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ELECTRIC AND SPECTROSCOPIC ANALYSIS OF A PURE NITROGEN MONO-FILAMENTARY DIELECTRIC BARRIER DISCHARGE (MF-DBD) AT 760 TORR

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Abstract

As far as we know, VUV emissions of nitrogen MF-DBDs have never been reported before.

Mono-filamentary dielectric barrier discharge (MF-DBD), occurring within 1-mm gap of atmospheric pressure pure nitrogen and operating with a sinusoidal electric supply at about 8 kHz, is studied in this paper. A thorough electrical analysis allows experimental determination of the ignition and extinction voltages, respectively (15750 ± 50) V and (2097 ± 7) V, the injected energy (158 ± 2) J and charge (17.22 ± 0.22) nC in a single filament. The mean axial reduced electric field is equal to (644 ± 2) Td at ignition. An empirical technique is proposed to evaluate these discharge parameters by avoiding bulky calculations. Optical emission spectroscopic measurements of the vacuum ultraviolet (VUV), ultraviolet (UV), visible and near infrared (IR) emissions are presented and discussed. Two atomic nitrogen lines attributed to the decay of the N[$2s^2p^23s^2P$] triplet towards N[2s²2p³ ²D^o] level are observed at 150 and 175 nm, together with the Lyman-Birge-Hopfield system $(N_2[a^{-1}\Pi_g] \rightarrow N_2[X^{-1}\Sigma_g^+])$ in the VUV range. The Second Positive system $(N_2[C^{-3}\Pi_u] \rightarrow N_2[B^{-3}\Pi_g])$ dominates the UV and visible-blue spectra. The (0-0) transition of the First Negative system $(N_2^+[B^2 \Sigma_u^+] \rightarrow$ $N_2^+[X^2\Sigma_g^+]$ peaking at 391.4 nm, the First Positive system ($N_2[B^3\Pi_g] \rightarrow N_2[A^3\Sigma_u^+]$) and the Herman infrared transitions $(N_2[C'' \Pi_u] \rightarrow N_2[A' \Sigma_g])$ also present. Both our VUV and near IR spectra are consistent with recently reported results in hollow cathode and cylindrical DBDs. The electrical and spectroscopic experimental results reported here are useful for ongoing and forthcoming modelling of filamentary nitrogen dielectric barrier discharges.

Key-words: Dielectric barrier discharge, Filamentary discharge, Nitrogen, Electrical analysis, Optical emission spectroscopy, Vacuum ultraviolet and ultraviolet radiations, Visible light and near Infrared emissions.

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1. INTRODUCTION

Cold non-equilibrium atmospheric plasmas achieved with Dielectric Barrier Discharges (DBDs) are of great interest for production of energetic photons (hv > 10 eV), radicals or ion sources for industrial use. Some of these applications concern photo-chemical treatment of water [*Azrague K et al 2005, Gonzalez et al 2004*], volatile organic compound (VOC) remediation [*Biomorgi J et al 2005, Koutsospyros A et al 2004*], biochemical decontamination and remediation of toxic gases [*Herrman H W et al 1999*], realisation of non-coherent vacuum ultraviolet (VUV: 100 nm < λ < 200 nm) or ultraviolet (UV: 200 nm < λ < 400 nm) sources [*Kurunczi P et al 1999, Masoud N et al 2004*], material deposition [*Babayan S E et al 1998*], H₂ generation for fuel cells and diesel reforming [*Qiu H et al 2004, Becker K H et al 2004*], exhaust treatment [*Dietz et al 2004*] ...

As far as non-coherent radiation DBD sources are concerned, VUV sources are of topical interest for a variety of industrial purposes such as plasma processing [Kogelschatz U et al 1999], surface cleaning [Korfatis G et al 2002] and modification [Wagner H-E et al 2003, Borcia G et al 2003], sterilization, decontamination and medical care. Recently some emerging scientific investigations were performed in neon [Carman R J et al 2010]. Rare gas DBDs can be used for excilamps [Zhang J-Y and Boyd I W 2000, Boichenko A M et al 2004], mercury-free lamps [Jinno et al 2005], for oxidation of silicon at 250 °C [Kogelschatz U et al 2000]. Operating in xenon or neon, they produce VUV or EUV (10 nm $< \lambda < 100$ nm) emissions with an efficiency as high as 50 to 60% [Vollkommer F and Hitzschke L 1996, Hitzschke L and Vollkommer F 2001, Mildren R P and Carman R J 2001, Carman R J and Mildren R P 2003, Carman R J et al 2004, Merbahi N et al 2007, Carman R J et al 2010]. In order to understand their behaviour, stable MF-DBDs were must be achieved at frequencies around tens of kilohertz [Adler F and Muller S 2000, Sewraj N et al 2009, Merbahi N et al 2004, Merbahi N et al 2010]. In such circumstances, for each half-period all the observed emissions can be unambiguously attributed to a single filament whose time-origin is well defined by the fast rising supply current.

During the last three decades a growing attention has been paid to atmospheric DBDs containing nitrogen because of their potentiality for a wide variety of applications [*Kogelschatz U et al 1999*]. Polymer surfaces are often smooth and it is rather difficult to wet them, or to paint or print on them or to deposit adherent metal coatings. Such surfaces can be activated by a DBD treatment in open air. An increasing number of investigations are devoted to the decomposition of nitrogen and sulphur oxides in flue gases, and of VOCs emanating from various large-scale industrial plants, for which dc corona discharges technique is used. However, for applications where high specific powers or low gas temperature are required, DBDs constitute a good alternative to corona discharges.

Atmospheric DBDs are usually diffuse (homogeneous) or multi-filamentary [*Okazaki S et al 1993*]. Homogenous nitrogen based DBDs are often devoted to surface treatment [*Wagner H-E et al 2003*]. Nevertheless, random filaments which may occur at ignition or, to a lesser extent, during the steady state are detrimental to the quality of the treatment. For both type of DBDs, a good knowledge of the electrical, spectroscopic and kinetic behaviour of mono-filamentary discharge is required in order to optimize specific applications. In order to study filamentary nitrogen DBDs a few authors have achieved laboratory multi-filamentary [*Lukas C et al 2001*] or mono-filamentary DBDs in mixtures [*Kozlov K V et al 2005*] containing nitrogen. In contrast to rare gas DBDs, those occurring in nitrogen yield a variety of emissions over an extended spectral range from VUV to infrared.

The intent of this paper is to perform a simultaneous experimental analysis of both the electrical and spectroscopic performance of a pure nitrogen MF-DBD. The original feature is that spectroscopic measurements are performed on spatially stable nitrogen MF-DBDs, where emissions are recorded from VUV to near infrared emissions. It also constitutes a first step in providing essential electrical and spectroscopic data which will allow validation of electro-dynamic and kinetic modelling of these MF-DBDs which can be more easily controlled and modelled than multi-filamentary ones.

2. THE EXPERIMENTAL SETUP



Fig. 1: The experimental setup for electric and emission spectroscopic analysis of a MF-DBD (including photography)

The experimental setup allowing electrical and spectroscopic investigations of pure nitrogen MF-DBDs is shown in figure 1. It consists of a DBD chamber (cell) associated to a pumping/filling system, an electric supply as well as the electrical and spectroscopic measuring devices. The MF-DBD cell is a closed cylindrical glass chamber provided with two diametrical side-on fittings, each one equipped with a high voltage vacuum feed-through isolated (15 kV) from the chamber. Each electrode is a 4-mm diameter cylindrical aluminium rod, covered with a $d_{die} = 0.5 \text{ mm}$ thick insulating MAKOR on its flat front-end (relative electrical permittivity: $\epsilon_R = 4.0$). Such small plate-to-plate double barrier configuration electrodes ensures establishment of a unique filamentary barrier discharge for every half-cycle of the driving sinusoidal voltage $u_{sup}(t)$ [*Merbahi N et al 2004*] and was previously used to investigate rare gas DBDs. The gap distance was maintained at d = 1 mm, so as to achieve stable MF-DBDs in pure nitrogen at a pressure of $P_{N2} = 760 \text{ Torr}$, with our electric supply at a frequency of f = 7.8 kHz. Daily refreshed, the DBD cell was filled with 99.999% pure nitrogen (Air Liquide) and before filling up, it was pumped down to about 10⁻⁷ Torr. The impurities initially present in our gas were: O₂, H₂, CO, CO₂,

CNHM (< 0.1 p.p.m., each) and H_2O (< 0.5 p.p.m.).

Fig. 2a shows the complete electrical driving circuit with the relevant electrical ratings. The tunable sinusoidal voltage, delivered by a function generator, was amplified by a CREST C4000 (4000 W, in the bridge mode at 1 kHz) power amplifier whose output voltage was stepped up via a ferrite transformer (*voltage ratio 1:150*). The supply voltage $u_{sup}(t)$ available at its secondary coil was measured with a high voltage probe (Tektronix P6015A, bandwidth: 75 MHz), while the supply current $i_{sup}(t)$ was recorded via a resistance $R = 50 \ \Omega$ in series with the grounded electrode. Both electrical waveforms were recorded by using a 2.5 GHz (2 channels) digital sampling 500 MHz bandwidth oscilloscope (Tektronix TDS 5054B).



Fig. 2: (a) The electric driving circuit of the MF-DBD (b) The equivalent electric circuit of the MF-DBD cell

The front-end of the reactor was equipped with a transparent MgF_2 window for spectroscopic measurements. To record the visible and near infrared emission spectra we used a 0.5 *m* imaging visible spectrophotometer (*V_MC*: Acton Spectra Pro 2556i, in the Czerny-Turner configuration). The detecting device was a 32-bits 2D Intensified Charge Coupled Device camera (*ICCD*: Princeton Instruments, 512x512 pixels) was placed at its exit port. Its spectral domain lies within 360 and 920 nm. The emitted photons were guided to the entrance slit of the V_MC spectrophotometer via a 3 m-long optical fibre (*OF*: UV-Silicon LG-455-020-3).

For VUV measurements the optical fibre (OF) and the visible spectrophotometer (V MC)were replaced by a 0.3 m VUV monochromator (VUV MC, in the Czerny-Turner configuration as well) equipped with a fast solar blind photomultiplier tube (*PMT*). The PMT operated in the photon counting mode and its anode pulses were delayed through a delay line (DL) then processed by a constant fraction discriminator (CFD-Sig) and finally sent to a multi-channel analyzer (Stanford Research SR430). In order to classify these photons according to their emission instant, the time origin of the discharge needs to be unambiguously detected. For each micro-discharge this origin instant was defined on the leading edge of the fast current pulse via a fast amplifier and a constant fraction discriminator (CFD-Trig). No significant difference is observed for the electric and spectroscopic behaviours of the MF-DBD between the positive half-cycle and the negative one. So, the VUV emission spectrum was recorded by summing all the photons of the luminescence decays accumulated over a great number of discharges of only the positive half-cycle of the supply voltage, for each central exit wavelength of the monochromator, between 120 and 200 nm. For all VUV records, we verified that the probability for more than one photon to be recorded in a single channel was always less than 1 %.

All the emission spectra were performed on the first interference order of the grating. The exit spectral width $\Delta \lambda_{exit}$ and the wavelength step $d\lambda_{step}$ were equal. Our *VUV*, *UV*, visible and near *IR* optical emission spectra are given without any correction for the wavelength-dependent of the relevant detection device. For VUV measurements (130-200 nm: 2400 grooves/mm grating), the MgF₂ exit window was the high-pass filter whereas for UV emissions (200-400 nm: 2400 grooves/mm grating), air was a natural high-pass filter. For the 400-600 nm (2400 grooves/mm) and 600-900 nm (1200 grooves/mm grating) spectral ranges, either the *Corning 0-53* (cutting wavelength: 340 nm) or the *MTO J517a* (cutting wavelength: 517 nm) high-pass filter was

respectively used. The exit spectral widths were respectively 0.5 nm, 0.016 nm, 0.020 nm and 0.073 nm for these spectra.

3. ELECTRICAL DIAGNOSIS OF A MF-DBD

Fig. 2b shows the equivalent electrical circuit, which fairly describes the dynamic behaviour of an axially symmetric DBD [*Liu S and Neiger M 2001*]. The filamentary discharge is represented by a non-linear dissipative current source $i_{dis}(t)$. The capacitor C_{die} is the total capacity of both dielectric layers whereas C_{gap} accounts for the inter-electrode region devoid of any discharge. $C_{cell} = \frac{C_{gap} \cdot C_{die}}{C_{gap} + C_{die}}$ represents the total cell's capacity. The capacitor C_p represents the total

stray capacitors of the driving circuit due to the electrical leads and probes. In this model, we assume that C_{die} remains linear and that C_{gap} keeps the same value whether a discharge occurs or not. These assumptions will be commented in section 4.1. C_{gap} and C_{die} are calculated from geometrical considerations: $C_{gap} = \frac{\varepsilon_0 S}{d} = 0.25 \ pF$ and $C_{die} = \frac{\varepsilon_0 \varepsilon_R S}{2 d_{die}} = 1.00 \ pF$. The stray capacitor C_p was determined from $u_{sup}(t)$ and $i_{sup}(t)$ when no discharge occurs: $C_p = 0.44 \ pF$. Time origin is considered at the instant when the discharge voltage $u_{dis}(t)$ is maximum. This instant is obtained from our calculations. In our experimental conditions, the voltage drop $u_R(t)$ across the resistance R (50 Ω) was about (10.0 ± 0.1) V and was negligible compared to the amplitude of the supply voltage $u_{sup}(t)$. The voltage $u_{cell}(t)$ applied to the DBD cell was:

$$u_{cell}(t) \approx u_{sup}(t) = U_{Max} \sin(2\pi f t + 0.150)$$
(1)

When the voltage across the gap $u_{dis}(t)$ is high enough to induce a primary electronic avalanche, a single micro-discharge occurs for each half-cycle. The measured supply current $i_{sup}(t)$ is then a fast-rising pulsed current superimposed on a slow displacement current [*Merbahi N et al* 2004]. The current $i_{cell}(t)$ flowing through the DBD cell can be evaluated:

$$i_{cell}(t) = i_{sup}(t) - i_{C_p}(t) = i_{sup}(t) - C_p \frac{du_{sup}(t)}{dt} \approx i_{sup}(t) - C_p \frac{du_{cell}(t)}{dt}$$
(2)

In our experimental conditions, the current flowing through the stray capacitor $i_{Cp}(t)$ is so weak that $i_{cell}(t)$, calculated from equation (1), could hardly be distinguished from $i_{sup}(t)$: the amplitude of $i_{Cp}(t)$ was less than 0.7 mA. The current $i_{cell}(t)$ can also be considered as a fast and short current pulse $i_{cell_Pulse}(t)$ (this component exists only during the discharge phase) superimposed on a slow weak displacement one: $i_{cell_Displ}(t)$ (itself due to the cell's capacity C_{cell}):

$$i_{cell}(t) = i_{cell_Pulse}(t) + i_{cell_Displ}(t)$$
(3)

According to the electrical scheme given in Fig. 2b, the voltage drop $u_{die}(t)$ across both dielectric layers can be calculated from $i_{cell}(t)$. The voltage drop $u_{dis}(t)$ across the gap is then easily deduced. In order to evaluate the electrical power injected in the micro-discharge, the discharge current $i_{dis}(t)$ should also be determined:

$$i_{dis}(t) = \left(1 + \frac{C_{gap}}{C_{die}}\right) i_{cell}(t) - C_{gap} \frac{du_{cell}(t)}{dt} \quad (4)$$

And substituting equation (3) in equation (4), $i_{dis}(t)$ is found to be proportional to the pulsed component $i_{cell_Pulse}(t)$ of $i_{cell}(t)$:

$$i_{dis}(t) = \left(1 + \frac{C_{gap}}{C_{die}}\right) i_{cell_Pulse}(t) \quad (5)$$

 $i_{cell_Pulse}(t)$ can be determined by subtracting the displacement component $i_{cell_Displ}(t)$ from $i_{cell}(t)$.

$$i_{cell_Displ}(t) = C_{cell} \frac{du_{cell}(t)}{dt}$$
(6)

The energy E_1 and the charge Q_1 injected into a single filament are respectively:

$$E_{1} = \int_{0}^{\frac{1}{2j}} i_{dis}(t) \cdot u_{dis}(t) dt$$
(7)
$$Q_{1} = \int_{0}^{\frac{1}{2j}} i_{dis}(t) dt$$
(8)

4. RESULTS AND DISCUSSION

4.1. Electrical behaviour of the nitrogen MF-DBD

In order to analyze the electrical behaviour of the MF-DBD, representative waveforms of u_{sup} and of i_{sup} were recorded and processed with the macroscopic model previously described so as to determine the discharge voltage u_{dis} and current i_{dis} . The supply voltage u_{sup} and current i_{sup} are given in fig. 3a for one signal period T = 1/f. As explained, the sinusoidal cell voltage u_{cell} was almost equal to u_{sup} , and i_{cell} could hardly be distinguished from i_{sup} at f = 7.8 kHz.

The recorded quantities (amplitudes, frequencies, rise- and fall- times) of our electric signals, denoted by *x*, correspond to mean values of the N = 101 samples. The expanded uncertainty Δx of each quantity was calculated, according to the general rules established by ISO in Guide to the Expression of Uncertainty in Measurement [*ISO/IEC Guide 98: 1995*], by multiplying the standard deviation σ_x of each quantity *x*, by a coverage factor of *two* ($\Delta x = 2 \sigma_x$) which defines a level of confidence of approximately 95%. The standard deviation was obtained by the appropriate functionality of the digital oscilloscope.



Fig. 3a: The cell voltage and supply current at f = 7.8 kHz (one period)

Fig. 3b: The cell voltage and supply current (with an expanded scale: $t_B < t < t_E$).

For the supply voltage: $U_{Max} = (13220 \pm 43) V$ and $f = (7.79 \pm 0.02) kHz$. The amplitude of the cell current is $I_{cell_Max} = (200.0 \pm 2.5) mA$, and its measured rise-time and fall-time are respectively $(36.6 \pm 3.5) ns$ and $(124 \pm 4.5) ns$. For the amplitude of u_{sup} and i_{sup} the recorded

relative uncertainties are respectively: $\frac{\Delta U}{II} = 0.32\%$ and $\frac{\Delta I}{I} = 1.23\%$. We assume that our voltages and currents are all known with the above relative uncertainties.

In Fig. 3b the supply voltage and current are depicted with an expanded time scale corresponding to the discharge phase, from the ignition time (t_B) to the extinction time (t_E) . In Fig. 4a, we represent the discharge voltage u_{dis} and current i_{dis} as well as the u_{cell} and the voltage drop across both dielectrics u_{die} , during the positive half cycle. The various currents and voltages of the circuit are displayed in Fig. 4b, with an expanded time-scale so that the different discharge phases can be followed. For the positive half-cycle, we can thus distinguish the following phases:



(over a half period)



4.1.1 Establishment of the mono-filamentary discharge

(i) $t_0 = -19246$ ns $< t < t_B = 0$ ns: Pure capacitive divider circuit. Initially ($t = t_0 = -19246$ ns) the voltage drop across both dielectrics $u_{die}(t_0)$ is negative, due to the net charge accumulated on the dielectric surfaces during the preceding micro-discharge (negative half cycle). The initial voltage across the discharge itself is: $u_{dis}(t = t_0) = U_0 = \frac{|q_{die}(t = t_0)|}{C_{die}} = (7.500 \pm 0.024) \, kV$. u_{dis} then

increases to its maximum value, according to capacitor divider equation (9), until the axial

electrical field across the gap is high enough to ignite the gas, at $t = t_B$. During this phase i_{cell_Pulse} and hence i_{dis} are both equal to 0 A.

$$u_{dis} = \left(\frac{C_{die}}{C_{gap} + C_{die}}\right) u_{cell} + U_0 = 0.8 u_{cell} + U_0 \quad (9)$$

(ii) t = 0 ns: Gas ignition. The gas ignites (i_{dis} starts) when the space charge is high enough to accelerate "seed electrons" present in the gas. At this instant t_B the gap voltage attains its maximum value, the ignition voltage $V_B = u_{dis}(t_B) = (15.752 \pm 0.051) kV$, which corresponds to a mean axial reduced field (E/N) of (644 ± 2) Td (where $1 Td = 10^{-17} V cm^2$). The cell and dielectric voltages are respectively: $u_{cell}(t_B) = (10.725 \pm 0.035) kV$ and $u_{die}(t_B) = (-5.027 \pm 0.016) kV$.

(iii) 0 ns < t < 849 ns: Townsend stabilization phase or background charge phase. After ignition, i_{dts} increases only slightly while u_{dts} decreases very slightly around 15.7 kV. This time lag, which is about 0.8 µs here, is usually observed in nitrogen filamentary DBDs. According to *Yurgelenas* and Wagner [YurgelenasY V and Wagner H-E 2006] who performed a computational model of a barrier discharge in air (d = 1.2 mm) at atmospheric pressure N₂/O₂ binary mixtures, residual charges play an important role in the micro-discharge (*MD*) formation. According to their calculations, in this first phase, which they labelled as a Townsend phase by stabilization, the "early" electrons are multiplied through ionization processes and then accumulated on the anode dielectric, thus leaving a net ionic density in the gap. It follows that ions dominate in the gap. The discharge current (and axial electron density) increases slowly until the charges accumulated on the dielectrics reduce the axial electric field. During this stabilization phase where background charges are accumulated [*Capeillère J 2009*], the axial field remains nearly constant along the gap's axis. It lasts several µs in air at atmospheric pressure, and the electron number density remains quite low.

In other gases like argon and xenon this phase is generally skipped [*Merbahi N et al 2004*] [*Merbahi et al 2007*]. So it seems that, in nitrogen based DBDs ignition processes followed by the avalanche propagation should be somewhat different from the classical Townsend breakdown usually encountered at such high *p.d* values (*76 Torr.cm*) in noble gases.

(iv) 849 ns < t < 894 ns: Electronic avalanche propagation or space-charge-dominated phase. It then follows another phase characterized by an exponential growth of the charged particles (electrons and ions) in every position of the gap. The current i_{dis} then increases much faster. The electrons attain the anode at about t = 894 ns, which corresponds to the first current peak around 30 mA. In the meanwhile u_{dis} decreases slowly. A semi-log plot of i_{dis} (not given here) shows that during the electronic avalanche i_{dis} rises quasi-exponentially with a time constant of $\tau_{avalanche} \approx$ 100 ns, quasi-independent of pressure from 100 Torr to 760 Torr.

(v) 894 ns $< t < t_M = 923$ ns: Propagation of the cathode directed ionization wave (IW) or

filamentary phase. Following the preceding space-charge-dominated phase, a fast cathode directed ionization wave (streamer) propagates towards the cathode due to secondary electrons emitted by the cathode dielectric. During this phase (duration = 29 ns), i_{dis} increases very rapidly. When the ionization wave reaches the cathode, the discharge current rise slows down because of the additional charge deposited on the cathode layer. The usual characteristic "flagging" of the discharge current [*Merbahi N et al 2004*] is not observed in our conditions probably because the streamer propagates too fast, the space charge being quite high at this pressure.

The charge deposited on both dielectrics $t > t_M$, lowers the axial electric field. However, the radial electric field becomes strong enough to initiate a surface discharge in the vicinity of the cathode dielectric. The formation of a thin luminous cathode layer starts. Up to $t = t_M$, this phenomenon gains over the lowering of the axial electric field.

(vi) $t_M < t < t_E = 2000 \text{ ns: Cathode layer formation}$. When $t = t_M$, i_{dis} attains its maximum value, the charge deposited on both dielectrics predominates over the effect of the increase of the conducting cross-section available at the cathode. The cathode layer grows up while the discharge current i_{dis} rapidly falls down. We verified that the current oscillations present near the end of this discharge extinction did not correspond to reflections of the pulsed current along the electrical leads.

(vii) $t = t_E$: Discharge quenching. At $t = t_E$ when plasma conductivity is greatly reduced mainly due to accumulation of electrons and ions on the opposite surface of the dielectrics, the discharge stops: $i_{dis} = 0 \ mA$. The gap voltage then attains its minimum value $V_E = (2097 \pm 7) \ V$ (extinction voltage). The corresponding mean axial reduced field is $(86.0 \pm 0.3) \ Td$. At this particular instant, the cell and dielectric voltages are respectively: $u_{cell}(t_E) = (11.365 \pm 0.037) \ kV$ and $u_{die}(t_E) =$ $(9.268 \pm 0.030) \ kV$.

(viii) $t > t_E$: Post-discharge. The luminescence decay of the main VUV emission at 150 nm shows that this emission lasts more than 1 µs after ignition (section 4.2.2), while the current is completely quenched. Between $t = t_E$ and $t = t_0 + T/2$, no discharge exists again ($i_{dis} = 0 \ mA$) and u_{dis} is consistent with: $u_{dis} = \left(\frac{C_{die}}{C_{gap} + C_{die}}\right) u_{cell} - U_0$.

Then, the discharge of the negative half-cycle occurs, similarly to the previously described successive phenomena.

4.1.2 Characteristics of the mono-filamentary discharge

During the positive filamentary discharge, the cell voltage varies from 10.72 to 11.36 kV (Fig. 3b), showing that, between t_B and t_E , the supply voltage drop is about 5.8%, around 11.04 kV. Moreover, the current i_{gap} flowing through C_{gap} is negative (u_{dis} decreases monotonically from V_B to V_E). Subsequently, $i_{dis} > i_{cell}$, as predicted by equation (5). Due to the respective values

of C_{gap} and C_{die} , the discharge's maximum current ((246.8 ± 3.0) mA) is 25% higher than the amplitude of i_{cell} : (198.2 ± 2.4) mA.

The different characteristic times of the discharge current i_{dis} are calculated for the electrical sample given here. The full wave half maximum (*FWHM*) of i_{dis} and the duration during which i_{dis} remains above 10% of its maximum value are respectively $t_half_cal = (21.6 \pm 2.1)$ ns and $t_full_cal = (169.2 \pm 6.2)$ ns. The rise- and fall- times of i_{dis} are respectively (35.2 \pm 3.4) ns and (125.5 \pm 4.6) ns.

The electric energy E_1 injected in a single discharge and its uncertainty $\Delta E_1 = E_1 \sqrt{\left(\frac{\Delta U_{Max}}{U_{Max}}\right)^2 + \left(\frac{\Delta I_{Max}}{I_{Max}}\right)^2} \text{ are also calculated: } E_1 = (157.9 \pm 2.0) \ \mu J. \text{ Compared to argon}$ [Merbahi N et al 2004], krypton [Merbahi N et al 2010] and xenon [Merbahi N et al 2007], for which typical values of E_1 do not exceed 5 μJ , such a high value is not astonishing for nitrogen, in regard to both a high ignition voltage V_B and discharge current amplitude, despite the short pulse duration. The electric charge Q_1 injected in one discharge (its uncertainty $\Delta Q_1 = Q_1 \frac{\Delta I_{Max}}{I_{Max}}$), is: $Q_1 = (17.72 \pm 0.22) \ nC.$

Much emphasis needs to be laid upon the fact that the displacement current of i_{cell} perfectly fits the cell current during the whole capacitive voltage divider phase, meaning that C_p is correctly determined. Besides, during the discharge phase, both C_{die} and C_{gap} seem to remain constant. As far as C_{die} is concerned, this is not astonishing because the charges are deposited on the anode and cathode surfaces. However, even if the *MD* occupies a certain gap volume due to the filament, the funnel shape anode charges and the disc of the cathode layer, the gap capacitance C_{gap} remains unaffected in the short duration of the discharge. Macroscopically, the gap behaves as a linear capacitor. Fig. 4b shows that after discharge quenching, u_{dis} never exceeds u_{die} during the positive half-cycle. This excludes the occurrence of a second *MD* before the negative half-cycle. If wrong values of C_{die} and C_{gap} are used, the calculated waveform u_{dis} can exceed V_B in the post-discharge phase and most of the discharge parameters (mainly V_B and V_E) will then be incorrectly determined. Indeed, the waveforms of u_{dis} and i_{dis} are very sensitive to C_p , C_{gap} and C_{die} . So our results were crosschecked by calculating the injected charge by two different manners.

$$Q_{1} = \int_{t_{B}}^{t_{E}} i_{dis}(t) dt = \int_{t_{B}}^{t_{E}} i_{cell}(t) dt - \int_{t_{B}}^{t_{E}} i_{gap}(t) dt = C_{die} \{ u_{die}(t_{E}) - u_{die}(t_{B}) \} + C_{gap} (V_{B} - V_{E}) = (14.31 + 3.42) \ nC = 17.72 \ nC$$

and

$$Q_{1} = \left(C_{gap} + C_{die}\right)\left(V_{B} - V_{E}\right) + C_{gap}\left\{u_{cell}(t_{E}) - u_{cell}(t_{B})\right\} = (17.08 + 0.64) nC = 17.72 nC$$

The last detailed estimation of Q_1 shows that the charge injected in a MD is merely (96%) the charge variation for the capacitor equivalent to two parallel capacitors C_{gap} and C_{die} . The injected energy is: $E_1 \approx \frac{1}{2} (C_{gap} + C_{die}) (V_B^2 - V_E^2) = 153 \,\mu J$. So, provided that t_B and t_E can be estimated by

visualising $u_{dis}(t) = u_{cell}(t) - \frac{\int_{0}^{t} i_{cell}(t') dt' + q_{die}(t_0)}{C_{die}}$, Q_1 and E_1 can be simply evaluated in the following

manner, without the determination of *i*_{dis}:

$$V_{B} = u_{dis}(t_{B}) \qquad V_{E} = u_{dis}(t_{E})$$

$$E_{1} \approx \frac{1}{2} (C_{gap} + C_{die}) (V_{B}^{2} - V_{E}^{2}) \qquad Q_{1} = (C_{gap} + C_{die}) (V_{B} - V_{E}) + C_{gap} \{ u_{cell}(t_{E}) - u_{cell}(t_{B}) \}$$

4.2. Spectroscopic results and discussion



Fig. 5 depicts the UV emission spectrum of the atmospheric pressure N₂ MF-DBD recorded within 200 and 400 nm. The most intense emissions observed between 300 and 400 nm can be attributed to the N₂ Second Positive System (SPS: N₂[C ${}^{3}\Pi_{u}$] \rightarrow N₂[B ${}^{3}\Pi_{g}$]) emission bands, with a dominant emission (0,0) band head at 337.1 nm. Between 200 and 300 nm (Fig. 5b), a few weak double head characteristic emissions are observed and are attributed to the NO γ bands, even if laboratory 99.999% pure nitrogen is used. We believe that oxygen is present in the discharge, due to unavoidable impurity traces and sputtering of our dielectric material.

4.2.2 Visible spectrum

The visible spectrum recorded between 400 and 600 nm (Fig. 6) is mainly dominated by the SPS emissions. The weak emission at 391.4 (Fig. 6b), whose rotational blue wing looks different to SPS ones, is attributed to the (0-0) transition of the N₂ First Negative System (FNS: N₂⁺[B ${}^{2}\Sigma_{u}^{+}] \rightarrow N_{2}^{+}[X {}^{2}\Sigma_{g}^{+}])$. The upper level of this emission is very high (19 eV), can be reached by either likely step ionisation via a relay metastable molecule or by less likely direct electronic collisions, in the very short cathode directed streamer phase (*814 ns < t' <923 ns*). Subsequently, the FNS is expected to be very weak, and its existence is not astonishing since our discharge is

filamentary. If any weak FNS (0-1) transition emitting at 427.8 nm is present it can't be distinguished from the characteristic SPS (1-5) band at 426.97 nm, bearing a characteristic SPS rotational blue wing. The SPS (2,5) band (band-head at 394.3 nm) is also present in our spectrum. Another very weak molecular emission peaking at 557.7 nm is also present. It may correspond to the $O(^{1}S)N_{2}$ excimer emitting according to : $O(^{1}S)N_{2} + \rightarrow O(^{1}D) + N_{2} + hv_{558}$. This emission was also observed by *Kozlov K V et al* [*Kozlov et al 2005*] in a diffused N₂/O₂ atmospheric pressure DBD. In atmospheric air MF-DBD, they did not observe this emission, but also recorded a very weak FNS (0-0) transition. More recently, *Meiners A et al* [*Meiners A et al 2009*] reported both (0-0) and (0-1) FNS transitions in a multi-filamentary N₂ BDB.



Fig. 6: Emission spectra ($\Delta \lambda_{exit} = 0.02 \text{ nm}$): (a) 400–600 nm (b) FNS 0-0 transition at 391.4 nm) (c) ON₂ excimer

4.2.3 Near IR emission spectrum

Fig. 7 shows the emission spectrum of the MF-DBD between 600 and 900 nm. The Herman IR System (HIRS: N₂[C" ${}^{5}\Pi_{u}$] \rightarrow N₂[A' ${}^{5}\Sigma_{g}{}^{+}$]) dominates the spectrum with emissions at 703.3 (3-1) nm, 709.4 nm (2-0), 747.2 nm (2-1), 755.8 nm (1-0), 810.1 nm (0-0), 839.7 nm (1-3) and 854.9 nm (1-2). The values given between brackets are the vibrational levels of the molecular transitions. At 760 Torr, N₂[C" ${}^{5}\Pi_{u}$] molecules can easily be populated via 2-body collisions

between the metastable N₂[A ${}^{3}\Sigma_{u}{}^{+}$]) and their vibrational levels depend on the electron energy distribution. The emissions recorded between 650 and 670 nm are attributed to the N₂ First Positive System (FPS: N₂[B ${}^{3}\Pi_{g}$] \rightarrow N₂[A ${}^{3}\Sigma_{u}{}^{+}$]), namely at 662.4 nm (6-3), 678.8 nm (4-1), 770.5 nm (5-2) and 775.4 nm (2-0).



Fig. 7: Emission spectrum in the 600 to 900 nm range ($\Delta \lambda_{exit} = d\lambda_{step} = 0.073$ nm)

All the UV and visible emissions shown in fig. 5 were reported by many authors ([*Meiners A et al 2009*], [*Kozlov et al 2005*]) for nitrogen or air filamentary plasmas. *Meiners et al* [*Meiners A et al 2009*] observed both the HIRS and FPS in the multi-filamentary N₂ dielectric barrier discharge with flowing nitrogen. *Cermak et al* [*Cermak et al 2007*] reported similar FPS and HIRS emissions in a plane-to-plane DBD. *Simek et al 2001* studied a pulsed positive primary atmospheric pressure nitrogen streamer in a wire-cylinder configuration corona discharge. Above 700 nm, their observations are quite similar to ours.

4.2.4 VUV emission spectrum

For the VUV emissions whose wavelengths are situated between 130 and 180 nm (fig. 8), we observed a main narrow emission (Full Wave Half Maximum of about 1 nm) at 150.0 nm together with some other weak emissions between 130 and 174 nm. Between 180 and 200 nm, did not record any emission because of the upper cut-off wavelength of our extremely solar blind

photomultiplier tube at 180 nm. No significant emission was present between 120 and 130 nm. These weak emissions are attributed to the Lyman-Birge-Hopfield System (LBHS) which arises from transitions issuing from the N₂[a¹ Π_g] states towards the N₂[X¹ Σ^+_g] ones. The transitions and wavelengths of the observed emissions are given in table A.1 (appendix), where the reported wavelengths and relative intensities correspond to values given in reference [*Lofthus and Krupenie 1977*]. A close look at these relative intensities suggests that the 0-2, 0-4, 1-0, 1-1, 1-4, 1-5 and 1-6 transitions of the LBH system are likely. If it occurs, the 0-3 transition is unexpectedly weak while the 0-5 one is not present in our spectrum. As specified in table A.1 and as expected due to the relative positions of the potential curves of the involved molecular nitrogen states [*Lofthus and Krupenie 1977*], the 0-2 transition should be the most intense one. The intense narrow emission peaking at 150 nm is therefore too strong to be attributed to the 0-1 LBHS transition. It is attributed to a nitrogen atomic line issuing from the N[2s²2p³ 2D^o] state.



Fig. 8: Emission spectrum in the VUV range: 130 to 200 nm ($\Delta \lambda_{exit} = d\lambda_{step} = 0.5$ nm)

The N[2s²p²3s ²P] excited atom can also undergo a radiative decay (175 nm) towards the N[2s²2p³ ²P°]. The N[2s²p²3s ²P] level has a very short lifetime of 2 ns. This means that if atomic

nitrogen is produced, it can emit two emissions at 150 nm and 175 nm, even at 760 Torr. The temporal luminescence decays of these atomic emissions are depicted with linear scales in fig. 9a, where the integrated values of the photon counts over the total emission width of each radiation are corrected for the quantum efficiency of the photomultiplier tube: *11% at 150 nm* and *1.6% at 175 nm*. The time lag of longer than 1 µs suggests that these emissions occur after the streamer propagation phase. They may follow dissociative recombination processes, such as for example reaction R12 of table 1. We can consider that $N_2^+(X^2\Sigma_g^+)$ ground state ions lead to the $N(2s^22p^23s^2P)$ excited atomic state, which may emit both the 150 or 175 nm line. According to *Rahman R et al [Rahman R et al 2004*], $N_2(A^3\Sigma_u^+)$ created by electron impact excitation is mainly responsible for populating N(2P) via collisions with N(⁴S) atoms followed by electron impact of the N(²P₀) atomic by-product. Note that the dissociation of ground state N₂ dimers requires 9.8 eV [*Luo J et al 1999*] which can be provided by the MF-DBD.



Fig. 9: Luminescence decays of the 150 and 175 nm emissions (5 ns resolution) with (a) linear (b) semi-log scales.

The temporal luminescence decays of the nitrogen atomic lines are plotted in a semi-log scale in fig. 9b. Both emissions have similar time constants but the sharp maxima clearly suggest that the evolution laws of the photon counts are non-exponential This indeed provides evidence in favour of electronic dissociative recombination processes.

To our knowledge, VUV emissions of nitrogen MF-DBDs have never been reported before. *Masoud N et al [Masoud N et al 2005]* observed these emissions in a 300 Torr nitrogen radio-frequency (13.56 MHz) cylindrical dielectric barrier discharge and also attributed them to nitrogen atomic lines. In their spectrum where emissions from impurities were present, the peak intensity of the 175 nm line is about half the 150 nm one.

Rahman R et al [Rahman R et al 2004] also attribute these emissions recorded in a radiofrequency (rf) driven Ar/N_2 hollow slot micro-plasma, operating in open air, to nitrogen atomic lines. In their experiments the recorded emissions could arise either from the active rf discharge or from the downstream afterglow region. Ambient air species such as O₂, H₂O could also intervene. Consequently they also observed some other nitrogen atomic lines (120 nm, 124.3 nm), hydrogen (121.60) and oxygen (130.5 nm) ones, as well as the OH and NH bands at respectively 308 nm and 315.8 nm.

Occurrence of the observed VUV, UV, visible and near IR emissions of our pure nitrogen MF-DBD at 760 Torr may be partly explained by the non-exhaustive reactions listed in table 1. Direct electronic excitation and ionisation are likely to occur during the electronic avalanche propagation and streamer phases where electrons are quite energetic. Metastable N₂($A^{3}\Sigma_{u}^{+}$: 6.17 eV, $\tau_{rad} = 13$ s) molecules can then be populated through reaction R1, while N₂⁺(X ² Σ_{g}^{+} : 15.5 eV) molecular ions can be created indirectly by reaction R10 (+9.5 eV) or to a lesser extent directly by reaction R8. Note that according to Cesses [*Cesses Y 2004*], in the steady operation metastable N₂(A ³ Σ_{u}^{+}) molecules are accumulated in the neighbourhood of both electrodes. Direct electronic impact can also lead to the population of higher excited molecular states N₂(B³ Π_{g} : 7.35 eV, τ_{rad} = 110 µs), N₂(C³ Π_{u} : 11.3 eV, τ_{rad} = 38 ns) by respectively reactions R2, R3 and R4. These molecules can also be easily populated via electronic excitation processes stepping from the long-life metastable molecules N₂(A ³ Σ_{u}^{+}) through reactions R5 (+1

eV), R6 (+2 eV) and R7 (+5 eV). During these phases, the highly excited $N_2(C^3 \Pi_u)$ molecule predominantly emits the SPS leading to $N_2(B^3 \Pi_g)$ which in turn emits the FPS. Furthermore, step-wise ionisation towards the $N_2^+(X^2 \Sigma_g)$ can also take place from $N_2(A^3 \Sigma_u^+: 9.5 \text{ eV} (R10))$. Further excitation leads to $N_2^+(B^2 \Sigma_u^+)$ which are responsible for the FNS.

Direct destantis mitation from the encound state N = 1 = 1										
<u>Direct electronic excitation from the ground state N_2 molecule :</u>										
$e - + N_2(X^T \Sigma_g^T)$	\rightarrow	$N_2(A^3\Sigma_u^{\dagger})$	+	e-	(6 - 7 eV)	(R1)				
$e - + N_2(X^{-1}\Sigma_g^{+})$	\rightarrow	$N_2(B^3\Pi_g)$	+	e-	(7.5 – 8 eV)	(R2)				
$e - + N_2(X^{-1}\Sigma_g^{+})$	\rightarrow	$N_2(a^1\Pi_g)$	+	e-	(8.5 – 9 eV)	(R3)				
$e - + N_2(X^{-1}\Sigma_g^{+})$	\rightarrow	$N_2(C^3\Pi_u)$	+	e-	(11 – 12 eV)	(R4)				
Step-wise electronic excitation	from th	e metastable molecu	ıle :							
$e- + N_2(A^3\Sigma_u^+)$	\rightarrow	$N_2(B^3\Pi_g)$	+	e-	(1 - 2 eV)	(R5)				
$e- + N_2(A^3 \Sigma_u^{+})$	\rightarrow	$N_2(a^1\Pi_g)$	+	e-	(2 – 3 eV)	(R6)				
$e- + N_2(A^3 \Sigma_u^{+})$	\rightarrow	$N_2(C^3\Pi_u)$	+	e-	(5 - 6 eV)	(R 7)				
Direct electronic ionisation from the ground state N_2 molecule :										
$e_{-} + N_2(X^{-1}\Sigma_g^{+})$	\rightarrow	$N_2^{+}(X^2\Sigma_g^{+})^{-}$	+	2e-	(15.5 eV)	(R8)				
$e_{-} + N_2(X^{-1}\Sigma_g^{+})$	\rightarrow	$N_2^+(B^2 \Sigma_u^+)$	+	2e-	(19 eV	(R9)				
Step-wise electronic ionisation	Step-wise electronic ionisation from the metastable N_2 molecule :									
$e- + N_2(A^3 \Sigma_u^+) \rightarrow$	N_2	$^{+}(X^{2}\Sigma_{g}^{+})$	+	2e-	(9.5 eV)	(R10)				
$e - + N_2(A^3 \Sigma_u^+) \rightarrow$	N_2	$^{+}(\mathrm{B}^{2}\Sigma_{\mathrm{u}}^{+})$	+	2e-	(13 eV)	(R11)				
Dissociative electronic recombi	nation	of the ground state	N ₂ mole	cular ion	:					
$e_{-} + N_2^{+}(X^2\Sigma_g^{+})$	\rightarrow	$N(2s^22p^23s\ ^2P)$	<u> </u>	+	$N(^{4}S^{0})$	(R12)				
Radiative decay of nitrogen mo	lecules	and atoms:								
$N_2(C^3\Pi_u, \upsilon)$	\rightarrow	$N_2(B^3\Pi_g,\upsilon')$		+	hv_{SPS}	(R13)				
$N_2^{+}(B^{-2}\Sigma_u^{+},\upsilon)$	\rightarrow	$N_2^{+}(X^{2}\Sigma_{g}^{+},\upsilon')$		+	$h\nu_{\rm FNS}$	(R14)				
$N_2(B^3\Pi_g,\upsilon)$	\rightarrow	$N_2(A^3\Sigma_u^+,\upsilon')$		+	$h v_{FPS}$	(R15)				
$N_2(C^{"5}\Pi_u,\upsilon)$	\rightarrow	$N_2(A' {}^5\Sigma_g^{+}, \upsilon')$		+	$h \mathbf{v}_{\mathrm{HIRS}}$	(R16)				
$N_2(a^{-1}\Pi_g, \upsilon)$	\rightarrow	$N_2(X \ ^1\Sigma_g \ ^+, \upsilon')$		+	$h\nu_{\text{LBHS}}$	(R17)				
$N(2s^22p^23s^2P)$	\rightarrow	$N(2s^22p^{3} {}^2D^{\circ})$		+	hv_{150}	(R18)				
N(2s ² 2p ² 3s ² P)	\rightarrow	$N(2s^22p^{3} {}^2P^{\circ})$		+	hv_{175}	(R19)				
<u>Pooling reaction of $N_2(A^3\Sigma^+_{\mu})$ metastables:</u>										
$\frac{1}{N_2(A^3\Sigma^+_u)} + N$	$T_2(A^3\Sigma)$	$\dot{u} \rightarrow N_{2}$	₂ (С" ⁵ П	u) +	$N_2(X \ ^1\Sigma^+_{\ g})$	(R20)				

Table 1: Main reactions occurring in the 760 Torr pure N_2 MF-DBD, deduced from the spectroscopic analysis

A full kinetic study of the post-excitation phase requires analysis of the pressure dependence of temporal decays of some visible emissions and is beyond the scope of this paper.

Conclusion

We report the electrical investigation of a high-pressure pure nitrogen MFDBD, together with its spectroscopic characteristics from the VUV to infrared emissions. As far as we know, VUV emissions of nitrogen MF-DBDs have never been reported before.

The observed emission spectrum can be attributed to a single filamentary discharge and is dominated by the second positive system in the UV and blue spectral ranges. The Herman infrared and first positive systems are emitted between 650 and 850 nm. Inevitable weak UV emissions of the NO- γ system are observed as well and "traces" of O(¹S)N₂ excimer emission at 557 nm. A weak emission of the (0-0) transition of the first negative system peaking at 391.4 nm is present and this is consistent with the visual observation of filamentary behaviour of our discharge. The corresponding ionic molecules are probably excited during the streamer propagation phase. Even at such a high pressure two significant nitrogen atomic lines are observed at 150 and 175 nm, lagging about 1 µs after discharge ignition, suggesting that the relevant exited nitrogen atoms likely result from recombination of jonic nitrogen molecules. Our emission spectra bear some discrepancies with the few results available for VUV and IR emissions. We believe that this is due to many combined reasons: the operating pressure, our operating discharge mode and possible impurities. The electrical analysis, performed with a macroscopic model, allows determination of the ignition and extinction voltages as well as the injected energy and electric charge. An empirical technique, based on the use of available digital oscilloscope, is proposed to rapidly evaluate all these quantities, without performing bulky calculations. The experimental results presented here provide valuable electrical and

spectroscopic *in-situ* data to validate electro-dynamic and kinetic modelling of filamentary discharges.

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Appendix

v' – v''	λ (nm) [Ref 1]	Intensity [Ref 1]	v' – v''	λ (nm) [Ref 1]	Intensity [Ref 1]
0-0	145.0	23.7	1 - 0	141.59	37
0 – 1	150.08*	72.2	1 – 1	146.42	62.6
0-2	155.45	100	1 – 2	151.53*	23.7
0-3	161.14**	92.6	1 - 4	162.66**	20
0-4	167.19	52.9	1-5	168.74	34.2
0-5	173.01	23	1 – 6	176.81	28.7

 Table A-1 : Emissions of the LBHS in the range of 130 – 180 nm (Ref 1: Lofthus and Krupenie 1977, Pg 204)

 (*): Cannot be distinguished from the more intense 150 nm emission of atomic nitrogen

 (**): Overlapping emissions due to our exit spectral width equal to 0.5 nm

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