

Hafnium Silicate dielectrics fabricated by RF magnetron sputtering.

L. Khomenkova, X. Portier, P. Marie, F. Gourbilleau

▶ To cite this version:

L. Khomenkova, X. Portier, P. Marie, F. Gourbilleau. Hafnium Silicate dielectrics fabricated by RF magnetron sputtering. Journal of Non-Crystalline Solids, 2011, 357 (8-9), pp.1860-1865. 10.1016/j.jnoncrysol.2010.12.048. hal-00737872

HAL Id: hal-00737872

https://hal.science/hal-00737872

Submitted on 26 Jun 2018

HAL is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers.

L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.

Hafnium silicate dielectrics fabricated by RF magnetron sputtering

L. Khomenkova *,1, X. Portier, P. Marie, F. Gourbilleau

CIMAP, CEA/CNRS/ENSICAEN/UCBN, 6 Boulevard du Maréchal-Juin, F-14050 Caen Cedex, France

ABSTRACT

Keywords: High-k dielectrics; Hafnium oxide; Hafnium silicate; RF magnetron sputtering; Structural and composition properties of hafnium silicate layers fabricated by RF magnetron sputtering were studied by means of spectroscopic ellipsometry, X-ray diffraction, transmission electron microscopy and attenuated total reflection infrared spectroscopy with respect to the deposition parameters and post-deposition annealing treatment. The variation of the deposition conditions allows the temperature of amorphous-crystalline phase transformation of pure hafnium oxide layers to be controlled. It is shown that the silicon incorporation in oxide matrix prevents the formation of interfacial silicon oxide layer and plays a major role in the stability of the structure of hafnium based layers remaining an amorphous state upon annealing at 900–950 °C.

1. Introduction

High-k oxides are promising for future CMOS devices owing to wide band gap (higher than 5 eV), good thermal stability on silicon and high permittivity (more than 20), which are vital to overcome the problem of high leakage current [1–5]. Based on thermodynamic studies [6] and band gap measurements [7], hafnium oxide is considered as the best candidate among high-k dielectrics to replace silicon oxide and its oxynitride in CMOS devices. However, the microstructure, optic and electrical properties of ultrathin films depend strongly on the fabrication process, growth conditions and post deposition treatment.

The most widespread techniques for fabrication of ultrathin layers were ALD, CVD and MOCVD [5,8-11]. Unfortunately, lower-temperature methods such as physical vapor deposition [12] and magnetron sputtering [13-16] were not often addressed. It is worth to note that magnetron sputtered HfO₂ layers were fabricated by sputtering of pure metallic Hf cathode either in argon–oxygen gas mixture [13], or in pure argon plasma and then submitted to oxidation. Only in a few cases, the layers were deposited from a pure HfO₂ target [14-18].

The use of pure HfO_2 as a gate dielectric in the direct contact with Si substrate suffered from the undesirable formation of an interfacial SiO_2 layer during deposition or annealing treatment [1,2]. To overcome this problem, it was proposed to form Si oxynitride or

nitride barrier between high-k material and the Si substrate. Another drawback of pure HfO_2 is its low crystallization temperature (about 350–500 °C) accompanied usually by a formation of monoclinic phase. At the same time, amorphous films are advantageous because of their lower leakage current, a better homogeneity and, moreover, a higher reproducibility for the electrical properties in comparison with polycrystalline gate dielectric films. The improvement of thermal stability of amorphous HfO_2 phase was an issue of numerous investigations. It was shown, for example, in Refs. [13,19] that the incorporation of nitrogen into high-k matrix either during the deposition or post-deposition processing allows amorphous HfON films to be thermally stable up to 1100 °C.

The dielectric properties of HfO₂ depend significantly on its microstructure: both amorphous and monoclinic phases exhibit a dielectric constant of 16–20, but cubic and tetragonal phases have higher relative permittivity (about 28 [20]) and they are preferable in terms of device application. The formation of tetragonal and cubic phases requires an incorporation of cationic dopants in the HfO₂ matrix as it was demonstrated both theoretically [21–23] and experimentally [24–26]. Note that the dopants have to be carefully chosen to prevent a significant decrease of the band gap energy and dielectric constant of the material.

The most promising candidate to fulfill these conditions is silicon. Its incorporation into HfO_2 matrix results in the formation of pseudobinary alloys such as $(HfO_2)_x(SiO_2)_{1-x}$. A significant decrease of leakage current and an improvement of thermodynamic stability of this alloy were demonstrated in Refs. [1–3,27]. However, the permittivity of this material depends significantly on its composition and in most cases it was found to be less than 10, whereas higher permittivity values are required for microelectronic applications. Considering the above mentioned discussion, the effect of the silicon

^{*} Corresponding author. Permanent address: V.Lashkaryov Institute of Semiconductor Physics, NAS Ukraine, 45 Pr. Nauky, 03028 Kyiv, Ukraine. Tel.: +380 44 5257234. E-mail addresses: larysa.khomenkova@ensicaen.fr, khomenkova@rambler.ru (L. Khomenkova), fabrice.gourbilleau@ensicaen.fr (F. Gourbilleau).

¹ Tel.: +33 231452671; fax: +33 231452557.

incorporation on the properties of HfSiO dielectric materials requires further investigation.

In the present study, HfO_2 -based thin layers have been fabricated by RF magnetron sputtering of a pure HfO_2 or composed $HfO_2 + Si$ targets in pure argon plasma. The properties of the layers have been analyzed by means of X-ray diffraction, phase-modulated spectroscopic ellipsometry, infrared absorption spectroscopy and high resolution transmission electron microscopy with respect to the deposition conditions and the annealing treatments.

2. Experimental

The layers investigated were grown on B-doped (100) oriented Si substrates with a resistivity of 15 Ω cm. The substrates were cleaned in a diluted hydrofluoric solution (10%) to remove native oxide, dried in nitrogen flow and immediately placed into the vacuum chamber of the deposition setup.

Pure and Si-rich HfO_2 layers were grown by RF magnetron sputtering of 4 inches HfO_2 target (99.9%) in pure argon plasma at RF power density $RFP = 0.74 - 1.2 \text{ W/cm}^2$. The substrate temperature was kept at 45 and 100 °C, the total plasma pressure and substrate-cathode distance were fixed at 0.04 mbar and 57 mm, respectively. Si-rich layers were fabricated by the co-sputtering of HfO_2 target topped by Si chips with a surface ratio of $R_{Si} = 12\%$. The latter was estimated as a ratio of the surface of all Si chips to the total surface of HfO_2 target. To study the thermal stability of the layers, an annealing treatment in nitrogen flow in a conventional furnace at different temperatures $T_A = 600 - 1100 \, ^{\circ}\text{C}$ during $t_A = 15$ and 30 min was performed.

Several techniques were used to analyze the properties of the layers. The thickness of the films was analyzed by means of X-ray reflectometry and phase-modulated spectroscopic ellipsometry. Besides, the last technique was used to determine optical parameters of the films. The data were collected by means of a Jobin-Yvon ellipsometer (UVISEL) where the incident light was scanned in the range of 1.5-4.5 eV under an incident angle of 66.3°. The fitting of the experimental data was performed using DeltaPsi2 software [28]. The uncertainty of this method in the estimation of the refractive index is $\Delta n = \pm 0.01$. ATR-FTIR technique was used to control the chemical composition of the layers. The spectra were measured in the range of 600–4000 cm⁻¹ by means of a 60° Ge Smart Ark accessory inserted in a Nicolet Nexus spectrometer. X-ray diffraction analysis was performed using a Phillips XPERT HPD Pro device with a Cu K_{α} radiation ($\lambda = 0.154 \text{ nm}$) at a fixed grazing angle incidence of 0.5°. Cross-sectional specimens were prepared for TEM examination by the standard procedure involving grinding, dimpling and Ar⁺ ion beam thinning until electron transparency. The samples were observed by conventional (CTEM) and high resolution electron microscopy (HRTEM) using a FEG 2010 JEOL instrument, operated at 200 kV. Image processing was done with the commercial Digital micrograph GATAN software.

3. Results

3.1. Spectroscopic ellipsometry analysis

Spectroscopic ellipsometry does not require special environment and can be easily integrated into a semiconductor processing [29,30] as a fast, sensitive and non-destructive method for film characterization. The fitting of the experimental data is usually based on the effective medium approximation (EMA) [29,31]. There are several EMA models (Bruggeman, Maxwell-Garnet, etc.). As for all mean field theories, they fail to predict the properties of a multiphase medium close to the percolation threshold due to its long-range correlation length. They suppose that the macroscopic system is homogeneous (with correlation length below the Lorentz cavity radius) whereas the properties of multiphase system are determined by the properties of

its components and their relative fractions. The most considered parameters are conductivity and/or dielectric constant.

In the present work, Bruggeman effective medium approximation (BEMA) is used [31]. To investigate the properties of the as-deposited and annealed layers, a four-phase optical model is applied [18]. It consists of a silicon substrate, an interfacial SiO_2 layer, an amorphous HfO-based layer (pure or Si-rich), and a surface rough layer, composed of a mixture of void space and HfO_2 -based layer. To fit spectroscopic ellipsometry data, the dispersion law for n and k was chosen as described in Ref. [18]. It is based on the Forouhi–Bloomer model elaborated for amorphous semiconductor and insulating materials [32] using an improved parameterization [33]. The model structure and optical properties of the films were optimized by least-square refinement approach (χ^2) from a fitting of the experimental data. As a result, the thickness of HfO-based film (pure or Si-rich) and interfacial SiO_2 layer, the dispersion law of refractive index and absorption coefficient were obtained.

The evolution of the refractive index of pure HfO_2 and HfSiO films with the annealing treatment is presented in Fig. 1. As one can see, asdeposited pure HfO_2 film demonstrates an n value around 2.03 at 2-eV light energy. The annealing treatment shows a slight decrease of the refractive index down to n=1.97 (Fig. 1a, curve 1). Using the Clausius–Mossotti relationship between optical dielectric constant and film density, ρ , the latter was estimated with the relative uncertainty of \leq 5%. As for pure HfO_2 films, they were considered to be composed of HfO_6 and/or HfO_8 unit cells [20] and their density was found to be in the range of 8.0–8.8 g/cm³. It increased slightly for higher temperatures (Fig. 1b) confirming the data described in Ref. [34].

The as-deposited HfSiO films exhibit the higher n value in comparison with pure HfO_2 films, i.e. 2.12 against 2.03 (Fig. 1a) whereas n decreases upon annealing (Fig. 1a). Thus, HfSiO films annealed at $T_A \ge 800$ °C exhibit lower n values (n = 1.85) than their pure counterparts (n = 1.97). Besides, they demonstrate the decrease of the extinction coefficient, k, followed by a shift of the "absorption"

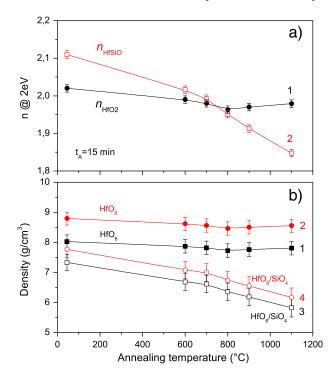


Fig. 1. The evolution of (a) the refractive index, n, and (b) the density of the films, ρ , with annealing temperature for pure HfO₂ and HfSiO films. The annealing time is 15 min. The density of the HfO₂ films is estimated for two unit cells as HfO₆ and HfO₈, the density of HfSiO films is presented considering two possible unit cells as HfO₆/SiO₄ and HfO₈/SiO₄.

edge" to higher energy side (not shown here). Since the optic model, proposed in the present study, allows the contributions of interfacial SiO₂ layer and HfO-based layer to be separated, this means that a decrease of *n* and *k* values observed for HfSiO upon annealing is more likely due to structural transformation of HfO-based layer itself rather than due to the increase of the contribution of interfacial SiO₂ one. In the case of pure HfO_2 films, the formation of voids in film volume can explain the decrease of the refractive index. It could be the same reason for the *n* evolution in the case of HfSiO films, but the formation of SiO₂ phase stimulated by high temperature annealing has also to be taken into account as a cause of the refractive index decrease. Moreover, as we showed earlier [16,17], an annealing at $T_A = 800 \,^{\circ}\text{C}$ for 15 min did not lead to the formation of any pores in the HfSiO volume. Consequently a formation of SiO₂ phase could be only the reason of the lowering of the refractive index upon annealing and this can explain the increasing of film transparency.

The density of the HfSiO films was also estimated based on the approach described in Ref. [18]. Similar to the case of pure HfO₂ films it was assumed that the HfO₂ phase is presented by HfO₆ and/or HfO₈ unit cells, whereas the SiO₂ phase consisted of SiO₄ unit cells. The results obtained, considering both possibilities, are presented in Fig. 1b. The film density decreases gradually from 7.3–7.8 g/cm³ down to 5.9–6.3 g/cm³. This can be explained by the SiO₂ phase formation for the films annealed at $T_A \le 800$ °C. However, annealing at higher temperatures could also cause pores formation. To discriminate the origin of the variation of the film density with annealing treatment, analyses of the chemical composition and structure of the films were made by ATR and XRD methods, whereas TEM study was performed to obtain information about the microstructure of the films.

3.2. ATR infrared spectra of the films

Among the nondestructive methods used to study thin films, ATR technique holds an important place, since it allows very thin layers to be analyzed. Unfortunately, referenced ATR data for HfSiO materials are not so numerous. The only known data are those regarding the monoclinic HfO_2 phase, while for tetragonal and cubic phases of HfO_2 and, moreover, for $HfSiO_4$ they are not available in the case of thin films. The interpretation of experimental data obtained for high-k silicate materials is usually based on the comparison of infrared spectra obtained for amorphous SiO_2 and HfO_2 , and their transformation due to an increase of the high-k material contribution. Information about the crystalline or amorphous nature of the layers can be extracted from the peak shape and the position of the band(s). Usually, Hf-O vibration bands are located in the range of 600-800 cm⁻¹ [36,37].

When crystalline HfO_2 phase is present in the layer, the well-defined peaks are detected at 770–780 cm⁻¹ and 675–685 cm⁻¹, the first peak being the signature of the monoclinic HfO_2 phase [36,37]. The observation of only one broad band with a maximum around 690–700 cm⁻¹ was usually attributed to the amorphous nature of the layers [16,17,36].

ATR experiments have been performed on the same pure and Si-rich layers described above. The presence of two main vibration bands in the range of $600-800~\rm cm^{-1}$ and $960-1260~\rm cm^{-1}$ was revealed for pure $\rm HfO_2$ layers (Fig. 2). The first band corresponds to the Hf–O vibrations, whereas the second one is due to Si–O vibrations. Its presence is caused by a $\rm SiO_x$ interfacial layer [1,2,4,8,16,17], whereas the shape and peak position depend on its thickness and composition [37].

It is known that continuous random network of SiO₂ consists of corner-coordinated SiO₄ tetrahedra and a disorder of amorphous structures comes from the changes of Si–O–Si bond angle. Usually, SiO₂ leads to four main peaks in the mid-infrared range coming mainly from motion of oxygen atoms. Under specific conditions several LO and TO phonons of Si–O bond can be detected in the 450–1350 cm⁻¹ spectral range: at 1076 cm⁻¹ (TO₃) and 1256 cm⁻¹ (LO₃), at 1160 cm⁻¹ (TO₄) and 1200 cm⁻¹ (LO₄), at 810 cm⁻¹ (TO₂) and 820 cm (LO₂) as

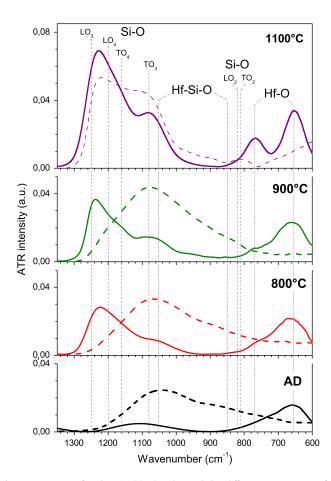


Fig. 2. ATR spectra of as-deposited (AD) and annealed at different temperatures HfO_2 (solid lines) and HfSiO (dashed lines) samples. Annealing temperatures are mentioned in the figure, annealing time is 15 min.

well as at 457 cm $^{-1}$ (TO₁)–507 cm $^{-1}$ (LO₁) [38,39]. The decrease of SiO₂ layer thickness to a nanometer scale is accompanied by a shift of the peak positions of all vibration bands to lower wavenumbers as well as by a decrease of the intensity of TO₄–LO₄ phonons that are usually considered as a feature of the disordering of SiO₂ matrix [39]. In the case of LO₃ and TO₃ phonons, this shift is about 25–30 cm $^{-1}$ [40]. However, the detection of LO₃ phonon is usually considered as an evidence of the formation of perfect Si/SiO₂ interface [41] and the spectral shift of LO₃ and TO₃ peak positions towards higher wavenumbers confirms the formation of a stoichiometric silicon oxide layer and increase of its thickness [17.37].

Annealing treatment of pure HfO₂ layers results in the different behaviours of Si-O and Hf-O vibration bands. Thus, initial broad Si-O band, centred at about 1100 cm⁻¹ (Fig. 2), shifts towards 1180 cm⁻¹ when $T_A = 800$ °C accompanied by an appearance of well-defined LO₃ and TO_3 phonon modes. Upon T_A increase, it results in a shift of both of them towards 1230 cm⁻¹ and 1070 cm⁻¹, respectively. This is due to the formation of stoichiometric SiO₂ interfacial layer caused by a diffusion of oxygen towards the film/substrate interface. This diffusion is usually considered as a diffusion of the oxygen from environment through the whole volume of high-k layer. However, it was shown for the case of sputtered layers [35], it occurs preferably from the HfO2 layer close to HfO2/Si interface. Since the films investigated were annealed in nitrogen atmosphere, the layer closed to HfO₂/Si interface could be enriched by hafnium caused the formation of oxygen deficient centres. The composition of asdeposited and annealed at T_A = 800 °C layers by energy filtered TEM analysis gave the evidence that HfO2 layer is stoichiometric (not shown here). However, the formation of oxygen-deficient centres

should occur at higher annealing temperatures due to the formation of monoclinic HfO_2 grains observed by ATR investigation. As one can see from Fig. 2, the evolution of the Hf-O vibration band is accompanied by the presence of a shoulder at about 770 cm⁻¹ transformed to a well-defined band at higher temperatures. Two separate bands at 770 cm⁻¹ and $680 \, \mathrm{cm}^{-1}$ detected for the layers annealed at $1100 \, ^{\circ}\mathrm{C}$ confirm the appearance of monoclinic HfO_2 phase.

The ATR spectra of as-deposited HfSiO films demonstrate featureless broad band peaked at 1050 cm⁻¹ with a shoulder at about 900 cm⁻¹ and a tail up to $600 \, \mathrm{cm}^{-1}$ (Fig. 2). This is the evidence of the formation of a HfSiO matrix. An annealing of the layers at T_A = 800 °C for 15 min leads to the increase of the intensity as well as to the slight narrowing and to the shift of the broad ATR peak from 1050 cm⁻¹ to 1070 cm⁻¹. On the contrary to pure HfO2 layers, the lower wavenumber part of the ATR spectrum does not change and is still featureless (Fig. 2). Further increase of the temperature up to $T_A = 900-1000$ °C results in the continuous increase of the peak intensity, shift of the peak position to $1120 \,\mathrm{cm}^{-1}$ and the appearance of a shoulder at about $1220 \,\mathrm{cm}^{-1}$. A treatment at $T_A = 1100$ °C causes the narrowing of this vibration band, a redistribution of the intensities of its components and an appearance of the band at about 820 cm⁻¹. All these three bands are ascribed to the SiO₂ phase [18]. The formation of an interfacial SiO₂ layer is confirmed by the well-defined LO₃ phonon whose intensity exceeds the TO₃ one (Fig. 2). At the same time, a comparable contribution of TO₃ and TO₄–LO₄ phonons to ATR spectrum enables us to assume that the increase of the thickness of interfacial SiO₂ layers as well as the formation of SiO₂ phase inside the film volume is more likely due to the phase separation of the HfSiO film on SiO₂ and HfO₂ phases. The latter is confirmed by an increase of the Hf-O vibrations' intensity in the range of 600-780 cm⁻¹ at $T_A \ge 1000$ °C. This broad and featureless band is usually ascribed to amorphous HfO₂ [11,12]. However, it is hardly believable that such high annealing temperature (1100 °C) would lead to an amorphous HfO2 phase.

To get more information about the structure of HfO_2 phase, we performed a XRD study of these samples. It is worth to note that the similarity of the ATR spectra obtained for as-deposited and annealed at $T_A \leq 900$ °C testifies the thermal stability of the HfSiO layers. The most probable reason explaining this phenomenon is the high flexibility of Si–O bonds that allows the amorphous nature of the materials to be kept (similar, for example, to fused silica). However, although HfSiO can be stable at such annealing temperatures, the slight transformation of Si–O vibration band can be due to the formation of a SiO₂ interfacial layer. The latter could occur at lower temperatures (less than 900 °C) than that corresponding to the phase separation in HfSiO film (1000–1100 °C). To clarify this issue, we performed TEM observations of cross-sections of the samples that will be presented below after the analysis of XRD data.

4. Discussion

4.1. XRD study of pure HfO2 and HfSiO layers

It is known that XRD patterns, obtained in asymmetric grazing geometry, allow studying the distribution of structural properties across thin layers. The monoclinic HfO₂ phase is usually revealed in XRD patterns by the most intense peaks in the range of $2\theta = 15-65^{\circ}$ at $2\theta = 28.37^{\circ}$ and 31.67° (for (111) planes), $2\theta = 34.34^{\circ}$, 34.66° and 35.55° (for (200) planes), and $2\theta = 49.59^{\circ}$, 50.49° and 50.91° (for (220) planes) [42]. The decrease of HfO₂ grain sizes results in the broadening of XRD peaks and for the case of amorphous structure only one broad peak in the range of $2\theta = 28-32^{\circ}$ is usually observed [14,15].

The as-deposited pure HfO_2 and HfSiO films show a broad peak in the range of $2\theta=25-35^\circ$ with a maximum intensity at about $2\theta=32^\circ$ (Fig. 3), which is consistent with the amorphous nature of both pure

and Si-rich films. The annealing of pure HfO_2 film at 800 °C results in the transformation of the shape of broad XRD peak and an appearance of two overlapped peaks at $2\theta = 28.5^{\circ}$ and $2\theta = 31.7^{\circ}$, that can be considered as a formation of monoclinic phase. This annealing does not lead to the crystallization of HfSiO films. At higher T_A , a formation of the HfO_2 monoclinic phase is evident by an increase of the observed peaks' intensity with their concomitant narrowing and an appearance of some peaks in the range of $2\theta = 40-65^{\circ}$ (Fig. 3). Annealing of HfSiO film at 1100 °C results in the appearance of the peaks at $2\theta = 30.3^{\circ}$, $2\theta = 35.3^{\circ}$ and $2\theta = 50.6^{\circ}$, that become to be more pronounced for longer annealing time. These peaks can be ascribed to the (111), (200) and (220) crystalline planes of the tetragonal HfO_2 phase. The peak at $2\theta = 59.8^{\circ}$ can be considered as an overlapping of the reflections from (311) and (222) planes of the same phase.

Usually, tetragonal HfO_2 phase is observed for ultrathin films, but annealing treatment at temperatures as high as 800 °C favors its transformation to monoclinic one [43]. However, in our case, longer annealing treatment (up to 30 min) at 1100 °C results in the increasing of the intensities of observed XRD peaks only and does not lead to an appearance of other peaks. This can be an evidence of thermal stability of the tetragonal HfO_2 phase in our samples. Since it has higher dielectric constant in comparison with that of monoclinic one, such a film can be used as a gate material for the structures that need high-temperature annealing process such as some nanomemory structures with embedded Si nanoparticles [44].

The analysis of optical properties of the HfSiO layers showed a decrease of their refractive index with annealing treatment. Regarding the structural properties, the phase separation is confirmed by the ATR spectra, whereas the crystallization of HfO_2 phase is demonstrated by the XRD data. Since an increase of n is expected with the HfO_2

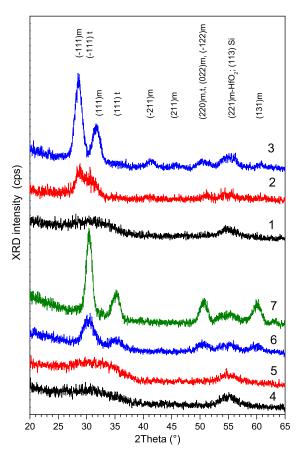


Fig. 3. XRD patterns measured for pure HfO_2 (1–3) and HfSiO (4–7) layers: as-deposited (1, 4) and annealed at T_A = 800 °C (2, 5) and 1100 °C (3, 6, 7) for 15 min (1–6) and 30 min (7).

crystallization, we concluded that the refractive index of HfSiO films decreases continuously due to an increasing contribution of the SiO₂ phase. At the same time, neither ATR nor XRD studies can reveal unambiguously where the SiO2 phase is located, i.e. at film/substrate interface or embedded in the film volume. However, based on these results and on the estimate of the film density, we can admit the presence of SiO₂ phase inside the volume of HfSiO layer. The latter will be confirmed by TEM observations of as-deposited and annealed films.

4.2. TEM observation

TEM images of cross-sections of as-deposited and annealed samples are shown in Fig. 4a. As-deposited samples are homogeneous, smooth enough (roughness does not exceed 1 nm) and do not have thick interfacial layer (the thickness being lower than 1 nm). These observations corroborate the results obtained by fitting of ellipsometry spectra. The increase of the annealing temperature up to 900 °C leads to the increase of total film thickness. However, this occurs not only due to the increase of the interfacial SiO₂ layer thickness, but also due to the expansion of HfSiO film itself (Fig. 4b). The appearance of bright and dark alternated lines parallel to the film/substrate interface is seen. This can be explained by the Hf segregation process, whereas the film remains amorphous (Fig. 4b).

Further increase of annealing temperature up to 1100 °C results in the formation of a thicker SiO₂ interfacial layer (up to 5 nm) as well as a higher contrast between bright and dark regions (Fig. 4b,c). They correspond to SiO₂-rich and HfO₂-rich phases that are an evidence of the HfSiO phase separation. Besides, the crystallites appear in the dark regions as demonstrated by Fig. 4c. The formation and crystallization

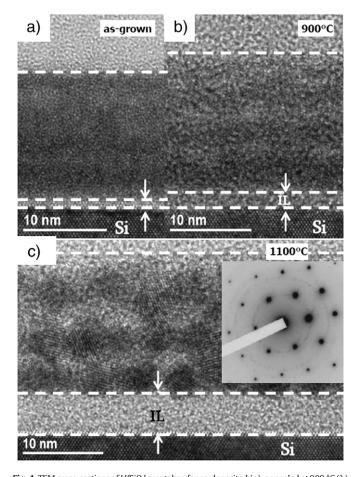


Fig. 4. TEM cross-sections of HfSiO layer taken for as-deposited (a), annealed at 900 °C (b) and 1100 °C (c) in nitrogen flow for 15 min. Electron diffraction pattern is shown for the sample annealed at 1100 °C. IL is interfacial SiO2 layer.

of HfO₂ phase usually leads to the increase of the film density. At the same time the appearance of SiO₂ phase results in the decrease of the film density and can be considered as a main reason of the decrease of the total density of the HfSiO film upon annealing.

It is known that thick pure HfO₂ has usually the monoclinic phase, whereas tetragonal and cubic phases appear at 1670-2200 °C [35]. When the layer thickness decreases down to a nanometer scale, the presence of the tetragonal phase can be observed in pure HfO₂ layers [36]. However, it transforms easily to the monoclinic one at moderate temperatures [36]. The analysis of crystallized HfO2 phase of HfSiO films was also performed. The inset in Fig. 4c presents a typical selected area electron diffraction (SAED) pattern corresponding to the film and a part of the [110] oriented silicon substrate. Apart from the well defined dots, rising from the silicon substrate, four weak rings appeared. The lattice plane distances estimated from SAED were 2.95 Å, 1.80 Å and 1.53 Å. These distances are compatible with lattice planes of the HfO₂ tetragonal phase: $d_{[111]} = 2.95 \text{ Å}$, $d_{[220]} = 1.82 \text{ Å}$. and $d_{[311]} = 1.59$ Å. Thus, TEM observations of HfSiO films show that the tetragonal phase is stable in such films upon annealing at 1100 °C, confirming the XRD study.

5. Conclusion

The effect of annealing treatment on the properties of HfO₂-based layers has been investigated. It is demonstrated that silicon incorporation into the HfO₂ matrix leads to a decrease in the thickness of interfacial layer down to 1 nm and improves the thermal stability of the layers upon annealing at 900–950 °C. The film density was estimated with respect to annealing temperature. Its decrease with temperature is explained by a phase separation of HfSiO on HfO2 and SiO2 phases over its volume and the increase of the contribution of the latter with annealing. The formation of the HfO₂ tetragonal phase was also revealed. The latter was found to be stable offering a possibility to use such HfSiO materials for microelectronic applications, for example, for nanomemory structures with embedded Si nanoparticles requiring high-temperature annealing. The formation of the Si-rich/Hf-rich multilayer structure upon annealing treatment was observed.

Acknowledgments

This work is supported by the French National Research Agency (ANR) through Nanoscience and Nanotechnology Program (NOMAD Project n°ANR-07-NANO-022-02) and by the Conseil Régional de Basse Normandie through the CPER project — Nanoscience axe (2007–2013).

References

- [1] G.D. Wilk, R.M. Wallace, J.M. Anthony, J. Appl. Phys. 87 (2000) 484.
- G.D. Wilk, R.M. Wallace, J.M. Anthony, J. Appl. Phys. 89 (2001) 5243.
- [3] C. Krug, G. Lucovsky, J. Vac. Sci. Technol. A. 22 (2004) 1301. [4] M. Houssa, L. Pantisano, R. Degraeve, T. Schram, G. Pourtois, S. De Gendt, G. Groeseneken, M.M. Heyns, Mat. Sci. Eng. R 51 (2006) 37.
- G. Pant, A. Gnade, M.J. Kim, R.M. Wallace, B.E. Gnade, M.A. Quevedo-Lopez, P.D. Kirsch, A. Gnade, S. Krishnan, Appl. Phys. Lett. 89 (2006) 032904.
- M. Gutowski, J.E. Jaffe, Ch.-L. Liu, M. Stoker, R.I. Hegde, R.S. Rai, P.J. Tobin, Appl. Phys. Lett. 80 (2002) 1897.
- [7] H. Kato, T. Nango, T. Miyagawa, T. Katagiri, K.S. Seol, Y. Ohki, Appl. Phys. Lett. 92 (2002) 1106.
- J.-H. Kim, V.A. Ignatova, M. Weisheit, Microelectron. Eng. 86 (2009) 357.
- S.K. Dey, A. Das, M. Tsai, D. Gu, M. Flyod, R.W. Carpenter, H. De Waard, C. Werkhoven, S. Marcus, J. Appl. Phys. 95 (2004) 5042. [10] B.K. Park, J. Park, M. Cho, C.S. Hwang, K. Oh, Y. Han, D.Y. Yang, Appl. Phys. Lett. 80
- (2002) 2368. [11] M. Lui, L.Q. Zhu, G. He, Z.M. Wang, J.X. Wu, J.-Y. Zhang, I. Liaw, Q. Fang, I.W. Boyd,
- Appl. Surf. Sci. 253 (2007) 7869.
- [12] K. Yamamoto, S. Hayashi, M. Kubota, M. Niwa, Appl. Phys. Lett. 81 (2002) 2053.
- G. He, Q. Fang, L.D. Zang, Mat. Sci. Semicond. Proc. 9 (2006) 870.
- L. Pereira, A. Marques, H. Águas, N. Nedev, S. Georgiev, E. Fortunato, R. Martins, Mat. Sci. Eng. B 109 (2004) 89.
- [15] L.-P. Feng, Z.-T. Liu, Y.-M. Shen, Vacuum 83 (2009) 902.
- [16] L. Khomenkova, C. Dufour, P.-E. Coulon, C. Bonafos, F. Gourbilleau, Mater. Res. Soc. Symp. Proc. 1160 (2009) 69-72.

- [17] L. Khomenkova, C. Dufour, P.-E. Coulon, C. Bonafos, F. Gourbilleau, Nanotechnology 21 (2010) 095704.
- L. Khomenkova, J. Cardin, X. Portier, F. Gourbilleau, Nanotechnology 21 (2010)
- [19] M.R. Visokay, J.J. Chambers, A.L.P. Rotondaro, A. Shanware, Appl. Phys. Lett. 80 (2002) 3183.
- [20] G.-M. Rignanese, J. Phys. Condens. Matter 17 (2005) R357.
- [21] D. Fischer, A. Kersch, Appl. Phys. Lett. 92 (2008) 012908.
 [22] Ch.-K. Lee, E. Cho, H.-S. Lee, Ch.S. Hwang, S. Han, J. Phys. Condens. Matter 78 (2008) 012102.
- [23] E. Cockayne, J. Appl. Phys. 103 (2008) 084103.
- [24] D. Tsoutsou, G. Apostolopoulos, S.F. Galata, P. Tsipas, A. Sotiropoulos, G. Mavrou, Y. Panayiotatos, A. Dimoulas, A. Lagoyannis, A.G. Karydas, V. Kantarelou, S. Harissopoulos, J. Appl. Phys. 106 (2009) 024107.
 [25] P. Majumder, G. Jursich, C. Takoudis, J. Appl. Phys. 105 (2009) 104106.
- [26] K. Tomida, K. Kita, A. Turiumi, Appl. Phys. Lett. 89 (2006) 142902.
- [27] R. Beyers, J. Appl. Phys. 56 (1984) 147.
- [28] http://www.horiba.com/scientific/products/ellipsometers/software/.
- [29] S. Charvet, R. Madelon, F. Gourbilleau, R. Rizk, J. Appl. Phys. 85 (1999) 4032.

- [30] O. Buiu, W. Davey, Y. Lu, I.Z. Mitorvic, S. Hall, Thin Solid Films 517 (2008) 453.[31] D.A.G. Bruggeman, Ann. Phys. 416 (1935) 636.
- A.R. Forouhi, I. Bloomer, Phys. Rev. B 34 (1986) 7018.
- G.E. Jelisson Jr., F.A. Modine, Appl. Phys. Lett. 69 (1996) 371.
- [34] T. Tan, Zh. Liu, H. Lu, W. Liu, H. Tian, Opt. Mater. 32 (2010) 432.
- [35] R. Jiang, E.Q. Xie, Z.F. Wang, J. Mater. Sci. 42 (2007) 7343.
 [36] N.V. Nguyen, A.V. Davydov, D. Chandler-Horowitz, M.F. Frank, Appl. Phys. Lett. 87 (2005) 192903.
- M.M. Frank, S. Sayan, S. Dörmann, T.J. Emge, L.S. Wielunski, E. Garfunkel, Y.J. Chabal, Mat. Sci. Eng. B 109 (2004) 6.
- C.T. Kirk, Phys. Rev. B 38 (1988) 1255.
- P. Lange, J. Appl. Phys. 66 (1989) 201.
- [40] H. Ono, T. Ikarashi, K. Ando, T. Kitano, J. Appl. Phys. 84 (1998) 6064.
- [41] J.E. Olsen, F. Shimura, J. Appl. Phys. 66 (1989) 1353.
- JCPDS No.78-0050.
- [43] J. Tang, F. Zhang, P. Zoogman, J. Fabbri, S.-W. Chan, P. Zhu, L.E. Brus, M.L. Steigerwald, Adv. Funct. Mater. 15 (2005) 1595.
- [44] M. Perego, G. Seguini, C. Wiemer, M. Fanciulli, P.-E. Coulon, C. Bonafos, Nanotechnology 21 (2010) 055606.