

Hyperfine structure of the a1 g - X 3Σ g – transitions of 160170, 170180 and 1702 by CRDS at 80 K

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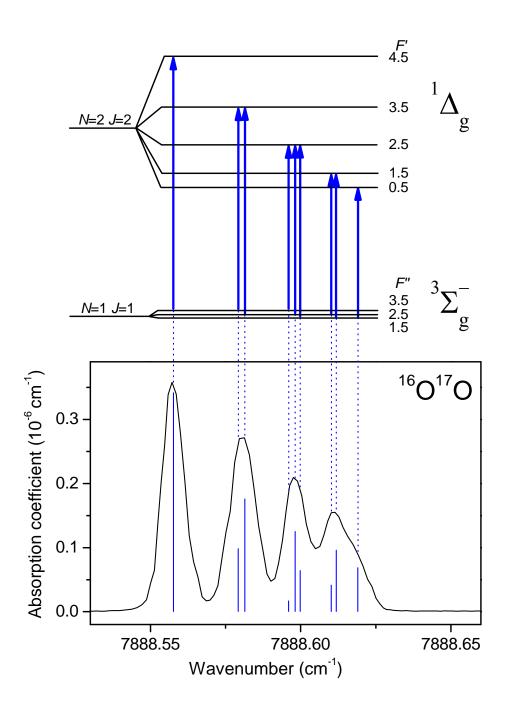
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The nuclear hyperfine structure of the transitions could be partly resolved.

The hyperfine parameter coupling in the $a^1\Delta_g$ state has been derived.



Hyperfine structure of the $a^1\Delta_g$ - $X^3\Sigma_g^-$ transitions of ${}^{16}O^{17}O$, ${}^{17}O^{18}O$ and ${}^{17}O_2$ by CRDS at 80 K

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Abstract

The high sensitivity absorption spectrum of the $a^1\Delta_g$ - $X^3\Sigma_g^-$ band of ${}^{16}O^{17}O$, ${}^{17}O^{18}O$ and ${}^{17}O_2$ has been recorded by CW-Cavity Ring Down Spectroscopy near 1.27 µm. The spectra were obtained between 7640 and 7917 cm⁻¹ with a ${}^{17}O$ -enriched sample at room temperature and at 80 K. Due to the I= 5/2 nuclear spin of the ${}^{17}O$ atom, the nuclear hyperfine structure of the transitions could be partly resolved at low temperature. The parameter coupling of the nuclear spin to the electronic angular momentum in the $a^1\Delta_g$ state has been derived for the three studied isotopologues of oxygen.

1. Introduction

Recent CRDS experiments carried out at the Grenoble University have allowed significant improvement of our knowledge of the spectral line parameters of the $a^1\Delta_g - X^3\Sigma_g^-$ band of oxygen. In Ref [1], the electric quadrupole line parameters were found to have sufficient intensities to be detected in the atmospheric spectrum and they were parameterized to complement magnetic dipole transitions provided in HITRAN database [2]. In Ref. [3], the measurements in ¹⁶O- and ¹⁸O- enriched mixtures of oxygen have allowed determination of the best to date spectroscopic constants in the $a^1\Delta_g$ band for five most abundant isotopologues of oxygen (¹⁶O₂, ¹⁶O¹⁸O, ¹⁸O₂, ¹⁶O¹⁷O and ¹⁷O¹⁸O). For the ¹⁷O-containing species significant broadening of rotational lines due to unresolved hyperfine structure was observed [3].

Indeed, unlike other isotopes of oxygen ¹⁷O has a non-zero nuclear spin (I=5/2) and the coupling of the nuclear spin to electron spin in the $X^{3}\Sigma_{g}^{-}$ state [3] and to the electronic angular momentum in the $a^{1}\Delta_{g}$ state [4] gives rise to a magnetic hyperfine (hf) structure. As a result of the Doppler broadening at room temperature, the magnetic hyperfine structure could not be resolved and it manifested itself only as a broadening of the ¹⁶O¹⁷O and ¹⁷O¹⁸O transitions.

The present study is devoted to a detailed experimental investigation of the hf structure of the ¹⁷O containing isotopologues of oxygen, on the basis of new CRDS spectra of the $a^1\Delta_g - X^3\Sigma_g^-$ band recorded at room temperature and at liquid nitrogen temperature (about 80 K, see below). By using a ¹⁷O-enriched sample, the ¹⁶O¹⁷O, ¹⁷O¹⁸O and ¹⁷O₂ isotopologues could be simultaneously studied. As illustrated below, the narrowing by a factor of two of the Doppler width at 80 K compared to room temperature was decisive to (partly) resolve and interpret the hf structure.

After the description of the CRDS spectrometer in the next section, the obtained spectra and the analysis of the hf structure will be presented in Section 3. The coupling parameters will be determined and compared to the values available in the literature [4].

2. Experimental

Room temperature measurements

The high-sensitivity CW-CRDS absorption spectrum of the $a^1\Delta_g - X \, {}^3\Sigma_g^-$ band was recorded at room temperature (RT) in the 7658-7917 cm⁻¹ region using a 17 O-enriched sample. The stated

atomic composition of the sample (from Sigma Aldrich) was 17 O: 55.8%, 16 O: 43.4% and 18 O: 0.8%.

The fibered distributed feedback (DFB) laser CW-CRDS spectrometer used for the recordings at room temperature is the same as the one used for the study of the natural and ¹⁸Oenriched samples [1, 3]. It has been described in details in Refs. [6-8]. Each DFB laser diode has a typical tuning range of about 40 cm⁻¹ by temperature tuning from -10°C to 60°C. A total of nine DFB laser diodes were needed to cover the 7658-7917 cm⁻¹ region. The electro-polished stainless steel ring down cell (l = 1.42 m, inner diameter $\Phi = 11$ mm) was fitted by a pair of super-mirrors. The reflectivity of these mirrors (about 99.997 %) corresponds to empty cell ring down times of about $\tau \sim 200 \mu$ s. About 40 ring down events were averaged for each spectral data point; the complete temperature scan of one DFB laser (15000 spectral points) required about 70 minutes. The achieved noise equivalent absorption was about $\alpha_{min} \sim 4 \times 10^{-11}$ cm⁻¹ over the whole spectrum. The pressure measured by a capacitance gauge (MKS 100 Torr full range with 0.1% accuracy) and the ring down cell temperature were monitored during the recordings. The pressure value of the RT spectra was fixed to 30.0 Torr. Additional recordings were performed at 5 Torr in the 7839-7917 cm⁻¹ region corresponding to the strongest transitions.

Measurements at low temperature

The cryogenic cell used for the CW-CRDS recordings at liquid nitrogen temperature (LNT) has been described in Refs. [9-12]. It was developed and extensively used to characterize the absorption spectrum of methane at low temperature in the 1.58 and 1.28 μ m methane transparency windows [9-12]. Briefly, it is based on an original design (see Fig. 1 of Ref. [9]) which dispenses with an external vacuum jacket by exploiting the fact that a low pressure gas sample constitutes itself a good thermal insulation. The cryostat is a 1.42 m long hollow cylinder both filled and completely surrounded by the sample gas volume. The liquid nitrogen filling tube constitutes its single suspension point which, combined with an almost perfect cylindrical geometry helps to eliminate stress normally present in a double-jacket cell configuration. Contrary to our previous setup [9], one of the mirrors was prealigned then glued on a piezo actuator allowing for cavity length modulation. Therefore a standard CW-CRDS data acquisition could be used with a spectral calibration fully relying on the wavemeter real time measurements. The distance between the high reflective mirrors and the cryostat ends is about 0.5 cm. A sensitivity comparable to that achieved at room temperature ($\alpha_{min} \sim 10^{-10}$ cm⁻¹) was obtained.

Coinciding temperature values (80 ± 2 K) were obtained from the Doppler profile of methane absorption lines [9,10] and the rotational intensity distribution of the $3v_2$ band of CH₃D [12]. This value of 80 ± 2 K is what we refer to as "liquid nitrogen temperature" (LNT). The LNT spectrum was recorded in the 7876-7893 cm⁻¹ region corresponding to the low

rotational transitions which exhibit the most resolved hyperfine structures of the ¹⁷O containing isotopologues. The pressure value was fixed to 1.5 Torr.

Fig. 1 shows the different line profiles obtained for the R1R1 transitions of the ${}^{16}O_2$, ${}^{16}O^{17}O$ and ${}^{17}O_2$ isotopologues (Note that we use $\Delta NN''\Delta JJ''$ notation where *N* and *J* are the quantum numbers relative to the rotational angular momentum and total angular momentum, respectively). While the ${}^{16}O_2$ line appears as single line with a Doppler limited profile, the ${}^{16}O^{17}O$ and ${}^{17}O_2$ lines are split in different hf components. The comparison between the 296 K and 80 K spectra (Fig.1) illustrates the much better resolution of the hf structure at 80 K due to the reduction by a factor of two of the Doppler broadening (HWHM of 0.008 and 0.004 cm⁻¹ at 296 K and 80 K, respectively).

Spectrum calibration

Each 40 cm⁻¹ wide spectrum recorded with one DFB laser was calibrated independently on the basis of the wavelength values provided by a Michelson-type wavemeter (Bristol 621A, 60 MHz resolution and 100 MHz accuracy). The calibration was further refined by stretching the whole spectrum in order to match accurate positions of the transitions of the $a^1\Delta_g - X^3\Sigma_g^-$ band of ${}^{16}O_2$ provided in the recent HITRAN update [2]. The uncertainty in the absolute values of the line positions is estimated to be 1×10^{-3} cm⁻¹. It results from an equivalent contribution of the uncertainties in the HITRAN values and in the values measured in the present study.

3. Results

Transitions due to the ¹⁷O₂, ¹⁶O¹⁷O, ¹⁶O₂, ¹⁷O¹⁸O and ¹⁶O¹⁸O isotopologues were found to contribute to the spectrum. The most abundant species are ¹⁷O₂, ¹⁶O¹⁷O and ¹⁶O₂ with relative concentration on the order of 0.33, 0.47 and 0.18, respectively. Overall, more than one thousand transitions were measured in the RT spectrum. The global rovibrational analysis including the complete line list, description of the assignment procedure and derivation of the spectroscopic parameters from a global fit will be published in a separate contribution. In the present Letter, we limit the discussion to the hf structure of the ¹⁷O₂, ¹⁶O¹⁷O and ¹⁷O¹⁸O species.

While ¹⁶O and ¹⁸O nuclei have a I=0 nuclear spin, ¹⁷O has a I=5/2 spin leading to the existence of an hyperfine structure in the 17 O containing species. The hf structure affecting the X ${}^{3}\Sigma_{g}^{-}$ ground state of ${}^{17}O_{2}$, ${}^{16}O^{17}O$ and ${}^{17}O^{18}O$ was previously reported from microwave and millimeter spectroscopy measurements [5, 13-16]. The observation of the ground state hyperfine structure by absorption spectroscopy in the near infrared or visible is particularly difficult as the Doppler width is significantly larger than the hf splitting. Very recently, Long et al. [14] could measure the ground hyperfine structure from an excess of the broadening of some transitions of the $b^{1}\Sigma_{g}^{+}$ - X ${}^{3}\Sigma_{g}^{-}$ band near 760 nm. The sub Doppler hyperfine structure could be evidenced from a detailed study of the line profile recorded by Frequency-stabilized CRDS [14]. It is important to note that the evidenced extra broadening of the $b^1\Sigma_g^+$ - $X^3\Sigma_g^-$ band transitions is due to the ground state hf structure as the $b^1\Sigma_g^+$ upper electronic state is a singlet Σ state with negligible hf structure. In the present study devoted to the $a^1\Delta_g$ - $X^{-3}\Sigma_g^-$ band, the situation is different because the nuclear spin can couple to the electronic angular momentum of the $a^1\Delta_g$ upper state. Our spectra (Fig. 1 for instance) show that the observed hf splitting is significantly larger than the Doppler broadening and then that the hf structure of the $a^1\Delta_g$ upper state is larger than that of the ground state. This is consistent with earlier results of electron paramagnetic resonance (EPR) measurements of ¹⁶O¹⁷O and ¹⁷O¹⁸O species in the $a^1\Delta_g$ state [4].

The general expression of the hyperfine structure Hamiltonian can be written as

$$\mathbf{H}_{hf} = a \mathbf{I}_{z} \mathbf{L}_{z} + b_{F} \mathbf{I} \cdot \mathbf{S} + c \left(\mathbf{I}_{z} \mathbf{S}_{z} - \frac{1}{3} \mathbf{I} \cdot \mathbf{S} \right) + \frac{eq \mathcal{Q} \left(3\mathbf{I}_{z}^{2} - \mathbf{I}^{2} \right)}{4I(2I-1)},$$
(1)

where the first term describes the interaction of the electronic orbital moment with the nuclear spin, the second and third terms represent the isotropic (Fermi contact interaction) and anisotropic parts of the electron spin-nuclear spin coupling, respectively. The forth term represents the electric quadrupole interaction due to asymmetric charge distribution around the nucleus.

In the $X^{3}\Sigma_{g}^{-}$ ground state, $\Lambda = 0$ and therefore the first term vanishes. The two other magnetic terms lead to an hf splitting of each (*N*, *J*) rotational level in different components:

$$W(F_1) = \frac{\mathbf{IJ}}{(N+1)} \{ b_F + c \frac{2N}{(2N+3)} \}$$
$$W(F_2) = \frac{\mathbf{IJ}}{N(N+1)} (b_F + \frac{2}{3}c)$$

where IJ = 1/2[F(F+1)-I(I+1)-J(J+1)] with F = J+I, J+I-1, ...|J-I| and F_1 , F_2 and F_3 : corresponds to J=N+1, J=N and J=N-1, respectively. Note that Eq. 2 ignores small contribution due to quadrupole coupling.

In the case of the ¹⁶O¹⁷O and ¹⁷O¹⁸O species, I=5/2 and then $F=J\pm5/2$, $J\pm3/2$ and $J\pm1/2$. The rotational levels are then split into six components for J>2.

In the case of the ¹⁷O₂ isotopologue, the vectorial addition of the two nuclear spins $(I_1=I_2=5/2)$ leads to six values of the total nuclear spin $I_{tot}=I_1+I_2,.../I_1-I_2/=5, 4, 3, 2, 1, 0$. In the $X^{3}\Sigma_{g}^{-}$ ground state because of the nuclear spin function symmetry, the even and odd values of N'' can only be associated with odd and even values of I_{tot} ", respectively [13]. Thus each J" level is split into $F=J+I_{tot}, J+I_{tot}-1, ...|J-I_{tot}|$ components with $I_{tot}=5, 3, 1$ for even N" values and with $I_{tot}=4, 2, 0$ for odd N" values. For instance, for N=1 the number of components is seven, while for N=2 the level splits into 13 components.

The $a^1\Delta_g$ upper state is a singlet state (*S*=0), so that only the first and fourth interaction terms remain in Eq. 1. This $a^1\Delta_g$ state obeys the Hund's electronic case (a) coupling and the dominant magnetic part of the hyperfine operator simplifies to yield the hyperfine energy [17]:

$$W = a \Lambda \Omega \mathbf{I} \mathbf{J} / J(J+1). \tag{3}$$

with $\Omega = \Lambda + \Sigma$, which in this case is equal to $\Lambda = 2$. From this expression, it is obvious that the hyperfine component separations decrease with increasing *J* values i.e. that, in the different branches, transitions from the first rotational levels are expected to show the most resolved hf structure. As an example, Figs. 2 and 3 illustrate the evolution of the hf structure with an increasing rotational excitation for ¹⁶O¹⁷O and ¹⁷O₂, respectively: the hf structure is partly resolved for the first *N*, *J* rotational levels but shows up as a small extra broadening for *N*, *J* values larger than 10. It is worth mentioning that Λ -doubling could not be resolved in this experiment.

The selection rules for ¹⁶O¹⁷O, ¹⁷O¹⁸O and ¹⁷O₂ magnetic dipole transitions are $\Delta J=0, \pm 1$ and $\Delta F = 0, \pm 1$. For the high *J* values, only the hyperfine transitions with $\Delta J = \Delta F$ are significant and the intensity of these transitions are proportional to (2*F*+1) [18]. Fig. 4 shows the energy diagram and allowed transitions of the R1R1 line together with the corresponding experimental spectra of ¹⁶O¹⁷O and ¹⁷O¹⁸O. The hf coupling terms being identical for these two species, the observed hf structures are very similar.

The values of the hf coupling parameters were determined from a global fit of all the transitions measured in this work together with the microwave (MW) lines from Cazzoli et al. [5, 13] by using the SPFIT software [19]. The expression of the rotational part of the Hamiltonians in the ground and excited electronic states is given in Ref. [3] (Eqs. 1 and 2). A detailed description of the fit including the weighting of the wavenumbers values of the hf components, the combination of RT and LNT datasets, the constraints of the fits of the spectroscopic parameters and comparisons with previously reported values will be detailed in a separate contribution [20]. In this Letter, we focus on the obtained hf coupling constants which are listed in Table 1 for both the ground and $a^1\Delta_g$ (last term of Eq. 1) but it could not be well determined. The corresponding constant in the ground state of ${}^{16}O{}^{17}O$ and ${}^{17}O{}^{18}O$ is about -8 MHz and according to the calculations of Minaev and Minaeva [21] it should be similar in the excited state and then too small to be determined from our spectra.

To the best of our knowledge, the only previous determination of the hf coupling constant in the $a^1\Delta$ state is due to Arrington et al. [4] who obtained the *a* value of ${}^{16}O{}^{17}O$ and ${}^{17}O{}^{18}O$ from a fit of EPR spectra [4]. Since in theory there is no isotopic dependence for this parameter, they performed a global fit for the two species and determined the coupling constant to be -424±1 MHz which is almost exactly a factor of two greater than our values (-215.6±2.3 and -212.4±3.5 MHz, for ${}^{16}O{}^{17}O$ and ${}^{17}O{}^{18}O$, respectively). This difference is merely due to the different definition of the coupling constant adopted by Arrington et al. [4], the value of Λ (Λ =2 in this case) being absorbed into the constant. The hf coupling parameter of ${}^{17}O_2$ could be also determined from the analysis of our LNT spectrum. As expected, the obtained value is very close to that of the ${}^{16}O{}^{17}O$ and ${}^{17}O{}^{18}O$ isotopologues.

We have superimposed to the spectra of Figs. 2 and 3 a simulation of the hf structure calculated with the fitted parameters values. The simulated spectra were obtained by affecting a Doppler limited line profile (at 80 K) to the stick spectrum provided by the SPFIT program. It leads to a very convincing agreement between the measured and calculated spectra.

The combination of the high sensitivity CRDS technique with a cryogenic cell at 80 K has allowed for the first observation of the partly resolved hyperfine structure of the $a^1\Delta_g$ - $X^3\Sigma_g^-$ band of the three ¹⁷O containing isotopologues of oxygen - ¹⁶O¹⁷O, ¹⁷O¹⁸O and ¹⁷O₂. The observed structure results from the hf splittings of both the $X^3\Sigma_g^-$ ground state and the $a^1\Delta_g$ upper state which have a different origin: the hf splitting of the $X^3\Sigma_g^-$ levels is due to the coupling of the ¹⁷O nuclear spin to the electronic spin while that of the $a^1\Delta_g$ upper states is larger and results from the coupling of the ¹⁷O nuclear spin to the electronic orbital angular momentum. The hf coupling parameters in the excited state have been determined from a global fit of the line positions. The simulation of the hf structure calculated using these parameters leads to a very satisfactory reproduction of the measured hf structure.

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Table 1.

The hf coupling parameters of the $X^{3}\Sigma_{g}^{-}$ and $a^{1}\Delta_{g}$ states of ${}^{16}O{}^{17}O$, ${}^{17}O{}^{18}O$ and ${}^{17}O_{2}$.

	Parameter (MHz)	¹⁶ O ¹⁷ O	¹⁷ O ¹⁸ O	¹⁷ O ₂
$a^1\Delta_{ m g}$	a (This work)	-217.33(195)	-211.50(261)	-211.93(126)
	a [4]	$-424(1)^{a}$	$-424(1)^{a}$	
$^{3}\Sigma_{g}^{-}$	b_F	-54.7444(92)	-54.745(21)	-54.758(23)
	С	140.079(24)	140.090(59)	140.0378(62)
	eQq	-8.13(24)	-8.23(47)	-8.307(40)

a) In Ref [4], there is a factor of 2 in the *a* parameter due to a different definition.

Figure Captions

Fig.1

The R1R1 transition of ${}^{16}O_2$, ${}^{16}O^{17}O$ and ${}^{17}O_2$ at 296 K and 80 K. Spectra recorded at the room (P=5.0 Torr) and liquid nitrogen temperature (P=1.5 Torr) are shown on the upper and lower panel, respectively.

Fig. 2

The hyperfine structure of different ${}^{16}O^{17}O$ transitions (T=80 K, P=1.5 Torr). The wavenumber values (v₀) corresponding to the centre of the spectral range of each panel are the following: R1R1 - 7888.5906, R2R2 - 7891.2606, R1Q2 - 7890.4704, Q6Q6 - 7882.3106 and Q13Q13 - 7879.6066 cm⁻¹.

The black solid line corresponds to the experimental spectrum while the red dash line is a simulation based on the calculated stick spectrum (in blue).

Fig. 3

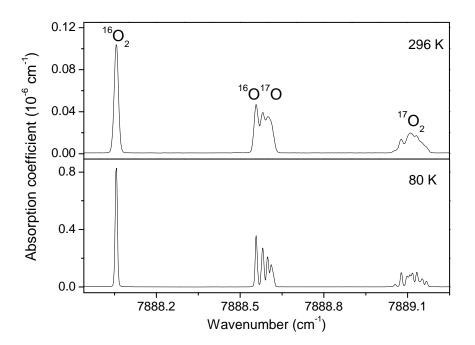
The hyperfine structure of different ${}^{17}O_2$ transitions (T=80 K, P=1.5 Torr). The wavenumber values (v_0) corresponding to the centre of the spectral range for each panel are the following: R1R1 – 7889.1152, R2R2 7891.7151, Q8R7 - 7884.4304 and Q13Q13 - 7880.4153 cm⁻¹. On the two upper panels the features at the left hand side are ${}^{16}O^{18}O$ transitions.

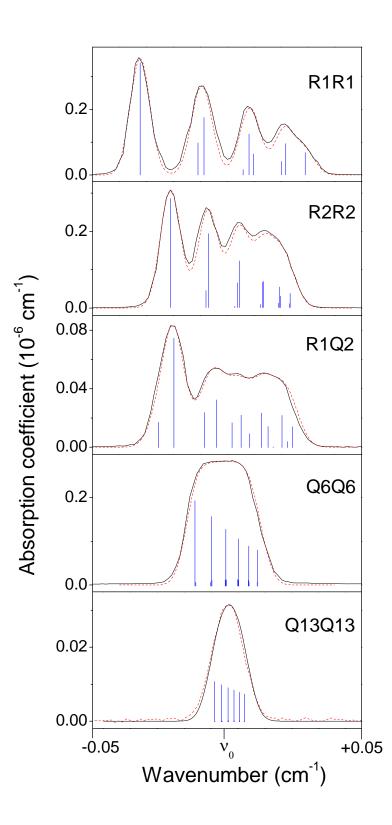
The black solid line corresponds to the experimental spectrum while the red dash line is a simulation based on the calculated stick spectrum (in blue).

Fig. 4

Energy diagram and spectrum of the hyperfine structure of the R1R1 transition of ${}^{16}O^{17}O$ and ${}^{17}O^{18}O$ in the spectrum recorded at 80 K. Note that the Λ -doubling is unresolved in the present experiment. The stick spectrum corresponds to the calculated hf structure.

Figure 1





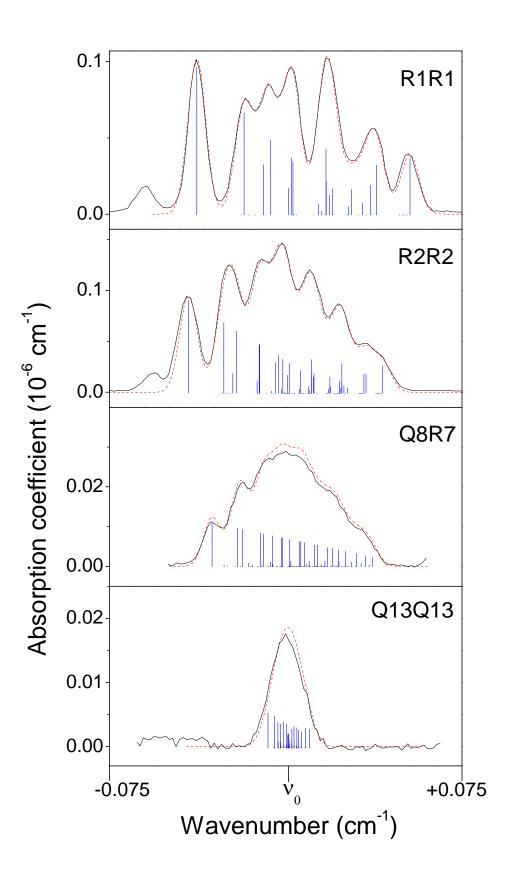


Figure 4

