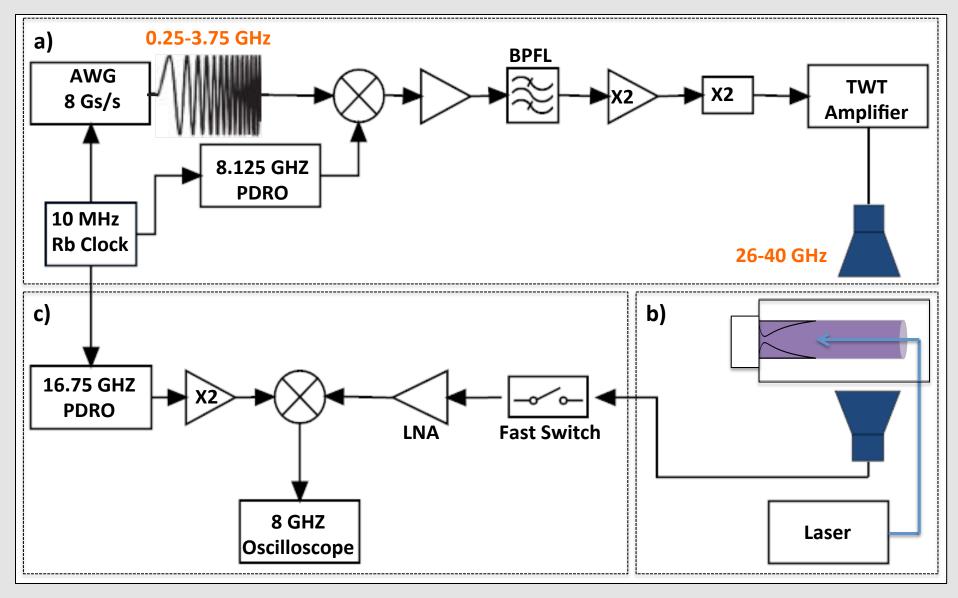
# Quantitative product branching for multichannel reactions using chirped pulse spectroscopy in pulsed uniform flow

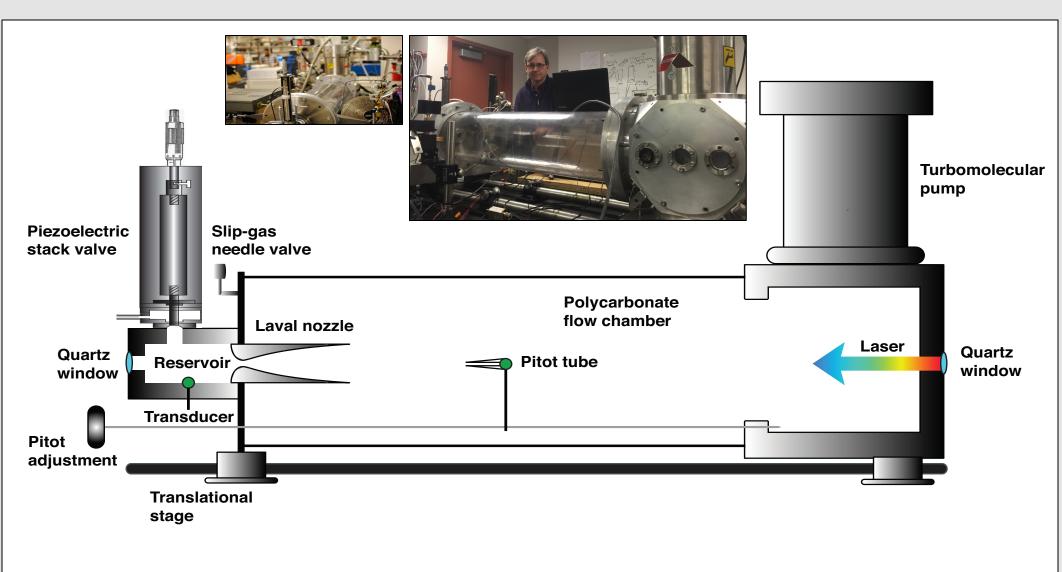


# **CPUF** Spectrometer

A new experimental apparatus is presented that combines two powerful technologies: chirped-pulse Fourier-transform microwave spectroscopy (CP-FTMW) and pulsed uniform supersonic flows to address the need of a detection technique that can give isomer-specific identification and structural characterization of reaction products and intermediates and their branching ratios in the field of reaction dynamics. CP-FTMW spectroscopy is a revolutionary new technique capable of producing broadband rotational spectrum in a single data acquisition. However, a high density (10<sup>16</sup> cm<sup>-3</sup>) of cold products (20-30 K) is needed for this approach to be effective. Uniform supersonic flows are an ideal environment for providing these conditions, as well as allowing for stabilization of reaction intermediates and determination of reaction kinetics with product detection. Presented here is a description of the new apparatus, chirped pulse in pulsed uniform supersonic flow (CPUF) spectrometer and its capabilities.



Schematic of the CPUF spectrometer. A chirped pulse is generated in an arbitrary waveform generator (AWG), upconverted, and amplified. The molecular sample is polarized by the radiation and the resulting emission, in the form of a free induction decay (FID), is collected by the oscilloscope. The FID is Fourier transformed to produce a spectrum. The spectrometer operates in the 26-40 and 60-90 GHz frequency regions.



A piezoelectric stack<sup>1</sup> valve is mounted on a small reservoir outside of the polycarbonate vacuum chamber. The Laval nozzle provides a uniform flow with the pressure monitored by a pair of transducers. The entire chamber is mounted on a translational stage, so different regions of the flow can be probed by the microwave radiation introduced via one of the feedhorns.

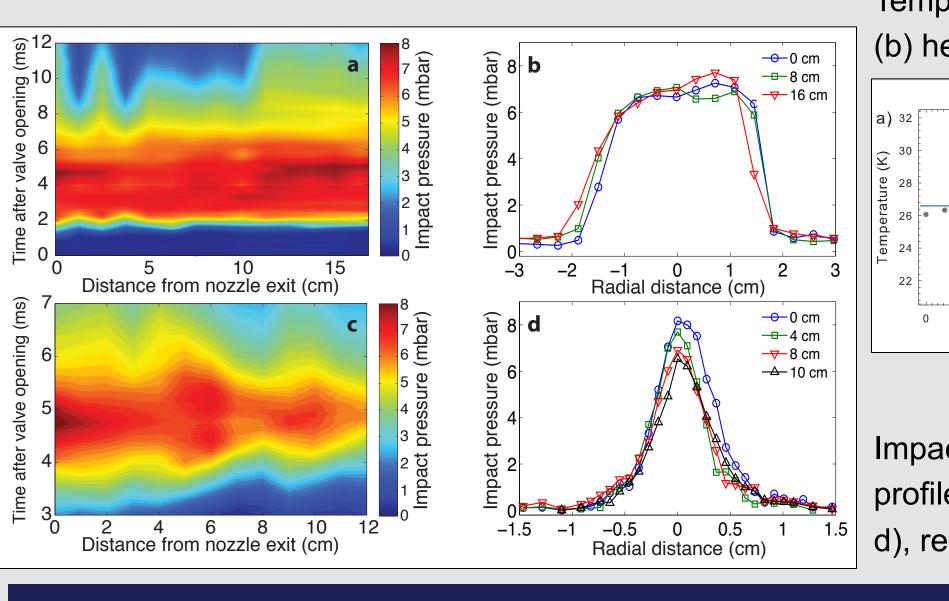
> adapter (Al half-cylinder (SS

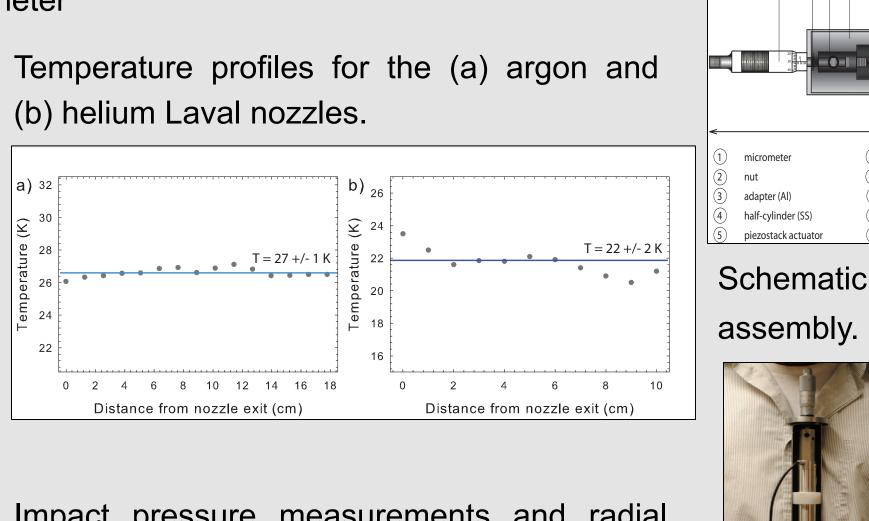
assembly.

valve body (SS

stagnation chamber

Schematic view of the valve





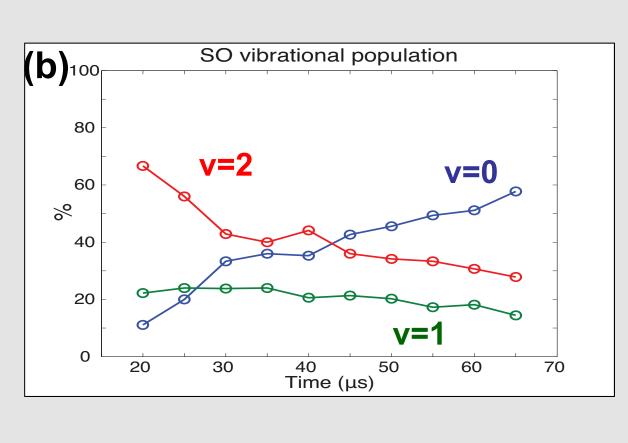
Impact pressure measurements and radial profiles for argon (a and b) and helium (c and d), respectively.

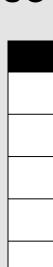
# **CPUF** Capabilities Photochemistry

SO  $(X^{3}\Sigma^{-})$ : N<sub>1</sub> = 1<sub>0</sub> - 0<sub>1</sub> v = 2 80 100 100 35 μs 300 350 175 45 μs 200 500 250 300 350 65 μs Λ 29.90 29.95 30.00 Frequency (GHz)

 $SO_2 + hv (193 \text{ nm}) \rightarrow SO (v, J) + O (^3P_J)$ 

Time evolution of SO spectra generated from the photodissociation of SO<sub>2</sub> with 193 nm radiation (a). SO begins to appear roughly 20 µs after the laser is fired, mostly in the v = 2 vibrational level, corresponding to the nascent vibrational populations determined from imaging studies. However, quick relaxation to the v = 0 state occurs (b)





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Nascent vibrational distributions for SO ( $X^{3}\Sigma^{-}$ ). <sup>3</sup>

V	Pop.
0	0.09
1	0.23
2	0.56
3	0.06
4	0.06

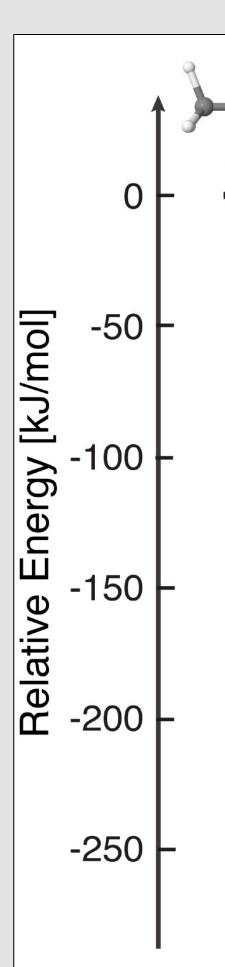
			Bimo					
	30	t = 10 μs	••					
	0 30	En 20	10 40 70 100					
		$t = 20 \ \mu s$	time (µs)					
	0	En Marine Ma	www.www.www.www.					
	30	t = 40 μs						
(۲۷)	0	E-many Markan Marka	Mundman with when the					
-> 	50	t = 50 μs						
าเตาอาเ	0 50	Engeneration and the second se						
		$t = 60 \ \mu s$	ma manne man Morrow Mar					
	0 50	t = 70 μs						
	0 75 0 150		when he we have a second					
		t = 80 μs						
		t = 90 μs						
	0							
	150	t = 100 μs						
	0 75	t = 110 μs						
	0	36.320 36.392	2					
		Frequency	(GHz)					

### Motivation

- been seen previously.

### **Results & Discussion**

- channels.



chamber cap (SS)

front plate (SS)

viton/kalrez o-ring

ferrule (Al) nozzle (SS 1mm)

## **lecular Reactions**

Spectra illustrating the time evolution of the J = 4 - 3rotational transition of HCCCN ( $X^{1}\Sigma^{+}$ ), generated via the bimolecular reaction of the CN radical ( $X^2\Sigma^+$ ) and  $C_2H_2$  $(X^{1}\Sigma_{a}^{+})$ , are displayed. The CN radical was produced by irradiation of BrCN with a 193 nm laser. HCCCN appears approximately 40–50 µs after the laser is fired. The inset shows the integrated intensity of the HCCCN line at each time point, with the peak intensity occurring at ~90  $\mu$ s. Each spectrum is an average of 22000 acquisitions.

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# **Product branching in the low temperature reaction of CN with Propyne**

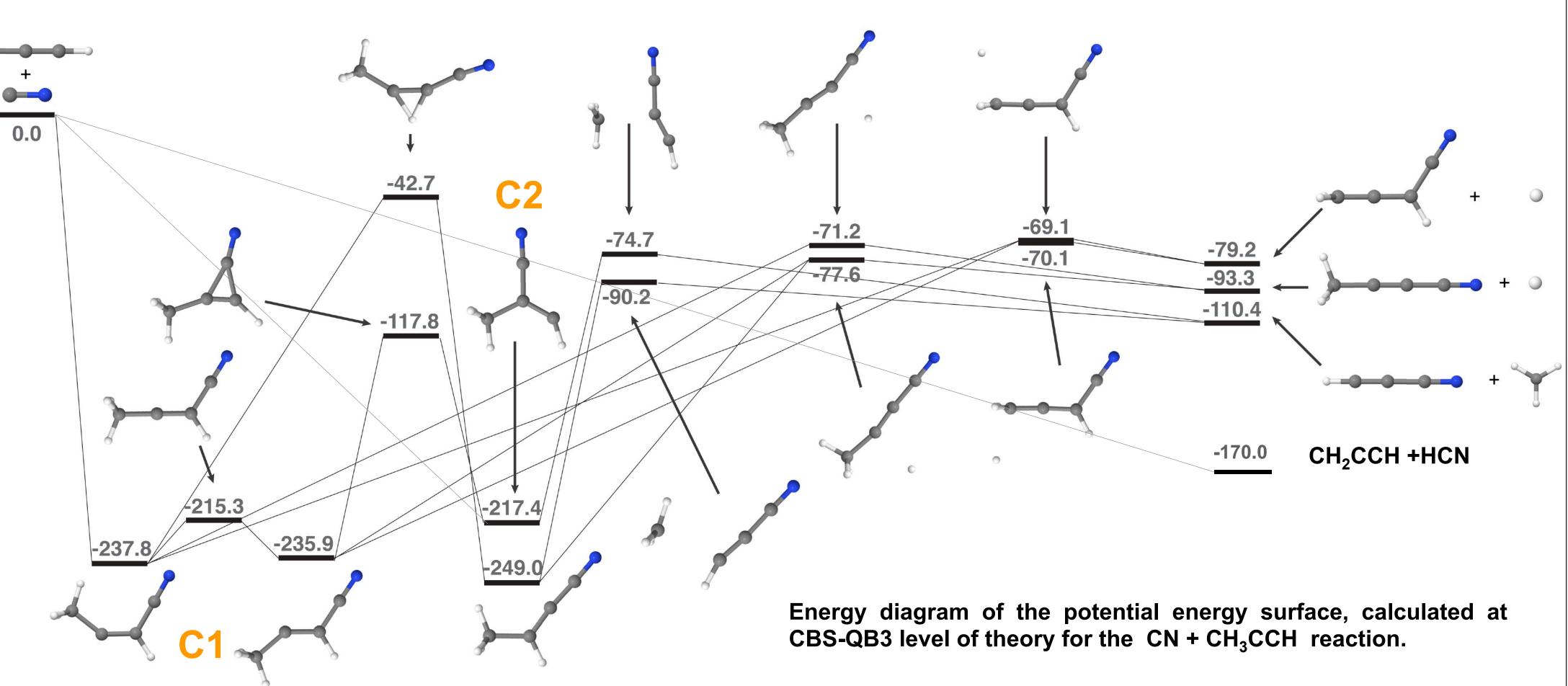
• The reaction between CN radical and methylacetylene (CH<sub>3</sub>CCH) is important in the growth of nitrile species in Titan's atmosphere and a prototypical system in the which to probe multichannel branching and the competition between direct and complex-mediated reactions.

Previous crossed-molecular beam experiments,<sup>4</sup> supported by ab initio calculations, reported the CN addition/H elimination channel and identified two product isomers.

Even though CH<sub>3</sub> elimination product HCCCN through C2 is more energetically stable, no clear evidence of this channel was indicated in previous work, nor has the direct abstraction channel

Both the direct abstraction and the indirect addition/elimination products have rotational transitions accessible within the detection frequency band of CPUF, thus a detailed study of the bimolecular reaction with supporting ab initio calculation was carried out.

Detection of all the possible products in a broad spectral window allows us to quantitatively determine the product branching ratios for complex bimolecular reactions with multiple product



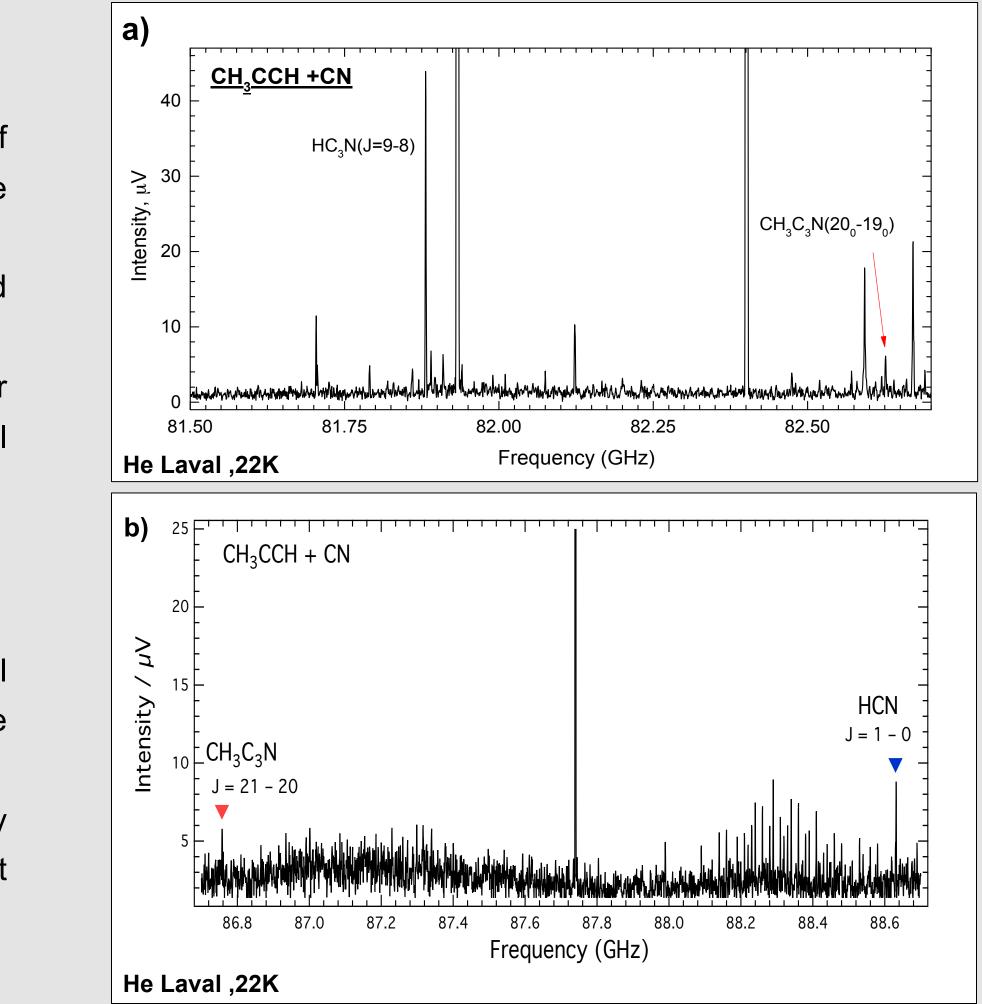
### $CN + HCCH \rightarrow HCCCN$

### Acknowledgements

		Addition/elimination			Direct Abstraction
%		HC <sub>3</sub> N	CH3C <sub>3</sub> N	CH <sub>2</sub> CCHCN	HCN
RRKM	C1 complex	60	33	7	
KKKIVI	C2 complex	71	24	5	
CPUF		72(4)	22(6)	<5	3.5(5)

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Calculated and experimental product branching for the CN with  $CH_3CCH$  reaction,  $2\sigma$  uncertainty to the last digit

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