

Dissolved carbon leaching from soil is a crucial component of the net ecosystem carbon balance

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3C signature. Leaching of DOC plus biogenic DIC was 11.9 ± C m-2 yr-1 for forests (median: 7.4 g C m-2 yr-1), 29.4 ± r grasslands (median: 34.2 g C m-2 yr-1), and 18.3 ± 3.5 g 2 yr-1 for croplands (median: 17.8 g C m-2 yr-1). The ge flux across land use systems was 19.4 ± 4.0 g C m-2 yr-1 an: 15.2 g C m-2 yr-1). Production of DOC in topsoils was vely related to their C/N ratio and its retention in B horizons of ils was inversely related to the ratio of organic carbon to iron duminium (hydr)oxides. Partial pressures of CO2 in soil air boil pH determined DIC concentrations and fluxes, but soil ons were often supersaturated with DIC in relation to soil air Leaching losses of biogenic carbon from grasslands equalled (median: 22%) of net ecosystem exchange (NEE) plus in inputs with fertilization minus carbon removal with harvest. In leaching increased the net losses from cropland soils by 24-1 (median: 25%). For the majority of forest sites, leaching of affected actual net ecosystem carbon balances because of neall solubility of CO2 in acidic forest soil solutions and large cosystem exchange. Leaching of CH4 proved to be insignificant ared with other fluxes of carbon or CO2-C equivalents. Ill, our results show that leaching losses are particularly tant for the carbon balance of agricultural systems.



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Abstract

Estimates of carbon leaching losses from different land use systems are few and their contribution to the net ecosystem carbon balance is uncertain. We investigated leaching of dissolved organic carbon (DOC), dissolved inorganic carbon (DIC), and dissolved methane (CH₄), at five forests, four grasslands, and three croplands across Europe. Biogenic contributions to DIC were estimated by means of its δ^{13} C signature. Leaching of DOC plus biogenic DIC was 11.9 ± 5.9 g C m⁻² yr⁻¹ for forests (median: 7.4 g C m⁻² yr⁻¹), 29.4 ± 8.0 for grasslands (median: $34.2 \text{ g C m}^{-2} \text{ yr}^{-1}$), and $18.3 \pm 3.5 \text{ g C m}^{-2} \text{ yr}^{-1}$ for croplands (median: 17.8 g C m⁻² yr⁻¹). The average flux across land use systems was 19.4 ± 4.0 g C m⁻² yr⁻¹ (median: 15.2 g C m⁻² yr⁻¹). Production of DOC in topsoils was positively related to their C/N ratio and its retention in B horizons of subsoils was inversely related to the ratio of organic carbon to iron plus aluminium (hydr)oxides. Partial pressures of CO₂ in soil air and soil pH determined DIC concentrations and fluxes, but soil solutions were often supersaturated with DIC in relation to soil air CO₂. Leaching losses of biogenic carbon from grasslands equalled 5–98% (median: 22%) of net ecosystem exchange (NEE) plus carbon inputs with fertilization minus carbon removal with harvest. Carbon leaching increased the net losses from cropland soils by 24–105% (median: 25%). For the majority of forest sites, leaching hardly affected actual net ecosystem carbon balances because of the small solubility of CO₂ in acidic forest soil solutions and large net ecosystem exchange. Leaching of CH₄ proved to be insignificant compared with other fluxes of carbon or CO₂-C equivalents. Overall, our results show that leaching losses are particularly important for the carbon balance of agricultural systems.

Introduction

Uptake and release of CO₂ by terrestrial ecosystems are crucial modulators of the climate system (e.g., Heimann & Reichstein, 2008). Reliable estimates of net ecosystem carbon balances (Chapin et al., 2006) are therefore fundamental to our understanding of global climate change and the development of appropriate mitigation strategies. Net ecosystem carbon balances can be determined by assessing changes in ecosystem carbon stocks over time (e.g., Liski et al., 2002; Sleutel et al., 2003). This approach is associated with large uncertainties, because annual changes are small compared to stock size, because of error propagation of measurements, spatial heterogeneity within ecosystems, and temporal or spatial extrapolation. A possible strategy to validate large-scale estimates of net ecosystem carbon balances is to compare them with independent estimates of carbon uptake derived from inverse modelling of CO₂ transport in the atmosphere (Janssens et al., 2003; Schulze et al., 2009). Such comparisons are hampered by the differing and uncertain system boundaries and the differing types of carbon fluxes measured. Atmosphere-based estimates of carbon uptake rates are often larger than increases in ecosystem carbon stocks because part of the carbon is exported in plant products (e.g. wood, grains, or forage), lost to the atmosphere in volatile organic forms, or leached in soluble forms (Janssens et al., 2003; Ciais et al., 2008). Siemens (2003) hypothesized that the leaching of dissolved inorganic and organic carbon from soils can explain a large part of the difference between atmosphere- and land-based estimates of the carbon uptake of European terrestrial ecosystems. Similarly, Ritchey et al. (2002), Cole et al. (2007), and Battin et al. (2009) argued that consideration of inland waters as components of terrestrial carbon budgets is necessary to assess the carbon cycle at the landscape scale. In line with these views, current estimates of carbon balances of European terrestrial ecosystems assume an average loss of 7 g C m⁻² vr⁻¹ with drainage, based on measurements of riverine fluxes (Schulze et al., 2009). Linking riverine carbon fluxes to those in terrestrial ecosystems is difficult because measurements of riverine fluxes not only integrate over a number of transformation processes in aquifers and surface waters, they often also average carbon fluxes from various land use systems within a particular catchment. Thus, better data on leaching from soils could considerably improve estimates of carbon budgets of terrestrial ecosystems. Such data are also essential to further assess the active processing of land-borne carbon in aquifers and inland waters (Cole *et al.*, 2007; Battin *et al.*, 2009).

There is evidence that carbon leaching from terrestrial ecosystems varies systematically with land use. Brye *et al.* (2001) reported a more than five-fold increase in dissolved carbon leaching during four years after the conversion of native tall-grass prairie to maize cultivation with fertilization and ploughing. Parfitt *et al.* (1997) found losses of dissolved inorganic carbon (DIC) of 14.7 g m⁻² yr⁻¹ from pasture and 0.7 g m⁻² yr⁻¹ from a pine forest under similar climatic conditions, but a reduction in soil pH(H₂O) from 5.6–5.9 under pasture to 5.3–5.5 under pine as a result of more intense acidification in the forest. Leaching of dissolved organic carbon (DOC) seems to be affected by ploughing and crop type (Vinther *et al.*, 2006). A better understanding of land use effects on dissolved carbon leaching is important because i) net ecosystem carbon balances are often grouped according to land use classes (e.g., cropland, grassland, forest; Janssens *et al.*, 2003) and ii) trends in riverine dissolved carbon fluxes seem to be related to land use (e.g., Raymond & Cole, 2003).

Carbon is dissolved in water not only as DOC and DIC but also as dissolved methane (CH₄). Despite its high global warming potential, dissolved CH₄ is commonly not included in measurements of dissolved carbon concentrations and has only been investigated in a small number of studies. For the Brocky Burn moorland catchment in Scotland, CH₄-C fluxes with stream water were less than 0.04% of CO₂-C fluxes (Dawson *et al.*, 2004). Even when considering a global warming potential factor of 25 for CH₄ (IPCC, 2007), its contribution to

fluxes of CO₂-C equivalents was smaller than 1%. Fiedler *et al.* (2006) found that CH₄ export (in CO₂-C equivalents) from a catchment in the Black Forest, Germany, was approximately 8% of the CO₂ export. The small contribution of CH₄ to overall carbon losses with stream water or spring water is due its poor solubility in water (1.5 10^{-5} mol 1^{-1} kPa⁻¹ at 20° C).

The goal of the present work was i) to determine leaching losses of DIC and DOC from soils of typical European land use systems and to relate them to the respective net ecosystem carbon balances and ii) to identify major factors controlling carbon leaching. In order to achieve a full carbon perspective, determinations of DOC and DIC leaching losses were complemented by analyses of CH₄ in soil air and soil water.

Materials and Methods

Instrumentation and Sampling

Twenty glass suction cups with a pore size <1 µm for soil water collection (ecoTech, Bonn, Germany) and eight Teflon suction cups (ecoTech, Bonn, Germany) for soil gas collection were installed at each of twelve European sites under different land use (Table 1, Fig. 1). Most of these sites were verification sites of the CarboEurope project (www.carboeurope.org) and/or flux sites of the NitroEurope project (Skiba et al., 2009). The probes were divided into two sets of 10 suction cups and 4 gas probes, which allowed for sampling of different treatments (e.g., intensive and extensive grazing at Laqueuille) or different landscape positions (e.g., slope and hollow at Easter Bush). Suction cups and gas probes were installed horizontally from a soil pit by inserting them into boreholes. Half the number of probes were installed beneath the A horizon (at approximately 30 cm depth), the other half was installed as deep as possible (65–120 cm depth; Table 1). Suction cups were connected to 2-liter glass bottles in an insulated aluminium box placed into the soil pit. Soil water was extracted by applying a vacuum of 40 kPa to the glass bottles after each sampling. In order to correctly measure DIC, degassing of CO₂ was minimized by using a headspace-free sample collection system similar to the one described by Suarez (1986). In this system, the soil water from the suction cup first flows through a 20-ml gas-tight vial before entering the evacuated 2-l bottle via a cannulae, which penetrates the septum of the vials just deep enough to allow the outflow of solution. At the Hainich and Wetzstein sites, soil water was additionally collected from two glass suction plates installed at 20 cm depth.

Sample collection started at different dates. Here, we present and discuss data that span the period from 1 October 2006 until 31 March 2008 at all sites except the Irish sites where soil water sampling commenced until 31 March 2009. Fluxes were calculated separately for summer (1 April until 30 September) and winter (1 October until 31 March).

Determination of soil solution pH, DIC, and CH₄ concentrations in soil water

Soil solution pH was determined using a micro glass electrode (blueline 16pH, Schott, Mainz, Germany) in the gas-tight 20 ml vials prior to DIC analysis. Concentrations of DIC of samples taken during the first winter 2006/2007 were analysed with a TOC-5050A analyser (Shimadzu Corp., Kyoto, Japan). Inorganic carbon was determined by acidification and sparging solution samples with CO₂-free synthetic air and subsequently detecting the evolving CO₂. Samples from summer 2007 and winter 2007/2008 were analysed using a gas chromatograph for simultaneous detection of CH₄ and CO₂. The Shimadzu GC-2014AF gas chromatograph was equipped with an AOC-5000 autosampler, a 1 m \times 1/8" HayeSep Q 80/100-mesh column, an electron capture detector and two flame ionization detectors, with one of them being coupled to a methanizer. Gas concentrations and total pressure were analysed in the headspace after shaking solutions at 90°C for 20 minutes. The gas concentrations in soil water were calculated from the headspace concentrations by applying Henry's law. Results were corrected for temperature, pressure-dependent residual gas concentrations and pH-dependent carbonate equilibrium (only for DIC). Tests with standard solutions showed that the Shimadzu TOC-5050A method and the GC method yielded comparable results for the DIC detection.

Dissolved inorganic carbon in soil water derives from either respiration, dissolution of carbonate minerals, or atmospheric CO₂. Because partial pressures of CO₂ in soil air are much higher than atmospheric partial pressures, the direct influence of atmospheric CO₂ on DIC concentrations in soil water is negligible. According to the stoichiometry of the reaction of respiratory CO₂ with calcium or magnesium carbonate (eq. [1]), the fraction of carbonateborne DIC in calcareous soils should approach 50% if chemical equilibrium is reached within the residence time of percolating water in the soil (e.g., Amiotte-Suchet et al., 1999).

 $CaCO_3 + CO_2 + H_2O = Ca^{2+} + 2 HCO_3^-$ [1]

In carbonate-free soils, respiratory CO₂ will react with silicates (e.g., feldspars) to form clay minerals and bicarbonate, which in this case is solely from soil respiration (e.g., eq. [2]; Amiotte-Suchet *et al.*, 1999). In these soils, the fraction of DIC from respiration should therefore be close to 100%.

198 2 NaAlSi₃O₈ + 2 CO₂ + 3 H₂O = Al₂Si₂O₅(OH)₄ + 2 Na⁺ + 2 HCO₃⁻ + 4 SiO₂ [2]

Since the δ^{13} C values of limestones are close to zero (+2 to -5% $_e$) and the δ^{13} C signature of CO $_2$ from soil respiration typically ranges from -21% $_e$ to -25% $_e$, the fraction of biogenic and respiratory DIC can be inferred from δ^{13} C values of DIC (Pawellek & Veizer, 1994; Amiotte-Suchet et~al., 1999), taking into account the isotopic fractionation during the dissolution of CO $_2$ in water (Zhang et~al., 1995). The enrichment of 13 C in soil air CO $_2$ compared to biomass (δ^{13} C of C3-plants: ~ -28% $_e$) is the result of isotopic enrichment during litter decay and enrichment of CO $_2$ during diffusion out of the soil into the atmosphere (Amiotte-Suchet et~al., 1999). To account for the uncertainty caused by the enrichment of 13 C in soil air CO $_2$, we used a value of -28% $_e$ to calculate a minimum fraction of biogenic DIC and the measured δ^{13} C of soil air CO $_2$ to calculate a maximum biogenic fraction. A value of 0% $_e$ was used for carbonate-C. In addition, a value of -5% $_e$ was used to assess the sensitivity of the calculated biogenic fraction to variations of this parameter.

The application of a mixing model with carbonates and CO_2 from soil respiration as the two end-members to derive the fraction of carbonate DIC relies on the assumption that the influence of isotopic exchange of DIC with atmospheric air ($\delta^{13}C \approx -8.5\%$) can be neglected.

Isotopic exchange of DIC with atmospheric CO_2 would shift the $\delta^{13}C$ of DIC towards less negative values, which would cause a systematic overestimation of the carbonate-borne fraction of DIC when applying the two end-member mixing model mentioned above.

Delta ¹³C values of DIC were determined using a Delta Plus XL isotope ratio mass spectrometer with GasBench II interface (Thermo-Finnigan MAT, Bremen, Germany, equipped with a Pal-80 autosampler, CEC Analytics, Zwingen, Switzerland), with an average deviation of 2% between replicate measurements. Measured δ^{13} C values of DIC were corrected for isotopic fractionation during the dissolution of CO₂ in water according to Zhang et al. (1995), considering the speciation of total dissolved CO₂ (H₂CO₃*, HCO₃-, CO₃²-) as a function of pH. Samples for the determination of δ^{13} C values of DIC were collected in October 2007. Because insufficient sample volume was collected during this time as a consequence of drought at Grignon, no δ^{13} C data are available for DIC samples from that site. Therefore, the biogenic fraction of DIC was calculated as the difference between the concentrations of DIC and the sum of Ca²⁺ and Mg²⁺ (in mmol l⁻¹) based on the stoichiometry of carbonate dissolution (eq. [1]), assuming i) that carbonate dissolution is the main source of Ca²⁺ and Mg²⁺ in the calcareous soil and ii) that chemical equilibrium is reached. A comparison of the method used for the Grignon site and the isotope method did not always yield identical results at the other sites. The release of Ca²⁺ and Mg²⁺ from litter in deciduous forests or fertilization of cropland with $Ca(NO_3)_2$ caused overestimates of the carbonate-borne DIC fraction exceeding 100%.

Analysis of DOC concentrations

Dissolved organic carbon concentrations of soil water samples were analysed using Pt-catalysed, high-temperature combustion (680°C) with a TOC-5050A analyser (Shimadzu

241	Corp., Kyoto, Japan). Prior to analysis, inorganic carbon was stripped off by adjusting the pH
242	to 2 with HCl and sparging with CO ₂ -free synthetic air.
243	
244	Determination of iron and aluminium (hydr)oxide concentrations in soil horizons
245	Concentrations of poorly-crystalline iron and aluminium oxides in the bulk soils at Wetzstein,
246	Sorø, Laqueuille, Easter Bush, Hainich, and Grignon were determined by ammonium-oxalate
247	extraction (pH 3) in the dark after Schwertmann (1964) and measurement of iron and
248	aluminium concentrations in extracts with ICP-OES.
249	
250	Analysis of CO ₂ and CH ₄ in the gas phase
251	Gas samples were analysed with the GC-2014AF gas chromatograph used for determination
252	of DIC concentrations (see above). The $\delta^{13}C$ values of CO_2 in soil air samples were analysed
253	using a HP 6890 gas chromatograph equipped with a BPX-5 column (50 m \times 0.32 mm \times 0.5
254	μm), linked via a combustion interface to a MAT 252 isotope ratio mass spectrometer
255	(Thermo-Finnigan MAT, Bremen, Germany).

256 Calculation of fluxes

Fluxes of dissolved C were calculated by multiplying concentrations of DOC, DIC, and CH₄,
with the volume of leached water, which was derived from a soil water model. We used a
capacity or "bucket" model which assumes that leaching from a soil layer commences when
the field capacity (FC in mm), is exceeded (eq. [3]),

262 if
$$SWC_i \le FC$$
, then $D_i = 0$, else $D_i = c(SWC_i - FC)$ [3],

with SWC_i denoting the soil water content at day i (mm), D_i the drainage volume at day i (mm

d⁻¹), and c a rate constant (d⁻¹) defining the fraction of excess water that can drain per day.

Changes in soil water contents over time were calculated from the soil water balance

according to eq. [4].

$$SWC_i = SWC_{i-1} + INF_i - ET_i$$
 [4]

with SWC_{i-1} indicating the soil water content of the preceding day (mm), INF_i denoting the infiltration at day i (mm), and ET_i the evapotranspiration at day i (mm). The soil water content was allowed to vary between the total pore volume of the soil (mm) and the water content at a soil matric potential of pF 4.2 (mm, wilting point). To this end, infiltration into the A horizon was restricted and diverted as "rapid runoff" when the amount of precipitation exceeded the available pore volume of the soil matrix (mm) or the infiltration capacity (mm d⁻¹). This "rapid runoff" might be discharged as surface runoff or flow through cracks and macropores. We did not consider carbon transport with this rapid runoff because the concentrations of DOC, DIC, and CH₄ therein were unknown. Water was replenished by capillary rise when SWC dropped below the soil water content at pF 4.2.

The model subdivided the soil pore volume into a fast and a slowly draining pool, each characterized by a specific total pore volume, field capacity and water content at pF 4.2. The proportion of precipitation entering the fast and the slow pool was assumed constant over time for each site. The model subdivided the soil profile into a topsoil layer and a subsoil layer. The fraction of water taken up by the plants from the subsoil was assumed constant over time and was set equal to the fraction of roots below the A horizon. An exception was the Carlow cropland for which only one soil layer above the gravelly subsoil was considered.

Values for total pore volume, field capacity, and water content at wilting point for the sum of the two soil water pools of each soil layer were matched to tabulated values (AG Boden, 1994), according to soil texture. In addition, the model parameters mentioned above were adjusted manually in a way that modelled soil water contents fitted *in situ* measured soil water contents. Input variables were precipitation, air temperature, and latent heat flux, which derived from eddy covariance measurements, either taken from the CarboEurope database (http://gaia.agraria.unitus.it/database/carboeuropeip/) or provided by the managers of the CarboEurope and NitroEurope sites. Actual evapotranspiration was calculated by dividing the latent heat flux derived from the database by the temperature-adjusted heat of evaporation. For the sites at Carlow and Laois, evapotranspiration was calculated according to approaches of Allen *et al.* (1998), Priestley & Taylor (1972), and Teklehaimanot & Jarvis (1991) because the gaps in eddy covariance data were too large to allow for calculation of reasonable evapotranspiration estimates. Details regarding the calculations of evapotranspiration at these sites are given in Walmsley (2010).

Statistics

Differences between land use types were tested for statistical significance using the nonparametric Kruskal-Wallis test, differences between sampling depths or terrain positions were

analyzed with the non-parametric Mann-Whitney *U*-test. The tests were carried out with the

Statistica version 8 software (Statsoft, Tulsa, USA).



Results

Concentrations of DOC and DIC in soil water

Dissolved inorganic carbon concentrations beneath topsoils of forest sites (2.5–23.9 mg Γ^{-1} ; mean: 8 mg l⁻¹, median: 4 mg l⁻¹; Fig. 2a) tended to be smaller than those of cropland (22.7– 50.5 mg Γ^{-1} ; mean: 35 mg Γ^{-1} , median: 33 mg Γ^{-1} ; Fig 2c, p = 0.09, Kruskal-Wallis test). No significant differences were found between mean concentrations of forests and grasslands (6– 70 mg l^{-1} ; mean: 28 mg l^{-1} , median: 17 mg l^{-1} ; Fig. 2b, p = 0.2, Kruskal-Wallis test). Also subsoil water DIC concentrations tended to increase in the order forest (2.5–67.7 mg 1^{-1} ; mean: 34 mg l^{-1} , median: 33 mg l^{-1}) < grassland (8.0–90.9 mg l^{-1} ; mean: 42 mg l^{-1} , median 34 $\text{mg } l^{-1}$) < cropland (28.3–84.8 $\text{mg } l^{-1}$; mean: 57 $\text{mg } l^{-1}$, median: 57 $\text{mg } l^{-1}$), but these differences were not statistically significant (Kruskal-Wallis test). Dissolved inorganic carbon concentrations were insensitive to management intensity at Laqueuille (Fig. 2b) or topography at Hainich, Sorø (Fig 2a), and Carlow grassland (Fig 2b). However, topography was important at Easter Bush, with significantly larger concentrations at the bottom of the valley than at the elevated position (Fig. 2b, shallow cups: p = 0.009; deep cups: p = 0.075, Mann-Whitney Utest). Theoretical concentrations of DIC can be calculated from the partial pressures of soil air CO₂ and soil solution pH by applying Henry's law and assuming chemical equilibrium between CO₂ in the gas phase and solution H₂CO₃*, HCO₃, and CO₃². Total concentrations of DIC were close to equilibrium with partial pressures of soil air CO₂ for the Loobos, Wetzstein, Hainich and Sorø forest sites (Fig. 3). Soil waters from all other sites were supersaturated with DIC relative to soil air CO₂, especially in subsoils, which might be partly due to selective

sampling of CO₂-poor soil air from larger pores. Overall, the measured DIC concentrations

were nearly equal to or greater than theoretical concentrations, indicating that outgassing of

soil water samples was successfully minimized by the sampling method.

Dissolved inorganic carbon concentrations were commonly much larger than DOC concentrations, particularly in subsoils (Fig. 2). Exceptions were the acidic soils at the Wetzstein and Loobos forest sites, with DIC representing less than 10% of the total dissolved carbon leached from topsoils and 16–30% in subsoils. Concentrations of DOC leached from topsoils of grasslands (median: 7 mg Γ^1 , mean: 8 mg Γ^1 ; range 1.9–17.1 mg Γ^1 , Fig 2b) and arable fields (median: 14 mg Γ^1 , mean: 12 mg Γ^1 , range: 3.9–17.3 mg Γ^1 , Fig 2c) tended to be smaller than those leached from forest topsoils (median: 24 mg Γ^1 , mean: 23 mg Γ^1 ; range: 7.1–43.1 mg Γ^1 , Fig. 2a), but the differences were not statistically significant (p = 0.16, Kruskal Wallis test). A close positive relationship was observed between DOC concentrations at shallow sampling depths and topsoil C/N ratios (Fig. 4).

Isotopic signature and sources of dissolved inorganic carbon

The δ^{13} C signatures of DIC in samples collected from subsoils clustered into two groups. Samples from the Hainich, Sorø, Laois, and Carlow grassland were all characterized by δ^{13} C values close to -15%, with little variation between different suction cups (Fig. 5b). At Früebüel, samples from eight suction cups were close to -15%, while the sample from one cup showed a value of -22% (Fig. 5b). Soils at Hainich, Sorø, Laois, Carlow, and Früebüel all developed from carbonate-containing parent materials, which is Muschelkalk limestone at the Hainich site and glacial till at the other sites. Soils at the Loobos, Wetzstein, Laqueuille, Easter Bush, and Klingenberg sites developed from carbonate-free parent material. Average

fields (median: 14 mg l^{-1}), and 9 mg l^{-1} for grasslands (median: 8 mg l^{-1}), with the latter value

being somewhat lower due to small DOC concentrations in the Laqueuille Andosol (Fig. 2b).

 δ^{13} C values of DIC were all below -20% at the Wetzstein, Laqueuille, Easter Bush and Klingenberg sites (Fig. 5b). Less negative values were found in the subsoil at the Loobos site. In contrast to subsoil, dissolved inorganic carbon from shallow depths of the Sorø and Früebüel sites had more negative δ^{13} C values below -20%, reflecting the dissolution of carbonates in topsoils during soil development (Fig. 5a).

Using a value of -28% for the biogenic end-member and a value of 0% for carbonate-C, a biogenic DIC fraction of 74–82% was calculated for the calcareous subsoils of the Hainich, Sorø, Laois, and Früebüel sites and used to estimate leaching losses of total biogenic carbon (Fig. 5b, upright triangles). The fraction was reduced by 1–11% if an extreme value of -5% was assumed for carbonate-C. Use of the measured δ^{13} C of soil air CO₂ increased the calculated biogenic fraction for DIC from the calcareous subsoils to 90–105%. An extreme value of 293% was calculated for the Früebüel site, but this unplausible value likely resulted from a contamination of soil air samples with atmospheric air during the difficult abstraction of air samples from the wet Früebüel soil.

Contrary to geochemical theory (eq. [2]), the biogenic DIC fractions calculated for the carbonate-free subsoils at the Loobos, Wetzstein, and Klingenberg sites, using the end-members -28% and 0%, were much lower than the expected 100%. This might be caused by more intense isotopic exchange of soil air and DIC with atmospheric air. Indeed, when using measured δ^{13} C values of soil air CO₂ in the calculation, the estimated biogenic fraction was >100% for the Klingenberg site. For the Lobos and Wetzstein sites, however, calculated biogenic fractions remained below 100%, which might indicate a contamination of the little DIC pool in the acidic samples from these sites with atmospheric CO₂ during sampling and sample handling. In line with geochemical theory (Eq [2]), we used a biogenic DIC fraction of

100% for calculating the total biogenic carbon leaching loss from the carbonate-free Loobos,
 Wetzstein, Laqueuille, Easter Bush, and Klingenberg soils.

Water balance of the research sites

Mean annual precipitation during our study period ranged from 710 mm yr⁻¹ at Grignon to 1771 mm yr⁻¹ at Früebüel with an average of 1009 mm yr⁻¹ across all sites (Table 2). The average loss of water by evapotranspiration equalled 436 mm yr⁻¹. The average relative deviation of modelled soil water contents from measured ones was 14% for the shallow depths and 6% for subsoils (Table 2). Figure 6 (c, d) provides a visual impression of the goodness of fit between measured and modelled soil water contents for the Carlow grassland site, where the quality of the model fit was exactly the average of all sites.

Especially at the sites with clayey soils (Hainich, Laois, and Früebüel), modelled water contents could only be matched to measured ones when large volumes of water (357–489 mm yr⁻¹) were discharged by rapid runoff via surface flow or preferential flow paths in the calibrated water balance model (Table 2). On average 429 mm yr⁻¹ of water were left over for drainage from the soil profile after subtracting evapotranspiration and rapid runoff from precipitation (Table 2).

Leaching losses of dissolved carbon

Across all sites, an average of 4.2 g DOC m⁻² yr⁻¹ was leached from soils, with rather small differences between land use systems (Table 3). Subtracting DOC leaching from subsoils from DOC fluxes from topsoils indicated that as much as 17 (Wetzstein) to 20 g m⁻² yr⁻¹ (Loobos) of DOC was retained in acidic forest B horizons (Table 3). Other forest B horizons retained only 2 (Laois) to 9 g DOC m⁻² yr⁻¹ (Sorø) and retention in the B horizons of grasslands or arable fields was generally smaller than 5.5 g DOC m⁻² yr⁻¹ (Table 3).

413	Retention of DOC in B horizons decreased exponentially with increasing ratio of organic
414	carbon to the sum of oxalate-extractable iron and aluminium present in the bulk soil (Fig. 7).
415	
416	The average loss of DIC from forest topsoils was significantly smaller than DIC leaching
417	from grassland topsoils ($p = 0.03$) and tended also to be smaller than leaching from cropland
418	topsoils ($p = 0.09$, Table 4). Leaching losses of DIC from subsoils, however, did not differ
419	significantly between land use types ($p = 0.31$, Kruskal Wallis test).
420	
421	As a result of the strong retention of DOC and the low DIC concentrations, total biogenic
422	dissolved carbon leaching from forest soils tended to be smaller than leaching from grasslands
423	or arable fields (Table 5). Differences, however, were not significant ($p = 0.23$, Kruskal
424	Wallis test) because the variability between sites under similar land use was large. The overall
425	average biogenic dissolved carbon loss from soils was 19.3 g C m ⁻² yr ⁻¹ . This value was
426	strongly influenced by large losses from the forest at Sorø and the grasslands at Easter Bush
427	and Carlow. Consequently, the median flux across all sites was smaller (15.2 g C m^{-2} yr $^{-1}$).
428	
429	Methane concentrations in soil air and its leaching losses
430	Methane concentrations in soil air at shallow depths were mostly smaller than global average
431	atmospheric concentrations of 1.7 μ l Γ^{-1} , indicating that most sites could have acted as sinks
432	for atmospheric CH ₄ (Table 6). Exceptions were the Laois and Klingenberg Stagnosols. In
433	subsoils, CH ₄ concentrations larger than atmospheric ones were only found in the Laois
434	Stagnosol and the Easter Bush Gleyic Cambisol in the hollow terrain position.
435	
436	Leaching losses of CH ₄ were <5 mg CH ₄ -C m ⁻² yr ⁻¹ at most sites. Even the largest loss of 26

 \pm 2 mg CH₄-C m⁻² yr⁻¹, measured at Klingenberg, was equivalent to a loss of only 0.23 g

- 438 CO₂-C_{eq} m⁻² yr⁻¹ when applying a warming potential of 25 (Table 6). Leaching of CH₄ was
- therefore not relevant for the greenhouse gas budget of the investigated sites.



Discussion

Uncertainties of calculated leaching losses

The errors in the estimates of leaching losses, indicated in brackets in tables 3–7, only reflect the uncertainty caused by the spatial variation of DIC, DOC, and dissolved CH₄ concentrations. Additional uncertainties arise from the modelling of the water balance, the spatial variation of water fluxes, and the calculation of the biogenic fraction of DIC from its δ^{13} C signature.

We assessed the uncertainty of evapotranspiration measurements using the eddy covariance method by analyzing the energy balance of the Klingenberg site for the years 2006 and 2007 (Vowinckel, 2008). A regression between net radiation and the sum of sensible and latent heat flux (assuming a negligible net soil heat flux over long-term periods) showed that approximately 68% of the available energy was consumed by turbulent heat fluxes only. Using the slope of this regression and monthly means of the Bowen ratio as well as the available energy, measured latent heat fluxes were subsequently corrected. Monthly correction factors ranged from 1.01 to 1.89 and averaged 1.66, which means an underestimation of the eddy covariance-based evapotranspiration at the Klingenberg site during our study period from October 2006 to March 2008 by approximately 182 mm.

This systematic underestimation of evapotranspiration is roughly compensated by a systematic underestimation of precipitation by standard rainfall collectors due to evaporation from the wet funnel surface and wind turbulence (Richter, 1995). Applying the empirical correction of Richter (1995) to precipitation data of the Klingenberg site (assuming that precipitation at temperatures <0°C occurs as snow) indicated an underestimation of the rainfall by 216 mm.

A major uncertainty of the soil water model is the calculated rapid runoff. At Laois, the relevance of lateral water flows could easily be deduced from the clayey soil texture and the occurrence of drainage ditches on the plot. Here, micro-catchments were constructed by separating areas of 10–12 m² by trenches and wooden boards in a way that lateral runoff could be collected in a down slope trench, so that modelled lateral flows could be matched to measured water volumes (Walsmley, 2010). No micro-catchments were available at Früebüel and Hainich, which were also characterized by clayey soils and rapid runoff. For these sites, the results of our water balance model were compared with results of the BLAU "bucket" model of Huwe (1991), which was extended with a surface runoff routine. For Früebüel, the results of the BLAU model deviated from the results in Table 2 by only 2% for surface/rapid runoff (440 vs. 449 mm) and by 0.3% for drainage from the soil profile (944 vs. 941 mm). A good agreement between results of the two models was also observed for the Hainich site; rapid/surface runoff and drainage from the soil profile deviated by only 4%.

Dissolved carbon fluxes associated with rapid runoff could not be calculated because concentrations in that runoff were not assessed. According to Kaiser *et al.* (2000), leaching losses of DOC from a shallow forest soil (Rendzic Leptosol) via preferential flow paths during heavy rainstorms were similar to mean annual fluxes as determined for soil matrix flow. This potential underestimation of DOC losses due to preferential flow and rapid runoff of soil water might partly have been compensated for in our study by an overestimation of DIC losses since DIC concentrations in soil water rapidly draining through large pores are probably smaller than concentrations in water in slowly draining pores of the soil matrix. We are not aware of any study that has analysed concentrations of DIC in preferential flow.

Biogenic DIC concentrations are also subjected to an error associated with the determination of the fraction of carbonate-borne DIC using the DIC δ^{13} C signature. The largest source of error is contamination or isotopic exchange of DIC with atmospheric air. Contamination and isotopic exchange with atmospheric air would shift the measured δ^{13} C values to less negative values, leading to systematic underestimation of biogenic DIC concentrations. According to the stoichiometry of carbonate dissolution, the fraction of biogenic DIC in soil water of calcareous soils must exceed 50% and, of course, it should be smaller than 100%. This limits the potential underestimation of biogenic DIC concentrations and fluxes from calcareous soils to values of 18–31%. The potential overestimation of biogenic DIC concentrations resulting from the assumption of an average δ^{13} C value of 0% for carbonates was 1–11% (average 5%). Our results indicated greater isotopic exchange between DIC and atmospheric air for samples from carbonate-free subsoils, especially for the acidic forest subsoils at Loobos and Wetzstein. For these subsoils, however, the assumption of 100% biogenic DIC according to the stoichiometry of weathering reactions is safe and the relevance of the small DIC concentrations at these sites for total carbon leaching is limited.

Factors controlling dissolved carbon concentrations and leaching losses

The positive correlation between soil C/N ratios and DOC concentrations and fluxes in terrestrial ecosystems across biomes was first published by Aitkenhead & McDowell (2000). In contrast to the observation of Aitkenhead & McDowell (2000), DOC concentrations and fluxes in our study correlated only to topsoil C/N ratios (upper 5–40 cm, Table 1) and not to C/N ratios of the entire soils down to 65–100 cm depth. Due to the accumulation of organic matter, topsoils and organic layers are regarded as the soil compartments that are the most important for the generation of DOC (e.g., Guggenberger & Kaiser, 2003). The topsoil C/N ratio is indicative of the litter quality and the intensity of microbial transformation processes. Large C/N ratios indicate poorly degradable litter and a slow transformation and recycling of

organic matter. Studies using δ^{13} C or 14 C to track sources and turnover of DOC (Schiff *et al.*, 1997; Flessa et al., 2000; Hagedorn et al., 2004; Fröberg et al., 2007) suggest that dissolved organic matter, which is transported over decimetres or metres down into subsoils, mainly represents highly altered residues of organic matter processing. Processing of N-poor organic matter seems to result in production of more soluble residues, which explains the positive correlation between soil C/N ratios and DOC concentrations. A positive correlation between DOC production and C/N ratios of soil organic matter was also observed in laboratory experiments (Gödde et al., 1996; Kalbitz & Knappe, 1997). A hypothesis put forward by Gödde et al. (1996) to explain this relationship was that microbial communities in soils with large C/N ratios have to process more organic matter to satisfy their N requirements than communities in soils with small C/N ratios. At C/N ratios << 10, most carbon associated with processed soil organic matter is either respired or re-assimilated so that only small amounts of DOC are produced. The regression equation in Fig. 4 suggests that above a critical C/N ratio of approximately 8, increasing amounts of soluble C residues are produced with increasing C/N ratio of soil organic matter. It is plausible to expect that DOC leaching from topsoils only occurs if a critical C/N ratio is exceeded.

Leaching of DOC from subsoils was controlled by retention in B horizons, with this retention being related to the ratio of organic carbon to the sum of oxalate-extractable iron plus aluminium (Fig. 7). Oxalate-extractable iron and aluminium are a measure of the concentration of poorly crystalline iron and aluminium (hydr)oxides with a high specific surface area in soils. These iron and aluminium(hydr)oxides are considered to be the most important sorbents for dissolved organic matter in soils (e.g., Kaiser *et al.*, 1996). The ratio of organic carbon to the concentration of iron and aluminium (hydr)oxides in B horizons is therefore an indicator of the saturation of the available sorption capacity of subsoils for the DOC they receive from topsoils. Hence, the relationship presented in Fig. 7 suggests that the

concept of a sorptive control on DOC output from mineral soils (e.g., Guggenberger & Kaiser, 2003) seems to apply for a wide range of soils and land use systems.

The close relationship between measured and calculated DIC concentrations leached from topsoils indicates that the partial pressure of CO₂ in soil air and the pH of soil solution control DIC concentrations (and fluxes), according to carbonate equilibrium. The almost constant degree of supersaturation of soil waters with respect to equilibrium concentrations probably reflects the diffusion gradient of CO₂ from microbial habitats through soil air into the atmosphere or the kinetic retardation of CO₂ outgassing from soil water into soil air. For subsoils, the degree of supersaturation was larger and more variable than for topsoils, which might reflect the longer diffusion path into the atmosphere and therefore the stronger build-up of disequilibria between CO₂ production in moist soil zones and the partial pressure of CO₂ in soil air in drier zones. On the other hand, any contamination of soil air samples with atmospheric air during sampling or sample handling would probably also lead to an underestimation of equilibrium DIC concentrations because CO₂ concentrations in soil air are commonly much larger than atmospheric concentrations.

We did not observe any relationship between the absolute or relative excess of measured compared to calculated DIC concentrations with mean annual precipitation, clay content, air capacity of the soils (air-filled pore volume at field capacity), soil water content, or water-filled pore space. Likely, the connectivity and spatial arrangement of soil (macro)pores rather than bulk soil properties control gas exchange with the atmosphere (Lange *et al.*, 2009) and the degree of supersaturation of soil waters (Walmsley, 2010).

Relevance of DOC plus biogenic DIC leaching for the net ecosystem carbon balance

Plot-scale balances of carbon fluxes of four forest sites in our study were not significantly affected by carbon leaching (Table 5). Only for the beech forest at Sorø, carbon leaching equalled 25% of the balance of atmospheric CO₂ fluxes (Table 5). Reasons for this large fraction are the comparably small NEE and high leaching losses of biogenic DIC due to the high pH of the calcareous subsoil. Neglecting leaching losses would lead to a substantial overestimation of the net ecosystem carbon balance of the Sorg forest (+145 g C m⁻² yr⁻¹ instead of +108 g C m⁻² yr⁻¹, Table 5). For forest ecosystems, in particular, the temporal scale of balances is extremely important for estimating the magnitude of carbon sequestration. The forest ecosystem carbon balances in Table 5 cover only few years (<13 years) in relation to a forest's life cycle. Therefore, carbon release as a result of logging or catastrophic events like fire or wind storms are not included. According to the latest estimates, logging reduces the average net carbon uptake of European forests by 63 ± 11 g m⁻² yr⁻¹ (Schulze et al., 2009). Another 5 ± 1 g C m⁻² yr⁻¹ are lost through fires. Due to these losses, the long-term average net biome productivity of European forests, excluding leaching losses, is estimated to 82 g C m⁻² yr⁻¹ (Schulze et al., 2009). The average leaching loss of 12 g C m⁻² yr⁻¹ we observed for forests equals 14% of the average net biome productivity of forests.

Grasslands strongly differed with regard to their balance of aboveground carbon fluxes (Table 5). While the grasslands at Laqueuille, Easter Bush, and Früebüel were characterized by a positive balance of NEE, fertilizer addition and carbon removal with harvest (indicating a carbon sink), the Carlow grassland showed a net negative balance due to large harvest of biomass (for silage). At Carlow, net carbon losses nearly doubled from 40 g m⁻² yr⁻¹ to 79 g m⁻² yr⁻¹ when leaching losses were considered. At the other sites, 5–29% of the net inputs of carbon into soil were leached. The estimated average net biome productivity of European grasslands, excluding leaching losses, is 64 g C m⁻² yr⁻¹ (Schulze *et al.*, 2009). Our average leaching loss for grasslands of 29 g C m⁻² yr⁻¹ is almost 50% of that estimate.

In contrast to forests or grasslands, all studied croplands were net sources of carbon (Table 5). Carbon losses with drainage were 24–105% of the sum of NEE plus fertilization minus harvest (median: 25% Table 5). Kutsch *et al.* (2010) reported an average net biome productivity (excluding leaching losses) of –97 g C m⁻² yr⁻¹ for five temperate European croplands (including the Carlow and Klingenberg sites) based on eddy covariance measurements and carbon fluxes with organic fertilizers and harvested biomass. The average leaching loss of 18 g C m⁻² yr⁻¹ we found equals 19% of the net carbon loss as given by Kutsch *et al.* (2010). According to Schulze *et al.* (2009), European croplands are a smaller net source of carbon (17 g m⁻² yr⁻¹ on average, excluding leaching losses), which is of a similar magnitude to the average leaching losses found in this study.

Downstream fate of dissolved carbon

The estimates of Schulze *et al.* (2009) for average greenhouse gas balances of European land use systems accounted for leaching losses of DOC and DIC by assuming a constant average loss of 7 g C m⁻² yr⁻¹ for all land use systems, which is close to the losses of 5.9 g m⁻² yr⁻¹ reported by Ciais *et al.* (2008). Both estimates are considerably smaller than the average (19.5 g m⁻² yr⁻¹) and the median (15.2 g m⁻² yr⁻¹) of leaching losses we found (Table 5). This difference is related to the fact that the losses given by Ciais *et al.* (2008) and Schulze *et al.* (2009) represent the carbon export with rivers to oceans because these studies aimed at the carbon budget of the whole continent. In the studies of Ciais *et al.* (2008) and Schulze *et al.* (2009), the riverine carbon export was divided by the area of the river catchments to derive an average loss per m², whereas our study directly measured leaching from the soil into groundwater. A comparison of our results with the estimates of Ciais *et al.* (2008) and Schulze *et al.* (2009) is therefore only possible when considering the downstream fate of dissolved carbon after being leached from soil.

Dissolved organic carbon leached from soil may partly be retained in the vadose zone before reaching aquifers (Siemens & Kaupenjohann, 2003). Adsorption or precipitation of DOC may slow down its biodegradation (Mikutta et al., 2007; Kalbitz & Kaiser, 2008; Scheel et al., 2008), but the extent of stabilization under field conditions is uncertain (Guggenberger & Kaiser, 2003; Siemens & Kaupenjohann, 2003). As a result of retention in the vadose zone, DOC concentrations in groundwater are often smaller than 5 mg l⁻¹ (e.g., Siemens & Kaupenjohann, 2003). For the range of groundwater recharge of 95–652 mm yr⁻¹we found in our study, these concentrations would lead to a small (<0.5–3.3 g DOC m⁻² yr⁻¹), albeit constant flux of DOC into surface waters. Significant additional amounts of DOC reach surface waters with runoff at or near the soil surface during storms (e.g., Hagedorn et al., 2000) or with drainage water from peatlands (e.g., Dawson et al., 2004). In lakes and rivers, allochtonous dissolved organic matter will be partly processed by aquatic organisms and finally mineralized to DIC (Tranvik, 1993; Tranvik & Bertilsson, 2001). The discharge of DIC-rich soil water and groundwater supersaturated with CO₂ (Stets et al., 2009), together with the oxidation of dissolved organic matter in surface waters, results in a large supersaturation of river and lake water relative to the atmosphere (e.g., Duarte & Prairie, 2005). Efficient outgassing of CO₂ has been observed from headwaters with a large fraction of young groundwater (e.g., Fiedler et al., 2006) and from estuaries (Frankignoulle et al., 1998; Ciais et al., 2008). The resulting evasion of CO₂ from inland waters and estuaries into the atmosphere was estimated to equal ~40% of carbon inputs from land on a global scale (Cole et al., 2007) and to ~66% of the carbon European rivers receive from terrestrial ecosystems (Ciais et al., 2008). Multiplication of these estimates with our mean carbon

leaching loss gives an evasion of 8–13 g C m⁻² yr⁻¹ and an evasion of 6–10 g C m⁻² yr⁻¹ when

using the median. Subtraction of this evasion from surface waters from our dissolved carbon flux from soils gives a flux of 5-12 g C m⁻² yr⁻¹ from the continent into the ocean.

A fraction of the carbon that is transferred from terrestrial ecosystems to inland waters is buried in the sediments of lakes and reservoirs. Burial of carbon in the sediments of inland waters has been estimated to equal 12% of inputs from land on a global scale (Cole et al., 2007) and 21% of the net ecosystem carbon transfer from European terrestrial ecosystems into rivers (Ciais et al., 2008). Particulate organic matter washed into inland waters due to erosion of soils certainly contributes to this burial of carbon in the sediments of lakes and reservoirs (Stallard, 1998). Since we did not include fluxes of organic matter with eroded soil material in our study, the estimates of carbon burial of Cole et al. (2007) and Ciais et al. (2008) are difficult to relate to the fluxes of dissolved carbon reported here. Assuming that all carbon buried in the sediments of lakes and reservoirs derived from dissolved carbon indicates that less than 2–4 g m⁻² yr⁻¹ of DOC and DIC leached from soils are probably stored in the sediments of inland waters, which would reduce the carbon lost from soils that reaches the ocean with rivers to 1–10 g C m⁻² yr⁻¹, but this reduction is uncertain. Overall, the carbon leaching losses reported here correspond to a riverine flux of DOC and biogenic DIC of 5.5 g m⁻² yr⁻¹ reported for Europe by Ludwig et al. (1998) and the river carbon fluxes of 5.9 and 7 g m⁻² yr⁻¹ given by Ciais et al. (2008) and Schulze et al. (2009), respectively, when considering outgassing from inland waters and potential burial in sediments.

Our set of research sites did not include sites from southern, northern, and eastern Europe. In the study of Ciais *et al.* (2008), the yield of river carbon fluxes reaching the continental shelf was twice as large for southern Europe (<42° N; 10.8 g m⁻² yr⁻¹) than for temperate Europe (42–50° N; 5.2 g m⁻² yr⁻¹). The rather good agreement of riverine carbon fluxes with carbon leaching losses in our study when considering outgassing and sedimentation is probably due

to the small effect of high C export from south European catchments on the overall export due to their small area and small river discharge in relation to temperate and northern Europe.

In summary, leaching represents a comparatively small, but continuous loss of carbon from terrestrial ecosystems. Considering these losses for ecosystem greenhouse gas balances is increasingly important when long time periods are examined and net changes of biomass stocks are small. Leaching losses are especially important to local and plot-scale studies, because outgassing of carbon into the atmosphere from surface waters probably occurs outside the boundaries of the studied systems. Leaching losses are particularly important for the net ecosystem carbon balance of grasslands and croplands where dissolved carbon leaching roughly equalled 25% of NEE plus fertilizer carbon minus harvested carbon. Topsoil C/N ratio, the saturation of the subsoils sorption capacity for organic carbon, and its carbonate content were important factors controlling the magnitude of carbon leaching losses.

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Tables

Table 1: Land use, soil characteristics, instrumentation, and temperature, of the investigated sites. Mean annual temperatures, mean annual precipitation, and mean drainage is given for the period of the observations of this study. This period was from 1 October 2006 until 30 March 2009 for Laois and Carlow and from 1 October 2006 until 30 March 2008 for all other sites.

Site	Site Land use		Soil texture	Soil organic carbon	pH of topsoil	Sampling depths	Mean temperature in observed period	Mean annual temperature in observed period
				$(kg m^{-2})$		(cm)	(° C)	(° C)
Hainich (slope and valley)	Beech forest	Cambisol	Loamy clay	11.9 ¹	5.7	20, 75	7.6	9.2
Sorø (near and far from tree)	Beech forest	Stagnic Luvisol	Loamy sand	9.8^{1}	4.6	15, 100	8.2	9.6
Laois	Spuce forest	Stagnosol	Clay	2.2^{3}	5.1	15, 70	9.1	10.9
Loobos	Pine forest	Haplic Arenosol	Sand	7.7^{2}	4.0	5, 120	9.5	10.9
Wetzstein	Spuce forest	Podzol	Sandy loam	8.7^{3}	3.8	20, 90	5.6	7.3
Laqueuille (intensive and extensive use)	Grassland	Andosol	Silt loam	23.1^{2}	5.4	30, 90	6.6	7.9
Carlow (g) (slope and valley)	Grassland	Calcic Luvisol	Sandy loam	2.3^{3}	7.5	20, 65	9.4	9.3
Easter Bush (slope and hollow)	Grassland	Gleyic Cambisol	Sandy loam	12.3 ¹	5.4	30, 100	8.1	9.0
Früebüel	Grassland	Endogleyic Cambisol	Loamy clay	5.8 ³	5.5	30, 100	6.6	8.1
Grignon	Cropland	Calcaric Cambisol	Silt loam	10.5^{1}	7.3	40, 90	9.9	11.2
Klingenberg	Cropland	Stagnosol	Silt loam	9.7^{1}	6.4	35, 75	7.1	8.8
Carlow (c)	Cropland	Eutric Cambisol	Sandy loam	7.3^{1}	6.9	40	9.0	9.3

down to 60 cm depth, ²down to 30 cm depth, ³down to 50 cm depth. Soil organic carbon data for Früebüel from Roth (2006). ⁶Infiltration into B horizon excluding a lateral runoff of 287 mm. Carlow (g): Carlow grassland, Carlow (c): Carlow cropland

Table 2: Water balance of the research sites during the investigated period. Drainage from A-horizon denoted fluxes deeper than the upper instrumentation depth given in Table 1, drainage from soil profile refers to water fluxes deeper than the lower instrumentation depth of Table 1.

site	Precipitation	Evapotranspiration	Rapid runoff	Dra	inage	Average relative deviation of modelled water contents from measured ones		
	F	_ · · · · · · · · · · · · · · · · · · ·		from A-	from soil			
				horizon	profile	shallow	deep	
Cumulative we	ater fluxes during	period of observations	(mm)					
Hainich	1363	374	589	572	326	14	no sensor	
Sorø	1345	318	141	1086	915	20	no sensor	
Laois	2594	1192	944^{2}	542 ¹	441	12	8	
Loobos	1452	819	59	1036	646	24	no sensor	
Wetzstein	1358	466	335	790	508	7	no sensor	
Laqueuille	1757	997	62	952	720	5	4	
Carlow (g)	2236	753	0	1627	1515	14	6	
Easter Bush	966	250	13	752	695	14	no sensor	
Früebüel	2178	797	449	1061	941	5	2	
Grignon	1027	662	189	438	189	17	8	
Klingenberg	1307	289	40	1038	975	17	no sensor	
Carlow (c)	2236	669	0	1595		13	_	
Mean annual	water fluxes (mm	yr^{-1})						
Hainich	1008	346	489	335	166	/	_	
Sorø	928	287	112	688	531		_	
Laois	1040	561	357^{3}	202^{1}	149	_	_	
Loobos	982	609	33	683	406	_	_	
Wetzstein	966	428	252	498	259	_	_	
Laqueuille	1295	773	55	670	486	_	_	
Carlow (g)	894	345	0	620	569	_	_	
Easter Bush	735	141	13	610	584	_	_	
Früebüel	1711	666	399	751	651	_	_	
Grignon	710	523	132	266	95	_	_	
Klingenberg	950	242	33	713	652	_	_	
Carlow (c)	894	308	0	598	_	_	_	

¹Total lateral runoff including surface runoff of 349 mm; ²Infiltration into B horizon excluding lateral runoff; ³Total lateral runoff including surface runoff of 135 mm yr⁻¹.

Table 3: Leaching losses of dissolved organic carbon from topsoils and subsoils of the research sites. The winter period is from 1 October until 30 March. Winter data are the average of winters 2006/2007 and 2007/2008 (and 2008/2009 for Irish sites). The summer period started on 1 April and ended on 30 September. For Irish sites, the summer value is the average of summers 2007 and 2008, for all other sites it is the value for summer 2007.

Site		\mathbf{L}	oss from top	soils	Lo	Loss from subsoils			
		winter	summer	total	winter	summer	total		
		$(g C m^{-2})$	$(g C m^{-2})$	$(g C m^{-2} yr^{-1})$	$(g C m^{-2})$	$(g C m^{-2})$	$(g C m^{-2} yr^{-1})$		
Hainich	slope	1.3 (0.1)	0.7 (0.0)	2.0 (0.1)	2.6 (0.0)	0.0 (0.0)	2.6 (0.0)		
	valley	1.8 (0.1)	1.4 (0.0)	3.2 (0.1)	1.9 (0.0)	0.0 (0.0)	1.9 (0.0)		
Sorø	near tree	8.7 (0.3)	7.5 (0.3)	16.2 (0.4)	5.7 (0.5)	2.7 (0.2)	8.3 (0.5)		
	between trees	9.3 (0.5)	8.4 (0.4)	17.8 (0.6)	4.7 (0.2)	2.6 (0.1)	7.3 (0.3)		
Laois		1.8 (0.8)	1.0 (0.4)	$2.8^1 (0.9)$	0.6 (0.2)	0.0 (0.0)	0.6 (0.2)		
Loobos		10.3 (0.5)	11.5 (0.9)	21.8 (1.1)	1.4 (0.1)	0.8 (0.1)	2.2 (0.1)		
Wetzstein		11.3 (0.5)	10.0 (0.4)	21.3 (0.7)	4.5 (0.6)	0.3 (0.2)	4.7 (0.7)		
Average fores	t	6.8 (2.1)	6.3 (2.2)	<i>13.1 (4.3)</i>	2.8 (0.9)	0.8 (0.5)	3.5 (1.3)		
Median forest	!	9.0	8.0	<i>17.0</i>	2.3	0.3	2.3		
Laqueuille	intensive use	0.5 (0.0)	1.5 (0.1)	2.0 (0.1)	0.3 (0.0)	0.7 (0.0)	1.1 (0.0)		
_	extensive use	0.6 (0.0)	1.4 (0.1)	2.0 (0.1)	0.4 (0.0)	0.7 (0.0)	1.1 (0.0)		
Carlow (g)	slope	3.1 (0.5)	2.2 (0.4)	5.3 (0.6)	1.3 (0.2)	0.7 (0.2)	2.0 (0.3)		
	valley	5.1 (0.6)	3.7 (0.4)	8.8 (0.7)	2.2 (0.3)	1.1 (0.2)	3.3 (0.4)		
Easter Bush	slope	6.4 (0.3)	4.9 (0.2)	11.2 (0.3)	5.0 (0.2)	3.8 (0.2)	8.7 (0.3)		
	valley	7.2 (0.4)	7.3 (0.7)	14.5 (0.8)	5.6 (0.1)	5.1 (0.3)	10.7 (0.3)		
Früebüel		1.2 (0.1)	1.9 (0.2)	3.1 (0.2)	3.4 (0.3)	4.2 (0.4)	7.6 (0.5)		
Average grass	sland	3.2 (1.4)	3.1 (1.0)	6.3 (2.5)	2.7 (1.1)	2.6 (1.0)	5.3 (2.0)		
Median grass	land	2.7	2.4	5.1	2.6	2.6	5.1		
Grignon		3.6 (0.1)	0.1 (0.0)	3.7 (0.1)	2.8 (0.1)	n.d	2.8 (0.1)		
Klingenberg		4.0 (0.2)	3.4 (0.2)	7.4 (0.3)	3.2 (0.3)	2.1 (0.1)	5.3 (0.3)		
Carlow (c)		1.7 (0.4)	0.9 (0.2)	2.6 (0.5)					
Average cropland		3.1 (0.7)	1.5 (1.0)	4.6 (1.4)	3.0 (0.2)	1.1 (1.1)	4.1 (1.3)		
Median cropl	and	3.6	0.9	3.8	3.0	2.1	4.1		
Total average		4.7 (1.1)	4.0 (1.1)	8.7 (2.2)	2.8 (0.5)	<i>1.6</i> (0.5)	4.2 (0.9)		
Median DOC	flux	3.8	2.4	5.4	2.8	0.9	2.8		

Median DOC flux

3.8

2.4

5.4

2.8

0.9

2.8

Numbers in brackets indicate standard deviations due to spatial variability of concentrations for individual sites and the standard error of the average across sites with the same land use system or across all sites; n.d.: not detectable; ¹excluding lateral runoff of 3.0 g DOC m⁻² yr⁻¹.

Topsoils reach down to the shallow instrumentation depth, subsoil reach down to the deeper instrumentation depth given in Table 1.

Table 4: Leaching losses of total dissolved inorganic carbon from topsoils and subsoils of the research sites. The winter period is from 1 October until 30 March. Winter data are the average of winters 2006/2007 and 2007/2008 (and 2008/2009 for Irish sites). The summer period started on 1 April and ended on 30 September. For Irish sites, the summer value is the average of summers 2007 and 2008, for all other sites it is the value for summer 2007.

Site			oss from top	soils	Loss from subsoils			
		winter	summer	total	winter	summer	total	
		$(g C m^{-2})$	$(g C m^{-2})$	$(g C m^{-2} yr^{-1})$	$(g C m^{-2})$	$(g C m^{-2})$	$(g C m^{-2} yr^{-1})$	
Hainich	slope	1.5 (0.2	0.6 (0.1)	2.1 (0.3)	9.1 (0.0)	0.4 (0.0)	9.4 (0.0)	
	valley	2.5 (0.1	1.2 (0.0)	3.6 (0.1)	5.9 (0.1)	0.4 (0.0)	6.2 (0.1)	
Sorø	near tree	1.4 (0.1	1.3 (0.2)	2.7 (0.2)	22.4 (1.1)	22.5 (4.4)	44.9 (4.6)	
	between trees	1.8 (0.2	1.6 (0.2)	3.4 (0.3)	19.4 (0.8)	14.5 (1.5)	33.9 (1.7)	
Laois		4.4 (2.5	2.1 (1.4)	6.5^1 (2.7)	8.5 (1.2)	0.5 (0.1)	9.1 (1.2)	
Loobos		0.7 (0.0	1.3 (0.2)	2.0 (0.2)	0.4 (0.0)	0.4 (0.0)	0.8 (0.0)	
Wetzstein		2.3 (0.0	1.4 (0.1)	3.7 (0.1)	0.9 (0.1)	0.1 (0.0)	1.0 (0.1)	
Average fores	t	2.2 (0.4	1.4 (0.2)	<i>3.6</i> (0.8)	7.6 (3.7)	4.0 (3.6)	<i>11.6 (7.2)</i>	
Median forest	t ·	2.0	1.4	3.1	7.5	0.4	<i>7.8</i>	
Laqueuille	intensive use	2.4 (0.3	6.3 (0.6)	8.7 (0.7)	1.7 (0.3)	4.4 (1.3)	6.1 (1.3)	
	extensive use	2.3 (0.2	6.2 (0.6)	8.4 (0.6)	2.0 (0.2)	2.7 (0.7)	4.7 (0.7)	
Carlow (g)	slope	30.6 (1.7	17.1 (0.7)	47.7 (1.8)	32.5 (2.4)	16.9 (1.6)	49.5 (2.9)	
	valley	25.7 (3.2	13.3 (1.9)	39.1 (3.7)	31.4 (1.3)	14.7 (0.9)	46.1 (1.7)	
Easter Bush	slope	9.8 (0.5	6.3 (0.7)	16.1 (0.9)	8.7 (0.5)	4.4 (0.3)	13.1 (0.6)	
	valley	15.5 (0.7	10.5 (1.7)	26.0 (1.8)	18.5 (0.7)	9.0 (0.7)	27.5 (1.0)	
Früebüel		3.9 (0.4	5.0 (0.4)	8.8 (0.6)	13.4 (1.1)	30.3 (2.0)	43.6 (2.3)	
Average grass	sland	11.8 (5.9	8.7 (2.3)	20.5 (8.2)	<i>14.5 (6.3)</i>	12.6 (4.7)	<i>27.1 (9.2)</i>	
Median grass	land	<i>8.3</i>	7.3	14.9	12.2	11.3	27.6	
Grignon		7.8 (0.2)	0.2 (0.0)	8.0 (0.2)	13.3 (0.3)	0.0 (0.0)	13.3 (0.3)	
Klingenberg		12.6 (0.8	5.5 (0.3)	18.0 (0.8)	12.8 (1.0)	6.6 (0.6)	19.4 (1.2)	
Carlow (c)		13.2 (4.2	7.9 (2.9)	21.1 (5.1)				
Average cropland		<i>11.2 1</i> .	7 4.5 2.3	13.7 3.0	11.3 1.3	3.3 2.7	<i>16.4</i> 2.5	
Median cropl		12.6	5.5	15.2	11.3	<i>3.3</i>	16.4	
Total average DIC flux		7.6 2 4.2			11.1 2.7	7.0 2.2	<i>18.1 4.6</i>	
Median DIC	Median DIC flux		3.6	8.3	10.7	3.6	13.3	

Numbers in brackets indicate standard deviations due to spatial variability of concentrations for individual sites and the standard error of the average across sites with the same land use system or across all sites; the errors do not reflect uncertainties associated with the calculation of the biogenic fraction of DIC with the isotope method (see text); n.d.: not detectable; ¹excluding lateral runoff of 3.7 g DIC m⁻² yr⁻¹ (Walmsley, 2009). Topsoils reach down to the shallow instrumentation depth, subsoil reach down to the deeper instrumentation depth given in Table 1.

Table 5: Leaching losses of DOC plus biogenic DIC in relation to the net ecosystem exchange (NEE) and net ecosystem carbon balance (NECB) of the study sites. Positive numbers indicate net uptake from atmosphere, negative numbers net release of carbon.

Site		Leach	ing los	s from ((sub)soils		Average	NEE +	Leached	NECB
							multi-year	fertilization -	fraction of	
							NEE^1	harvest	balance	
	Wi	nter	Sun	nmer	Tot			excluding leaching		including leaching
	(g C	m^{-2})	(g C	m^{-2}	(g C m	$^{-2} yr^{-1}$)	$(g C m^{-2} yr^{-1})$	$(g C m^{-2} yr^{-1})$	(%)	$(g C m^{-2} yr^{-1})$
Hainich	7.6	(0.1)	0.3	(0.0)	8.0	(0.1)	$+294^{2}$	$+294^{2}$	3	+286
Sorø	19.8	(1.2)	15.5	(3.9)	35.2	(4.1)	$+143^{3}$	$+143^{3}$	25	+108
Laois	7.0	(1.1)	0.4	(0.1)	7.4	(1.1)	+873 ⁴	+873 ⁴	1	+866
Loobos	1.8	(0.1)	1.2	(0.1)	3.0	(0.1)	+402 ⁵	+402 ⁵	1	+399
Wetzstein	5.4	(0.6)	0.3	(0.2)	5.7	(0.7)	+177 ⁶	$+177^{6}$	3	+171
Average forest	<i>8.3</i>	(3.0)	3.5	(3.0)	11.9	(5.9)	378(132)	378(132)	6(5)	366(134)
Median forest	7.0		0.4		<i>7.4</i>		294	294	3	286
Laqueuille	2.2	(0.3)	4.3	(1.4)	6.5	(1.5)	+181 ⁷	+141 ⁷	5	+135
Carlow (g)	25.8	(2.2)	12.6	(1.6)	38.4	(2.7)	+1857	-40^{7}	98	- 79
Easter Bush	18.9	(0.9)	11.1	(0.8)	30.0	(1.3)	+2217	+212 ⁷	14	+182
Früebüel	14.1	(1.1)	28.4	(1.9)	42.5	(2.1)	+3208	+1468	29	+104
Average grassl.	<i>15.3</i>	<i>(5.0)</i>	<i>14.1</i>	(5.1)	29.4	(8.0)	227(32)	115(54)	36(21)	85(57)
Median grassland	<i>16.5</i>		11.9		34.2		203	144	22	119
Grignon	12.5	(0.3)	n.d.	-	12.5	(0.3)	+472 ⁹	-51^{9}	25	-64
Klingenberg	16.0	(1.1)	8.7	(0.6)	24.7	(1.2)	+64 ¹⁰	-103^{10}	24	-128
Carlow (c)	11.2	(3.4)	6.6	(2.3)	17.8	(4.1)	+2224	-17^{4}	105	-35
Average cropland	13.2	(1.4)	7.7	(1.1)	18.3	(3.5)	<i>253(119)</i>	<i>-57(25)</i>	51(27)	-75(27)
Median cropland	12.5		7.7		17.8		222	-51	25	-64
Total average	11.9	(2.1)	<i>8.1</i>	(2.8)	19.4	<i>(4.0)</i>	<i>296(61)</i>	<i>181(76)</i>	28(10)	<i>162(78)</i>
Lower quartile	6.6		0.8		7.2		180	-23	3	-42
Upper quartile	<i>16.7</i>		11.9		31.3		340	232	26	208
Total median	11.9		6.6		15.2		222	145	19	121

net ecosystem exchange = gross primary productivity minus ecosystem respiration excluding C deprivation with harvest, fires etc., Numbers in brackets indicate standard deviations due to spatial variability of concentrations for individual sites and the standard error of the average across sites with the same land use system or across all sites. mean 2000-2007 Kutsch et al. (in prep.), mean 1997–2008 Pilegaard et al. (in prep.), mean 2003–2007 Saunders et al. (in prep.), mean 1997–2008 Moors et al. (unpublished data), mean 2002–2007 Rebmann et al. (2009), Laqueuille: 2003–2006; Easter Bush: 2004–2007, Carlow: 2003–2007 Klumpp et al. (in prep.), mean 2006–2007 Zeeman et al. (2010), mean 2005–2008 Lehuger et al. (2009), mean of four crop periods 2004/2005–2008; a net loss of -255 g m⁻² yr⁻¹ was observed when the application of organic fertilizer to rape in 2004 is excluded from the balance, Prescher et al. (2010).

Table 6: Concentrations of methane (CH₄) in soil air and leaching losses of dissolved CH₄.

		pCH ₄ beneath	pCH ₄ subsoil	CH ₄ leaching from	CH ₄ leaching from
		topsoil		topsoil	subsoil
	μl l ⁻¹			$(g CO_2-C_{eq})$	$m^{-2} yr^{-1}$
Hainich		1.20 (0.13)	0.86 (0.15)	0.005 (0.000)	0.001 (0.000)
Sorø		0.56 (0.08)	0.66 (0.11)	0.004 (0.001)	0.002 (0.000)
Laois		4.82 (1.34)	3.26 (0.54)	0.037 (0.045)	0.018 (0.053)
Loobos		1.18 (0.11)	0.73 (0.06)	0.007 (0.000)	0.004 (0.001)
Wetzstein			0.77 (0.06)	0.007 (0.000)	0.002 (0.000)
Laqueuille	intensive	1.76 (0.11)	1.53 (0.01)	0.009 (0.001)	0.007 (0.000)
-	extensive	1.73 (0.01)	1.73 (0.01)	0.005 (0.000)	0.004 (0.000)
Carlow (g)		1.51 (0.34)	1.45 (0.43)	0.002 (0.000)	0.002 (0.000)
Easter Bush	slope	1.09 (0.03)	1.71 (0.01)	0.012 (0.002)	0.007 (0.000)
	hollow	1.63 (0.13)	29.45 (27.08)	0.031 (0.006)	0.038 (0.005)
Früebüel		1.66 (0.11)	2.04 (0.16)	0.014 (0.001)	0.011 (0.000)
Grignon		0.92 (0.11)	0.69 (0.11)	0.002 (0.000)	0.002 (0.000)
Klingenberg		5.36 (2.27)	2.04 (0.41)	0.147 (0.008)	0.233 (0.009)
Carlow (c)		0.78 (0.09)		0.001 (0.001)	

Numbers in brackets indicate standard deviations due to spatial variability of concentrations. No data can be given for soil air concentration beneath the Wetzstein topsoil because wild boars destroyed the shallow soil air samplers. Topsoils reach down to the shallow instrumentation depth, subsoil reach down to the deeper instrumentation depth given in Table 1.

Figure captions

Fig. 1: Map of the research sites and their land use.



Fig. 2: Concentrations of total dissolved inorganic carbon (open), dissolved organic carbon (hatched), and total dissolved carbon (crosshatched) below the soil A horizon (white) and below the main rooting zone (gray). Concentrations represent the arithmetic mean of the median concentrations of all suction cups in one depth over time. Error bars represent the standard error of 5 suction cups (terrain positions) or 10 suction cups (all cups) and indicate the uncertainty due to spatial variability of concentrations.



Fig. 3: Comparison of measured average total DIC concentrations with average concentrations calculated from average partial pressures of soil air CO₂ and average soil solution pH using Henry's law. No calculated concentration can be given for topsoil leachates of the Wetzstein site, because wild boars destroyed the shallow soil air samplers.



Fig. 4: Average concentrations of DOC leached from topsoils as a function of topsoil C/N ratio. All sites for which C/N ratios were available were included into the relationship.

Averages were calculated as the arithmetic mean of median concentrations of suction cups and suction plates over time. Topsoils reach down to the shallow depth of instrumentation given in Table 1.



Fig. 5: δ^{13} C values (box plots and arithmetic mean as white dot) and the biogenic fraction of DIC (black triangles). The lower value of the biogenic fraction was calculated using a δ^{13} C of -28% for the biogenic end-member and a δ^{13} C of 0% for carbonates. The upper value of the biogenic fraction was calculated using the δ^{13} C of value of soil air CO₂ at the time of soil water collection and a δ^{13} C of 0% for carbonates. Use of a δ^{13} C of -5% for carbonate-C reduced the calculated biogenic fraction on average by 5% (biogenic end-member of -28%).



Fig. 6: Precipitation (a), evapotranspiration (b), measured and modeled soil water volumes (c, d), and modelled groundwater recharge (e) for the Carlow grassland. The average relative deviation of the modelled soil water volume was 14% for the A-horizon and 6% for the B-horizon (Table 2).



Fig. 7: Retention of DOC in B horizons of soils related to the ratio of organic carbon (OC) and the sum of oxalate-extractable iron (Fe_o) and aluminium (Al_o) of B horizons. Data for oxalate-extractable iron and aluminium were only available for the displayed sites.



Figures

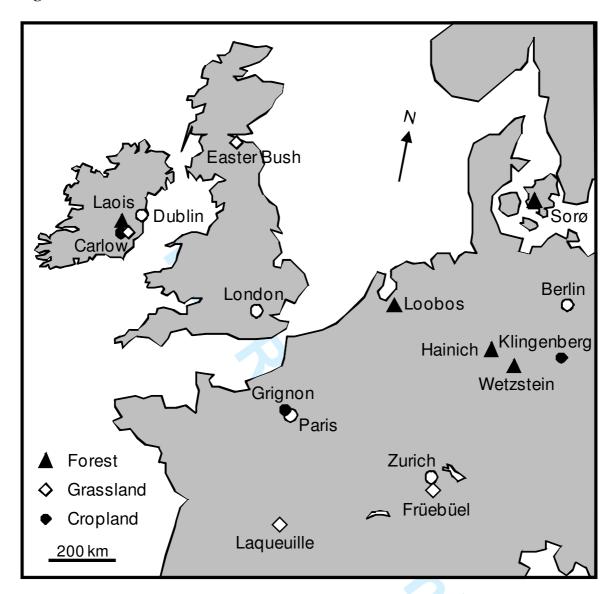


Figure 1

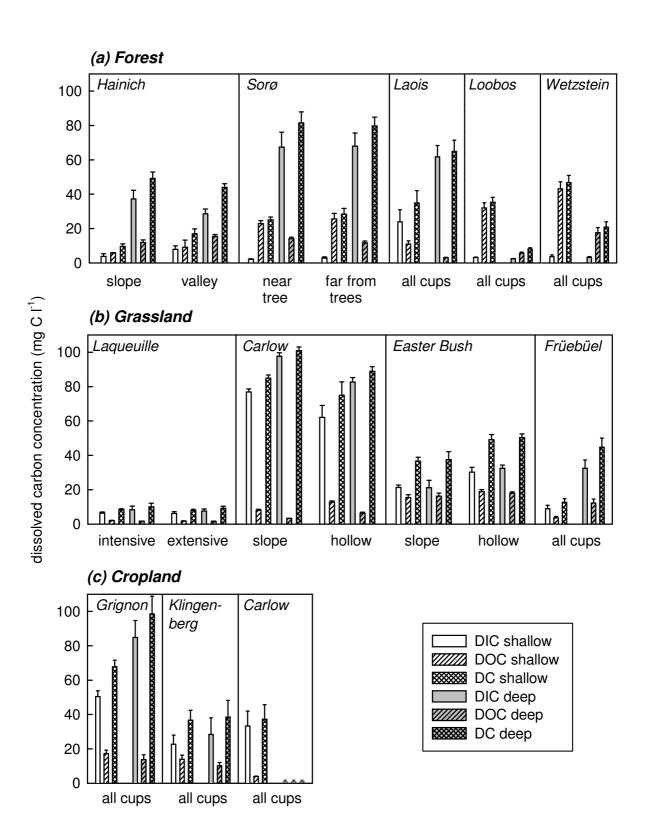


Figure 2

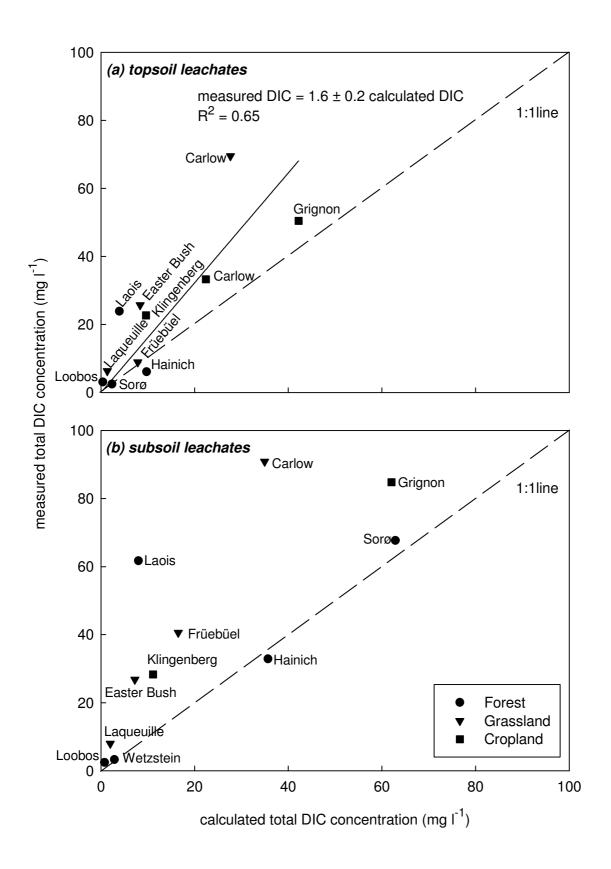


Figure 3

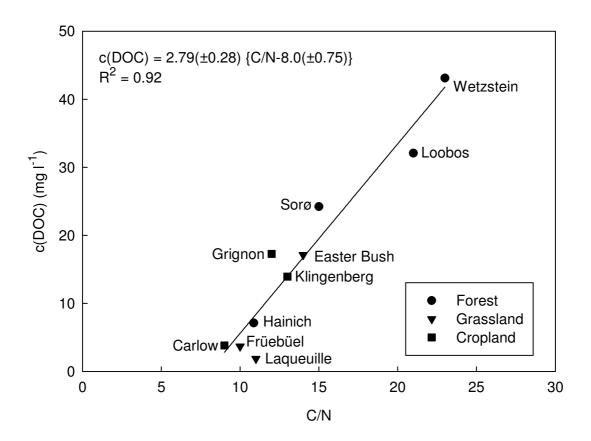


Figure 4

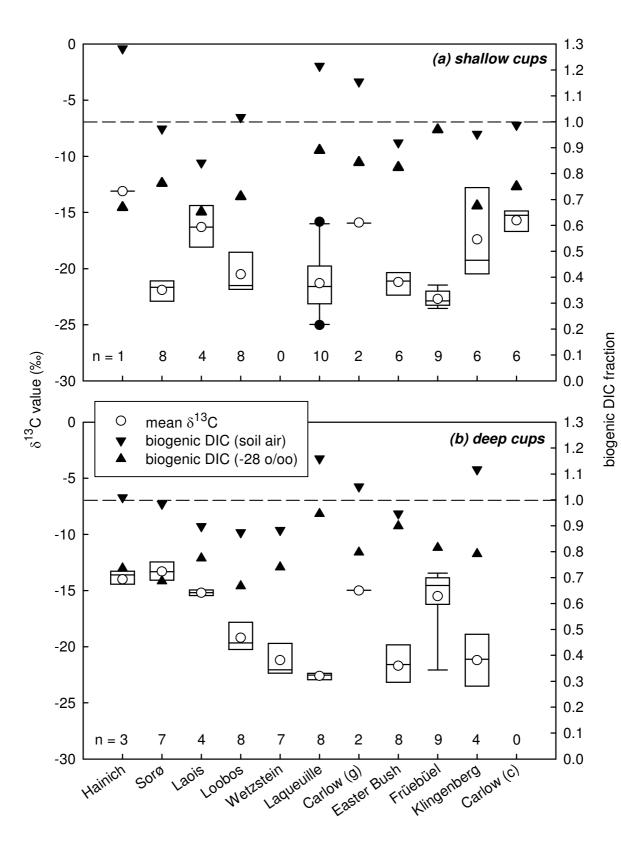
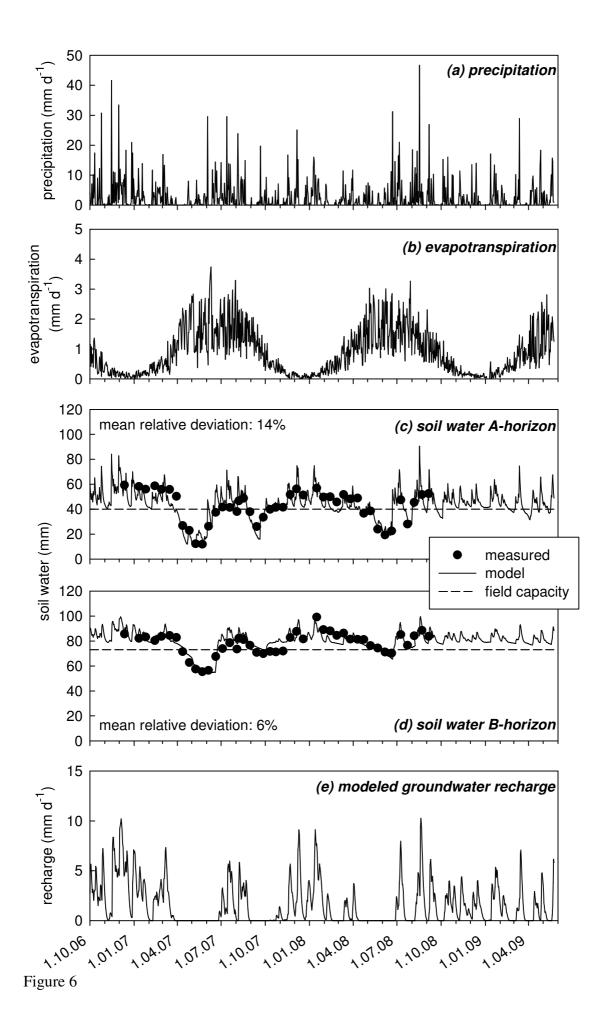


Figure 5



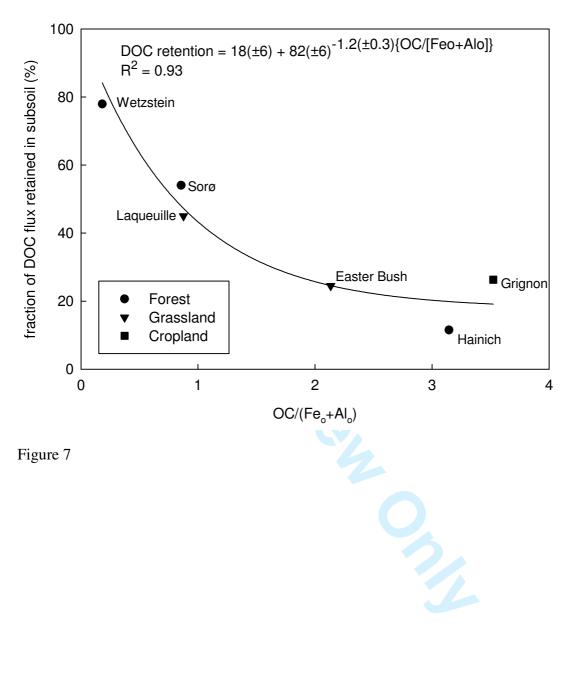


Figure 7