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# Photonic glass-ceramics: consolidated outcomes and prospects

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## ABSTRACT

Transparent glass-ceramics are nanocomposite materials which offer specific characteristics of capital importance in photonics. This kind of two-phase materials is constituted by nanocrystals embedded in a glass matrix and the respective composition and volume fractions of crystalline and amorphous phase determine the properties of the glass-ceramic. Among these properties transparency is crucial, in particular when confined structures, such as dielectric optical waveguides and optical fibers, are considered, and the number of papers devoted to this topic is continuously increasing. Another important point is the role of the nanocrystals when activated by luminescent species, as rare earth ions, and their effect on the spectroscopic properties of the glass-ceramic. The presence of the crystalline environment around the rare earth ion allows high absorption and emission cross sections, reduction of the non-radiative relaxation thanks to the lower phonon cut-off energy, and tailoring of the ion-ion interaction by the control of the rare earth ion partition. This last point is crucial and still object of intense experimental and theoretical studies. The composition of the glass matrix also impacts the properties of the rare earth ions located in nanoparticles. Moreover, some kinds of nanocrystals can play as effective rare earth sensitizers. Fabrication, assessment and application of glass-ceramic photonic systems, especially waveguides, deserve an appropriate discussion which is the aim of this paper, focused on luminescent glass-ceramics. In this work, a brief historical review, consolidated results and recent advances in this important scientific and technological area will be presented, and some perspectives will be outlined.

**Keywords:** Transparent glass ceramics; Bottom-up and top-down fabrication techniques; Confined structures; RF-Sputtering; Sol-gel; Oxide nanocrystals; Rare-earth ions; Photoluminescence

## INTRODUCTION

Glass photonics is pervasive in a huge number of human activities and drive the research in the field of enabling technologies. The fruitful exploitation of glass photonics is not restricted only to the area of Information and Communication Technology. Many other photonic devices, with a large spectrum of applications covering Health and Biology, Structural Engineering, and Environment Monitoring Systems, have been developed during the last years. Glass materials and photonic structures are the cornerstones of scientific and technological building in integrated optics. Photonic glasses, optical glass waveguides, planar light integrated circuits, waveguide gratings and arrays, functionalized waveguides, photonic crystal heterostructures, and hybrid microresonators are some examples of glass-based integrated optical devices that play a significant role in optical communication, sensing, biophotonics, processing, and computing. This cross-disciplinary approach leads to constructed luminescent structures that can perform sensing and functionalized structures to successfully address socioeconomic challenges, such as security, cost-effective healthcare, energy savings, efficient and clean industrial production, environmental protection, and fast and efficient communications. Photonics, with its pervasiveness, has already been identified as an enabling technology, and through advanced research in glass-based integrated optics systems, photonics can contribute to finding new technical solutions to still unsolved problems, and pave the way to applications not yet imagined.

Among the different glass-based systems, transparent glass-ceramics are nanocomposite materials which offer specific characteristics of capital importance in photonics. These two-phase materials are constituted by nanocrystals embedded in a glass matrix and the respective composition and volume fractions of crystalline and amorphous phase determine the properties of the glass-ceramic. Transparency is crucial property, in particular for dielectric optical waveguides and optical fibers. Another important point is the effect of the nanocrystals activated by rare earth ions on the spectroscopic properties. Fabrication, assessment and application of glass-ceramic photonic systems, deserve an appropriate discussion which is the aim of this paper, focused on luminescent glass-ceramics.

### A SHORT HISTORY OF GLASS-CERAMICS

The scientific and technological activity involving the development of materials at nano-micro scale and related converging technologies allows progress in the conception, design, and realisation of systems and devices with substantially improved performance and significant scientific results. This activity is coming from far both in terms of time and motivation. Table I presents some milestones concerning a brief history of glass-ceramic materials. René-Antoine Ferchault de Réaumur, the brilliant physicist and naturalist, was also in charge of the manufacture of porcelain. He tried to produce a porcelain which could compete with the porcelain of China, without being able to fully solve the problem but creating that opaque glass that bears the name of "porcelain Reaumur" (1739) [1]. We have to wait until the year 1952 to appreciate the invention of glass-ceramics thanks to the experiment performed by the Corning Glass Works scientist Stanley Donald Stookey. In the description of the experiment we read that the furnace containing the glass accidentally overheated. Instead a melted glass Stookey observed a white material with the shape of the original glass. He recognized the obtained material as a specific new material that we know as "glass-ceramic" and was initially described as a polycrystalline ceramic materials, formed through the controlled nucleation and crystallization of glass, where the amount of residual glassy phase is usually less than 50% [2]. The commercial outcome of this material was the immediate exploitation of its unique and amazing thermal and mechanical properties leading to the cookware pieces still now known as Pyroflam and Vision products. These strong and thermal shock resistant glass-ceramics have monopolized the research for a long time until the interest in transparent glass-ceramics has made its way thanks to both fundamental research and the required novel application in optics. Is only in 1968 that Schott AG produced a transparent glass-ceramic called Zerodur employed for a number of very large telescope mirrors and also used as sample reference to test the validity of innovative glass structural models [3]. The first application of transparent glass ceramics in the field of photonics appears in 1995 with the paper of Tick et al. concerning the exploitation of transparent glass ceramics for optical amplifiers operating at 1300 nm [4]. In the abstract of the paper authors say: "The properties of an oxyfluoride glass ceramic that possesses high transparency after ceramming are described. Approximately 25 vol% of this material is comprised of cubic, fluoride nanocrystals and the remainder is a predominantly oxide glass. When doped with Pr<sup>3+</sup>, the fluorescence lifetime at 1300 nm is longer than in a fluorozirconate glass, suggesting that a significant fraction of the rare-earth dopant is preferentially partitioned into the fluoride crystal phase. This material has the added advantage of being compatible with ambient air processing". It's worthy to note that this paper contains the more significant topics that will be developed in the next 20 years in the field of photonic glass ceramics, i.e. (i) the fundamental role of oxyfluoride as parent glass; (ii) the role of the nanocrystals in reducing the concentration quenching; (iii) the role of

nanocrystals with low cut-off energy in reducing non-radiative transitions [4]. After this work we can find in literature a huge amount of papers regarding rare earth activated oxyfluoride based transparent glass ceramics [5-8].

Table 1. A brief history of glass-ceramics

Milestone	Date
René-Antoine Ferchault de Réaumur produced polycrystalline materials from silica-soda-lime	1739
Stanley Donald Stookey - Corning's scientists - discovered glass-ceramics	1952
VISIONS and PYROFLAM Registered Corning Glass Works Trademark	1958
ZERODUR by Schott	1968
Transparent glass ceramics for 1300 nm amplifier applications	1995
The relationship between structure and transparency in glass-ceramic materials	2000

However, the main result in photonic glass ceramics has been obtained in waveguiding application. A rare earth-activated glass-ceramic waveguide constitutes a potential significant system to behave as an effective optical medium for light propagation and luminescence enhancement [9]. Looking at this specific application the problem of the transparency becomes crucial and in fact in 1998 Tick published a paper with the explicit title “Are low-loss glass-ceramic optical waveguides possible?” [10]. The answer was positive and he concluded the paper suggesting that the minimum transmission loss limit of the investigated effective medium glass ceramics is in the order of tens of decibels per kilometre (dB/km), once all of the impurities are eliminated. In the same paper some general criteria for light propagation are given. These criteria concern nanocrystals size, narrow particle-size distribution, inter-particle spacing, and clustering. In 2000 Tick, Borrelli and Reaney published another significant paper discussing the relationship between structure and transparency in glass-ceramic materials into a near single-mode optical waveguide fiber geometry [11]. In this significant paper authors investigate the transparency analysing the composition dependence of the lattice parameters and refractive index of the nanocrystals, their size distribution, the particle separation and the actual crystalline volume fraction [11]. A detailed scattering analysis was performed using two different approaches. One is as a contiguous two-phase system with the scattering characterized by a correlation function. The other is a particle approach, where one uses an effective medium scattering model. The authors conclude that although the latter predicts the correct order of magnitude of the scattering, it cannot distinguish the scattering behaviour as a function of ceramming conditions used. This point is still of crucial interest and several efforts have been done to achieve general rules to correlate scattering attenuation with appropriate ceramming protocols. From the point of view of simulation many techniques have been developed for analyzing scattering problems and several interesting solutions have been discussed [12]. Moving to fabrication, although a general behaviour is not yet determined excellent results have been obtained for specific compositions [13-18] and we will present some examples in this paper.

## GEO<sub>2</sub> TRANSPARENT GLASS CERAMIC PLANAR WAVEGUIDES

Recently effort is devoted on growth, characterization on GeO<sub>2</sub> based system for his employment for applications on various modern fields such as electronics and photonics. GeO<sub>2</sub> exhibits, in fact, many interesting properties that make this material suitable for applications in optical, electronic and optoelectronic devices [19, 20, 22]. It is of strategic importance to develop fabrication protocols allowing obtain GeO<sub>2</sub> transparent glass ceramic planar waveguides exhibiting low attenuation coefficients and simultaneously embed GeO<sub>2</sub> nanocrystals with a specific phase [22]. Among the various techniques can be used to fabricate these particular kinds of nanostructured systems such as sol-gel techniques with top-down and bottom-up approaches, and physical vapor deposition, we have demonstrate that RF-sputtering (RFS) is a suitable route to fabricate optical planar waveguides and photonic microcavities operating in the visible and NIR regions [20, 23]. In order to produce transparent glass-ceramic, a common procedure employ heat treatments of the systems, but lately efforts was performed for the development of alternative treatments such as laser annealing (LA) that presents advantages in terms of temporal and spatial annealing control in respect to the conventional thermal annealing [20, 24]. The CO<sub>2</sub> laser annealing has been successfully used to reduce scattering losses in different kinds of amorphous optical planar waveguides [20], and for the fabrication of glass-ceramic coating and glass-ceramic waveguides with low attenuation coefficient [20]. Anyway it is of crucial importance a precise characterization of the structure of the samples to assess the effect of the LA process and tailor the irradiation protocols. Moreover, it is well known that important structural modification, leading to artificial results, can be induced in GeO<sub>2</sub> based materials by electron irradiation [20]. For these reasons, information on the structural properties of the waveguides before and after

the laser annealing process are obtained by various techniques and in particular from positron annihilation spectroscopy (PAS), specifically by the Doppler broadening spectroscopy (DBS). PAS is a well-known non-destructive spectroscopy technique employed to investigate materials structure [25]. Experimental details could be found in [19].

In Table 2 are reported the optical parameters for the samples as prepared and after CO<sub>2</sub> laser irradiation for 2h. The waveguides have a thickness of about 1μm and support one mode at 1319 nm and 1542 nm. Before and after LA the refractive indices measured in TE and TM polarization modes are equal within the experimental uncertainty, so the birefringence can be considered negligible. Comparing the refractive indices in the GeO<sub>2</sub> waveguides before and after LA we observe an increase of about 0.04 with the irradiation at all the wavelengths.

Table 2. Optical parameters for the samples as prepared and after CO<sub>2</sub> laser irradiation for 2h.

Laser wavelength (nm)		Refractive Index		thickness (μm)	Attenuation coefficient (dB/cm)
		TE	TM		
632	before irradiation	1.614±0.001	1.616±0.001	1.1±0.1	1.9±0.2
	after irradiation	1.652±0.001	1.653±0.001	1.1±0.1	1.1±0.2
1319	before irradiation	1.590±0.01	1.590±0.01	1.0±0.1	1.4±0.2
	after irradiation	1.631±0.01	1.634±0.01	1.0±0.1	0.7±0.2
1542	before irradiation	1.585±0.01	1.585±0.01	1.0±0.1	0.9±0.2
	after irradiation	1.623±0.01	1.624±0.01	1.0±0.1	0.5±0.2

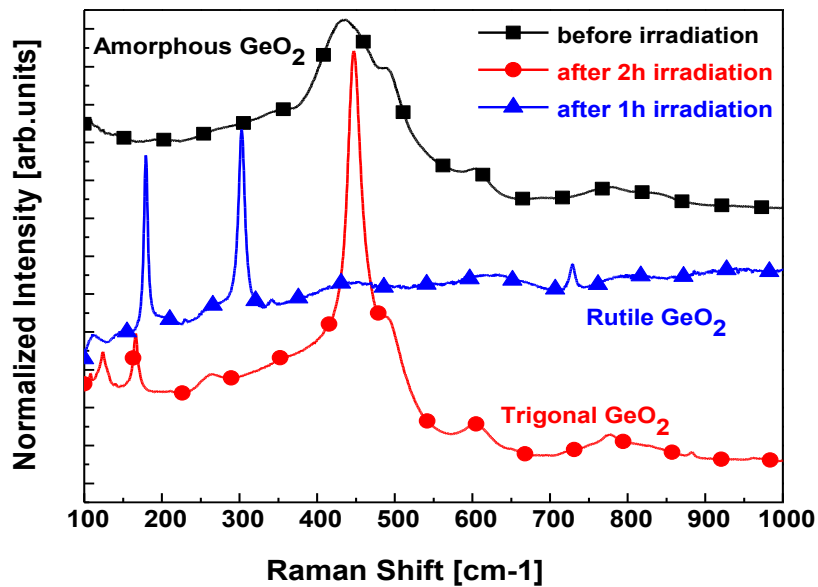
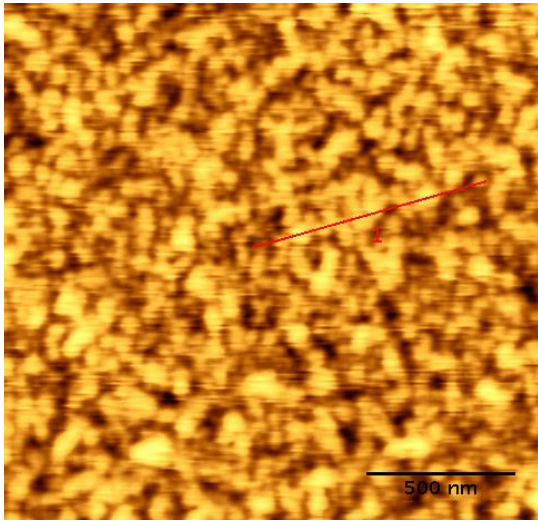
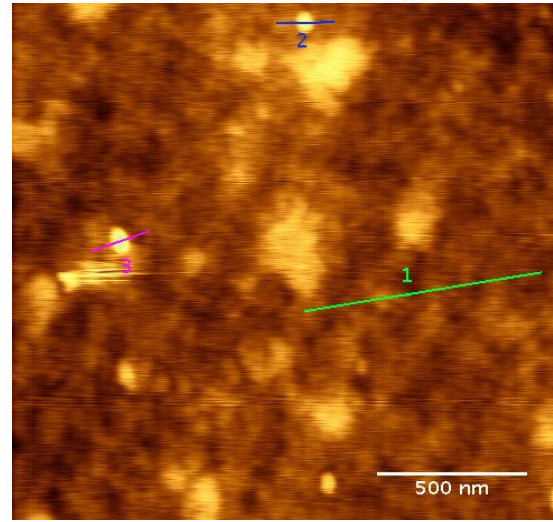


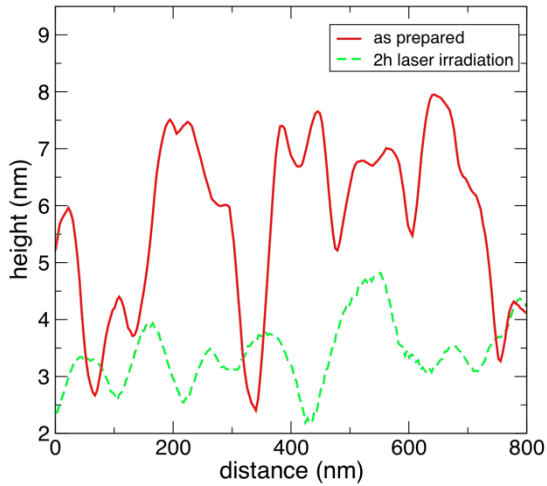
Figure 1 Micro Raman measurements carried out at room temperature for GeO<sub>2</sub> planar waveguide.



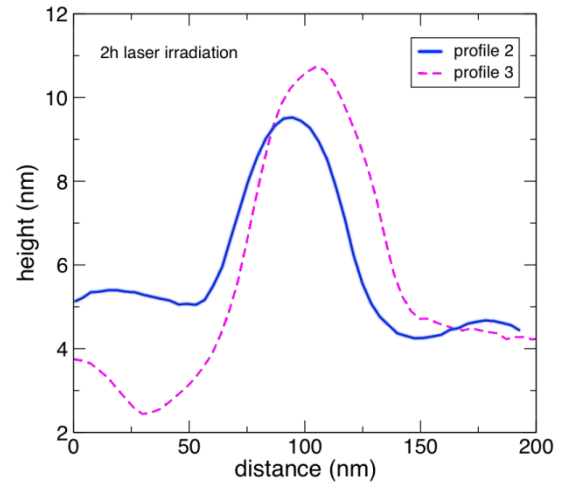
a)



b)



c)



d)

Figure 2. AFM images of representative  $1.8 \times 1.8 \mu\text{m}^2$  areas of the samples in the conditions: (a) as prepared and (b) after 2h of  $\text{CO}_2$  laser irradiation. Z scale 10 nm. In (c) a comparison of typical height profiles in these two conditions, corresponding to sections 1 in panels (a) and (b), showing a decrease in roughness after irradiation is presented. (d) profiles of the nanometric structures found after irradiation - sections 2 and 3 in panel (b).

The laser annealing allowed significant reduction of the attenuation coefficients. In fact, we observe an attenuation coefficient at 1542 nm of 0.5 dB/cm for the irradiated system while for the system before LA is 0.9 dB/cm. As shown in Table 2, the reduction of the attenuation coefficient with the  $\text{CO}_2$  laser irradiation is present at all the considered wavelengths. The decrease in the total attenuation coefficient for the  $\text{CO}_2$  laser irradiated systems, that take in account of all the contribution to the losses such as volume and surface scattering and absorption, has been attributed by Dutta et al. [24] to the reduction of the surface roughness

The optical parameters measured after 1h of LA are equivalent to those reported for 2h. EDXS analysis was used to monitor the chemical composition of the sample, as deposited and after LA. EDXS confirm that the composition of the sample is not affected by the laser irradiation and the correct stoichiometry between germanium and oxygen is always present.

In figure 1, the Micro Raman spectra measured at room temperature for GeO<sub>2</sub> planar waveguides before and after LA at different times are reported. The Raman spectrum of the as prepared GeO<sub>2</sub> waveguide before irradiation is typical of a GeO<sub>2</sub> amorphous system. The amorphous nature of the GeO<sub>2</sub> film is clearly pointed out by the broad peak at 440 cm<sup>-1</sup> [20]. After 1 h of CO<sub>2</sub> laser irradiation, the Raman spectra indicate that in the amorphous GeO<sub>2</sub> matrix there is the presence of a rutile-like GeO<sub>2</sub> crystalline phase [20]. After 2 h of CO<sub>2</sub> laser irradiation, the Raman spectrum shows that in the amorphous system there is also the presence of a trigonal GeO<sub>2</sub> crystal phase [20].

AFM images of representative 1.8 x 1.8 μm<sup>2</sup> areas for the as prepared sample and for that CO<sub>2</sub> laser irradiated for 2h are reported in figure 2 (a) and 2 (b) respectively. AFM analysis put in evidence a roughness of 1.05 ± 0.05 nm in the as prepared sample and a roughness of 0.80 ± 0.02 nm for the waveguide after 2h of CO<sub>2</sub> laser irradiation. This result correlates with the reduction of the attenuation coefficient with the laser annealing and confirms surface morphology as an important source of losses. Moreover, on the surface of the sample, the AFM image of figure 2 (b) shows the presence of nanometric structures with sizes ranging from 10 to 50 nm (see also figure 2 (c)). These structures are assigned to GeO<sub>2</sub> nanocrystals. It is worth noting that, although the presence of these scattering points are expected to increase the attenuation coefficient by increasing the scattering losses of the waveguide, the protocol here developed for the CO<sub>2</sub> irradiation allows reducing the attenuation coefficient by decreasing the surface scattering losses [24].

Table 2.  $S_n$  values characterizing each layer in the as prepared and the two irradiated samples, as obtained by fitting the positron depth profiles. The density value of the GeO<sub>2</sub> film was used as a guess parameter into the frame of the fitting procedure. The boundary depths are measured from the surface of the sample.

Sample	layer I $\rho = 3.15 \text{ g/cm}^3$		layer II $\rho = 3.15 \text{ g/cm}^3$		layer III $\rho = 3.15 \text{ g/cm}^3$		bulk $\rho = 2.1 \text{ g/cm}^3$
	$S_n$	depth (nm)	$S_n$	depth (nm)	$S_n$	depth (nm)	$S_n$
<b>before irradiation</b>	1.006	11 ± 2	0.952	983 ± 20	--	--	1.000
<b>after 1h irradiation</b>	1.003	2 ± 0.2	0.946	320 ± 25	0.966	983	1.000
<b>after 2h irradiation</b>	0.980	5 ± 0.1	0.950	175 ± 32	0.960	983	1.000

In the as prepared film, positrons detect a first thin superficial layer of about 10 nm which is characterized by a very high  $S_n$  value of 1.006. This thin layer is followed by another GeO<sub>2</sub> layer (labelled layer II) uniform up to the silica substrate and characterized by a  $S_n = 0.952$ . With the laser irradiation, the superficial layer reduces to a half of its initial thickness, and the uniform second layer observed in the as prepared film splits now into two regions, named layers II and III, respectively. The first region is structurally similar to the second layer of the non-irradiated sample having about the same  $S_n$  value (~ 0.950). In the sample irradiated during 1 h the thickness of the second layer is around 300 nm; and in the case of the sample irradiated 2 h, this thickness decreases up to approximately 170 nm. On the other hand, the second region (i.e., layer III) is characterized by a higher  $S_n$  value (~ 0.960) than the previous one and it extends up to the silica interface. The detection by DBS of a thin superficial layer that decreases its thickness with the laser irradiation treatment, well-correlates with the results obtained by AFM. Probably, this layer is highly defected due to their  $S_n$  values by which it is characterized (higher than the  $S_n$  values found in the bulk of the films). Both, the decrease in roughness and thickness of the defected superficial layer could be linked to the decrement of the attenuation coefficient. From the results reported in Table 2, it can be seen that the  $S_n$  values obtained for layer III are slightly higher than those of layer II, indicating a structural change of the GeO<sub>2</sub> samples. Besides, the thickness of this modified layer is larger in the sample annealed for a longer irradiation time. In the case of layer II, for the laser treated films  $S_n$  values are almost equal to that

not treated sample. Moreover we can note that: a) the bulk of the film interested by the structural change increases by increasing the irradiation time from 1 to 2 h; b) the change starts from the film/substrate interface moving towards the surface of the films. Raman spectroscopy reveals the presence of different GeO<sub>2</sub> phases. The rutile GeO<sub>2</sub> phase present after 1 h of irradiation exhibits a higher refractive index than that of the trigonal phase observed after 2 h of irradiation [20]. However, DBS results would allow inferring that the lack of substantial changes in the macroscopic refractive index measured after 1 and 2 h of irradiation, respectively, could be attributed to a balance between the contributions of the different phases to the modified matrixes. The behavior described in point b) could be assigned to the heating of the substrate by the CO<sub>2</sub> laser inducing a progressive modification of the film from the substrate to the surface. The SiO<sub>2</sub> matrix, in fact, due to the lattice vibration in the region of 940 cm<sup>-1</sup>, presents a higher absorbance in the region of the CO<sub>2</sub> laser line with respect to that of the GeO<sub>2</sub> system [20]. For that reason the CO<sub>2</sub> radiation is mainly absorbed by the SiO<sub>2</sub> substrate that induces the change in the GeO<sub>2</sub> film.

### Er<sup>3+</sup>-ACTIVATED SnO<sub>2</sub> SOL-GEL-DERIVED GLASS-CERAMICS: AN ENERGY TRANSFER CASE STUDY

During the last years several work has been performed in order to develop energy transfer based systems for photoluminescence application. In this scientific area we have up- and down-converters for planar waveguides for integrated optics [17], visible laser light sources and solar cells efficiency enhancement [18,26], specific nanostructured system used as sensitizers of rare earth luminescence [27]. Recently, a detailed work regarding the effect of the local environment on the spectroscopic properties of rare-earth-activated tin dioxide glass ceramic was presented [28, 29].

Tin dioxide, which is a high-refractive index semiconductor ( $E_g = 3.6$  eV at 300K), and is stable both chemically and mechanically, is an excellent choice for the crystalline component of the system. As it is also transparent through the visible and infrared regions, it covers the emission range for active ions like erbium. Compatibility is enhanced by the fact that tin oxide nano-crystals can be excited by a broad range of UV wavelengths, thus with the use of broad band arc lamps, while Er<sup>3+</sup> ions have very narrow excitation peaks.

Studies on silica-tin bulk glass ceramics have shown that the concentration of SnO<sub>2</sub> has a definite effect on the structure

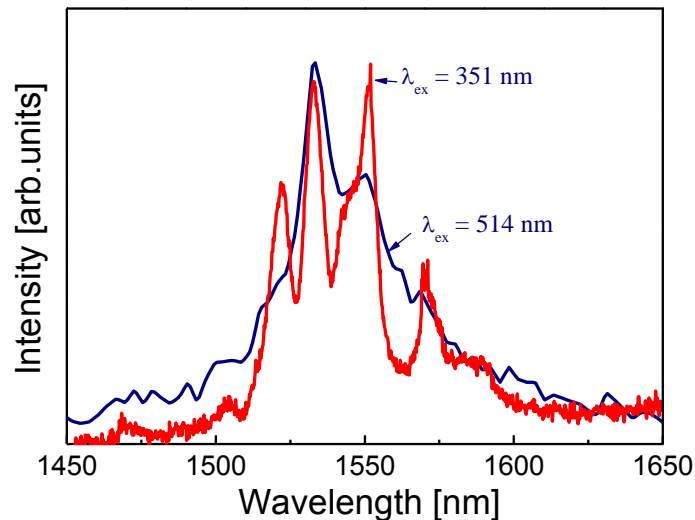


Figure 3: Photoluminescence spectra of a (100-30)SiO<sub>2</sub> - 30SnO<sub>2</sub> waveguide, doped with 0.5 mol% Er<sup>3+</sup> and heat-treated at 1200°C, upon excitations at 351 and 514.5 nm

of the SiO<sub>2</sub> glass phase [30,31]. In addition, while rare earth ion concentrations as high as 1 mol% can be doped into the glass ceramic, Er<sup>3+</sup> ions are found mainly in the glass phase, which results in high levels of quenching effects. In order to investigate any possible energy transfer between erbium ions and SnO<sub>2</sub> nanocrystals, the environment of the Er<sup>3+</sup> ions was studied via infrared photoluminescence measurements which were recorded using 351 nm and 514.5 nm excitation



wavelengths. These two lines correspond to the excitation of the SnO<sub>2</sub> crystals and to the excitation of the  $^4I_{15/2} \rightarrow ^2H_{11/2}$  transition of Er<sup>3+</sup> ions across their bandgap, respectively.

Figure 3 shows the photoluminescence spectra of the  $^4I_{13/2} \rightarrow ^4I_{15/2}$  transition of Er<sup>3+</sup> ion for the samples containing 30 mol% SnO<sub>2</sub>, doped with 0.5 mol% Er<sup>3+</sup> and heat-treated at 1200°C. Under excitation at 514.5 nm, one obtains an emission spectrum characteristic of Er<sup>3+</sup> ions in an amorphous medium, with a broad band centered around 1535 nm. However, excitation at 351 nm results in a completely different spectrum, in which the presence of narrow bands at 1521, 1531, 1549 and 1571 nm are assigned to the Stark transitions between  $^4I_{13/2}$  and  $^4I_{15/2}$  levels. When the erbium ion occupies the Sn<sup>4+</sup> sites in the cassiterite structure, the electrostatic field of the surrounding environment of ions removes degeneracy of the free ion J levels and makes them split into doubly degenerated Stark level [28,29]. This emission results from energy transfer between SnO<sub>2</sub> nanoparticles and the rare-earth ions by Forster's mechanism. Accordingly, this spectrum indicates that most part of Er<sup>3+</sup> ions are embedded in SnO<sub>2</sub> nanocrystals. Consideration of these spectral features indicates that the Er<sup>3+</sup> ions exist in two types of sites: those embedded in tin oxide nanocrystals and those within the amorphous silica matrix. The influence of the incorporation of erbium in the tin dioxide nanoparticles on luminescence were studied by the photoluminescence measurements of films with different Er<sup>3+</sup> concentrations. With increasing erbium concentration, the near-infrared emission intensity at 1540 nm is steadily enhanced. The PL intensity of the 2 mol% Er-doped sample is more than three orders of magnitude greater than that of the 0.5 mol% Er-doped one. This behavior indicates that quenching of luminescence did not occur in the 2 mol% erbium doped systems.

## CONCLUSIONS

Transparent glass ceramic photonic systems have demonstrated to be crucial for several applications and especially in integrated optics. The importance of these nanocomposite materials is evident when luminescence efficiency and attenuation losses are the more important requirements. Concerning the role of nanocrystals as rare earth sensitizers, SnO<sub>2</sub> nanocrystals constitutes a textbook example. SnO<sub>2</sub> is a wide band-gap ( $E_g = 3.6$  eV at 300 K) n-type semiconductor and exhibits a maximum phonon energy of about 630 cm<sup>-1</sup>. The rare earth ion can be incorporated in the SnO<sub>2</sub> nanocrystal and substitutes for the Sn<sup>4+</sup> ions in the cassiterite-structured SnO<sub>2</sub> nanocrystals. Among the different fabrication techniques we have mentioned the laser annealing used in the case of GeO<sub>2</sub> transparent glass ceramic planar waveguides. The GeO<sub>2</sub> planar waveguide after 2h of CO<sub>2</sub> laser irradiation exhibits an increase of 0.04 in the refractive index, measured at 1542 nm. We demonstrate that the laser annealing significantly reduce propagation loss in GeO<sub>2</sub> planar waveguides that laser annealing decreases the surface roughness. From another side Raman results have shown that laser treatment for 1 h and 2 h results in realization of structures at nanometric scale on the surface of the samples with dimension from 10 to 50 nm attributed to the presence of GeO<sub>2</sub> nanocrystals. These behaviors evidence the role of the laser heat treatment on the glass inducing a progressive modification of the films.

We can therefore conclude that the prospects of glass photonic are still very good, with continuously growing applications in the area of integrated optics, lasing, lightning, frequency converters and sensing. Optimization of the synthesis processes of glasses tailored for the specific application and of the confined structures fabrication technologies may guarantee that new record performances of rare earth doped glass based photonic devices will be achieved. In respect to the development of photonic glass ceramics the immediate problems to solve are: (i) reproducible fabrication protocols; (ii) modelling of transparency constraints including disorder; (iii) enhance solubility for the rare earth ions.

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