

Secondary metabolites of Bagassa guianensis Aubl. wood: A study of the chemotaxonomy of the Moraceae family

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Mariana Royer, Gaëtan Herbette, Véronique Eparvier, Jacques Beauchêne, Bernard Thibaut, et al.. Secondary metabolites of Bagassa guianensis Aubl. wood: A study of the chemotaxonomy of the Moraceae family. Phytochemistry, 2010, 71, pp.1708-1713. 10.1016/j.phytochem.2010.06.020 . hal-00857200

HAL Id: hal-00857200 https://hal.science/hal-00857200v1

Submitted on 3 Sep 2013

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3	of the Moraceae family.
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24 Abstract

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In effort to explain wood durability of Moraceae plants family, a phytochemical study was 26 undertaken on Bagassa guianensis. The phytochemical investigation of the ethyl acetate 27 extract obtained from the heartwood led to the isolation of 18 secondary metabolites, 28 including 6 moracins [the new 6-O-methyl-moracin M (3), 6- O-methyl-moracin N (4) and 29 moracin Z (5); the known moracin M (1), moracin N (2) and moracin P (6)], 8 phenolic 30 derivatives [the new (-)-epialboctalol (12), arachidin 4 (10) and the known alboctalol (11), 31 trans-resveratrol (7), arachidin 2 (9), trans-oxyresveratrol (8) and artogomezianol (13)], the 3 32 known flavonoids steppogenin (14), katuranin (15), dihydromorin (16), the β -sitosterol (17) 33 and the resorcinol (18). Comparison with literature data indicates that stilbenoids are 34 35 presumably responsible for the natural durability of the wood. In addition, chemical composition points out that B. guianensis is closely related to Morus sp. in the phylogeny and 36 37 should be placed within the Moreae s. s. tribe in the Moraceae family.

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Keywords: *Bagassa guianensis*, Moraceae, secondary metabolites, stilbenes, moracins,
flavonoids, natural durability

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1. Introduction

Wood as a material is used extensively in construction and other applications where it can be 45 degraded by many different organisms, mainly fungi and insects. However, some trees have 46 specialized considerably long-lasting heartwoods. It has been demonstrated in the past that 47 wood natural durability can be ascribed to the presence of extractives (Smith et al., 1989; 48 49 Wang et al., 2005; Hsu et al., 2007), although structural components of the cell wall may also contribute to its resistance to biodegradation (Silva et al., 2007). Heartwood natural durability 50 can also result from synergetic or additive effects of compounds with various modes of action 51 52 (toxic, hydrophobic, free radical scavengers and so on) (Suttie and Orsler, 1996; Okitani et al., 1999; Schultz and Nicholas, 2000; Schultz et al. 2007; Binbuga et al., 2008). Future processes 53 to preserve wood constructions may involve returning to mankind's historical use of naturally 54 55 durable heartwood as well as discovering eco-friendly wood protection agents inspired from long-lasting woods (Schultz et al., 2007). 56

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Bagassa guianensis Aubl. (Moraceae) commercially known as tatajuba is a large rather infrequent unbuttressed canopy tree naturally occurring in French Guiana. *Bagassa guianensis* is a member of Moraceae family, which is divided in 5 unequal tribes when comparing the number of species in these tribes (Mabberley, 2002). *Bagassa guianensis* (the only member of its genus) was originally classified in the Artocarpeae tribe, but Weiblen genoma-based classifications have suggested recently that this species would better be included in Moreae tribe (Datweyler and Weiblen, 2004; Zerega et al, 2005).

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66 Species in the Moraceae family have important economic and medicinal value. They are 67 widely acknowledged as a rich source of bioactive secondary metabolites such as flavonoids, stilbenes, triterpenoids and xanthones (Lee et al., 2009; Ngadjui et al., 2005; Han et al., 2006; Jayasinghe et al., 2008). Also, some of them like *Maclura pomifera* and *B. guianensis* are capable of specializing very long-lasting woods (Scheffer and Morrell, 1998; Schultz et al., 1995), although in the latter case, the substances responsible for this high durability were unknown. We therefore embarked upon identifying secondary metabolites of tatajuba wood that may responsible for its natural durability. In addition, our secondary goal here was to confirm (or refute) botanical classification of the *Bagassa* genus by chemotaxonomy.

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2. Results and Discussion

The dried heartwood of *Bagassa guianensis* was extracted with ethyl acetate. This extract was
fractionated by silicagel column chromatography to give 9 fractions. Subsequent preparative
HPLC purifications of these fractions allowed us to isolate compounds 1-18 (figure 1).

80

Figure 1 Compounds 1-18 isolated from *Bagassa guianensis* (Moraceae). (a) New
compounds; (b) New names.

83

Compounds 1 to 6 shared several common spectral characteristics. The ¹H and ¹³C NMR 84 spectral data (Table 1) indicate the presence of two independent aromatic systems with a 3,5-85 dihydroxyphenyl and a substituted benzofuran. For example, 3 exhibited the 3,5-86 dihydroxyphenyl with characteristic ¹H spectrum composed of one doublet at δ 6.78 for H-87 2'/H-6' and a triplet at δ 6.25 for H-4'. These protons are coupled to each other with a ${}^{4}J$ 88 coupling of 2.1 Hz. In addition, ¹³C spectrum indicates the presence of two equivalent aryl 89 hydroxyl groups at δ 159.7. The 3,5-dihydroxyphenyl moiety was linked to C-2 by the 90 observation of a long range ${}^{1}\text{H}{}^{-13}\text{C}$ correlation between H-2'/H-6' and C-2 at δ 156.5. The 91 92 second aromatic system appeared characteristic of a 6-monosubstituted benzofuran with

93 signals of protons H-4, H-5 and H-7 being a broad doublet at δ 7.43 (J = 8.5 Hz), a doublet of doublet at δ 6.85 (J = 8.5 and 2.0 Hz) and a doublet at δ 7.09 (J = 2.0 Hz), respectively. On 94 the furan ring H-3 gives a doublet at δ 6.95 (J = 0.6 Hz) due to a long range ⁵J coupling with 95 H-7 (confirmed by the presence of crosspeak between H-3 and H-7 on COSY NMR 96 spectrum). When compared to moracin M (1), it became obvious from signal at δ 3.85 (3H, s) 97 and the presence of crosspeak at δ 56.2 in the ¹H-¹³C HSQC spectra that compound **3** was a 98 moracin M methyl ether. The ¹H-¹³C long-range HMBC spectra gave a crosspeak with C-6 at 99 δ 159.6 unambiguously placing the methoxy group on C-6. HREIMS of **3** allowed us to 100 ascertain molecular formula C₁₅H₁₂O₄ further confirming that we had isolated the new 6-O-101 methyl-moracin M (3). 102

103

104 **Table 1** ¹H and ¹³C NMR spectroscopic data for moracins 3-5 in CD₃OD

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Compound 4 was isolated as yellowish amorphous powder. The HREIMS indicated a 106 molecular formula $C_{20}H_{20}O_4$ deduced from the ion peack at m/z 325.1437 $[M + H]^+$ (calcd 107 325.1434). The ¹H and ¹³C NMR spectral data of **4** were closely related to those of moracin N 108 (2) (Lee et al., 2001) except for the replacement of hydroxyl group by a methoxy group as 109 described for the above compound **3**. Indeed, the ¹H NMR data of **4** (Table 1) demonstrated 110 the presence a methoxy group on C-6 in the benzofuran ring, with a signal at δ 3.88 (3H, s), a 111 crosspeak at δ 56.2 in the ¹H-¹³C HSQC experiment and a crosspeak with C-6 at δ 157.4 in 112 the ¹H-¹³C long-range HMBC spectra. This novel molecule was named 6-O-methyl-moracin 113 114 N.

Compound 5 was isolated as an amorphous brown powder. The molecular formula $C_{20}H_{22}O_5$ 116 was deduced from the HREIMS at m/z 343.1542 [M + H]⁺ (calcd 343.1540). The ¹H- and ¹³C-117 NMR spectral data of 5 were closely related to those of 6-O-methyl-moracin N (4) (Table 1). 118 The main difference was observed in the prenyl moiety at C-5. The double bond is absent in 5 119 and it was unambiguously established that side chain at C-5 is hydrated and is therefore a 3-120 hydroxy-3-methylbutyl group, with the upfield shifts of methylene group H-1'' from δ 3.34 to 121 δ 2.73 and the apparition of a methylene H-2" at δ 1.74 in place of the vinyl proton at δ 5.52; 122 in addition, the two methyl groups H-4" and H-5" became equivalent at δ 1.27 (Table 1). 123 The ¹H-¹³C long-range HMBC spectra exhibited a crosspeak between the methylene group H-124 1" and H-2" with C-5 at δ 128.6 proving the linkage C-1"/C-5 between the 3-hydroxy-3-125 methylbutyl moiety and the benzofuran ring. This molecule is a hydrate of 6-O-methyl-126 moracin N and was named moracin Z. 127

128

Spectral data along with HREIMS of 1, 2 and 6 allowed us to determine and ascertain by
comparison with literature data that we had also isolated moracin M (1) (Basnet et al. 1993,
Zhou et al., 1999), moracin N (2) (Lee et al. 2001) and moracin P (6) (Dat et al., 2009).

132

Stilbenoids *trans*-resveratrol (7) (Lee et al. 2001; Su et al., 2002), *trans*-oxyresveratrol (8) (Likhitwitayawuid and Sritularak, 2001; Lee et al., 2001; Su et al., 2002; Li et al., 2007), arachidin 2 (9) (Orsini et al., 2004) and artogomezianol (13) (Likhitwitayawuid and Sritularak, 2001) were identified by comparison of the respective spectral and chemical data with those described in the literature (Figure 1).

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139 Compound **10** was a colorless syrup with molecular formula $C_{19}H_{22}O_4$ as deduced from the 140 HREIMS at m/z 315.1592 [M + H]⁺ (calcd 395.1591). The ¹H spectral data of **10** were closely

related to those of arachidin 2 (9) (Table 2) and suggested a stilbenoid compound with a para-141 disubstituted aromatic ring A, a trans double bond between the aromatic rings, and a 142 1',3',4',5'-tetrasubstituted aromatic ring B. Ring A is symmetrical, with 2 doublets at δ 7.32 143 (J = 8.7 Hz, H-2/H-6) and $\delta 6.75 (J = 8.7 \text{ Hz}, \text{H-3/H-5})$. The trans configuration of the double 144 bond can be ascertained by the very large coupling constant between the two protons at δ 6.90 145 $(J = 16.5 \text{ Hz}, \text{H}-\alpha)$ and $\delta 6.74 (J = 16.5 \text{ Hz}, \text{H}-\beta)$, and the B ring is symmetrical as well and 146 was characterized by a singlet at δ 6.46 (H-2'/H-6'). In the same way as we identified a 147 hydrated side chain in the moracins series, the main difference here between 9 and 10 is in the 148 149 side chain in position 4', the double bond of which is also hydrated. This has been established by the observation of methylene group H-1" at δ 2.66 instead of δ 3.28 and the apparition of 150 a second methylene H-2" at δ 1.68. In addition, the two methyl groups H-4" and H-5" 151 became equivalent at δ 1.25. The chromatography collected quantities was too low to observe 152 heteronuclear ¹H-¹³C HSOC / HMBC correlations and direct ¹³C chemicals shifts by 153 ¹³C/DEPTQ sequence. However, the above-described data in comparison with those of 154 arachidin 2 are sufficient to ascertain identification of compound 10 as trans-4'-(3-hydroxy-3-155 156 methylbutyl)-oxyresveratrol. We named this new compound arachidin 4.

157

Table 2 ¹H and ¹³C NMR spectroscopic data for stilbenes 9 and 10 in CD_3OD

159

160 Compounds **11** and **12** both isolated as brownish syrups presented the ion peak at m/z161 489.1540 [M + H]⁺ in HREIMS indicating that they are isomers with molecular formulas 162 $C_{28}H_{24}O_8$ (calcd 489.1544). The ¹H-NMR allowed us to identify a 3,5-dihydroxyphenyl group 163 and two distinct 2,4-dihydroxyphenyl groups in both compounds. By comparison of the 164 respective spectral and chemical data with those described in the literature, compound **11** was

identified as alboctalol (Bates et al., 1997). Compound **12** has an $\left[\alpha\right]_{D}^{20}$ value of -7.4° (c 165 0.004, CH₃OH). It was clear that **12** was a diastereoisomer of **11** with equivalent H-18/H-22 166 protons at δ 6.01 (Table 3). In **11**, H-18/H-22 pair gives a doublet at a strong upfield shift of δ 167 5.77 typical of the π -stacking effect of the neighboring 2,4-dihydroxyphenyl groups. In 168 addition, on this aliphatic ring, the main differences with 11 are on methylene H-5 and 169 methines H-6, H-7 and H-8. H-5_{ax} at δ 3.19 exhibited a broad triplet with large couplings (J =170 13.7 Hz) with the gem H-5_{eq} and the vicinal H-6 suggesting that the 6-aryl group should be 171 equatorial and proton H-6 axial. This observation was corroborated by the multiplicity of H-172 5_{eq} signal at δ 2.72. This signal is a doublet of doublet with a large coupling constant J = 15.6173 Hz with H-5_{ax} and a small coupling constant J = 3.0 Hz with H-6_{ax}. Signal of H-6_{ax} at δ 3.51 174 is a broad triplet of doublet with two larges coupling constants J = 11.6 Hz with H-5_{ax} and H-175 7 and a small coupling constant J = 2.1 Hz with H-5_{eq}. This pattern indicates that the 7-aryl 176 group is equatorial and H-7 axial. H-7_{ax} at δ 3.41 exhibited one doublet of doublet with one 177 large coupling constant (J = 11.3 Hz) with H-6_{ax} and a second rather large coupling constant 178 (J = 8.2 Hz) with H-8 indicating that the 8-aryl group might be equatorial and proton H-8 179 axial. These assumptions were confirmed by NOESY experiment with cross peaks observed 180 between H-5_{eq} and H-6_{ax}, H-6_{ax} and H-8_{ax}, H-6_{ax} and H-18, H-8_{ax} and H-22 and between H-5_{ax} 181 and H-16, H-7ax and H-16, H-7ax and H-28 (Figure 2). All data permitted to confirm that we 182 had isolated a new epimer of alboctalol (11) therefore named (-)-epialboctalol (12). 183

184

Table 3 ¹H and ¹³C NMR spectroscopic data for distilbenes **11** and **12** in CD_3OD

186

187 Figure 2 Pertinent NOE interactions observed for (-)-epialboctalol (12) from NOESY
188 experiment

In addition to these moracins and stilbenoids, we isolated flavanones steppogenin (14) (Lee et al., 2001), katuranin (15) (Lee et al., 2001) and dihydromorin (16) (Su et al., 2002), together with β -sitosterol (17) (Basnet et al., 2003, Aldrich Library of 13C and 1H FT NMR spectra, 193 1992) and resorcinol (18) (Aldrich Library of 13C and 1H FT NMR spectra, 1992). These known compounds were identified by comparison of the respective spectral and chemical data with those described in the literature.

196

197 Essentially three classes of compounds were isolated in this study: moracins, stilbenes and 198 flavanones. Only resorcinol **18** and β -sitosterol **17** do not belong to these classes. These two 199 compounds are widely distributed in nature and cannot be viewed as chemotaxonomic 200 markers.

201

Moracin N, M and P have been isolated before from *Morus alba*. In general, it was found from the literature that *Morus* genus is purveyor of moracins (Tagasuki et al., 1979; Hirakura et al., 1986; Basnet et al., 1993; Nguyen et al., 2009). The only one exception is the isolation of moracin M from *Artocarpus dadah* (Su et al., 2002).

206

Among stilbenes, *trans*-oxyresveratrol was isolated from various plants including *Morus* sp. and *Artocarpus* sp. (Hirakura et al, 1986; Su et al, 2002; Shimizu et al., 1998; Song et al, 2009). *Trans*-resveratrol was isolated from many sources including the Moraceae *Cudrania javanensis* classified today as *Maclura cochinchinensis* (Murti et al., 1972, Chapman & Hall, 2006). The distylbene artogomezianol **13** is a constituent of *Artocarpus gomezianus* roots and albolactol **11** was isolated from heartwood of *Morus alba* (Likhitwitayawuid and Sritularak 2001, Ferlinahayati et al., 2008). Regarding flavonoids, it has been described that many Moraceae can produce steppogenin
(El-Sohly et al, 1999; Su et al, 2002; Sheu et al., 2005). Katuranin was also isolated from
various biological sources in *Morus* and *Maclura* genera (El-sohly et al., 1999, Lee at al.,
2009) and dihydromorin was isolated from *Morus*, *Artocarpus*, and *Maclura* genera (Shimizy
et al., 1998, El-Sohly et al, 1999, Su et al., 2002).

220

It has been hypothesized before that stilbenes are the major types of compounds isolated from 221 Moraceae and may be useful chemotaxonomic markers (Rowe and Conner, 1979). Also, 222 Schultz has shown that stilbenoids play an important role in the high natural durability of 223 Maclura pomifera wood (Schultz et al., 1990). Stilbenes are known as fungicide, termicides 224 and bactericide (Hart and Shrimpton, 1979; Likhitwitayawui and Sritularak, 2001; Javasinghe 225 226 et al., 2004), and may also exhibit antioxidant properties (Dani et al., 2008; Iacopini et al., 2008; Luo et al., 2005). If it is reasonable to believe that stilbenes are responsible for Bagassa 227 228 guianensis heartwood natural durability based on literature precedents, stilbenes can be 229 considered as a secondary chemotaxonomic marker here indicating that Bagassa is related to Morus, Artocarpus, and Maclura genera. In Weiblen classification, Artocarpus belongs to the 230 231 Artocarpeae tribe and Maclura belongs to the Moreae sensu largo tribe, and both Moreae and Artocarpeae tribes are rather closely related genetically. 232

The peculiarity of *B. guianensis* in comparison with other Moraceae is the very high proportion of moracins. In this matter, it can be hypothesized that *Bagassa* genus is closely related to *Morus* and that moracins are specific to these two genera. These findings are in agreement with Weiblen genetic-based classification where both *Bagassa* and *Morus* belong to the Moreae sensu stricto tribe. It should be mentioned that the *Sorocea* genus, which also belongs to the Moreae s. s. tribe, has been investigated before in the literature and apparently

does not contain moracins (see for example Ferrari et al., 2003; Ross et al., 2008). This
observation speaks in favor of a very close relationship between *Bagassa* and *Morus*.

241

242 3. Concluding remarks

Studies of defensive wood chemicals in *Bagassa guianensis* allowed us to identify large amount of diversely functionalized stilbenes presumably responsible for wood natural durability. In addition, it was found based on the presence of moracins that *Bagassa* is very closely related to Morus genus, therefore corroborating Weiblen phylogenetic classification where *B. guianensis* belongs to the Moreae s. s. tribe rather than to the Artocarpeae tribe.

248

249 **4. Experimental**

250 *4.1 General experimental procedure*

The ¹H and ¹³C-NMR spectra were recorded on a Bruker Avance DRX500 spectrometer (¹H-251 500.13 MHz) equipped with a 5 mm triple resonance inverse Cryoprobe TXI (¹H-¹³C-¹⁵N), 252 with z gradient. Spectra were recorded with 1.7 mm NMR capillary tube in 40 µL of 99.99% 253 CD₃OD solvent (δ_{1H} 3.31 ppm - δ_{13C} 49.00 ppm) at 300 K. The 1H (500 MHz) and ^{13}C NMR 254 (125 MHz) data are reported in ppm downfield from tetramethylsilane. Coupling constants are 255 in Hz and s stands for singlet, d for doublet, t for triplet, q for quartet, m for multiplet and br 256 257 for broad. Hydrogen connectivity (C, CH, CH₂, CH₃) information was obtained from edited HSQC and/or DEPTQ-135 experiments. Proton and carbon peak assignments were based on 258 2D NMR analyses (COSY, NOESY, HSQC and HMBC). HREI-MS were performed using a 259 QStar Elite mass spectrometer (Applied Biosystems SCIEX, Concord, ON, Canada) equipped 260 with an ESI source operated in the positive ion mode. The capillary voltage was set at 5,500 261 V, the cone voltage at 20 V and air was used as the nebulizing gas (20 psi). In this hybrid 262 instrument, ions were measured using an orthogonal acceleration time-of-flight (oa-TOF) 263

mass analyzer. Analyst software version 2.1 was used for instrument control, data acquisition 264 and data processing. The accurate mass measurements were performed in triplicate with two 265 internal calibrations. Direct sample introduction was performed at a 5 µL/min flow rate using 266 a syringe pump. The UV spectra were recorded on a Perkin-Elmer Lambda 5 267 spectrophotometer. Optical rotations were measured with a Perkin-Elmer 241 polarimeter 268 equipped with a sodium lamp (589 nm) and a 1 dm cell. The HPLC separations were 269 performed on a Supelco Discovery[®] HS PEG column ($250 \times 21.1 \text{ mm}, 5 \mu \text{m}$) using a Waters 270 system equipped with a W600 pump and a W2996 photodiode array absorbance detector. The 271 samples were injected manually through a Rheodyne injector and the flow rate was 15 272 mL.min⁻¹. Silica gel 60 (35-70 µm) and analytical TLC plates (Si gel 60 F 254) were 273 purchased from SDS (France). All other chemicals and solvents were analytical grade and 274 purchased from SDS (France). 275

276 *4.2 Plant Material*

Bagassa guianensis was collected in Régina, French Guiana. A voucher specimen is kept at
the herbarium of Cayenne (CAY-RA13), French Guiana.

279 *4.3 Extraction and isolation*

The dried powdered heartwood of Bagassa guianensis (140 g) was extracted with ethyl 280 acetate $(3 \times 500 \text{ mL})$ at room temperature to give a crude extract which was fractionated first 281 on a silica gel column chromatography with polarity gradient of hexane/ethyl acetate 282 mixtures: 80/20; 50/50; 20/80; 0/100. 9 fractions numbered F1 to F9 were obtained. Fractions 283 F1 to F5 were purified on HPLC with a linear gradient of hexane/isopropanol, by the 284 following method: 70:30 changing over 2 min to 60:40, then to 40:60 at 10 min and pure 285 isopropanol at 15 min and remaining as is for 5 min. The fractions F6 and F9 were analyzed 286 and purified with an isocratic method: 30:70 hexane/isopropanol. These methods allowed us 287 to isolate moracin M 1 (6.2 mg; w/w 0.019%), moracin N 2 (6.7 mg; w/w 0.020%), 6-O-288

methyl-moracin M 3 (3.3 mg; w/w 0.010%), 6-O-methyl-moracin-N 4 (9.1 mg; w/w 0.027%), 289 moracin Z 5 (5.2 mg; w/w 0.016%), moracin P 6 (1.2 mg; w/w 0.003), trans-resveratrol 7 290 (12.6 mg; w/w 0.038%), trans-oxyresveratrol 8 (112.3 mg; w/w 0.343%), arachidin 2 9 (5.1 291 mg; w/w 0.015%), arachidin 4 10 (0.4 mg; w/w 0.001%), alboctalol 11 (0.5 mg; w/w 292 0.001%), (-)-epialboctalol 12 (5.4 mg; w/w 0.016%), artogomezianol 13 (12.7 mg; w/w 293 0.038%), steppogenin 14 (11.5 mg; w/w 0.035%), katuranin 15 (1.5 mg; w/w 0.004%), 294 dihydromorin 16 (20.4 mg; w/w 0.062%), the β -sitosterol 17 (8.4 mg; w/w 0.025%) and the 295 resorcinol 18 (1.8 mg; w/w 0.005%). Compounds 1-6, 9-10 and 17-18 were obtained from 296 the purification of the fractions F1-F5 while compounds 7-8 and 11-16 were isolated from the 297 fractions F6-F9. 298

- 299 *4.3.1* 6-*O*-Methyl-moracin M (**3**)
- 300 Yellowish amorphous powder; HR-EIMS $[M + H]^+ m/z$ 257.0805 $[M + H]^+$ (calcd 257.0808);
- 1 H and 13 C NMR (500 MHz; CD₃OD) see table 1.
- 302 *4.3.2* 6-*O*-Methyl-moracin N (4)
- 303 Yellowish amorphous powder; HR-EIMS $[M + H]^+ m/z$ 325.1437 $[M + H]^+$ (calcd 325.1434);
- 1 H and 13 C NMR (500 MHz; CD₃OD) see table 1.
- 305 *4.3.3* Moracin Z (**5**)
- 306 Yellowish amorphous powder; HR-EIMS $[M + H]^+ m/z$ 343.1542 $[M + H]^+$ (calcd 343.1540);
- 1 H and 13 C NMR (500 MHz; CD₃OD) see table 1.
- 308 *4.3.4* Arachidin 4 (**10**)
- 309 Colorless syrup; HR-EIMS $[M + H]^+ m/z$ 315.1592 $[M + H]^+$ (calcd 315.1591); ¹H and ¹³C
- 310 NMR (500 MHz; CD_3OD) see table 2.
- 311 *4.3.5* (–)-Epialboctalol (**12**)
- 312 Brownish syrup; $[\alpha]_D^{20}$ –7.4° (c 0.004, CH₃OH); HR-EIMS $[M + H]^+ m/z$ 489.1540 $[M + H]^+$
- 313 (calcd 489.1544); ¹H and ¹³C NMR (500 MHz; CD_3OD) see table 3.

314	The 3 known moracins M (1), N (2) and P (6) and the other known compounds 7-9, 11, and					
315	13-18 were identified by comparison of their physical and spectral data with those reported in					
316	the literature.					
317						
318	Role of the funding source					
319	The authors gratefully acknowledge the Programme Amazonie du CNRS for financial support,					
320	as well as the CNRS and the Région Guyane, France, for the research fellowship attributed to					
321	MR.					
322						
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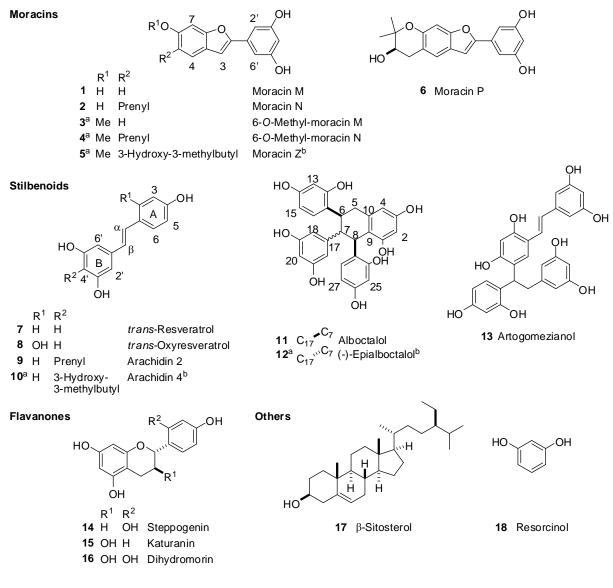
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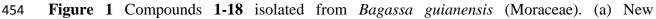
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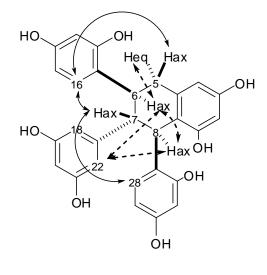
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Figures and legends





⁴⁵⁵ compounds; (b) New names.



459 Figure 2 Pertinent NOE interactions observed for (-)-epialboctalol (12) from NOESY
460 experiment

Atom	3		4		5	
Atom	δ_{C}	$\delta_{\rm H}$ (<i>J</i> in Hz)	$\delta_{\rm C}$	$\delta_{\rm H} (J \text{ in Hz})$	δ _C	$\delta_{\rm H} (J \text{ in Hz})$
2	156.5	-	156.2	-	156.2	-
3	96.5	6.95, d (0.6)	102.1	6.90, s	102.0	6.91, d (0.6)
4	121.9	7.43, d (8.5)	121.3	7.25, s	121.6	7.30,s
5	112.9	6.85, dd (8.5, 2.1)	127.5	-	128.6	-
6	159.6	-	157.4	-	157.5	-
7	102.0	7.09, brd (2.0)	94.7	7.09, s	94.7	7.09, s
8	157.0	-	155.7	-	155.8	-
9	123.7	-	123.0	-	123.1	-
1'	133.6	-	133.9	-	133.6	-
2'/6'	104.0	6.78, d (2.1)	103.9	6.77, d (2.1)	103.5	6.77, d (2.1)
3'/5'	159.7	-	159.9	-	160.0	-
4'	103.5	6.25, t (2.1)	103.5	6.24, t (2.1)	103.4	6.25, t (2.1)
1"	-	-	29.7	3.34, brd (7.3)	26.7	2.73, m
2"	-	-	124.3	5.32, tm (7.3)	45.5	1.74, m
3"	-	-	132.7	-	71.5	-
4"	-	-	17.8	1.73 brs	28.9	1.27, s
5''	-	-	26.0	1.74 brs	28.9	1.27, s
MeO	56.0	3.85, s	56.2	3.88 s	56.0	3.89, s

Table 1 ¹H and ¹³C NMR spectroscopic data for moracins 3-5 in CD₃OD

Tables

A 4 9 199	9		10
Atom	$\delta_{\rm C}$	$\delta_{\rm H} (J \text{ in Hz})$	$\delta_{\rm H} (J \text{ in Hz})$
1	130.6	-	-
2	128.6	7.31, d (8.6)	7.32, d (8.7)
3	116.5	6.75, d (8.6)	6.75, d (8.7)
4	158.1	-	-
5	116.5	6.75, d (8.6)	6.75, d (8.7)
6	128.6	7.31, d (8.6)	7.32, d (8.7)
α	128.3	6.88, d (16.3)	6.90, d (16.5)
β	127.2	6.74, d (16.3)	6.74, d (16.5)
1'	137.6	-	-
2'	105.7	6.46, s	6.46, s
3'	157.2	-	-
4'	116.0	-	-
5'	157.2	-	-
6'	105.7	6.46, s	6.46, s
1"	23.3	3.28, d (7.1)	2.66, m
2"	124.6	5.23, tm (7.1)	1.68, m
3''	131.4	-	-
4''	26.0	1.62, brs	1.25, s
5''	18.0	1.75, brs	1.25, s

Table 2 1 H and 13 C NMR spectroscopic data for stilbenes 9 and 10 in CD₃OD

470	
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12 11 Atom $\delta_{\rm H}$ (*J* in Hz) δ_{C} $\delta_{\rm H} (J \text{ in Hz})$ 1 156.7 -_ 2 101.8 6.10, d (2.2) 6.32, d (2.2) 3 156.2 4 107.3 6.19, d (2.2) 6.32, d (2.2) 40.1 3.19, brt (13.7) 2.98, dd (16, 14) 5_{ax} 2.72, dd (15.6, 3.0) 2.53, dd (16.3, 4.3) 5_{eq} 3.51, brtd (11.6, 2.1) 3.75, dt (14, 3.7) 6_{ax} 40.3 7_{ax} 56.2 3.41, dd (11.3, 8.2) 3.28, d (3.3) 7_{eq} -44.1 4.67, brs 8_{ax} 4.42, d (8.2) 9 119.6 -10 142.2 -123.7 11 _ 156.4 12 -? 13 103.3 6.16, d (2.2) 6.12, dd (8.4, 2.3) 14 156.4 6.13, dd (8.4, 2.3) 107.3 15 129.7 16 6.82, d (8.2) 6.44, d (8.2) 17 149.2 18 108.4 6.01, d (1.9) 5.77, d (1.9) 19 157.9 20 100.8 5.93, t (2.2) 6.02, t (2.2) 21 157.9 _ 22 108.4 6.01, d (1.9) 5.77, d (1.9) 23 125.2 24 156.1 6.76, d (8.2) 6.25, d (8.2) 25 103.2 6.23, dd (8.2, 2.5) 6.04, dd (8.2, 2.5) 26 156.7 -27 6.25, d (2.2) 108.1 6.19, d (2.2) 28 131.2

471 **Table 3** ¹H and ¹³C NMR spectroscopic data for distilbenes **11** and **12** in CD₃OD