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Miguel Angel Huerta-Diaz, Francisco Delgadillo-Hinojosa, Martín Hernández-Ayón, José Antonio Segovia-Zavala, Zaúl García-Esquivel, et al.. Diagnosis of trace metal contamination in sediments: the example of Ensenada and El Sauzal, two harbors in Baja California, Mexico. Marine Environmental Research, 2008, 66 (3), pp.345. 10.1016/j.marenvres.2008.05.008. hal-00563036

HAL Id: hal-00563036 https://hal.science/hal-00563036

Submitted on 4 Feb 2011

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Accepted Manuscript

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Marine Environmental Research

 PII:
 S0141-1136(08)00162-1

 DOI:
 10.1016/j.marenvres.2008.05.008

 Reference:
 MERE 3260

To appear in:

Received Date:8 May 2007Revised Date:1 May 2008Accepted Date:15 May 2008



Please cite this article as: Huerta-Diaz, M.A., Delgadillo-Hinojosa, F., Hernández-Ayón, M., Segovia-Zavala, J.A., García-Esquivel, Z., López-Zárate, H., Siqueiros-Valencia, A., Galindo-Bect, S., Diagnosis of trace metal contamination in sediments: the example of Ensenada and El Sauzal, two harbors in Baja California, Mexico, *Marine Environmental Research* (2008), doi: 10.1016/j.marenvres.2008.05.008

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24 Abstract

25

26	Total metal concentrations in sediments from within Ensenada and El Sauzal harbors are
27	generally higher than at the mouths. Grain-size analyses suggested that this enrichment could
28	be due to the presence of fine-grained sediments in the inner part of the harbors rather than to
29	anthropogenic perturbations. The $(Me/AI)_{sample}$ ratios for Pb, Co, Ni and Fe were significantly
30	higher for Ensenada Harbor relative to El Sauzal Harbor, whereas the ratios for Cd, Mn, Zn and
31	Cu were statistically equivalent for both harbors. Calculated enrichment factors $[EF_{Me} =$
32	(Me/AI) _{sample} /(Me/AI) _{shale}] indicated that the metals showing slight enrichment were those
33	associated with anthropogenic contamination (Pb, Zn), or probably related to primary productivity
34	in the water column (Cd, Co). The levels of most of the metals were not greatly enriched, a
35	consideration that is of the utmost importance when contamination issues are at stake.
36	
37	Keywords: enrichment factor, geochemistry, grain size, metals, sediment pollution, trace
38	elements, Baja California.
	ACCEPTER

39 1. Introduction

40

41 Harbors are enclosed and low-energy water bodies where fine-grained sediment tends to 42 accumulate. They are also prone to receive significant metal inputs from marinas, boat hull 43 maintenance, wholesale fish markets, shipping activities, sacrificial anodes, and industrial, storm 44 and urban discharges. Once in the water column, metals are quickly adsorbed onto particulates 45 and eventually removed to bottom sediments (de Groot et al., 1982; Santschi et al., 1984; Blake 46 et al., 2004), thus producing conspicuous trace metal enrichments. Moreover, trace metal 47 enrichments in sediments produced by normal harbor activities can be further enhanced by the 48 larger surface area of the finer sediments that are naturally deposited in these areas of minimum 49 hydrodynamic energy. It is therefore important to measure the extent of metal enrichment in 50 harbor sediments since they can act as point sources of contamination during dredging 51 operations or any other activity by which contaminated sediments can be transported out of the 52 harbors and into neighboring bays or open ocean areas. 53 Ensenada and El Sauzal harbors are the main harbors on the Pacific coast of northern 54 Mexico. They are located in Todos Santos Bay (Figure 1) and plans are well underway to 55 modernize both harbors by increasing their extension and container handling capacity, especially 56 in the case of El Sauzal Harbor, whose area will be increased approximately ten times. 57 Additionally, Ensenada Harbor has a cruise terminal capable of handling at least two cruise ships 58 simultaneously. All this expansion work and increase in shipping traffic are associated with 59 dredging activities that may produce a net export of metals out of the harbors. Hence, 60 understanding the relative contributions of the different trace metals is necessary in order to 61 assess the impact of future developments in these harbors and their surrounding waters, and to 62 develop effective strategies to protect them and other harbors elsewhere. 63 In this paper we will try to answer the following specific questions: Are sediments from 64 Ensenada and El Sauzal harbors enriched in trace metals (Cd, Co, Cu, Fe, Mn, Ni, Pb, Zn)? If 65 so, to what extent and with which metals? Which harbor presents the highest metal

66 enrichments? If present, are these trace metal enrichments anthropogenically derived? To

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67 answer these questions, trace element levels (normalized with AI) will be compared with those 68 reported for other sediments from the Southern California Bight and with average shale values of 69 sedimentary rocks to evaluate the extent of anthropogenic perturbations in the sediments of these 70 two important Mexican harbors. Additionally, pollution load indexes (Tomlinson et al., 1980; 71 Angulo, 1996) and geoaccumulation indexes (Müller, 1979) will be calculated for both harbors. 72 Finally, total concentrations will be compared with the U.S. NOAA's sediment quality guidelines 73 (as proposed by Long et al., 1995) to estimate the possible consequences of the metal levels 74 analyzed in this study to the local biota. Because of the absence of similar studies in the region, 75 this paper will also provide baseline data for future research on anthropogenic impacts in the 76 region.

77

78 2. Materials and methods

79 2.1. Study area

80

81 Ensenada and El Sauzal harbors are situated in Todos Santos Bay, which is located in 82 the northwestern part of the Baja California Peninsula, Mexico (Figure 1a), 100 km south of the US-Mexico border (31º 40', 31º 56' N; 116º 36', 116º 50' W). This 240-km² bay is limited to the 83 84 south by Punta Banda, to the north by Punta San Miguel, to the southwest by two small islands 85 located near the mouth of the bay, and to the northeast by the city of Ensenada (around 400,000 86 inhabitants). According to Pavía (2004), the climate of the region is Mediterranean, with scarce 87 rainfall (annual average of 250 mm) that occurs mainly during the winter (average of 200 mm). 88 The regional surface circulation pattern is produced by a predominant northwest component of the wind and surface currents have been reported to average 15 cm s⁻¹ and 5 cm s⁻¹ during 89 90 summer and winter, respectively (Alvarez-Sánchez et al., 1988). This wind-driven circulation 91 pattern promotes the transport of water from the adjacent Pacific Ocean towards the inner bay, 92 generating a cell circulation system (dotted arrows; Figure 1b). Part of this flow is directed toward 93 Ensenada harbor and the other part toward Punta Banda, in the southwestern part of the bay

94 (Figure 1b). Upwelling is intense all year round, but especially in spring and summer (Gonzalez95 Morales and Gaxiola-Castro, 1991).

96 Ensenada Harbor, an international marine terminal, was built in 1956 and covers an area of 1.95 km² (i.e., approximately 13 times bigger than El Sauzal Harbor). The mouth of the harbor 97 98 is formed by a 1,640-m-long rock breakwater that is connected to the coast and by a 855-m-long 99 jetty (El Gallo; Figure 1d). Water depths range from 1.5 to 11 m, with the deepest parts located 100 adjacent to the loading and terminal docks and the main navigation channel. In addition to the 101 presence of fine-grained sediments (approximately 80% <62.5 µm), this harbor tends to 102 accumulate metals due to inputs from (a) a wholesale fish market, (b) marine vessels normally 103 docked in the port and marina areas. (c) urban runoff discharged through Arroyo Ensenada 104 (Figure 1d) during the winter rains, and (d) vessel reparation and maintenance activities 105 (liberation of antifouling paint, sand blasting operations) within the port zone.

106 El Sauzal Harbor, built in the mid-1980s and located 10 km north of Ensenada in the 107 town of El Sauzal de Rodriguez (approximately 7,500 inhabitants), has an area of 0.15 km² (Flores-Vidal et al., 2005) and a 100-m-wide mouth formed by two breakwaters that are 500 m 108 109 and 250 m long (Figure 1c). Water depths range from 1 to 9 m, with maximum depth occurring in 110 the main navigation channel (Flores-Vidal et al., 2005). Inputs to this harbor are essentially from 111 fishery companies dedicated to extraction, processing and production activities, as well as from 112 blood and organic wastes discharged by docked fishing boats, and from the powder generated during the unloading of cement from large boats or vessels. 113

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115 2.2. Field and laboratory methods

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Sediment samples from Ensenada Harbor were obtained from one core (core 0E)
collected at approximately 6 m water depth on November 2, 1998 (Carreón-Martínez et al., 2001).
This core was collected within the dredged channel, close to Arroyo Ensenada and in front of the
cruise terminal, which was still not constructed at that time. Six more cores (cores 1E to 6E) were
collected on January 8, 2002 at water depths of approximately 9 m (Figure 1d) and beneath the

122 docks, considered the areas least affected by sediment resuspension and dredging. Sediments 123 had a hydrogen sulfide smell, were black along the length of the cores and did not show any 124 oxidized portion close to the sediment-water interface. Cores from El Sauzal Harbor were 125 collected on September 29, 2004 (cores 1S to 4S) at water depths ranging from 6 to 9 m (Figure 126 1c). A polycarbonate plastic core liner (7.2 cm internal diameter, 60 cm in length) was introduced 127 into the sediment, retrieved and capped underwater by a diver who transferred it to a boat. Once 128 on board, the core liner was capped with a plastic cap, sealed with electrical tape and transported 129 to shore, where it was extruded and sliced every 1 cm with a plastic spatula. Each section was 130 then transferred to a 50-mL centrifuge polypropylene tube and stored at -20°C for further 131 laboratory analysis. Only core 0E was sliced in a glove bag under nitrogen atmosphere to avoid 132 precipitation of the reduced Fe and Mn dissolved in the anoxic portion of the interstitial water. 133 The 10-13 cm sediment depths of core 2S consisted almost entirely of fish scales.

134 All laboratory materials were washed with phosphate-free soap, rinsed three times with 135 distilled water and left for 24 hours in a 5% HCl solution. The material was then rinsed three 136 times with deionized water (Milli-Q grade) and left semi-closed to dry at room temperature. Total 137 metal concentrations were obtained after complete digestion of 0.5 g of sediment in Teflon 138 beakers with concentrated HNO₃, HCIO₄ and HF (Carignan and Tessier, 1988). Trace metal 139 concentrations were measured by atomic absorption spectrometry (Thermo Jarrel Ash model 140 Smith Hieftje 12 or Varian model SpectrAA 220 Fast Sequential). Certified Reference Material 141 (CRM) Beaufort Chemistry Standard Sediment (BCSS-1; National Research Council of Canada) 142 was used to ascertain the accuracy and precision of the total extraction procedure. Recovery 143 percentages of the BCSS-1 CRM ranged from 90% for Cd to 104% for Cu (Table 1). Blanks were 144 routinely run and analyzed in the same manner as the samples. Limits of detection for the 145 different trace metals, calculated as three times the standard deviation of the procedural blanks. 146 were (in µmol g⁻¹): 2.4 (Al), 0.0076 (Cd), 0.026 (Co), 0.024 (Cu), 0.014 (Fe), 0.0047 (Mn), 0.11 147 (Ni), 0.035 (Pb), 0.013 (Zn). In this paper, and for practical purposes, all metals associated with 148 sediments will be designated as trace metals, regardless of their concentration level. Percentage 149 of sediment grain size <62.5 µm (%GS) was measured using a Horiba laser scattering particle

150 size distribution analyzer model LA-910, with size interval of 0.02-1000 µm. The analytical 151 efficiency of the particle analyzer was determined with 0.9-µm sieved polystyrene spheres (CRM 152 NIST 8010D and NIST 1690). Organic carbon (org-C) concentrations in the sediments were 153 estimated using the loss on ignition (LOI) method (550 ℃ for 5 h). These measurements were 154 transformed to org-C concentrations through a calibration curve (org-C $\% = (0.206 \pm 0.010)^{*}$ LOI; $r^2 = 0.929$, p<0.001, n = 33) using org-C values from Todos Santos Bay sediments measured by 155 a LECO model CHNS-932 elemental analyzer. Due to the lack of available samples, org-C and 156 157 grain size were not measured in core 0E.

To reduce the complexity within the similarity matrix, the dataset (including metal concentrations, org-C content and %GS) was transformed into a simpler factor matrix by principal components analysis (PCA). For both Ensenada and El Sauzal harbors, the PCA was calculated based on the correlation matrix of the standardized data using the SYSTAT 8.0 statistical package. For this analysis, eigenvalues >1 were considered significant and, in addition, the orthogonal varimax rotation was chosen for factor rotation.

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165 3. Results and discussion

166 3.1. Trace metal distributions

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Trace metal profiles for both Ensenada and El Sauzal harbors showed no substantial 168 changes in concentration with sediment depth (Figures 2 and 3), except for Cu and Zn in core 1E, 169 in which peaks of maximum concentration (229 μ mol g⁻¹ and 30 μ mol g⁻¹, respectively; Figures 2c 170 171 and 2i) were observed at a depth of 5.5 cm for both elements. The concentration of Cu was also high in the first 6 cm of core 4E (2.66 \pm 0.49 μ mol g⁻¹), while Cd concentrations increased with 172 depth (maximum value = 0.018 μ mol g⁻¹ at 19.5 cm) in the first 20 cm of core 0E. Significant 173 174 trace metal enrichments close to the sediment-water interface are absent in both harbors, further 175 proof that the cores were anoxic up to the sediment-water interface. The lack of significant 176 features in all trace metal profiles is probably a consequence of the absence of the oxic layer 177 (and probably the anoxic non-sulfidic layer as well) and of the homogeneity of the %GS (Figure

178 2j). Bioturbation can be discarded as the cause since all the cores were anoxic-sulfidic up to the179 sediment-water interface.

180 Overall, trace metals in Ensenada Harbor showed a wide range of average 181 concentrations (Table 2), the lowest corresponding to Cd (overall average 32 ± 16 nmol g⁻¹) and the highest to AI (overall average 3.81 ± 0.75 mmol g⁻¹). On average, core 1E showed the 182 highest values for Cu (15 ± 40 μ mol g⁻¹), Fe (1.33 ± 0.16 mmol g⁻¹), Mn (12.7 ± 2.5 μ mol g⁻¹) and 183 Zn $(5.8 \pm 4.5 \,\mu\text{mol g}^{-1})$; core 0E for Co $(1.79 \pm 0.45 \,\mu\text{mol g}^{-1})$ and Ni $(2.136 \pm 0.046 \,\mu\text{mol g}^{-1})$; 184 core 2E for Cd (50.6 \pm 2.5 nmol g⁻¹) and Al (4.94 \pm 0.25 mmol g⁻¹); and core 6E for Pb (371 \pm 14) 185 186 (Table 2). Metals (Co, Cu, Ni, Zn, Fe and, to a certain extent, Al, Mn and Cd) and org-C generally 187 decreased in concentration from the inner part of Ensenada Harbor to its mouth (Figure 4). This 188 result is not unexpected since the inner portion of the harbor shows finer grain size (Figure 4), 189 more shipping activity and less hydrodynamic energy than the outer parts. In this harbor, the 190 deepest areas range from 8 to 12 m and the shallower parts from 2 to 6 m. Cores 3E, 4E and 5E 191 were taken just at the edge of the deep main channel and thus presented higher sand 192 composition but also low metal concentrations (Figure 4). Sediments located at the mouth of the 193 harbor presented low %GS (reflecting a high level of hydrodynamic energy) and, consequently, 194 high sand contents that diluted trace metal concentrations. Core 0E, however, usually presented 195 anomalous concentrations when compared with the other cores from Ensenada Harbor (Figure 196 4); for example, it had considerably higher median concentrations of Co, Fe and Ni, and lower 197 median concentrations of Cd and Pb relative to the other cores. This anomalous behavior can be 198 attributed to the location of the core, which was collected inside the dredged channel and close to 199 the mouth of Arroyo Ensenada. Sediment resuspension produced by continuous shipping traffic, 200 repeated dredgings (which can alter the distribution of some key components of sediments, like 201 nutrients and carbon; Nayar et al., 2007) and occasional particulate and dissolved contributions 202 from the seasonal stream have evidently altered the distribution and levels of trace metals in core 203 0E. These alterations are more noticeable for Cd, Co, Fe, Mn, Pb and Zn, as reflected in the 204 variability (spread) of their concentrations (Figures 2a,b,d,f,h and i, respectively) and by the 205 increase in length of the box plots shown in Figure 4.

206	Average metal concentrations in El Sauzal Harbor (Table 2) were consistently high in
207	cores 2S (Cd = 28.2 \pm 1.5 nmol/g, Cu = 1.24 \pm 0.13 μ mol g ⁻¹ , Ni = 0.476 \pm 0.017 μ mol g ⁻¹ , Pb =
208	183 \pm 18 nmol g $^{\text{-1}}$, Zn = 4.75 \pm 0.70 μmol g $^{\text{-1}}$) and 3S (Al = 2.81 \pm 0.24 mmol g $^{\text{-1}}$, Co = 0.495 \pm
209	0.037 μ mol g ⁻¹ , Fe = 0.668 ± 0.085 mmol g ⁻¹ , Mn = 7.6 ± 1.1 μ mol g ⁻¹). As in Ensenada Harbor,
210	trace metal enrichments close to the sediment-water interface are conspicuously absent for
211	practically all metals, suggesting that the cores were anoxic up to the sediment-water interface.
212	Median concentration values were consistently high in cores 2S, 3S and 4S from the inner
213	stations and low in core 1S from the harbor mouth, a reflection of the coarser grain size that
214	characterizes this last core (Figure 4). The bathymetry for this harbor (Flores-Vidal et al., 2005)
215	shows that cores 3S and 4S were located at approximately 9 m depth (i.e., the deepest inner
216	area), whereas cores 1S and 2S were collected at water depths of approximately 6 m. Judging
217	from the grain size composition, station 1S was located where the hydrodymic energy was higher.
218	Figure 4 shows that total concentrations of Cd, Co, Fe, Mn, Ni and Pb are generally lower
219	in El Sauzal Harbor than in Ensenada Harbor. These results probably cannot be ascribed to
220	differences in org-C concentrations for Ensenada (1.50 \pm 0.38 mmol g ⁻¹ ; n = 139) and El Sauzal
221	$(1.78 \pm 0.97 \text{ mmol g}^{-1}; n = 74)$ harbors, since statistically both have similar average org-C levels
222	($p = 0.142$, Mann-Whitney rank sum test). The metal concentration results may suggest that
223	Ensenada Harbor, which is approximately 30 years older than El Sauzal Harbor, tentatively has
224	had more time to accumulate trace metal contaminants in its sediments. Alternatively, the
225	sediments of both harbors may have, on average, different particulate sizes. A Mann-Whitney
226	rank sum test of the %GS indicated that the average fraction <62.5 μm from Ensenada Harbor
227	(80 ± 18%, n = 182) was significantly higher ($p \le 0.001$) than the one measured for El Sauzal (73
228	\pm 14%, n = 74). The differences in %GS distribution between the two harbors are also noticeable
229	in Figure 5, in which plots of metal concentrations versus %GS are displayed. This figure shows
230	that the sediments from El Sauzal (closed symbols) form distinctly different groups for Co, Fe, Ni,
231	Mn, Pb and Cd, which tend to cluster in the lower portion of the concentration range and
232	generally below the concentrations measured in Ensenada Harbor (open symbols). One of the
233	most important factors controlling the natural distribution of trace metals in sediments is the

234 variation in grain size (Cauwet, 1987; Grant, 1990; Ergin, 1995; Lin et al., 2002). It is widely 235 accepted that metals are enriched in the fine (<62.5 μ m) fraction of sediments (Gibbs, 1977; 236 Helmke et al., 1977; de Groot et al., 1982; Förstner, 1982; Cauwet, 1987; Horowitz et al., 1989; 237 Zhang et al., 2002) and that metals exhibit positive linear correlations with the fine-grained sizes 238 (de Groot et al., 1982; Ackermann et al., 1983; Ergin, 1995; Meador et al., 1998; Charlesworth 239 and Service, 2000; Zhang et al., 2002). Although grain size analysis is not suitable for 240 differentiating natural baseline versus anthropogenic enrichments of trace elements, such an 241 analysis may help to account for the natural diluting effect of sand.

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243 3.2. Trace metals and grain size

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The size-class distribution in sediments is intimately related to the hydrodynamics of the 245 246 systems. Hence, comprehension of this process in combination with coastal topography can help 247 to understand the behavior of sediments in terms of flocculation, adsorption or precipitation. 248 Based on textural parameters, Pérez-Higuera and Chee-Barragán (1984) inferred the existence 249 of a southward dominant transport in the littoral zone of Todos Santos Bay. In the case of Ensenada Harbor, the water flow across the mouth produced by the tide effect is considered to 250 251 be the main hydrodynamic forcing in the movement of particulate material, sediments and 252 pollutants (Coronado-Méndez et al., 2003). The high sand contents (and low metal 253 concentrations) are just a reflection of the high level of hydrodynamic energy at both harbor 254 mouths.

Figure 5 suggests that for sediments showing similar %GS, the metal concentrations measured in Ensenada and El Sauzal harbors were higher than those measured at the Point Loma Ocean Outfall (PLOO) discharge point and its surroundings, off San Diego Bay (City of San Diego, 2005). This is surprising, considering that sediments from the PLOO area have been found to be contaminated with a number of trace metals (Sb, As, Ba, Be, Cd, Cr, Cu, Fe, Pb, Mn, Hg, Ni; City of San Diego, 2005). Furthermore, data reported for the Southern California Bight (SCB; Schiff, 2000) show that average Cu, Ni and Zn values for the whole SCB (3,420.3 km²)

262 follow the same trends as the ones observed for Ensenada and El Sauzal harbors (Figures 263 5c,f,h). Only Cd and Pb (Figures 5a, g) exhibited a differential behavior, with average SCB 264 concentrations well below the ones observed for Ensenada and El Sauzal sediments (Co, Fe and 265 Mn were not measured by Schiff, 2000). Similar results (Figure 5) were obtained for the average 266 values of Santa Monica Bay (SMB, 457.4 km²), outfalls of publicly owned treatment works (POTWs, 292.8 km²) and stormwater areas (STW, 81.1 km²), the latter located within 3 km of the 267 268 11 largest river and creek mouths draining directly into the SCB (Schiff, 2000). Pristine 269 sediments from Strangford and Carlingford loughs (Charlesworth and Service, 2000), as well as 270 from anthropogenically impacted regions of Mexico (Mazatlan Harbor: contaminated with oil, 271 sewage and industrial discharges; Osuna-López et al., 1986) and Europe (Belfast Lough: 272 domestic and industrial inputs from the city of Belfast; Charlesworth and Service, 2000) follow 273 trends similar to the ones observed for Ensenada and El Sauzal sediments (Figure 5). Only Ni 274 and, to a certain extent. Zn in the Belfast samples show higher concentrations than those 275 measured for sediments with similar %GS. Hence, apparently the grain size:metal relationship is 276 not a useful indicator of trace metal enrichment: sediments with similar %GS, but far more 277 contaminated by sewage effluents and industrial inputs (e.g., San Diego-PLOO, Los Angeles-278 SMB, Belfast) than our study area have metal concentrations that are generally below or at the same level as the ones measured in Ensenada and El Sauzal harbors (Figure 5). 279

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The concentrations of org-C in Ensenada Harbor showed considerable variation with depth (except core 2E, Figure 2k), but without any noticeable trend. In contrast, El Sauzal Harbor showed no substantial changes with depth, except core 2S. This core had the highest concentrations of the four cores, probably because it was collected where docked fishing boats wash their decks, discharging in the process blood and organic wastes into the water and underlying sediment. The concentrations of org-C in core 2S were especially high between 8 and 13 cm (Figure 3k), probably reflecting past high inputs of organic wastes, whose degraded

²⁸¹ *3.3. Organic carbon*

290 remains are represented by the 10-13 cm sediment depth, consisting almost entirely of fish 291 scales. Additionally, this core showed anomalously high org-C concentrations with respect to the 292 %GS µm (Figure 5i), suggesting that anthropogenic contributions are considerable in the west 293 corner of the harbor (core 2S, Figure 1). There was a significant correlation between %GS and 294 org-C for both El Sauzal (r = 0.54, p ≤ 0.001 , n = 74) and Ensenada (r = 0.71, p ≤ 0.001 , n = 139) 295 harbors (Table 3); however, the correlation for El Sauzal increased substantially when core 2S 296 was eliminated (r = 0.84, $p \le 0.001$, n = 61; data not shown), further supporting the contention of 297 the anthropogenic origin of the sedimentary org-C.

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299 *3.4. Correlations between the different sediment components*

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301 Results of the PCA analysis (Table 4) show that two components explained 79.9% and 302 81.4% of the total variance for Ensenada and El Sauzal, respectively. In the case of Ensenada, a 303 firmly related group (PC1) was formed by AI, Cd, Co, Fe, Mn, Ni, Pb and %GS (loadings >0.63), 304 while the second group (PC2) was formed by Cu, Zn and org-C (loadings >0.75). For El Sauzal 305 Harbor, Cd, Cu, Zn and org-C (loadings >0.89) constituted the first group (PC1), and Al, Co, Fe, 306 Mn and %GS the second (PC2; loadings >0.76). The strong correlation among Cu, Zn, Cd and 307 org-C found for both harbors reflects the complex nature of organic matter as well as the 308 importance of the role played by the diagenetic reactions driven by its degradation, which, in turn, 309 regulates the behavior of these reactive metals. The strong association among Al, Fe, Mn, Co 310 and %GS suggests that the concentrations of these metals are strongly controlled by the 311 sediment grain size of these two Mexican harbors. The PCA results are essentially in agreement 312 with the Pearson correlation coefficient matrix (Table 3), which probably indicates the association 313 of Cd, Cu and Zn with organic matter and the existence of a strong lithogenous component (Al, 314 Fe and %GS) controlling the distribution of trace metals in sediments. 315

316 3.5. Enrichment factors

317

318 Another approach by which trace element enrichments can be evaluated consists of the 319 geochemical normalization of metal concentrations with a conservative element. This approach 320 is based on the assumption that there are certain elements that represent proxies for the clay 321 mineral concentration (Kersten and Smedes, 2002). Geochemical normalization has several 322 advantages. For example, it compensates for the mineralogical as well as the natural grain size 323 variability of trace element concentrations in sediment (Loring, 1991; Birch, 2003). Aluminum has 324 been widely used as a normalizing agent for evaluating metal concentrations in estuarine and 325 coastal sediments (e.g., Goldberg et al., 1979; Schropp et al., 1990; Summers et al., 1996; 326 Weisberg et al., 2004). Some of the main advantages of using AI as a geochemical normalizer are its high natural concentration (second most abundant metal in the Earth's crust), minimal 327 328 anthropogenic contamination, it is a structural element of clays, and the metal to AI proportions in 329 the crust are relatively constant (Schropp et al., 1990; Summers et al., 1996). Although Fe has 330 also been used to normalize metal concentrations (e.g., Schiff and Weisberg, 1999), Al is a better 331 agent because it is more strongly associated with the aluminosilicate matrix (one of the most 332 important metal-bearing phases of sediments). In contrast, Fe can be significantly affected by 333 diagenetic processes (redistribution, anthropogenically induced formation of Fe sulfides and Fe 334 oxides; Summers et al., 1996), which can distort total Fe concentrations (Loring, 1991).

335 Metal/aluminum (Me/Al) ratios for both Ensenada and El Sauzal harbors increase in the 336 following order: Cd/AI < Pb/AI < Ni/AI ≈ Co/AI ≈ Cu/AI < Zn/AI < Mn/AI << Fe/AI (Figure 6). The 337 Me/AI ratios of Pb, Co, Ni and Fe were significantly higher for Ensenada Harbor relative to El 338 Sauzal Harbor, whereas the ratios for Cd, Mn, Zn and Cu were statistically equivalent for both 339 harbors. Hence, geochemical normalization suggests that metal concentrations in sediments 340 from El Sauzal show some similarities with those from Ensenada Harbor, but that the latter 341 sediments are more enriched in some elements. Proximity to a larger urban settlement (City of 342 Ensenada) may explain the higher Pb/AI ratio for Ensenada; however, explaining the higher 343 values for the other three metal ratios is more difficult. It can be hypothesized that the presence 344 of more intense reducing conditions and greater flushing times (due to its smaller size) in El 345 Sauzal Harbor may explain the deficiency in Fe, Pb, Co, Ni and, to a certain extent, Mn. Fishing

346 boats are routinely cleaned inside this harbor and blood and organic waste end up accumulating 347 in the sediments. For example, the last 6 cm (7-13 cm interval) of core 2S were almost entirely 348 made up of fish scales, an indirect indication of the magnitude of organic matter contribution to 349 the sediments. All this waste could conceivably drive the sediment redox conditions toward a 350 reducing condition, which is probably not so intense in Ensenada Harbor due to the absence of 351 this type of wastes. Fe and Mn are dynamic participants in the redox cycle and their more active 352 reductive dissolution in the El Sauzal sediments with subsequent diffusion to the water column. 353 coupled with a rapid flushing out of this harbor may explain the differences between Fe/Al and 354 Mn/Al in El Sauzal and Ensenada sediments. However, since the reaction of Mn with oxygen is 355 several orders of magnitude slower than that for reduced Fe (Jørgensen and Boudreau, 2001). 356 Mn will be preferentially lost to the water column due to its slower oxidation kinetics of 357 precipitation. Cobalt enrichment could be associated with primary productivity since this element 358 is associated with vitamin B12 (cobalamin), an essential component for the growth of many 359 phytoplankton species (Okbamichael and Sañudo-Wilhelmy, 2004). Furthermore, high 360 concentrations of dissolved B12 are generally found in shallow harbors and bays with limited 361 flushing (Okbamichael and Sañudo-Wilhelmy, 2004), characteristics that should favor the 362 accumulation of Co in Ensenada Harbor.

363 It is generally accepted that for the calculation of metal enrichments it is preferable to use
364 regional pre-industrial background concentrations. However, since no such information was
365 available for the study area, the values of Turekian and Wedepohl (1961) and Li and
366 Schoonmaker (2005) for the average composition of shale were used instead. The enrichment
367 factor (EF_{Me}) was simply calculated as:

368

369
$$EF_{Me} = \left(\frac{(Me/AI)_{sample}}{(Me/AI)_{shale}}\right)$$
(1)

370

371 If $EF_{Me} > 1$, the metal concentration in the sample is enriched relative to the average shale values 372 and, maybe, anthropogenically impacted. If $EF_{Me} < 1$, then the metal is impoverished relative to

373 the average shale values and, probably, subjected to a diagenetic process that reduces its 374 concentration in the sample. Finally, if $EF_{Me} = 1$, then the metal concentration has the same 375 value as the reference value (normalized with Al). The results of this exercise are shown in 376 Figure 7, where each core from Ensenada and El Sauzal harbors was plotted against the 377 corresponding EF_{Me} value. Values ranged from 0.41 (EF_{Cu}, core 5E) to 13 (EF_{Cu}, core 1E), and 378 from 0.36 (EF_{Ni}, core 4S) to 12 (EF_{Cd}, core 2S). Based on the EF_{Me} values, the elements were 379 arbitrarily divided into four groups: (1) those highly enriched ($EF_{Me} > 2.5$) in the harbor sediments, 380 represented solely by Cd (average EF_{Me} value of 10 ± 3); (2) those slightly enriched (1.5 > EF_{Me} > 381 2.5), represented by Zn, Co and Pb (average EF_{Me} values of 1.7 ± 0.9, 2.1 ± 0.9 and 2.3 ± 0.5, 382 respectively); (3) those neither enriched nor impoverished (0.6 > EF_{Me} > 1.5), represented 383 (excluding core 1E) by Fe and Cu (average EF_{Me} values of 0.9 ± 0.2 and 1.2 ± 0.6, respectively); 384 and (4) those slightly impoverished ($EF_{Me} < 0.6$), represented by Mn and Ni (average EF_{Me} values 385 of 0.50 ± 0.06 and 0.6 ± 0.3 , respectively).

386 The high EF_{Me} values of Cd may be a consequence of the Cd-enriched plankton (10 to 20 387 ppm; Martin and Broenkow, 1975) found in Baja California waters, which probably transfer this 388 element to the sediments after their death and subsequent sedimentation. Additional Cd 389 enrichment in the sediments can be produced by upwelling processes (Segovia-Zavala et al., 390 1998; Muñoz-Barbosa et al., 2004), which are common off the northwest coast of Baja California 391 during spring and summer (Alvarez-Borrego and Alvarez-Borrego, 1982). The metals showing 392 slight enrichment were those generally associated with anthropogenic contamination (Pb, Zn), or 393 probably related to primary productivity and its relationship with vitamin B12, as was previously 394 discussed for Co. However, it is surprising that Cu was among the metals neither enriched nor 395 impoverished, since granular copper mine tailings were routinely employed to sandblast ships in 396 a dry dock facility located adjacent to Ensenada Harbor. It is highly probable that the high 397 enrichment value ($EF_{Cu} = 13$) obtained for core 1E (Figure 7) was due to the use of this 398 sandblasting tailing. Impoverishment of Mn and Ni can be the result of the slow kinetics of 399 precipitation of the former (Jørgensen and Boudreau, 2001), and the known association of the 400 latter with Mn oxides (Murray, 1975; Sclater et al., 1976; Balistrieri and Murray, 1986; Tessier et

al., 1996; Kay et al., 2001; Trivedi and Axe, 2001), at least for the case of Ensenada Harbor,
where a significant correlation was found between these two elements (Tables 3 and 4). Hence,
Ni is probably released into the interstitial water upon reductive dissolution of the Mn oxides and,
most likely, diffuses towards the sediment-water interface and is lost to the water column together
with the dissolved Mn.

406 The average ratios of the Ensenada:El Sauzal enrichment factors (defined as 407 $EF_{Me(Ensenada)}/EF_{Me(El Sauzal)}$ ranged from 0.63 ± 0.48 for Zn to 2.9 ± 12.4 for Cu (Table 5). 408 According to these results, only Zn and Cd concentrations were higher in El Sauzal Harbor, 409 whereas the rest of the metals were more concentrated in Ensenada Harbor, especially Cu with 410 a ratio of 2.9 ± 12.4 . However, if the anomalously high EF_{Me} results for some of the cores are 411 removed from the calculation of the average ratio (e.g., Cd and Ni without core 0E and Cu without 412 core 1E), then the values are very close to unity for most of the metals (Table 5). Since a value 413 of 1.0 would indicate that total metal concentrations in the sediments of both harbors are equal, 414 these results suggest that Zn is the only metal whose concentrations are generally higher in El 415 Sauzal, while Co is higher in Ensenada $(0.63 \pm 0.48 \text{ and } 1.7 \pm 0.8, \text{ respectively; Table 5})$. 416 Furthermore, the EF_{Me} values calculated from reported concentrations of pristine (Strangford and 417 Carlingford loughs; Charlesworth and Service, 2000) and anthropogenically impacted areas 418 (Belfast Lough and PLOO; Charlesworth and Service, 2000; City of San Diego, 2005) were 419 approximately equal or generally higher for Cu, Fe, Mn, Ni, Pb and Zn than the values measured for Ensenada and El Sauzal harbors (Figure 7). 420

421

422 3.6. Pollution Load Index

423

The pollution load index (PLI) for a given core or zone is calculated from the contamination factors (CF_{Me} = concentration in sediment/base value for that metal) of each of its constituent samples, according to the following equation (Tomlinson et al., 1980):

427

428
$$PLI = (CF_{Cd} \times CF_{Co} \times CF_{Cu} \times CF_{Ni} \times CF_{Pb} \times CF_{Zn})^{1/6}$$

(2)

429	
430	where the metal base value represents its average concentration in shale (Turekian and
431	Wedepohl, 1961; Li and Schoonmaker, 2005). According to Tomlinson et al. (1980), PLI values
432	of zero, one, or larger than one suggest absence of baseline pollutants, presence of them, or
433	progressive deterioration of sediment quality, respectively. Figure 8 shows that for Ensenada
434	Harbor, the core closer to the value of baseline pollutants (1.24) was 5E (at the harbor mouth),
435	whereas the innermost core 1E had the highest PLI value (4.29). The other cores showed values
436	between 1.97 and 2.89, with an overall value for the zone of 2.47 ("Ens" black bar in Figure 8).
437	The PLI values calculated for EI Sauzal Harbor suggest that this zone is appreciably less
438	impacted by metal pollutants than Ensenada, ranging from 1.01 (core 1S at the harbor mouth) to
439	1.99 (core 2S), with a zonal value of 1.51 ("Sauz" black bar in Figure 8), equivalent to 61% of the
440	overall value for Ensenada Harbor. Hence, as a rule, the cores collected at both harbor mouths
441	had the lowest PLI values, whereas the inner cores had the highest values.
442	
443	3.7. Geoaccumulation index
444	
445	The geoaccumulation index (I_{geo}) , introduced by Müller (1981), has been used to
446	quantitatively measure metal pollution in aquatic sediments (e.g., Santos Bermejo et al., 2003),
447	based on a pollution intensity classification (I_{geo} class) that consists of 7 grades or classes (0 to 6,
448	Table 6), the highest grade reflecting a 100-fold enrichment above baseline values. The I_{geo} can
449	be calculated using the equation (Müller, 1979):
450	\mathcal{C}
451	$I_{geo} = \log_2 \frac{[Me]_{studied area}}{1.5 [Me]_{baseline}} $ (3)
452	
453	where [Me] _{baseline} represents the metal concentration in average shales taken from Turekian and
454	Wedepohl (1961) and Li and Schoonmaker (2005), with a "1.5" factor included because of
455	possible variations in the background data due to lithogenic effects (Salomons and Förstner,

1984). Results from this exercise indicated that 79.3% (Ensenada Harbor) and 86.1% (El Sauzal

456

457 Harbor) of the elements belonged to I_{geo} classes 0 and 1 (unpolluted to moderately polluted), with 458 only 6.3% (Ensenada Harbor only) positioned in I_{qeo} class 4 (0% for El Sauzal Harbor; Table 6). 459 The only element that consistently ranked high in the I_{aeo} classes (3 or 4), for both harbors and all 460 cores (except 0E), was Cd (Table 6). It is clear that Ensenada has more metals in the "polluted" 461 I_{deo} classes (2 to 6; Table 6) than El Sauzal (20.6% vs. 13.9%, respectively). None of the 462 analyzed elements were positioned in the last two, more polluted classes (5 and 6). 463 464 3.8. Sediment quality guidelines 465 466 Comparison of our total concentrations with the NOAA's sediment quality guidelines 467 (Long et al., 1995) can be used to evaluate the possible biological consequences of the levels of 468 metals in Ensenada and El Sauzal harbors. Long et al. (1995) matched biological and chemical data compiled from numerous modeling, laboratory, and field studies performed in marine and 469 470 estuarine sediments. Using these data, effects range-low (ERL: lower 10th percentile of the 471 effects data) and range-median (ERM: the median, or 50th percentile, of the effects data) guideline values were determined for a number of metals (Table 7: guideline values available only 472 473 for Cd, Cu, Ni, Pb and Zn). These two guideline values (ERL and ERM) delineate three 474 concentration ranges (<ERL, ERL-ERM, >ERM) for a particular metal. Hence, concentrations 475 below the ERL value will represent a minimal-effects range, a range intended to estimate 476 conditions in which biological effects would be rarely observed. Concentrations equal to and 477 above the ERL, but below the ERM, will represent a possible-effects range within which effects 478 would occasionally occur. Finally, concentrations equivalent to and above the ERM value will 479 represent a probable-effects range within which effects would frequently occur (Long et al., 1995). 480 Table 7 shows that there is a considerable difference between the two harbors in the allocation of 481 Cu, Ni and Pb levels among the different concentrations ranges. In El Sauzal Harbor (n = 74), Cu 482 was distributed between the <ERL (29.7%) and ERL-ERM (70.3%) ranges, whereas in Ensenada 483 Harbor (n = 147), it occurred between the $\langle \text{ERL} (10.4\%), \text{ERL-ERM} (75.3\%)$ and $\rangle \text{ERM} (14.3\%)$

ranges (Table 7). In El Sauzal, Ni was also distributed between the <ERL and ERL-ERM ranges
(17.6 and 82.4%, respectively), whereas in Ensenada it was between the ERL-ERM and >ERM
ranges (47.8 and 52.2%, respectively) (Table 7). Finally, Pb was distributed only between the
first two concentration ranges in both harbors, though the proportion of this metal in the <ERL
range was only 14.3% in Ensenada Harbor, but 98.6% in El Sauzal Harbor (Table 7). The rest of
the metals (Cd and Zn) showed similar behaviors in their distribution ranges (Table 7).

490

491 4. Conclusions

492

493 Despite the high total metal concentrations measured in the sediments of Ensenada and 494 El Sauzal harbors, their Al-normalized values generally correspond to what would be expected 495 from their aluminosilicate content. Results from EF_{Me}, PLI, I_{geo} and effects range (Figures 7 and 496 8, and Tables 6 and 7, respectively) clearly demonstrate that the inner parts of both harbors 497 contain higher levels of metals relative to the outer parts (with cores 1E and 2S generally the 498 most enriched), and that Ensenada is the more trace metal-enriched harbor of the two. In terms 499 of particular elements, Cd ranked among the most enriched in both harbors (e.g., Figure 7, 500 Tables 6 and 7); however, Cu and Ni in Ensenada and, to a much lesser extent, Zn in both 501 harbors, are the metals that can potentially cause adverse effects to local living resources. 502 Overall, PLI values for both harbors were reasonable (≤2.47), with an important percentage (79.3-503 86.1%) of the analyzed metals belonging to I_{geo} classes 0 and 1 (unpolluted to moderatelly 504 polluted). Hence, despite the apparently high total trace element concentrations measured in 505 both Mexican harbors, the levels of most of the metals are not greatly enriched in these 506 sediments and do not represent a threat to the local biota, a consideration that is of the utmost 507 importance when contamination issues are at stake.

508

509 Acknowledgments

510

511	We thank Rafae	el Blanco-Betancourt,	who collected all core	s for this study	and two anonymous
		,			

- 512 reviewers, whose insightful comments greatly improved the content of the manuscript. This
- 513 research was partially supported by grants from Fondo Sectorial SEMARNAT-CONACYT
- 514 (SEMARNAT-2002-C01-0016) and Fondo CONACYT-BC (SINVE 001-EC) to M.A. Huerta-Diaz.
- 515
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- 517 References
- 518
- 519 Ackermann, F., Bergmann, H., Schleichert, U., 1983. Monitoring of heavy metals in coastal and
- 520 estuarine sediments A question of grain size: <20 μm versus <60 μm. Environmental
 521 Technology Letters, 4, 317-328.
- 522 Alvarez-Borrego, J., Alvarez-Borrego, S., 1982. Temporal and spatial variability of temperature in 523 two coastal lagoons. CalCOFI Reports, 23, 188-198.
- Alvarez-Sánchez, L.G., Hernández, R., Durazo, R., 1988. Drift patterns of lagrangian tracers in
 Todos Santos Bay. Ciencias Marinas, 14, 135-162.
- 526 Angulo, E., 1996. The Tomlinson Pollution Load Index applied to heavy metal, 'Mussel-Watch'
- 527 data: a useful index to assess coastal pollution. Science of the Total Environnment, 187, 19-528 56.
- 529 Balistrieri, L.S., Murray, J.W., 1986. The surface chemistry of sediments from the Panama Basin:
- 530 The influence of Mn oxides on metal adsorption. Geochimica et Cosmochimica Acta, 50,531 2235-2243.
- 532 Birch, G.F., 2003. A test of normalization methods for marine sediment, including a new post-
- 533 extraction normalization (PEN) technique. Hydrobiologia, 492, 5-13.
- 534 Blake, A.C., Chadwick, D.B., Zirino, A., Rivera-Duarte, I., 2004. Spatial and temporal variations
- 535 in copper speciation in San Diego Bay. Estuaries, 27, 437–447.
- 536 Carignan, R., Tessier, A., 1988. The co-diagenesis of sulfur and iron in acid lake sediments of
- 537 southwestern Québec. Geochimica et Cosmochimica Acta, 52, 1179-1188.

- 538 Cauwet G., 1987. Influence of sedimentological features on the distribution of trace metals in
- 539 marine sediments. Marine Chemistry, 22, 221-234.
- 540 Carreón-Martínez, L.B., Huerta-Diaz, M.A., Nava-López, C., Siqueiros-Valencia, A., 2001.
- 541 Mercury and silver concentrations in sediments from the Port of Ensenada, Baja California,
- 542 Mexico. Marine Pollution Bulletin, 42, 415-418.
- 543 Cauwet, G., 1987. Influence of sedimentological features on the distribution of trace metals in
- 544 marine sediments. Marine Chemistry, 22, 221-234.
- 545 Charlesworth, M., Service, M., 2000. An assessment of metal contamination in northern Irish
- 546 coastal sediments. Proceedings of the Royal Irish Academy, 100B, 1-12.
- 547 City of San Diego, 2005. Annual Receiving Waters Monitoring Report for the Point Loma Ocean
- 548 Outfall, 2004. City of San Diego Ocean Monitoring Program, Metropolitan Wastewater
- 549 Department, Environmental Monitoring and Technical Services Division, San Diego, CA., 157
- 550 p.
- 551 Coronado-Méndez, C.A., Ramírez-Aguilar, M.A., Czitrom Baus, S., 2003. Actualización de la
- 552 batimetría del puerto de Ensenada, B.C. México. Cruise PE110502 of May 2002, Technical

553 Report 1-31 A from CICESE, 26 p.

- de Groot, A., Zshuppe, K., Salomons, W., 1982. Standardization of methods of analysis for
 heavy metals in sediments. Hydrobiologia, 92, 689-695.
- .
- 556 Ergin, M., 1995. Effects of fine grain size on distribution of Mn in shallow and deep water Black
- 557 Sea sediments: A comparison between oxic and anoxic depositional environments. Geo558 Marine Letters, 15, 51-58.
- 559 Flores-Vidal X., Ramírez-Aguilar I. and Czitrom-Baus S., 2005. Numerical simulation of the
- 560 hydrodynamics of a port and effect of a wave-driven seawater pump. Ciencias Marinas, 31,561 57-77.
- Förstner, U., 1982. Accumulative phases for heavy metals in limnic sediments. Hydrobiologia,
 91-92, 269-284.
- Gibbs, R.J., 1977. Transport phases of transition metals in the Amazon and Yukon rivers.
- 565 Geological Society of America Bulletin, 88, 661-679.

- 566 Goldberg, E.D., Griffin, J.J., Hodge, V., Koide, M., Windom, H., 1979. Pollution history of the
- 567 Savannah River Estuary. Environmental Science and Technology, 13, 588-594.
- 568 Gonzalez-Morales, A.T., Gaxiola Castro, G., 1991. Daily variation of physico-chemical
- 569 characteristics, biomass, and phytoplankton primary production in an upwelling costal zone of
- 570 Baja California. Ciencias Marinas, 3, 21-37.
- 571 Grant, A., 1990. Multivariate statistical analyses of sediment geochemistry. Marine Pollution
- 572 Bulletin, 21, 297-299.
- 573 Helmke, P.A., Koons, R.D., Schomberg, P.J., Iskandar, I.K., 1977. Determination of trace
- 574 element contamination of sediments by multielement analysis of clay-size fraction.
- 575 Environmental Science and Technology, 11, 984-989.
- 576 Horowitz, A.J., Elrick, K.A., Hooper, R.P., 1989. The prediction of aquatic sediment-associated
- 577 trace element concentrations using selected geochemical factors. Hydrological Processes, 3,
 578 347-364.
- Jørgensen, B.B., Boudreau, B.P., 2001. Diagenesis and sediment-water exchange. In:
- 580 Boudreau, B.P., Jørgensen, B.B. (Eds.). The Benthic Boundary Layer. Oxford University

581 Press, Oxford, 211-244 pp.

- 582 Kay, J.T., Conklin, M.H., Fuller, C.C., O'Day, P.A., 2001. Processes of nickel and cobalt uptake
- 583 by a manganese oxide forming sediment in Pinal Creek, Globe Mining District, Arizona.
- 584 Environmental Science and Technology, 35, 4719-4725.
- 585 Kersten, M., Smedes, F., 2002. Normalization procedures for sediment contaminants in spatial
 586 and temporal trend monitoring. Journal of Environmental Monitoring, 4, 109–115.
- 587 Li, Y.-H., Schoonmaker, J.E., 2005. Chemical composition and mineralogy of marine sediments.
- 588 In: Mackenzie, F.T. (Ed.) Sediments, Diagenesis, and Sedimentary Rocks. Vol 7, First
- 589 Edition: Treatise on Geochemistry, Holland, H.D., Turekian, K.K., Executive Directors, 1-35
- 590 pp.
- Lin, S., Hsieh, I-J., Huang, K.-M., Wang, C.-H., 2002. Influence of the Yangtze River and grain
- 592 size on the spatial variations of heavy metals and organic carbon in the East China Sea
- 593 continental shelf sediments. Chemical Geology, 182, 377-394

- 594 Long, E.R., Macdonald, D.D., Smith, S.L., Calder F.D., 1995. Incidence of adverse biological
- 595 effects within ranges of chemical concentrations in marine and estuarine sediments.596 Environmental Management, 19, 81-97.
- 597 Loring, D.H., 1991. Normalization of heavy-metal data from estuarine and coastal sediments.
- 598 Journal of Marine Science, 48, 101-115.
- 599 Martin, J.H., Broenkow, W.W., 1975. Cadmium in plankton: Elevated concentrations off Baja
- 600 California. Science, 190, 884-885.
- Meador, J.P., Robisch, P.A., Clark, R.C., Jr., Ernest, D.W., 1998. Elements in fish and sediment
- from the Pacific coast of the United States: Results from the national benthic surveillance
- 603 project. Marine Pollution Bulletin, 37, 56-66.
- Müller G., 1979. Schwermetalle in den sedimenten des Rheins Veranderungen seit. Umschan
 79, 778-783.
- 606 Muñoz-Barbosa, A., Gutiérrez-Galindo, E.A., Segovia-Zavala, J.A., Delgadillo-Hinojosa, F.,
- Sandoval-Salazar, G., 2004. Trace metal enrichment in surficial sediments of the northwest
 coast of Baja California, Mexico. Marine Pollution Bulletin, 48, 596-603.
- 609 Murray, J.W., 1975. The interaction of metal ions at the manganese dioxide-solution interface.
- 610 Geochimica et Cosmochimica Acta, 39, 505-519.
- 611 Nayar, S., Miller, D.J., Hunt, A., Goh, B.P.L., Chou, L.M., 2007. Environmental effects of
- 612 dredging on sediment nutrients, carbon and granulometry in a tropical estuary.
- 613 Environmental Monitoring and Assessment, 127, 1-13.
- Okbamichael, M., Sañudo-Wilhelmy, S.A., 2004. A new method for the determination of Vitamin
 B₁₂ in seawater. Analytica Chimica Acta, 517, 33-38.
- 616 Osuna-López, J.I., Ortega-Romero, P., Páez-Osuna, F., 1986. Cd, Co, Cr, Cu, Fe, Ni, Pb y Zn
- en los sedimentos del Puerto y antepuerto de Mazatlán. Ciencias Marinas, 12, 35-45.
- 618 Pavía, E.G., 2004. La influencia de los ciclones tropicales en la lluvia del noroeste de Baja
- 619 California y suroeste de California. Geos, 24, 69-72.

- 620 Peña-Manjarrez, J.L., Helenes, J., Gaxiola-Castro, G., Orellana-Cepeda, E., 2005. Dinoflagellate
- 621 cysts and bloom events at Todos Santos Bay, Baja California, México, 1999–2000.

622 Continental Shelf Research, 25, 1375-1393.

- Pérez -Higuera, I., Chee-Barragán, A., 1984. Sediment transport in Todos Santos Bay. Ciencias
 Marinas, 10, 31-52.
- 625 Salomons, W., Förstner, U., 1984. Metals in the Hydrocycle (pp. 80-81). Berlin: Springer-Verlag.
- 626 Santos Bermejo, J.C., Beltrán, R., Gómez Ariza, J.L., 2003. Spatial variations of heavy metals
- 627 contamination in sediments from Odiel river (Southwest Spain). Environmental International,628 29, 69-77.
- 629 Santschi, P.H., Nixon, S., Pilson, M., 1984. Accumulation of sediments, trace metals (Pb, Cu)
- and hydrocarbons in Narragansett Bay, Rhode Island. Estuarine, Coastal and Shelf Science,19, 427-450.
- Schiff, K.C., 2000. Sediment chemistry on the mainland shelf of the Southern California Bight.
 Marine Pollution Bulletin, 40, 268-276.
- 634 Schiff, K., Weisberg, S.B., 1999. Iron as a reference element for determining trace metal
- 635 enrichment in California coastal shelf sediments. Marine Environmental Research, 48, 161-
- 636 176.
- 637 Schropp, S.J., Lewis, F.G., Windom, H.L., Ryan, J.D., Calder, F.D., Burney, L.C., 1990.
- 638 Interpretation of metal concentrations in estuarine sediments of Florida using aluminum as a
- 639 reference element. Estuaries, 22, 227-235.
- Sclater, F.R., Boyle, E., Edmond, J.M., 1976. On the marine geochemistry of nickel. Earth and
 Planetary Science Letters, 31, 119-128.
- 642 Segovia-Zavala, J.A., Delgadillo-Hinojosa, F., Alvarez-Borrego, S., 1998. Cadmium in the coastal
- 643 upwelling area adjacent to the California Mexico Border. Estuarine, Coastal and Shelf
- 644 Science, 46, 475-481.
- 645 Summers, J.K., Wade, T.L., Engle, V.D., Malaeb, Z.A., 1996. Normalization of metal
- concentrations in estuarine sediments from the Gulf of Mexico. Estuaries, 19, 581-594.

- 647 Tessier, A., Fortin, D., Belzile, N., DeVitre, R.R., Leppard, G.G., 1996. Metal sorption to
- 648 diagenetic iron and manganese oxyhydroxides and associated organic matter: Narrowing the
- 649 gap between field and laboratory measurements. Geochimica et Cosmochimica Acta, 60,
- 650387-404.
- Tomlinson, D.L., Wilson, J.G., Harris, C.R., Jeffrey D.W., 1980. Problems in the assessment of
- heavy-metal levels in estuaries and the formation of a pollution index. Helgoland Marine
- 653 Research, 33, 566-575.
- Trivedi, P., Axe, L., 2001. Predicting divalent metal sorption to hydrous Al, Fe, and Mn oxides.
- Environmental Science and Technology, 35, 1779-1784.
- 556 Turekian, K.K., Wedepohl, K.H., 1961. Distribution of the elements in some major units of the
- 657 Earth's crust. Geological Society of America Bulletin, 72, 175-192.
- Weisberg, S.B., Wilson, H.T., Heimbuch, D.G., Windom, H.L., Summers, J.K., 2004. Comparison
- of sediment metal: aluminum relationships between the eastern and gulf coasts of the United
- 660 States. Environmental Monitoring and Assessment, 61, 373-385.
- Zhang, C., Wang, L., Li, G., Dong, S., Yang, J., Wang, X., 2002. Grain size effect on multi-
- 662 element concentrations in sediments from the intertidal flats of Bohai Bay, China. Applied
- 663 Geochemistry, 17, 59-68.

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665 Figure legends

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667

Figure 1. Location of Ensenada and El Sauzal harbors. The circulation pattern shown in (b) was
taken from Peña-Manjarrez et al. (2005).

670

Figure 2. Profiles of total trace metal concentrations (on a dry weight basis), percentage of grain
 size <62.5 μm (%GS) and organic carbon (org-C) for each of the seven cores collected in

673 Ensenada Harbor. Gray symbols represent concentrations below the detection limit.

674

Figure 3. Profiles of total trace metal concentrations (on a dry weight basis), percentage of grain
size <62.5 μm (%GS) and organic carbon (org-C) for each of the four cores collected in El Sauzal
Harbor.

678

Figure 4. Box plots of total trace metal concentrations, organic carbon (org-C) and percentage of grain size <62.5 μ m (%GS) for each one of the sediment cores collected in Ensenada and El Sauzal harbors (open and gray boxes, respectively). Cd concentrations below the detection limit in core 0E were not considered for the calculation of basic statistics. In this figure, the boundary of the box closest to zero indicates the 25th percentile, a line within the box marks the median, and the boundary of the box farthest from zero indicates the 75th percentile. Bars above and below the box indicate the 90th and 10th percentiles, respectively.

686

Figure 5. Plots of percentage of grain size <62.5 μ m (%GS) vs. metal concentration and organic carbon (org-C) for sediment cores collected in Ensenada and El Sauzal harbors. Concentrations (± one standard deviation) reported for Point Loma Ocean Outfall (PLOO; City of San Diego,

690 2005), Mazatlan Harbor (MH; Osuna-López et al., 1986), northern Irish coastal sediments (NICS;

691 Charlesworth and Service, 2000), Southern California Bight (SCB), Santa Monica Bay (SMB),

outfalls of publicly owned treatment works (POTWs) and stormwater areas (STW) (Schiff, 2000),

- are also included for comparison purposes
- 694

Figure 6. Box plots and average values (± one standard deviation) of total metal (Me)

696 concentrations and Me/AI ratios for sediments collected in Ensenada and EI Sauzal harbors,

697 Metals were arranged in increasing order of average values of Ensenada Harbor. Note the log

698 scales in the y-axes.

699

700	Figure 7. Enrichment factors, $EF_{Me} = (Me/AI)_{sample}/(Me/AI)_{shale}$, for each one of the sediment
701	cores collected in Ensenada (E) and El Sauzal (S) harbors (black and gray bars, respectively).
702	The average composition of shale was taken from Turekian and Wedepohl (1961) and Li and
703	Schoonmaker (2005). Concentrations (± one standard deviation) reported for Point Loma Ocean
704	Outfall (PLOO; City of San Diego, 2005), Belfast Lough (B), and Strangford and Carlingford
705	loughs (S&C) (Charlesworth and Service, 2000) are also included for comparison purposes.
706	Dashed lines represent the value of 1.0 [Me/AI) _{sample} = $(Me/AI)_{shale}$]. Error bars for E and S
707	harbors were calculated by error propagation analysis.
708	
709	Figure 8. Pollution load index for each one of the sediment cores (gray bars) and for Ensenada
710	(Ens) and El Sauzal (Sauz) areas (black bars). PLI values of zero, one (dashed line), or larger

than one suggest absence of baseline pollutant, presence of them, or progressive deterioration of

712 sediment quality, respectively.



Figure 1



Figure 2a

Figure 2b





Figure 3a







Figure 4



Figure 5







Figure 7



Figure 8

Table 1. Measured (n = 21) and certified concentrations (\pm two standard deviations), and percentages of recovery of Beaufort Chemistry Standard Sediment (BCSS-1; National Research Council of Canada).

Meas Element concer (µg		Certified concentration (µg g ⁻¹)	Percentage of recovery (%)
Al Cd Cu Fe Mn Ni Pb Zn	$(6.5 \pm 1.0) \times 10^{4} \\ 0.23 \pm 0.03 \\ 10.8 \pm 0.2 \\ 19.2 \pm 0.7 \\ (3.23 \pm 0.04) \times 10^{4} \\ 220 \pm 10 \\ 52.5 \pm 0.3 \\ 22.2 \pm 1.8 \\ 112 \pm 5$	$(6.3 \pm 0.2) \times 10^4 \\ 0.25 \pm 0.04 \\ 11.4 \pm 2.1 \\ 18.5 \pm 2.7 \\ (3.3 \pm 0.1) \times 10^4 \\ 229 \pm 15 \\ 55.3 \pm 3.6 \\ 22.7 \pm 3.4 \\ 119 \pm 12$	103 90 95 104 98 96 95 98 94
		MAN	

	Al (mmol g ⁻¹)	Cd (nmol g ⁻¹)	Co (µmol g⁻¹)	Cu (µmol g⁻¹)	Fe (mmol g ⁻¹)	Mn (µmol g⁻¹)	Ni (µmol g⁻¹)	Pb (nmol g ⁻¹)	Zn (µmol g⁻¹)
Avg 0E (43) Range 0E	$\begin{array}{c} 3.49 \pm 0.47 \\ 2.84 \text{-} 4.56 \end{array}$	6.8 ± 4.2 <7.6-18	1.79 ± 0.45 1.11-2.81	$\begin{array}{c} 1.32 \pm 0.19 \\ 0.92 1.59 \end{array}$	1.31 ± 0.16 1.03-1.68	9.3 ± 1.3 6.5-12	2.136 ± 0.046 2.069-2.293	231 ± 62 135-423	2.50 ± 0.78 1.34-4.22
Avg 1E (32) Range 1E	$\begin{array}{c} 4.53 \pm 0.28 \\ 3.99\text{-}5.07 \end{array}$	$\begin{array}{c} 41.8 \pm 2.4 \\ 37.7 \text{-} 46.6 \end{array}$	1.169 ± 0.085 1.052-1.411	15 ± 40 4-229	1.33 ± 0.16 1.11-1.82	12.7 ± 2.5 10-20	$\begin{array}{c} 0.948 \pm 0.044 \\ 0.8921.10 \end{array}$	$\begin{array}{c} 366\pm38\\ 295\text{-}429 \end{array}$	$\begin{array}{c} 5.8\pm4.5\\ 3.530\end{array}$
Avg 2E (27) Range 2E	$\begin{array}{c} 4.94 \pm 0.25 \\ 4.43 \text{-} 5.52 \end{array}$	$\begin{array}{c} 50.6 \pm 2.5 \\ 45.5\text{-}55.2 \end{array}$	$\begin{array}{c} 1.006 \pm 0.043 \\ 0.918 \text{-} 1.083 \end{array}$	1.09 ± 0.21 0.91-1.58	1.200 ± 0.046 1.09-1.26	12.61 ± 0.35 11.87-13.15	$\begin{array}{c} 0.907 \pm 0.040 \\ 0.832 1.02 \end{array}$	$\begin{array}{c} 353\pm32\\ 307\text{-}490 \end{array}$	$\begin{array}{c} 3.36 \pm 0.38 \\ 2.99 4.18 \end{array}$
Avg 3E (36) Range 3E	$\begin{array}{c} 3.36 \pm 0.22 \\ 3.00\text{-}3.99 \end{array}$	$\begin{array}{c} 35.3 \pm 3.4 \\ 30.2 \text{-} 41.4 \end{array}$	$\begin{array}{c} 0.733 \pm 0.059 \\ 0.637 \text{-} 0.919 \end{array}$	$\begin{array}{c} 0.79 \pm 0.12 \\ 0.53 \text{-} 0.94 \end{array}$	0.931 ± 0.089 0.742-0.126	9.28 ± 0.38 8.27-10.0	0.593 ± 0.054 0.518-0.744	$\begin{array}{c} 288 \pm 26 \\ 233\text{-}337 \end{array}$	1.73 ± 0.23 1.33-2.17
Avg 4E (21) Range 4E	$\begin{array}{c} 3.46 \pm 0.23 \\ 2.93 \hbox{-} 3.81 \end{array}$	$\begin{array}{c} 38.9 \pm 3.7 \\ 34.8 \text{-} 47.8 \end{array}$	$\begin{array}{c} 0.823 \pm 0.074 \\ 0.695 \text{-} 0.963 \end{array}$	1.51 ± 0.90 0.37-3.01	0.900 ± 0.089 0.739-1.084	11.2 ± 1.9 9.2-16.4	$\begin{array}{c} 0.645 \pm 0.055 \\ 0.560 \hbox{-} 0.750 \end{array}$	$\begin{array}{c} 285\pm40\\ 223\text{-}336 \end{array}$	1.92 ± 0.58 1.13-3.03
Avg 5E (14) Range 5E	$\begin{array}{c} \textbf{2.64} \pm \textbf{0.20} \\ \textbf{2.26-2.90} \end{array}$	25.0 ± 2.4 19.6-29.3	0.518 ± 0.028 0.470-0.574	$\begin{array}{c} 0.262 \pm 0.087 \\ 0.148 \text{-} 0.418 \end{array}$	0.56 ± 0.14 0.42-0.97	$\begin{array}{c} 7.12 \pm 0.62 \\ 6.14 \hbox{-} 8.19 \end{array}$	$\begin{array}{c} 0.492 \pm 0.042 \\ 0.428 \text{-} 0.600 \end{array}$	257 ± 16 241-298	0.89 ± 0.14 0.70-1.2
Avg 6E (9) Range 6E	$\begin{array}{c} 3.84 \pm 0.18 \\ 3.52 4.07 \end{array}$	$\begin{array}{c} 37.8 \pm 2.0 \\ 35.1 \text{-} 42.0 \end{array}$	$\begin{array}{c} 0.918 \pm 0.043 \\ 0.870 \hbox{-} 0.999 \end{array}$	1.76 ± 0.13 1.57-1.99	1.175 ± 0.011 1.16-1.19	$\begin{array}{c} 8.41 \pm 0.41 \\ 7.73 \text{-} 9.07 \end{array}$	$\begin{array}{c} 0.671 \pm 0.035 \\ 0.637 \text{-} 0.742 \end{array}$	$\begin{array}{c} 371 \pm 14 \\ 350\text{-}400 \end{array}$	2.72 ± 0.10 2.57-2.85
Overall Avg (182): Overall range:	3.81 ± 0.75	32 ± 16 <7.6-55	1.10 ± 0.48	3.5 ± 17.2 0.15-229	1.11 ± 0.26 0.419-1.82	10.4 ± 2.3 6.14-19.9	1.07 ± 0.62	299 ± 66 135-490	2.9 ± 2.4 0.70-30
		40.0 + 4.0	0.240 + 0.045	0.054 + 0.004	0.402 + 0.020			100 5 1 0 5	1 00 + 0 10
Range 1S	2.33 ± 0.08 2.18-2.44	19.2 ± 1.2 17.7-21.6	0.348 ± 0.015 0.327 - 0.382	0.251 ± 0.031 0.210-0.320	0.483 ± 0.036 0.423-0.536	5.21 ± 0.37 4.58-5.68	0.363 ± 0.032 0.327-0.411	129.5 ± 8.5 112-146	1.33 ± 0.19 1.08-1.62
Avg 2S (13) Range 2S	2.61 ± 0.12 2.30-2.81	28.2 ± 1.5 25.9-30.6	$\begin{array}{c} 0.431 \pm 0.020 \\ 0.390 0.463 \end{array}$	1.24 ± 0.13 1.09-1.54	$\begin{array}{c} 0.570 \pm 0.026 \\ 0.527 0.622 \end{array}$	$\begin{array}{c} 6.06 \pm 0.24 \\ 5.75 \hbox{-} 6.53 \end{array}$	$\begin{array}{c} 0.476 \pm 0.017 \\ 0.453 \text{-} 0.504 \end{array}$	183 ± 18 156-231	$\begin{array}{c} 4.75 \pm 0.70 \\ 4.07\text{-}6.66 \end{array}$
Avg 3S (29) Range 3S	$\begin{array}{c} \textbf{2.81} \pm \textbf{0.24} \\ \textbf{2.49-3.44} \end{array}$	23.8 ± 3.5 18.4-32.1	$\begin{array}{c} 0.495 \pm 0.037 \\ 0.409 \text{-} 0.592 \end{array}$	0.72 ± 0.23 0.41-1.2	$\begin{array}{c} 0.668 \pm 0.085 \\ 0.561 \text{-} 0.848 \end{array}$	7.6 ± 1.1 6.2-9.6	$\begin{array}{c} 0.448 \pm 0.041 \\ 0.361 \text{-} 0.507 \end{array}$	158 ± 10 139-181	2.92 ± 0.64 1.90-4.75
Avg 4S (19) Range 4S	$\begin{array}{c} 2.60 \pm 0.14 \\ 2.36 \text{-} 2.90 \end{array}$	$\begin{array}{c} 23.8 \pm 3.4 \\ 19.128.8 \end{array}$	$\begin{array}{c} 0.444 \pm 0.044 \\ 0.380 0.549 \end{array}$	0.75 ± 0.17 0.52-1.0	$\begin{array}{c} 0.565 \pm 0.069 \\ 0.485 \text{-} 0.729 \end{array}$	$\begin{array}{l} \textbf{6.55} \pm \textbf{0.59} \\ \textbf{5.91-8.04} \end{array}$	$\begin{array}{c} 0.370 \pm 0.031 \\ 0.310 \hbox{-} 0.437 \end{array}$	154 ± 82 133-167	$\begin{array}{c} 3.15 \pm 0.75 \\ 2.30 \hbox{-} 5.23 \end{array}$
Overall Avg (74): Overall	2.64 ± 0.24	23.8 ± 3.9	0.445 ± 0.062	0.74 ± 0.34	0.592 ± 0.095	6.6 ± 1.2	0.418 ± 0.057	157 ± 19	3.0 ± 1.2
range:	2.18-3.44	17.7-32.1	0.327-0.592	0.21-1.5	0.423-0.848	4.6-9.6	0.310-0.507	112-231	1.1-6.7

Table 2. Average total concentrations (Avg) and range of values for sediments from Ensenada (cores 0E to 6E) and El Sauzal (cores 1S to 4S) harbors. Numbers in parentheses refer to the number of samples. Cd concentrations below the detection limit in core 0E were not considered for the calculation of basic statistics.

Table 3. Pearson correlation matrix for the sedimentary geochemical data from Ensenada and El Sauzal harbors. Core 0E was not included in the analysis because the percentage of grain size <62.5 µm (%GS) and organic carbon (org-C) were not available. Cu and Zn concentrations at 5.5 cm depth in core 1E were also not considered because they represent outliers.

	AI	Cd	Со	Cu	Fe	Mn	Ni	Pb	Zn	%GS
				EI S	Sauzal Ha	rbor (n =	70)			
Cd Co Cu Fe Mn Ni Pb Zn %GS Org-C	0.14 0.85*** 0.20 0.90*** 0.51*** 0.53*** 0.43*** 0.21 0.76*** 0.32	0.19 0.87*** 0.08 -0.10 0.53*** 0.52*** 0.84*** 0.45 0.69***	0.26 0.89*** 0.72*** 0.57*** 0.47*** 0.25 0.83*** 0.31	0.17 -0.04 0.63*** 0.74*** 0.94*** 0.57*** 0.84***	0.56*** 0.57*** 0.43*** 0.14 0.77*** 0.28	0.24 0.21 0.00 0.46*** 0.07	0.61*** 0.57*** 0.68*** 0.70***	0.70*** 0.59*** 0.77***	0.55*** 0.84***	0.54***
				Ense	enada Ha	rbor (n =	120)			
Cd Co Cu Fe Mn Ni Pb Zn %GS Org-C	0.85*** 0.84*** 0.31*** 0.84*** 0.66*** 0.92*** 0.72*** 0.75*** 0.63*** 0.49***	0.71*** 0.16 0.74*** 0.63*** 0.77*** 0.60*** 0.57*** 0.61*** 0.37***	0.56*** 0.88*** 0.77*** 0.93*** 0.73*** 0.89*** 0.75*** 0.69***	0.46*** 0.44*** 0.50*** 0.36*** 0.74*** 0.35*** 0.50***	0.60*** 0.86*** 0.79*** 0.82*** 0.79*** 0.74***	0.73*** 0.43*** 0.67*** 0.53*** 0.30***	0.73*** 0.87*** 0.62*** 0.58***	0.72*** 0.63*** 0.69***	0.62*** 0.72***	0.71***
9%GS 0.63*** 0.61*** 0.75*** 0.35*** 0.79*** 0.53*** 0.62*** 0.63*** 0.62*** 0.62*** 0.72*** 0.71*** org-C 0.49*** 0.37*** 0.69*** 0.50*** 0.74*** 0.30*** 0.58*** 0.69*** 0.72*** 0.71*** ***p≤0.001										

Table 4. Results of PCA applied to sedimentary geochemical data gathered at Ensenada and El Sauzal harbors. Core 0E was not included in this analysis because the percentage of grain size <62.5 μ m (%GS) and organic carbon (org-C) were not available. Cu and Zn concentrations at 5.5 cm depth in core 1E were also not considered because they represent outliers.

Harbor	Component	Eigenvalues	Explained variance (%)	Accumulated Variance (%)
Ensenada Ensenada	1 2	5.40 3.39	49.1 30.8	49.1 79.9
El Sauzal El Sauzal	1 2	4.80 4.16	43.6 37.8	43.6 81.4
	Com	ponent loadings	.9	<i>.</i>
Geochemical	El Sauza	al Harbor	Ensena	ida Harbor
factor	PC1	PC2	PC1	PC2
AI	0.17	0.90	0.92	0.25
Cd	0.89	-0.03	0.94	0.04
Со	0.20	0.95	0.77	0.58
Cu	0.97	0.06	0.07	0.87
Fe	0.13	0.94	0.77	0.53
Mn	-0.11	0.76	0.71	0.28
Ni	0.66	0.50	0.83	0.45
Pb	0.75	0.36	0.64	0.50
Zn	0.95	0.05	0.58	0.75
%GS	0.52	0.76	0.63	0.48
	0.89	0.18	0.35	0.78

2

Table 5. Average ratios of Ensenada (n = 182) to El Sauzal (n = 74) enrichment factors ($EF_{Me(Ensenada)}/EF_{Me(El Sauzal)}$) for the different elements analyzed. The associated errors were calculated by error propagation analysis. For the special case of Cd, Ni and Cu, results are presented with and without some of the cores that showed anomalously elevated EF_{Me} values.

Element	EF _{Me(Ensenada})/EF _{Me(El Sauzal)} (molar ratio)	Propagated error
Zn	0.63	0.48
Cd	0.91	0.43
Cd (without core 0E)	1.1	0.2
Mn	1.1	0.3
Fe	1.3	0.3
Pb	1.3	0.3
Со	1.7	0.8
Ni	1.8	1.2
NI (without core 0E)	1.2	0.2
	2.9	12.4
	- C	
	2460	

Table 6. Metals distributed according to their geoaccumulation index (I_{geo}). The I_{geo} classes (*in parentheses*) consist of 7 grades of pollution intensity (Santos Bermejo et al., 2003): 0 = unpolluted, 1 = unpolluted to moderately polluted, 2 = moderately polluted, 3 = moderately to strongly polluted, 4 = strongly polluted, 5 = strongly to very strongly polluted, and 6 = very strongly polluted.

Coro	I _{geo} (I _{geo} class)							
Core	< 0 (<i>0</i>)	0-1 (<i>1</i>)	1-2 (<i>2</i>)	2-3 (<i>3</i>)	3-4 (<i>4</i>)	4-5 (5)	>5 (6)	
0E	Al,Mn	Fe,Zn,Cu,Ni,Cd,Pb	Со			Q`		
1E	Mn,Ni	AI,Fe	Co,Pb,Zn	Cu	Cd	·		
2E	Mn,Ni,Fe	Cu,AI,Zn	Co,Pb		Cd			
3E	Ni,Mn,Cu,Fe,Al,Zn	Co,Pb			Cd			
4E	Ni,Mn,Fe,Al,Zn	Cu,Co,Pb			Cd			
5E	Cu,Ni,Mn,Zn,Fe,Al	Co,Pb		Cd)			
6E	Mn,Ni,Cu,Fe,Zn,Al	Co,Pb		Cd				
CP*	47.6 [30]	31.7 [20]	9.5 [6]	4.8 [3]	6.3 [4]	0	0	
1S	Ni,Mn,Cu,Fe,Al,Zn,Co,Pb			Cd				
2S	Mn,Ni,Fe,Al,Co	Cu,Pb	Zn	Cd				
3S	Ni,Mn,Fe,Al,Cu	Co,Pb,Zn		Cd				
4S	Ni,Mn,Fe,Al,Cu,Co	Pb,Zn		Cd				
CP*	66.7 [24]	19.4 [7]	2.8 [1]	11.1 [4]	0	0	0	
OP **	54.5 [54]	27.3 [27]	7.1 [7]	7.1 [7]	4.0 [4]	0	0	

*Percentage for each I_{geo} class [number of metals]. **Overall percentage for each I_{geo} class [number of metals].

ACEPTE

Table 7. Effects range-low (ERL) and range-median (ERM) guideline values for trace metals (on a dry weight basis) and percent incidence of sediment concentration values in concentration ranges defined by the two guideline values.

Harbor	Element	ERL (µmol/g)	ERM (µmol/g)	Percent (ratios) incidence for each concentration range ^b		
		Guidelines ^ª		<erl< th=""><th>ERL-ERM</th><th>>ERM</th></erl<>	ERL-ERM	>ERM
Ensenada	Cd	0.011	0.085	19.2 (35/182)	80.8 (147/182)	0.0 (0/182)
El Sauzal	Cd	0.011	0.085	0.0 (0/74)	100 (74/74)	0.0 (0/74)
Ensenada	Cu	0.54	4.25	10.4 (19/182)	75.3 (137/182)	14.3 (26/182)
El Sauzal	Cu	0.54	4.25	29.7 (22/74)	70.3 (52/74)	0.0 (0/74)
Ensenada	Ni	0.225	1.05	0.0 (0/182)	47.8 (87/182)	52.2 (95/182)
El Sauzal	Ni	0.225	1.05	17.6 (13/74)	82.4 (61/74)	0.0 (0/74)
Ensenada	Pb	0.356	0.879	14.3 (26/182)	85.7 (156/182)	0.0 (0/182)
El Sauzal	Pb	0.356	0.879	98.6 (73/74)	1.4 (1/74)	0.0 (0/74)
Ensenada	Zn	2.29	6.27	47.3 (86/182)	51.6 (94/182)	1.1 (2/182)
El Sauzal	Zn	2.29	6.27	24.3 (18/74)	74.3 (55/74)	1.4 (1/74)

^aTaken from Long et al. (1995); ^bPercentage and number of data entries within each concentration range divided by the total number of entries for all ranges.