



HAL
open science

Nitrous oxide distribution and its origin in the central and eastern South Pacific Subtropical Gyre

J. Charpentier, L. Farias, N. Yoshida, N. Boontanon, Patrick Raimbault

► **To cite this version:**

J. Charpentier, L. Farias, N. Yoshida, N. Boontanon, Patrick Raimbault. Nitrous oxide distribution and its origin in the central and eastern South Pacific Subtropical Gyre. *Biogeosciences*, 2007, 4 (5), pp.729-741. hal-00330324

HAL Id: hal-00330324

<https://hal.science/hal-00330324>

Submitted on 18 Jun 2008

HAL is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers.

L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.

Nitrous oxide distribution and its origin in the central and eastern South Pacific Subtropical Gyre

J. Charpentier^{1,2}, L. Farias², N. Yoshida^{3,4}, N. Boontanon^{4,5}, and P. Raimbault⁶

¹Programa de Postgrado, Departamento de Oceanografía, Facultad de Ciencias Naturales y Oceanográficas, Universidad de Concepción, Chile

²Departamento de Oceanografía & Centro Oceanográfico del Pacífico Sur (COPAS), Universidad de Concepción, Concepción, Chile

³Frontier Collaborative Research Center, Tokyo Institute of Technology, Midori-ku, Yokohama, Japan

⁴SORST project, JST, Kawaguchi, Saitama, Japan

⁵Faculty of Environment and Resource Studies, Mahidol University 999 Phuttamonthon 4 Road, Phuttamonthon, Salaya, Nakhon Pathom 73170, Thailand

⁶Laboratoire d'Océanographie et de Biogéochimie (CNRS UMR 6535), Centre d'Océanologie de Marseille, Campus de Luminy, Marseille Cedex, France

Received: 10 May 2007 – Published in Biogeosciences Discuss.: 25 May 2007

Revised: 22 August 2007 – Accepted: 12 September 2007 – Published: 13 September 2007

Abstract. The mechanisms of microbial nitrous oxide (N_2O) production in the ocean have been the subject of many discussions in recent years. New isotopomeric tools can further refine our knowledge of N_2O sources in natural environments. This study compares hydrographic, N_2O concentration, and N_2O isotopic and isotopomeric data from three stations along a coast-perpendicular transect in the South Pacific Ocean, extending from the center (Sts. GYR and EGY) of the subtropical oligotrophic gyre ($\sim 26^\circ\text{S}$; 114°W) to the upwelling zone (St. UPX) off the central Chilean coast ($\sim 34^\circ\text{S}$). Although $\text{AOU}/\text{N}_2\text{O}$ and NO_3^- trends support the idea that most of the N_2O (mainly from intermediate water (200–600 m)) comes from nitrification, N_2O isotopomeric composition (intramolecular distribution of ^{15}N isotopes) expressed as SP (site preference of ^{15}N) shows low values (10 to 12‰) that could be attributed to the production through of microbial nitrifier denitrification (reduction of nitrite to N_2O mediated by ammonium oxidizers). The coincidence of this SP signal with high – stability layer, where sinking organic particles can accumulate, suggests that N_2O could be produced by nitrifier denitrification inside particles. It is postulated that deceleration of particles in the pycnocline can modify the advection - diffusion balance inside particles, allowing the accumulation of nitrite and O_2 depletion suitable for nitrifier denitrification. As lateral advection seems to be relatively insignificant in the gyre, in situ nitrifier denitrification could account for 40–50% of the N_2O produced in this

layer. In contrast, coastal upwelling system is characterized by O_2 deficient condition and some N deficit in a eutrophic system. Here, N_2O accumulates up to 480% saturation, and isotopic and isotopomer signals show highly complex N_2O production processes, which presumably reflect both the effect of nitrification and denitrification at low O_2 levels on N_2O production, but net N_2O consumption by denitrification was not observed.

1 Introduction

N_2O exists in the atmosphere at trace levels. However, this “greenhouse gas” is of great environmental importance. It is 170 to 300 times more efficient (per molecule) than CO_2 (Manne and Richels 2001) in trapping infrared radiation and accounts for 5–6% of the greenhouse effect (Law and Ling 2001). The oceans are a net source of N_2O for the atmosphere, with an estimated average annual emission of 6 Tg N year⁻¹, corresponding to 20% of the global emissions (Nevison et al., 1995).

The relative importance of the biological processes producing N_2O remains unclear. Nitrification is a chemotrophic process by which NH_4^+ and NO_2^- are aerobically oxidized to fix inorganic carbon. These reactions are carried out in two stages by different groups of microorganisms (Ward 2000). The first stage is ammonium oxidation to nitrite, carried out by microorganisms called ammonium-oxidizers. In this reaction, hydroxylamine acts as an intermediate and has been proposed as a precursor of N_2O , although the biochemical

Correspondence to: J. Charpentier
(jcharpentier@prof.udec.cl)

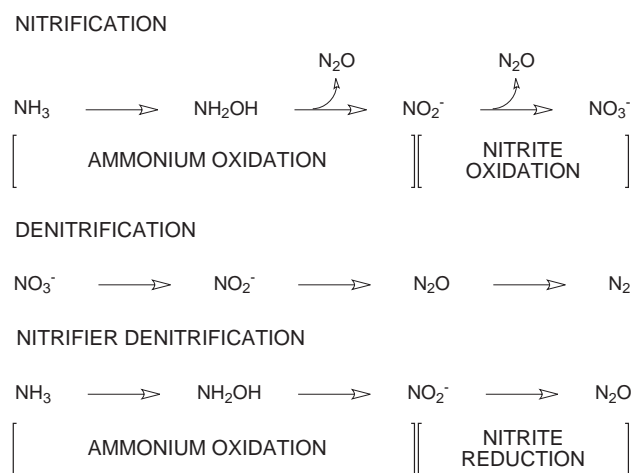


Fig. 1. Outline of different pathways of N_2O production. Adapted from Wrage et al. (2001).

mechanism by which it is formed is not clear (Naqvi and Noronha, 1991; Ostrom et al., 2000). Recent works have shown that certain *Archaea* can also oxidize ammonium to nitrite with a metabolism similar to that of bacterial ammonium oxidation (Konneke et al., 2005). The second stage is nitrite oxidation to nitrate, with nitric oxide acting as an intermediate and possible precursor of N_2O . This process is carried out by organisms called nitrite-oxidizers.

Conventional denitrification is the reduction of oxidized inorganic nitrogen (NO_3^- , NO_2^- , NO , N_2O) to gaseous nitrogen forms (N_2 , N_2O), and involves a loss of fixed-nitrogen from the system. This process is carried out by several organisms as a respiration process under suboxic (<0.1 ml/L O_2) conditions, with nitrate as the electron acceptor (Knowles, 1982). For this reason, denitrification is enhanced in O_2 minimum areas (Codispoti et al., 2001; Gruber and Sarmiento, 1997) or in areas where large accumulation of particulate organic matter takes place, such as the pycnocline (Allredge and Cohen 1987). Denitrification occurs in several stages, during which N_2O is an intermediate, unlike nitrification. Thus, N_2O can be both produced and consumed during denitrification (Elkins et al., 1978). Under suboxic conditions (<0.1 ml/L), N_2O produced by denitrification is almost entirely reduced to N_2 , whereas at the oxycline, where the O_2 concentration is near 0.5 ml/L, the N_2O production rate significantly increases and the reduction of N_2O to nitrogen is inhibited (Castro and Farías, 2004).

Moreover, certain species of nitrifying bacteria can produce N_2O under O_2 stress conditions (< 1 ml/L) by means of ammonium oxidation to nitrite, which in turn is reduced to N_2O ; this process is called “nitrifier denitrification” (Poth and Focht, 1985). This process is carried out by ammonium oxidizers like *Nitrosomona europaea* (Ritchie and Nicholas, 1972; Shaw et al., 2006). Biochemical studies show that the second part of this process indeed corresponds to denitrifica-

tion, and that the involved enzymes could be the same (Poth and Focht, 1985). Otherwise, the genomes of ammonium oxidizing archaea also present nitrite reductase and nitric oxide reductase genes, potentially allowing them to carry out nitrifier denitrification (Hallam et al., 2006; Treusch et al., 2005). Figure 1 shows a general outline of these processes.

Recently, some additional processes have been recognized as a part of the marine nitrogen cycle, particularly the anammox reaction ($\text{NO}_2^- + \text{NH}_4^+ \rightarrow \text{N}_2 + 2\text{H}_2\text{O}$), which result in a loss of nitrogen from the system as nitrogen gas (N_2). Kartal et al. (2007) found that *Kuenenia stuttgartensis*, an anammox bacterium, can produce small quantities of N_2O as a byproduct of anammox.

The typical vertical distribution of N_2O in the open ocean shows that it is directly correlated with nitrate and inversely correlated with O_2 (Nevison et al., 2003; Oudot et al., 2002). Since nitrification is favored at low O_2 concentrations (Carlucci and McNally, 1969; Goreau et al., 1980), this distribution has usually been interpreted as an indirect evidence that nitrification is the dominant process in oceanic N_2O formation. Isotopic data, however, show that $\delta^{15}\text{N}(\text{N}_2\text{O})$ values are higher than expected for N_2O produced by nitrification in areas where the relationship between O_2 , N_2O , and nitrate indicates that nitrification is the dominant process (Dore et al., 1998; Kim and Craig, 1990; Naqvi et al., 1998; Ostrom et al., 2000; Yoshida, 1988; Yoshida et al., 1989).

The determination of N_2O isotopomers (i.e., the intramolecular distribution of ^{15}N in the linear NNO molecule) offers a useful and relatively new tool for elucidating N_2O production mechanisms (Toyoda and Yoshida 1999). $\delta^{15}\text{N}^\alpha$ is the relative isotopic abundance of ^{15}N for the central position and $\delta^{15}\text{N}^\beta$ for the terminal position. Site preference (SP) is defined as follows:

$$S.P. = \delta^{15}\text{N}^\alpha - \delta^{15}\text{N}^\beta \quad (1)$$

Since N_2O precursors contain only one nitrogen atom (NO_2^- , NO , NH_2OH) and excluding the possibility that different chemical species combine to form N_2O , the isotopomer distribution in N_2O should be independent of the $\delta^{15}\text{N}$ of the precursors. It can, therefore, be determined by the biochemical reaction step in which equivalence between both nitrogen atoms is lost (Toyoda et al., 2002). This is the main advantage of the N_2O isotopomer ratio over the conventional $\delta^{15}\text{N}$ ratio. Nevertheless, the lack of knowledge of the specific biochemical mechanisms involved in N_2O production makes it difficult to “predict” which isotopomer signature corresponds to which particular processes (Schmidt et al., 2004). The few works that present isotopomer results in marine environments have found positive SP values associated with nitrification (Popp et al., 2002; Toyoda et al., 2002) or denitrification (Westley et al., 2006; Yamagishi et al., 2005). The main problem in interpreting these results is the context of N_2O production processes and how these processes influence the isotopomer signal.

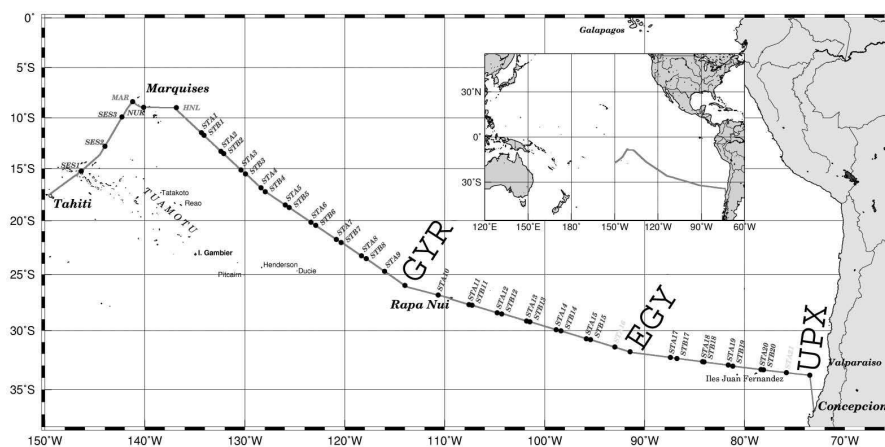


Fig. 2. Sampled stations during BIOSOPE Cruise, GYR, EGY and UPX are indicated in bigger letters.

The South Pacific Central Gyre has been described as the most oligotrophic zone in the world ocean (Claustre and Maritorena, 2003), with an extreme nutrient (N, Fe) limitation (Bonnet et al., 2007); it is also one of the least studied areas of the ocean (Daneri and Quiñones, 2001). In contrast, the central Chile upwelling system is one of the most productive marine environments in the world, with a high supply of nitrate and other nutrients due to an upwelling regime and coastal contribution through land runoff, which supports a high rate of new production (Daneri et al., 2000). In addition, this area contains the southernmost extension of the O₂ minimum zone (OMZ), fed from the equator by the Peru Chile Undercurrent (PCUC); the OMZ has a significant influence on the water column community (Morales et al., 1996). Thus, the transect stretching from the center of the gyre to the Chilean coast is an ideal area for studying the variety of processes that produce N₂O along an extreme gradient.

The main objective of this work is to try to elucidate the changes in N₂O production processes in relation to different environmental conditions in the water column through a comparison of the different environments encountered during the BIOSOPE cruise (Fig. 2). To accomplish this, we employ a combination of isotopic and isotopomeric N₂O measurements along with oceanographic and biogeochemical data.

2 Methods

2.1 Sampling

All samples were collected during the BIOSOPE cruise (October–December 2004 or early austral spring) on board the R/V L'Atalante. The stations chosen in this work – GYR (26.07° S, 113.99° W), EGY (31.90° S, 91.41° W), and UPX (34.58° S, 72.43° W) – are representatives of different zones along the trophic gradient. GYR is within the South Pacific Central Gyre, EGY is on the eastern border of the gyre, and

UPX is in the coastal upwelling zone, 33 km from the shore (Fig. 2).

Temperature and salinity profiles were obtained with a Seabird SBE 911 + CTD. Fluorescence (as Relative Fluorescence Unit, RFU) was measured in situ from a Chelsea AquaTracka fluorometer, attached to the CTD rosette-sampler system. Sampling was done one hour before sunrise, with twelve-Liter Niskin bottles attached to a CTD-O rosette. Particle content was measured by an Underwater Video Profiler (UVP) system as described in Gorsky et al. (2000). Particle size was measured as Equivalent Spherical Diameter (ESD) in mm (Stemmann et al., 2000).

Seawater samples were collected for isotopic and N₂O concentration analyses as well as for O₂ and nutrient analyses. The samples for isotopes, isotopomers, and N₂O analyses were transferred directly into 125-ml glass flasks, preserved with HgCl₂, and sealed with butyl rubber stoppers following the method described in Yamagishi et al. (2001).

2.2 Nitrous oxide, oxygen, nitrate, and phosphate determination

N₂O was determined in the seawater samples used for the isotopic analysis with gas chromatography and mass spectrometry detection. The method is well described in Toyoda et al. (2005).

O₂ was measured in situ with a Seabird SBE 43 O₂ meter. The O₂ sensor was calibrated with Winkler titrations. Nitrate was immediately analyzed on board following two methods depending on the concentration range. For low concentrations (nM range), the method described in Raimbault et al. (1990) was used, whereas for higher concentrations (μM range), the method of Armstrong et al. (1967) was used. Phosphate was analyzed immediately on board by the method described in Tréguer and LeCorre (1975). The Apparent O₂ Utilization (AOU) value was obtained by subtracting the measured value of the O₂ concentration from the

saturation value computed for the temperature and salinity of the seawater (Murray and Riley, 1969), whereas the difference between the N_2O saturation concentration and its measured concentration (ΔN_2O) in the seawater was used to infer its production (positive) or consumption (negative) (Yoshinari, 1976).

2.3 Isotopic and isotopomeric determinations

Isotopic and isotopomeric determinations were carried out at the Tokyo Institute of Technology using a Finnigan MAT 252 mass spectrometer following the method described in Toyoda and Yoshida (1999). For this, N_2O was extracted from samples by sparging with helium and then introduced into a preconcentration-gas chromatography-isotopic ratio mass spectrometry system. $\delta^{15}N$ (α and β) and $\delta^{18}O$ (N_2O) were determined in relation to the atmospheric nitrogen and Vienna Standard Mean Ocean Water (VSMOW), respectively. N_2O isotopomers were determined based on the analysis of ionic mass fragments (NO^+ and N_2O^+) formed by the electronic impact of N_2O . This determination was possible since NO^+ fragments contain the central nitrogen (α), which allows the conversion of the fragment ratios into isotopic ratios of $^{14}N^{15}NO$ and $^{15}N^{14}NO$. Although there is a rearrangement reaction during the ionic fragmentation process, its magnitude can be determined and corrections applied. Precision of the measurements is typically better than 0.5‰ for $\delta^{15}N_{\text{bulk}}$ and $\delta^{18}O$, and better than 1‰ for $\delta^{15}N^\alpha$ and $\delta^{15}N^\beta$ (Toyoda et al., 2005). Isotopomeric ratios were calculated with the following expressions:

$$^{15}R^\alpha = \frac{[^{14}N^{15}N^{16}O]}{[^{14}N^{14}N^{16}O]} \quad (2)$$

$$^{15}R^\beta = \frac{[^{15}N^{14}N^{16}O]}{[^{14}N^{14}N^{16}O]} \quad (3)$$

$$\delta^{15}N^\alpha = \left\{ \frac{^{15}R^\alpha}{^{15}R^\alpha(\text{std}) - 1} \right\} \times 1000 \quad (4)$$

$$\delta^{15}N^\beta = \left\{ \frac{^{15}R^\beta}{^{15}R^\beta(\text{std}) - 1} \right\} \times 1000 \quad (5)$$

Furthermore, the definition of $\delta^{15}N_{\text{bulk}} = (\delta^{15}N^\alpha + \delta^{15}N^\beta)/2$ allowed us to compare the relative abundance of the isotopomers α and β with the relative isotopic abundance of ^{15}N .

2.4 Data analysis

The Brunt-Vaisälä frequency (BVF) was determined using temperature and salinity data. For better data interpretation, BVF profiles were visually fitted to an eight-term Gaussian model included in MATLAB software. The mixed layer was determined by averaging the depths of

four criteria as described in the official data source of the Biosope cruise (http://www.obs-vlfr.fr/proof/php/bio_log_basicfiles.php). The photic layer is defined as the depth at which the surface light irradiance is reduced to 1% of its surface value. This determination was made by averaging readings from four different sensors.

3 Results and discussion

3.1 General water column characteristics

The three studied stations are located along a gradient from very oligotrophic to highly eutrophic systems and are found in different oceanographic regimes with the following characteristics. The GYR station is located at the center of South Pacific Gyre, with severe oligotrophy (Claustre and Mariotorena, 2003). The fluorescence peak (0.23 RFU) at this station occurs at ~ 180 m, just below the bottom of the photic layer (164 m). The shallower pycnocline is located at 10 m. A second, thicker pycnocline is observed between 100 and 500 m, as shown by the BVF plot in Fig. 3a.

The EGY station is located on the eastern border of the South Pacific Gyre, in a less oligotrophic environment. The fluorescence peak (0.34 RFU at 52 m) is shallower and more pronounced than at the previous station, well above the depth of the photic layer (92 m). The shallower pycnocline is located at 26 m and the BVF plot shows a second, thicker pycnocline between 120 and 400 m (Fig. 3b). Both GYR and EGY stations are located in a zone in which surface and subsurface waters (0–500 m) are mainly dominated by East South Pacific Central Water (ESPCW), which is characterized by wide temperature and salinity ranges. The ESPCW is formed in the middle of the South Pacific close to 40° S (Emery and Meincke 1986).

The UPX station is located off the coast of central Chile in a highly productive environment. The chlorophyll fluorescence peak (0.81 RFU) at this station is found at 32 m, but fluorescence is still high at the surface (0.70 RFU). The peak is close to the base of the photic layer (37 m) and the mixed layer (36 m). The BVF plot showed a high stability zone from 20 to 60 m depth (Fig. 3c). Changes in the values and shapes of the chlorophyll profiles from west to east clearly showed the expected differences in productivity due to the enhanced nutrient supply. The central Chile coast is characterized by the upwelling of nutrient-rich, low O_2 Equatorial Subsurface Water (ESSW) (Strub et al., 1995).

Circulation at this coastal station is greatly affected by horizontal advection that brings nutrient-enriched waters to this latitude in the form of ESSW, which is transported by the PCUC. Indeed, the PCUC has been detected as far south as 48° S (Silva and Neshyba, 1979). The SW wind stress on the surface layer causes upwelling of nutrient-rich, low-oxygen waters followed by increased productivity that provides favourable conditions for N_2O production (Cornejo

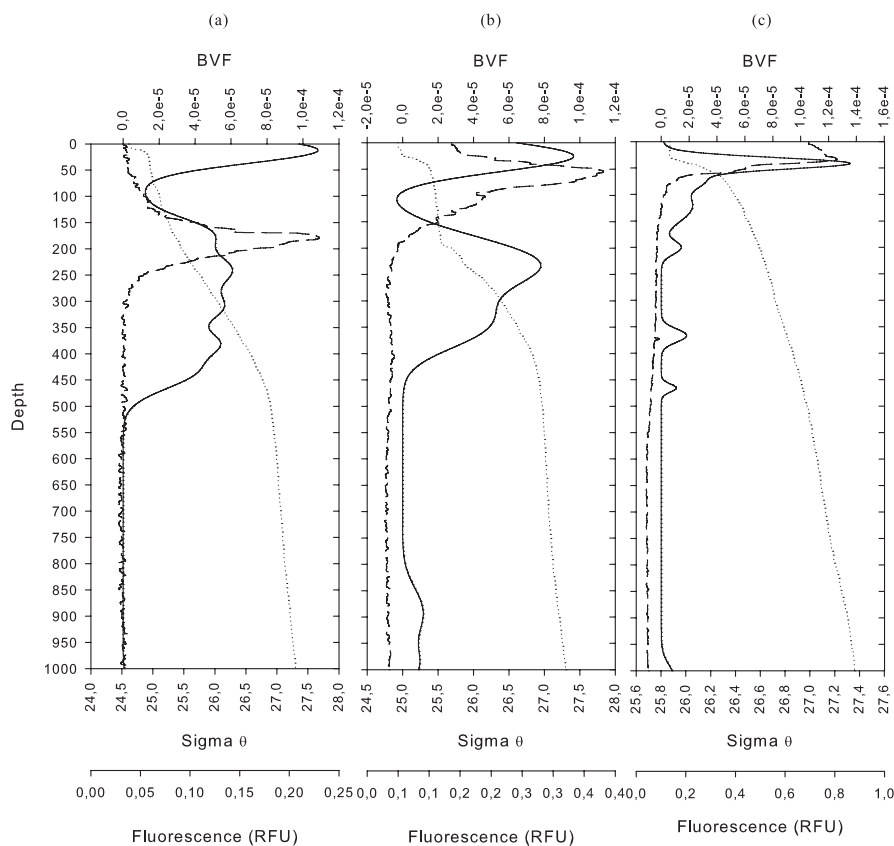


Fig. 3. Vertical profiles at stations (a): GYR, (b): EGY and (c): UPX. The parameters shown are: Potential density (dotted line), Brunt-Vaisälä frequency (continuous line), and Fluorescence (dashed line).

et al., 2006). Thus, N_2O produced in subsurface waters is brought to the surface through advection and mixing, enhancing N_2O fluxes to the atmosphere.

3.2 Nitrous oxide, oxygen, and nutrient behavior

Vertical distributions of N_2O and O_2 are shown in Fig. 4. N_2O is slightly oversaturated at GYR and EGY from the surface to ≈ 200 m ($\approx 114\%$ and $\approx 105\%$, respectively), and is highly oversaturated below this depth ($\approx 230\%$, at both stations) (Figs. 4a, b). N_2O profiles at GYR and EGY are almost mirrored by the O_2 profiles, indicating that the amount of N_2O produced in the water column is strongly driven by the O_2 concentration. At the UPX station, the whole water column is highly N_2O -oversaturated from 230% at the surface to 480% at the N_2O maximum at 350 m. At this station, the O_2 concentrations display a strong depletion up to $\approx 10 \mu\text{mol/Kg}$ between 150 m and 300 m and do not show a clear relationship with the N_2O profile.

Nitrate and phosphate profiles are shown in Fig. 5. At the GYR and EGY stations, both nutrients have very similar profiles, with low concentrations in the first 200 m. Below this depth, the nitrate concentrations gradually increase to $40 \mu\text{mol/Kg}$ (1000 m) at GYR and $35 \mu\text{mol/Kg}$ (400 m)

at EGY. At both stations (GYR, EGY), abrupt increases in nitrate and phosphate concentrations close to 400 m coincide with maxima in the N_2O concentrations, suggesting an increase in the rate of nitrification at this depth. At the UPX station, nitrate concentrations are high at the surface (Fig. 5c) due to the upwelling of nutrient-rich water (Daneri et al., 2000). Although the nitrate and phosphate profiles show similar trends, the profiles appear to be more complicated than at the oceanic stations. N_2O and nitrate show very similar profiles at this station, suggesting that the N_2O production-consumption processes are the same as for nitrate.

The PO_4^{3-} vs NO_3^- plots (Figs. 5a, b, insets) suggest that at stations GYR and EGY both nutrients behave according to the Redfield ratio, indicating that nitrogen oxidative processes are predominant. At UPX, the plot (Fig. 5c, inset) suggests nitrate removal by denitrification since there is a phosphate excess in relation to nitrate according to the Redfield ratio.

In order to obtain information about the processes involved in N_2O cycling throughout the water column, scatter diagrams of AOU vs. ΔN_2O and AOU vs. NO_3^- below the photic layer (200–600 m) at GYR and EGY are presented (Figs. 6a, b). ΔN_2O correlated positively with AOU at both

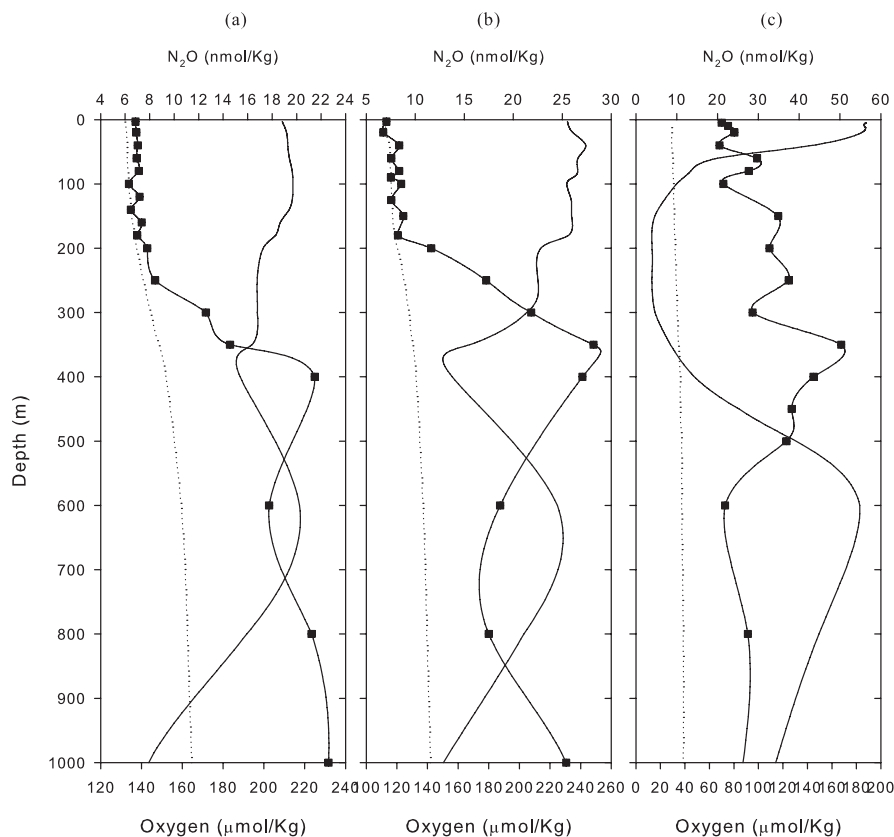


Fig. 4. Vertical profiles at stations (a): GYR, (b): EGY (c): UPX. The parameters shown are: N₂O saturation (dotted line), N₂O (squares) and O₂ (continuous line).

stations, indicating that more N₂O is produced while O₂ is consumed. NO₃⁻ also correlated positively with AOU, supporting the view that, N₂O and nitrate production are produced mainly from nitrification. The ratio of N₂O production to O₂ consumption (on a molar basis), referred to as the N₂O oxidative ratio, was 1.76×10^{-4} and 1.42×10^{-4} for GYR and EGY station respectively. Since NO₃⁻ regeneration was also linearly related to AOU, it was possible to calculate the ratio of N₂O production to NO₃⁻ production (expressed as a percentage on the mol-N basis) as 0.088% and 0.15% for GYR and EGY stations respectively. The N₂O oxidative ratio (i.e., produced N₂O per mol of respired O₂) and the N₂O reaction yields (i.e., produced N₂O per mol of produced NO₃⁻) are close to the oceanic values reported by Cohen and Gordon (1979) for the North East Pacific.

3.3 $\delta^{15}\text{N}_{\text{bulk}}$, $\delta^{18}\text{O}$, and isotopomers of nitrous oxide

The $\delta^{15}\text{N}_{\text{bulk}}$ has been defined as equivalent to conventional $\delta^{15}\text{N}$ (Toyoda and Yoshida 1999). Surprisingly, variations of this parameter at the three stations are relatively small. In contrast, the SP signal shows great variability, which provides more information about the processes occurring in the water column. At the GYR and EGY stations, the SP values

in the mixed layer are close to the expected atmospheric N₂O value ($19 \pm 2\text{‰}$) (Yoshida and Toyoda, 2000), indicating that surface N₂O originates mainly from ocean-atmosphere interactions (Figs. 7a, b). At UPX, however, the SP value in the mixed layer ($\approx 14\text{‰}$) is lower than in the air, indicating the influence of N₂O upwelled with subsurface waters (Fig. 7c). Although the accepted value for the SP of atmospheric N₂O was determined in air samples from the North West Pacific, the high mixing rates of tropospheric gases and the high residence times of atmospheric N₂O over 120 years (Liao et al., 2004) make the value given by Yoshida and Toyoda (2000) a good approximation.

Site preference between the shallow pycnocline and the base of the photic layer shows high variability. Assimilatory nitrate reduction has been mentioned as a source of N₂O in surface waters (Oudot et al., 1990). However, not enough information is available to establish whether this process is actually occurring. At UPX, where the mixed layer coincides with the photic layer, SP and $\delta^{18}\text{O}$ (N₂O) values over the pycnocline are quite stable.

Below the photic zone, SP profiles at GYR and EGY show conspicuous minima (11.5‰, 8.5‰) at 350 m and 250 m, respectively, followed by a gradual increase up to 22‰ in

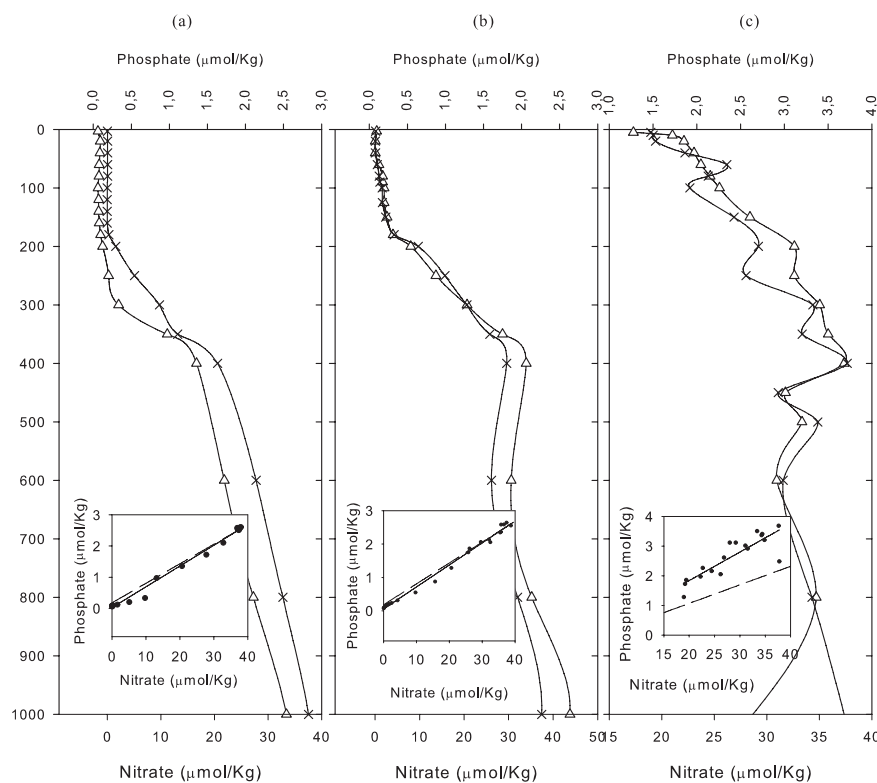


Fig. 5. Vertical profiles at stations (a): GYR, (b): EGY (c): UPX. The parameters shown are: Nitrate (cross) and phosphate (open triangle). Insets shows nitrate v/s phosphate ratio. Solid line shows linear regression of data, dashed line shows the equation deduced from Redfield N:P ratio (16:1) (Gruber and Sarmiento, 1997).

deeper waters. Such a minimum is also observed at UPX at a very shallow depth (40 m) within the upper pycnocline. Below the pycnocline, SP values do not show a clear depth trend. Since SP is independent from the $\delta^{15}\text{N}$ of its precursors, or the extent of the reaction, and is only dependant on the reaction mechanism, the observed changes in SP in the water column must be associated with changes in the mechanistic sources of N_2O (Toyoda et al., 2002). The SP minima at oceanic and coastal stations must be influenced by a particular process that produces low SP N_2O .

It is unlikely that the SP minimum values are being transported from nutrient-rich coastal waters. The Ekman transport in east oceanic margins is determined by the Rossby radius of deformation, whose maximum reach at mid-latitudes is 200 to 300 km for very unstratified coastal waters (Csanady, 1982). Otherwise, as has been previously mentioned, surface and subsurface waters (0–500 m) of the Subtropical South Pacific Gyre are dominated by ESPCW, produced close to 40°S (Emery and Meincke, 1986).

The coincidence that all SP minima (at three stations) with high density gradient layers (see Figs. 3, 7) where particles are likely to accumulate indicates that N_2O production at this point should be, at least in part, associated with the accumulation of such particles. Studies have demon-

strated that temporal oxygen-depleted microsites are possible in marine snow in the absence of light (Alldredge and Cohen, 1987; Ploug, 2001; Ploug et al., 1997). The O_2 consumption around sinking particles strongly depends on the advection-diffusion balance driven by sinking velocities (Csanady, 1986; Kiørboe et al., 2001), with higher O_2 depletion associated with slower-falling particles.

Although the stations GYR and EGY are located in a very oligotrophic area, the accumulation of sinking particles at the steepest density gradient layer inside the pycnocline (see Brunt-Vaisälä frequency plots, Fig. 3) should allow the formation of aggregates suitable for nitrogen reductive processes, as proposed by Yoshida et al. (1989). Figure 8 shows the particle count between 0.052 mm and 8.438 mm of ESD (a wide range of marine snow sizes) at three sampled stations (C. Gorsky and M. Picheral, personal communication). At the GYR and EGY stations, small particle accumulation is observed at 300 m (≈ 35 part/L) and 50 m (≈ 45 part/L), respectively, roughly coincident with high stability zones. The particle content at EGY below the photic zone shows no significant accumulation; the mean particle content between 200 m and 400 m is ≈ 22 part/L. High particle contents at the UPX station are also closely related to the stability profile.

Toyoda et al. (2002) proposed that nitrification should

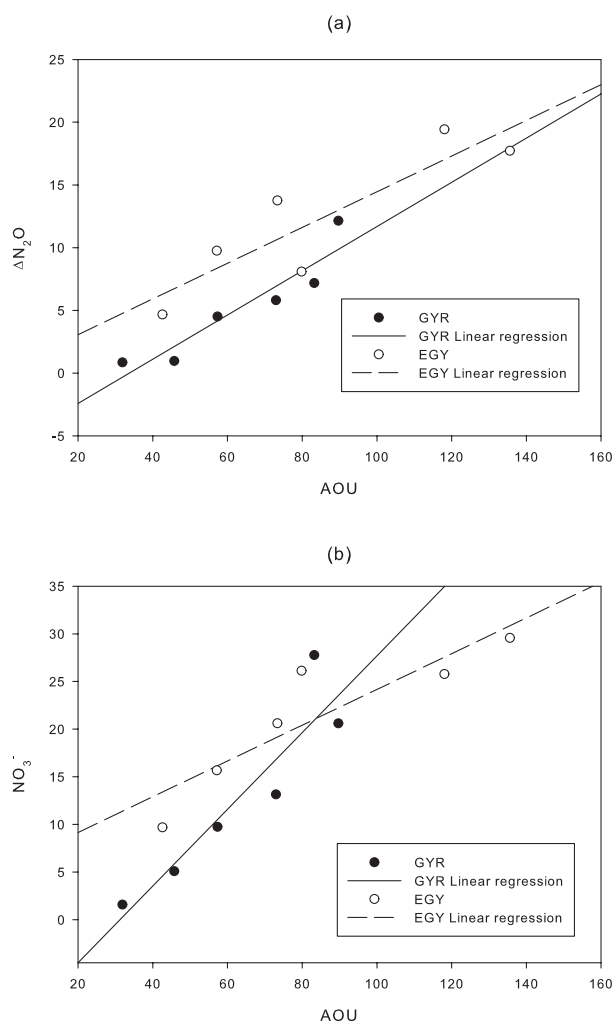


Fig. 6. (a) AOU v/s ΔN_2O at GYR and EGY stations from 200 to 600 m, the linear regression model for each line was GYR: $y = -5.94 + 0.176x$, $r^2 = 0.87$; EGY: $y = 0.22 + 0.142x$, $r^2 = 0.79$. (b) AOU v/s NO_3^- at GYR and EGY stations from 200 to 600 m, the linear regression model for each line was GYR: $y = -12.0 + 0.403x$, $r^2 = 0.85$; EGY: $y = 5.4 + 0.188x$, $r^2 = 0.81$.

yield $\delta^{15}N^\alpha$ enriched N_2O due to the existence of a hyponitrite (-ONNO-) intermediary followed by the selective breakage of the NO bond where the lightest isotopes are located, based on the ZPE (zero point energy) for N_2O (Zuñiga et al., 2003). This mechanism has been described for some nitric oxide reductases (Hendriks et al., 2000; Wasser et al., 2002). Culture experiments carried out by Sutka et al. (2006) show that ammonium oxidation carried out by ammonium oxidizers (i.e. *Nitrosomona europaea*, *Nitrosospira multiformis*) produces N_2O enriched in $\delta^{15}N^\alpha$, with average SP values of 33‰. Despite the lack of conclusive evidence about this specific mechanism in nitrification, if we assume that nitrification is the main process of N_2O production, at least at the oceanic stations, positive SP values below the photic layer can be attributed to nitrification.

These results, as well as other culture experiments carried out by Sutka et al. (2003, 2006) show that, under O_2 stress, nitrifier denitrification mediated by ammonium oxidizers (i.e. *Nitrosomona europaea* and *Nitrosospira multiformis*), produces N_2O with an average SP value of zero. This means that there is no selectivity of nitrogen atoms during N_2O production. Evidence of sequential mechanisms of N_2O production by nitrite reductive processes in bacteria has been reported several times in the biochemical literature (Aeressens et al., 1986; Averill and Tiedje, 1982; Weeg-Aeressens et al., 1988; Zafriou et al., 1989). Such mechanisms involve the successive addition of one molecule of precursor (NO_2^- or NO) to the enzyme. The resulting ^{15}N distribution must be statistically determined by the entrance of the precursor, contrary to the selective effect of simultaneous mechanisms proposed by Toyoda et al. (2002).

Although it has been pointed out that nitrifier denitrification occurs under O_2 stress, Shaw et al. (2006) demonstrated that bacteria of the genera *Nitrosospira* and *Nitrosomona* are capable of producing N_2O via nitrifier denitrification in the presence of nitrite, even in aerobic environments. Nitrifier denitrification has not been studied in marine environments, and its possible ecological role is far from being established. Wrage et al. (2001) proposed thermodynamic reasons for the occurrence of nitrifier denitrification due to the negative Gibbs free energy for the $NH_4^+ \rightarrow NO_2^- \rightarrow N_2O$ processes, which are favoured at low pH. Poth and Focht (1985) have proposed that nitrifier denitrification in soil can act as a nitrite detoxification mechanism. Nitrite has been shown to be toxic for ammonium oxidizers since it reduces ammonium monooxygenase activity (Stein and Arp, 1998). Therefore, we propose that nitrite accumulation around sinking particles due to the sudden loss of speed in the pycnocline potentially activates nitrifier denitrification enzymes on ammonium oxidizers.

It is important to emphasize that $\delta^{15}N^\alpha$ enrichment due to denitrifying N_2O consumption in anoxic microsites of particles in oceanic environments, as previously pointed out (Yoshida et al., 1989), is not expected because N_2O reduction only occurs at very low O_2 concentrations (Castro and Farías, 2004; Elkins et al., 1978). Very large aggregates are necessary for the occurrence of highly anoxic microsites in particles and this is unlikely in open oligotrophic or even mesotrophic environments. If this process occurs, its effect must be negligible.

$\delta^{18}O$ (N_2O) is always hard to elucidate due to its dependence on the isotopic signature of precursors. The $\delta^{18}O$ (N_2O) profile below the mixed layer at GYR and EGY and below 500 m at UPX are similar in shape to the SP, as previously observed by Toyoda et al. (2002) and Popp et al. (2002), indicating that $\delta^{18}O$ (N_2O) is also driven by N_2O production mechanisms. The primary source of O in the N_2O produced by ammonium oxidation is dissolved O_2 (Ostrom et al., 2000), since it is expected that $\delta^{18}O$ (N_2O) will be greater than 40‰. Furthermore, the preferential breakage of

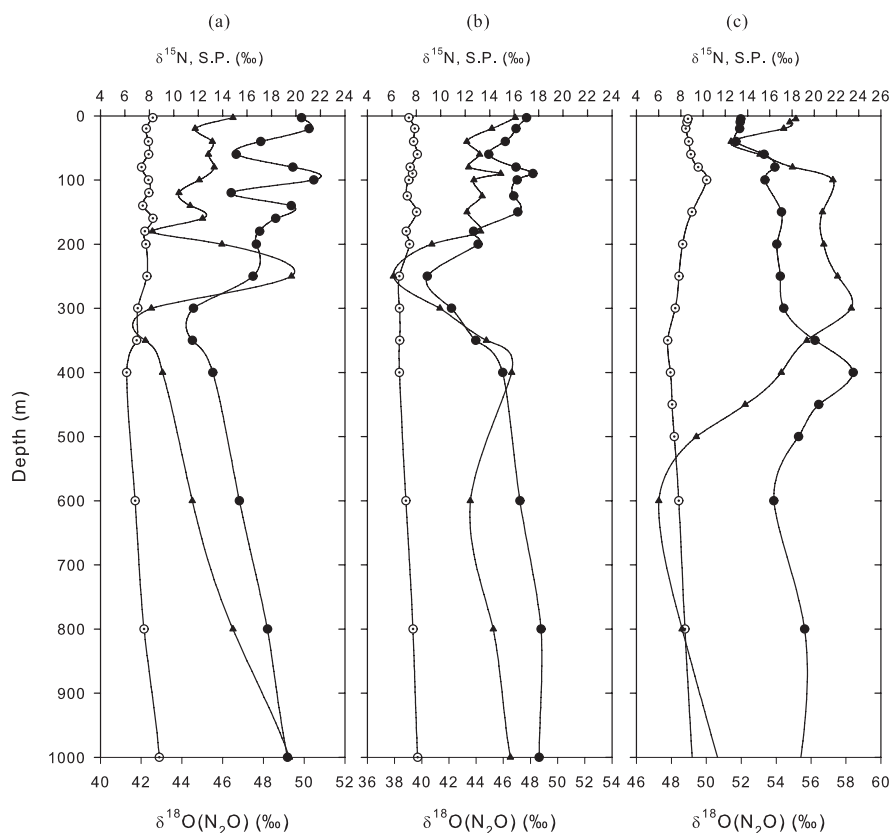


Fig. 7. Vertical profiles at stations (a): GYR, (b): EGY (c): UPX. The parameters shown are: $\delta^{15}\text{N}_{\text{bulk}}$, (open circle): Site preference (solid circle), $\delta^{18}\text{O}(\text{N}_2\text{O})$ (solid triangle).

bonds of the lightest isotopes during the intermediate hypoxic step should drive the $\delta^{18}\text{O}(\text{N}_2\text{O})$, as occurs with $^{15}\text{N}^{\alpha}$ enrichment (Schmidt et al., 2004). The source of the oxygen atom in N_2O produced by nitrite reduction is nitrite (Aeressens et al., 1986; Averill, 1996). Therefore, if this (nitrite) is produced in situ by ammonium oxidizers, it must be isotopically depleted compared to ammonium and O_2 . Thus, the resulting N_2O must be depleted in ^{18}O . This explains why $\delta^{18}\text{O}(\text{N}_2\text{O})$ is lighter in the deeper pycnocline, even when dissolved O_2 should be isotopically enriched due to the respiration of organic matter accumulated in this pycnocline, supporting nitrifier denitrification hypotheses as a N_2O source in these regions.

The behavior of SP and $\delta^{18}\text{O}(\text{N}_2\text{O})$ below the pycnocline at UPX is more complicated than at the oceanic stations. Maximum values of $\delta^{18}\text{O}(\text{N}_2\text{O})$ between 56‰ and 60‰ coincide with the O_2 minimum zone (OMZ). This can be interpreted as the influence of the isotopic enrichment of O_2 due to respiration. From this, it can be concluded that the main source of N_2O is nitrification. Otherwise, SP in the OMZ ($\approx 17\text{‰}$) is diminished related to the N_2O released by nitrification at the oceanic station ($\approx 22\text{‰}$), probably due to the influence of nitrifier denitrification. In any case, these two

processes could be running simultaneously. The SP maximum is observed at 400 m, coinciding with the nitrate maximum and the rising O_2 concentration below the OMZ, and slightly deeper than the N_2O maximum ($50.29 \text{ nmol Kg}^{-1}$). This suggests an enhancement of the extent of nitrification at this depth. It must be considered that the coastal zone is a highly dynamic environment, with significant meridional transport and isotopic, isotopomeric, and concentration values that could be influenced by non-local processes. In fact, minima of $\delta^{18}\text{O}(\text{N}_2\text{O})$, SP, and N_2O are found at UPX (600 m), as is an O_2 maximum; this parameter seems to be related to the core of a water mass identified at this depth ($S \approx 34.3$ and $T \approx 5.5^\circ\text{C}$).

3.4 Contribution of nitrifier denitrification

At GYR and EGY, it is possible to identify at least 3 sources of N_2O according to the shape of the SP profiles. A first source can be attributed to mixing with the air in the shallow mixed layer, where the average SP at both stations is about 17‰, slightly below the value expected for the atmosphere ($19 \pm 2\text{‰}$) (Yoshida and Toyoda, 2000). A second source is represented by the SP signal below 1000 m, which averaged 22‰, and a third source is located at the SP minimum

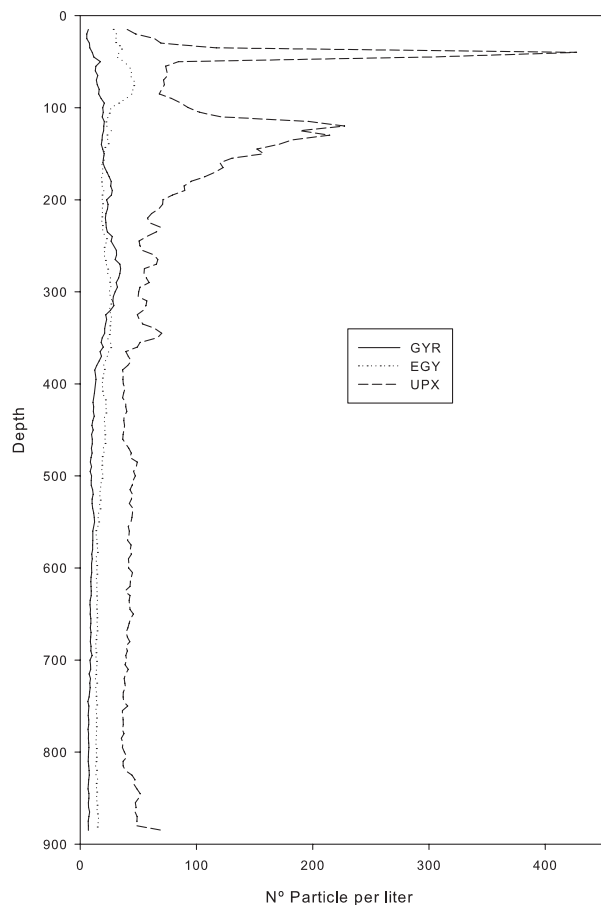


Fig. 8. Particle content at the stations : GYR (continuous line), EGY (dotted line) and UPX (dashed line).

located at 320 m and 250 m at the GYR and EGY stations, respectively.

Assuming that there is no significant contribution from air-sea transfer below the pycnocline, and that SP values below 1000 m are representative only of ammonium oxidation, it is possible to build a simple two-box model to elucidate the contribution of ammonium oxidation and nitrite reduction to the minimum SP zone at GYR and EGY using SP values. The fraction of N_2O produced by nitrifier denitrification (F) can be determined as follows:

$$F = \frac{SP_{IM} - SP_{ao}}{SP_{nr} - SP_{ao}} \quad (6)$$

where SP_{IM} is the SP value at the site preference minimum, SP_{ao} is the SP value attributable to ammonium oxidation (equal to the SP value below 1000 m), and SP_{nr} is the SP value attributable to nitrite reduction, equal to zero, based on the biochemical literature discussed above, and the values obtained in cultures with induced nitrifier denitrification (Sutka et al., 2004; Sutka et al., 2006). The assumption that

SP values below 1000 m are representative of nitrification is supported by the $\delta^{15}N^{\alpha}$ enrichment predicted for this process and the agreement of the nitrate-phosphate ratio with the Redfield ratio for those stations shown in Fig. 5. If we assume that our hypothesis about N_2O production through nitrifier denitrification is correct, the calculated F values of 0.4 and 0.5 (GYR and EGY, respectively) indicate that nitrifier denitrification pathways can be an important source of N_2O , even in oligotrophic and well-oxygenated waters.

At UPX, the N_2O production processes seems to be more complicated than at the oceanic stations. Coastal environments are commonly highly dynamic and highly influenced by horizontal advection. Because of this, the model applied here to the oceanic station should not be valid for the coastal station. Nevertheless, nitrifier denitrification could be an important source of N_2O to the atmosphere, in view of the SP-depleted N_2O present in the mixed layer (related to the expected atmospheric SP values).

4 Conclusions and perspectives

Our results show that N_2O sources in the ocean are far from being identified. Even in very stable oligotrophic environments, different mechanisms could coexist. In this sense, new tools like N_2O isotopomer determination provide new information about nitrogen cycling processes.

The SP signal supporting the evidence of the occurrence of nitrifier denitrification in highly oligotrophic environments is one of the most striking contributions of this work. Particular chemical dynamics of this process under such conditions, as well the role of particles should be the subject of further challenging studies.

The importance of the occurrence of nitrifier denitrification in oceanic environments for the whole nitrogen cycle may be enormous, considering that the Central South Pacific Gyre represents a major part of the South Pacific. Moreover, such conditions are found in the other oceans, thus rendering this process a potentially significant contributor to the oceanic N_2O budget.

Acknowledgements. Dominique Tailliez and Claudie Bournot are warmly thanked for their efficient help in CTD rosette management and data processing. The authors thank G. Gorsky and M. Picheral for allowing us to use their UVP data; O. Ulloa, G. Alarcon, M. Gallegos, and the crew of the R. V. L'Atalante for their help during the BIOSOPE Cruise. J. Charpentier thanks V. Molina, A. Montecinos, and S. Hormazabal for their help during the writing of this manuscript. This is a contribution of the BIOSOPE project of the LEFE-CYBER program. This research was funded by the Centre National de la Recherche Scientifique (CNRS), the Institut des Sciences de l'Univers (INSU), the Centre National d'Etudes Spatiales (CNES), the European Space Agency (ESA), the National Aeronautics and Space Administration (NASA), and the Natural Sciences and Engineering Research Council of Canada (NSERC). Financial assistance was provided by the FONDAP-COPAS center (Project No. 150100007). J. Charpentier was supported by a grant

from the MECESUP UCO002 project. This work was supported by the Chilean National Commission for Scientific and Technological Research through FONDECYT grant 1050743 (L. Farias).

Edited by: S. W. A. Naqvi

References

- Aeressens, E., Tiedje, J., and Averill, B.: Isotope labelling studies on the mechanism of N-N bond formation in denitrification, *J. Biol. Chem.*, 261, 9652–9656, 1986.
- Allredge, A. L. and Cohen, Y.: Can Microscale Chemical Patches Persist in the Sea? Microelectrode Study of Marine Snow, Fecal Pellets, *Science*, 235, 689–691, 1987.
- Armstrong, F. A., Stearns, C. R., and Strickland, D. H.: The measurement of upwelling and subsequent biological processes by means of the Technicon Autoanalyzer and associated equipment, *Deep Sea Res.*, 14, 381–389, 1967.
- Averill, B. A.: Dissimilatory Nitrite and Nitric Oxide Reductases, *Chem. Rev.*, 96, 2951–2964, 1996.
- Averill, B. A. and Tiedje, J. M.: The chemical mechanism of microbial denitrification, *FEBS Lett.*, 138, 8–12, 1982.
- Bonnet, S., Guieu, C., Bruyant, F., Pril, O., Van Wambeke, F., Raimbault, P., Grob, C., Moutin, T., Gorbunov, M. Y., Zehr, J. P., Masquelier, S. M., Garczarek, L., and Claustre, H.: Nutrients limitation of primary productivity in the Southeast Pacific (BIOCOPE cruise), *Biogeosciences Discuss.*, 4, 2733–2759, 2007, <http://www.biogeosciences-discuss.net/4/2733/2007/>.
- Carlucci, A. F. and McNally, P. M.: Nitrification by Marine Bacteria in Low Concentrations of Substrate and Oxygen, *Limnol. Oceanogr.*, 14, 736–739, 1969.
- Claustre, H. and Maritorena, S.: OCEAN SCIENCE: The Many Shades of Ocean Blue, *Science*, 302, 1514–1515, 2003.
- Castro, M. and Farias, L.: N₂O cycling at the core of the oxygen minimum zone off northern Chile, *Mar. Ecol. Prog. Ser.*, 280, 1–11, 2004.
- Codispoti, L. A., Brandes, J. A., Christensen, J. P., Devol, A. H., Naqvi, S. W. A., Paerl, H. W., and Yoshinari, T.: The oceanic fixed nitrogen and nitrous oxide budgets: Moving targets as we enter the anthropocene?, *Sci. Mar.*, 65, 85–105, 2001.
- Cohen, Y. and Gordon, L. I.: Nitrous oxide production in the ocean, *J. Geophys. Res.*, 84, 347–353, 1979.
- Cornejo, M., Farias, L., and Paulmier, A.: Temporal variability in N₂O water content and its air-sea exchange in an upwelling area off central Chile (36 degrees S), *Mar. Chem.*, 101, 85–94, 2006.
- Csanady, G. T.: Circulation in the coastal ocean. Dordrecht, D. Reidel, pp. 279, 1982.
- Csanady, G. T.: Mass Transfer to and from Small Particles in the Sea, *Limnol. Oceanogr.*, 31, 237–248, 1986.
- Daneri, G., Dellarossa, V., Quinones, R., Jacob, B., Montero, P., and Ulloa, O.: Primary production and community respiration in the Humboldt Current System off Chile and associated oceanic areas, *Mar. Ecol. Prog. Ser.*, 197, 41–49, 2000.
- Daneri, G. and Quinones, R.: Undersampled Ocean systems : a plea for an international study of biogeochemical cycles in the Southern Pacific Gyre and its boundaries, *U.S. JGOFS Newsletter*, 11, 9, 2001.
- Dore, J. E., Popp, B. N., Karl, D. M., and Sansone, F. J.: A large source of atmospheric nitrous oxide from subtropical North Pacific surface waters, *Nature*, 396, 63–66, 1998.
- Elkins, J. E., Wofsy, S. C., McElroy, M. B., Kolb, C. E., and Kaplan, W. A.: Aquatic sources and sinks for nitrous oxide., *Nature*, 275, 602–606, 1978.
- Emery, W. J. and Meincke, J.: Global Water masses: summary and review, *Oceanol. Acta*, 9, 383–391, 1986.
- Goreau, T. J., Kaplan, W. A., Wofsy, M. B., McElroy, M. B., Valois, F. W., and Watson, S. W.: Production of NO₂⁻ and N₂O by nitrifying bacteria at reduced concentrations of oxygen, *Appl. Environ. Microbiol.*, 40, 526–532, 1980.
- Gorsky, G., Picheral, M., and Stemmann, L.: Use of the Underwater Video Profiler for the Study of Aggregate Dynamics in the North Mediterranean, *Est. Coast. Shelf Sci.*, 50, 121–128, 2000.
- Gruber, N. and Sarmiento, J. L.: Global patterns of marine fixation and denitrification, *Global Biogeochem. Cycles*, 11, 235–266, 1997.
- Hallam, S. J., Mincer, T. J., Schleper, C., Preston, C. M., Roberts, K., Richardson, P. M., and DeLong, E. F.: Pathways of Carbon Assimilation and Ammonia Oxidation Suggested by Environmental Genomic Analyses of Marine Crenarchaeota, *PLoS Biology*, 4, e95, 2006.
- Hendriks, J., Oubrie, A., Castresana, J., Urbani, A., Gemeinhardt, S., and Saraste, M.: Nitric oxide reductases in bacteria, *Biochim. Biophys. Acta*, 1459, 266–273, 2000.
- Kartal, B., Kuypers, M. M. M., Lavik, G., Schalk, J., Op den Camp, H. J. M., Jetten, M. S. M., and Strous, M.: Anammox bacteria disguised as denitrifiers: nitrate reduction to dinitrogen gas via nitrite and ammonium, *Environ. Microbiol.*, 9, 635–642, 2007.
- Kim, K. R. and Craig, H.: Two-isotope characterization of N₂O in the Pacific Ocean and constraints on its origin in deep water, *Nature*, 347, 58–61, 1990.
- Kjørboe, T., Ploug, H., and Thygesen, U.: Fluid motion and solute distribution around sinking aggregates. I. Small-scale fluxes and heterogeneity of nutrients in the pelagic environment, *Mar. Ecol. Prog. Ser.*, 211, 1–13, 2001.
- Knowles, R.: Denitrification, *Microbiol. Rev.*, 46, 43–70, 1982.
- Konneke, M., Bernhard, A. E., de la Torre, J. R., Walker, C. B., Waterbury, J. B., and Stahl, D. A.: Isolation of an autotrophic ammonia-oxidizing marine archaeon, *Nature*, 437, 543–546, 2005.
- Law, C. S. and Ling, R. D.: Nitrous oxide flux and response to increased iron availability in the Antarctic Circumpolar Current, *Deep-Sea Res. II*, 48, 2509–2527, 2001.
- Liao, T., Camp, C., and Yung, Y.: The seasonal cycle of N₂O, *Geophys. Res. Lett.*, 31, L17108, doi:10.1029/2004GL020345, 2004.
- Manne, A. S. and Richels, R. G.: An alternative approach to establishing trade-offs among greenhouse gases, *Nature*, 410, 675–677, 2001.
- Morales, C., Blanco, J., Braun, M., Reyes, H., and Silva, N.: Chlorophyll-a distribution and associated oceanographic conditions in the upwelling region off northern Chile during the winter and spring 1993, *Deep Sea Res. I*, 43, 267–289, 1996.
- Murray, C. N. and Riley, J. P.: The solubility of gases in distilled water and sea water–II. Oxygen, *Deep Sea Res.*, 16, 311–320, 1969.
- Naqvi, S. W. A. and Noronha, R. J.: Nitrous oxide in the Arabian Sea, *Deep-Sea Res. A*, 38, 871–890, 1991.

- Naqvi, S. W. A., Yoshinari, T., Jayakumar, D. A., Altabet, M. A., Narvekar, P. V., Devol, A. H., Brandes, J. A., and Codispoti, L. A.: Budgetary and biogeochemical implications of N₂O isotope signatures in the Arabian Sea, *Nature*, 394, 462–464, 1998.
- Nevison, C., Butler, J. H., and Elkins, J. W.: Global distribution of N₂O and the Δ N₂O-AOU yield in the subsurface ocean, *Global Biogeochem. Cycles*, 17, 30–1–30–18, 2003.
- Nevison, C. D., Weiss, R. F., and Erickson, D. J., III.: Global oceanic emissions of nitrous oxide, *J. Geophys. Res.*, 100, 809–820, 1995.
- Ostrom, N. E., Russ, M. E., Popp, B., Rust, T. M., and Karl, D. M.: Mechanisms of nitrous oxide production in the subtropical North Pacific based on determinations of the isotopic abundances of nitrous oxide and di-oxygen, *Chemosphere - Global Change Sci.*, 2, 281–290, 2000.
- Oudot, C., Andrie, C., and Montel, Y.: Nitrous oxide production in the tropical Atlantic Ocean, *Deep-Sea Res.*, 37, 183–202, 1990.
- Oudot, C., Jean-Baptiste, P., Fourre, E., Mormiche, C., Guevel, M., Ternon, J.-F., and Le Corre, P.: Transatlantic equatorial distribution of nitrous oxide and methane, *Deep-Sea Res. I*, 49, 1175–1193, 2002.
- Ploug, H.: Small-Scale Oxygen Fluxes and Remineralization in Sinking Aggregates, *Limnol. Oceanogr.*, 46, 1624–1631, 2001.
- Ploug, H. and Kühl, M., Buchholz-Cleven, B. and Jørgensen, B.: Anoxic aggregates – an ephemeral phenomenon in the pelagic environment?, *Aquat. Microb. Ecol.*, 13, 285–294, 1997.
- Popp, B., Westley, M., Toyoda, S., Miwa, T., Dore, J., Yoshida, N., Rust, T., Sansone, F., Russ, M., Ostrom, N. and Ostrom, P.: Nitrogen and oxygen isotopomeric constraints on the origins and sea-to-air flux of N₂O in the oligotrophic subtropical North Pacific gyre, *Global Biogeochem. Cycles*, 16, 1064–1073, 2002.
- Poth, M. and Focht, D.: ¹⁵N Kinetic analysis of N₂O production by *Nitrosomonas europaea*: an examination of nitrifier denitrification, *Appl. Environ. Microbiol.*, 49, 1134–1141, 1985.
- Raimbault, P., Slawyk, G., Coste, B., and Fry, J.: Feasibility of using an automated colorimetric procedure for the determination of seawater nitrate in the 0 to 100 nM range: Examples from field and culture, *Mar. Biol.*, 104, 347–351, 1990.
- Ritchie, G. A. and Nicholas, D. J.: Identification of the sources of nitrous oxide produced by oxidative and reductive processes in *Nitrosomonas europaea*, *Biochem. J.*, 126, 1181–1191, 1972.
- Schmidt, H. L., Werner, R. A., Yoshida, N., and Well, R.: Is the isotopic composition of nitrous oxide an indicator for its origin from nitrification or denitrification? A theoretical approach from referred data and microbiological and enzyme kinetic aspects, *Rapid Commun. Mass Spectrom.*, 18, 2036–2040, 2004.
- Shaw, L. J., Nicol, G. W., Smith, Z., Fear, J., Prosser, J. I., and Baggs, E. M.: *Nitrosospira* spp. can produce nitrous oxide via a nitrifier denitrification pathway, *Environ. Microbiol.*, 8, 214–222, 2006.
- Silva, N. and Neshyba, S.: On the southernmost extension of the Peru-Chile undercurrent, *Deep Sea Res.*, 26, 1387–1393, 1979.
- Stein, L. Y. and Arp, D. J.: Loss of Ammonia Monooxygenase Activity in *Nitrosomonas europaea* upon Exposure to Nitrite, *Appl. Environ. Microbiol.*, 64, 4098–4102, 1998.
- Stemann, L., Picheral, M., and Gorsky, G.: Diel variation in the vertical distribution of particulate matter (>0.15 mm) in the NW Mediterranean Sea investigated with the Underwater Video Profiler, *Deep-Sea Res. I*, 47, 505–531, 2000.
- Strub, T., Mesias, J. M., Monecino, V., Rutlland, J., and Salinas, S.: Coastal Ocean Circulation off Western South America, in: *The Sea*, edited by: Robinson, A. R. and Brinks, K. H., New York, John Wiley and Sons, 11, 273–213 1995.
- Sutka, R., Ostrom, N., Ostrom, P., Gandhi, H., and Breznak, J.: Nitrogen isotopomer site preference of N₂O produced by *Nitrosomonas europaea* and *Methylococcus capsulatus* Bath, *Rapid Commun. Mass Spectrom.*, 17, 738–745, 2003.
- Sutka, R., Ostrom, N., Ostrom, P., Gandhi, H., and Breznak, J.: Erratum: Nitrogen isotopomer site preference of N₂O produced by *Nitrosomonas europaea* and *Methylococcus capsulatus* Bath, *Rapid Commun. Mass Spectrom.*, 18, 1411–1412, 2004.
- Sutka, R. L., Ostrom, N. E., Ostrom, P. H., Breznak, J. A., Gandhi, H., Pitt, A. J., and Li, F.: Distinguishing Nitrous Oxide Production from Nitrification and Denitrification on the Basis of Isotopomer Abundances, *Appl. Environ. Microbiol.*, 72, 638–644, 2006.
- Toyoda, S., Mutoke, H., Yamagishi, H., Yoshida, N., and Tanji, Y.: Fractionation of N₂O isotopomers during production by denitrifier, *Soil Biol. Biochem.*, 37, 1535–1545, 2005.
- Toyoda, S. and Yoshida, N.: Determination of Nitrogen Isotopomers of Nitrous Oxide on a Modified Isotope Ratio Mass Spectrometer, *Anal. Chem.*, 71, 4711–4718, 1999.
- Toyoda, S., Yoshida, N., Miwa, T., Matsui, Y., Yamagishi, H., Tsunogai, U., Nojiri, Y., and Tsurushima, N.: Production mechanism and global budget of N₂O inferred from its isotopomers in the western North Pacific, *Geophys. Res. Lett.*, 29, 25–28, 2002.
- Tréguer, P. and LeCorre, P.: Manuel d'analyses des sels nutritifs dans l'eau de mer (Utilisation de l'Autoanalyser II). Brest, Laboratoire de Chimie Marine, Université de Bretagne Occidentale, pp. 110 pp, 1975.
- Treusch, A. H., Leininger, S., Kletzin, A., Schuster, S. C., Klenk, H.-P., and Schleper, C.: Novel genes for nitrite reductase and Amo-related proteins indicate a role of uncultivated mesophilic crenarchaeota in nitrogen cycling, *Environ. Microbiol.*, 7, 1985–1995, 2005.
- Ward, B.: Nitrification and the marine nitrogen cycle, in: *Microbial Ecology of the Oceans*, edited by: Kirchman, D., New York, Wiley – Liss, 427–453, 2000.
- Wasser, I. M., de Vries, S., Moënné-Loccoz, P., Schröder, I., and Karlin, K.: Nitric Oxide in Biological Denitrification: Fe/Cu Metalloenzyme and Metal Complex NO_x Redox Chemistry, *Chem. Rev.*, 102, 1201–1234, 2002.
- Weeg-Aerssens, J. M., Tiedje, J. M., and Averill, B. A.: Evidence from isotope labeling studies for a sequential mechanism for dissimilatory nitrite reduction, *J. Am. Chem. Soc.*, 110, 6851–6856, 1988.
- Westley, M. B., Yamagishi, H., Popp, B. N., and Yoshida, N.: Nitrous oxide cycling in the Black Sea inferred from stable isotope and isotopomer distributions, *Deep Sea Res. II*, 53, 1802–1816, 2006.
- Wrage, N., Velthof, G. L., van Beusichem, M. L., and Oenema, O.: Role of nitrifier denitrification in the production of nitrous oxide, *Soil Biol. Biochem.*, 33, 1723–1732, 2001.
- Yamagishi, H., Miwa, T., Toyoda, S., Tsunogai, U., and Yoshida, N.: A method for the Measurement of Dissolved Nitrous Oxide Isotopomers in Natural Waters. 1st International Symposium on Isotopomers, Yokohama, Japan, 2001.
- Yamagishi, H., Yoshida, N., Toyoda, S., Popp, B., Westley, M.,

- and Watanabe, S.: Contributions of denitrification and mixing on the distribution of nitrous oxide in North Pacific, *Geophys. Res. Lett.*, 32, L04603, doi:10.1029/2006JG000227, 2005.
- Yoshida, N.: ^{15}N -depleted N_2O as a product of nitrification, *Nature*, 335, 528–529, 1988.
- Yoshida, N., Morimoto, H., Hirano, M., Koike, I., Matsuo, S., Wada, E., Saino, T., and Hattori, A.: Nitrification rates and ^{15}N abundances of N_2O and NO_3^- in the western North Pacific, *Nature*, 342, 895–897, 1989.
- Yoshida, N. and Toyoda, S.: Constraining the atmospheric N_2O budget from intramolecular site preference in N_2O isotopomers, *Nature*, 405, 330–334, 2000.
- Yoshinari, T.: Nitrous oxide in the sea, *Mar. Chem.*, 4, 189–202, 1976.
- Zafiriou, O., Hanley, Q., and Snyder, G.: Nitric oxide and nitrous oxide production and cycling during dissimilatory nitrite reduction by *Pseudomonas perfectomarina*, *J. Biol. Chem.*, 264, 5694–5699, 1989.
- Zuñiga, J., Bastida, A., and Requena, A.: Theoretical calculations of vibrational frequencies and rotational constants of the N_2O isotopomers, *J. Mol. Spectrosc.*, 217, 43–58, 2003.