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Enhanced anomalous diffusion of sputtered atoms in nanosized pores

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Abstract

Plasma sputtering deposition of platinum catalysts in porous anodic aluminum oxide (AAO) templates is shown to generate an anomalous superdiffusion concentration profile. The growth of an overlayer between the hexagonal array of pores is shown to enhance the diffusion into the pores, leading to a diffusion coefficient having superlinear time dependence. The Pt clusters in the pores have a mean size of 10 nm and almost concentrate on the inner pore surface, and are present up to a depth of 6 μ m.

Keywords: atom anomalous diffusion, cylindrical pore, atom transport, concentration profile

1. Introduction

Efficient development of highly functional porous materials, used as catalysts in the automotive, fuel cells, batteries, etc., is particularly welcome in reducing the losses of the precious metals and enhancing their catalytic efficiency [1, 2, 3]. The catalytic efficiency of these advanced porous materials strongly depends on the distribution of the nanoparticles, the concentration and size distribution in the porous media [3, 4, 5], hence these quantities

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must be predicted. Plasma sputtering deposition is known to provide a relevant way for designing catalytic thin films [1, 2, 6, 7, 8, 9]. The growth of the active layer is thus obtained by transport/diffusion/clustering steps of the sputtered atoms impinging the porous substrate. Usually, the diffusion in a porous medium is called anomalous and is characterized by the mean square displacement $\langle z^2(t) \rangle$ which evolves according to a power law in time t^{α} different from the well known linear behavior ($\alpha = 1$) for normal diffusion[10]. This anomalous diffusion can be observed in many domains such as transport in fractal media [11], solid surface diffusion [12] or hydrodynamics (rotating flows [13], turbulence [14] or diffusion in an array of convection rolls [15, 16]). Different models that involve space-dependent diffusion coefficients can be used to describe this anomalous diffusion process, [17, 18, 19, 20, 21]. In the present article, we study plasma sputtering platinum deposition on porous AAO templates by measuring the density profiles of the deposited matter at different times and compare them with concentration profiles issued from anomalous diffusion.

2. Experiment

Platinum is deposited by a radio frequency inductive plasma sputtering system which has been previously described [2, 3]. Briefly, an argon plasma is created in a 80 liter cylindrical stainless steel chamber by an external planar antenna powered by a 13.56 MHz using a Radio Frequency generator at 100 W input power. AAO templates are placed on a rotating substrate holder in front of the sputtering target at a distance of 7.0 cm. The Pt target is biased with a continuous DC potential V_b = -100 V and the deposition time t is in the range of 10 min to 50 min at argon pressure of 3.10^{-3} mbar. The morphology of the Pt coated AAO templates were analyzed by high resolution scanning electron microscopy (HRSEM, ZEISS). The platinum depth profiles were measured by Rutherford Backscattering Spectroscopy, with a Van de Graaf accelerator (CEMHTI-CNRS, Orléans, France) using a 2 MeV 4 He $^+$ ion beam. The scattering geometry is : scattering angle = 165 °, exit angle is 15 °, incident angle is 0 °. The detector size is 25 mm 2 and sample to detector distance is 90 mm.

3. Results and discussion

The AAO templates (purchased Pu-Yan Nanotechnologies, China) are made of cylindrical pores with average pore size of 90 nm, a porosity of 25%,

and the density of the pore is 4.7 10⁹ cm⁻². The pore are perpendicular to the external surface. The Pt deposited AAO templates have been observed by SEM (Fig. 1). The bright areas corresponds to the presence of platinum. The particle flux delivered by the plasma sputtering cannot be completly absorbed by the porous AAO template. Part of the Pt atoms is accumulated on the surface and part is deposited into the porous AAO and adsorbed on the inner walls as can be be shown in Fig. 1. As deposition time is increased from 10 min to 50 min, the amount of Pt clusters staying on the surface is increased. In addition, Pt clusters on the channel inner surface can be observed (Fig. 1d). The clusters have 10 nm size. Moreover EDS line-scan measurements (Fig. 2 along a pore cross- section confirms that Pt is significanltly incorporated inside the pores. To get more information about the depth concentration profiles of the Pt atoms in the porous AAO template, the samples are further investigated by RBS. A simulated spectrum is built using SIMNRA software [22]. it is obtained by considering a target made of successive layers with a given concentration of Pt and Al₂O₃. An iteration process is carried out until the simulated spectrum matches the experimental spectrum. The Pt depth concentration profiles are recovered when converting the Al₂O₃ concentration in a layers to an equivalent thickness interval, taking into account the porosity. Moreover, because the incident beam is perpendicular to the AAO surface, thus the Pt line shape mainly reflects the concentration profile, i.e. intensity at lower backscattered α particle energies do correspond to adsorbed Pt atoms on the pore wall. Because incident angle is 0°, a fraction of the incident beam is not scattered so the charge should be corrected by the fraction of pore area, i.e. 25 %.

Typical RBS spectra of Pt deposited AAO are shown in Fig. 3. The left side of the spectrum, at low backscattering energies, corresponds to Al, O atoms, whereas on the right side, the asymmetrical peak originates from α particles backscattered by Pt atoms. The area under the Pt peak is directly related to the number of the platinum atoms in the porous substrate. The tail on the left side of the Pt peak indicates that atoms have penetrated into the porous substrate. This significant increase of the area under the peak and the extending of the tail reveal that the number of Pt atoms in the AAO templates is increasing and more Pt atoms are present in the channels as the sputtering time is increased. This is consistent with EDS measurements.

The depth density profiles (Fig. 4) of several successive Pt deposited AAO templates are obtained by fitting the experimental spectrum with a spectrum using a defined concentration profile. The best choice is given by

the solution of anomalous diffusion[10, 17, 18, 19, 20]:

$$\rho(z,t) = Z_1(t), -z_0(t) < z \le 0$$

$$\rho(z,t) = Z_1(t)e^{-\frac{z^{2+\theta}}{Z_2(t)}}, z > 0 (1)$$

In Eq. 1, the depth profile is composed of a surface layer of height $z_0(t)$ and by a stretched gaussian concentration profile. θ is a dimensionless coefficient which characterizes the anomalous diffusion behaviour. If $\theta < 0$ then the behaviour is superdiffusive else if $\theta > 0$ it is subdiffusive. In our experiments, θ equals to $-\frac{5}{3} \pm 0.15$, which means that Ar plasma sputtered Pt atoms are subjected to superdiffusion in the cylindrical pores. More over, $Z_1(t)$ and $Z_2(t)$ exhibit power law dependence: $Z_1(t) \propto t^{0.75\pm0.05}$ and $Z_2(t) \propto t^{1.25\pm0.05}$. The maximum reached depth of Pt is increasing from 1.5 μ m to 6 μ m as the deposition time is increasing from 10 min to 50 min. The density of the Pt atoms on the surface (overlayer) is increased when increasing deposition time. Figure 5 only displays a small depth range (< 400 nm) for clarity.

We obtain an almost linear evolution of the total mass $(t^{1.\pm0.05})$ of deposited platinum, as expected by the experimental process: the flux of Pt atoms impinging on the AAO template is constant. The mass of platinum inside the porous medium is found to scale as $t^{1.5\pm0.05}$, while the Pt mass of the surface overlayer scales as $t^{0.7\pm0.05}$ as shown on Fig. 5. The rate of nonlinear mass increase on the surface is smaller than rate of total mass one, while the rate of mass increase inside the pores is larger than the one of total mass. It indicates that part of the atoms deposited on the surface are gradually diffusing into the channels as deposition time is increasing, while part of the incoming flux is directly entering the pores i.e. with an amount corresponding to the ratio of the pores area to the total area. This growing external layer provides an additional source for the transport/diffusion of atoms inside the porous AAO and is dependent on the deposition time. In the process of deposition, Pt atoms arriving with a relatively high mean kinetic energy (around 7 eV [3]) and Ar ions from the plasma are also impinging the substrate [23, 24, 25]. So this superlinear process can be due to the flux of sputtered Pt atoms which do not only have collisions with AAO template, but also with the Pt atoms or Pt clusters deposited before and thus caused these Pt atoms to be driven into the channels. In addition, the surrounding plasma also has an interaction on these surface Pt atoms, mainly metastable atoms and ions that can cause another energy transfer [26, 27]

favouring indiffusion when colliding with already deposited Pt atoms. Thus, the outer surface Pt atoms behave like an additional source for the diffusion of atoms inside the channels. Following ref. [21], we find a time dependant diffusion coefficient $D(z,t) = D_0 \frac{t^{\gamma}}{z^{\theta}}$ with $\gamma = \frac{1}{4}$ (see Appendix). This means that the diffusion is increased when increasing deposition time, i.e. when increasing the deposited mass. One can also notice, this behaviour is opposite to deposition into porous carbon [21] in similar conditions, for which mass inside porous layer is increasing slower than linearity. Because the porosity of the AAO is lower than in the porous carbon layer, we can invoke that the lower tortuosity of the AAO (the tortuosity of a random sphere packing is higher than for a cylindrical pore lattice) can be responsible for a superlinear inside mass increase.

4. Conclusion

During sputtering deposition, the diffusion of sputtered Pt atoms in the porous AAO substrate is found to be anomalous and it displays superdiffusion behaviour. Pt particles are not only deposited on the surface, part of them also diffuse into the porous substrate. This external growing layer provided an additional dynamical source for the diffusion of atoms inside the porous substrate and caused the increase of Pt mass inside the porous substrate in a superlinear way. The Pt particles with a mean size of 10 nm almost concentrate on the depth less than 400 nm and the reached maximum penetration depth increased to 6 μ m while the deposition time is increased to 50 min.

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Appendix

In Ref.[21], scaling laws for anomalous diffusion equation with space and time dependent diffusion coefficient $D(z,t)=D_0\frac{z^{\theta}}{t^{\gamma}}$ are :

$$\alpha = \frac{\gamma + 1}{2 + \theta}$$
 and $\alpha + \beta = 1 + \epsilon$ (2)

with α and β are issued from experiments considering that the concentration profile $\rho(z,t)$ can be considered as a self-similar solution of the anomalous diffusion equation, $\rho(z,t) \propto t^{\beta} f\left(\frac{z}{t^{\alpha}}\right)$ and that the experimental fitting

functions are $\rho(z,t) = Z_1(t)e^{-\overline{Z_2(t)}}$ with $Z_1(t) \propto t^m$ and $Z_2(t) \propto t^p$. So $\alpha = \frac{p}{2+\theta}$ and $\beta = m$. And thus parameters γ and ϵ can be deduced using the set of equations (2).

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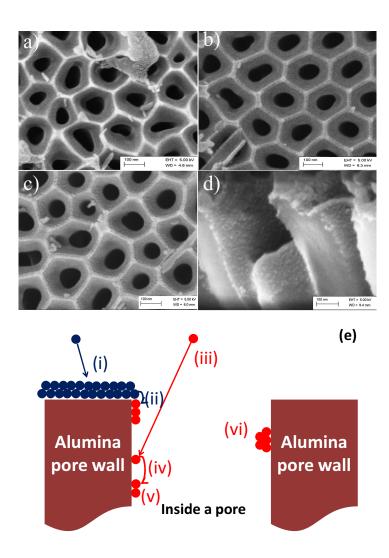


Figure 1: Scanning Electron Microscopy images of Pt deposited AAO template: a) t=5 min; b) t=30 min; c) t=50 min; d) Cross-section of Pt deposited 50 min; e) Basic mechanism involved in the sputter deposition process i) deposition on the top surface separating pores ii) possible diffusion from top surface to inner pore surface iii) direct deposition into a pore iv) inner pore surface diffusion v) nucleation process in a pore vi)cluster growth inside a pore

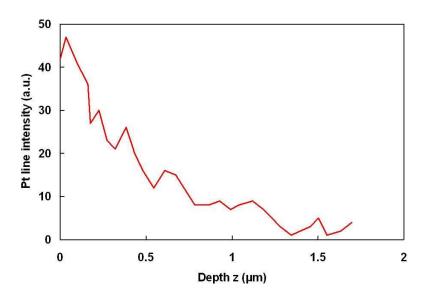


Figure 2: Typical Pt EDS line scan profile along a pore. z = 0 is the outermost surface.

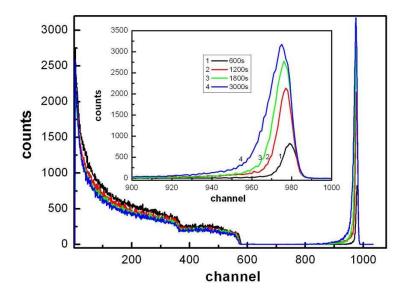


Figure 3: RBS spectra of different time Pt deposited AAO templates.

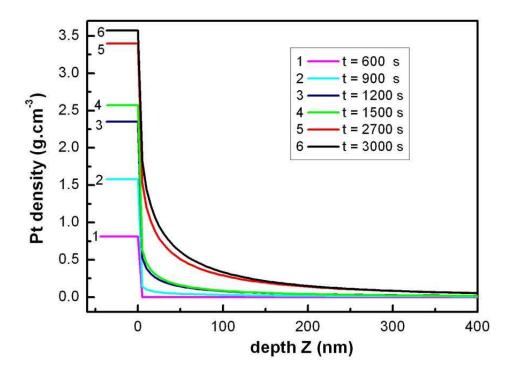


Figure 4: Pt depth density profiles in AAO template at different time in a range less than $400~\mathrm{nm}$.

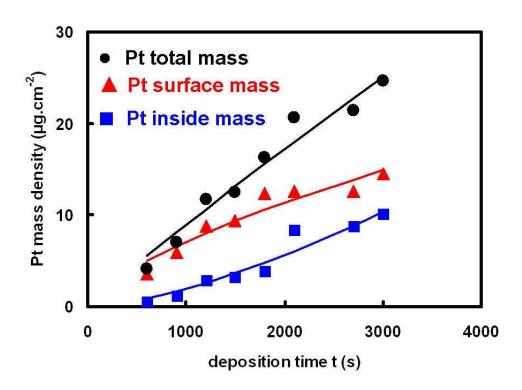


Figure 5: Time evolution of the surface, inside and total mass of deposited platinum. The lines are the power law fits.