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Anne Poszwa, Bruno Ferry, Benoît Pollie, Catherine Grimaldi, Pierre Charles-Dominique, et al.. Variations of plant and soil $^{87}\text{Sr}/^{86}\text{Sr}$ along the slope of a tropical inselberg. *Annals of Forest Science*, 2009, 66 (5), pp.1-13. <10.1051/forest/2009036>. <hal-00883478>

HAL Id: hal-00883478

<https://hal.science/hal-00883478v1>

Submitted on 11 May 2020

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Variations of plant and soil $^{87}\text{Sr}/^{86}\text{Sr}$ along the slope of a tropical inselberg

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(Received 30 July 2008; accepted 15 December 2008)

Keywords:

inselberg /
nutrient cycling /
 $^{87}\text{Sr}/^{86}\text{Sr}$ /
tropical soil /
rainforest

Abstract

- From the summit downslope a granitic inselberg in French Guiana, soils and vegetation evolve from bare granite covered by cyanobacteria, to a savannah-type vegetation on thin patchy sandy accumulations, then to a low forest on shallow young soils and to a high forest on deep highly weathered ultisols.
- We have used element budgets and Sr isotopic variations in soils and plants to investigate the mineral nutrient supply sources of the different plant communities.
- Granite and atmospheric deposition have $^{87}\text{Sr}/^{86}\text{Sr}$ ratios of 1.3 and 0.71, respectively. The $^{87}\text{Sr}/^{86}\text{Sr}$ ratio of cyanobacteria (0.72) suggests granite weathering by cyanobacteria crusts. The $^{87}\text{Sr}/^{86}\text{Sr}$ ratio of the savannah-type vegetation is 0.73 and varies between 0.75 and 0.76 in the low and high forest leaf litter regardless of soil depth, age and degree of impoverishment.
- These almost constant ratio suggest that forest Sr comes from rainwater and from the summit of this inselberg, where it is released and redistributed along the slope, by surface flow, lateral redistribution of litter, and mineral particles. However, because of its very low content in the rock and soils, Ca is supplied to plants by atmospheric deposition.

Mots-clés :

inselberg /
cycle des nutriments /
 $^{87}\text{Sr}/^{86}\text{Sr}$ /
sols tropicaux /
rainforest

Résumé – Variation des rapports $^{87}\text{Sr}/^{86}\text{Sr}$ dans les sols et les plantes d'un inselberg tropical.

- Du sommet vers la base d'un inselberg granitique (Nouragues, Guyane Française), les sols et la végétation évoluent depuis des savanes sur des îlots sableux entre les affleurements rocheux couverts de cyanobactéries, vers une forêt basse sur sols peu épais, riches en minéraux altérables puis une forêt haute sur sols très profonds et altérés.
- Les variations isotopiques du strontium des sols et des plantes ont été mesurées pour rechercher les sources de nutriments des différentes communautés végétales.
- Les rapports $^{87}\text{Sr}/^{86}\text{Sr}$ du granite et des dépôts atmosphériques sont respectivement de 1,3 et 0,71. Le rapport $^{87}\text{Sr}/^{86}\text{Sr}$ des cyanobactéries (0,72) suggère une libération de Sr par altération du granite. Le rapport $^{87}\text{Sr}/^{86}\text{Sr}$ de la savanne est de 0,73 et varie entre 0,75 et 0,76 dans les litières de forêt basse et haute, quelle que soit la profondeur, et la richesse en minéraux altérables des sols.
- La faiblesse et l'homogénéité surprenante de ces rapports suggèrent une alimentation en Sr des forêts essentiellement à partir de dépôts atmosphériques et des sols de la partie haute de l'inselberg, via des écoulements de surface, des redistributions latérales de litière et de particules minérales lors de crises érosives. Cependant, en raison de l'extrême pauvreté de la roche et des sols en calcium, le Ca des communautés végétales provient de la pluie.

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1. INTRODUCTION

Two primary sources provide mineral nutrients to forested ecosystems: the atmosphere and soil mineral weathering. The sustainability of forest ecosystems depends on these primary fluxes, as well as on the efficiency of nutrient uptake and cycling within the ecosystems. Tropical rainforests at low elevations generally cover deep highly weathered acidic soils, extremely impoverished in nutrients. The common view is that plants take up most of their nutrient requirements from the upper horizons, recycling the elements released by litter decomposition (Jordan, 1982) and atmospheric deposition (Chadwick et al., 1999). But the possible contribution of erosion processes bringing fresh rock fragments or less impoverished soils horizons close to the surface has recently been stressed (Bern et al., 2005; Porder et al., 2005).

Inselbergs are large rock outcrops dominating tropical landscapes. The savanna-like vegetation associated to these landforms is rather uncommon in present tropical landscapes but was much more widespread during the Würm period (22 000–13 000 BP) and the shorter dry period of the Holocene (De Granville, 1982) especially in South America. The present persistence of savannahs is explained by large-scale forest fires which took place during the last few thousand years (Charles-Dominique et al., 1998) leading to forest opening and recurrent erosive crisis (Rosique et al., 2000). Thus, forests surrounding inselbergs are aggrading forests, tending to colonize inselberg slopes, where soil formation is initiated. Along the slope of inselbergs, soil depth, age and properties evolve gradually from bare rock patches and thin eroded soils at the summit to deep highly weathered soils in the high forest. These small scale variations in soil thickness and microclimatic conditions lead to an extreme variation in plant diversity (Burke, 2001; Larpin et al., 2000; Porembski and Barthlott, 2000; Sarthou and Villier, 1998), and offer a unique opportunity to study the contribution of rock and soil mineral weathering to plant nutrition.

Tracers, and especially isotopic tracers, are useful when the contribution of different sources to tree supply is to be established. Strontium (Sr) is not essential to the plant, but the chemical structures of calcium (Ca) and strontium are close and Sr concentrations correlate with Ca concentrations in stream waters and plant organs (Poszwa et al., 2000). In rocks, ^{87}Sr derives from the natural radioactive decay of ^{87}Rb (half-life $4.88 \cdot 10^{10}$ y, Faure, 1986). Rubidium is abundant in minerals rich in potassium such as K-feldspar and biotite. The ^{86}Sr isotope is non radiogenic and primarily found in minerals rich in Ca such as plagioclase and apatite. As a result, the $^{87}\text{Sr}/^{86}\text{Sr}$ ratio of the Sr released by mineral weathering is high in old rocks, which are rich in K and Rb and poor in Ca, such as granites. In comparison, the $^{87}\text{Sr}/^{86}\text{Sr}$ ratio of atmospheric deposition is low and close to the oceanic $^{87}\text{Sr}/^{86}\text{Sr}$ ratio (0.709). This difference has been used to quantify the relative contribution of atmospheric deposition and mineral weathering to base cation fluxes in various forested ecosystems (Åberg 1995; Bailey et al., 1996; Capo et al., 1998; Graustein, 1989; Graustein and Armstrong, 1983; Miller et al., 1993).

The strontium isotope method was applied in the Nouragues natural reserve in French Guiana, dominated by a large granitic inselberg. The ecology of this pristine area has been the focus of a large research program for many years (Bongers et al., 2001). In this study, the nutrient content and $^{87}\text{Sr}/^{86}\text{Sr}$ isotopic ratio of selected and representative plant leaves, bulk litters and soils along the slope of the inselberg were measured in order to compare the contributions of atmospheric deposition and weathering to the different plant communities. Our hypothesis was that the contribution of the rock to plant Sr and other nutrients would decrease progressively from the communities living on the shallow soils of the summit to the high forest on the lower part of the slope, as soils get deeper and more depleted in weatherable primary minerals.

2. MATERIALS AND METHODS

2.1. Study site

The study was carried out at the Nouragues Reserve research station ($4^{\circ} 3' \text{ N}$, $52^{\circ} 40' \text{ W}$, French Guiana), 100 km west of the Atlantic coast, in the “Montagnes Balenfois” massif. The dense pristine tropical rainforest is dominated by an isolated and partly denuded granitic inselberg, which emerges from a plateau of deeply weathered metavolcanic rocks of the Paramaca formation (Choubert, 1974). The inselberg bedrock is a 1.9 to 2.1 Ga old “Caribbean” granite (Depagne and Lelong, 1965; Gruau et al., 1985). This rock is a pinkish monzonitic-type granite. The mineral content is 76% SiO_2 , 12.9% Al_2O_3 , 0.7% FeO , 4–5% K_2O , 3.8–4.2% Na_2O , 0.12% MgO , 0.45% CaO , and 0.1% P_2O_5 (Sarthou and Grimaldi, 1992; this work). The major minerals are quartz, microcline and albite. Apatite and muscovite are accessory minerals. Average annual precipitation is 3000 mm with large inter-annual variations (range: 2500–4000 mm) (Boyé et al., 1979; Grimaldi and Riera, 2001). Rain occurs on average 280 d y^{-1} , with two drier periods, one from September to October and a shorter one in March. The relative humidity is 20% in the dry season and 100% in the rainy season (Sarthou and Grimaldi, 1992) and the average daily temperature is $25\text{--}27^{\circ}\text{C}$. However, the temperature of bare rock surfaces may vary widely ($18\text{--}55^{\circ}\text{C}$) diurnally (Rascher et al., 2003; Sarthou and Grimaldi, 1992), because of the dark cyanobacteria covering these surfaces.

The inselberg is dome-shaped and elongated along the SE-NW axis. From the top of the inselberg, the landscape is composed of four main units (Fig. 1a):

- The “rock-savannah” unit (Sarthou, 1992) extends from the summit (411 m) to the high terrace (200–140 m). The relief of this unit is a smooth pattern of domes and small depressions, forming elementary micro-catchments, each of them up to several hundred square meters in size. The bare rock is covered by a black continuous crust of cyanobacteria. Species include (Rascher et al., 2003; Sarthou et al., 1995): *Scytonema guyanense*, *S. multiramosum*, *S. myochrous*, *Stigonema flexuosum*, *S. hormoides*, *S. ocellatum*, *S. panniforme*, *Hapalosiphon fontinalis*, *H. luteolu* and *Gloeocapsa sanguinea*, most of which are nitrogen fixers (Dojani et al., 2007). The endolithic cyanobacteria *Choococidiopsis* sp. (Rascher et al., 2003) penetrate the granitic surface to 1–5 mm in depth. Small depressed areas are occupied by dense patches of *Bromeliaceae* mats of varying size,

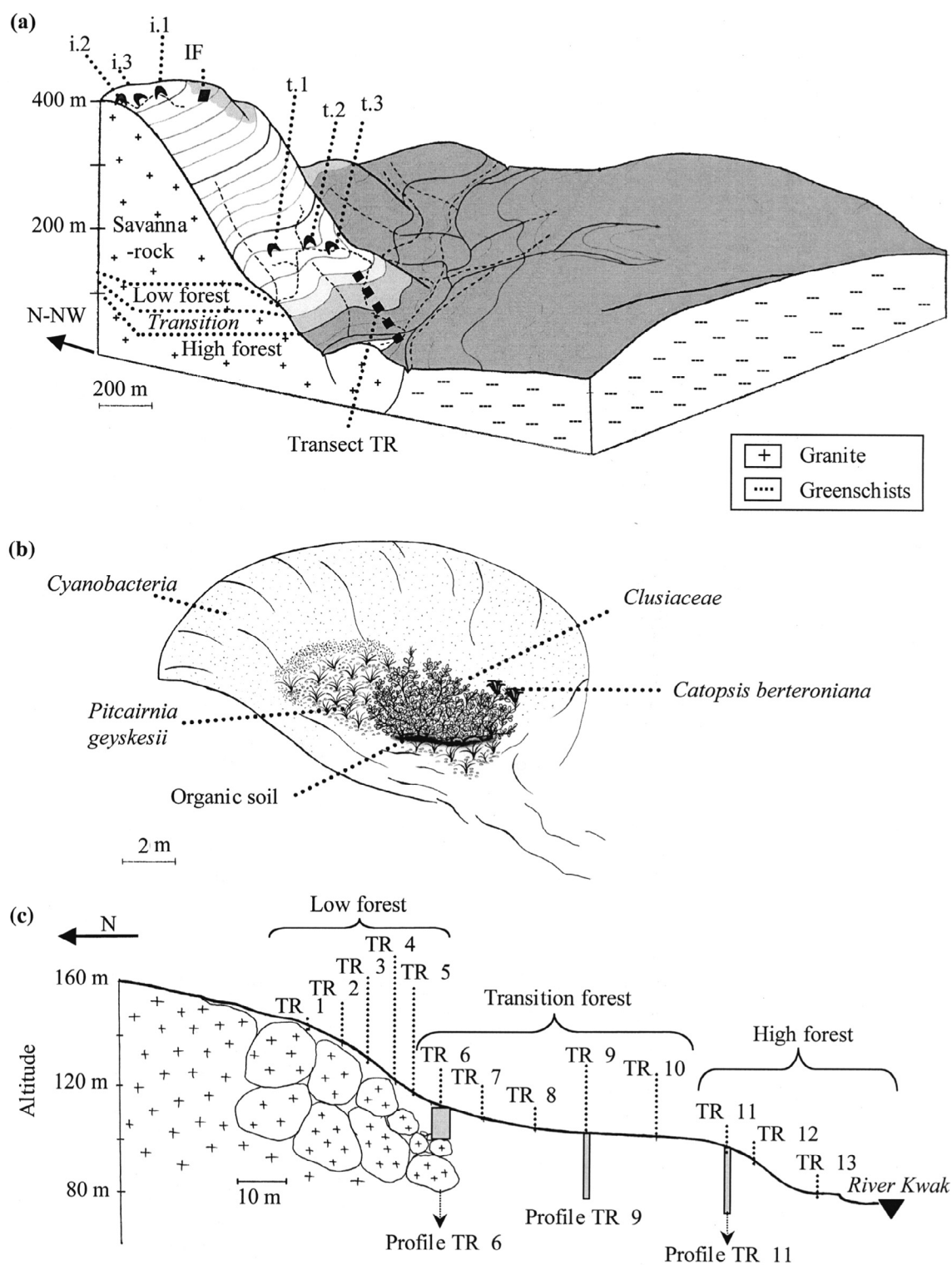


Figure 1. Nouragues site. (a) General view of relief, vegetation units and sampling locations on the Nouragues inselberg; (b) Savannah rock main vegetation sampling locations on depressions at the summit of the inselberg and on the high terrace; (c) Soil and litter sampling locations on the transect (TR) from the high terrace to the basis of the inselberg.

mainly represented by *Pitcairnia geyskesii* and shrubs of *Clusiaceae* (*Clusia minor* and *C. nemorosa*) and *Myrtaceae* (*Myrcia saxatilis* and *M. guyanensis*...) (Sarhou and Villiers, 1998). Black organic cyanobacteria residues generally accumulate at the upper edge of these patches (Vačulik et al., 2004). The soils are very thin and acidic (pH 4–5.5), formed by *Pitcairnia* and *Clusiaceae* litter accumulation over sandy accumulations (0–20 cm thick). During rain events, waters running on bare rocks downslope from these patches are especially enriched in K and N compared to rainwater (Sarhou and Grimaldi, 1992; Dojani et al., 2007).

- the low forest (tree height up to 8–15 m) extends between the rock savannah and the high forest, as a continuous belt less than 100 m wide, from an elevation of 140 down to 120 m. It also occurs as isolated inclusions within the high forest and the rock savannah, for example on the northeastern flank close to the summit (Fig. 1a). The main canopy species are *Eriotheca surinamensis* (*Bombacaceae*) and large individuals of *Clusia minor*. The understory vegetation is composed of the herb *Stelestylis surinamensis*, numerous epiphytes, as well as shrubs and small trees predominantly belonging to the *Rubiaceae*, *Myrsinaceae* and *Nyctaginaceae* families. Close to the summit, soils below the low forest are very shallow and vary from sandy to sandy-clay colluvium. On the slope, yellowish brown sandy clayed soils with many weathered granite boulders develop below the low forest.
- The low forest evolves downslope through a transition forest (> 15 m height) to a high forest (height > 30 m) on slopes above the river Nouragues. In the transition forest, many species, which have a multi-stemmed habit in the low forest, show tall and vertical stems. Some species typical of the high forest occur, but typical features of the low forest, like the abundance of herbs, persist. Downslope, in the high forest, only 25% of the low forest species are still found. The total number of tree species is 550, and includes 63 families (Poncy et al., 2001). *Eschweilera coriacea*, *Vouacapoua americana* and *Eperua falcata* are the most abundant. Soils below the transition and high forest are red ultisols, more than 3 m deep, depleted in base cations (pH 4.1–4.8).
- On the other side of the Kwak river, the Grand Plateau (GP) is covered by a pristine high forest. Soils derived from metavolcanic rocks are thick reddish ultisols extremely rich in Al-Fe nodules (Grimaldi and Riéra, 2001).

2.2. Sampling

The field work was conducted during the dry season, in October 1999. We sampled cyanobacteria crusts on bare rock edges (Fig. 1b) in three small depressions (i.1, i.2, i.3) at the summit of the inselberg (at about 400 m), and three others (t.1, t.2, t.3) on the high terrace (200–140 m), where the typical savannah-rock vegetation was present. Rain supplies water directly to the cyanobacteria on the summit of the inselberg whereas the cyanobacteria located on the high terrace may also receive a fraction of the surface flow draining along rock gullies throughout the savannah-rock vegetation unit. Leaves from different plants of *Pitcairnia geyskesii* were collected in the small depressions. Leaves of *Clusiaceae*, downstream of *Pitcairnia* mats, were sampled, all around shrubby thickets. Leaves of some epiphytes (*Catopsis berteroniana*) growing on the *Clusiaceae* were also sampled. In each depression, mineral soils associated with *Pitcairnia* mats and organic soils under *Clusiaceae* were sampled.

Because of the high plant diversity, we collected composite samples (100 g) of fresh leaf litter from the soil surface, within circular areas ten meters in diameter. Two leaf litter samples were collected within a patch of low forest (IF) on the North Eastern summit of the inselberg. From the upper limit of the high terrace, at an elevation of 140 m, leaf litter was sampled along a several NW-SE transects (TR, Fig. 1a) below the low forest (TR1 to TR6), below the transition forest (TR7 to TR10) and below the high forest, (TR11 to TR13) down to the bank of the Kwak river (TR13). In addition, leaf litter was sampled on the other side of the Kwak river, below the high forest growing on thick ultisols developed from metavolcanic rocks (site GP).

Below each type of forest, we dug and described characteristic soil profiles at the centre of some of the circular areas where leaf litter were collected (see Fig. 1c). The stone content was visually assessed in each profile. Soil bulk density was not measured. To estimate exchangeable and total element stores in the fine earth, we assumed density values from 1 in surface horizons to 1.4 at depth. On the northeastern summit of the inselberg, at site IF, in a deep bleached sandy colluvium where the groundwater level rises close to the soil surface in the rain season, soils were sampled per horizon, down to 1.80 m; after sieving, fine roots (< 2 mm) were separated from the bulk soil at each depth. Small patches of soils between boulders in the low forest transect were not sampled. Down the slope, granite boulders progressively disappear from the ground surface, as the transition forest appears. In the lower part of the low forest (profile TR6), soils were sampled per horizon, down to the weathered bedrock (80 cm), sieved in the field and fine roots (diameter smaller than 2 mm) were separated. Between 120 and 100 m, below the high forest, soils were also sampled at two sites (TR9 and TR11) with an auger down to 280 cm (granite was not met at this depth). One soil sample (10–20 cm) from site GP (on metavolcanic rocks) was also taken. Three rock samples were taken. From the summit of the inselberg, two fragments of granite were broken from large solid isolated boulders, and we sampled the heart of these fragments. From Profile TR6, on the slope, a piece of weathered granite boulder was taken at –80 cm depth. One rainwater, one gully-water and two groundwater (from pit IF) samples were collected at the summit of the inselberg. Seven water samples, all of them draining the granitic inselberg, were taken from the main river Nouragues, from several springs in the high forest, and from a tributary of the Kwak river.

2.3. Soils and vegetation analysis

Soil texture, pH, carbon (C), nitrogen (N) and C/N ratio were determined by conventional methods. Soil exchangeable base cations from all soil samples were extracted by two sequential extractions of 10 g of soil (< 2 mm) in 50 mL 1 M NH_4Cl . Aliquots of the fine earth (< 2 mm) from the major horizons and bulk rock samples were ground in an agate mortar. Mineralogy of the fine earth was assessed for all soil horizons by X-ray diffraction. In addition, minor mineral phases enriched in Sr, were investigated in the granite and in several soil horizons using a scanning electron microscope (SEM) coupled to an energy dispersive spectrometer (EDS). Ground samples from selected horizons of profiles IF, TR6 and TR11 (Tab. I), were digested overnight with 1 mL of HNO_3 purified by double distillation and 1 mL of high purity HF. After evaporation, to ensure total digestion of the samples we added a few drops of HClO_4 and H_2O_2 , to remove Ca fluorides and organic matter residues. Cyanobacterial crusts were carefully detached from bare rock surfaces and adhering mineral

Table I. Total and exchangeable element concentrations in profiles IF, TR6 and TR11.

Profile	Depth (cm)	Exchangeable			Total										
		(*mg kg ⁻¹ **μg kg ⁻¹)			(*mg g ⁻¹ **μg g ⁻¹)										
		Ca*	Sr**	K*	Ca *	Sr**	K*	Na*	Ba**	Rb**	Si*	Al*	Fe*	Ti*	⁸⁷ Sr/ ⁸⁶ Sr
Granite	Inselberg				2.8	23.8	17.8	13.95	72.6	161	363	33.8	2.8	0.59	1.275
IF	0–10	46.51	695.4	36.1											
10–30	2.80	6.4	11.1	0.05	10.5	0.4	< dl	10.9	6.9	435	7.4	1.3	3.04	0.758	
150–180	2.83	5.1	4.0	0.20	9.0	1.4	< dl	11.8	20.0	418	10.5	5.1	3.31	0.868	
TR 6	0–5	34.77	476.1	66.1											
10–20	4.77	48.9	28.5	0.16	14.1	9.7	0.26	121	88.9	301	46.9	15.4	3.53	1.142	
30–50	2.04	34.7	13.8	0.52	13.8	8.3	0.26	107	77.9	302	50.2	16.5	3.47	1.094	
80	0.67	154.8	11.8	0.66	28.8	18.8	3.12	309	168.7	214	83.3	17.3	3.12	1.150	
W. rock	8.76	990.6	18.1	2.01	39.7	19.2	8.38	319	165.3	348	36.6	5.7	1.06	1.041	
TR 11	0–10	5.65	73.7	32.5											
10–20	3.38	48.9	18.9	0.08	9.2	0.8	< dl	23.6	12.6	289	54.6	20.2	3.53	0.783	
20–60	1.98	14.8	11.7	0.10	10.5	0.7	< dl	23.4	12.5	284	61.7	23.1	3.69	0.776	
100–140	0.50	0.17	5.5												
GP	10–20	10.56	49.0	10.4	0.10	33.1	0.3	< dl	34.1	5.2	112	81.3	84.8	15.27	0.714

grains were removed using a paintbrush under a magnifying glass. The fine roots collected in profiles IF and TR 6 were cleaned of soil particles using ultra pure water, and a paintbrush. Roots, cyanobacteria, *Bromeliaceae*, *Clusiaceae* and epiphyte leaves as well as leaf litter samples were dried at 65 °C in an oven, then finely ground. The total digestion of 0.5 g of each sample was performed in 10 mL of high purity HNO_3 (Merk suprapur, 65%), in a digestion block system (Tecator 40 1016).

Calcium, Sr, Mg, P, and K in vegetation samples; Ca, Sr, Mg, K, Al, and Mn in soil NH_4Cl extracts; and Ca, Sr, Mg, K, and Si in rain and river samples were analysed by ICP AES. Major and trace element contents of rock and soil samples were measured after acid digestion by ICP-MS. For isotopic measurements, aliquots of samples in solution (NH_4Cl extract, water, rock, plant and soil digest by HNO_3) containing about 500 ng of Sr, were evaporated, treated with high purity HNO_3 and H_2O_2 in order to remove organic matter remains, then evaporated to a solid in Teflon beakers. Samples were then dissolved with HCl 2 M and passed through quartz glass ion exchange columns containing 8 mL of a Temex 50WX8, 200–400 mesh resin. Strontium was collected after sequential elution with 2 M and 4 M HCl and evaporation. Strontium contents in chemical blanks were lower than 2 ng. The $^{87}\text{Sr}/^{86}\text{Sr}$ isotopic ratio of samples was measured by mass spectrometry (Finnigan Mat MS 261). NBS 987 standard was used to control and standardize the measurements. Measurement accuracy was always below 2×10^{-5} .

2.4. Determination of element losses by mineral weathering

Element losses during the process of rock weathering and soil formation are usually calculated using a conservative tracer, which is least affected by dissolution and leaching processes, such as titanium (Aubert et al., 2001), zirconium (Bain and Bacon, 1994; Hodson, 2002), thorium (Braun et al., 1993) or niobium (Porder et al., 2006). The more immobile the element, the more its concentration increases

from the parent material to the soil surface. Titanium (Ti) concentration was found to increase more than zirconium, niobium and thorium in these soils, and therefore was used to estimate mineral element losses by weathering from the granitic rock. Assuming the only source of element X in the bulk soil is the rock, the proportion of this element X_{Ti} (in reference to Ti) originating from the rock at a given soil depth was computed as :

$$X_{\text{Ti}}(\%) = 100 \times (R_{\text{soil}}/R_{\text{granite}})$$

with R_{soil} = concentration ratio ($\mu\text{g g}^{-1}$) of X to Ti in the bulk soil (fine earth).

and R_{granite} = concentration ratio ($\mu\text{g g}^{-1}$) of X to Ti in the granite. This computation was applied to Sr and other mineral elements.

Of the two solid rock samples analysed, one contained very low amounts of Ca (1.0 mg g^{-1}), Sr ($5 \mu\text{g g}^{-1}$) and Ti ($369 \mu\text{g g}^{-1}$), much lower than the other (see Tab. I), but also lower than the weathered granite sample of profile TR6. Its $^{87}\text{Sr}/^{86}\text{Sr}$ ratio was 1.374. This sample, which was not weathered, contained low amounts of feldspar, and no muscovite. Therefore, it was considered not representative of the rock composition and it was not used in the computations.

3. RESULTS

3.1. Soil and stream water composition

Soils developed in small depressions in the savannah-rock unit were thin (5–20 cm) bleached sandy layers, rich in organic carbon (5 to 50%). The cation exchangeable capacity ($5\text{--}100 \text{ mmol}_{(+)}\text{kg}^{-1}$) and the sum of exchangeable base cations were strongly correlated with the total C content (Sarhou and Grimaldi, 1992). Base saturation was below 30% (Sarhou and Grimaldi, 1992). Below forest, soil textures varied from sandy-clay (up to 70% sand) in TR6 (low and transition forest) to clayey-sand (up to 50% clay) in TR9 and TR11, in the

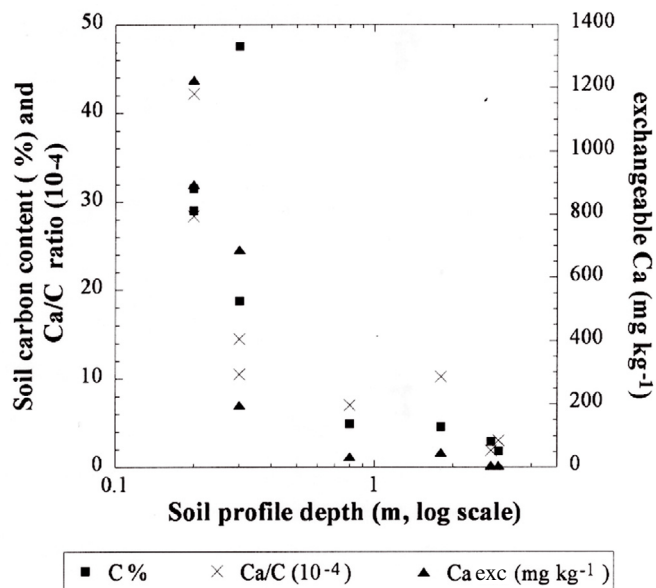


Figure 2. Soil total carbon content and exchangeable Ca concentration in the horizons 0–10 cm in relation to the total depth of soil profiles.

high forest. The silt fraction was always below 10% (Grimaldi and Riera, 2001). The profile IF contained about 20% gravels (< 1 cm) down to 150 cm depth but gravels were absent from the deeper sandy layer. The stone content at TR6 was about 50 % at 80 cm and between 30% and 15% in upper horizons. No stone was observed at TR11. The C content of the upper layer (0–5 cm) decreased from 5% in TR6 (transition forest) to 3% in the high forest. Soil pH increased from 4.2 in upper horizons to 5 in deep horizons. The cation exchangeable capacity was about 10 mmol₍₊₎kg⁻¹ in upper horizons. Base saturation decreased sharply at depth, except in the weathered granite from the deepest horizon in profile TR6 (Tab. I). Relative to the C content, the base saturation in the upper soil layer (0–10 cm) decreased along the slope to the high forest (Fig. 2).

From the granite to the topsoil of profile TR6, titanium concentration increased from 0.6 mg g⁻¹ to 3.5 mg g⁻¹. Titanium concentrations in the topsoil horizons of profiles IF (3 mg g⁻¹), TR6 and TR11 (3.5 mg g⁻¹) were in the same range. The total Ca concentration decreased drastically from the granite (2.8 mg g⁻¹) and the weathered granite (2 mg g⁻¹) to the topsoil horizons of all profiles (from 0.16 mg g⁻¹ in TR6 to 0.05 mg g⁻¹ in IF). Sodium concentration decreased even more, from 13.95 mg g⁻¹ to 0.26 mg g⁻¹ at TR6 and below 0.1 mg g⁻¹ in the other profiles. Contrastingly, Sr concentration decreased much less, from 24 µg g⁻¹ in the granite to 14, 10.5 and 9.2 µg g⁻¹ in the topsoil of TR6, IF and TR11 respectively (Tab. I). In TR6, K, Ba and Rb concentrations decreased from the bottom to the top in parallel to Sr. But the concentrations of K, Ba and Rb in surface horizons of the deeply weathered profiles IF and TR11 were almost ten times lower than in TR6. The proportions of Ca and K in soil profiles IF, TR6 and TR11, in reference to Ti, are illustrated in Figures 3a, 4a

and 5a, respectively. Less than 1% of the Ca and Na from the initial rock content remained in the upper soil horizons at all sites, while about 10% of the rock original Sr content was preserved.

From X-ray diffraction patterns we identified in the granite quartz, albite, and microcline as well as low amounts of muscovite. The sample used as reference contained low amounts of gibbsite and kaolinite, showing incipient weathering. Apatite was not identified but could be deduced from the total rock content in P. In profile TR6, quartz (4.25 Å and 3.34 Å), gibbsite (4.84 Å), minerals of the kaolinite/halloysite group (large peaks around 7.20 Å and 3.58 Å) and microcline (4.22 Å and 3.25 Å) were the main minerals. Muscovite (10 Å) was present in the weathered rock and in the fine earth of the deepest horizon. It disappeared in the 30–50 cm layer and was again present (but with a low intensity) in 0–10 and 10–20 cm layers. Albite was present in the deepest horizon (4.03 Å and 3.19 Å) but disappeared higher in the profile. In shallow sandy soils below *Pitcairnia* patches, minerals were quartz, microcline and albite, while the mica component was almost absent. In the sandy colluvium of profile IF, the main remaining mineral was quartz, with low amounts of gibbsite, and microcline especially in the topsoil, while albite was almost absent. In profile TR11, the main minerals were halloysite, gibbsite and quartz, with small amounts of microcline in the upper horizon (10–20 cm). Small amounts of albite and microcline were present as well in this deepest horizon sampled (220–280 cm), the texture of which was sandy. The upper horizon of profile GP, on metavolcanic rocks, was composed of aluminous goethite (peaks at 4.12 Å, 2.65 Å and 1.69 Å), gibbsite and minerals of the kaolinite-halloysite family, with quartz but no other primary minerals.

Along the river Kwak and upstream waters draining the slope of the inselberg, Ca (1–3.5 µmol L⁻¹), K (2.3–15.6 µmol L⁻¹), Sr (0.0007–0.0045 µmol L⁻¹) and Si (40–97 µmol L⁻¹) concentrations were very low and Sr was best correlated to K ($r = 0.93$). Groundwater from IF as well as gully water samples collected at the summit of the inselberg have concentrations generally close to rainwater (Négre and Lachassagne, 1997) but much higher concentrations in K and NO₃, as already observed by Sarthou and Grimaldi (1992).

3.2. Element concentrations in vegetation samples

Ca and Sr are poorly translocated in plants (Marschner, 1995), so that leaf and litter concentrations are approximately similar. The highest Ca and Sr concentrations (Tab. II) were measured in *Clusiaceae* leaves, and in leaf litter from the low forest on the upper part of the transect (greater than 5 mg g⁻¹ and 50 µg g⁻¹, respectively). Calcium and Sr concentrations of litter from intermediate and high rainforest trees on granite were less than 5 mg g⁻¹ and 50 µg g⁻¹, respectively. Lower Ca and Sr concentrations were found in *Pitcairnia* and *Catopsis* leaves, and especially in cyanobacterial crusts (0.5 mg g⁻¹ and 6.3 µg g⁻¹ respectively). The lowest Sr concentrations (4.4 µg g⁻¹) were measured in epiphyte leaves. Strontium and Ca concentrations were well correlated in plant

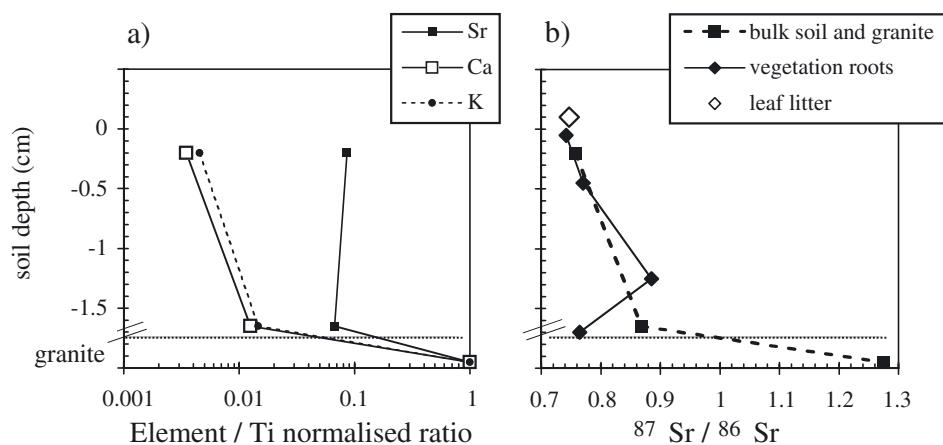


Figure 3. Soil profile IF. (a) Proportion of Ca, Sr, and K remaining after weathering and dissolution according to soil depth; (b) $^{87}\text{Sr}/^{86}\text{Sr}$ isotopic ratio of granite, bulk soil and roots according to depth.

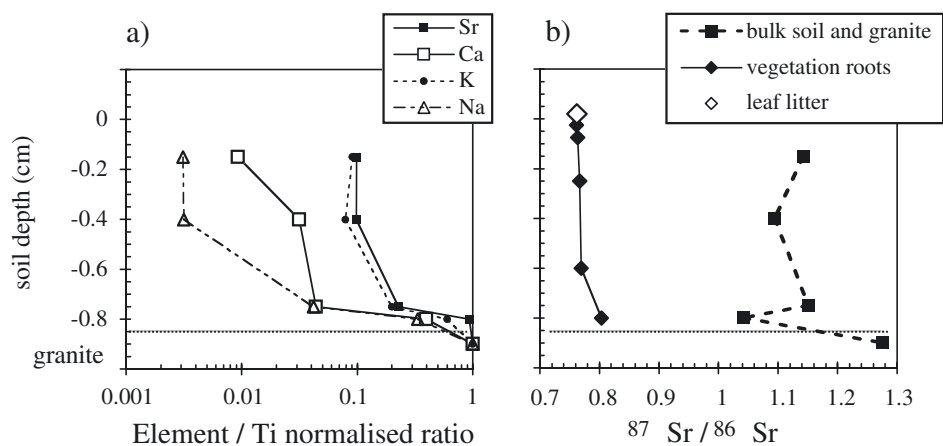


Figure 4. Soil profile TR6. (a) Proportion of Ca, Sr, K and Na remaining after weathering and dissolution according to soil depth; (b) $^{87}\text{Sr}/^{86}\text{Sr}$ isotopic ratio of granite, bulk soil and roots according to depth.

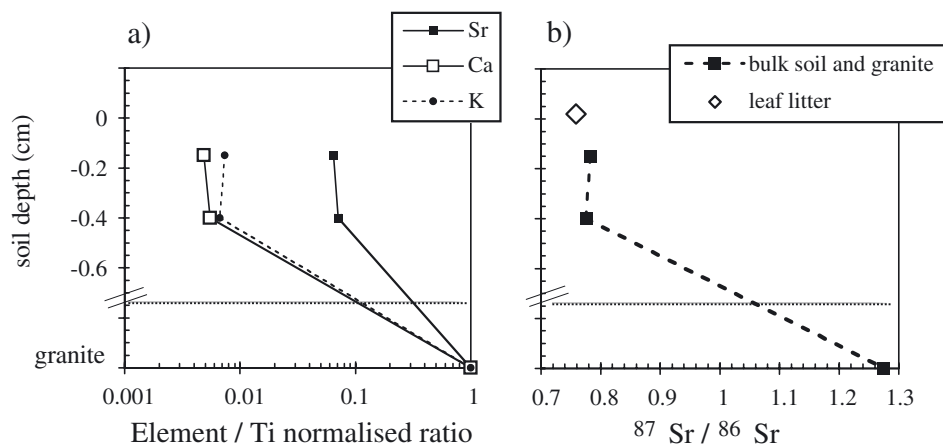


Figure 5. Soil profile TR11. (a) Proportion of Ca, Sr, and K remaining after weathering and dissolution according to soil depth; (b) $^{87}\text{Sr}/^{86}\text{Sr}$ isotopic ratio of granite, bulk soil and roots according to depth.

Table II. Strontium, Ca, K, P concentrations and $^{87}\text{Sr}/^{86}\text{Sr}$ isotopic ratio in leaves and leaf litters of the different plant species sampled on the Nouragues site.

Sample	Vegetation	Site	Sr $\mu\text{g g}^{-1}$	Ca mg g^{-1}	K mg g^{-1}	P mg g^{-1}	$^{87}\text{Sr}/^{86}\text{Sr}$
Crust	Cyanobacteria	i1	6.60	0.53	2.20	0.14	0.723
		i2	6.00	0.48	1.91	0.14	0.729
Leaves	<i>Pitcairnia geyskesii</i>	i1	11.78	1.95	11.52	0.45	0.723
		i2	13.99	1.91	12.96	0.36	0.724
		i3	11.21	2.07	13.85	0.42	
		t1	6.67	1.41	14.26	0.47	0.751
		t2	9.72	2.40	13.26	0.50	0.728
		t3	6.28	1.82	15.99	0.50	
	<i>Clusiaceae</i>	i1	44.79	5.28	9.98	0.34	0.728
		i2	51.71	7.26	9.34	0.32	0.730
		i3	60.52	8.59	11.84	0.30	
		t1	60.56	7.15	12.37	0.33	0.753
		t2	56.10	6.14	10.95	0.35	0.730
		t3	43.63	5.08	11.91	0.38	
	<i>Catopsis berteroniana</i>	i1	3.17	0.65	4.32	0.29	0.722
		i2	5.72	0.93	3.17	0.20	0.720
		i3	4.45	0.87	3.94	0.22	
	<i>Vriesea splendens</i>	IF	6.07	0.67	17.69	0.35	0.752
Leaf litter	Low forest	IF	38.28	2.84	1.59	0.07	0.747
		IF	49.47	3.12	2.53	0.08	0.756
		TR1	84.78	10.06	2.01	0.08	0.758
		TR2	55.06	7.12	3.41	0.06	
		TR3	68.55	6.75	2.46	0.09	0.764
		TR4	31.46	4.83	2.66	0.18	0.760
		TR5	29.98	4.52	3.09	0.19	
	Transition forest	TR6	37.60	4.79	3.64	0.20	0.762
		TR7	47.27	4.86	4.14	0.15	
		TR8	46.59	4.05	2.66	0.13	
		TR9	45.31	3.89	2.88	0.14	0.757
	High forest	TR10	52.69	5.13	2.84	0.20	0.757
		TR11	52.89	5.10	2.79	0.20	0.759
		TR12	49.76	4.81	3.86	0.21	0.760
		TR13	47.44	6.41	2.88	0.18	0.758
		GP	77.73	10.18	2.55	0.13	0.713

leaves, forest leaf litters and forest fine roots for the whole site ($r = 0.93$, Fig. 6). Potassium concentrations in *Pitcairnia* leaves were generally higher than in *Clusiaceae* leaves (9–12 mg g^{-1}), whatever the sampling site, and much higher than in other plant leaves and cyanobacteria crusts (below 4.3 mg g^{-1}). Phosphorus concentrations in *Pitcairnia* leaves (0.4–0.5 mg g^{-1}) were generally higher than in *Clusiaceae* leaves (0.3–0.35 mg g^{-1}), with slightly higher contents on the high terraces than on the summit of the inselberg. Leaves of the epiphyte *Catopsis* had lower P content (0.25 mg g^{-1}) than leaves of *Clusiaceae* on the stem of which they develop. Phosphorus and K contents in leaves from the savannah-rock unit and leaf litters from the forest cannot be directly compared because of internal translocation before leaf senescence. In leaf litters, P concentrations (0.1–0.2 mg g^{-1}) increase from the low forest (inselberg and transect) to the intermediate and then to the high forest.

3.3. $^{87}\text{Sr}/^{86}\text{Sr}$ variations in soils and plants

The Sr isotopic ratios of the two granite samples taken at the summit were 1.275 and 1.374 respectively. In profile TR6, the $^{87}\text{Sr}/^{86}\text{Sr}$ ratio of the bulk soil varies between 1.04 and 1.14 from the weathered granite to the topsoil (Fig. 4b and Tab. I). In the bleached sandy colluvium of profile IF, bulk soil $^{87}\text{Sr}/^{86}\text{Sr}$ decreases from 0.87 in the deepest horizon (150–180 cm) to 0.76 close to the soil surface (Fig. 3b). In the thick profile TR11 below the high forest, the Sr isotopic signature of the bulk soils is 0.78 between 0 and 60 cm depth (Fig. 5b). Considering all soil samples, bulk soil $^{87}\text{Sr}/^{86}\text{Sr}$ ratio increases with K (or Rb) concentrations.

At the summit of the inselberg, the Sr isotopic ratio of cyanobacteria crusts (i1, i2) attached to hard rock surfaces varied from 0.723 to 0.729. The Sr isotopic ratio of *Pitcairnia* (0.723 to 0.724) and *Clusiaceae* (0.728 to 0.730) leaves

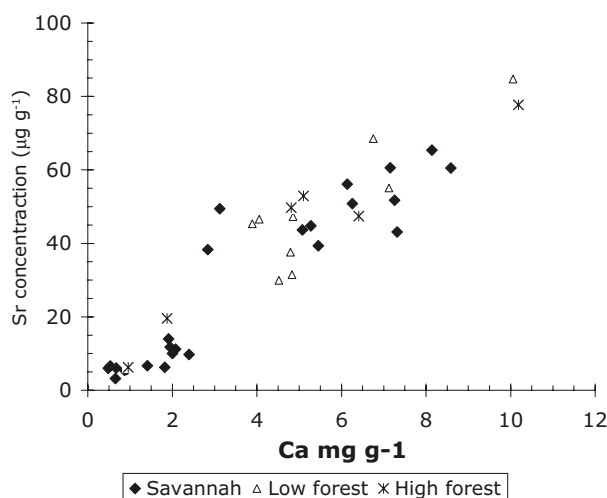


Figure 6. Correlation between Ca and Sr concentrations in the different plant organs sampled on the Nouragues site (savannah, low and high forest).

growing in the small depressions varied in the same range as that of the cyanobacteria (Tab. II). The Sr isotopic ratio of leaf samples of the epiphyte *Catopsis*, growing on upper branches of *Clusia* shrubs and collecting rain water was in the same range but slightly lower (0.720–0.722). The $^{87}\text{Sr}/^{86}\text{Sr}$ of *Pitcairnia* and *Clusiaceae* collected on the high terrace (t1, t2, t3) was higher and more variable (0.728 to 0.753) than on the summit. Below the low forest patch close to the summit (IF), the Sr isotopic ratio measured in leaf litters were 0.747 and 0.756. An intermediate $^{87}\text{Sr}/^{86}\text{Sr}$ ratio (0.752) was measured in leaves of *Vriesea splendens*, a *Bromeliaceae* growing on the soil surface in the same area. The Sr isotopic ratio of fine roots in the IF profile increased from 0.742 at 10 cm to 0.885 at 150 cm depth, but decreased to 0.764 at 160–180 cm (Fig. 3b).

At TR6, the litter $^{87}\text{Sr}/^{86}\text{Sr}$ ratio is 0.762. Fine root $^{87}\text{Sr}/^{86}\text{Sr}$ ratio increased from 0.761 in the upper layer to 0.769 at 70 cm, then more sharply to 0.803 in the weathered granite (Fig. 4b). On the inselberg slope, from the high terrace to the valley, the Sr isotopic ratio in leaf litter is between 0.757 and 0.764 (Tab. II), regardless of site, forest type (low or high) and soil thickness. We did not measure the Kwak river $^{87}\text{Sr}/^{86}\text{Sr}$ ratio but it may be close to that of the litter collected from trees growing along the bank (0.758).

4. DISCUSSION

Plant $^{87}\text{Sr}/^{86}\text{Sr}$ ratio varied along the slope within a rather narrow range of values, much closer to rain than to granite $^{87}\text{Sr}/^{86}\text{Sr}$ ratio. In order to explain this, we first discuss the fluxes and pools of mineral elements, then we suggest isotopic ratio values for the atmosphere and weathering end members in relation to rock and soil composition, and finally we propose a set of possible mechanisms.

4.1. Fluxes of Sr and Ca at the Nouragues site

At Cayenne in French Guiana, close to the seashore, Négrel et al. (1997) reported a volume weighted average Ca concentration of $15.3 \mu\text{mol L}^{-1}$ in precipitation during the dry season, which provided 11% of the annual precipitation, and $8.2 \mu\text{mol L}^{-1}$ for the whole year. The annual Ca deposition was 148 mol ha^{-1} . These Ca concentrations are probably overestimated for our site as Ca (and Sr) concentrations decrease inland as a function of distance to the ocean (Meybeck, 1986). In fact, in the Maroni catchment (200–300 km inland from the coast), Négrel and Lachassagne (2000) measured an average Sr concentration in rain of $0.017 \mu\text{mol L}^{-1}$ and Ca concentration of $7.5 \mu\text{mol L}^{-1}$ during the dry season. Multiplying these concentrations by the mean annual precipitation (3000 mm) at our site leads to a deposition flux of 225 mol ha^{-1} of Ca, and 510 mmol ha^{-1} of Sr. A probably better estimate, which takes into account the lower concentrations of rains during the rain season compared to the dry season could be obtained by multiplying the deposition fluxes, calculated above for the Maroni catchment, by the ratio of Ca concentration between the wet and the dry season at Cayenne reported by Négrel et al. (1997) (8.2/15.3). This leads to annual deposition rates of 120 mol ha^{-1} ($4.8 \text{ kg ha}^{-1} \text{ y}^{-1}$) of Ca, and 273 mmol ha^{-1} (24 g ha^{-1}) of Sr. In comparison, in the Osa peninsula on the Pacific coast of Costa Rica, Bern et al. (2005) measured an annual Sr deposition of $135 \text{ mmol ha}^{-1} \text{ y}^{-1}$. In this study, we measured Sr concentrations of 0.0042 and $0.0026 \mu\text{mol L}^{-1}$ during two rain events in the dry season at Nouragues which suggests that the fluxes calculated above may be overestimated.

The calculated exchangeable Sr pools in the upper 60 cm of profiles IF, TR6 and TR11 are about 590, 420 and 180 g ha^{-1} respectively, and exchangeable Ca stores are about 50, 30 and 20 kg ha^{-1} . The annual fine litterfall in the high forest is estimated to be $7 \text{ tons ha}^{-1} \text{ y}^{-1}$, based on values measured at Piste Sainte Elie (French Guiana), near the coast, by Puig and Delobelle (1988) and from a forest biomass inventory at Nouragues (Chave et al., 2001) combined with relationships between forest biomass and litterfall in tropical forests compiled by Clarke et al. (2001). Using this value and average litter concentrations measured in the transition and high forest, we computed an annual Sr and Ca cycling in litterfall of 320 g Sr ha^{-1} and 34 kg Ca ha^{-1} respectively. No values are available for the low forest, but litterfall is probably less. Hence, the exchangeable Sr pool is about 7–25 times the annual atmospheric deposition rate, exchangeable soil Ca is about 5–10 times the annual deposition rate, but annual Sr and Ca fluxes in litterfall (at least in the high forest) are in the same range as exchangeable pools in the upper 60 cm of the soils.

Strontium concentrations in drainage waters are lower than the rain concentrations measured by Négrel and Lachassagne (2000). Even if these streams only collect a fraction of the water draining the inselberg, the low Sr concentration suggests a continuous build-up of Sr stores in the vegetation, as soil exchangeable Sr pools are very low, and implies a low rate of Sr release by weathering.

4.2. Isotopic ratio of the atmospheric deposition and mineral weathering end-members

Measuring the mean Sr isotopic ratio of atmospheric deposition is complicated because of the inter-rainfall event variations of this ratio, the deposition of mineral dust with a different ratio (Grousset et al., 1992), and the risk of contamination by locally produced dust or animals. In French Guiana, Négrel and Lachassagne (2000) measured an average Sr isotopic ratio of 0.7118 in individual rain events during the dry period in the Maroni catchment, the highest ratio being 0.7149. In western Costa Rica (Bern et al., 2005) and in Hawaii (Chadwick et al., 1999), the Sr isotopic ratio of rain events was close to the seawater ratio (0.709). In Hawaii, deposition of dust from Asia (Nakai et al., 1993) with a Sr isotopic ratio of 0.718, contributed up of 3% of the total Sr load deposited from the atmosphere (Kennedy et al., 1998). Swap et al. (1992) showed that North African dust is the main source of dust to the Atlantic and the Central Amazon Basin. Grousset et al. (1992) measured a Sr isotopic ratio for North African dust of 0.716. Unfortunately, the relative contribution of dust to Sr deposition on the Amazon is unknown. At Nouragues, taking into account that the rainy season provides 80% of annual precipitation and are mostly of oceanic origin, we may hypothesize an annual average Sr isotopic ratio of 0.710. In order to provide an integrated value of the $^{87}\text{Sr}/^{86}\text{Sr}$ ratio in atmospheric deposition, we sampled the epiphyte *Catopsis* growing on upper branches of *clusia* leaves, because this plant has no functional roots and looks like a rain gauge, accumulating rain in its leaves. But this carnivorous plant use this water reservoir to attract insects which are used as a source of nutrients. We believe that the high $^{87}\text{Sr}/^{86}\text{Sr}$ ratio obtained (0.721), which is close to the ratio of plants in the savannah-rock unit, reflects mostly the contamination of the water reservoir by insects, as well as frogs and birds, which use it as a source of water. Finally, owing to the elevated age, stable position and extreme depletion of soils of the Grand Plateau on metavolcanic rocks, we had sampled recently fallen leaf litter with the idea that it might yield an upper estimate for the atmospheric deposition isotopic ratio. But the value obtained ($^{87}\text{Sr}/^{86}\text{Sr} = 0.713$) may still be influenced by soil sources of Sr although it was close to that (0.714) measured by Poszwa et al. (2002) in trees growing on sandy quartzitic patches near the Atlantic coast.

In the wet and warm climate of the tropics, most granite minerals such as feldspar and mica are dissolved and clay minerals such as gibbsite and kaolinite/halloysite precipitate (Tardy, 1993). Even quartz is dissolved, but its dissolution rate is about two orders of magnitude lower than that of K feldspar (White and Brantley, 1995). A few accessory minerals, containing Ti (ilmenite, rutile) or Zr (zircon), resist dissolution. In granite, a fraction of the Sr may be located in Ca-rich minerals such as plagioclase and apatite, with another fraction in K-rich minerals with a higher $^{87}\text{Sr}/^{86}\text{Sr}$ ratio like microcline and muscovite. Compared to feldspar, the Sr isotopic ratio of mica is very high, but its content in Sr is very low (Aubert et al., 2001; Clow et al., 1997; Probst et al., 2001), and especially that of muscovite. Muscovite is weathered in the deep horizon at TR6 and is absent from the other profiles. There-

fore, its contribution to the Sr isotopic ratio of soils and to Sr release by weathering is most likely weak.

The Sr isotopic ratio of the solid granite sample ($^{87}\text{Sr}/^{86}\text{Sr}$ ratio = 1.27) collected at the summit of the inselberg is higher than that of the weathered granite sample ($^{87}\text{Sr}/^{86}\text{Sr}$ ratio = 1.04) from the deeper horizon of profile TR6, whereas Na/Ti, K/Ti and Sr/Ti were respectively 66%, 40% and 7% lower in the weathered granite. This isotopic shift might be related either to preferential dissolution of a very radiogenic mineral such as muscovite or to local variations in the isotopic composition of the granite. The Ca/Ti, Na/Ti, K/Ti and Sr/Ti of the soil (10–20 cm layer) of profile TR6 are 98%, 99%, 85% and 90% respectively lower than those of the weathered granite (Fig. 4a). This large loss changes the bulk soil $^{87}\text{Sr}/^{86}\text{Sr}$ ratio from 1.04 to 1.14, which shows that the ratio of the Sr released by weathering is close to that of bulk soil Sr. Strontium is released from the dissolution of K and Na feldspars and apatite. Apatite dissolution in the upper layer may explain the decrease of Ca while Na remains stable and the Sr isotope ratio increases. We may assume that the isotopic signature of the Sr released from TR6 is in the range 1–1.3, because this range covers the variation of the isotopic composition of the parent rock and soil. Evidence for the release of radiogenic Sr by rock weathering in profile TR6 is provided by the high root $^{87}\text{Sr}/^{86}\text{Sr}$ ratio (0.803) in the weathered granite (Fig. 4b). The progressive decrease of root $^{87}\text{Sr}/^{86}\text{Sr}$ ratio to 0.761 towards the soil surface reflects the decreasing input of Sr delivered by weathering versus the increasing input of atmospheric Sr, as well as the intensity of Sr cycling.

In the bulk soil of the deeply weathered profiles IF and TR11, the Sr concentrations (between 10.5 and 9.2 $\mu\text{g g}^{-1}$) are slightly lower than the Sr concentration (14 $\mu\text{g g}^{-1}$) in the upper horizon of TR6, and Ti concentrations are similar in these profiles. However, the $^{87}\text{Sr}/^{86}\text{Sr}$ ratio of the top soils at IF and TR11 is much lower (0.758–0.783) than in TR6 (1.14), approaching that of bioavailable Sr (plant Sr: 0.752–0.759). This change is not related to Sr adsorption on clays, because the total amount of Sr in the mineral soil (between 45 kg ha^{-1} at IF and 62 kg ha^{-1} at TR6, in the 60 cm upper layer) is 2–3 orders of magnitude greater than the content of exchangeable Sr (between 590 g ha^{-1} at IF and 180 g ha^{-1} at TR11). This means that either large amounts of Sr have been incorporated into clays (as non exchangeable Sr), or that a fraction of non radiogenic Sr is stored in primary minerals extremely resistant to weathering. The first hypothesis was used by Derry and France-Lanord (1998) to explain variations in the $^{87}\text{Sr}/^{86}\text{Sr}$ ratio of vermiculite and smectite in sediments. But this explanation does not stand in profile IF, which only contains small amounts of gibbsite. On the other hand, quartz is by far the most abundant primary mineral at IF, and at TR11. Sr concentrations, Sr/Ti and $^{87}\text{Sr}/^{86}\text{Sr}$ ratio get close in these two profiles. Quartz might contain Sr in solid or fluid inclusions and is very resistant to weathering. Unfortunately such Sr rich inclusions were not detected by SEM. The contribution of quartz to the release of Sr is difficult to assess but the stability of the Sr/Ti ratio in profiles IF and TR11 suggest that the rate of Sr release is extremely low. The $^{87}\text{Sr}/^{86}\text{Sr}$ ratio of this phase should be below that of the bulk soil, as X ray diffraction

shows that some K feldspar remains in the upper horizons of IF and TR11. The persistence of a minor K-feldspar fraction in the upper horizons of profiles IF and especially TR11 might be related to the lateral redistribution of soil particles eroded from bare rock surfaces and shallow soils of the summit of the inselberg during erosion crisis.

The increase of root Sr isotopic ratio with depth, from 0.742 in the top soil to 0.770 at the depth 30–60 cm and 0.885 at 150 cm (Fig. 3b) suggests a large contribution of atmospheric Sr at the top and the release of Sr by weathering of radiogenic minerals, possibly K-feldspar (muscovite and albite are absent), in the bleached horizons of profile IF. Nevertheless, these horizons contained about 20% granitic gravel which dissolution may release radiogenic Sr and contribute to raise the root isotopic ratio. The low root isotopic ratio below the depth of 150 cm might be related to the absence of gravel in this sandy horizon, and to the accumulation of groundwater coming from surrounding bare rock surfaces.

4.3. Sources of Sr in the different vegetation types on the inselberg

The source contribution of Sr to the “rock-savannah” communities have been calculated using two end members : atmospheric deposition ($^{87}\text{Sr}/^{86}\text{Sr}$ ratio = 0.71), and bedrock $^{87}\text{Sr}/^{86}\text{Sr}$ ratio (between 1 and 1.3). At the summit of the inselberg (i1, i2), low isotopic ratios of cyanobacteria indicate that only 2–7% of Sr in cyanobacteria is supplied by rock weathering. Nevertheless, considering that the granite surface is compact on the one hand, and the very large rainfall amount on the other hand, this proportion seems significant. An explanation is that the inselberg is penetrated by a small layer of endolithic cyanobacteria (mainly *Choococidiopsis* sp.) which can dissolve the granite by increasing the pH up to 10 during photosynthesis (Büdel et al., 2004). This process may cause the dissolution of most granite minerals, including quartz. This rock contribution might be underestimated because cyanobacteria may preferentially dissolve apatite, which strontium isotopic ratio is generally low. But apatite is generally included in or framed by other silicate minerals. Furthermore, Sarthou and Grimaldi (1992) showed that waters running over the cyanobacteria crusts were enriched in K, which origin is related to K feldspar dissolution.

The isotopic ratio of *Pitcairnia* leaves is generally very close to that of cyanobacteria in the same area, which indicates that these plants take up the Sr supplied by waters draining the cyanobacterial crusts. The $^{87}\text{Sr}/^{86}\text{Sr}$ ratio of *Clusiaceae* leaves, always slightly elevated relative to that of cyanobacteria and *Pitcairnia* leaves, indicates a slightly higher Sr supply from mineral weathering in shallow organic soils. The similarity between the Sr isotopic ratios of low forest tree leaves and *Vriesea splendens*, usually an epiphyte, but growing here on the soil surface, below *Clusia* trees, indicates a direct supply from decomposing leaf litter. The $^{87}\text{Sr}/^{86}\text{Sr}$ ratio increases slightly from *Clusia* patches to the low forest.

At the lower end of the low forest (TR6), the computation suggests that weathering contributes 8–19% of tree Sr. The

high contribution of atmospheric deposition could be expected from the ratio between the annual flux of Sr deposited from the atmosphere and the total amount of soil exchangeable Sr (6%).

The stability of tree Sr isotopic ratio along the slope from the low to the high forest is very striking, given the considerable change in soil depth and weathering stage. We propose several possible explanations.

- Trees absorb Sr mainly in upper horizons, so that the influence of the underlying bedrock is limited. Because rain Sr is deposited on the soil surface, whereas soil weathering releases Sr from soil minerals, the isotopic ratio of bioavailable Sr increases with soil depth, which, in comparison to leaf Sr, may be used to measure the mean depth of Sr uptake (Dambrine et al., 1997, Wickman and Jacks, 1993). Comparing root and leaf $^{87}\text{Sr}/^{86}\text{Sr}$ ratio suggests that plant take up Sr from the upper horizons. The variation of root $^{87}\text{Sr}/^{86}\text{Sr}$ ratio in relation to depth was not evaluated in the high forest, but in a study of tropical forests on deeply weathered soils, Poszwa et al. (2002) observed very steep gradients in soils and demonstrated that almost all the Sr was taken up from the uppermost soil horizon.
- The low forest is supplied with Sr by gully waters drained laterally from the savannah rock unit. These waters provide a large influx of Sr with a low isotopic ratio (0.72–0.73) which decreases the isotopic ratio of bioavailable Sr by diluting the flux of Sr released by weathering. This lateral input of elements was illustrated by Dojani et al. (2007), who measured at the same site the change in soil $\delta^{15}\text{N}$ along the slope. Soil $\delta^{15}\text{N}$ was +6‰ in the high forest, decreased abruptly in the low forest and reached, –1.9‰ and –1.3‰ respectively in cyanobacteria crusts and pitcairnia soils. Intermediate values of $\delta^{15}\text{N}$ in the low forest were explained by the flow of nitrogen fixed from the atmosphere by cyanobacteria and drained laterally as nitrate and amino acids toward the low forest.
- Sr, as other nutrients, is taken up by roots and sprayed by litterfall over large areas. Tropical tree roots commonly prospect the surface of soils over tens of meters. Litter can also be dispersed from high trees over the same range. Along the slope, this lateral flux may homogenize chemical patterns and isotopic ratios over long distances. But this process cannot fully explain the stability of the Sr isotopic ratio along the slope.
- Down the slope, surface flow disappears, as soils get deeper. The stable value of the Sr isotopic ratio in the high forest should express the mixing of the atmospheric input with a small but consistent flux derived from weathering. Otherwise this ratio should be close to that of atmospheric Sr, as it is in the Grand Plateau. This small flux may be provided by the weathering of soil particles originating from the upper slope and redistributed downslope during erosion crisis. The observation of low amounts of K feldspar in the upper layer of TR11 provides an independent argument supporting this hypothesis. Given the low isotopic ratio of the Sr phase, extremely resistant to weathering and probably included in quartz, its contribution to Sr release is most likely very low.

5. CONCLUSION

In contrast with our original hypothesis, plant Sr isotope ratio did not decrease from the summit to the bottom of the inselberg as soils get deeper and more depleted in primary minerals. Instead, the Sr isotope ratio in plants remained nearly constant from the cyanobacterial crusts on bare granite to the high forest on deep, highly weathered soils. Possible explanations include: the preferential uptake of Sr by plants in the upper soil layers; the lateral redistribution of nutrients in surface water, by litterfall, and in the form of primary minerals dispersed along the slope during recurrent erosion crisis affecting the top of the inselberg. All these processes tend to supply nutrients, especially K, from bare rock surfaces to the forest downslope. However, given the very low Ca in the bedrock, and the differences between Ca and Sr dynamics during pedogenesis, Sr isotope variations are not well suited for investigating Ca release from the granite in this tropical setting. The extremely low soil Ca content in soil minerals implies that Ca is almost exclusively supplied to the high forest by atmospheric deposition.

Acknowledgements: We warmly thank the team present at the research site “Les nouragues” (UPS 656, CNRS) for their help, hospitality and friendship. The isotopic analysis was supervised by Pierre Brunet at the university of Toulouse. We thank Pascal Bonnaud for X-ray analysis, and C. Rose for SEM observations. Linda Pardo clarified our expressions, many thanks. This project was funded by GIP-ecofor.

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