

Comparison of extraction conditions for milk and hen's egg allergens

Mirva Steinhoff, Markus Fischer, Angelika Paschke

► To cite this version:

Mirva Steinhoff, Markus Fischer, Angelika Paschke. Comparison of extraction conditions for milk and hen's egg allergens. Food Additives and Contaminants, 2011, pp.1. 10.1080/19440049.2010.545957. hal-00670743

HAL Id: hal-00670743 https://hal.science/hal-00670743

Submitted on 16 Feb 2012

HAL is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers. L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.

Food Additives and Contaminants



Comparison of extraction conditions for milk and hen's egg allergens

Journal:	Food Additives and Contaminants
Manuscript ID:	TFAC-2010-207.R1
Manuscript Type:	Original Research Paper
Date Submitted by the Author:	25-Nov-2010
Complete List of Authors:	Steinhoff, Mirva; Food Chemistry, Chemistry, University of Hamburg Fischer, Markus; Food Chemistry, Chemistry, University of Hamburg Paschke, Angelika; University of Hamburg, Chemistry, Food Chemistry
Methods/Techniques:	Extraction, Immunoassays
Additives/Contaminants:	Allergens, Process contaminants
Food Types:	Bakery products, Eggs, Milk



http://mc.manuscriptcentral.com/tfac Email: fac@tandf.co.uk

Comparison of extraction conditions for milk and hen's egg allergens

MIRVA STEINHOFF, MARKUS FISCHER, ANGELIKA PASCHKE-KRATZIN¹

University of Hamburg, Institute of Food Chemistry, Grindelallee 117, 20146 Germany;

8 Abstract

The evaluation of recovery rates by extracting dried milk and egg powder using eleven different extractants gave approximately similar results for both foods. Compared to the other extraction solutions investigated, "1% Tween 20 and 0.4% Triton X-100" and "4% SDS" are the most suitable extractants to isolate proteins of hen's egg or milk. Comparing calculated protein recovery rates of egg and milk powder extracts the results clearly indicated that the choice of a suitable extractant is of particular importance. Qualitative investigation of the extracts via LDS - PAGE followed by silver staining as well as immunoblotting confirmed the results of protein quantification. Hence, the immunoblots showed that the extraction agents had no negative influence on the antigenicity of the extracted allergenic proteins. In this study, variation of extraction temperature neither led to any benefit in extraction quality nor to degradation. Changing pH did not reveal any trends, but progressive protein hydrolysis under strong alkaline conditions. Evaluation of recovery rates as well as results of unspecific and specific staining of the extracts showed that an extraction time of one hour is sufficient for an appropriate sample preparation. For investigations with and without food matrix different results were obtained. In summary, wheat starch did not influence the extraction quality within all examined materials and different extractants. In contrast, using fat powder and dry cake mix, respectively, led to different results in the extraction procedure. When fat powder and dry cake mix were used as food matrices, some protein recovery rates decreased and

¹ angelika.paschke@uni-hamburg.de

some increased depending on the allergen material. These results highlight the fact that the
suitability of the extractant not only depends on the properties of the allergen but furthermore
on the type of matrix containing the allergen.

Keywords: Food allergy, hidden allergens, extraction of allergens, hen's egg, cow's milk

35 Introduction

Cow's milk and hen's egg as well as their products belong to eight food or food groups which are responsible for about 90 % of allergic reactions (Hefle and Taylor, 2004). The consumption of small amounts can already cause life-threatening or lethal reactions (Vieluf et al., 2002). Nevertheless, allergic persons can unintentionally consume allergenic material if these are unexpectedly part of a food product e.g. because of food contamination / cross contact within the manufacturing process (Hefle and Taylor, 2004). The threshold doses for triggering symptoms after ingestion of the offending food are strongly dependent on the patient's individual susceptibility and the allergic potential of the particular food. Clinical data shows that exposure to even about 1 mg of the allergen can elicit adverse reactions up to anaphylactic shocks in individuals with IgE-mediated food allergies (Besler et al., 2001; Moneret-Vautrin and Kanny, 2004). The legislation on food labelling only concerns allergenic ingredients that are knowingly and deliberately added to food products.

50 The analytical proof of the presence of small amounts of allergens, must use a technique 51 sensitive enough to specifically detect the allergens in those amounts that might trigger 52 allergic reactions in sensitized individuals. The determining step in allergen detection is the 53 isolation of the allergens from the food which can be influenced by numerous factors. The 54 diversity of foods in terms of composition and texture as well may require appropriate

Food Additives and Contaminants

extraction methods. The extraction solution must provide efficient isolation of the allergenic proteins from the sample into a liquid phase and minimize background effects due to nonspecific binding. In food many ingredients are available to participate in complex physical and chemical reactions with the allergenic proteins (Davis et al., 2001). Especially during the manufacturing process the matrix might cause covalent modification of proteins through food processing. As well as antigenicity, the extractability of allergens can be decreased in these reactions, they might be unchanged or even increased (Besler et al., 2001).

The aim of this study was to evaluate and compare extraction solutions for the isolation of milk and egg allergens. For this purpose extractability of allergenic proteins of dried milk powder, sweet whey, acid whey, whole egg, egg yolk and egg white with eleven different extraction agents was examined. Extractants were selected by collecting recipes already described in literature to isolate allergens. In addition, chosen extraction solutions represent a huge diversity in pH, ionic strength, and detergent content. As extraction quality describing parameters protein yield, the composition of the proteins as well as the antigen patterns were determined. To evaluate, if food matrix influences the extractability of proteins, wheat starch, fat powder and an experimental dry cake mix were employed in the extraction experiments.

73 Materials and Methods

74 Reagents and materials

Milk powder, dried sweet whey, dried acid whey, whole egg powder, powdered egg yolk and powdered egg white were obtained from Kampffmeyer Food Innovation GmbH (Hamburg, Germany). Powdered egg white was purchased from Sigma-Aldrich Chemie GmbH (Munich, Germany). To obtain powdered egg yolk, liquid whole egg was separated mechanically and lyophilized referring to homogenization. Commercially available wheat starch (protein-free, Heinrich Klenk GmbH & Co. KG, Schwebheim, Germany), fat powder (protein-free coconut oil, Kampffmeyer Food Innovation GmbH, Hamburg, Germany) and an experimental dry cake mix were used as matrix. The dry cake mix contained sucrose (70.5 %, Kampffmeyer Food

Innovation GmbH, Hamburg, Germany), fat powder (17.7 %), wheat starch (9.8 %), sodium
hydrogen carbonate (1.6 %, Merck Chemicals, Darmstadt, Germany) and sodium chloride
(0.4 %, Merck Chemicals, Darmstadt, Germany).

Sheep anti-casein serum and biotinylated rabbit anti-sheep serum were purchased from antibodies-online GmbH (Aachen, Germany). Rabbit anti-whey and rabbit anti-ovalbumin sera were both acquired from Sigma-Aldrich Chemie GmbH (Munich, Germany). Biotinylated goat anti-rabbit antibodies were bought from Rockland (Gilbertsville, PA, USA) and streptavidin-horseradish peroxidase was obtained from Bio-Rad Laboratories GmbH (Munich, Germany). All other reagents were of analytical grade and purchased from Sigma-Aldrich Chemie GmbH (Munich, Germany), Fluka (Buchs, Switzerland) and Serva Electrophoresis GmbH (Heidelberg, Germany).

96 Composition of extractants

In this investigation, eleven different extractants were used to extract protein from dairy and egg products: A: 0.01 M phosphate-buffered saline containing 10 mM Na₂HPO₄, 1 mM KH₂PO₄, 3 mM KCl and 140 mM NaCl, pH 7.4 (Schubert, 2003); B: 0.01 M phosphate-buffered saline containing 2 mM NaH₂PO₄, 8 mM Na₂HPO₄ and 150 mM NaCl, pH 7.4 (Hefle and Lambrecht, 2004); C: 0.01 M phosphate-buffered saline containing 150 mM NaCl, 10 mM Na₂HPO₄, pH 7.2 (Leduc et al., 1999); D: physiological saline containing 150 mM NaCl, pH 6.6 (Langeland, 1982); E: 1 % Tween 20[®], pH 6.0 (Fernandez et al., 1999); F : 8 M urea, pH 7.7 (Natarajan et al., 2005); G: 100 mM acetate buffer, pH 3.8 (Hirose et al., 2004); H: 86 mM NaCl, 30 mM NaHCO₃, 0.4 % phenol, pH 8.2 (Coca, 1922); I: 1 % Tween 20 [®], 0.4 % Triton X-100[®], 280 mM NaCl, 40 mM NaH₂PO₄, pH 7.4 (Yeung et al., 2000); J: 4 % Sodium dodecyl sulphate, pH 8.5 (Kato et al., 2001a); K bidestilled water. All extraction solutions were prepared using bidestilled water and were stabilized with 0.02 % NaN₃. For investigations with the aim to optimize pH-value of selected extractants, J (pH 8.5) and I (pH

Food Additives and Contaminants

- 7.4) were additionally prepared with pH-values adjusted to 10.5 and 12.5 and 9.4 and 11.4, respectively. Adjustment of pH was carried out with 0.1 M NaOH.

Protein isolation

Principally, extraction procedure was carried out as follows. According to the protein content measured via Kjeldahl method, 30 mg milk powder, 100 mg sweet whey, 100 mg acid whey, 25 mg whole egg powder, 15 mg dried egg white and 30 mg dried egg yolk, respectively, were weighed into a 50 mL polypropylene centrifuge tube (Nunc GmbH & Co. KG, Wiesbaden, Germany). In the case of matrix addition, 1 g matrix (wheat starch, fat powder, dry cake mix) was also weight in the centrifuge tube and the dry material was mixed 5 sec with a vortex mixer (VWR international, Darmstadt, Germany; mixing frequency 2,500 rpm) before extractant was added. Protein extraction was carried out by suspending sample material with 20 mL of extraction solution. The homogenate was extracted over night on a laboratory shaker (Bühler, Tübingen, Germany; shaking frequency 150 rpm) at room temperature and centrifuged afterwards 30 min at 10,000 rpm (Sigma-Aldrich Chemie GmbH, Munich, Germany). Solids were removed by filtration (Macherey-Nagel GmbH & Co. KG, Düren, Germany; filter paper grade MN 616). Investigations with the aim to optimize extraction temperature were additionally carried out at 4 °C on a laboratory shaker in a standard refrigerator (Linde, München, Germany) and at 40 ℃ using a laboratory incubator (IRC-1-U Clim-o-shake, Adolf Kühner AG, Birsfelden, Swiss; shaking frequency 150 rpm). For examination of extraction duration, protein isolation as described above was interrupted after 0.5, 1, 2, 4 and 8 hours. Each extraction process was performed in triplicate.

Protein quantification

Absolute protein content of used dry materials was determined via Kjeldahl method analysing the nitrogen content. The conversion factors from nitrogen to protein were as recommended 6.38 (dairy products) and 6.25 (egg products), respectively. The protein concentration of the extracts was measured photometrically according to the method of Lowry (1951) using

bovine serum albumin (BSA) and ovalbumin respectively as standards. Photometrical measurement of protein content was favoured over Kjeldahl method because in comparison the latter is of high expenditure of time and technical effort. The Folin's reagent used in the Lowry method consists amongst others of phosphomolybdic / phosphotungstic acid. A standard curve was produced for each run using a 500 mg/L protein solution prepared using the appropriate extractant. The protein content of the extracts prepared with "Coca's solution" (containing phenol) was measured by the method of Bradford (1976) because phenol interferes with the Lowry method. Bradford reagent consists of Coomassie Brilliant Blue G-250 and phosphoric acid. Quantitative assessments of extracts and standards were performed by measuring optical density at 720 nm (Lowry method) and 590 nm (Bradford method) using a spectrometer (MRX Plate Reader, Dynex Technologies, Frankfurt, Germany).

Lithium Dodecyl Sulphate-Polyacrylamide Gel Electrophoresis (LDS - PAGE)

Proteins were separated with precast NuPAGE® Bis-Tris 12 % gels (Invitrogen GmbH, Karlsruhe, Germany) under denaturing conditions at 200 V for 55 min using NuPAGE® MOPS [3-(N-morpholino) propanesulfonic acid] SDS running buffer (Invitrogen GmbH, Karlsruhe, Germany). Following the manufacturers instructions obtained extracts were diluted 1:1 in LDS (lithium dodecyl sulphate) sample buffer (Invitrogen GmbH, Karlsruhe, Germany) containing dithiothreitol for reduction at room temperature. Sample load was 20 µL per lane. Proteins were further silver stained or transferred by blotting method onto a nitrocellulose membrane.

Silver staining

The proteins were detected by standard silver staining procedure according to (Heukeshoven and Dernick, 1986) following LDS - PAGE. After fixing the gel (30 % ethanol, 10 % acetic acid in water; 30 min) it was sensitized in 0.1 % sodium thiosulfate (0.1 % Na₂S₂O₃, 30 % ethanol, 6.8 % sodium acetate in water). After washing with water the gel was incubated for 20 min in 0.2% silver nitrate solution (0.2 % AgNO₃, 0.02% formaldehyde in water).

Food Additives and Contaminants

Afterwards the gel was developed in 2.5 % sodium carbonate (2.5 % Na₂CO₃, 0.01%
formaldehyde in water). Staining was terminated using 1.8 % EDTA-solution (1.8 %
ethylenediaminetetraacetate in water).

10 171 *Immunoblotting*

The isolated proteins separated by LDS - PAGE were transferred electrophoretically from the gel onto the surface of a nitrocellulose membrane (2 µm, Whatman, Kent, UK) using semidry blotting method. In brief, dried membranes were blocked with blocking solution (containing 0.5 % Tween 20[®], 0.9 % sodium chloride and 0.6 % tris base) and incubated overnight with sheep anti-casein (1:400), rabbit anti-whey (1:200000) and rabbit anti-ovalbumin (1:200000) antibodies, respectively, at room temperature. Membranes were then incubated consecutively with biotinylated rabbit anti-sheep serum (1:100000; milk powder) or biotinylated goat anti-rabbit serum (1:7500; powdered sweet whey, powdered acid whey, whole egg powder, powdered egg yolk, powdered egg white) and streptavidin - horseradish peroxidase. Staining solution containing TMB and DONS (0.06 % 3, 3', 5, 5' -Tetramethylbenzidine, 0.2 % dioctylsodiumsulphosuccinate, 25 % ethanol, 0.7 % citric acid, 1 % Na₂HPO₄ in water) was used for staining.

39 184

41 185 **Results**

186 Protein quantification

Protein recovery rates in the extracts were estimated by relating theoretical calculated protein content measured via the Kjeldahl method to total protein yield of the extract measured with Lowry or Bradford method (data not shown). The estimated recovery rates are presented in Table 1 and Table 2. Extracting milk proteins from milk powder, recovery rates range from 69 to 109 % except for solution G yielding 16 % protein. Highest recoveries in protein isolation were achieved using solution I and J as extractants. Estimated recovery rates for extraction of milk proteins from milk powder in dry cake mix ranged from 40 to 87 % except for solution G recovering 14 % protein. Within these extracts, extractants B, D and J gave the highest protein recoveries. Extracting proteins from milk powder in wheat starch

and fat powder, the highest protein contents were obtained in the extracts prepared usingsolution I, J and C or I and D, respectively (Table 1).

For all the egg materials which were examined, the extraction agents I and J aqueous solution gave the highest protein yields (Table 2). In contrast to the extractants mentioned above, the protein contents of the extracts prepared using saline solutions (A, B, C, D), G or H to extract egg proteins from egg material led to recovery rates lower than 30 %. Where egg proteins were isolated from egg powder in dry cake mix, solutions I and J gave the highest protein recovery rates. Extracting proteins from egg material in wheat starch and fat powder as food matrix, the highest protein contents were measured in the extracts prepared using solutions I and J. But also E achieved high recovery of protein from egg material in wheat starch (Table 2).

In this study, optimization of pH-value of extractants as well as extraction temperature was carried out. Therefore, solution I (pH 7.4) and J (pH 8.5) were additionally prepared having a pH of 9.4 and 11.4 (I) as well as 10.5 and 12.5 (J), respectively. Extraction procedures were simultaneously performed at three different extraction temperatures (4 $^{\circ}$ C, RT = room temperature and 40 $^{\circ}$ C). The estimated recovery rates are presented in Tables 3 and 4. Comparing the protein yields there is not much of a difference. The deviation of the obtained recovery rates is of the order of the calculated standard deviation.

47 216

With the aim to investigate optimum extraction duration, extraction procedures were performed at different extraction temperatures using solution I (pH 7.4) and J (pH 8.5) as extractants to isolate milk and egg proteins from milk powder and whole egg powder and from wheat starch and cake mix, respectively. Extraction procedures were stopped after 30 min, 1 h, 2 h, 4 h and 8 h and protein contents were measured subsequently. Resulting recovery rates are shown in Tables 5 and 6. Regardless of extractant and extraction

Food Additives and Contaminants

temperature there is not much of a difference comparing protein recoveries. The extracts
 prepared of solution J have slightly higher protein contents than those made of by solution I.

226 Lithium Dodecyl Sulphate-Polyacrylamide Gel Electrophoresis (LDS - PAGE)

The composition of the isolated protein extracts were analyzed by electrophoretic separation and silver staining. Comparison of (i) the number of protein bands and (ii) the intensities of characteristic protein bands illustrates considerable protein diversity between the examined extracts. LDS - PAGE and silver staining patterns of proteins extracted from egg powder and milk powder using different extractants are shown in Figure 1. Those extracts prepared from egg powder using solution I, J and E gave particularly strong stained bands of conalbumine, ovalbumine and ovomucoid. Extracts prepared using solution G or H gave weakly stained bands of characteristic egg proteins. Similar results were obtained extracting powdered dairy products. The extractants I, J and E gave strongly stained bands of characteristic milk proteins α-lactalbumin, β-lactoglobulin and casein, whilst solution G gave weakly detected bands representing characteristic milk proteins. Analysis of protein patterns of extracts obtained extracting allergen material out of food matrix did not reveal wide differences in band intensity of characteristic milk and egg allergens.

0 240

The influence of the pH-value of the extractants was investigated. Solution I was prepared having pH-values of 7.4, 9.4 and 11.4, respectively. The pH-value of solution J was adjusted to 8.5, 10.5 and 12.5, respectively. Detection of the electrophoretically separated extracts via silver staining demonstrated that extracted proteins out of egg products and milk products led to similar results. Figure 2 shows the LDS - PAGE and silver staining patterns of proteins, adjusted to different pH-values and used for extraction at 4 °C, room temperature and 40 °C. Comparing the band intensities of the characteristic proteins in the extracts prepared with solution I pH 7.4 and pH 9.4 no differences appeared. Likewise, the band intensities of the characteristic proteins in the extracts obtained using solution J pH 8.5 were as strong as those prepared at pH 10.5. Any characteristic proteins could be visualized in the extracts

made of solution J with a pH of 12.5. Conversely, any characteristic proteins detected after electrophoretic separation of the extracts made from solution I with a pH of 11.4 or the detection of the characteristic proteins was much weaker.

The band intensities obtained via silver staining of milk and egg proteins isolated at different temperatures were almost identical. Extracts obtained at the end of varied extraction durations using solution I (pH 7.4) and J (pH 8.5) were electrophoretically separated and proteins were detected via silver staining. Regardless of extractant and extracted material, band intensities of characteristic proteins did not differ in terms of extraction duration (data not shown).

Immunoblotting

Whether the estimated extractants ensure comprehensive isolation of intact allergens specific immunochemical staining on nitrocellulose membrane after LDS - PAGE and Western Blotting was performed (Figures 3 and 4). The extracts derived with solution I, J and E present most strongly stained bands of ovalbumine compared to solutions G and H in the immunoblots. The results of unspecific staining procedure as well as the determination of the protein concentration are reflected in the results of immunochemical staining. Rabbit anti-ovalbumin antibody used for detection of albumin also detects conalbumin (76 kDa) as characteristic egg protein. Quite similar results were obtained by immunochemical detection of milk proteins. Band intensities of milk proteins of purchased extracts looked rather similar to the extracts described in this paper and Immunoblots showed only slight variations in protein recovery rates. Anti-casein antibody used for detection of casein also detects bovine serum albumin (BSA; 76 kDa) as characteristic milk protein.

The influence of wheat starch and dry cake mix on extraction efficiency and quality was investigated by comparing the extract properties prepared with matrix addition and without as described above. Addition of 8 M urea as extractant to wheat starch or dry cake mix

Food Additives and Contaminants

279 containing samples caused a strong enhancement of extract viscosity. Those solutions were 280 unemployable for further analysis. Band intensities referring to silver and immunochemical 281 staining do not differ noteworthy comparing the extracts made with matrix addition or without 282 (Figures 3 and 4). Considering dilution factors, extracts of solution I, J as well as E present 283 most strongly stained bands of ovalbumine compared to solutions G and H in the 284 immunoblots. Purchased extracts showed similar band intensities. The immunoblots affirm 285 the slight variation in protein recovery rates obtained for extracts made of dairy products.

287 Comparing the band intensities of the characteristic proteins in the extracts prepared with 288 solution I (pH 7.4 and pH 9.4) and J (pH 8.5 and 10.5), respectively, no differences 289 appeared. Any characteristic proteins could be visualized in the extracts made of solution J 290 pH 12.5 whilst the detection of the characteristic proteins extracted via solution I" pH 11.4 291 was much weaker.

The band intensities obtained via immunochemical staining of milk and egg proteins isolated at different temperatures were almost identical. Regardless of extractant and extracted material, the band intensity of the characteristic proteins did not differ in terms of extraction duration (data not shown).

2 297

298 Discussion

299 The ultimate prerequisite for sensitive analytical allergen detection is an efficient and reliable
 300 extraction method.

1 301 Protein recovery rates

⁵³ 302 Protein recovery rates were calculated by comparing the total protein concentration of the
⁵⁴ 303 food sample with the protein yields of the extracts. The photometric detection, which was
⁵⁷ 304 much more suitable for the screening of the extracts than the Kjeldahl procedure, only
⁵⁹ 305 provides a relative but not an absolute quantification of the proteins. Nevertheless, the
³⁰⁶ relative quantification was sufficient to assess the efficiency of the different extractants.

1 2		
3 4	307	
5 6	308	Ranking the best extractant
7 8	309	The evaluation of recovery rates of extracted milk and egg material using eleven different
9 10	310	solutions led to similar results. The solutions I and J achieved the highest protein yields.
11 12	311	In the case of milk samples, the extractants had only minor impact on the extraction
13 14 15	312	efficiency. On the other hand, the extraction efficiency of egg material was much more
16 16 17	313	sensitive to the solution used for the protein extraction. The major difference between milk
18 19	314	and egg is the fat content. Because of the higher fat content of egg material, extractants
20 21	315	containing a detergent yielded higher protein recovery rates as compared to those without.
22 23	316	Emulsifying properties of egg-lecithin alone are not sufficient to extract egg powder properly.
24 25 26	317	These results clearly indicate that the composition of the extractant is of particular
20 27 28	318	importance for the efficiency and quality of the extract.
29 30	319	
31 32	320	Quality of the protein extracts
33 34	321	Analysis of the extracts by LDS - PAGE followed by silver staining as well as immunoblotting
35 36	322	confirmed the results of protein quantification. Solution I as well as J led to the strongest
37 38 39	323	bands of characteristic milk and egg proteins, respectively. However, it has to be considered,
40 41	324	that silver staining generally possesses a lower sensitivity than photometrical protein
42 43	325	quantification. Slight variations concerning protein content in the extracts cannot be
44 45	326	discriminated after electrophoretical separation.
46 47	327	
48 49 50	328	Detection of selected proteins by western blotting clearly demonstrated that both extraction
50 51 52	329	solutions had no negative influence onto the antigenicity of the extracted allergenic proteins.
53 54	330	Solution I and J are the most qualified extractants to isolate proteins of milk. In case of hen's
55 56	331	eggs, they are the only suitable solutions for the efficient extraction of proteins. Hence, hen's
57 58	332	egg is more sensitive to the extractant than milk.

334 Optimization of the extraction procedure: temperature and pH

http://mc.manuscriptcentral.com/tfac Email: fac@tandf.co.uk

Page 13 of 28

Food Additives and Contaminants

For the elucidation of the optimal temperature range, extractions were carried out at 4 °C, room temperature and 40 °C. According to Howard et al. (1988) solubility of proteins can be increased at 40 °C compared to lower temperatures. In contrast, thermal treatment of allergens can influence conformational epitopes, which might loose their binding capacity to specific antibodies (Besler et al., 2001). In the worst case, false-negative results are obtained. In this study, variation of extraction temperature neither led to any benefit in extraction quality nor to degradation. Comparison of recovery rates obtained via extraction at different temperatures resulted in essentially similar results. Also unspecific staining as well as immunoblots did not show any differences in band intensity comparing the extracts prepared at different extraction temperature. The influence of pH-values of extractants was investigated using solution I and J prepared having different pH-values. Solubility of proteins strongly depends on the pH-value of the medium. Proteins have minimum solubility in water or salt solutions at the pH which corresponds to their isoelectric point (pl). Because milk proteins as well as egg proteins have pl lower than 7 (except lysozyme), solution I (pH 7.4) was additionally prepared having a pH of 9.4 and 11.4 and solution J (pH 8.5) was prepared having pH-values of 10.5 and 12.5. In this study, evaluation of the influence of the protein content in the extracts changing the pH did not show any tendencies. In contrast, analysis of unspecific staining of the electrophoretically separated extracts demonstrated, that solution I with a pH of 11.4 caused a partially hydrolysis of the proteins, whilst solution J with a pH of 12.5 brought the proteins about to a total hydrolysis. Alkaline hydrolysis of proteins causes progressive disintegration up to small peptides so that they are not detectible neither via unspecific nor specific staining procedure. As a consequence, these extractants are not applicable for further analysis damaging allergens antigenicity, although, solution I pH 7.4 and 9.4 as well as J pH 8.5 and 10.5 did not show this hydrolysis effect in the immunoblots.

Optimization of the extraction duration

To establish an extraction procedure in routine analyses, sample preparation should not be too time-consuming. Otherwise, extraction duration has to guarantee maximum recovery of proteins. Evaluation of recovery rates, results of unspecific and specific staining of the extracts showed that an extraction time of 1 hour is sufficient for an appropriate sample preparation.

Food matrix

Comparing the results of qualitative and quantitative investigations of extraction of milk and egg allergens out of certain dried milk and egg products with and without the presence of several other food matrices, different findings were obtained. In summary, wheat starch influences the extraction yield and the extraction guality with regard to the protein and antigen pattern for all materials which were tested and for different extractants. In contrast, using fat powder and dry cake mix, respectively, different extraction procedures led to different results. Whilst fat powder did not influence recovery rates of milk powder, for sweet whey and whole egg powder, up to twice as much protein was isolated from powdered egg yolk. However, isolation of egg proteins from egg yolk combined with dry cake mix resulted in protein recovery rates quite similar as those obtained for extracts without any matrix. On the other hand extraction of proteins from whey as well as powdered egg white blended with fat powder and dry cake mix, respectively, for some extractants lower protein recovery rates were found. Detection of electrophoretically separated proteins confirmed the results of protein quantification in giving weaker stained bands. These results highlight the fact that the suitability of extractant not only depends on allergen properties but furthermore on the surrounding food matrix.

What is the best procedure?

In this study, in the case of isolating proteins out of milk material "1% Tween 20 and 0.4% Triton X-100" pH 7.4 was the most suitable extraction agent. To extract egg proteins out of egg material the best results were obtained using 4 % SDS aqueous solution at pH 8.5. If the allergens are to be extracted out of complex food matrices containing sugars, fat and proteins, the extractant mentioned above are still suitable, but partly less quantitative. In the Page 15 of 28

Food Additives and Contaminants

1 2		
3 4	391	case of egg allergens in a starch and fat containing matrix Tween 20 $^{\scriptscriptstyle (B)}$ as extractant gave
5 6	392	high protein recoveries. Regardless of allergen material and food matrix, carrying out the
7 8	393	extraction procedure for 1 hour at room temperature guarantees high protein yields without
9 10	394	any negative influence on the allergens antigenicity.
11 12	395	
13 14	396	Acknowledgments
15 16	397	This study was supported by the AiF-project AiF 14805 N (AiF: Arbeitsgemeinschaft
17 18 19	398	industrieller Forschungsvereinigungen). Special thanks go to B. Jacobsen, K. Ramcke and S.
20 21	399	Schwarz for technical support.
22 23	400	
24 25	401	References
26 27	402	Besler M, Steinhart H, Paschke A. 2001. Stability of food allergens and allergenicity of
28 29	403	processed foods. J Chromatogr B Biomed Sci Appl. 756: 207-228.
30 31	404	Bradford MM. 1976. A rapid and sensitive method for the quantitation of microgram
32 33	405	quantities of protein utilizing the principle of protein-dye binding. Anal Biochem. 72:
34 35	406	248-54.
36 37		
38 39	407	Coca AF. 1922. Studies in specific hypersensitiveness. V. The preparation of fluid extracts
40 41 42	408	and solutions for use in the diagnosis and treatment of the allergies with notes on the
42 43	409	collection of pollens. J Immunol. 7: 163-78.
44 45	410	Davis PJ, Smales CM, James DC. 2001. How can thermal processing modify the antigenicity
46 47	411	of proteins? Allergy. 56 Suppl 67: 56-60.
48 49	412	Fernandez SS, Padilla AP, Mucciarelli S. 1999. Protein extraction from Atriplex lampa
50 51 52	413	leaves: potential use as forage for animals used for human diets. Plant Foods Hum
52 53 54	414	Nutr. 54: 251-259.
55 56	415	Hefle SL, Taylor SL. 2004. Food allergy and the food industry. Curr Allergy Asthma Rep. 4:
57 58	416	55-9.
59 60		

- Hefle SL, Lambrecht DM. 2004. Validated sandwich enzyme-linked immunosorbent assay for
 casein and its application to retail and milk-allergic complaint foods. J Food Prot. 67:
 1933-1938.
- 420 Heukeshoven J, Dernick R. 1986. Neue Ergebnisse zum Mechanismus der Silberfärbung.
 11
 12 421 Paper presented at: Electrophoresis Forum '86; Technische Universität München.
- Hildebrandt S, Steinhart H, Paschke A. 2008. Comparison of different extraction solutions for
 the analysis of allergens in hen's egg. Food Chem. 108: 1088-1093.
- Hirose J, Kitabatake N, Kimura A, Narita H. 2004. Recognition of native and/or thermally
 425 induced denatured forms of the major food allergen, ovomucoid, by human IgE and
 426 mouse monoclonal IgG antibodies. Biosc Biotechnol Biochem. 68: 2490-2497.
- 427 Howard SB, Twigg P, Baird JK, Meehan EJ. 1988. The solubility of hen egg-white lysozyme.
 428 J Cryst Growth. 90: 94-104.
- 429 Kato Y, Oozawa E, Matsuda T. 2001a. Decrease in antigenic and allergenic potentials of
 430 ovomucoid by heating in the presence of wheat flour: dependence on wheat variety
 431 and intermolecular disulfide bridges. J Agric Food Chem. 49: 3661-3665.

Kato Y, Suginohara K, Fujiwara M. 2001b. A novel and simple method of insolubilization of ovomucoid in cookies prepared from batter containing egg white. Food Sci Technol Res. 7: 35-38.

- 42 435 Langeland T. 1982. A clinical and immunological study of allergy to hen's egg white. II.
 436 Antigens in hen's egg white studied by crossed immunoelectrophoresis (CIE). Allergy.
 437 37: 323-33.
- 48
 49
 438
 438
 438
 50
 51
 439
 52
 53
 54
 54
 54
 54
 54
 54
 54
 54
 54
 55
 56
 57
 58
 58
 59
 59
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50
 50</l
- 441 Lowry OH, Rosebrough NJ, Farr AL, Randall RJ. 1951. Protein measurement with the Folin
 442 phenol reagent. J Biol Chem. 193: 265-75.
- 59 60

1 2 3

4 5

6 7

8 9

13

24

1 2		
3 4	443	Moneret-Vautrin DA, Kanny G. 2004. Update on threshold doses of food allergens:
5 6	444	Implications for patients and the food industry. Curr Opin Allergy Clin Immunol. 4:
7 8	445	215-219.
9 10	446	Natarajan S, Xu C, Caperna TJ, Garrett WM. 2005. Comparison of protein solubilization
11 12	447	methods suitable for proteomic analysis of soybean seed proteins. Anal Biochem.
13 14	448	342: 214-220.
15 16 17	449	Schubert S. 2003. Charakterisierung und Isolierung von Allergenen der Tomate
18 19	450	(Lycopersicon lycopersicum (L.) Karst. ex Farw.) [dissertation]. [Institut für Biochemie
20 21	451	und Lebensmittelchemie]: Universität Hamburg.
22 23	452	Vieluf I, Besler M, Paschke A, Steinhart H, Vieluf D. 2002. Practical approach to adverse
24 25	453	food reactions. New Trends in Allergy V, 5th, Davos, Switzerland, Sept. 15-17, 2000.
26 27 28	454	190-202.
28 29 30	455	Yeung JM, Newsome WH, Abbott MA. 2000. Determination of egg proteins in food products
31 32	456	by enzyme immunoassay. J AOAC Int. 83: 139-143.
33 34	457	
35 36	458	
37 38		
39 40 41		
41 42 43		
44 45		
46 47		
48 49		
50 51		
52 53		
54 55		
56 57 58		

Table 1: Protein recovery rates (%) estimated using different extractants to isolate milk proteins out of milk material or out of milk material in matrix

Milk material	Matrix		Extractant														
		A	В	С	D	E	F	G	Н	I	J	K					
	WM	88 ± 4	85 ± 3	90 ± 2	89 ± 3	85 ± 2	86 ± 3	16 ± 1	69 ± 4	109 ± 4	94 ± 4	86 ± 8					
milk powder	WS	87 ± 1	91 ± 4	92 ± 4	88 ± 4	79 ± 7	-	10 ± 0	69 ± 4	116 ± 6	92 ± 5	89 ± 2					
	FP	86 ± 3	79 ± 5	83 ± 5	98 ± 5	44 ± 0	-	27 ± 5	78 ± 2	106 ± 4	85 ± 5	88 ± 2					
	СМ	71 ± 4	82 ± 9	72 ± 13	87 ± 13	40 ± 3	-	14 ± 0	81 ± 3	74 ± 5	82 ± 8	74 ± 4					
	WM	81 ± 1	85 ± 3	91 ± 3	92 ± 3	85 ± 3	97 ± 4	53 ± 1	37 ± 1	138 ± 12	112 ± 3	85 ± 4					
Sweet whey	WS	81 ± 4	89 ± 7	90 ± 1	93 ± 1	89 ± 6	-	45 ± 3	36 ± 1	142 ± 4	105 ± 2	91 ± 8					
Oweet whey	FP	80 ± 1	84 ± 5	80 ± 2	101 ± 2	47 ± 3	B	65 ± 11	42 ± 1	87 ± 1	92 ± 3	95 ± 2					
	СМ	83 ± 4	84 ± 3	63 ± 10	110 ± 10	47 ± 2	-	77 ± 4	32 ± 3	75 ± 2	86 ± 6	89 ± 5					
	WM	99 ± 10	95 ± 1	102 ± 3	79 ± 3	79 ± 2	93 ± 4	41 ± 2	46 ± 0	140 ± 6	112 ± 5	60 ± 3					
Acid whey	WS	94 ± 5	107 ± 9	106 ± 7	59 ± 7	83 ± 3	-	31 ± 2	55 ± 3	137 ± 9	113 ± 8	30 ± 1					
Acid Wiley	FP	81 ± 2	82 ± 1	79 ± 6	96 ± 6	41 ± 2	-	71 ± 2	43 ± 2	96 ± 3	91 ± 2	30 ± 2					
	СМ	81 ± 1	81 ± 3	82 ± 12	101 ± 12	47 ± 2	-	97 ± 1	44 ± 1	87 ± 2	91 ± 5	66 ± 5					
	1						1	1	1								

http://mc.manuscriptcentral.com/tfac Email: fac@tandf.co.uk

Food Additives and Contaminants

Table 2: Protein recovery rates (%) estimated using different extractants to isolate egg proteins out of egg material or out of egg material in matrix

(WM = without matrix; WS = wheat starch; FP = fat powder; CM = cake mix)

		1	2	3	4	5	6	7	8	9	10	11	12
	WM	27 ± 2	26 ± 2	20 ± 1	23 ± 2	108 ± 3	74 ± 5	26 ± 1	22 ± 1	119 ± 11	112 ± 3	64 ± 2	87
Whole egg	WS	22 ± 1	22 ± 2	22 ± 2	21 ± 1	108 ± 10	-	21 ± 2	27 ± 2	100 ± 4	117 ± 4	70 ± 0	84
Whole egg	FP	20 ± 1	22 ± 1	18 ± 2	41 ± 2	75 ± 1	-	22 ± 2	30 ± 2	125 ± 0	134 ± 5	64 ± 6	97
	СМ	29 ± 2	25 ± 1	21 ± 4	28 ± 3	78 ± 1	-	27 ± 3	30 ± 3	100 ± 2	124 ± 4	45 ± 2	112
	WM	37 ± 2	36 ± 1	38 ± 0	30 ± 2	86 ± 0	63 ± 4	79 ± 2	49 ± 2	129 ± 6	113 ± 5	16 ± 0	71
Egg yolk	WS	38 ± 1	38 ± 2	39 ± 4	30 ± 2	83 ± 4	-	62 ± 1	45 ± 1	111 ± 9	108 ± 2	20 ± 1	73
	FP	81 ± 6	70 ± 4	70 ± 11	75 ± 4	77 ± 2		86 ± 6	33 ± 2	151 ± 9	139 ± 4	94 ± 7	90
	СМ	38 ± 2	38 ± 1	50 ± 3	45 ± 4	85 ± 5		107 ± 8	35 ± 3	113 ± 12	133 ± 6	51 ± 1	83
	WM	76 ± 2	76 ± 7	61 ± 11	53 ± 1	121 ± 3	116 ± 8	37 ± 2	20 ± 0	138 ± 16	123 ± 5	77 ± 2	86
Egg white	WS	76 ± 3	79 ± 1	71 ± 1	69 ± 1	120 ± 3	-	39 ± 1	21 ± 2	135 ± 9	132 ± 2	88 ± 2	95
	FP	15 ± 1	14 ± 0	19 ± 0	16 ± 0	90 ± 1	-	39 ± 2	32 ± 1	130 ± 10	145 ± 4	24 ± 2	111
	СМ	16 ± 2	19 ± 2	22 ± 8	26 ± 4	78 ± 0	-	52 ± 6	44 ± 1	120 ± 4	141 ± 9	41 ± 2	102
											71		

Table 3: Protein recovery rates (%) using solution I and J, respectively, at different pH-values to isolate milk proteins from milk material and from milk material in certain matrices at different temperatures. (WM = without matrix; WS = wheat starch; CM = cake mix)

Dry milk	Extractant	Matrix	pН	Rec	overy rates	[%]
material	Extractant	IVIALITX	рп	4 °C	RT	40 °C
			7.4	95 ± 6	103 ± 5	96 ± 4
		WM	9.4	72 ± 7	82 ± 9	89 ± 7
Milk powder			11.4	90 ± 4	91 ± 2	95 ± 3
		WS	7.4	95 ± 2	71 ± 4	85 ± 6
		vv3	9.4	63 ± 4	59 ± 2	68 ± 6
		СМ	7.4	97 ± 2	89 ± 1	89 ± 3
		CIVI	9.4	60 ± 3	62 ± 6	57 ± 5
			8.5	73 ± 3	75 ± 5	74 ± 5
		WM	10.5	83 ± 2	83 ± 3	78 ± 3
			12.5	99 ± 2	93 ± 4	90 ± 5
	J	WS	8.5	90 ± 1	96 ± 6	96 ± 5
		vv3	10.5	92 ± 4	91 ± 5	92 ± 1
		СМ	8.5	98 ± 3	97 ± 5	99 ± 6
		CIVI	10.5	112 ± 7	108 ± 2	115 ± 6
		6	7.4	85 ± 2	87 ± 3	100 ± 3
		WM	9.4	82 ± 1	90 ± 3	79 ± 6
			11.4	111 ± 4	107 ± 1	105 ± 2
Sweet whey	l l	WS	7.4	80 ± 5	70 ± 4	80 ± 3
		VV 3	9.4	76 ± 6	87 ± 5	81 ± 3
		СМ	7.4	89 ± 3	83 ± 3	87 ± 1
		CIVI	9.4	76 ± 3	70 ± 1	75 ± 6
			8.5	67 ± 3	72 ± 1	77 ± 4
		WM	10.5	88 ± 0	91 ± 1	89 ± 4
			12.5	129 ± 11	118 ± 11	110 ± 2
	J	WS	8.5	103 ± 5	102 ± 3	109 ± 7
		VV 3	10.5	101 ± 1	100 ± 1	101 ± 1
		СМ	8.5	113 ± 4	111 ± 1	114 ± 1
		CIVI	10.5	118 ± 2	113 ± 2	111 ± 1
			7.4	81 ± 5	95 ± 2	96 ± 1
		WM	9.4	91 ± 1	91 ± 10	89 ± 4
			11.4	109 ± 6	128 ± 12	109 ± 8
	I	WS	7.4	84 ± 3	87 ± 4	87 ± 3
		W3	9.4	81 ± 2	49 ± 5	88 ± 7
		СМ	7.4	80 ± 7	97 ± 4	95 ± 3
Acid whey			9.4	87 ± 7	85 ± 4	86 ± 2
			8.5	71 ± 2	75 ± 4	76 ± 1
		WM	10.5	87 ± 2	98 ± 6	83 ± 2
			12.5	113 ± 2	113 ± 3	101 ± 4
	J	WS	8.5	100 ± 2	102 ± 2	101 ± 3
		110	10.5	96 ± 3	100 ± 5	101 ± 1
		СМ	8.5	118 ± 12	115 ± 4	111 ± 3
		ON	10.5	108 ± 2	119 ± 7	108 ± 11

Food Additives and Contaminants

Table 4: Protein recovery rates (%) using solution I and J, respectively, at different pH-values to isolate egg proteins from egg material and from egg material in certain matrices at different temperatures. (WM = without matrix; WS = wheat starch; CM = cake mix)

Dry egg	Extractant	Matrix	ъЦ	Red	covery rates	6 [%]
material		IVIALITX	рН	4 ℃	RT	40 °C
			7.4	94 ± 1	89 ± 7	99 ± 5
		WM	9.4	102 ± 4	98 ± 8	101 ± 2
Whole egg -			11.4	110 ± 3	103 ± 4	94 ± 6
		WS	7.4	93 ± 4	93 ± 1	92 ± 4
		VV3	9.4	102 ± 3	101 ± 1	101 ± 8
		СМ	7.4	100 ± 3	105 ± 9	105 ± 2
		Civi	9.4	105 ± 1	105 ± 2	106 ± 3
			8.5	114 ± 3	120 ± 2	126 ± 4
		WM	10.5	109 ± 2	117 ± 3	114 ± 2
			12.5	110 ± 3	111 ± 7	111 ± 6
	J	WS	8.5	111 ± 5	117 ± 7	114 ± 9
		VV3	10.5	102 ± 6	108 ± 4	113 ± 5
		СМ	8.5	111 ± 7	113 ± 5	106 ± 1
		Civi	10.5	92 ± 4	91 ± 4	100 ± 2
			7.4	114 ± 8	109 ± 9	117 ± 3
		WM	9.4	116 ± 4	115 ± 3	127 ± 6
			11.4	126 ± 6	123 ± 7	118 ± 3
Egg white	I	WS	7.4	107 ± 6	115 ± 5	116 ± 5
		VV3	9.4	123 ± 7	123 ± 9	126 ± 1
		СМ	7.4	114 ± 2	114 ± 4	121 ± 7
		Civi	9.4	118 ± 5	120 ± 4	127 ± 10
			8.5	102 ± 4	122 ± 2	131 ± 2
		WM	10.5	119 ± 7	116 ± 1	126 ± 6
			12.5	118 ± 6	124 ± 6	120 ± 7
	J	WS	8.5	124 ± 8	119 ± 8	127 ± 2
		W5	10.5	126 ± 4	116 ± 5	119 ± 3
		СМ	8.5	123 ± 11	122 ± 3	118 ± 5
		Civi	10.5	105 ± 6	109 ± 6	114 ± 4
			7.4	121 ± 2	121 ± 2	132 ± 11
		WM	9.4	125 ± 10	111 ± 6	103 ± 5
			11.4	116 ± 1	<u>111 ± 7</u>	112 ± 7
	I	WS	7.4	106 ± 1	101 ± 0	103 ± 7
		V 0	9.4	118 ± 7	109 ± 5	114 ± 8
		СМ	7.4	105 ± 3	106 ± 3	105 ± 7
Egg yolk			9.4	115 ± 3	121 ± 7	113 ± 2
			8.5	108 ± 1	110 ± 10	121 ± 8
		WM	10.5	117 ± 2	113 ± 1	117 ± 6
			12.5	103 ± 6	109 ± 7	105 ± 3
	J	WS	8.5	115 ± 4	112 ± 3	114 ± 5
			10.5	111 ± 1	110 ± 2	111 ± 4
		СМ	8.5	113 ± 4	113 ± 10	111 ± 5
			10.5	92 ± 6	87 ± 3	96 ± 6

Table 5: Protein recovery rates obtained by varied extraction durations at three temperatures using solution I (pH 7.4) and J (pH 8.5) to isolate milk proteins from skim milk (WM = without matrix; CM = cake mix).

Extractant	Matrix	Temp.		E	xtraction	duration	[h]	
	IVIALITX	remp.	0	0.5	1	2	4	8
		4 ℃		82 ± 0	90 ± 1	85 ± 2	85 ± 6	82 ± 2
	WM	RT	81 ± 1	81 ± 2	97 ± 1	89 ± 3	80 ± 7	85 ± 4
J		40 °C		82 ± 3	89 ± 3	78 ± 1	82 ± 3	84 ± 3
(pH 8.5)		4 °C		89 ± 8	98 ± 3	92 ± 7	99 ± 2	88 ± 5
	СМ	RT	80 ± 1	80 ± 6	90 ± 9	98 ± 5	75 ± 7	86 ± 0
		40 °C		82 ± 4	88 ± 7	87 ± 1	96 ± 2	91 ± 7
		4 ℃	2	88 ± 1	101 ± 9	84 ± 2	86 ± 1	82 ± 0
	WM	RT	96 ± 1	94 ± 2	102 ± 4	86 ± 0	114 ± 9	86 ± 1
I		40 °C		84 ± 7	112 ± 8	83 ± 1	87 ± 1	88 ± 8
(pH 7.4)		4 °C		90 ± 4	83 ± 1	80 ± 5	86 ± 2	83 ± 1
	СМ	RT	78 ± 4	80 ± 0	90 ± 6	83 ± 4	80 ± 6	77 ± 7
		40 °C		83 ± 3	92 ± 5	76 ± 7	93 ± 9	78 ± 3

Table 6: Protein recovery rates obtained by varied extraction durations at three temperatures using solution I and J to isolate egg proteins from whole egg powder (WM = without matrix; CM = cake mix).

Extractant	Matrix	Temp.		E	xtraction c	luration [h]	
Extractant	IVIALITA	remp.	0	0.5	1	2	4	8
		4 °C		106 ± 5	105 ± 1	104 ± 4	106 ± 3	104 ± 4
	WM	RT	107 ± 4	110 ± 3	108 ± 3	104 ± 2	104 ± 1	109 ± 2
J		40 °C		110 ± 0	109 ± 2	109 ± 4	106 ± 3	107 ± 1
(pH 8.5)		4 °C		116 ± 6	105 ± 3	112 ± 1	105 ± 4	113 ± 8
	СМ	RT	115 ± 6	119 ± 12	117 ± 5	112 ± 4	111 ± 3	112 ± 6
		40 °C		116 ± 0	122 ± 3	120 ± 2	123 ± 0	111 ± 6
		4 °C		96 ± 2	99 ± 2	99 ± 5	104 ± 2	104 ± 3
	WM	RT	94 ± 6	96 ± 4	93 ± 3	98 ± 2	94 ± 3	96 ± 6
I		40 °C		99 ± 3	103 ± 3	95 ± 2	101 ± 0	101 ± 3
(pH 7.4)	-	4 °C		92 ± 1	93 ± 1	92 ± 1	89 ± 6	91 ± 0
	СМ	RT	91 ± 2	90 ± 0	99 ± 1	93 ± 2	94 ± 4	93 ± 5
		40 °C		112 ± 3	112 ± 6	117 ± 3	117 ± 2	108 ± 0

Fig. 1: LDS - PAGE and silver staining patterns of proteins extracted from egg powder (left)
and milk powder (right), respectively, using different extractants. MW: molecular weight
marker; CON: Conalbumin (76 kDa); OVA: Ovalbumin (44.5 kDa); OVO: Ovomucoid (28
kDa); LYS: Lysozyme (14.3 kDa); BSA: Bovine serum albumin (66.3 kDa); CAS: Caseins
(20.6 -25.2 kDa); LG: β-lactoglobulin (18.3 kDa); LA: α-lactalbumin (14.2 kDa).

Fig. 2: LDS - PAGE and silver staining patterns of proteins extracted from egg powder using
solution J (left) and I (right), respectively, adjusted to different pH-values and used for
extraction at 4 °C, room temperature (RT) and 40 °C. MW: molecular weight marker; CON:
Conalbumin (76 kDa); OVA: Ovalbumin (44.5 kDa); OVO: Ovomucoid (28 kDa); LYS:
Lysozyme (14.3 kDa).

Fig. 3: Left: LDS - PAGE and Immunoblotting of ovalbumin extracted from egg powder using different extractants and detection via anti-ovalbumin antibody. Right: LDS - PAGE and Immunoblotting of casein extracted from milk powder using different extractants and detection via anti-casein antibody. MW: molecular weight marker; CON: Conalbumin (76 kDa); OVA: Ovalbumin (44.5 kDa); OVO: Ovomucoid (28 kDa); LYS: Lysozyme (14.3 kDa); BSA: Bovine serum albumin (66.3 kDa); CAS: Casein (20.6 -25.2 kDa).

Fig. 4: Left: LDS - PAGE and Immunoblotting of ovalbumin extracted from egg powder out of
dry cake mix as matrix using different extractants and detection via anti-ovalbumin antibody.
Right: LDS - PAGE and Immunoblotting of casein extracted from milk powder out of dry cake
mix as matrix using different extractants and detection via anti-casein antibody. MW:
molecular weight marker; CON: Conalbumin (76 kDa); OVA: Ovalbumin (44.5 kDa); OVO:
Ovomucoid (28 kDa); LYS: Lysozyme (14.3 kDa); BSA: Bovine serum albumin (66.3 kDa);
CAS: Casein (20.6 -25.3 kDa).

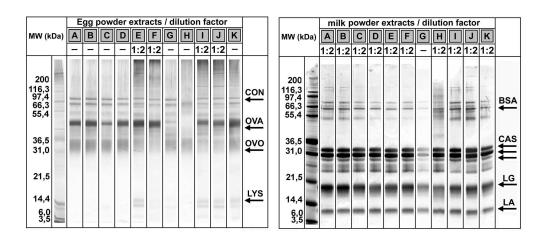


Figure 1 299x129mm (183 x 183 DPI)

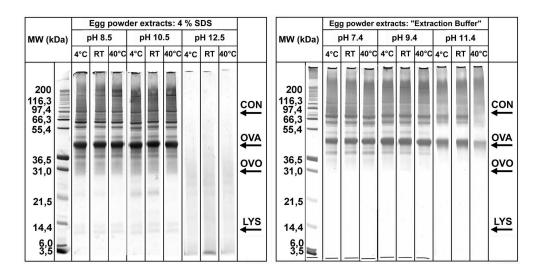


Figure 2 259x129mm (183 x 183 DPI)

http://mc.manuscriptcentral.com/tfac Email: fac@tandf.co.uk

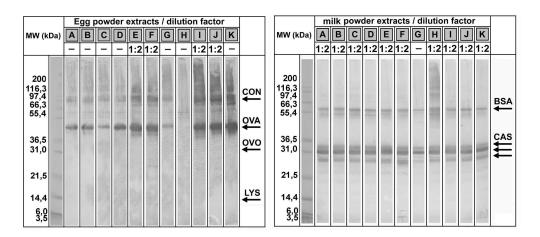


Figure 3 299x129mm (183 x 183 DPI)

		E	gg I	oow	der	extra	acts	/ di	lutic	on fa	icto	r			n	nilk	pow	der	extra	acts	/ di	lutio	n fa	cto		
MW (kD	a)	Α	в	С	D	Ε	F	G	Н	I	J	κ		MW (kDa)	Α	В	С	D	Ε	F	G	Н	T	J	κ	
		-	-	-	-	1:2		-	-	1:2	1:2	-			1:2	21:2	1:2	1:2	1:2		-	1:2	1:2	1:2	1:2	
200 116,3 97,4 66,3 55,4 36,5 31,0	I I I I I I I I		A The second sec		1. 12	Carlo Stationer							ÇON QVA QVO	200 116,3 97,4 66,3 55,4 36,5 31,0		N - H - K			The subscript of the							BSA CAS
21,5 14,4 6,0 3,5	1 1 1 F		: 0			14						and the second	LYS ◀	21,5 14,4 6,0 3,5											A State of the sta	

Figure 4 299x129mm (183 x 183 DPI)