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Thermodynamic insights into the structural basis governing the donor substrate recognition by the human β1,4-galactosyltransferase 7

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Short Title: Substrates recognition by β1,4-GalT7

Human β1,4-galactosyltransferase7 (β1,4-GalT7) is involved in the biosynthesis of the tetrasaccharide linker protein region (GlcA β 1-3Gal β 1-4Xyl β 1) of proteoglycans, by catalysing the transfer of galactose from the uridine 5'-diphosphogalactose, to xylose residue. This reaction is rate-limiting in the glycosaminoglycan biosynthesis. In the present study, we established a large scale production system of the β1.4-GalT7 fused with the maltose-binding protein to study substrate recognition. Calorimetric binding studies showed that the binding of the donor substrate UDP-Gal largely promoted binding of the acceptor substrate. To identify the structural basis governing substrate recognition, we used a fragment-based approach involving the artificial breaking down of donor substrate into smaller fragments and the characterisation of their respective binding to the enzyme by isothermal titration calorimetry. The β -phosphate, and more moderately the α -phosphate, largely contributed to the binding energy. However the uridine moiety was found to be essential for an optimal positioning of the donor substrate within the binding site. Unexpectedly, the contribution of the galactose moiety in substrate recognition was found negligible. Indeed, UDP-Gal, but also various to β 1,4-GalT7. **UDP-sugars** could bind Surprisingly, in contrast other galactosyltransferases, soluble \$1,4-GalT7 was able to transfer Glc, Xyl, and more moderately GlcA and GlcNAc to acceptor sugars while UDP-Man and UDP-GalNAc were not substrates.

Key words: β1,4-galactosyltransferase; substrate recognition; substrate specificity; isothermal titration calorimetry; kinetics; glycosaminoglycan

Abbreviations used: PG, proteoglycan; GAG, glycosaminoglycan; ITC, isothermal titration calorimetry; GalT, galactosyltransferase; MBP, maltose-binding protein; Gal, galactose; Glc, glucose; Man, mannose; GlcA, glucuronic acid; GlcNAc, *N*-acetyl glucosamine; GalNAC, *N*-acetyl galactosamine; Xyl, xylose; pNP-Xyl, 4-Nitrophenyl-β-D-xylopyranoside; 4MU-Xyl, 4-Methylumbelliferyl-β-D-xylopyranoside

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INTRODUCTION

Proteoglycans (PGs) represent one of the largest and most complex molecular structures in animals, plants and others. They are composed of polysaccharide glycosaminoglycan (GAG) chains covalently bound to a protein core. The different types of PGs are distinguished by their various core proteins and the composition of their polysaccharide chains (1). Their biological roles are closely related to the presence of these GAG chains, which contain binding sites for various growth factors, cytokines, morphogens, enzymes and other signaling molecules (2-5). These glycoproteins actively partipate in numerous physiological processes at the cell-tissue-organ interface (6). It is becoming clear that they play key roles in cell growth and development (7), angiogenesis (8), viral invasion (9) and anticoagulation (10). Therefore, GAG synthesis has received increasing attention in the recent past. GAG synthesis is initiated by the glycosyltransferase-dependent formation of a common carbohydrate sequence, GlcAβ1-3Galβ1-3Galβ1-4Xylβ1, bound to specific serine residues in the core protein (11). From this linker tetrasaccharide, the sugar chains are extended by addition of two alternate monosaccharides, an aminosugar and GlcA. The reaction leads to the formation of heparin and heparan-sulfate when the aminosugar is N-acetyl-glucosamine and to chondroitin-sulfate and dermatan-sulfate when it is N-acetyl-galactosamine. Because of the wide distribution of GAG chains and their common synthetic pathway, defects in their biosynthesis should therefore cause serious abnormalities in a wide variety of tissues and organs. Aberrant and insufficient GAG substitution of PGs was observed in patients affected with the progeroid type of Ehlers Danlos syndrome (12, 13). On the contrary, many pathological processes, especially those which are accompanied by an accumulation of extracellular matrix components and connective tissue remodelling, such as fibrotic and sclerotic processes, are characterised by increased PG biosynthesis and deposition (14-16). Human β1,4-galactosyltransferase 7 (β1,4-GalT7) (EC 2.4.1.133, family GT7 in CAZy database, www.cazv.org) is a Golgi resident type II membrane protein that catalyses the Mn²⁺-dependent transfer of galactose from UDP-Gal to the nonreducing end of the xylose residue of the common tetrasaccharide linker region of GAGs (17). β1,4-GalT7 is a member of the β1,4-GalT-related family of enzymes (18, 19). These glycosyl transferase family members are classified as "inverting" enzymes because they transfer galactose, from the donor UDP- α -D-galactose to an acceptor substrate, generating a $\beta 1 \rightarrow 4$ linkage product (20). β1,4-GalT7 was shown to be a rate-limiting enzyme in the biosynthesis of GAG chains (21, 22) and defects in β1,4-GalT7, due to mutations (23-25), are involved in severe pathological processes. Therefore, specific inhibitors or activators of this enzyme might represent a good way to modulate GAG biosynthetic pathway. Indeed, β-D-xyloside analogs are able to prime GAGs, because they serve as substrates for β1,4-GalT7 (26).

The characterisation of potential ligand-binding sites on a target protein is a prerequisite for successful design of molecules able to inhibit or activate GAGs biosynthesis. In this study, we have used a fragment-based approach that involved the breaking-down of the donor substrate into smaller fragments, to probe for interactions at the substrate-binding site. To assess the ability of substrate fragments to bind to β 1,4-GalT7, we used a combination of kinetic, inhibition and isothermal titration calorimetry (ITC) binding studies.

MATERIALS AND METHODS

Cloning and expression of human β1,4-galactosyltransferase 7 in HeLa cells

The coding sequence corresponding to the human β 1,4-GalT7 was amplified by RT-PCR from human brain total RNA using specific primers corresponding to nucleotides 1 to 27 (forward primer) and 961 to 984 (reverse primer) of the reported human β 1,4-GalT7 cDNA (GenBank accession number NM007255). Forward and reverse primers included KpnI and



EcoRI restriction sites at their extremities, respectively. The reverse primer included a 30 nucleotide sequence coding the amino acid sequence EQKLISEEDL, allowing addition of the myc-tag to the *C*-terminal extremity of the recombinant protein. The purified PCR product amplified with AccuPrime Taq DNA Polymerase (Invitrogen, Cergy Pontoise, France) was then subcloned into pcDNA3.1 (Invitrogen, Cergy Pontoise, France) and sequenced on both strands. The sequence was 100 % identical to that previously reported.

To express membrane β 1,4-GalT7, *HeLa* cells were cultured in complete Dubelcco's modified Eagle's medium (Gibco, Cergy Pontoise, France) supplemented with glucose (4.5 g/L), fetal bovine serum (10 %), glutamine (1mM) and penicillin-streptomycin (1 %). The expression plasmid pcDNA- β 1,4-GalT7myc (10 μ g) was transfected into *HeLa* cells on 100 mm plates using Exgen 500 (Euromedex, Souffelweyersheim, France) according to the manufacturer's recommendations. Two days after transfection, cells were harvested and resuspended in Hepes-sucrose buffer (5 mM Hepes, 0.25 M sucrose, pH 7.4) and briefly sonicated (1 x 5 s). Protein concentration of the resulting homogenate was determined by the method of Bradford (27), and β 1,4-GalT7myc protein expression was evaluated by western blot analysis using anti-myc antibodies (Invitrogen, Cergy Pontoise, France) according to the manufacturer's recommendations. The resulting homogenate was used for β 1,4-galactosyltransferase activity assay.

Cloning and expression of a soluble form of the β1,4-galactosyltransferase 7

 β 1,4-GalT7 was expressed as a fused-protein with the maltose-binding protein (MBP). A truncated form of β 1,4-GalT7, lacking the first *N*-terminal 81 amino acids of the β 1,4-GalT7 was amplified by PCR using specific primers corresponding to nucleotides 244 to 264 (forward primer) and 961 to 984 (reverse primer) including XbaI and HindIII restriction sites, respectively. The PCR product was then subcloned into the pMAL-c2x vector (New England Biolabs, Saint Quentin Yvelines, France) and sequenced on both strands. The sequence was 100 % identical to that previously reported.

To express the soluble MBP- β 1,4-GalT7 fused-protein, *Escherichia coli* BL21(DE3) cells (Invitrogen, Cergy Pontoise, France) were transformed with pMAL- β 1,4-GalT7 plasmid and grown at 37°C in Terrific Broth (Sigma, Saint Quentin Fallavier, France) supplemented with ampicillin (100 μ g/mL) to an absorbance from 0.6 to 0.8 at 600 nm. Isopropyl β -thio galactopyranoside (Euromedex, Souffelweyersheim, France) was then added to a final concentration of 0.2 mM, and the cells were further incubated overnight at 37°C. Cells were harvested by centrifugation at 10,000 g for 10 min and resuspended in MOPS buffer (20 mM) containing 150 mM NaCl and 100 μ g/mL lysozyme. After incubation for 30 min at 4°C, the resuspended cells were disrupted by sonication (i.e., five 30-s bursts separated by periods of cooling). The solution was centrifuged at 30,000 g for 30 min at 4°C. The resulting supernatant was loaded onto an amylose column (New England Biolabs, Saint Quentin Yvelines, France).

Purification of MBP- β1,4-GalT7 fused-protein

All purification procedures were carried out at 4°C. The resulting supernatant was applied (1 mL/min) onto an amylose column (~30 mL) equilibrated with buffer A (20 mM MOPS, pH 7.0 containing 150 mM NaCl). The column was washed with five column-volumes of buffer A and bound proteins were then eluted with maltose (20 mM) in MOPS buffer. Eluted fractions were analysed by SDS-polyacrylamide gel electrophoresis. The fractions containing the MBP- β 1,4-GalT7 fusion proteins were pooled and dialysed extensively against buffer A and then conserved at -80°C. The concentration of the purified protein MBP- β 1,4-GalT7 was determined by using the 2100 Bioanalyser (Agilent, Massy, France) according to the manufacturer's recommendations.



Isothermal Titration Calorimetry

ITC experiments were carried out at 30°C in MOPS (20 mM) buffer, pH 7.0 containing NaCl (150 mM) with or without MnCl₂ (5 mM) using a VP-ITC MicroCalorimeter (Micro Cal, Inc.). Ligand solutions were diluted in the final dialysis used for the protein to achieve the expected final concentration (UDP and UDP-Gal, 0.5 mM; TDP, UTP and UDP-Glc, 1 mM; UDP-GlcA, UDP-Xyl and UDP-Man, 2 mM; UDP-GlnAc and UDP-GlnAc, 5 mM; UMP, CDP, 10 mM; pNP-Xyl, 40 mM) (Sigma, Saint Quentin Fallavier, France). Substrate solution was injected into the reaction cell containing the degassed purified protein MBP-β1,4-GalT7 (1.4 mL, 30 μM). Thirty 4 to 10 μL-injections, lasting 8 to 20 s respectively, were done at 240 s intervals. During the titration, the enzyme was stirred continuously at 260 rpm. Control runs were performed by titrating ligands into buffer. Thermodynamic parameters were obtained from individual experiments performed with three different batches of purified MBP-\beta1,4-GalT7 and for each batch, determination was performed in duplicate. Thermodynamic parameters n (stoichiometry), ΔH (enthalpy change) and K_a (association constant) were obtained by nonlinear least-squares fitting of experimental data using the single-site model of the Origin 7.0 software package provided with the instrument. The dissociation constant (K_d) , the free energy of binding (ΔG) and entropy change (ΔS) were deduced from the K_a and ΔH

The thermodynamic parameters for donor substrate analogs that have very low affinity were determined by the displacement titration method (28). Titration of the enzyme solution (30 μ M) was performed with UDP (0.5 mM) in the presence of various substrate analogs (uridine, uracil, ribose, ADP and GDP, 5 to 10 mM). Thermodynamic parameters ΔH and K_a for low binding substrate analogs were obtained by nonlinear least-squares fitting of experimental data using the competitive binding model of the Origin 7.0 software package using the thermodynamic parameters obtained for UDP in an independent experiment on the same batch of enzyme.

β1,4-galactosyltransferase activity

4-Methylumbelliferyl-β-D-xylo-pyranoside (4MU-Xyl) and 4-Nitrophenyl-β-D-xylo-pyranoside (pNP-Xyl) (Sigma, Saint Quentin Fallavier, France) were used as acceptor substrate for the measurement of activity of recombinant human β 1,4-GalT7. Standard reactions were performed in 100 mM cacodylate buffer (pH 7.0) containing 10 mM MnCl₂, 0.6 g.L⁻¹ of total proteins containing membrane β 1,4-GalT7 or 0.2 to 6 μM of purified truncated MBP-β1,4-GalT7, and various concentrations of 4MU-Xyl or pNP-Xyl and UDP-Gal. After incubation at 37°C for 30 min, reaction products were analysed by high performance liquid chromatography (HPLC) on a reverse phase C-18 column (4.6 x 150 mm, 4 μm, Waters, Milford, MA) at a detection wavelength of 320 nm. The mobile phase was composed of 15% (v/v) acetonitrile, 0.02% (v/v) trifluoroacetic acid in water and used at a flow rate of 1 mL/min.

Kinetic parameters for 4MU-Xyl or pNP-Xyl and UDP-Gal or other donor substrates were obtained from at least two separate experiments with three different batches of membrane or purified enzymes after the addition of increasing amounts of acceptor or donor substrates (0.02 to 10 mM of 4MU-Xyl or pNP-Xyl in the presence of 10 mM of UDP-Gal; 0.025 to 20 mM of donor substrates in the presence of 10 mM of 4MU-Xyl) to the standard reaction. The estimation of the kinetic parameters was achieved by the incubation of at least 10 different concentrations of substrate. Incubations were performed in duplicate. Data were fitted by nonlinear regression to the Michaelis-Menten model using the curve fitter program of Sigmaplot 9.0Tm.



In order to evaluate the inhibiting effect of UDP analogs on β 1,4-galactosyltransferase activity, inhibitors were added to the standard reaction. The estimation of the inhibition constant (K_i) was achieved by the incubation of at least 8 different concentrations of donor substrate in the presence of three different concentrations of inhibitor. Data were fitted by non-linear regression to the competitive inhibition model using Sigmaplot 9.0Tm software.

Reaction products characterisation

For reaction products characterisation, 4 mg of 4MU-Xyl were glycosylated in the presence of saturating concentration of UDP-Gal, UDP-Glc or UDP-Xyl with 8 to 100 nmol of purified MBP-β1,4-GalT7, using HPLC as described previously to monitor reaction progress. The reaction products (i.e., 4MU-Xyl-Gal, 4MU-Xyl-Glc and 4MU-Xyl-Xyl) were purifed by HPLC on a reverse phase C-18 column (10 x 250 mm, 10 μm, Merck, Darmstadt, Germany) at a detection wavelength of 320 nm. The reaction products and 4MU-Xyl were deuterium exchanged by repeated lyophilisation from D₂O and then dissolved in 0.5 mL of D₂O (99,8%) for NMR analysis. Spectra were recorded at 298K using a Bruker Avance 300 NMR spectrometer (TOCSY, COSY, 2D-NOESY) and a Bruker DRX 600 spectrometer (one-dimensional ¹H and HSQC). All ¹H chemical shifts were referenced to methyl signals of 3-(trimethylsilyl)propionic-2,2,3,3-d4 acid sodium salt as an internal standard. ¹³C chemical shifts were referenced to acetone as an external reference (30 ppm). Water (HOD) residual resonance suppression in the COSY, TOCSY and NOESY experiments was achieved using presaturation during relaxation delay. Data processing was performed using XWIN-NMR software.

RESULTS

Expression, purification and characterization of human β1,4-GalT7

To easily produce and purify pure recombinant human β1,4-GalT7 on a large scale, we expressed in Escherichia coli a soluble form of the enzyme fused with the MBP. Such a soluble form of β1,4-GalT7 was designed by the deletion of the transmembrane and stem regions, corresponding to the first 81 N-terminal amino acids of the β1,4-GalT7. The resulting fused MBP-β1,4-GalT7 protein was 16-fold purified onto amylose resin (Table 1). The relative mobility of the purified MBP-β1,4-GalT7 protein in SDS-PAGE (Figure 1) indicated a homogeneous protein with a molecular weight of approximately 70 kDa, consistent with the theorical molecular mass of the truncated \(\beta 1.4\)-GalT7, lacking the first \(N\)-terminal 81 amino acids and fused with the maltose-binding protein. The purified MBP-\beta1,4-GalT7 was found to be active. The formation of the product, using 4MU-Xyl or pNP-Xyl as acceptor substrate, was a linear function of protein concentration. To determine whether the truncation and/or the fusion with the MBP affected the kinetic properties of the MBP-β1,4-GalT7, we determined its apparent affinity constant $K_{\rm m}$ toward the donor (i.e., UDP-Gal) and the acceptor substrates (i.e., 4MU-Xyl and pNP-Xyl) compared to those of the membrane β1,4-GalT7 expressed in HeLa cells. K_m values of the MBP-β1,4-GalT7 toward UDP-Gal and 4MU-Xyl or pNP-Xyl were 0.23 mM, 0.27 mM and 1.27 mM, respectively and were similar to those obtained for the membrane β 1,4-GalT7 (Table 2). The K_{cat} values of the MBP- β 1,4-GalT7 with 4MU-Xyl or pNP-Xyl as acceptor substrate were 68 min⁻¹ and 91 min⁻¹, respectively (Table 2).

Substrate binding to the human β1,4-GalT7

To get insight into the mechanism of β 1,4-GalT7, the binding of substrates analogs, UDP and pNP-Xyl, was investigated by ITC. The pNP-Xyl was used as acceptor substrate because of the limited solubility of the 4MU-Xyl when high concentrations are necessary. Prior to this study, we verified the absence of binding of these compounds on the MBP part of the fused protein in the conditions used for titrations with the fused protein. Therefore we expressed and



purified the MBP protein and analysed the signal detected by ITC after injections of UDP or pNP-Xyl. No calorimetric signal was detected neither with UDP nor with pNP-Xyl.

We first explored the binding of UDP to $\beta1,4$ -GalT7. Figure 2A shows a representative calorimetric profile of UDP binding to the enzyme with and without Mn²⁺. Binding of UDP to the enzyme was not detected in the absence of Mn²⁺. On the other hand, when Mn²⁺ (5 mM) was present, an exothermic binding profile was obtained. From this, the thermodynamic parameters were determined (Figure 2A). A dissociation constant of 9 μ M was measured. The binding of the donor substrate to MBP- $\beta1,4$ -GalT7 was driven by an enthalpy change of 23 \pm 5 kcal.mol⁻¹, which is partially countered by an unfavourable entropy change (Table 3). Moreover, whatever the batch of enzyme, the stoichiometry of UDP binding was between 0.4 to 0.6. This result prompted us to determine the oligomeric state of the protein. Gel filtration analysis under native conditions showed that MBP- $\beta1,4$ -GalT7 eluted with a molecular mass of about 160 kDa (curves not shown). All these data support a dimeric state of the enzyme with only one subunit active within the dimer. Further analyses are needed to explain this unexpected behavior. In that context, it will be necessary to determine the oligomeric state of the enzyme *in vivo*.

The binding of the acceptor substrate analog (pNP-Xyl) to the enzyme was then analysed with and without UDP (Figure 2B). In both cases, an exothermic signal was detected. However the magnitude of the exothermic signal detected in the presence of UDP was greater than that obtained without UDP, suggesting the requirement of UDP to promote the binding of the acceptor substrate to the enzyme. The signal detected without UDP might be due to nonspecific binding. Only, the signal obtained with UDP reached saturation and generated a significant binding isotherm. The acceptor substrate analog pNP-Xyl bound with relatively weak affinities to MBP- β 1,4-GalT7. The K_d was higher than the protein concentration, resulting in c-values (c = $[Prot]_t/K_d$) lower than 1. Under low c-value conditions, both ΔG and ΔH can be measured with confidence provided that, at least, 80% saturation is achieved and that the stoichiometry n is known, as described by Turnbull and Danaras (29). In this experiment, it was possible to reach saturation at sufficiently high ligand concentration. Therefore, we used the stoichiometry value determined independently by the active site titration experiments with UDP under higher c-values conditions, for fitting the low c-value titrations. Consequently, the dissociation constant and enthalpy change of binding could be measured and the entropy of binding calculated with reasonable confidence (Figure 2B). The binding of acceptor substrate was driven by an enthalpy change (-11.8 kcal/mol) with an unfavourable entropy change pNP-Xyl was found to bind weakly to the enzyme, compared to UDP with an estimated dissociation constant of about 4.6 mM (Figure 2B). This poor binding might be explained by the fact that pNP-Xyl is not the natural ligand and is consistent with the high $K_{\rm m}$ value determined for pNP-Xyl (Table 2).

Identification of the structural determinants of the donor substrate recognition

To determine the contributions of individual chemical groups of the donor substrate to the binding affinity and enthalpic and entropic components of the free energy of binding, the thermodynamics of the interaction of MBP-β1,4-GalT7 with UDP analogs were characterised using ITC, by using the stoichiometry value observed for UDP binding (Table 3).

The smallest fragment for which direct binding could be detected was uridine. However, the binding affinity estimated was very weak ($K_d > 8$ mM), and a low heat signal prevented an accurate determination of the thermodynamic parameters. In contrast, binding of UMP and UTP was fully characterised by ITC (Table 3). Dissociation constants of 562 and 99 μ M were obtained, respectively, and were 62- and 11-fold higher than that obtained for UDP binding. No heat released or absorbed was measured with ribose or uracil when these ligands were added in the enzyme solution. Therefore, the binding of ribose or uracil and uridine to the



enzyme was further demonstrated or confirmed using competitive ITC experiments (Table 3). This indirect approach allows characterisation of weak interaction of the ligand of interest by displacement with a known high-affinity ligand. Titrations of MBP- β 1,4-GalT7 with UDP were performed in the absence or in the presence of uracil, ribose or uridine under otherwise identical conditions. The K_d and ΔH for UDP obtained from the two titrations were used to estimate the K_d and ΔH for the weak competitor ligand, by applying the displacement model described by Zhang and Zhang (30), and Sigurskjold (28). As expected, uridine weakly inhibited the binding of UDP to MBP- β 1,4-GalT7. Besides, both uracil and ribose were also able to weakly inhibit the binding of UDP to MBP- β 1,4-GalT7, but with dissociation constants higher than 8 mM (Table 3). To further investigate the requirement for the uridine moiety, other nucleotides were also tested (Table 3). TDP and CDP were found to bind weaker to the enzyme, compared to UDP, with estimated dissociation constants 11- and 94-fold higher, respectively. No binding was observed for ADP and GDP, even by using competitive ITC experiments (Table 3).

To determine the contributions of the sugar moiety of the donor substrate to the binding, the thermodynamics of the interaction of MBP- β 1,4-GalT7 with various UDP-sugars were characterised using ITC by direct titration (Table 4). First, we compared the binding of UDP and UDP-Gal to MBP- β 1,4-GalT7. The binding of UDP-Gal was also exothermic (Table 4). Dissociation constant of 8.7 μ M and enthalpy of binding of – 25.0 kcal.mol⁻¹ were obtained and both values were similar to those measured when UDP was used as ligand. Because of the low contribution of the galactose moiety on the donor substrate binding, we investigated the enzyme ability to bind other UDP-sugars (Table 4). Direct binding to MBP- β 1,4-GalT7 was observed with all the analogs tested and binding parameters are indicated in Table 4. Except for UDP-Glc, other UDP-sugars were found to bind weakly to the enzyme, compared to UDP-Gal (Table 4). Furthermore the magnitude of ΔH for the binding of all these UDP-sugars was less than that for UDP-Gal.

Kinetic studies

The ITC results prompted us to test the ability of MBP- β 1,4-GalT7 to transfer the sugar part from other naturally occurring UDP-sugars (2.5 mM) to the acceptor substrate. In this section, 4MU-Xyl was used as acceptor substrate because of the better $K_{\rm m}$ obtained for MBP- β 1,4-GalT7 toward this subtrate (i.e., 0.27 mM) compared to pNP-Xyl (i.e., 1.27 mM). Surprisingly, MBP- β 1,4-GalT7 catalysed the transfer of Glc, Xyl, GlcA and GlcNAc onto the acceptor substrate (Table 5). In contrast UDP-Man and UDP-GalNAc were not substrates. Kinetic parameters toward UDP-sugars were thus determined and are summarised in Table 5. $K_{\rm m}$ values of the purified MBP- β 1,4-GalT7 toward UDP-Glc and UDP-Xyl were 0.28 mM and 0.16 mM, respectively and were comparable to that obtained for UDP-Gal (0.23 mM). Thus, our results showed that MBP- β 1,4-GalT7 exhibited a markedly higher efficiency toward UDP-Gal compared to UDP-Glc and UDP-Xyl due to the decrease in $k_{\rm cat}$ values (respectively 58- and 109-fold lower than that obtained with UDP-Gal).

MBP- β 1,4-GalT7 was also able to catalyse the transfer of GlcA and GlcNAc onto the acceptor substrate. But the $K_{\rm m}$ values obtained for both were so high that the determination of $k_{\rm cat}$ values was not achieved.

To validate the ITC results and to investigate whether the identified ligands were inhibitors, inhibition analyses of the β 1,4-galactosyltransferase activity using UDP-analogs were carried out and when inhibitory effects were observed, K_i values were determined as indicated in *Materials and methods* (Table 6). As expected, UDP was the most potent inhibitor and inhibited the β 1,4-galactosyltransferase activity in a competitive manner with a K_i value of 0.61 mM (Table 6). No significant inhibition of the β 1,4-galactosyltransferase activity was detected in the presence of donor substrate fragments such as UMP, uridine, uracil or ribose



(data not shown). On the other hand, inhibitors such as UTP and TDP, which are structurally related to UDP, inhibited significantly the β 1,4-galactosyltransferase activity in a competitive manner. K_i values were, respectively, 1.65 mM and 1.93 mM (Table 6). The presence of the methyl-group in C5 of thymidine and the addition of one phosphate to UDP had only a negligible effect on the donor substrate binding, as indicated by K_i values. On the contrary, the exocyclic amine group of cytidine limited the inhibition potency and the K_i value was 16-fold higher than that obtained for UDP.

Although MBP- β 1,4-GalT7 was not able to catalyse the transfer of Man or GalNAc from the donor substrates (i.e., respectively, UDP-Man and UDP-GalNAc) onto the acceptor substrate, these donor substrate analogs inhibited the β 1,4-galactosyltransferase activity, but only at high concentration. However K_i values were too high to be determined (Table 6).

Structural characterisation of the reaction products

Structural characterisation of the reaction products (i.e., 4MU-Xyl-Gal, 4MU-Xyl-Glc and 4MU-Xyl-Xyl) showed that MBP-β1,4-GalT7 formed the β1-4 linkage between the Gal, Glc or Xyl from the UDP-Gal, UDP-Glc or UDP-Xyl donor substrates, respectively, and the Xyl from the 4MU-Xyl acceptor substrate. Comparison of the three reaction products and the 4MU-Xyl one-dimensional ¹H NMR spectra clearly showed an additional anomeric (H'₁) proton resonance (4.515 ppm, 4.581 ppm and 4.501 ppm respectively) with a coupling constant (${}^3J_{1',2'} = 7.9$ Hz, ${}^3J_{1',2'} = 7.9$ Hz and ${}^3J_{1',2'} = 7.7$ Hz respectively) typical of a sugar residue linked in the β-configuration (see supplementary data 1A & 2A). These results for 4MU-Xyl-Gal were in agreement with those obtained in the same conditions by Almeida et al. (17). All ¹H and ¹³C resonances were unambiguously assigned by the TOCSY, COSY and HSQC experiments (Table 7). The linkages in 4MU-Xyl-Glc and 4MU-Xyl-Xyl were established by observation of interglycosidic H'₁-H₄ correlations in the 2D-NOESY spectra (see supplementary data 1B & 2B). For these two compounds, correlations between H'1 (anomeric proton of Glc or Xyl linked to 4MU-Xyl) and protons of Xyl from 4MU-Xyl clearly indicated a NOE correlation between H'₁ and H₄, and thus a C'₁-C₄ linkage. The absence of H'₁-H₃ and H'₁-H₂ NOE correlations excluded C'₁-C₃ and C'₁-C₄ linkages. Finally the H'₁-H_{5eq} NOE correlation showed a spatial proximity of these two protons but was inconsistent with a C'₁-C₅ linkage because of the absence of hydroxyl on this carbon (the endocyclic oxygen linked to C₅ is blocked because of the 4MU aglycon). Consistent with the C'₁-C₄ linkage, the 4MU-Xyl C₄ resonance showed a substantial downfield glycosylation induced shift ($\Delta\delta$ = 7.21 ppm and 7.04 ppm for 4MU-Xyl-Glc and 4MU-Xyl-Xyl respectively) that was unique when comparing products to substrate, whereas C₃ and C₅ exhibited small upfield shift changes. All of the proton resonances of the 4MU-Xyl residue exhibited downfield glycosylation-induced shifts: H_4 was the largest ($\Delta \delta = 0.208$ ppm and 0.157 ppm for 4MU-Xyl-Glc and 4MU-Xyl-Xyl respectively). These results confirmed the linkage structure of the product as β 1,4.

DISCUSSION

PGs play an important role in numerous physiological and pathological processes. Progress in the elucidation of the PG biosynthetic pathway has emphasised the importance of glycosyltransferases. These enzymes are responsible for the biosynthesis of the GAG chains, which are largely involved in the biological roles of PGs. Specific compounds able to modulate the activity of these enzymes might represent attractive compounds able to control the GAG biosynthesis rate. Therefore, structure-function studies of glycosyltransferases appear as a crucial step for future design of potentially valuable drugs. Herein, we focus our work on the identification of the structural determinants involved in the substrate binding by the human recombinant $\beta 1,4$ -GalT7. This enzyme catalyses the rate-limiting transfer of



galactose to the xylosylated core protein (17, 31) and, indeed, defects of this enzyme result in severe phenotypes (i.e., the progeroid variant of Ehlers-Danlos syndrome) (23, 24).

To examine precisely the kinetic and physicochemical properties of β1,4-GalT7, a convenient expression system allowing the preparation of large amounts of pure recombinant enzyme has been developed. To date, few studies have reported the expression of the β1,4-GalT7 (17, 32). Almeida et al. (17) previously expressed and partially characterised a semi-purified truncated form lacking the first N-terminal 63 amino acids of the β1,4-GalT7. In our previous study (32), we expressed in yeast the β1,4-GalT7 as a membrane form, but, because no method was available to estimate the concentration of the β1,4-GalT7 in the membrane fraction, the expressed form was also partially characterised. Herein expression of a soluble β1,4-GalT7 was achieved successfully in Escherichia coli by engineering a fusion protein with MBP and the catalytic domain of β 1,4-GalT7. A similar approach has been used by Pedersen et al. (33) to express and purify Exostose Tumor Like 2 (EXTL2). Because β1,4-GalT7 fusion protein is enzymatically active and highly soluble, we decided to characterise the MBP- β1,4-GalT7 in the current study. This strategy was already used for the studies of other membrane enzymes (34), for which expression and solubility are limited. Consequently, we verified the effect of the MBP fusion on the apparent affinity (K_m) of the enzyme for UDP-Gal and the acceptor substrates 4MU-Xyl and pNP-Xyl. The $K_{\rm m}$ of the MBP- β 1,4-GalT7 toward UDP-Gal was similar to those determined for the membrane β1,4-GalT7 expressed in HeLa cells (i.e., 0.31 mM) and for the membrane β 1,4-GalT7 previously expressed in yeast (32). The $K_{\rm m}$ of the MBP-β1,4-GalT7 toward 4MU-Xyl and pNP-Xyl were also similar to those of the membrane β1,4-GalT7 expressed in *HeLa* cells.

β1,4-GalT7, as other glycosyltransferases, is a two-substrate enzyme leading to the formation of two different products in the presence of divalent cations. Therefore, kinetic analysis of such enzymes is complex. In many studies, a saturated concentration of one of the two substrates has been used to reduce the kinetic analysis of glycosyltransferases to a pseudosingle-substrate reaction. However, the data generated under these circumstances do not reflect the general properties of the enzyme. ITC is a direct method for characterising binding reactions in solution and has been largely employed in binding studies of saccharides with their partners such as lectins (35, 36) or proteins (37). In this study, ITC was used to determine whether the binding of the acceptor substrate was affected by the binding of the donor substrate. ITC results showed that the binding of the donor substrate largely promoted the binding of the acceptor substrate analog, which is fully consistent with the mechanism already described for other "inverting" glycosyltransferases (38). However, the acceptor substrate seems able to bind to the enzyme without UDP-Gal, but with a very low affinity (K_d > 15 mM, Figure 2). Since only pNP-Xyl was used as acceptor substrate in this ITC study, it is not possible to determine whether this phenomenon also occurs with the natural substrate, Xyl-core protein. However, the 3D-structure of the human β1,4-GalT1 (family GT7 in CAZy database) (39), which is related to the β1,4-GalT7 (19), indicates that the presence of the UDP-sugar is a prerequisite for acceptor substrate binding, consecutively to a donor substrateinduced conformational change. β1,4-GalT1 has two flexible loops, one short and one long, highly conserved among β1,4-GalT family members, which play crucial roles in the catalytic mechanism. Upon binding a metal ion and UDP-Gal, the loops change from an open to a closed conformation, creating the acceptor binding site. This conformational change can be induced by either low concentrations of sugar-nucleotide donor and Mn²⁺, or high concentrations of acceptor (39). This assumption has to be confirmed in the case of the β1,4-GalT7, but can be postulated by analogy with the β1,4-GalT1. This would explain why in this study we detected binding of acceptor substrate analog without UDP, because very high concentration of pNP-Xyl was necessary to reach saturation.



Although specific inhibitors of various glycosyltransferases have been primarily designed to mimic the acceptor substrates (40, 41), the elucidation of the binding order of subtrates to β1,4-GalT7 suggests that the design of a potent inhibitor or activator should incorporate structural elements of the donor substrate, as previously proposed for other glycosyltransferases (42, 43). Nevertheless structural elements of the acceptor substrate should be also subsequently considered to obtain molecules targeting selectively the \beta_1,4-GalT7. Herein we applied a fragment-based approach to probe for interactions at the donor substrate-binding site by ITC, which is particularly amenable to detect low binding (44). The differences in the enthalpy change of binding between compounds (i.e., the $\Delta\Delta H$ values) (Figure 3) provide insights into the roles of the different structural portions of the donor substrate on its recognition by the β 1,4-GalT7. Indeed, ΔH is derived from the net heats of making and breaking of non-covalent bonds. For the binding of uracil-containing substrates, sequential loss in enthalpy change of binding occurred with the systematic removal of phosphates, with the largest drop upon the removal of β -phosphate. The increase in the K_d for UMP compared to UDP arises from a major decrease of the enthalpy of binding (with value of 9.3 kcal/mol), suggesting that the β-phosphate might be the key group for donor substrate recognition. Nevertheless, we might suppose also a significant contribution of the αphosphate in the interaction because UMP is the smallest compounds for which significant direct binding or inhibition could be detected. As reported for the human β1,4-GalT1, the anionic oxygen atoms of the diphosphate group of UDP certainly interact with β1,4-GalT7 through Mn²⁺ via a DXD sequence highly conserved in all the β1,4-GalT family members (45). The major contribution of the β-phosphate in the recognition, observed in this study, might be due to its interaction with β 1,4-GalT7 through hydrogen bonds, as reported for the bovine β1,4-GalT1 (family GT7 in CAZy database) (46).

Surprisingly, despite the major contribution of the phosphate groups in the binding of the donor substrate to \$1,4-GalT7, modifications of the uridine moiety change significantly the binding affinity (Figure 3). Binding or inhibition of ADP and GDP could not be detected in this study. Replacing the pyrimidic nucleotides by puric nucleotides might induce a steric hindrance, preventing the binding of the puric analogs. On the other hand, β1,4-GalT7 has the ability to bind other pyrimidic nucleotides than UDP. Nevertheless enthalpies of binding are largely modified, particularly when CDP is used. The donor pyrimidic analogs, CDP or TDP, might interact with β1,4 GalT7 in a completely different way from UDP. Interestingly, the presence of the methyl-group in C5 of TDP does not change the binding affinity as much as the additional exocyclic amine group of cytidine. In the 3D-structure of the bovine β1,4galT1, the uracil ring is held in position by hydrophobic stacking interaction occurring between a Phe side-chain and an Arg side-chain, but also by hydrogen-bonding contacts between O2 and N3 atoms of the uracil ring and the main chain nitrogen and oxygen atoms of Arg189 residue, respectively (46). Since binding of TDP by β1,4-GalT7 is better (8-fold) than that of CDP, same kind of interactions may be supposed for \beta1,4-GalT7. Thus, design of inhibitors or activators should take into account the importance of the positioning of the donor analogs in the binding site and this positioning might be different when the uridine portion is modified, as suggested by the fact that UDP is a better inhibitor than TDP, which is better than CDP itself.

Contrary to the uridine moiety, surprisingly, the role of the sugar part seems to be negligible in the binding of the donor substrate. Indeed, when comparison was made between binding of UDP and UDP-Gal by β 1,4-GalT7, no significant difference of K_d or ΔH was detected. It is likely that the galactose part does not interact or very weakly with β 1,4-GalT7. Glycosyltransferases have been previously reported to be quite specific for the binding of the nucleotide portion while exhibiting some tolerance for different sugars (47). In this way, UDP



binds nearly as well to bovine α1,3-GalT (family GT6 in CAZy database) as the natural substrate UDP-Gal (38). Nevertheless, in the 3D-structure of the human β1,4-GalT1, the Gal O2 and O3 hydroxyl groups interact with the side chain carboxylate group of the highly conserved DXD motif and the Gal O4 axial hydroxyl group hydrogen binds with the side chain carboxyl oxygen atoms of the EDDD motif largely conserved between β1,4-GalT family members. These interactions which are not compatible with the Glc moiety of UDP-Glc, because the O4 hydroxyl group in Glc is equatorial, partially govern the strict donor substrate specificity of β1,4-GalT1 (39). Contrary to β1,4-GalT1, β1,4-GalT7 has the ability to bind other UDP-sugars than UDP-Gal, confirming the low contribution of the galactose moiety in binding and suggesting that the binding donor site has enough space to accommodate larger and more ionic substrates than UDP-Gal. Surprisingly and contrary to what was reported for the other six β1,4-GalTs (48) which have exclusive specificity for UDP-Gal, β1,4-GalT7 is also able to transfer these sugar parts, except for Man or GalNAc, generating $\beta 1 \rightarrow 4$ linkage products. The reason why the transfer of Man or GalNAc residue is not possible, while the binding is effective, might be explained by a positioning of these substrates in a non competent mode for efficient catalysis. These findings would certainly be useful for developing specific inhibitors of β1,4-GalT7.

Thermodynamic analysis of the β 1,4-GalT7 in the present study allowed a better knowledge of the binding contribution of individual fragments of the donor substrate to the β 1,4-GalT7. Such a study provides, based on their ligand efficiency, important insights into potential hot spots at the donor substrate-binding site for future design of drugs able to modulate GAG biosynthesis.

FOOTNOTES

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TABLES

Table 1 Purification of MBP-β1,4-GalT7

Purification of MBP-β1,4-GalT7 was performed onto an amylose resin from a 2L culture of recombinant *E. coli* BL21 transformed with the pMAL-β1,4-GalT7 plasmid, as described in Materials and Methods.

Purification step	Total activity (µmol.min ⁻¹)	Total protein (mg)	Specific activity (nmol.min ⁻¹ .mg ⁻¹)	Purification fold
Supernatant	55.1	950	58	1
Amylose Elution	22.3	23	970	16

Table 2 K_m values of the membrane β 1,4-GalT7 expressed in Hela cells (WT-c) and the purified MBP- β 1,4-GalT7 (WT-p) expressed in *E. coli* toward UDP-Gal and toward 4MU-Xyl or pNP-Xyl.

Standard reactions were performed in 100 mM cacodylate buffer (pH 7.0) containing 10 mM MnCl₂, 0.6 g.L⁻¹ of total proteins containing membrane β 1,4-GalT7 or 0.2 μ M of purified MBP- β 1,4-GalT7. Kinetic parameters for 4MU-Xyl and UDP-Gal were obtained after the addition of increasing amounts of acceptor or donor substrates (0.02 to 10 mM of 4MU-Xyl or pNP-Xyl in the presence of 10 mM of UDP-Gal; 0.025 to 10 mM of donor substrates in the presence of 10 mM of 4MU-Xyl) to the standard reaction. Each data point represent the mean \pm 2 SD of three individual determinations. nd, not determined.

	UDP-Gal	4MU-Xyl		pNP-Xyl	
	$K_{\rm m}$ (mM)	$K_{\rm m}$ (mM)	$K_{\rm cat} ({\rm min}^{-1})$	K_{m} (mM)	$K_{\rm cat} ({\rm min}^{-1})$
WT - c	0.31 ± 0.03	0.16 ± 0.02	nd	0.85 ± 0.07	nd
WT - p	0.23 ± 0.075	0.27 ± 0.056	68 ± 8	1.27 ± 0.15	91 ± 5



Table 3 Thermodynamic parameters for the binding of various UDP-analogs to the MBP- β 1,4-GalT7

ITC experiments were carried out at 30°C in MOPS (20 mM) buffer, pH 7.0 containing NaCl (150 mM) with MnCl₂ (5 mM) using a VP-ITC MicroCalorimeter (Micro Cal, Inc.). Each data point represent the mean \pm 2 SD of three individual determinations. nd, not determined by direct titration; —, no binding detected.

	ΔG (cal/mol)	∆H (kcal/mol)	ΔS (cal/mol/K)	-TAS (kcal/mol)	K _d (μM)
UDP	- 6880 ± 160	- 22.9 ± 4.8	- 54	16.07	9.0
UTP	-5460 ± 100	- 20.4 ± 5.1	- 50	14.91	99
UMP	- 4430 ± 210	- 13.6 ± 0.6	- 31	9.13	562
Uridine	nd	nd	nd	nd	> 8000
Uracil	nd	nd	nd	nd	> 10000
Ribose	nd	nd	nd	nd	> 10000
TDP	-5440 ± 80	- 13.5 ± 3.4	- 27	8.10	101
CDP	-4190 ± 340	-5.4 ± 0.6	-4.2	1.23	847
GDP	_	-	_	_	_
ADP	_	_	_	_	_



Table 4 Thermodynamic parameters for the binding of various UDP-sugar to the MBPβ1,4-GalT7

ITC experiments were carried out at 30°C in MOPS (20 mM) buffer, pH 7.0 containing NaCl (150 mM) with $MnCl_2$ (5 mM) using a VP-ITC MicroCalorimeter (Micro Cal, Inc.) Each data point represent the mean \pm 2 SD of three individual determinations.

	ΔG (cal/mol)	ΔH (kcal/mol)	ΔS (cal/mol/K)	-T∆S (kcal/mol)	$K_{\rm d}$ (μM)
UDP	- 6880 ± 160	- 22.9 ± 4.8	- 54	16.07	9.0
UDP-Gal	- 6890 ± 180	-25.0 ± 6.2	- 61	18.11	8.7
UDP-Glc	- 6500 ± 280	- 11.5 ± 1.5	- 17	5.02	17
UDP-Man	-5400 ± 45	- 15.3 ± 0.03	- 33	9.89	109
UDP-Xyl	- 5610 ± 30	-12.1 ± 0.8	- 22	6.52	77
UDP-GlcA	- 5520 ± 230	-12.6 ± 3.8	- 24	7.06	89
UDP-GalNAc	- 5300 ± 100	- 11.4 ± 2.1	- 21	6.13	129
UDP-GlcNAc	- 5340 ± 10	-9.43 ± 0.6	- 14	4.10	121

Table 5 Kinetic parameters of the purified $\beta 1,4$ -GalT7 toward various donor substrate. Standard reactions were performed in 100 mM cacodylate buffer (pH 7.0) containing 10 mM MnCl₂, 0.2 to 6 μ M of purified MBP- $\beta 1,4$ -GalT7. Kinetic parameters were obtained after the addition of increasing amounts of donor substrates (0.025 to 20 mM) in the presence of 10 mM of 4MU-Xyl to the standard reaction. Each data point represent the mean \pm 2 SD of three individual determinations. n is the number of determinations ND, not detected. *p < 0.02 compared to the results obtained with UDP-Gal as the donor substrate.

	n	K _m (mM)	$k_{cat} \ (min^{-l})$	k_{cat} / K_m $(min^{-1}.mM^{-1})$
UDP-Gal	6	0.23 ± 0.07	68.1 ± 8.2	296 (100 %)
UDP-Glc	3	0.28 ± 0.02	1.2 ± 0.2	4.2* (1.5%)
UDP-Man		ND	ND	ND
UDP-Xyl	3	0.16 ± 0.02	0.6 ± 0.1	3.9* (1.4%)
UDP-GlcA	3	> 4	> 0.2	_
UDP-GalNAc	_	ND	ND	ND
UDP-GlcNAc	3	> 10	> 0.2	_



Table 6 K_i values for various inhibitors.

The estimation of the K_i was achieved by the incubation in the standart reaction of at least 8 different concentrations of UDP-Gal (from 0.1 to 16 mM) in the presence of four different concentrations of inhibitor (for UDP, from 0 to 2 mM; for UTP and TDP, from 0 to 4 mM; for CDP, UDP-Man and UDP-GalNAc, from 0 to 10 mM).

Inhibitor	K _i (mM)
UDP	0.61 ± 0.10
UTP	1.65 ± 0.22
TDP	1.93 ± 0.31
CDP	> 9
UDP-Man	> 9
UDP-GalNAc	> 9



Table 7 ¹H, ¹³C chemical shifts δ (ppm) and ¹H-¹H coupling constants J (Hz) for 4MU-Xyl acceptor substrate and biosynthetic 4MU-Xyl-Gal, 4MU-Xyl-Glc and 4MU-Xyl-Xyl products in D₂O at 25°C.

¹H chemical shifts are given in ppm downfield from 3-(trimethylsilyl)propionic-2,2,3,3-d4 acid sodium salt. ¹³C

¹H chemical shifts are given in ppm downfield from 3-(trimethylsilyl)propionic-2,2,3,3-d4 acid sodium salt. ¹³C chemical shifts are referenced to acetone as external reference (30 ppm). N.D. not determined because of signal overlapping.

	4MU-Xyl	4MU-Xyl-Gal	4MU-Xyl-Glc	4MU-Xyl-Xyl
H' ₁ / H ₁	- / 5.202	4.515 / 5.212	4.581 / 5.137	4.501 / 5.210
H_2' / H_2	-/3.619	3.551 / 3.671	3.324 / 3.653	3.295 / 3.658
H'_3/H_3	- /3.606	3.671 / 3.766	3.523 / 3.755	3.448 / 3.727
H'_4/H_4	- / 3.746	3.934 / 3.975	3.425 / 3.954	3.647 / 3.903
H' _{5ax} / H _{5ax}	- /3.560	3.731 / 3.645	3.497 / 3.626	3.330 / 3.630
$\mathrm{H'}_{\mathrm{5eq}}/\mathrm{H}_{\mathrm{5eq}}$	- / 4.066	- / 4.215	-/4.208	3.998 / 4.203
H'6 _a	_	3.825	3.951	_
H'6 _b	_	3.771	3.751	_
³ J _{1',2'} , / ³ J _{1,2}	- / 7.0	7.9 / 7.6	7.9 / 7.5	7.7 / 7.7
$^{3}J_{2^{,},3^{,}}$ / $^{3}J_{2,3}$	- /9.1	10.6 / 9.1	8.8 / 8.8	9.2 / 9.0
$^{3}J_{3^{,}4^{,}}$ / $^{3}J_{3,4}$	- / 8.8	3.5 / 9.4	9.3 / 9.3	9.2 / 9.0
$^{3}J_{4',5'ax} / ^{3}J_{4,5ax}$	- /10.6	ND / 9.7	9.3 / 10.8	9.0 / 9.5
$^{3}J_{4',5'eq}$ / $^{3}J_{4,5eq}$	- /5.3	- /5.3	- /5.1	5.5 / 5.3
2 J $_{5$ 'ax,5'eq / 2 J $_{5$ ax,5eq}	- /-11.4	- /-11.7	- /-11.6	-11.7 / -11.7
³ J _{5',6'a}	-V	8.2	ND	_
3 J _{5',6'b}		3.5	ND	_
$^{2}J_{6^{\circ}a,6^{\circ}b}$	(-)	-11.7	ND	_
C' ₁ / C ₁	-/99.91	101.91 / 99.88	101.01 / 99.82	101.66 / 99.85
C_2/C_2	-/72.35	70.63 / 72.40	72.70 / 72.34	72.56 / 72.33
C' ₃ / C ₃	- / 75.20	72.40 / 73.38	75.43 / 73.35	75.39 / 73.24
C' ₄ / C ₄	- / 68.82	68.47 / 75.99	69.42 / 76.03	69.05 / 75.86
C' ₅ / C ₅	- / 65.04	75.19 / 62.98	75.78 / 62.92	65.03 / 62.93
C' ₆	_	61.06	60.48	_



FIGURE LEGENDS

Figure 1 Analysis of the purified human MBP- β 1,4-GalT7 fused-protein by SDS-PAGE Human β1,4-GalT7, lacking the first *N*-terminal 81 amino acids, was expressed in *Escherichia coli* as a fused-protein with the MBP, and then, purified onto an amylose column. The supernatant loaded on the amylose column (10μg, lane 1) and the purified MBP-β1,4-GalT7 (2μg, lane 2) were electrophoresed on polyacrylamide gel (10% w/v) in the presence of SDS and then, stained with Coomassie blue.

Figure 2 Calorimetric profiles of the binding of donor and acceptor substrate analogs to MBP-β1,4-GalT7

A/ Binding of UDP to MBP- β 1,4-GalT7. The reaction cell contained MBP- β 1,4-GalT7 (1.4 mL, 30 μ M) in 20 mM MOPS buffer containing 150 mM NaCl in the absence (gray line) or presence (5 mM, black line) of MnCl₂. The syringe contained 0.5 mM UDP in the same buffer. *Top*, raw calorimetric data obtained from the injection of 4 μ L (11 injections), then 8 μ L (19 injections) of UDP at 4-min intervals. The bottom plots show the integrated binding isotherm with the experimental points and the corresponding best fit, providing values for thermodynamic parameters calculated in the presence of MnCl₂. B/ Binding of pNP-Xyl to MBP- β 1,4-GalT7. The reaction cell contained MBP- β 1,4-GalT7 (1.4 mL, 30 μ M) in 20 mM MOPS buffer containing 150 mM NaCl and 5mM MnCl₂ in the absence (gray line) or presence (5 mM, black line) of UDP. The syringe contained 40 mM pNP-Xyl in the same buffer. *Top*, raw calorimetric data obtained from the injection of 4 μ L (11 injections), then 10 μ L (25 injections) of pNP-Xyl at 4-min intervals. The bottom plots show the integrated binding isotherm with the experimental points and the corresponding best fit, providing values for thermodynamic parameters calculated in the presence of UDP.

Figure 3 Free enthalpy change differences ($\Delta\Delta H$) A / for UDP, UMP and Uracil binding to MBP- β 1,4-GalT7 as compared to UDP-Gal; and B / for TDP and CDP as compared to UDP.

A/ $\triangle\Delta H = \Delta H$ obtained for the binding of UDP, UMP or Uracil - ΔH obtained for the binding of UDP-Gal. B/ $\Delta\Delta H = \Delta H$ obtained for the binding of TDP or CDP - ΔH obtained for the binding of UDP. nd, not determined.





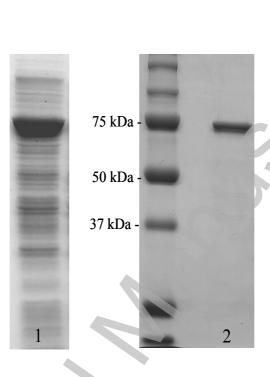


Figure 1



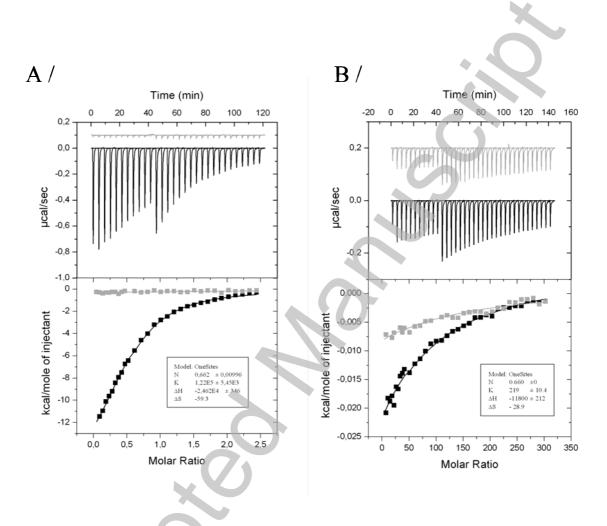


Figure 2



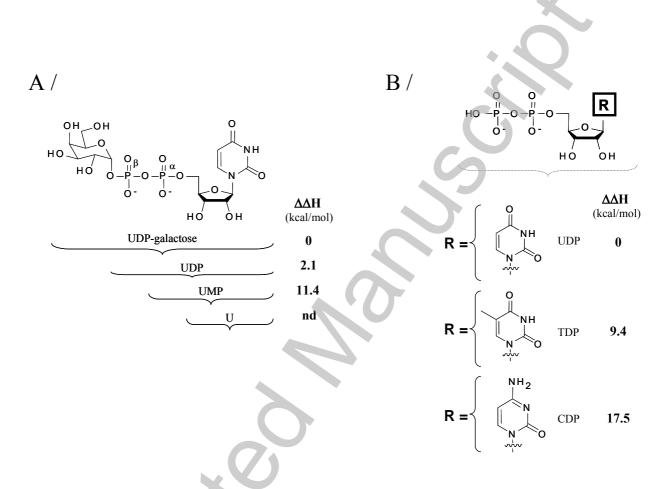


Figure 3