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Original article

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Pharmacological profile of inhibition of the chloride channels activated by extracellular acid in cultured rat Sertoli cells

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Abstract – Sertoli cells from mammalian testis are key cells involved in the development and maintenance of stem cell spermatogonia as well as in the secretion of a Cl⁻ and K⁺-rich fluid into the lumen of seminiferous tubules. The pharmacology and contribution of Cl⁻ channels to the physiology of Sertoli cells were investigated using whole-cell patch clamp and iodide efflux experiments applied to cultured rat Sertoli cells. We characterized an outwardly rectifying Cl⁻ current stimulated by various acid species including the physiologically relevant lactic acid. Using the iodide efflux technique, the pharmacological properties of this Cl⁻ current, noted ICl_{acid}, revealed Ca²⁺-independent inhibition by DIDS (IC₅₀ = 27 μ M), glibenclamide (IC₅₀ = 31 μ M) and DPC (IC₅₀ = 86 μ M). ICl_{acid} is DIDS \approx glibenclamide > DPC >> calix[4]arene, 9-AC. For comparison, the inhibitory profile of the swelling- and ATP-activated Cl⁻ currents in Sertoli cells is DPC = DIDS >> glibenclamide = 9-AC for ICl_{swell} and DPC = 9-AC = DIDS >> glibenclamide for ICl_{ATP}. This description provides new insights into the physiology and pharmacology of the endogenous Cl⁻ channels expressed and potentially involved in fluid secretion in Sertoli cells.

Cl⁻ channels / pharmacology / proton-activation / ATP / cell volume / glibenclamide

1. INTRODUCTION

In the mammalian testis, the somatic Sertoli cells, discovered by the Italian physiologist and histologist Enrico Sertoli in 1865, play a key role in the initiation and maintenance of spermatogenesis, in the development of a functional testis, and hence in the expression of the male phenotype [1-3]. The structure of Sertoli cells and the specialized junctions between

them and the neighboring germ cells create an immunologically protected space [4], a barrier to protect developing germ cells against harmful agents and a sophisticated microenvironment providing all the nutrients, a variety of hormones [5] and growth factors required for the full development of germ cells [1–3]. In the rat, Sertoli cells proliferate during the fetal and early neonatal periods before assuming a terminally differentiated state [6, 7].

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Sertoli cells are also responsible for the secretion of a chloride and potassium-

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rich fluid in the lumen of seminiferous tubules [8]. Cytosolic calcium $[Ca^{2+}]_i$ also plays an important role in the biochemistry and physiology of immature Sertoli cells [9, 10]. Cytosolic calcium signaling is involved in the regulation of intracellular pH, which appears to play an important role in secretory processes in response to hormones [9, 11]. In addition, an acidic microenvironment is crucial for male fertility since sperm cannot fertilize an egg without proper acidification of the acrosome [12]. Sertoli cells secrete lactate in the presence of glucose [13]. Because germ cells degenerate in the absence of lactate [14], isolated spermatocytes and round spermatids may utilize lactate secreted by Sertoli cells, as the main source of energy [13].

To fulfil these important physiological functions, Sertoli cells must have developed complex transport systems of ions and monocarboxylate acids. Indeed Sertoli cells express a variety of membrane transporters such as Na⁺/H⁺ and Na⁺-dependent Cl⁻/HCO₃⁻ exchangers and Na⁺/HCO₃⁻ co-transporters [9]. Electrophysiological and molecular analysis revealed the expression of several types of ionic channels, among them voltage-dependent Ca²⁺ channels [15], Ca²⁺-dependent Cl⁻ channels [16], CFTR Cl- channels [17] and voltage-dependent Cl- channels including rClC-2, rClC-3, rClC-6, rClC-7 [18]. In addition, we recently identified a voltage-dependent Clcurrent activated by extracellular acidic pH (hereafter noted IClacid) in rat Sertoli cells [18]. This novel acid-activated Clcurrent may be of particular interest for the physiology of Sertoli cells considering the importance of endogenous production of lactate [13, 14].

We report here the pharmacology of an outwardly rectifying Cl⁻ current activated by extracellular acid pH in rat Sertoli cells and compared some of its pharmacological properties to those of two other different endogenous Cl⁻ channels activated by extracellular ATP (ICl_{ATP}) or in response to osmotic swelling (ICl_{swell}). In particular, we found the sulfonylurea agent glibenclamide as a discriminator of ICl_{acid} compared to ICl_{ATP} or ICl_{swell} whereas 9-AC is a good discriminator of ICl_{ATP} compared to ICl_{acid} or ICl_{swell}. Collectively this information will help to promote our understanding of the biophysical characteristics and physiological roles of the endogenous Cl⁻ channels in the Sertoli cell, a cell at the blood-testis barrier.

2. MATERIAL AND METHODS

2.1. Preparation of rat Sertoli cells

Experiments were performed on cultured Sertoli cells isolated in sterile conditions from 13- to 15-day-old Wistar rats. Animals, raised at constant temperature (20 °C) under a 12L:12D light cycle, were killed in agreement with a procedure approved by the local animal ethics committee and the testes were removed aseptically. Then, the parenchyma of 6–10 testes were submitted at 34 °C to two enzymatic treatments by collagenase and pancreatin. The interstitial tissue was separated from tubules by incubation and continuous shaking (90 cycles/min, 10 min) in 25 mL medium A supplemented with 4.7 mg collagenase (Worthington Biochemical Corporation; 390 u·mg⁻¹). Medium A was a Ca^{2+} - and Mg^{2+} - free modified Earle solution containing (in mM) the following: 116.3 Na⁺, 5.4 K⁺, 121.7 Cl⁻, 0.9 $H_2PO_4^-$, and 5.5 glucose, supplemented with 53.5 mannitol and 20 Hepes and with streptomycin sulfate (100 mg \cdot mL⁻¹) and penicillin G (100 $IU \cdot mL^{-1}$) (Sigma). The pH was adjusted to 7.4 and the osmolarity to 300 mOsmol by mannitol. The medium was sterilized by filtration through a 0.22-µm-pore filter (Millipore, Molsheim, France). Tubules were then allowed to sediment for 5 min, and the supernatant was removed. Then they were resuspended

in 10 mL of medium A and shaken by hand for 10 s. The washing procedure was repeated four or five times. The tissue was minced with scissors and washed five times with 10 mL of medium A, then allowed to sediment for 5 min. The second dissociation was carried out to detach peritubular cells, comprising myoid cells and fibroblasts embedded in a collagen network, from the tubular wall. This was performed under continuous shaking (90 cycles/min, 5 min) in 25 mL of medium A containing 12.5 mg of pancreatin (grade VI; Sigma). The pancreatin solution was then discarded, and peritubular cells were separated from fragments of seminiferous tubules by gentle shaking in the presence of 10 mL of medium A (five times). Suspended fragments of free seminiferous tubules were discarded after centrifugation. Sedimented fragments were resuspended in 10 mL of medium A and passed about ten times through a syringe needle (17 gauge) at a slow rate. Cell suspension was adjusted to a volume of 10 mL with the following culture medium: DMEM/NUT mix F-12 (HAM) (Gibco BRL, UK) supplemented with insulin (0.01 mg·mL⁻¹; Sigma), BSA (1 mg·mL⁻¹; Sigma), transferrin (0.005 mg/ml; Sigma), streptomycin sulfate (100 mg·mL⁻¹; Sigma) and peni-cillin G (100 IU·mL⁻¹; Sigma). Final cell preparations were plated at low density in 35-mm plastic dishes (Nunclon) for patchclamp or in 24-well plates (5 \times 10⁵ cells) for iodide efflux experiments. After two days, a hypotonic shock (20 mM TrisHCl solution) was applied to eliminate germ cells. Culture dishes were incubated at 34 °C a further 2-6 days in a humidified CO₂ incubator (5% CO₂:95% ambient air). From day 2 on, the medium was renewed at 2-day intervals.

2.2. Electrophysiology

Whole-cell currents were measured in the broken-patch, whole-cell config-

uration of the patch clamp [19] using an RK300 patch clamp amplifier (Biologic, France). The holding potential was -40 mV in all whole-cell experiments. Current/voltage (I-V) relationships were built by clamping the membrane potential to -40 mV and by pulses from -100 mV to +100 mV in 20 mV increments. Pipettes with resistance of 3–5 M Ω were prepared by pulling borosilicate glass capillary tubes (GCL150-TF10, Clark Electromedical Inc., Reading, UK) using a two-step vertical puller (Narishige, Japan). They were connected to the head stage of the patch clamp amplifier through an Ag-AgCl pellet. Seal resistances ranging from 3 to 15 G Ω were obtained. Pipette capacitance was electronically compensated for in a cell-attached mode. Membrane capacitance (25.8 \pm 2.1 pF; n = 13) and series resistances $(3.4 \pm 1.1 \text{ M}\Omega; n = 13)$ were measured in the whole-cell mode by fitting capacitance currents, obtained in response to a hyperpolarization of 6 mV, with a firstorder exponential, and by integrating the surface of the capacitance current. They were not compensated. Voltage-clamp signals, allowing the membrane potential to be held at different values, were applied and recorded with the resulting currents by means of a microcomputer equipped with an analog/digital-digital/analog conversion board (TM-40; Tekmar, Cincinnati, OH). The results were analysed with the pCLAMP5.5 package software (pCLAMP, Axon Instruments). The external bath solution contained (in mM) the following: 161 NaCl, 5 CsCl, 2 CaCl₂, 2 MgCl₂ and 10 HEPES-NaOH (pH 7.4, 370 mOsmol). The intrapipette solution contained (in mM) the following: 1 NaCl, 39 CsCl, 81 CsGlutamate, 9.1 CaCl₂, 3 MgCl₂, 10 EGTA, 10 HEPES (pH 7.2, 300 mOsmol). The EGTA-calcium buffer was used to hold the intracellular free calcium concentration at 1 µM. The pipette solution was always hypotonic (with respect to the bath solution) to prevent cell

swelling and activation of the volumesensitive chloride channels [18]. All experiments were conducted at room temperature (20 °C). The calculated chemical equilibrium potential for chloride (E_{Cl-}) is -25 mV and for sodium (E_{Na+}) is +130 mV. The relative permeability of the channel to sodium and chloride (P_{Cl}:P_{Na}) was estimated from the reversal potential of the current-voltage relationship using the classical GHK equation. Osmolarity was corrected with mannitol. For hypo-osmotic challenge, we used the external bath solution in which the osmolarity was modified to 200 mOsmol. Extracellular pH was adjusted with HCl, lactic or aspartic acid as indicated in the text. The cells were stimulated with the appropriate compound at the concentration indicated in the text (dissolved in dimethyl sulfoxide Me₂SO; final concentration: 0.1%). All current traces shown in the figures are single traces.

2.3. Radiotracer efflux experiments

Chloride channel activity was assayed by measuring the rate of iodide (125I) efflux from a population of Sertoli cells. Cells were cultured in multiwell plates in order to perform parallel experiments and comparison analysis using a robot-based system (MultiProbe II[®], PerkinElmer Life Sciences, Courtabœuf, France) adapted to the iodide efflux method [20,21]. The cells were washed with an efflux buffer containing (in mM) the following: 136.9 NaCl, 5.4 KCl, 0.3 KH₂PO₄, 0.3 NaH₂PO₄, 4.2 NaHCO₃, 2 CaCl₂, 0.5 MgCl₂, 0.4 MgSO₄, 5.6 glucose and 10 HEPES, pH 7.4. The cells were then incubated in an efflux buffer containing 1 µM KI and 1 µCi Na¹²⁵I.mL⁻¹ (NEN, Boston, MA) during 1 h at 34 °C to permit the ¹²⁵I to reach equilibrium. The cells were then washed with an efflux medium to remove extracellular ¹²⁵I. The loss of intracellular ¹²⁵I was determined by removing the medium

with the efflux buffer every 1 min for up to 10 min. The first three aliquots were used to establish a stable baseline in the efflux buffer alone. A medium containing the appropriate drug was used for the remaining aliquots. Residual radioactivity was extracted with 0.1 N NaOH/SDS 0.1%, and determined using a gamma counter (CobraTM II, PerkinElmer Life Sciences, Courtaboeuf, France). The fraction of initial intracellular ¹²⁵I lost during each time point was determined and time-dependent rates (k, \min^{-1}) of ¹²⁵I efflux were cal-culated from $k = \ln ({}^{125}I_{t1}/{}^{125}I_{t2})/(t_1 - t_2)$ where ${}^{125}I_t$ is the intracellular ${}^{125}I$ at time t, and t_1 and t_2 are successive time points [22, 23]. Curves were constructed by plotting k versus time. To compare different sets of data from separate experiments, the relative rates (r) were calculated from $r = k_{\text{peak}} - k_{\text{basal}}$ (min⁻¹). In all experiments the peak rate (k_{peak}) corresponds to the maximum value of the rate of efflux whereas the basal rate (k_{basal}) represents the third point on the graphs except for Figure 4 where k_{basal} is the fourth experimental point. All inhibitors were pre-incubated during 30 min before the beginning of the experiment.

2.4. Chemicals

Glibenclamide, DPC (diphenylamide-2-carboxylic acid), 9-AC (Anthracene-9-Carboxylate), DIDS (4,4'-diisothiocyanatostilbene-2,2'-disulfonic acid), ATP, and L-(+)-Lactic acid were obtained from Sigma Chemicals (St Louis, MO). The inhibitor of outwardly rectifying chloride channels TS-TM calix[4]arene (5,11,17,23-tetrasulfonato-25,26,27,28-tetramethoxy-calix[4]arene) was generously given by Drs. Singh and Bridges (University of Pittsburgh, Pittsburgh, USA). For experiments using chloride transport inhibitors, DPC, DIDS, glibenclamide, 9-AC, were dissolved in dimethylsulfoxide Me_2SO (final concentration 0.1%). In

control experiments, the currents or efflux were not altered by Me_2SO .

2.5. Statistics

The results are expressed as means \pm SEM and *n* refers to the number of independent experiments. Mean statistical analysis was performed on raw data. In order to compare sets of data, we used either analysis of variance (ANOVA) or the student *t*-test. Differences were considered statistically significant when P < 0.05, ns: non significant difference, * P < 0.05, simon significant difference, * P < 0.05,

3. RESULTS

3.1. Activation of a chloride current by various extracellular acids in rat Sertoli cells

Figure 1 illustrates whole-cell recordings resulting from the stimulation of cultured rat Sertoli cells bathed in an extracellular medium maintained at pH 5 either by HCl (Figs. 1A and 1B) or lactic acid (Figs. 1C and 1D). In control experiments (i.e. in an extracellular solution at pH 7.4), we recorded a very small basal current (Figs. 1A and 1C, top traces). Pronounced outwardly rectifying currents were consistently activated in Sertoli cells (n = 21) following the perfusion of an extracellular solution maintained at pH 5 by HCl (Fig. 1A, bottom traces). Similar ionic currents were recorded in further cells (n = 9) after the perfusion of an extracellular solution maintained at pH 5 by lactic acid (Fig. 1C, bottom traces). In both experimental situations, the acid-induced currents were activated instantaneously by depolarising pulses followed by a further slow activation. Because the average predicted reversal potential of the current is -14 ± 1.3 mV (n = 30), a value near the equilibrium potential for Cl⁻ (E_{Cl⁻} = -25 mV), we concluded that Cl⁻ is the main ion carrying the current. Moreover, the calculated relative permeability between chloride and sodium (P_{Cl}:P_{Na}) fitted by the GHK equation is 6.05 ± 1.07 (n = 9) indicating that Cl⁻ is 6-fold more permeable than Na⁺. Therefore, this current will be hereafter denoted ICl_{acid}.

For each experimental condition, we expressed the amplitude of the current as the density of current (in pA/pF). Steady-state current-density curves of HCl or lactic acid-induced IClacid generated by our pulse protocol revealed a strong outward rectification (Figs. 1B and 1D). For example, the basal current density measured at +60 mV from 30 cells was 1.4 ± 0.15 pA/pF. The averaged current densities for ICl_{acid} measured at +60 mV were 67.45 ± 9.95 pA/pF (*n* = 21) and 69.65 ± 9.22 pA/pF(n = 9) with HCl and lactic acid, respectively. These values were significantly different from the basal value (P < 0.001). However, the amplitudes of IClacid stimulated either by HCl or lactic acid were not significantly different. There were also no significant differences between the calculated mean conductances for inward and outward currents (G_{in} at -60 mV and G_{out} , at +60 mV respectively) for cells stimulated by either HCl ($G_{in} = 1.1 \pm 0.2 \text{ nS}$; $G_{out} = 35.5 \pm 2 \text{ nS}; n = 9)$ or lactic acid $(G_{in} = 1.2 \pm 0.3 \text{ nS}; G_{out} = 36 \pm 4 \text{ nS};$ n = 9).

In a second set of experiments, we performed iodide (¹²⁵I) efflux to further study and compare the Cl⁻ transports after stimulating rat Sertoli cells by various extracellular acids. In control experiments, i.e. basal condition at pH 7.4, ¹²⁵I efflux relative rate (r) was 0.066 ± 0.007 min⁻¹ (n = 20). A significant efflux was observed (Fig. 1E) with cells incubated in an acid solution maintained at pH 5 either with HCl (n = 25), lactic acid (n = 20) or aspartic acid (n = 4). It can be seen from the



Figure 1. External acidification activates Cl⁻ channels in Sertoli cells. (**A**) Whole cell patch clamp recordings performed on rat Sertoli cells bathed with a NaCl-rich solution at pH 7.4 (up) and pH 5 (down). (**B**) The corresponding current density/voltage relationship is given for each condition (n = 21). Data are presented as mean \pm SEM. (**C**) Whole cell patch clamp recordings performed on rat Sertoli cells bathed with control solution at pH 7.4 (up) and with 9 mM lactic acid (down) to maintain a pH 5. (**D**) The corresponding current density/voltage relationship is given for these conditions (n = 9). (**E**) Time-dependent iodide efflux performed on rat Sertoli cells bathed with a NaCl-rich solution at pH 5 with HCl (n = 25), with 9 mM lactic acid (n = 20) and with 6 mM aspartic acid (n = 4). pH 7.4 corresponds to the basal condition (n = 20). k: rate of iodide efflux. (**F**) Bar graphs presenting the effects of HCl, lactic acid and aspartic acid on iodide efflux in Sertoli cells. All results are means \pm SEM. ns, no significant difference; *** P < 0.001.

time-dependent representation of the efflux rate (Fig. 1E) that the stimulation was almost immediate within the first minute after adding the acid (corresponding to t =3 min). The rate constants were maximum after 1 min and then decreased and returned to the basal value because more than 80% of the radiotracer had left the cell through the opening of acid-activated Cl⁻ channels. The stimulation either with HCl, lactic acid or aspartic acid reached a comparable maximum level (Fig. 1F). The corresponding relative rates were 0.127 \pm 0.007 min^{-1} (*n* = 25) for HCl, $0.132 \pm 0.018 \text{ min}^{-1}$ (*n* = 20) for lactic acid and $0.128 \pm 0.013 \text{ min}^{-1}$ (n = 4) for aspartic acid (black bars, Fig. 1F). By contrast, when the solutions containing HCl or lactic acid were buffered at pH 7.4, no stimulation could be detected. In this experimental condition, the calculated relative rates were $0.06 \pm 0.004 \text{ min}^{-1}$ (*n* = 25) for HCl, $0.07 \pm 0.004 \text{ min}^{-1}$ (*n* = 20) for lactic acid and $0.075 \pm 0.01 \text{ min}^{-1}$ (*n* = 4) for aspartic acid (open bars, Fig. 1F, P <0.001). Therefore these experiments and the whole-cell recordings presented above collectively demonstrate the existence of specific acid-activated Cl⁻ channels in cultured rat Sertoli cells.

3.2. Inhibition by glibenclamide, DIDS, DPC but not by TS-TM calix[4]arene nor 9-AC

One of the properties that distinguish Cl^- channels is the sensitivity to a panel of inhibitors. Several inhibitors were then used to decipher a pharmacological hallmark of the ICl_{acid} . In the absence of a specific inhibitor of Cl^- channels, we decided to use a range of chemically unrelated derivatives. The cells were stimulated at pH 5 and Cl^- channel inhibitors were added in the bath solution containing 2 mM extracellular Ca^{2+} .

Sulfonylurea glibenclamide, besides its well known property of inhibition of the

sulfonylurea receptor, has been shown to inhibit, at high concentration, Cl- channels including the CFTR Cl- channel [24-26]. Whole-cell patch-clamp recordings demonstrate the inhibition of IClacid (Fig. 2A) by 100 µM glibenclamide (Fig. 2B). The current amplitude measured at +60 mV in 4 Sertoli cells bathed in a solution at pH 5 was 3640 \pm 213.7 pA (Fig. 2C) and decreased to 1598 ± 197.4 pA (n = 4, Fig. 2C) in the presence of 100 µM glibenclamide. Previous work [18] has shown no-Ca²⁺ dependent activation of IClacid. Here, we tested the Ca²⁺ effect on ICl_{acid} inhibition. The concentration-dependent inhibition of the iodide efflux (Fig. 2D) indicates an $IC_{50} =$ 31 ± 1.2 μ M (n = 4) with 2 mM extra-cellular Ca²⁺ and an IC₅₀ = 39 ± 1.4 μ M in free-extracellular Ca²⁺ (n = 4). Because these two values are not significantly different, we conclude that ICl_{acid} is inhibited by glibenclamide in a Ca^{2+} -independent manner. Surprisingly, the effects of glibenclamide appeared different when measured using whole-cell recording or alternatively radiotracers (50% block with 0.1 mM using patch-clamp and 80% inhibition using iodide efflux). This discrepancy was due to the methods: (i) the cells were incubated with glibenclamide during 30 min before activation by pH 5 with the iodide efflux technique while in the patch-clamp technique the inhibitor was added following this activation, (ii) the cell number tested with these two techniques was different (one cell using a whole-cell vs. a cell population using an iodide efflux).

The two classical inhibitors DIDS and DPC were then tested (Fig. 3). Concentration-response relationships were constructed from separate experiments using increasing concentrations of inhibitors from 3 μ M to 300 μ M (n = 4 for each concentration). Half-maximal effective concentrations for inhibition (IC₅₀) of the acid-activated Cl⁻ transport were 27 \pm 1 μ M for DIDS and 86 \pm 5 μ M for DPC



Figure 2. Effect of glibenclamide on ICl_{acid} . (**A** and **B**) Whole cell patch clamp recordings performed on a rat Sertoli cell bathed with a NaCl-rich solution at pH 5 without (**A**) or with 100 μ M glibenclamide (**B**). (**C**) The corresponding current/voltage relationship is given for each condition (*n* = 4). (D) Concentration-response relationships between the concentration of glibenclamide and the percentage of maximal activation of the iodide efflux in Sertoli cells in the presence or absence of extracellular calcium. All results are means ± SEM of 4 experiments for each protocol.

(Figs. 3A and 3B). We observed an inhibition > 80% by 500 μ M of the disulfonic stilbene DIDS and > 77% by 500 μ M of the arylaminobenzoate DPC in the presence or absence of extracellular calcium (n = 4, Fig 3C). However, in the presence of 1 mM Anthracene-9-Carboxylate (9-AC) an inhibitor of ClC-1 [27] and other Cl- channels [25], no inhibition was found (n = 4, Fig. 3C). The sulfonated calixarene TS-TM calix[4]arene (5,11,17,23-tetrasulfonato-25,26,27,28-tetramethoxy-calix[4]arene) [28] is a highly potent inhibitor of outwardly rectifying Cl⁻ channels (so called ORCC). Figure 3C shows that 100 nM TS-TM calix[4]arene did not block the iodide efflux activated

by HCl in the presence of extracellular calcium. Thus the Cl⁻ channel inhibitors DIDS, DPC and glibenclamide inhibit ICl_{acid}, independently of the presence of extracellular calcium. The profile for inhibition of ICl_{acid} was therefore DIDS \approx glibenclamide > DPC >> TS-TM calix[4]arene, 9-AC. The same inhibitory profile was obtained when ICl_{acid} was stimulated by lactic acid (Fig. 3D).

3.3. Pharmacological comparison of Cl⁻ transports regulated by acid, cell volume and ATP

A global analysis of the Cl⁻ channels in Sertoli cells using iodide efflux techniques





Figure 3. Effect of DIDS, DPC, 9-AC and calixarene on the iodide efflux. (**A** and **B**) Concentrationresponse relationships for DIDS (**A**) and DPC (**B**). The data are expressed as a percentage of maximal activation of the iodide efflux in Sertoli cells in the presence of different concentrations of the inhibitor. (**C**) Histograms presenting the effect of 1 mM 9-AC, 500 μ M DPC, 500 μ M DIDS and 100 nM calixarene on HCl activation in the presence or absence of extracellular calcium. (**D**) Histograms presenting the inhibitory profile of the pH-induced chloride current stimulated by lactic acid in the presence or absence of calcium. All results are means ± SEM of 4 experiments for each protocol. *ns*, no significant difference; *** *P* < 0.001; ND, not determined.

and whole cell patch-clamp has revealed the presence of several other Cl⁻ conductances [18]. A molecular and whole cell patch clamp analysis of rat Sertoli cells already identified the following voltagedependent Cl⁻ channels: rClC-2, rClC-3, rClC-6, rClC-7 and a volume-activated Clcurrent (referred as ICl_{swell}) [18]. ICl_{swell} is activated following a hypo-osmotic challenge with a voltage-dependent activation at depolarizing pulses followed by a pronounced inactivation [18]. In addition, because the nucleotide ATP induces an increase of the intracellular calcium concentration $[Ca^{2+}]_i$ [16,29], we investigated the effect of 100 µM ATP on the Cl- currents in Sertoli cells.

Iodide efflux experiments were performed to compare Cl⁻ transports activated by ATP or hypo-osmotic challenge. Figure 4 shows that the two experimental conditions, i.e. a hypo-osmotic challenge or addition of 100 µM ATP in the extracellular bath induced two different responses in Sertoli cells. With cells exposed to a hypo-osmotic solution, the iodide efflux was stimulated within the first minute but with a maximal response two minutes after the addition of the hypo-osmotic solution with a relative rate of 0.180 \pm 0.037 min^{-1} (n = 6, Fig. 4A). This response was not inhibited by 9-AC, the relative rate was $0.167 \pm 0.055 \text{ min}^{-1}$ (*n* = 8, Fig. 4A). With cells exposed to $100 \,\mu\text{M}$ ATP in the extracellular bath, the iodide efflux was stimulated with a relative rate of $0.121 + 0.025 \text{ min}^{-1}$ after one minute (faster than with the hypo-osmotic solution) followed by a rapid decline of the iodide efflux (n = 12, Fig. 4C), respectively. The control relative rate calculated in the absence of stimulation was $0.014 \pm$ 0.002 min^{-1} (n = 22). The differences between the two experimental conditions with respect to the control was found to be statistically significant (P < 0.001). 9-AC produced a significant inhibition of the relative rate $(0.052 \pm 0.002 \text{ min}^{-1})$, P < 0.01; n = 8, Fig. 4C). We then compared the effect of glibenclamide (100 µM), DPC (500 µM) and 9-AC (1 mM) on the ATPand hypo-osmotic-activated iodide efflux. Glibenclamide neither inhibits the hypoosmotic-activated iodide efflux (n = 10, Fig. 4B) nor the ATP-activated iodide efflux (n = 14, Fig. 4D). Whereas 500 μ M DPC (Fig. 4B and D) and 500 µM DIDS (not shown) inhibited both effluxes, the compound 9-AC only inhibited the ATPactivated iodide efflux (Fig. 4D) but not the hypo-osmotic-activated iodide efflux (Fig. 4B) (n = 8 for each experimen-)tal condition). Figure 4E shows a typical whole-cell experiment identifying outwardly rectifying Cl⁻ currents stimulated by 100 µM ATP added to the bath perfusion and presenting a slow activation at positive potentials. This current denoted IClATP is fully blocked by 500 µM DPC (Fig. 4F). By contrast, with a bath perfusion containing the adenylate cyclase activator forskolin (10 µM) used to increase the cell cAMP level, no further Cl⁻ currents could be stimulated (not shown). We determined that the inhibition profile was DIDS \approx DPC = 9-AC >> glibenclamide for the ATP-activated Cl⁻ transport and DIDS \approx DPC >> glibenclamide, 9-AC for the hypo-osmotic response.

4. DISCUSSION

The present study was designed to characterize the pharmacological properties of the acid-activated Cl⁻ current, ICl_{acid}, recently identified in cultured rat Sertoli cells and to compare its pharmacological signature to that of two other endogenous Cl⁻ channels activated by extracellular ATP (ICl_{ATP}) and hypo-osmotic solution (ICl_{swell}). The principal findings of this study were (i) ICl_{acid} can be stimulated by various acid species (HCl, aspartic acid and the physiologically relevant lactic acid) suggesting that the most





Figure 4. Stimulation of Cl⁻ channels by 100 μ M ATP and hypo-osmotic challenge. (**A**) Iodide efflux performed on rat Sertoli cells bathed with a hypo-osmotic challenge (n = 6) and in the presence of 1 mM 9-AC (n = 8). (**B**) Bar graphs presenting the effects of 100 μ M glibenclamide, 500 μ M DPC and 1 mM 9-AC on the hypo-osmotic-stimulated efflux. (**C**) Iodide efflux from rat Sertoli cells bathed with 100 μ M ATP (n = 12) and in the presence of 1 mM 9-AC (n = 8). (**D**) Bar graphs presenting the effects of 100 μ M glibenclamide, 500 μ M DPC and 1 mM 9-AC (n = 8). (**D**) Bar graphs presenting the effects of 100 μ M glibenclamide, 500 μ M DPC and 1 mM 9-AC (n = 8). (**D**) Bar graphs presenting the effects of 100 μ M glibenclamide, 500 μ M DPC and 1 mM 9-AC on the ATP-mediated efflux. (**E**) Whole cell patch clamp recordings performed on a rat Sertoli cell without (noted basal) or with 100 μ M ATP in the bath medium. The full inhibition of the outwardly rectifying Cl⁻ current by 500 μ M DPC is illustrated on the right traces. (**F**) The corresponding current density/voltage relationship is given for each condition (n = 4). Data are presented as mean \pm SEM. *ns*, no significant difference; ** P < 0.01.

important determinant is the concentration of protons itself rather than the nature of the acid; (ii) the profile for inhibition of ICl_{acid} was DIDS \approx glibenclamide > DPC \gg TS-TM calix[4]arene, 9-AC; (iii) an ATP-activated Cl⁻ channel ICl_{ATP} was inhibited by DIDS \approx DPC \approx 9-AC \gg glibenclamide; (iv) a swellingactivated Cl⁻ channel ICl_{swell} was inhibited by DIDS \approx DPC \gg 9-AC, glibenclamide. Thus glibenclamide appears to be a relatively good discriminator of ICl_{acid} compared to ICl_{ATP} or ICl_{swell} whereas 9-AC is a good discriminator of ICl_{ATP} compared to ICl_{acid} or ICl_{swell}.

Several members of the voltagedependent ClC family of Cl⁻ channels are activated, inhibited or gated by pH [27,30]. In rat Sertoli cells, electrophysiological and molecular analysis have revealed the expression of several types of Cl⁻ channels among them Ca²⁺-dependent [16], swelling-activated [18] and four voltage-dependent Cl⁻ channels, i.e. ClC-2, ClC-3, ClC-6 and ClC-7 [18]. We have already ruled out the possibility that any of these members could contribute to ICl_{acid} [18]. To our knowledge, there are only few examples of native Cl⁻ channels activated by external protons. The best studied pH-sensitive Cl⁻ current is the inwardly rectifying Cl⁻ channel ClC-2 for which numerous studies on both native and cloned channels indicated an extreme sensitivity to extracellular protons and an inhibition by Cd²⁺ [30–34]. A recent study characterized an endogenous Cl- current in transformed human embryonic kidney HEK-293 cells, activated by extracellular pH [35]. The corresponding current shares some characteristics with the IClacid found in Sertoli cells, i.e an outward rectification with a slow voltage-dependent activation, a lack of dependence on intraor extracellular calcium, blocking by DIDS and DPC [18, 35]. However, we determined a different anionic selectivity i.e. Cl⁻>Br⁻>I⁻>gluconate [18], whereas

in HEK-293 the current has the following selectivity I^{->} Br^{->} Cl^{->}gluconate [35]. Moreover in their conclusions, these authors suggested that IClacid and IClswell could represent two different transition states of the same channel suggesting that the extracellular acidification could reflect a novel stimulus for activating ICl_{swell} resulting in an alteration of its biophysical properties [35]. In Sertoli cells, this hypothesis is unlikely, both IClacid and ICl_{swell} co-exist but we clearly determined different biophysical properties (slow voltage-dependent activation for IClacid, fast voltage-dependent activation, and time-dependent inactivation for ICl_{swell}) as well as different pharmacological properties (glibenclamide-sensitivity for IClacid, glibenclamide-insensitivity for ICl_{swell}) (this study). Lambert and Oberwinkler [36] identified the proton-activated, outwardly rectifying currents expressed in HEK293 cells. They also described distinct and separable properties from those of the volume regulated anion currents, even under identical experimental conditions. These authors concluded the proton-activated anion currents and the volume-regulated anion currents are linked to physically distinct populations of ion channels. To clarify if IClacid and IClswell are provided by two different channels, it is necessary to identify the molecular properties of the proteins responsible for these two conductances.

Previous works have shown that an acidic pH is required for the maturation process of sperm cells [37, 38] and that the secretion of a large amount of lactate by Sertoli cells could be responsible for the establishment of the acid environment [39, 40]. In the present study, we characterized in rat Sertoli cells some elements of the pharmacology and regulation of novel outwardly rectifying Cl⁻ channels, ICl_{acid}, activated by extracellular acid, including lactic acid. We have previously described the pH-dependence of this current

and we estimated the half-maximal stimulation around pH 5.5 [18]. Although it is premature to conclude that ICl_{acid} plays a role in the physiology of Sertoli cells regarding the regulation of lactic acid, it nevertheless represents a good potential candidate. Moreover, another interesting issue is that ICl_{acid} is noticeable only at positive voltages. Further studies will be necessary to clear this point.

Finally, the hormonal regulation of the secretion of a Cl⁻ and K⁺-rich fluid by Sertoli cells is important in spermatogenesis. It involves multiple signaling pathways, including regulation of adenyl cyclase with increasing or decreasing cAMP levels, calcium homeostasis and purinergic receptor stimulation [41]. Many endogenous ion channels have been described in Sertoli cells [41] but sparse information concerning their role in the physiology of Sertoli cells and spermatogenesis are available. The identification and pharmacological characteristics of the three types of Cl⁻ channels identified in our rat Sertoli cell model may help to promote our understanding of the biophysical characteristics and physiological roles of ion channels in the Sertoli cell, a cell at the blood-testis barrier.

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