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Atomic scale modeling to understand how matter organizes during growth of ultrathin materials in close relation with elaboration process parameters: climbing the scales

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Abstract—A multi-levels modeling approach focusing on the elaboration of directly integrated materials is described. Atomic scale calculations are conducted to identify local mechanisms occurring during the growth of material and are then used to parameterize a macroscopic code, developed to simulate vapor like deposition process. The objective is to establish the link between the microstructure of materials and the way they are achieved, i.e. process parameters.

I. Introduction

An absolute control at the atomic scale of the material growth during the technological processes, to tailor composition of the layers, and especially to master the formation of interfaces is necessary since this is this atomic arrangement at the nanoscale level which will impact decisively and ensure final properties to the achieved materials. Developing the means of predicting and controlling the structures of integrated materials at the nanometre scale will thus help to push forward their elaboration.

In this context, predictive simulation of the deposition process is strategic. We present a multi-levels approach conducted with the objective to provide an original TCAD tool endowed by an atomistic granularity, as a new tool-for-the-engineer dedicated to the design of tailored integrated material. A multi-levels modeling study is detailed where atomic-scale observations are used as input data to parameterize a macroscopic code, based on a kinetic Monte Carlo methodology. This code is able to simulate vapor-like deposition process and to predict interface composition as a function of applied process parameters. The developed code will help to understand and evaluate the influence of experimental conditions on the material nanostructuring, linking design and manufacturing. In this paper, the deposition of CuO on Al(111) substrate is described.

II. CLIMBING THE SCALES: A MULTI-LEVELS STRATEGY

A. Atomistic investigations of the growth local mechanisms using atomic scale calculations

Massive ab initio calculations are performed to identify local mechanisms at the atomic scale as they could occur during the growth.

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All the atomic diffusions are thus precisely characterized in terms of energetics (activation barriers and energy gains). All the results presented in this work have been conducted using DFT methodology as implemented into VASP 5.3.3 software.

B. Simulation of the deposition process

Atomic scale data are then implemented into a homemade code, based on a kinetic Monte Carlo methodology (kMC). This code mixes all the atomic scale events. It simulates CuO deposition on Al(111) substrate as a function of realistic process parameters taking into account gases and substrate temperatures, deposition duration, gases pressures... This level of modeling reaches size and time scales characteristic of the process scale [1].

At each step of the kMC cycle, each site of the system is scanned in order to identify all the possible events that can occur in the given state. For each possible event, an occurrence probability is associated. Two kinds of events are defined: Atomic diffusion and Arrival. The occurrence probability of an atomic diffusion event is calculated using the Arrhenius Law formula given in (1):

$$P_i = v. e^{\left(\frac{-E_i}{k_B T}\right)} \tag{1}$$

where $\nu=10^{12}~\text{s}^{-1}$ is the attempt frequency, E_i is the activation barrier of the diffusion i, k_B is the Boltzmann constant and T is the temperature.

The occurrence probability of an Arrival event is calculated using Boltzmann Statistics given in (2):

$$P_i = C \frac{PS}{\sqrt{MT}} \tag{2}$$

where T is the chamber temperature, P the gas pressure, M the molar mass, S the elementary surface unit and C a constant. Using this approach, we introduce a matter supply into the simulation and the process parameters P and T, as carried out during deposition.

Then the Poisson formula (3) is used to associated an occurrence time to the occurrence probability as:

$$t_i = t_{simulation} - \frac{\ln(z)}{P_i}$$
 (3)

with z is a random number. The formula introduces a time dependency in the simulation. This time can be seen as the time for one event to be realized.

III. RESULTS AND DISCUSSION

A. Complex chemistry during the formation of interfacial layer

All these atomic scale results highlight the complexity of the interface formation due to the chemistry of materials themselves (Fig. 1). The adsorption of CuO on Al(111) is dissociative. O atoms exhibit a thermally activated surface diffusion behavior whereas Cu will diffuse back and forth on the surface due to a non-existing diffusion barrier when isolated. Both Cu and O atoms tend to form island on the Al(111) surface. But while Cu prefers to penetrate deep into the substrate [2], O will extract Aluminum atoms [3] to form oxide nucleus. Oxidized part on the surface will trap Cu species, avoiding them to diffuse into the bulk [4].

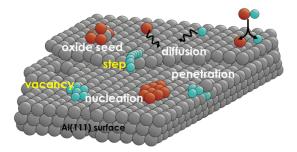


Fig. 1. Schematic of investigated local mechanisms using DFT.

B. Design material elaboration thanks to the control of kinetics at the atomic scale

We compare simulations of CuO deposition at low temperature (120K) and room temperature (300K).

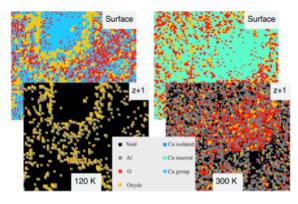


Fig. 2. Simulations of CuO deposition at 120K (left) and 300K (right).

At low temperature, Cu atoms do not succeed to insert into the Al(111) substrate due to the height of their activation barriers. We can notice on Fig. 2, that a large part of the surface is covered by non-inserted Cu atoms (in blue) surrounded by oxidized parts (in yellow).

At room temperature, a massive insertion of Cu atoms is observed into the substrate, leading to rich-copper zone up to 15-20 layers below the surface. Here, oxidized parts are distributed homogeneously on the topmost surface.

We notice here that copper is a key species to control interfacial layer morphology. Notably, using this deposition

simulation tool, we can thus evaluate the effect of the temperature on the nanostructuring of the interface. Authorizing copper diffusion thanks to temperature enables to control the thickness and the composition of the interfacial layer.

IV. CONCLUSION

We propose a computationally-driven approach for the design of material to predict their microstructure and morphology during technological process using a multi-levels modeling approach coupling atomic scale understanding of the local mechanisms occurring during the growth and predictive simulations of deposition process. This tool will enable virtual experiments of the deposition to foresee deposition parameters influence on final materials morphology towards a chemically controlled structure of materials and interfaces.

Here, we point out the diversity of the atomic scale diffusions associated to the complex chemistry encountered during the interface formation. We also highlight that applying controlled parameters such as temperature during deposition can lead to tailored interfaces thanks to the atomic scale investigation and understanding, as a promising way to control matter toward dedicated applications.

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