

Spectroscopic and structural investigation of undoped and Er3+ doped hafnium silicate layers.

Larysa Khomenkova, Yong Tao An, Dimitri Khomenkov, Xavier Portier,

Christophe Labbé, Fabrice Gourbilleau

▶ To cite this version:

Larysa Khomenkova, Yong Tao An, Dimitri Khomenkov, Xavier Portier, Christophe Labbé, et al.. Spectroscopic and structural investigation of undoped and Er3+ doped hafnium silicate layers.. Physica B: Condensed Matter, 2014, 453, pp.100-106. 10.1016/j.physb.2014.03.087 . hal-01078069

HAL Id: hal-01078069 https://hal.science/hal-01078069

Submitted on 26 Jun 2018

HAL is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers. L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.

Spectroscopic and structural investigation of undoped and Er³⁺ doped hafnium silicate layers

L. Khomenkova^{a,b,*}, Y.-T. An^a, D. Khomenkov^c, X. Portier^a, C. Labbé^a, F. Gourbilleau^a

^a CIMAP CEA/CNRS/ENSICAEN/UCBN, 6 Blvd. Maréchal Juin, 14050 Caen Cedex 4, France

^b V. Lashkaryov Institute of Semiconductor Physics at NASU, 41 Pr. Nauky, Kyiv 03028, Ukraine

^c Taras Shevchenko National University of Kyiv, Faculty of Physics, 4 Pr. Hlushkov, Kyiv 03022, Ukraine

Keywords: Hafnium silicate films, Erbium, Photoluminescence, TEM, Raman scattering, FTIR.

ABSTRACT

This paper demonstrates the functionality of radio-frequency magnetron sputtering for the fabrication of undoped and Er-doped Si-rich-HfO₂ films with specific structural and spectroscopic properties. The effect of post-deposition treatment on film properties was investigated by means of Fourier-transform infrared spectroscopy, Raman scattering and photoluminescence methods, as well as Transmission Electron microscopy. It was observed that annealing treatment at 850–1000 °C causes phase separation process and the formation of HfO₂, SiO₂ and pure Si phases. This process stimulates also an intense light emission in the 700–950-nm spectral range under broad band excitation. The phase separation mechanism as well as the nature of radiative transitions were discussed. Photoluminescence was ascribed to carrier recombination in silicon clusters and host defects. The appearance of silicon clusters was also confirmed by the comparison of luminescent properties of pure HfO₂, SiO₂, Si-rich-HfO₂ and Si-rich-HfO₂ films. Additional argument for Si clusters' formation was obtained under investigation of Erdoped Si-rich HfO₂ films. These latter demonstrated 1.54- μ m Er³⁺ luminescence under non-resonant excitation originating from an energy transfer from Si clusters Er³⁺ ions.

1. Introduction

Hafnia-based materials are mainly considered as alternative dielectrics to SiO₂ in Si-based CMOS technology [1–3]. Along with promising electrical properties, HfO₂ has a high optical transparency in the ultraviolet-infrared spectral range, high refractive index (2.0–2.1 at 1.95 eV), wide optical bandgap (\sim 5.8 eV) and low phonon cut-off energy (\sim 780 cm⁻¹) offering a low probability of phonon assisted relaxation. Despite these advantages, spectroscopic investigation and optical applications of HfO₂-based materials are not numerous [4,5].

Intrinsic luminescence of pure HfO_2 is usually observed at 4.2– 4.4 eV and ascribed to self-trapped exciton [6,7], whereas visible emission (2.5–3.5 eV) is attributed to different oxygen vacancies with trapped electrons [6]. An emission of HfO_2 materials doped with rare-earth elements was also demonstrated [8,9], but the mechanism of the excitation of rare-earth ions and their interaction with the host defects were not clarified. The development of rare-earth doped materials suffers from their lower absorption cross-sections for 4f–4f transitions $(10^{-18}-10^{-20} \text{ cm}^{-2})$ and requires high-power excitation sources. Meanwhile, 4f–5d transitions have higher cross-section ($\sim 10^{-12} \text{ cm}^{-2}$), but corresponding excitation levels belong to UV and vacuum UV spectral range, restricting many applications of these materials. Thus, to enhance an excitation of 4f–4f transitions, a host mediated excitation via energy transfer is needed.

Trivalent erbium ion (Er³⁺) is one of the most popular lanthanides due to its radiative transitions in the green (${}^{4}S_{3/2} \rightarrow {}^{4}I_{15/2}$, 558 nm) and in the infrared (${}^{4}I_{13/2} \rightarrow {}^{4}I_{15/2}$, 1.54 µm) being extensively used as an eye-safe source in atmosphere, laser radar, medicine and surgery (${}^{4}I_{11/2} \rightarrow {}^{4}I_{13/2}$) [10,11]. Earlier, a lot of efforts have been concentrated on Er³⁺-doped Si-rich-SiO₂ materials [12–14]. An enhancement of Er³⁺absorption cross-section from 10^{-21} cm^{-2} [15] up to 10^{-16} cm^{-2} [13,14] was achieved due to an effective energy transfer from Si-nanoclusters (Si-ncs) towards Er³⁺ ions under visible broad-band excitation. Only a few studies of Er-doped SiO₂-HfO₂ materials were reported [8,16,17], whereas Er³⁺ ion excitation mechanism was not considered well.

In this work we present the results of a spectroscopic investigation of different HfO_2 -based materials, i.e. pure and Si-rich HfO_2 materials (either undoped or doped with Er^{3+} ions). The effect of annealing treatment on the formation of Si-ncs in Si-rich HfO_2

^{*} Corresponding author at: V. Lashkaryov Institute of Semiconductor Physics at NASU, 41 Pr. Nauky, Kyiv 03028, Ukraine.

E-mail addresses: khomen@ukr.net, khomen@isp.kiev.ua (L. Khomenkova).

host as well as the interaction mechanism of Si-ncs, Er^{3+} ions and host defects are investigated to achieve and to control light emitting properties of these materials.

2. Experimental details

The layers investigated were grown RF magnetron sputtering of pure HfO₂ target (99.9%) topped by calibrated Si and/or Er₂O₃ chips on pure Si and fused quarts substrates. Both types of substrates were placed on the same sample holder to obtain the layers grown at the same conditions. Before deposition, all Si wafers were etched in 10%-water HF solution to remove thermal SiO₂ laver from their surface. Substrate temperature was kept at 100 °C. The films were deposited in pure argon plasma with RF power density of 0.74 W/cm² during 200 min. The thickness of all films investigated was about 850 nm. After deposition, the wafers were cut on $1 \times 1 \text{ cm}^2$ pieces. Thermal treatments of the samples were performed in a conventional furnace in nitrogen flow at $T_A = 800$ -1100 °C and $t_A = 10-60$ min. Besides, pure and Si-rich SiO₂ films were grown by similar approach from a pure SiO_2 target (99.99%) and processed with the same conditions. More experimental details can be found in Ref. [3]. This allowed a comparative investigation of Si-ncs formation in Si-rich HfO2 films and Sirich-SiO₂ films to be performed.

The microstructure of the layers was analyzed by means of Fourier-transform infrared (FTIR) spectroscopy. The spectra were recorded in the range of 600–4000 cm⁻¹ using a Nicolet Nexus spectrometer under normal and Brewster angle (65°) configuration. Raman scattering spectra were investigated with back-scattered geometry using a dispersive Raman spectrometer equipped with a CCD camera and a laser source at 532 nm. Transmission electron microscopy (TEM) was used to observe the evolution of the film structure with annealing. Photoluminescence (PL) and PL excitation (PLE) spectra in the 200–800 nm

spectral range were studied with a Jobin–Yvon Fluorolog3-22 setup equipped with a Xe lamp as excitation source and a R928 photomultiplayer tube. All measurements were performed at room temperature.

3. Results and discussion

3.1. Microstructure of the samples versus annealing treatment

FTIR and Raman scattering spectra we measured to study the microstructure evolution of the films with thermal treatment. Both methods allowed following the phase separation process upon annealing.

Fig. 1 represents the FTIR spectra obtained for as-deposited (AD) and annealed samples. AD samples show two broad bands (Fig. 1a and b). One of them, detected in the range of the 500–700 cm⁻¹, is caused by Hf–O vibrations [18–25]. Usually, amorphous films show a broad Hf–O vibration band [18–20,25], whereas their crystalline counterparts demonstrate the peaks at 760–780 cm⁻¹, 640–670 cm⁻¹ and 500–560 cm⁻¹ [20,23–25]. Precise spectral positions and corresponding intensities of these bands depend on the HfO₂ crystalline nature (monoclinic, tetragonal, orthorhombic or cubic) [24,25].

Another vibration band, observed for AD films, locates at $\sim 1010-1030 \text{ cm}^{-1}$ (Fig. 1a) and 940–960 cm⁻¹ (Fig. 1b). It can be ascribed to Hf–Si–O stretching vibration mode taking into account the transformation of FTIR spectra of pure HfO₂ films with Si incorporation towards HfSiO_x host formation [2,20,27].

Thermal treatment at T_A =800–950 °C and t_A =60 min causes an increase of the intensity of O–Si–Hf band and its spectral shift up to 1070–1080 cm⁻¹, while Hf-O vibration band remains unchanged (Fig. 1a and b). FTIR spectra of the films annealed at T_A =1000 °C and t_A =60 min show several peaks at about 670 cm⁻¹ and 760 cm⁻¹ (Fig. 1a) and 560 cm⁻¹, 650 cm⁻¹ and

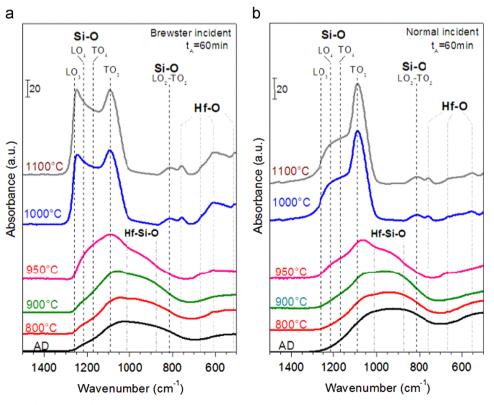


Fig. 1. Evolution of FTIR spectra of Si-rich-HfO₂ films with *T_A*. Spectra were measured at Brewster (65°) (a) and normal incident (b) of the light. Annealing time is *t_A*=60 min.

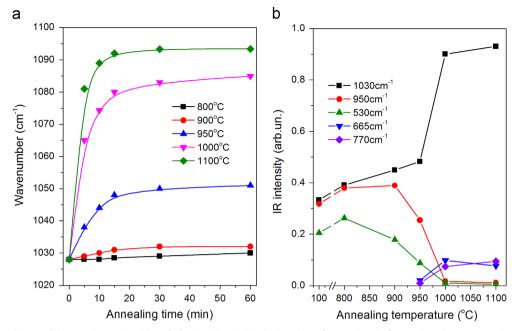


Fig. 2. (a) Effect of annealing conditions on the peak position of Hf–Si–O vibration band. The values of T_A are given in figure. The points at $t_A=0$ min correspond to AD films. (b) Variation of the intensities of different vibration bands corresponding to Hf–Si–O (1030 cm⁻¹ and 950 cm⁻¹) and Hf–O (530 cm⁻¹, 665 cm⁻¹ and 770 cm⁻¹) vibration bands with T_A ; $t_A=60$ min. The points at $T_A=100$ °C correspond to AD films. Analysis was performed on FTIR spectra measured at Brewster (65°) incidence.

760 cm⁻¹ (Fig. 1b) that are due to Hf–O vibrations [20,24–27]. Besides, the bands centered at 820 cm⁻¹, 1090 cm⁻¹ and 1250 cm⁻¹ (Fig. 1a) and at 820 cm⁻¹, 1085 cm⁻¹ and 1210 cm⁻¹ (Fig. 1b) are also observed. They can be explained by the different optical SiO₂ phonons [2,28,29]. This evolution of FTIR spectra can be caused by a phase separation process leading to the formation of SiO_x and HfO_x phases. However, the shoulder at about 950 cm⁻¹ could be an evidence of the presence of HfSiO_x phase in the films annealed at $T_A < 1000$ °C, meaning that the phase separation is not complete. An annealing at T_A =1100 °C and t_A =60 min results in the narrowing of all vibration bands as well as in the disappearance of the shoulder at 950 cm⁻¹ that reflects a complete phase separation process (Fig. 1).

The effect of the duration of annealing treatment on FTIR spectra was also studied. The analysis of the spectra measured at Brewster incidence was performed because it gives more details on the evolution of FTIR spectra with annealing. As one can see from Fig. 2a, the shift of Hf-Si-O peak position (observed at about 1030 cm^{-1} for AD film) is more pronounced for the samples annealed at $T_A \ge 950$ °C, whereas lower annealing temperature does not affect this peak position. The increase of t_A up to 15 min results in a pronounced shift of Hf-Si-O peak position to higher wavenumbers (Fig. 2a) as well as a significant increase of its intensity (Fig. 2b). This demonstrates that the main variation of FTIR spectra occurs at short-time treatment (\leq 15 min) that is in agreement with the results of our earlier work [20]. At the same time, longer annealing time results in a slight shift of this peak (less than 2 cm^{-1}) (Fig. 2a). The analysis of FTIR spectra shows also the decrease of intensity of 950-cm⁻¹ shoulder (considered as a contribution of the HfSiO_x phase) and its vanishing for $T_A \ge 1000$ °C. Along with this, the magnitude of 530 cm^{-1} band decreases, whereas the bands at 665 cm^{-1} and 770 cm^{-1} appear and increase in intensity (Fig. 2b). However, no significant variation of these peaks' positions was observed. This behavior of FTIR spectra means that there are two steps of the evolution of samples' microstructure. Short-time treatment results in the formation of SiO_x and HfO_x phases, whereas longer annealing causes the crystallization of HfO₂ phase and further SiO_x decomposition. However, it is worth to note

that the films annealed at T_A =800–950 °C, whatever t_A , demonstrate the HfSiO_x phase stability (Fig. 2a).

Raman scattering experiments of AD and annealed films were carried out to obtain additional information about microstructure evolution. For this purpose, the films deposited on Si and quarts substrates were investigated. Since corresponding Raman scattering spectra were found to be similar, hereafter we will discuss the spectra of the films grown on quarts wafers to eliminate the contribution of Si TO-phonon mode observed usually at 521 cm⁻¹.

Fig. 3 shows only a broad band with a maximum at \sim 499 cm $^{-1}$ for AD samples. Upon annealing at T_A =800 °C, this peak position shifted up to \sim 502 cm⁻¹ (Fig. 3) that can be due to the formation of rather pure silicon phase than HfO₂ phase. In fact, HfO₂ related phonons are usually observed at 115 cm^{-1} , 295 cm^{-1} , 470 cm^{-1} , 595 cm^{-1} and 665 cm^{-1} (for tetragonal phase) and at 123 cm^{-1} , 146 cm^{-1} , 257 cm^{-1} , 316 cm^{-1} , 390 cm^{-1} , 496 cm^{-1} , 570 cm^{-1} , 623 cm^{-1} and 660 cm^{-1} (for monoclinic phase) [22–26]. It is worth to note that the band peaked either at 470 cm^{-1} (tetragonal phase) or 498 cm⁻¹ (monoclinic one) are the most intense phonon responses (Fig. 3). Since any formation of nanocrystalline or amorphous phase results in the broadening and low-energy shift of the phonons of bulk material, one can expect that nanocrystalline HfO₂ phase have to follow this prediction [26,27]. Since the contribution of bulk Si TO phonon is eliminated, and SiO₂ phonon is peaked at \sim 495 cm⁻¹ (Fig. 3) and the HfO₂ related phonons are not observed at 500–525 cm⁻¹, one can conclude that the peak observed at 502 cm⁻¹ for the film annealed at $T_A = 800$ °C is due to Si TO phonon and attests for the formation of pure Si phase. Higher annealing temperatures result in a continuous shift of the Raman peak position up to 516 cm⁻¹ (T_A =900 °C) as well as an increase of its intensity. This behavior can be ascribed to the formation of Si clusters as well as to the increase of their sizes and density. Their mean size was estimated to be 3.7–4.0 nm using the model described in Ref. [30].

Further T_A increase up to 950–1100 °C results in the enhancement of Raman signal intensity as well as in the appearance of welldefined peaks at 455 cm⁻¹, 506 cm⁻¹, 523 cm⁻¹, 560 cm⁻¹, 575 cm⁻¹ and 625 cm⁻¹. Higher T_A values result in the enhancement

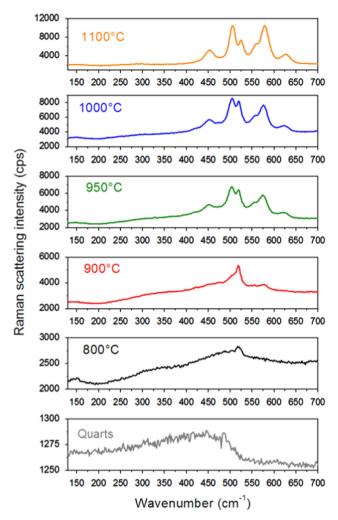


Fig. 3. Evolution of Raman scattering spectra of Si-rich HfO₂ films with annealing temperature measured for the samples grown on quarts substrate. Corresponding spectrum for quartz substrate is presented for the comparison. The T_A values are presented on the figure; t_A = 60 min.

of all phonon modes as well as in the variation of their relative contribution to the spectrum (Fig. 3). All these bands correspond to the crystallized HfO_2 phase, but some of them are overlapping with the Si TO phonon mode, hampering to conclude about the evolution of Si clusters' distribution (Fig. 3). To reveal the presence of Si clusters, cross-sections of annealed samples were observed by TEM.

Fig. 4 presents a high-resolution TEM image of the sample annealed at T_A =950 °C and t_A =60 min and showing the brightest PL emission. It is clearly seen some peculiarities of phase separation in the film volume. An irregular structure is observed in the film depth close to the film/substrate interface (Fig. 4, image 1). Dark grey regions are due to HfO₂ phase, whereas white regions are due to SiO₂, respectively. Light grey regions, separating both phases, can be due to either pure Si phase (or SiO_x phase with high Si content), or $HfSiO_x$ phase, considering the relatively large volume of this phase. The structure in the middle part of the film (image 3) appears to be unaffected by the thermal treatment. This can explain the significant contribution of Hf-Si-O vibration bands in FTIR spectra of samples annealed at T_A =950 °C (Figs. 1 and 2). The cross-section of upper regions showed the formation of alternated layers with different contrasts that can be related to the formation of Hf-rich and Si-rich layers (Fig. 4, image 4). Observed features of the structure can be explained taking into account the possible mechanism of HfSiO_x decomposition.

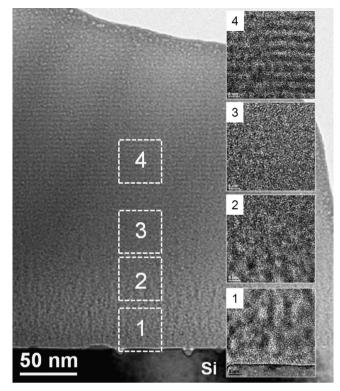


Fig. 4. Cross-sectional bright-field TEM images of the sample annealed at 900 °C. The images 1–4 represent magnified sections of different regions of the film depth.

Several mechanisms are known to explain the processes of material decomposition. One of them is Ostwald ripening that described the change of an inhomogeneous structure over time, when some clusters (grains) grow, only to be later dissolved, and other clusters are increasing in size due to the diffusion of corresponding species [31]. However, the initial formation of the microscopic clusters involves a large free energy barrier, and so can be very slow. Contrary to Ostwald ripening, spinodal decomposition is the mechanism for the rapid unmixing of a mixture of liquids or solids from one thermodynamic phase [32,33]. A clustering reaction occurs in a homogeneous, supersaturated solid solution, which is unstable against infinitesimal fluctuations in density or composition. This solution therefore separates spontaneously into two phases, starting with small fluctuations and proceeding with a decrease in the Gibbs energy without a nucleation barrier [32,33]. It is known that HfSiO_x materials have a metastable miscibility gap due to the covalent nature of Si-O bonds on one hand and ionic nature of Hf-O ones, on the other hand. The decrease of film free energy occurs usually through a phase separation process between HfO₂-rich and SiO₂-rich phases [34] when the HfO₂ phase forms at first. Thus, Ostwald ripening could not explain the formation of alternated multilayer structure in our films. At the same time, spinodal decomposition supposes the existence of compositional wavelength, λ_{C} , in the mixture, causes multilayer formation. This λ_{C} value, can be defined as an average distance between the centres of two successive HfO₂rich layers [34] and for our films $\lambda_{\rm C}$ was found to be about 4–5 nm.

For very thick films (compared to bulk materials), this composition wave starts to form at their surface and decays towards the volume. However, the substrate/film interface stimulates also formation of composition wave. Thus, the interaction of the waves penetrating in opposite directions, i.e. from film/substrate interface towards film surface and vice versa [34], will determine the result of the phase separation. The formation of alternated layer structure close to the film surface is an evidence of significant role of the surface in the phase separation process. Besides HfO_2 and SiO_2 phases, light grey regions separating the two phases were also observed (Fig. 4, images 1,2). They can correspond to either pure Si phase (or SiO_x phase with low amount of oxygen), or $HfSiO_x$ phase, considering the relatively large volume of this phase. From the comparison of FTIR (Figs. 1 and 2) and Raman scattering data (Fig. 3), one can suppose that the contribution of silicon-rich phase exceeds that of $HfSiO_x$. It is interesting to notice that the diameters of light-grey regions do not exceed 5 nm (Fig. 4). This is in agreement with Si-ncs mean size estimated from Raman scattering spectra (Fig. 3). However, any sequences of crystalline planes were not revealed during TEM-HREM observation. The amorphous nature was also confirmed by selected angle electron diffraction data (not shown here).

One of the reasons could be a non-completed phase separation process. In this case, not only the HfO₂ and SiO₂ phases are formed, but also SiO_x (with a high Si content). Besides HfSiO_x is observed and found to be stable at 900–950 °C (Fig. 2). It is clear that T_A =950 °C can be too low to stimulate the crystallization of Si-rich phase. This fact is in agreement with the Raman scattering data showing peak position of TO Si phonon mode at about 516 cm⁻¹ instead of 521 cm⁻¹ for crystallized Si phase (Fig. 3). The presence of amorphous Si clusters was also demonstrates by atomic-probe tomography study for similar samples [35].

The increase of the annealing temperature up to 1000 °C results mainly in the crystallization of HfO_2 -rich phase and in the formation of HfO_2 nanocrystals. Besides, the crystallization of Si clusters occurs [36]. However, this process starts at 1000–1050 °C [33] before the temperatures usually required for the formation of Si-ncs in SiO₂ host (1100–1150 °C) [3,14,37,38].

It is known that formation of Si-ncs results in a bright visible PL emission [3,14,38]. Along with this, the peak position of corresponding PL band demonstrates usually the "blue" shift with the size decrease [3,38]. By analogy, the similar behavior of PL emission from Si clusters embedded in HfO₂ can be also expected. Thus, the comparison of light emitting properties of Si-rich-HfO₂ samples with that of Si-rich-SiO₂ films, grown with similar approach [3,38], can give information about the formation of Si clusters in HfO₂-based host.

3.2. Light emission from the films

AD samples showed blue PL emission under UV (260–360 nm) excitation only [17]. Any luminescence was not detected under visible

(450–535 nm) excitation. Annealing treatments at $T_A > 800$ °C result in the appearance of PL emission in the red spectral range (Fig. 5a). It is worth to note that we used 532-nm excitation to separate the contribution from host defects and formed Si clusters.

The comparison of pure SiO₂, pure HfO₂, Si-rich-SiO₂ and Sirich HfO₂ films (Fig. 5a) showed that this red emission is due to Si cluster formation. Their presence was also predicted by Raman scatting spectra (Fig. 3). Red emission was found to be very similar to the case of Si-rich-SiO₂ materials (Fig. 5a) and can be thus attributed to Si clusters formed in Si-rich-HfO₂ samples upon annealing treatment. The peak position of this PL band shifts to shorter wavelengths (from 820 nm to 690 nm) with T_A increase from 850 °C to 1100 °C at constant t_A (Fig. 5b). This can be ascribed to the decrease of mean size of Si clusters. The t_A increase at constant T_A results in the shift of PL peak position to longer wavelengths [17] that can be caused by an increase of Si cluster sizes similar to the case observed for Si-rich SiO₂ films [3,38]. The PL intensity shows a non-monotonous behavior (Fig. 5b). The most efficient emission was detected for samples annealed at $T_A=900-$ 950 °C, $t_A = 60$ min.

It is known that Si clusters embedded in SiO₂ host doped with rare-earth ions are effective sensitizers of these latter [12–14]. Thus, an achievement of the efficient PL emission from rare-earth elements under non-resonant excitation can be an additional argument of Si-ncs presence in HfO₂ host. For this purpose, pure and Si-rich HfO₂ films were co-doped with Er^{3+} ions and their PL properties have been studied versus annealing treatment.

Among usually addressed Er^{3+} emission in 1.4–1.7 µm from ${}^{4}\text{I}_{13/2} \rightarrow {}^{4}\text{I}_{15/2}$ transitions, the spectra were recorded in the 630–700-nm range corresponding to ${}^{4}\text{F}_{9/2} \rightarrow {}^{4}\text{I}_{15/2}$ transitions at resonant (488 nm or 532 nm) and non-resonant (476 nm) excitations.

Fig. 6 shows a comparison of PL spectra of Er^{3+} -doped and $(\text{Er}^{3+}, \text{Si})$ -codoped HfO₂ films annealed at T_A =900 °C, t_A =60 min. Since pure HfO₂ does not emit under 532-nm excitation wavelength (Fig. 6), this means that, if any oxygen vacancies are present in Er-doped HfO₂ films, their emission cannot be excited by this light wavelength. Moreover, self-trapped excitons, requiring an ultraviolet excitation [6], are not considered as a sensitizer of Er^{3+} ions, since used optical excitation is not efficient in this case. Thus, well-defined PL peaks (corresponding to ${}^{4}\text{F}_{9/2} \rightarrow {}^{4}\text{I}_{15/2}$) observed for Er-doped HfO₂ films under 532-nm excitations (Fig. 6) are due to direct excitation of Er^{3+} ions. This latter was confirmed by very low PL intensity (as a noise level) observed under "non-resonant" 476-nm-illumination [17].

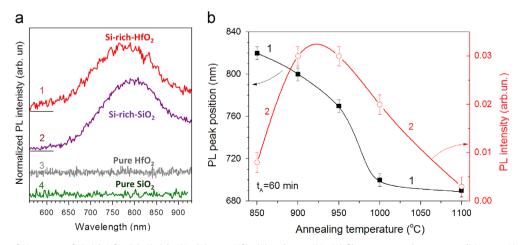


Fig. 5. (a) Comparison of PL spectra of Si-rich HfO₂ (1), Si-rich-SiO₂ (2), pure HfO₂ (3) and pure SiO₂ (4) films grown at the same conditions on Si substrates. T_A =900 °C, t_A =60 min. Excitation wavelength λ_{exc} =488 nm. It is seen that both pure HfO₂ (3) and pure SiO₂ (4) layers do not demonstrate emission in the 600–950 nm spectral range. At the same time, Si-rich-HfO₂ (1) and Si-rich-SiO₂ (2) films show the emission with similar PL peak position that allows to conclude about important contribution of excess Si in the red emission. (b) Evolution of PL peak position (1) and PL intensity (2) of the "red" PL band of Si-rich-HfO₂ films with T_A . Annealing time is t_A =60 min. (For interpretation of the references to color in this figure legend, the reader is referred to the web version of this article.)

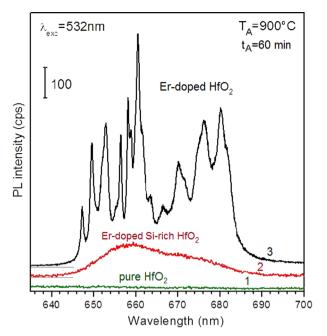


Fig. 6. PL spectra of pure HfO₂ (1), Er-doped HfO₂ (2) and Er-doped Si-rich-HfO₂ (3) films grown with the same conditions on Si substrate and annealed at T_A =900 °C and t_A =60 min. The spectra were measured under excitation with λ_{exc} =532 nm.

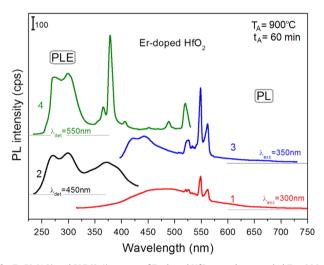


Fig. 7. PL (1,3) and PLE (2,4) spectra of Er-doped HfO₂ samples annealed T_A =900 °C and t_A =60 min. PL spectra were measured under excitation with λ_{exc} =300 (1) and 350 nm (3). PLE spectra were detected at λ_{det} =450 (2) and 550 nm (4). The spectra are shifted in vertical direction for clarity. (For interpretation of the references to color in this figure legend, the reader is referred to the web version of this article.)

Fig. 7 presents PL spectra of Er-doped HfO₂ films under UV excitation (curves 1,3) and corresponding PLE spectra (curves 2,4). Several broad PLE bands are seen at 280 nm (4.42 eV), 300 nm (4.13 eV) and 370 nm (3.30 eV), when detected at λ_{det} =450 nm (Fig. 7, curve 2). The overlapping of all these bands can affect their peak positions and a slight shift of their maxima can be observed. Two first peaks are usually ascribed to negatively charged Hf-related oxygen vacancies with different coordination, i.e.V₀₍₃₎⁻² (4.42–4.44 eV) and V₀₍₄₎⁻² (4.2–4.23 eV) [36]. The third PLE band can be attributed to either neutral or positively charge oxygen vacancies, i.e. V₀₍₃₎⁰ or V₁₍₃₎⁺² based on the data of Ref. [39].

The broad PL band peaked at 400–460 nm is due to the relaxation of the vacancies (Fig. 7, curves 1,3). It results in the blue light emission under UV light excitation. Sharp peaks in

the 520–570 nm range (Fig. 7, curve 3) are caused by ${}^{4}S_{3/2} \rightarrow {}^{4}I_{15/2}$ transitions in Er^{3+} ions that is confirmed by the PLE spectrum recorded at λ_{det} =550 nm, showing the corresponding sharp excitation lines (Fig. 7, curve 4). The PLE spectrum detected at λ_{det} =450 nm (curve 2) has broad PLE bands with maxima at 280 nm and 300 nm, whereas the peak at about 380 nm is overlapped with the absorption bands corresponded to ${}^{4}I_{15/2} \rightarrow {}^{4}G_{11/2}$, ${}^{4}H_{11/2}$ transitions in Er^{3+} ions. Thus, one can conclude that, using UV-blue light, Er^{3+} ions can be effectively excited not only via direct excitation but also via an efficient energy transfer from oxygen vacancies in the HfO₂ matrix.

Similar host mediate excitation was observed for Er-doped Sirich-HfO₂ samples. Fig. 8 shows the comparison of PL and PLE spectra obtained for Er-free (curves 1,5) and Er-doped films (curves 2–4,3–6). The variation of the excitation wavelength in the 260–380 nm spectral range does not result in a shift of the PL peak position. Thus, this PL band is most probable due to defects in the Si-rich-HfO₂ matrix.

In general, PLE spectra of all the samples are broad. Two main peaks are observed similar to those of Er-free Si-rich-HfO₂ samples (Fig. 8). The first one is situated at about 290 nm (4.27 eV), whereas the position of the second peak depends on the detection wavelength. Meanwhile, it is observed between 370 nm and 400 nm (3.35-3.10 eV). Similar shift of PLE peak position was reported in Ref. [38] and explained by a depth distribution of the emitting defects. It is interesting that the Er-free samples show broader PL spectrum (Fig. 8, curve 1) than their Er-doped counterparts (curves, 2–4). This can be due to quenching of some radiative channels transferring efficiently their energy towards Er³⁺ ions. This assumption is supported by the observation of some PL peaks in the green (\sim 570 nm) and orange (\sim 620 nm) spectral ranges (Fig. 8, curves 2-4) when 260-280 nm-excitation is used. The PLE spectrum, measured at λ_{det} = 615-nm wavelength (Fig. 8, curve 8), confirms the Er^{3+} origin of the green and orange emissions.

It is interesting to note that PLE spectrum, measured at λ_{det} =720 nm (corresponding to PL maximum of Si cluster's emission [17]), was found to be similar to the spectrum detected at λ_{det} =650 (Fig. 8, curve 6). The same data were obtained in Ref. [40] demonstrating that oxygen-related defects can take part in the excitation of Si clusters in Si-rich-SiO₂ films.

PL study of Er-doped Si-rich-ZrO₂ films [41,42] revealed the Er^{3+} emission under non-resonant excitation and ascribed it to an energy transfer from Si clusters. However, the presence of these latter was not clearly demonstrated. Meanwhile, it was shown that oxygen-deficient silicon centers (Si-ODCs) and oxygen interstitial defects are effective excitation paths for Er^{3+} ions [40]. It is also known that Si- (Ge-)ODCs can be effective sensitizers of rare earth ions embedded in Si-rich [40] or Ge-rich [43] silicon oxide host.

Thus, one can conclude that the energetic balance of interacting "Er-ions, Si clusters and defects" system is complicated. All three of them can emit light under "corresponding" excitation, whereas Si clusters and defects can be effective sensitizers for rare-earth ions. This can allow achieving broad-band emission from hafniumsilicates doped by other rare earth ions.

4. Conclusions

In the present study, the properties of RF magnetron sputtered hafnium silicate films were investigated by means of FTIR, Raman scattering, TEM, PL and PLE methods. It was observed that high temperature annealing governs a phase separation process as well as the formation of silica, hafnia and silicon phases. The appearance of a PL emission in the visible-near-infrared spectral range occurred. The evolution of the PL peak position was found to be correlated to the Si-ncs size. The properties of Si-doped-HfO₂ films

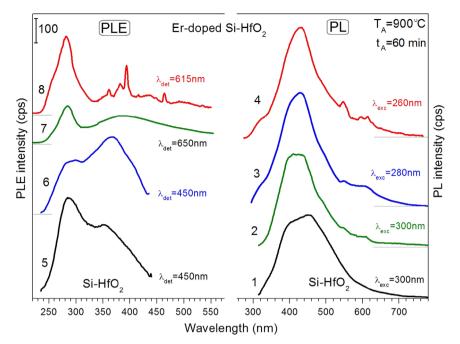


Fig. 8. PL (1-4) and PL excitation (5-8) spectra of Si-rich-HfO2 (1,5) and Er-doped-Si-rich-HfO2 (2-4, 6-8) films. PL spectra were measured under excitation with λ_{exc} =300 nm (1,2), 280 nm (3) and 260 nm (4). PLE spectra were detected at λ_{det} =450 nm (5,6), 650 nm (7) and 615 nm (8) wavelengths. All films were annealed at T_A=900 °C and t_A=60 min. The spectra are shifted in vertical direction for clarity. (For interpretation of the references to color in this figure legend, the reader is referred to the web version of this article.)

are compared to those of their counterparts doped with Er^{3+} ions. The investigation of the effect of annealing treatment on luminescent properties revealed that the enhancement of Er³⁺ PL emission occurs due to an effective energy transfer from Si-nanoclusters and defects of hafnia based host (such as oxygen vacancies).

Acknowledgments

This work was partially supported by the French National Agency (ANR) through Nanoscience, Nanotechnology Program (NOMAD project, No. ANR-07-NANO-022-02) and the Conseil Regional de Basse Normandie through the CPER project of the program "Nanoscience axe" (2007-2013), as well as by the National Academy of Sciences of Ukraine (project III-4-11).

References

- [1] G. He, L.Q. Zhu, Z.Q. Sun, Q. Wan, L.D. Zhang, Prog. Mater. Sci. 56 (2011) 475.
- [2] L. Khomenkova, X. Portier, J. Cardin, F. Gourbilleau, Nanotechnology 21 (2010) 285707
- [3] L. Khomenkova, B.S. Sahu, A. Slaoui, F. Gourbilleau, Nanoscale Res. Lett. 6 (2011) 172.
- [4] J.M. Khoshman, A. Khan, M.E. Kordesch, Surf. Coat. Technol. 202 (2008) 2500.
- [5] O. Stenzel, S. Wilbrandt, S. Yulin, N. Kaiser, M. Held, A. Tünnermann, J. Biskupek, U. Kaiser, Opt. Mater. Express 1 (2011) 278.
- [6] M. Kirm, J. Aarik, M. Jürgens, I. Sildos, Nucl. Instr. Meth. A 537 (2005) 251.
- [7] K. Smits, L. Grigorjeva, D. Millers, A. Sarakovskis, J. Grabis, W. Lojkowski,
- J. Lumin. 131 (2011) 2058. [8] V. Kiisk, I. Sildos, S. Lange, V. Reedo, T. Tatte, M. Kirm, J. Aarik, Appl. Surf. Sci.
- 247 (2005) 412. [9] L.X. Liu, Z.W. Ma, Y.Z. Xie, Y.R. Su, H.T. Zhao, M. Zhou, J.Y. Zhou, J. Li, E.Q. Xie,
- J. Appl. Phys. 107 (2010) 024309.
- [10] C. Stoneman, L. Esterowitz, Opt. Lett. 15 (1990) 486.
- [11] L. Feng, J. Wang, Q. Tang, L.F. Liang, H.B. Liang, Q. Su, J. Lumin. 124 (2007) 187.
- [12] A.J. Kenyon, Semicond. Sci. Technol. 20 (2005) R65. [13] M. Wojdak, M. Klik, M. Forcales, O.B. Gusev, T. Gregorkiewicz, D. Pacifici,
- G. Franzò, F. Priolo, F. Iacona, Phys. Rev. B 69 (2004) 233315.
- [14] S. Cueff, C. Labbé, J. Cardin, J.-L. Doualan, L. Khomenkova, K. Hijazi, O. Jambois, B. Garrido, R. Rizk, J. Appl. Phys. 108 (2010) 064302.
- [15] J. Miniscalco, J. Lightwave Technol. 9 (1991) 234.

- [16] G.C. Righini, S. Berneschi, G. Nunzi Conti, S. Pelli, E. Moser, R. Retoux, P. Féron, R.R. Gonçalves, G. Speranza, Y. Jestin, M. Ferrari, A. Chiasera, A. Chiappini, C. Armellini, J. Non-Cryst. Solids 355 (2009) 1853.
- [17] L. Khomenkova, Y.-T. An, C. Labbé, X. Portier, F. Gourbilleaua, ECS Trans. 45 (2012) 119.
- [18] N.V. Nguyen, A.V. Davydov, D. Chandler-Horowitz, M.F. Frank, Appl. Phys. Lett. 87 (2005) 192903.
- [19] M.M. Frank, S. Sayan, S. Dörmann, T.J. Emge, L.S. Wielunski, E. Garfunkel, YJ. Chabal, Mater, Sci. Eng., B 109 (2004) 6. [20] L. Khomenkova, C. Dufour, P.-E. Coulon, C. Bonafos, F. Gourbilleau, Nanotech-
- nology 21 (2010) 0955704.
- [21] M. Lui, L.Q. Zhu, G. He, Z.M. Wang, J.X. Wu, J.-Y. Zhang, I. Liaw, Q. Fang, I.W. Boyd, Appl. Surf. Sci. 253 (2007) 7869.
- [22] B.-K. Kim, H.-O. Hamaguchi, Mater. Res. Bull. 32 (1997) 1367.
- [23] A.C. Marques, R.M. Almeida, J. Sol-Gel Sci. Technol. 40 (2006) 371.
- [24] M. Yashima, H. Takahashi, K. Ohtake, T. Hirose, M. Kakihana, H. Arashi, Y. Ikuma, Y. Suzuki, M. Yoshimura, J. Phys. Chem. Solids 57 (1996) 289.
- [25] X. Zhao, D. Vanderbilt, Phys. Rev. B 65 (2002) 233106.
- [26] H. Arashi, J. Am. Ceram. Soc. 75 (1992) 844.
- [27] M. Mattarelli, M. Montagna, F. Rossi, C. Tosello, N.D. Afify, M. Bettinelli, A. Speghini, C. Armellini, Y. Jestin, F. Rocca, S. Gialanella, Opt. Mater. 31 (2009) 1362.
- [28] C.T. Kirk, Phys. Rev. B 38 (1988) 1255.
- [29] P. Lange, J. Appl. Phys. 66 (1989) 201.
- [30] H. Richter, Z.P. Wang, L. Ley, Solid State Commun. 39 (1981) 625.
- [31] I.M. Lifshitz, V.V. Slyozov, J. Phys. Chem. Solids 19 (1961) 35.
- [32] J.W. Cahn, Acta Metall. 9 (1961) 795.
- [33] V.P. Skripov, A.V. Skripov, Sov. Phys. Usp. 22 (1979) 389.
- [34] J. Lui, X. Wu, W.N. Lennard, D. Landheer, M.N.C. Dharma-Wardana, J. Appl. Phys. 107 (2010) 123510.
- [35] E. Talbot, M. Roussel, L. Khomenkova, F. Gourbilleau, P. Pareige, Mater. Sci.
- Eng., B 177 (2012) 717. [36] Y.-T. An, C. Labbé, L. Khomenkova, M. Morales, X. Portier, F. Gourbilleau, Nanoscale Res. Lett. 8 (2013) 43.
- E. Talbot, R. Lardé, F. Gourbilleau, C. Dufour, P. Pareige., Eur. Phys. Lett. 87 [37] (2009) 26004.
- [38] L. Khomenkova, N. Korsunska, V. Yukhimchuk, B. Jumayev, T. Torchynska, A. Vivas Hernandez, A. Many, Y. Goldstein, E. Savir, J. Jedrzejewski, J. Lumin. 102-103 (2003) 705.
- [39] H.-K. Noh, B. Ruy, E.-A. Choi, J. Bang, K.J. Chang, Appl. Phys. Lett. 95 (2009) 082905.
- [40] S. Cueff, C. Labbé, B. Dierre, F. Fabbri, T. Sekiguchi, X. Portier, R. Rizk, J. Appl. Phys. 108 (2011) 113504.
- [41] C. Rozo, L.F. Fonseca, J. Phys.: Condens. Matter 20 (2008) 315003.
- [42] C. Rozo, D. Jaque, L.F. Fonseca, J.G. Solé, J. Lumin. 128 (2008) 1197.
- [43] A. Kanjilal, L. Rebohle, M. Voelskow, W. Skorupa, M. Helm, Appl. Phys. Lett. 94 (2009) 051903.