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Microdevice for studying the *in situ* permeabilization and characterization of *Chlamydomonas reinhardtii* in lipid accumulation phase

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ABSTRACT

Microalgae are considered as a renewable source of lipid-rich biomass feedstock for biofuels due to their high 19 fatty acids content when cultivated in stress conditions (nitrogen starvation). Nevertheless the use of solvents 20 in conventional extraction methods raises important environmental, health and safety issues. The application 21 of Pulsed Electric Field (PEF) to electroporate microalgae is a promising alternative to traditional processes in- 22 volved in lipid recovery, as it might permeabilize cell membrane, easing the access out of the cytoplasm, and re- 23 ducing the use of solvents. In order to study the PEF effects on Chlamydomonas reinhardtii, we developed a 24 microdevice that allows real time visualization during such electrical solicitation. A high number of electropora- 25 tion chambers are designed on this biochip to characterize, in real-time, and in parallel, the permeabilization of 26 cells subjected to PEF using the propidium iodide (PI). Several conditions were investigated (pulse energy, pulse 27 duration and electrical field amplitude). Reduced energy consumption, heat effects and electrochemical reactions 28 are obtained when applying short pulses (5 μ s) of high electric field (4 to 6 kV \cdot cm⁻¹). Moreover, an increase is 29 observed in cell diameter and lipid content over time in nitrogen stress conditions. The cell sensitivity to the PEF 30 seems to be affected by the cell diameter. Finally, for the first time, lipid droplet redistribution was observed with- 31 in the cytoplasm during the treatment, showing that 5 μ s pulses lead to additional intracellular electroporation 32 effects. 33

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43 **1. Introduction**

44 1.1. Microalgae as a renewable source

45 Due to their chemical composition (proteins, pigments, starch, fatty 46 acids) algae can be used for food, feed, medical and energy purposes [1, 47 2]. Many algal strains are able to accumulate high amounts of fatty acids 48 as triacylglycerols (TAGs). The accumulation reaches up to 20–50% of 49 their dry weight in certain conditions such as high light intensity and ni-50 trogen limitation stress [3]. Thus, algae are considered as a renewable 51 source of lipid-rich biomass feedstock for biofuels [4]. This generation of biofuels demonstrates high productivity and no competition with 52 food crops in comparison with the 1st and 2nd generations [5]. 53

The overall energy consumption for each step of the production, 54 from algae culture to downstream processes (harvesting, molecules ex- 55 traction) is a decisive factor on the price of the end product [6]. The en- 56 vironmental impact of each step, regardless of the energetic 57 consumption, is also a key factor [7–9]. Downstream processes, such 58 as, classic mechanical disruption and harvesting methods, are constant- 59 ly challenged because of their numerous drawbacks. Moreover, extrac- 60 tion methods are generally ineffective when applied on wet intact 61 cells. New technologies have been studied to weaken algal cells prior 62 to wet extraction, including: microwaves [10,11], ultrasounds [10,12] 63 and electrical fields [13]. All these technologies show very low energy 64 consumptions. However, before their use in the algae industry, their im- 65 provement at laboratory scale is mandatory. Thus, studies concerning 66 pulsed electric fields (PEF) applied to algae cells are increasing consider- 67 ably, as this process weakens cell membranes and improves the extrac- 68 tion of soluble compounds, such as, proteins and carbohydrates [14], or 69 large hydrophobic molecules [15]. 70

PEF is broadly used for other applications in biology, *e.g.*, DNA trans-71 fection [16], drug delivery into tissue cells [17,18]; and in food process-72 es: treatment of fruit juices [19,20], pasteurization [21,22], sugar 73

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Abbreviations: a.u., arbitrary units; Cm (μ F·cm⁻²), membrane capacitance; DMSO, dimethyl sulfoxide; E (kV·cm⁻¹), Electric field; E₅₀ (kV/cm), electric field corresponding to 50% of permeabilization; FDA, fluorescein diacetate; IPA, isopropyl alcohol; Np, number of pulses delivered during a burst of PEF treatment; nPl₀, number of cells stained by Pl in an electroporation chamber before treatment; nPl₁, number of cells stained by Pl in an electroporation chamber after treatment; ntot₁, number of cells counted in an electroporation chamber after treatment; ntot₁, number of cells counted in an electroporation chamber after treatment; Nto₁, number of cells counted in an electroporation chamber after treatment; Nto₁, number of cells counted in an electroporation chamber after treatment; Nto₁, number of cells counted in an electroporation chamber after treatment; Nto₁, number of cells counted in an electroporation chamber after treatment; Nto₁, number of cells counted in an electroporation chamber after treatment; Nto₁, number of cells counted in an electroporation chamber after treatment; Nto₁, number of cells counted in an electroporation chamber after treatment; Nto₁, number of cells counted in an electroporation chamber after treatment; Nto₁, number of cells counted in an electroporation chamber after treatment; Nto₁, number of cells counted in an electroporation chamber after treatment; Nto₁, medium tris-acetate-phosphate; TAPN-, medium tris-acetate-phosphate depleted in nitrogen; W (kJ·m⁻³), energy delivered; Δt_{pu} (μ), pulse width; ΔT_{pu} (°C), heating resulting from on pulse; ΔT_{burst} (°C), heating resulting from on burst; $\Delta \Psi_i$ (V), induced trans-membrane voltage; σ (S·m⁻¹), conductivity; θ (°), angular position on the cellular membrane facing the electrodes.

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extraction from beets [16,23] or pigments extraction from potatoes [24].

The use of PEF on microalgae aims several targets: extraction of lipids [15,25–27], pigments [14,28] and proteins [14,29], or lysis of toxic

77 algae [30].

78 1.2. Electropermeabilization of microalgae

The application of PEF raises the transmembrane potential up to a critical value and in turn induces membrane permeabilization [31] due to creation of pores, their size and reversibility of which depend on the treatment intensity and duration [32]. Thus, depending on the shape, amplitude (E), duration (Δt_{pu}) and number (N_p) of electric field pulses, different effects on algae can be modulated. Parameter values found in the literature are shown in Table 1.

Conditions found in the literature vary considerably (Table 1). Usual pulse duration ranges from several μ s to several ms, while the pulse amplitude varies from 1 to 50 kV · cm⁻¹ [33]. The achievement of a nonlethal level of poration is dependent on both parameters [18].

90 1.3. Electroporation parameters

91The level of electroporation can be estimated from the Schwan equa-92tion (Eq. (1)) [34], which gives the potential ($\Delta\Psi$ i) induced on cells sub-93mitted to PEF.

$$\Delta \Psi i = -\frac{3}{2} r E \cos(\theta) \cdot \left(1 - e^{\frac{-\Delta t_{pu}}{\tau}}\right) \tag{1}$$

where r is the cell radius, Δt_{pu} the duration of an electric field pulse, τ the charging time of the cellular membrane and θ the angular position on the cellular membrane facing the electrodes.

⁹⁸ The induced potential ($\Delta \Psi i$) required to trigger permeabilization is ⁹⁹ known to be in the range 0.2–1.5 V for mammalian cells [33]. Algae ¹⁰⁰ strains differ in cell radius (r) and cell properties (affecting τ – ¹⁰¹ Eqs. (2) and (3)). These features are paramount for the selection of ¹⁰² the electric field intensity E inducing permeabilization (Eq. (1)) [29].

Furthermore, conventional electroporation uses pulses with a longer 103duration (Δt_{pu}) than the plasma membrane charging time (τ). The 104 membrane charging time (τ) , estimated from 0.4 to 1 µs for mammalian 105cells [35], depends on the membrane specific capacitance (Cm), the cell 106 107radius (r), and the medium (σ_{med}), cell wall (σ_{cw}), cell membrane (σ_m) and cytoplasm (σ_{cyto}) conductivities, respectively, as shown in Eq. (2) 108(deduced from the single shell model [36]) or in Eq. (3) (deduced 109110from the double shell model where the cell wall is considered [37]).

$$\tau = r Cm \left(\frac{1}{\sigma cyt} + \frac{1}{2 \sigma med} \right)$$
(2)

95

$$\tau = r Cm \left(\frac{1}{\sigma cyt} + \frac{\sigma med + \sigma cw}{2 \sigma med \sigma cw} \right).$$
(3)

115

The full charge of the membrane requires a duration of the electric field application longer than several times the charging time τ ($\Delta t_{pu} > 5 \tau$ to reach 95% of the final charge, when considering Eq. (1)). For this reason, very short pulses ($\Delta t_{pu} < 1 \mu$ s) may lead to cell apoptosis by affecting internal organelles, while the membrane charge does not reach the permeabilization level [38]. It has been shown that longer 120 pulse durations ($\Delta t_{pu} > 1 \mu s$) can lead to increased pore radius and 121 resealing time [32,39] (time needed for the membrane to recover, if 122 the permeabilization is reversible). Besides, millisecond pulses can be 123 applied in order to weaken the mechanical resistance of the cells, such 124 as, cytoskeleton [40] or cell wall [29]. As shown in Eq. (1), the induced 125 transmembrane potential $\Delta \Psi_i$ is non-homogenous on the cell surface, 126 as it depends on the angle θ of the cell with the applied field direction. 127 This may focus the effect of PEF to a small region of the membrane [41]. 128

1.4. Electroporation side effects 129

Applying an electric field in a liquid medium may result in undesir- 130 able side effects, such as, Joule heating, water electrolysis and redox re- 131 actions at the electrodes [42]. The thermal aspect should indeed be 132 considered when applying PEF, as the Joule effect occurs in the conduc- 133 tive medium. The energy delivered during the PEF treatment W 134 (expressed in J·m⁻³), can be expressed thanks to Eq. (4): 135

$$W = |E|^2 \,\Delta t_{pu} \,\sigma \tag{4}$$

137

142

where σ is the medium conductivity in S·m⁻¹.

By neglecting thermal external exchanges (diffusion, convection), 138 the temperature elevation (ΔT_{pu}) induced by a pulse, due to the Joule effect, can be over-estimated as shown in Eq. (5) [33].

$$\Delta T p u = \frac{W}{C \rho} \tag{5}$$

where $C \cdot (J \cdot m^{-3} \cdot K^{-1})$ is the heat capacity of the medium, and ρ its density.

During the treatment, if the frequency of pulse delivery is too high 144 and does not enable the temperature to decrease between pulses [43], 145 temperature may increase by several dozens of degrees Celsius, affecting cell viability and degrading valuable compounds, such as, lipids, pigments or proteins. To prevent this heating, some studies on microalgae are performed in a low conductivity buffer [28,29] or with a cooling system [15,25]. However, it is well known that the conductivity of the medium increases during PEF treatment because of the leakage of ions out of the cells [26,39], which might enhance the temperature increase.

Water electrolysis may also occur on electrodes. This leads to gas153production at the cathode (hydrogen) and at the anode (oxygen).154These mechanisms may have many consequences during PEF applica-155tion, including: interference with the electric field distribution, change166in the medium conductivity, generation of reactive oxygen species and157evolution of the medium pH close to the electrodes. Finally, both tem-158perature [44] and reactive oxygen species [45] can affect the perme-159abilization threshold (the electric field needed to open pores in the160membrane) of cell membranes.161

In this paper, the effects of PEF application on algae will be observed 162 in real time, using a dedicated microdevice [46]. This microdevice allows 163 real time observation of the cell wall behavior and the distribution of internal lipid droplets during and after pulses application [47,48]. Moreover, the efficiency of the treatment is discussed with respect to the energetic cost of the treatment and heating aspects. 167

t1.1	Table 1
t1.2	PEF parameters (pulse duration, electric field, pulse shape) used on several algae strains.

1.3	Pulse duration (µs)	Electric field ($kV \cdot cm^{-1}$)	Pulse shape	Pulse number	Strain	Cells diameter (µm)	Study
1.4	1	23-43	Square	20-110	Auxenochlorella protothecoides	5–8	[26]
1.5	6-150	10-25	Square	50	Chlorella vulgaris	2-4	[28]
1.6	10	20	Exponential decay	1-600	Nannochloropsi s	2-3	[14]
1.7	100	2.7	Square	21	Chlorella vulgaris	2-4	[27]
1.8	2 000	3	Square bipolar	30	Chlorella vulgaris	2-4	[29]

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168 2. Material and methods

169 2.1. Microalgae culture

Algae strains were obtained from the Culture Collection of Algae at 170the University of Goettingen (EPSAG, Nikolausberger Goettingen, 171 Germany). Cultures assessed in this investigation were: Chlamydomonas 172reinhardtii SAG 34.89 (wild type), C. reinhardtii SAG 83.81 (Cell wall de-173174ficient mutant strain from the wild type 34.98). C. reinhardtii wild type and the mutant were cultivated in TAP Medium [49] for the growth 175phase and in TAP nitrogen depleted medium for the lipid accumulation 176(NH₄Cl removed from the TAP medium). 177

178 2.1.1. Culture conditions

179 Cells were cultivated photo-autotrophically in exponential growth 180 in 250 mL Erlenmeyer flasks with a culture volume ranging from 20 to 181 60 mL: rotation speed 150 rpm, temperature 25 °C, enriched air 1.5% 182 CO₂, white light 20 μ mol·m⁻²·s⁻¹ at the surface of the flasks. In the ex-183 ponential phase, microalgae concentrations ranged from 5 · 10⁴ to 184 5 · 10⁷ cells·mL⁻¹.

In order to trigger the accumulation of neutral lipids in algae cells, 185 the cells were centrifuged (6000 g, 6 min) and resuspended, at a con-186 centration of $2 \cdot 10^6$ cells \cdot mL⁻¹ in a 40 mL TAP nitrogen depleted me-187 dium. The following conditions were applied: light intensity of 188 150 μ mol \cdot m⁻² \cdot s⁻¹, agitation at 50 rpm, temperature set at 24 \pm 189 1 °C, 0.0035% CO₂. Cell concentration ranges from 2 \cdot 10⁶ cells \cdot mL^{-1} 190to $1 \cdot 10^7$ cells \cdot mL⁻¹. In terms of dry weight, the amount of biomass 191 after 7 days of stress was 1.23 ± 0.06 g·L⁻¹ for 7.6 $\cdot 10^{6}$ cells·mL⁻¹. 192

193 2.2. Microalgae characterization

Cell concentration (in cell \cdot mL⁻¹) was monitored using the Guava 194easyCyte 5 flow cytometer (Millipore Corporation 25801 Industrial 195Blvd Hayward, CA 94545). The analysis method was developed on 196Incyte 2.7 using the following parameters: Forward Scatter gain = 8; 197 Side Scatter gain = 8; RED gain (680 nm) = 8; 5 decades acquisitions; 198 total counts = 2000 events; flow rate = $0.59 \,\mu\text{L}\cdot\text{s}^{-1}$.; cell concentration 199 for analyses: from 1 to $3 \cdot 10^5 \cdot \text{cell} \cdot \text{mL}^{-1}$; Forward Scatter threshold: 200 2.10^2 a.u.; C. reinhardtii gate: Forward Scatter from $3 \cdot 10^2$ to 201 $2 \cdot 10^4$ a.u. and Side Scatter from $2 \cdot 10^2$ to $2 \cdot 10^4$ a.u.; threshold on 202 203RED fluorescence at $3 \cdot 10^1$ a.u.

Cell size measurements were performed using an inverted micro-204 scope Nikon Eclipse Ti–U, with a $40 \times$ magnification objective (Nikon 205Achromat LWD 40 \times Ph1 ADL NA = 0.6) and a CMOS colored camera 206 Nikon DS-Ri2 (4908 \times 3264 pixels, 36 \times 23.9 mm) equipped with an ad-207208ditional 2.5 \times magnification lens. Cell diameter (220 \pm 20 cells) was measured manually from bright field images using the camera software 209NIS-Elements D 4.30. The mean cell size (referred in Fig. 3) is defined as 210the longest diameter of the spheroidal shaped cells. 211

To follow the neutral lipid content, Bodipy 505/515 212 213(Lifetechnologies, D-3921) was used to stain lipid droplets in the cell cytoplasm. Bodipy was diluted in DMSO to a concentration of $100 \text{ mg} \cdot \text{L}^{-1}$. 214Then 10 μ L of the stock solution was added to 10^6 cells \cdot mL⁻¹ (final con-215centration of 1 μ g·mL⁻¹). Stained cells were incubated by rotative 216mixing in the dark during 10 min. The cells were then centrifuged at 2172415 g for 5 min and resuspended in a TAP medium depleted in 218nitrogen. 219

The accumulation of Bodipy-stained lipid droplets was monitored 220 using the Guava easyCyte 5 flow Cytometer (Millipore Corporation 221 25801 Industrial Blvd Hayward, CA 94545). The following analysis 222method is used to gate C. reinhardtii cells: The mean Bodipy fluorescence 223was estimated on a green detector 525/30 nm (Green gain = 1) from 224gated C. reinhardtii cells. The green autofluorescence value (fluores-225cence without Bodipy) was subtracted from the mean fluorescence 226227 value. Confocal scanning laser microscopy (CLSM) confirmed that Bodipy stained exclusively the lipid droplets and not the phospholipids 228 of the cytoplasmic membrane, as previously reported [50]. 229

The experiments were performed using a pulse generator Betatech 231 Electrocell B10 HVLV. The generator parameters were: U = -1000 V 232 to +1000 V (bipolar pulses), $\Delta t_{pu} = 5-50,000 \ \mu s$ (pulse duration), 233 $T_i = 5-50,000 \ \mu s$ (duration between positive and negative pulse), 234 $P = 80-9999 \ 500 \ \mu s$ (period of one pulse cycle), burst of pulses: $1-235 \ 10,000 \ pulses$.

2.4. Device microfabrication

The biodevice was designed to trap microalgae cells in visualization 238 chambers. Those chambers were made of thick photoresist, as detailed 239 hereafter, and were patterned above a parallel electrode network. 240 Cells were monitored by microscopy during and after application of 241 pulse electric fields in bright or fluorescence fields. The biochip 242 consisted of two layers: an electric layer to deliver pulsed electric fields 243 and a fluidic layer to trap the cells. It was fabricated using a conventional 244 microfabrication process in a clean room [51]. 245

For the electrodes, a 20 nm chromium adhesion layer, covered by 246 150 nm gold layer, was sputtered on a quartz substrate. A first photoli-247 thography step was employed to pattern the electric layer. This process 248 includes the deposition of S1805 Shipley photoresist by spin coating 249 (t = 30 s, v = 1000 rpm), followed by a prebake (T = 115 °C, t = 250 1 min) and UV exposure (365 nm, intensity = 120 mJ·cm⁻²) 251 (Fig. 1a). A developing step was then performed (developer 351, t = 252 1 min) (Fig. 1b), followed by gold etching with KI (4 g KI, 1 g I₂, 40 mL 253 H₂O, t = 7 s) and chromium etching (ChromeEtch18 micro resist tech-254 nology, t = 45 s) (Fig. 1c). A cleaning step was done using acetone and 255 isopropanol (IPA).

The microfluidic level was patterned with a second photolithogra- 257 phy step: the 25 µm high microfluidic chamber was defined by negative 258 thick photoresist (SU8 2025, Microchem©) spin coated in two steps: 259 t1 = 5 s, v1 = 500 rpm and t2 = 30 s, v2 = 2700 rpm. A soft-baking 260 (t = 3' at 65 °C and t = 15' at 95 °C), followed by UV exposure 261 (365 nm, intensity = 240 mJ·cm⁻²) (Fig. 1d) gave the pattern for the 262 microdevice design (Fig. 1e). A post exposure bake was then done 263 (t = 3' at 65 °C and t = 15' at 95 °C) before development (MicroChem 264 © SU8 developer, t = 6 min) associated to a hard-baking step (T = 265 175 °C, t = 30', slope 5 °C·min⁻¹) to fix the thick resist level (3 h). 266

The electrode gap was small enough to reach very high electric fields 267 [46], up to 30 kV \cdot cm⁻¹ or 60 kV \cdot cm⁻¹ with a gap of 300 µm or 150 µm 268 using an accessible voltage generator (up to 1 kV). The microfluidic 269 layer was designed as a set of 6 lines of 15 chambers for a total of 90 270 closed chambers (Fig. 2a). There were 3 lines of 45 large chambers 271 (gap of 300 µm) or 3 lines of 45 smaller chambers (gap of 150 µm). 272 Each large chamber had an internal dimension of 720 µm × 480 µm, de- 273 signed to be observed with a 20 × objective. Small chambers possessed 274 an internal dimension of 360 µm × 150 µm designed for observation 275 with a 40 × objective. 276

The electroporation microdevice was connected to the pulse gener-277 ator by a homemade supporting base fastened above the microscope278 objectives and designed to support up to 2 kV DC voltage (Fig. 2b). 279

2.5. Characterization of the permeabilization

Cell permeabilization was characterized using propidium iodide 281 (PI). This molecule of 668 Da is an intercalating agent that binds to the 282 DNA of permeable cells. It exhibits excitation and emission peaks of re-283 spectively 536 and 617 nm. When the cell is electroporated, the PI mol-284 ecules can diffuse through the permeable membrane and stain their 285 DNA, while intact cells remain unstained. PI was dissolved in a PBS buff-286 er at a concentration of 1.5 mM and diluted in the sample of microalgae 287

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Fig. 1. Microfabrication process of the biodevice. The electrode layer is patterned on a glass substrate in 3 steps: photoresist patterning by UV light exposure (a), photoresist development (b), and Au/Cr wet etching and photoresist removing (c). A fluidic layer is patterned above the electrode layer by a second UV light exposure (d) to obtain several multiple electroporation chambers shown on the scheme (e) and picture (f).



Fig. 2. a. Scheme of the overall design of the electroporation micro-system. Orange: electric level, white: fluidic level. (*) 45 large chambers (internal dimensions: 720 µm × 480 µm, electrode gap: 300 µm) designed to be observed with a 20 × objective, (**) 45 small chambers (internal dimensions: 360 µm × 150 µm, electrode gap: 150 µm) designed to be observed with a 40 × objective. b. Picture of the chip connected to the generator, placed above a microscope objective.

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cells (with a cell concentration of $2 \cdot 10^6 - 1 \cdot 10^7$ cells $\cdot ml^{-1}$) to a final concentration of 100 μ M. Then 15 μ L of this sample was placed on the surface of the device.

A glass slide (24 mm \times 30 mm, d = 1 mm) covered the droplet (15 µL) of algae solution, spreading it in a way that it filled multiple electroporation chambers. All experiments were performed using TAP culture medium. The medium conductivity was 0.213 S \cdot m⁻¹ at 20 °C.

The microscopic observations were conducted with a Nikon eclipse Ti-U inverted microscope using a $20 \times$ objective (Nikon Plan fluor $20 \times$ DIC N2 NA = 0.6) $40 \times$ objective (Nikon Achromat LWD $40 \times$ Ph1 ADL NA = 0.5).

Three chambers were utilized for the quantitative analysis of PI pen-299 etration. First, the cells in the chambers were observed with a $20 \times$ ob-300 jective and the images were recorded using bright field mode (no 301 fluorescence filter, exposure time of 10 ms and gain 1) and CY3 fluores-302 cence (exposure time of 700 ms, and gain 4). The ratio of the number of 303 stained cells before the PEF treatment and the total number of cells 304 305 within the chamber was calculated. PEF treatment was then applied. It was observed that PI penetrated permeabilized cells within a few sec-306 onds. The counting of permeabilized cells was done, more than 30 s 307 after the treatment, to let treated cells settle down within the focus of 308 309 the objective. As in the previous treatment, bright field and fluorescent 310 images were recorded, respectively, in order to estimate the total number of cells and finally the percentage of permeabilized cells in the 311 chambers. 312

It should be pointed out that some cells might be naturally perme-313 abilized because of adverse culture conditions or natural mortality. In 314 315 the calculation of the PI uptake (Eq. (9)), for each measurement, the total number of cells stained by PI before treatment (nPI₀ consisting of 316 317 naturally permeabilized cells) were subtracted from the stained cell 318 number after treatment (nPI₁). The total cell number counted in the 319bright field $(ntot_1)$ was also taken into account in the calculation, as 320 shown in Eq. (9).

PI uptake (%) =
$$\frac{nPI1 - nPI0}{ntot1 - nPI0} \times 100.$$

322

The average number of cells per chamber was approximately 60. PI uptake estimation is calculated on 3 independent chambers and in duplicate experiments.

2.6. Permeability and viability tests in electroporation cuvettes

Additional measurements were performed in 1 mm electroporation 326 cuvettes (Dutscher, ref. 4905020) in order to assess cell viability after 327 electroporation treatment. Viability was monitored using Fluorescein 328 Diacetate (FDA, Thermo fisher scientific F1303) dye that measures cell 329 enzymatic activity. This non-fluorescent dye, after diffusing through 330 the cell is hydrolysed by esterase enzymes and releases fluorescein 331 (which exhibits excitation and emission peaks of respectively 430 nm 332 and 525 nm). Additionally, the cytoplasmic membrane must be intact 333 in order to retain the fluorescent dye inside the cell [52,53]. The follow-334 ing staining protocol was applied: FDA diluted in acetone (final concen-335 tration of 5.5 μ M, 6 min of incubation) was added to an algae solution 336 (1-2 \cdot 10⁵ cells \cdot mL⁻¹). FDA positive and negative cells were distin-337 guished by flow cytometry at 525 nm.

Cell permeabilization after PEF treatment was measured using Sytox 339 Green (SG, Thermo fisher scientific S7020), that binds to the DNA of 340 permeabilized cells. This dye and propidium iodide present similar molecular weights (SG: 660 Da, PI: 668 DA [54]). Sytox green (ratio of 342 1 μ mol SG to 1 \cdot 10⁵ · cells, 5 min of incubation) is added to an algal solution before PEF treatment. Permeabilized cells were thereafter counted by cytometry at 583 nm. 345

3. Results and discussion

In the present study, the effect of PEF on *C. reinhardtii* cells accumu- 347 lating lipids is characterized and quantified using a dedicated 348 microdevice. The increase in cell size over time, in the lipid accumula- 349 tion phase, was first evaluated (Fig. 3), as this parameter directly influ- 350 ences the level of permeabilization. The cell lipid content was thereafter 351 measured. 352

3.1. Evolution of microalgae cell size and lipid content in stress conditions 353

The mean cell diameter in exponential growth was $6.8 \pm 1.4 \mu m$, $_{354}$ which increased to $9.2 \pm 1 \mu m$ after 4 days of stress and then was con- $_{355}$ stant at $10 \pm 1 \mu m$ between day 7 and day 15 (Fig. 3). This is in agree- $_{356}$ ment with results reported by [55] that mentioned a diameter $_{357}$ increase of 2 μm (from 5.7 to 7.2 μm) after 96 h of nitrogen stress for $_{358}$ the wild type strain *Clamydomonas* CC124. No further increase in the $_{359}$



Fig. 3. Evolution of cell lipid content with time. Mean green fluorescence evaluated by flow cytometry.

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Fig. 4. Pl uptake (%) resulting from several treatments with different electric fields E (0.63, 1.13, 2 and 3.56 kV \cdot cm⁻¹) and pulse durations Δt_{pu} (5, 16, 50, 158 and 500 µs). Experiments performed with *Chlamydomonas reinhardtii* cells with 7–8 days of lipid accumulation.

cell diameter is observed after 7 days. This may be due to the presence of
 a cell wall, which seems to prevent cell swelling, as previously reported
 [56].

The evolution of lipid content in the cells was followed from day 0 (beginning of the stress) to day 16 (Fig. 3). The lipid content clearly increased during stress (nitrogen starvation).

From the previous observations, we can assume that lipid droplet ac cumulation started after 3–4 days of stress, which is in agreement with
 many results of the literature [57]. The lipid accumulation rate seems to
 decrease after 8 days of stress.

Moreover, data showed that nitrogen related stress induced a diameter increase in *C. reinhardtii*, mostly in the first 4 days, followed by a strong increase of the lipid content from day 4 to day 8. Therefore, we conclude that 8 days of stress correspond to the best time to submit cells to PEF for lipid extraction.

3.2. Effect of the PEF parameters on the electroporation level

We first investigated the relation between the PEF spent energy, for 376 a given level of pulse duration, and the efficiency of the treatment estimated by the propidium iodide uptake. The electric field amplitude was 378 therefore adapted for five pulse durations Δt_{pu} (5, 16, 50, 158 and 379 500 µs), in order to achieve different levels of energy (135, 426, 380 1243 kJ·m⁻³). Joule heating corresponding to each situation was 381 overestimated (cooling not considered) using Eqs. (4) and (5) 382 (Section 1.4). For each treatment, the heat effect ΔT , in °C, was calculatas d for one pulse (ΔT_{pu}).

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All experiments were performed using a burst of 10 Hz monopolar 385 pulses on *C. reinhardtii* SAG 34.89 (wild type) in 7 to 8 days of lipid ac-386 cumulation (the optimal treatment time determined previously 387 (Section 3.1)). 388



Fig. 5. PI uptake (%) resulting from several treatments varying electric field E (from 0.36 to 20 kV·cm⁻¹) and pulse duration Δt_{pu} (5, 16, 50, 158 and 500 µs) corresponding to 3 levels of energy: 135, 426 and 1342 kJ·cm⁻³. Experiments performed with *Chlamydomonas reinhardtii* cells with 7–8 days of lipid accumulation.

t2.1 Table 2

t2.2 Pulse duration and the corresponding electric field inducing 50% of permeabilization (E_{50}) t2.3 determined for *Chlamydomonas reinhardtii* (in 7–8 days of lipid accumulation phase).

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135
0.03
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The results of permeabilization at four levels of electric fields E, and for several treatment durations Δt_{pu} , are presented in Fig. 4, while those for several levels of energy W with respect to the pulse duration Δt_{pu} are presented in Fig. 5.

Considering the Schwan equation (Eq. (1)), the critical electric field 393 that induces electroporation is evaluated to be within the range of 0.3-394 2.0 kV \cdot cm⁻¹ for a cell radius of 5 μ m. The charging time τ is within the 395 range of 0.1–0.9 μs (calculated from Eqs. (2) and (3) in which σ_{cyt} de-396 pends on the number of lipid droplets in the cytoplasm [36]). Indeed ex-397perimental results (Fig. 4) confirm that the minimal electric field 398 required to obtain cell permeabilization is in the range of 1-399 2 kV·cm⁻¹ for the longer pulses ($\Delta t_{pu} > 16 \ \mu s$). Pulses of 5 μs require higher electric field (above 2 kV·cm⁻¹) to permeabilize the cells in 400 401 402 order to compensate the charging time of the membrane.

In addition, the electric field inducing 50% of cell permeabilization
 (E₅₀) depends on pulse duration (Table 2).

The obtained electric fields (Table 2) were lower than those reported 405 406 in the literature, regardless of the applied pulse duration (Table 1, Section 1.2). Indeed, for $\Delta t_{pu} = 5 \,\mu s$, values of 10–20 kV·cm⁻¹ have 407 been reported [14,28]. Moreover, cell electroporation has been reported 408 at 2.7 kV·cm⁻¹ for $\Delta t_{pu} = 100 \,\mu s$ [27], while our results demonstrate 409 50% cell permeabilization for a pulse duration between 50 and 500 µs 410 at 1–2 $kV \cdot cm^{-1}$. Finally, by extrapolating our results, approximately 411 5-6 kV·cm⁻¹ for 1 μ s pulse (instead of 23-43 kV·cm⁻¹ [26]) and 412 0.8–0.9 kV \cdot cm⁻¹ for 2 ms pulses (instead of 3 kV \cdot cm⁻¹ [29]) would 413be required to induce cell electroporation. 414

We can easily explain those differences by at least two reasons: (*i*) *C. reinhardtii* cells are 3 to 4 fold larger than *Chlorella vulgaris* and *Nannochloropsis.* This increase in cell diameter contributes to the reduction of the permeabilization threshold by a factor of 3–4 fold according

to the Schwan equation (Eq. (1), Section 1.3) and (*ii*), in most cases, in- 419 cluding references [26,29], the intensity of the field that is actually ap- 420 plied for algae electroporation is set much higher (2–3 times) than the 421 necessary E_{50} in order to ensure the permeabilization of 100% of the 422 cells. 423

Fig. 5 shows that, for a similar level of cell permeabilization, more 424 energy is used for long pulses. For instance, 1342 kJ·m⁻³ energy is de- 425 livered, using 10 pulses of 500 µs, to permeabilize more than 50% of 426 the cells. Nevertheless, shorter pulses require higher electric field to 427 trigger permeabilization (Table 2 and Eq. (4), Section 1.4). Interestingly, 428 by reducing the pulse duration by 10 times, the electric field is only in- 429 creased by a factor slightly less than two. Under these conditions, the 430 energy needed for cell electroporation is drastically reduced by a factor 431 of 3.15 (Table 2). Moreover, since the temperature increase is propor- 432 tional to energy, little heat production is generated when using short 433 pulses with optimized electric field, as demonstrated by our data 434 (Table 2). It should be pointed out that the temperature increase of 435 the entire treatment ΔT_{burst} after 10 pulses, remains close to the tem- 436 perature increase corresponding to one pulse ΔT_{pu} , due to the natural 437 cooling that occurs between pulses under our frequency conditions 438 and with the large surface to volume ratio [46]. Indeed, simulations 439 showed that for any treatment condition, in our system, the cooling 440 time is shorter than 0.1 s. Nevertheless, this approximation should be 441 reconsidered when using frequencies higher than 10 Hz. 449

Moreover, it should be noted that for high energy treatments 443 $(W \ge 1342 \text{ kJ} \cdot \text{m}^{-3})$, gas appeared at the vicinity of electrodes, probably 444 due to water electrolysis or redox reactions. Reducing the energy used 445 during the PEF treatment is thus a way to avoid such effects. 446

With our conditions (10 pulses), the optimal parameters (inducing 447 high permeabilization for *C. reinhardtii* after 7 days of stress with a 448 low energy consumption) were obtained for the lowest pulse duration 449 tested: 5 μ s with a corresponding field amplitude of 3.5 kV·cm⁻¹. 450

It must be noted that the energy consumption should not be the only 451 criterion to select pulse width and electric field intensity. It was indeed 452 shown in the literature that the pulse width impacts the size of the 453 pores created by the treatment; short pulses result in a multiplicity of 454 small pores while long pulses result in a few large pores [32]. Moreover, 455 the use of micro-pulses, with a duration inferior to 100 µs remains quite 456 unexplored and may have unexpected intracellular effects such as 457 impacting cells viability [33].



Fig. 6. Permeabilization efficiency of *Chlamydomonas reinhardtii* cells (wild type) over electrical field intensity (PEF treatment conditions: 10 unipolar pulses, 5 µs long, 10 Hz). The efficiency is expressed as the total number of stained cells by PI after treatment (this includes the PI stained cells prior to PEF) over the total number of cells in the sample. The colors correspond to different conditions: normal growth (green square), 4 days of stress (yellow triangle), 7 days of stress (orange diamond) and 14 days of stress (red circle). (For interpretation of the references to color in this figure legend, the reader is referred to the web version of this article.)

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Fig. 7. *Chlamydomonas reinhardtii* (wild type) permeabilized by a PEF treatment of 6 kV·cm⁻¹. (a) bright field, (b) PI detected using CY3 fluorescence filter, (c) lipid droplets stained with bodipy 505/515 using B2A fluorescence filter. Observation at the beginning of the treatment (d), 1 s after (e) and 2 s after (f). Displacement and appearance of large droplets are observed (red circles). Lipid displacement and fusion of intracellular lipid droplets can be observed in a video provided in supplementary data. (For interpretation of the references to color in this figure legend, the reader is referred to the web version of this article.)

459 3.3. Sensitivity of C. reinhardtii cells to PEF at different stages of lipid 460 accumulation

461 *C. reinhardtii* at different stages of lipid accumulation (after 4, 7 and 462 14 days of stress conditions) and in exponential growth phase has been 463 subjected to PEF treatment (burst of 10 unipolar pulses of 5 μ s with a 464 frequency of 10 Hz and an amplitude ranging from 0 to 5 kV·cm⁻¹). Re-465 sults showed that *C. reinhardtii* cell permeabilization occurred at electric 466 field intensities ranging from 2 to 5 kV·cm⁻¹ (Fig. 6).

Without PEF treatment (0 kV/cm) and in the case of cells at exponential growth and after 4 days of stress, the percentage of cells stained by PI (Fig. 6) is below 5%. Permeability achieves values in the range of 10 to 20% after 7 days of stress, and dramatically increases above 40% after 14 days of stress. These results show that *C. reinhardtii* cells are naturally permeabilized depending on the time of application and the starvation 472 stress condition. 473

Furthermore, lipid accumulation does not seem to affect the perme- 474 abilization threshold. Indeed after 4, 7 or 14 days of stress, E_{50} remains 475 within the range of 3–3.5 kV·cm⁻¹ despite the expected reduction 476 (by a factor of 6) of the cytoplasm conductivity under these conditions, 477 as mentioned in reference [36].

In addition, we found that the cells in exponential growth seem to 479 be less sensitive to the treatment. The difference is mostly notable 480 for $4 \text{ kV} \cdot \text{cm}^{-1}$; the total number of cells stained by propidium iodide 481 after PEF treatment was $50.5 \pm 7.8\%$ for exponential growth condi-482 tions, while more than 80% of the total cells were stained under 483 stress conditions. This difference in cell behavior may be explained 484 by the difference in the size of cells between day 0 and day 4 (see 485)



Fig. 8. Chlamydomonas reinhardtii cell wall deficient mutant in exponential growth conditions (a) before and (b) 2 min after 6 kV·cm⁻¹ treatment.

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Fig. 9. Red triangle – permeability (measured by Sytox Green uptake) during electroporation in cuvettes; green diamond – enzymatic activity (measured by FDA staining, 1 h after treatment). Treatment: burst of 10 pulses, 10 Hz, pulse duration: 5 µs, cells: *Chlamydomonas reinhardtii* after 7 days of stress. (For interpretation of the references to color in this figure legend, the reader is referred to the web version of this article.)

486 Section 3.1, Fig. 2), which affects the permeabilization threshold 487 (Eq. (1), Section 1.3).

488 3.4. Qualitative observations

PI staining of *C. reinhardtii* cells (Fig. 7b) after PEF treatment ($6 \text{ kV} \cdot \text{cm}^{-1}$, burst of 5 µs 10 Hz unipolar pulses) demonstrates membrane permeabilization. However, the cell structure was maintained as suggested by the bright field observation (Fig. 7a). In addition, bodipy fluorescence showed that the lipid droplets remain in the cytoplasm (Fig. 7c). The cell wall may prevent lipid exit from the cytoplasm or cell disintegration even after electrical treatment.

However, a rearrangement of lipid droplets in the cytoplasm was ob-496 served a few seconds after the PEF treatment (Fig. 6, d = 0 s, e = 1 s, f =497 2 s after treatment); some droplets have merged 2 s after treatment. 498 Lipid displacement and fusion of lipid droplets within the cytoplasm 499 can be observed in real time during the PEF treatment in a video provid-500ed in supplementary data. On the whole, observations performed at re-501peated times in bright and fluorescence fields showed that very short 502503pulses (5 µs) permeabilize the cytoplasm membrane (shown by PI penetration) with limited effects on the cell wall (Fig. 6a), but with clear ad-504ditional effects on the intracellular lipids. 505

In addition, the effects of PEF were also tested on a cell wall deficientstrain (SAG 34.98) (Fig. 8).

A strong effect on cell integrity was observed in the mutant compared with the wild type strain (Fig. 8). This experiment underlines the fact that when a cell is permeabilized, the cell wall maintains the cell integrity (Fig. 7.a), while a cell wall deficient strain, as *C. reinhardtii* cw15, is disintegrated in such fields (Fig. 8) [56].

513 3.5. Permeability and viability after PEF treatment

Results show that 50% of cells were permeabilized for electrical fields ranging from 3 to 3.5 kV·cm⁻¹ (Fig. 9, sytox green uptake experiment), which is in agreement with data in Fig. 6. Besides, we could demonstrate that the enzymatic activity, measured through fluorescein diacetate (FDA), is affected by electric fields higher than 5 kV·cm⁻¹. Particularly at 7 kV·cm⁻¹ where more than 90% of the cells have lost their enzymatic activity after 1 h of PEF treatment.

Interestingly, although cell permeabilization occurs at electrical fields ranging from 3 to 5 kV·cm⁻¹, data show that cells recover from the electrical stress and are able to keep their enzymatic activity at 523 least for 1 h after the treatment. 524

Further studies will be undertaken to confirm these results and optimize the process. 526

3.6. Lipid extraction from permeabilized cells, preliminary results 527

Previous works already demonstrated enhancement of lipid extrac- 528 tion thanks to electroporation ([15,27]), but with other algae strains 529 (respectively *Ankistrodesmus falcatus* and *C. vulgaris*), and with other 530 electrical conditions (respectively $E = 45 \text{ kV} \cdot \text{cm}^{-1}$ for $\Delta t_{pu} = 360 \text{ ns}$ 531 and $E = 2.7 \text{ kV} \cdot \text{cm}^{-1}$ for $\Delta t_{pu} = 100 \text{ µs}$). 532

We performed preliminary experiments to evaluate the potential of 533 PEF treatment to improve lipid extraction by solvents, under our condi-534 tions. *C. reinhardtii* cells harvested after 20 days of nitrogen stress (cells 535 full of lipid droplets) were electroporated (5 μ s pulse duration, 536 7 kV·cm⁻¹). Bodipy fluorescence of the non-treated cells, PEF-treated 537 cells, solvent-treated cells and of cells treated by both PEF and solvent 538 were measured and compared. Bodipy fluorescence was reduced by a 539 factor of 1.7 after 20 min of solvent exposure, and by a 4.5 factor after 540 treatment by both PEF and solvent exposure. We interpret this higher 541 florescence reduction as an effect of PEF treatment towards the facilita-542 tion of lipid extraction. This preliminary experiment shows that the 543 electroporation treatment may ease the lipid extraction performed by 544 solvent. Further studies will be undertaken to confirm these results 545 and optimize the process. 546

4. Conclusion

The effect of PEF treatment on microalgae cells was observed and 548 quantified in real-time using a specifically designed microdevice. With 549 this dedicated microdevice, it was possible to (1) estimate the permeabilization threshold for various treatment conditions (electric field amplitude, pulses duration, and number), (2) evaluate the sensitivity of 552 microalgae cells from different culture conditions to PEF and (3) analyze the PEF treatment at the microalgae cell scale, synchronized with the PEF application. 555

C. reinhardtii, the chosen microalgae strain, was attractive for this 556 study because, as confirmed by our results, this strain is particularly sensitive to PEF (Schwan equation [34]), thanks to the large diameter of the 558 cell (approximately 7–10 μ m) compared with many other strains used 559 for lipid production and extraction. 560

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561The energy of the PEF technology cost was estimated; the use of 562short pulses with high field intensity seems to be more favorable. This choice also results in lower heat effects. 563

564Visualization of lipid droplets during the treatment showed their clear displacement in the cytoplasm, and droplets merging in some 565cases. To the best of our knowledge, this is the first demonstration of 566 lipid droplets behavior under an electrical field. Nevertheless, there 567was no lipid leakage out of the cell and no cell lysis, even though cells 568 569were permeabilized as confirmed by PI or Sytox Green uptake.

570These results suggest that electroporation can be used as a pre-571treatment of microalgae in order to improve extractions by weakening 572the cell membrane. Extraction remains challenging for large molecules 573extractions such as triacylglycerols. However, mild electroporation, as 574a pre-treatment, might enhance the efficiency of solvent extractions since membrane permeabilization would improve the diffusion of a sol-575vent across the membrane. 576

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Appendix A. Supplementary data 582

Supplementary data to this article can be found online at http://dx. 583 doi.org/10.1016/j.algal.2016.03.023. 584

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