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1 **Spatio-temporal variability in benthic silica cycling in two macrotidal**
2 **estuaries: causes and consequences for local to global studies**

3

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10 ABSTRACT

11

12 The high heterogeneity of silica cycling in coastal margins and the lack of silica data
13 (compared to nitrogen and phosphorus) prevent the estimation of global silica retention in
14 estuaries. In this study, the spatial and temporal variability of pore water silicic acid ($\text{Si}(\text{OH})_4$)
15 profiles - that integrate benthic transport and reaction processes - was investigated at different
16 spatial (meter, longitudinal and cross-section, intra-estuary) and temporal (tidal, seasonal) scales
17 in two macrotidal estuaries, very close geographically but differing essentially in their shape.
18 Studying the spatial and temporal variability of $\text{Si}(\text{OH})_4$ concentrations in pore waters provided
19 evidence for the importance of transport processes, *e.g.* bio-irrigation, tidal pumping,
20 resuspension and any combination of these processes, in affecting $\text{Si}(\text{OH})_4$ concentrations and
21 fluxes and hence, temporary or permanent retention along the land-ocean continuum. We confirm
22 that aSiO_2 transported by rivers and estuaries clearly deserves to be better characterized as it
23 provides an important source of reactive aSiO_2 to sediments. This study brings crucial
24 information which allows to (1) deconvolute spatial and temporal scales, although both are most
25 often in complete interaction, (2) design the most appropriate sampling schemes to be
26 representative of any given system and to extrapolate at the scale of the whole estuary, (3)
27 quantify uncertainty associated to the estimations of $\text{Si}(\text{OH})_4$ stocks and fluxes in this type of
28 ecosystem, essential for budget calculations. We showed that two adjacent small macrotidal
29 estuaries, may exhibit different behaviors regarding Si retention. Temporary retention has been
30 observed in the meanders of the Aulne Estuary and not along the more linear Elorn Estuary,
31 demonstrating the importance of the morphology and hydrodynamism component of the estuarine

32 filter. This effort needs to be pursued in other systems and climatic zones, but our study suggests
33 that the typology should not only account for the different types of land-ocean continuum (fjord,
34 delta, mangrove...), but also incorporate the physical or biological attributes of the estuarine
35 filter.

36

37 *Keywords:* heterogeneity, season, silicate, sediment, land-sea interface, upscaling

38 1. Introduction

39

40 Studying the silica (Si) cycle - and especially estimating the transient and permanent
41 retention of Si - has important ecological and biogeochemical implications at small to global
42 scales. At the local scale, higher Si retention leads to Si limitations and decreasing Si:N and Si:P
43 ratios. These environmental modifications are responsible for shifts in phytoplanktonic
44 communities dominated by diatoms to other species *e.g.* cyanobacteria or toxic dinoflagellates,
45 which have repercussions on higher trophic levels (Officer and Ryther, 1980; Conley et al., 1993;
46 Howarth et al., 2011). Such Si limitations and eutrophic events have however been prevented by
47 benthic recycling and fluxes associated to transient retention in several estuarine and coastal
48 shallow ecosystems (Yamada and D'Elia, 1984; Ragueneau et al., 2002a; Struyf et al., 2006;
49 Laruelle et al., 2009). At the global scale, permanent retention decreases the export of Si to the
50 ocean, while bioavailable Si has an essential role in enhancing the biological carbon pump by
51 increasing particle sedimentation rate, aggregate formation, and protection from organic matter
52 degradation (Smetacek, 1985; Moriceau et al., 2007; Moriceau et al., 2009). Studying Si cycling
53 along land-sea continuum is particularly essential as coastal margins - including estuaries - must
54 strongly contribute to Si sinking (Bernard et al., 2010).

55 Nevertheless, the heterogeneity of Si cycling is still few quantified from local to global
56 scales (Jansen et al., 2010; Ragueneau et al., 2010; Dürr et al., 2011a; Moosdorf et al., 2011),
57 especially in estuaries (Dürr et al., 2011a). The difficulty in estimating estuarine heterogeneity is
58 related to the high spatial and temporal environmental variability. Estuaries are generally
59 characterized by the most heterogeneous and changing environmental conditions because of

60 numerous external forces and human activities (Nichols et al., 1986; Dalrymple and Choi, 2007).
61 In order to decrease uncertainties in estuaries, heterogeneity must thus be explored. Investigating
62 estuarine heterogeneity is moreover essential to handle research objectives and to design
63 associated sampling schemes (Dutilleul, 1993), to validate interpretations of ecosystem
64 functioning (Livingston, 1987; Hunt et al., 1997) and to ensure the quality of long-term records
65 (Wolfe et al., 1987). In addition to that, these quantifications are necessary to estimate incertitude
66 during downscaling and upscaling, and prevent significant errors incurred by failing to resolve
67 spatial and temporal variation (Swaney and Giordani, 2007; Swaney et al., 2012).

68 The variability of Si cycling, including retention, is strongly controlled by transport and
69 reaction processes. Transport processes are either physical, *e.g.* deposition and erosion of
70 amorphous silica (aSiO₂) (Arndt and Regnier, 2007), or biological, *e.g.* bioturbation, bioirrigation
71 (Aller, 1980; Berner, 1980), while reaction processes are either chemical, *e.g.* dissolution and/or
72 reprecipitation of aSiO₂ (Berner, 1980; Michalopoulos and Aller, 2004), or biological, *e.g.*
73 filtration of aSiO₂ by benthic filter feeders (Ragueneau et al., 2002a), uptake of dissolved silica
74 (Si(OH)₄) by benthic diatoms or sponges (Ni Longphuir et al., 2009), all these processes
75 depending on environmental factors. Even if the riverine flux of aSiO₂ to the ocean has generally
76 been neglected compared to the Si(OH)₄ flux (Tréguer et al., 1995), it may constitute some 16-40
77 % of the total Si inputs to estuaries (Conley, 1997; Smis et al., 2010) which may, at least partly,
78 settle in sediments. Depending on watersheds and seasons, estuarine aSiO₂ can derive from
79 terrestrial ecosystems, *e.g.* forest, grassland, wetland, soil (Conley, 2002; Blecker et al., 2006;
80 Gérard et al., 2008; Struyf and Conley, 2009), and/or from diatoms growing in rivers (Conley,
81 1997).

82 The numerous and heterogeneous environmental parameters and processes occurring in
83 estuaries moreover lead to variations of Si stocks and fluxes at different spatial and temporal
84 scales. At small scales, processes such as tidal resuspension or biological activity, lead to hourly,
85 daily, seasonal and spatial variations of benthic Si(OH)_4 fluxes (Sakamaki et al., 2006; Ni
86 Longphuir et al., 2009; Leynaert et al., 2011). At larger scales, annual and regional variations of
87 Si fluxes and retention are generated by different intrinsic properties, *e.g.* lithology, land cover,
88 climate, runoff (Jansen et al., 2010; Dürr et al., 2011a; Moosdorf et al., 2011), and human
89 activities, *e.g.* eutrophication, dam building, population density, invasive species (Conley et al.,
90 1993; Humborg et al., 1997; Roy et al., 1999; Ragueneau et al., 2002a).

91 The rapid response of estuarine ecosystems to environmental parameter variations makes
92 the characterization of material fluxes (*i.e.* aSiO_2 and Si(OH)_4) within the sediment and at the
93 sediment-water interface very difficult (Matisoff et al., 1975; Arndt and Regnier, 2007).
94 Investigating spatial and temporal variations of pore water Si(OH)_4 concentrations allows
95 however to integrate benthic process variations related to changing environmental factors. Pore
96 water profiles of Si(OH)_4 - or other dissolved metabolites - are then used to estimate
97 biogeochemical rates and fluxes (Khalil et al., 2007; Lettmann et al., 2012). Measurements of
98 pore water Si(OH)_4 profiles and their variability is thus useful to evaluate the variability of
99 benthic processes, but also the representativeness and uncertainty of pore water Si(OH)_4
100 concentrations in estuarine samplings.

101 The Bay of Brest is an example of a macrotidal system under oceanic climate,
102 downstream of a silicified watershed impacted by strong anthropogenic nitrogen enrichment (Del
103 Amo et al., 1997a). This shallow coastal embayment is characterized by a coastal - biological -

104 silicate pump associated to benthic fluxes (Del Amo et al., 1997b), increased by the presence of
105 benthic filter-feeders which prevent dinoflagellate blooms (Ragueneau et al., 2000; Laruelle et
106 al., 2009). While numerous studies have been performed in the bay, no studies were undertaken
107 on benthic Si cycling in the brackish estuaries. Describing benthic spatio-temporal variability and
108 the main transport and reaction processes involved is then crucial to evaluate the uncertainty
109 associated to benthic processes and fluxes. The two main estuaries flowing into the Bay (*e.g.*
110 Elorn and Aulne) are very close but characterized by different shapes. They are thus a good
111 example to determine the impact of linear and S-shape morphology on benthic Si cycle.

112 In this study, we investigated spatial and temporal variability of benthic Si(OH)_4
113 concentrations in two estuaries at various scales (tide, cross-section, intra-estuary, season). We
114 described and compared the variability of pore water Si(OH)_4 concentrations in the two estuaries,
115 and emphasized the main environmental factors, transport and reaction processes explaining this
116 variability. We discussed then the implications of these results for local, regional and global
117 investigations: Does the small scale variability allow studying seasonal and intra-estuary
118 variations? Is the heterogeneity similar between two macrotidal and temperate estuaries? What is
119 the impact on estimations of model uncertainty?

120 **2. Material and methods**

121

122 *2.1. Study site*

123

124 The Elorn and Aulne estuaries are located at the interface between their drainage basins
125 and the semi-enclosed Bay of Brest in Northwestern France (Fig. 1). These two estuaries supply
126 up to 85% of fresh water inputs to the Bay of Brest. They are characterized by similar watershed
127 lithology, climate and tidal regime, but have a different size, morphology and land use. The
128 lithology of the two watersheds is similarly dominated by silica-enriched rocks, mainly granite
129 and schist (Lague et al., 2000). The two watersheds are both characterized by intensive
130 agriculture activities. Urbanization is stronger in the Elorn watershed. The oceanic climate of the
131 region generates precipitations of 1145 mm yr⁻¹ (average for the 30-year period 1971-2000;
132 World Weather Organization, <http://www.wmo.int>). Monthly temperatures for this period ranged
133 from range 6.7 °C in January to 16.8 °C in August. The Aulne watershed is 8 times larger than
134 the Elorn watershed (1822 versus 280 km²). The Aulne River discharge is thus than four times
135 larger than the Elorn River one. The Aulne and Elorn river discharges decrease from winter (42
136 and 189 m³ s⁻¹ in 2009) to summer (1.0 and 1.7 m³ s⁻¹ in 2009; Banque Hydro,
137 <http://www.hydro.eaufrance.fr>). As for the watershed area, the length of the Elorn Estuary is
138 smaller (~ 15 km) compared to the Aulne Estuary (~ 35 km). The morphology of the Elorn
139 Estuary is straight and more directly exposed to marine hydrodynamic influence, while the Aulne
140 estuary is meandering and more protected by the Bay of Brest. The semi-diurnal tidal amplitude

141 of 4 m (7.5 m during spring tides) in the Bay of Brest results in intense variations of water depth
142 in these shallow estuaries. All these properties, and principally the watershed lithology and the
143 river flow regime, lead to high Si(OH)_4 fluxes to the Bay of Brest. Due to a larger drainage basin
144 area and river flow, the Aulne River Si(OH)_4 fluxes are generally higher than the Elorn River
145 ones. The relative Si(OH)_4 fluxes to the Bay of Brest range from 40% (in summer) to 80% (in
146 winter)¹.

147

148 2.2. Sampling design

149

150 Sampling was performed along the Elorn and Aulne estuaries (Fig. 1) in February, May,
151 July, and October 2009. In order to interpret intra-estuary, inter-estuary and seasonal variations,
152 small spatial and temporal variability were investigated first. All the different scales are
153 summarized in Table 1.

154

155 2.2.1. Small scale sampling

156 *Centimeter scale:* Vertical pore water profiles were analysed at a resolution of 0.5 cm in
157 surface to 4 cm at 20 cm-depth in each sediment core.

158 *Meter scale:* Variability at the meter scale was investigated through the sampling of three
159 cores at each station (E1, E2, E3, A1, A2, A3 and A4) and season (n=3).

¹ Data not shown. Si(OH)_4 flux was calculated by multiplying Si(OH)_4 concentration with the river flow at the
exutory of Elorn and Aulne rivers. Relative Si(OH)_4 fluxes to the Aulne Estuary was calculated as the proportion
of the Aulne Si(OH)_4 flux divided by the sum of Elorn and Aulne Si(OH)_4 fluxes.

160 *Longitudinal transect:* In both Elorn and Aulne estuaries, longitudinal variations were
161 quantified by sampling sediment cores at three stations located at a same tidal level on the left
162 subtidal shores (stations up, E2/A2, dw) in February 2009 (Fig. 1). Distance between the three
163 stations was 100-1000 m depending on stations.

164 *Cross-section:* In both estuaries, sediment cores were sampled in the channel (stations c)
165 in February 2009, and at five stations from the left to the right borders (stations E2/A2, b, c, d and
166 e) in May 2009. Cross-section widths were ~75 and ~200 m in intermediate Elorn and Aulne
167 estuaries, respectively.

168 *Tide:* High-frequency sampling was performed on board the *Hésione* every 2 h over 12 h
169 in July 2009 in the outer Aulne Estuary (station A4; Fig. 1).

170

171 2.2.2. Intra-estuary, inter-estuary and seasonal sampling

172 *Intra- and inter-estuary sampling:* Intra- and inter-estuarine variations were investigated
173 by sampling at three stations located in upper (stations E1 and A1), intermediate (stations E2 and
174 A2) and outer (stations E3 and A3) estuaries (Fig. 1). Sampling was always performed at mid-
175 tide on subtidal sediments located between the channel and the border.

176 *Seasonal sampling:* Sampling at the three stations located along each estuary was
177 performed in February, May, July, and October 2009.

178

179 2.3. Sediment and water processings

180

181 Core sampling was achieved using a gravity corer (UWITEC[®]) with Plexiglas[®] cores (9.5
182 cm diameter x 60 cm long). Corer weight was adjusted to allow a penetration of 30 cm into the
183 sediment. This gravity corer facilitated acquisition of an undisturbed sediment–water interface.
184 Overlying water temperature and salinity were immediately measured with a salinometer after
185 sampling. Sediment cores were immediately sliced every 0.5 cm in the first 2 cm, every 1 cm
186 down to 4 cm, every 2 cm down to 12 cm, and every 4 cm down to 20 cm. Sediment sections
187 were placed in sealed 50-ml centrifugation tubes containing Vectaspin 20 filters (0.45 µm pore
188 size, Whatman[®]) as described in Andrieux-Loyer et al. (2008). Interstitial waters were extracted
189 by centrifugation at 3500 rpm for 10 min (2 times) at 4°C. Overlying and pore waters were
190 acidified to pH = 2 with HCl. An aliquot was preserved at 4 °C for analyses of Si(OH)₄
191 concentrations.

192 An aliquot of non-centrifuged bulk sediment was stored at 4 °C for less than 15 d for
193 granulometry measurements. Centrifuged sediments were freeze-dried for 48 h, placed at 60 °C
194 to ensure complete sediment dryness, and slightly powdered for further analyses of amorphous
195 silica (aSiO₂) concentrations in the solid fraction and dissolution experiments.

196 Surface waters were also collected at 1 m depth along the salinity gradient of the two
197 estuaries. An aliquot of 100-200 ml of surface water was filtered on a polycarbonate filter (0.6
198 µm pore size) and stored at 4 °C until Si(OH)₄ measurements. The membrane was dried for 48h
199 at 60°C and stored at room temperature until aSiO₂ analyses.

200

201 2.4. Laboratory analyses

202

203 Sediment grain size analyses were performed with a laser-based particle size analyser (LS
204 Beckman Coulter).

205 Pelagic aSiO₂ concentrations were determined by using the sequential alkaline digestion
206 method of Ragueneau et al. (2005) and benthic aSiO₂ contents were quantified by using the
207 method of DeMaster (1981). Both methods allowed correcting amorphous silica concentrations
208 from lithogenic silica interference which is essential in environments rich in aluminosilicates -
209 *e.g.* estuaries. Even if results can be relatively different depending on the extraction method,
210 comparisons of different methods have been performed by Rebreau (2009) which showed that
211 the method of DeMaster (1981) was particularly suitable in estuarine sediments in the Scheldt
212 continuum.

213 For pelagic aSiO₂ concentrations, polycarbonate filters were used for two sequential
214 extractions at pH 13.3 for 40 min. aSiO₂ concentrations were calculated with the equation defined
215 by Ragueneau et al. (2005):

216
$$[\text{aSiO}_2]_{\text{corr}} = [\text{Si}]_1 - [\text{Al}]_1 \times ([\text{Si}]_2 / [\text{Al}]_2)$$

217 $[\text{Si}]_1$: Si(OH)₄ concentrations in supernatant after the first digestion

218 $[\text{Al}]_1$: Al concentrations in supernatant after the first digestion

219 $[\text{Si}]_2$: Si(OH)₄ concentrations in supernatant after the second digestion

220 $[\text{Al}]_2$: Al concentrations in supernatant after the second digestion

221 Triplicate measurements of Al concentrations were performed using the manual fluorescent
222 method, by adding lumogallion which forms a fluorescent complex with Al. Excitation and
223 emission wave lengths are 492 nm and 565 nm, respectively.

224 For the quantification of benthic aSiO₂ concentrations, 30 mg of surficial sediments (0-0.5
225 cm) was added to 50 ml of Na₂CO₃ 5% and incubated at 85°C for 6h. At 0.5, 1, 2, 3, 4, 5, 6h, the
226 vials were centrifugated for 15 min at 4000 rpm. 0.5 ml of supernatant was put in 9.2 ml of
227 ultrapure water to prevent problems with the reagent, and neutralized at pH 7 with HCl 10%. The
228 mass pourcentage of aSiO₂ was graphically represented as a function of time. According to the
229 DeMaster (1981) method, the content in aSiO₂ was given by the y-intercept of the linear part of
230 the plot (after 2h). Note that benthic aSiO₂ concentrations are expressed as % in this study to
231 refer to µg gDW⁻¹.

232 Si(OH)₄ concentrations were determined with an AutoAnalyzer III (Bran+Luebbe®) using
233 the method of Tréguer and Le Corre (1975). The precision of the analysis was 0.5%. Note that
234 the overall dataset of Si(OH)₄ concentrations is available in Annexe.

235

236 2.5. Statistical methods

237

238 All statistical analyses described below were performed with R software ([http://cran.r-](http://cran.r-project.org)
239 [project.org](http://cran.r-project.org)). Before each test, the normality and homoscedasticity of data sets were assessed with
240 the Shapiro-Wilk and Bartlett tests, respectively. Parametric ANOVA and non-parametric
241 Kruskal-Wallis tests were performed to examine station and seasonal differences in 10-cm

242 integrated $\text{Si}(\text{OH})_4$ concentrations ($\int \text{Si}(\text{OH})_4$) occurring for longitudinal, cross-section, tidal,
243 and seasonal samplings in the Elorn and Aulne estuaries. Tukey post-hoc tests and multiple
244 comparison tests were performed after ANOVA and Kruskal-Wallis tests, respectively, to
245 identify the significant different groups. The Wilcoxon test was used to find the significant
246 differences in intra- and inter-tidal measurements of $\int \text{Si}(\text{OH})_4$ performed in February and July.
247 For all tests, a probability of 0.05 was used to determine statistical significance.

248 The minimum (min), maximum (max) and average (avg) of pore water $\text{Si}(\text{OH})_4$
249 concentrations, as well as standard deviation (SD) and coefficient of variation (CV, the standard
250 deviation as a percentage of the mean) were summarized for the different scales investigated in
251 Table 2. Clustering (the Ward method) was used to investigate differences between the sum of
252 squared distances integrated over 10 cm at stations E2 and A2 to compare variations in $\text{Si}(\text{OH})_4$
253 profiles at the meter, pluri-meter (longitudinal and cross-section), and seasonal scales.

254 **3. Results**

255

256 *3.1. Benthic variability at small spatial and temporal scales*

257

258 3.1.1. Longitudinal transect

259

260 In the intermediate Elorn Estuary, pore water Si(OH)_4 profiles were similar at all stations
261 sampled at a same tidal level on the left shore (Fig. 3A). Pore water Si(OH)_4 concentrations
262 increased from 80-130 $\mu\text{mol l}^{-1}$ at the surface to 300-330 $\mu\text{mol l}^{-1}$ at 20 cm depth. In the
263 intermediate Aulne Estuary, a very different profile was monitored at station A2 with subsurface
264 Si(OH)_4 concentrations reaching 580 $\mu\text{mol l}^{-1}$ (Fig. 3B). With the exception of this distinct
265 profile, pore water Si(OH)_4 profiles were similar at all stations sampled on the left shore in the
266 Aulne Estuary. In general, pore water Si(OH)_4 concentrations were higher in the Aulne Estuary
267 than in the Elorn Estuary (Table 2, avg) and increased from 110-230 $\mu\text{mol l}^{-1}$ at the surface to
268 380-560 $\mu\text{mol l}^{-1}$ at 20 cm depth. The 10-cm integrated Si(OH)_4 concentrations ($\int \text{Si(OH)}_4$) was
269 not significantly different along longitudinal transects than between triplicates in both
270 intermediate estuaries (Table 3). The coefficient of variation (CV) was however very low in the
271 Elorn Estuary (2-3%), but high in the Aulne Estuary (20-49%), regardless of triplicate or
272 longitudinal sampling in intermediate estuaries (Table 2).

273

274 3.1.2. Cross-section

275

276 In the intermediate Elorn Estuary, pore water Si(OH)_4 profiles were similar across the
277 estuarine section (Fig. 3C). Pore water Si(OH)_4 concentrations regularly increased from 80-100
278 $\mu\text{mol l}^{-1}$ at the surface to 250-400 $\mu\text{mol l}^{-1}$ at 20 cm depth. Slightly higher concentrations were
279 observed at 1-5 cm depth on the right shore (station e) and at 10-20 cm depth on the left shore
280 (station E2). Higher cross-section variations were observed in the intermediate Aulne Estuary
281 (Fig. 3D). As observed in February, pore water Si(OH)_4 concentrations reached more than 800
282 $\mu\text{mol l}^{-1}$ at 3-5 cm depth in May. While pore water Si(OH)_4 concentrations were similar at 0-5 cm
283 depth across the section in the intermediate Aulne Estuary, they varied between 100 and 600
284 $\mu\text{mol l}^{-1}$ down to 5 cm depth. Similar to station E2, the highest concentrations were measured in
285 left and right subtidal shores (400-1000 $\mu\text{mol l}^{-1}$ at stations A2 and e). $\int \text{Si(OH)}_4$ values were not
286 significantly different at the stations of the cross-section and at station E2 in the Elorn Estuary,
287 but were slightly lower at the stations of the cross-section in the Aulne Estuary than at station A2
288 (Table 3). As for the longitudinal transect, the CV was lower in the Elorn Estuary (6-17%) than in
289 the Aulne Estuary (42-59%; Table 2). The CV was slightly higher along the cross-section than on
290 the left subtidal shore in the Elorn Estuary. In the Aulne Estuary, the subtidal shore meter
291 variability at station A2 was even higher than the cross-section variability.

292 Comparisons between channel and subtidal shore Si(OH)_4 concentrations in pore waters at
293 different stations and seasons are shown in Fig. 4. Pore water Si(OH)_4 concentrations at 0-10 cm
294 depth were lower in the channel than in subtidal shores in both estuaries. In surficial sediments,

295 similar pore water Si(OH)_4 concentrations in the channel and the subtidal shore were observed in
296 the intermediate Elorn Estuary in February (Fig. 4A) and Aulne Estuary in May (Fig. 4D). Higher
297 differences in pore water Si(OH)_4 concentrations between the channel and the subtidal shore were
298 observed in the intermediate Aulne Estuary in February (Fig. 4B), in the intermediate Elorn
299 Estuary in May (Fig. 4C), and in the outer Aulne Estuary in July (Fig. 4E). A slight but
300 significant correlation was found between pore water Si(OH)_4 concentrations and the proportion
301 of fine particles ($\% < 63 \mu\text{m}$) in surficial sediments of the whole dataset (Pearson, $r^2 = 0.35$, $p <$
302 0.0001).

303

304 3.1.3. Tidal variability

305

306 Tidal cycles were visible through increasing salinity and water depth and decreasing
307 bottom water temperature and Si(OH)_4 concentrations from low tide to high tide, with the inverse
308 trend until the next low tide (Fig. 5A). In sediments, variations in pore water Si(OH)_4
309 concentrations were also observed over tidal cycles (Fig. 5B). Pore water Si(OH)_4 concentrations
310 at 0-8 cm depth increased from low to high tide and then decreased until low tide. The CV was
311 high (29-32%; Table 2) and $\int \text{Si(OH)}_4$ variations were significant over the tidal cycle (Table 3).

312

313 3.2. Benthic variability at estuarine and seasonal scales

314

315 3.2.1. Intra- and inter-estuary variations

316

317 Overall pore water Si(OH)_4 concentrations were lower in the Elorn Estuary than in the
318 Aulne Estuary (Fig. 6). Deep Si(OH)_4 concentrations (at 20 cm depth) slightly increased from
319 upper to outer Elorn Estuary (Fig. 6; E1, E2, E3). Regardless of the season, pore water Si(OH)_4
320 concentrations were stable between 0 and 20 cm depth at station E1. At station E2, Si(OH)_4
321 profiles were characterized by a depletion at intermediate depths (5-15 cm depth), while at station
322 E3, typical asymptotic Si(OH)_4 profiles were observed. No trend was observed however along the
323 Aulne Estuary (Fig. 6; A1, A2, A3). At stations A1, A2 (in July and October) and A3, Si(OH)_4
324 profiles exponentially increased with depth, except the very different profiles observed in
325 February and May at station A2 (described in Section 3.1).

326 Despite of higher pore water Si(OH)_4 concentrations in the Aulne Estuary, a similar range
327 of pore water Si(OH)_4 concentrations was, however, observed at saline stations E3 and A3 (300-
328 $600 \mu\text{mol l}^{-1}$). $\int \text{Si(OH)}_4$ values varied between stations and seasons and with the interaction of
329 these two factors (ANOVA; Table 4). Tukey post-hoc tests (not shown) indicated significantly
330 lower $\int \text{Si(OH)}_4$ values at stations E1, E2, and E3 than at stations A1 and A2, but no significant
331 differences with station A3. $\int \text{Si(OH)}_4$ was significantly lower at station E2 than at E3 (Kruskal-
332 Wallis and multiple comparison test), but no spatial differences were observed in the Aulne
333 Estuary.

334 Very high pelagic aSiO_2 concentrations were also found in February in upper estuaries
335 (Fig. 7). They ranged between 23-56 $\mu\text{mol l}^{-1}$ and 28-77 $\mu\text{mol l}^{-1}$ at salinity 0-5 in the upper Elorn

336 and Aulne estuaries, respectively.

337

338 3.2.2. Seasonal variations

339

340 The lowest seasonal variations in Si(OH)_4 profiles were observed in the Elorn Estuary
341 (stations E1, E2, and E3) and at station A3 compared to stations A1 and A2 (Fig. 6). Seasonal
342 variations were not significant in the Elorn Estuary (ANOVA; Table 4). The shape of Si(OH)_4
343 profiles however varied at station E2, with an increase of Si(OH)_4 depletion depth from May (5
344 cm depth) to October (15 cm depth). In contrast, high and significant seasonal variations of
345 $\int \text{Si(OH)}_4$ were observed in the Aulne Estuary (Kruskal-Wallis; Table 4). At stations A1 and A2,
346 $\int \text{Si(OH)}_4$ values were significantly lower in October than in February and May (Tukey post-hoc
347 test, not shown).

348

349 3.3. Comparison of benthic variation amplitudes at the different scales

350

351 The Fig. 8 shows the similarity and difference between meter, longitudinal, cross-section
352 and seasonal variations of $\int \text{Si(OH)}_4$ in intermediate Elorn and Aulne estuaries. In the Elorn
353 Estuary, $\int \text{Si(OH)}_4$ measured in February was different from other seasons, but seasonal
354 variability between May, July, and October was low (Fig. 8A). The variability of $\int \text{Si(OH)}_4$ along

355 longitudinal transect in February was lower than seasonal variations. The low seasonal variations
356 did however not allow to distinguish seasonal and cross-section variations. Even if $\int si(OH)_4$
357 was similar from May to October, it is important to note that the shape of the profiles were
358 different depending on seasons (Fig. 6). In the intermediate Aulne Estuary, seasonal variations of
359 $\int si(OH)_4$ were significant (Table 4) but were lower than triplicate, longitudinal and cross-
360 section variability (Fig. 8B). As for the Elorn Estuary, $\int si(OH)_4$ was different in February
361 compared to the other seasons. One season (May) was characterized by a high dispersion of
362 triplicates in the whole cluster. This heterogeneity was also observed in the cross-section transect,
363 indicating high meter and pluri-meter variability in the intermediate Aulne Estuary. In both
364 estuaries, July and October were individualized with always one replicate more distant from the 2
365 others.

366 **4. Discussion**

367

368 *4.1. Factors controlling the variability of pore water Si(OH)₄ concentrations in macrotidal*

369 *estuaries*

370

371 *4.1.1. Analytical precision*

372

373 The coefficient of variation (CV) of pore water Si(OH)₄ concentrations ranges between 1
374 and 59 %, regardless of the scale studied in the Elorn and Aulne estuaries (Table 2). These values
375 are always higher compared to the precision of the analytical method (0.5 %), indicating that
376 variability caused by analytical methods is negligible compared to the field variability. This
377 confirmed the observations made in the Indian River Estuary in Florida, where laboratory
378 variability was 0.3 % while meter benthic heterogeneity reached 40 % (Montgomery et al., 1979).

379

380 *4.1.2. Reaction and transport processes*

381

382 *Benthic properties controlling asymptotic Si(OH)₄ concentrations:* Despite variation at the
383 different scales in the Elorn and Aulne estuaries, similar pore water profiles and asymptotic
384 Si(OH)₄ concentrations (Figs. 3A and 6-station E2) suggest homogeneous benthic properties at
385 meter and longitudinal scales. The main benthic sediment and pore water properties expected to
386 be homogeneous at a small scale (1-100 m) are pH, aluminium and detrital contents, as well as

387 specific sediment surface area, which are known to strongly control deep pore water Si(OH)_4
388 concentrations (Dixit et al., 2001; Dixit and Van Cappellen, 2002). At larger scale (km), the small
389 increase in asymptotic Si(OH)_4 concentrations from the upper to the lower Elorn Estuary is
390 consistent with decreasing aluminium and detrital contents in fresh to marine sediments (Hydes
391 and Liss, 1976; Odum, 1984). The low seasonal variations of asymptotic Si(OH)_4 concentrations
392 observed in the Elorn Estuary are explained by the relative stability of deep sediment properties
393 compared to surficial ones (Berner, 1980). The low intra-estuary variations and the high seasonal
394 variations in the Aulne Estuary suggest however that benthic variability in deep Si(OH)_4
395 concentrations might be less constrained by deep sediment properties than by other processes
396 described below.

397

398 *Hydrodynamic regime:* The lower pore water Si(OH)_4 concentrations observed in the
399 channel compared to subtidal shores, especially in the top 5 cm of the sediment (Fig. 4), was
400 explained by sediment resuspension. Wave and tidal currents indeed lead to intense mixing,
401 which strongly enhances diffusive transport and generates low pore water Si(OH)_4 concentrations
402 in upper estuarine sediments (*e.g.* until 3 to 5 cm depth in the Scheldt Estuary; Vanderborcht et
403 al., 1977; Rebreanu, 2009). The lowest differences in pore water Si(OH)_4 concentrations between
404 the channel and the subtidal shore sediments - and the low CV of $\int \text{Si(OH)}_4$ - in May compared
405 to February, can be explained by the lower river discharge which decreases sediment
406 resuspension, and thus prevents the decrease in pore water Si(OH)_4 concentrations in the channel
407 in May. Resuspension must have important consequences in increasing Si(OH)_4 export to pelagic
408 waters, either through the direct enhancement of benthic fluxes, or through the increase of aSiO_2

409 dissolution by keeping Si(OH)_4 concentrations from equilibrium with aSiO_2 dissolution.
410 Resuspension thus mainly creates spatial heterogeneity in estuarine areas characterized by high
411 tidal currents, and especially in the channel which leads to potential cross-section heterogeneity.

412 The overall positive correlation between Si(OH)_4 concentrations and the proportion of
413 fine particles ($\% < 63 \mu\text{m}$) in surficial sediments ($r^2 = 0.35$, $p < 0.0001$) suggests that lower
414 benthic Si(OH)_4 concentrations - and thus aSiO_2 concentrations - are associated to coarser
415 sediments, which has already been suggested in the Scheldt Estuary (Rebreanu, 2009). These
416 observations reinforce that sediment redistribution and sorting is important in determining the
417 spatial variability of benthic Si contents. Even if aSiO_2 have been deposited with sediment
418 particles - as also observed in intertidal marshes (Struyf et al., 2006) - the low correlation
419 however highlights that other processes than sediment grain size control benthic Si(OH)_4
420 concentrations, *e.g.* aSiO_2 quantity and quality.

421

422 *Biological processes:* In addition to the role of sediment properties and diffusive transport
423 in pore water Si(OH)_4 distributions, biological processes (*e.g.* bioirrigation) play an important
424 role in the modulation of benthic Si(OH)_4 concentrations. The exponential increase in pore water
425 Si(OH)_4 concentrations - commonly observed in the absence of bioirrigation (McManus et al.,
426 1995) - was consistent with the absence of benthic fauna in the intermediate Aulne Estuary
427 (station A2; Emma Michaud, pers. comm.). A typical example of the role of bioirrigation
428 however occurs in the intermediate Elorn Estuary (station E2; Fig. 6). Pore water Si(OH)_4
429 concentrations are characterized by Si(OH)_4 depletion in sediments at this station, where the
430 depletion depth increases seasonally from 5 to 15 cm depth. The shape of pore water Si(OH)_4

431 profiles is typical of sediments where transport processes induced by bioirrigation may overcome
432 the importance of dissolution (Aller, 1980). Regardless the irrigation strategies, these depletions
433 are consistent with the enhancement of diffusive transport of pore waters - which effect is similar
434 to diffusive transport mediated by sediment resuspension - and/or non-local transport of water
435 (Mermillot-Blondin et al., 2005). Modelling the benthic Si cycle at this station (Raimonet, 2011),
436 together with direct identification and measurement of bioirrigation (Michaud et al., in prep) have
437 confirmed the role of benthic fauna in producing these profiles. Bioirrigation often increases
438 sediment oxygenation (Waldbusser et al., 2004), which favours the degradation of benthic
439 organic matter, and in turn, the dissolution of benthic aSiO_2 . By flushing burrows, bioirrigation
440 prevents the enhancement of pore water concentrations induced by dissolution (Boudreau and
441 Marinelli, 1994). Decreasing pore water Si(OH)_4 concentrations - and keeping them away from
442 equilibrium - increases dissolution rates. Bioirrigation thus simultaneously enhances the
443 dissolution of aSiO_2 and decreases pore water Si(OH)_4 concentrations due to the net export of
444 pore waters to bottom waters.

445

446 *Interaction of processes:* The different processes described above simultaneously impact
447 benthic Si(OH)_4 concentrations, as highlighted in our tidal cycle. Even if tidal variations in pore
448 water Si(OH)_4 concentrations are expected to be limited in cohesive sediments - as they are
449 generally dominated by diffusive rather than advective processes (Boudreau, 1997) - changes in
450 Si(OH)_4 concentrations were observed over 8 cm depth (Fig. 5). Resuspension - detailed above –
451 is often limited to the first 3-5 cm (Vanderborgh et al., 1977; Rebreanu, 2009). Tidal variations
452 observed down to 8 cm depth suggest that either stronger currents are present in subtidal shores

453 (which is not expected in subtidal sediments), or that other processes - detailed above -
454 simultaneously happen.

455 Rapid and deep variations in Si(OH)_4 concentrations have already been related to tidal
456 pumping in permeable sandy and coastal sediments, which leads to the vertical transport of pore
457 waters over several centimeters (Shum and Sundby, 1996; Jahnke et al., 2003; Chatelain, 2010).
458 A recent modelling study highlighted that tidal pumping occurs in cohesive sediments, in
459 particular close to the creekbank (Wilson and Morris, 2012). This model was applied to a shallow
460 estuary (< 3 m depth) characterized by a mean tidal amplitude of 1.4 m, confirming that shallow
461 and macrotidal estuaries could be much more impacted by such tidal pumping than expected.
462 Tidal pumping has also been highlighted in macrotidal intertidal mudflats where advective fluxes
463 were 400 times higher than diffusive fluxes when water rose (Leynaert et al., 2011). It is
464 therefore reasonable to attribute the deep variations of pore water Si(OH)_4 concentrations over
465 tidal cycles to tidal pumping.

466 Considering that tidal pumping occurs in these muddy sediments, pore water Si(OH)_4
467 concentrations are expected to be higher at low tide and not at high tide, as we observe in Fig. 5.
468 The presence of burrows built by benthic fauna is hypothesized to enhance the transport of pore
469 waters and/or bottom waters out and/or into benthic sediments (Aller, 1980; Berner, 1980;
470 Stieglitz et al., 2000). Aller (1980) showed indeed that burrows lead to significant decreases in
471 pore water Si(OH)_4 concentrations. Between high and low tides, the flushing of burrows with
472 bottom waters might have led to lower pore water Si(OH)_4 concentrations. Localized lower
473 Si(OH)_4 concentrations observed at 4, 6, or 8 cm depth (Fig. 5) particularly highlight burrows
474 localized at specific depths, as it has been shown in permeable sediments (Meysman et al.,

475 2006). As the Elorn and Aulne estuaries are shallow (< 10 m depth) and characterized by a
476 macrotidal regime, changes in water column height and current speed are very important at tidal
477 scales, strengthening the potential role of tidal pumping and resuspension in bioirrigated
478 sediments of macrotidal estuaries. We hypothesize that these processes significantly control
479 Si(OH)_4 concentrations and benthic fluxes (1) at the tidal scale, in particular in shallow
480 macrotidal estuaries, due to wide tidal amplitude, and (2) at the seasonal scale, due to the
481 seasonal variation of benthic macrofauna (maximal activity and biomass during summer) and
482 temperature (maximal production and dissolution in summer).

483

484 Benthic reaction and transport processes and their controlling factors thus play an
485 important role in controlling pore water Si(OH)_4 concentrations and their variability at different
486 scales. However, the aSiO_2 transported by the river and estuary, as well as the factors that control
487 the area of aSiO_2 deposition and their variability, also play a role in the distribution of pore water
488 Si(OH)_4 concentrations in estuarine sediments.

489

490 *4.1.3. Quantity and quality of aSiO_2 inputs*

491

492 Up to $80 \mu\text{mol l}^{-1}$ of aSiO_2 were measured in fresh waters in the upper Aulne Estuary in
493 February (Fig. 7). These concentrations are similar to those of large rivers like the Amazon River
494 ($74 \mu\text{mol l}^{-1}$) and high compared to most of the rivers ($2.9\text{-}38 \mu\text{mol l}^{-1}$; Conley, 1997; review in
495 Vieillard et al., 2011). This confirms that, although Si has long been thought to flow from rivers

496 mostly in its dissolved form (Tréguer et al., 1995), inputs of Si in the form of particulate
497 amorphous silica (aSiO₂) may be significant (Conley, 1997; Smis et al., 2010). The rapid and
498 concave decrease in pelagic aSiO₂ concentrations along the salinity gradient highlights a transient
499 increase in aSiO₂ concentrations in freshwaters as already demonstrated through a modelling
500 approach (Scheldt Estuary; Regnier et al., 1998), and estuarine deposition of aSiO₂ (Danube
501 Delta; Ragueneau et al., 2002b). The high concentration in pelagic aSiO₂ and its deposition may
502 explain our observations of these subsurface maxima in the pore waters in the meander of the
503 intermediate Aulne Estuary, as detailed below.

504

505 *Control of the hydrodynamic regime on aSiO₂ deposition:* The high aSiO₂ concentrations
506 in the upper Aulne Estuary in February are associated with high winter river discharge, which
507 leads to the downward displacement of fluid muds, associated to the maximal turbidity zone
508 (MTZ; Hermann and Heip, 1999; Meire et al., 2005), close to station A2. The transport of MTZ
509 might lead to the deposition of suspended matter - including aSiO₂ - in meanders located close to
510 the MTZ, in the intermediate estuary (this study). River discharge variability - which can be
511 essential in maintaining biodiversity and stability of the ecosystem (Poff, 2009) - is thus also
512 crucial in determining the spatial and temporal variability of sediment deposition, and thus of
513 benthic Si contents. During high river discharge, the quantity of exported particulate matter and
514 aSiO₂ is higher and the MTZ moves downward in the estuary, which contributes to the settlement
515 of particles from the MTZ in the meander.

516 The presence of this high subsurface maximum in February and May at station A2,
517 suggests that the deposition of aSiO₂ is localized in the point bar, the internal part of the meander

518 (Fig. 6, station A2). The contribution of groundwater (e.g. Wilson and Morris, 2012) was
519 discarded by constant salinity over sediment depth in February and May. By accumulating
520 sediments, point bars are potential recorders of the seasonality of river and estuarine loads, e.g.
521 aSiO₂, organic matter, and other particulate loads, from the watershed, river and estuary. The
522 seasonal evolution of Si(OH)₄ profiles highlights deposition and erosion dynamics in point bars.
523 The deepening of the subsurface maximum indicates a deposition rate of ~ 2 cm between
524 February and May (*i.e.* ~ 0.5-0.6 cm month⁻¹). This rate is particularly high for coastal
525 ecosystems and characteristic of preferential deposition areas (McKee et al., 1983) and consistent
526 with generally favoured deposition in point bars. In July and October, the absence of soft
527 sediments, however, highlights sediment erosion in the point bar between May and July,
528 confirming our hypothesis that transient deposition and erosion events occur at this station.
529 Erosion - which has occurred in late spring (after May), probably after a small storm event in
530 June - indicates that the potential storage of aSiO₂ in intermediate point bars can either be
531 translocated, which may have important implications for the ecological functioning of estuarine
532 ecosystems. The highest spatial heterogeneity at a 1-100 m scale, and even at seasonal scales, is
533 moreover observed in the meander, which highlights localized deposition and strong
534 heterogeneity associated to high dynamic regimes. Studying deposition-erosion dynamics in
535 estuaries, and more especially in meanders, is thus essential to investigate transient retentions in
536 estuaries.

537

538 *Quality of aSiO₂*: Seasonal changes in pore water Si(OH)₄ concentrations in the point bar
539 of an intermediate estuarine meander also indicate the seasonality of the quality of aSiO₂

540 transported into the estuary. The high subsurface Si(OH)_4 concentrations observed at station A2
541 in February and May is most likely due to transient deposition of highly reactive aSiO_2 that
542 dissolved quickly in the subsurface sediment layers. Dissolution rates must be high enough to
543 sustain such subsurface maxima. In the Danube Delta (Ragueneau et al., 2002b; Becquevort et
544 al., 2002) or in the Scheldt Estuary (Roubeix et al., 2008), it has been demonstrated that aSiO_2
545 dissolution is closely coupled to the microbial degradation of organic matter. Bacteria are very
546 efficient at degrading the organic matrix that is associated with diatom frustules or any other form
547 of aSiO_2 , exposing the silica surfaces to surrounding waters and thus increasing the dissolution of
548 aSiO_2 (Bidle and Azam, 1999). Hence, the effect of bacteria, in particular in the MTZ, has been
549 shown to overwhelm the effect of salinity on dissolution in the MTZ of the Scheldt Estuary
550 (Roubeix et al., 2008).

551 Highly reactive aSiO_2 in estuarine sediments in February and May either results from the
552 deposition of (1) a diatom bloom, or (2) allochthonous winter loads. Even if diatoms are often
553 characterized by higher dissolution rates than sedimentary and degraded aSiO_2 (Rickert et al.,
554 2002), the deposition of a diatom bloom grown during the previous year is not a likely
555 explanation. A loss in reactivity and solubility of aSiO_2 is indeed observed over time due to
556 coatings or detritals in sediments, and leads to lower pore water Si(OH)_4 concentrations (Van
557 Cappellen and Qiu, 1997a,b; Rickert et al., 2002). The origin of reactive aSiO_2 observed in
558 surficial sediments in the intermediate Aulne Estuary is rather consistent with high pelagic aSiO_2
559 concentrations in the upper Aulne Estuary in February, providing evidence of allochthonous aSiO_2
560 inputs to the estuary. Loads of aSiO_2 after fall may, at least partly, be associated with the export
561 of terrestrial or tidal marsh detritus, such as dead plants or phytoliths (Smis et al., 2010; Querné,

562 2011). As phytoliths have not already suffered dissolution loops, they are generally characterized
563 by higher solubility (Frayse et al., 2009), and higher dissolution rates than benthic aSiO₂
564 (Querné, 2011).

565 Our results point out that riverine aSiO₂ can be very reactive and lead to high pore water
566 Si(OH)₄ concentrations in local deposition areas. aSiO₂ is even shown to be very reactive during
567 winter, while winter dissolution is often assumed to be lower due to the absence of primary
568 production, less bacteria and low temperature limiting biological activities. This reinforces the
569 idea that a better quantitative and qualitative characterization of riverine and estuarine aSiO₂ is
570 needed to better understand its fate in estuaries. This will help us in understanding better the
571 temporal and spatial variability of pelagic and benthic properties and processes; such information
572 is crucial when discussing, at different scales, temporary or permanent retention of Si along the
573 land ocean continuum.

574

575 *4.2. Implications of this benthic variability for local ecological studies and perspectives for*
576 *regional and global biogeochemical approaches of Si retention*

577

578 This study has different implications from local to global scales, either from the
579 ecological or biogeochemical point of view. At the local scale, sampling at various temporal and
580 spatial scales is not often performed in estuarine studies. It is however essential to study small
581 scale variations as spatial heterogeneities can sometimes prevent to study seasonal trends
582 (Mouret, 2009). The absence of significant differences in pore water Si(OH)₄ concentrations

583 along and across estuarine sections in the linear Elorn Estuary (Table 3) shows that triplicates in
584 subtidal shores are representative of pore water $\text{Si}(\text{OH})_4$ concentrations in a section of 100-1000
585 meter scale in small linear estuaries. The coefficient of variation (CV) of pore water $\text{Si}(\text{OH})_4$
586 concentrations in the Elorn Estuary (CV of triplicates = 2-25%; Table 2) is in the range of values
587 observed in other estuaries (10-40 %; Matisoff et al., 1975; Montgoméry et al., 1979). The
588 smaller variability of pore water $\text{Si}(\text{OH})_4$ concentrations in the Elorn Estuary compared to the
589 Aulne Estuary (1-59%; Table 2) suggests that linear estuaries might have more homogeneous
590 pore water properties, implying that a discrete sampling along the salinity gradient in small linear
591 estuaries is enough to investigate variations along estuarine salinity gradients.

592 The lower variability at meter and plurimeter scales in the linear Elorn Estuary compared
593 to the meandering Aulne Estuary has implications for local or regional modelling studies (*e.g.* on
594 eutrophication), which use available, and often sparse, data to calibrate and/or validate models.
595 When transposing small spatial variability to data uncertainty, this study highlights that data
596 obtained along salinity gradients of linear estuaries are associated to lower uncertainty (2-25 %)
597 than those of meandering estuaries (1-59 %), providing more confidence in model calibration
598 and/or results in the linear systems. These quantifications are useful to estimate uncertainty
599 during upscaling, but also to prevent significant errors incurred by failing to resolve spatial and
600 temporal variation (Swaney and Giordani, 2007; Swaney et al., 2012). Webster et al. (2000) even
601 highlighted that inappropriate temporal and spatial averaging could lead to errors of up to 30%
602 and 100% in estuarine budgets.

603 The high pore water $\text{Si}(\text{OH})_4$ concentrations observed in these estuaries, especially in the
604 upper and intermediate Aulne Estuary, and the transient retention in point bars confirm that

605 continental margins, and more precisely estuaries, constitute significant areas for Si retention.
606 From an ecological point of view, this indicates a potential stock of Si available for recycling and
607 net benthic fluxes of Si(OH)_4 to pelagic waters. A rough estimate of the contribution of benthic
608 diffusive fluxes of Si(OH)_4 (calculated from diffusive gradients at the sediment-water interface
609 and integrated over the whole estuarine area) to the dissolved Si(OH)_4 inputs to pelagic waters
610 varied from < 3 % to > 50 % of the Si(OH)_4 flux (*i.e.* river and benthic flux) in winter and
611 summer, respectively. This confirms that sediments of shallow estuaries are potentially important
612 for pelagic coastal ecosystems, in particular in summer under low river flow conditions. The
613 contribution of benthic recycling to sustain the growth of pelagic diatoms has already been shown
614 in different shallow coastal areas (Yamada and D'Elia, 1984; Struyf et al., 2006), and especially
615 during limiting conditions during summer in the Bay of Brest (Del Amo et al., 1997b; Ragueneau
616 et al., 2002a), where it prevents the development of flagellates (Laruelle et al., 2009).

617 From a regional and global biogeochemical point of view, our study brings new data on
618 the Si cycle, including retention, in two small macrotidal estuaries that belong to the same
619 estuarine type according to most of estuarine typologies (*e.g.* Dürr et al., 2011b). This study is in
620 line with the approach proposed by Ragueneau et al. (2010), *i.e.* developing a typology of land-
621 ocean continuums and studying Si retention in one type of continuum under different climatic
622 zones, or different types of continuum under the same climatic zone, to derive the most important
623 mechanisms driving Si retention. At this stage, such an effort has been undertaken in regional
624 seas by Meybeck et al. (2007), but it needs to be pursued in other systems (*e.g.* fjords, deltas,
625 mangroves, etc; Dürr et al., 2011a) as not enough data exist for the Si cycle to do so (Ragueneau,
626 2004). Our study suggests that the concept of typology may have to be refined. Indeed, even if

627 the Elorn and Aulne estuaries belong to the same macrotidal estuarine type (Dürr et al., 2011), the
628 two estuaries show different variability patterns which have implications in terms of Si retention.
629 The higher variability in the Aulne Estuary at various scales (CV; Table 2) confirms that
630 meanders generate heterogeneity in the benthic Si cycle, as well as transient retention, at a small
631 local scale. Such differences have also been observed at the regional scale where $\text{Si}(\text{OH})_4$ yields
632 can be different for similar lithology and must be regionally calibrated (Jansen et al., 2010). Thus,
633 estuarine sub-types need to be taken into account in typologies, on the basis of new traits. From
634 this study, meandering could be a useful descriptor, considering that more studies on Si cycle in
635 meanders are first needed to precise the impact of transient and/or permanent Si retention
636 occurring in meandering estuaries.

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638

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883 **Figure captions**

884

885 Figure 1: Study area and location of stations sampled along the Elorn and Aulne estuaries.

886 Sampling at stations E1, E2, E3, A1, A2, and A3 was performed in February, May, July, and

887 October 2009. In intermediate estuaries, longitudinal sections (stations dw, E2/A2, and up) and

888 transversal sections (stations E2/A2, b, c, d, and e) were sampled in February and May 2009,

889 respectively. High-frequency sampling was performed over tidal cycles at station A4 (black star)

890 in July 2009.

891 Figure 2: Vertical profiles of pore water Si(OH)_4 concentrations ($\mu\text{mol l}^{-1}$) in the first 20 cm of

892 sediments in February 2009 along a longitudinal transect performed at the same tidal level

893 (stations up, E2/A2, and dw; Fig. 1) in intermediate Elorn and Aulne estuaries.

894 Figure 3: Vertical profiles of pore water Si(OH)_4 concentrations ($\mu\text{mol l}^{-1}$) in the first 20 cm of

895 sediments in May 2009 in a transversal transect (stations E2/A2, b, c, d, and e; Fig. 1) performed

896 in intermediate Elorn and Aulne estuaries.

897 Figure 4: Vertical profiles of pore water Si(OH)_4 concentrations ($\mu\text{mol l}^{-1}$) in the first 20 cm of

898 sediments in subtidal shores (black color) and in the channel (grey color). Sampling was

899 performed in February 2009 at station E2 (A) and station A2 (B), in May 2009 at station E2 (C)

900 and station A2 (D), and in July 2009 at station A4 (E). Note that data were missing at 2-6 cm

901 depth in May in the Elorn Estuary (C), and that concentrations were only monitored in the first 8

902 cm during the tidal cycle (E). SD were not shown for triplicates in this figure but were

903 summarized in Table 2.

904 Figure 5: Salinity S (-), temperature T ($^{\circ}$ C), and Si(OH)_4 concentrations $\text{Si(OH)}_{4\text{bw}}$ ($\mu\text{mol l}^{-1}$) of
905 bottom waters, water depth D (m) (A), and vertical profiles of pore water Si(OH)_4 concentrations
906 ($\mu\text{mol l}^{-1}$) in the first 20 cm of sediments (B) every 2h during 12h in July 2009.

907 Figure 6: Vertical profiles of pore water Si(OH)_4 concentrations ($\mu\text{mol l}^{-1}$) in the first 20 cm of
908 sediments in February, May, July, and October 2009 ($n=3$) in the Elorn and Aulne estuaries
909 (stations E1, E2, E3, A1, A2, and A3). SD were not shown in this figure but were summarized in
910 Table 2.

911 Figure 7: Pelagic aSiO_2 concentrations ($\mu\text{mol l}^{-1}$) and benthic surficial aSiO_2 concentrations (%)
912 in February 2009 in the Elorn Estuary (A), and in the Aulne Estuary (B).

913 Figure 8: Cluster of the sum of squared distances (SSD) between Si(OH)_4 concentrations between
914 0 and 10 cm depth at different spatial and temporal scales in intermediate Elorn Estuary (A), and
915 Aulne Estuary (B). See Fig. 1 and Table 1 for abbreviations.

916 **Tables**

917

918 Table 1: List of the spatial and temporal variations of pore water Si(OH)₄ concentrations
 919 investigated at small and large scales in this study. The associated unit scale and figure numbers
 920 are indicated in the last two columns.

		Unit scale	Name	Figs.
Spatial variations	Small scale	cm	vertical distribution	2, 3, 4, 5, 6
		m	triplicate	5, 6, 7
	Large scale	10-100 m	longitudinal transect	2
		10-100 m	cross-section	3
Temporal variations	Small scale	km	salinity gradient	6
		km	inter-estuary	6, 7
	Large scale	h-d	tide	5
		d-yr	season	6, 7

921

922

923 Table 2: Statistical values of pore water Si(OH)₄ concentrations (mmol m⁻³) and vertically-
 924 integrated pore water Si(OH)₄ concentrations (mmol m⁻²) at the different spatial and temporal
 925 scales in the Elorn and Aulne estuaries. Abbreviations: n=number of samples; min=minimum;
 926 max=maximum; avg=average; SD=standard deviation; CV=coefficient of variation (the standard
 927 deviation as a percentage of the mean).

Estuary	Scale name	Unit	n	min	max	median	avg	SD	CV
Elorn	depth	mmol m ⁻³	144	17	429	192	206	81	40
	triplicate		3x12	14-24	16-27	14-26	15-24	0-5	2-25
	longitudinal section		3	23	24	24	23	1	3
	cross-section	mmol m ⁻²	5	11	17	16	15	2	17
	salinity gradient		3	18	23	20	20	2	11
	tide		-	-	-	-	-	-	-
	season		4	19	21	20	20	1	4
Aulne	depth	mmol m ⁻³	144	37	830	279	300	141	47
	triplicate		3x12	15-41	19-81	17-51	17-51	0-30	1-59
	longitudinal section		3	13	40	38	30	15	49
	cross-section	mmol m ⁻²	5	20	51	24	29	12	42
	salinity gradient		3	27	32	32	30	3	9
	tide		6	19	30	23	24	4	17
	season		4	22	38	31	30	8	26

928

929

930 Table 3: Results of statistical tests for the comparison between the smallest spatial and temporal
 931 scales in the Elorn and Aulne estuaries. The scales compared are indicated in two columns by n°1
 932 and n°2. Abbreviations: df=degree of freedom; K² and V=statistical value for each statistical test;
 933 $p = p$ value.

Estuary	Scale n°1	Scale n°2	Statistical test	df	K ² or V	p
Elorn	triplicate	longitudinal section	Kruskal-Wallis	1	0.05	0.83
		cross-section	Kruskal-Wallis	1	0.20	0.6547
Aulne	triplicate	longitudinal section	Kruskal-Wallis	1	0.05	0.83
		cross-section	Kruskal-Wallis	1	3.75	0.05
		tide	Wilcoxon	-	21	0.03

934

935

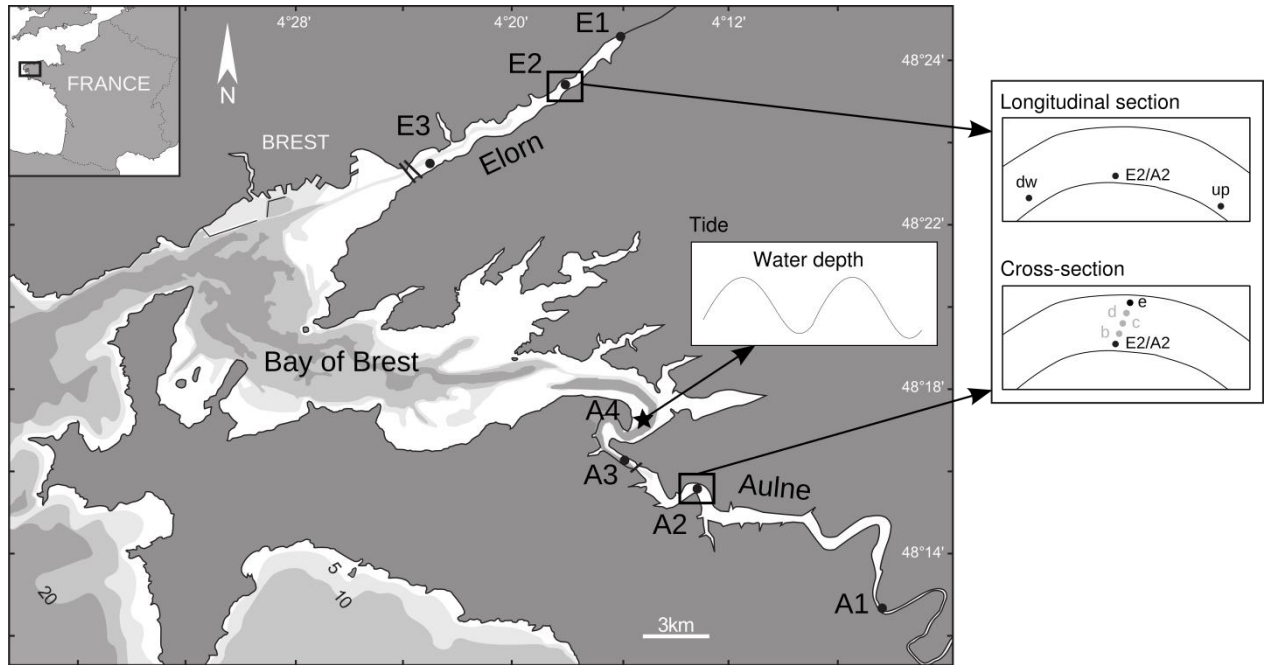
936 Table 4: Results of statistical tests for seasonal variations of $\int si(OH)_4$ along Elorn and Aulne
 937 estuaries. The station and season factors and the combination of these two factors were tested on
 938 the $\int si(OH)_4$ dataset. ANOVA or Kruskal-Wallis test were chosen depending on dataset criteria
 939 described in Section 2.5. Abbreviations: df=degree of freedom; F and K^2 =statistical value for
 940 each statistical test; p = p value.

Factors	Statistical test	df	F or K^2	p
<i>Elorn</i>				
salinity gradient	Kruskal-Wallis	2	7.49	0.02
season	Kruskal-Wallis	3	1	0.8
salinity gradient x season	Kruskal-Wallis	-	-	-
<i>Aulne</i>				
salinity gradient	ANOVA	2	1.04	0.37
season	ANOVA	3	6.44	0.002
salinity gradient x season	ANOVA	6	2.68	0.04

941

942 **Figures**

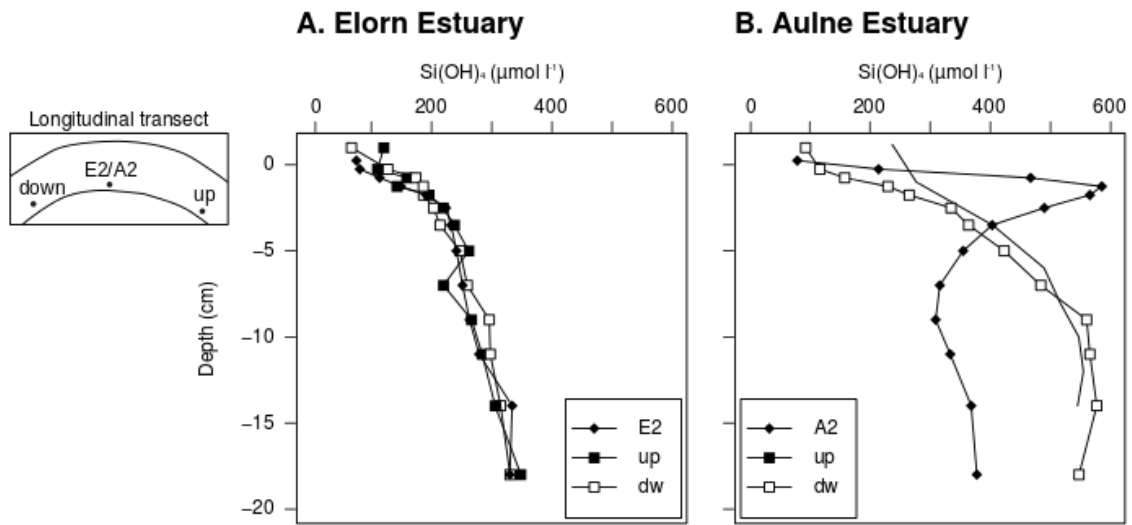
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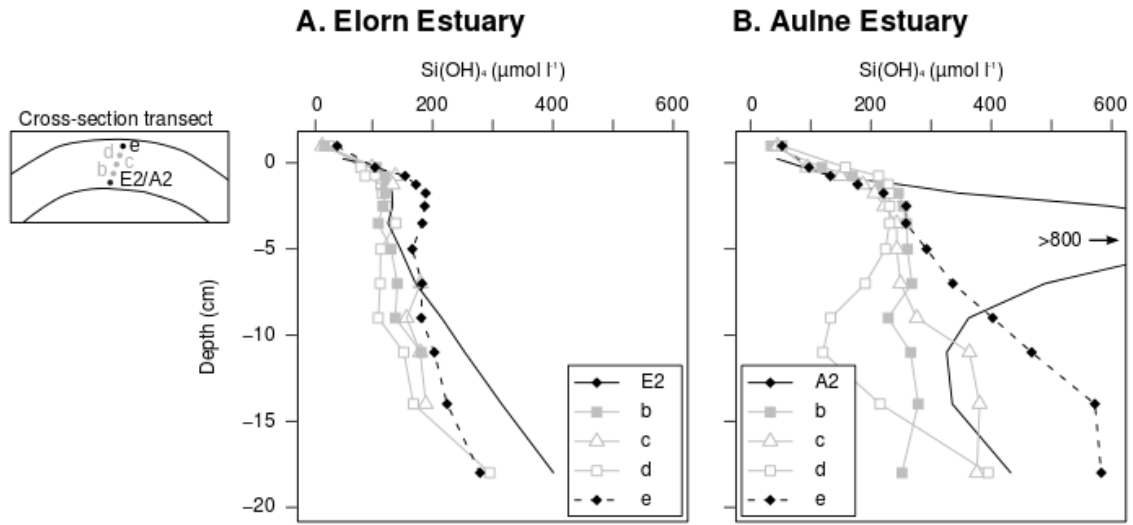
945 **Figure 1**

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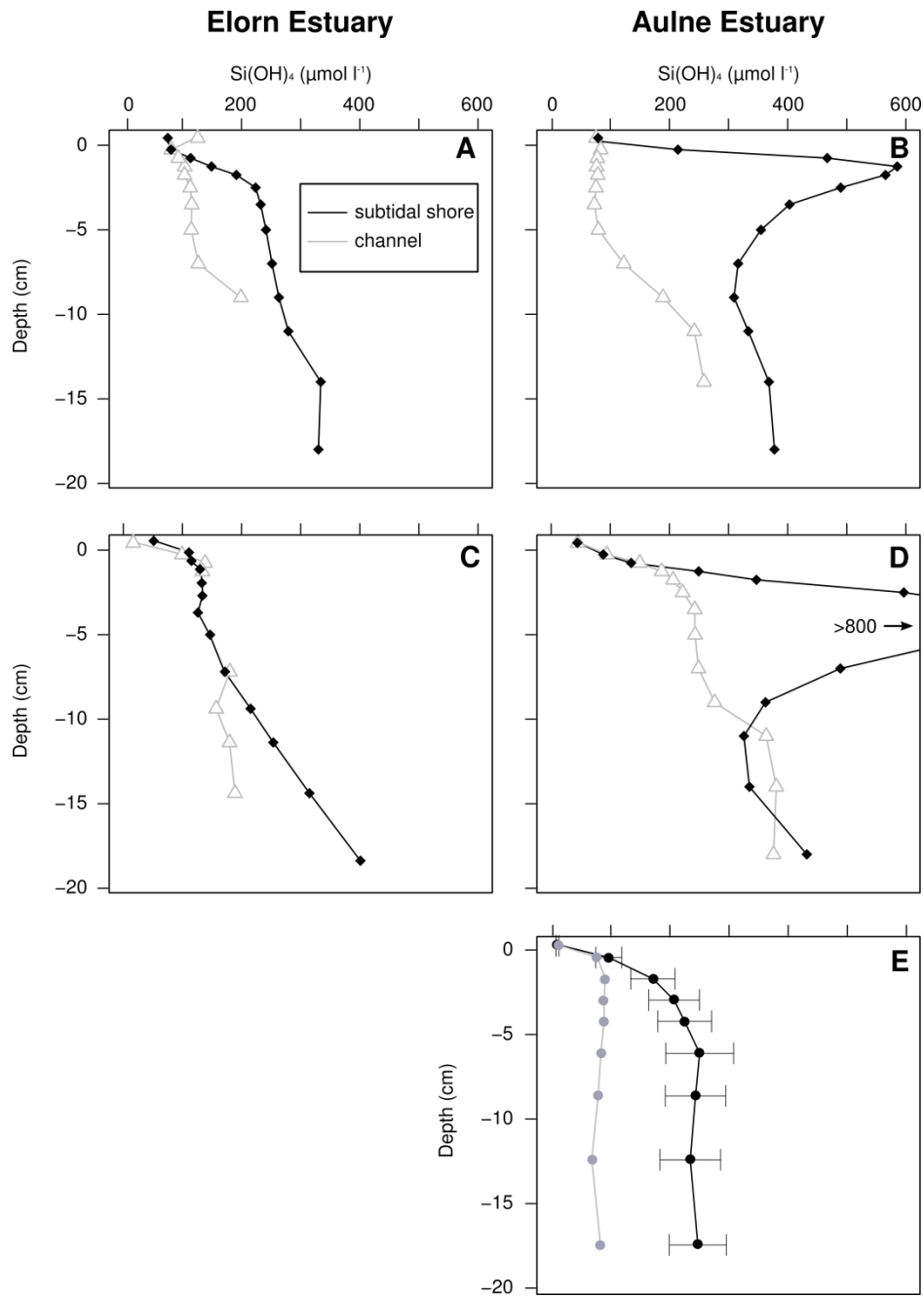
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948 Figure 2



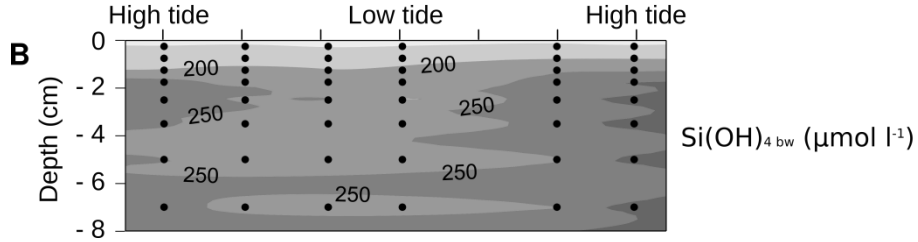
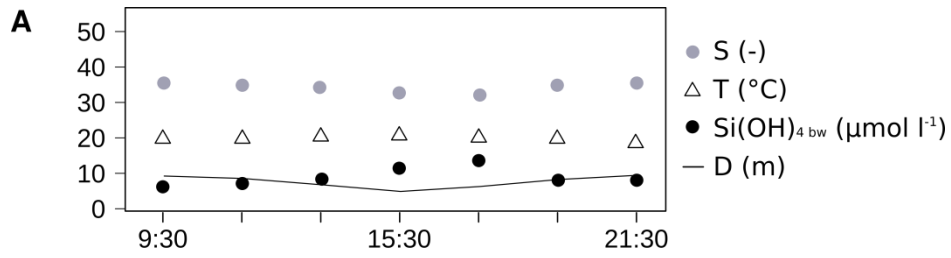
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950 Figure 3



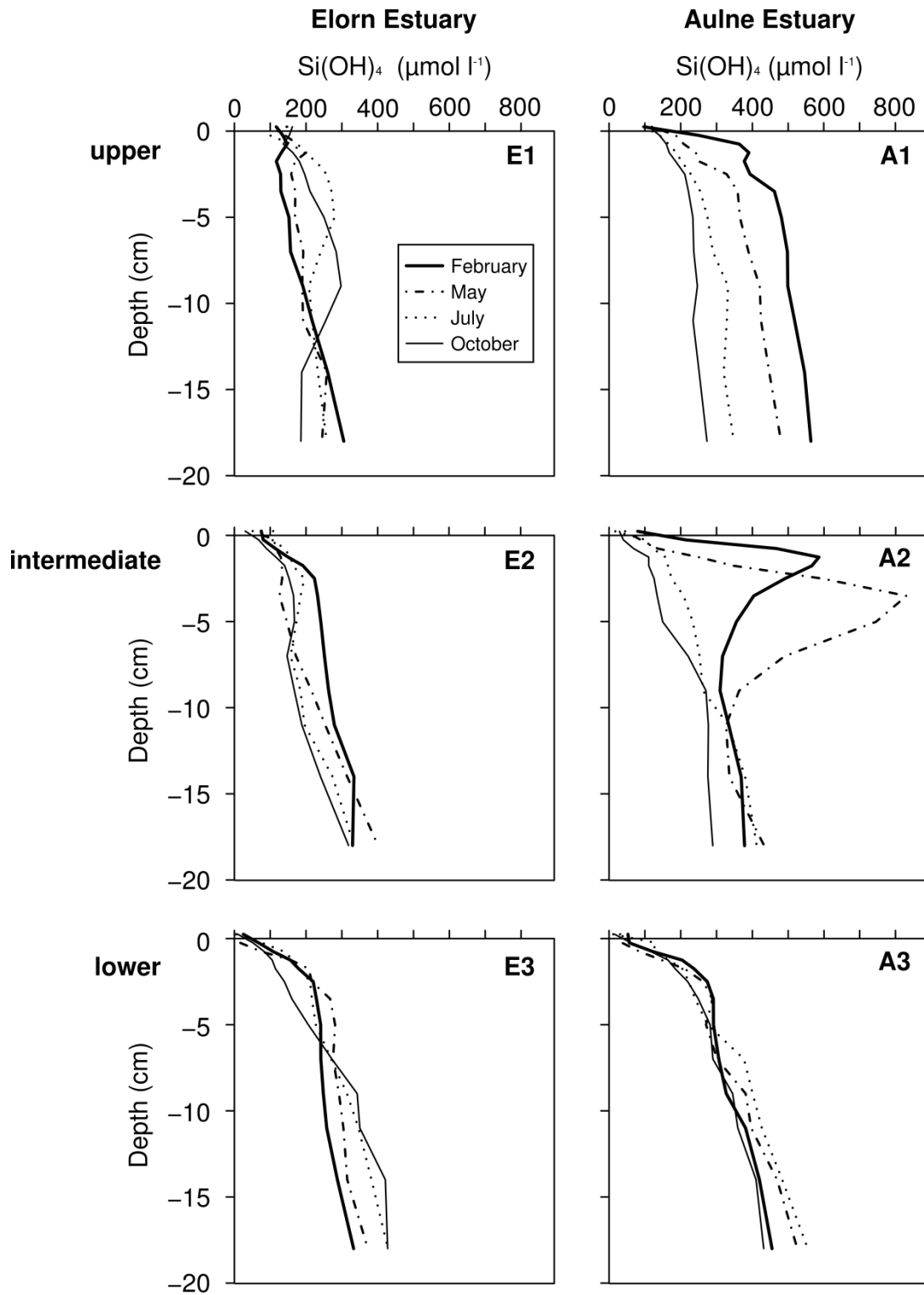
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952 Figure 4



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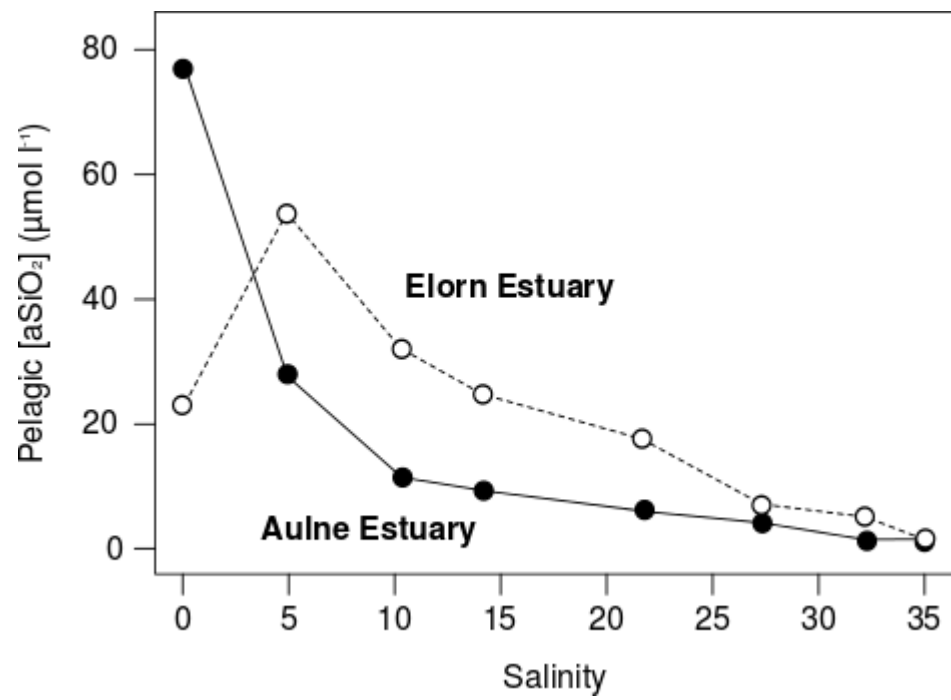
954 Figure 5



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956 Figure 6

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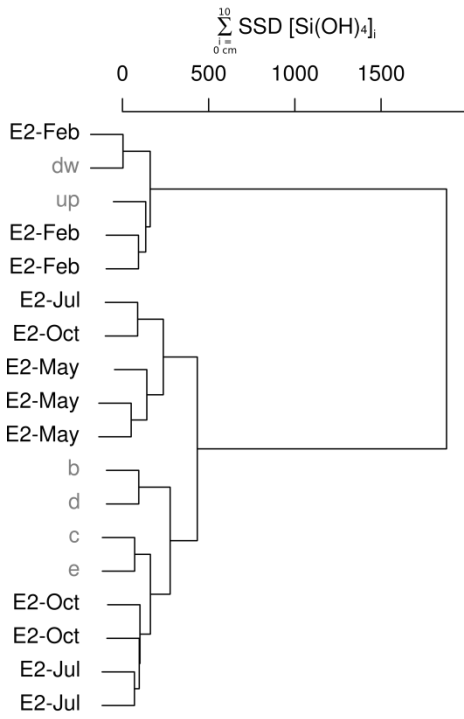


958

959 Figure 7

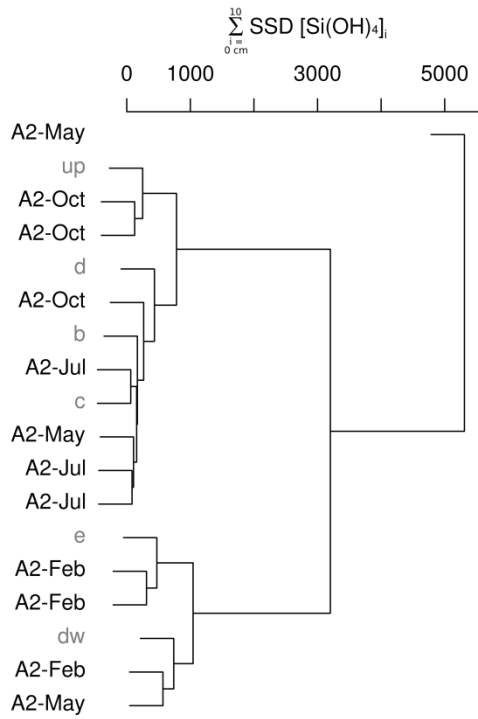
960

A. Intermediate Elorn Estuary



961

B. Intermediate Aulne Estuary



962 Figure 8

963

964 **Annexe:** Dataset of surficial and pore water Si(OH)₄ concentrations (μmol l⁻¹) measured during
 965 lateral and cross-section sampling, tidal sampling and seasonal sampling along the Elorn and
 966 Aulne estuaries.
 967

968

Estuary/Station	Depth (cm)	Si(OH) ₄									
		February					May				
		up	c	dw	b	c	b	c	d	e	
Intermediate Elorn Estuary	0.5	119	124	69	20	17	28	17	28	41	
	-0.25	109	80	88	110	102	84	102	84	107	
	-0.75	158	92	114	124	141	91	141	91	157	
	-1.25	142	103	119	126	137	117	137	117	175	
	-1.75	195	103	127	125	n.d.	119	125	119	192	
	-2.5	219	112	174	120	n.d.	125	120	125	190	
	-3.5	237	115	170	113	n.d.	142	113	142	186	
	-5	261	114	205	134	n.d.	117	134	117	169	
	-7	219	126	203	144	183	116	144	183	185	
	-9	266	198	242	141	160	113	141	160	184	
	-11	283	n.d.	307	184	183	155	184	183	206	
	-14	305	n.d.	352	n.d.	192	171	n.d.	192	227	
	-18	347	n.d.	335	n.d.	n.d.	298	n.d.	n.d.	282	
	Intermediate Aulne Estuary	0.5	227	74	92	34	45	52	45	52	53
-0.25		278	85	118	123	99	163	99	163	102	
-0.75		405	79	159	174	155	218	155	218	138	
-1.25		491	78	231	222	192	234	192	234	183	
-1.75		517	80	266	251	211	235	211	235	226	
-2.5		548	77	336	259	227	236	227	236	264	
-3.5		556	74	365	264	248	236	248	236	263	
-5		546	81	424	266	248	229	248	229	297	
-7		n.d.	124	485	273	254	196	254	196	341	
-9		n.d.	190	561	234	281	138	234	281	407	
-11		n.d.	243	567	271	369	125	271	369	472	
-14		n.d.	260	578	284	386	220	284	386	577	
-18		n.d.	n.d.	549	257	381	400	257	381	588	

Estuary/Station	Depth (cm)	Time	Tide	Si(OH) ₄					
				July			channel		
				#1	#2	#3			
Outer Aulne Estuary (A4)	0.5	09:30	LT	5	5	5	n.d.		
	-0.25			105	92	93	n.d.		
	-0.75			199	163	157	n.d.		
	-1.25			249	166	170	n.d.		
	-1.75			278	199	201	n.d.		
	-2.5			292	n.d.	n.d.	n.d.		
	-3.5			261	n.d.	n.d.	n.d.		
	-5			262	n.d.	n.d.	n.d.		
	-7			289	n.d.	n.d.	n.d.		
	0.5			11:30		5	5	5	n.d.
	-0.25					100	116	126	n.d.
	-0.75					173	189	208	n.d.
	-1.25					205	249	230	n.d.
	-1.75					223	256	258	n.d.
	-2.5	231	n.d.			n.d.	n.d.		
	-3.5	222	n.d.			n.d.	n.d.		
	-5	202	n.d.	n.d.	n.d.				
	-7	215	n.d.	n.d.	n.d.				
	0.5	13:30		7	7	7	n.d.		
	-0.25			98	81	85	n.d.		
	-0.75			169	134	138	n.d.		
	-1.25			244	164	175	n.d.		
	-1.75			260	173	203	n.d.		
	-2.5			270	n.d.	n.d.	n.d.		
	-3.5			243	n.d.	n.d.	n.d.		
	-5	226	n.d.	n.d.	n.d.				
	-7	220	n.d.	n.d.	n.d.				
	0.5	15:30	HT	10	10	10	n.d.		
	-0.25			93	104	103	n.d.		
	-0.75			180	178	177	n.d.		
	-1.25			193	206	219	n.d.		
	-1.75			187	220	245	n.d.		
	-2.5			177	n.d.	n.d.	n.d.		
	-3.5			183	n.d.	n.d.	n.d.		
	-5	189	n.d.	n.d.	n.d.				
	-7	211	n.d.	n.d.	n.d.				
	0.5	19:30		7	7	7	n.d.		
	-0.25			146	106	95	n.d.		
	-0.75			241	186	178	n.d.		
	-1.25			272	252	221	n.d.		
	-1.75			277	272	229	n.d.		
	-2.5			286	n.d.	n.d.	n.d.		
-3.5	279			n.d.	n.d.	n.d.			
-5	234	n.d.	n.d.	n.d.					
-7	242	n.d.	n.d.	n.d.					
0.5	21:30	LT	7	6	6	8			
-0.25			99	87	106	75			
-0.75			215	186	195	89			
-1.25			272	235	221	87			
-1.75			305	271	232	87			
-2.5			315	n.d.	n.d.	81			
-3.5			321	n.d.	n.d.	76			
-5	331	n.d.	n.d.	66					
-7	334	n.d.	n.d.	81					

Estuary	Station	Depth (cm)	Si(OH) ₄													
			February			May			July			October				
			#1	#2	#3	#1	#2	#3	#1	#2	#3	#1	#2	#3		
Elorn Estuary	Upper (E1)	0.5	116	118	117	148	148	147	142	151	144	165	169	152		
		-0.25	124	119	163	167	139	119	133	53	113	104	136	219		
		-0.75	100	115	227	194	181	153	193	125	156	132	130	148		
		-1.25	117	145	n.d.	229	179	192	216	171	198	160	168	188		
		-1.75	116	121	115	152	145	185	262	206	215	182	190	172		
		-2.5	127	134	126	141	185	150	286	206	280	178	205	195		
		-3.5	129	141	119	129	233	149	307	196	310	194	228	212		
		-5	148	166	143	136	211	160	303	191	272	216	228	263		
		-7	157	162	152	136	255	187	234	187	303	237	311	305		
		-9	189	194	191	167	208	192	179	193	249	249	306	338		
		-11	211	253	187	175	n.d.	209	216	205	278	201	286	286		
		-14	256	304	222	275	n.d.	241	179	255	265	162	219	183		
		-18	350	260	n.d.	225	n.d.	264	196	243	334	195	184	178		
		Intermediate (E2)		0.5	77	78	70	55	56	102	105	105	84	74	66	29
				-0.25	77	73	91	108	100	125	116	114	84	74	66	63
				-0.75	113	110	117	103	108	135	140	154	123	95	95	78
				-1.25	171	154	122	124	119	145	161	185	125	118	130	130
				-1.75	219	225	130	141	120	136	175	120	129	131	154	136
-2.5	269			225	177	141	125	134	200	202	184	132	162	163		
-3.5	288			236	173	136	120	123	203	181	170	128	192	176		
-5	268			247	208	159	140	141	199	160	161	159	160	149		
-7	271			277	206	152	156	207	194	135	143	142	180	120		
-9	260			284	245	204	214	227	222	154	177	170	191	141		
-11	265			263	310	251	271	239	212	188	188	190	204	172		
-14	292			354	355	325	316	303	316	255	250	226	257	238		
-18	251			401	338	391	435	377	344	335	323	323	335	298		
Outer (E3)				0.5	24	25	27	5	5	7	6	6	6	9	10	10
				-0.25	64	76	63	17	17	15	79	93	60	46	42	69
				-0.75	104	133	87	60	57	108	130	186	99	69	68	112
				-1.25	141	194	134	130	115	229	167	220	108	92	94	131
				-1.75	144	232	162	159	168	295	188	232	125	108	92	144
		-2.5	166	274	220	200	232	n.d.	205	269	160	135	127	158		
		-3.5	202	252	233	209	236	357	211	264	174	154	130	197		
		-5	224	222	277	254	254	335	269	240	173	154	191	273		
		-7	297	169	258	246	305	n.d.	315	269	227	230	207	381		
		-9	289	211	246	250	343	283	352	304	287	339	281	408		
		-11	286	235	252	240	335	338	378	350	311	333	330	388		
		-14	313	285	265	255	349	341	415	395	338	442	355	468		
		-18	356	342	301	368	385	364	445	437	403	466	331	486		

Estuary	Station	Depth (cm)	Si(OH) ₄													
			February			May			July			October				
			#1	#2	#3	#1	#2	#3	#1	#2	#3	#1	#2	#3		
Aulne Estuary	Upper (A1)	0.5	94	96	101	118	121	120	97	99	98	111	112	111		
		-0.25	214	322	218	163	188	208	146	139	139	141	140	143		
		-0.75	302	454	335	188	194	244	152	172	174	162	171	147		
		-1.25	343	455	374	212	213	302	180	187	197	169	183	152		
		-1.75	355	390	387	242	283	283	171	212	224	193	201	163		
		-2.5	351	366	463	322	263	400	207	224	253	216	220	198		
		-3.5	408	496	481	345	302	429	219	276	269	225	227	213		
		-5	449	506	489	371	309	416	265	240	320	235	240	227		
		-7	454	509	531	376	318	474	322	264	283	243	230	237		
		-9	471	522	503	426	366	470	316	353	327	241	239	260		
		-11	482	544	528	443	377	452	321	378	288	228	223	252		
		-14	492	581	565	443	399	500	297	385	278	256	229	269		
		-18	533	587	571	467	422	550	269	517	258	237	313	270		
		Intermediate (A2)		0.5	77	88	74	41	47	74	17	1750	17	28	30	2770
				-0.25	149	249	245	60	136	69	96	126	87	35	42	43
				-0.75	348	420	633	70	241	95	115	151	98	50	81	73
				-1.25	474	502	780	75	467	207	171	173	128	113	n.d.	108
				-1.75	505	449	743	124	500	418	175	170	148	95	n.d.	126
-2.5	428			363	679	202	801	787	193	175	156	95	151	130		
-3.5	385			314	510	219	683	1587	213	215	195	144	138	120		
-5	381			302	382	207	478	1548	226	235	239	142	142	114		
-7	362			243	344	276	521	670	242	242	273	216	274	171		
-9	269			285	375	225	454	410	n.d.	274	249	278	274	258		
-11	329			312	360	198	429	352	n.d.	340	325	282	266	282		
-14	373			343	389	224	414	368	n.d.	380	379	275	269	283		
-18	368			325	440	n.d.	392	473	n.d.	412	n.d.	300	270	297		
Outer (A3)				0.5	56	50	53	31	33	28	12	12	15	16	16	
				-0.25	56	52	63	25	41	44	75	158	158	78	59	49
				-0.75	76	128	161	69	92	95	110	131	131	131	117	101
				-1.25	144	240	228	123	151	167	172	165	165	168	150	177
				-1.75	183	289	239	176	225	231	197	208	208	182	169	201
		-2.5	244	330	249	224	281	282	227	221	221	196	198	265		
		-3.5	279	338	256	286	295	285	223	249	249	179	236	332		
		-5	284	331	259	291	301	221	230	298	298	180	290	375		
		-7	276	328	315	306	340	260	315	408	408	203	333	334		
		-9	314	384	283	380	426	337	384	409	409	287	424	324		
		-11	348	402	393	426	377	387	430	425	425	347	366	366		
		-14	386	444	429	471	471	465	494	478	478	431	395	406		
		-18	431	477	456	520	531	n.d.	574	545	545	457	414	424		