

# Radioisotopes demonstrate the contrasting bioaccumulation capacities of heavy metals in embryonic stages of cephalopod species

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1	Radioisotopes demonstrate the contrasting bioaccumulation				
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#### 24 ABSTRACT

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26 Cephalopods play a key role in many marine trophic food webs and also constitute alternative fishery 27 resources in the context of the ongoing decline in finfish stocks. Most coastal cephalopod species of commercial importance migrate into shallow waters during the breeding season to lay their eggs, and 28 29 are consequently subjected to coastal contamination. Eggs of common cuttlefish Sepia officinalis, European squid Loligo vulgaris, common octopus Octopus vulgaris and the sepiolid Rossia 30 macrosoma were exposed during embryonic development to dissolved <sup>110m</sup>Ag, <sup>109</sup>Cd, <sup>60</sup>Co, <sup>54</sup>Mn and 31 <sup>65</sup>Zn in order to determine their metal accumulation efficiencies and distribution among different egg 32 33 compartments. Cuttlefish eggs, in which hard shells enclose the embryos, showed the lowest 34 concentration factor (CF) values despite a longer duration of exposure. In contrast, octopus eggs, which are only protected by the chorionic membrane, accumulated the most metal. Uptake appears to 35 be linked to the selective retention properties of the egg envelopes with respect to each element. The 36 study also demonstrated that the octopus embryo accumulated <sup>110m</sup>Ag directly from the dissolved 37 phase and also indirectly through assimilation of the contaminated yolk. These results raise questions 38 39 regarding the potential contrasting vulnerability of early life stages of cephalopods to the metallic 40 contamination of coastal waters.

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### 43 INTRODUCTION

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45 Cephalopods in European waters have attracted increasing attention from marine biologists and fishery scientists [1]. Indeed, this group is of high ecological significance for the understanding of 46 47 the food webs in marine ecosystems. This reflects their central position as they are as much a predator 48 of fish and invertebrates as they are prey for numerous fish, seabirds and marine mammals [2,3]. 49 Secondly, cephalopod fisheries have increased significantly over the last six decades, with landings 50 nowadays reaching 120 000 tons per year, i.e., 4-fold the landings made in the 1950s [4], and they 51 therefore constitute alternative resources in the context of the ongoing decline in finfish stocks. The 52 European squid Loligo vulgaris, the common octopus Octopus vulgaris and the common cuttlefish Sepia officinalis are the main targets of these fishery activities in Europe [1]. For these reasons, an 53 54 increasing number of studies has focused on their biology, ecology and their responses to the 55 environmental factors that drive their population dynamics.

These coastal cephalopods undergo extensive (e.g., S. officinalis) or limited migration (e.g., O. 56 57 vulgaris) in the breeding season to reach shallow waters, where they lay their eggs on benthic 58 substrates. The eggs therefore develop in an environment that is exposed to more intense 59 anthropogenic activities than adult environments, and to the release of various contaminants including 60 metals. Cephalopod embryos are thus exposed to dissolved trace elements during development, which 61 represents 10–15% of the cephalopod life span [5]. Previous studies of S. officinalis demonstrated that 62 its embryos, which are encapsulated in an eggshell composed of nidamental-, oviducal-originated 63 layers and a chorion, are partially protected against metal penetration [6-8]. Interestingly, these 64 shielding properties varied with respect to the particular trace element, resulting in, for instance, high 65 Ag-accumulation in the embryonic tissues but total protection against Pb-penetration [9-10]. This contrasting metal-dependent permeability of the eggshell can thus limit the exposure of the embryo to 66 67 the potential toxicological effects of particular contaminants during development, but not others [11].

68 Within the cephalopods, egg encapsulation varies in terms of size, architecture and structure [12]. 69 Briefly, common cuttlefish and sepiolids lay individual medium-sized encapsulated eggs that are 70 covered by gelatinous envelopes stained with ink or by a jelly covered by a rigid capsule, respectively 71 [12]. European squid lay several small-sized eggs that are united by a common oviducal and 72 nidamental jelly sheet, whereas the common octopus produces small eggs that are devoid of an 73 envelope other than the chorion [13]. In this context, this study aimed to determine the potential inter-74 specific differences in metal bioaccumulation in the embryos and hatchlings of four species (three of 75 which are of high economic interest) that differ appreciably in their eggshell structure. By exposing the eggs to dissolved <sup>110m</sup>Ag, <sup>109</sup>Cd, <sup>57</sup>Co, <sup>54</sup>Mn and <sup>65</sup>Zn throughout development, we determined the 76 77 concentration and distribution of radiotracers in the eggshell envelopes, vitellus and embryo to help 78 explain the bioaccumulation efficiencies observed in hatchlings of four species, R. macrosoma, S. 79 officinalis, L. vulgaris, and O. vulgaris, representing the three main Orders of the Class Cephalopoda.

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MATERIALS AND METHODS

83 Adult cuttlefish and octopus, collected by net fishing off Monaco in March 2006 and June 84 2008, respectively, were maintained in 300-1 open-circuit tanks at the IAEA-EL premises and fed 85 daily with crabs and frozen fish. After spawning, cuttlefish eggs (n = 20) were collected and 86 immediately separated to optimise their oxygenation. Octopus egg masses (six strings of ~ 200 eggs 87 each) were kept with the female for brooding during the first week to optimise embryo survival before 88 their collection for experiments. Freshly laid eggs of squid (n = six strands of ~ 40 eggs each) and 89 sepiolids (n = 12 eggs) were collected in May 2008 and March 2009, respectively, off Monaco at a 90 depth of 20-300 m, and placed immediately under experimental conditions.

91 The eggs were placed in four 20-1 glass aquaria containing filtered (0.45 µm) and UV-sterilized 92 seawater (constantly aerated closed circuit; salinity 37 p.s.u.; light/dark cycle: 12 h/12 h) at 16 °C, or 19 °C for the octopus eggs. They were then exposed to dissolved radiotracers <sup>110m</sup>Ag, <sup>109</sup>Cd, <sup>57</sup>Co 93 (Amersham, UK) and <sup>54</sup>Mn and <sup>65</sup>Zn (Isotope Product Laboratory, USA) throughout development 94 95 (except octopus eggs, which were exposed one week after spawning), according to the experimental 96 procedure described by Lacoue-Labarthe et al. [14]. Due to the sensitivity of the radiotracer tool, the 97 low levels of radioactivity used in this work allowed us to monitor trace element accumulation 98 behaviour under environmentally realistic levels of metal exposure [15]. The concentrations of 99 radiotracers in seawater, exposure durations and concentrations of added metals are reported in Table 100 1.

101 The eggs were sampled at intermediate developmental stages [16] (see Table 2) (n = 6 and 4 102 eggs for sepiolids and cuttlefish, respectively; n = 9 pools of 10–20 eggs from three strands of squid 103 and five pools of 20 eggs for octopus, to obtain sufficient biomass to allow the detection of significant 104 levels of radiotracers) and dissected to determine the distribution of the radiotracers among the 105 different egg compartments, i.e. the different eggshell layers, the embryo, the vitellus and the 106 perivitelline fluid. At hatching, six and four sepiolid and cuttlefish hatchlings, respectively, and nine 107 pools of 10-20 individuals and five pools of 25 individuals of squid and octopus hatchlings, 108 respectively, were weighed and counted.

109 Radioanalyses were carried out using a high-resolution  $\gamma$ -spectrometry system consisting of 110 four coaxial Germanium (N- or P-type) detectors (EGNC 33-195-R, Canberra® and Eurysis®) 111 connected to a multi-channel analyser and a computer equipped with spectra analysis software 112 (Interwinner® 6). Radioactivities were determined by comparison with standards of known activity 113 and of appropriate geometry. Measurements were corrected for counting efficiency and physical 114 radioactive decay. The counting time was adjusted to obtain a propagated counting error of less than 115 5% [17]. 116 The accumulation of each radiotracer was expressed as a concentration factor (CF), which is 117 the ratio between radiotracer activity in the egg or egg compartment (Bq  $g^{-1}$ ) and the time-integrated 118 activity in seawater (Bq  $g^{-1}$ ). The results are expressed as mean  $\pm$  SD.

Whole-body autoradiography was used to reveal the distribution of <sup>110m</sup>Ag in octopus eggs exposed to dissolved <sup>110m</sup>Ag for 1 week. Eggs were embedded in a 2.5% carboxymethylcellulose gel and flash-frozen in a slurry of dry ice in hexane. From these egg pools, 20-µm thick sections were cut with a specially designed cryomicrotome (Leica CM3600), freeze-dried, and placed on phosphor screens (Perkin-Elmer) for 4 to 7 d. After exposure, the screens were scanned with a Cyclone Phosphor Imager (Perkin-Elmer) and <sup>110m</sup>Ag activity in the egg compartments was quantified as previously described [18].

A Kruskal-Wallis test was applied to determine the differences between the CFs calculated for the
different egg compartments. All experiments complied with the ethical and welfare considerations
relating to the use of cephalopods as experimental animals [19].

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## 131 **RESULTS AND DISCUSSION**

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Many studies have shown that cephalopods have the capacity to accumulate trace elements at high levels in their tissues [20-28]. For this reason, the health concerns related to the consumption of cephalopod flesh have been evaluated for different commercial species from different locations [29-33]. In cephalopods, the effects of metals range from direct effects during embryogenesis to various toxic effects on the early juveniles [11,34-36]. Nonetheless, both maternal transferred and dissolved metals can have an impact during the embryo's development, and may be particularly critical for the success of population recruitment in cephalopods.

140 This study revealed that metal accumulation (considering all radioisotopes) in embryos varied between the four considered species, such that  $CF_{octopus} > CF_{squid} \ge CF_{cuttlefish} > CF_{sepiolid}$  (Table 2). 141 142 Although the maternal transfer of several elements has previously been demonstrated through the 143 vitellus [37], our results are consistent with the natural metal concentrations recorded in hatchlings 144 from the field [38]. This suggests that metal concentrations in hatchlings reflect the environmental conditions in which the eggs develop and depend on egg morphology. Once the eggs have been laid, 145 the observed inter-specific differences in bioaccumulation therefore result from contrasting 146 147 waterborne metal uptake that is likely to be due to the type of encapsulation of the embryos, which is specific to the different species. The CF values recorded in embryos at intermediate stages of 148 149 development (Table 2) suggest differences in the permeabilities of the eggshells towards metals. Indeed, all radiotracers accumulated in the embryonic tissues in squid and octopus whereas only <sup>54</sup>Mn 150 and <sup>110m</sup>Ag, and <sup>54</sup>Mn and <sup>65</sup>Zn, respectively, were detected at significant levels in the internal 151 152 compartment of sepiolid and cuttlefish eggs.

153 The selective permeability of the cuttlefish eggshell towards trace elements has previously been interpreted as being linked to the expansion of the chorionic membrane, resulting in egg swelling 154 155 [9,10,14], which induces the shrinkage and stretching of the envelopes due to the notable increase in 156 egg volume [12]. Nevertheless, the composition of the eggshell, which is rich in carboxyl- and sulphydryl-groups [13], might explain why most metals were associated with this compartment (> 157 90% for  ${}^{109}$ Cd,  ${}^{57}$ Co,  ${}^{54}$ Mn,  ${}^{65}$ Zn and > 64% for  ${}^{110m}$ Ag; results not shown), therefore hindering metal 158 penetration into the internal compartments. More precisely, <sup>110m</sup>Ag, <sup>109</sup>Cd, <sup>54</sup>Mn and <sup>65</sup>Zn were more 159 highly concentrated in the nidamental layers than in the pooled oviducal envelopes and chorion, but 160 <sup>57</sup>Co was found at similar concentrations in these two compartments ( $P_{Kruskall-Wallis} > 0.05$ ; Table 2), 161 162 suggesting that different components of the eggshell provide different levels of shielding according to the metal. The contrasting affinities of metals for the different eggshell parts were more obvious in 163 squid eggs. Thus, the chorion may act as a barrier against <sup>109</sup>Cd and <sup>57</sup>Co penetration as it exhibited CF 164 values 4 and 10-fold higher ( $P_{Kruskall-Wallis} < 0.01$ ; Table 2) than the nidamental layers and retained 41 165 and 60% of the total amount of metal accumulated, respectively. In contrast, <sup>110m</sup>Ag, <sup>54</sup>Mn and <sup>65</sup>Zn 166 were mainly bound to the nidamental- and oviducal-originated layers ( $P_{Kruskall-Wallis} < 0.01$ ). These 167 results highlight the fact that the diffusion capacities of the different metals through the egg envelopes 168 169 differ with respect to the element considered and to the eggshell composition and/or structure. Thus, the soft and gelatinous eggshell of squid appears more permeable than this of cuttlefish leading to 170 171 higher accumulation of heavy metals in the embryo during development.

With regard to the "well-protected" egg of sepiolids, more than 99% (results not shown) of 172 <sup>110m</sup>Ag, <sup>109</sup>Cd, <sup>57</sup>Co and <sup>65</sup>Zn was bound to the outer "nidamental" envelope, with the highest CF 173 values recorded in this compartment among the four species investigated (Table 2). In contrast to 174 175 squid and cuttlefish eggs, the outer coating in sepiolids is a rigid sphere. Therefore, its volume does 176 not change due to the egg swelling that occurs during development. Moreover, its structure is 177 characterized by several layers, each covered by an electron microscopic "dense" membrane [39] that 178 might favour the retention of metal ions. Such strong shielding properties of the eggshell probably 179 explain the low metal contamination found in the hatchlings despite long-term exposure (i.e., 82 d) to radioisotopes. It is therefore likely that radiotracer uptake occurs rapidly during the last days of 180 embryonic life when the eggcase starts to break before hatching. However, <sup>54</sup>Mn was detected in the 181 182 internal jelly and chorion, suggesting that this essential element can pass selectively through the outer 183 capsule before accumulating in the embryo.

Octopus eggs lack an eggshell and the eggs are only protected by a chorionic membrane, which retained 88, 84 and 72% of the <sup>57</sup>Co, <sup>54</sup>Mn and <sup>65</sup>Zn, respectively (results not shown). However, the retention capacity of the chorion for <sup>110m</sup>Ag and <sup>109</sup>Cd was low (i.e., 4 and <7%, respectively). Consequently, both metals became highly concentrated in the internal compartments (Table 2), resulting in at least 4- and 7-fold higher CF values in the octopus hatchlings than in the other species (Table 1). Our results are consistent with the 5-fold higher Ag concentrations recorded in hatchlings of 190 O. vulgaris compared to those of S. officinalis and L. vulgaris [38]. In cuttlefish, it is noteworthy that 191 the embryo accumulated more of the metals that penetrated through the eggshell than the other compartments (*i.e.*  $CF_{emb} > CF_{yolk} > CF_{PVF}$ ;  $P_{Kruskall-Wallis} < 0.05$ ; Table 2). In contrast, no significant 192 differences between the embryo and the yolk were observed in the CF values of <sup>65</sup>Zn in cuttlefish, 193 194 <sup>57</sup>Co, <sup>54</sup>Mn, <sup>65</sup>Zn in squid and <sup>109</sup>Cd in octopus ( $P_{Kruskall-Wallis} > 0.05$ ; Table 2). More surprising was that the highest CF values of <sup>109</sup>Cd and <sup>110m</sup>Ag were observed in the yolk of squid and octopus, 195 196 respectively, compared to the other compartments ( $P_{Kruskall-Wallis} < 0.01$ ; Table 2). The autoradiogram of octopus eggs exposed to dissolved <sup>110m</sup>Ag revealed that metal distribution was closely related to the 197 198 position of the outer and inner yolk sacs (Figure 1). This clearly indicates that Ag is efficiently 199 incorporated in the vitelline reserves. Thus, the octopus embryo accumulates waterborne Ag both directly from seawater and from the yolk, which absorbs the metal. This would explain the high CF 200 values recorded in octopus paralarvae at hatching (CF<sub>Ag</sub> of  $20760 \pm 3785$ ; Table 2). In this context, the 201 202 embryos of octopus appeared to be the most highly exposed to environmental contamination by 203 metals, and especially by Ag.

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In conclusion, this work shows the strongly contrasting efficiencies of metal uptake by the 205 206 four studied species, which appears to be correlated with the different embryo encapsulation strategies 207 observed in these cephalopods [12]. The lowest CF values were found in the sepiolid and cuttlefish 208 embryos (despite their longer exposure to radiotracers in this study; see Table 2), both of which are 209 protected by hard shells during their longer embryonic life, unlike those of squid, which are embedded 210 in soft envelopes, or those of octopus that lack any protective eggshell [12]. The large eggs of R. 211 macrosoma and S. officinalis produce large, benthic hatchlings that weigh 30 to 100 times more than 212 the small, planktonic and active swimming hatchlings of L. vulgaris and O. vulgaris [5,40] (see Table 213 2). The different mode of life of these early stages suggests the possibility of a contrasting metabolism 214 between the embryos that could modulate the rate of uptake of metals [41]. In fact, metals were 215 incorporated in inverse relation to body size (i.e. lower for the large-sized S. officinalis and higher for 216 the small-sized O. vulgaris). It can thus be tentatively hypothesized that the surface-volume 217 relationship may influence metal absorption [42,43]. Finally, these results demonstrate different 218 strategies regarding metal requirements and/or protection with respect to the essential and non-219 essential elements [37].

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- 342

### 343 FIGURE CAPTION

345 Figure 1. Tissue section (top) and autoradiogram (bottom) of *Octopus vulgaris* eggs at stage

346 XVI [16] after exposure to dissolved <sup>110m</sup>Ag for 1 week; c: chorion, e: embryo, iy: inner yolk sac, oy:
347 outer yolk sac.



351 352 353 Table 1. Exposure duration expressed in days (d) and radiotracer activities in seawater (mean  $\pm$  SD) during experimental exposure.

	R. macrosoma	S. officinalis	L. vulgaris	O. vulgaris
Exposure duration (d)	82	44	23	28
$^{110\hat{m}}Ag (kBq.l^{-1})$	$0.45\pm0.33$	$0.66\pm0.37$	$0.55\pm0.37$	$0.54\pm0.37$
$^{109}$ Cd (kBq. $l^{-1}$ )	$1.41\pm0.58$	$1.56\pm0.77$	$1.71\pm0.37$	$0.94\pm0.47$
${}^{57}$ Co (kBq.1 <sup>-1</sup> )	$1.05\pm0.38$	$1.02\pm0.44$	$0.73\pm0.13$	$0.80\pm0.31$
$^{54}$ Mn (kBq.l <sup>-1</sup> )	$1.10\pm0.46$	$1.07\pm0.40$	$0.75\pm0.11$	$0.93\pm0.14$
$^{65}$ Zn (kBq.l <sup>-1</sup> )	$0.80\pm0.46$	$0.93\pm0.45$	$0.78\pm0.12$	$0.64\pm0.31$
Ag (ng. $l^{-1}$ )	108	1.3	86	86
$Cd (ng.l^{-1})$	16	0.1	na	na
$Co(ng.l^{-1})$	$2.10^{-3}$	1.2	170	170
$Mn (ng.l^{-1})$	$7.10^{-2}$	$25.10^{-2}$	na	na
$Zn (ng.l^{-1})$	1.2	65.10 <sup>-3</sup>	64	64

354 355 na: non available data

Table 2. Concentration factors (mean  $\pm$  SD) of  $^{110m}$ Ag,  $^{109}$ Cd,  $^{57}$ Co,  $^{54}$ Mn, and  $^{65}$ Zn in the hatchlings (wet weight; mg) and in the compartments of eggs sampled at intermediate embryonic developmental 356

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358 stages (according to Naef [16]) of Rossia macrosoma, Sepia officinalis, Loligo vulgaris and Octopus vulgaris exposed to dissolved radiotracers in seawater during embryonic development (days; d).

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360	
300	

Species	R. macrosoma	S. officinalis	L. vulgaris	O. vulgaris				
Hatchlings								
Wet weight	$121.6\pm3.9$	$137.1 \pm 11.3$	$3.9\pm0.9$	$1.3 \pm 0.1$				
<sup>110m</sup> Ag	$398 \pm 44$	$2025 \pm 176$	$5504 \pm 2781$	$20760 \pm 3785$				
<sup>109</sup> Cd	41 + 19	83 + 17	150 + 75	1111 + 366				
<sup>57</sup> Co	6 + 4	161 + 13	54 + 18	221 + 73				
<sup>54</sup> Mn	29 + 3	92 + 4	66 + 19	120 + 23				
<sup>65</sup> Zn	$62 \pm 65$	$656 \pm 53$	$1515 \pm 348$	$2669 \pm 183$				
Egg compartments								
Developmental stages	XVII	XVIII	XVII	XVI				
	5274 . 788	1946 + 222	505 × 05	$0147 \pm 101$				
whole eggshell	$52/4 \pm 788$	$1840 \pm 223$	$393 \pm 93$	$2147 \pm 181$				
Nidamental	$690/\pm /81$	$2180 \pm 245$	$894 \pm 241$	-				
Oviducal	< d1	$310 \pm 154$	$307 \pm 262$	-				
Chorion	< dl		$114 \pm 95$	$2147 \pm 181$				
Embryo	< dl	$5253 \pm 891$	$997 \pm 1245$	$1525 \pm 284$				
Vitellus	< dl	$1242 \pm 477$	$398 \pm 393$	$6953 \pm 1067$				
PVF	< d1	$27.8 \pm 9.1$	$8.3 \pm 7.0$	na				
<sup>109</sup> Cd CF								
Whole eggshell	$3158 \pm 635$	$590 \pm 238$	$74.4 \pm 7.6$	< 300				
Nidamental	12295 + 1990	663 + 274	40 + 8	-				
Oviducal	134 + 28		73 + 24	-				
Chorion	< d1	$237 \pm 73$	169 + 33	< 300				
Embryo	< d]	< d1	514 + 333	595 + 186				
Vitellus	< d1	< d1	$136 \pm 104$	$433 \pm 165$				
PVF	< dl	< dl	$22.2 \pm 8.0$	na				
<sup>57</sup> C CE								
	1240 200	1522 200	100 07	1151 155				
Whole eggshell	$1249 \pm 308$	$1522 \pm 298$	$109 \pm 27$	$1151 \pm 177$				
Nidamental	$4009 \pm 1055$	$1550 \pm 311$	$33 \pm 10$	-				
Oviducal	$28 \pm 22$	1151 + 385	$75 \pm 44$	-				
Chorion	< dl		$332 \pm 109$	$1151 \pm 177$				
Embryo	< d1	< d1	$25.0 \pm 9.3$	$48.4 \pm 89.0$				
Vitellus	< dl	< d1	$21.0 \pm 10.9$	< 10				
PVF	< dl	< d1	$24.3\pm9.2$	na				
<sup>54</sup> Mn CF								
Whole eggshell	$217 \pm 9$	$174 \pm 45$	$137 \pm 40$	$489 \pm 160$				
Nidamental	646 + 65	200 + 54	159 + 53	-				
Oviducal	56 + 12		121 + 26	-				
Chorion	133 + 77	$52 \pm 11$	75 + 16	489 + 160				
Embryo	$27.9 \pm 1.7$	336 + 41	352 + 198	< 20				
Vitellus	27.5 = 1.7 $25 \pm 1.3$	$52.0 \pm 1.1$ $523 \pm 2.71$	$265 \pm 107$	< 5				
PVF	$5.4 \pm 2.7$	$3.9 \pm 0.6$	$3.4 \pm 4.8$	na				
657 m CE								
	EE21 . 007	1250 . 170	146 - 25	2020 - 270				
whole eggshell	5531 ± 997	$1359 \pm 169$	$146 \pm 25$	$2929 \pm 370$				
Nıdamental	$12474 \pm 2116$	$1490 \pm 678$	$182 \pm 31$	-				
Oviducal	$42 \pm 36$	678 + 205	$125 \pm 44$	-				
Chorion	< dl		$66 \pm 14$	$2929 \pm 370$				
Embryo	< dl	$335 \pm 185$	$1136 \pm 428$	$164 \pm 30$				
Vitellus	< dl	$194 \pm 133$	$1032 \pm 385$	$99 \pm 26$				
PVF	< dl	$1.2 \pm 0.4$	$10.0 \pm 1.5$	na				

na: not analysed; dl: detection limit; PVF, perivitelline fluid 361