



HAL
open science

Development of nickel phosphorus coatings containing micro particles of talc phyllosilicates

Joël Alexis, Clélia Gaussens, Bernard Etcheverry, Jean-Pierre Bonino

► To cite this version:

Joël Alexis, Clélia Gaussens, Bernard Etcheverry, Jean-Pierre Bonino. Development of nickel phosphorus coatings containing micro particles of talc phyllosilicates. *Materials Chemistry and Physics*, 2013, pp. 723-733. 10.1016/j.matchemphys.2012.09.049 . hal-00828234

HAL Id: hal-00828234

<https://hal.science/hal-00828234v1>

Submitted on 30 May 2013

HAL is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers.

L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.



Open Archive Toulouse Archive Ouverte (OATAO)

OATAO is an open access repository that collects the work of Toulouse researchers and makes it freely available over the web where possible.

This is an author-deposited version published in: <http://oatao.univ-toulouse.fr/>
Eprints ID: 8491

To link to this article: DOI:10.1016/j.matchemphys.2012.09.049
<http://dx.doi.org/10.1016/j.matchemphys.2012.09.049>

To cite this version:

Alexis, Joël and Gaussens, Clélia and Etcheverry, Bernard and Bonino, Jean-Pierre *Development of nickel phosphorus coatings containing micro particles of talc phyllosilicates*. (2013) *Materials Chemistry and Physics* (n° 137). pp. 723-733. ISSN 0254-0584

Any correspondence concerning this service should be sent to the repository administrator:
staff-oatao@inp-toulouse.fr

Development of nickel–phosphorus coatings containing micro particles of talc phyllosilicates

Joël Alexis ^{a,*}, Clélia Gaussens ^a, Bernard Etcheverry ^b, Jean-Pierre Bonino ^c

^a Université de Toulouse, INPT-ENIT, Laboratoire Génie de Production, Ecole Nationale d'Ingénieurs de Tarbes, 47 avenue d'Azereix, BP 1629, 65016 Tarbes cedex, France

^b ALSTOM TRANSPORT, BP4, Rue du Docteur Guinier, 65601 SEMEAC cedex, France

^c CIRIMAT-CNRS (UMR n°5085), Université Paul Sabatier, 118 route de Narbonne, 31062 Toulouse cedex 4, France

H I G H L I G H T S

- ▶ Nickel phosphorus composite coatings NiP-talc particles are developed.
- ▶ The content of particles increase with talc content of the bath and are distributed randomly on the coatings.
- ▶ A significant decrease in rigidity is observed by adding small amounts of particles of talc.
- ▶ The toughness increases with the insertion of talc when the deposit is heat treated at 420 °C
- ▶ Rough or heat treated at 600 °C deposits have a ductile behavior unlike the treated at 420 °C deposits.

A B S T R A C T

The present work aims to characterized nickel phosphorus coatings co-deposited with talc particles on steel. The NiP-talc composite deposits were developed to serve as hard coatings with a lubricating effect at 600 °C. This process, which is free of hexavalent chromium, could provide a reliable substitute for the electrodeposition of hard chromium coating used in industrial applications. Local responses to static and dynamic mechanical loading have been obtained by nano- and microhardness, microtensile and nano-scratch testing. The hardness and stiffness values slightly decrease when the amount of talc increases for untreated coatings. In contrast, a 420 °C heat treatment leads to high hardness and Young's modulus values due to crystallization. Moreover, a 600 °C heat treatment lowers these values through overageing. A 420 °C heat treatment greatly improves the adherence and the cohesion of the coatings containing talc.

Keywords:
Mechanical properties
Coatings
Adhesion
Composite materials
Hardness

1. Introduction

Nickel electroless plating is an autocatalytic deposition process in which nickel ions in aqueous solution are reduced at the catalytic surface of the substrate by a reducing agent that has also been added to the solution. The reducing agents used are phosphorus salts, such as NaPH₂O₂ [1], and the coatings obtained on the substrate are metallic solid solutions of nickel and phosphorus. By comparison with electroplating, electroless deposition offers the advantage of obtaining a quite uniform thickness, regardless of the roughness of the substrates. After heat treatments of the NiP solid solutions

involving the precipitation of nickel phosphides in a nickel matrix [2], an increase in hardness and wear resistance is induced; this behavior is one of the major reasons for the widespread use of electroless NiP coatings. Composite coatings containing hard or lubricating particles have also received attention for tribological applications [3]. In this case, the process consists of incorporating particles into the NiP matrix from an electrolyte containing particles in suspension. The particles adsorbed on the substrate during the growth of the matrix are definitely incorporated into the coating.

The insertion of particles in the NiP matrix is intended to increase the hardness and wear resistance, to reduce friction or to increase the resistance to corrosion. Their nature and shape differ depending on the goals. The reinforcement of the second phase can be hard oxides (Al₂O₃ and TiO₂) [4] or carbide particles (SiC [3], WC [5], B₄C [6], diamond [7] and nitride [8]). Solid lubricants (PE [9], PTFE [10], graphite [11] and MoS₂ [12]) can reduce the friction coefficient. Nanoparticles (SiO₂, CeO₂ [13], Zn₃(PO₄)₂, ZnSnO₃ and

* Corresponding author. Tel.: +33 0 562442723; fax: +33 0 562442708.

E-mail addresses: joel.alexis@enit.fr (J. Alexis), clélia.gaussens@enit.fr (C. Gaussens), bernard.etcheverry@transport.alstom.com (B. Etcheverry), boninojp@chimie.ups-tlse.fr (J.-P. Bonino).

Table 1
Composition and deposition parameters of the electroless NiP (Europlate Ni216) commercial bath used.

	NiP	NiP-talc
Nickel metal (g L ⁻¹)	6 (±1)	6 (±1)
Nickel hypophosphite (g L ⁻¹)	23 (±3)	23 (±3)
Sodium orthophosphate (g L ⁻¹)	0–180	0–180
Talc (g L ⁻¹)	–	20–120
Temperature (°C)	88 (86–92)	88 (86–92)
pH	5.0 (4.8–5.4)	5.0 (4.8–5.4)
Speed deposition (μm hL ⁻¹)	18–22	18–22
Surface treated (dm ² L ⁻¹)	1.3	1.3

ZnSiO₃ [14]) have been deposited to improve the corrosion resistance. Some electroless nickel co-deposits, such as Ni-PTFE, have been developed by the industry because of their significant resistance to corrosion and wear, but most co-deposits are still at the experimental stage [8,15–18]. The aim of this study was to obtain amorphous electroless nickel–phosphorus (NiP) coatings containing talc micro-particles (formula: Mg₃Si₄O₁₀(OH)₂) and to investigate both the coating structure and its mechanical properties. The effect of the talc particle content on hardness and adherence was particularly analyzed, before and after various 1-h heat treatments. The influence of the phosphorus rate is cited but is not the subject of this study because it is well known [19–21]. The deposits are developed for tribological high-temperature applications to replace hard chromium. Talc was chosen because it is resistant until 900 °C. The goal is to develop a hard (1000 HV) and lubricating coating [22]. Indeed, the first tribological tests have shown the beneficial role of talc on the reduction of the friction coefficient for heat-treated deposits [23].

2. Experimental procedure

2.1. Coatings and sample preparation

The composite electroless NiP-talc coatings were deposited on low-alloy steel substrates (36NiCrMo16). The substrates were heat-treated to obtain a hardness of 450 HV₁₀. The samples were polished and pretreated in an alkaline bath (pH = 10) and acetone

and then pickled in a diluted hydrofluoric acid bath with intermediate water rinses. The samples are either round sheets with a 1.7 cm² surface area or standardized microtensile samples. The bath composition (Europlate Ni216 commercial bath) and operation conditions used for preparing the NiP-talc composite coatings are given in Table 1. This bath was chosen because it yielded the optimal hardness. The bath was stirred by a rotating stirring rod (to keep particles from settling) with a rate of 500 rev min⁻¹. The talc particles employed were pretreated with carboxyl methyl cellulose (CMC) (long-chain polysaccharide), a well-known polymer depressant, to reduce the floatability of talc [24]. Talc is a 2:1 layer clay (phyllosilicate family) that links two tetrahedral sheets with one octahedral sheet. The average diameter when diluted in osmosed water was measured by photon correlation spectroscopy (Zetasizer 4, Malvern Instrument[®]) at 1.3 μm, and the talc particle concentrations in the bath varied from 20 to 120 g L⁻¹ talc. The deposition process was carried out for 2 h at a constant temperature of 88 °C with a fixed pH of 5 with or without talc particles. The electroless composite NiP-talc layers and the NiP layers were deposited to a thickness of approximately 30–40 μm. The number of incorporated particles by unit area was evaluated by image analysis (with Imagetools software) of micrographs obtained by Scanning Electron Microscopy (SEM). In Fig. 1, an increase in the number of particles incorporated into coatings with an increase in the talc content of the bath can be observed, while the phosphorus contained in the NiP matrix simultaneously increased slightly from 4.69 to 5.65 wt. % for 120 g L⁻¹ of talc in the bath. Indeed, for other electroless and electroplating composite coatings, such as NiP–Si₃N₄, CeO₂, or TiO₂, it has been shown by Balaraju [25] that an effective factor for trapping particles in the coating matrix is the particle content of the bath. The samples were heat-treated for 1 h at 420 °C or 600 °C under a controlled atmosphere and then were allowed to cool down in a high-vacuum environment for at least 20 min prior to their exposure to the atmosphere.

2.2. Observations and measurements

The surface morphology of both NiP and NiP-talc layers was examined by means of scanning electron microscopy (Philips SEM

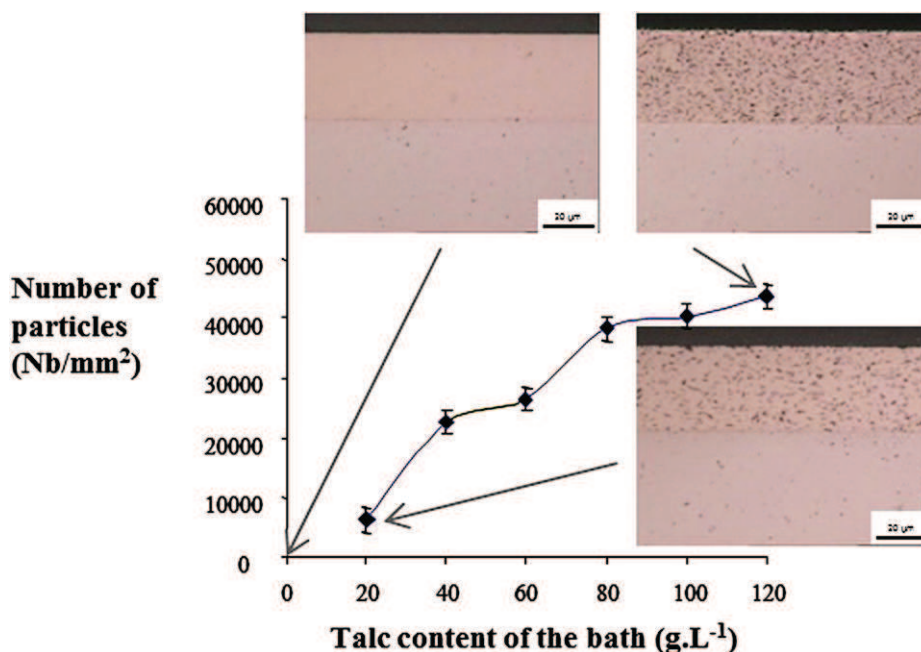


Fig. 1. Talc content of the coating as function of the talc content of the bath.

515[®]) equipped with an electron dispersive spectrometer that was used for the determination of the chemical composition of the deposits. An X-ray diffraction analysis has also been performed using a Cu K_α X-ray source (Philips X'Pert MRD[®]) to study the coatings' structure changes and the residual stresses. A VYCO NT1100 optical profiler (Veeco[®]) was employed to provide roughness data. Five replicate scans were performed on different zones. The average roughness parameter (Sa) and the Surface Area Ratio (Sdr) were measured. The Surface Area Ratio, Sdr, is the ratio between the surface area (taking the z height into account) and the area of the flat x-y plane. This parameter, also called the Developed Interfacial Area Ratio, is expressed as the percentage of additional surface area contributed by the texture compared with an ideal plane the size of the measurement region. Nanohardness tests were performed with the MTS XP[®] nanoindenter in dynamic mode, making it possible to obtain the Hardness and Young's Modulus values of the coatings. The indenter used is a Berkovitch pyramidal indenter. Nanoindentation tests were also performed on deposits treated at 600 °C to determine the influence of over-aging. For this study, two types of adhesion tests were used, the well-known and common microscratch test and a microtensile test. The microscratch tests were realized with a CSM[®] microscratch tester equipped with a diamond Rockwell indenter with a 200 μm radius. The technique involves generating a controlled scratch with a diamond tip on the sample under test. The tip, either made of diamond or sharp metal, is drawn across the coated surface under a progressive load. At a certain critical load, the coating will start to fail. The critical loads are very precisely detected by means of an acoustic sensor connected to the load arm together with observations from a built-in optical microscope. The critical load data are used to quantify the adhesive properties of different film–substrate combinations. Three scratches are made on each coating with the following parameters: initial load: 0.1 N; final load: 20 or 30 N; scratch length: 2 mm; scratch speed: 1 mm min⁻¹; acoustic emission threshold: 9; space between two scratches: 2 mm. Physico–chemical analyses by X-ray energy spectrometry are made to determine the type of rupture (adhesive or cohesive).

The microtensile test is used on systems made up of a ductile, coated substrate to analyze the mechanical stability and the damage to the deposits [26,27]. These tests require various experimental measurements, such as the tensile strength and spacing inter-crack characteristic of film split up. During the tensile tests, a low rate of deformation equal to 0.06 mm min⁻¹ has been implemented. According to the Agrawal and Raj model [28], the maximum interfacial shearing force, τ , developed for the interface between the deposit and the metal, is related to the tensile strength of the coating, σ , by equation (1), where $\sigma = E\varepsilon_f$, with ε_f representing the stress corresponding to the beginning of cracking of the deposit. E and δ are, respectively, the Young modulus and the thickness of the coating. Finally, λ is the wavelength of the sine function that defines the maximum interfacial shearing force. It is equivalent to a virtual spacing inter-crack characteristic of saturation in transverse cracks, in spite of the increase in the tensile stress. This equation is valid for a system film/coating free from residual stresses. This model was then modified to extend its validity for a thicker and harder deposit with residual stresses σ_r (equations (2) and (3)) [29].

$$\tau = \frac{\pi\delta\sigma_f}{\lambda}, \quad (1)$$

$$\sigma_e = \sigma_f + \sigma_r, \quad (2)$$

$$\tau = \frac{\pi\delta\sigma_e}{\lambda}, \quad (3)$$

3. Results and discussion

3.1. Morphology & topography

The SEM micrographs of representative NiP and NiP-talc deposits are presented in Fig. 2. The presence of talc particles in the NiP layer induced modifications of the surface morphology and an increase in the roughness of the deposits (Fig. 3). The nodule quantity increased with the presence of talc in the coatings. Moreover, it would seem

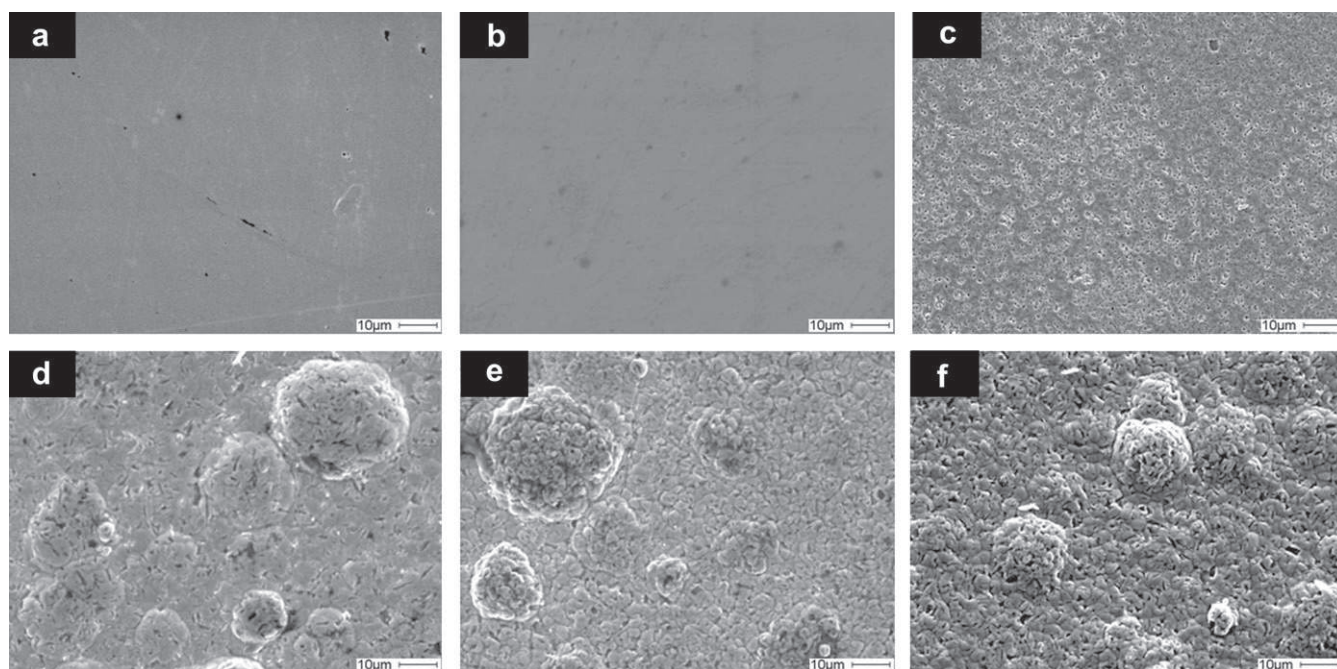


Fig. 2. Surface morphology of (a) as-plated NiP, (b) NiP heat-treated at 420 °C, (c) NiP heat-treated at 600 °C, (d) as-plated NiP-120 g L⁻¹ talc, (e) NiP-120 g L⁻¹ talc heat-treated at 420 °C and (f) NiP-120 g L⁻¹ talc heat-treated at 600 °C.

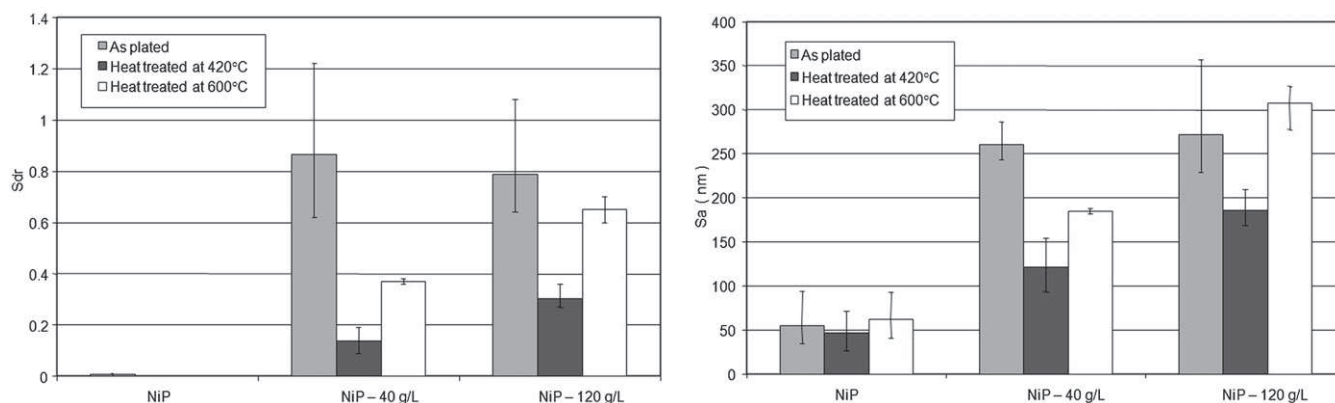


Fig. 3. Roughness measurements for the as-plated and heat-treated NiP and NiP-talc composite coatings.

that the insertion of talc particles is responsible for the lower cohesion of the coatings. Indeed, micro-porosities are present at the surface of the deposits containing talc particles. The surface examination after heat treatments shows that the deposits become slightly smoother (Fig. 3). The heat treatments have a slight effect on the morphology (Fig. 2); only the 600 °C heat treatment tends to increase the sub-micron pores observed on the untreated deposits.

3.2. Microstructure

The diffractograms of the various deposits according to their content of talc particles are shown in Fig. 4. These measures show that whatever the content of talc, the deposits are poorly crystalline

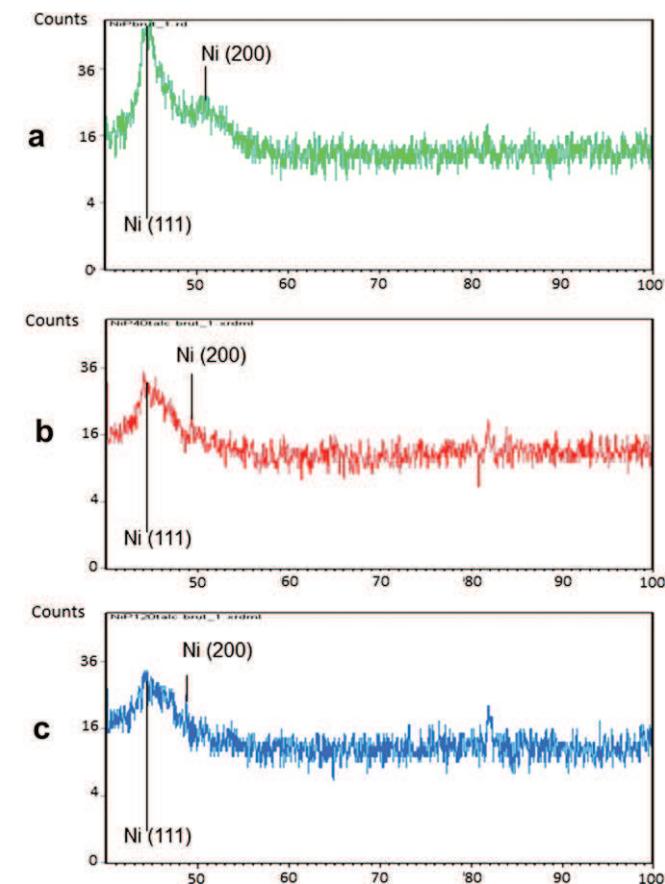


Fig. 4. XRD patterns of as-plated (a) NiP coating, (b) NiP-40 g L⁻¹ talc coating and (c) NiP-120 g L⁻¹ talc coating.

due to their phosphorus content (≈ 5 wt. %) (Table 2). With the increase of the talc content of the bath, a slight modification of the expansion or location of the diffraction peaks is observed. The plane (200) gradually disappears. The intensity of the peak (111) is larger for rough deposits without talc particles, and the peak width at half height is less important; this is a sign of greater crystallinity. Therefore, the particles of talc appear to play a role in the crystallinity of the coatings. Indeed, the addition of talc in coatings is accompanied by a very slight increase of phosphorus in coatings that become increasingly amorphous.

It has been confirmed by several studies that structural changes from an amorphous state to a crystalline state can be obtained in NiP electroless coatings over 300 °C [30,31]. Controlled atmosphere heat treatments (420 °C–1 h and 600 °C–1 h) of both NiP-talc and NiP lead to rapid crystallization of the amorphous matrix. Based on the DSC and TEM analysis, many authors have shown that the thermal stability of low-phosphorus coatings seems superior to that of high-phosphorus coatings, but the formation of Ni₃P and the crystallization is completed at 450 °C [32–34]. The introduction of talc in the coating does not alter the nature of the phases present but seems to have an effect on texture (Fig. 5). It has been observed in these composite deposits that there are no other phases except Ni and Ni₃P. The intensity ratio between the diffraction peaks varies with the insertion of talc particles. Similar structural changes have been reported for NiP–SiC electroless composite coatings [35].

3.3. Mechanical properties

Nanoindentation tests are performed to determine the hardness and the rigidity of the coatings. The loading-unloading curves (Fig. 6) do not vary significantly according to the talc content. Only the load reached for an imposed displacement of 2000 nm decrease with the insertion of talc. The springback of approximately 450 nm is the same regardless of the amount of talc coating present. A significant decrease in rigidity is observed by adding small numbers of particles (Fig. 7). Beyond 20 g L⁻¹ of talc in the bath, the rigidity continues to decline to $E = 120$ GPa for the NiP-120 g L⁻¹ talc. A similar trend was observed for hardness, which decreases sharply

Table 2
Deposit composition as a function of the talc content of the bath.

Talc content of the bath (g L ⁻¹)	Coating composition (wt. %)				
	Ni	P	Mg	Si	O
0	95.31	4.69	–	–	–
40	88.09	5.46	1.28	1.17	4.00
120	85.13	5.65	1.51	1.54	6.17

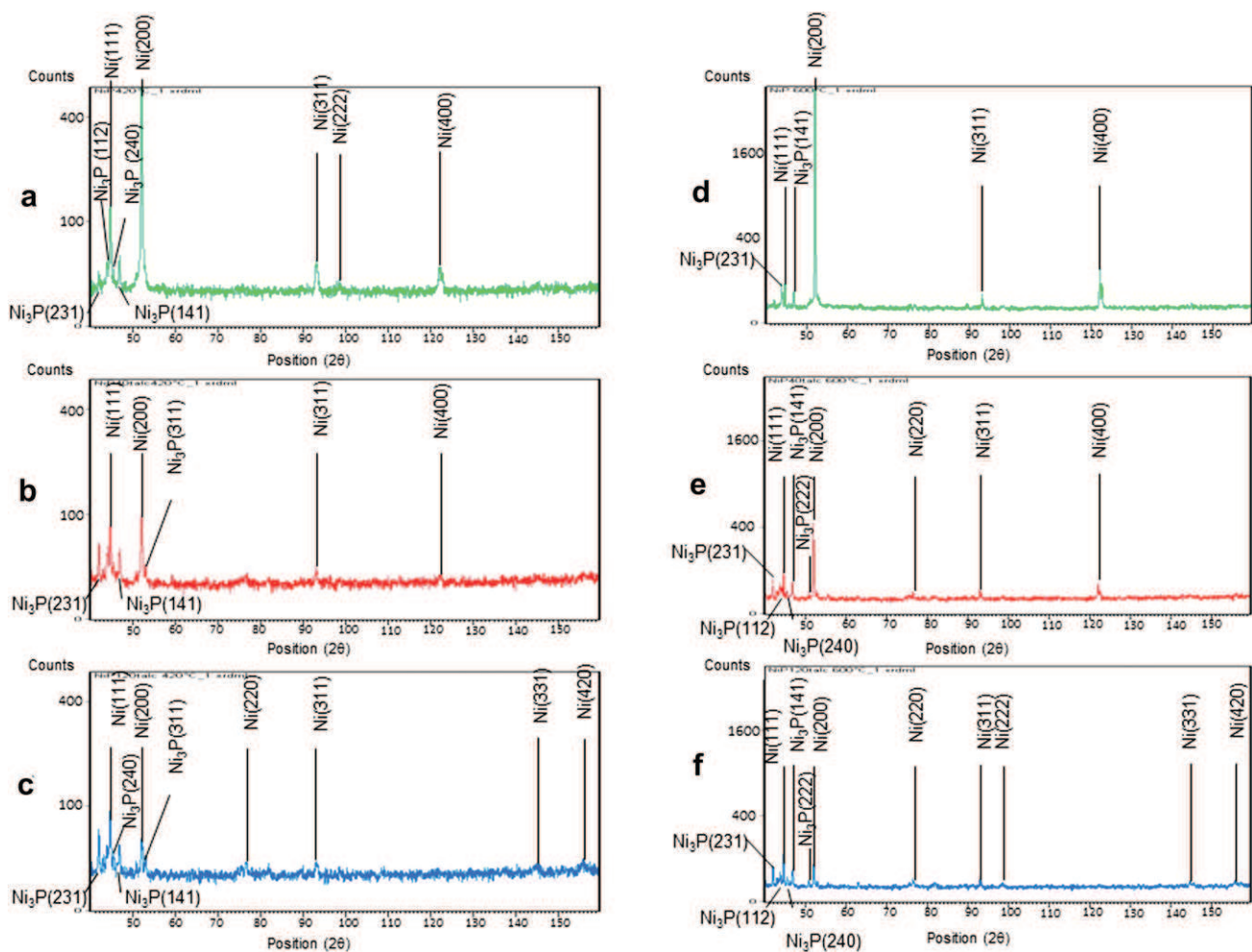


Fig. 5. XRD patterns of (a) NiP, (b) NiP-40 g L⁻¹ talc and (c) NiP-120 g L⁻¹ talc coatings heat-treated at 420 °C and (d) NiP, (e) NiP-40 g L⁻¹ talc and (f) NiP-120 g L⁻¹ talc coatings heat-treated at 600 °C.

with the insertion of a small quantity of talc ($H_{\text{NiP}} = 8.5$ GPa and $H_{\text{NiP-20}} = 6.7$ GPa). The hardness then decreases slowly beyond this concentration in the bath ($H_{\text{NiP-Talc120}} = 5.7$ GPa). This decrease can be explained only by the incorporation of particles with very low mechanical properties (H and E), such as talc (Fig. 8), and not by the slight increase in the phosphorus content in the matrix. Indeed, some authors [36] have reported that an increase of phosphorus involves a hardening of the NiP solid solution.

The heat treatments have a significant influence on the hardness and rigidity of the coatings. Indeed, the curves of loading–unloading vary depending on the heat treatment (Fig. 6). The springback is greater after treatment at 420 °C and lowest after treatment at 600 °C. Furthermore, the load reached for an imposed displacement of 2000 nm increases after treatment at 420 °C and decreases after treatment at 600 °C. Note the large dispersion of measurements recorded for deposits treated at 600 °C. The variation of Young's modulus and the hardness as a function of temperature treatment are presented in Fig. 7. As noted earlier, the deposit without talc is noteworthy. However, Young's modulus and the hardness increase when the rough as-deposited coating is treated at 420 °C, irrespective of the talc content. These hardness curves are typical of a precipitation hardening because the hardness H increases up to a maximum (420 °C-aging) and then decreases (600 °C-overaging). The increase in hardness and stiffness are due to the precipitation of nickel phosphides (Ni₃P) and crystallization of the Ni matrix.

Beyond 420 °C and until 600 °C, Young's modulus of the coating produced without talc fell slightly but remained above that of the rough deposit. The small reduction shows that no significant change in structure occurs between these two temperatures. The same is not true for deposits containing talc. The hardness also decreases between a deposit treated at 420 °C and another treated at 600 °C. This sharp decrease in hardness is due to grain growth, to

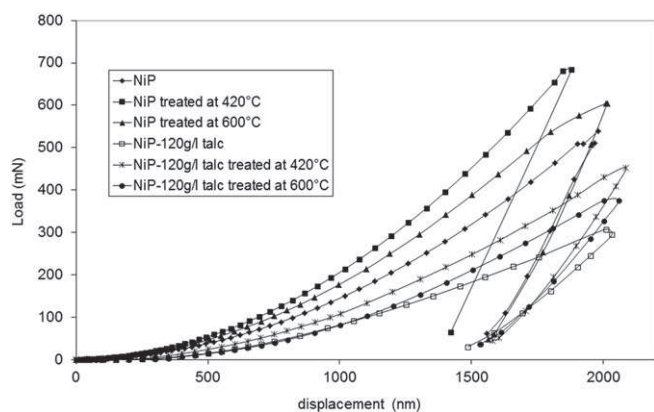


Fig. 6. Nanoindentation load–depth (P – h) curves of NiP and NiP-120 g L⁻¹ talc coatings as deposited and treated at various temperatures.

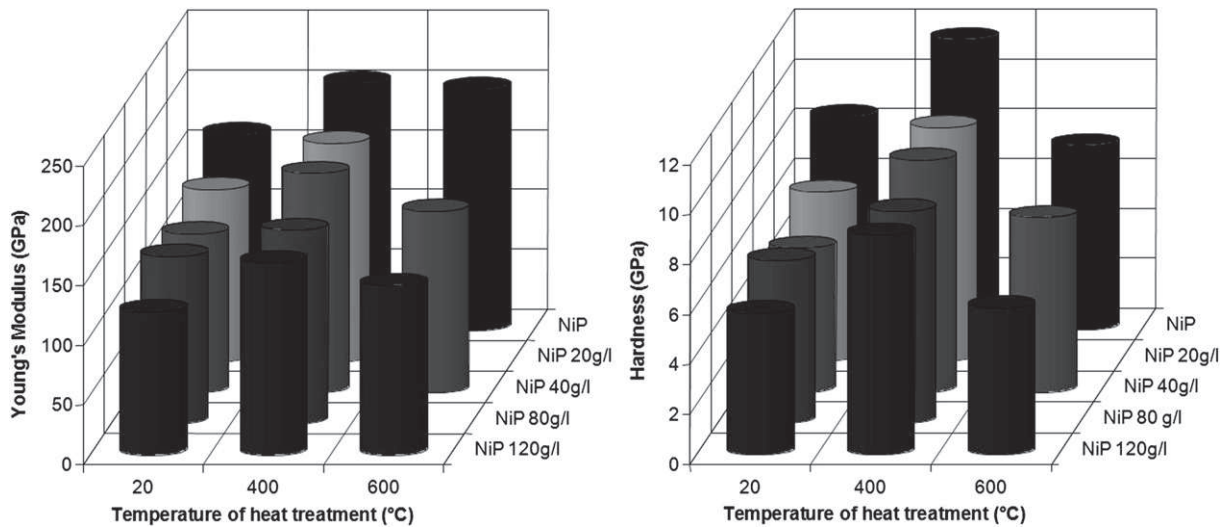


Fig. 7. Young's modulus and hardness of NiP coating and NiP-talc coatings as deposited and treated at various temperatures.

the coalescence of the precipitates and/or to degradation or debonding of mineral fillers.

The toughness of the coatings was determined from Vickers hardness tests. The crack length was measured as a function of the applied load [37,38]. The method of successive surface polishing is used to confirm that the cracking is Palmqvist-type cracking, i.e., the beginning of the crack deviates from the edge of the footprint as clarified by the polishing steps. The Palmqvist-type cracking justifies the choice of the model of Shetye, Wright, Mincer and

Clauer for determining the toughness (equation (4)) [39]. This model was used because it has the advantage of not being a function of Young's modulus or the hardness of the sample.

$$Kc = 0.0319 P / (al^{1/2}) \quad (4)$$

The toughness increases with the insertion of talc when the deposit is heat-treated at 420 °C (Table 3). In addition, heat

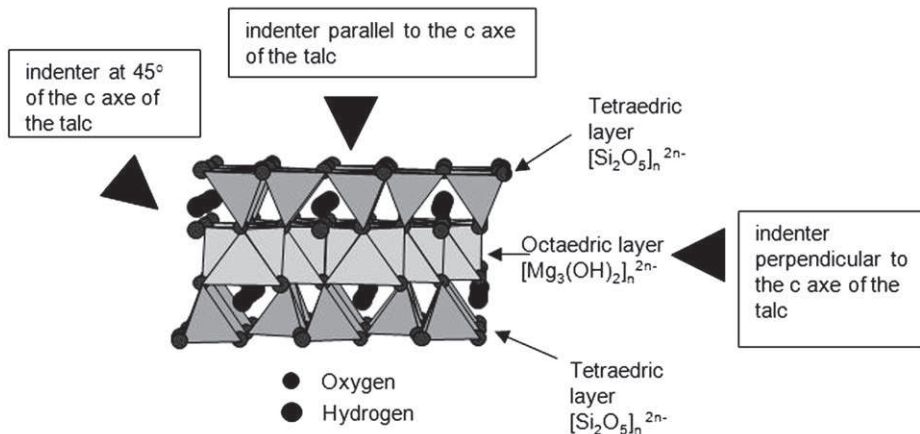
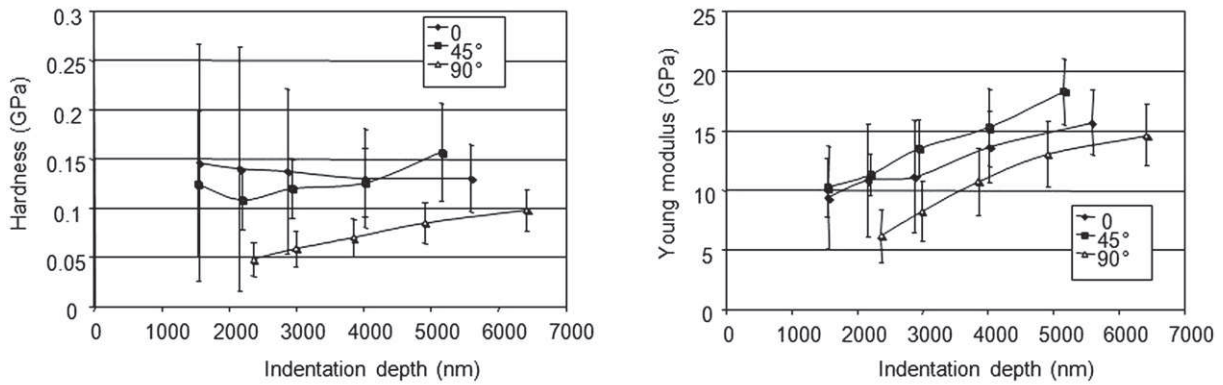


Fig. 8. Evolution of the hardness and Young's modulus measured on a macro-particle of talc according to the orientation of the layered structure (0°, 45° and 90° with regard to the perpendicular axis of the 111 plane).

Table 3

Toughness and residual stresses of NiP, NiP-40 g L⁻¹ and NiP-120 g L⁻¹ coatings after treatment at 420 °C and 600 °C.

	Type of coatings	Kc (Mpa m ^{1/2})	Residual stresses (MPa)
Heat treated 420 °C	NiP	2.37	39 ± 5
	NiP-40 g l ⁻¹	3.65	143 ± 24
	NiP-120 g l ⁻¹	4.02	186 ± 10
Heat treated 600 °C	NiP	3.75	9 ± 3
	NiP-120 g l ⁻¹	3.46	211 ± 10
	NiP-40 g l ⁻¹	–	247 ± 20

treatment at 600 °C increases the coating's toughness. The presence of talc does not affect the toughness of deposits processed at 600 °C. A. Roman et al. [40] found comparable values: $K = 1.5 \text{ MPa}\sqrt{\text{m}}$ for the 300 °C heat treatment and $2.1 \text{ MPa}\sqrt{\text{m}}$ for the 600 °C heat treatment.

The residual stresses in the heat-treated coatings have been determined by X-ray diffraction. Evaluation of the residual stresses by the $\sin^2 \psi$ method was performed. A copper anode X-ray tube was used, and measurements were made on the (420) nickel diffraction peak at a 2θ angle of approximately 155.6° with an

irradiated area of approximately 4 mm^2 . Seven ψ offsets were measured for the two principal in-plane stress directions. A Gaussian function was used to fit the diffraction peaks. The parameters are $S1 = 1/\text{TPa}$ and $S2 = 6.32/\text{TPa}$. All types of deposits are tested in a slight state of tension (Table 3). Talc tends to increase this state of tension.

The curves and optical observations of scratches are presented in Fig. 9. The response of the coating and substrate depends on the heat treatment and the amount of talc in the bath. Two mechanical behaviors stand out: ductile behavior for as-deposited coating and deposits heat-treated at 600 °C and fragile behavior for deposits treated at 420 °C (Fig. 10). The ratio between the thickness of the deposit at the side of the track and at the center is a discriminating parameter for ductile (ratio > 1.4) and brittle behavior (compared to 1). The influence of heat treatment is determined on coatings with 120 g L^{-1} talc; a ratio of 1.61 is obtained for the untreated coating, a ratio of 1.07 for the heat-treated at 420 °C and a ratio of 1.44 for the coating heat-treated at 600 °C. These results show that the untreated and treated coatings at 600 °C are more ductile. The plastic deformation is absorbed mainly by the coating. For the coating treated at 420 °C, the coatings' thickness remains constant, regardless of the zone observed; the plastic deformation is located

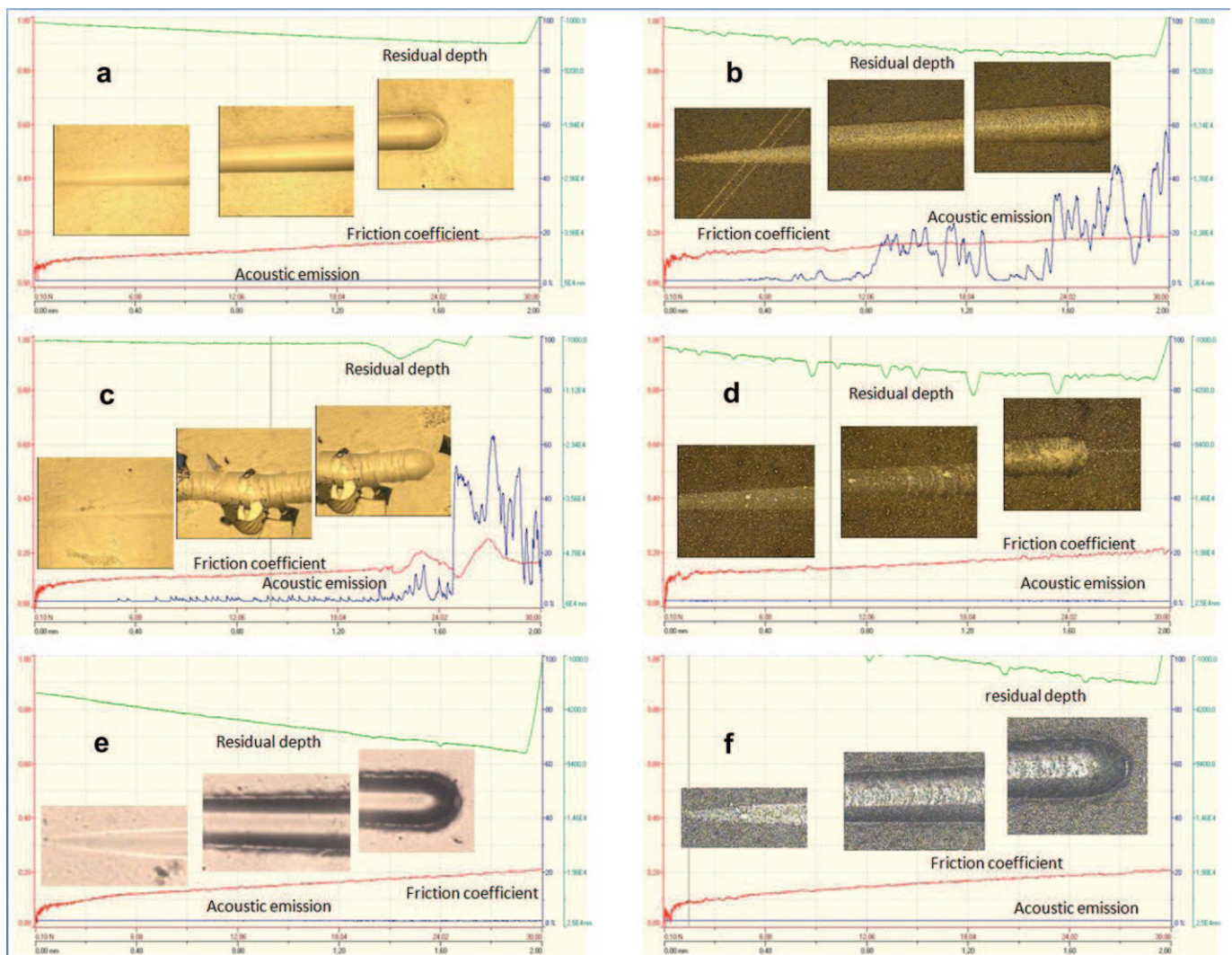


Fig. 9. Curves and optical observations of scratches on NiP and NiP-120 g L⁻¹ coatings, respectively, as deposited (a, b), after treatment at 420 °C (c, d) and after treatment at 600 °C (e, f).

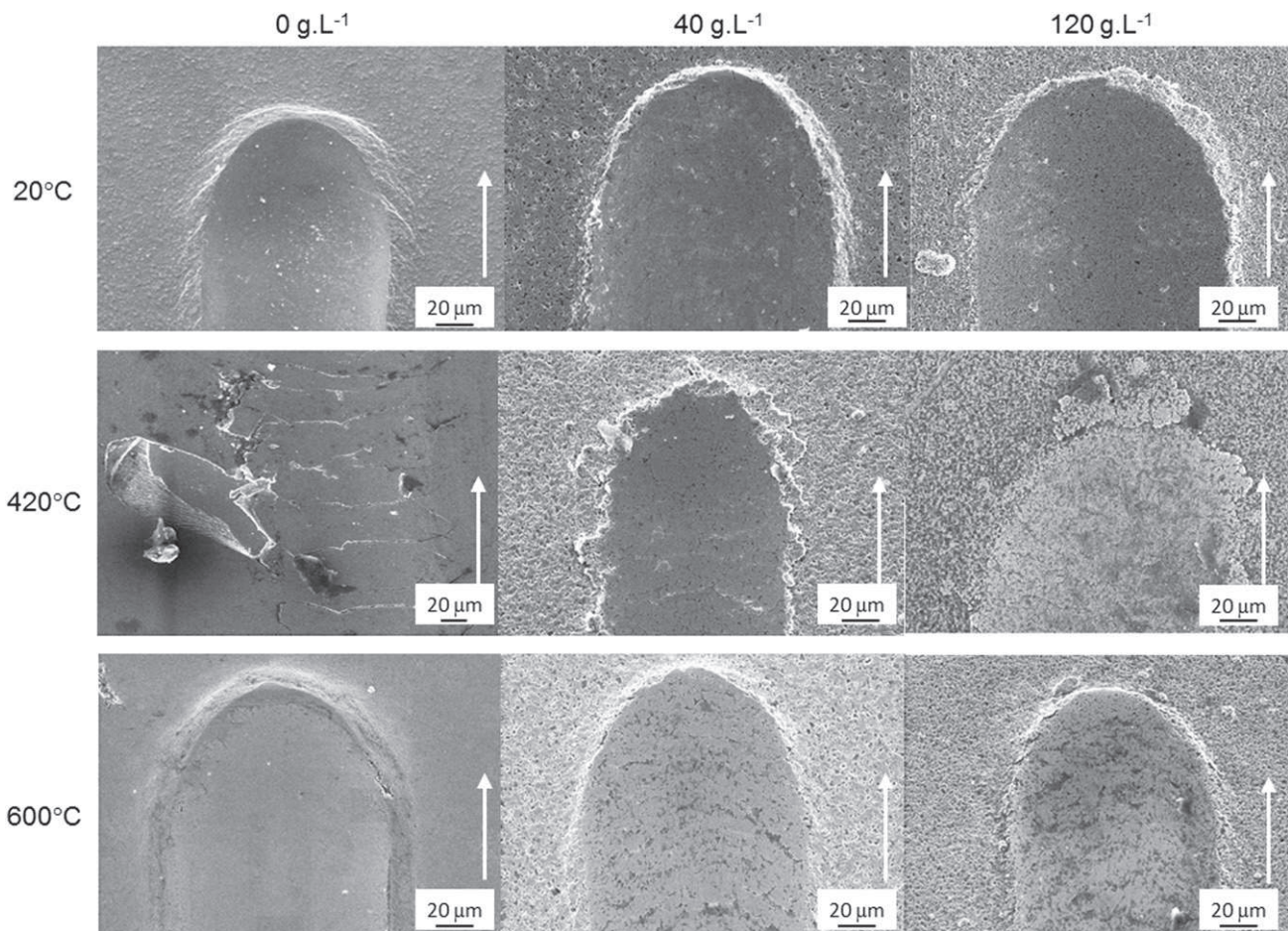


Fig. 10. SEM observations of the scratches on NiP, NiP-40 g L⁻¹ talc and NiP-120 g L⁻¹ talc coatings as deposited, after treatment at 420 °C and after treatment at 600 °C.

in the substrate rather than in the fragile coating. The deposit only transmits the mechanical stress introduced by the indenter. The influence of the amount of talc is determined from the surfaces treated at 600 °C; the values are 1.57, 1.46 and 1.44 for 0, 40 and

120 g L⁻¹ of talc, respectively. These values decrease slightly when the amount of talc is increased. To determine the critical loads, the parameters used were the acoustic emission, depth and residual optical observations. All critical loads are determined depending on

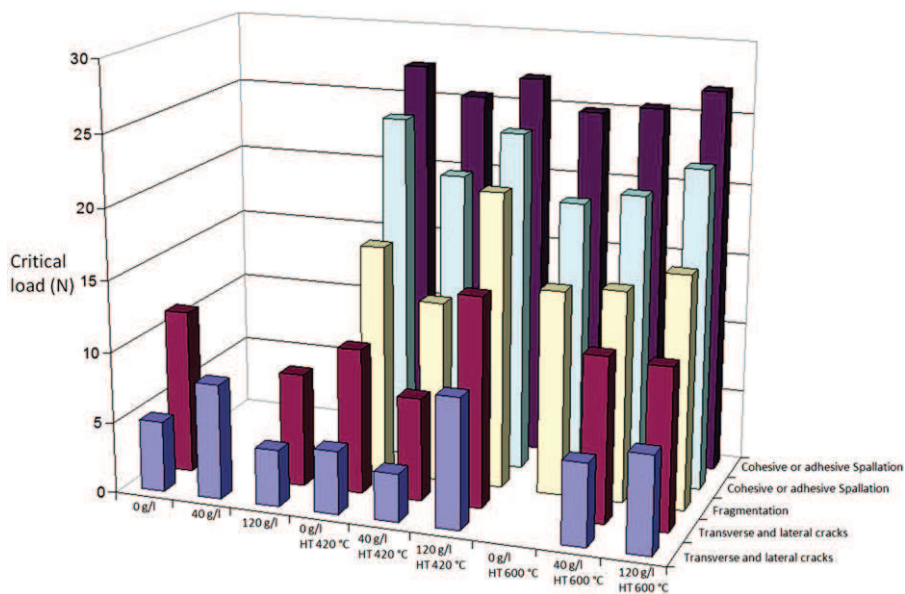


Fig. 11. Critical loads obtained from scratch curves of NiP, NiP-40 g L⁻¹ and NiP-120 g L⁻¹ coatings as deposited, after treatment at 420 °C and after treatment at 600 °C.

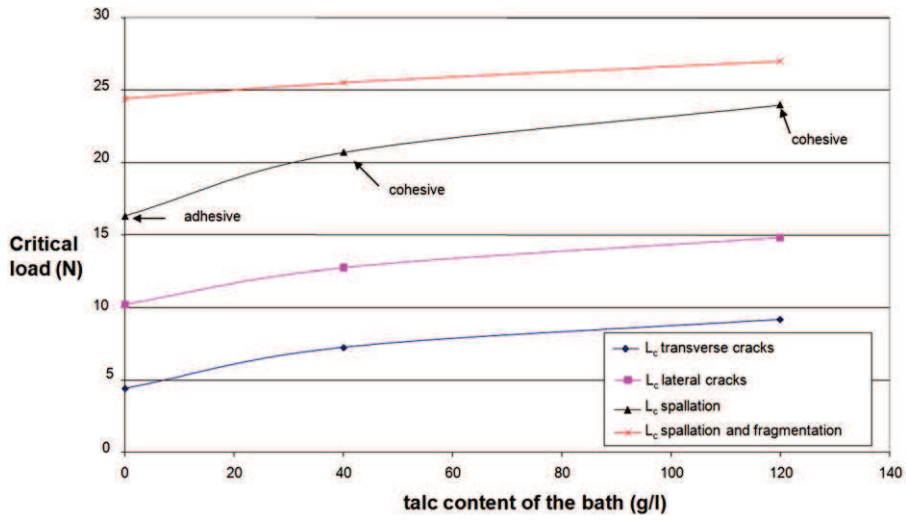


Fig. 12. Critical loads as a function of the talc content in the bath (g L^{-1}) for coatings heat-treated at 420°C .

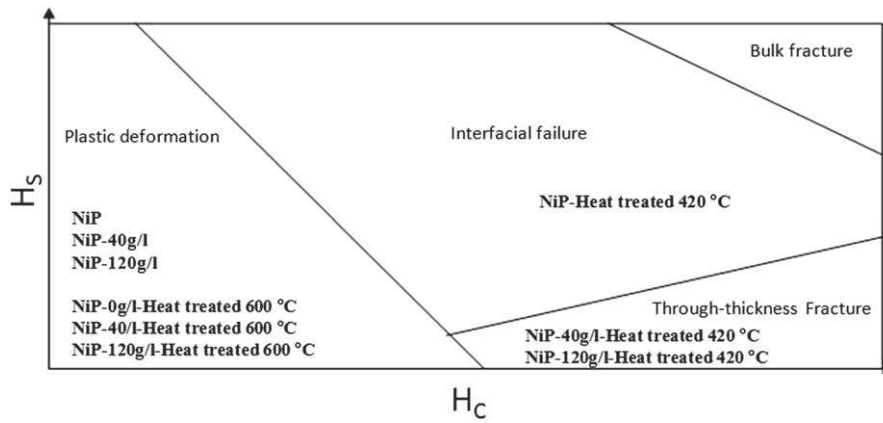


Fig. 13. Graphic representation showing the various scratch test failure modes that dominate as a function of the coating and the substrate hardness of the coatings (H_c and H_s , respectively) for the different NiP and NiP-talc coatings.

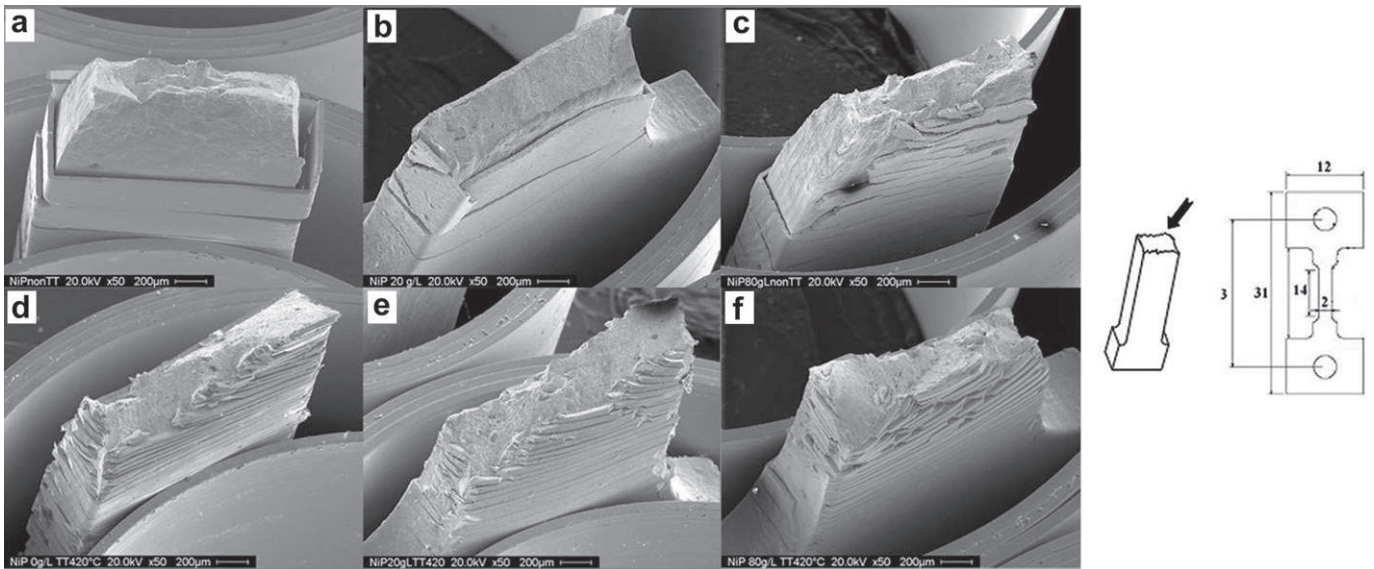


Fig. 14. SEM micrographs after tensile tests of NiP and NiP- 20 g L^{-1} talc and NiP- 80 g L^{-1} coatings, respectively, as deposited (a)(b)(c) and after treatment at 420°C (d)(e)(f).

Table 4
Critical stresses (τ_{A-R}) determined by the model of Agrawal and Raj for each type of coating.

Treatments	Coatings	$e_f(\mu\text{m})$	$\lambda_0(\mu\text{m})$	$E_f(\text{GPa})$	$\epsilon_f(\%)$	$\sigma_R(\text{GPa})$	$\tau_{A-R}(\text{GPa})$	$\tau_{res}(\text{GPa})$	Rupture mode
Untreated	NiP	35.4	1562	164	1.37	—	0.08	—	Adhesive
	NiP-20 g l ⁻¹	38.6	200	140	3.75	—	0.28	—	Mixed
	NiP-40 g l ⁻¹	35.4	451	134	1.94	—	0.32	—	Mixed
	NiP-80 g l ⁻¹	39.3	100	130	3.51	—	3.03	—	Cohesive
	NiP-120 g l ⁻¹	47.8	44	119	4.34	—	9.96	—	Cohesive
Heat treated at 420 °C	NiP	34.3	27	201	0.4	0.039	1.1	1.14	Cohesive
	NiP-20 g l ⁻¹	37.5	10	180	1.67	—	3.89	—	Cohesive
	NiP-40 g l ⁻¹	36.8	31	181	2.06	0.143	4.45	4.59	Cohesive
	NiP-80 g l ⁻¹	39.4	15	160	1.24	—	2.45	—	Cohesive
	NiP-120 g l ⁻¹	47.7	34	172	2	0.186	6.35	6.54	Cohesive
Heat treated at 600 °C	NiP	36.8	148	202	1.54	0.009	1.21	1.22	Adhesive
	NiP-40 g l ⁻¹	42.2	299	152	2	0.211	0.8	1.01	Adhesive
	NiP-120 g l ⁻¹	29.2	139	142	2.06	0.247	0.96	1.21	Adhesive

the type of coating: talc content and heat treatment. Moreover, scanning electron microscopy observations were performed to attribute physical phenomena to each critical load (L_c): simple cracking ($L_{C1\&2}$: lateral or transverse), multiple cracking (L_{C3} : fragmentation) and spalling ($L_{C4\&5}$: adhesive or cohesive). The results are summarized in Fig. 11. For coatings heat-treated at 420 °C, the talc particles delay the onset of phenomena such as spalling (Fig. 12). This behavior is due to a better mechanical accommodation. For coatings heat-treated at 600 °C, talc seems to provide some stress accommodation, and plastic deformation is slightly less important. These behaviors are consistent with the fracture map patterns describing the hardness of the coating and the substrate provided by Bull (Fig. 13) [41].

From the tensile tests (Fig. 14), the crack density due to the strain rate was measured during the test by video acquisition. Talc seems to modify the adherence of the coatings. It was shown that the rate of cracking to saturation increases with the talc content for untreated coatings. Moreover, the cracks are deviated on coatings containing talc. The observations by SEM of the rupture surface confirm this result and attest to the brittle fracturing of the coatings. The rupture is adhesive for untreated deposits without talc and becomes mixed with the insertion of talc. The cohesion of the particles/matrix seems less strong than that of the deposit/substrate. The crack density due to the rate of deformation increases significantly after a heat treatment. The observations made in situ by an optical microscope and made post mortem by SEM confirm the increase in the adherence of the heat-treated deposits. The coatings remain adherent close to the rupture zone of the substrate (Fig. 14). The influence of talc on the deviation of the cracks is less obvious because of the better cohesion of the particles/matrix but more importantly because of the improved rigidity of the deposits. The ruptures of coatings are adhesive without talc and are cohesive with talc. The interfacial shear stress was thus given by the model of Agrawal and Raj without taking into account the residual stresses (τ_{A-R} and τ_{res}) of the coatings (Table 4). Note that for the studied coatings, τ_{A-R} and $\tau_{residual}$ are almost identical because the values of $\sigma_{residual}$ (a few hundreds of MPa) are negligible compared with $\sigma_{applied}$ (on the order of GPa). The heat treatment increases the adherence of the deposit considerably. It also seems that the talc addition improves the behavior of the coating. These results confirm preceding experimental results and testify to the relevance of using the model to quantitatively investigate the adherence of the studied coatings.

4. Conclusion

The present investigation has demonstrated that talc powder could be incorporated successfully in a Ni–P. The observations

carried out with optical microscopy showed good homogeneity of the deposits. The roughness of the deposits increases with the incorporation of talc particles in the bath due to an important morphological change observed by scanning electron microscopy. In fact, the NiP structure changes radically to a globular structure with the addition of dispersoids in the solution. The heat treatments do not significantly influence the roughness but do modify the microstructure. This study highlights several interesting mechanical properties of the composite NiP coatings with talc dispersoids. The values of the hardness and Young's modulus decrease slightly with the increase in talc content when the coating was not treated thermally. In contrast, with a recrystallization heat treatment at 420 °C, the hardness and Young's modulus remain quasi-constant, regardless of the content of dispersoids (20–120 g L⁻¹ in solution). This treatment produces desirable hardness and Young's modulus values (from 8 to 11 GPa for nano-hardness and Young's modulus values from 160 to 210 GPa as a function of the talc content) due to the Ni₃P precipitation. Finally, a 600 °C treatment degrades these parameters considerably. The hardness increases with the insertion of talc when the deposit is heat-treated at 420 °C or with a heat treatment at 600 °C. Two separate mechanical behaviors are shown by a scratch test as a function of the intrinsic properties of the coatings: ductile behavior for rough deposits and those heat-treated at 600 °C and fragile behavior for deposits treated at 420 °C (Fig. 10). The microtensile tests showed that a 420 °C treatment causes an important improvement in the adhesion level of the coating, more precisely, of the matrix NiP. The talc content also increases the adhesion of the untreated or heat-treated coatings. The rupture is cohesive when the coatings contain talc dispersoids.

Acknowledgments

This study received the financial support of the “Conseil Régional de Midi Pyrénées” under the project heading “multipurpose composite Coatings for the industry of transport”.

References

- [1] T.S.N. Sankara Narayanan, S. Selvakumar, A. Stepen, Surface and Coatings Technology 172 (2) (2003) 298–307.
- [2] A. Brenner, P. Couch, E. Williams, Research paper n°2061, Journal Research National Bureau of Standards 44 (1) (1950) 109–122.
- [3] I.R. Aslanyan, J.P. Bonino, J.P. Celis, Surface and Coatings Technology 200 (2006) 2909–2916.
- [4] J.N. Balaraju, Kalavati, K.S. Rajam, Surface and Coatings Technology 200 (12–13) (2006) 3933–3941.
- [5] A. Zoildis-Karathanasis, E.A. Pavlatou, N. Spyrellis, Electrochimica Acta 54 (2009) 2563–2570.
- [6] S.M.M. Vaghefi, A. Saatchi, M. Ebrahimian-Hoseinabadi, Surface and Coatings Technology 168 (2003) 259–262.

- [7] V.V.N. Reddy, B. Ramamoorthy, P.K. Nair, *Wear* 239 (2000) 111–116.
- [8] O.A. Leon, M.H. Staia, H.E. Hintermann, *Surface and Coatings Technology* 108 (1998) 461–465.
- [9] Z. Abdel Hamid, I.M. Ghayad, *Materials Letters* 53 (2002) 238–243.
- [10] M.D. Ger, B.J. Hwang, *Materials Chemistry and Physics* 76 (2002) 38–45.
- [11] Y.T. Wu, L. Lei, B. Shen, W.B. Hu, *Surface and Coatings Technology* 201 (2006) 441–445.
- [12] G. Straffellini, D. Colombo, A. Molinari, *Wear* 236 (1–2) (1999) 179–188.
- [13] R. Xu, J. Wang, Z. Guo, H. Wang, *Journal of Rare Earths* 26 (4) (2008) 579–583.
- [14] M.H. Ma, J.X. Chen, X.H. Li, *Journal of Inorganic Chemistry* 17 (1) (2001) 101–107.
- [15] Q. Zhao, Y. Liu, H. Muller-Steinhagen, G. Liu, *Surface and Coatings Technology* 155 (2002) 279–284.
- [16] E. Pena-Munoz, P. Berçot, A. Grosjean, M. Rezrazi, J. Pagetti, *Surface and Coatings Technology* 107 (1998) 85–93.
- [17] P. Berçot, E. Pena-Munoz, J. Pagetti, *Surface and Coatings Technology* 157 (2002) 282–283.
- [18] Y.S. Huang, X.T. Zeng, I. Annergren, F.M. Liu, *Surface and Coatings Technology* 167 (2003) 207–211.
- [19] D.B. Lewis, G.W. Marshall, *Surface and Coatings Technology* 78 (1996) 150–156.
- [20] C.C. Hu, A. Bai, *Materials Chemistry and Physics* 77 (2003) 215–225.
- [21] M.H. Seo, J.S. Kim, W.S. Hwang, D.J. Kim, S.S. Hwang, B. Sun Chun, *Surface and Coatings Technology* 176 (2004) 135–140.
- [22] B. Etcheverry, PhD thesis, Institut National Polytechnique de Toulouse, 2006.
- [23] B. Etcheverry, F. Le Coz, J. Alexis, J.Y. Paris, J.A. Petit, *Proceeding of the 18th International Conference on Surface Modification Technologies*, 2006, 355–363.
- [24] P. Bacchin, J.-P. Bonino, F. Martin, M. Combacau, P. Barthes, S. Petit, J. Ferret, *Colloids and Surfaces A Physicochemical and Engineering Aspects* 272 (2006) 211–219.
- [25] J.N. Balaraju, PhD thesis, IIT Madras, Chennai, 2000.
- [26] A. Mezin, J. Lepage, N. Pacia, D. Paulmier, *Thin Solid Films* 172 (2) (1989) 197–209.
- [27] P.H. Wojciechowski, M.S. Mendiola, *Journal of Vacuum Science Technology A7* (3) (1989) 1282–1287.
- [28] D.C. Agrawal, R. Raj, *Acta Metallurgica* 37 (1989) 1265–1270.
- [29] B.F. Chen, J. Hwang, I.F. Chen, G.P. Yu, J.H. Huang, *Surface and Coatings Technology* 126 (2000) 91–95.
- [30] S. Vaillant, PhD thesis, Institut National Polytechnique de Toulouse, 2002.
- [31] G. Jiaqiang, W. Yating, L. Lei, S. Bin, H. Wenbin, *Materials Letters* 59 (2005) 1665–1669.
- [32] I. Apachitei, F.D. Tichelaar, J. Duszczyk, L. Katgerman, *Surface and Coatings Technology* 149 (2002) 263–278.
- [33] Th. Hentschel, D. Isheim, R. Kirchheim, F. Muller, H. Kreye, *Acta Materialia* 48 (2000) 933–941.
- [34] B. Farber, E. Cadel, A. Menand, G. Schmitz, R. Kirchheim, *Acta Materialia* 48 (2000) 789–796.
- [35] J. Alexis, B. Etcheverry, J.D. Beguin, J.-P. Bonino, *Materials Chemistry and Physics* 120 (2010) 244–250.
- [36] C.C. Hu, A. Bai, *Materials Chemistry and Physics* 77 (2002) 215–225.
- [37] C.B. Ponton, R.D. Rawlings, *Materials Science Technology* 5 (1989) 865–872.
- [38] C.B. Ponton, R.D. Rawlings, *Materials Science Technology* 5 (1989) 961–976.
- [39] D.K. Shetty, I.G. Wright, P.N. Mincer, A.H. Clauer, *Journal of Material Science* 20 (1985) 1873–1882.
- [40] A. Roman, D. Chicot, J. Lesage, *Surface and Coatings Technology* 155 (2002) 161–168.
- [41] S.J. Bull, E.G. Berasetegui, *Tribology International* 39 (2006) 99–114.