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Microbial responses to mustard gas dumped in the Baltic Sea
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#### 14 Abstract

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Microbiological studies were carried out on chemical weapon dump sites in the 16 Baltic Sea. The effect of mustard gas hydrolysis products (MGHPs) on marine 17 microbiota and the ability of microorganisms to degrade MGHPs were studied. Many 18 stations at the dump sites demonstrated reduced microbial diversity, and increased 19 growth of species able to use mustard gas hydrolysis products as sole source of 20 carbon. Significant amounts of MGHP-degrading bacteria were revealed in the near-21 bottom water. The MGHP-degrading microorganisms identified as Achromobacter 22 sp., *Pseudomonas* sp., and *Arthrobacter* sp. were isolated. These microorganisms 23 were capable of utilizing the major product of hydrolysis, thiodiglycol, as the sole 24 source of carbon and energy. The bacteria were capable of metabolizing MGHPs at a 25 low temperature. The metabolic pathway for thiodiglycol degradation was proposed. 26 27 The results suggest the potential for MGHPs biodegradation by naturally occurring populations of near-bottom-water and sediment microorganisms. 28

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*Keywords:* Bacteria, Baltic Sea, Chemical weapon, Sea dumping, Mustard gas,
 Degradation, Self purification.

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Abbreviations: CWA, chemical warfare agents; MGHPs, mustard gas hydrolysis
 products; TDG, thiodiglycol; OCS, organochlorine substances; TDGA, thiodiglycolic
 acid; TGA, thioglycolic acid; HETA, [(2-hydroxyethyl)thio]acetic acid.

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#### 36 **1. Introduction**

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Tens of thousands of tons of captured German chemical munitions have been dumped in the Baltic Sea since World War II. It is estimated that the chemical munitions contained about 13,000 tons of chemical warfare agents (HELCOM, 1994). The dumping of munitions at sea by the allied countries was done in the southern and western Baltic, mainly in the Gotland Basin, the Bornholm Basin, the Little Belt and the Skagerrak Strait (Fig. 1).

In most cases the warfare agents were thrown overboard, either loose (bombs, 44 shells) or in containers, but some ships were also sunk (HELCOM, 1996). During 45 transportation to the dumping areas some munitions were thrown overboard while the 46 ships were en route. Warfare agents are, therefore, assumed to be spread over a 47 considerable area along the transport routes (Schulz-Ohlberg et al., 2001). Chemical 48 munitions have been spread even further by fishing vessels when trawling. Table 1 49 gives an overview of the quantities of chemical munitions and warfare agents dumped 50 in the Baltic Sea and Skagerrak (Chemical munitions, 1993). 51

The largest number of chemical agents was dumped in the Bornholm Basin, east of the island of Bornholm. Fishermen still recover bombs, shells and lumps of mustard gas in their nets in this area (HELCOM, 1993).

Over the period 2000-2005 up to 1,010 kg of chemical munitions a year were netted by fishermen east of Bornholm (HELCOM, 2005). The netted munitions were completely corroded and resembled clay-like lumps of mustard gas. The highest number of incidents during the last 20 years was in 1991 with 103 catches amounting to 5,378 kg of chemical warfare agents. The most recent report of recovery of bomb containing mustard gas and tear gas Clark I came from Sweden in 2007 (Åstot et al., 2007).

Most accidents with warfare agents so far have involved mustard gas. Because of the admixture of thickeners, viscous mustard gas is the only warfare agent occurring in large lumps that are mechanically relatively stable and can exist unchanged for decades on the sea bed. All injuries to fishermen have occurred when liquid or solid mustard gas has escaped from rusted munitions and come into contact with the skin. The mustard gas causes severe blistering of the skin and irritates the eyes and throat (Glasby, 1997).

Mustard gas is extremely toxic for all species (Opresko et al., 1998). Because of 69 its high lipid solubility mustard gas rapidly penetrates cells. It is able to alkylate 70 DNA, RNA, and proteins affecting a variety of cell functions (Papirmeister et al., 71 1991). The products of mustard gas hydrolysis have a wide spectrum of toxic activity 72 on soil microbiota (Medvedeva et al., 2008). They are known to change significantly 73 the number and the specific composition of soil microorganisms (Medvedeva et al., 74 2006). Contamination of soil with mustard gas causes significant ecological shifts, 75 such as a decrease in the similarity index and the diversity index of microbiota, and a 76 decrease in the dominance index. Microbial succession is accompanied by a decrease 77

<sup>78</sup> in the fermentative activity of soils (Medvedeva et al., 2000).

To assess the risk relating to dumped chemical weapons, it is necessary to 79 evaluate the ability of the ecosystem itself (and microbiota as its component) for self-80 purification of contaminated water and sediments. The ability of microorganisms to 81 carry out mineralization of organic substances, in particular, toxicants of various 82 chemical natures is well known and widely used in practice and scientific research 83 (Oldenhuis et al., 1992). Because of its high degree of toxicity, mustard gas is 84 generally resistant to microbial attack. However, several studies have demonstrated 85 that mustard gas is subject to biodegradation by a limited number of naturally 86 occurring species. 87

Biodegradation of mustard gas by basidiomycetes, *Coriolus versicolor* and *Tyromyces palustris*, has been reported (Wariishi et al., 2002). Bacteria *Micrococcus sp*. 6-2 and *Pseudomonas sp*. 8-2 were able to metabolize the products of mustard gas hydrolysis, particularly those that were chlorinated (Medvedeva et al., 2000). Bacterium *Rhodococcus rhodochrous* IQTS8 had the ability to utilize a chemical analog of mustard gas 2-chloroethyl-ethylsulfide as the only carbon and energy source for growth (Kilbane & Jackowski, 1996).

More studies are concerned with thiodiglycol (TDG), the major product of 95 mustard gas hydrolysis (Munro et al., 1999). Thiodiglycol is a fairly stable substance 96 with weak toxicity. Nevertheless, it is one of the war gas precursors that must be 97 destroyed, according to the Chemical Weapons Convention. In this connection, 98 biodegradation of mustard gas is closely related to the utilization of TDG by 99 microorganisms. Soil bacteria Pseudomonas pikettii SH18 and Alcaligenes 100 xylosoxydans ssp. xylosoxydans SH42 (Harvey et al., 1993), Micrococcus sp. 6-2 and 101 Pseudomonas sp. 8-2 (Medvedeva et al., 2000), Pseudomonas sp. Y-13 (Medvedeva 102 et al., 2006), Alcaligenes xylosoxydans PGH 10 (Garcia-Ruiz et al., 2002), 103 Alcaligenes xylosoxydans subsp. denitrificans TD 1 (Ermakova et al., 2002), 104 Alcaligenes xylosoxydans ssp. xylosoxydans SH91 (Lee et al., 1996) are capable of 105 using TDG as the sole source of carbon. While degradation of thiodiglycol has been 106 characterized in terrestrial isolates, little is known about marine-degrading 107 microorganisms. 108

The goal of our study was to investigate the effect of warfare agents on the Baltic Sea microbiota. We report in this paper on the isolation, characterization and metabolic activities of the bacteria able to use mustard gas hydrolysis products as the sole source of carbon and energy.

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115 **2. Materials and methods** 

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117 2.1. Sampling

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119 Seawater samples were collected from Baltic Sea chemical weapon dump sites 120 in 1994-2007. The water samples were taken from the Bornholm dump site, Gotland 121 Basin, and Skagerrak Strait. Water samples (100 ml) were collected from the whole 122 water column using sterile bathometers. Near-bottom water (100 ml) was taken from 123 Niemistö corers and put into sterile plastic flasks.

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#### 125 2.2. Enumeration of microorganisms

The number of heterotrophic microorganisms and content of heterotrophic 127 microorganisms able to degrade mustard gas hydrolysis products (MGHPs) were 128 analyzed immediately after sampling. The total number of microorganisms was 129 determined after conservation with formalin using the standard technique (Tsyban, 130 1980). The number of heterotrophic microorganisms was estimated using inoculation 131 on nutrient agar (1:10) (Kuznetsov & Dubinina, 1989). The incubation time was 3-5 132 days. The content of heterotrophic microorganisms able to degrade MGHPs was 133 determined by inoculation on selective agar with water samples. The incubation time 134 was 7 - 10 days. The selective agar had the following composition:  $(NH_4)_2SO_4$  4.0 g/l, 135 KH<sub>2</sub>PO<sub>4</sub> 1.5 g/l, K<sub>2</sub>HPO<sub>4</sub> 1.5 g/l, MgSO<sub>4</sub> 0.2 g/l, NaCl 5.0 g/l and agar 20.0 g/l. 136 Mustard gas hydrolysis products (16 mg/l of organochlorine substances (OCS) and 137 0.35 mg/l of thiodiglycol) were added to this selective medium as the sole source of 138 carbon. 139

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#### 141 2.3. Isolation of microorganisms, media, and culture conditions

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Samples from the flasks were streaked onto plates containing mineral salts medium (4.0 g/l (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>, 1.5 g/l KH<sub>2</sub>PO<sub>4</sub>, 1.5 g/l K<sub>2</sub>HPO<sub>4</sub>, 0.2 g/l MgSO<sub>4</sub>, 5.0 g/l NaCl) solidified with 1.5% agar. Mustard gas hydrolysis products, including 25 mg/l of organochlorine substances (OCS), were used as the sole source of carbon. The plates were incubated at 20 °C for up to 3 weeks. A rapidly growing, visually distinct colony and a separate, morphologically unique isolate were selected for further analysis and purified by repeated plating.

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The selected strains of microorganisms were cultivated by three methods:

(1) By inoculation into a solid medium with the following composition (g/l): (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> 4.0, KH<sub>2</sub>PO<sub>4</sub> 1.5, K<sub>2</sub>HPO<sub>4</sub> 1.5, MgSO<sub>4</sub> 0.2, NaCl 5.0, thiodiglycol 2.0, CaCO<sub>3</sub> 2.5, agar 20.0. The control parameters included the occurrence of growth and dissolution of CaCO<sub>3</sub>, which gives evidence of the formation of acidic products of thiodiglycol degradation;

156 (2) Cultivation under static conditions: in 20-ml test tubes filled with 17 ml of 157 seawater containing MGHPs (32 mg OCS/l and 0.96 g/l TDG), pH 7.8  $\pm$  0.2. The 158 MGHPs solutions, pre-sterilized separately by filtering through membrane filters with 159 a pore diameter of 0.22 µm, were aseptically introduced into the medium, and 160 simultaneously the test tubes were inoculated with the bacterial culture under

161 scrutiny.

(3) Cultivation under aeration: in 250-ml Erlenmeyer flasks on a rotary shaker CERTOMAT BS-1 (B.Braun Biotech Int. GmbH, Germany) at 240 rpm on a medium consisting of (g/l):  $(NH_4)_2SO_4$ , 6.0;  $KH_2PO_4$ , 5.0;  $K_2HPO_4$ , 5.0;  $MgSO_4$ , 0.2; and CaCO<sub>3</sub>, 10.0, containing MGHPs (20 - 750 mg /l OCS; 6 – 22.5 g/l TDG). Control parameters included the biomass growth, the change in pH of the culture broth, the presence of TDG, and the content of organochlorine substances.

The biomass growth was routinely monitored by measuring  $D_{540}$  with a Genesys 105 spectrophotometer (Thermo Electron Corp., USA).

170 Considering that chemical hydrolysis of organochlorine substances can proceed 171 in water medium, the studies of microbial degradation of MGHPs were carried out in 172 comparison with the control probe (with the process performed under the same 173 conditions but without microorganisms).

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175 2.4. DNA extraction from bacterial cultures

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DNA was extracted from the culture samples using incubation with 0.75 ml of 177 1% (w/v) potassium ethyl xanthogenate containing 1% SDS (w/v), 20 mM EDTA, 178 100 mM TRIS and 800 mM CH<sub>3</sub>COONH<sub>4</sub>, pH 7.4 (70 ℃, 1 h), followed by vortexing 179 (10 s), a second incubation (4 %, 30 min) and centrifuging (17 530 x g, 4 %, 10 min). 180 DNA was precipitated using 0.75 ml of 2-propanol (22-24 °C, 10 min), pelleted (12 181 000 x g, 22-24 °C, 10 min), washed twice with 200  $\mu$  of 70% (v/v) ethanol, air dried 182 (20-24 ℃) and dissolved in 50-200 µ of 10 mM TRIS, 1 mM EDTA, pH 8.0 (TE 183 buffer). The DNA isolates were further purified using Montage cartridges (Millipore, 184 Billerica, CA, USA) and stored at -20°C until amplification with polymerase chain 185 reaction (PCR). 186

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188 2.5. DNA amplification189

The primers were from Oligomer (Helsinki, Finland). Partial 16S rRNA genes 190 were amplified from the culture samples using a forward primer 27F (5'-191 AGAGTTTGATCCTGGCTCAG-3'; general bacterial primer; Lane 1991) and a 192 reverse primer 1525R (5'-AAGGAGGTGATCCAGCC-3'). PCR was performed in 20 193 μl scale with 1 μl of template DNA, 0.4 μM of forward and reverse primers, 200 μM 194 of each deoxynucleoside triphosphate (dNTP, Sigma-Aldrich), 1 U of MasterTaq 195 polymerase (Eppendorf, Hamburg, Germany), 1 x MasterTaq polymerase buffer 196 containing 1.5 mM Mg(OAc)<sub>2</sub>, 0.5 x Taq Master Enhancer (Eppendorf), and 3 g l<sup>-1</sup> 197 BSA. A Thermo PCR Sprint (Milford, MA, USA) thermal cycler was used. The 198 initial step in the temperature programme was 70 °C (5 min) and 94 °C (5 min). DNA 199 polymerase was added during the initial 70 °C period (hot start). The following 200 denaturing, annealing and extension programmes were used: 9 cycles of [94 °C (45 s), 201

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<sup>202</sup> 57 °C (45 s), 68 °C (2 min)], 24 cycles of [94 °C (45 s), 54 °C (45 s), 68 °C (2 min)] <sup>203</sup> and 68 °C (7 min). The formation of PCR products (approx. 1,500 bp) was checked <sup>204</sup> using agarose gel electrophoresis. The PCR products were stored at −20 °C. Products <sup>205</sup> used in cloning were stored for a maximum of 24 h.

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2.6. Cloning and sequencing

The cloning of amplified DNA was performed with a Qiagen PCR cloning kit 209 (pDrive plasmid vector) at the Sequencing Laboratory of the Institute of 210 Biotechnology, University of Helsinki. After a 24-h culturing bacterial pellets were 211 lysed and plasmid DNA purified using 2-propanol precipitation and centrifuging, and 212 stored in TE (-20 °C). The plasmids were further amplified with primers UP and RP 213 (5'-TAACGCCAGGGTTTTCCCAGTCACGA-3' and 5'-214 GTGAGCGGATAACAATTTCACACAGG-3'), amplifying the insert from the 215 plasmid. The products were isolated, subjected to sequencing reactions (two 216 directions; primers t7 and sp6; 5'-TAATACGACTCACTATAGGG-3' and 5'-217 ATTTAGGTGACACTATAGAA-3') and sequenced (ABI3730 XL 48-capillary 218 automated sequencer). Sequences without overlapping or unclear bases were 219 subjected to preliminary analysis using the Basic Local Alignment Search Tool 220 (BLAST) programme (National Center for Biotechnology Information, Bethesda, 221 MD, USA). 222

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#### 224 2.7. *Chemicals*

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A model mixture of mustard gas hydrolysis products was prepared by heating an aqueous mixture of 0.65 M TDG (ICN 103039 RT, 98% purity) and 0.65 M HCl at 90 °C for 8 h. The resulting MGHPs mixture contained 2 500 ± 300 mg/l OCS and (0.57 ± 0.01) M TDG. Thiodiglycolic acid (TDGA), thioglycolic acid (TGA) and βmercaptoethanol were obtained from Merck KgaA, Germany. All the chemicals were of the highest purity commercially available. [(2-hydroxyethyl)thio]acetic acid (HETA) was synthesized and purified according to Black et al. (1993).

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2.8. Analytical methods

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The quantitative determination of OCS was performed spectrophotometrically 236 (Franke, 1967). Thiodiglycol, thiodiglycolic acid, [(2-hydroxyethyl)thio]acetic acid 237 and thiodiglycol sulphoxide content were determined by reversed-phase HPLC on an 238 1090 Hewlett-Packard chromatograph under the following HP conditions: 239 temperature 25 °C, Luna (Phenomenex) column; flow rate 1.5 ml/min; detection at 240 215 nm; eluent 5% acetonitrile and 0.1% TFA (trifluoroacetic acid); sample volume 241 20 µl. 242

Hydrolysis of MGHPs was also determined by measuring the release of chloride ions with a model 96-17 B combination electrode (Orion Research Inc., USA). The electrode was calibrated by adding known concentrations of sodium chloride standards to the mineral salts medium used in the experiments. Uninoculated flasks were run in each experiment as controls.

Thioglycolic acid was detected by using a Hewlett-Packard 1090 HPLC liquid chromatograph with a diode array detector (USA) equipped with a Vydac column (USA) 218TP54 (RP-18, 4.6×250 mm, particle size 5  $\mu$ m) under isocratic conditions with 20 mM tetrabutyl ammonium (TBA) buffered with 20 mM Na phosphate (pH=6.0), containing 15% (v/v) acetonitrile at 50 °C. The flow rate was 1.5 ml/min and detection of the compounds was at 210 nm, 250 nm, and 265 nm. The injection volume was 10  $\mu$ l and the analysis lasted 10 min.

Detection of β-mercaptoethanol after its modification by Elman (1959) was carried out on a Hewlett-Packard 1090 HPLC liquid chromatograph with UV Monitor Model 1740 BIO-RAD (USA) using Waters C-18 column, ( $\Delta$ -Pack, 3.9×150 mm, particle size 5 µm) gradient mode (1.5 ml/min). The eluents were: A, 0.1% TFA and B, 100% acetonitrile. For the analysis, the following linear gradient was used: from 20% B (0 min), 37% B (8.5 min) at 20 °C. The total analysis time was 12 min. The compounds were detected at 313 nm and 338 nm.

- 262
- 263 2.9. Statistical analyses
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All statistical analyses were performed with Statistica software (version 6; Statsoft). The differences were considered significant at P<0.05.

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#### 269 **3. Results and discussion**

- 271 3.1. Effect of MGHPs on marine microbiota
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Over the last few decades microbiological studies have been carried out at the 273 chemical munitions dump sites in the Baltic Sea. In 1994-2000, we monitored the 274 ecological situation at the dump sites under the Marine Ecological Patrol Program. 275 The analysis of the near-bottom water samples taken while we were sailing in the 276 Skagerrak Strait, as well as near Liepaja and at the Bornholm dump site yielded 0.5 • 277  $10^6$  to 7.1 •  $10^6$  cells/ml for the total amount of microbiota, 0.4 to 6.5 mg/l for 278 bacterial biomass, and  $0.2 \cdot 10^4 - 4.9 \cdot 10^4$  cells/ml for the amount of heterotrophic 279 microorganisms (Table 2). 280

At many stations significant amounts of bacterial cultures, able to grow in the medium containing mustard gas hydrolysis products (both chlorinated and nonchlorinated), were observed in the near-bottom water. However, there was no significant growth of these microorganisms following inoculation of water samples to

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the same agar medium without MGHPs. It was therefore concluded that the source of 285 carbon for these bacteria were MGHPs. As Table 2 shows, at some stations they 286 account for 20-98% of the total heterotrophic microbiota. No MGHP-degrading 287 microorganisms were found in surface water layers and horizons 1 m above the sea 288 bed. The formation of an enrichment culture of microorganisms at some of the 289 observation stations can be due to the appearance in the environment of a new 290 compound, mustard gas, acting as a selective agent. The largest number of 291 sites was observed in the Bornholm Basin. MGHP-degrading "abnormal" 292 microorganisms appeared at about 40% of stations in this area (Medvedeva et al., 293 1998). 294

In 2007, the microbiological studies were carried out in the Bornholm dump site 295 area in the framework of the EC-FP6 project "MERCW" (Modelling of 296 Environmental Risks Related to Sea-dumped Chemical Weapons). The studies 297 focused on the primary dumping area located in the southern part of the Bornholm 298 Basin, east of the island of Bornholm, in water depths ranging from 70 to over 95 m 299 (Fig. 2). Four shipwrecks were identified in this area in 1994-2000 (Paka and 300 Spiridonov, 2001). The wrecks were extensively damaged. One of the wrecks was 301 covered by trawling nets, which were suspended in the water column above the 302 wrecks by intact floats (Fig. 3A). Partly corroded artillery missiles were observed on 303 the deck of the other wreck (Fig. 3B). The microbiological survey was undertaken in 304 two research areas (Fig. 2): the first research area was in the vicinity of the 305 shipwrecks and the second in the zone marked by the detection of chemical warfare 306 agents on previous research trips (Paka, 2004). 307

Near-bottom water contained MGHP-degrading microorganisms that were up to 58% of the total number of heterotrophic microorganisms at more than a half of the stations (Fig. 4). Most of such "abnormal" points with a high concentration of MGHPdegrading microorganisms were detected in the first research area near the sunken ships. 7 "abnormal" points were found in the second research area, where CWA had been detected previously.

The "abnormal" sites of mustard gas possibly leaking into the environment differed from other stations in the species diversity of microorganisms (Table 3).

The microbiota of the near-bottom water out of the dump site (30 km) was represented by microorganisms predominantly of the genera *Pseudomonas, Bacillus, Sphingomonas, Flavobacterium, Micrococcus, Alcaligenes, Achromobacter*, with *Pseudomonas* and *Bacillus* genera dominating (the latter accounted for up to 25% of the total number of heterotrophs).

The primary dump site showed a reduced diversity of bacteria. At many stations the dominance of a minimal number of *Pseudomonas*, *Bacillus*, *Arthrobacter*, and *Achromobacter* species in near-bottom water was observed. The species diversity of the microbial population at the dump site was reduced compared with other sites, most probably because of an increase in the number of MGHP-degrading

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microorganisms. The reduction in the species diversity of bacterioplankton suggests a
 decrease in the stability of the biosystem as a whole.

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#### 3.2. Isolation and characterization of MGHP-degrading microorganisms

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During the period 1994-2007, over 200 MGHP-degrading bacteria were obtained from the near-bottom water samples taken at the Gotland Basin, Skagerrak Strait, and Bornholm dump site. Three of the most active strains capable of utilizing mustard gas hydrolysis products, 75-1, 23-3 and 93-2, were obtained from the Bornholm dump site.

All three isolated bacterial strains were aerobic, oxidase and catalase-positive, – two were gram-negative, and one was gram-positive. The gram-positive strain 23-3, established by phenotypic assays, underwent a rod-coccus morphological change. Strain 75-1 was the gram-negative, nonmotile rod of ~ 0.5 by 1.5 to 2.0  $\mu$ m. Strain 93-2 was the gram-negative, rod-shaped motile organism about 1.0 by 1.5  $\mu$ m. Motility was imparted by a single polar flagellum. All three strains grew between 0 °C and 30 °C, with the optimal growth temperature around 25 °C.

The 16S rRNA gene sequences (420-470 bp) showed the greatest similarities (99%) with strain 75-1 and were related to those of the *Achromobacter* genera. Strain 23-3 was most closely related to bacteria from the genus *Arthrobacter* (with sequence similarities ranging from 98 to 99%). The results indicated that strain 93-2 was phylogenetically closest to *Pseudomonas*.

The genera previously implicated in the degradation of mustard gas hydrolysis 348 products include the genera Alcaligenes (Lee et al., 1996, Mulbry et al., 1998), 349 Pseudomonas (Harvey et al., 1993; Medvedeva et al., 2007), and Micrococcus 350 (Medvedeva et al., 2000). These bacteria have been isolated from contaminated 351 terrestrial environments or from soil enrichment cultures by using mustard gas 352 hydrolysis products. The present results indicate there are also marine MGHP-353 degrading bacteria of genera Arthrobacter, Achromobacter, and Pseudomonas. 354 Although Achromobacter (also known as Alcaligenes) and Pseudomonas strains have 355 been characterized as microorganisms capable of metabolizing the products of 356 mustard gas hydrolysis, MGHP-degrading Arthrobacter has been found for the first 357 time. 358

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#### *360 3.3. MGHP-degradation range*

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Comparative cultivation of degrader microorganisms was carried out (Table 4). The highest MGHPs-dechlorinating capability was exhibited by *Achromobacter sp.* 75-1. The maximum dechlorination rates were 26 mg OCS/g biomass/day under static conditions, and under aeration, 37 mg OCS/g biomass/day. All strains had shown lower dechlorination rates under static conditions than under aeration. The process of

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dechlorination was accompanied by evolution of the stoichiometric amount of mineralchlorine in the medium.

The degrader microorganisms also grew more actively under aeration. For example, the biomass gain for *Achromobacter sp.* 75-1 under aeration was 8 times that under static conditions. Figure 5 shows the dynamics of growth of *Achromobacter sp.* 75-1 in a medium supplemented with MGHPs as the sole carbon and energy source. The growth of microorganisms correlates with the decrease in thiodiglycol content. The complete utilization of MGHPs (60 mg OCS /l and 0.18% TDG) is observed after 6 days.

Figure 6 shows the effect of different concentrations of mustard gas hydrolysis products on the bacterial growth under aeration. The growth of strain 75-1 (Fig. 6a) was proportional to the concentration of OCS up to 360 mg/l in the medium. However, concentrations above 360 mg/l induced a lag phase in bacterial growth. This effect was more pronounced at OCS concentrations above 500 mg/l. Concentrations equal to or greater than 750 mg/l totally suppressed the growth.

Increasing concentrations of OCS adversely affected the biomass growth of both 382 strain 93-2 (Fig. 6b) and strain 23-3 (Fig. 6c). The concentration of OCS also affected 383 the lag phase before the beginning of TDG degradation by Arthrobacter sp. 23-3 and 384 Pseudomonas sp. 93-2. The higher the concentration of OCS, the longer the lag phase 385 lasted before the start of degradation (data not shown). The lag period before TDG 386 degradation at high OCS concentrations was shown to be a lag phase in cell growth. 387 Concentrations of OCS equal to or greater than 240 mg/l totally suppressed the 388 growth of Arthrobacter sp. 23-3 and Pseudomonas sp. 93-2. 389

The highest specific growth rate  $(\mu_{max})$  of 0.89 day<sup>-1</sup> was at 360 mg of OCS/l for strain *Achromobacter sp.* 75-1 (Fig. 7).

The highest growth rates for strains *Arthrobacter sp.* 23-3 and *Pseudomonas sp.* 93-2 were 0.76 - 0.73 day<sup>-1</sup> at 120 and 60 mg of OCS/l, respectively. The growth rates decreased in proportion to an increasing OCS concentration, and the lowest growth rate of  $0.1 \text{ h}^{-1}$  was observed at 730 mg of OCS/l (strain 75-1) and 260 mg of OCS/l (strains 93-2 and 23-3). The growth of strains 75-1, 93-2 and 23-3 is characterized by the peak of the maximum possible growth rate. Such growth is typical of the growth of microorganisms on a toxic substrate (Karasevitch, 1982).

The concentration of MGHPs produces a considerable effect on the process of the dechlorination of OCS, as well as on the growth of bacteria. The ultimate concentration of OCS (360 mg per litre) for the growth of *Achromobacter sp.* 75-1 corresponds to the highest rate of dechlorination. The increase in the concentration of OCS above this level leads to the sharp decrease in the rate of dechlorination as well as in the specific growth rate.

The growth rate of *Arthrobacter sp.* 23-3 and *Pseudomonas sp.* 93-2 reaches its maximum value at an OCS concentration of 60 - 120 mg/l, but the most rapid dechlorination takes place at higher concentrations of the toxicant (240 mg/l). At this concentration the growth of bacteria is practically absent, and the whole substrate is

409 consumed for energy of maintenance. The low maximum growth rate, long generation
410 time and considerable outlays for the maintenance of both cultures are characteristic
411 of the growth of microorganisms on toxic substrates (Pirt, 1975).

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#### 3.4. Effect of environmental conditions on biodegradation

The pH range for good growth of degrader bacteria under aeration is 7.0 to 8.0 (Fig. 8). There was little difference in growth in cultures with an initial pH between 6.0 and 8.0 (averaging about 20%). The growth of *Achromobacter sp.* 75-11 and *Arthrobacter sp.* 23-3 was slightly more rapid at a neutral pH. There was a gradual decrease in the growth rate of *Arthrobacter sp.* as the initial culture pH was increased from 8 to 9. *Pseudomonas sp.* 93-2 grew better in a slightly alkaline rather than acidic condition. The maximum growth rate of strain 93-2 was observed at pH 8.

We found that the fastest growth of degrader-microorganisms occurred at 20 and 422 25 °C (Table 5). The growth rate at 20°C ( $\mu_{max} = 0.23 - 0.31$  g/g/d) was about 3-5 423 times more rapid than the growth rate at 5 °C. Cultures also took longer to degrade 424 TDG at 5 °C (Fig. 9, Fig. 10, Fig. 11). The time required for thiodiglycol degradation 425 was 30-45 days at 20 and 25 °C, but 180-210 days were required to degrade TDG at 5 426 °C. The temperature had little effect on the biomass yield (Y) of the degrader-427 microorganisms (Table 5). However, the data indicated that the optimum temperature 428 for biomass yield was 20-25 °C. 429

The ability of microorganisms to degrade thiodiglycol at a low temperature is
 particularly important in cold Baltic Sea conditions.

- 432
- 433 *3.5. Metabolic pathways*
- 434

The metabolic pathway for the degradation of MGHPs by marine bacteria was studied. The intermediate products from the growth of strains 93-2, 23-3, and 75-1 in the medium containing thiodiglycol as a source of carbon were analyzed to elucidate the possible pathways of degradation. The disappearance of TDG was accompanied by an accumulation of acids. HPLC analysis revealed thiodiglycolic acid (TDGA) and [(2-hydroxyethyl)thio]acetic acid (HETA) in the culture supernatants of all the strains (Fig. 12). Thiodiglycol sulphoxide was not detected.

The appearance of acid products suggests that biodegradation of TDG begins with oxidation of its primary alcoholic groups. Earlier, oxidation of the thiodiglycol molecule to TDGA by alcohol dehydrogenase was reported (Lee et al., 2000, Ermakova et al., 2002). Incubation of all strains for more than 7 days led to a decrease in HETA and TDGA concentrations to an undetectable level.

<sup>447</sup> TGA and acetate were detected in the culture supernatant from the growth of all <sup>448</sup> bacteria on the TDGA-containing medium, which suggests C-S bond cleavage in <sup>449</sup> TDGA to yield these products.  $\beta$ -mercaptoethanol was detected as an intermediate <sup>450</sup> product of TDG transformation in only two strains: *Pseudomonas sp.* 93-2 and

#### 451 Achromobacter sp. 75-1.

These results indicate that the major pathway for TDG degradation by all the 452 strains studied is the oxidation of its primary alcoholic groups to HETA, TDGA and 453 TGA with the formation of acetate upon further transformation. This pathway for 454 TDG degradation is similar to that reported earlier for Alcaligenes and Pseudomonas 455 (Lee et al., 2000; Garcia-Ruiz et al., 2002, Medvedeva et al., 2006). However, the 456 formation of  $\beta$ -mercaptoethanol was detected for *Pseudomonas sp.* 93-2 and 457 Achromobacter sp. 75-1. Previously, this intermediate product has also been detected 458 among the products of TDG transformation by soil bacterium Pseudomonas sp. 459 (Medvedeva et al., 2007). Thus, the results suggest two possible pathways of the TDG 460 biodegradation, one of which yields HETA, TDGA and TGA, and the other,  $\beta$ -461 mercaptoethanol, which is further transformed to TGA (Fig. 13). 462

463 464

#### 4. Conclusions

465

Microbiological analysis of near-bottom water samples near the sea bed of the 466 dump sites for chemical weapons in the Baltic Sea revealed high levels of 467 microorganisms able to use mustard gas hydrolysis products as source of carbon 468 especially at the Bornholm dump site, as well as in the Gotland Deep, and at the 469 Skagerrak dump site. The "abnormal" sites showed reduced diversity of microbiota. 470 High numbers of MGHP-degrading microorganisms in near-bottom water indicate a 471 possible leakage of CW agents into the environment and contamination of water and 472 sediments with mustard gas and products of its hydrolysis in the research areas. 473

Bacterial cultures isolated from near-bottom water were able, with differing 474 biodegradation rates, to dechlorinate and utilize mustard gas hydrolysis products as 475 the source of carbon. The Arthrobacter, Achromobacter and Pseudomonas genera 476 cultures were the most active. The main features of Arthrobacter sp. 23-3, 477 Achromobacter sp. 75-1, and Pseudomonas sp. 93-2 are the tolerance to mustard gas 478 hydrolysis products, and the ability to degrade a high concentration of OCS. They can 479 use thiodiglycol as the sole source of carbon and energy. The isolated bacteria can 480 completely dechlorinate OCS at up to 750 mg/l in 4-9 days. But the complete 481 utilization of the products of hydrolysis (mainly thiodiglycol) is observed at lower 482 concentrations of OCS in the medium (up to 360 mg). The major pathway of TDG 483 degradation was oxidation to HETA, TDGA and TGA plus eventual formation of 484 acetate. The degradation of TDG by Pseudomonas sp. and Achromobacter sp. 75-1 485 can also occur via C-S bond cleavage yielding ßmercaptoethanol. The 486 microorganisms cannot only detoxify the contaminants, but also utilize them 487 completely. The rate of biodegradation of mustard gas hydrolysis was temperature-488 dependent but also occurred at low temperatures. 489

The ability of microorganisms to carry out the mineralization of thiodiglycol at a low temperature is important for the marine environment, particularly for the dump sites, where water temperatures rarely exceed 7°C. The results suggest the potential

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for self-purification of contaminated water and sediments by natural populations ofmicroorganisms.

495

#### 496 Acknowledgements

497

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#### 506 **References**

- Åstot, C., Hägglund, L., Nilsson, C., 2007. World War I Gas Bomb Found in Harbour
   Pool. Abstracts of the 9th International Symposium on Protection against
   Chemical and Biological Warfare Agents (Gothenburg, Sweden, 22-25 May,
   2007): 169.
- Beaudry, W.T., Bossle, P.C., Harvey, S.P., Kolakowski, J.E., Procell, L.R., et al.,
  1995. Neutralization/biodegradation of HD. In: Analytical Methods for
  Environmental Sampling of Chemical Warfare Agents and Their Degradation
  Products (Watson AP, Kistner SL, eds). ORNL/M-4315. Oak Ridge, TN:Oak
  Ridge National Laboratory.
- Black, R.M., Brewstar, K., Clarke, R.J., Hambrook, J.L., Harrison, J.M., Howells,
  D.J., 1993. Metabolism of thiodiglycol (2,2'thiobis-ethanol): isolation and
  identification of urinary metabolites following intraperitoneal administration to
  rats. Xenobiotica 23, 473-481.
- 521 Chemical munitions in the Southern and Western Baltic Sea. Complication, 522 assessment and recommendations. 1993. Federal Maritime and Hydrographic 523 agency, Hamburg, Germany.
- Ellman, G.L., 1959. Tissue sulphydryl groups. Archives of Biochemistry and
   Biophysics 82, 70-77.
- Ermakova, I.T., Starovoitov, I.I., Tikhonova, E.B., Slepen'kin, A.V., et al., 2002.
  Bioutilization of tiodiglycol, the product of mustard detoxification: isolation of
  degrading strains, study of biodegradation process and metabolic pathways.
  Process Biochemistry 38, 31-39.
- Franke S., 1967. Lehrbuch der Militarchemie. Berlin: Deutscher Militarverlag, B.1,
  440 (Dutch).
- Glasby, G.P., 1997. Disposal of Chemical Weapons in the Baltic Sea. The Science of
   the Total Environment. 206, 267-273.
- 534 Garcia-Ruiz, V., Martin-Otero, L. E., and Puyet, A., 2002. Transformation of

535	thiodiglycol by resting cells of Alcaligenes xylosoxidans PGH 10. Biotechnology
536	Progress 18, 252 - 256.
537	Harvey, S.P., DeFrank, J.J., 1993. Biodegradation of chemical warfare agents:
538	demilitarization applications, in: Kamely, D., Sasmore, R. (Eds.), Army science:
539	The New Frontiers, Military and Civilian Application, Woodlands TX: Borg
540	Biomedical Services.
541	HELCOM, 1993. Chemical munitions in the southern and western Baltic Sea -
542	compilation, assessment and recommendations. Federal Maritime and
543	Hydrographic Agency, Hamburg, Germany, May 1993.
544	HELCOM, 1994. Report on Chemical Munitions Dumped in the Baltic Sea, Report to
545	the 16th Meeting of the Helsinki Commission 8-11th March 1994 from the Ad
546	Hoc Working Group on Dumped Chemical Munitions, HELCOM CHEMU.
547	HELCOM, 1996. Third Periodic Assessment of the State of the Marine Environment
548	of the Baltic Sea, 1989-1993. Executive Summary. Balt. Sea Envir. Proc. No 64.
549	HELCOM, 2005. Press release regarding chemical weapons dumps in the Baltic.
550	Helsinki Commission, May 2.
551	www.helcom.fi/press_office/news_helcom/en_GB/ChemicalMunitions1115039886140/.
552	Karasevitch, Ju.N., 1982. Osnovi selekzii microorganizmov, utilizirujushich
553	sinteticheskiye organicheskiye soedineniya [Fundamental of Selection of
554	Synthetic Organic Substances Utilizing Microorganisms], Nauka, Moscow, 144.
555	Kilbane, J.J., Jackowski, K., 1996. Biocatalytic detoxification of 2-chloroethyl ethyl
556	sulfide. Journal of Chemical Technology & Biotechnology 65, 370 – 374.
557	Kuznetsov, C.I., Dubinina, G.A., 1989. Metody izucheniya vodnykh
558	mikroorganizmov (Methods of studying aquatic microorganisms), Moscow,
559	Nauka, 286 (in Russian).
560	Lee, T.S., Pham, M.Q., Weigand, W.A., Harvey, S.P., Bentley, W.E., 1996.
561	Bioreactor strategies for the treatment of growth inhibitory waste: an analysis of
562	thiodiglycol degradation, the main hydrolysis product of sulfur mustard.
563	Biotechnology Progress 12, 533 - 539.
564	Lee, T.S., Chan, S.H., Weigand, W.A., Bentley, W.E., 2000. Biocatalytic
565	transformation of [(2-Hydroxyethyl)thio]acetic acid and thiodiglycolic acid from
566	thiodiglycol by Alcaligenes xylosoxidans ssp. xylosoxidans (SH91).
567	Biotechnology Progress 16, 363-367.
568	Medvedeva, N.G., Spiridonov M.A., Polyak, Yu.M., Gridneva, Yu.A., Zaytseva,
569	T.B., Ribalko, A.E. 1998. Ecological inspections of the Baltic Sea chemical
570	weapon dump sites. Ecological Chemistry 7, 20-26 (in Russian).
571	Medvedeva, N.G., Zaytseva, T.B., Zinovieva, S.V., Sukharevich, V.I., Orlova, O.G.
572	2006. Destruction of mustard gas hydrolysis products by <i>Pseudomonas sp.</i> Y –
573	13. Biotekhnologiya 2, 50-56 (in Russian).
574	Medvedeva, N., Polyak, Y., Zaytceva, T., Zinovieva, S., 2000. Microbiological
575	destruction of mustard in soil, in: Wise, D.D., Trantolo, D.J., Cichon, E.J., et.al.
576	(Eds.), Bioremediation of contaminated soils, Marcel Dekker, New York -

15

- 577 Basel, pp. 151-176.
- Medvedeva, N., Polyak, Yu., Zaytseva, T., Zinovieva, S., 2007. Soil bacterium
   *Pseudomonas sp.*: destroyer of mustard gas hydrolysis products. Biotechnology
   Journal 2, 1033-1039.
- Medvedeva, N., Polyak, Yu., Kuzikova, I., Orlova, O., Zharikov, G., 2008. The effect
   of mustard gas on the biological activity of soil. Environmental Research 6, 289
   295.
- Mulbry, W., Rainina, E., 1998. Biodegradation of chemical warfare agents. ASM news 64, 325-331.
- Munro, N.B., Talmage, S.S., Griffin, G.D., Waters, L.C., et al., 1999. The sources,
   fate, and toxicity of chemical warfare agent degradation products. Environmental
   Health Perspectives 107, 933-973.
- Oldenhuis, R., 1992. Microbial Degradation of Chlorinated Compounds: Application
   of Specialized Bacteria in Treatment of Contaminated soil and Waste Water,
   Groningen, 214 p.
- Opresko, D.M., Young, R.A., Faust, R.A., Talmage, S.S., Watson, A.P., Ross, R.H.,
   Davidson, K.A., King, J., 1998. Chemical warfare agents: estimating oral
   reference doses. Reviews of Environmental Contamination & Toxicology 156,
   1–184.
- Paka, V., Spiridonov, M., 2001. Research of dumped chemical weapons made by R/V
  "Professor Shtokman" in the Gotland, Bornholm & Skagerrak dumpsites, in:
  Missiaen, T., Henriet, J.-P. (Eds.), Chemical Munition Dump Sites in Coastal
  Environments, pp. 27–42.
- Paka, V., 2004. Sea-dumped chemical weapons: state of problem. Russian Chemical
   Journal (Journal of Mendeleev Russian Chemical Society) XLVIII (2), 99–109.
- Papirmeister, B., Feister, A.F., Robinson, S.I., et al., 1991. Medical defense against
   mustard gas: Toxic mechanisms and pharmacological implications, CRC Press,
   Boca Raton, FL.
- Pirt, C.J., 1975. Principles of Microbe and Cell Cultivation, Blackwell Scientific
   Publications, London, 287.
- Schulz-Ohlberg, J., Lemke, W., and Tauber, F., 2001. Tracing dumped chemical munitions in Pomeranian Bay (Baltic Sea) at former transport routes to the dumping areas off Bornholm Island, in: Missiaen, T., Henriet, J.-P. (Eds), Chemical Munition Dumpsites in Coastal Environments, pp. 43-52.
- Tsyban A.V., 1980. Rukovodstvo po metodam biologicheskogo analiza morskoi vody
   i donnykh otlozhenii (Manual on methods of biological analysis of seawater and
   bottom sediments), Gidrometeoizdat, Leningrad, 192 (in Russian).
- Wariishi, H., Itoh, N., Yoshida, M., Miyamoto, T., Tanaka, H., 2002. Complete
  degradation of yperite, a chemical warfare agent, by basidiomycetes.
  Biotechnology Letters 24, 501-505.
- Yang, Y.C., Baker, J.A., Ward, J.R., 1992. Decontamination of chemical warfare
   agents. Chemical Reviews 2, 1729-1743.

#### 619 Figure legends

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Fig. 1 - Dump site locations map: (A) the Bornholm site; (B) the Gotland Deep site;(C) the Skagerrack site

Fig. 2 - Location of the chemical munitions dump site at the Bornholm Basin. The circle marks the primary dump site area; the rectangle marks the larger dump site boundary. 1: the first research area, 2: the second research area

Fig. 3 – Images of the shipwrecks at the Bornholm dump site (after Paka and Spridonov, 2001). A: wreck covered with fishing nets. B: missile found on the deck of the wreck

631

Fig. 4 - Concentration of MGHP-degrading microorganisms (mean  $\pm$ SD) present at the primary dump site within the first research area (1), within the second research area (2), and out of the primary dump site (3)

635

Fig. 5 - Growth and utilization of TDG by *Achromobacter sp.* 75-1. The data are means  $\pm$  SD of three independent measurements.

638

Fig. 6 - Effect of MGHPs concentrations on the growth of *Achromobacter sp.* 75-1
(a), *Arthrobacter sp.* 23-3 (b), and *Pseudomonas sp.* 93-2 (c).

641

Fig. 7 - Effect of OCS on dechlorination rate (A) and growth rate (B) of *Achromobacter sp.* 70-11, *Arthrobacter sp.* 41-3 and *Pseudomonas sp.* 3-1. The data are means  $\pm$  SD of three independent measurements.

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Fig. 8 - Effect of pH on growth of *Achromobacter sp.* 75-1, *Arthrobacter sp.* 23-3 and *Pseudomonas sp.* 93-2 (25°C). The data are means  $\pm$  SD (n=3).

- Fig. 9 Effect of temperature on degradation of TDG by *Achromobacter sp.* 75-1 under static conditions: (a) TDG concentration; (b) biomass growth  $(OD_{540})$ . The symbols indicate the means of triplicate experiments. Vertical bars indicate standard deviations.
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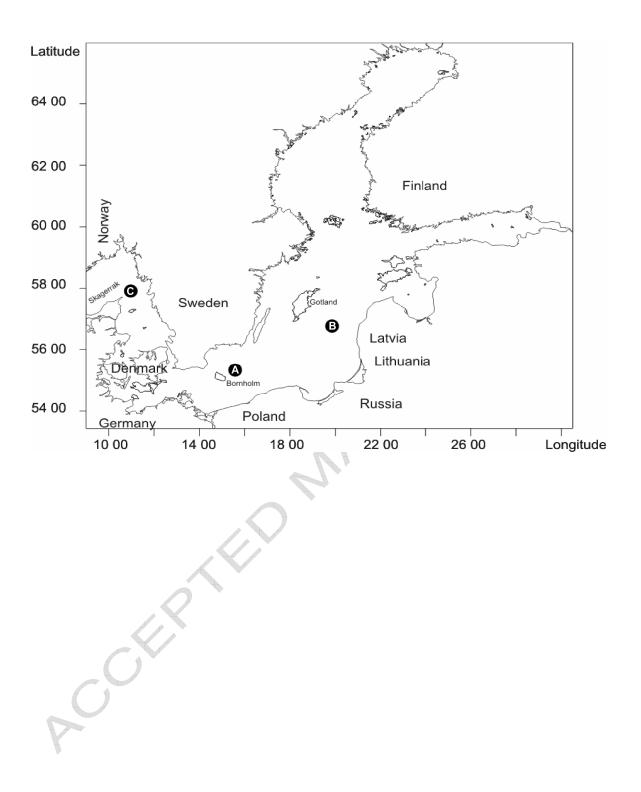
Fig. 10 - Effect of temperature on degradation of TDG by *Arthrobacter sp.* 23-3 under static conditions: (a) TDG concentration; (b) biomass growth  $(OD_{540})$ . The symbols indicate the means of triplicate experiments. Vertical bars indicate standard deviations.

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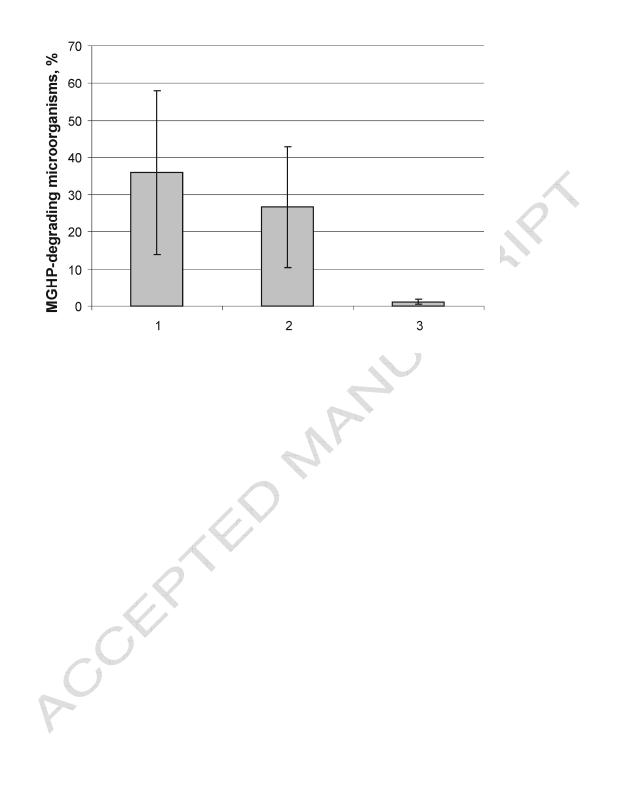
Fig. 11 - Effect of temperature on degradation of TDG by *Pseudomonas sp.* 93-2 under static conditions: (a) TDG concentration; (b) biomass growth  $(OD_{540})$ . The

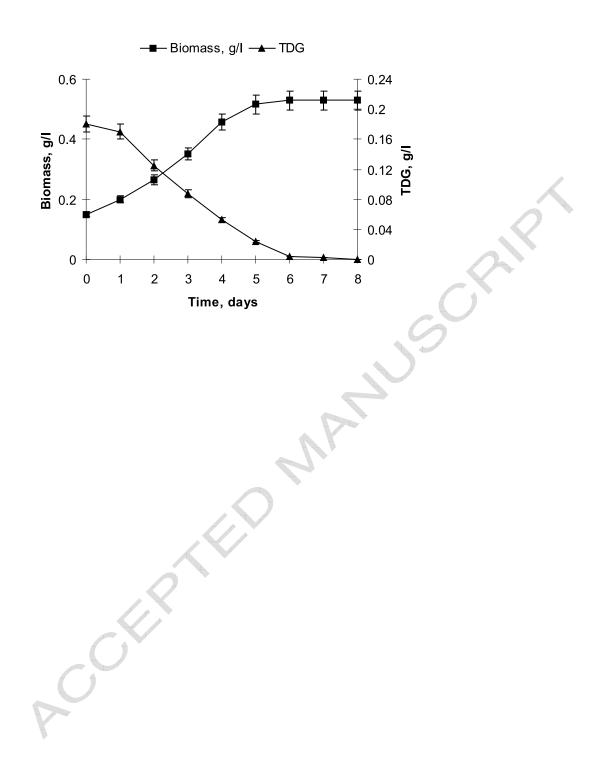
- 661 symbols indicate the means of triplicate experiments. Vertical bars indicate standard 662 deviations.
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- <sup>664</sup> Fig 12 HPLC chromatogram of *Arthrobacter sp.* 23-3 supernatant: (1) at initiation of
- the incubation in TDG-containing media; (2) after 72 hours of incubation
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- Fig. 13 Metabolic pathways for TDG utilization by bacteria isolated from the Baltic Sea

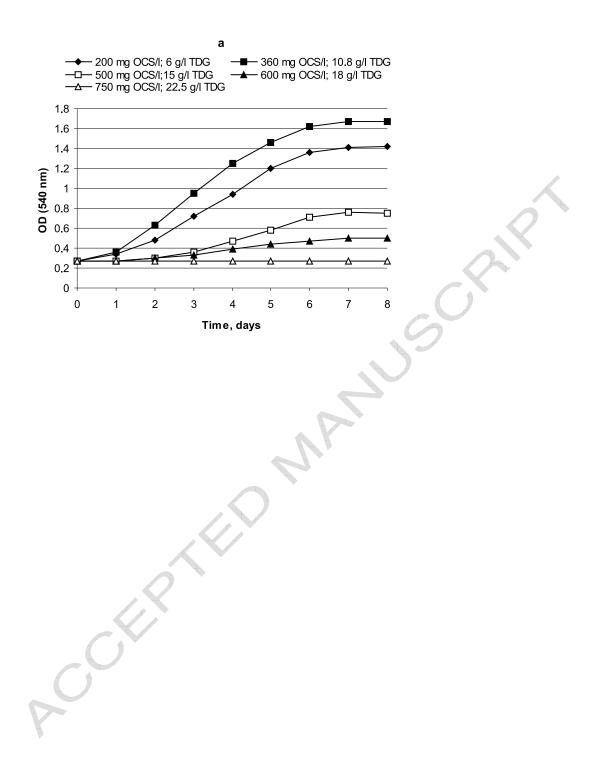
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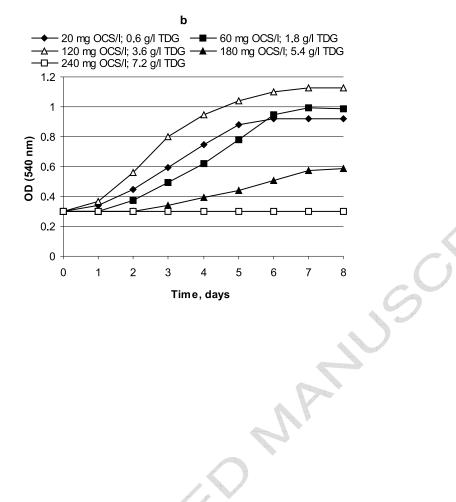




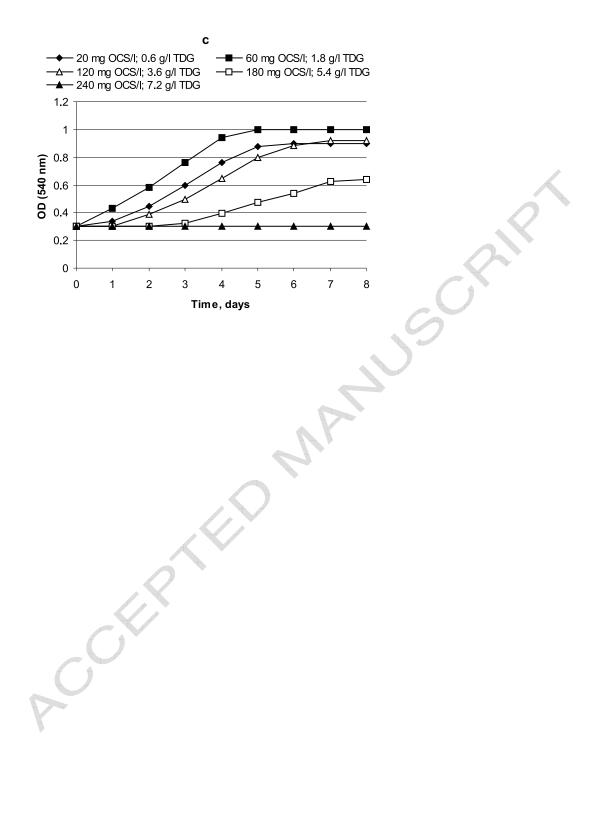


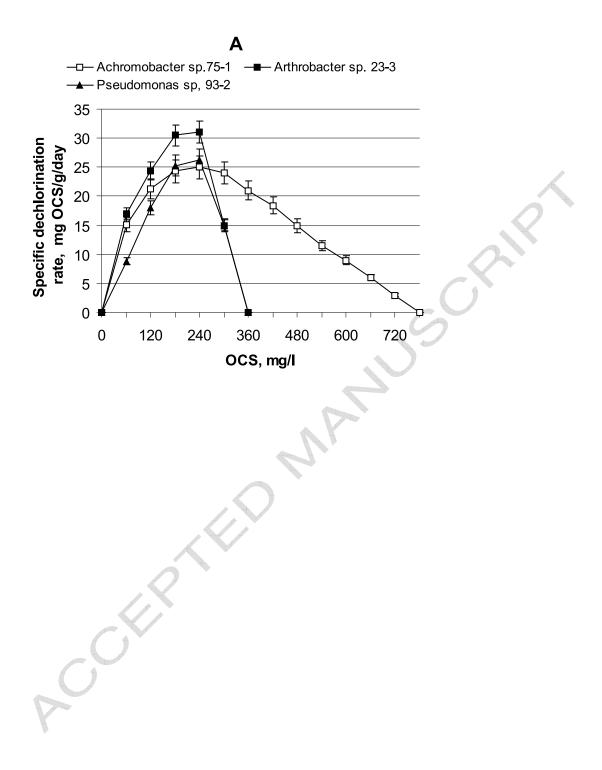


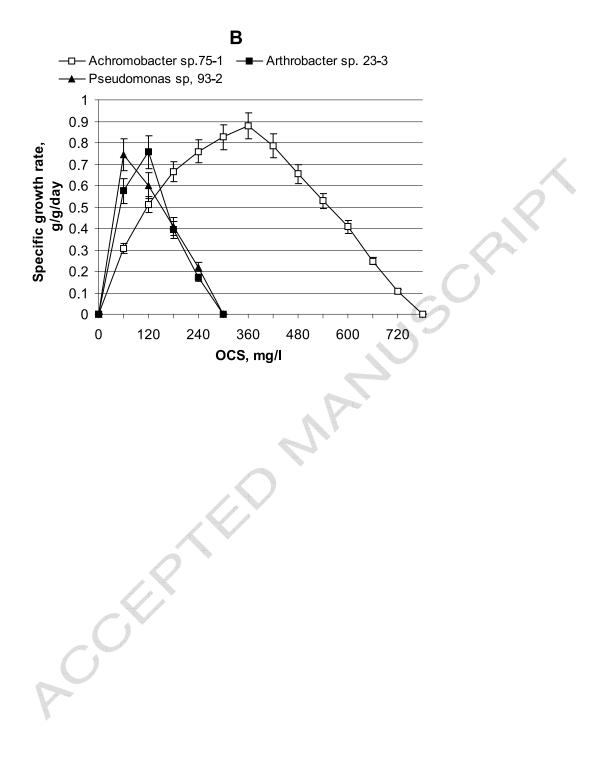


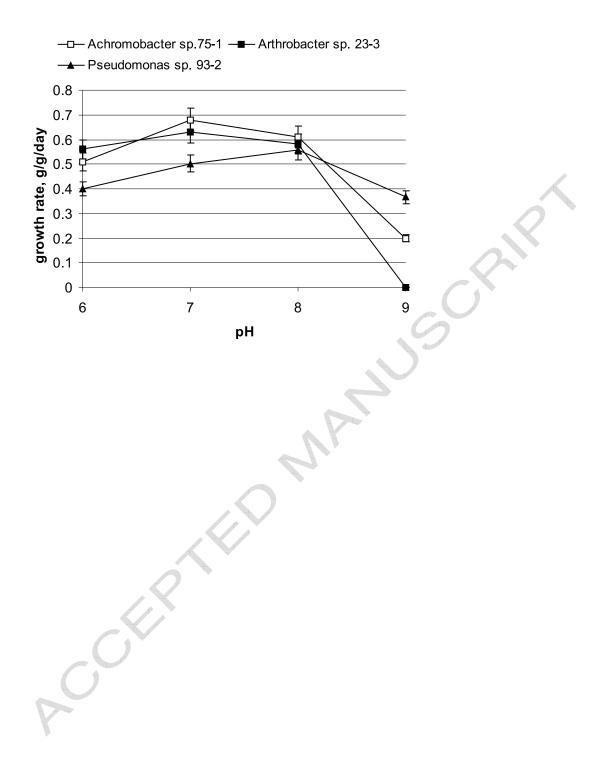


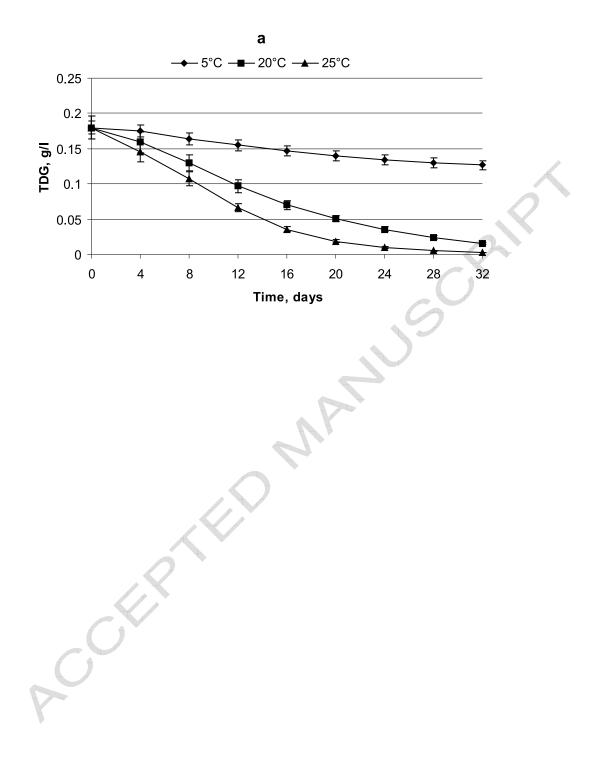
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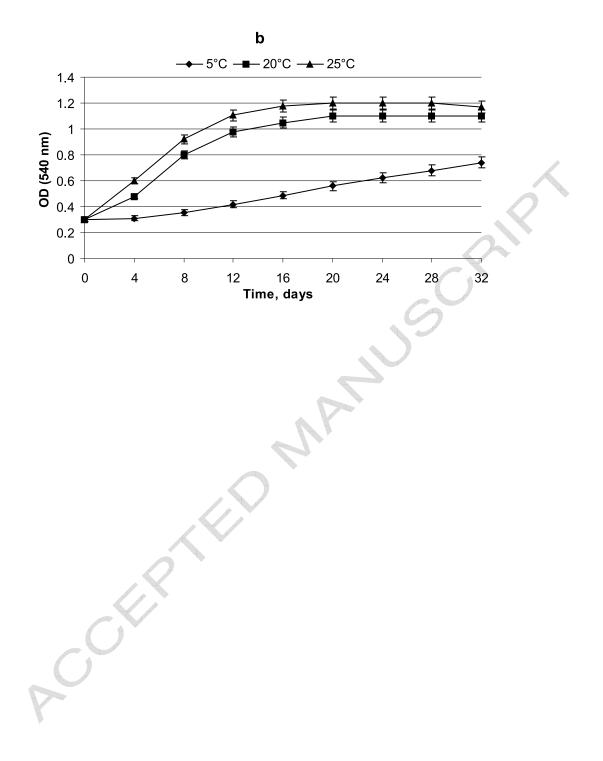


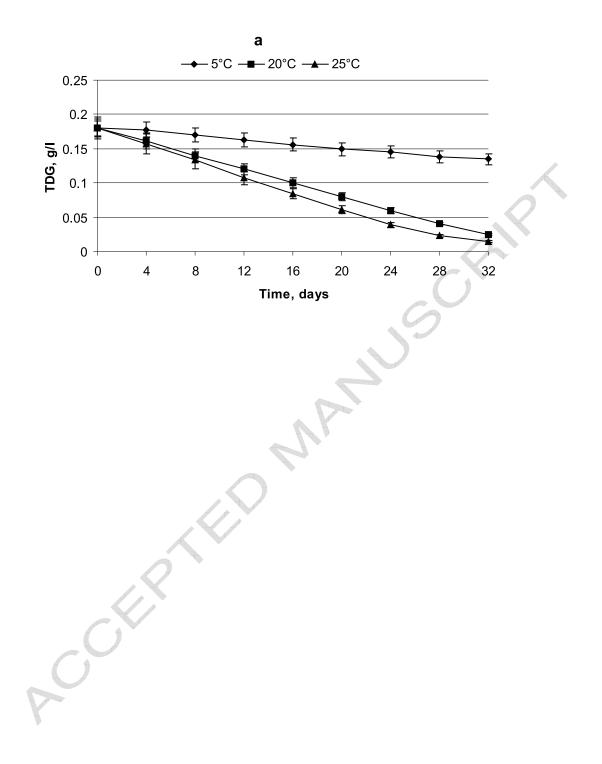


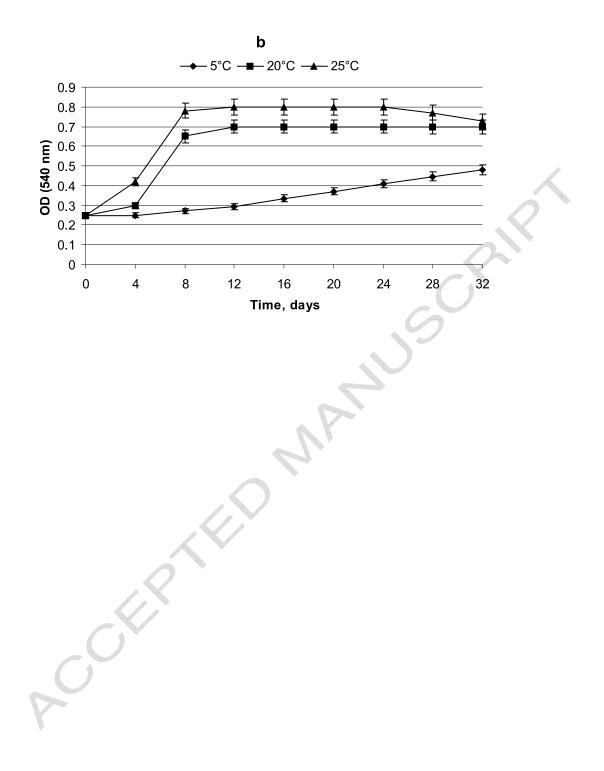


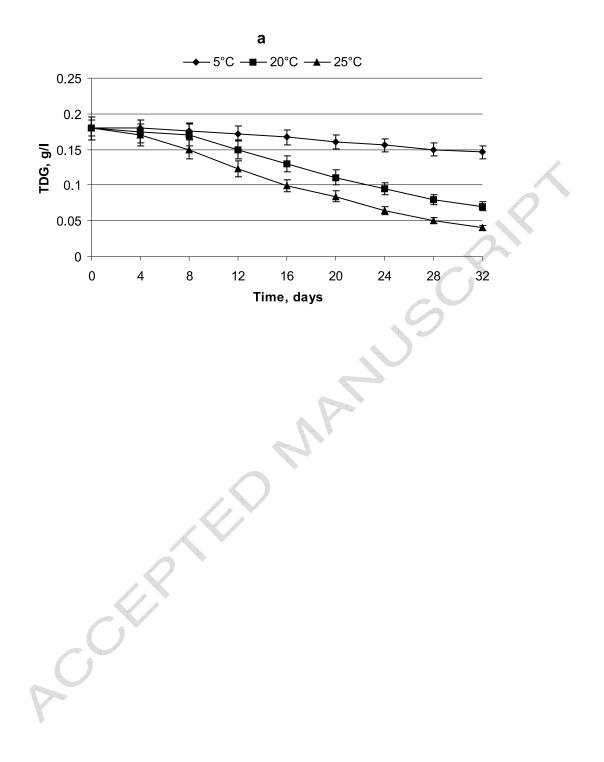


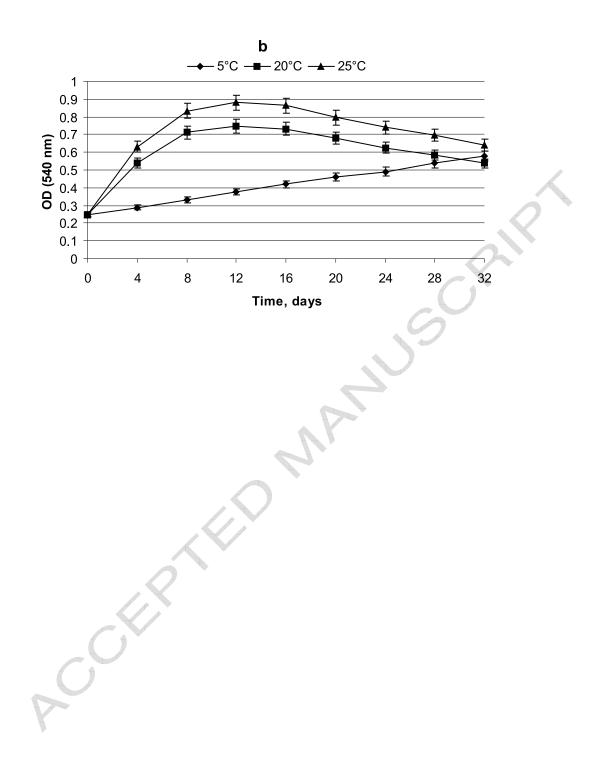


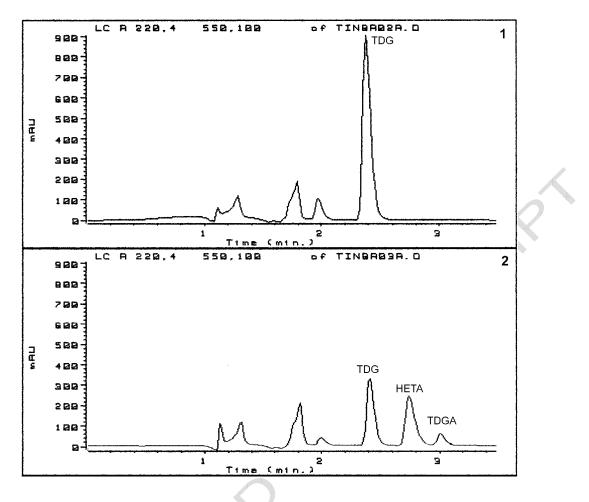




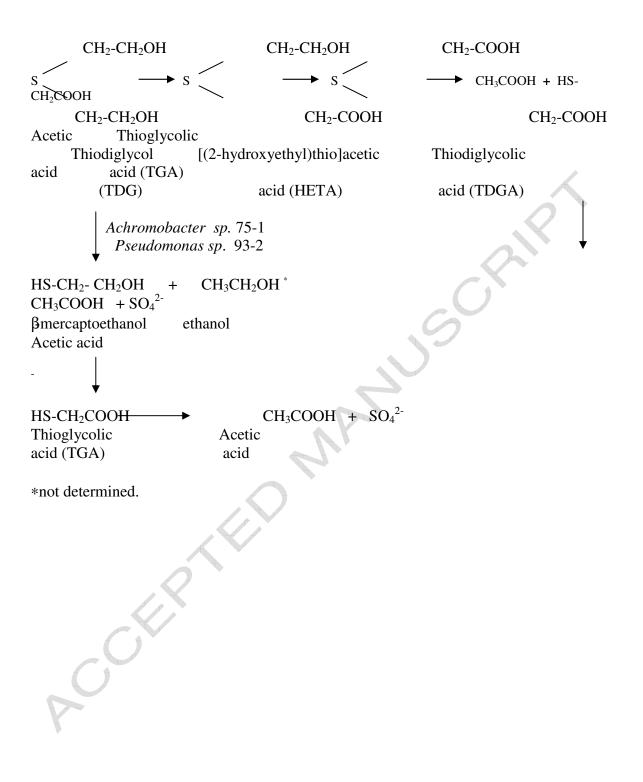








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#### Table 1

Quantities of the chemical munitions and chemical war gases dumped in the southern and western Baltic Sea and Skagerrak (west of Sweden)

Area	Quantities of		Types of war gases
	chemical munitions	chemical war gases	
Bornholm	from 35,300 tons	from 5,300 tons	Mustard gas, viscous mustard
Basin	(reliable data)	to 6,500 tons	gas, Clark I, II, adamsite,
(E of	to 43,300 tons		chloroacetophenone (possibly,
Bornholm)	(unreliable data)		phosgene, nitrogen mustard)
Area SW of	up to15,000 tons	2 250 tons	no data
Bornholm	(unreliable data)		
Gotland Basin	2,000 tons	300 tons	no data
Little Belt	5,000 tons	750 tons	tabun, phosgene
Skagerrak (W	app. 20,000 tons	-	mustard gas
of Sweden )	(quantity not verified)		(other types not verified)

Microbiological characteristics of the Baltic Sea near-bottom water					
	Total bacterial	Bacterial	Number of heterotrophs		
Dumping site	number,	biomass,	Total,	MGHP-degrading microorganisms,	
	10 <sup>6</sup> /ml	mg/ml	$10^4$ /ml	% of total number of heterotrophs	
Gotland Basin	1.3 - 4.2	1.1 - 3.0	0.6 – 2.0	up to 20	
Bornholm Basin	0.7 – 5.3	0.6 – 4.9	0.5 – 3.2	up to 85	
Skagerrak Strait	0.5 - 7.1	0.4 - 6.5	0.2 – 4.9	up to 98	

 Table 2

 Microbiological characteristics of the Baltic Sea near-bottom water

 Table 3

 Taxonomic structure of the bacterioplankton in Southern Baltic Sea

 Out of the dump site

 Bornholm dump site

Out of the dump site	Bornholm dump site	
Bacillus	Pseudomonas	
Pseudomonas	Bacillus	
Alcaligenes	Achromobacter	
Sphingomonas	Arthrobacter	
Achromobacter	Others: <15%	
Flavobacterium		
Arthrobacter		
Acinetobacter		9
Micrococcus		
Flectobacillus		
Planococcus		
Others: < 25%		
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#### Table 4

Comparative cultivation of the degrader microorganisms on the MGHPs containing medium (60 mg OCS/l and 0.18% TDG, 20°C)

	Static conditions		Aeration (15 mM O <sub>2</sub> /l/h)	
Microorganisms	Biomass gain	$q_{max}^*$ (mean ±	Biomass gain	$q_{max}^{*}$ (mean ±
	(mean ±SD), %	SD), mg/g/d	(mean ±SD), %	SD), mg/g/d
Pseudomonas sp. 93-2	10.4±0.6	9±1	212.1±6.4	17±2
Arthrobacter sp. 23-3	17.1±0.8	19±2	226.0±9.4	26±3
Achromobacter sp. 75-1	24.0±1.3	26±3	267.5±12.2	37±4

inas. \* q<sub>max</sub> – maximum specific dechlorination rate, mg OCS/g biomass/day.

ter of temperature on growth of degrader-interoorganishis under static conditions				
Microorganisms	Temperature,	Y <sup>a</sup> (mean ±SD),	$\mu_{max}^{b}$ (mean $\pm$ SD),	
	°C	g/g	g/g/d	
	5	0.37±0.02	0.07±0.01	
Achromobacter sp. 75-1	20	0.39±0.03	0.23±0.02	
	25	0.39±0.04	0.20±0.01	
	5	0.23±0.02	0.07±0.01	
Arthrobacter sp. 23-3	20	$0.29 \pm 0.02$	0.34±0.03	
	25	0.31±0.03	0.30±0.02	
	5	0.32±0.03	0.09±0.01	
Pseudomonas sp. 93-2	20	0.33±0.02	0.31±0.02	
	25	0.35±0.03	0.35±0.03	

## Table 5 Effect of temperature on growth of degrader-microorganisms under static conditions

<sup>a</sup>Y - biomass yield, g biomass produced [dry wt]/ g substrate utilized.

 ${}^{b}\mu_{max}$  - maximum specific growth rate, g /g biomass/day.

Data are means ± SD of three independent measurements.