

Structure and reorientational dynamics of 1-F-adamantane

Bacem Ben Hassine, Philippe Négrier, M. Romanini, M. Barrio, R Macovez,

A. Kallel, Denise Mondieig, Josep Ll Tamarit

▶ To cite this version:

Bacem Ben Hassine, Philippe Négrier, M. Romanini, M. Barrio, R Macovez, et al.. Structure and reorientational dynamics of 1-F-adamantane. Physical Chemistry Chemical Physics, 2016, 18 (16), pp.10924-10930. 10.1039/c6cp01144f . hal-01341793

HAL Id: hal-01341793 https://hal.science/hal-01341793v1

Submitted on 4 Jul 2016

HAL is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers. L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.



Distributed under a Creative Commons Attribution - ShareAlike 4.0 International License

Structure and reorientational dynamics of 1-F-adamantane[†]

B. Ben Hassine,^{ab} Ph. Negrier,^a M. Romanini,^c M. Barrio,^c R. Macovez,^c A. Kallel,^b D. Mondieig^a and J. Ll. Tamarit*^c

The polymorphism and the dynamics of a simple rigid molecule (1-fluoro-adamantane) have been studied by means of X-ray powder diffraction and broadband dielectric spectroscopy. At temperatures below the melting point, the molecule forms an orientationally disordered Phase I with a cubic-centered structure (Phase I, *Fm 3m*, *Z* = 4). This phase possesses eight equilibrium positions for the fluorine atom, with equal occupancy factors of 1/8. A solid solid phase transition to a low-temperature tetragonal phase (Phase II, *P* 42₁*c*, *Z* = 2) reduces the statistical disorder to only four possible equivalent sites for the fluorine atom, with fractional occupancies of 1/4. The dynamics has been rationalized under the constraints imposed by the space group of the crystal structure determined by powder X-ray diffraction. The dielectric spectroscopy study reveals that the statistical disorder in Phase II is dynamic in character and is associated with reorientational jumps along the two- and three-fold axes. In the dielectric loss spectra, the cooperative (α) relaxation exhibits a shoulder on the high-frequency side. This remarkable finding clearly reveals the existence of two intrinsic reorientational processes associated with the exchange of the F atom along the four sites. In addition to such "bimodal" relaxation, a secondary Johari Goldstein relaxation is detected at lower temperatures.

Introduction

The dynamics of amorphous systems is one of the areas of greatest interest and challenging studies in the field of condensed matter physics.¹⁻⁴ The physics of disordered materials is intriguing; in particular, a full understanding of the origin of the glass transition phenomenon in dynamically disordered systems (*i.e.*, systems where the disorder is due to time-varying structural, orientational or molecular conformations) is still lacking.^{5,6}

The dynamics of glass-forming liquids has been the subject of many experimental investigations, which have shown very diverse dynamic behavior.^{4,6,7} Most of the research has been focused on "canonical" glasses, namely those obtained from the freezing (through a decrease in temperature or an increase in pressure) of both translational and orientational degrees of freedom.⁸ The disorder in a condensed phase, however, can also be only partial, involving only a subset of all degrees of freedom. Partial disorder gives rise to different states of matter:

Département Physique, Faculté des Sciences de Sfax, 3000 Sfax, Tunisie

liquid crystals are translationally disordered but orientationally ordered, while, oppositely, plastic crystals are translationally ordered and orientationally disordered (OD).⁹⁻¹³ For the latter materials, composed of globular shaped molecules, the molecular orientation is completely undetermined, as the molecules undergo free-rotor motions. The class of solids with dynamic orientational disorder also comprises systems in which, at least in a certain temperature range, the thermal energy is not high enough to allow free rotations of the molecules, which rather perform reorientational jumps between well-defined, distinguishable orientations (compatible to the symmetry operations of the site in which they are located at the lattice). Decreasing the temperature slows down the reorientational jumps and can lead to the formation of a so-called "glassy crystal", the non-ergodic state of the OD phase. Nevertheless, there is also the possibility that a phase transition to a lower symmetry crystal occurs, which entails partial or complete reduction of the orientational disorder. A reduction of disorder through the transition to a low-temperature phase has been detailed recently for low-symmetry (monoclinic) phases of 2-adamantanone,¹⁴ CBrCl₃ and CBr₂Cl₂,^{15,16} CCl₃-CF₂Cl¹⁷ and CHCl2-CCl2H.18,19

The molecular dynamic processes in these systems have been found to display the same general features of those of canonical glasses. A universal feature of all glass formers is, for example, the existence of collective dynamics whose slowing down marks the transition to the glassy state, *i.e.*, the well-known α -(primary) relaxation process. A less cooperative dynamic relaxation,

^a Univ. Bordeaux, LOMA, UMR 5798, F 33400 Talence, France

^b Laboratoire des Matériaux Céramiques Composites et Polymères,

^c Grup de Caracterització de Materials, Department de Física, ETSEIB, Diagonal 647, Universitat Politècnica de Catalunya, 08028 Barcelona, Catalonia, Spain. E mail: josep.lluis.tamarit@upc.edu; Tel: +34 934016564

 $[\]dagger$ CCDC 1455093 and 1455094, tetragonal phase II (at 100 K, CCDC 1455093) and cubic phase I (at 295 K, CCDC 1455094) of 1 fluoro adamantane. For crystallographic data in CIF or other electronic format see DOI: 10.1039/c6cp01144f

the so-called Johari-Goldstein (JG) process,²⁰ is also commonly observed, at time scales faster than the primary relaxation. Such a process is generally considered a universal feature of glassy-like materials, but its origin is still a matter of debate.²¹ Some models assign this feature to the existence of microscopic domains (islands of mobility) in which molecular mobility is higher due to a lower value of the local density,²⁰ while some others ascribe it to smallangle reorientations displayed by all molecules alike.²² In particular. for strong glasses according to the strong/fragile classification of Angell,²³ some models claim the nonexistence of the JG-relaxation.²⁴ In contrast, recent experimental results have shown the emergence of glass-like α - and β -dynamic processes even in a quite ordered system, namely the low-temperature monoclinic phase of 2-adamantanone, in which these processes were ascribed, respectively, to a strongly restricted collective reorientational motion and the local non-cooperative molecular motion, that is, to a genuine JG-relaxation. A clear β -relaxation was also observed in the low-temperature phases of $CBr_nCl_{(4-n)}$, n = 0, 1, 1and 2 compounds, a group of strong glasses with a well-defined disorder related to the occupancy of halogen sites. Such relaxation was ascribed to the presence of dynamic heterogeneities.^{15,16}

These unexpected results regarding very simple and "wellordered" systems suggest that OD solids can be exploited as model playgrounds in which the physics behind the glass transition and the associated molecular relaxations can be more thoroughly investigated, with the hope of disentangling the origin of local, non-cooperative JG-relaxation. In order to pursue this goal, the present work discusses the dynamics of the lowtemperature phase of a simple rigid molecule displaying statistical disorder. The material under scrutiny is 1-fluoro-adamantane ($C_{10}H_{15}F$, hereinafter referred to as 1F-ada). Adamantane derivatives display rich polymorphism that has stimulated a large number of experimental and theoretical investigations aimed at understanding the dynamics of the high-temperature OD phases and, to a less extent, the low-temperature phases.^{25–28}

1F-ada is a globular molecule with C_{3v} (3m) point-group symmetry, obtained from adamantane $(C_{10}H_{16})$ by replacing one of the hydrogen atoms attached to a three-coordinated (tertiary) carbon with a fluorine atom. Such substitution confers on the molecule a noticeable dipole moment (μ = 2.1 D).²⁹ Calorimetric (DSC) measurements revealed a solid-solid phase transition at 222 K, with an enthalpy and an entropy of transition of 1.50 kJ mol $^{-1}$ and 6.77 J K $^{-1}$, respectively. The reported melting temperature is 525 K.30 The solid phase of 1-F-ada between 525 K and 222 K is an OD crystal with a cubic structure ($Fm\bar{3}m$ space group, hereafter Phase I),³¹ whose structure and dynamics have been widely studied by numerous techniques, namely thermal analysis,³⁰ X-ray diffraction,³¹ NMR,³² single-crystal Raman scattering,³³ dielectric analysis,²⁹ and incoherent quasi-elastic neutron scattering.³⁴ It was shown that the fluorine atom occupies 8 distinguishable equilibrium positions corresponding to as many orientations along the [111] direction of the cubic lattice in such a way that the threefold molecular and crystal axes are coincident.³¹ The relaxation observed by dielectric spectroscopy was assigned to jumps of the molecular dipole between the neighboring [111] directions of the cubic lattice.²⁹

Incoherent quasielastic neutron scattering measurements showed the existence of two characteristic correlation times, one corresponding to the dipolar tumbling observed by dielectric spectroscopy and the other to uniaxial 120° rotations between three equilibrium positions around the molecular symmetry axis (along the C–F dipole direction).³⁴ Much less is known about the low temperature phase stable below 222 K (hereafter Phase II), on which we focus in this work.

Experimental details

1-Fluoro-adamantane was purchased from ABCR, with a purity of 99% and used as received. To study the dynamics of the low-temperature Phase II of 1F-ada, we have combined X-ray diffraction with broadband dielectric spectroscopy. Powder X-ray diffraction data were collected by means of a horizontally mounted INEL cylindrical position-sensitive detector (CPS 120). The detector, used in Debye-Scherrer geometry, is constituted by 4096 channels, providing an angular step of 0.029° (2 θ) between 4° and 120° . Monochromatic Cu K₂₁ radiation ($\lambda = 1.5406$ Å) was selected with an asymmetric focusing incident-beam curved quartz monochromator. The samples were introduced into 0.5 mmdiameter Lindemann capillaries which rotate along their longitudinal axes during data collection to prevent the effects of the preferred orientations. The system is equipped with a liquid nitrogen 600 series cryostream cooler from Oxford Cryosystems with a temperature accuracy of 0.1 K and similar fluctuations.

The relaxation times were measured by means of broadband dielectric spectroscopy conducted using a Novocontrol analyzer $(10^{-2} \text{ to } 10^7 \text{ Hz})$ and an HP4291 impedance analyzer $(10^6 \text{ to } 1.8 \times 10^9 \text{ Hz})$, equipped with a Quatro temperature controller $(\pm 0.1 \text{ K})$ for measurements down to *ca.* 110 K and a closed-cycle helium cryostat for measurements down to *ca.* 60 K. The capacitor was prepared by pressing the 1F-ada powder between two stainless steel disks using a hydraulic press (120 kN).

To fit the dielectric spectra, a combination of a Havriliak–Negami (HN) function for the α relaxation and a Cole–Cole (CC) function for the α' and β relaxations was used. The general expression for the HN function is:³⁵

$$\varepsilon_{\rm NH}(f) = \varepsilon_{\infty} + \frac{\varepsilon_{\rm S} \quad \varepsilon_{\infty}}{\left[1 + (i2\pi f \tau)^{\alpha}\right]^{\beta}}$$

Here, ε_s and ε are the high-frequency and the static low-frequency limits of the real permittivity and the exponents α and β are shape parameters with values between 0 and 1. τ_{HN} is a fitting parameter from which the characteristic time τ at which the dielectric loss of the primary relaxation process is maximum is obtained as:

$$\tau = \tau_{max} = \tau_{HN} \left(\sin \frac{\alpha \pi}{2 + 2\beta} \right)^{-1/\alpha} \left(\sin \frac{\alpha \beta \pi}{2 + 2\beta} \right)^{1/\alpha}$$

The equation of the Cole–Cole function can be expressed as:³⁶

 $\varepsilon_{\rm CC}(f) = \varepsilon_{\infty} + \frac{\varepsilon_{\rm S} \quad \varepsilon_{\infty}}{1 + (i \, 2\pi f \, \tau_{\rm CC})^{\alpha}}$

In this case the characteristic time τ corresponds to the fitting parameter $\tau_{\rm CC}.$

Results and discussion

The lattice and the possible space groups of Phases I and II were determined at 295 and 100 K, respectively, by means of X-Cell software, available in the module Powder Indexing of Materials Studio,37 according to a reasonable density and compatibility between the volume change at the II-I transition. The Pawley refinement,³⁸ which minimizes the weighted *R*-factor, $R_{\rm wp}$, describing the agreement between the experimental and the simulated patterns, helps confirm the indexing result and the systematic absences for the space group assessment. In order to determine the structure of the two crystalline phases we have built up the molecule of 1F-ada using the parameters defining the adamantane molecule C10H16,39 replacing one hydrogen atom in the molecule of adamantane by a fluorine. The carbon atom C₁ and the fluorine atom F are placed on the diagonal of the cube and the carbon atom C₂ is placed on an axis of the cube. The distance between the atoms and the centre of mass (origin of the cube) "O" has the following values: $OF_1 = 2.90 \text{ Å}$, $OC_1 = 1.530 \text{ Å}$, and $OC_2 = 1.765$ Å. The parameters defining the rigid molecule are: C–C = 1.529 Å, C–F = 1.370 Å and C–H = 0.970 Å. The final Rietveld refinement,⁴⁰ the position and orientation of the molecule, within the rigid-body constraint, with a single overall isotropic displacement parameter and the preferred orientation by using the Rietveld–Toraya function⁴¹ were refined. The final refined patterns are depicted in Fig. 1a and b for Phases I and II, respectively, together with the experimental and refined pattern difference. The Rietveld refinement converged to the final R_{wp} values of 6.19% and 4.07% for Phases I and II, respectively.

The refined structure of the high-temperature Phase I of 1F-ada is face-centred cubic ($Fm\bar{3}m$), with four molecules per unit cell and a = 9.5522(4) Å at 295 K (see Fig. 2a and b). This phase possesses eight equilibrium positions for the fluorine atom, with an equal fractional occupancy of 0.125. These results match those of the earlier study by means of single crystal diffraction mentioned in the Introduction.³¹

When cooling from Phase I to Phase II previous results indicated that the single crystal breaks and thus powder diffraction is the only way to determine the structure of the low-temperature Phase II. We determined the lattice symmetry of Phase II to be tetragonal, space group $P\bar{4}2_1c$, with two molecules per unit cell, Z = 2, then Z' = 1/8 per asymmetric unit, and lattice parameters a = b = 6.7776(3) Å, and c = 8.8235(5) Å at 100 K. The most striking feature of the structure of Phase II is the existence of statistical intrinsic disorder concerning the site occupancy of the fluorine atom. As can be seen in Fig. 2c and d, four possible sites are occupied by the fluorine atom with temperature-independent fractional occupancies of 0.25.

In Fig. 2c, we present four cells along the c axis of the structure of this tetragonal phase in order to highlight the resemblance between the stacking of both tetragonal and cubic structures. It can be noticed that both structures have a very

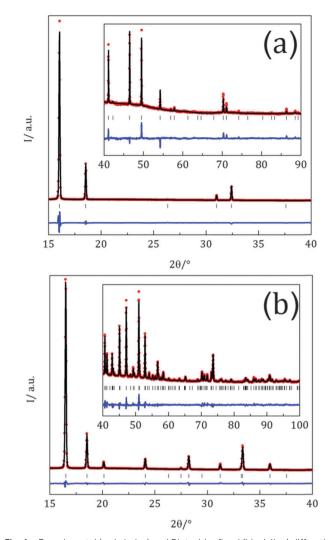


Fig. 1 Experimental (red circles) and Rietveld refined (black line) diffraction patterns along with the difference profile (blue line) and Bragg reflections (vertical sticks): (a) of the cubic $Fm\overline{3}m$ space group Phase I at 295 K of 1 fluoro adamantane and (b) of the tetragonal $P\overline{4}2_{1c}$ space group Phase II at 100 K. The insets show the high angle portion of each pattern (scaled to enhance their visibility): (a) between 40° and 90°, magnified 45 times and (b) between 40° and 100°, magnified 25 times.

close stacking with a slight difference: a rotation of 6.65° of the molecule along the *c* axis in Phase II with respect to Phase I. As a consequence of the four different molecular orientations in the low-temperature Phase II, large-angle rotations around a twofold and a threefold axis (see insets in Fig. 5), which would exist for a molecule devoid of the fluorine atom, appear. These rotations result in an effective time-average fluctuation of the molecular dipole which contributes to the dielectric susceptibility.

The structural solutions for Phases I and II determined, respectively, at 295 and 100 K were also verified for other temperatures. To do so, X-ray powder diffraction measurements were conducted from 90 to 360 K. Diffraction patterns were collected every 20 K and 5 K close to the Phase II–I transition. The variation as a function of temperature of the

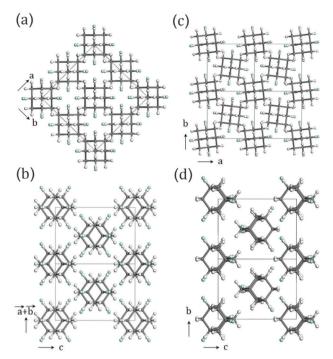


Fig. 2 Crystal structures of 1 fluoro adamantane: Phase I at 295 K (a) along *c* and (b) along a + b; Phase II at 100 K (c) along *c* and (d) along *a*. Blue atoms correspond to the fluorine atom within their eight sites of fractional occupancies 0.125 for cubic phase (I) and four sites of fractional occupancies 0.25 for the tetragonal phase (II).

lattice parameters $a_{\rm I}$ for Phase I, and $a' = a_{\rm II}\sqrt{2}$ and $c_{\rm II}$ for Phase II, is drawn as Fig. 3a.

The variation of the volume per molecule, V/Z, for both Phases I and II is displayed in Fig. 3b. The discontinuity across the II–I transition confirms its first-order character.

While diffraction techniques cannot provide information about the dynamics of the dipole reorientation, they highlight the disorder and quantify unequivocally the possible molecular orientations within the commensurate long-range lattice. It is important to note that, since the fractional occupancies of the four orientational sites have been shown to be independent of temperature, this means that a sort of local symmetry is preserved. This can only occur if correlations exist among motions of different molecules.

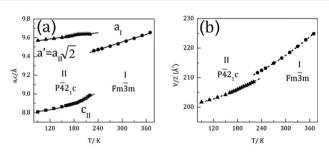


Fig. 3 Variation with temperature of the unit cell parameters $(a' = a_{II}\sqrt{2}, c_{II}, a_{I})$ of 1 fluoro adamantane (\blacktriangle and \blacksquare Phase II; \blacklozenge Phase I) (for the sake of comparison, the unit cell parameter a_{II} has been multiplied by $\sqrt{2}$) (left panel) and of the volume per molecule (*V*/*Z*) for the two solid Phases I and II ($Z_{I} = 4$ and $Z_{II} = 2$, respectively) (right panel).

In order to confirm the dynamic character of the occupational disorder of the F atom within the tetragonal Phase II and fully characterize such dynamics, a dielectric spectroscopy study was undertaken. Molecular motions are readily detected with this technique owing to the dipole moment of the 1F-ada molecule $(\mu = 2.1 \text{ D})$. Fig. 4 shows several representative examples of dielectric loss spectra of 1F-ada from 216 K down to 64 K. The most prominent feature is the existence of a clear relaxation peak (α -relaxation) whose maximum continuously shifts to lower frequencies with decreasing temperature, thus revealing the freezing of the collective molecular dynamics. The frequency of maximum loss is close to the early dielectric studies of Amoureux *et al.*²⁹ In that work the dielectric loss spectra had only very sparse points and were fitted with two different types of functions, namely a Cole-Cole for T > 130 K and a Cole-Davison for T < 130 K. Our data clearly show the existence, in addition to the main peak (α relaxation), of an excess wing (hereinafter called α' relaxation) appearing as a shoulder to the main peak, as well as of a β -relaxation appearing in the low-temperature spectra (below 88 K).

The fitting of the most prominent feature with only one Havriliak–Negami function did not reproduce the spectra, precisely due to the presence of the shoulder. As shown in Fig. 4, two relaxation processes are needed for describing the experimental data in this frequency range. Accordingly, spectra were fitted using a Havriliak–Negami function for the main α relaxation and a Cole–Cole function for the shoulder, the α' relaxation (a fit of the α' relaxation with the HN function always resulted in $\beta_{\rm HN}$ very close to 1). Fits are displayed with continuous lines in Fig. 4.

A representative example of both contributions to the fit is shown for the spectrum at 128 K by dashed lines. The exponents resulting from the HN fits lie in the ranges 0.71 < $\alpha_{\rm HN}$ < 1 and 0.85 < $\beta_{\rm HN}$ < 1 for the α -relaxation, and 0.61 < $\alpha_{\rm CC}$ < 1 for the α' relaxation. According to the Alvarez–Alegria–Colmenero

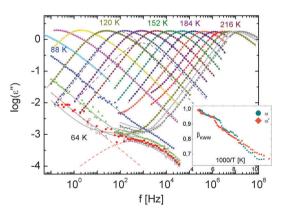


Fig. 4 Dielectric loss spectra of 1F ada from 216 to 64 K, every 8 K. Symbols are experimental points, while the solid lines represent the fits of the spectra as the sum of two components. For the spectrum at 128 K dashed lines indicate the two components (HN and CC) used for the fit, while for the spectrum at 64 K the linear component and the CC functions are depicted with red dashed lines. Inset: the stretching exponent β^{KWW} as a function of reciprocal of temperature.

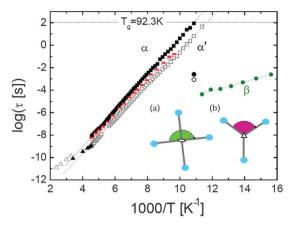


Fig. 5 Relaxation times as a function of reciprocal of temperature for the different relaxation processes. Phase II: α (full black squares), α' (empty black squares) and β (full green circles) from this work. Full empty red squares are relaxation times obtained from Amoureux *et al.*²⁹ Empty and full black circles are the predicted β relaxations according to the coupling model.²¹ Phase I: α relaxation obtained from quasielastic neutron scatter ing in ref. 34 (empty black triangles) and from dielectric spectroscopy in ref. 29 (full black triangles). Dashed lines represent the Arrhenius fits. Insets show the projection of the F atom representing the large angle rotations around the twofold axis (a) and the threefold axis (b) of the tetragonal structure (space group $P\bar{4}2_1c$).

equation,⁴² the stretching exponent β^{KWW} describing the spectral relaxation function in the time domain can be calculated as $\beta^{\text{KWW}} = (\alpha_{\text{HN}} \cdot \beta_{\text{HN}})^{\frac{1}{1.23}}$. This exponent changes from values around 1 at high-temperature, *i.e.* close to the II to I transition, to values of *ca.* 0.67 at temperatures close to the glass transition temperature (92.3 K) for both α and α' relaxations. This experimental fact clearly evidences that orientational correlations between the nearest neighbors ascribed to many-body molecular motions increase with decreasing temperature. It is worth mentioning that values close to 1 for the stretching exponent β^{KWW} were also found in the high-temperature Phase I, indicating that the average packing does not change noticeably across the Phase I to II transition.²⁹

The relaxation times obtained for both α and α' relaxations are collected in Fig. 5 as a function of the reciprocal of temperature together with previously published relaxation times.^{29,34} Both α and α' processes follow an Arrhenius temperature behaviour, with activation energies of 3709 \pm 10 K (30.84 \pm 0.08 kJ mol⁻¹) and 3517 \pm 11 K (29.24 \pm 0.10 kJ mol^{-1}) and pre-exponential factors of $\tau_o(\alpha) = 2.9 \times 10^{-15}$ s and $\tau_o(\alpha') = 0.17 \times 10^{-15}$ s, respectively. Both activation energies are significantly higher than that characterizing the high-temperature Phase I, determined to be 24.4 \pm 0.08 kJ mol⁻¹ using dielectric spectroscopy²⁹ or 19.0 kJ mol⁻¹ using quasielastic neutron scattering.³⁴ The difference is likely due to the higher steric hindrance in the lowtemperature phase. According to the α -relaxation values, the glass transition is determined to be at $T_{g}(\tau \alpha = 100 \text{ s}) = 92.3 \text{ K}$, in agreement with previously published values.²⁹ It is worth noting that the relaxation times from Amoureux et al.,²⁹ who reported only one relaxation process, match quite well with values of the present work for the α relaxation at high-temperatures,

but diverge at low temperatures (see Fig. 4) in such a way that the values are much closer to those of the α' relaxation as temperature decreases. Due to such difference the activation energy was reported to be 3502 K, which is instead close to the value of the faster α' relaxation.

The experimental "bimodal" (α and α) relaxation scheme can be rationalized based on the structural disorder concerning the statistical site occupancy of the fluorine atom. Such a disorder, identified in our X-ray diffraction analysis, is related to the existence of a twofold and a threefold symmetry axis around which the 1-F-ada molecule can perform reorientational motions (see insets in Fig. 5), which are accompanied by largeangle rotations of the molecular dipole contributing to the dielectric susceptibility. Probably, as it has been demonstrated for the high-temperature Phase I,³⁴ rotations around the C-F molecular symmetry axis are also present, but they leave invariant the dipole moment and thus cannot be observed by dielectric spectroscopy. Because rotation of only one single molecule around the twofold axis cannot occur due to steric hindrance, the observed disorder must result from a highly cooperative motion, which is energetically expensive. Based on this, we ascribe the primary α -relaxation to twofold rotations, whereas the faster α' relaxation, which also has lower activation energy, is ascribed to rotations around the threefold axis, as these are characterized by lower steric hindrance in the structure of Phase II.

Because 1F-ada is a rigid molecule,^{43,44} it cannot exhibit any "internal" relaxation associated with intramolecular dipole moment changes within the analyzed frequency range. This means that the fast β relaxation (whose characteristic relaxation time is plotted in Fig. 5) should be ascribed to the Johari– Goldstein (JG) relaxation.

This process involves only local environments in which molecules perform small-angle thermally activated reorientational motions according to an homogenous interpretation²² or motions due to the existence of "islands of mobility" due to the structural inhomogeneities.^{20,45} Several reasons are pled for this argument. First, there are no other possible whole molecular rotations along different symmetry axes compatible with the space group symmetry of the tetragonal phase, in addition to those giving rise to the α - and α' -relaxations. Second, the number of molecules in the asymmetric unit is just one, so the existence of heterogeneous dynamics appearing as a consequence of different relaxations from different molecules is not possible.^{15,16}

The JG character of the relaxation may in some cases be checked by comparing the experimental data with the predictions of the so-called coupling model (CM),²¹ according to which the JG relaxation process should exhibit very similar relaxation time as the precursor motion of the main α relaxation. According to the CM, the β_{JG} and the α -relaxation times should be roughly linked through the equation:

$$\log (\tau_{\rm JG}) = (1 \qquad \beta^{\rm KWW}) \log(t_{\rm c}) + \beta^{\rm KWW} \log (\tau_{\alpha})$$

where t_c is a cross-over time of the order of 2 ps.

The coupling model is expected to hold at high temperatures, namely above the glass transition. Unfortunately, the strength of

the $\beta_{\rm JG}$ relaxation is quite small (see Fig. 4), so that it is difficult to perform a reliable fit at temperatures higher than the glass transition temperature T_g . We can still compare the prediction of the coupling model with the experimental data close to the glass transition temperature. As visible in Fig. 5, the calculated values for the JG relaxation times according to the CM employing either the α or α' fit parameters are not far away from the experimental $\beta_{\rm JG}$ relaxation times, although a discrepancy is observed. Similar deviations at sub- T_g temperatures have been found in structural glasses and attributed to the non-equilibrium state of the system.^{46,47}

The $\beta_{\rm JG}$ relaxation in Phase II follows, as is usually found, the Arrhenius law below the glass transition temperature with an activation energy of 919 K (7.64 \pm 0.38 kJ mol⁻¹) and a pre-exponential factor of $\tau_{\rm o}(\beta) = 1.4 \times 10^{-9}$ s (see Fig. 5). The situation at $T > T_{\rm g}$ is less clear because $\beta_{\rm JG}$ relaxation is generally swallowed by the nearby α -process. Moreover, in the present case, the existence of a shoulder (α' relaxation) adds an additional difficulty.

Conclusions

The results reported herein provide clear evidence that universal properties pertaining to the glass behavior also apply to solid state disordered matter, as we show for the particular case of 1-fluoro-adamantane. We demonstrate that the low-temperature Phase II of 1-fluoradamantane (space group $P\bar{4}2_1c$) displays an intrinsic statistical disorder concerning the site occupancy of the fluorine atom. We further show that such statistical disorder is dynamic in character, and is in fact the consequence of the existence of large-angle rotations around a twofold and a threefold axis, which would exist for a molecule devoid of the fluorine atom. Such rotations give rise to a bimodal primary relaxation, namely to a more prominent α relaxation, visible as a loss peak in the imaginary dielectric susceptibility, and to an α' relaxation visible as a shoulder to the latter peak. Moreover, a secondary β -relaxation, identified as the Johari–Goldstein relaxation (due to the rigid character of the molecule, no intramolecular relaxations exist), has been found at temperatures below the glass transition. Both α and α' relaxations display an Arrhenius behavior, thus indicating a strong character of this glasslike system according to Angell's classification, in spite of the low value of the stretching β^{KWW} parameter of both relaxations close to the glass transition.

To the best of our knowledge, this is the first time two distinct primary relaxations are observed for separate rotational degrees of freedom of the same rigid molecule. Such a feature can only be shared by OD systems where the intermolecular interactions are strongly anisotropic and not by other types of glass formers such as supercooled liquids or plastic crystals, which display isotropic rotations. Our work demonstrates orientationally disordered phases as a class of systems with richer and more varied relaxation dynamics than "conventional" glass formers. Considering that many rigid molecular structural glass-formers are prone to crystallization and not amenable for study, orientationally disordered solids displaying the universal features of glasses are thus ideal systems in which the structure and dynamics should allow a deeper understanding of the underlying physics. In particular, these systems are a unique opportunity to elucidate the isomorph invariance of the structure and dynamics theory recently generalized for classical crystals.^{48,49}

Acknowledgements

This work was partially supported by the Spanish Ministry of Science and Innovation through the project FIS2014-54734-P and by the Generalitat de Catalunya under the project 2014 SGR-581.

References

- 1 P. G. Debenedetti and F. H. Stillinger, Nature, 2001, 410, 259.
- C. A. Angell, K. L. Ngai, G. B. McKenna, P. F. McMillan and S. W. Martin, *J. Appl. Phys.*, 2000, 88, 3113.
- 3 K. L. Ngai, J. Non-Cryst. Solids, 2007, 353, 709-718.
- 4 P. Lunkenheimer, U. Schneider, R. Brand and A. Loidl, *Contemp. Phys.*, 2000, **41**, 15.
- 5 P. W. Anderson, Science, 1995, 267, 1615.
- 6 M. D. Ediger, C. A. Angell and S. R. Nagel, *J. Phys. Chem.*, 1996, **100**, 13200–13212.
- 7 S. N. Tripathy, Z. Wojnarowska, J. Knapik, H. Shirota, R. Biswas and M. Paluch, *J. Chem. Phys.*, 2015, **142**, 184504.
- 8 M. Jiménez-Ruiz, A. Criado, F. J. Bermejo, G. J. Cuello, F. R. Trouw, R. Fernandez-Perea, H. Löwen, C. Cabrillo and H. E. Fischer, *Phys. Rev. Lett.*, 1999, 83, 2757.
- 9 H. M. Huffman, S. S. Todd and G. D. Oliver, *J. Am. Chem. Soc.*, 1949, **71**, 584.
- 10 K. Adachi, H. Suga and S. Seki, *Bull. Chem. Soc. Jpn.*, 1972, 45, 1960.
- 11 M. Sorai and S. Seki, Mol. Cryst. Liq. Cryst., 1973, 23, 299.
- 12 H. Suga and S. Seki, J. Non-Cryst. Solids, 1974, 16, 171-194.
- C. A. Angell, L. E. Busse, E. I. Cooper, R. K. Kadiyala,
 A. Dworkin, M. Ghelfenstein, H. Szwarc and A. Vassal,
 J. Chim. Phys. Phys.-Chim. Biol., 1985, 82, 267.
- 14 M. Romanini, P. Negrier, J. Ll. Tamarit, S. Capaccioli, M. Barrio, L. C. Pardo and D. Mondieig, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2012, 85, 134201.
- 15 M. Zuriaga, L. C. Pardo, P. Lunkenheimer, J. Ll. Tamarit, N. Veglio, M. Barrio, F. J. Bermejo and A. Loidl, *Phys. Rev. Lett.*, 2009, **103**, 075701.
- 16 M. J. Zuriaga, S. C. Perez, L. C. Pardo and J. Ll. Tamarit, J. Chem. Phys., 2012, 137, 054506.
- 17 P. Negrier, M. Barrio, J. Ll. Tamarit, L. C. Pardo and D. Mondieig, *Cryst. Growth Des.*, 2012, **12**, 1513–1519.
- 18 S. C. Pérez, M. Zuriaga, P. Serra, A. Wolfenson, P. Negrier and J. L. Tamarit, *J. Chem. Phys.*, 2015, **143**, 134502.
- 19 P. Negrier, M. Barrio, J. L. Tamarit, D. Mondieig, M. Zuriaga and S. C. Pérez, *Cryst. Growth Des.*, 2013, **13**, 2143–2148.
- 20 G. P. Johari and M. Goldstein, J. Chem. Phys., 1970, 53, 2372.
- 21 K. L. Ngai, *Relaxation and Diffusion in Complex systems*, Springer Science & Business Media, New York, 2011.

- 22 F. H. Stillinger, Science, 1995, 267, 1935-1939.
- 23 C. A. Angell, Science, 1995, 267, 1924–1935.
- 24 H. Tanaka, *Phys. Rev. E: Stat., Nonlinear, Soft Matter Phys.*, 2004, **69**, 021502.
- 25 Y. Huang, D. F. R. Gilson, I. S. Butler and F. Morin, *J. Phys. Chem.*, 1991, **95**, 2151–2156.
- 26 N. Arul Murugan, R. S. Rao, S. Yashonath, S. Ramasesha and
 B. K. Godwal, *J. Phys. Chem. B*, 2005, **109**, 17296–17303.
- 27 J. P. Amoureux, M. Bee and J. C. Damien, Acta Crystallogr., Sect. B: Struct. Crystallogr. Cryst. Chem., 1980, 36, 2633–2636.
- 28 N. Arul Murugan and S. Yashonath, *J. Phys. Chem. B*, 2005, **109**, 2014–2020.
- 29 J. P. Amoureux, M. Castelain, M. Bee, M. D. Benadda and M. More, *Mol. Phys.*, 1985, 55, 241–251.
- 30 T. Clarck, T. Mc. O. Kknox, H. Mackle and M. A. Mckervey, *J. Chem. Soc., Faraday Trans.* 1, 1977, 73, 1224.
- 31 J. P. Amoureux, M. Bee and J. L. Sauvajol, Acta Crystallogr., Sect. B: Struct. Crystallogr. Cryst. Chem., 1982, 38, 1984–1989.
- 32 J. P. Amoureux, R. Decressain and M. Castelain, in *Dynamics of molecular crystals*, ed. J. Lascombe, Elsevier Science Publishers B.V., Amsterdam, 1987, vol. 46, pp. 559–563.
- 33 Y. Guinet, J. L. Sauvajol and M. Muller, *Mol. Phys.*, 1988, **65**, 723–738.
- 34 M. Bee and J. P. Amoureux, Mol. Phys., 1983, 50, 585.

- 35 S. Havriliak and S. Negami, J. Polym. Sci., Part C: Polym. Symp., 1966, 16, 99–117; S. Havriliak and S. Negami, Polymer, 1967, 8, 161–210.
- 36 K. S. Cole and R. H. Cole, J. Chem. Phys., 1941, 9, 341.
- 37 MS Modeling (Materials Studio), version 5.5. http://accelrys. com/products/collaborative-science/biovia-materials-studio.
- 38 G. S. Pawley, J. Appl. Crystallogr., 1981, 14, 357-361.
- 39 J. P. Amoureux and M. Foulon, Acta Crystallogr., Sect. B: Struct. Sci., 1987, 43, 470–479.
- 40 H. M. Rietveld, J. Appl. Crystallogr., 1969, 2, 65.
- 41 H. Toraya and F. Marumo, J. Mineral., 1981, 10, 211.
- 42 F. Alvarez, A. Alegría and J. Colmenero, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1991, 44, 7306.
- 43 N. T. Kawai, D. F. R. Gilson and I. S. Butler, *Can. J. Chem.*, 1991, **69**, 1758.
- 44 A. C. Legon, J. Tizard and Z. Kisiel, J. Mol. Struct., 2002, 612, 83-91.
- 45 G. P. Johari, J. Non-Cryst. Solids, 2002, 317, 307-310.
- 46 P. Lunkenheimer, L. C. Pardo, M. Köhler and A. Loidl, *Phys. Rev. E: Stat., Nonlinear, Soft Matter Phys.*, 2008, 77, 031506.
- 47 D. Prevosto, S. Capaccioli, M. Lucchesi, P. A. Rolla and K. L. Ngai, *J. Chem. Phys.*, 2004, **120**, 4808.
- 48 D. E. Albrechtsen, A. E. Olsen, U. R. Pedersen, T. B. Schroder and J. C. Dyre, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2014, **90**, 094106.
- 49 J. C. Dyre, J. Phys. Chem. B, 2014, 118, 10007-10024.