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Morphology of gold nanoparticles determined by full-curve fitting of the light absorption spectrum. Comparison with X-ray scattering and electron microscopy data^{\dagger}

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UV-Vis absorption spectroscopy is frequently used to characterize the size and shape of gold nanoparticles. We present a fullspectrum model that yields reliable results for the commonly encountered case of mixtures of spheres and rods in varying proportions. We determine the volume fractions of the two populations, the aspect ratio distribution of the nanorods (average value and variance) and the interface damping parameter. We validate the model by checking the fit results against small-angle X-ray scattering and transmission electron microscopy data and show that correctly accounting for the polydispersity in aspect ratio is essential for a quantitative description of the longitudinal plasmon peak.



1 Introduction

The importance of gold nanoparticles in physics, biology and materials science over the last decade can hardly be overstated. Their manifold applications (see Ref. 1 and references therein) motivated a sustained effort^{2,3} towards the reproducible synthesis of particles (more commonly, spheres and rods) with controlled shape and size. Once produced, the particles must be characterized as quickly and as thoroughly as possible. In general, this is first done by UV-Vis-IR absorption spectroscopy (AS), a widely available technique that provides valuable information, since the absorption spectrum of noble metal nanoparticles is extremely sensitive to their shape (though much less to their absolute size).

The optical properties of metal nanoparticles have been extensively studied⁴. For particles much smaller than the wavelength, in the electrostatic approximation, the simplest way of describing the AS data is by the Gans theory⁵, with a finitesize correction to the dielectric function of the metal⁶. This approach was widely used for noble metal nanospheres^{7–9} and nanorods^{10,11}. In general, the authors only compare the experimental plasmon peak position with the simulation result, but some of the models also describe its width, by including the effect of polydispersity, for the sphere radius⁹ and for the aspect ratio of rods¹². With many synthesis conditions, the final state consists of rod and sphere mixtures, each population being polydisperse in size and (for the rods) in aspect ratio¹³. Any realistic model should take these complications into account.

The analysis is usually complemented by transmission electron microscopy (TEM), a technique with the major advantages of imaging the particles directly and of requiring no assumptions as to their environment. However, accurate statistical analysis of the TEM images is very work-intensive and relies on the assumption that the distribution of deposited and analysed particles is the same as in the original solution. Moreover, TEM does not give access to the particle concentrations.

[†] Electronic Supplementary Information (ESI) available: TEM images of nanoparticles and detailed analysis, simplified relations for the AS model, alternative estimate for the concentration and discussion of the dielectric constant chosen for bulk gold. See the end of this file.

Small-angle X-ray scattering (SAXS) does not perturb the solution, yields results averaged over millimetre-size volumes and over times ranging from milliseconds to minutes and is practically insensitive to the organic components (whose contribution is negligible compared to that of the inorganic nanoparticles). The modelling process makes no particular assumptions beyond solution homogeneity and is very sensitive to the particle size, but when different types of particles are present it is quite difficult to discriminate between them.

In this article, we develop a model that allows fitting the entire AS curve, while taking into account the polydispersity of the aspect ratio. We use it to study solutions obtained in diverse synthesis conditions, yielding mixtures of nanospheres and nanorods with varying radius and aspect ratio. Furthermore, the polydispersities of the two populations and their relative proportion change from sample to sample. This provides an ideal testing ground for the accuracy and discriminating power of the model.

After analyzing the absorption spectra by full-curve fitting we validate the results through a comparison with SAXS and TEM data. More specifically:

- Fitting the AS curves with reasonable starting values for the radii of the spheres and rods ($R_S = 10$ nm and $R_R = 5$ nm, respectively) yields the volume fractions of spheres and rods (ϕ_S and ϕ_R) and the aspect ratio X of the rods, defined as the length-to-diameter ratio: $X = L/(2R_R)$ as well as its polydispersity σ_X .
- The volume fractions and *X* are used as starting parameters for fitting the SAXS intensity I(q), which then yields the radii for the spheres and rods and their polydispersities (σ_S and σ_R). R_S and R_R are then injected back into the AS fitting for a refined analysis.
- Finally, the results are checked against the values derived from TEM images for (*R_S*, σ_S), (*R_R*, σ_R) and (*L*, σ_L).

The comparison of the three characterization techniques shows that the full-curve model for the AS data is quite robust and yields accurate values for the average aspect ratio of the rods and its variance, as well as for the particle concentrations. It can thus be employed for rapid evaluation of synthesis protocols and provides detailed information even in the absence of time-consuming techniques such as TEM. For ease of use, we also provide simplified relations between the features of the AS curves, on the one hand, and the morphological parameters of the particles and their concentration, on the other hand.

Sample	A1	A2	A3	A4	A5	A6
AgNO ₃	36	36	36	36	36	36
HCl	2	2	3	4.3	5	6
Sample	B1	B2	B3	B4	B5	B6
1						
AgNO ₃	24	24	36	24	12	30

Table 1 Synthesis conditions. Amounts of AgNO₃ and HCl solutions added to the growth solution (in mL).

2 Methods

2.1 Gold nanorod synthesis

The nanoparticles were synthesized by a seed-growth method, in the presence of either sodium oleate ¹⁴ (series A) or bromosalicylic acid ¹⁵ (series B); we changed the pH and the amount of seed solution to obtain varied particle characteristics (e.g. the aspect ratio of the nanorods). For the seed solution, 5 mL of 0.5 mM HAuCl₄ solution was mixed with 5 mL of 0.2 M hexadecyltrimethylammonium bromide (CTAB) solution. 1 mL of a fresh 6 mM NaBH₄ solution was injected into the HAuCl₄-CTAB mixture under vigorous stirring at 1200 rpm. The solution color changed from yellow to brownishyellow, and the stirring was stopped after 2 min. The seed solution was aged at room temperature for 30 min before injection into the growth solution.

The growth solution was prepared by mixing 9.0 g of CTAB either with 1.234 g of sodium oleate (series A) or with 1.1 g of 5-bromosalicylic acid (series B) dissolved in 250 mL of distilled water at 30 °C. A varying amount of 4 mM AgNO₃ solution was then added (see Table 1). The mixture was kept undisturbed at 30 °C for 15 min, after which 250 mL of a 1 mM HAuCl₄ solution and a small amount of 37 wt.% HCl solution were added (see Table 1).

After slowly stirring at 400 rpm for 15 min, 1.25 mL of 64 mM ascorbic acid solution was added and the mixture was strongly stirred until it became colorless. Finally, 0.4 mL (series A) or 0.8 mL (series B) seed solution was injected into the growth solution. The resulting mixture was stirred for 30 s and left undisturbed at 30 °C for 12 h, allowing the particles to grow. The particles were washed twice, by centrifuging (at 6000 g for 30 min) and redispersed in 0.1 M CTAB.

2.2 Absorbance spectroscopy: measurement and analysis

We used a Cary 5000 spectrometer (Agilent) to measure the absorption spectrum of dilute particle solutions between 400 and 1400 nm. The synthesis solutions were diluted 25 to 100 times (yielding a total Au volume fraction $\phi = \phi_S + \phi_R$ of the



Figure 1 Fits to the absorbance spectroscopy data with the model (1). The curves were shifted vertically for clarity.

order of 10^{-6}) and held in polystyrene cuvettes with 10 mm optical path.

The absorption spectra of gold nanorods consist of two bands corresponding to the oscillation of the free electrons either parallel or perpendicular to the long axis of the nanorods. The transverse plasmon peak (TPP) has a maximum around 520 nm and depends only weakly on the nanorod diameter. The longitudinal plasmon peak (LPP) is much stronger than the TPP and shifts to larger wavelength as the aspect ratio X increases. To describe the spectrum of the nanorods we treated them as ellipsoids and used Gans' formula⁵, which is a standard approach in the literature, to obtain Equation (1). However, since the synthesis rarely provides monodisperse nanorods, we account for their polydispersity by integrating over an aspect ratio distribution f(X) (first term) as well as for the presence of an additional particle population, modelled as spheres^{*} (second term), finally yielding for the absorbance γ:

$$\gamma = \frac{2\pi\varepsilon_m^{3/2}}{3\ln 10} \frac{l}{\lambda} \left[\phi_R \int f(X') dX' \sum_{i=1}^3 \frac{\varepsilon_2 P_i^{-2}}{\left(\varepsilon_1 + \frac{1-P_i}{P_i}\varepsilon_m\right)^2 + \varepsilon_2^2} \right]$$
(1)
+ $27\phi_S \frac{\varepsilon_2}{(\varepsilon_1 + 2\varepsilon_m)^2 + \varepsilon_2^2}$

where λ is the wavelength in vacuum, l is the optical path, $\varepsilon_m(\lambda)$ is the relative dielectric constant of the surrounding medium (water)¹⁶, ε_1 and ε_2 are the real and imaginary parts of the dielectric constant of the particles $\varepsilon(\lambda, R)$, derived from the (wavelength-dependent) value $\varepsilon_{\text{bulk}}(\lambda)$ for bulk gold[†] and corrected for the additional dissipation at the interface, which introduces the size dependence via the damping parameter A^{17} :

^{*}The spheres contribute to the AS spectrum a peak at approximately the same position as the TPP of the nanorods.

[†]See the ESI for a detailed discussion.

$$\varepsilon(\lambda, R) = \varepsilon_{\text{bulk}}(\lambda) + \frac{\omega_p^2}{\omega^2 + i\omega\gamma_0}$$
(2)
$$- \frac{\omega_p^2}{\omega^2 + i\omega(\gamma_0 + Av_F/R)},$$

where $\omega = 2\pi c/\lambda$ (with *c* the speed of light in vacuum), $\omega_p = 2.183 \times 10^{15} \text{ s}^{-1}$ the plasmon frequency, $\gamma_0 = 6.46 \times 10^{12} \text{ s}^{-1}$ the bulk damping rate, $v_F = 1.4 \times 10^6 \text{ m/s}$ the Fermi velocity and *R* stands for *R_S*, *R_R* or *L*/2, depending on the contribution to be described.

The geometrical factors P_i for a prolate ellipsoid along the major axis and along the transverse axes are respectively given by:

$$P_{1} = \frac{1 - e^{2}}{e^{2}} \left[\frac{1}{2e} \ln\left(\frac{1 + e}{1 - e}\right) - 1 \right],$$

$$P_{2} = P_{3} = \frac{1 - P_{1}}{2},$$
(3)

and depend on the aspect ratio via the eccentricity $e = \sqrt{1 - X'^{-2}}$. We assumed that the aspect ratio distribution f(X') of the nanorods is given by a Schulz function¹⁸, with average value X and standard deviation σ_X . Each particle participates to the absorption (1) in proportion to its volume, but f(X') is not volume-weighted since the volume of the nanorods is not correlated with their aspect ratio. This feature has already been described in the literature¹² and is fairly well verified by the analysis of our TEM data (see the ESI).

2.3 SAXS experiments

The SAXS data was collected at the SWING beamline of the SOLEIL synchrotron (Saint-Aubin, France), at a beam energy of 12 keV and two sample-to-detector distances (1 and 6.5 m). The samples were contained in cylindrical glass capillaries (Mark-Rörchen) 1.5 mm in diameter. Preliminary data treatment (angular averaging and normalization) was done using the software FOXTROT developed at the beamline and yielded the intensity as a function of the scattering vector I(q) in absolute units. Subsequent data fitting was done in IGOR PRO using the NCNR SANS package¹⁹.

2.4 **TEM experiments**

The TEM was done using a Topcon instrument, at a voltage of 100kV and a magnification of 14000. The objects were identified automatically using the image treatment routines available in IGOR PRO, individually validated by visual inspection and sorted into "nanospheres/nanocubes" and "nanorods" according to their aspect ratio. For each synthesis batch we analyzed at least 100 objects of each type.

3 Results and discussion

3.1 Absorbance spectroscopy

We obtain very good fits to the AS data using Equation (1), with free parameters ϕ_S , ϕ_R , *X*, σ_X and *A* (the radii are fixed at $R_S = 10$ nm and $R_R = 5$ nm). The experimental data and the models are shown in Figure 1.

3.2 SAXS

The scattered intensity I(q) is described by the sum of two contributions: spheres with polydisperse radius (mean value R_S and variance σ_S^2) and cylinders (or rods) with a fixed length L (the SAXS is not very sensitive to this parameter over the available q-range) and polydisperse radius (with mean value R_R and variance σ_R^2). We assume that both R_S and R_R are Schulz-distributed¹⁸. The scattering length densities for gold and water are: $\rho_{Au} = 1.082 \times 10^{-4} \text{\AA}^{-2}$ and $\rho_{H_2O} = 9.444 \times 10^{-6} \text{\AA}^{-2}$. [‡] We performed the SAXS on concentrated solutions, with a total Au volume fraction $\phi \sim 10^{-4}$.



Figure 2 Ratio of the two experimental estimates for the gold volume fraction $\phi(AS)/\phi(SAXS)$. $\phi(AS) = \phi_R + \phi_S$ is obtained by fitting the absorption spectrum with Equation (1) (diamonds) and by using the atomic extinction coefficient from the literature (open dots); see the ESI for more details. $\phi(SAXS)$ is obtained from the integrated SAXS intensity as described in the text.

We started by estimating the volume fraction of gold particles in solution from the intensity integral over the entire reciprocal space. This SAXS "invariant" is given by: $Q = \int_{0}^{\infty} dq q^2 I(q)$ and, for a two-phase system (in our case, gold and solvent) is related to their volume fractions by $Q = 2\pi^2 (\rho_{Au} - \rho_{H_2O})^2 \phi (1 - \phi)^{20}$. This treatment involves no assumption as to the particle morphology.

[‡]From http://www.ncnr.nist.gov/resources/activation/.



Figure 3 SAXS data (dots) and fits (solid lines). For clarity, the curves are shifted vertically by arbitrary amounts and only every fifth point is shown.

The resulting gold volume fraction should be equal to that determined from fitting the absorption spectrum (accounting for the dilution ratio between the two samples).

In Figure 2 we show the ratio $\phi(AS)/\phi(SAXS)$ for all studied samples (solid diamonds). Its average is about 1.25, an overestimation that should be kept in mind when using Equation (1) for determining the total particle concentration. We also use a different estimate for $\phi(AS)$, based on literature results for the spheres^{21,22} and the rods²³ obtained by inductively coupled plasma (ICP). In this case, the ratio $\phi(AS)/\phi(SAXS)$ (open dots) is remarkably close to 1, showing excellent agreement between ICP and SAXS. The detailed calculation is given in the ESI.

We then perform a full fit to the SAXS data (Figure 3) with $\phi = \phi_R + \phi_S$ as obtained from the invariant estimation, while the ratio ϕ_S/ϕ_R and the nanorod aspect ratio *X* are fixed at the values extracted from the AS measurements. The fits are generally excellent, with only (R_S, σ_S) and (R_R, σ_R) as free parameters; see Table 2 for the values.

The AS model is only sensitive to the damping parameter *A* in the combination A/R_S , via the corrected dielectric constant (2) (The contribution of the transverse mode of the rods, which involves A/R_R , is much weaker than that of the spheres). Having determined R_S independently from the SAXS data, we are now in a position to evaluate *A* (see Figure 4). Its average value is about A = 0.35, to be compared with the literature values of 0.6 - 2 for gold spheres^{24–26}.



Figure 4 Interface damping parameter *A* defined in Equation (2). A/R_S is determined from the fits to the AS data in Figure 1, while R_S is extracted from the SAXS data (Figure 3).

3.3 TEM

The TEM images show the presence of two populations, separated by aspect ratio: elongated ($X \ge 2$) "nanorods" and isometric particles ($1 \le X \le 1.5$), which can be either rounded "nanospheres" or faceted "nanocubes".

In Figure 5 we compare the aspect ratio *X* and its relative dispersion σ_X/X for the nanorods, determined from the AS data (diamonds) and from the TEM images (bars).

Finally, in Figure 6 we compare the values for the radii R_R



Figure 5 Aspect ratio for the nanorods, as determined from the fits to the AS data in Figure 1 (diamonds) and from the TEM images (bars). The error bars (AS) and bar height (TEM) indicate the average value \pm standard deviation: $X \pm \sigma_X$.

and R_S of both particle populations, as obtained from SAXS and TEM. These graphs show that modelling the data obtained by the three different techniques provides values of the particle dimensions that are in good agreement.

The results of the three techniques (AS, SAXS and TEM) are summarized in Table 2 for all samples.

4 Simplified relations

Full-curve fitting of the AS data is done using a very fast and stable routine, implemented either in MATLAB (available on request from the first author, KS) or in IGOR PRO (available on request from the last author, DC). It is nevertheless useful to extract simplified relations among the relevant parameters.

From extensive simulations we thus obtained polynomial interpolations for the position and width of the LPP, $\lambda_{\max}(X, \varepsilon_X)$ and $W(X, \varepsilon_X)$, as well as for the converse relations $X(\lambda_{\max}, W)$ and $\varepsilon_X(\lambda_{\max}, W)$, where we introduced the relative polydispersity $\varepsilon_X = \sigma_X/X$. The latter relations, which are the most useful for evaluating the experimental results, are shown in Figure 7. If the parameters λ_{\max} and W are known, the aspect ratio and its polydispersity can be read directly from the two graphs in the Figure.

The complete relations, as well as formulas for the concentration can be found in the ESI.



Figure 6 Radii of the rods R_R and spheres R_S as determined from the fits to the SAXS data in Figure 3 (diamonds) and from the TEM images (shaded squares). The error bars (AS) and bar height (TEM) indicate the average value \pm standard deviation: $R_S \pm \sigma_S$ and $R_R \pm \sigma_R$.

5 The importance of polydispersity

In Figure 8 we compare the simplified model discussed above to the data in this paper and to results published by other groups for the dependence of λ_{max} on *X*.

For perfectly monodisperse systems, the dependence $\lambda_{\max}(X,0)$ (dotted line in Figure 8) is very close to that given by Yan *et al.*²⁹. However, a quantitative description of the experimental data requires accounting for the polydispersity: $\varepsilon_X \le 0.15$ for our experimental data (in good agreement with the values presented in Table 2), $\varepsilon_X \simeq 0.15$ for those of Al-Sherbini²⁷ and Link *et al.*⁸ and $\varepsilon_X \simeq 0.5$ for Pérez-Juste *et al.*²⁸. It is obvious from Figure 8 that a unique curve $\lambda_{\max}(X)$ cannot describe all the experimental data. We emphasize that the polydispersity correction is very significant: for instance, $\lambda_{\max}(7, 0.5) - \lambda_{\max}(7, 0.0) \simeq 180$ nm!

6 Conclusions

We characterized solutions of gold nanoparticles using three different techniques, summarized in Figure 9.

Our goal was to show that fitting the full AS spectrum al-



Figure 7 Aspect ratio *X* (left) and relative variance ε_X (right) of the nanorod population simulated by Equation (1) in the main text as a function of λ_{max} and *W* (for $R_R/A = 14.3 \text{ nm}$) represented as color code.



Figure 8 Experimental data for the dependence of λ_{max} on *X*, from several authors: Al-Sherbini²⁷ (Figure 1b) (\diamond), Link *et al.*⁸ (Figure 4, without reshaping) (\bullet), Pérez-Juste *et al.*²⁸ (Tables S1 and S2) (+) and the present work (\bullet). Predictions of the simplified model (see the ESI), for various relative polydispersities $\varepsilon_X = \sigma_X/X = 0$ (dotted line), 0.25 (solid line) and 0.5 (dashed line). For all curves, $R_R/A = 14.3$ nm.

lows retrieving detailed information about particle dimensions and their polydispersity in mixtures of gold nanospheres and nanorods. After comparison with the SAXS and TEM measurements and with other results in the literature we conclude that:

- The total volume fraction of particles $\phi = \phi_R + \phi_S$ inferred from the AS data overestimates by about 25% the SAXS results (which we consider as reliable).
- The aspect ratio of the nanorods and its dispersion (X, σ_X) are determined accurately by AS and validated by a comparison with the TEM data.
- The radii for both the rods and spheres (R_R, R_S) are measured by SAXS and confirmed by the TEM data. Combining *X* and R_R then yields a reliable estimate for the length *L*, which is not directly accessible via either AS or SAXS alone.
- The model for $\lambda_{\max}(X, \sigma_X)$ successfully describes the experimental results from various research groups; in particular, the polydispersity σ_X plays an important role.

From the AS model, we extracted polynomial approximations for *X*, σ_X , ϕ_R and ϕ_S as a function of the features of the AS spectrum (height, position and width of the longitudinal peak and height of the transverse peak). These formulas can be easily implemented in any data treatment software and should therefore provide a fast and simple way of characterizing mixtures of gold nanoparticles, both *ex situ*, but also during synthesis, to characterize the kinetics of particle formation and to optimize the preparation method.

Some points we have not considered in the present paper, but that would deserve development in future work are:

• The role of the finite size of the particles with respect to the wavelength. With increasing size, the electrostatic approximation becomes less and less accurate. Corrections are then needed to describe the position and amplitude of the plasmon peaks.



Figure 9 The three techniques at a glance for sample A2. (a) Selection from a TEM image. (b) TEM analysis: aspect ratio vs. radius for all analyzed particles, with histograms of both parameters. The solid lines are Schulz distributions with the average values and standard deviations given in Table 2. (c) AS data from Figure 1. (d) SAXS data from Figure 3.

- The influence of the organic layer at the interface between the particle and the medium. The nature of the ligands adsorbed onto the surface of the nanoparticles modifies their optical response: for homogeneous molecular layers, this effect might be described by a coated ellipsoid model.
- Extending the approach to other particle shapes, either rounded (dumbbells, dogbones) or faceted (cubes, prisms, plates). Changing the synthesis parameters leads to a wide variety of shapes, each of them exhibiting interesting features and potential applications. It would thus be extremely useful to extend the model in this direction.

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TEM	σZ	[uu]	1.0	1.2	1.5	1.3	1.3	1.7	2.6	2.8	2.2	3.2	3.3	1.9
	R_S	[mm]	9.0	8.9	7.0	6.5	6.5	8.0	9.9	10.7	8.6	12.7	13.2	8.6
	σ_R	[uu]	0.7	0.6	0.5	0.3	0.5	0.5	0.8	1.0	0.6	0.7	1.2	0.8
	R_R	[mm]	5.4	4.7	3.7	3.3	3.9	4.2	4.6	4.5	3.7	4.3	9.3	4.2
	σ_X	I	0.8	0.8	1.1	1.2	1.3	1.7	0.6	1.1	0.9	1.3	0.5	1.6
	X	I	3.4	5.6	5.9	4.8	5.4	6.0	3.9	5.4	4.5	6.7	2.5	6.9
	σ_S	[mm]	2.4	2.5	2.4	1.9	2.4	2.5	3.9	4.1	2.4	3.8	3.6	2.8
	R_S	[mm]	6.5	10.3	8.0	6.4	8.0	9.6	9.7	10.1	12.8	13.2	15.8	7.3
SAXS	σ_R	[mm]	0.7	0.3	0.2	0.2	0.2	0.6	0.6	0.8	0.8	1.1	3.5	0.7
• •	R_R	[mm]	6.7	6.0	4.6	4.0	4.4	4.6	5.8	4.9	4.2	4.6	7.0	4.3
	ϕ_{SAXS}^{d}	10^{-5} vol.	51.6	30.3	38.8	29.7	57.4	40.4	15.2	0.88	22.6	6.58	4.11	1.73
	$\phi_{AS}{}^{c}$	10^{-5} vol.	66.7	42.0	35.5	48.2	67.0	49.5	20.7	1.04	23.6	10.0	8.67	2.78
	ϕ_S	10^{-5} vol.	7.76	12.9	24.2	27.4	45.8	37.8	10.5	0.63	16.2	4.57	6.71	1.44
AS	ϕ_R	10^{-5} vol.	59.0	29.1	11.3	20.8	21.2	11.7	10.2	0.41	7.42	5.46	1.96	1.34
	$A^{\boldsymbol{b}}$	I	0.23	0.39	0.25	0.32	0.41	0.47	0.20	0.39	0.41	0.31	0.40	0.36
	σχ	I	0.83	0.46	0.63	0.79	0.74	0.76	0.39	0.74	0.68	0.97	0.55	1.54
	X	!	3.76	5.43	6.00	6.38	6.51	6.78	3.92	5.80	6.00	7.45	2.62	8.41
	$\lambda_{\max}{}^{a}$	[mm]	748	926	982	1022	1042	1072	780	982	066	1137	667	1216
Technique	Parameter	Unit	A1	A2	A3	A4	A5	A6	B1	B2	B3	B4	B5	B6

Table 2 Parameters obtained by the three techniques (AS, SAXS and TEM) for all samples.

^{*a*}Position of the longitudinal plasmon peak of the nanorods. ^{*b*}Determined using R_S (obtained by SAXS). ^{*c*}Total volume fraction: $\phi_{AS} = \phi_R + \phi_S$. ^{*d*}Determined from the invariant.

References

- [1] S. Eustis and M. A. El-Sayed, *Chemical Society Reviews*, 2006, **35**, 209–217.
- [2] M.-C. Daniel and D. Astruc, *Chemical Reviews*, 2004, 104, 293–346.
- [3] M. Grzelczak, J. Perez-Juste, P. Mulvaney and L. M. Liz-Marzan, *Chem. Soc. Rev.*, 2008, **37**, 1783–1791.
- [4] U. Kreibig and M. Vollmer, *Optical properties of metal clusters*, Springer, 1995.
- [5] R. Gans, Annalen der Physik, 1912, 342, 881–900.
- [6] L. Genzel, T. P. Martin and U. Kreibig, *Zeitschrift für Physik B Condensed Matter*, 1975, **21**, 339–346.
- [7] P. Mulvaney, Langmuir, 1996, 12, 788-800.
- [8] S. Link, M. B. Mohamed and M. A. El-Sayed, *The Journal of Physical Chemistry B*, 1999, **103**, 3073–3077.
- [9] V. Amendola and M. Meneghetti, *The Journal of Physical Chemistry C*, 2009, **113**, 4277–4285.
- [10] S. Link and M. A. El-Sayed, *The Journal of Physical Chemistry B*, 1999, **103**, 4212–4217.
- [11] A. Brioude, X. C. Jiang and M.-P. Pileni, *The Journal of Physical Chemistry B*, 2005, 109, 13138–13142.
- [12] S. Eustis and M. A. El-Sayed, *Journal of Applied Physics*, 2006, **100**, 044324.
- [13] F. Hubert, F. Testard, G. Rizza and O. Spalla, *Langmuir*, 2010, **26**, 6887–6891.
- [14] X. Ye, C. Zheng, J. Chen, Y. Gao and C. B. Murray, *Nano Letters*, 2013, 13, 765–771.
- [15] X. Ye, L. Jin, H. Caglayan, J. Chen, G. Xing, C. Zheng, V. Doan-Nguyen, Y. Kang, N. Engheta, C. R. Kagan and C. B. Murray, ACS Nano, 2012, 6, 2804–2817.
- [16] M. Daimon and A. Masumura, *Applied Optics*, 2007, 46, 3811–3820.
- [17] H. Hövel, S. Fritz, A. Hilger, U. Kreibig and M. Vollmer, *Phys. Rev. B*, 1993, **48**, 18178–18188.
- [18] M. Kotlarchyk and S.-H. Chen, *The Journal of Chemical Physics*, 1983, **79**, 2461–2469.
- [19] S. R. Kline, *Journal of Applied Crystallography*, 2006, 39, 895–900.
- [20] G. Porod, *Small Angle X-ray Scattering*, Academic Press, 1982, ch. 2.
- [21] S. Link, Z. L. Wang and M. A. El-Sayed, *The Journal of Physical Chemistry B*, 1999, 103, 3529–3533.
- [22] P. Mulvaney, M. Giersig and A. Henglein, *The Journal* of *Physical Chemistry*, 1992, **96**, 10419–10424.
- [23] C. J. Orendorff and C. J. Murphy, *The Journal of Physical Chemistry B*, 2006, **110**, 3990–3994.
- [24] M. Quinten, Zeitschrift f
 ür Physik B Condensed Matter, 1996, 101, 211–217.
- [25] M. M. Alvarez, J. T. Khoury, T. G. Schaaff, M. N.

Shafigullin, I. Vezmar and R. L. Whetten, *The Journal* of *Physical Chemistry B*, 1997, **101**, 3706–3712.

- [26] L. B. Scaffardi and J. O. Tocho, *Nanotechnology*, 2006, 17, 1309–1315.
- [27] A.-S. A.-M. Al-Sherbini, *Colloids and Surfaces A: Physicochemical and Engineering Aspects*, 2004, **246**, 61–69.
- [28] J. Pérez-Juste, B. Rodríguez-González, P. Mulvaney and L.-M. Liz-Marzán, *Advanced Functional Materials*, 2005, 15, 1065–1071.
- [29] B. Yan, Y. Yang and Y. Wang, *The Journal of Physical Chemistry B*, 2003, **107**, 9159–9159.

Supporting information for: Morphology of gold nanoparticles determined by full-curve fitting of the absorption spectrum. Comparison with X-ray scattering and electron microscopy data

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TEM data

Particle analysis

In Figure S1, we show the aspect ratio X as a function of the radius R for all the particles used in the analysis (each panel displays the objects from two synthesis batches).



Figure S1: Aspect ratio as a function of the radius for all the objects (both rods and spheres).

Particle volume and correlation with the aspect ratio

The theoretical model assumes that the volume V and the aspect ratio X of the rods are uncorrelated. In Figure S2 we plot $V = \pi R_R^2 L$ versus $X = L/(2R_R)$ for all the rods (symbols) and the best linear fit to V(X). The slope dV/dX is given in Figure S3. Although this parameter is significantly different from zero for most samples, its value is randomly distributed over the entire series. Setting it to zero (i.e. neglecting the correlation between V and X) is thus a reasonable pragmatic decision when aiming for a general model.



Figure S2: Estimated volume $V = \pi R_R^2 L$ as a function of the aspect ratio $X = L/(2R_R)$ for all the rods (various symbols) and linear fits to V(X). The fit slopes are shown in Figure S3.

AS model: simplified relations

From extensive simulations of the theoretical model (Equation (1) in the main text) we extracted simplified relations between the relevant parameters:

- the absorption spectrum is characterized by the position λ_{max} and amplitude H_{lp} of the longitudinal peak as well as its full width at half maximum W, and by the amplitude of the transverse peak H_{tp} .
- the nanoparticle suspensions are characterized by the aspect ratio of the nanorods



Figure S3: Slope dV/dX of the linear fits in Figure S2. The error bars indicate the fit uncertainty (one standard deviation).

X and its relative dispersion $\varepsilon_X = \sigma_X/X$, as well as the concentration of rods and spheres, ϕ_R and ϕ_S . The results also depend weakly on R_R/A and R_S/A , where R_R and R_S are the radii of the rods and spheres and A is the damping parameter defined in Equation (2) of the main text.

Each simulated spectrum was fitted with the following function:

$$A_{\lambda,fit}(\lambda) = B_1 \exp\left(-a_1\lambda\right) + B_2 \exp\left(-a_2\lambda\right) + \frac{A_1}{\sigma_1\lambda\sqrt{2\pi}} \exp\frac{\left(\ln\lambda - \mu_1\right)^2}{2\sigma_1^2} + \frac{A_2}{\sigma_2\lambda\sqrt{2\pi}} \exp\frac{\left(\ln\lambda - \mu_2\right)^2}{2\sigma_2^2} \tag{1}$$

where the wavelength λ is expressed in nm and B_1 , a_1 , B_2 , a_2 , A_1 , μ_1 , σ_1 , A_2 , μ_2 , σ_2 are adjustable parameters. The first and the third terms describe the peak at 518 nm (sum of the plasmonic resonance in the spheres and the transverse resonance in the rods) and the second and the fourth terms describe the longitudinal resonance in the rods. Parameters λ_{max} and W are obtained as:

$$\lambda_{\max} = \exp\left(\mu_2 - \sigma_2^2\right) \tag{2a}$$

$$W^{2} = 8 \log 2 \exp \left(2\mu_{2} + \sigma_{2}^{2}\right) \left[\exp\left(\sigma_{2}^{2}\right) - 1\right]$$
(2b)



Figure S4: Absorption spectrum simulated using Equation (1) in the main text and fit with the function $\gamma_{\rm fit}(\lambda)$ (1) for two aspect ratios: X = 4 (left) and X = 10 (right). The other parameters are identical for the two simulations: $\varepsilon_X = 0.1$, $\phi_R = \phi_S = 4.4 \cdot 10^{-7}$, $R_R/A = 14.3$ nm and $R_S/A = 100$ nm. The three terms in the fit function are also shown.

The dependence of λ_{max} and W (both in nm units) on X and ε_X for a fixed value $R_R/A = 14.3 \text{ nm}$ is shown in Figure S5. The polynomial interpolation is given by (3). The converse dependence (of X and ε_X on λ_{max} and W) is more useful in practice. It is presented in Figure S6 (also included in the main text as Figure 7) and the corresponding interpolation is given in (4).



Figure S5: Position λ_{max} (left) and width W (right) of the longitudinal peak of the nanorods simulated by Equation (1) in the main text as a function of X and ε_X (for $R_R/A = 14.3 \text{ nm}$) represented as color-coded level curves.

 $\lambda_{\max}(X,\varepsilon_X) = 394.4 + 98.69X + 0.07853X^2 + 134.4\varepsilon_X - 394.7\varepsilon_X^2 - 24.28\varepsilon_X X + 171\varepsilon_X^2 X$ (3a)

$$W(X, \varepsilon_X) = 29.5 + 9.269X + 0.2907X^2 + 307.9\varepsilon_X - 1407\varepsilon_X^2 + 148.9\varepsilon_X X + 725.7\varepsilon_X^2 X + 2.618\varepsilon_X X^2$$
(3b)

Figure S6: Aspect ratio X (left) and relative variance ε_X (right) of the nanorod population simulated by Equation (1) in the main text as a function of λ_{max} and W (for $R_R/A = 14.3 \text{ nm}$) represented as color code. Reproduces Figure 7 in the main text.

$$X(\lambda_{\max}, W) = -8.155 + 0.060\lambda_{\max} - 0.6061W - 0.000109\lambda_{\max}^{2} + 0.0013\lambda_{\max}W - 0.00056W^{2} + 5.08 \cdot 10^{-8}\lambda_{\max}^{3} - 5.69 \cdot 10^{-7}\lambda_{\max}^{2}W + 3.74 \cdot 10^{-7}\lambda_{\max}W^{2} - 1.93 \cdot 10^{-8}W^{3}$$

$$\varepsilon_{X}(\lambda_{\max}, W) = 1.408 - 0.2256\lambda_{\max} + 0.9768W + 0.00041\lambda_{\max}^{2} - 0.00197\lambda_{\max}W + 0.000704W^{2} - 1.73 \cdot 10^{-7}\lambda_{\max}^{3} + 8.89 \cdot 10^{-7}\lambda_{\max}^{2}W - 5.43 \cdot 10^{-7}\lambda_{\max}W^{2} + 7.097 \cdot 10^{-8}W^{3}$$
(4a)
(4b)

For a given morphology, the height of the longitudinal peak H_{lp} is proportional to the volume concentration of the nanorods ϕ_R . We can write the latter as:

$$\phi_R = 10^{-7} \frac{H_{\rm lp}}{H_{\rm lp}^{\star}},\tag{5}$$

where H_{lp}^{\star} is the amplitude of the longitudinal peak simulated by the model for $\phi_R = 10^{-7}$ (with $R_R/A = 14.3$ nm), see Figure S7 and is interpolated by the polynomials (6). We emphasize that both H_{lp} and H_{lp}^{\star} are *amplitudes*, i.e. the heights of the peaks with respect to their backgrounds, and not the raw peak values. The distinction is particularly relevant for small aspect ratio, see e.g. Figure S4.

$$H_{lp}^{\star}(X,\varepsilon_{X}) = -0.03127 + 0.05944X - 0.05944\varepsilon_{X} - 0.002907X^{2} - 0.241X\varepsilon_{X}$$

+ $5.427\varepsilon_{X}^{2} + 7.765 \cdot 10^{-5}X^{3} + 0.003201X^{2}\varepsilon_{X} + 0.3961X\varepsilon_{X}^{2} - 12.18\varepsilon_{X}^{3}$ (6a)
 $H_{lp}^{\star}(\lambda_{\max},W) = 0.1748 + -0.001527\lambda_{\max} + 0.001773W + 3.541 \cdot 10^{-6}\lambda_{\max}^{2}$
- $6.213 \cdot 10^{-6}\lambda_{\max}W + 1.943 \cdot 10^{-6}W^{2} - 2.376 \cdot 10^{-9}\lambda_{\max}^{3} + 3.627 \cdot 10^{-9}\lambda_{\max}^{2}W$
+ $1.712 \cdot 10^{-9}\lambda_{\max}W^{2} - 2.815 \cdot 10^{-9}W^{3} + 5.515 \cdot 10^{-13}\lambda_{\max}^{4} - 7.542 \cdot 10^{-13}\lambda_{\max}^{3}W$
- $7.711 \cdot 10^{-13}\lambda_{\max}^{2}W^{2} + 7.116 \cdot 10^{-13}\lambda_{\max}W^{3} + 3.672 \cdot 10^{-13}W^{4}$ (6b)

The volume concentration of the spheres is:

$$\phi_S = 3.28 \cdot 10^{-6} \left[H_{\rm tp} - H_{\rm tp}^{\star} \, 10^7 \cdot \phi_R \right] = 3.28 \cdot 10^{-6} \left[H_{\rm tp} - H_{\rm tp}^{\star} \frac{H_{\rm lp}}{H_{\rm lp}^{\star}} \right],\tag{7}$$

where ϕ_R is given by (5) and with:

$$H_{\rm tp}^{\star} = 0.012 + 0.0035 \cdot X^{-1} + 0.05 \cdot X^{-2} \tag{8}$$

and $H_{\rm tp}$ is the absorbance at the position of the transverse peak (518 nm).

Figure S7: Amplitude $H_{\rm lp}^{\star}$ of the longitudinal peak for $\phi_R = 10^{-7}$. Left: as a function of (X, ε_X) , from Equation (6a) (color-coded level curves). Right: as a function of $(\lambda_{\rm max}, W)$, from Equation (6b) (color code).

So far, we have set $R_R/A = 14.3$ nm, a value close to the experimental results. Furthermore, λ_{max} and W only weakly depend on this parameter. For completeness, we give below formulas for this dependence:

$$\lambda_{\max} = \lambda_{\max}^{14.3} (1.003 - 0.045 \, A/R_R) \tag{9a}$$

$$W = W^{14.3}(0.880 + 1.694 \, A/R_R) \tag{9b}$$

where $\lambda_{\text{max}}^{14.3}$ and $W^{14.3}$ are the values for $R_R/A = 14.3 \text{ nm}$ obtained in Equations (3).

Alternative estimate for the concentration

Spheres The literature value for the atomic extinction coefficient of spherical gold nanoparticles (at the wavelength of the plasmon peak) is $\epsilon_{\rm S} = 3900 - 4000 \,\mathrm{M^{-1} cm^{-1}}$.^{S1,S2} The corresponding absorbance is $A_{\rm S} = lc_{\rm Au}\epsilon_{\rm S}$, where $c_{\rm Au}$ is the molarity of gold atoms and l the optical path through the sample (1 cm in our experiments). The volume concentration of spheres then reads:

$$\phi_S = \frac{c_{\rm Au} M_{\rm Au}}{10^3 \rho_{\rm Au}} = \frac{A_{\rm S} M_{\rm Au}}{10^3 l \epsilon_{\rm S} \rho_{\rm Au}},\tag{10}$$

where $M_{\rm Au} = 197 \,\mathrm{g}$ is the molar mass and $\rho_{\rm Au} = 19.3 \,\mathrm{g/cm^3}$ is the density of gold.

Rods In the case of nanorods, the absorbance depends on the molar extinction coefficient of the particles as $A_{\rm R} = lc_{\rm Au}\epsilon_{\rm R}$ and the formula for the volume fraction is in all points similar to (10), except for the absorbance being evaluated at the position of the longitudinal plasmon resonance:

$$\phi_R = \frac{c_{\rm Au} M_{\rm Au}}{10^3 \rho_{\rm Au}} = \frac{A_{\rm R} M_{\rm Au}}{10^3 l \epsilon_{\rm R} \rho_{\rm Au}},\tag{11}$$

Here, however, the extinction coefficient $\epsilon_{\rm R}$ depends on the aspect ratio of the particles. Fitting the literature data^{S3} to a straight line yields the approximation:

$$\epsilon_{\rm R} = (7263 \, X - 6224) \, {\rm M}^{-1} {\rm cm}^{-1} \tag{12}$$

By extrapolating this dependence to the aspect ratio of our nanorods (taken from the TEM data) we obtain ϕ_R . The sum of ϕ_S and ϕ_R , normalized by the value ϕ_{SAXS} obtained from the SAXS data is plotted as open dots in Figure 2 of the main text.

The dielectric constant of bulk gold

In describing the absorbance of the nanoparticles, the dielectric constant of bulk gold $\epsilon_{\text{bulk}}(\lambda) = \epsilon_1(\lambda) + i\epsilon_2(\lambda)$ is a fundamental ingredient. Several sets of data were published, ^{S4–S8} showing

significant differences, in particular in the imaginary component ϵ_2 , which is strongly affected by the presence of defects (Figure S8).

Figure S8: Wavelength dependence of the complex dielectric constant of gold, as given by several authors (lines and symbols): a – real part $\epsilon_1(\lambda)$; b – imaginary part $\epsilon_2(\lambda)$. The values used in this work are shown as solid green line and correspond to model (13) with parameters given in Table S1

A very convenient analytical model, combining the Drude-Lorentz contribution and two interband transitions (DLBB) was proposed by Etchegoin et al.^{S9},^{S10}. For instance, with appropriate values for the parameters, the model describes very well the data of Johnson and Christy^{S5}. We therefore use it in the following. The model is given in Equation (13) and the parameters in Table S1.

$$\epsilon_{\text{bulk}}(\lambda) = \epsilon_{\infty} - \frac{1}{\lambda_p^2 [1/\lambda^2 + i/(\gamma_p \lambda)]} + \sum_{1,2} \frac{A_i}{\lambda_i} \left[\frac{\exp(i\varphi_i)}{(1/\lambda_i - 1/\lambda - i/\gamma_i)} + \frac{\exp(-i\varphi_i)}{(1/\lambda_i + 1/\lambda + i/\gamma_i)} \right]$$
(13)

We chose the parameter values such that $\epsilon_2(\lambda)$ is roughly in the middle of the experimental points from various authors (Figure S8b) and $\epsilon_1(\lambda)$ agrees with the experimental data below

Table S1: Parameter values for the analytical DLBB model (13), for the data of Johnson and Christy^{S5} (from S9 and S10) and those used in this paper. The λ and γ are given in nanometres, while φ_1 and φ_2 are in radians. Values modified in our treatment with respect to S5 are given in bold.

Parameter	ϵ_{∞}	λ_p	γ_p	A_1	φ_1	λ_1	γ_1	A_2	φ_2	λ_2	γ_2
Johnson and Christy ^{S5}	1.54	143	14500	1.27	$-\pi/4$	470	1900	1.1	$-\pi/4$	325	1060
This work	4	146	25000	1	$-\pi/4$	470	1900	0.5	$-\pi/4$	325	1060

500 nm (Figure S8a). Above this value, the behaviour of $\epsilon_1(\lambda)$ is mainly determined by λ_p , which we fix at 146 nm, rather than the 143 nm given by Johnson and Christy^{S5}, by calibrating our model against recent data^{S11} for the dependence of λ_{max} on the aspect ratio X of gold nanorods (see Figure S9).

Figure S9: Experimental data for the dependence of λ_{max} on X, from Ye *et al.*, ^{S11} Figure S27b (\diamond). The error bars represent the variance σ_X . Linear fit (dotted line) and prediction of our model (Equation 1 in the main text) with $R_R/A = 14.3 \text{ nm}$, for $\lambda_p = 146 \text{ nm}$ (solid line) and for $\lambda_p = 143 \text{ nm}$ (dashed line).

References

(S1) Link, S.; Wang, Z. L.; El-Sayed, M. A. Alloy Formation of Gold-Silver Nanoparticles and the Dependence of the Plasmon Absorption on Their Composition. *The Journal* of Physical Chemistry B 1999, 103, 3529–3533.

- (S2) Mulvaney, P.; Giersig, M.; Henglein, A. Surface chemistry of colloidal gold: deposition of lead and accompanying optical effects. *The Journal of Physical Chemistry* **1992**, 96, 10419–10424.
- (S3) Orendorff, C. J.; Murphy, C. J. Quantitation of metal content in the silver-assisted growth of gold nanorods. The Journal of Physical Chemistry B 2006, 110, 3990–3994.
- (S4) Irani, G.; Huen, T.; Wooten, F. Optical constants of silver and gold in the visible and vacuum ultraviolet. JOSA 1971, 61, 128–129.
- (S5) Johnson, P. B.; Christy, R. W. Optical constants of the noble metals. *Phys. Rev. B* 1972, 6, 4370.
- (S6) Palik, E. D. Handbook of Optical Constants of Solids; Academic Press, 1991.
- (S7) Ordal, M. A.; Long, L. L.; Bell, R. J.; Bell, S. E.; Bell, R. R.; Alexander, R. W., Jr.;
 Ward, C. A. Optical properties of the metals Al, Co, Cu, Au, Fe, Pb, Ni, Pd, Pt, Ag,
 Ti, and W in the infrared and far infrared. *Applied Optics* 1983, 22, 1099.
- (S8) Thèye, M.-L. Investigation of the Optical Properties of Au by Means of Thin Semitransparent Films. *Physical Review B* 1970, 2, 3060–3078.
- (S9) Etchegoin, P. G.; Le Ru, E. C.; Meyer, M. An analytic model for the optical properties of gold. *The Journal of Chemical Physics* **2006**, *125*, 164705.
- (S10) Etchegoin, P. G.; Le Ru, E. C.; Meyer, M. Erratum: "An analytic model for the optical properties of gold" [J. Chem. Phys.125, 164705 (2006)]. The Journal of Chemical Physics 2007, 127, 189901.
- (S11) Ye, X.; Jin, L.; Caglayan, H.; Chen, J.; Xing, G.; Zheng, C.; Doan-Nguyen, V.; Kang, Y.; Engheta, N.; Kagan, C. R.; Murray, C. B. Improved Size-Tunable Synthesis of Monodisperse Gold Nanorods through the Use of Aromatic Additives. ACS Nano 2012, 6, 2804–2817.