in dry and degassed toluene. The resulting yellow solution was stirred at 60 °C for 30 min; the yellow color disappeared. Then, the norbornene derivative, alkyne (3 equiv), and dry and degassed toluene were added. The resulting mixture was stirred at 60 °C for 72 h. Then, the volatiles were removed under reduced pressure. The crude product was purified on silica gel chromatography to obtain the desired product.

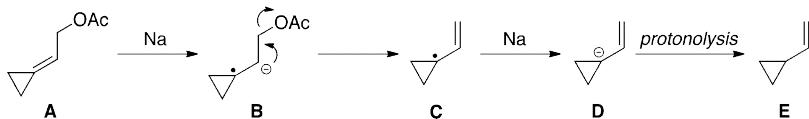
(*1R*,4S*,5S**)-Tricyclo[3.2.1.0^{2,4}]oct-6-en-3-ylidene)ethyl Acetate **1a**.¹¹ Following the general procedure A, after purification on silica gel chromatography, the product was obtained as a colorless oil (653 mg, 77% yield). ¹H NMR (400 MHz, CDCl₃): δ = 6.36 (br s, 2H), 5.74 (tt, J = 1.1 and 6.5 Hz, 1H), 4.60–4.70 (m, 2H), 3.03 (d, J = 6.0 Hz, 2H), 2.05 (s, 3H), 1.57 (br s, 2H), 1.03 (d, J = 8.5 Hz, 1H), 0.91 (d, J = 8.5 Hz, 1H). ¹³C NMR (100 MHz, CDCl₃): δ = 170.9 (s), 144.7 (s), 139.5 (d), 139.3 (d), 111.3 (d), 64.3 (t), 44.4 (d), 44.2 (d), 42.2 (t₂), 26.9 (d), 26.5 (d), 21.0 (q).

Table 4. Study of the Leaving Group^a

Entry	Substrate	Product	Yield (d.r.)
1			86% (8.5:1)
2 ^b			77% (5:1)
3			78% (5:1)
4			65%
5			60% (4:1)
6 ^b			43% (6:1)
7			56% (>20:1)
8			99% (>20:1)

^aReaction conditions: substrate (0.5 mmol), Na (29 mg, 1.25 mmol, 2.5 equiv), NH₃ (20 mL). ^b6 equiv of Na were used.

Scheme 4. Postulated Reaction Mechanism for the Reductive Isomerization



(1*R*^{*},2*S*^{*},5*S*^{*})-Di-*tert*-butyl 3-(2-acetoxymethylene)-6,7-diazatricyclo[3.2.1.0^{2,4}]octane-6,7-dicarboxylate 1b. Following the general procedure A, after purification on silica gel chromatography, the product was obtained as a white solid (1.46 g, 97% yield). Mp = 84–85 °C. ¹H NMR (400 MHz, CDCl₃) δ = 5.90 (m, 1H), 4.82 (br s, 1H), 4.58–4.72 (m, 2H), 4.55 (br s, 1H), 2.08 (s, 3H), 1.97 (br s, 1H), 1.49 (s, 18H), 1.18–1.36 (m, 2H), 0.80–0.90 (m, 1H). ¹³C NMR (100 MHz, CDCl₃) δ = 170.5 (s) 157.1 (br s), 156.1 (br s), 127.9 (s), 115.8 (d), 81.3 (s, 2C), 63.5 (t), 62.0 (d, 2C), 30.3 (t), 28.0 (q, 6C), 20.7 (q), 19.4 (br d), 18.0 (br d). HRMS (ESI-MS) *m/z*: [M + H]⁺, calcd for C₂₀H₃₁N₂O₆⁺, 395.2177; found, 395.2173.

(1*R*^{*},2*S*^{*},5*S*^{*})-Diethyl 3-(2-acetoxymethylene)-6,7-diazatricyclo[3.2.1.0^{2,4}]octane-6,7-dicarboxylate 1c. Following the general procedure A, after purification on silica gel chromatography, the product was obtained as a colorless oil (465 mg, 71% yield). ¹H NMR (400 MHz, CDCl₃) δ = 5.91, (m, 1H), 4.86 (br s, 1H), 4.50–4.74 (m, 3H), 4.15–4.32 (m, 4H), 2.06 (s, 3H), 1.51 (d, *J* = 10.8 Hz, 1H), 1.24–1.35 (m, 8H), 0.81–0.93 (m, 1H). ¹³C NMR (100 MHz, CDCl₃) δ = 170.7 (s) 158.3 (br s, 2C), 127.7 (s), 116.2 (d), 63.6 (t), 62.5 (t, 2C), 62.5 (br d, 2C), 30.9 (d), 30.5 (t), 23.8 (d), 20.9 (q, 2C), 14.5 (q). HRMS (ESI-MS) *m/z*: [M + H]⁺, calcd for C₁₆H₂₃N₂O₆⁺, 339.1551; found, 339.1553.

2-((1*a*S^{*},2*S*^{*},2*a*S^{*},5*a*S^{*},6*R*^{*})-2,2*a*,5,5*a*,6,6*a*-Hexahydro-2,6-methanocyclopropa[f]inden-1(1*a*H)-ylidene)ethyl Acetate 1d. Following the general procedure A, after purification on silica gel chromatography, the product was obtained as a colorless oil (469

mg, 69% yield). The product was isolated as a mixture of 2 diastereomers. ¹H NMR (400 MHz, CDCl₃) δ = 5.84 (m, 1H), 5.67 (m, 1H), 5.59 (m, 1H), 4.54–4.67 (m, 2H), 3.14 (m, 1H), 2.58–2.67 (m, 1H), 2.49–2.55 (m, 1H), 2.35–2.43 (m, 1H), 2.21–2.35 (m, 2H), 2.05 (s, 3H), 1.44 (m, 1H), 1.12–1.17 (m, 1H), 1.10 (pseudo d, *J* = 9.8 Hz, 1H), 1.00 (pseudo d, *J* = 9.8 Hz, 1H). ¹³C NMR (100 MHz, CDCl₃) major diastereomer δ = 170.9 (s), 134.62 (s), 131.78 (d), 131.40 (d), 113.9 (d), 64.6 (t), 53.6 (d), 42.22 (d), 41.3 (d), 40.2 (d), 33.0 (t), 31.86 (t), 21.0 (q), 18.5 (d), 16.0 (d); minor diastereomer δ = 170.9 (s), 134.59 (s), 131.81 (d), 131.35 (d), 113.9 (d), 64.6 (t), 53.6 (d), 42.20 (d), 41.2 (d), 40.3 (d), 33.0 (t), 31.90 (t), 21.0 (q), 18.8 (d), 15.6 (d). HRMS (ESI-MS) *m/z*: [M + H]⁺, calcd for C₁₅H₁₈O₂Na⁺, 253.1199; found, 253.1200.

(1*R*^{*},2*S*^{*},5*S*^{*},*Z*)-Dimethyl 3-(2-acetoxymethylene)tricyclo[3.2.1.0^{2,4}]oct-6-ene-6,7-dicarboxylate 1e. Following the general procedure A, after purification on silica gel chromatography, the product was obtained as a yellow oil (378 mg, 54% yield). ¹H (400 MHz, CDCl₃) δ 5.87 (m, 1H), 4.60–4.72 (m, 2H), 3.80 (s, 3H), 3.80 (s, 3H), 3.44 (d, *J* = 6.5 Hz, 2H), 2.07 (s, 3H), 1.92–1.99 (m, 2H), 1.15–1.27 (m, 2H). ¹³C (100 MHz, CDCl₃) δ 170.8 (s), 165.29 (s), 165.27 (s), 148.3 (s), 148.2 (s), 141.0 (s), 114.08 (d), 63.8 (t), 52.1 (q, 2C), 47.8 (d), 47.7 (d), 40.6 (t), 26.8 (d), 26.6 (d), 21.0 (d). HRMS (ESI-MS) *m/z*: [M + H]⁺, calcd for C₁₆H₁₉O₆⁺, 307.1177; found, 307.1176.

2-((1*R*^{*},2*S*^{*},5*S*^{*},6*S*^{*},7*R*^{*})-6,7-Bis(methoxymethyl)-8-oxatricyclo[3.2.1.0^{2,4}]octan-3-ylidene)ethyl Acetate 1f. Following the general

procedure A, after purification on silica gel chromatography, the product was obtained as a yellow oil (221 mg, 33% yield). ^1H NMR (400 MHz, CDCl_3) δ = 5.90 (m, 1H), 4.62 (m, 2H), 4.48 (d, J = 9.0 Hz, 2H), 3.33–3.36 (m, 2H), 3.32 (s, 6H), 3.19–3.26 (m, 2H), 2.25–2.40 (m, 2H), 2.04 (s, 3H), 1.67–1.75 (m, 2H). ^{13}C NMR (100 MHz, CDCl_3) δ = 170.8 (s), 128.5 (s), 115.1 (d), 79.9 (d, 2C), 70.7 (t, 2C), 64.1 (t), 58.8 (q, 2C), 46.2 (d), 46.1 (d), 22.2 (q), 21.7 (d), 21.0 (d). HRMS (ESI-MS) m/z : [M + Na]⁺, calcd for $\text{C}_{15}\text{H}_{22}\text{O}_5\text{Na}^+$, 305.1359; found, 305.1359.

2-((1*R,4*S**,5*S**,6*S**,7*R**)-6,7-Bis(cyanomethyl)-8-oxatricyclo[3.2.1.0^{2,4}]octan-3-ylidene)ethyl Acetate **1g**.** Following the general procedure A, after purification on silica gel chromatography, the product was obtained as a yellow oil (432 mg, 56% yield). ^1H NMR (400 MHz, CDCl_3) δ = 5.97 (tt, J = 1.2, 6.4 Hz, 1H), 4.59–4.70 (m, 2H), 4.50 (d, J = 7.0 Hz, 2H), 2.52–2.59 (m, 2H), 2.39–2.49 (m, 4H), 2.07 (s, 3H), 1.81–1.88 (m, 2H). ^{13}C NMR (100 MHz, CDCl_3) δ = 170.6 (s), 125.4 (s), 118.3 (s, 2C), 116.8 (d), 81.8 (d, 2C), 63.6 (t), 43.2 (d), 43.1 (d), 21.5 (d), 21.2 (d), 20.9 (q), 17.0 (t, 2C). HRMS (ESI-MS) m/z : [M + NH₄]⁺, calcd for $\text{C}_{15}\text{H}_{20}\text{N}_3\text{O}_3^+$, 290.1499; found, 290.1502.

1-Phenyl-2-((1*R,2*S**,5*S**)-tricyclo[3.2.1.0^{2,4}]octan-3-ylidene)ethyl Acetate **1h**.** Following the general procedure A, the product was obtained as a colorless oil in a 1:1 mixture of 2 diastereomers and used without further purification (510 mg, 99% yield). ^1H NMR (400 MHz, CDCl_3) δ = 7.33–7.36 (m, 4H), 7.28–7.31 (m, 1H), 6.35 (t, J = 6.8 Hz, 1H), 5.89 (tt, J = 1.2, 6.5 Hz, 1H), 2.41 (m, 0.5H), 2.38 (m, 0.5H), 2.35 (m, 0.5H), 2.26 (m, 0.5H), 2.11 (s, 3H), 1.45–1.53 (m, 2H), 1.31–1.39 (m, 2H), 1.18–1.29 (m, 2H), 0.95 (dm, J = 9.8 Hz, 0.5H), 0.80 (dm, J = 9.8 Hz, 0.5H), 0.76 (dm, J = 9.8 Hz, 0.5H), 0.70 (dm, J = 9.8 Hz, 0.5H). ^{13}C NMR (100 MHz, CDCl_3) δ 170.10 (s), 170.07 (s), 140.2 (s), 140.1 (s), 132.9 (s), 132.8 (s), 128.4 (d, 4C), 127.8 (d), 127.7 (d), 126.8 (d, 2C), 128.7 (d, 2C), 118.2 (d), 118.0 (d), 76.3 (d), 75.9 (d), 38.1 (d), 38.0 (d), 37.92 (d), 37.86 (d), 30.7 (t), 30.6 (t), 28.65 (t), 28.59 (t), 28.5 (t, 2C), 21.81 (d), 21.76 (d), 21.44 (d), 21.41 (d), 20.94 (d), 20.87 (d). HRMS (ESI-MS) m/z : [M + Na]⁺, calcd for $\text{C}_{18}\text{H}_{20}\text{O}_2\text{Na}^+$, 291.1356; found, 291.1358.

2-((1*aS**,2*S**,2*a**S**,5*a**S**,6*R**)-2,2*a*,5,5,6,6*a*-Hexahydro-2,6-methanocyclopropa[f]inden-1(1*a*H)-ylidene)ethyl Benzoate **1i**.** Following the general procedure A, after purification on silica gel chromatography, the product was obtained as a colorless oil (408 mg, 80% yield). The product was isolated as a mixture of 2 diastereomers. ^1H NMR (400 MHz, CDCl_3) δ = 8.03–8.09 (br d, 2H), 7.52–7.59 (m, 1H), 7.40–7.48 (m, 2H), 5.93–6.00 (m, 1H), 5.66–5.72 (m, 1H), 5.57–5.63 (m, 1H), 4.82–4.94 (m, 2H), 3.11–3.19 (m, 1H), 2.59–2.68 (m, 1H), 2.51–2.56 (m, 1H), 2.38–2.43 (m, 1H), 2.21–2.38 (m, 2H), 1.48 (dd, J = 7.0, 15.3 Hz, 1H), 1.12–1.23 (m, 2H), 1.00 (pseudo d, J = 10.0 Hz, 1H). ^{13}C NMR (100 MHz, CDCl_3) *major diastereomer* δ = 166.4 (s), 134.6 (s), 132.7 (d), 131.7 (d), 131.4 (d), 130.5 (s), 129.5 (d, 2C), 128.2 (d, 2C), 113.9 (d), 65.0 (t), 53.6 (d), 42.2 (d), 41.3 (d), 40.2 (d), 33.0 (t), 31.9 (t), 18.6 (d), 15.9 (d); *partial minor diastereomer* δ = 131.8 (d), 131.3 (d), 41.2 (d), 40.3 (d), 31.9 (t), 18.7 (d), 15.7 (d). HRMS (ESI-MS) m/z : [M + NH₄]⁺, calcd for $\text{C}_{20}\text{H}_{24}\text{NO}_2^+$, 310.1802; found, 310.1799.

(2*R,2*a**S**,5*a**S**,6*S**,6*a**S**)-1-(2-Phenoxyethylidene)-1,1*a*,2,2*a*,3,5,6,6*a*-Octahydro-2,6-methanocyclopropa[f]inden-1*a**H*-ylidene **1j**.** Following the general procedure A, after purification on silica gel chromatography, the product was obtained as a colorless oil (288 mg, 62% yield). The product was isolated as a mixture of 2 diastereomers. ^1H NMR (400 MHz, CDCl_3) δ = 7.24–7.30 (m, 2H), 6.88–6.95 (m, 3H), 5.94–6.00 (m, 1H), 5.66–5.72 (m, 1H), 5.63–5.75 (m, 1H), 4.54–4.66 (m, 2H), 3.11–3.19 (m, 1H), 2.59–2.68 (m, 1H), 2.49–2.54 (m, 1H), 2.22–2.41 (m, 3H), 1.44–1.50 (m, 1H), 1.13–1.20 (m, 1H), 1.11 (pseudo d, J = 9.8 Hz, 1H), 1.00 (pseudo d, J = 9.8 Hz, 1H). ^{13}C NMR (100 MHz, CDCl_3) *major diastereomer* δ = 158.6 (s), 133.7 (s), 131.8 (d), 131.4 (d), 129.3 (d, 2C), 120.5 (d), 115.2 (d), 114.8 (d, 2C), 67.9 (t), 53.6 (d), 42.3 (d), 41.3 (d), 40.1 (d), 33.1 (t), 31.9 (t), 18.6 (d), 16.1 (d); *minor diastereomer (partial)* δ = 131.8 (d), 131.4 (d), 129.4 (d, 2C), 41.1

(d), 40.3 (d), 18.9 (d), 15.7 (d). HRMS (ESI-MS) m/z : [M + Na]⁺, calcd for $\text{C}_{19}\text{H}_{20}\text{ONa}^+$, 287.1406; found, 287.1406.

(2*R,2*a**S**,5*a**S**,6*S**,6*a**S**)-1-(2-(Benzyl oxy)ethylidene)-1,1*a*,2,2*a*,3,5,6,6*a*-Octahydro-2,6-methanocyclopropa[f]inden-1*a**H* **1k**.** Following the general procedure A, after purification on silica gel chromatography, the product was obtained as a colorless oil (397 mg 63% yield). The product was isolated as a mixture of 2 diastereomers. ^1H NMR (400 MHz, CDCl_3) δ = 7.32–7.36 (m, 4H), 7.26–7.31 (m, 1H), 5.90 (m, 1H), 5.69 (m, 1H), 5.60 (m, 1H), 4.48 (m, 2H), 4.09 (m, 2H), 3.15 (m, 1H), 2.58–2.68 (m, 1H), 2.48–2.55 (m, 1H), 2.32–2.42 (m, 2H), 2.22–2.23 (m, 1H), 1.43 (m, 1H), 1.09–1.15 (m, 1H), 1.17 (pseudo d, J = 9.8 Hz, 1H), 1.02 (pseudo d, J = 9.8 Hz, 1H). ^{13}C NMR (100 MHz, CDCl_3) *major diastereomer* δ = 138.7 (s), 132.9 (s), 131.8 (d), 131.4 (d), 128.3 (d, 2C), 127.7 (d, 2C), 127.4 (d), 116.5 (d), 71.5 (t), 70.2 (t), 53.7 (d), 42.3 (d), 41.2 (d), 40.2 (d), 33.1 (t), 31.9 (t), 18.3 (d), 16.0 (d); *minor diastereomer (partial)* δ = 131.5 (d), 53.6 (d), 42.3 (d), 40.1 (d), 18.8 (d), 15.4 (d). HRMS (ESI-MS) m/z : [M + Li]⁺, calcd for $\text{C}_{20}\text{H}_{22}\text{OLi}^+$, 285.1826; found, 285.1819.

2-Methyl-1-((1*R,4*S**,5*S**)-tricyclo[3.2.1.0^{2,4}]oct-6-en-3-ylidene)-propan-2-ol **1l**.**¹¹ Following the general procedure A, after purification on silica gel chromatography, the product was obtained as a colorless oil (531 mg, 99% yield). ^1H NMR (400 MHz, CDCl_3) δ = 6.36 (s, 2H), 5.80 (s, 1H), 3.00 (m, 2H), 1.79 (br s, 1H), 1.64 (pseudo d, J = 7.8 Hz, 1H), 1.46 (pseudo d, J = 7.3, 1H), 1.38 (s, 6H), 1.07 (d, J = 8.5 Hz, 1H), 0.92 (d, J = 8.3 Hz, 1H). ^{13}C NMR (100 MHz, CDCl_3) δ 139.4 (d), 138.9 (d), 135.8 (s), 125.0 (d), 71.7 (s), 44.2 (d), 44.0 (d), 41.8 (t), 29.8 (q), 29.7 (q), 27.2 (d), 24.0 (d). HRMS (ESI-MS) m/z : [M + Na]⁺, calcd for $\text{C}_{12}\text{H}_{16}\text{ONa}^+$, 199.1093; found, 199.1093.

(1*R,2*S**,5*S**)-3-(Benzylidene)tricyclo[3.2.1.0^{2,4}]octane **1m**.**¹¹ Following the general procedure A, after purification on silica gel chromatography, the product was obtained as a colorless oil (649 mg, 86% yield). ^1H NMR (400 MHz, CDCl_3) δ = 7.46–7.51 (m, 2H), 7.28–7.35 (m, 2H), 7.16–7.22 (m, 1H), 6.63 (s, 1H), 2.61–2.65 (m, 1H), 2.48–2.52 (m, 1H), 1.55–1.65 (m, 3H), 1.42–1.52 (m, 2H), 1.34 (d, J = 7.3 Hz, 1H), 1.01 (d, J = 9.9 Hz, 1H), 0.81 (d, J = 9.9 Hz, 1H). ^{13}C NMR (100 MHz, CDCl_3) δ = 138.1 (s), 130.7 (s), 128.4 (d), 126.5 (d), 126.4 (d), 119.5 (d), 38.4 (d), 38.0 (d), 30.9 (t), 28.9 (t), 28.6 (t), 23.1 (d), 19.5 (d).

2-((1*aS**,2*S**,2*a**S**,5*a**S**,6*R**)-2,2*a*,5,5,6,6*a*-Hexahydro-2,6-methanocyclopropa[f]inden-1(1*a*H)-ylidene)ethyl Ethanol **1n**.** To a solution of **1d** (230 mg, 1 mmol) in MeOH (10 mL) was added K_2CO_3 (690 mg, 5 mmol, 5 equiv) at room temperature. The reaction was followed by TLC, and after 2 h, the conversion was completed. The reaction mixture was evaporated to dryness, and the residue was dissolved with Et_2O (10 mL) and H_2O (10 mL). The product was extracted with Et_2O (3 × 10 mL), and the combined organic layers were dried over anhydrous Na_2SO_4 . The solvents were removed under reduced pressure, and after purification on silica gel chromatography, the product was obtained as a colorless oil (141 mg, 75% yield). The product was isolated as a mixture of 2 diastereomers. ^1H NMR (400 MHz, CDCl_3) δ = 5.94 (m, 1H), 5.68 (m, 1H), 5.51 (m, 1H), 4.20 (m, 2H), 3.14 (m, 1H), 2.58–2.68 (m, 1H), 2.51 (m, 1H), 2.20–2.40 (m, 3H), 1.35–1.48 (m, 2H), 1.08–1.19 (m, 2H), 1.01 (pseudo d, J = 9.8 Hz, 1H). ^{13}C NMR (100 MHz, CDCl_3) *major diastereomer* δ = 131.9 (s), 131.8 (d), 131.4 (d), 118.9 (d), 63.2 (t), 53.6 (d), 42.3 (d), 41.2 (d), 40.11 (d), 32.9 (t), 31.8 (t), 18.1 (d), 15.0 (d); *minor diastereomer (partial)* δ = 17.8 (d), 15.2 (d). HRMS (ESI-MS) m/z : [M + Na]⁺, calcd for $\text{C}_{13}\text{H}_{16}\text{ONa}^+$, 211.1093; found, 211.1094.

2-((1*R,2*R**,4*S**,5*S**,8*R**)-3-Oxatetracyclo[3.3.1.0^{2,4}.0^{6,8}]nonan-7-ylidene)ethyl Acetate **1o**.** A solution of meta-chloroperbenzoic acid (50–55% purity, 1 g, 2.90 mmol, 1.1 equiv) in 13 mL of dry DCM was added slowly at –20 °C in 40 min to a solution of cycloadduct **1a** (500 mg, 2.63 mmol) in 13 mL of dry DCM. The mixture was stirred at –20 °C for 24 h and let to reach slowly 0 °C. After basic hydrolysis with aqueous Na_2CO_3 solution and extraction with DCM, the organic phase was washed with brine and dried over Na_2SO_4 . The white crude solid was purified by flash chromatography

on silica gel to afford the expected product as a pale oil (426 mg, 79% yield). ^1H (400 MHz, CDCl_3) δ = 5.85 (tt, J = 1.3, 6.5 Hz, 1H), 4.56–4.57 (m, 2H), 3.38–3.42 (m, 2H), 2.65–2.71 (m, 2H), 2.06 (s, 3H), 1.53–1.60 (m, 2H), 0.99 (d, J = 10.4, Hz, 1H), 0.53 (d, J = 10.4 Hz, 1H). ^{13}C (100 MHz, CDCl_3) δ = 170.8 (s), 134.1 (s), 115.1 (d), 63.9 (t), 54.8 (d, 2C), 38.8 (d), 38.7 (d), 21.0 (d), 20.6 (q), 20.3 (d), 17.2 (t). HRMS (ESI-MS) m/z : [M + Na] $^+$, calcd for $\text{C}_{12}\text{H}_{14}\text{O}_3\text{Na}^+$, 229.0835; found, 229.0835.

(1*R*^{*,2*S*^{*,5*S*^{*}}}-*D*-*tert*-butyl 3-(2-Hydroxyethylidene)-6,7-diazatricyclo[3.2.1.0^{2,4}]octane-6,7-dicarboxylate 1*p*. Following the general procedure A, after purification on silica gel chromatography, the product was obtained as a white solid (703 mg, 83% yield). Mp = 142–144 °C. ^1H (400 MHz, CDCl_3) δ = 6.00 (m, 1H), 4.82 (br s, 1H), 4.56 (br s, 1H), 4.25 (br d, J = 5.3, 2H), 1.62–2.03 (m, 3H), 1.49 (s, 18H), 1.20–1.33 (m, 2H). ^{13}C NMR (100 MHz, CDCl_3) δ = 156.8 (br s, 2C), 125.0 (s), 120.8 (d), 81.6 (s, 2C), 62.5 (t), 62.2 (br d, 2C), 30.4 (t), 28.1 (q, 6C), 19.3 (br d), 17.8 (br d). HRMS (ESI-MS) m/z : [M + H] $^+$, calcd for $\text{C}_{18}\text{H}_{29}\text{N}_2\text{O}_5^+$, 353.2071; found, 353.2071.

General Procedure B: Reductive Isomerization. In a Schlenk flask equipped with a gas condenser and cooled at –60 °C was condensed 10 mL of ammonia. The methylene cyclopropane derivative **1** (1 mmol) was dissolved in 3 mL of THF (or Et_2O for volatile adduct) and then introduced into the Schlenk. Sodium (58 mg, 2.5 mmol) was added, resulting in an immediate dark blue coloration. After almost 30 min, the solution turned pale yellow, indicating total consumption of free sodium radical. Solid NH_4Cl (50 mg) was then added, and ammonia was evaporated by slowly warming the flask at room temperature. After addition of water (5 mL), the crude product was extracted with AcOEt (3 × 5 mL) and purified over silica.

(1*R*^{*,2*R*^{*,5*S*^{*}}}-*D*-*tert*-butyl 3-Vinyl-6,7-diazatricyclo[3.2.1.0^{2,4}]octane-6,7-dicarboxylate 5*b*. Following the general procedure B, after purification on silica gel chromatography, the product was obtained as a white solid (221 mg, 66% yield). Mp = 73–75 °C. ^1H NMR (400 MHz, CDCl_3) δ = 5.25 (ddd, J = 8.3, 10.0, 17.1 Hz, 1H), 5.08 (dd, J = 1.5, 17.1 Hz, 1H), 4.93 (dd, J = 1.5, 10.0 Hz, 1H), 4.78 (br s, 1H), 4.53 (br s, 1H), 1.49 (s, 18H), 1.35–1.50 (m, 4H), 1.11–1.25 (m, 1H). ^{13}C NMR (100 MHz, CDCl_3) δ = 157.2 (br s, 2C), 136.8 (d), 114.0 (t), 81.2 (s, 2C), 61.0 (br d, 2C), 28.0 (q, 6C), 22.5 (br d), 21.1 (d, 2C), one C (t) is missing. HRMS (ESI-MS) m/z : [M + H] $^+$, calcd for $\text{C}_{18}\text{H}_{29}\text{N}_2\text{O}_4^+$, 337.2122; found, 337.2119.

(1*R*^{*,2*R*^{*,5*S*^{*}}}-6,7-Bis(ethylperoxy)-3-vinyltricyclo[3.2.1.0^{2,4}]oct-6-ene 5*c*. Following the general procedure B, after purification on silica gel chromatography, the product was obtained as a colorless oil (154 mg, 55% yield). ^1H NMR (400 MHz, CDCl_3) δ = 5.22 (ddd, J = 8.3, 10.0, 17.1 Hz, 1H), 5.06 (dd, J = 1.5, 17.1 Hz, 1H), 4.91 (dd, J = 1.3, 10.3 Hz, 1H), 4.80 (br s, 1H), 4.58 (br s, 1H), 4.21 (br s, 4H), 1.48–1.53 (m, 1H), 1.45 (d, J = 11.5 Hz, 1H), 1.27 (t, J = 6.8 Hz, 6H), 1.17–1.25 (m, 3H). ^{13}C NMR (100 MHz, CDCl_3) δ = 158.4 (br s, 2C), 136.6 (d), 114.5 (t), 62.4 (t, 2C), 61.3 (br d, 2C), 28.3 (t), 21.8 (d), 21.2 (d, 2C), 14.4 (q, 2C). HRMS (ESI-MS) m/z : [M + H] $^+$, calcd for $\text{C}_{14}\text{H}_{21}\text{N}_2\text{O}_4^+$, 281.1496; found, 281.1496.

(2*R*^{*,2*AS*^{*,5*A**S*^{*,6*aR*^{*}}}}**-1-Vinyl-1,1*a*,2,2*a*,3,5*a*,6,6*a*-octahydro-2,6-methanocyclopenta[f]inden-5*d*. Following the general procedure B, after purification on silica gel chromatography, the product was obtained as a 10:1 mixture of diastereomers (148 mg, 86% yield). ^1H NMR (400 MHz, CDCl_3) δ = 5.68–5.74 (m, 1H), 5.52–5.58 (m, 1H), 5.26 (ddd, J = 9.0, 10.0, 17.1 Hz, 1H), 5.00 (dd, J = 1.8, 17.1 Hz, 1H), 4.79 (dd, J = 1.8, 10.0 Hz, 1H), 3.04–3.12 (m, 1H), 2.49–2.58 (m, 1H), 2.39–2.44 (m, 1H), 2.27–2.34 (m, 2H), 2.16–2.26 (m, 1H), 1.48 (dt, J = 9.0, 2.5 Hz, 1H), 1.12 (d, J = 10.3 Hz, 1H), 0.85–0.93 (m, 2H), 0.59–0.66 (m, 1H); *partial minor diastereomer* δ = 6.05 (ddd, J = 4.5, 10.3, 17.3 Hz, 1H), 5.64–5.68 (m, 1H), 5.46–5.51 (m, 1H), 5.14 (dm, J = 17.3 Hz, 1H). ^{13}C (100 MHz, CDCl_3) δ = 140.9 (d), 132.2 (d), 130.8 (d), 111.4 (t), 54.6 (d), 43.0 (d), 39.5 (d), 38.2 (d), 31.4 (t), 31.3 (t), 22.0 (d), 19.0 (d), 17.3 (d); *minor diastereomer* δ = 137.6 (d), 131.8 (d), 131.5 (d), 116.0 (t), 52.2 (d),

43.6 (d), 39.4 (d), 38.4 (d), 33.5 (t), 31.6 (t), 20.0 (d), 19.5 (d), 16.6 (d). HRMS (ESI-MS) m/z : [M + H] $^+$, calcd for $\text{C}_{13}\text{H}_{17}^+$, 173.1325; found, 173.1325.

(1*R*^{*,2*R*^{*,5*S*^{*,6*S*^{*,7*R*^{*}}}}}-6,7-Bis(methoxymethyl)-3-vinyl-8-oxatricyclo[3.2.1.0^{2,4}]octane 5*f*. Following the general procedure B, after purification on silica gel chromatography, the product was obtained as a 10:1 mixture of diastereomers (143 mg, 64% yield). ^1H NMR (400 MHz, CDCl_3) *major diastereomer* δ = 5.36 (ddd, J = 9.0, 10.3, 17.1 Hz, 1H), 5.03 (dd, J = 1.2, 17.1 Hz, 1H), 4.85 (dd, J = 1.3, 10.3 Hz, 1H), 4.28 (s, 2H), 3.31 (s, 6H), 3.29–3.33 (m, 2H), 3.18–3.24 (m, 2H), 2.15–2.21 (m, 2H), 1.55 (dt, J = 2.8, 8.8 Hz, 1H), 1.14 (d, J = 2.8 Hz, 2H); *partial minor diastereomer* δ = 6.17 (ddd, J = 7.3, 10.3, 17.3 Hz, 1H), 5.10 (dd, J = 1.6, 17.3 Hz, 1H), 4.96 (dd, J = 1.6, 10.3 Hz, 1H), 4.41 (s, 3H). ^{13}C (100 MHz, CDCl_3) *major diastereomer* δ = 137.5 (d), 113.0 (t), 78.6 (d, 2C), 70.7 (t, 2C), 58.7 (q, 2C), 46.5 (d, 2C), 24.1 (d, 2C), 21.3 (d); *partial minor diastereomer* δ = 135.7 (d) 115.4 (t), 78.5 (d, 2C), 58.7 (q, 2C), 46.8 (d, 2C), 23.1 (d, 2C), 21.3 (d). HRMS (ESI-MS) m/z : [M + H] $^+$, calcd for $\text{C}_{13}\text{H}_{21}\text{O}_3^+$, 225.1485; found, 225.1484.

2,2'-((1*R*^{*,4*R*^{*,5*S*^{*,6*S*^{*,7*R*^{*}}}}}-3-Vinyl-8-oxatricyclo[3.2.1.0^{2,4}]octane-6,7-diyli)dacetonitrile 5*g*. Following the general procedure B, after purification on silica gel chromatography, the product was obtained as a 3.3:1 mixture of diastereomers (53 mg, 25% yield). Pale yellow solid. Mp = 103–105 °C. ^1H NMR (400 MHz, CDCl_3) *major diastereomer* δ = 5.25 (ddd, J = 8.8, 10.3, 17.1 Hz, 1H), 5.01 (dd, J = 1.0, 17.1 Hz, 1H), 4.87 (dd, J = 1.5, 10.3 Hz, 1H), 4.26 (s, 2H), 2.37–2.45 (m, 6H), 1.58 (ddd, J = 2.8, 3.0, 8.5 Hz, 1H), 1.21 (d, J = 2.8 Hz, 2H); *partial minor diastereomer* δ = 6.05–5.96 (m, 1H), 5.09 (dd, J = 1.8, 17.6 Hz, 1H), 4.96 (dd, J = 2.0, 10.5 Hz, 1H). ^{13}C NMR (100 MHz, CDCl_3) *major diastereomer* δ = 135.9 (d), 118.4 (s, 2C), 114.4 (t), 80.8 (d, 2C), 43.6 (d, 2C), 23.6 (d, 2C), 18.2 (d), 17.0 (t, 2C); *partial minor diastereomer* δ = 134.0 (d), 118.4 (s), 116.9 (t), 80.9 (d, 2C), 43.8 (d, 2C), 22.7 (d, 2C). HRMS (ESI-MS) m/z : [M + NH] $^+$, calcd for $\text{C}_{13}\text{H}_{18}\text{N}_3\text{O}^+$, 232.1444; found, 232.1444.

(1*R*^{*,2*S*^{*,5*S*^{*}}}-3-((E)-styryl)tricyclo[3.2.1.0^{2,4}]octane (minor) 5*h* and (1*R*^{*,2*R*^{*,5*S*^{*}-3-Phenethyltricyclo[3.2.1.0^{2,4}]octane (major) 7*h*. Following the general procedure B, after purification on silica gel chromatography, products were obtained as an inseparable mixture (157 mg, 12% yield for **5h** and 62% yield for **7h**). ^1H NMR (400 MHz, CDCl_3) *major product* δ = 7.24–7.31 (m, 2H), 7.14–7.20 (m, 3H), 2.65 (dd, J = 7.5, 8.0 Hz, 2H), 2.24 (br s, 2H), 1.33–1.44 (m, 4H), 1.18–1.25 (m, 2H), 0.97–0.99 (m, 1H), 0.66–0.73 (m, 1H), 0.57–0.62 (m, 1H), 0.43–0.48 (m, 2H); *partial minor product* δ = 6.41 (d, J = 15.8 Hz, 1H), 5.68 (dd, J = 9.3, 15.8 Hz, 1H), 2.37 (br s, 2H), 1.83–1.89 (m, 1H), 1.12–1.18 (m, 1H), 1.01–1.08 (m, 1H). ^{13}C NMR (100 MHz, CDCl_3) *major product* δ = 142.7 (s), 128.4 (d, 2C), 128.1 (d, 2C), 125.5 (d), 36.0 (d, 2C), 35.9 (t), 34.0 (t), 29.7 (t, 2C), 28.6 (t), 22.9 (d, 2C), 13.7 (d); *partial minor product* δ = 137.8 (s), 132.8 (d), 128.2 (d), 127.1 (d), 126.4 (d), 36.3 (d), 30.3 (t), 29.4 (t), 25.0 (d), 17.8 (d). HRMS (ESI-MS) *major product* m/z : [M + NH] $^+$, calcd for $\text{C}_{16}\text{H}_{24}\text{N}^+$, 230.1903; found, 230.1904.}}

(1*R*^{*,4*S*^{*,5*S*^{*}}}-3-(2-Methylprop-1-en-1-yl)tricyclo[3.2.1.0^{2,4}]oct-6-ene 5*l*. Following the general procedure B, after purification on silica gel chromatography, the product obtained as a 6:1 mixture of diastereomers (69 mg, 43% yield). ^1H NMR (400 MHz, CDCl_3) *major diastereomer* δ = 6.35–6.38 (m, 2H), 4.61–4.66 (dm, J = 9.3 Hz, 1H), 2.84 (br s, 2H), 2.55 (dt, J = 9.5, 2.5 Hz, 1H), 1.74 (d, J = 1.0 Hz, 3H), 1.66 (d, J = 1.0 Hz, 3H), 1.22 (dm, J = 9.3 Hz, 1H), 0.95 (d, J = 2.3 Hz, 2H), 0.88 (dm, J = 9.3 Hz, 1H); *partial minor diastereomer* δ = 6.50 (m, 2H), 1.71 (dm, J = 5.5 Hz, 2H), 1.17 (dm, J = 7.3 Hz, 1H). ^{13}C NMR (100 MHz, CDCl_3) *major diastereomer* δ = 140.4 (d, 2C), 129.5 (s), 125.9 (d) 41.8 (d, 2C), 38.8 (t), 31.5 (d), 30.3 (d, 2C), 25.5 (q), 18.3 (q); *minor diastereomer* δ = 142.2 (d, 2C), 133.2 (s), 122.3 (d), 42.2 (d, 2C), 41.5 (t), 37.0 (d), 28.7 (d, 2C), 25.2 (q), 18.2 (q). HRMS (ESI-MS) m/z : [M + Ag] $^+$, calcd for $\text{C}_{12}\text{H}_{16}\text{Ag}^+$, 267.0297; found, 267.0298.

(1*R*^{*,2*R*^{*,4*S*^{*,5*S*^{*,8*R*^{*}}}}}-7-Vinyl-3-oxatetracyclo[3.3.1.0^{2,4,0^{6,8}}]nonane 5*o*. Following the general procedure B, after purification on

silica gel chromatography, the product obtained as a colorless oil (146 mg, 99% yield). ^1H NMR (400 MHz, CDCl_3) δ = 5.31 (ddd, J = 8.5, 10.3, 17.0 Hz, 1H), 5.01 (dd, J = 1.5, 17.0 Hz, 1H), 4.83 (dd, J = 1.5, 10.3 Hz, 1H), 3.32 (s, 2H), 2.60 (s, 2H), 1.94 (dt, J = 8.5, 1.6 Hz, 1H), 1.00 (d, J = 2.5 Hz, 2H), 0.96 (pseudo d, J = 11.0 Hz, 1H), 0.57 (pseudo d, J = 11.0 Hz, 1H). ^{13}C (100 MHz, CDCl_3) δ 138.6 (d), 112.4 (t), 55.0 (d, 2C), 37.1 (d, 2C), 23.8 (d, 2C), 23.5 (d), 14.7 (t). HRMS (ESI-MS) m/z : [M + H] $^+$, calcd for $\text{C}_{10}\text{H}_{13}\text{O}^+$, 149.0961; found, 149.0963.

Di-tert-butyl ((1*R,2*R**,4*S**)-6-vinylbicyclo[3.1.0]hexane-2,4-diyldicarbamate 6b.** Following the general procedure B, after purification on silica gel chromatography, the product obtained as a white solid (186 mg, 55% yield). M_p = 144–146 °C. ^1H NMR (400 MHz, CDCl_3) δ = 5.29 (ddd, J = 8.5, 10.3, 17.1 Hz, 1H), 4.96 (dd, J = 1.3, 17.1 Hz, 1H), 4.83 (dd, J = 1.3, 10.3 Hz, 1H), 4.01 (br s, 2H), 1.72–1.85 (m, 1H), 1.61 (d, J = 15.6 Hz, 1H), 1.52 (m, 2H), 1.47 (m, 2H), 1.43 (s, 18H), 1.04 (m, 1H). ^{13}C NMR (100 MHz, CDCl_3) δ = 155.4 (s, 2C), 137.8 (d, 112.9 (t), 79.3 (br s, 2C), 52.3 (br d, 2C), 38.3 (t), 38.3 (d), 28.4 (q, 6C), 23.4 (d, 2C). HRMS (ESI-MS) m/z : [M + H] $^+$, calcd for $\text{C}_{18}\text{H}_{31}\text{N}_2\text{O}_4^+$, 339.2278; found, 339.2277.

(1*R,2*R**,5*S**)-3-Benzyltricyclo[3.2.1.0^{2,4}]octane 7m.** Following the general procedure B, after purification on silica gel chromatography, the product obtained as a colorless oil (196 mg, 99% yield). ^1H NMR (400 MHz, CDCl_3) δ = 7.18–7.34 (m, 5H), 2.43 (d, J = 7.0 Hz, 2H), 2.30 (br s, 2H), 1.41–1.47 (m, 2H), 1.23–1.30 (m, 2H), 0.95–1.03 (m, 2H), 0.61–0.67 (m, 3H). ^{13}C NMR (100 MHz, CDCl_3) δ = 142.3 (s), 128.3 (d, 2C), 128.2 (d, 2C), 125.7 (d), 37.7 (t), 36.0 (d, 2C), 29.6 (t, 2C), 28.6 (t), 23.2 (d, 2C), 14.8 (d). HRMS (ESI-MS) m/z : [M + Ag] $^+$, calcd for $\text{C}_{15}\text{H}_{18}\text{Ag}^+$, 305.0454; found, 305.0451.

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Notes

The authors declare no competing financial interest.

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■ REFERENCES

- (1) For reviews on vinylcyclopropanes, see: (a) Rubin, M.; Rubina, M.; Gevorgyan, V. *Chem. Rev.* **2007**, *107*, 3117–3179. (b) Hudlicky, T.; Reed, J. W. *Angew. Chem., Int. Ed.* **2010**, *49*, 4864–4876. (c) Jiao, L.; Yu, Z.-X. *J. Org. Chem.* **2013**, *78*, 6842–6848.
- (2) For reviews, see: (a) Donaldson, W. A. *Tetrahedron* **2001**, *57*, 8589–8627. (b) Chen, D. Y.-K.; Pouwer, R. H.; Richard, J.-A. *Chem. Soc. Rev.* **2012**, *41*, 4631–4642. For selected examples, see: (c) Liu, P.; Jacobsen, E. N. *J. Am. Chem. Soc.* **2001**, *123*, 10772–10773. (d) Yue, G.; Yang, L.; Yuan, C.; Du, B.; Liu, B. *Tetrahedron* **2012**, *68*, 9624–9637.
- (3) For selected examples, see: (a) Jarstfer, M. B.; Blagg, B. S. J.; Rogers, D. H.; Poulter, C. D. *J. Am. Chem. Soc.* **1996**, *118*, 13089–13090. (b) Larock, R. C.; Yum, E. K. *Tetrahedron* **1996**, *52*, 2743–2758. (c) Adams, L. A.; Aggarwal, V. K.; Bonnert, R. V.; Bressel, B.; Cox, R. J.; Shepherd, J.; de Vicente, J.; Walter, M.; Whittingham, W. G.; Winn, C. L. *J. Org. Chem.* **2003**, *68*, 9433–9440. (d) Zhang, J.-L.; Chan, P. W. H.; Che, C. M. *Tetrahedron Lett.* **2003**, *44*, 8733–8737. (e) Jiang, H.; Sun, X.-L.; Zhu, C.-Y.; Dai, L.-X.; Tang, Y. *Tetrahedron* **2008**, *64*, 5032–5035. (f) Barluenga, J.; Quiñones, N.; Tomás-Gamasa, M.; Cabal, M.-P. *Eur. J. Org. Chem.* **2012**, *2012*, 2312–2317. (g) Yang, Y.; Li, J.; Du, B.; Yuan, C.; Liu, B.; Qin, S. *Chem. Commun.* **2015**, *51*, 6179–6182.
- (4) (a) Craig, D.; Gore, S. J.; Lansdell, M. I.; Lewis, S. E.; Mayweg, A. V. W.; White, A. J. P. *Chem. Commun.* **2010**, *46*, 4991–4993. (b) Chang, M.-Y.; Chen, Y.-C.; Chan, C.-K. *Tetrahedron* **2014**, *70*, 8908–8913.
- (5) (a) Pyun, H.-J.; Chaudhary, K.; Somoza, J. R.; Sheng, X. C.; Kim, C. U. *Tetrahedron Lett.* **2009**, *50*, 3833–3835. (b) Zhu, B.-H.; Zhou, R.; Zheng, J.-C.; Deng, X.-M.; Sun, X.-L.; Shen, Q.; Tang, Y. *J. Org. Chem.* **2010**, *75*, 3454–3457.
- (6) For selected recent references, see: (a) Zhang, H.; Curran, D. P. *J. Am. Chem. Soc.* **2011**, *133*, 10376–10378. (b) Liang, Y.; Jiang, X.; Yu, Z.-X. *Chem. Commun.* **2011**, *47*, 6659–6661. (c) Tang, W.; Wei, X.; Yee, N. K.; Patel, N.; Lee, H.; Savoie, J.; Senanayake, C. H. *Org. Process Res. Dev.* **2011**, *15*, 1207–1211. (d) Rassadin, V. A.; Sokolov, V. V.; Khlebnikov, A. F.; Ulin, N. V.; Kozhushkov, S. I.; de Meijere, A. *Synthesis* **2012**, *44*, 372–376. (e) Hay, E. B.; Zhang, H.; Curran, D. P. *J. Am. Chem. Soc.* **2015**, *137*, 322–327.
- (7) (a) Catellani, M.; Chiusoli, G. P. *J. Organomet. Chem.* **1982**, *233*, C21–C24. (b) Catellani, M.; Chiusoli, G. P. *J. Organomet. Chem.* **1983**, *250*, S09–S15.
- (8) Miura, T.; Sasaki, T.; Harumashi, T.; Murakami, M. *J. Am. Chem. Soc.* **2006**, *128*, 2516–2517.
- (9) Tseng, N.-W.; Mancuso, J.; Lautens, M. *J. Am. Chem. Soc.* **2006**, *128*, 5338–5339.
- (10) Achard, T.; Lepronier, A.; Gimbert, Y.; Clavier, H.; Giordano, L.; Tenaglia, A.; Buono, G. *Angew. Chem., Int. Ed.* **2011**, *50*, 3552–3556.
- (11) Bigeault, J.; Giordano, L.; De Riggi, I.; Gimbert, Y.; Buono, G. *Org. Lett.* **2007**, *9*, 3567–3570.
- (12) (a) Bigeault, J.; Giordano, L.; Buono, G. *Angew. Chem., Int. Ed.* **2005**, *44*, 4753–4757. (b) Bigeault, J.; de Riggi, I.; Gimbert, Y.; Giordano, L.; Buono, G. *Synlett* **2008**, *2008*, 1071–1075. (c) Thota, R.; Lesage, D.; Gimbert, Y.; Giordano, L.; Humbel, S.; Milet, A.; Buono, G.; Tabet, J.-C. *Organometallics* **2009**, *28*, 2735–2743. (d) Gatineau, D.; Moraleda, D.; Naubron, J.-V.; Bürgi, T.; Giordano, L.; Buono, G. *Tetrahedron: Asymmetry* **2009**, *20*, 1912–1917. (e) Clavier, H.; Lepronier, A.; Bengobesse-Mintsa, N.; Gatineau, D.; Pellissier, H.; Giordano, L.; Tenaglia, A.; Buono, G. *Adv. Synth. Catal.* **2013**, *355*, 403–408.
- (13) (a) Ge, G.-C.; Mo, D.-L.; Ding, C.-H.; Dai, L.-X.; Hou, X.-L. *Org. Lett.* **2012**, *14*, 5756–5759. (b) Mo, D.-L.; Yuan, T.; Ding, C.-H.; Dai, L.-X.; Hou, X.-L. *J. Org. Chem.* **2013**, *78*, 11470–11476.
- (14) For the preparation of platinum-phosphinito-phosphinous acid complexes and their application in catalysis, see: (a) Bigeault, J.; Giordano, L.; de Riggi, I.; Gimbert, Y.; Buono, G. *Org. Lett.* **2007**, *9*, 3567–3570. (b) Achard, T.; Giordano, L.; Tenaglia, A.; Gimbert, Y.; Buono, G. *Organometallics* **2010**, *29*, 3936–3950.
- (15) (a) Mellor, J. M.; Smith, M. N. *J. Chem. Soc., Perkin Trans. I* **1984**, 2927–2931. (b) Denmark, S. E.; Nicaise, O.; Edwards, J. P. *J. Org. Chem.* **1990**, *55*, 6219–6223. (c) Storsberg, J.; Nandakumar, M. V.; Sankaranarayana, S.; Kaufmann, D. E. *Adv. Synth. Catal.* **2001**, *343*, 177–180. (d) Bournaud, C.; Robic, D.; Bonin, M.; Micouin, L. *J. Org. Chem.* **2005**, *70*, 3316–3317.
- (16) For a review on bicyclic hydrazines, see: (a) Bournaud, C.; Chung, F.; Pérez Luna, A.; Pasco, M.; Errasti, G.; Lecourt, T.; Micouin, L. *Synthesis* **2009**, *2009*, 869–887. For alternative methods

of N–N cleavage of bicyclic hydrazine, see: (b) Pérez Luna, A.; Ceschi, M.-A.; Bonin, M.; Micouin, L.; Husson, H.-P. *J. Org. Chem.* **2002**, *67*, 3522–3524. (c) Bournaud, C.; Bonin, M.; Micouin, L. *Org. Lett.* **2006**, *8*, 3041–3043. (d) Moumné, R.; Pasco, M.; Prost, E.; Lecourt, T.; Micouin, L.; Tisné, C. *J. Am. Chem. Soc.* **2010**, *132*, 13111–13113. (e) Pasco, M.; Moumné, R.; Lecourt, T.; Micouin, L. *J. Org. Chem.* **2011**, *76*, 5137–5142. (f) Blond, A.; Moumné, R.; Bégis, G.; Pasco, M.; Lecourt, T.; Micouin, L. *Tetrahedron Lett.* **2011**, *52*, 3201–3203. (g) Blond, A.; Dockerty, P.; Álvarez, R.; Turcaud, S.; Lecourt, T.; Micouin, L. *J. Org. Chem.* **2013**, *78*, 12236–12242.

(17) For a rare example of C–O bond reduction using Na in NH₃, see: Birch, A. *J. J. Chem. Soc.* **1945**, 809–813.

(18) For reviews on methylenecyclopropanes, see: (a) Brandi, A.; Goti, A. *Chem. Rev.* **1998**, *98*, 589–636. (b) Audran, G.; Pellissier, H. *Adv. Synth. Catal.* **2010**, *352*, 575–608. (c) Pellissier, H. *Tetrahedron* **2010**, *66*, 8341–8375. (d) Brandi, A.; Cicchi, S.; Cordero, F. M.; Goti, A. *Chem. Rev.* **2014**, *114*, 7317–7420. (e) Pellissier, H. *Tetrahedron* **2014**, *70*, 4991–5031.

(19) (a) Bouveault, L.; Blanc, G. *C. R. Hebd. Séances Acad. Sci.* **1903**, *136*, 1676–1678. (b) Bouveault, L.; Blanc, G. *Bull. Soc. Chim.* **1904**, *31*, 666–672.

(20) Scott, N. D.; Walker, J. F.; Hansley, V. L. *J. Am. Chem. Soc.* **1936**, *58*, 2442–2444.

(21) (a) Metz, P. *Tetrahedron* **1989**, *45*, 7311–7316. (b) Marchand, A. P.; Reddy, G. M. *Synthesis* **1991**, *1991*, 198–200.

(22) See [Supporting Information](#).

(23) Perrin, D. D.; Armarego, W. L. F. *Purification of Laboratory Chemicals*, 3rd ed.; Pergamon Press: Oxford, U.K., 1988.

(24) Fulmer, G. R.; Miller, A. J. M.; Sherden, N. H.; Gottlieb, H. E.; Nudelman, A.; Stoltz, B. M.; Bercaw, J. E.; Golberg, K. I. *Organometallics* **2010**, *29*, 2176–2179.