

Sources and input mechanisms of hafnium and neodymium in surface waters of the Atlantic sector of the Southern Ocean

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Sources and input mechanisms of hafnium and neodymium in surface waters of the Atlantic sector of the Southern Ocean

Torben Stichel^{a,b,*}, Martin Frank^a, Jörg Rickli^c, Ed C. Hathorne^a, Brian A. Haley^d, Catherine Jeandel^e, Catherine Pradoux^e

^a GEOMAR, Helmholtz Centre for Ocean Research Kiel, Wischhofstr. 1-3, 24148 Kiel, Germany

^b SOEST, University of Hawaii at Manoa, 1680 East-West Rd., Honolulu, HI 96822, USA

^d COAS, Oregon State University, 104 Ocean Admin. Bldg., Corvallis, OR 97331, USA

^e LEGOS (UMR 5566 CNRS, CNES, IRD, UPS) Observatoire Midi-Pyrénées, 14 Avenue Edouard Belin, 31400 Toulouse, France

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13 Abstract

14 Radiogenic isotopes of hafnium (Hf) and neodymium (Nd) are powerful tracers for water mass transport and trace metal cycling in the present and past oceans. However, due to the scarcity of available data the processes governing their distribu-15 16 tion are not well understood. Here we present the first combined dissolved Hf and Nd isotope and concentration data from 17 surface waters of the Atlantic sector of the Southern Ocean. The samples were collected along the Zero Meridian, in the Wed-18 dell Sea and in the Drake Passage during RV Polarstern expeditions ANTXXIV/3 and ANTXXIII/3 in the frame of the Inter-19 national Polar Year (IPY) and the GEOTRACES program. The general distribution of Hf and Nd concentrations in the 20 region is similar. However, at the northernmost station located 200 km southwest of Cape Town a pronounced increase of 21 the Nd concentration is observed, whereas the Hf concentration is minimal, suggesting much less Hf than Nd is released 22 by the weathering of the South African Archean cratonic rocks. From the southern part of the Subtropical Front (STF) 23 to the Polar Front (PF) Hf and Nd show the lowest concentrations (<0.12 pmol/kg and 10 pmol/kg, respectively), most prob-24 ably due to the low terrigenous flux in this area and efficient scavenging of Hf and Nd by biogenic opal. In the vicinity of 25 landmasses the dissolved Hf and Nd isotope compositions are clearly labeled by terrigenous inputs. Near South Africa Nd isotope values as low as $\varepsilon_{Nd} = -18.9$ indicate unradiogenic inputs supplied via the Agulhas Current. Further south the iso-26 topic data show significant increases to $\varepsilon_{\rm Hf} = 6.1$ and $\varepsilon_{\rm Nd} = -4.0$ documenting exchange of seawater Nd and Hf with the Antarctic Peninsula. In the open Southern Ocean the Nd isotope compositions are relatively homogeneous ($\varepsilon_{\rm Nd} \sim -\frac{8}{10}$ to $-\frac{8.5}{100}$) 27 28 towards the STF, within the Antarctic Circumpolar Current, in the Weddell Gyre, and the Drake Passage. The Hf isotope 29 30 compositions in the entire study area only show a small range between $\varepsilon_{Hf} = +6.1$ and +2.8 support Hf to be more readily 31 released from young mafic rocks compared to old continental ones. The Nd isotope composition ranges from $\varepsilon_{Nd} = -18.9$ to 32 -4.0 showing Nd isotopes to be a sensitive tracer for the provenance of weathering inputs into surface waters of the Southern 33 Ocean.

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1. INTRODUCTION

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1.1. Hafnium and neodymium isotopes as tracers in seawater 37

* Corresponding author at: SOEST, University of Hawaii at Manoa, 1680 East-West Rd., Honolulu, HI 96822, USA *E-mail address:* tstichel@hawaii.edu (T. Stichel).

0016-7037/\$ - see front matter @ 2012 Published by Elsevier Ltd. http://dx.doi.org/10.1016/j.gca.2012.07.005 Dissolved radiogenic hafnium (Hf) and neodymium 38 (Nd) isotope compositions (expressed as ε_{Hf} and ε_{Nd} by 39

^c Department of Earth Sciences, University of Bristol, Wills Memorial Building, Bristol BS8 1RJ, UK

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normalizing the ¹⁷⁶Hf/¹⁷⁷Hf and the ¹⁴³Nd/¹⁴⁴Nd to that of 40 CHUR in parts per 10,000 (Jacobsen and Wasserburg, 41 1980; Nowell et al., 1998)) in seawater have been shown 42 43 to provide insights into present and past ocean circulation and weathering conditions on land (e.g. Piotrowski et al., 44 45 2000; Frank et al., 2002; van de Flierdt et al., 2002). The global average residence time of Nd in the ocean is between 46 47 360 and 1500 years, allowing the use of Nd isotopes as a 48 tracer for advection and mixing of water masses over long 49 distances (Jeandel et al., 1995; Tachikawa et al., 1999; Sid-50 dall et al., 2008; Arsouze et al., 2009; Rempfer et al., 2011). 51 Hafnium, on the other hand, has an oceanic residence time. 52 which is most likely on the order of only few hundred years (Rickli et al., 2009; Zimmermann et al., 2009b), implying 53 54 that radiogenic Hf isotopes are useful tracers for water 55 masses restricted to basin scales. However, there are also 56 estimates for the residence time of Hf in the deeper ocean longer than that of Nd, whereas in the surface ocean the 57 58 opposite was proposed (Godfrey et al., 2008, 2009). This 59 discrepancy is mainly due to the lack of information avail-60 able on the distribution of dissolved Hf in seawater and the 61 isotopic range of the input sources. While the Nd input into 62 the ocean is known to be governed by river and groundwa-63 ter contributions, as well as exchange between sediments and seawater (Elderfield et al., 1990; Greaves et al., 1994; 64 65 Jeandel et al., 1995; Haley et al., 2004; Lacan and Jeandel, 66 2005), the inputs contributing to the marine Hf budget are 67 still debated. Hafnium in seawater is possibly delivered mainly by fluvial input (e.g. Bayon et al., 2006; Godfrey 68 69 et al., 2008) but also partial dissolution of dust (Albarède et al., 1998; David et al., 2001; Pettke et al., 2002; Godfrey 70 et al., 2009; Rickli et al., 2010), exchange with marine sed-71 02 72 iments similar to Nd (Lacan and Jeandel, 2005), and hydrothermal inputs (Bau and Koschinsky, 2006) may play a 73 74 role.

Comparison of the relationship between the Hf and Nd 75 76 isotopes in terrestrial rocks, referred to as the "terrestrial 77 array" (Patchett et al., 1984; Vervoort et al., 1999), with available seawater data (Godfrey et al., 2009; Rickli et al., 78 2009, 2010; Zimmermann et al., 2009a,b) reveals fundamen-79 tal differences (Fig. 1A). The Hf and Nd isotope composi-80 tion of terrestrial rocks is governed by a very similar 81 lutetium-hafnium (Lu/Hf) and samarium-neodymium 82 83 (Sm/Nd) elemental fractionation during igneous processes. The radiogenic isotope systems of Hf and Nd are controlled 84 by the decay of their respective parent isotopes ¹⁷⁶Lu and 85 ¹⁴⁷Sm: Continental crustal rocks are thus characterized by 86 87 low Lu/Hf and low Sm/Nd leading to unradiogenic bulk Hf and Nd isotope compositions (low ε-values) over time. 88 Mafic rocks (i.e. mantle-derived rocks) have higher Lu/Hf 89 90 and Sm/Nd and are therefore more radiogenic (higher *e-va-*91 lue) in their bulk Hf and Nd isotope compositions. Compared to the terrestrial array, the available seawater Hf 92 93 and Nd isotope data ("seawater array") are shifted to more 94 radiogenic Hf isotope compositions for a given Nd isotope 95 composition. The most plausible explanation is that during 96 incongruent weathering of continental crust, unradiogenic 97 Hf is retained in weathering resistant minerals, such as zir-98 cons, a process often referred to as the "zircon effect" 99 (Patchett et al., 1984; Albarède et al., 1998; Piotrowski

et al., 2000; van de Flierdt et al., 2002, 2007). In addition, 100 minerals with highly radiogenic Hf isotope composition 101 have been shown to be more susceptible to weathering, thus 102 causing preferential supply of radiogenic Hf to seawater 103 (Bayon et al., 2006, 2009; van de Flierdt et al., 2007; Chen 104 et al., 2011). Rickli et al. (2010) suggested that areas with 105 high atmospheric dust loads, such as the Sahara region, 106 also act as sources of Hf to seawater. Comparing the sur-107 face ocean Hf isotope compositions to those of dust indi-108 cates that Hf is preferentially released from traces of 109 radiogenic apatite and clays resulting in a surface water 110 Hf isotope composition, which is $\sim 10 \varepsilon_{\rm Hf}$ units more radio-111 genic than the bulk dust composition in the Atlantic Ocean. 112 Therefore, both fluvial and atmospheric inputs are likely to 113 contribute to the radiogenic Hf isotope composition in the 114 ocean. An alternative explanation, although so far not sup-115 ported by direct data, is the input of radiogenic Hf through 116 hydrothermal venting (Bau and Koschinsky, 2006; White 117 et al., 1986). This source has been shown not to deliver 118 any radiogenic Nd to deep waters (German et al., 1990; 119 Halliday et al., 1992) and thus could also explain the ob-120 121 served offset to more radiogenic Hf isotope compositions of seawater. 122

Despite of the existence of Hf and Nd isotope data from 123 different parts of the global ocean the continental contribu-124 tions and their supply pathways are still largely uncon-125 strained, in particular in the Southern Ocean. In addition, 126 the Southern Ocean is characterized by a high biogenic opal 127 productivity, which may play an important role for the cy-128 cling of Hf and the rare earth elements (REE). Here we 129 present the first combined surface water Hf and Nd isotope 130 compositions and concentrations complemented by REE 131 distributions from the Atlantic sector of the Southern 132 Ocean. With this new dataset we aim at better understand-133 ings of the external inputs of Hf and Nd and of the mech-134 anisms, which control their elemental and isotopic 135 distribution in the surface ocean. Comparison of REE pat-136 terns from surface waters with those of potential sources al-137 low the processes governing their distribution to be 138 evaluated. 139

1.2. Study area

The surface waters of the Southern Ocean (Fig. 2) are 141 separated from those of the subtropical region further 142 north by a pronounced meridional temperature and salinity 143 gradient. This hydrographic boundary is referred to as the 144 Subtropical Front (STF, Hofmann, 1985). The STF is the 145 only front in the Southern Ocean that is not circumpolar, 146 given that no subtropical waters are found in the Drake 147 Passage between the Antarctic Peninsula and South Amer-148 ica (Sievers and Emery, 1978; Sievers and Nowlin, 1984). 149 The STF marks the transition between the relatively cold 150 and fresh Subantarctic Surface Water (SASW) and the 151 much warmer and saltier Subtropical Surface Water 152 (STSW). The front develops in the western South Atlantic 153 Ocean and is associated with an eastward transport of up 154 to 37 Sv (1 Sv = $10^6 \text{ m}^3 \text{s}^{-1}$). This flow decreases to less than 155 15 Sv towards South Africa, commonly called the South 156 157 Atlantic Current (SAC) where it turns north to feed the

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Fig. 1. Hf and Nd isotope compositions of terrestrial rocks ("terrestrial array"), seawater and ferromanganese crusts ("seawater array"). The terrestrial array (Vervoort et al., 1999) is characterized by a larger ε_{Hf} variability than that of ε_{Nd} (A). The seawater array displays more radiogenic ε_{Hf} for a given ε_{Nd} value (B). Atlantic Hf isotope compositions (Godfrey et al., 2009; Rickli et al., 2009, 2010) plot at the unradiogenic end of the array, whereas Pacific compositions (Zimmermann et al., 2009a) form the radiogenic part. The Hf and Nd isotope compositions of Southern Ocean seawater as well as of ferromanganese crusts and nodules (van de Flierdt et al., 2006) plot between Atlantic and Pacific compositions.

158 Benguela Current (Stramma and Peterson, 1990). South of 159 the STF, a region of weak flow separates the SAC from the 160 Antarctic Circumpolar Current (ACC), which dominates 161 geostrophic eastward flow around the Antarctic continent. 162 The ACC is bound to the south by cyclonically circulating 163 gyres, such as the Weddell Gyre (WG, Deacon, 1979; Klatt et al., 2005). Within the ACC, bands of pronounced hori-164 165 zontal density gradients, also associated with strong lateral 166 currents, characterize the Subantarctic (SAF) and the Polar (PF) Fronts (Emery, 1977; Nowlin and Whitworth, 1977; 167 Whitworth, 1980; Peterson and Stramma, 1991). From 168 169 the PF to the Antarctic continent the relatively homoge-170 nous Antarctic Surface Water (AASW) is found, which 171 forms the topmost layer above the warmer and saltier 172 Upper (U) and Lower (L) Circumpolar Deep Water 173 (CDW).

174 The Atlantic sector of the Southern Ocean is bounded 175 by the landmasses of South Africa in the east, the Antarctic 176 continent in the south and South America in the west 177 (Fig. 2). The geology of South Africa mainly comprises 178 the Archaean Kaapvaal Craton, which is surrounded by 179 several orogenic belts of Proterozoic and Paleozoic age. 180 On top of this old basement the sedimentary Karoo Basin 181 developed during the Paleozoic (Cole, 1992). The geological 182 structure of the Antarctic continent (Fig 2A) can be broadly 183 divided into the three tectonic domains comprising the East Antarctic Shield (EAS), the Transantarctic Mountains 184 (TAM) and West Antarctica (WA). The EAS comprises a 185 Precambrian to Ordovician igneous and sedimentary base-186 ment overlain by Devonian to Jurassic sediments. East Ant-187 arctica, in particular the Dronning Maud Land (DML), 188 consists of approximately the same tectonic facies as South 189 Africa, in both cases originating from the Pan-African 190 orogeny which formed during the collision of East and 191 West Gondwana about 600-500 Ma ago (Thomas et al., 192 1993; Rogers et al., 1994). The TAM mark the boundary 193 between East and West Antarctica. They consist of sedi-194 mentary deposits from the Late Cambrian Ross orogeny, 195 Devonian to Permian sediment covers of Gondwana, and 196 Jurassic tholeiites. The TAM underwent uplift since the 197 early Cretaceous (Dalziel, 1992; Fitzsimons, 2000). West 198 Antarctica has five crustal blocks, of which three are rele-199 vant for this study: the Haag Nunataks (HN), the Ellsworth 200 Whitmore Mountains (EWM) and the Antarctic Peninsula 201 (AAP). The HN represents a small crustal block between 202 the AAP and the EWM consisting of Proterozoic amphib-203 olites and orthogneisses (Millar and Pankhurst, 1987). The 204 EWM form a 415 km long, NNE-SSW trending mountain 205 chain being a terrane once situated at the margin of 206 Gondwana prior to its break-up during the Jurassic 207 (Dalziel, 1992; Storey, 2005). The AAP mainly consists of 208 magmatic rocks of varying origin and is interpreted as a 209

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Fig. 2. Overview of the study area. The upper map (A) shows the bedrock configuration of Antarctica together with the major shelf ice contours (grey areas). The main geologic terrains are displayed: Antarctic Peninsular (AAP), East Antarctic Shield (EAS), Ellsworth Whitmore Mountains (EWM), Dronning Maud Land (DML), Haag Nunataks (HN), Transantarctic Mountains (TAM) and West Antarctica (WA). The map was constructed using GeoMapApp[©] (http://www.geomapapp.org) with the topography after Ryan et al. (2009). (B) Shows the sample sites and the hydrographic surface conditions and (C) is an enlargement of the area near the Antarctic Peninsula and King George Island (KGI). The main hydrographic features are the Agulhas Current (AC), the Antarctic Circumpolar Current (ACC), the Benguela Current (BC), the South Atlantic Current (SAC) and the Weddell Gyre (WG). The dashed and dotted lines in (B) and (C) represent the approximate locations of the Southern Ocean frontal system from north to south: the Subtropical Front (STF, the Subantarctic Front (SAF), the Polar Front (PF) and the Southern ACC boundary (BDY) as described in Orsi et al. (1995).

210 long-lived magmatic arc partly built on Cambrian continen211 tal crust (Millar et al., 2001). Most of the exposed rocks,
212 however, are of Mesozoic and younger ages resulting from
213 subduction on the western margin of the Peninsula. These
214 consist of the AAP Batholith, which reflects central arc

magmatism, and back-arc sequences on the eastern side215of the Peninsula bounded by the Weddell Sea. Across the216Drake Passage to the north these back-arc sequences con-
tinue as the southern part of South America (Patagonia).218Together with the AAP, Patagonia forms one of the largest
igneous provinces on Earth (Pankhurst et al., 1998).220

2. METHODS 221

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2.1. Sample collection and onboard procedures

Most samples were collected during the International 223 Polar Year (IPY) cruise ANTXXIV/3 on the German re-224 search vessel FS Polarstern from February to April 2008. 225 This cruise was also part of the international GEOTRAC-226 ES program and all samples were collected and measured 227 for their Nd and Hf concentrations and isotope composi-228 tions according to the agreed GEOTRACES protocols 229 (http://www.obs-vlfr.fr/GEOTRACES/libraries/documents/ 230 Intercalibration/Cookbook.pdf). Between 100 and 130 L of 231 seawater were taken for each sample mostly when the ship 232 233 was underway. The collection was carried out in three different ways: (1) using a towed stainless steel fish with a Tef-234 lon head, through which surface water was pumped directly 235 into the ship's laboratory under trace metal clean condi-236 tions, (2) using the ship's own seawater intake system made 237 of polypropylene tubes, or (3) using Niskin bottles on a 238 CTD rosette collecting water from 25 to 480 m water depth 239 during station time (Table 1 and Fig. 2). For this study we 240 refer to samples shallower than 100 m as surface samples. 241 All samples were filtered through a 142 mm MilliPore[®] fil-242 ter disc with a pore size of 0.45 µm into acid cleaned LDPE 243 cubitainers. Filtration was performed within 12 h after col-244 lection or directly during collection when possible. The 245 samples were then acidified to pH \sim 2 with double distilled, 246 concentrated HNO₃. From each sample an aliquot of 2 L 247 was kept for concentration analysis. About 500-600 mg 248 iron (Fe), in the form of previously cleaned dissolved Fe-249 chloride, was added to the large volume samples and left 250 for 24 h for equilibration. In a second step, ammonia solu-251 tion (25%, Merck Suprapur®) was added to bring the pH 252 back to 8 in order to co-precipitate the dissolved trace met-253 als (i.e. Hf and Nd) with Fe hydroxide (FeOOH). The 254 supernatant was discarded and the residual FeOOH precip-255 itate was transferred into pre-cleaned 2 L LDPE-bottles. 256

In addition, five Nd samples collected in the Drake Pas-257 sage during the expedition ANTXXIII/3 on Polarstern were 258 included in this study (see Table 1). For each of these sam-259 ples 10 L of seawater were filtered and acidified as described 260above. The pre-concentration was realized by co-precipita-261 tion with manganese hydroxide (MnOOH) after increasing 262 the pH to ~ 8 . The MnOOH precipitate was filtered on 263 board in a Taminar flow hood and the filters were stored 264 away for further analysis on shore. 265

2.2. Further procedures in the home laboratory

The FeOOH precipitates obtained during ANTXXIV/3 267 were centrifuged in the home laboratory of GEOMAR in 268 Kiel (Germany) and were rinsed at least twice with de- 269

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270 ionized water (MilliQ, 18.2 M) to wash out major ions. The 271 precipitate was removed from the centrifuge tubes and 272 transferred into 60 ml Teflon vials with 5 ml 6 M HCl 273 (quartz distilled). After evaporation, 4 ml of aqua regia 274 were added and the closed vials were heated at 110 °C for 275 at least 24 h to destroy organic compounds. The samples 276 were then transformed to Cl-form and finally dissolved in 277 4 ml of 6 M HCl. The majority (more than 99%) of the 278 Fe was removed prior to chromatographic separation by solvent extraction with di-ethyl ether (Stichel et al., 2014). 279 Q3 280 The residual was then evaporated and re-dissolved in 4 ml 281 3 M HCl.

282 Most of the samples formed a jelly-like silica gel in HCl, which scavenged about 90% of the Hf from the sample. 283 284 This silica gel was isolated and dissolved with 2 M HF. 285 After evaporation of the 2 M HF, this fraction was recom-286 bined with the HCl soluble fraction of the same sample. 287 The purification of Hf and separation of REE followed a 288 modified recipe of Münker et al. (2001). Some remaining traces of ytterbium (Yb), which would have caused an iso-289 290 baric interference on ¹⁷⁶Hf, were subsequently removed on 291 an additional column loaded with 1.4 ml of cation-exchange resin (Biorad[®] AG50W-X8, 200-400 µm mesh-size) 292 293 using 1 M HCl and 0.05 M HF to elute Hf before the rare 294 earth elements (REE). Purification after Münker et al. 295 (2001) also allowed the separation of the REE from most 296 of the sample matrix. The REE were eluted in 6 M HCl 297 immediately after loading the sample. The REE cut was 298 further processed using LN-Spec resin following Pin and 299 Zalduegui (1997). The respective Hf and Nd cuts were treated with 100 µl HNO₃ (concentrated, guartz distilled) and 300 100 µl H₂O₂ (30 wt.%, Merck Suprapur[®]) in preparation 301 302 for mass spectrometry to avoid potentially disturbing ma-303 trix effects by traces of resin or other residual organic com-304 pounds. Finally the samples were dissolved in 0.5 M HNO3 (Nd) and $0.5 \text{ M HNO}_3 + 0.1 \text{ M HF}$ (Hf) for measurement 305 306 on the MC-ICPMS.

307 The filtered MnOOH precipitates from the additional samples from expedition ANTXXIII/3 were dissolved and 308 further treated at LEGOS in Toulouse (France). Extraction 309 of the "REE fraction" from the MnOOH co-precipitate was 310 conducted on an anion exchange resin following the proce-311 312 dure described in Jeandel et al. (2011). Neodymium was 313 then purified following the two-step procedure of Lacan 314 and Jeandel (2001).

315 **2.3. Concentration measurements**

Hafnium and neodymium concentrations were obtained 316 317 by isotope dilution (ID) similar to the method described by 318 Rickli et al. (2009). For the five additional samples, only Nd concentrations were determined following Lacan and Jean-319 del (2001). Weighed spike solutions of ¹⁷⁸Hf and a mixture 320 of ¹⁵⁰Nd/¹⁴⁹Sm were added to an acidified 0.5 L aliquot of 321 each sample. The samples were left for 4-5 days to com-322 pletely equilibrate. Subsequently FeCl₃ was added to the 323 324 samples and Hf and Nd were co-precipitated with FeOOH 325 by adding ammonia to raise the pH to 8. For mass spectro-326 metric analysis, purification of Hf and Nd using a single 327 cation chromatographic separation step (1.4 ml resin bed,

Biorad[®] AG50W-X8, 200-400 µm mesh-size) was sufficient. 328 Hafnium was loaded and collected in a mixture of 1 M HCl 329 and 0.05 M HF (0.5 ml loading solution, 2 additional ml for 330 elution), whereas Nd was collected in 5 ml 6 M HNO₃, fol-331 lowing the elution of barium in 12 ml of 2 M HNO₃. The 332 respective cuts containing Nd and Hf were evaporated to 333 dryness and then oxidized by adding 200 µl of a 1:1 mixture 334 of 0.5 M HNO_3 and H_2O_2 (30 wt.%) to destroy remaining 335 organic compounds originating from the resin before mea-336 surement on the MC-ICPMS. Replicates for each element 337 were processed and yielded an external reproducibility of 338 better than 1% for Nd and between $\frac{3}{6}$ and $\frac{10}{6}$ for Hf 339 depending on concentration. Laboratory blanks were quan-340 tified by processing 0.5 L of MQ-water in the same way as 341 the samples and corresponded to less than 1% for Nd, 342 where no blank corrections were applied. The Hf content 343 of the samples being very low was closer to the blank level. 344 The blank amounted to 5 ± 0.7 pg (n = 11, corresponding 345 to 50% of the smallest sample), which was subtracted from 346 all samples to achieve a correct seawater concentration. 347

The concentrations of all the naturally occurring rare 348 earth elements (REE) concentrations were determined 349 using a seaFAST system (Elemental Scientific Inc.) con-350 nected to an Agilent 7500ce ICP-MS. The seaFAST system 351 performs preconcentration, matrix removal and online elu-352 tion into the ICP-MS enabling direct analysis of trace met-353 als from 4 ml of undiluted seawater (Hathorne et al., 2012). 354 The overall reproducibility for all REE concentrations was 355 better than 8% (1 SD), estimated from repeated measure-356 ments of a seawater sample (n = 10). The agreement for 357 Nd concentrations between the seaFast measurements and 358 isotope dilution is better than 10% for most of the samples 359 except for stations S3, S7 and 222 (180 m), where the sea-360 FAST produced 13.7%, 13.0% and 15.3% higher 361 concentrations. 362

2.4. Isotope measurements

The Hf isotope compositions were measured on a Nu 364 Plasma HR MC-ICPMS at GEOMAR applying manual 365 time resolved data acquisition. The samples were dissolved 366 in 250-500 µl 0.5 M HNO₃/0.1 M HF to obtain a Hf con-367 centration of ~ 20 ppb corresponding to a total beam of 368 at least 4 V. The measured Hf isotope ratios were integrated 369 over the time of analysis (3–5 min) and corrected for instru-mental mass fraction to 179 Hf/ 177 Hf = 0.7325 applying an 370 371 exponential mass fractionation law. All ¹⁷⁶Hf/¹⁷⁷Hf ratios 372 were normalized to the accepted value for JMC 475 of 373 0.28216 (Nowell et al., 1998) using the average of repeated 374 measurements of the JMC 475 standard in each session. 375 The external reproducibility for the set of samples in this 376 study was $\pm 0.8 \epsilon_{\rm Hf}$ (2 SD) and was estimated by 4–6 re-377 peated measurements of the JMC 475 Hf standard during 378 every session at the same concentration as the samples. 379 The Nd isotope composition was either also measured on 380 the Nu plasma instrument, though in automatic mode, or 381 on a Thermo Scientific Triton1 TIMS. The measured isoto-382 pic composition was corrected for instrumental mass bias to 383 146 Nd/ 144 Nd = 0.7219 applying an exponential mass fractionation law. All 143 Nd/ 144 Nd ratios were normalized to 384 385

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386 the JNdi-1 standard with the accepted ratio of 0.512115 387 (Tanaka et al., 2000) using the average of the repeatedly measured standard of the respective session. The external 388 reproducibility was between $\pm 0.3~\epsilon_{Nd}$ and $\pm 0.39~\epsilon_{Nd}$ 389 (2 SD) estimated by repeated measurements of the JNdi-1 390 standard, as well as of an internal laboratory standard. 391 392 Duplicates measured on both instruments resulted in the 393 same value within this error. The blanks for both elements (laboratory work only) were ~ 20 pg (Hf) and ~ 50 pg (Nd), 394 which in both cases was less than 1% of the smallest sample. 395 396 The additional samples from ANTXXIII/3 were mea-397 sured on a MAT 261 TIMS (Observatoire Midi-Pyrénées, 398 Toulouse, France), using the same method as described 399 above to correct for instrumental mass bias. These samples were normalized to the La Jolla Nd standard with the ac-400 cepted ¹⁴³Nd/¹⁴⁴Nd value of 0.511859, yielding an external 401 reproducibility between ± 0.2 and 0.6 ε_{Nd} units (2 SD). Both 402 laboratories participated in the 2008-2010 GEOTRACES 403 404 intercalibration efforts and their results agreed well within the overall 2 SD variation of $\epsilon_{Nd} \pm 0.6$ (van de Flierdt 405 406 et al., 2012).

407 2.5. Particles

408 We compared our data to results for Nd isotopes and 409 REE concentrations in particulates from the Southern 410 Ocean published by Hegner et al. (2007), whose study area 411 largely overlapped with ours. Hafnium concentrations for 412 marine particulates, which were measured on the same sam-413 ples as in Hegner et al. (2007), have not been published in that study. Therefore, we analyzed these Hf concentrations 414 415 for samples that were most proximal to our study sites. The methodology followed the one described in Hegner et al. 416 417 (2007).

418

3. RESULTS

419 The results for dissolved Hf and REE concentrations 420 and Hf and Nd isotope compositions from this study are 421 presented in Tables 1 and 2. The previously published par-422 ticulate Nd concentrations (Hegner et al., 2007) and the 423 new particulate Hf concentrations are listed together in Ta-424 ble 3. This table also shows those stations of ANTXXIV/3 425 closest to those of Hegner et al. (2007).

426 **3.1. Hf and Nd concentrations in the dissolved fraction**

427 Both Hf and Nd concentrations show similar distribu-428 tion patterns in the studied area (Fig. 3) with some distinct 429 exceptions as will be detailed below. The Zero Median sec-430 tion displays the following elemental distributions from 431 north to south (Fig. 3A): the highest Nd concentration of 24.2 pmol/kg is observed at station S1, about 200 km 432 433 southwest of Cape Town. This high Nd concentration is 434 in marked contrast to the very low Hf concentration at this station with only 0.06 pmol/kg (Table 1). The Nd concen-435 tration decreases southward to a minimum value 436 437 6.7 pmol/kg at the STF (station S2). Towards the SAF, 438 the concentrations for both elements remain at a low level 439 corresponding to ~6 pmol/kg for Nd and ~0.05 pmol/kg for Hf, respectively. A pronounced southward increase in the concentrations of both elements is observed between the SAF (station 104/105) and the BDY (station 113 and 116). From there onwards, the Hf and Nd concentrations remain relatively constant across the Weddell Gyre (WG) yielding typical concentrations of ~ 0.2 pmol/kg for Hf and ~ 15 pmol/kg for Nd. 440 441 442 443 444 445

The elemental distribution in the section through the 447 Weddell Sea (east to west) and the Drake Passage (south 448 to north) is characterized (Fig. 3B) by Hf and Nd concen-449 trations in the Weddell Sea that are similar to those ob-450 served around the Zero Meridian in the WG. Starting 451 approximately at station S6 the Hf concentrations, and, 452 to a lesser extent, the Nd concentrations, rise continuously 453 until station S7 north of the AAP. This broad maximum in 454 concentrations with a peak value of 0.38 pmol/kg Hf and 455 22.6 pmol/kg Nd at station 7 occurs near King George Is-456 land. Further north in the Drake Passage Hf and Nd con-457 centrations drop again to values similar to those found in 458 the eastern Atlantic sector of the ACC. 459

3.2. REE patterns of the dissolved fraction

Fig. 4 shows PAAS-normalized (Post-Archean Average 461 Shale, Taylor and McLennan, 1985) REE patterns for the 462 30 analyzed surface seawater samples (see data in Table 463 2). With the exception of two samples, the surface waters 464 show shale normalized REE patterns similar to previous 465 studies with heavy REE (HREE) enrichment compared to 466 the light REE (LREE, e.g. Elderfield and Greaves, 1982; 467 de Baar et al., 1983, 1985; Piepgras and Jacobsen, 1992; 468 German et al., 1995). Stations S2 and particularly S1 show 469 only small enrichments of the HREEs. The cerium anoma-470 lies Ce/Ce^* ($Ce/Ce^* = \log [3 \times Ce_N/(2 \times La_N + Nd_N)]$, 471 whereby values ≤ 0 represent negative anomalies and REE_N 472 is normalized to that of PAAS, Taylor and McLennan, 473 1985) of all samples are negative between -1.31 and 474 -0.24 (Table 2). The smallest negative Ce anomaly was 475 measured at station S1, close to South Africa. The most 476 negative Ce/Ce^{*} values of -1.31 to -1.03 and -1.13 to 477 -0.61 were found at stations within the Weddell Gyre 478 and the ACC, respectively. The pronounced negative anom-479 alies in the WG weaken close to Antarctica and only corre-480 spond to -0.56 at station 222 west of the AAP and to 481 -0.76 at station 154 close to the DML. PAAS-normalized 482 La-Yb ratios (La_N/Yb_N) range from 0.19 to 0.68 (Fig. 3 483 and Table 2). The highest La_N/Yb_N ratios were measured 484 at S1(0.68) and S2(0.53). South of the STF, in the ACC 485 and the Weddell Gyre, these ratios drop to 0.19-0.38. Like 486 the Ce/Ce^{$^}$, the La_N/Yb_N ratios close to the AAP are sim-</sup> 487 ilar to those in the Weddell Gyre (0.28-0.36). 488

3.3. Dissolved Hf and Nd isotope composition

Overall, the Hf isotope composition of the surface 490 waters is relatively constant and most samples are between 491 $\varepsilon_{\rm Hf} = +2.8$ and +5.0, whereas the range of Nd isotopes is 492 almost 15 $\varepsilon_{\rm Nd}$ units ($\varepsilon_{\rm Nd} = -18.9$ to -4.0, Fig. 3). It should 493 be noted that for the stations north of the Polar Front (PF) 494 the Hf concentration in the samples was too low for Hf 495

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Table 1

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Dissolved trace metal data from ANTXXIV/3 and ANTXXIII/3. The hydrographic data (temperature and salinity) are from the ships own PODAS system.

Station					Device	Depth (m)	Temperature (°C)	Salinity	₽ _{Nd}	Error (2σ)	Nd (pmol/ kg)	8 _{Hf}	Error (2σ)	Hf (pmol/ kg)
S 1	34° 53.15'	S	16° 40.70'	Е	Fish	5	21.03	35.47	-18.9	0.3	24.2			0.06
97	36° 59.60'	S	12° 45.40	Е	CTD	100	18.23	35.55	-15.9	0.4				
S2	38° 48.00'	S	11° 35.80'	Е	Fish	5	17.53	34.92	-11.0	0.3	6.7			0.04
101	42° 20.60'	S	8° 59.20'	Е	CTD	75	10.21	34.55	-9.7	0.4	8.6			
104	47° 39.80'	S	4° 16.20'	Е	CTD	75	6.41	33.72	-7.9	0.4	8.2			
105	47° 39.00'	S	4° 16.60'	Е	Fish	5	6.50	33.71	-7.9	0.3	7.4			0.08
113	52° 59.80'	S	0° 02.00'	Е	CTD	75	1.22	33.79	-7.9	0.4	16.1			
113	52° 59.80'	S	0° 02.00'	Е	CTD	150	0.20	34.02	-8.0	0.4	15.0			
116	54° 21.00'	S	0° 01.00'	Е	Fish	5	1.04	33.80	-8.0	0.3	15.8	4.5	0.8	0.18
133	59° 14.00'	S	0° 02.00'	Е	Fish	5	0.21	33.95	-8.6	0.3	18.2			0.19
142	62° 20.00'	S	0° 00.00'	Е	Fish	5	0.53	33.94	-8.6	0.3	17.3	5.0	0.8	0.11
153	65° 19.00'	S	0° 00.00'	Е	Fish	5	-0.18	33.97	-8.5	0.3	17.3	4.9	0.8	0.18
154	70° 34.50'	S	8° 07.38'	W	CTD	135	-1.69	34.02	-9.9	0.4	19.4			0.22
156	67° 08.00'	S	0° 24.00'	Е	Fish	5	-0.67	33.97	-8.6	0.3	19.5	3.0	0.8	0.25
161	66° 29.20'	S	0° 00.00'	Е	CTD	100	0.77	34.64	-8.7	0.4	17.9			0.40
GvN	68° 31.80'	S	4° 39.00'	W	Fish	5	-1.18	33.84	-8.8	0.3	16.8	3.4	0.8	0.20
S3	69° 02.00'	S	15° 42.00'	W	Snorkel	5	_	_	-8.6	0.3	16.9	4.0	0.8	0.16
186	69° 03.00'	S	17° 25.00'	W	CTD	25	-1.84	33.93	-8.4	0.3	17.2	4.3	0.8	0.20
191	67° 21.00'	S	23° 38.00'	W	CTD	25	-1.85	34.12	-8.5	0.3	18.5	4.3	0.8	0.21
S4	65° 34.00'	S	36° 46.00'	W	Snorkel	5	-1.44	34.15	-9.0	0.3	18.5			0.18
S 5	64° 59.00'	S	42° 00.00'	W	Snorkel	5	-1.55	33.91	-8.4	0.3	18.3	2.8	0.8	0.16
S 6	64° 20.00'	S	46° 04.40'	W	Snorkel	5	-1.76	33.25	-8.6	0.3	18.1	3.8	0.8	0.22
210	64° 02.80'	S	48° 15.40'	W	CTD	25	-1.83	33.79	-8.5	0.3	18.3	3.3	0.8	0.27
222	63° 21.00'	S	52° 51.00'	W	CTD	25	-1.82	34.08	-8.9	0.3	19.9	3.7	0.5	0.25
222	63° 21.00'	S	52° 51.00'	W	CTD	50	-1.82	34.10	-8.9	0.4	21.2			0.29
222	63° 21.00'	S	52° 51.00'	W	CTD	100	-1.58	34.37	-8.9	0.4	19.9			0.30
222	63° 21.00'	S	52° 51.00'	W	CTD	180	-1.22	34.49	-8.7	0.4	20.4			0.32
222	63° 21.00'	S	52° 51.00'	W	CTD	280	-0.90	34.54	-8.7	0.4	21.3			0.35
222	63° 21.00'	S	52° 51.00'	W	CTD	400	-0.90	34.56	-9.1	0.3	22.5	3.9	0.8	0.34
223	63° 17.00'	S	53° 13.00'	W	CTD	25	-1.82	34.18	-8.3	0.3	21.1	4.2	0.9	0.31
S 7	62° 08.00'	S	57° 31.00'	W	Snorkel	5	1.04	34.06	-4.0	0.3	22.6	6.1	0.5	0.38
S 8	60° 03.00'	S	55° 24.00'	W	Snorkel	5	1.78	33.85	-7.2	0.3	14.3	3.1	0.9	0.20
230	60° 06.00'	S	55° 16.40'	W	CTD	50	0.79	34.17	-6.9	0.4	18.9			0.31
230	60° 06.00'	S	55° 16.40'	W	CTD	150	0.57	34.34	-6.4	0.4	19.9			0.35
241	57° 37.20'	S	60° 53.80'	W	CTD	50	2.84	33.79	-8.4	0.4	11.8			0.17
244	56° 53.80'	S	62° 28.00'	W	CTD	25	5.38	33.92	-8.2	0.4	10.2			0.12
244	56° 53.80'	S	62° 28.00'	W	CTD	50	5.38	33.92	-8.2	0.4	9.6			0.09
250	55° 45.50'	S	64° 26.20'	W	CTD	150	3.64	34.02	-8.2	0.4	9.7			
DRA001 ^a	55° 07.00'	S	65° 3 5 ,04'	W	CTD	50	8.32	33.74	-5.7	0.6	8.5			
DRA006 ^a	55° 45.00'	S	64° 30.00'	W	CTD	20	5.10	33.90	-8.6	0.3	8.9			
DRA030 ^a	58° 52.00'	S	58° 18.00'	W	CTD	20	2.50	33.89	-8.2	0.5	14.7			
DRA052 ^a	61° 50.00'	S	55° 26.00'	W	CTD	31	_		-6.3	0.3	21.4			
DRA062 ^a	60° 39.00'	S	55° 47.00'	W	CTD	20	0.62	34.33	-7.2	0.4	17.7			

^aSamples from ANTXXIII/3 (see text).

496 isotope measurements. For Nd, stations north of the STF 497 close to South Africa have the least radiogenic Nd isotope compositions between $\varepsilon_{Nd} = -18.9$ and -11.0 (Fig. 3A). 498 Another distinctly unradiogenic Nd isotope value of 499 $\varepsilon_{\rm Nd} = -9.9$ is observed in the WG at the southernmost sta-500 501 tion 154 close to the Antarctic continent. The unique devi-502 ation from the above range of Hf isotope compositions is 503 observed at station S7, where the most radiogenic value 504 of $\varepsilon_{\rm Hf} = +6.1$ occurs close to King George Island (KGI, 505 Figs. 2 and 3B). At the same station a similar excursion 506 to more radiogenic values is also observed for Nd isotopes 507 $(\epsilon_{Nd} = -4.0)$. At the nearby stations DRA 052, 062 and S8

north of King George Island, Nd isotope compositions be-508 tween $\varepsilon_{Nd} = -6.3$ to -7.2 were measured. The Hf isotope 509 compositions on nearby stations, on the other hand, are 510 not elevated. A further radiogenic Nd isotope excursion 511 in the ACC to $\varepsilon_{Nd} = -5.7$ is measured at station 512 DRA001, close to the southern tip of Chile. Apart from 513 these deviations the Nd isotope compositions, similar to 514 Hf, are rather uniform within the ACC and the WG at 515 values between $\varepsilon_{Nd} = -7.8$ and -8.6. 516

Combined Hf and Nd isotope compositions show that 517 surface waters from the Southern Ocean are shifted towards 518 more radiogenic Hf values for given Nd values when 519

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Table 2

Dissolved REE concentration in pmol/kg, together with PAAS-normalized La/Yb ratios and Ce-anomalies (Ce/Ce^{*}). The REE concentrations were measured with a seaFAST system (see Section 2.3). For comparison the Nd concentration in pmol/kg obtained by isotope dilution is also provided.

Station	Depth		La	Ce	Pr	Nd	Sm	Eu	Gd	Tb	Dy	Ho	Er	<mark>Tm</mark>	Yb	Lu	La/Yb	Ce/	Nd	
	(m)																(norm)	Ce	(ID)	
SAC &	ACC on	Zero Meridian:																		
S1	0	34° 53.15' S 16° 40.7'	Е	29.5	35.3	6.02	25.1	4.68	1.24	5.54	0.75	4.97	1.11	3.64	0.46	5 2.57	0.43	0.68	-0.24	24.2
S2	0	38° 48' S 11° 35.8'	Е	12.1	6.47	1.52	6.77	1.53	0.29	2.00	0.29	2.25	0.71	2.03	0.24	1.43	0.17	0.50	-0.53	6.68
101	75	42° 20.60' S 8° 59.20'	Е	12.3	5.90	2.05	8.99	2.14	0.34	2.49	0.43	3.00	0.94	3.18	8 0.37	2.52	2 0.45	0.29	-0.61	8.56
104	75	47° 39.80' S 4° 16.20'	Е	14.1	3.53	1.95	8.61	1.82	0.49	2.60	0.45	3.63	1.05	3.74	0.55	3.54	0.70	0.24	-0.87	8.16
105	9	47° 39.00' S 4° 16.60'	Е	11.9	2.60	1.54	7.21	1.41	0.42	1.99	0.42	3.83	1.12	3.81	0.51	3.73	0.64	0.19	-0.93	7.38
113	70	52° 59.80' S 0° 02.00'	Е	23.6	5.64	3.34	15.1	2.80	0.87	4.59	0.74	5.73	1.47	5.37	0.81	5.26	0.97	0.27	-0.90	16.1
116	9	54° 21.00' S 0° 01.00'	Е	24.5	8.34	3.69	16.6	2.77	0.76	4.68	0.64	5.71	1.49	5.16	5 0.90	5.39	0.94	0.27	-0.75	15.8
Wedde	ll Gyre:																			
133	9	59° 14.00' S 0° 02.00'	Е	31.9	3.74	4.50	19.8	3.60	0.93	4.86	0.85	6.22	1.63	5.91	0.88	6.01	1.05	0.31	-1.20	18.2
142	9	62° 20.00' S 0° 00.00'	Е	30.3	3.60	4.08	18.3	3.41	0.86	4.61	0.76	5.95	1.73	5.64	0.88	6.02	2 1.04	0.30	-1.20	17.3
153	9	65° 19' S 0° 00.00'	Е	28.7	3.63	3.94	18.0	3.19	0.83	4.50	0.70	5.94	1.57	5.77	0.80	5.62	2 1.15	0.30	-1.17	17.3
154	135	70° 34.50' S 8° 07.38'	W	33.9	11.0	4.95	20.9	3.46	0.91	5.08	0.88	6.02	1.65	5.97	0.97	6.71	1.14	0.30	-0.76	19.4
156	5	67° 08.00' S 0° 24.00'	Е	28.4	3.75	4.03	19.5	2.95	0.80	5.00	0.71	6.05	1.62	5.80	0.84	5.45	5 1.01	0.31	-1.16	19.5
161	100	66° 29.20' S 0° 00.00'	Е	31.1	4.53	4.40	18.6	3.19	1.02	4.67	0.81	6.24	1.72	6.43	0.89	6.50	1.23	0.28	-1.11	17.9
GvN	5	68° 31.80' S 4° 39.00'	W	29.5	3.37	4.01	16.8	3.04	0.71	6.06	0.70	6.10	1.58	5.37	0.91	5.09	1.05	0.34	-1.21	16.8
S3	0	69° 02.00' S 15° 42.00	' W	30.6	4.72	4.25	19.2	3.14	0.84	5.01	0.77	6.28	1.57	5.84	0.94	5.98	3 1.00	0.30	-1.09	16.9
186	0	69° 03.00' S 17° 25.00	' W	29.9	4.14	4.11	19.0	3.06	0.88	5.10	0.71	5.44	1.65	6.02	2 0.84	5.86	0.97	0.30	-1.13	17.2
191	5	67° 21.00' S 23° 38.00	' W	32.1	4.01	4.43	18.9	3.70	0.90	5.72	0.84	6.23	1.72	6.00	0.96	6 4.96	5 1.09	0.38	-1.17	18.5
S4	0	65° 34.00' S 36° 46.00	' W	32.6	2.92	4.36	18.7	3.48	0.87	5.05	0.80	5.97	1.77	6.03	0.92	6.06	5 1.10	0.32	-1.31	18.5
S5	0	64° 59.00' S 42° 00.00	' W	31.0	3.43	4.48	20.2	3.73	0.96	5.65	0.74	6.55	1.88	6.11	0.83	5.98	3 1.05	0.31	-1.23	18.3
S6	0	64° 20.00' S 46° 04.40	' W	30.9	3.48	4.30	19.2	3.43	0.84	5.66	0.83	5.75	1.66	5.91	0.84	6.21	1.06	0.29	-1.22	18.1
210	0	64° 02.80' S 48° 15.40	' W	30.9	5.45	4.75	20.0	3.69	0.84	4.55	0.78	5.95	1.80	5.71	0.92	6.26	5 1.13	0.29	-1.03	18.3
AAP &	b Drake p	bassage:																		
222	50	63° 21.00' S 52° 51.00	' W	39.1	19.3	5.41	21.0	3.73	0.98	5.27	0.75	6.43	1.74	6.29	0.90	6.52	2 1.14	0.36	-0.56	21.2
222	180	63° 21.00' S 52° 51.00	' W	34.2	9.49	4.96	23.6	4.03	0.92	4.95	0.87	6.22	1.80	6.22	2 0.92	6.37	1.18	0.32	-0.84	20.4
222	280	63° 21.00' S 52° 51.00	' W	36.1	14.1	5.19	22.1	4.11	1.05	5.89	0.94	6.97	1.83	6.72	2 0.99	6.61	1.18	0.32	-0.68	21.3
222	Bottom	63° 21.00' S 52° 51.00	' W	37.4	10.1	5.58	23.2	4.49	0.95	5.72	0.95	7.16	1.90	6.54	0.99	7.10	1.29	0.31	-0.84	22.5
S 7	0	62° 08.00' S 57° 31.00	' W	33.6	16.2	5.80	25.6	4.98	1.19	6.49	0.96	7.24	1.99	6.60	0 1.08	7.22	2 1.29	0.28	-0.61	22.6
S 8	0	60° 03.00' S 55° 24.00	' W	26.3	4.70	3.44	15.6	2.74	0.78	3.90	0.69	5.00	1.47	5.04	0.74	4.41	0.82	0.35	-1.02	14.3
230	50	60° 06.00' S 55° 16.40	' W	30.0	9.18	4.55	20.8	3.85	1.09	4.99	0.78	6.39	1.75	6.05	5 0.88	5.96	5 1.17	0.30	-0.80	18.9
230	150	60° 06.00' S 55° 16.40	' W	31.8	9.46	4.58	21.0	3.81	1.11	5.56	0.98	6.59	1.93	6.55	5 1.02	6.77	1.22	0.28	-0.81	19.9
241	50	57° 37.20' S 60° 53.80	, w	20.1	3.75	2.95	12.4	2.21	0.63	3.44	0.60	4.68	1.41	4.80	0.70	4.39	0.82	0.27	-1.00	11.8
244	25	56° 53.80' S 62° 28.00	, W	14.7	2.06	2.09	9.5	1.69	0.46	2.94	0.46	3.73	1.21	3.98	8 0.64	3.79	0.70	0.23	-1.13	10.2
244	50	56° 53.80' S 62° 28.00	, M	16.3	2.78	2.23	9.47	1.69	0.40	3.08	0.52	3.97	1.19	3.98	8 0.66	6 4.21	0.72	0.23	-1.03	9.57

compared to the "terrestrial array" (Fig. 1B). Similar to 520 521 previously published seawater data (Godfrey et al., 2009; Rickli et al., 2009, 2010; Zimmermann et al., 2009a,b), the surface waters in this study plot on the "seawater 522 523 524 array", as previously also derived from analyses of ferro-525 manganese crust and nodule surfaces (Albarède et al., 1998; David et al., 2001; van de Flierdt et al., 2006). All 526 samples form an isotopically tightly constrained field, with 527 528 the only exception of the samples collected in the vicinity of KGI, which plot on the more radiogenic part of the seawa-529 530 ter array.

531 4. DISCUSSION

532 4.1. Weathering inputs of Hf and REEs

533 4.1.1. South Africa

While the general distribution of Hf and REE concentrations, as well as Hf and Nd isotopes are similar in most
parts of the study area, the data are distinctly different close

to South Africa. At station S1 the Nd concentration of 537 24.2 pmol/kg is the highest measured in this study, whereas 538 Hf is at its lowest level (0.06 pmol/kg). The high Nd con-539 centration together with the shale-normalized La_N/Yb_N of 540 0.68 and an only slightly negative Ce anomaly (Ce/ 541 $Ce^* = -0.24$) suggest enhanced terrigenous inputs from 542 the South African continent. A shale-normalized REE pat-543 tern comparable to that of particulate data (Hegner et al., 544 2007) further supports increased contributions to the dis-545 solved REEs originating from terrigenous material, proba-546 bly via partial dissolution of particles (Fig. 4). A 547 continental input is also indicated by the very negative sea-548 water Nd isotope composition of $\varepsilon_{Nd} = -18.9$ at S1. How-549 ever, the isotope composition of suspended particles 550 yielding ε_{Nd} of -14.5 (Hegner et al., 2007) is too radiogenic 551 to explain the low seawater Nd isotope composition at S1. 552 Surface sediments deposited around the South African 553 coast are on average also more radiogenic ($\varepsilon_{Nd} = -14$ to 554 -12, Franzese et al., 2006). Therefore, an additional source 555 supplying unradiogenic Nd is necessary to explain the 556

Table 3

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randulate ri, ind and biogenic opar data from regner et al. (2007) closest to the sample sites of this study.												
Sample ^a	LAT ^a (degrees N)	LON ^a (degrees E)	Biogen. SiO ₂ ^a (%)	Particulate Nd ^a (pmol/l)	Error (%)	Particulate Hf ^b (pmol/l)	Error (%)	Nd/Hf particles	Closest station to ANTXXIV/3 ^c			
z137	-39.53	18.84		0.460	4.2	0.015	13.7	30.9	S2			
z01	-49.75	5.87	29	1.145	3.5	0.080	7.1	14.2	105			
z02	-53.47	9.00	42	0.405	4.4	0.039	6.4	10.4	116			
z04	-60.58	3.00	17	4.635	1.8				133			
z05	-63.63	-0.31	8	4.028	2.6	0.148	5.9	27.2	142/153			
z06	-67.27	-3.93	9	1.170	3.3	0.056	9.3	21.1	156/GvN			
z71 (2)	-68.40	-26.14	58	0.231	12.3	0.042	14.2	5.4	191			
z72	-66.81	-32.00	47	0.156	12.0				S4			
z75	-63.03	-44.17	37	0.675	3.1	0.092	5.4	7.3	S5/S6			
z76	-61.86	-52.42	27	0.076	6.0	0.009	7.5	8.5	210			
z77	-62.23	-55.47		0.100	7.0	0.010	7.7	9.7	222/223			
z84	-62.30	-58.65		24.438	3.1	2.882	3.7	8.5	S 7			
z87	-59.81	-59.69	42	0.195	3.8	0.027	14.2	7.2	S8			
z89	-56.68	-63.21	29	0.142	3.6	0.010	6.7	14.8	244			

^aFrom Hegner et al. (2007).

^bFrom the same samples of Hegner et al. (2007).

°This study.

557 unradiogenic value at S1, most plausibly surface waters ad-558 vected by the Agulhas Current. The Agulhas waters may 559 have been influenced by particle dissolution close to the 560 eastern coast of South Africa and carry unradiogenic Nd into the Atlantic, which is supported by the provenance 561 of deposited detritus in the southern Cape Basin (Rutberg 562 563 et al., 2005; Franzese et al., 2006). Unfortunately, there 564 are no further data available on the marine particle compo-565 sition from the Indian Ocean side of South Africa to support this assumption. As mentioned above, the Hf 566 567 concentration at S1 is one of the lowest of the entire sample 568 set, which does not support an additional input from parti-569 cle dissolution for Hf in that area. Thus, a release of Hf 570 from particles appears to be of no importance or Hf is sim-571 ply scavenged more readily. Unfortunately the lack of a Hf 572 isotope value at this station does not allow to investigate if boundary exchange has been involved instead, through 573 574 which the radiogenic isotope signature can change but not 575 necessarily the concentration (e.g. Lacan and Jeandel, 576 2005). Possibly, Hf is largely retained in refractory miner-577 als, such as zircon, which make it less available during par-578 ticle dissolution. If Hf scavenging was high at S1, the effect 579 on the REEs seems to be very low given the high concentra-580 tion of these elements. Even though recent studies have 581 shown that Hf is scavenged in surface waters more readily than e.g. Nd, the overall scavenging behavior of these two 582 elements is similar (Godfrey et al., 2009; Rickli et al., 583 584 2010). Thus, it seems rather unlikely that a very high Nd 585 concentration accompanied with a very low Hf concentra-586 tion is only the result of different scavenging behavior between both elements. 587

Further south at station S2, just north of the STF, the HREE are 40% lower while the LREE are between 60% and 84% lower, and Hf is at its minimum value of 0.04 pmol/kg. The REE pattern shifts from a flatter shale normalized shape at S1 to a steeper seawater-like pattern at S2 with LREEs relatively depleted compared to HREEs ($La_N/Yb_N = 0.5$) and a lower Ce/Ce^{*} of -0.53 (Fig. 4 and

Table 2). Although not very pronounced, this pattern, to-595 gether with the overall low abundance of the REEs suggest 596 either a low contribution from the dissolution of terrige-597 nous material or enhanced scavenging due to higher pro-598 ductivity at the STF, most likely a combination of both 599 processes. Hegner et al. (2007) reported low aluminum 600 (Al) and thorium (Th) concentrations in particles docu-601 menting very low terrigenous inputs to this area thus pro-602 viding an explanation for the low dissolved REE 603 concentrations. These authors also proposed that scaveng-604 ing of Nd from seawater on the order of only 5-10% of 605 the available dissolved load would be sufficient to yield 606 the observed Nd concentrations adsorbed to the particles. 607 However, the slightly lower La_N/Yb_N ratios from our study 608 between 0.19 and 0.29 from the STF to the BDY compared 609 to the ones from the Weddell Gyre (~ 0.29 to 0.38) may 610 indicate more ample scavenging, since LREE are preferen-611 tially scavenged to biogenic particles (e.g. Elderfield and 612 Greaves, 1982). The low dissolved Nd concentrations in 613 our surface water samples from the ACC can thus be ex-614 plained by limited supply but additional scavenging also 615 seems to be involved. 616

4.1.2. Antarctica and South America

The elemental distribution of the sampled stations near 618 the Antarctic continent does not indicate strong terrigenous 619 supply. Within the entire Weddell Gyre Hf and REE con-620 centrations are elevated relative to the ACC but with no sig-621 nificant increases similar to those observed for S1 close to 622 the Antarctic continent (Figs. 3 and 4). However, at sta-623 tions S7 and 222, close to the volcanic King George Island 624 (KGI, Fig. 2C) and east of the Antarctic Peninsula (AAP), 625 respectively, the Hf and REE concentrations are clearly ele-626 vated. In particular, at S7 the Hf concentration of 627 0.38 pmol/kg is almost twice as high as it is on average in 628 the Weddell Gyre, whereas the REE concentrations are 629 on average only about 10-50% higher. At station 222 no 630 significant increase of the Hf concentration in the surface 631

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Fig. 3. Geochemical composition of the surface samples (<100 m water depth) along the Zero Meridian (A) and through the Weddell Sea and the Drake Passage (B), note the different scales in ε_{Nd} and temperature between (A) and (B). Major hydrographic transitions are marked by horizontal dashed lines (see text). Hafnium and Nd concentrations largely follow a similar pattern in both sections with the exception that S1 in (A) shows a significantly elevated Nd concentration. The Hafnium and Nd isotope compositions also follow similar trends where combined data are available. The only exceptions are stations 156 and GvN, where the ε_{Hf} drops to less radiogenic values, whereas ε_{Nd} does not. The La/Yb ratios are normalized to PAAS (Taylor and McLennan, 1985). Horizontal bars indicate the analytical uncertainties, which in most cases are smaller than symbol size.

632 water is observed. The more radiogtenic Hf and Nd isotope compositions at S7 ($\epsilon_{\rm Hf} = +6.1$ and of $\epsilon_{\rm Nd} = -4.0$, respec-633 tively) clearly document external inputs of these elements 634 635 from the basaltic rocks of KGI. The observed radiogenic Nd isotope composition accompanied only by a small in-636 637 crease of the REE concentrations implies a boundary ex-638 change process with the KGI. A similar, though not as 639 pronounced exchange of Nd isotopes has obviously influ-640 enced station 154, which has been sampled in 135 m water depth close to the seafloor at the Antarctic ice shelf. Here 641 the Nd isotope composition of $\varepsilon_{Nd} = -9.9$ is more than 1 642 ε-unit lower than those in the nearby surface stations 643 (Fig. 3B). A simple mass balance calculation, using the 644 645 Nd isotope composition of station 154 ($\varepsilon_{Nd} = -9.9$), the 646 average isotope composition of the Weddell Sea and that of glacio-marine sediments $(\epsilon_{Nd} = -8.6 \text{ and } -15.0, \text{ respec-}$ 647 tively, this study; Roy et al., 2007; Stichel et al., 2011), sug-648 gests that about 20% of the Nd must be derived from shelf 649 sediments off Dronning Maud Land. However, the increase 650

of the Nd concentration and most of the REEs is less than 651 10% compared to the average surface concentration of 652 17.9 pmol/kg. In order to test if processes other than parti-653 cle or sediment release are involved, we simply calculate a 654 hypothetical ε_{Nd} signature only caused by exchange with 655 particles based on the above considerations and compare 656 it with the measured value. This is an approach similar to 657 the one of Lacan and Jeandel (2001) using the following 658 equation: $\varepsilon_{\text{Nd-calc}} = ([\text{Nd}]_{\text{initial}} * \varepsilon_{\text{Nd-initial}} + \Delta[\text{Nd}] * \varepsilon_{\text{Nd-}}$ 659 SED)/[Nd]_{meas}, where [Nd]_{initial} is the average concentration 660 of surface waters in the Weddell Sea (18 pmol/kg), [Nd]meas 661 the one of station 154 (19.4 pmol/kg), Δ [Nd] the difference 662 in concentration (1.4 pmol/kg) and $\varepsilon_{\text{Nd-initial}}$ is the average 663 isotope composition of the Weddell Sea surface waters 664 $(\varepsilon_{\rm Nd} = -8.6)$. The isotope composition of the sediment 665 $(\overline{\epsilon_{\text{Nd-SED}}} = -15.0)$ is taken from Roy et al. (2007) close to 666 station 154 of our study. If the calculated value is the same 667 as the one observed, the difference in the isotope composi-668 tion between these waters seems to be derived entirely by 669

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670 particle or sediment release. The calculated isotope composition of $\varepsilon_{Nd} = -9.1$, however, is about 0.8 ε_{Nd} units more 671 672 radiogenic, which is just about higher than the external 673 reproducibility of ± 0.4 , but still suggests that boundary ex-674 change processes do influence the isotope composition at this location. If particle dissolution was the only process 675 to explain the isotope composition at station 154, the corre-676 677 sponding Δ [Nd] must be ~4.5 pmol/kg, whereas the ob-678 served Δ [Nd] is only 1.4 pmol/kg.

The only significant drop towards less radiogenic Hf iso-679 680 tope compositions is observed at station $\frac{153}{(\epsilon_{Hf} = +4.9)}$ 681 and stations 156 and GvN nearest to Dronning Maud Land 682 $(\varepsilon_{\text{Hf}} = +3.0 \text{ and } \varepsilon_{\text{Hf}} = +3.4$, respectively; Fig. 3A). The 683 proximity of these stations to the Antarctic shelf suggests 684 an unradiogenic Hf input from the Dronning Maud Land 685 (e.g. van de Flierdt et al., 2006, 2007). Although a shift to unradiogenic Nd is not observed for these stations the 686 687 unradiogenic Nd isotope composition at the southernmost station 154 ($\varepsilon_{Nd} = -9.9$, ε_{Hf} is not available) also suggests 688 689 Nd inputs from the Dronning Maud Land. In general, 690 Nd isotopes in the WG are slightly less radiogenic than in 691 the ACC (Table 1), also pointing at some Nd fluxes from 692 Antarctica to the Weddell Sea. The less radiogenic Hf sig-693 natures observed in the vicinity of Dronning Maud Land 694 may be a consequence of a relatively congruent release of 695 Hf from the exposed rocks on land due to the glacial grind-696 ing of weathering resistant zircons. Evidence for a more 697 congruent release of Hf under physical weathering condi-698 tions has previously been found in ferromanganese crusts, 699 which recorded less radiogenic Hf isotope compositions than those of the seawater array in regions and during peri-700 701 ods of prevailing enhanced physical weathering (Piotrowski 702 et al., 2000; van de Flierdt et al., 2002, 2007). These authors 703 concluded that the onset of Northern Hemisphere Glacia-704 tion and the accompanying enhanced physical weathering 705 led to a more congruent release of Hf isotopes from the 706 old continental crust in northern Canada and Greenland. 707 However, the shift of the Hf isotope compositions in these 708 studies was several ε_{Hf} units compared to only 1 ε_{Hf} unit in 709 our case. Our data from the Weddell Sea in the proximity of 710 Dronning Maud Land thus suggest that the local Hf iso-711 tope composition in this case is not a sensitive tracer for 712 physical weathering inputs. Furthermore, the combined 713 Hf and Nd isotope data do not plot significantly closer to 714 the terrestrial array (Fig. 1).

715 The exchange with Antarctica is clearly more prominent 716 at station S7, close to King George Island (KGI). Like most 717 of the South Sandwich Islands, KGI is composed of relatively young, mafic island arc rocks (see Section 1.2). Previ-718 719 ous studies on James Ross Island, which has the same 720 geological origin as KGI (e.g. Machado et al., 2005) re-721 ported Hf and Nd isotope compositions of the bulk rocks of $\varepsilon_{\rm Hf} \sim +7$ (Sims et al., 2008) and $\varepsilon_{\rm Nd} \sim +5$ (Hole et al., 722 723 1994; Kosler et al., 2009). Enhanced weathering input from 724 these rocks is also indicated by the high particulate concentrations of Hf and Nd (Hegner et al., 2007; Table 3), which 725 726 are likely to modify the isotope composition of the surface 727 waters. We thus also calculated the hypothetical Nd isotope 728 composition only controlled by partial dissolution of parti-729 cles at station S7, analogous to the approach above. We

used the Nd isotope composition and concentration of sta-730 tion 230/50 m ($\epsilon_{Nd} = -8.2$; 18.9 pmol/kg) as a representa-731 tive value of surface waters from the Drake Passage. The 732 isotope composition of KGI ranges from $\varepsilon_{Nd} = +1$ to +5733 (Hole et al., 1994; Kosler et al., 2009), from which a hypo-734 thetical Nd isotope composition of $\varepsilon_{Nd} = -6.7$ to -6.0 is 735 736 calculated. This is 2–2.7 ε_{Nd} units less radiogenic than the observed signatures (Fig. 3B). The observed Nd isotope 737 composition of $\varepsilon_{\rm Nd} = -4.0$ is therefore largely a result of 738 boundary exchange rather than dissolution of weathered 739 volcanic material. The REE pattern of S7 is similar to that 740 of LCDW (Hathorne et al., 2012) and also suggests that ele-741 mental input derived from volcanic material (REE values 742 from Machado et al., 2005) is of minor importance (Fig. 4). 743

The observed radiogenic Hf isotope composition at KGI 744 of $\varepsilon_{\rm Hf} = +6.1$ (S7) also indicates a significant influence from 745 weathering inputs on the surface water isotope composi-746 tion. The Hf concentration of 0.38 pmol/kg is also signifi-747 cantly higher than the average in the Weddell Sea or the 748 ACC (Fig. 3 and Table 1). Applying the same approach 749 as for Nd reveals that the hypothetical Hf isotope composi-750 tion should be between +3.8 and +5.4, using average com-751 positions of the Weddell Gyre and the ACC ($\varepsilon_{Hf-initial} = +3$ 752 to +4 and $[Hf]_{initial} = 0.2$ to 0.3 pmol/kg). The observed Hf 753 isotope composition at S7, however, is $\varepsilon_{\rm Hf} = +6.1$, indicat-754 ing again that boundary exchange processes govern the iso-755 tope composition rather than addition of dissolved Hf from 756 particles or weathered material. 757

The peak radiogenic Hf and Nd isotope values sharply 758 decrease towards less radiogenic levels of $\varepsilon_{Nd} = -6.3$ 759 (DRA052) and $\varepsilon_{Nd} = -7.2$; $\varepsilon_{Hf} = +3$ (S8 and DRA062) to-760 wards the Drake Passage, implying that the influence of 761 KGI is locally limited and that the boundary exchange sig-762 nals are rapidly diluted by surface ocean waters from the 763 Drake Passage. The imprint on the Hf isotope compositions 764 is even more locally confined and is only detected at station 765 7. This is explained by the lower isotopic contrast for Hf be-766 tween station 7 ($\varepsilon_{\rm Hf} = +6.1$) and the "typical" ACC surface 767 ocean ($\varepsilon_{\rm Hf} \sim +4$, Fig. 3B) relative to the contrast observed 768 for Nd isotopes, but may also reflect a shorter surface ocean 769 residence time for Hf than for Nd as suggested before (Ric-770 kli et al., 2010). A more radiogenic Nd isotope signal is 771 again observed in the northern part of the Drake Passage 772 (Fig. 3B). At station DRA001 the Nd isotope composition 773 of $\varepsilon_{\rm Nd} = -5.7$ implies radiogenic input from the southern 774 tip of South America, possibly derived from the Pali Aike 775 Volcanic Field, which consist of lavas with a Nd isotope 776 composition as radiogenic as $\varepsilon_{Nd} \sim +5$ (D'Orazio et al., 777 2000). The imprint is only observed close to the coast and 778 the Nd isotope composition of the open Drake Passage is 779 indistinguishable from the ACC ϵ_{Nd} signatures of ${\sim}{-8}$ 780 along the Zero Meridian. 781

4.2. Cycling of Hf and Nd in the Southern Ocean

Previous studies have suggested that Hf and REEs are 783 prone to scavenging by biogenic opal in the Southern 784 Ocean (e.g. Rickli et al., 2009; Stichel et al., 2011). It was 785 suggested that the large specific surface area of diatom frustules acts as an effective scavenger for Hf and Nd. Stichel 787

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et al. (2011) found that Hf and Nd concentrations correlate 788 789 well with dissolved Si content in the Southern Ocean and suggested that the observed linear increase of Nd concen-790 791 tration with water depth represents a steady release of Nd, while the frustules dissolve. This feature was not ob-792 served for Hf, indicating a different remineralization behav-793 794 ior. A comparison of particulate Nd/Hf ratios from our 795 study with the particulate opal content in the eastern Atlan-796 tic sector of the ACC indicates a strong negative correlation between these parameters (Nd and opal data from Hegner 797 798 et al. (2007); Fig. 5 and Table 3). This suggests that Hf is 799 more efficiently scavenged from surface waters than Nd 800 when siliceous frustules of diatoms are present. This means 801 that diatom productivity acts as a more effective sink for Hf 802 than for Nd in the Southern Ocean given that a significant 803 fraction of the opal is buried in the sediment. However, one would expect that such an effect would influence the dis-804 805 solved fraction as well, which is not observed (Fig. 5B). Except for stations S1, S2 and 142 the elemental distribution 806 of Hf and Nd is very similar and does not support any frac-807 tionation by scavenging. One plausible explanation is that 808 809 the particulate and dissolved fractions were collected sepa-810 rately on different expeditions. The particulate samples 811 were taken between November 1997 and March 1998 (Heg-812 ner et al., 2007), whereas the dissolved fraction was sampled 813 from February to April 2008. To estimate if and to what ex-814 tent the diatoms contribute to a preferential scavenging or 815 elemental fractionation, parallel collection of the particu-816 late and dissolved fractions is required in future studies.

817 South of the ACC the dissolved concentrations for both elements are elevated and rather homogenous, with the 818 819 exception of station 142, where the Hf concentration 820 amounts to only 0.11 pmol/kg, i.e. only half of the average 821 Hf concentration observed for other samples in this area. 822 Hegner et al. (2007) also observed higher particulate Nd 823 concentrations in the WG (more than 2 pmol/L) together 824 with unradiogenic detrital Nd isotope compositions and 825 high Al concentrations of up to 7.2 nmol/L, supporting supply of terrigenous material by icebergs. However, the 826 elevated concentrations of Nd (~18 pmol/kg) and Hf 827 (~0.21 pmol/kg) observed within the Weddell Gyre in our 828 829 study compared to those north of the PF are more likely 830 caused by accumulation of deep sourced Hf and Nd rather 831 than derived from icebergs, as discussed below. The highest 832 dissolved Hf and Nd concentrations south of the STF of 833 0.38 pmol/kg and 22.1 pmol/kg, respectively, were mea-834 sured near the volcanic King George Island. This matches 835 the highest concentrations in particulate Hf and Nd (2.88 pmol/L and 24.44 pmol/L, respectively, Hegner 836 837 et al., 2007; Table 3), implying a strong input from this or 838 adjacent islands.

839 Terrigenous influence is further supported by the isotope composition of particles in the Weddell Gyre yielding values 840 $\varepsilon_{\rm Nd} = -15$ or lower, suggesting an enhanced input of mate-841 rial from East Antarctica via ice rafting (Hegner et al., 842 2007). Alternatively, the Weddell Gyre's average dissolved 843 844 Nd and Hf isotope composition is nearly identical to that of CDW ($\epsilon_{Nd} = -8.5 \pm 0.3$, $\epsilon_{Hf} = +4.4 \pm 0.6$, Stichel 845 et al., 2011) in the studied area. This similarity may suggest 846 847 that the isotope compositions of surface waters are the result

Fig. 4. Shalenormalized REE patterns of surface samples (<100 m depth) from north of the Polar Front on the Zero Meridian (A), the Weddell Gyre (B) and the Antarctic Peninsular including the Drake Passage (C). The color-coding corresponds to the zones illustrated in Fig. 3. Additional patterns are shown for comparison: The composition of suspended particles close to South Africa (A, Hegner et al., 2007), Lower Circumpolar Deep Water (LCDW) at 2000 m in the Weddell Gyre (A–C, Hathorne et al., 2012) and the composition of volcanic rocks of King George Island (C, Machado et al., 2005). Seawater values are normalized to PAAS multiplied by 10^6 , whereas the normalized particulate values are multiplied by 0.01 to fit on the same scale.

of vigorous mixing and accumulation of these elements within the WG rather than reflecting the result of low terrigenous input. A supply of Hf and Nd from deeper waters is supported by relatively high surface water concentrations of ~ 0.2 pmol/kg and ~ 18 pmol/kg, respectively. Addition-

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ally, the REE patterns in the Weddell Gyre are very similar 853 854 to the pattern of LCDW in Hathorne et al. (2012). On the 855 basis of the distribution of the highly particle reactive elements Th and protactinium (Pa) in the Weddell Gyre, Rut-856 857 gers van der Loeff and Berger (1993) documented that 858 particle export and thus scavenging is very low leading to 859 their accumulation in the water column of the Weddell Gyre. The general homogeneity of Hf and REE within the 860 861 Weddell Gyre also suggests accumulation of these elements. 862 This observation can be used to estimate the minimum res-863 idence time of Hf and REE in surface waters within the 864 Weddell Gyre. The average residence time of a parcel of surface water there is around 2.5 years (Gordon and Huber, 865 1990; Hoppema et al., 1995), implying that the residence 866 867 times of Hf and the REEs is at least in that order.

868 5. SUMMARY AND CONCLUSIONS

In this study the first combined surface water Hf and Nd
isotope compositions and Hf and REE concentrations of
surface waters from the Atlantic sector of the Southern
Ocean are presented. The data set comprises a wide range
of Hf and Nd concentrations from low values in the South
Atlantic Current (0.04 and 6.7 pmol/kg) to maximum values
ues near King George Island (0.38 and 22.6 pmol/kg).

- 876 (1) The highest Nd concentrations were found in surface 877 waters southwest of Cape Town. In the same sample, 878 a very low Hf concentration indicates restricted 879 release of Hf from the South African continental crust. High La_N/Yb_N ratios between 0.50 and 0.68 880 and markedly flat PAAS-normalized REE patterns 881 support enhanced terrigenous input from South Afri-882 883 ca. This is accompanied by the least radiogenic Nd 884 isotope composition as low as $\varepsilon_{Nd} = -18.9$, which was probably advected via the Agulhas Current and 885 886 documents inputs from weathering of Archean cratonic rocks exposed in South Africa. 887
- 888 (2) The low elemental concentrations just north of the Subantarctic Front most likely result from a combi-889 890 nation of limited terrigenous supply and high biolog-891 ical productivity. A good correlation of the particulate Nd/Hf ratio with biogenic SiO₂ content 892 893 suggests a preferential scavenging of Hf onto sili-894 ceous frustules. The Hf and Nd concentrations are lower north of the PF and increase southward to 895 more constant values within the WG indicating accu-896 897 mulation resulting from low biogenic particle fluxes 898 and thus a weak scavenging efficiency of trace metals.
- (3) We observe the influence of volcanic rocks by 899 900 marked increases of Hf and Nd concentrations in 901 the vicinity of King George Island. These are, accom-902 panied by a pronounced excursion to more radiogenic isotope compositions for both elements 903 ($\epsilon_{\rm Hf} = +6.1$; $\epsilon_{\rm Nd} = -4.0$). Mass balance consider-904 905 ations suggest that not only elemental release originating from particle dissolution but also boundary 906 907 exchange processes govern the isotope composition 908 of surface waters. A similar influence from mantle-

Fig. 5. Nd/Hf ratio in particles (A) and in the dissolved fraction (B) plotted against the biogenic opal content in particles (Hegner et al., 2007).

derived rocks is observed at the southern tip of Chile. Boundary exchange processes most likely also take place on the Antarctic shelf where less radiogenic Nd than the average composition of the Weddell Gyre is found at 135 m water depth close to the seafloor.

- (4) The Hf and Nd isotope compositions of the surface samples are shifted towards more radiogenic Hf values for given Nd values when compared to the "terrestrial array" and plot on the "seawater array".
- The majority of the studied areas comprising the (5)919 ACC and the Weddell Gyre shows essentially 920 homogenous isotope compositions of $\varepsilon_{\rm Hf} \sim +4$ and 921 $\varepsilon_{\rm Nd} \sim -8.5$, which is similar to the deep water com-922 positions studied previously and documents strong 923 vertical and lateral mixing of Hf and Nd south of 924 the Polar Front, indicating a residence time at least 925 as long as the turnover time of surface waters in 926 the Weddell Gyre. 927

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