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## Food Additives and Contaminants

## RESEARCH ARTICLE

# Determination of sterigmatocystin in cheese by high performance liquid chromatography - tandem mass spectrometry 

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#### Abstract

Different cheese samples, produced in Latvia (8) and Belgium (13), were analysed for their sterigmatocystin (STC) content. Only two ( $9.5 \%$ ) of the samples were positive for STC with concentration levels of 1.23 and $0.52 \mu \mathrm{~g} \mathrm{~kg}^{-1}$ respectively. Five ( $24 \%$ ) samples contained STC above the limit of detection ( $0.03 \mu \mathrm{~g} \mathrm{~kg}$ ) but below the limit of quantification $\left(0.1 \mu_{\mathrm{g} \mathrm{kg}} \mathrm{kg}^{-1}\right)$, while five samples contained STC above LOD, but below LOQ. A sensitive liquid chromatography electrospray positive ionization - tandem mass spectrometry (LC-MS/MS) method, which was previously developed for the analysis of STC in grains, was modified and applied to the analysis of STC in cheese. This method involved sample extraction with acetonitrile:water (90:10, v/v), defatting with n-hexane, solid phase extraction, separation on a reversed phase $\mathrm{C}_{18}$ column and STC detection by LC-MS/MS.


Keywords: Sterigmatocystin, mycotoxin, cheese, LC-MS/MS, Aspergillus spp., Aspergillus versicolor

[^0]
## Introduction

Mould growth on cheese and, to a less extent, other fermented dairy products stored at low temperatures is a common and recurrent problem for the cheese manufacturer during aging and for the retailer and consumer during storage. Mould growth on cheese causes economic losses from discoloration, poor appearance, and off-flavours. However, of even more serious concern is that some moulds are capable of producing toxic metabolites known as mycotoxins (van Egmond 1989). Mycotoxins are toxic secondary metabolites produced by various fungal species. The presence of these substances in dairy products can have two origins: a) indirect contamination, which results from lactating animals ingesting contaminated feed and b) direct contamination, which occurs because of intentional or accidental growth of moulds on dairy products (van Egmond 1989).

Some of the fungal starters used in cheese manufacturing have been shown to produce various mycotoxins on culture media (Engel and Teuber 1989). Other moulds besides starters may be present in mould cheeses as contaminants, since various fungi are encountered in the environment of the cheese factory and some of these organisms can be potential mycotoxin producers (Lafont et al. 1990; Lund et al. 1995). Reports indicating the occurrence of mycotoxins due to contaminating fungi in mould cheeses are infrequent, in different cheese samples from Finnish supermarkets only mycophenolic acid and roquefortine C were detected in blue cheese samples (Kokkonen et al. 2005). The presence of mycotoxins in dairy products is suggested to be related more to environmental factors causing mould growth than to ingestion of mouldy feed by cows (Lafont et al. 1990).

Aflatoxin producers - Aspergillus flavus (A. flavus) and Aspergillus parasiticus (A. parasiticus) were seldom found in cheese. However, Aspergillus versicolor (A. versicolor), a species known to be capable of producing sterigmatocystin (STC), seemed to be frequently present. A. versicolor isolates from cheese and air in the cheese ripening room were all able to produce STC (Lund et al. 1995). STC is of interest because of its relationship to aflatoxin. It contains the same 7,8-dihydrofuro[2,3-b]furan moiety, which for STC is fused to a hydroxyl- and methoxy- substituted xanthone (Rodricks 1969). STC is a metabolite not only of A. versicolor, but also of A. flavus, A. Sydowi, A. nidulans, Bipolaris, Chaetomiun and Emiricella spp. (Davis 1981; Terao 1983; Scott 1994; Scott 2004; Frisvad and Thrane 2004). Its acute toxicity, mutagenity, cytotoxicity and carcinogenity have been reviewed by Terao (1983) and it is classified as a 2 B carcinogen by the International Agency for Research on Cancer (IARC 1987). The structure of the adduct of STC at the N-guanine of oligodeoxynucleotides was determined (Gopalakrishnan et al. 1992). A 'no
significant risk' level for humans of $8 \mu \mathrm{~g} \mathrm{~kg}$ - of body weight per day was estimated by the State of California (Scudamore et al. 1996). STC has been found in cheese contaminated with A. versicolor (Scott 1989). Lactose, glycerol, high humidity and temperature are stimulating factors in STC production by A. versicolor on cheese (Veringa et al. 1989). Northolt et al. (1980) reported the occurrence of STC in cheese ripening in warehouses in the Netherlands. However, low temperatures $\left(5^{\circ} \mathrm{C}\right.$ to $\left.7^{\circ} \mathrm{C}\right)$ should prevent the growth of A. versicolor and production of STC. At these temperatures, the moulds that develop will be Penicillium species, which do not produce aflatoxins or STC (Bullerman 1981).

A number of analytical methods have been published to examine STC in cheese matrix. Thin layer chromatography (TLC) with limits of detection (LODs) around $20 \mu \mathrm{~g} \mathrm{~kg}^{-1}$. UV or postcolumn fluorescence with aluminium chloride was described for liquid chromatography (LC) detection. Aluminium chloride was also used as a spray reagent for two dimensional TLC with densitometric detection (LOD $1 \mathrm{gg} \mathrm{kg}^{-1}$ in feed) (Domagala et al. 1997). The most sensitive method for STC detection in cheese was a liquid chromatography - atmospheric pressure chemical ionization - mass spectrometry (LC-APCI-MS) method with a LOD of $2.4 \mu \mathrm{~g} \mathrm{~kg}^{-1}$ (Scudamore et al. 1996). Therefore no information is available on lower contamination levels which are also of interest.

In this study, a previously developed LC - electrospray positive ionization (ESI) - tandem mass spectrometry (MS/MS) method for the determination of STC in different grains (Versilovskis et al., 2007) (LOD $0.15 \mu \mathrm{~g} \mathrm{~kg}^{-1}$ ) was modified and applied to the analysis of STC in different cheese samples. This is the first described LC-ESI-MS/MS method for STC determination in cheese with solid phase extraction (SPE) and also, the first time cheese samples produced in Latvia and Belgium were checked for STC contamination.

## Materials and Methods

## Cheese samples

Cheese samples produced in Latvia (8) and Belgium (13) were purchased during first two month of 2008 year from different local supermarkets in Latvia and Belgium. Samples were homogenized by a food processor (Valentin, SEB, Selongey Cedex, France), and kept frozen ( $-20^{\circ} \mathrm{C}$ ) before analysis.

## Chemicals and reagents

Methanol (HPLC-grade), acetonitrile (HPLC-grade), n-hexane (HPLC-grade) and chloroform (HPLC-grade) were purchased from Merck (Darmstadt, Germany). Formic acid was from Fluka
(Buchs, Switzerland). Deionized water was purified with a Millipore Milli-Q Plus system (Millipore, Brussels, Belgium). Argon (Air Liquide, Belgium) was used as collision gas in mass spectrometry. STC standard was purchased from Sigma (St Louis, MO, USA). A stock solution of $500 \mu \mathrm{~g} \mathrm{ml}{ }^{-1}$ was prepared by dissolving 5 mg of STC in 10 ml of chloroform. An aliquot ( $500 \mu \mathrm{l}$ ) of the stock solution was evaporated to dryness under oxygen-free nitrogen at ambient temperature and immediately redissolved in acetonitrile ( 5 ml ) getting a standard solution of $50 \mu \mathrm{~g} \mathrm{ml}{ }^{-1}$. This was further used to prepare a standard solution of $5 \mu \mathrm{~g} \mathrm{ml}^{-1}$ of STC in acetonitrile:water (60:40, $\mathrm{v} / \mathrm{v})$. This solution was used to spike samples for recovery experiments and to prepare working standards $1.6 \mu \mathrm{~g} \mathrm{ml}^{-1}, 0.8 \mu \mathrm{~g} \mathrm{ml}^{-1}, 0.2 \mu \mathrm{~g} \mathrm{ml}^{-1}, 0.08 \mu \mathrm{~g} \mathrm{ml}^{-1}, 0.04 \mu \mathrm{~g} \mathrm{ml}^{-1}, 0.008 \mu \mathrm{~g} \mathrm{ml}^{-1}$ and 0.004 $\mu \mathrm{g} \mathrm{ml}^{-1}$ as equivalents to $200 \mu \mathrm{~g} \mathrm{~kg}{ }^{-1}, 100 \mu \mathrm{~g} \mathrm{~kg}{ }^{-1}, 25 \mu \mathrm{~g} \mathrm{~kg}^{-1}, 10 \mu \mathrm{~g} \mathrm{~kg}{ }^{-1}, 5 \mu \mathrm{~g} \mathrm{~kg}^{-1}, 1 \mu \mathrm{~g} \mathrm{~kg}{ }^{-1}$ and $0,5 \mu \mathrm{~g} \mathrm{~kg}^{-1}$ in cheese matrix.

## Sample preparation

The sample preparation was a slight modification of the method of Veršilovskis et al. (2007) for the determination of STC in grains. Ten grams of cheese were weighed and 50 ml of acetonitrile:water ( $90: 10, \mathrm{v} / \mathrm{v}$ ) and 30 ml of n -hexane were added. The sample was homogenized using a shaker (Agitelec, J. Toulemonde \& Cie, Paris, France) for 30 minutes and then centrifuged ( $3500 \mathrm{rmp}, 10$ $\min ,+15^{\circ} \mathrm{C}$ ). The acetonitrile phase was filtered through filter paper 589 (Schleicher \& Schuell, Dassel, Germany). Filtered extract ( 10 ml ) was diluted with 30 ml of water and purified using a Strata X ( 60 mg ) SPE column (Phenomenex, Torrance, CA, USA). The SPE column was conditioned with 3.5 ml methanol followed by 3.5 ml of water prior to use. Then, 40 ml of diluted extract was loaded on to the column (using loading reservoirs of 75 ml (UCT, Bristol, PA, UK) and manifold (Waters, Milford, MA, USA)). The column was twice washed with 3.5 ml of acetonitrile:water (30:70, v/v) and STC was eluted with 3 ml of $100 \%$ acetonitrile. The resulting eluate was evaporated to dryness under nitrogen at $37^{\circ} \mathrm{C}$ and re-dissolved in $250 \mu 1$ mobile phase ( $0.02 \%$ formic acid in acetonitrile and $0.02 \%$ formic acid in water ( $60: 40 \mathrm{v} / \mathrm{v}$ )). The calibrants were prepared by spiking blank matrix with the standard and prepared in the same way as the samples. Spiked cheese samples were prepared by adding $400 \mu \mathrm{l}, 200 \mu \mathrm{l}, 50 \mu \mathrm{l}, 20 \mu \mathrm{l}$ and $10 \mu \mathrm{l}$ of the $5 \mathrm{\mu g} \mathrm{ml}^{-1}$ STC standard solution using a digital pipette to 10 g of sample, which was left for 1 h at ambient temperature with occasional agitation to allow the acetonitrile to evaporate. These volumes of standard were equivalent to levels of $200 \mu \mathrm{~g} \mathrm{~kg}$, $100 \mu \mathrm{~g} \mathrm{~kg}{ }^{-1}, 25 \mu \mathrm{~g} \mathrm{~kg}^{-1}, 10 \mu \mathrm{~g} \mathrm{~kg}{ }^{-1}$ and $5 \mu \mathrm{~g} \mathrm{~kg}{ }^{-1}$ of STC in cheese respectively. Eight replicates at each concentration level were prepared for recovery experiments.

## LC-MS/MS analysis

The LC - MS/MS system consisted of a Waters Alliance 2695 separation module with a $100-\mu \mathrm{l}$ injection loop (Waters, Milford, MA, USA) coupled to a Quattro LC triple quadrupole mass spectrometer (Waters-Micromass, Manchester, UK) equipped with an electrospray interface. An electrospray ionization (ESI) probe in the positive mode was used in the analysis of STC. The mobile phase was used in isocratic regime. A Phenomenex Luna $C_{18}(2)(5 \mu \mathrm{~m}), 150 \times 3.0 \mathrm{~mm}$ (Phenomenex, Torrance, CA, USA) column was used. The flow rate was $0.3 \mathrm{ml} \mathrm{min}^{-1}$, column temperature was $30^{\circ} \mathrm{C}$ and the injection volume was $50 \mu$ l. The parameters of the mass spectrometer were optimized using the STC standard. The best response was recorded with the following parameters: cone voltage 30 V , capillary voltage 3.5 kV , extractor 2 V , radio frequence ( RF ) lens 0.2 V , source temperature $120^{\circ} \mathrm{C}$ and desolvation temperature $350^{\circ} \mathrm{C}$, cone gas flow $1081 \mathrm{~h}^{-1}$, desolvation gas flow $660 \mathrm{l} \mathrm{h}^{-1}$, collision energy 30 eV , dwell 0.2 sec .
For the structural identification in selected reaction monitoring (SRM) mode, the molecular ion $[\mathrm{M}+\mathrm{H}]^{+}(\mathrm{m} / \mathrm{z}=325)$ was fragmented within the MS to its product ions ( $325>310$ and $325>281$ ). Argon at a pressure of 3.5 bar was used as collision gas. A calibration curve was constructed using external standardization in matrix. The product ion with $m / z=281$ was used for the quantification of STC. The ratio between peaks of STC obtained on two MRM channels (Peak area (325 > 310) / Peak area ( $325>281$ )) was used for confirmation of the analyte. This ratio should be $0.7 \pm 0.1$ for the compound to be confirmed. The MRM chromatograms of blank cheese sample (a), spiked cheese sample ( $0.5 \mu \mathrm{~g} \mathrm{~kg}^{-1}$ ) (b), naturally contaminated cheese sample (c) and sample containing traces of STC (d) are shown in Figure 1. The following validation parameters were determined: linearity, LOD, limit of quantification (LOQ), recovery and repeatability. The LOD and LOQ of STC were determined by using standards prepared in the blank matrix and calculated using "Masslynx" software integrated calculation (LOD $=3.3 \times \mathrm{S} / \mathrm{N}-\mathrm{ratio} ; \mathrm{LOQ}=10 \times \mathrm{S} / \mathrm{N}$-ratio).

## Results and Discussion

## Method performance and validation parameters

In LC-MS methods the food matrix often causes changes in responses because matrix components can interfere with the ionization of the analytes (Tang and Kebarle 1993; Spanjer et al. 2007). Because of this matrix effect, the calibrants were prepared in blank matrix (Gouda cheese produced in Belgium), in two linear ranges from $0.5 \mu \mathrm{~g} \mathrm{~kg}^{-1}$ to $25 \mu \mathrm{~g} \mathrm{~kg}$ and from $10 \mu \mathrm{~g} \mathrm{~kg}^{-1}$ to $200 \mu \mathrm{~g} \mathrm{~kg}$. The method was linear for STC from $0.5 \mu \mathrm{~g} \mathrm{~kg}^{-1}$ to $25 \mu \mathrm{~g} \mathrm{~kg}{ }^{-1}\left(\mathrm{y}=2921 \mathrm{x}+82, \mathrm{R}^{2}>0.991\right)$ and from 10 to $200 \mu \mathrm{~g} \mathrm{~kg}^{-1}\left(\mathrm{y}=2656 \mathrm{x}+617, \mathrm{R}^{2}>0.990\right)$. A tolerance of $\pm 10 \%$ was accepted for the different calibration points for good linearity. The lowest validated level for STC in cheese was 0.5
$1 \quad \mu \mathrm{~g} \mathrm{~kg}^{-1}(\mathrm{~S} / \mathrm{N}=25)$. The LOD and LOQ were $0.03 \mu \mathrm{~g} \mathrm{~kg}^{-1}$ and $0.1 \mu \mathrm{~g} \mathrm{~kg}^{-1}$, respectively. A good recovery was achieved for STC and the relative standard deviation (RSD, \%) demonstrated acceptable repeatability (Table 1).

## Presence of STC in cheese samples

The results on the presence of STC in cheese are shown in Table 2. STC above LOQ was only detected in two ( $15 \%$ ) of the 13 Belgian cheese samples. In four ( $50 \%$ ) of the 8 Latvian cheese samples STC was detected above LOD, but below LOQ. Five ( $24 \%$ ) of the 21 analysed samples contained STC above LOD, but below LOQ and five ( $24 \%$ ) samples contained traces of STC. In the past only limited studies on the occurrence and stability of STC in cheese were carried out. STC was usually detected in Dutch hard cheeses like Gouda or Edam (van Egmond et al. 1982). Northolt et al. (1980) found that nine of 39 cheeses were positive for STC. These positive samples were obtained from all different cheese categories with levels ranging from 5 to $600 \mu \mathrm{~g} \mathrm{~kg}{ }^{-1}$. In other studies the stability of STC in cheese was checked and it was found to be stable for 3 months at various temperatures (van Egmond 1986). The occurrence, distribution and stability of STC were also studied in Ras cheese in Egypt and $35 \%$ was found positive with a mean concentration of 22 $\mu \mathrm{g} \mathrm{kg}{ }^{-1}$. In Ras cheese contaminated with spores of $A$. versicolor, toxin production started after 45 days of ripening, reaching a maximum after 90 days. Aged cheese (more than 6 month) inhibited toxin production (Abd Alla et al. 1996). Migration studies of STC in Wilstermarsh cheese after inoculation with $A$. nidulans and incubation at $15^{\circ} \mathrm{C}$ or room temperature for 40 days was limited to 0.5 cm from the cheese surface (Engel and Teuber 1980). Similar results were obtained with six naturally contaminated Gouda cheeses on which A. versicolor was growing (van Egmond et al. 1982).

In our study cheese samples were different from soft (like Kamambers (camembert type) and Bouquet Des Moines) to hard (like Parito, Montenegro and Breugel). Most of hard cheeses do not contain any typical moulds producing STC, so contamination can only be possible during cheese making or cheese storage if they are stored at temperatures above $7^{\circ} \mathrm{C}$. Also contamination of camembert type cheese is only possible during cheese making and storage, because Penicillium camemberti does not produce neither spores nor STC. But in one Latvian camembert type cheese sample (Kamambers 1) STC was detected, probably due to contamination during cheese making or during storage. Also other cheese samples contained STC traces or STC above LOD.

Our results show that a risk of contamination of cheese with STC exists. These very low concentrations will not acutely affect consumers' health. Only long-term intake of small
concentrations of toxins might have possible adverse effects or synergistic effects with other toxins.

## Conclusions

This is the first simple and sensitive LC-ESI-MS/MS method for the determination of STC in cheese that can be applied for routine and confirmation analyses. This is the first study of STC in Latvian and Belgium cheese that indicates a low exposure rate to STC.

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(a)

(b)

(c)

(d)

| Figure 1. SRM chromatograms of blank cheese sample (a), spiked cheese sample ( 0.5
Deleted: M $\mu \mathrm{g} \mathrm{kg}^{-1}$ ) (b), naturally contaminated cheese sample (Chevre Affine - contaminated with $1.23 \mu \mathrm{~g} \mathrm{~kg}^{-1}$ ) (c) and cheese sample containing traces of STC (Krievijas) (d).

Table 1
Average recovery of STC from spiked cheese samples obtained during two days
Deleted: R

| Added <br> concentration, <br> $\mu \mathrm{g} \mathrm{kg}^{-1}$ | Found <br> concentration <br> $($ Mean $\pm$ S.D. $)$, <br> $\mu \mathrm{g} \mathrm{kg}$ | Mean recovery <br> $(\mathrm{n}=8), \%$ | Precision <br> RSD, $\%$ |
| :---: | :---: | :---: | :---: |
| 5 | $5 \pm 1$ | 97 | 23 |
| 10 | $10 \pm 3$ | 96 | 27 |
| 25 | $26 \pm 7$ | 103 | 25 |
| 100 | $104 \pm 25$ | 104 | 24 |
| 200 | $205 \pm 36$ | 103 | 17 |

Table 2
STC presence in different Latvian and Belgian cheese samples

| Sample № | Sample Name | Cheese type | Produced in | $\begin{gathered} \text { Found STC } \\ \text { concentration, } \\ \mu \mathrm{g} \mathrm{~kg}^{-1} \end{gathered}$ | Final decision |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | Zelta | Semi hard | Latvia | 0.04 | <LOQ |  |
| 2 | Krievijas | Semi hard | Latvia | $<L^{\text {L }}$ | <LOD | Deleted: (traces) |
| 3 | Hollandes | Hard | Latvia | 0.04 | <LOQ |  |
| 4 | Parito | Hard | Latvia | 0.07 | <LOQ |  |
| 5 | Montenegro | Hard | Latvia | $<L^{\text {L }}$ | <LOD | Deleted: (traces) |
| 6 | Kamambers 1 | Soft | Latvia | 0.03 | <LOQ |  |
| 7 | Kamambers 2 | Soft | Latvia | N.D. | <LOD |  |
| 8 | Trikatas Blue | Soft | Latvia | N.D. | <LOD |  |
| 9 | Breugel | Hard | Belgium | <LOD | <LOD | Deleted: (traces) |
| 10 | Brugs Broodje | Semi soft | Belgium | N.D. | <LOD |  |
| 11 | Brugse Blomme | Soft | Belgium | N.D. | <LOD |  |
| 12 | Bouquet des Moines | Soft | Belgium | N.D. | <LOD |  |
| 13 | Chevre Affine | Soft | Belgium | 1.23 | Positive |  |
| 14 | Chimay | Semi hard | Belgium | $<L^{\text {L }}$, | <LOD | Deleted: (traces) |
| 15 | Orval | Semi soft | Belgium | 0,03 | <LOQ |  |
| 16 | Oud Brugge | Semi hard | Belgium | N.D. | <LOD |  |
| 17 | Passendale | Semi soft | Belgium | N.D. | <LOD |  |
| 18 | St.Paulin | Semi soft | Belgium | $<L^{\text {L }}$ | <LOD | Deleted: (traces) |
| 19 | Nazareth | Semi hard | Belgium | N.D. | <LOD |  |
| 20 | Watou Special | Soft | Belgium | 0.52 | Positive |  |
| 21 | Gouda | Hard | Belgium | N.D. | <LOD |  |
| LOD $=0.03 \mu \mathrm{~g} \mathrm{~kg}{ }^{-1}, \mathrm{LOQ}=0.1 \mu_{\mathrm{gkg}}$-1, N.D. - not detected |  |  |  |  |  | Deleted: 27 |


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