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Crystal structure and physical properties of the new ternary compounds Nd₃Pt₂₃Si₁₁ and Pr₃Pt₂₃Si₁₁

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Abstract

The two ternary compounds, $Pr_3Pt_{23}Si_{11}$ and $Nd_3Pt_{23}Si_{11}$ have been synthesized for the first time. They both crystallize in the same Fm-3m cubic space group as $Ce_3Pt_{23}Si_{11}$. Their lattice parameters are $a_{Pr} = 16.8633(3)$ Å and $a_{Nd} = 16.8498(4)$ Å for $Pr_3Pt_{23}Si_{11}$ and $Nd_3Pt_{23}Si_{11}$, respectively. From magnetic and heat capacity measurements, $Nd_3Pt_{23}Si_{11}$ is found to present a magnetic transition at 1.6 K, whereas for $Pr_3Pt_{23}Si_{11}$, the magnetic susceptibility shows typical paramagnetic Van Vleck behavior at low temperatures.

Keywords: Rare earth compounds, Crystal structure determination, Magnetic properties, Heat capacity.

1. Introduction

The ternary system RE-Pt-Si (where RE is a rare earth element) has triggered a lot of interest in the last decade because some of its compounds present interesting physical properties such as unconventional superconducting properties, heavy-fermion states or magnetic ordering. The best known compound in this system is CePt₃Si [1]. Very recent studies of the novel RE₃Pt₂₃Si₁₁ series have shown that the ferromagnetic Ce₃Pt₂₃Si₁₁ compound (T_C = 440 mK) presents unusual magnetic behavior [2, 3] and that some features in Yb₃Pt₂₃Si₁₁ suggest an unstable character of the Yb 4*f* shell [4]. As part of our studies on this ternary Re₃Pt₂₃Si₁₁ series, we have successfully synthesized two novel compounds: Pr₃Pt₂₃Si₁₁ and Nd₃Pt₂₃Si₁₁. In this paper, we present a study of the structural and physical properties performed on polycrystalline samples of Pr₃Pt₂₃Si₁₁ and Nd₃Pt₂₃Si₁₁. These properties are compared to those of Ce₃Pt₂₃Si₁₁ and to the non-magnetic isomorphous compound La₃Pt₂₃Si₁₁.

2. Sample development and experimental details

High quality polycrystalline samples of $Pr_3Pt_{23}Si_{11}$ and $Nd_3Pt_{23}Si_{11}$ have been prepared in an induction furnace. Stoichiometric proportions of the different constituents: Pr or Nd (99.9%, Johnson Matthey), Pt (99.95%, Alfa Aesar) and Si (99.9999%, Alfa Aesar), were melted in a cold copper crucible under a high purity argon atmosphere. The samples were melted several times to improve the homogeneity. Mass losses during this first step were less than 0.1%. Sample structure and quality were checked by conventional X-ray powder diffraction (Cu-K α radiation on a Philipps PW1730 diffractometer).

In the temperature range 1.8 - 300 K, magnetization measurements were performed using two magnetometers based on the extraction method: a commercial Quantum Design MPMS magnetometer and a in-house experimental setup allowing measurements in fields of up to 10.5 T. In the temperature range 0.38 - 300 K, the heat capacity was measured using the relaxation method on a Quantum Design-Physical Property Measurement System (QD-PPMS).

3. Determination of the Crystal structure

Figure 1 shows the powder diffraction patterns collected for Pr₃Pt₂₃Si₁₁ and Nd₃Pt₂₃Si₁₁, together with the Rietveld refinements carried out using the FullProf program [5]. For both compounds, the results are consistent with the face-centered cubic structure (Fm-3m space group) already reported for La-, Ce- and Yb₃Pt₂₃Si₁₁ [6, 3, 4]. Details of the crystallographic analyses are presented in tables 1, 2 and 3. Pr³⁺ and Nd³⁺ ions are located on the 24*d* sites. The values of these refined lattice parameters are 16.8633(3) Å and 16.8498(4) Å for Pr₃Pt₂₃Si₁₁ and Nd₃Pt₂₃Si₁₁, respectively. As seen in figure 2, these lattice parameters, supplemented by other known compounds of the series, show behavior consistent with the volume contraction in the lanthanide series. Nevertheless, like in many lanthanide compounds, this behavior is not a smooth linear variation.

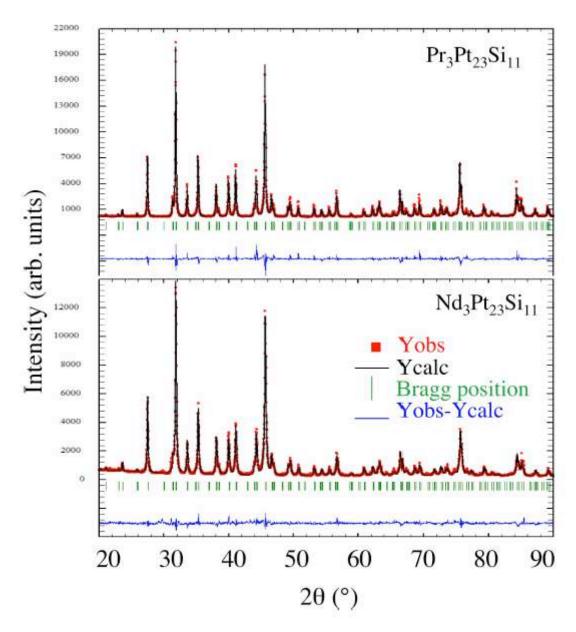


Figure 1: X-ray powder diffraction patterns collected for $Pr_3Pt_{23}Si_{11}$ and $Nd_3Pt_{23}Si_{11}$. The X-ray wavelengths are λ CuK α_1 = 1.54051 Å and λ CuK α_2 = 1.54433 Å. Full squares represent the observed intensities while the solid lines represent the best fit calculated from a Rietveld refinement. The vertical bars show the ideal Bragg positions while the bottom curves shows the difference between the observed and the calculated intensities.

 Table 1: Crystallographic details.

Formula from refinements	$Pr_3Pt_{23}Si_{11}$	$Nd_3Pt_{23}Si_{11}$
Space group	Fm-3m (No 225)	Fm-3m (No 225)
Structure type	$Ce_3Pt_{23}Si_{11}$	$Ce_3Pt_{23}Si_{11}$
Z	8	8
Lattice constant (Å)	16.8633(3)	16.8498(4)
Cell volume (Å ³)	4795.4(2)	4783.9(2)
$D_{cal} (g/cm^3)$	14.46	14.52
Reflection measured	284	272
Number of refined parameters	30	30
χ^2	4.58	7.96

Table 2: Atomic coordinates and isotropic displacement parameters for $Pr_3Pt_{23}Si_{11}$.

Atom	Wyckoff position	x/a	y/b	z/c	$U_{iso}(A^2)$	Occupation
Des	244	0	1/4	1/4	1.2	1
Pr	24 <i>d</i>	0	1/4	1/4	1.3	1
Pt ₁	32 <i>f</i>	0.0830(1)	0.0830(1)	0.0830(1)	1.2	1
Pt ₂	32 <i>f</i>	0.3089(1)	0.3089(1)	0.3089(1)	1.1	1
Pt ₃	24 <i>e</i>	0.3741(2)	0	0	0.8	1
Pt ₄	96 <i>k</i>	0.0843(1)	0.0834(1)	0.2523(1)	1.2	1
Siı	24 <i>e</i>	0.176(1)	0	0	1.7	1
Si ₂	32 <i>f</i>	0.1682(6)	0.1682(6)	0.1682(6)	1.7	1
Si ₃	32 <i>f</i>	0.3929(6)	0.3929(6)	0.3929(6)	1.7	1

Table 3: Atomic coordinates and isotropic displacement parameters for Nd₃Pt₂₃Si₁₁.

Atom	Wyckoff position	x/a	y/b	z/c	Uiso (Å ²)	Occupation
Nd	24 <i>d</i>	0	1/4	1/4	1.0	1
Pt1	32 <i>f</i>	0.0830(2)	0.0830(2)	0.0830(2)	1.2	1
Pt2	32 <i>f</i>	0.3079(1)	0.3079(1)	0.3079(1)	1.3	1
Pt3	24e	0.3744(3)	0	0	0.9	1
Pt4	96 <i>k</i>	0.0839(1)	0.0839(1)	0.2504(2)	1.7	1
Si1	24e	0.1789(9)	0	0	1.6	1
Si2	32 <i>f</i>	0.1676(9)	0.1676(9)	0.1676(9)	1.6	1
Si3	32 <i>f</i>	0.3878(9)	0.3878(9)	0.3878(9)	1.6	1

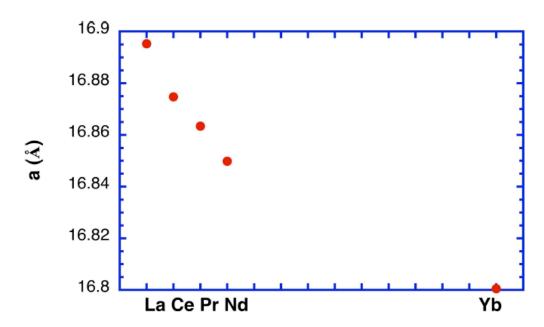


Figure 2: Lattice parameter variation in the $RE_3Pt_{23}Si_{11}$ series. The Yb data comes from reference [4].

4. Magnetic properties

4.1. Pr₃Pt₂₃Si₁₁

Figure 3 shows the thermal variation of the inverse of the magnetic susceptibility of Pr₃Pt₂₃Si₁₁. Dots represent the inverse susceptibility deduced from the thermal variation of the magnetization under an applied field of 0.1 T, χ=M/H. At low temperatures, the open diamonds represent the inverse susceptibility obtained from the Arrott plots, M²=f(H/M) or M/H=f(H²). The magnetization curves shown in the inset of figure 3, show that down to 1.8 K, the magnetization varies linearly with the field and that under an applied field of 5 T, its value is far lower than the 3.2 $\mu_{\rm B}$ expected for the saturated magnetic moment of the ${\rm Pr}^{3+}$ ion. This is in agreement with the paramagnetic behavior of the Pr^{3+} ions. In the temperature range 50-300 K, $1/\chi$ varies linearly with the temperature, consistent with the Curie-Weiss law. From the linear fit of the experimental data, one deduces an effective moment of 3.55 µ_B per Pr ion, in good agreement with the theoretical value, $\mu_{eff} = 3.58 \mu_B$. The solid line in Fig. 3 has been calculated using the Curie-Weiss law: $1/\chi = T/C - n$, where C is the theoretical Curie constant for a Pr^{3+} ion, C=2.865 μ_B K/T ($\mu_{\rm eff} = 3.58 \mu_{\rm B}$) and n is the molecular field coefficient. The experimental data are well reproduced for $n = -4.8 \text{ T/}\mu_B$. A negative value of n suggests antiferromagnetic exchange interactions. We note however, that at the highest temperatures, the experimental data tends to slightly deviate from the Curie-Weiss line. Below 50 K, 1/χ deviates from the Curie-Weiss law and tends to a constant value $1/\chi$ (T = 0 K) of 11.81 T/ μ _B per Pr. This Van Vleck-type behavior is a direct consequence of the crystal electric field (CEF) effects that can select a non-magnetic ground state for a non-Kramers trivalent ion located in a site of cubic symmetry. In the present case, one would expect either a singlet $(\Gamma 1)$ or a triplet $(\Gamma 3)$ non-magnetic ground state [7].

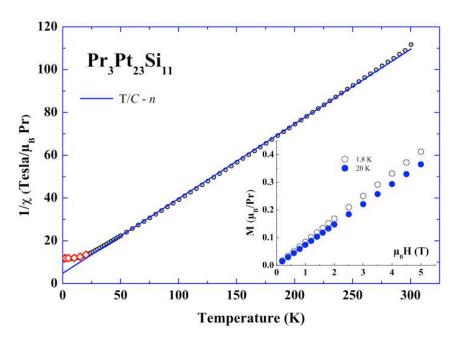


Figure 3: Thermal variation of the inverse magnetic susceptibility of Pr₃Pt₂₃Si₁₁: open dots are deduced from the thermal variation of the magnetization under an applied field of 0.1 T while the open diamonds are from the Arrott plots. The line shows the calculated Curie-Weiss law: $1/\chi$ = T/C - n, with C=2.865 μ _B K/T and n= - 4.8 T/ μ _B. The inset shows the field dependent magnetization curves obtained at 1.8 and 20 K.

4.2. Nd₃Pt₂₃Si₁₁

Figure 4 shows the thermal variation of the inverse magnetic susceptibility of $Nd_3Pt_{23}Si_{11}$ deduced from Arrott plots. The magnetization curves shown in the inset of figure 4 confirm that $Nd_3Pt_{23}Si_{11}$ remains paramagnetic down to the lowest attainable temperature, 1.8 K. However, in contrast to $Pr_3Pt_{23}Si_{11}$, the magnetization of the Nd compound tends to saturate. Under an applied field of 8 T, the magnetization tends to $\approx 1.45~\mu_B$ per Nd ion.

The thermal variation of the inverse susceptibility presents strong similarities with that of the Pr compound. $1/\chi$ varies linearly with temperature down to approximately 25 K, where it shows a downward curvature. A linear fit of the experimental data yields an effective moment of 3.53 μ_B per Nd ion, slightly smaller than the theoretical value $\mu_{eff} = 3.62 \ \mu_B$, but nevertheless, in agreement with trivalent Nd ions. The line in Fig. 4 shows the calculated Curie-Weiss law using the theoretical Curie constant, C=2.93 μ_B K/T and a value of \approx -4.8 T/ μ_B for the molecular field coefficient. At high temperatures, the experimental data deviate clearly from the calculated

Curie-Weiss line. In diluted rare earth systems, the paramagnetic signal is very weak at high temperatures and may become the same order of magnitude as the magnetic signal of the matrix. This signal, which results from several contributions such as the Pauli paramagnetism of conduction electrons, the Landau diamagnetism of these electrons and the magnetic signal of the allied atoms, can be well accounted for by the signal of the isomorphous La compound. In the present series, the La₃Pt₂₃Si₁₁ compound has been found to be diamagnetic [3]. This may explain the reduction of the experimental susceptibility with respect to the Curie-Weiss law.

Below 25 K, $1/\chi$ presents a downward curvature and tends to zero as T tends to 0 K. As for $Pr_3Pt_{23}Si_{11}$, this deviation from the Curie-Weiss law of $1/\chi$ is ascribed to CEF effects. This is also confirmed by the value at which the magnetization saturates; 1.45 μ_B instead of 3.27 μ_B expected for the $^4I_{9/2}$ ground state multiplet. As Nd³⁺ is a Kramers ion, the CEF levels are always magnetic.

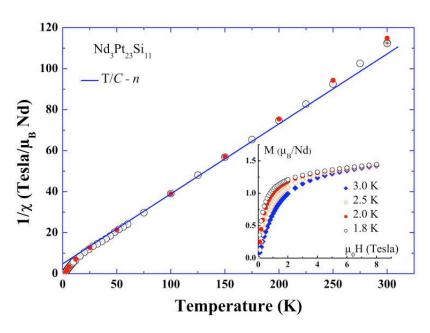


Figure 4: Thermal variation of the inverse magnetic susceptibility of Nd₃Pt₂₃Si₁₁. The experimental data have all been deduced from Arrott plots: open dots come from the measurements performed with the in-house experimental setup and the full dots from the measurements in the Quantum Design MPMS. The line shows the calculated Curie-Weiss law: $1/\chi = T/C - n$, with $C = 2.93~\mu_B~K/T$ and $n = -4.8~T/\mu_B$. The field dependent magnetization curves between 1.8 and 3 K are shown in the inset.

5. Heat capacity

In figure 5, the specific heat curves of the Pr₃Pt₂₃Si₁₁, Nd₃Pt₂₃Si₁₁ and La₃Pt₂₃Si₁₁ compounds are compared in temperature range 0.37 - 30 K. For the Pr compound, the specific heat measurements show that no transition exists down to 0.37 K. For the Nd compound on the other hand, the specific heat shows a lambda anomaly at 1.6 K. According to the magnetization curves, this most likely corresponds to a magnetic transition. To determine the nature of this transition, magnetization measurements below 1.6 K are required. In the inset of Fig. 5, the logarithmic scale emphasizes the broadness of the lambda anomaly at the transition. A non-negligible signal extends up to 3 K. Such a feature can be ascribed to short-range magnetic correlations that may develop well above the temperature at which the long-range magnetic order stabilizes. Magnetic results have shown that the CEF effects are important in Pr₃Pt₂₃Si₁₁ and Nd₃Pt₂₃Si₁₁. A tentative determination of the entropy from the experimental specific heat at the transition yields a value very close to 3Rln2 for Nd₃Pt₂₃Si₁₁ (see figure 6). Such a result is in fair agreement with a magnetic doublet (Γ 6) CEF ground state [7]. Actually, the value of 3Rln2 is only reached around 4 K, this correlates with the broadness of the lambda anomaly (see Fig. 5) and is consistent with the existence of short-range correlations. Fig. 6 also shows that the entropy of Pr₃Pt₂₃Si₁₁ remains negligible up to 5-6 K. This is consistent with a singlet Γ_1 CEF ground state in this compound. At higher temperatures, it is very probable that the excess of specific heat for the Pr and Nd compounds, compared to the La compound, is due to the CEF Schottky contribution.

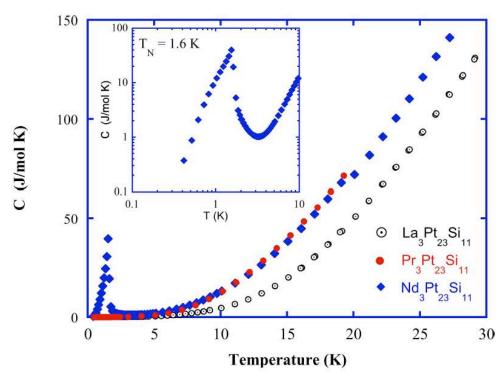


Figure 5: Comparison of the specific heat curves of $Pr_3Pt_{23}Si_{11}$, $Nd_3Pt_{23}Si_{11}$ and the non-magnetic isomorphous compound $La_3Pt_{23}Si_{11}$. The inset shows the magnetic transition of $Nd_3Pt_{23}Si_{11}$ at 1.6 K (logarithmic scales).

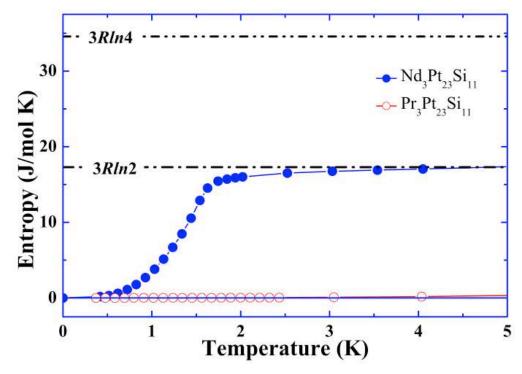


Figure 6: Thermal variation of the entropy per mole for $Pr_3Pt_{23}Si_{11}$ (open dots) and $Nd_3Pt_{23}Si_{11}$ (solid dots).

6. Conclusion

The two ternary compounds, Pr₃Pt₂₃Si₁₁ and Nd₃Pt₂₃Si₁₁, have been synthesized for the first time. The crystallographic studies have confirmed that both compounds crystallize in the same Fm-3m cubic space group as Ce₃Pt₂₃Si₁₁, La₃Pt₂₃Si₁₁ and Yb₃Pt₂₃Si₁₁. In this series, one observes a reduction of the lattice parameter with the increase of the rare earth atomic number, in agreement with the contraction of the lanthanides. Magnetic and heat capacity measurements reveal that the crystal electric field effects are important in the Pr and Nd compounds. For instance in Pr₃Pt₂₃Si₁₁, the CEF selects a non magnetic ground state leading to a Van Vleck paramagnetic behavior at low temperatures, while in Nd₃Pt₂₃Si₁₁, it strongly reduces the value of the magnetic moment. For this last compound a transition, ferromagnetic or antiferromagnetic, is observed at 1.6 K in the specific heat data.

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References

- [1] E. Bauer, H. Kaldarar, A. Prokofiev, E. Royanian, A. Amato, J. Sereni, W. Bramer-Escamilla, I. Bonalde, J. Phys. Soc. Jpn. 76 (5) (2007) 051009.
- [2] C. Opagiste, C. Paulsen, E. Lhotel, P. Rodiere, R.-M. Galera, P. Bordet, P. Lejay, J. Magn. Magn. Mater. 321 (6) (2009) 613 618.
- [3] C. Opagiste, R.-M. Galéra, M. Amara, C. Paulsen, S. Rols, B. Ouladdiaf, Phys. Rev. B 84 (13) (2011) 134401.
- [4] D. Kaczorowski, A. Gribanov, S. Safronov, P. Rogl, Y. Seropegin, J. Alloys Compd. 509 (37) (2011) 8987-8990.
- [5] J. Rodriguez-Carvajal, Satellite Meeting on Powder Diffraction of the XV Congress of the IUCr, Book of Abstracts, Toulouse, France, 1990, p. 127.
- [6] A. I. Tursina, A. V. Gribanov, Y. D. Seropegin, K. V. Kuyukov, O. I. Bodak, J. Alloys Compd. 347 (1-2) (2002) 121 123.
- [7] K. Lea, M. Leask, W. Wolf, J. Phys. Chem. Solids 23 (10) (1962) 1381 1405.