

Radioisotopes demonstrate the contrasting bioaccumulation capacities of heavy metals in embryonic stages of cephalopod species

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To cite this version:

Thomas Lacoue-Labarthe, Roger Villanueva, Claude Rouleau, François Oberhänsli, Jean-Louis Teyssié, et al.. Radioisotopes demonstrate the contrasting bioaccumulation capacities of heavy metals in embryonic stages of cephalopod species. PLoS ONE, 2011 , 6 (11), pp.e27653. $10.1371/jour$ nal.pone.0027653. hal-00645126

HAL Id: hal-00645126 <https://hal.science/hal-00645126v1>

Submitted on 26 Nov 2011

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24 **ABSTRACT**

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26 Cephalopods play a key role in many marine trophic food webs and also constitute alternative fishery 27 resources in the context of the ongoing decline in finfish stocks. Most coastal cephalopod species of 28 commercial importance migrate into shallow waters during the breeding season to lay their eggs, and 29 are consequently subjected to coastal contamination. Eggs of common cuttlefish *Sepia officinalis*, 30 European squid *Loligo vulgaris*, common octopus *Octopus vulgaris* and the sepiolid *Rossia* 31 *macrosoma* were exposed during embryonic development to dissolved 110m Ag, 109 Cd, 60 Co, 54 Mn and ⁶⁵ Zn in order to determine their metal accumulation efficiencies and distribution among different egg 33 compartments. Cuttlefish eggs, in which hard shells enclose the embryos, showed the lowest 34 concentration factor (CF) values despite a longer duration of exposure. In contrast, octopus eggs, 35 which are only protected by the chorionic membrane, accumulated the most metal. Uptake appears to 36 be linked to the selective retention properties of the egg envelopes with respect to each element. The 37 study also demonstrated that the octopus embryo accumulated $110m$ Ag directly from the dissolved 38 phase and also indirectly through assimilation of the contaminated yolk. These results raise questions 39 regarding the potential contrasting vulnerability of early life stages of cephalopods to the metallic 40 contamination of coastal waters. 41

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⁴³**INTRODUCTION**

44

45 Cephalopods in European waters have attracted increasing attention from marine biologists 46 and fishery scientists[1]. Indeed, this group is of high ecological significance for the understanding of 47 the food webs in marine ecosystems. This reflects their central position as they are as much a predator 48 of fish and invertebrates as they are prey for numerous fish, seabirds and marine mammals [2,3]. 49 Secondly, cephalopod fisheries have increased significantly over the last six decades, with landings 50 nowadays reaching 120 000 tons per year, i.e., 4-fold the landings made in the 1950s [4], and they 51 therefore constitute alternative resources in the context of the ongoing decline in finfish stocks. The 52 European squid *Loligo vulgaris,* the common octopus *Octopus vulgaris* and the common cuttlefish 53 *Sepia officinalis* are the main targets of these fishery activities in Europe [1]. For these reasons, an 54 increasing number of studies has focused on their biology, ecology and their responses to the 55 environmental factors that drive their population dynamics*.*

56 These coastal cephalopods undergo extensive (e.g., *S. officinalis*) or limited migration (e.g., *O.* 57 *vulgaris*) in the breeding season to reach shallow waters, where they lay their eggs on benthic 58 substrates. The eggs therefore develop in an environment that is exposed to more intense 59 anthropogenic activities than adult environments, and to the release of various contaminants including 60 metals. Cephalopod embryos are thus exposed to dissolved trace elements during development, which 61 represents 10–15% of the cephalopod life span [5]. Previous studies of *S. officinalis* demonstrated that 62 its embryos, which are encapsulated in an eggshell composed of nidamental-, oviducal-originated 63 layers and a chorion, are partially protected against metal penetration [6-8]. Interestingly, these 64 shielding properties varied with respect to the particular trace element, resulting in, for instance, high 65 Ag-accumulation in the embryonic tissues but total protection against Pb-penetration [9-10]. This 66 contrasting metal-dependent permeability of the eggshell can thus limit the exposure of the embryo to 67 the potential toxicological effects of particular contaminants during development, but not others [11].

68 Within the cephalopods, egg encapsulation varies in terms of size, architecture and structure [12]. 69 Briefly, common cuttlefish and sepiolids lay individual medium-sized encapsulated eggs that are 70 covered by gelatinous envelopes stained with ink or by a jelly covered by a rigid capsule, respectively 71 [12]. European squid lay several small-sized eggs that are united by a common oviducal and 72 nidamental jelly sheet, whereas the common octopus produces small eggs that are devoid of an 73 envelope other than the chorion [13]. In this context, this study aimed to determine the potential inter-74 specific differences in metal bioaccumulation in the embryos and hatchlings of four species (three of 75 which are of high economic interest) that differ appreciably in their eggshell structure. By exposing 76 the eggs to dissolved 110m Ag, 109 Cd, 57 Co, 54 Mn and 65 Zn throughout development, we determined the 77 concentration and distribution of radiotracers in the eggshell envelopes, vitellus and embryo to help 78 explain the bioaccumulation efficiencies observed in hatchlings of four species, *R. macrosoma, S.* 79 *officinalis, L. vulgaris,* and *O. vulgaris*, representing the three main Orders of the Class Cephalopoda.

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⁸¹**MATERIALS AND METHODS**

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83 Adult cuttlefish and octopus, collected by net fishing off Monaco in March 2006 and June 84 2008, respectively, were maintained in 300-l open-circuit tanks at the IAEA-EL premises and fed 85 daily with crabs and frozen fish. After spawning, cuttlefish eggs (n = 20) were collected and 86 immediately separated to optimise their oxygenation. Octopus egg masses (six strings of \sim 200 eggs 87 each) were kept with the female for brooding during the first week to optimise embryo survival before 88 their collection for experiments. Freshly laid eggs of squid ($n = six$ strands of ~ 40 eggs each) and

89 sepiolids (n = 12 eggs) were collected in May 2008 and March 2009, respectively, off Monaco at a 90 depth of 20–300 m, and placed immediately under experimental conditions.

91 The eggs were placed in four 20-l glass aquaria containing filtered (0.45 µm) and UV-sterilized 92 seawater (constantly aerated closed circuit; salinity 37 p.s.u.; light/dark cycle: 12 h/12 h) at 16 °C, or 93 19 °C for the octopus eggs. They were then exposed to dissolved radiotracers 110m Ag, 109 Cd, 57 Co 94 (Amersham, UK) and 54 Mn and $52n$ (Isotope Product Laboratory, USA) throughout development 95 (except octopus eggs, which were exposed one week after spawning), according to the experimental 96 procedure described by Lacoue-Labarthe et al. [14]. Due to the sensitivity of the radiotracer tool, the 97 low levels of radioactivity used in this work allowed us to monitor trace element accumulation 98 behaviour under environmentally realistic levels of metal exposure [15]. The concentrations of 99 radiotracers in seawater, exposure durations and concentrations of added metals are reported in Table 100 1.

101 The eggs were sampled at intermediate developmental stages [16] (see Table 2) ($n = 6$ and 4 102 eggs for sepiolids and cuttlefish, respectively; $n = 9$ pools of 10–20 eggs from three strands of squid 103 and five pools of 20 eggs for octopus, to obtain sufficient biomass to allow the detection of significant 104 levels of radiotracers) and dissected to determine the distribution of the radiotracers among the 105 different egg compartments, *i.e.* the different eggshell layers, the embryo, the vitellus and the 106 perivitelline fluid. At hatching, six and four sepiolid and cuttlefish hatchlings, respectively, and nine 107 pools of 10–20 individuals and five pools of 25 individuals of squid and octopus hatchlings, 108 respectively, were weighed and counted.

109 Radioanalyses were carried out using a high-resolution γ-spectrometry system consisting of 110 four coaxial Germanium (N- or P-type) detectors (EGNC 33-195-R, Canberra® and Eurysis®) 111 connected to a multi-channel analyser and a computer equipped with spectra analysis software 112 (Interwinner® 6). Radioactivities were determined by comparison with standards of known activity 113 and of appropriate geometry. Measurements were corrected for counting efficiency and physical 114 radioactive decay. The counting time was adjusted to obtain a propagated counting error of less than 115 5% [17].

116 The accumulation of each radiotracer was expressed as a concentration factor (CF), which is 117 the ratio between radiotracer activity in the egg or egg compartment (Bq g^{-1}) and the time-integrated 118 activity in seawater (Bq g^{-1}). The results are expressed as mean \pm SD.

119 Whole-body autoradiography was used to reveal the distribution of $110th$ Ag in octopus eggs 120 exposed to dissolved 110m Ag for 1 week. Eggs were embedded in a 2.5% carboxymethylcellulose gel 121 and flash-frozen in a slurry of dry ice in hexane. From these egg pools, 20-µm thick sections were cut 122 with a specially designed cryomicrotome (Leica CM3600), freeze-dried, and placed on phosphor 123 screens (Perkin-Elmer) for 4 to 7 d. After exposure, the screens were scanned with a Cyclone Phosphor 124 Imager (Perkin-Elmer) and $110m$ Ag activity in the egg compartments was quantified as previously 125 described [18].

126 A Kruskal-Wallis test was applied to determine the differences between the CFs calculated for the 127 different egg compartments. All experiments complied with the ethical and welfare considerations 128 relating to the use of cephalopods as experimental animals [19].

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¹³¹**RESULTS AND DISCUSSION**

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133 Many studies have shown that cephalopods have the capacity to accumulate trace elements at 134 high levels in their tissues [20-28]. For this reason, the health concerns related to the consumption of 135 cephalopod flesh have been evaluated for different commercial species from different locations [29- 136 33]. In cephalopods, the effects of metals range from direct effects during embryogenesis to various 137 toxic effects on the early juveniles [11,34-36]. Nonetheless, both maternal transferred and dissolved 138 metals can have an impact during the embryo's development, and may be particularly critical for the 139 success of population recruitment in cephalopods.

140 This study revealed that metal accumulation (considering all radioisotopes) in embryos varied 141 between the four considered species, such that $CF_{octopus} > CF_{squid} \ge CF_{cuttlefish} > CF_{sepidid}$ (Table 2). 142 Although the maternal transfer of several elements has previously been demonstrated through the 143 vitellus [37], our results are consistent with the natural metal concentrations recorded in hatchlings 144 from the field [38]. This suggests that metal concentrations in hatchlings reflect the environmental 145 conditions in which the eggs develop and depend on egg morphology. Once the eggs have been laid, 146 the observed inter-specific differences in bioaccumulation therefore result from contrasting 147 waterborne metal uptake that is likely to be due to the type of encapsulation of the embryos, which is 148 specific to the different species. The CF values recorded in embryos at intermediate stages of 149 development (Table 2) suggest differences in the permeabilities of the eggshells towards metals. 150 Indeed, all radiotracers accumulated in the embryonic tissues in squid and octopus whereas only ⁵⁴Mn 151 and $110m$ Ag, and 54 Mn and 55Zn , respectively, were detected at significant levels in the internal 152 compartment of sepiolid and cuttlefish eggs.

153 The selective permeability of the cuttlefish eggshell towards trace elements has previously 154 been interpreted as being linked to the expansion of the chorionic membrane, resulting in egg swelling 155 [9,10,14], which induces the shrinkage and stretching of the envelopes due to the notable increase in 156 egg volume [12]. Nevertheless, the composition of the eggshell, which is rich in carboxyl- and 157 sulphydryl-groups [13], might explain why most metals were associated with this compartment (> 158 90% for 109° Cd, 57° Co, 54 Mn, 55° Zn and > 64% for 110° Ag; results not shown), therefore hindering metal 159 penetration into the internal compartments. More precisely, 110m Ag, 109 Cd, 54 Mn and 65 Zn were more 160 highly concentrated in the nidamental layers than in the pooled oviducal envelopes and chorion, but 161 ⁵⁷Co was found at similar concentrations in these two compartments ($P_{Kruskall-Wallis} > 0.05$; Table 2), 162 suggesting that different components of the eggshell provide different levels of shielding according to 163 the metal. The contrasting affinities of metals for the different eggshell parts were more obvious in 164 squid eggs. Thus, the chorion may act as a barrier against 109 Cd and 57 Co penetration as it exhibited CF 165 values 4 and 10-fold higher ($P_{Kruskall-Wallis}$ < 0.01; Table 2) than the nidamental layers and retained 41 166 and 60% of the total amount of metal accumulated, respectively. In contrast, 110m Ag, 54 Mn and 65 Zn 167 were mainly bound to the nidamental- and oviducal-originated layers ($P_{Kruskall-Wallis} < 0.01$). These 168 results highlight the fact that the diffusion capacities of the different metals through the egg envelopes 169 differ with respect to the element considered and to the eggshell composition and/or structure. Thus, 170 the soft and gelatinous eggshell of squid appears more permeable than this of cuttlefish leading to 171 higher accumulation of heavy metals in the embryo during development.

172 With regard to the "well-protected" egg of sepiolids, more than 99% (results not shown) of 173 110m Ag, 109 Cd, 57 Co and 65 Zn was bound to the outer "nidamental" envelope, with the highest CF 174 values recorded in this compartment among the four species investigated (Table 2). In contrast to 175 squid and cuttlefish eggs, the outer coating in sepiolids is a rigid sphere. Therefore, its volume does 176 not change due to the egg swelling that occurs during development. Moreover, its structure is 177 characterized by several layers, each covered by an electron microscopic "dense" membrane [39] that 178 might favour the retention of metal ions. Such strong shielding properties of the eggshell probably 179 explain the low metal contamination found in the hatchlings despite long-term exposure (i.e., 82 d) to 180 radioisotopes. It is therefore likely that radiotracer uptake occurs rapidly during the last days of 181 embryonic life when the eggcase starts to break before hatching. However, ⁵⁴Mn was detected in the 182 internal jelly and chorion, suggesting that this essential element can pass selectively through the outer 183 capsule before accumulating in the embryo*.*

184 Octopus eggs lack an eggshell and the eggs are only protected by a chorionic membrane, 185 which retained 88, 84 and 72% of the ${}^{57}Co$, ${}^{54}Mn$ and ${}^{65}Zn$, respectively (results not shown). However, 186 the retention capacity of the chorion for 110^m Ag and $10⁹$ Cd was low (i.e., 4 and <7%, respectively). 187 Consequently, both metals became highly concentrated in the internal compartments (Table 2), 188 resulting in at least 4- and 7-fold higher CF values in the octopus hatchlings than in the other species 189 (Table 1). Our results are consistent with the 5-fold higher Ag concentrations recorded in hatchlings of

190 *O. vulgaris* compared to those of *S. officinalis* and *L. vulgaris* [38]. In cuttlefish, it is noteworthy that 191 the embryo accumulated more of the metals that penetrated through the eggshell than the other 192 compartments (*i.e.* $CF_{emb} > CF_{yolk} > CF_{PVF}$; $P_{Kruskall-Wallis} < 0.05$; Table 2). In contrast, no significant 193 differences between the embryo and the yolk were observed in the CF values of ^{65}Zn in cuttlefish, 194 ${}^{57}Co$, ${}^{54}Mn$, ${}^{65}Zn$ in squid and ${}^{109}Cd$ in octopus (P_{Kruskall-Wallis} > 0.05; Table 2). More surprising was that 195 the highest CF values of 109 Cd and 110 m Ag were observed in the yolk of squid and octopus, 196 respectively, compared to the other compartments (P*Kruskall-Wallis* < 0.01; Table 2). The autoradiogram of 197 octopus eggs exposed to dissolved $110m$ Ag revealed that metal distribution was closely related to the 198 position of the outer and inner yolk sacs (Figure 1). This clearly indicates that Ag is efficiently 199 incorporated in the vitelline reserves. Thus, the octopus embryo accumulates waterborne Ag both 200 directly from seawater and from the yolk, which absorbs the metal. This would explain the high CF 201 values recorded in octopus paralarvae at hatching (CF_{Ag} of 20760 \pm 3785; Table 2). In this context, the 202 embryos of octopus appeared to be the most highly exposed to environmental contamination by 203 metals, and especially by Ag.

204

205 In conclusion, this work shows the strongly contrasting efficiencies of metal uptake by the 206 four studied species, which appears to be correlated with the different embryo encapsulation strategies 207 observed in these cephalopods [12]. The lowest CF values were found in the sepiolid and cuttlefish 208 embryos (despite their longer exposure to radiotracers in this study; see Table 2), both of which are 209 protected by hard shells during their longer embryonic life, unlike those of squid, which are embedded 210 in soft envelopes, or those of octopus that lack any protective eggshell [12]. The large eggs of *R.* 211 *macrosoma* and *S. officinalis* produce large, benthic hatchlings that weigh 30 to 100 times more than 212 the small, planktonic and active swimming hatchlings of *L. vulgaris* and *O. vulgaris* [5,40] (see Table 213 2). The different mode of life of these early stages suggests the possibility of a contrasting metabolism 214 between the embryos that could modulate the rate of uptake of metals [41]. In fact, metals were 215 incorporated in inverse relation to body size (i.e. lower for the large-sized *S. officinalis* and higher for 216 the small-sized *O. vulgaris*). It can thus be tentatively hypothesized that the surface–volume 217 relationship may influence metal absorption [42,43]. Finally, these results demonstrate different 218 strategies regarding metal requirements and/or protection with respect to the essential and non-219 essential elements [37].

220 221

222 **ACKNOWLEDGMENTS**

223

224 The IAEA is grateful to the Government of the Principality of Monaco for the support provided to its 225 Environment Laboratories.

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- 342

FIGURE CAPTION

Figure 1. Tissue section (top) and autoradiogram (bottom) of *Octopus vulgaris* eggs at stage

346 XVI [16] after exposure to dissolved ^{110m}Ag for 1 week; c: chorion, e: embryo, iy: inner yolk sac, oy: outer yolk sac.

351 Table 1. Exposure duration expressed in days (d) and radiotracer activities in seawater (mean \pm SD) 351
352
353

352 during experimental exposure.

354 na: non available data

356 Table 2. Concentration factors (mean \pm SD) of 110m Ag, 109 Cd, 57 Co, 54 Mn, and 65 Zn in the hatchlings

357 (wet weight; mg) and in the compartments of eggs sampled at intermediate embryonic developmental

358 stages (according to Naef [16]) of *Rossia macrosoma*, *Sepia officinalis*, *Loligo vulgaris* and *Octopus* 359 *vulgaris* exposed to dissolved radiotracers in seawater during embryonic development (days; d).

361 na: not analysed; dl: detection limit; PVF, perivitelline fluid