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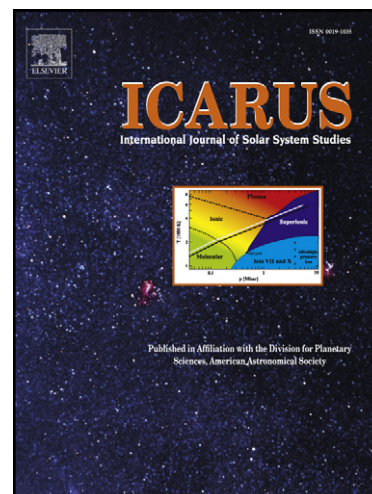
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1 Dark spots and cold jets in the polar regions of Mars: new
2 clues from a thermal model of surface CO₂ ice ☆

3 C. Pilorget^{a,1,*}, F. Forget^a, E. Millour^a, M. Vincendon^b, J.B. Madeleine^a

4 ^a*Laboratoire de Météorologie Dynamique, CNRS/UPMC/IPSL, 4 place Jussieu, BP99, 75252 Paris*
5 *Cedex 05, France*

6 ^b*Department of Geological Sciences, Brown University, Providence, RI 02912, USA*

7 **Abstract**

Observations of the Martian CO₂ ice cap in late winter and spring have revealed exotic phenomena. Unusual dark spots, fans and blotches form as the south-polar seasonal CO₂ ice cap retreats. The formation mechanisms of these features are not clearly understood. Theoretical models suggest that photons could penetrate deep into the CO₂ ice down to the regolith, leading to basal sublimation and gas and dust ejection. We have developed a detailed thermal model able to simulate the temporal evolution of the regolith-CO₂ ice layer-atmosphere column. It takes into account heat conduction, radiative transfer within the ice and the atmosphere, and latent heat exchange when there is a phase transition. We found that a specific algorithm, fully coupling these 3 components, was needed to properly predict ice sublimation below the surface. Our model allows us to determine under what conditions basal sublimation is possible and thus when and where it can occur on Mars. Our results show that basal sublimation is possible if we consider large pathlengths and very little dust content within the ice. Moreover, the model can explain how dark spots can appear very early after the end of the polar night at high latitudes. We also evaluate the importance of the different parameters in our simulations. Contrary to what was suggested by theoretical models, the role of seasonal thermal waves is found to be limited. Solar radiation alone can initiate basal sublimation, which therefore only depends on the CO₂ ice properties. Three main modes were identified: one where condensation/sublimation only occurs at the surface (in the case of small grains and/or high dust content), one where basal sublimation is possible (large pathlengths and very little dust content) and an intermediate mode where sublimation within the ice may occur. We suggest that these different modes could be keys to understanding

many processes that occur at the surface of Mars, like the anticyclonic area behavior or the recent reported activity in gullies.

- 8 *Keywords:* Mars, polar caps, climate, ices
-

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^{*}A proposed running head: Modeling Mars polar cold jets in CO₂ ice

^{*}Corresponding author. Tel.: +33 1 69 85 87 32. E-mail address: Cedric.Pilorget@lmd.jussieu.fr

Email addresses: Cedric.Pilorget@lmd.jussieu.fr (C. Pilorget),
Francois.Forget@lmd.jussieu.fr (F. Forget), Ehouarn.Millour@lmd.jussieu.fr (E. Millour),
Mathieu_Vincendon@brown.edu (M. Vincendon), Jean-Baptiste.Madeleine@lmd.jussieu.fr
(J.B. Madeleine)

¹now at Institut d'Astrophysique Spatiale, Universite de Paris Sud 11, 91405 Orsay, France

9 1. Introduction

10 On Mars, about one third of the atmosphere (composed of 95% CO₂) condenses dur-
11 ing fall and winter to form polar caps. Because of pressure and temperature conditions
12 at the surface, there is no CO₂ liquid phase and solid CO₂ directly condenses and sub-
13 limates. During spring, the insolation increases and the seasonal polar cap disappears
14 gradually (except in the southern very high latitudes where a small perennial polar cap
15 remains). The Thermal Emission Spectrometer (TES) onboard the Mars Global Sur-
16 veyor (MGS) spacecraft has characterized the recession date as defined by the “crocus
17 date” and the thermal and visual properties of the seasonal polar caps (Kieffer *et al.*
18 2000; Kieffer and Titus 2001). The Observatoire pour la Mineralogie, l’Eau, les Glaces,
19 et l’Activite (OMEGA) onboard Mars Express has acquired repeated spectral data of
20 the polar ices and dust in the visible and near infrared and mapped their distribution
21 (Bibring *et al.* 2004; Langevin *et al.* 2006; Langevin *et al.* 2007). The MGS Mars Or-
22 biter Camera (MOC) and Mars Reconnaissance Orbiter (MRO) High Resolution Imaging
23 Science Experiment (HiRISE) have returned high-resolution images of the polar regions
24 revealing numerous exotic features. During seasonal polar cap recession, dark spots and
25 fans appear on the surface. Many of the features look like exposures of subfrost soil,
26 and have been interpreted as such (Malin and Edgett 2000; Supulver *et al.* 2001), but
27 the Thermal Emission Imaging System (THEMIS) observations showed that they remain
28 near the temperatures expected for solid CO₂ for days to months, and therefore must be
29 in intimate contact with, or composed of, solid CO₂ (Kieffer *et al.* 2006). A proposed
30 explanation is that solar radiation penetrates into the CO₂ slab, heats the regolith and
31 sublimates the base of the CO₂ slab, which then levitates above the ground by gas pres-
32 sure. Gas tries to escape, and when a path to the surface is created, CO₂ gas and dust
33 are ejected and form these dark spots (Piqueux *et al.* 2003; Kieffer 2007; Portyankina
34 *et al.* 2010). This process results from the characteristics and behavior of solid CO₂ on
35 Mars, which are unlike those for any natural ice on Earth. In this paper, we present a
36 model that helps us to investigate these features and some of their possible formation
37 mechanisms. Our main objectives are:

- 38 (1) To study the different sublimation processes that can occur during spring.

- 39 (2) To test the “cold jets” theory and see under which conditions the CO₂ slab can
40 sublimate at its base.
- 41 (3) To try to understand why dark spots appear in some locations and not everywhere
42 at the same time.
- 43 (4) To study the earliest seasonal dark spots and see how much time they need to form,
44 and under which conditions. A case study was made of the Manhattan Island area
45 at 85° S.
- 46 (5) To test and improve the scenarios of this “dark spots cycle” with our results.

47 **2. Model description**

48 *2.1. General description*

49 To simulate the detailed seasonal evolution of a layer of CO₂ ice at a given location
50 on Mars, we have developed a 1D model able to compute the temporal evolution of a
51 column composed of an underlying regolith, a CO₂ ice layer, and the atmosphere above.
52 It is an extension of a 1D version of the General Circulation Model from Laboratoire
53 de Meteorologie Dynamique (Forget *et al.* 1999). The model is fully described in the
54 following sections.

55 In practice, the atmospheric model allows one to simulate a radiative convective equi-
56 librium atmospheric profile and is used to compute the incident radiation on the surface
57 and the sensible heat exchange between the surface and the atmosphere (see below). Be-
58 low the surface, we have developed a model that solves heat exchange, radiative transfer
59 and mass evolution equations in the CO₂ slab as well as in the regolith. We used a finite
60 volume approach to solve the equations, with a timestep of 0.001 Martian day (89 s).
61 The model takes into account the solar flux (0.1-5 μm), the incident thermal flux, the
62 thermal emission, the sensible heat flux, the geothermal heat flux, and the latent heat
63 flux when there is a phase transition (Fig. 1).

64 [Figure 1 about here.]

65 *2.2. Vertical grid*

66 The vertical computational grid used in our model is a dynamic layer grid where
67 individual layers have distinct thermal and optical properties. Their total number is

68 fixed; therefore, the number of regolith layers is equal to the difference between the total
 69 number of layers and the number of CO₂ ice layers which evolves with time as CO₂
 70 condenses and sublimates. When a layer sublimates totally, this layer disappears and
 71 another layer of regolith is added at the bottom of the domain at the temperature of
 72 the above layer. In the case of condensation, if the first layer is full of CO₂ ice, another
 73 layer of CO₂ ice is added above it, and the last layer of regolith is removed. Tests
 74 were performed to check the energy balance and showed that it has no effect on the
 75 results. The layer thickness is set to 1 cm. Simulations with thinner layers showed no
 76 significant change in the results. Since measured thermal inertias on Mars range from
 77 30 to 2000 J s^{-1/2} m⁻² K⁻¹, the grid used had to be such that it captures diurnal and
 78 annual thermal waves in such media. For the regolith part, the skin depth corresponding
 79 to oscillation period P would be for an homogeneous (infinite) soil (volumetric specific
 80 heat C and conductivity λ):

$$\delta_P = \frac{I}{C} \left(\frac{P}{\pi} \right)^{1/2} \quad (1)$$

81 If we want to take into account the seasonal thermal waves, we thus have to consider a
 82 minimum depth of 0.13 m for I=30 and 8.69 m for I=2000 (C=10⁶ J K⁻¹ m⁻³). In our
 83 simulations, we thus use a 10 m thick surface composed of a thousand 1 cm layers.

84 2.3. Incident solar visible radiation

85 2.3.1. Atmospheric dust scattering

86 Visible and near infra-red radiative effects of dust are taken into account by computing
 87 the radiative transfer in two different bands: 0.1-0.5 and 0.5-5 μm with a Toon *et al.*
 88 (1989) radiative transfer code. The Delta-Eddington approximation is applied in order
 89 to better simulate the forward scattering peak of dust particles. We use the Ockert-Bell
 90 *et al.* (1997) dust properties in our model. However, other dust properties could be used
 91 if needed (e.g. Wolff *et al.* 2009).

92 [Figure 2 about here.]

93 Atmospheric dust scattering plays a major role when the sun is low on the horizon,
 94 as is common in polar regions. In the case of the Manhattan Island area (85° S), the
 95 solar zenith angle is never less than about 60°. More interestingly, Fig. 2 shows that the

96 solar zenith angle is never less than about 85° when the first dark spots appear around
 97 Ls 175° (Kieffer *et al.* 2006; Aharonson *et al.* 2004).

98 [Figure 3 about here.]

99 Simulations with a Monte-Carlo code using a spherical geometry (Vincendon and Langevin
 100 2010) show that at that time, for an atmospheric dust optical depth set to 0.1 (Vincendon
 101 *et al.* 2008) and dust confined to the first 6 km of the cold polar atmosphere, at least
 102 one half of the solar flux that reaches the surface is scattered by atmospheric dust. The
 103 incidence angles of these photons can be very different from the incidence angle of the
 104 direct solar beam. This is important since in the slab model, for instance, the Fresnel
 105 reflection coefficient quickly rises from 0.5 to 1 for solar angles higher than 85° (Fig.
 106 3). Simulations performed with the Monte-Carlo code give us the distribution of these
 107 incidence angles for various solar angles. To simplify, we computed an effective incidence
 108 angle for these photons by calculating an effective albedo due to specular reflection and
 109 then derived the corresponding angle

$$R_{\text{eff}} = \frac{\int_0^{\frac{\pi}{2}} f(i)R(i)di}{\int_0^{\frac{\pi}{2}} f(i)di} \quad (2)$$

110 with R_{eff} the effective albedo, $f(i)$ the fraction of photons reaching the surface with an
 111 incidence angle i and $R(i)$ the Fresnel reflection coefficient for an incidence angle i . In
 112 our model, we calculate the penetration of photons from the direct flux with an incidence
 113 angle equal to the solar angle, and the penetration of photons from the scattered flux
 114 with this computed effective incidence angle. The incident solar flux can be nonzero
 115 when the sun is below the skyline. For an atmospheric dust optical depth set to 0.1
 116 and dust confined to the first 6 km of the atmosphere, solar flux cannot be neglected at
 117 incidence angles up to 95° (at 91° , about 1 percent of the incoming solar flux reaches
 118 the surface and at 95° it is less than 0.1 percent).

119 We tuned our model to fit with these simulation results when the solar zenith angle
 120 is greater than 90° . The cosine of the solar zenith angle is also corrected to take into
 121 account the spherical geometry and the refraction by the atmosphere (Forget *et al.* 1999):

$$\mu_0 = \frac{\sqrt{1224 \mu_0^2 + 1}}{35} \quad (3)$$

122 [Figure 4 about here.]

123 *2.3.2. Effect of slope*

124 We use the model developed by Spiga and Forget (2008) to compute the influence
125 of the slope and its orientation on our results. This allows us to compute the direct
126 incoming flux D , the scattered flux by the atmosphere S and the reflected flux from the
127 neighboring terrains R .

$$F_0 = D + S + R \quad (4)$$

128 We introduce here σ_s which is the sky-view factor that quantifies the proportion of the
129 sky in the half-hemisphere “seen” by the slope that is not obstructed by the surrounding
130 terrains (assumed to be flat).

$$\sigma_s = \frac{1 + \cos(\theta)}{2} \quad (5)$$

131 where θ is the inclination of the slope. The thermal incident flux is computed assuming
132 that the atmospheric thermal radiation is isotropic

$$F_{th} = \sigma_s F_{th0} + (1 - \sigma_s)E \quad (6)$$

133 with F_{th0} the incident thermal flux that would reach a flat terrain, E the thermal flux
134 emitted by the surrounding terrains.

135 *2.4. Solar flux penetration*

136 Two models have been used to test the different scenarios: a CO_2 “snow” model
137 assuming that CO_2 is composed of scattering and absorbent grains and a CO_2 “slab
138 ice” model assuming that CO_2 is an homogeneous absorbent medium. This latter model
139 can be used to test different scenarios where the photon pathlengths are in the range of
140 several tens of centimeters, as revealed in several areas by TES (Kieffer *et al.* 2000) and
141 OMEGA (Langevin *et al.* 2007) observations. We ran the simulations with a CO_2 ice
142 density of 1606 kg m^{-3} (which corresponds to a CO_2 ice without porosity).

143 *2.4.1. CO_2 snow model*

144 In this model, the grain size is specified, and then remains constant during the sim-
145 ulation rather than evolving with time. The mass fraction of dust grains in the CO_2 ice
146 can be specified. However, to simplify the computation, dust scattering was treated as
147 if dust particles were surrounded by vacuum. As for the atmosphere, we use the Ockert-
148 Bell *et al.* (1997) optical parameters. Based on Langevin *et al.* (2007) we assume that

149 H₂O ice grain content is negligible in the areas where the Kieffer (2007) scenario is likely
 150 to occur. CO₂ ice optical properties are taken from Quirico and Schmitt (1997) and
 151 Schmitt *et al.* (1998). Radiative transfer within the CO₂ ice is calculated with a Toon
 152 *et al.* (1989) radiative transfer code as in Williams *et al.* (2008). Our model uses Mie
 153 theory for single scattering by individual snow grains, together with the Delta-Eddington
 154 method for multiple scattering in a snowpack as in Warren *et al.* (1990), Hansen (1999)
 155 and Kieffer *et al.* (2000). To compute the optical depth of CO₂ and dust at a given
 156 wavelength, we assume

$$\tau_{\text{CO}_2} = \frac{3 M_{\text{CO}_2} Q_{\text{ext}}^{\text{CO}_2}}{4 \rho_{\text{CO}_2} r_{\text{eff}}^{\text{CO}_2}} \quad (7)$$

$$\tau_{\text{dust}} = \frac{3 M_{\text{CO}_2} q_{\text{dust}} Q_{\text{ext}}^{\text{dust}}}{4 \rho_{\text{dust}} r_{\text{eff}}^{\text{dust}}} \quad (8)$$

158 where M_{CO_2} is the mass of CO₂ ice, q_{dust} the mass fraction of dust, Q_{ext} the extinction
 159 coefficient, ρ the density, and r_{eff} the effective radius. In our simulations we assume
 160 $r_{\text{eff}}^{\text{dust}} = 1.5 \mu\text{m}$, $\rho_{\text{dust}} = 2000 \text{ kg m}^{-3}$ and $\rho_{\text{CO}_2} = 1606 \text{ kg m}^{-3}$. Finally the radiative
 161 transfer code returns the upward and downward fluxes at each level, which leads to

$$E_{\text{pen}}(z_2 - z_1) = F_{\text{up}}(z_2) - F_{\text{up}}(z_1) + F_{\text{down}}(z_1) - F_{\text{down}}(z_2) \quad (9)$$

162 with $E_{\text{pen}}(z_2 - z_1)$ the radiative energy deposited between z_1 and z_2 in W m^{-2} , $F_{\text{up}}(z)$
 163 the upward flux at depth z , $F_{\text{down}}(z)$ the downward flux the depth z .

164 2.4.2. Translucent slab model

165 This model is an extreme case of the previous model, in which the CO₂ slab is an
 166 homogeneous absorbent medium. As photons reach the surface, some are reflected (Fres-
 167 nel reflection) and the others penetrate into the ice. The latter are absorbed following a
 168 Beer-Lambert law. However, the way solar radiation penetrates and propagates in the
 169 ice depends not only on the properties of the medium, but also on the wavelength. To
 170 simplify the calculation we compute an effective CO₂ real optical index (Warren 1986;
 171 Quirico and Schmitt 1997; Schmitt *et al.* 1998; Hansen 2005) and Beer-Lambert coef-
 172 ficient so that the reflection only depends on the incidence angle and the absorption of
 173 the solar radiation in the ice only depends on the distance covered by the photons. The
 174 penetration of solar energy was calculated as

$$\int_0^{\infty} S_{\lambda} (1 - R_{\lambda}) e^{-p/l_{\lambda}} d\lambda \quad (10)$$

175 where S_λ is the incident solar flux at the surface at a certain wavelength, R_λ the Fresnel
 176 reflection coefficient, p the path length from the surface along the refracted path and
 177 $l_\lambda = \lambda / 4\pi n_i$ the absorption length. With these results, we computed an effective absorp-
 178 tion coefficient k that does not depend on the wavelength and use it in our model. The
 179 solar radiation reaches the regolith whose albedo is fixed (mean value in the visible). The
 180 reflected fraction is partly absorbed on its way back and the rest escapes from the slab.
 181 In theory, as it reaches the surface, some of the flux is reflected into the slab. However,
 182 when the regolith albedo is low (less than 0.3 for instance) this effect is small. For this
 183 reason, we do not take into account multiple reflections in the slab. Dust can also be
 184 added in the slab (see Fig. 6) . We have for the Fresnel reflection

$$R = \frac{r_s^2 + r_p^2}{2} \quad (11)$$

185 where r_s is the Fresnel coefficient if the light is polarized with the electric field perpen-
 186 dicular to the incident plane and r_p the Fresnel coefficient if the light is polarized with
 187 the electric field parallel to the incident plane.

$$r_s = \frac{n_{\text{atm}} \cos(\theta_{\text{inc}}) - n_{\text{CO}_2} \cos(\theta_{\text{ref}})}{n_{\text{atm}} \cos(\theta_{\text{inc}}) + n_{\text{CO}_2} \cos(\theta_{\text{ref}})} \quad (12)$$

$$r_p = \frac{n_{\text{atm}} \cos(\theta_{\text{ref}}) - n_{\text{CO}_2} \cos(\theta_{\text{inc}})}{n_{\text{atm}} \cos(\theta_{\text{ref}}) + n_{\text{CO}_2} \cos(\theta_{\text{inc}})} \quad (13)$$

189 with $n_{\text{atm}} = 1$ and $n_{\text{CO}_2} = 1.4$ (mean value in the visible (Warren 1986; Quirico and
 190 Schmitt 1997; Schmitt *et al.* 1998; Hansen 2005). θ_{inc} and θ_{ref} are respectively the
 191 incident angle and the refracted angle, which are linked by the relation

$$n_{\text{atm}} \sin(\theta_{\text{inc}}) = n_{\text{CO}_2} \sin(\theta_{\text{ref}}) \quad (14)$$

192 For an incident flux, F_0

$$F_{\text{pen}}(1) = F_0(1 - R) \quad (15)$$

193 where R is the reflection coefficient for unpolarized light, $F_{\text{pen}}(1)$ the flux that penetrates
 194 into the first layer in W m^{-2} and F_0 the incident flux in W m^{-2} . For the following layers,
 195 with z_1 and z_2 , two different depths under the surface, and $z_1 < z_2$:

$$F_{\text{pen}}(z_2) = F_{\text{pen}}(z_1) e^{\left(\frac{-k(z_2 - z_1)}{\cos(\theta_{\text{ref}})}\right)} \quad (16)$$

196 with $F_{\text{pen}}(z)$ the flux that penetrates into a surface at a depth z in W m^{-2} , and k
 197 a constant depending on the properties of the material between z_1 and z_2 . Kieffer
 198 (2007) also computes the flux that reaches the regolith with respect to the depth and the
 199 incidence angle. Our results are consistent with theirs.

200 After solar radiation reaches the regolith, one fraction is reflected and the other penetrates
 201 into the regolith where it is absorbed in the first millimeters. We set the reflected angle
 202 to 45° , the average angle for a lambertian reflection (see e.g. Vincendon *et al.* (2007),
 203 Fig. 1). The reflected fraction is then partly absorbed into the ice before reaching the
 204 surface. We thus have at the interface between CO_2 ice and regolith

$$F_{\text{penref}}(z_{\text{interface}}) = F_{\text{pen}}(z_{\text{interface}})A_{\text{grd}} \quad (17)$$

205 and

$$F_{\text{penref}}(z_1) = F_{\text{penref}}(z_2)e^{\left(\frac{-k(z_2-z_1)}{\cos(\theta_{\text{ref}})}\right)} \quad (18)$$

206 with $z_{\text{interface}}$ the depth of the interface between CO_2 ice and the regolith in m, A_{grd} the
 207 regolith albedo, $F_{\text{penref}}(z)$ the reflected flux that penetrates into a surface at a depth z
 208 in W m^{-2} .

209

210 At the end we have:

$$E_{\text{pen}}(z_2 - z_1) = (F_{\text{pen}}(z_1) - F_{\text{pen}}(z_2)) + (F_{\text{penref}}(z_1) - F_{\text{penref}}(z_2)) \quad (19)$$

211 with $E_{\text{pen}}(z_2 - z_1)$ the radiative energy deposited between z_1 and z_2 in W m^{-2} .

212 2.4.3. Synthesis

213 Figure 5 shows the reflected and absorbed fluxes in the two models. When we increase
 214 the grain size in the CO_2 snow model, the albedo decreases. More energy is absorbed
 215 by the ice plus regolith. When the size is over a few centimeters, the solar radiation can
 216 reach the regolith and the part of the flux that is absorbed in the ice decreases. The slab
 217 ice model appears to be a good way to simulate pathlengths in the range of several tens
 218 of centimeters.

219

[Figure 5 about here.]

220 For both models, we compute the energy that is absorbed in each layer and with that
 221 result, we calculate the temperature increase between z_1 and z_2

$$\frac{\partial T}{\partial t} = \frac{E_{\text{pen}}}{\rho C_p (z_2 - z_1)} \quad (20)$$

222 with E_{pen} the radiative energy deposited between z_1 and z_2 in W m^{-2} , C_p the heat
 223 capacity of the layer in $\text{J K}^{-1} \text{kg}^{-1}$, ρ the density, $z_2 - z_1$ the thickness of the layer in m

224 [Figure 6 about here.]

225 However, the pressure is not constant within the slab, which is a critical point with
 226 regards to the phase transition (see section 2.9).

227 2.5. Thermal infrared radiation absorption and emission

228 2.5.1. Downward atmospheric thermal flux

229 Gaseous CO_2 and airborne dust emit a thermal flux that reaches the surface. This
 230 thermal flux cannot penetrate more than a few millimeters into the CO_2 ice (Kieffer
 231 2007). In our model, it is treated as surface absorption.

232 2.5.2. Thermal emission

233 Thermal emission is equal to $\epsilon \sigma T_{\text{surf}}^4$ with σ the Stephan-Boltzmann constant ($\sigma = 5.67$
 234 $10^{-8} \text{ W m}^{-2} \text{ K}^{-4}$), T_{surf} the surface temperature and ϵ the spectrally averaged emissivity.
 235 We set $\epsilon = 0.9$ in the case of CO_2 ice covering the soil and $\epsilon = 0.95$ when there is no
 236 CO_2 ice (Forget *et al.* 1998). Surface temperature T_{surf} is determined assuming that
 237 CO_2 frost at the surface is in vapor pressure equilibrium with atmospheric CO_2 gas at
 238 surface pressure P_{surf} assuming a CO_2 mixing ratio of 95%.

239 2.6. Sensible heat flux

240 The sensible exchange of heat between the surface and the atmosphere is calculated as
 241 the product between the vertical gradient of temperature (estimated between the surface
 242 value and that in the first atmospheric layer), the surface drag coefficient (computed
 243 assuming a roughness length of 1 cm), and the wind velocity in the lowest level. This
 244 velocity is estimated using the boundary layer scheme described in Forget *et al.* (1999),
 245 assuming that the background wind velocity in the free atmosphere is 10 m s^{-1} . In

246 practice the sensible heat flux is very small because of the low density of the Martian
247 atmosphere and the fact that the horizontal advection of heat is neglected.

248 2.7. Geothermal flux

249 Geothermal flux is set to a constant upward heat flux of 0.030 W m^{-2} , though esti-
250 mates vary by a few mW m^{-2} (Schubert *et al.* 1992). The order of magnitude of this
251 flux is generally negligible in the cumulative frost balance when compared with the other
252 terms. Nevertheless, for facets in the shadow, it could accelerate the occurrence of the
253 “crocus” day (i.e. the recession date) by typically one martian day.

254 2.8. Thermal conduction in ice and ground

255 For a solid, the time dependent diffusion equation leads to

$$\rho C_p \frac{\partial T}{\partial t} = -\nabla \vec{F}_c \quad (21)$$

256 where C_p is the specific heat (unit: $\text{J kg}^{-1} \text{K}^{-1}$), ρ the material’s density in kg m^{-3} , \vec{F}_c is
257 the conductive heat flux: $\vec{F}_c = -\lambda \vec{\nabla} T$ (λ is the solid’s heat conductivity, in $\text{W m}^{-1} \text{K}^{-1}$).

$$\rho C_p \frac{\partial T}{\partial t} = \nabla [\lambda \nabla T] \quad (22)$$

258 Thermal conduction is here considered as a one dimensional (1D) process. Temperature
259 T of the soil is thus a function of time t and depth z , which must satisfy the following
260 equation

$$\rho C_p \frac{\partial T}{\partial t} = \frac{\partial}{\partial z} \left(\lambda \frac{\partial T}{\partial z} \right) \quad (23)$$

261 In the context of physical processes in the soil, the boundary conditions for this
262 problem are

$$-\lambda \frac{\partial T}{\partial z} (z = 0) + \Sigma F_{\text{other}} - \epsilon \sigma T^4 = 0 \quad (24)$$

263 at the surface and

$$-\lambda \frac{\partial T}{\partial z} (z = z_0) = F_{\text{geothermal}} \quad (25)$$

264 at the bottom of the domain.

265

266 Thermal conduction in the soil is computed as follows:

- 267 • Time integration is done via an implicit (first order) Euler scheme which approxi-
268 mates the differential equation $dT(t)/dt = F(T, t)$ as:

$$\frac{T^{(i+1)} - T^{(i)}}{\delta t} = F(T^{(i+1)}, t^{(i+1)}) \quad (26)$$

269 where superscripts denote time levels and δt the time step.

- 270 • The lower boundary condition is included as such in the solver, but the upper
271 boundary condition is not. Surface temperature is technically linked to atmospheric
272 and ground processes which are coupled (via boundary condition in Eq. 24). Rather
273 than solving the coupled problem, it is uncoupled in a way that allows one to solve
274 atmospheric and ground processes separately. The essential feature of the artificial
275 uncoupling is that the atmospheric problem is first solved, yielding in the processes
276 the value of surface temperature $T_{\text{surf}}^{(i+1)}$ at time $t^{(i+1)}$, which is then used as an
277 input boundary condition for the soil heat diffusion problem.

278 2.9. CO_2 phase transition

279 2.9.1. Sublimation temperature

280 CO_2 condensation/sublimation temperature is taken from Hourdin *et al.* (1993).
281 This relation tends to slightly underestimate the sublimation temperature in the pressure
282 range compared to other relations (James *et al.* 1992). However a maximum difference
283 of 2 K was noticed and simulation results are not significantly impacted.

$$T_{\text{sub}}(z) = \frac{1}{0.0734 - 0.000324 \log(0.01 P_{\text{CO}_2}(z))} \quad (27)$$

284 with P_{CO_2} the partial pressure of CO_2 in Pa. We take $P_{\text{CO}_2} = 0.95 P_{\text{surf}}$ at the surface.
285 We assume that the pressure inside the slab at depth z is given by:

$$P_{\text{CO}_2}(z) = P_{\text{surf}} + \rho g z \quad (28)$$

286 with ρ the density of the ice in kg m^{-3} , g the gravity on Mars (3.72 m s^{-2}) and z the
287 depth in m. Figure 7 shows the evolution of the sublimation temperature with depth. In
288 our case, CO_2 ice thickness can range from 10 cm at mid-latitudes to more than 70 cm
289 in the polar regions. The difference between the sublimation temperature at the surface
290 and the one at the bottom of the slab can be higher than 20 K in typical polar conditions.

291 [Figure 7 about here.]

292 *2.9.2. CO₂ sublimation/condensation*

293 Latent heat is released or absorbed as CO₂ condenses or sublimates such that:

$$L = -\frac{dm_{\text{CO}_2}}{dt} L_{\text{sub}}^{\text{CO}_2} \quad (29)$$

294 with $L_{\text{sub}}^{\text{CO}_2}$ the CO₂ latent heat of sublimation ($L_{\text{sub}}^{\text{CO}_2}=5.9 \cdot 10^5 \text{ J kg}^{-1}$) and m_{CO_2} the
 295 mass of CO₂. For each layer, the difference between the temperature of the layer, after
 296 adding the increase due to solar energy absorption and computing conduction, and the
 297 sublimation temperature leads us to the amount of CO₂ ice that sublimates (if CO₂ ice
 298 is present) or condenses (if CO₂ gas is available):

$$dm_{\text{CO}_2}(i) = \frac{T_{\text{sub}}(i) - T_{\text{soil}}(i)}{L_{\text{sub}}^{\text{CO}_2} \Delta t} \rho C_p \Delta z(i) \quad (30)$$

299 where $dm_{\text{CO}_2}(i)$ is the CO₂ mass variation in $\text{kg s}^{-1} \text{ m}^{-2}$, $T_{\text{soil}}(i)$ the predicted temper-
 300 ature (without taking into account the phase transition) in layer i in K, $T_{\text{sub}}(i)$ the subli-
 301 mation temperature in K, Δt the timestep in s, C_p the CO₂ heat capacity in $\text{J K}^{-1} \text{ kg}^{-1}$,
 302 $\Delta z(i)$ the thickness of layer i in m. We actually compare the temperature of the layer,
 303 which is calculated at the middle of the layer, with the sublimation temperature, which
 304 is calculated at the base of the layer (or at the top of the layer for the first one, since
 305 sublimation occurs at the surface). Decreasing the thickness of the layers shows that this
 306 approximation had no significant effect on the results.

307 *2.10. Algorithm*

308 As we said previously, our model solves heat, radiation and mass evolution equations
 309 in the CO₂ ice and in the regolith. If we consider a translucent slab, solar radiation can
 310 easily penetrate. The regolith quickly heats and so does the CO₂ ice at the bottom of the
 311 slab. If the sublimation temperature is reached, we first need to convert the energy in
 312 excess to the amount of CO₂ that sublimates before computing the thermal conduction
 313 in the above layers. Thus we need to couple heat, radiation and mass evolution. The
 314 algorithm that we chose consists in computing the solar energy deposition in each layer,
 315 then doing a loop with the following processes:

- 316 1. Find the layer where the excess of energy is the highest compared to the sublimation
 317 point (or the lack of energy if we consider CO₂ layer where CO₂ gas is present, like

318 the first layer for instance). It is the layer where sublimation (or condensation) is
 319 the most likely to occur.

- 320 2. Compute the latent heat related to this sublimation (or condensation).
- 321 3. Calculate the new thermal conduction coefficients and compute the new tempera-
 322 tures.
- 323 4. Return to step 1. as long as there is still an excess of energy (or a lack in the
 324 specified conditions).

325 These fluxes will be used later to calculate the sublimated or condensed amounts of CO₂
 326 ice in each layer. Calculating processes this way guarantees that the thermal conduction
 327 is computed using realistic temperatures in the CO₂ ice that are never higher than the
 328 sublimation point.

329 2.11. Effect of gas in the layers

330 If basal sublimation occurs, a layer of gas forms between the CO₂ slab and the
 331 regolith, which modifies the thermal behavior of the slab. The conductivity of CO₂
 332 gas is much lower than the conductivity of CO₂ ice ($\lambda_{\text{ice}}^{\text{CO}_2} = 0.65202 \text{ W m}^{-1} \text{ K}^{-1}$ at
 333 144K, $\lambda_{\text{gas}}^{\text{CO}_2} = 0.00590 \text{ W m}^{-1} \text{ K}^{-1}$ (Kieffer 2007)). Moreover, the distance between
 334 the ice and the regolith will increase. Both radiation and convection can play a role
 335 in the transfer of heat through this gaseous layer. We can simulate this behavior as
 336 the conductive, radiative and convective transfer between two horizontal plates with the
 337 temperature of the plate below higher than the temperature of the plate above. The
 338 total flux of energy between the two plates is

$$\Sigma Flux = C_d + R_d + C_v \quad (31)$$

339 where C_d is the conductive flux, R_d the radiative flux and C_v the convective flux.

340 We can estimate

$$C_d = \lambda_{\text{gas}}^{\text{CO}_2} \frac{\partial T}{\partial z_{\text{gas}}} \quad (32)$$

341 z_{gas} takes into account the gas expansion. For the radiative exchange, we suppose here
 342 that ϵ is equal to 1 for the CO₂ ice and for the regolith at the interface in order to
 343 simplify the calculation:

$$R_d = \Delta(\sigma T^4) \quad (33)$$

344 where σ is the Stephan-Boltzmann constant and T the temperature at the regolith/CO₂
345 gas interface and at the CO₂ ice/CO₂ gas interface.

346 For the convective flux, we have first to determine the Rayleigh number (Ra), which
347 is the product of the Grashof number (Gr) and the Prandtl number (Pr).

$$Ra = Gr Pr \quad (34)$$

$$Gr = \frac{g\beta\Delta T\Delta z_{\text{gas}}^3}{\nu^2} \quad (35)$$

348

$$Pr = \frac{C_p \mu}{\lambda} \quad (36)$$

349 where ν is the kinematic viscosity of CO₂, β the volumetric thermal expansion coefficient
350 (equal to approximately 1/T for ideal fluids, where T is the temperature), g the gravity
351 on Mars and Δz_{gas} the distance between the regolith/CO₂ gas interface and the CO₂
352 ice/CO₂ gas interface. $\nu = \frac{\mu}{\rho}$ where μ is the CO₂ gas viscosity (1.1 10⁻⁵ Pa s at 160 K)
353 and ρ the CO₂ gas density (0.125 kg m⁻³ at 162.8 K under a 0.6 m thick CO₂ ice slab
354 for surface pressure equal to 507 Pa). We can estimate

$$Ra = 4.81 * 10^6 \Delta T \Delta z_{\text{gas}}^3 \quad (37)$$

355 Since we are in the case of Rayleigh-Benard convection, the critical value of the Rayleigh
356 number is estimated at 1707. We can assume a reasonable gas layer width of 2 cm
357 (Kieffer 2007). In this case, the critical value is obtained for $\Delta T=44\text{K}$, which is one
358 order of magnitude above the modeled gradient. We therefore assumed in our study that
359 convection could be neglected.

360 To compute the effect of gas in the layer using the conduction equation (Eq. 22), we
361 kept the calculation as before, as if the ice and the regolith were in conductive contact,
362 but used an equivalent conductivity between them which takes into account

363 (1) the expansion of the gas and its conductivity

364 (2) the radiative flux.

365 We have

$$\lambda_{\text{eqgas}} \frac{\Delta T}{(1 - q_{\text{CO}_2})\Delta z} = \lambda_{\text{gas}} \frac{\Delta T}{(1 - q_{\text{CO}_2})\gamma\Delta z} + \sigma\Delta(T^4) \quad (38)$$

366 where γ is the expansion coefficient of CO₂ ($\gamma = \frac{\rho_{\text{solid}}}{\rho_{\text{gas}}}$) and q_{CO_2} the fraction of solid
367 CO₂ in the layer. Rearranging,

$$\lambda_{\text{eqgas}} = \frac{\lambda_{\text{gas}}}{\gamma} + (1 - q_{\text{CO}_2})\Delta z\sigma \frac{\Delta(T^4)}{\Delta T} \quad (39)$$

368 When basal sublimation occurs, the last layer of CO₂ has both ice and gas. We have to
369 calculate its new conductivity, which is given by

$$\lambda_{\text{eq}} = \frac{\Delta z}{\frac{q_{\text{CO}_2}\Delta z}{\lambda_{\text{solid}}} + \frac{(1-q_{\text{CO}_2})\Delta z}{\lambda_{\text{eqgas}}}} \quad (40)$$

370 2.12. Gas ejection

371 In our model, we assume that in case of basal sublimation, when the gas amount
372 reaches a threshold value (which can be chosen), the ice breaks, the gas is ejected in
373 the atmosphere and the amount of gas at the base of the CO₂ slab is set to zero. This
374 threshold value is set to 2 cm in the simulations, a reasonable value according to Kieffer
375 (2007). However, it has no significant impact since it corresponds to a very small amount
376 of gas which can form in only a few hours. Assuming no heat exchange during the ejection
377 we have

$$\frac{T_{\text{exit}}}{T_{\text{soil}}(z)} = \left(\frac{P_{\text{surf}}}{P(z)}\right)^{\left(\frac{R_{\text{CO}_2}}{C_p}\right)} \quad (41)$$

378 where T_{exit} is the temperature of the gas when it is ejected in the atmosphere, $T_{\text{soil}}(z)$ the
379 temperature at the depth z where CO₂ ice is sublimated, P_{surf} the surface pressure, $P(z)$
380 the pressure of the gas at the depth z where CO₂ ice is sublimated, R_{CO_2} the specific
381 gas constant of CO₂, C_p the specific heat of CO₂ gas. Let us take a typical example
382 with a 0.6 m thick CO₂ slab. Pressure at the base is close to 3985 Pa if the surface
383 pressure is equal to 400 Pa (a typical surface pressure at 85° S latitude and Ls 180°).
384 The temperature at the base of the slab is equal to the sublimation temperature in these
385 conditions, i.e. 162.75 K. The calculation gives us an ejection temperature close to 98 K
386 which means that, using this simple approach, we can estimate that about 7 percent of
387 the ejected gas may recondense during the ejection:

$$\% \text{ of recondensation} = \frac{T_{\text{cond}} - T_{\text{exit}}}{L_{\text{cond}}^{\text{CO}_2}} C_p^{\text{gas}} \quad (42)$$

388 3. Results and application to the Manhattan Island region

389 3.1. The Manhattan Island region

390 The “Manhattan Island” region is centered at 99° E, 86.25° S, and follows the classic
391 TES “cryptic” behavior of low albedo while remaining near the CO_2 ice temperature
392 (Kieffer *et al.* 2006). At this latitude, polar night begins at $\text{Ls } 10^\circ$ and ends at $\text{Ls } 168^\circ$.
393 The first dark spots appear very quickly after the end of the polar night. They can be
394 seen in THEMIS images at $\text{Ls } 176^\circ$ (Kieffer *et al.* 2006). In this region, OMEGA spectra
395 show that CO_2 ice consists of a thick slab, contaminated by dust in the upper layers
396 (Langevin *et al.* 2007). This spectrum is typical of the cryptic region. OMEGA data
397 were acquired between $\text{Ls } 183^\circ$ and $\text{Ls } 192^\circ$ and since dust may have been brought by
398 the venting process, it is possible that a clean slab lay on the ground earlier, at the end
399 of the polar night. According to our model, such a slab would be very translucent in the
400 visible. For instance, with a solar incidence angle of 70° , about $3/4$ of the flux reaches
401 the base of the slab. We thus tried our model with this particular area. The main point
402 here was to check if basal sublimation could occur and when.

403 3.2. Evolution of temperature

404 The following baseline parameters were chosen: slab model; uniform regolith thermal
405 inertia: 200 SI; regolith albedo: 0.24; atmospheric dust optical depth: 0.1 (Vincendon *et*
406 *al.* 2008; Vincendon *et al.* 2009).

407 3.2.1. Simulation results on a flat terrain:

408 [Figure 8 about here.]

409 [Figure 9 about here.]

410 Just after the end of the polar night, as the sun is low on the horizon, the Fresnel
411 reflection severely limits the direct solar flux penetration and temperatures remain near
412 145 K in the slab, as shown in Fig. 8. The total amount of CO_2 ice keeps growing even
413 if sublimation at the surface has already begun during the day. As soon as the sun is
414 high enough in the sky, solar radiation penetrates down to the regolith interface where it
415 is absorbed in the first few millimeters. Temperature increases quickly, and by thermal

416 conduction, at the base of the CO₂ slab. Sublimation temperature at the base of the
417 slab is reached at Ls 191 °, during the afternoon. The ejection time occurs a few tens
418 of minutes after the sublimation begins. Gas continues to form and to be vented out
419 until the night comes and the temperature gets too low. The simulation results show
420 that some energy is kept during the night in the slab. Therefore, each following day the
421 energy that is needed for basal sublimation to occur decreases and the first gas ejection
422 of the day is earlier than for the previous day. As shown by Fig. 8, adding dust, even
423 a few ppm, has a significant impact on the results. Dust radically changes the slab's
424 properties. For instance, a slab with a mass fraction of dust of 10 ppm absorbs about 5
425 time more solar radiation. That is the reason why temperature is higher in the slab when
426 dust is present. However, a smaller fraction of solar radiation can reach the regolith and
427 temperature increase at the interface is slower. "Crocus" day occurs at about Ls 245 ° in
428 these simulations, which is quite early compared to the observations (more in the range
429 of Ls 250-280 °) (Kieffer *et al.* 2000; Langevin *et al.* 2007). We have to keep in mind
430 that we simulate here a slab with constant properties, as we are mainly interested in
431 simulating the onset of spot formation. In reality dust and CO₂ frost that are ejected
432 change the optical properties of the slab. This optically thick layer of dust and frost stops
433 the solar radiation and sublimation can occur only at the surface until wind removes this
434 dust away and frost sublimates. Solar radiation can then penetrate in the slab again and
435 basal sublimation can go on. After the "crocus" day, the surface temperature increases
436 very quickly up to about 260 K as shown by Fig. 9. During fall, the surface temperature
437 decreases down to the condensation point and CO₂ ice begins to form.

438 *3.2.2. Simulation results on a 30 ° north oriented slope:*

439 [Figure 10 about here.]

440 [Figure 11 about here.]

441 For these simulations, we took into account a 30 ° north oriented slope (equator fac-
442 ing). Processes involved in this simulation (Fig. 10 and 11) are quite similar to the
443 previous one. However, since the angle between the direct incoming solar flux and the
444 local vertical is slightly lower, Fresnel reflection is not that significant and a large part of
445 the solar radiation can penetrate very early into the slab. As a result, first gas ejections

446 can occur as early as Ls 174° which is consistent with the observations. We can notice
447 (Fig. 10) that at Ls 174° (12 AM), gas ejection has not yet occurred (it will occur a few
448 hours later). As for the flat terrain case, each day, the first gas ejection occurs earlier
449 than for the previous day; at Ls 180°, the gas ejection already occurred. Since the gas
450 layer acted like an insulator, energy was kept in the first regolith layers and was released
451 after the gas ejection. This 30° north oriented slope case can actually be seen as a good
452 way to represent a small-scale irregularity on a flat terrain.

453 4. Dark spots formation on Mars

454 We have shown in the previous section that basal sublimation could occur in the
455 Manhattan Island area. We then investigated more general cases and tried to see how
456 significant the different parameters are for initiating the venting process.

457 4.1. Mean effective grain size and the dust content

458 CO₂ ice properties are the key to the venting process. Figure 12 shows that different
459 scenarios will occur, depending on these properties (CO₂ ice grain size and dust content).
460 In every case, sublimation occurs at the surface, but if solar radiation can penetrate deep
461 in the ice, basal sublimation becomes possible.

462 [Figure 12 about here.]

463 4.1.1. Surface sublimation

464 Very clean small-grained ice has an albedo so high that there is not enough energy
465 to entirely sublime the seasonal CO₂ ice layer and ice accumulates year after year (Fig.
466 12, zone 4). If we increase the grain size or add dust (Fig. 12, zone 3), we decrease the
467 albedo and a CO₂ cycle consistent with the observed one is possible. Figure 13 shows
468 that the temperature remains near 145 K in the CO₂ slab until the “crocus” day. Solar
469 radiation is absorbed in the first millimeters and sublimation only occurs at the surface.

470 [Figure 13 about here.]

471 *4.1.2. Sublimation within the ice*

472 If we continue to increase the effective grain size (more than a few millimeters) and
473 limit the mass fraction of dust to less than a few hundreds of ppm (Fig. 12, zone 2),
474 solar radiation begins to penetrate more deeply into the ice. Still, most of the energy is
475 absorbed by the ice and very little can reach the regolith. Sublimation temperature can
476 be reached within the ice, as shown on Fig. 14 and Fig. 15. CO₂ gas bubbles can form
477 around and ice could eventually crack.

478 [Figure 14 about here.]

479 [Figure 15 about here.]

480 *4.1.3. Basal sublimation*

481 When the effective grain size is in the range of about ten centimeters or more, and
482 the mass fraction of dust is less than about 10 to 15 ppm (Fig. 12, zone 1), the fraction
483 of solar radiation that reaches the regolith is sufficient for basal sublimation to occur.
484 As we explained before, surface sublimation continues at the same time.

485 *4.1.4. Diurnal cycle*

486 Slab CO₂ ice and small grained CO₂ ice have totally different behaviors with regard
487 to the diurnal cycle, as can be seen on Fig. 16 and Fig. 17. With small grained CO₂
488 ice, everything occurs at the surface, where the CO₂ ice layer is in equilibrium with the
489 atmosphere and the CO₂ ice amount evolution is directly linked to the diurnal cycle. It
490 decreases during the day and increases during the night. However, in the case of the
491 translucent slab scenario, the diurnal cycle has less influence on the sublimation and
492 condensation processes. Only a small fraction of the solar flux is absorbed by the CO₂
493 ice. Most of the energy is absorbed by the regolith. Thus during the day, sublimation at
494 the surface is limited. Thermal conduction, but also solar flux absorption in the CO₂ ice,
495 leads to energy storage in the slab and condensation during the night is reduced (Fig.
496 17).

497 [Figure 16 about here.]

498 [Figure 17 about here.]

499

[Figure 18 about here.]

500 *4.2. Effect of slope and orientation*

501 Slope and orientation change the incidence angle of the photons, and therefore the
502 Fresnel reflection which limits the solar flux penetration. At high latitudes, the solar
503 zenith angle is generally between 80° and 90° at the beginning of spring. Slope and
504 orientation thus determine when a large fraction of the solar radiation begins to penetrate
505 into the slab and thus when the first gas ejection will occur. In the case of Manhattan
506 Island (85° S), Fig. 18 shows that the CO_2 ice maximum is about 15% smaller on a
507 30° north oriented slope than on a flat terrain, and the sublimation temperature at the
508 base of the slab is therefore about 1 K lower. Furthermore, at $\text{Ls } 175^\circ$ for instance, solar
509 zenith angle minimum is at about 86° (Fig. 2), which means that a maximum of 40%
510 of the direct flux can penetrate into the slab for a flat terrain whereas more than 90%
511 penetrates in the case of a 30° north oriented slope.

512 *4.3. Regolith thermal inertia*

513 Aharonson (2004) and Kieffer (2007) suggested that the seasonal thermal wave could
514 be sufficient to cause CO_2 ice sublimation during the polar night. We did not observe any
515 gas ejection in our simulations during the polar night, even when we considered a high
516 thermal inertia (up to 2000 SI) that could be representative of an ice-rich near-surface.
517 High thermal inertia certainly limits the CO_2 condensation but also makes the regolith
518 heating much more difficult. Since the sublimation temperature at the interface is much
519 greater than at the surface it is very difficult to reach it. Figure 19 shows that increasing
520 the thermal inertia of the regolith delays by up to 10° of Ls the starting date of CO_2 jet
521 activity. Seasonal thermal waves can therefore not initiate the venting process on their
522 own. Furthermore, according to our results, regolith thermal inertia does not have a large
523 impact on the time of the first gas ejection (a few degrees of Ls) since both previously
524 mentioned processes seem to compensate each other.

525

[Figure 19 about here.]

526 4.4. Regolith albedo

527 We ran tests with albedo values between 0.05 and 0.3 (realistic range for the martian
528 soil). Figure 20 shows a small increase (less than 5 ° of Ls for the albedo range) of the
529 time needed to heat up the regolith as we increased the albedo.

530 [Figure 20 about here.]

531 4.5. Latitude

532 Simulations were also run at different latitudes (Fig. 21). Atmospheric dust optical
533 depth was kept to 0.1. Surface pressure was set to 400 Pa in the southern hemisphere
534 and to 1000 Pa in the northern hemisphere. This higher pressure in the north tends
535 to increase the maximum amount of CO₂ ice. This explains why CO₂ ice maxima are
536 rather close in north and south polar regions in spite of the fact that the southern polar
537 night is longer. The time of initiation of the venting process evolves with latitude, as a
538 function of the incident solar radiation.

539 [Figure 21 about here.]

540 5. Discussion

541 5.1. Manhattan Island region scenario

542 CO₂ ice may have some dust inclusions when the first solar beams penetrate, about
543 10 ° of Ls before the end of the polar night. The solar radiation remains too low to
544 significantly heat the regolith at that time, but its heating may clean the slab, as described
545 in Kieffer (2007) and Portyankina *et al.* (2010). Thus, at the end of the polar night, when
546 the solar radiation is significantly higher, the slab is probably translucent and the clean
547 slab scenario is appropriate. A large fraction of solar radiation can penetrate as soon
548 as polar night ends on small-scale irregularities (equator facing slopes) and initiates the
549 venting process. The regolith quickly heats up and sublimation temperature is reached
550 at the base of the CO₂ slab as early as Ls 174 °. During the day, gas continues to form
551 and to be ejected. At night, sublimation stops until the following day when sublimation
552 temperature can be reached again. When frost and dust cover the ice, solar radiation
553 cannot penetrate as easily anymore. However, basal sublimation can continue outside

554 the dark spot. As the slab has already been weakened by the first gas ejection, it is
555 likely that sublimated CO₂ will try to escape by the same path. As the CO₂ ice amount
556 decreases, pressure difference between the surface and the base of the slab decreases.
557 Gas and dust ejection is therefore less and less violent and less and less dust or substrate
558 material is expelled outside. Near Ls 260°, “crocus” day occurs: CO₂ disappears and
559 regolith heats up to 260 K in a few days.

560 5.2. Venting process on Mars

561 Several areas with slab ice have been observed at different latitudes on Mars. Our
562 simulations tend to show that for similar conditions, basal sublimation can occur at every
563 latitude. Of course, the effect will not be the same at 85° S where there may be a 60 cm
564 thick slab and at 50° S where it will not be thicker than 10 cm. Indeed the difference
565 of pressure between the bottom of the slab and the surface will be totally different. At
566 mid latitudes, where there is much less ice, gas ejection will be less energetic and there
567 will be less (or no) ejected dust and substrate material. The main result is that as long
568 as we have similar conditions (ice properties, albedo, regolith thermal inertia, etc.), gas
569 ejections are as likely to occur in the north as in the south. Kieffer and Titus (2001)
570 and Piqueux *et al.* (2008) report evidence of active venting processes in the north which
571 confirm our results. However, water ice is much more abundant in north polar regions
572 and certainly plays a role. Adding water ice inclusions in our model should allow us to
573 investigate more deeply dark spot formation in these regions.

574 5.3. Dark spots on dunes

575 One interesting question with dark spots is that when we look at an area with dune
576 fields, they often seem to appear only on dunes (see Fig. 2 in Kieffer (2007) and Fig.
577 22 in this article). Can we explain this behavior with our model? Dunes have generally
578 a lower albedo than the surrounding terrains (less than 0.2), a lower thermal inertia
579 due to smaller grained material (as low as 50 SI) and present some slopes (Fig. 23).
580 Model results show that gas ejections on the surrounding terrains should occur about
581 10 to 20° of Ls after the ones on the dunes. The model cannot explain why dark spots
582 seem to occur only on dunes. Whether gas ejection occurs or not is determined by CO₂
583 ice properties, which are expected to be similar both on dunes and the surroundings

584 (see previous sections). We think that gas ejections also occur around dunes but are
585 much more difficult to detect due to differences in the substrate material. Dunes are
586 composed of dark sand and the dust grains that cover the surrounding terrains have a
587 higher albedo, as can be seen on Fig. 23. In the case of a gas ejection, some substrate
588 material is brought up by the gas and ejected. The substrate material forms a dark spot
589 surrounded by bright CO₂ frost. We thus suggest that gas ejections first occur on dunes
590 and that dark and large spots appear. A little bit later, gas ejection begins to occur on
591 surrounding terrains and dust is ejected. Considering the difference of albedo, it is more
592 difficult to detect dust than dark sand on the CO₂ ice deposit. Still, in both cases bright
593 patches due to CO₂ recondensation should form. The results that we present here are
594 consistent with previous HiRISE observations of dunes fields (Fig. 24).

595 [Figure 22 about here.]

596 [Figure 23 about here.]

597 [Figure 24 about here.]

598 5.4. Cracks within the ice

599 As explained previously, we found that the sublimation temperature could be reached
600 within the CO₂ ice under certain conditions. We suggest the following interpretation for
601 such a phenomenon: CO₂ gas bubbles form at the interfaces, which means around the
602 dust, H₂O and CO₂ small grains. Pressure around these interfaces increases and causes
603 cracks in the ice. These cracks anneal during the cooler portion of the diurnal cycle,
604 especially as the slab becomes cleaner following the likely ejection of the dust particles
605 that induced them. This means that CO₂ ice properties do not remain constant: the
606 CO₂ ice effective grain radius decreases as cracks appear within the ice. Figure 12 should
607 therefore be seen as a dynamic diagram. CO₂ ice composition and properties evolve with
608 time. For instance, if sublimation within the ice was possible at a given time, cracks
609 would cause a decrease of the effective radius, leading to totally different ice properties.
610 This mechanism could be a clue to understand the decrease of the mean effective grain
611 radius observed during spring in some areas of the southern polar cap (Langevin *et al.*
612 2007). Different ice properties would lead to different sublimation processes, that could

613 themselves change these ice properties. This interpretation is highly speculative and
614 needs further study to be repudiated or confirmed, but raises an interesting concern
615 about the CO₂ ice dynamics on Mars

616 6. Conclusion

617 (1) Only in the case of very large CO₂ grains and very little dust contamination, can
618 a large fraction of solar radiation reach the regolith. In this latter case, the model
619 shows that the regolith surface can heat and sublimation temperature at the base of
620 the CO₂ ice can be reached, which is consistent with Kieffer (2007).

621 (2) Simulation results show that the occurrence of basal sublimation during spring only
622 depends on the CO₂ ice properties, considering that the substrate on which the CO₂
623 ice lies has an albedo lower than 0.3. Other parameters (thermal inertia, slope angle),
624 however, have a significant impact on the starting date of the jet activity.

625 (3) Early dark spot formation in Manhattan Island region (85° S) cannot be explained
626 by seasonal thermal waves. However, our results show that the venting process can
627 under certain conditions be initiated very early by solar radiation penetration and
628 that dark spots could form as early as Ls 174°.

629 (4) We suggest that the venting process also occurs around the dunes. However, sub-
630 strate material nature make them certainly difficult to observe compared to the dark
631 spots on the dunes.

632 (5) We suggest that the decrease of the CO₂ ice effective grain radius observed during
633 spring in some of the southern polar regions could be caused by cracks in the ice,
634 which would be the result of sublimation within the ice.

635 In our simulations, we considered CO₂ ice with homogeneous and constant prop-
636 erties (grain size, dust inclusions) through time. Adding the possibility to have non-
637 homogeneous and evolving ice should allow us to investigate in the future the differences
638 between the cryptic and in the anticryptic regions. Furthermore, better knowledge about
639 the nature and properties of the CO₂ ice during winter and its evolution with time would
640 also be required to improve the simulations.

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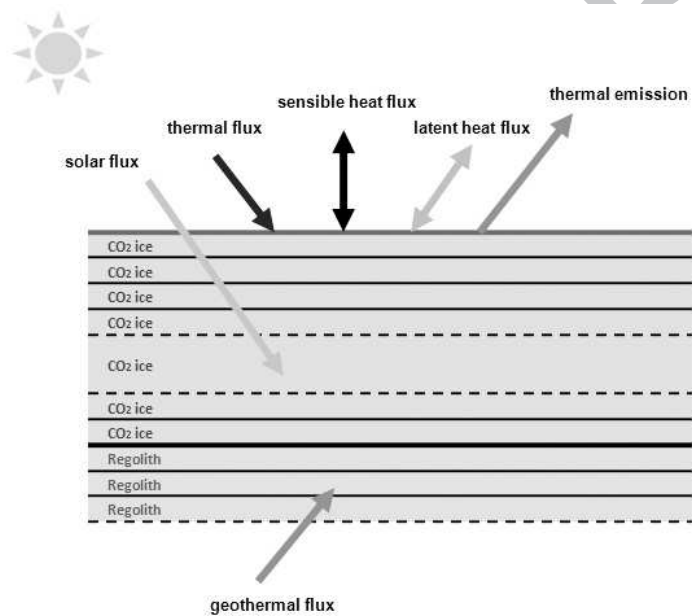


Figure 1: Energy terms used in our model (solar flux ($0.1-5 \mu\text{m}$), incident thermal flux, thermal emission, sensible heat flux, geothermal heat flux, and latent heat flux when there is a phase transition).

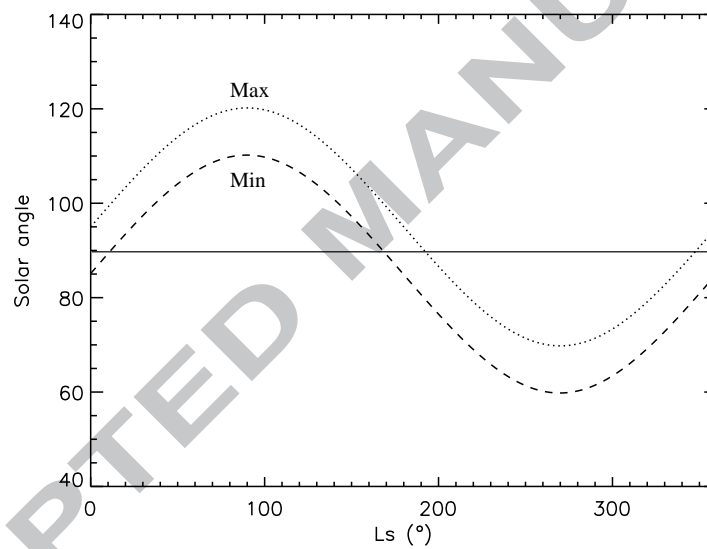


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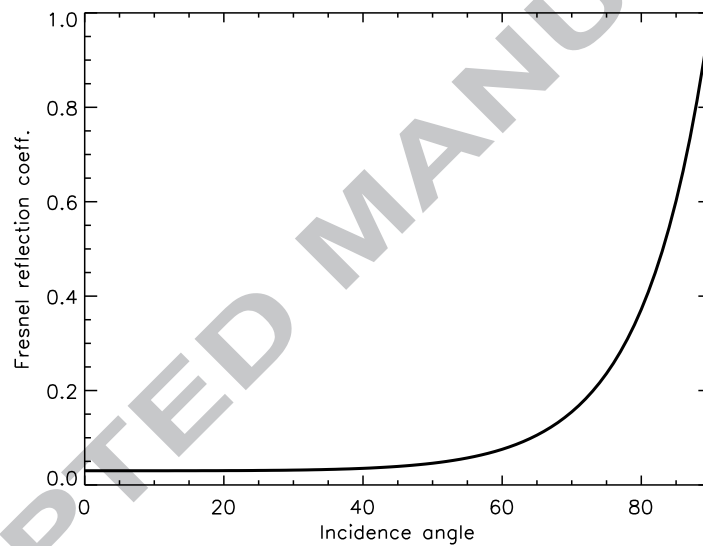


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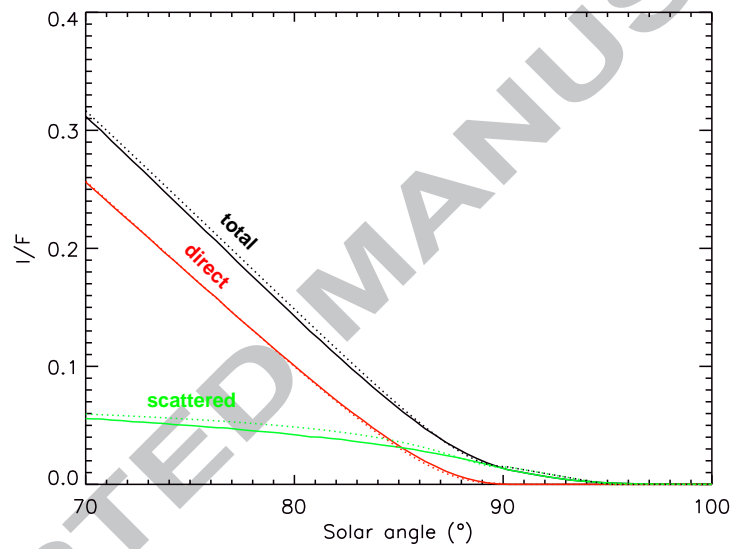


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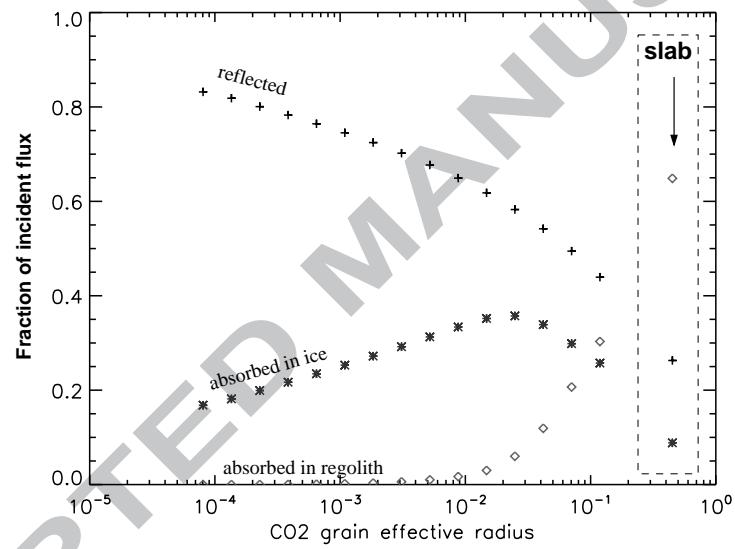


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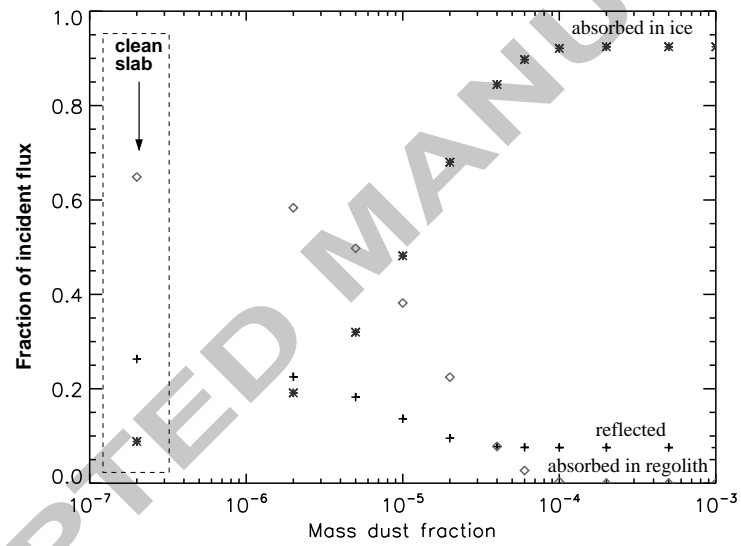


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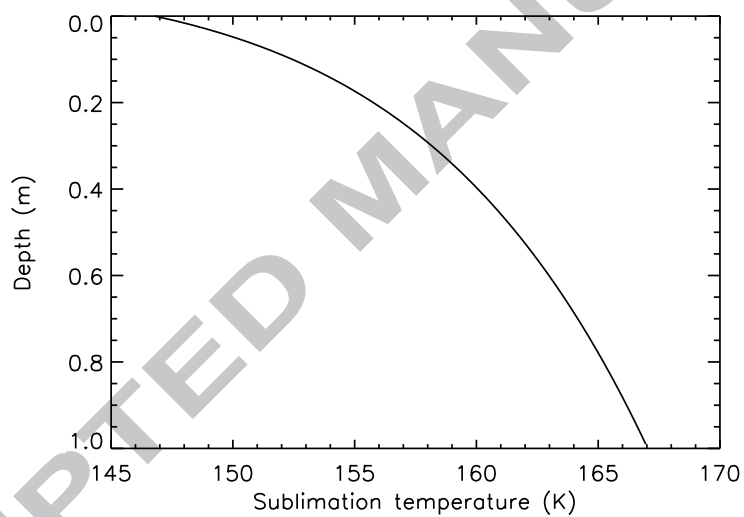


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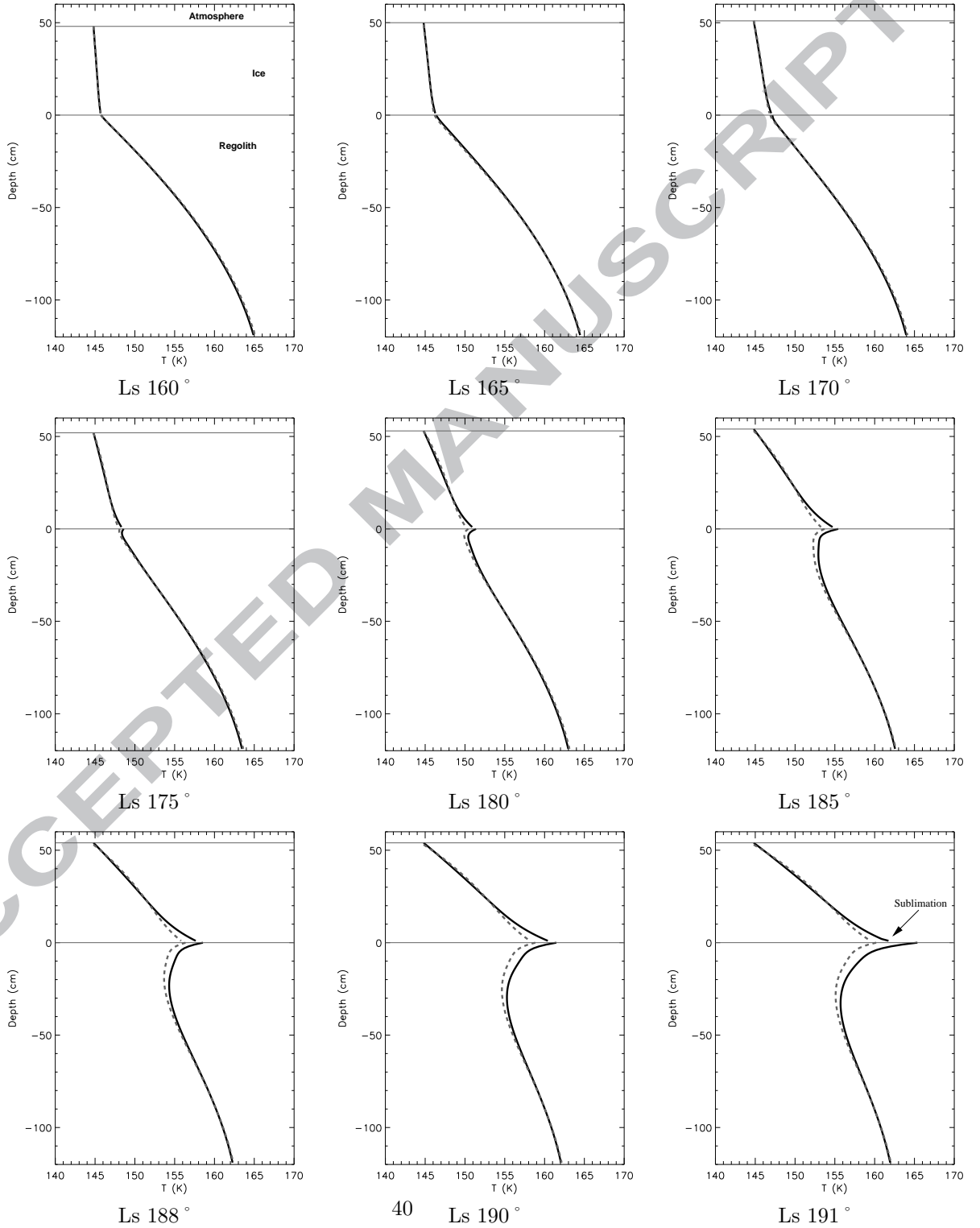


Figure 8: Temperature profile at different Ls for clean CO₂ ice on a flat terrain (local time: 12AM) at 85° S (late winter / early spring). Solid lines are simulations with a clean slab, dashed lines are simulations with a slab contaminated by 10 ppm of dust. Temperature at the regolith interface increases after the end of the polar night and basal sublimation begins at Ls 191°.

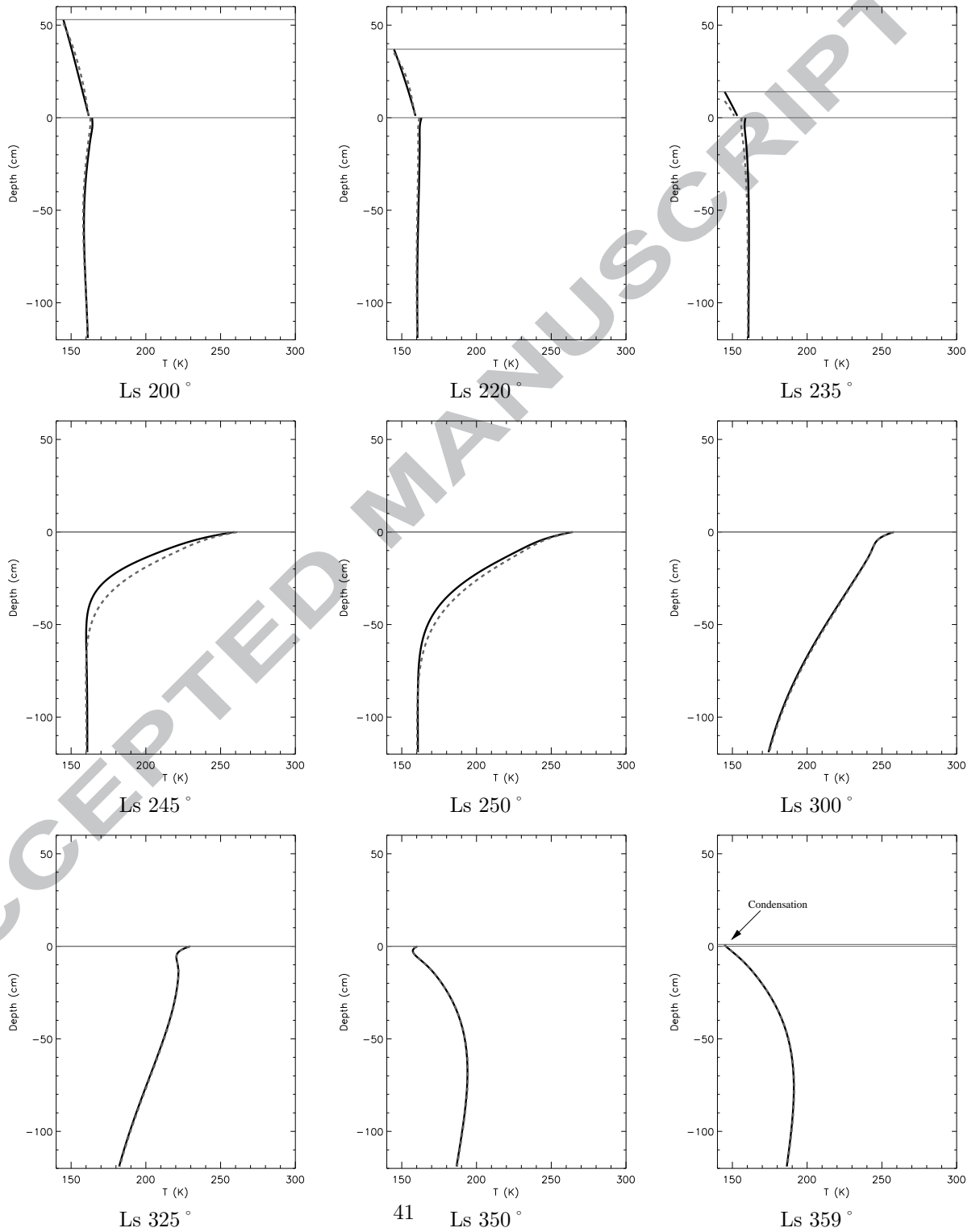


Figure 9: Continuation of Fig. 8 (mid-spring to fall equinox). The x-axis limits have been changed to 145-300K for clarity. CO₂ ice disappears at about Ls 240°. Then the surface quickly heats and reaches 260 K after a few martian days. During fall, surface temperature decreases down to the condensation temperature and CO₂ ice begins to accumulate.

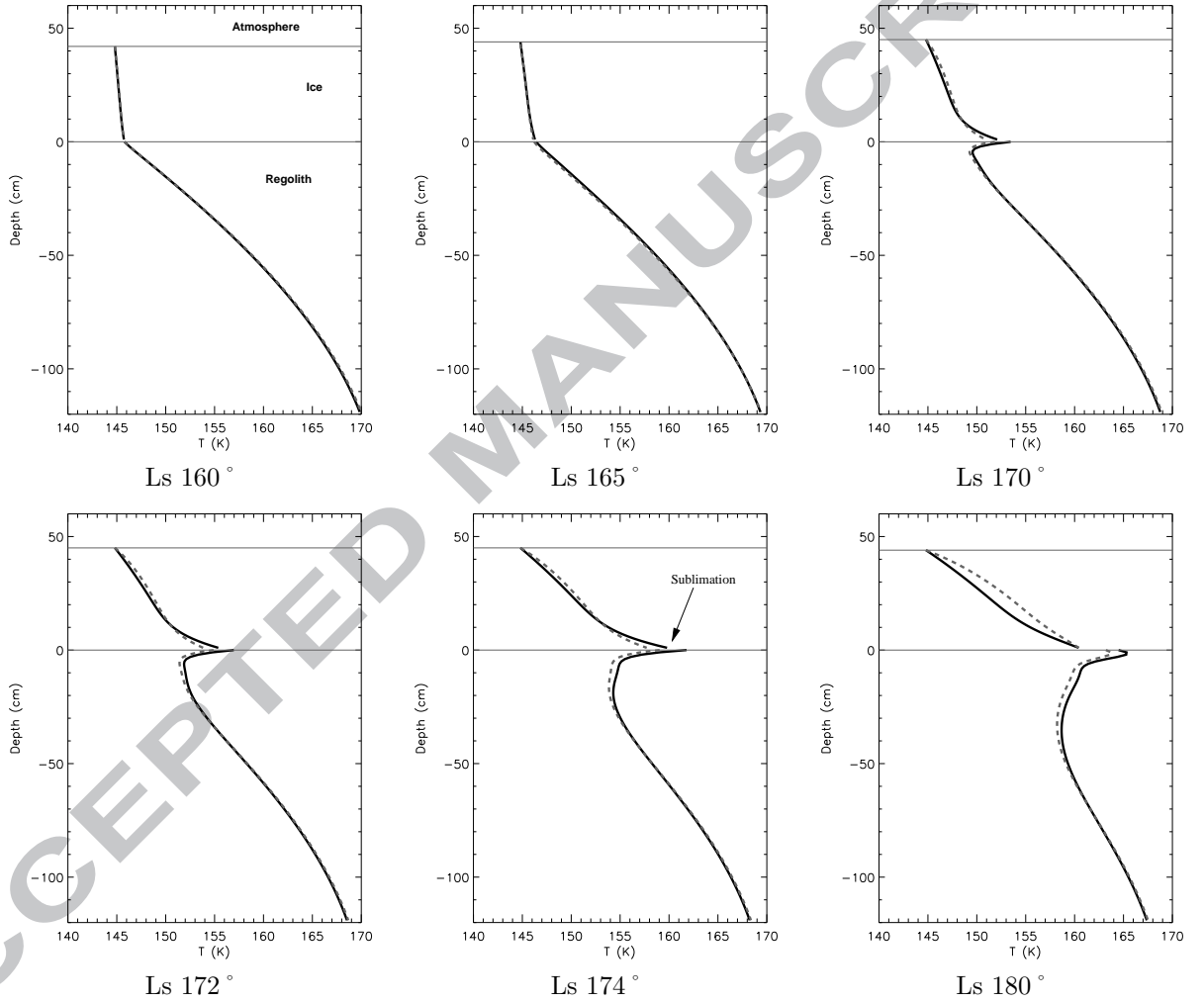


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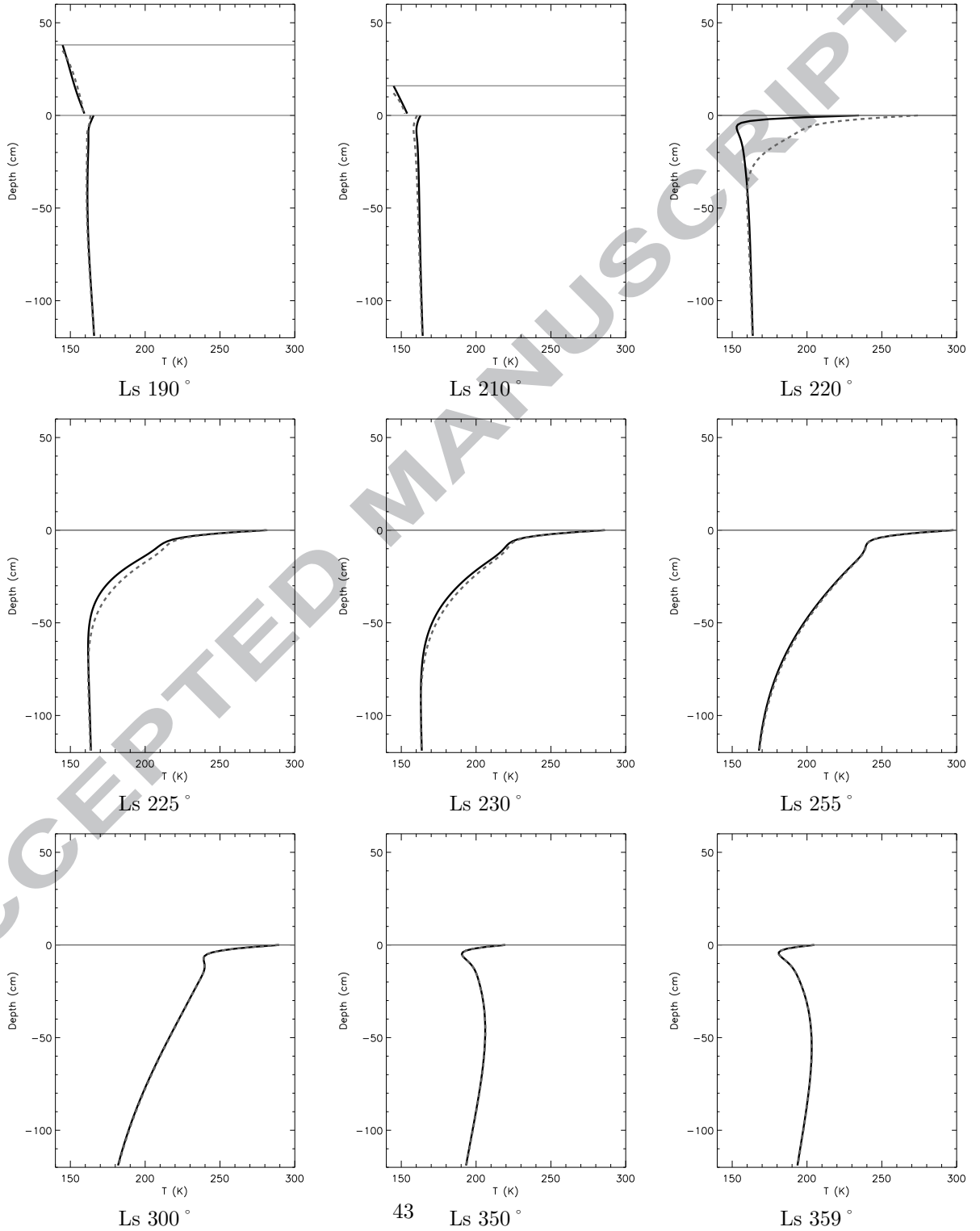


Figure 11: Continuation of Fig. 10 (mid-spring to fall equinox). CO₂ ice disappears at about Ls 220°. Then the surface temperature quickly heats up and reaches 260 K after a few martian days and 290 K after about 50 martian days.

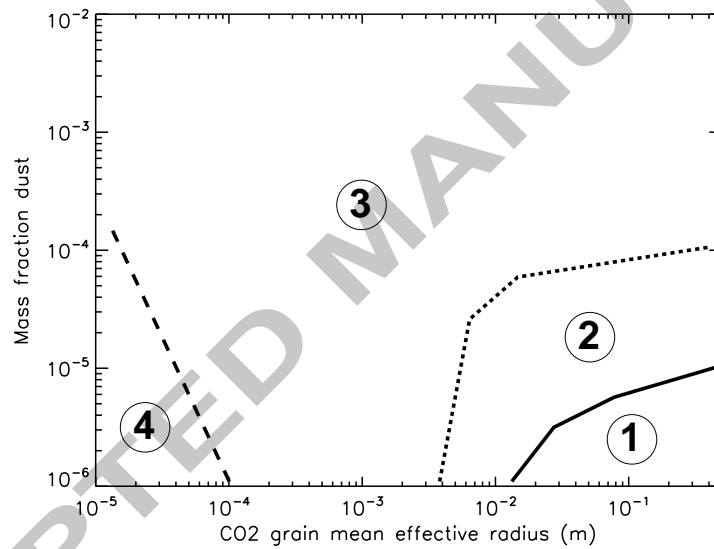


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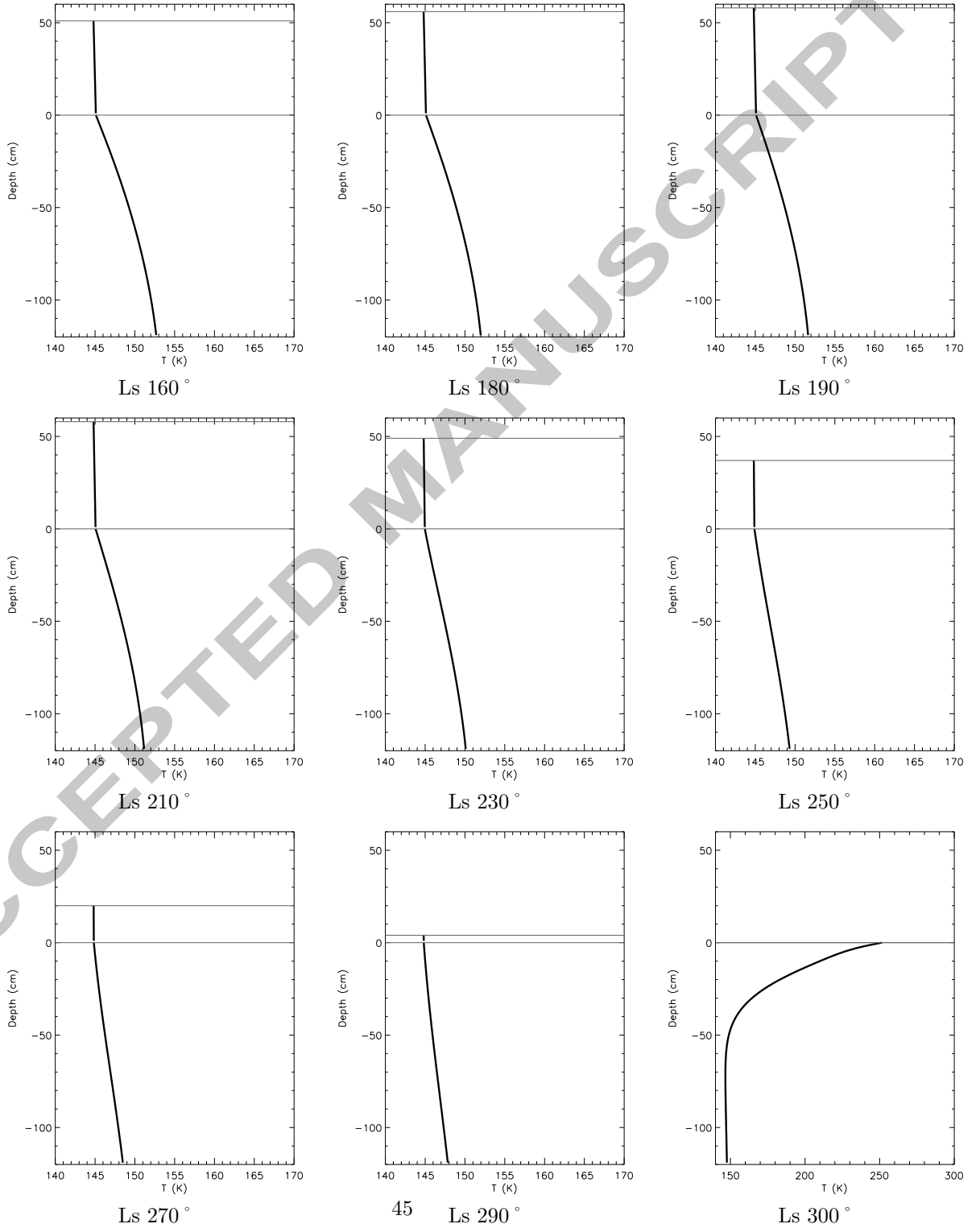


Figure 13: Same as Fig. 8 but for small-grained CO₂ ice (effective radius: 130 μm , 200 ppm of dust). As the photons are not able to penetrate into the CO₂ ice, temperature within the ice stays close to 145 K. In this case, the albedo is higher than for slab ice and CO₂ ice does not disappear until about Ls 295°.

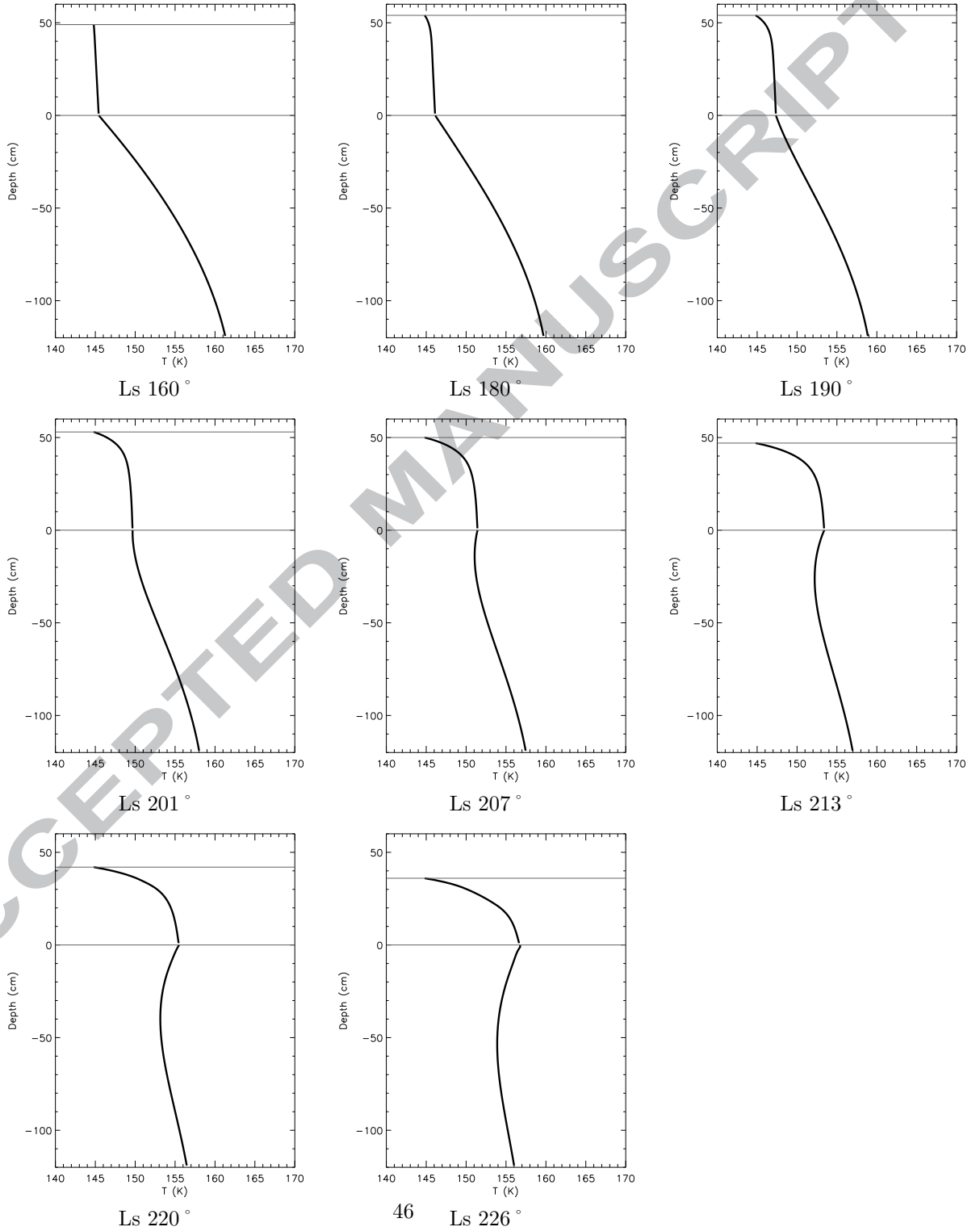


Figure 14: Same as Fig. 8 but for CO₂ ice with an effective grain radius of 2.5 cm and a dust contamination of 10 ppm. Sublimation temperature is reached within the ice at about Ls 218°. We limit the covered period here, since we think that gas bubbles will form around and ice will crack. CO₂ ice properties and involved processes will therefore change after that point.

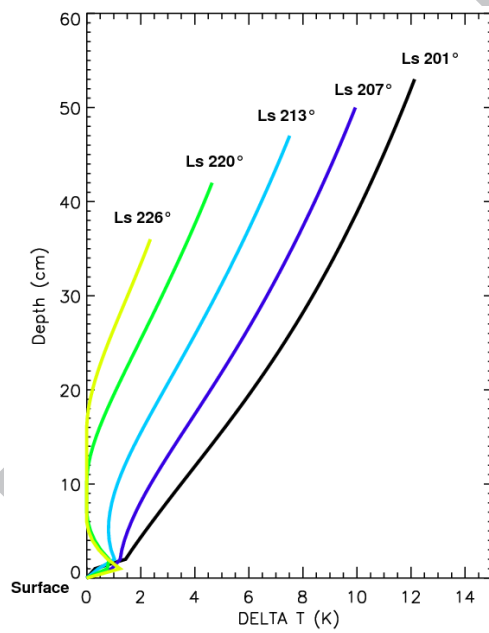


Figure 15: Evolution of the difference between the sublimation temperature and the temperature within the CO₂ ice (same parameters as in Fig. 14). Sublimation temperature is reached at about 9 cm depth.

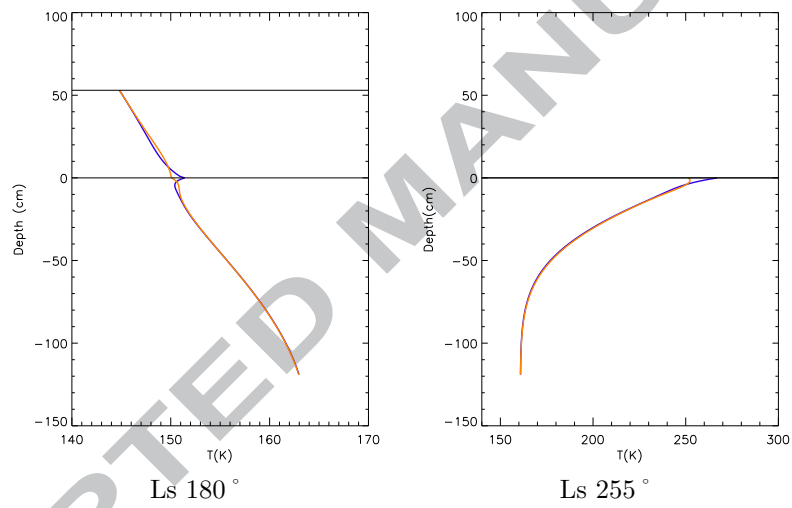


Figure 16: Temperature profile at local 12AM (blue) and 12PM (orange) for a clean CO_2 ice slab on a flat terrain at 85°S . Energy storage in the slab is apparent. Furthermore, the largest temperature difference between 12AM and 12PM occurs at the regolith interface. Once CO_2 ice has disappeared, there are daily thermal waves.

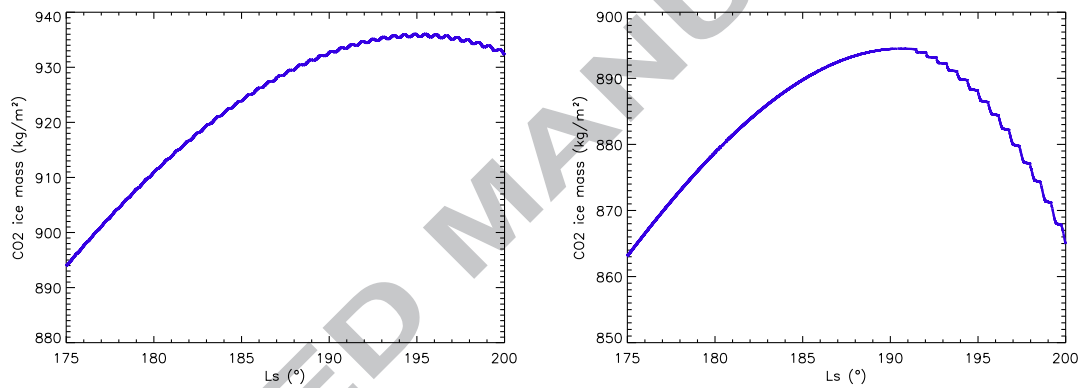


Figure 17: Evolution of the amount of CO₂ ice for small-grained dusty (100 ppm) CO₂ ice (left) and a clean slab (right) at 85° S. Considering the plot on the left and the left part of this plot (before reaching the maximum), we can see that CO₂ ice condenses at night (the CO₂ ice amount increases) and CO₂ ice sublimates during the day (the CO₂ ice amount decreases). However, in the case of translucent CO₂ ice, the amount of CO₂ ice never decreases before reaching the maximum. This is due to the fact that the solar flux penetrates in the case of a translucent slab and does not appreciably slow the condensation process whereas this same flux sublimates the ice in the case of small grains with little penetration. After the end of the polar night, in the case of slab ice, energy storage makes condensation during the night very difficult to achieve.

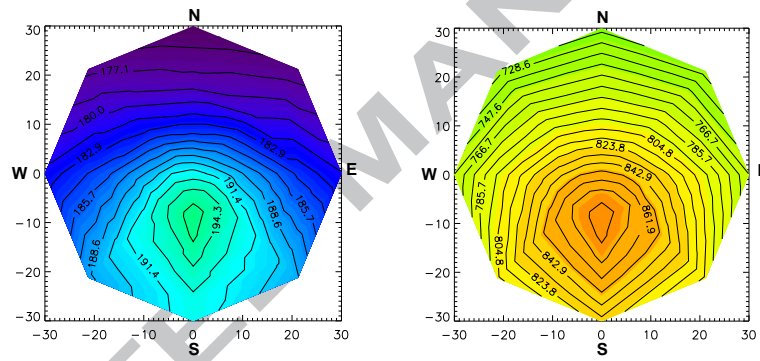


Figure 18: Evolution of the solar longitude L_s (deg) of the first gas ejection with slope and orientation (left) and evolution of CO_2 ice maximum amount (kg m^{-2}) (right) at 85°S (slab model is used). Both are directly linked to the insolation and therefore present similar patterns. We can notice that basal sublimation can occur on pole-facing slopes.

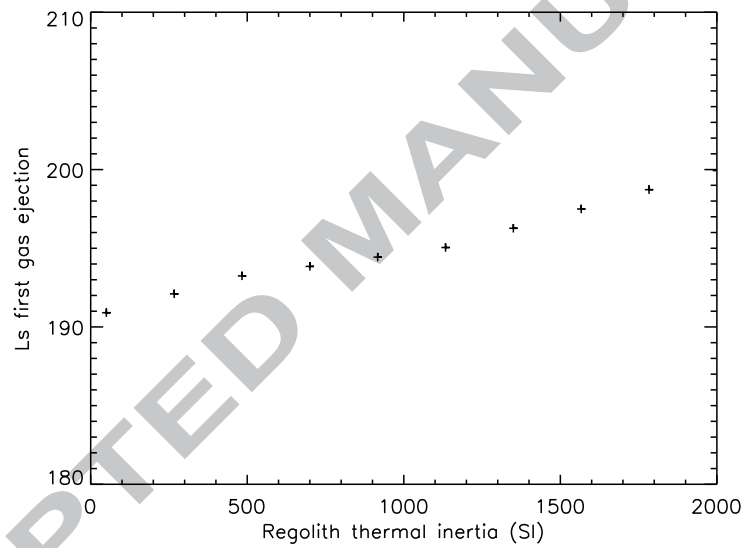


Figure 19: Evolution of the solar longitude L_s (deg) of the first gas ejection with respect to the regolith thermal inertia at 85° S (slab model). Contrary to what was expected, increasing the regolith thermal inertia does not accelerate the beginning of the venting process but delays it (less than 10° of L_s in the considered range).

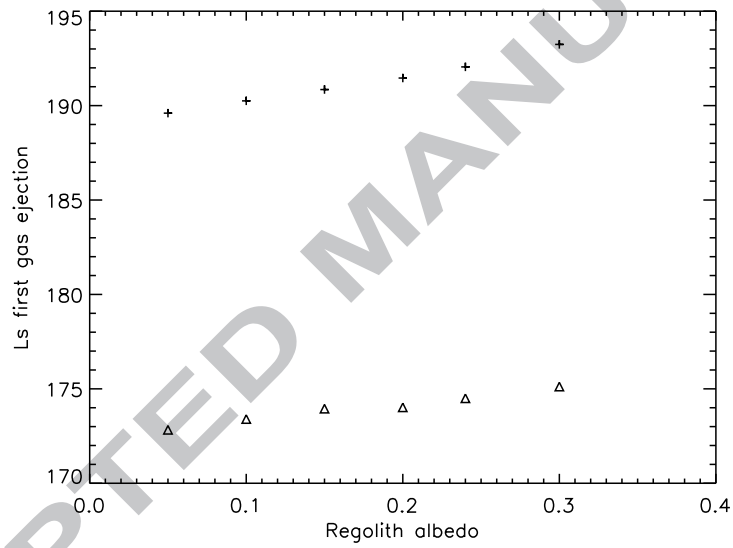
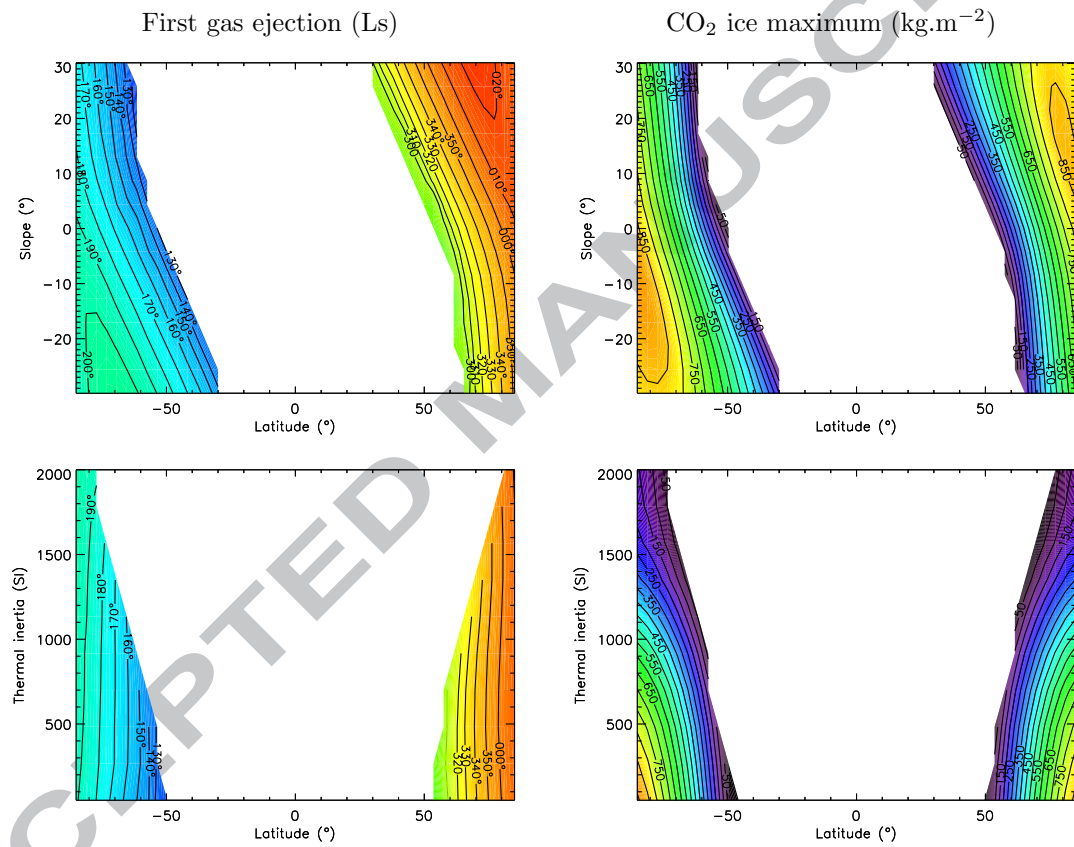


Figure 20: Evolution of the solar longitude L_s (deg) of the first gas ejection with respect to the regolith albedo at 85° S for a flat terrain (crosses) and for a 30° north oriented slope (triangles). Slab model is used. Since the slab is translucent, a large part of the energy reaches the regolith. Therefore, when we decrease the albedo, the fraction of the solar flux that penetrates into the regolith does not increase enough to accelerate significantly the beginning of the venting process.



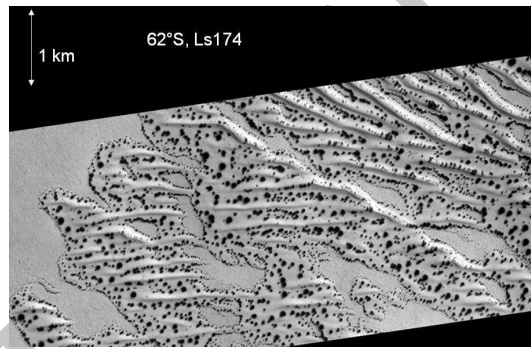


Figure 22: MOC image of a dune field at 62° S and Ls 174° . Dark spots seem to be present only on dunes.

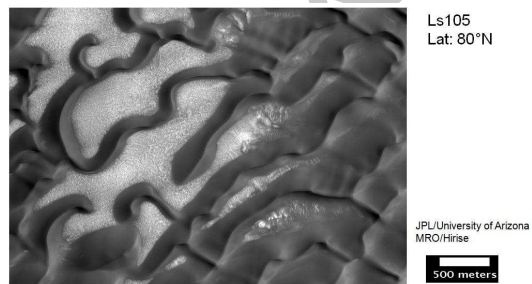


Figure 23: HiRISE image of a dune field at 80° N and Ls 105°. We can notice the albedo difference between dune material and the surroundings, since the seasonal CO₂ ice cap has disappeared.

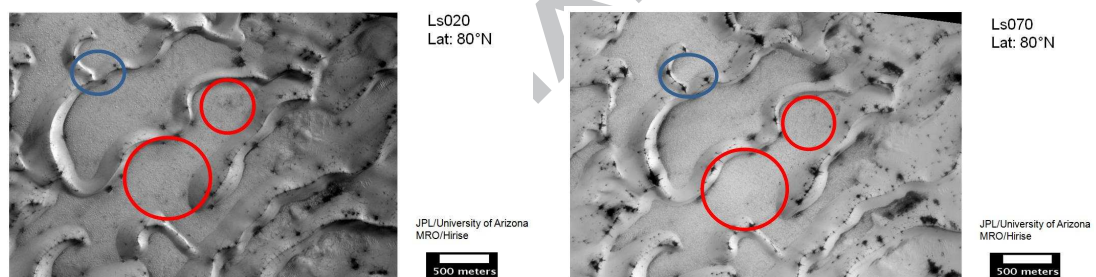


Figure 24: HiRISE image of the same dune field as on Fig. 23, but at different Ls. Dark spots are clearly visible on dunes (blue circle). Some gray patterns are also visible around at Ls 20° and possibly at Ls 70° (red circles) in the surrounding terrain.

Research highlights:

- CO₂ ice basal sublimation is possible under certain conditions on Mars.
- Different CO₂ ice properties lead to different sublimation processes.
- Solar radiation alone can initiate basal sublimation in polar regions.

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