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Petrology and U-Th-Pb geochronology of the sapphirine-quartz-bearing metapelites from Rajapalayam, Madurai Block, Southern India: Evidence for polyphase Neoproterozoic high-grade metamorphism

I. Braun ^{a,*}, B. Cenki-Tok ^b, J.-L. Paquette ^c, M. Tiepolo ^d

Strongly migmatitic sapphirine-bearing granulites occur as dm- to m-sized enclaves in garnetiferous leucogranites in the southeastern part of the Madurai Block (MB), the largest section of Neoproterozoic crust in the Southern Granulite Terrain of India (SGT). Early mineral assemblages (biotite-sillimanite-quartz, garnet-biotite-quartz) break down via successive dehydration melting reactions to high-*T* and high-*P* phase assemblages (e.g. garnet-melt, orthopyroxene-sillimanite-melt, orthopyroxene-cordierite-melt). Melt-consuming reactions along the retrograde path led to partial resorption of garnet to orthopyroxene-sillimanite, orthopyroxene-sapphirine and/or orthopyroxene-cordierite as well as orthopyroxene-sillimanite assemblages to cordierite-biotite-quartz symplectites. The occurrence of UHT metamorphism is indicated by the presence of small inclusions of sapphirine-quartz intergrowths in garnet. This reaction sequence is interpreted in a KFMASH partial petrogenetic grid taking both melt-producing and melt-consuming reactions into account which yields a clockwise heating-cooling-decompression *P-T* path culminating at 8–11 kbar and 1000–1100 °C.

In-situ U-Th-Pb monazite dating has been performed to more precisely constrain the ages given by chemical electron-microprobe on monazite in various textural positions. Monazite inclusions in garnet often yield a bimodal distribution of early (950-850 Ma) and late Neoproterozoic (600-500 Ma) ages which is in good agreement with the results of U-Pb and EPMA zircon dating in other places of the southeastern Madurai Block. By contrast, monazite inclusions in other mafic minerals (orthopyroxene, cordierite, biotite) or as interstitial phase exclusively yield ages between 550 and 480 Ma. The early Neoproterozoic population is interpreted to reflect the timing of migmatisation and peak UHT metamorphism, whereas the late Neoproterozoic ages are inferred to date cooling and exhumation with granitic magmas to a higher crustal level of the Madurai Block.

Keywords: UHT metamorphism; South India; Sapphirine; Quartz; Monazite; U-Th-Pb geochronology

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1. Introduction

The equilibrium assemblage sapphirine+quartz provides evidence for so-called ultrahigh-temperature (UHT) metamorphism with P-T conditions of ca. 1000-1100 °C and ca. 8-12 kbar. Since their first reports (Dallwitz, 1968; Ellis, 1980) such UHT assemblages in lower crustal rocks have attracted wide interest as their extreme P-T conditions indicate unusual geodynamic settings (e.g. Harley, 1998). However, such rocks are often difficult to trace and to identify since the appropriate assemblages usually have been largely replaced by lower-T assemblages along the retrograde path and thus may only be preserved as small isolated bodies. Therefore, a major challenge arises when trying to decipher if such retrogressed UHT granulites reflect a single or multiple orogenic cycles. To address such a question requires the combined use of in-situ dating techniques and thermobarometry which allows the P-T-t conditions of formation of specific phase assemblages to be constrained.

In this paper, we provide a detailed description of the P-T evolution of sapphirine–quartz bearing granulites that have been sampled from a previously undescribed locality in the southeastern part of the Madurai Block, near Rajapalayam in southern India. Tateishi

et al. (2004) recently published a first account on the petrology of these rocks and could show that they witnessed UHT conditions. Braun and Appel (2004, 2006) carried out detailed EPMA monazite dating of the granulites and their host rocks. Their results revealed a bimodal distribution of Mid- to early Neoproterozoic and late Neoproterozoic ages in monazite inclusions in the granulite, whereas monazite in the magmatic host rocks exclusively yielded Neoproterozoic values. Based on these reconnaissance studies we present the outcome of U-Pb TIMS and in-situ LA-ICPMS dating of monazite from these rocks in order to constrain the individual ages more precisely and also to check the validity of the early Neoproterozoic values (800-900 Ma) which so far have not been reported from this part of the Madurai Block. Along with this we present new thermobarometric results and petrogenetic grids, which are used to constrain the petrogenesis of the UHT granulites and their role in the Proterozoic tectono-metamorphic evolution of the Madurai Block more comprehensively.

2. Geological setting

The Madurai Block (MB) forms the largest crustal domain in the Proterozoic part of the Southern Granulite

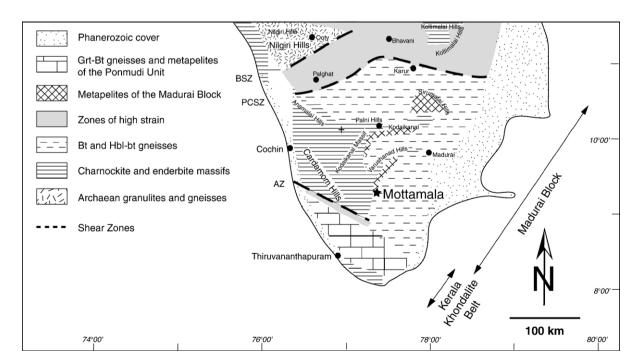


Fig. 1. Simplified geological map of southern India showing the main lithological units and structural elements as well as the position of the sample location (taken from: Braun and Appel (2006). BSZ — Bhavani Shear Zone; PCSZ — Palghat—Cauvery Shear Zone; AZ — Achankovil Zone.

Terrain and is limited to the north by a complex set of tectonic lineaments commonly referred to as the Palghat—Cauvery Shear Zone (PCSZ; Fig. 1).

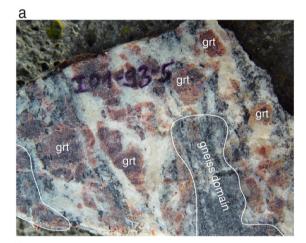
Details on the crustal architecture and the P-T-t evolution of the MB are rare. The Cardamon, Palni and Anamalai Hills (Fig. 1) predominantly consist of massive charnockites and enderbites. Their transition to the gneissic basement in the eastern part of the MB is marked by a steep slope along which numerous metasedimentary complexes occur. These predominantly consist of granulitic metapelites and grt—bt gneisses but also contain lesser quartzites, marbles and calc-silicate rocks.

Isotope and geochronological studies on basement rocks from the MB yield Late Archean to Paleoproterozoic Nd model ages ($T_{\rm DM}$ =ca. 2–3 Ga; Harris et al., 1994; Jayananda et al., 1995; Bartlett et al., 1995, 1998; Bhaskar Rao et al., 2003). Pb-Pb zircon evaporation and U-Pb TIMS zircon upper intercept ages between 2 and 2.4 Ga (Jayananda et al., 1995; Bartlett et al., 1995, 1998; Ghosh et al., 2004) are interpreted to reflect a stage of Late Paleoproterozoic magmatism and/or metamorphism. In addition, U-Pb and Pb-Pb zircon ages as well as the results of EPMA dating of monazite from various localities of the MB point to a polyphase Neoproterozoic metamorphic evolution and emphasize the strong imprint of Pan-African high-grade metamorphism (Bartlett et al., 1995; Jayananda et al., 1995; Miller et al., 1997; Bartlett et al., 1998; Santosh et al., 2003; Ghosh et al., 2004).

Tectonically, the Madurai Block is located between two W-E and WNW-ESE late (D3) high strain zones, the Palghat Cauvery Shear Zone (PCSZ) in the north and the Achankovil Zone (AZ) in the south (Cenki and Kriegsman, 2005). The PCSZ is dominated by dextral strike slip deformation (Chetty et al., 2003), whereas the AZ is an inclined zone dominated by intense flattening and pure shear, with a minor reverse component (Cenki and Kriegsman, 2005). The dominant foliation forming event (D1) is synchronous with peak metamorphism. An intermediate deformation phase (D2) is dominated by large scale upright folding of the S1 gneissic layering and affects the regional pattern of the MB. D2 trends are N-S in the western part of the MB (the Kerala highlands), and NE-SW in its north-eastern part (the Tamil Nadu lowlands). These trends are considered to be coeval forming an arc shape pattern (Fig. 1). The studied quarry close to Rajapalayam is located at the hinge of the easternmost inferred arc.

UHT conditions (~ 1000 °C) at pressures characteristic for the continental lower crust (8–11 kbar) were

mainly derived from sapphirine-bearing granulites from the Kodaikanal Massif but occurrences of UHT mineral assemblages are increasingly reported from other places in the MB (Mohan and Windley, 1993; Brown and Raith, 1996; Raith et al., 1997; Sajeev et al., 2004; Tateishi et al., 2004). These estimates of peak metamorphic conditions are slightly higher than those reported for rocks from the Kerala Khondalite Belt (KKB) (Cenki et al., 2002 and references therein) but similar to temperature estimates from marbles further southeast in the MB (Satish-Kumar, 2000). Contrary to this, the vast majority of the gneissic basement in the eastern MB preserves amphibolite-facies assemblages with rare granulite-facies assemblages. It is unclear whether the latter are relics of widespread retrogression



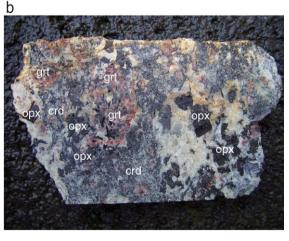


Fig. 2. Hand specimen of migmatitic granulites from the Mottamala quarry. a. Sample (I01-93-5) which has preserved gneissic domains and contains garnet-bearing leucosomes; b. Strongly restitic sample (I02-46-5) with incongruent euhedral orthopyroxene in the leucosome. Large garnet porphyroblast (left) shows partial replacement through coarse-grained cordierite and orthopyroxene. Width of the photographs is 15 cm.

under amphibolite-facies conditions or if granulite-facies conditions were achieved only locally. In addition, the regional significance of UHT assemblages remains ambiguous and it is not known if these reflect the P-T evolution of a different crustal fragment which developed independently from the rocks which now seem to host the UHT granulites.

3. Petrography

The sapphirine-bearing granulites presented in this study were found in several active quarries close to the village Mottamala which is situated *ca.* 1 km to the east of Rajapalayam (Fig. 1). They occur as meter-sized enclaves in garnetiferous leucogranites that intruded into granitic (garnet-)biotite gneisses and charnockites. Structures indicate a high degree of migmatisation and melt extraction resulting in strongly restitic and coarse-grained metapelitic granulites (Fig. 2). Folded leucosomes within the enclaves indicate that the metapelites were migmatised prior to leucogranite magma emplacement. This raises the question, whether the migmatisation is related to the generation of the granites or if it represents an earlier and previously unrecognized stage of high-grade metamorphism in this part of the MB.

Hand specimen of the weakly deformed metapelitic granulites display a large textural variability which is controlled by the degree of melt segregation and mineral-melt back reactions (Fig. 2a and b). Those samples that experienced significant melt segregation are dark blueish-black rocks composed of cm-sized cordierite and orthopyroxene crystals with homogeneously distributed sillimanite needles. Samples that still contain remnants of the in-situ melt, host a first generation of large (≤ 1 cm) euhedral orthopyroxene, which most likely formed as an incongruent phase during partial melting. Garnet and biotite are also present in small amounts in most of these samples. Garnet locally forms porphyroblasts but mostly appears as small, relic grains rimmed by retrograde cordierite and, rarely, orthopyroxene. Biotite is present in the mafic domains of all samples and either occurs randomly distributed and finegrained, or forms medium-grained cluster as breakdown product of cordierite and orthopyroxene.

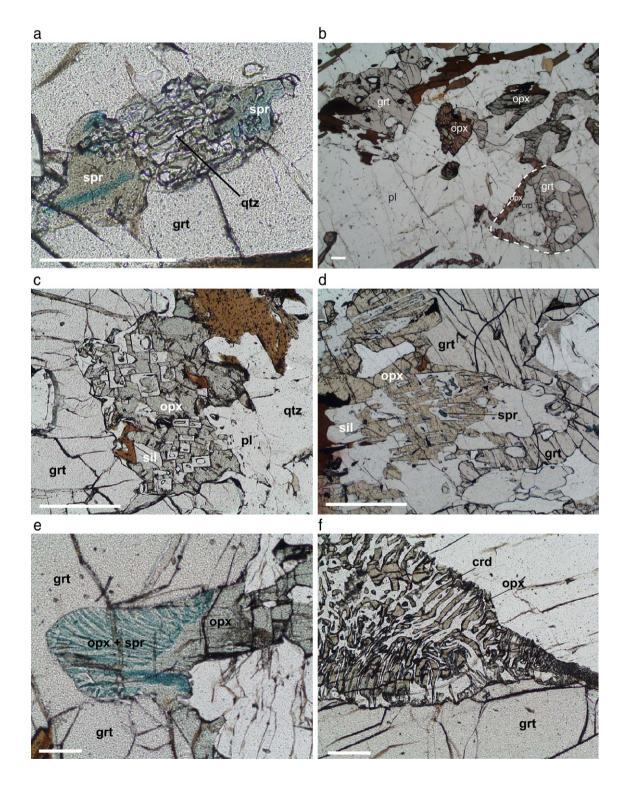
In thin section, poikiloblastic garnet hosts abundant inclusions of biotite, plagioclase, quartz, opaque and accessory phases. Most spectacular among these inclusion assemblages is the occurrence of tiny (<500 µm) sapphirine-quartz intergrowths (Fig. 3a). Sapphirine is strongly pleochroic, locally optically zoned and contains rare sillimanite inclusions. Locally spinel is associated to sapphirine and quartz without forming intergrowths. Poikilitic cordierite contains inclusions of biotite, sillimanite, garnet, sapphirine and rarely spinel. Large mm-sized orthopyroxene is locally present and contains rare small sillimanite and biotite inclusions. Accessory phases are spinel, zircon, monazite, ilmenite and occasionally rutile. The latter exclusively occurs as inclusion in garnet, whereas the other minerals can be found both as inclusions (predominantly garnet, but also in orthopyroxene, sillimanite and, rarely, quartz and alkali feldspar) and along grain boundaries in the restitic parts and the leucosomes.

The investigated samples display a large variety of reaction textures that are interpreted to reflect the multistage consumption of garnet, orthopyroxene-sillimanite and orthopyroxene-cordierite assemblages through repeated interaction with melt and fluid phases. Generally restitic domains have a low amount of quartz whereas retrograde reactions that may have involved a melt phase occur close to leucosome areas. Garnet is commonly strongly embayed and surrounded by symplectites and coronas of variable mineralogy: orthopyroxene and sillimanite (Fig. 3c), orthopyroxene, sillimanite and sapphirine (Fig. 3d), orthopyroxene and sapphirine (Fig. 3e), and orthopyroxene and cordierite (sometimes forming coarse intergrowth mimicking a garnet shape (Fig. 3b), sometimes along cracks in garnet, occasionally with sapphirine inclusions in cordierite; Fig. 3f and g). By contrast, garnet is euhedral at the contact with the leucosome. Quartz is usually found in the leucosome. It forms large anhedral grains and shows undulose extinction. Large orthopyroxene and sillimanite are separated by cordierite (Fig. 3h), cordierite-quartz±sapphirine symplectites, or cordierite-biotite symplectites (Fig. 3i). Locally double coronas (e.g. orthopyroxene-sillimanite symplectites between garnet and cordierite-biotite or orthopyroxene-cordierite symplectites between garnet and biotite) developed with grain-size coarsening from

Fig. 3. Microphotographs (plane polarized light) showing the various reaction textures preserved in the granulites. Scale (lower left corner of each picture) for all photographs is 200 μm. a. Inclusion of a sapphirine–quartz intergrowth in garnet; b. partial replacement of garnet through orthopyroxene+cordierite. The dotted line delineates the shape of the replaced garnet crystal; c. partial replacement of garnet through orthopyroxene+sillimanite; d. strongly embayed garnet relic which is rimmed by symplectitic intergrowths of orthopyroxene, sillimanite and sapphirine; e. symplectitic intergrowth of orthopyroxene and cordierite replacing garnet; g. partial resorption of garnet at the expense of orthopyroxene, cordierite and sapphirine; h. growth of cordierite between orthopyroxene and sillimanite; i. formation of late cordierite and biotite along orthopyroxene–sillimanite grain boundaries and around garnet; j. coronitic growth of cordierite around sapphirine grains in contact with garnet; k. late breakdown of biotite to ilmenite–cordierite intergrowths.

the leucosome towards the garnet. Tiny halos of cordierite between garnet and sapphirine crystals indicate garnet and sapphirine breakdown to cordierite in the presence of melt (Fig. 3j). Orthopyroxene and/or cordierite are separated

from the leucosome by irregular biotite aggregates indicating a breakdown of both the first and second generation of orthopyroxene and cordierite into biotite. Additionally, tiny symplectitic intergrowths of ilmenite



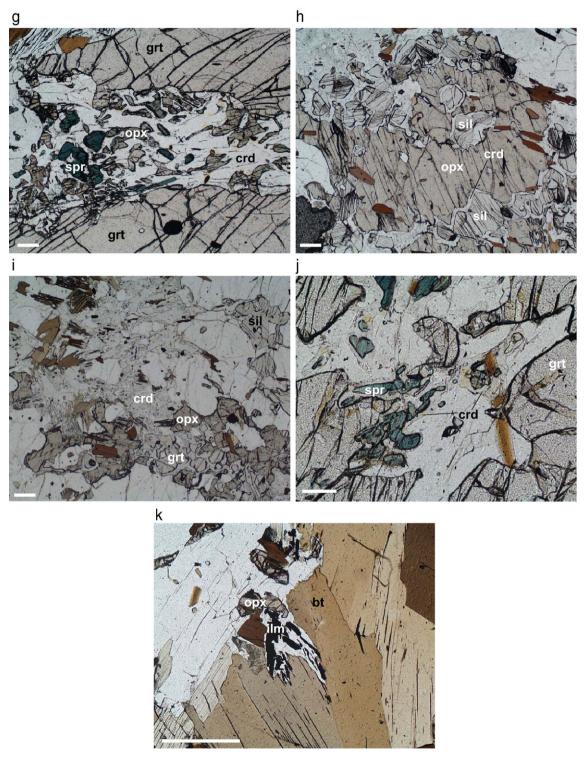


Fig. 3 (continued).

and cordierite form at the expense of biotite (Fig. 3k). As the rocks are strongly restitic, leucosomes are rare. In case they occur they are dominated by spindleperthite and quartz. Plagioclase is almost absent and does not play a key role in reaction textures.

4. Mineral chemistry

Mineral compositions have been determined with a Cameca Camebax Microbeam electron probe at the Department of Mineralogy, University of Bonn, operating at 20 kV accelerating voltage and 15 nA beam current. Counting times were 20 s (peak) and 10 s (background). Data processing was performed with the PAP correction procedure (Pouchou and Pichoir, 1984).

Cordierite has a homogeneous magnesium-rich composition ($X_{\rm Mg}$ =0.86–0.87) regardless of its textural position and the reaction through which it was formed. In contrast to this, different *orthopyroxene* generations can be easily distinguished by their chemical composition. Those grains that form subhedral porphyroblasts and symplectites together with sillimanite around garnet yield the highest magnesium abundances ($X_{\rm Mg}$ =0.68–0.73) and Al₂O₃ concentrations (9.22–9.56 wt.-%; Table 1) while intergrowths with cordierite as well as late opx that formed from biotite breakdown have distinctly lower values ($X_{\rm Mg}$ =0.66–0.67; Al₂O₃=7.12–7.25 wt.-%). Chemical zoning has not been observed in any of the different generations.

Garnet is a pyrope–almandine solid solution with both end-members in almost equal amounts ($X_{\rm Mg}$ =0.48–0.51) and very minor contributions from the other constituents. Chemical zoning is very weakly developed and expressed by slightly decreasing $X_{\rm Mg}$ values (\leq 0.05).

Biotite is TiO_2 -rich (5–5.5 wt.-%) and displays significant variations in X_{Mg} values according to its textural position. Biotite in the leucosome, which most likely crystallized from the granitic melt, yields X_{Mg} values of ca. 0.69. By contrast, those grains which rimorthopyroxene are significantly more magnesian ($X_{\text{Mg}} \sim 0.80$).

Sapphirine has a Mg-rich composition in between the 2:2:1 and 7:9:3 end-members and displays strongly varying $X_{\rm Mg}$ (=Mg/(Mg+Fe)) values between 0.76–0.82. These are mainly due to changes in Fe₂O₃ concentrations which are highly variable (3.18–6.80 wt.-%) and negatively correlated with Al₂O₃ and SiO₂. However, there is no observable correlation with its textural context as inclusions in garnet or replacing orthopyroxene–sillimanite intergrowths.

Alkali feldspar has a mesoperthitic composition with high anorthite contents (\sim 9 mol-%) which points to very high temperatures also during migmatisation.

5. Reaction sequence

The non-equilibrium phase assemblage of the Mottamala rocks comprises garnet (grt), orthopyroxene (opx), cordierite (crd), sapphirine (spr), spinel (spl), biotite (bt), ilmenite (ilm), alkali feldspar (kfs), quartz (qtz) and melt (liq) (mineral abbreviations after Kretz, 1983). Many of these minerals occur in one or more textural contexts and inferred generations. Following the method used to describe the Achankovil cordierite gneisses located further south in the Southern Granulite Terrain (Cenki et al., 2002), all the assemblages can be described in terms of melt-producing and melt-consuming reactions. In the following section, reactions are described in a KFMASH system with K-feldspar in excess (i.e. K-feldspar is not presented and balanced in the reactions).

5.1. Melt-producing reactions

Sapphirine–quartz (Fig. 3a) and sapphirine–spinel–quartz assemblages are only present as inclusions in garnet and therefore do not provide unequivocal textural evidence for the reaction sequence that was responsible for this association. There is evidence that at least some of the garnet was formed quite early at upper amphibolite to granulite-facies conditions during several melt-producing reactions. Small inclusions (10–200 µm) of biotite, quartz, alkali feldspar and fibrolitic sillimanite in garnet porphyroblasts indicate garnet formation through the KFMASH divariant dehydration melting reaction:

$$bt + sil + qtz + kfs = grt + liq. (1)$$

Large (0.2–1 mm) subhedral orthopyroxene crystals containing sillimanite and biotite inclusions may have formed through the multivariate KFMASH melting reaction:

$$bt + sil(+qtz + kfs) = opx + liq.$$
 (2)

Garnet is broken down at the contact with orthopyroxene and sillimanite crystals but is euhedral at the contact with the leucosome. Large incongruent orthopyroxene and sillimanite aggregates mimicking garnet shapes (Fig. 3c) may have formed via the invariant KFMASH garnet-consuming reaction:

$$grt + bt(+qtz + kfs) = opx + sil + liq.$$
 (3)

Large orthopyroxene and cordierite patches containing sillimanite and biotite inclusions (Fig. 3b) could

Table 1 Representative microprobe analyses of mineral phases from the granulite enclaves

Cordierite												
Sample	I02-40	I02-46-5b										
	11		12	1	8	21	21					
Na ₂ O	(0.01	0.03		0.01	0.0)4	0.03				
MgO	1	1.92	11.96	11.96 11.89			11.68					
Al_2O_3	3-	4.92	34.26		34.83	34.3	4	34.48				
SiO ₂	50	0.54	50.72		49.97	50.6	48.19					
FeO		3.09	3.08		3.04	3.2	3.52					
Σ	10	0.62	100.07		99.81	100.1	2	98.25				
Na		0.001	0.005		0.001	0.0	800	0.006				
Mg	174	4	1758		1754	1720		1796				
Al^{VI}	404	0	3984		4062	3999		4105				
Si	496	1	5004		4945	5000		4868				
Fe ²⁺		0.254	0.254		0.252	0.2	268	0.297				
cat.#	11,01		11,008	1	1,022	11,072		11,084				
$X_{ m Mg}$		0.87	0.87		0.87	0.8	37	0.86				
Orthopyroxen	e											
Sample	I02-46-5b						I02-46-5c					
	3 17		19	20	8	13	3	4				
	Rim to		Symplectitic i									
	grt	grt	crd	bt	sil							
SiO ₂	47.92	48.12	48.36	49.00	47.80	47.23	47.76	46.99				
TiO ₂	0.18	0.16	0.13	0.13	0.22	0.19	0.16	0.17				
Al_2O_3	9.22	9.53	7.25	7.12	9.56	9.56 9.39		9.26				
Cr_2O_3	0.04	0.02	0.06	0.00	0.03	0.03 0.00		0.11				
Fe_2O_3 (c)	3.48	1.23	2.27	1.07	1.68	2.54	4.37	3.78				
FeO (c)	15.64	17.51	19.49	19.07	18.08	17.14	15.47	16.60				
MnO	0.08	0.02	0.10	0.09	0.11	0.13	0.06	0.04				
MgO	23.37	22.51	21.41	22.03	21.88	21.97	23.35	22.19				
CaO	0.04	0.01	0.04	0.06	0.04	0.03	0.03	0.03				
Σ	99.98	99.12	99.15	98.61	99.43	98.65	100.07	99.21				
Si	1749	1772	1805	1827	1763	1755	1746	1740				
Ti	0.005	0.004	0.004	0.004	0.006	0.005	0.004	0.005				
Al^{IV}	0.251	0.228	0.195	0.173	0.237	0.245	0.254	0.260				
$\mathrm{Al}^{\mathrm{VI}}$	0.145	0.185	0.124	0.140	0.179	0.166	0.122	0.144				
Cr	0.001	0.001	0.002	0.000	0.001	0.000	0.004	0.003				
Fe^{3+}	0.096	0.034	0.064	0.030	0.047	0.071	0.120	0.105				
Fe ²⁺	0.477	0.539	0.608	0.594	0.558	0.533	0.473	0.514				
Mn	0.002	0.001	0.003	0.003	0.003	0.004	0.002	0.001				
Mg	1271	1235	1191	1224	1203	1217	1273	1225				
Ca	0.001	0.000	0.002	0.002	0.001	0.001	0.001	0.001				
cat.#	4000	4000	4000	4000	4000	4000	4000	4000				
Wo	0.08	0.02	0.10	0.14	0.08	0.08	0.06	0.06				
En	72.64	69.60	66.13	67.22	68.27	69.50	72.87	70.40				
Fs	27.28	30.38	33.78	32.64	31.65	30.42	27.08	29.54				
$X_{\rm Mg}$	0.73	0.70	0.66	0.67	0.68	0.70	0.73	0.70				

Table 1 (continued)

Garnet						
Sample	I02-46-5b	<u>102-46-5c</u>	I02-46-5b		102-46-50	
	7	1	2	14	2	
	Core		Rim to			
			opx	opx	opx	
SiO ₂	39.69	39.90	39.37	38.97	39.16	
TiO ₂	0.02	0.01	0.04	0.00	0.02	
Al_2O_3	22.79	22.71	23.02	22.72	22.96	
Cr ₂ O ₃	0.02	0.04	0.00	0.03	0.03	
Fe_2O_3 (c)	0.00	0.58	0.90	0.57	1.85	
FeO (c)	23.62	22.94	22.76	23.52	22.77	
MnO	0.26	0.21	0.30	0.23	0.26	
MgO	12.69	13.40	13.10	12.38	13.02	
CaO	0.56 0.53		0.55	0.60	0.43	
Σ	99.65 100.32		100.04	99.04	100.52	
Si	2992	2982	2955	2964	2933	
Ti	0.001	0.001	0.002	0.000	0.001	
Al^{IV}	0.008	0.018	0.045	0.036	0.067	
Al^{VI}	2016	1982	1990	2001	1960	
Cr	0.001	0.002	0.000	0.002	0.002	
Fe ³⁺	0.000	0.032	0.051	0.033	0.104	
Fe^{2+}	1489	1434	1428	1496	1427	
Mn	0.017	0.013	0.019	0.015	0.016	
Mg	1425	1493	1465	1404	1454	
Ca	0.045	0.042	0.044	0.049	0.035	
cat.#	7995	8000	8000	8000	8000	
Pyrope	47.89	50.06	49.55	47.36	49.60	
Almandin	50.02	48.08	48.32	50.48	48.67	
Spessartine	0.56	0.45	0.65	0.51	0.56	
Andradite	0.00	1.61	2.49	1.61	5.05	
Uvarovite	0.06	0.12	0.01	0.09	0.09	
Grossulaire	1.47	-0.30	-1.01	-0.05	-3.96	
X_{Mg}	0.49	0.51	0.51	0.48	0.50	
Alkalifeldspar						
Sample		I02-46	-5b		I02-46-5c	
r		1	<u> </u>		7	
		•				
					Rim to	
		Leucos			opx-sil	
SiO ₂		65.3			64.60	
Al_2O_3		20.8			20.20	
CaO		1.9			1.12	
Na ₂ O		6.5			6.11	
K ₂ O		6.2			8.87	
Σ		101.1	1		101.03	
Si		2902			2906	
Al		1091			1071	
Ca		0.0			0.054	
Na		0.5			0.533	
K		0.3	53		0.509	
cat.#		5006			5078	
Ab		55.6			48.63	
An		9.3	0		4.93	
Or		35.0	6		46.45	

(continued on next page)

Table 1 (continued)

Sample		I02-4	6-5c					
		5		6				
SiO ₂		36	19		37.27			
TiO ₂		5	5.38					
Al_2O_3		15	15.46					
Cr_2O_3		0	0.06					
Fe_2O_3 (c)		0	00		0.00			
FeO (c)		12	19		8.11			
MnO		0	0.00					
MgO		15	68		18.79			
CaO			01		0.00			
Na ₂ O			18		0.19			
K_2O		10	04		10.46			
H ₂ O			44		4.56			
Σ		99			100.56			
Si		2444			2449			
Ti			273		0.248			
Al ^{IV}			675		0.688			
Al ^{VI}			556		0.551			
Cr			003		0.004			
Fe ³⁺			000		0.000			
Fe ²⁺			689		0.446			
Mn			000		0.004			
		1579	000		1841			
Mg Co			001		0.000			
Ca								
Na			024		0.024			
K			865		0.877 2000			
OH			2000					
cat.#		9110	20		9132			
$X_{ m Mg}$		U	30		0.19			
Sapphirine	102.46.51							
Sapphirine Sample	I02-46-5b		0	15	16			
	4	5	9	15	16			
	4 Inclusion in							
Sample	4 Inclusion in grt	sil	sil	grt	grt			
Sample SiO ₂	Inclusion in grt	sil 12.08	sil 13.17	grt 13.26	grt 13.05			
Sample SiO ₂ TiO ₂	4 Inclusion in grt 12.56 0.06	sil 12.08 0.05	sil 13.17 0.07	grt 13.26 0.07	grt 13.05 0.13			
SiO ₂ TiO ₂ Al ₂ O ₃	4 Inclusion in grt 12.56 0.06 58.00	sil 12.08 0.05 57.75	sil 13.17 0.07 59.24	grt 13.26 0.07 59.74	grt 13.05 0.13 59.88			
SiO ₂ TiO ₂ Al ₂ O ₃ Cr ₂ O ₃	4 Inclusion in grt 12.56 0.06 58.00 0.13	sil 12.08 0.05 57.75 0.14	sil 13.17 0.07 59.24 0.12	grt 13.26 0.07 59.74 0.10	grt 13.05 0.13 59.88 0.07			
SiO ₂ TiO ₂ Al ₂ O ₃ Cr ₂ O ₃ Fe ₂ O ₃ (c)	4 Inclusion in grt 12.56 0.06 58.00 0.13 5.69	sil 12.08 0.05 57.75 0.14 6.80	sil 13.17 0.07 59.24 0.12 3.71	grt 13.26 0.07 59.74 0.10 3.18	grt 13.05 0.13 59.88 0.07 3.88			
SiO ₂ TiO ₂ Al ₂ O ₃ Cr ₂ O ₃ Fe ₂ O ₃ (c) FeO (c)	4 Inclusion in grt 12.56 0.06 58.00 0.13 5.69 6.53	sil 12.08 0.05 57.75 0.14 6.80 6.75	sil 13.17 0.07 59.24 0.12 3.71 8.56	grt 13.26 0.07 59.74 0.10 3.18 8.28	grt 13.05 0.13 59.88 0.07 3.88 7.49			
SiO ₂ TiO ₂ Al ₂ O ₃ Cr ₂ O ₃ Fe ₂ O ₃ (c) FeO (c) MnO	4 Inclusion in grt 12.56 0.06 58.00 0.13 5.69 6.53 0.05	sil 12.08 0.05 57.75 0.14 6.80 6.75 0.01	sil 13.17 0.07 59.24 0.12 3.71 8.56 0.00	grt 13.26 0.07 59.74 0.10 3.18 8.28 0.00	grt 13.05 0.13 59.88 0.07 3.88 7.49 0.00			
SiO ₂ TiO ₂ Al ₂ O ₃ Cr ₂ O ₃ Fe ₂ O ₃ (c) FeO (c) MnO MgO	4 Inclusion in grt 12.56 0.06 58.00 0.13 5.69 6.53 0.05 15.69	sil 12.08 0.05 57.75 0.14 6.80 6.75 0.01 15.21	sil 13.17 0.07 59.24 0.12 3.71 8.56 0.00 15.15	grt 13.26 0.07 59.74 0.10 3.18 8.28 0.00 15.39	grt 13.05 0.13 59.88 0.07 3.88 7.49 0.00 15.69			
Sample SiO ₂ TiO ₂ Al ₂ O ₃ Cr ₂ O ₃ Fe ₂ O ₃ (c) FeO (c) MnO MgO Σ	4 Inclusion in grt 12.56 0.06 58.00 0.13 5.69 6.53 0.05 15.69 98.73	sil 12.08 0.05 57.75 0.14 6.80 6.75 0.01 15.21 98.80	sil 13.17 0.07 59.24 0.12 3.71 8.56 0.00 15.15 100.02	grt 13.26 0.07 59.74 0.10 3.18 8.28 0.00 15.39 100.03	grt 13.05 0.13 59.88 0.07 3.88 7.49 0.00 15.69 100.24			
SiO ₂ TiO ₂ Al ₂ O ₃ Cr ₂ O ₃ Fe ₂ O ₃ (c) FeO (c) MnO MgO	4 Inclusion in grt 12.56 0.06 58.00 0.13 5.69 6.53 0.05 15.69 98.73 1536	sil 12.08 0.05 57.75 0.14 6.80 6.75 0.01 15.21	sil 13.17 0.07 59.24 0.12 3.71 8.56 0.00 15.15	grt 13.26 0.07 59.74 0.10 3.18 8.28 0.00 15.39	grt 13.05 0.13 59.88 0.07 3.88 7.49 0.00 15.69			
Sample SiO ₂ TiO ₂ Al ₂ O ₃ Cr ₂ O ₃ Fe ₂ O ₃ (c) FeO (c) MnO MgO Σ Si Ti	4 Inclusion in grt 12.56 0.06 58.00 0.13 5.69 6.53 0.05 15.69 98.73 1536 0.006	sil 12.08 0.05 57.75 0.14 6.80 6.75 0.01 15.21 98.80 1487 0.005	sil 13.17 0.07 59.24 0.12 3.71 8.56 0.00 15.15 100.02 1594 0.006	grt 13.26 0.07 59.74 0.10 3.18 8.28 0.00 15.39 100.03 1599 0.006	grt 13.05 0.13 59.88 0.07 3.88 7.49 0.00 15.69 100.24 1569 0.012			
Sample SiO ₂ TiO ₂ Al ₂ O ₃ Cr ₂ O ₃ Fe ₂ O ₃ (c) FeO (c) MnO MgO Σ Si Ti Al	4 Inclusion in grt 12.56 0.06 58.00 0.13 5.69 6.53 0.05 15.69 98.73 1536	sil 12.08 0.05 57.75 0.14 6.80 6.75 0.01 15.21 98.80 1487 0.005 8376	sil 13.17 0.07 59.24 0.12 3.71 8.56 0.00 15.15 100.02 1594 0.006 8450	grt 13.26 0.07 59.74 0.10 3.18 8.28 0.00 15.39 100.03 1599 0.006 8492	grt 13.05 0.13 59.88 0.07 3.88 7.49 0.00 15.69 100.24 1569 0.012 8487			
Sample SiO ₂ TiO ₂ Al ₂ O ₃ Cr ₂ O ₃ Fe ₂ O ₃ (c) FeO (c) MnO MgO Σ Si Ti Al Cr	4 Inclusion in grt 12.56 0.06 58.00 0.13 5.69 6.53 0.05 15.69 98.73 1536 0.006	sil 12.08 0.05 57.75 0.14 6.80 6.75 0.01 15.21 98.80 1487 0.005	sil 13.17 0.07 59.24 0.12 3.71 8.56 0.00 15.15 100.02 1594 0.006	grt 13.26 0.07 59.74 0.10 3.18 8.28 0.00 15.39 100.03 1599 0.006	grt 13.05 0.13 59.88 0.07 3.88 7.49 0.00 15.69 100.24 1569 0.012			
Sample SiO ₂ TiO ₂ Al ₂ O ₃ Cr ₂ O ₃ Fe ₂ O ₃ (c) FeO (c) MnO MgO Σ Si Ti Al Cr Fe ³⁺	4 Inclusion in grt 12.56 0.06 58.00 0.13 5.69 6.53 0.05 15.69 98.73 1536 0.006 8375	sil 12.08 0.05 57.75 0.14 6.80 6.75 0.01 15.21 98.80 1487 0.005 8376	sil 13.17 0.07 59.24 0.12 3.71 8.56 0.00 15.15 100.02 1594 0.006 8450	grt 13.26 0.07 59.74 0.10 3.18 8.28 0.00 15.39 100.03 1599 0.006 8492	grt 13.05 0.13 59.88 0.07 3.88 7.49 0.00 15.69 100.24 1569 0.012 8487			
Sample SiO ₂ TiO ₂ Al ₂ O ₃ Cr ₂ O ₃ Fe ₂ O ₃ (c) FeO (c) MnO MgO Σ Si Ti Al Cr	4 Inclusion in grt 12.56 0.06 58.00 0.13 5.69 6.53 0.05 15.69 98.73 1536 0.006 8375 0.013	sil 12.08 0.05 57.75 0.14 6.80 6.75 0.01 15.21 98.80 1487 0.005 8376 0.014	sil 13.17 0.07 59.24 0.12 3.71 8.56 0.00 15.15 100.02 1594 0.006 8450 0.011	grt 13.26 0.07 59.74 0.10 3.18 8.28 0.00 15.39 100.03 1599 0.006 8492 0.010	grt 13.05 0.13 59.88 0.07 3.88 7.49 0.00 15.69 100.24 1569 0.012 8487 0.007			
Sample SiO ₂ TiO ₂ Al ₂ O ₃ Cr ₂ O ₃ Fe ₂ O ₃ (c) FeO (c) MnO MgO Σ Si Ti Al Cr Fe ³⁺ Fe ²⁺	4 Inclusion in grt 12.56 0.06 58.00 0.13 5.69 6.53 0.05 15.69 98.73 1536 0.006 8375 0.013 0.525 0.669	sil 12.08 0.05 57.75 0.14 6.80 6.75 0.01 15.21 98.80 1487 0.005 8376 0.014 0.630 0.695	sil 13.17 0.07 59.24 0.12 3.71 8.56 0.00 15.15 100.02 1594 0.006 8450 0.011 0.338 0.867	grt 13.26 0.07 59.74 0.10 3.18 8.28 0.00 15.39 100.03 1599 0.006 8492 0.010 0.288 0.835	grt 13.05 0.13 59.88 0.07 3.88 7.49 0.00 15.69 100.24 1569 0.012 8487 0.007 0.351 0.753			
Sample SiO ₂ TiO ₂ Al ₂ O ₃ Cr ₂ O ₃ Fe ₂ O ₃ (c) FeO (c) MnO MgO Σ Si Ti Al Cr Fe ³⁺ Fe ²⁺ Mn	4 Inclusion in grt 12.56 0.06 58.00 0.13 5.69 6.53 0.05 15.69 98.73 1536 0.006 8375 0.013 0.525 0.669 0.005	sil 12.08 0.05 57.75 0.14 6.80 6.75 0.01 15.21 98.80 1487 0.005 8376 0.014 0.630 0.695 0.001	sil 13.17 0.07 59.24 0.12 3.71 8.56 0.00 15.15 100.02 1594 0.006 8450 0.011 0.338 0.867 0.000	grt 13.26 0.07 59.74 0.10 3.18 8.28 0.00 15.39 100.03 1599 0.006 8492 0.010 0.288 0.835 0.000	grt 13.05 0.13 59.88 0.07 3.88 7.49 0.00 15.69 100.24 1569 0.012 8487 0.007 0.351 0.753 0.000			
Sample SiO ₂ TiO ₂ Al ₂ O ₃ Cr ₂ O ₃ Fe ₂ O ₃ (c) FeO (c) MnO MgO Σ Si Ti Al Cr Fe ³⁺ Fe ²⁺	4 Inclusion in grt 12.56 0.06 58.00 0.13 5.69 6.53 0.05 15.69 98.73 1536 0.006 8375 0.013 0.525 0.669	sil 12.08 0.05 57.75 0.14 6.80 6.75 0.01 15.21 98.80 1487 0.005 8376 0.014 0.630 0.695	sil 13.17 0.07 59.24 0.12 3.71 8.56 0.00 15.15 100.02 1594 0.006 8450 0.011 0.338 0.867	grt 13.26 0.07 59.74 0.10 3.18 8.28 0.00 15.39 100.03 1599 0.006 8492 0.010 0.288 0.835	grt 13.05 0.13 59.88 0.07 3.88 7.49 0.00 15.69 100.24 1569 0.012 8487 0.007 0.351 0.753			

have formed via a multivariant KFMASH reaction related to the univariant KFMASH reaction:

$$bt + sil(+qtz) = opx + crd + liq.$$
 (4)

The occurrence of sapphirine—quartz inclusions in garnet (Fig. 3a) may be explained in three different ways. If garnet exclusively formed through reaction (1) it may be assumed that spr—qtz formed previously during a stage of UHT metamorphism and then became enclosed in garnet. We consider this hypothesis as unlikely since it would require that the symplectitic intergrowths of sapphirine and quartz did not recrystallize during cooling from UHT conditions and the subsequent garnet-forming reactions.

Alternatively, sapphirine—quartz intergrowths could have formed after the growth of garnet at UHT conditions and a possible mechanism to explain this would be the reaction between the host garnet and sillimanite inclusions following the multivariate KFMASH reaction:

$$grt(+sil) = spr + qtz.$$
 (5a)

Finally, sapphirine-quartz intergrowths and host garnets could both have formed at UHT conditions

and with increasing temperature in domains with high bulk Al/Si ratios. First, intergrowths were formed through the univariant KFMASH reaction

$$opx + crd + sil = spr + qtz + liq. (5b)$$

At even higher temperatures garnet (over)growth on them could have taken place in such domains according to the multivariate KFMASH reaction

$$opx + sil + qtz = grt. (5c)$$

5.2. Melt-consuming reactions

The large garnet grains always display strongly embayed grain boundaries and are rimmed by coronas of various breakdown assemblages. Among these, the most common are coarse-grained and occasionally euhedral orthopyroxene—sillimanite intergrowths (Fig. 3c and d) which formed according to the KFMASH divariant reaction

$$grt + liq(+spr) = opx + sil$$
 (6)

which is located at slightly higher pressure than the univariant reaction opx+sil+kfs=grt+spr+qtz+liq. In

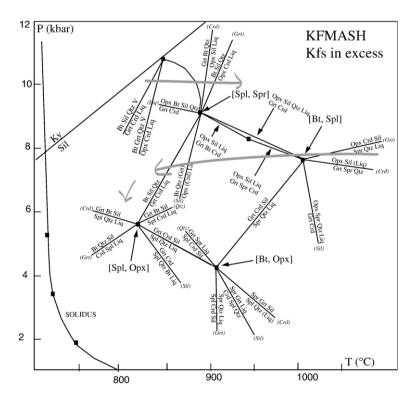


Fig. 4. Partial petrogenetic grid locating in P-T space the qualitative pseudosection presented in Fig. 5. The arrows indicate the inferred P-T path (see text for explanations). Abbreviations after Kretz (1983).

addition to this, garnet replacement has also led to the development of coarse to fine-grained symplectitic orthopyroxene—sapphirine (Fig. 3e) and orthopyroxene—cordierite-(sapphirine) assemblages (Fig. 3d, f, g). Appropriate multivariate KFMASH reactions are

$$grt + liq = opx + spr (7a)$$

and

$$grt + liq = opx + crd(+spr)$$
 (7b)

which are located on the lower pressure side of the orthopyroxene-sillimanite stability joint.

Locally, tiny halos of cordierite between sapphirine and garnet crystals (Fig. 3j) close to the quartzofeldspathic matrix indicate a multivariate reaction closely related to the KFMASH divariant reaction:

$$grt + spr + liq = crd + qtz.$$
 (7c)

Breakdown of the various orthopyroxene—sillimanite-bearing assemblages leads to the growth of cordierite as well as several symplectitic intergrowths involving cordierite and other phases through a variety of di- or trivariant melt-mineral KFMASH back reactions (Fig. 3h and i) located above the orthopyroxene—sillimanite joint:

$$opx + sil + liq = crd (8a)$$

$$opx + sil + liq = crd + bt(+qtz)$$
 (8b)

$$opx + sil + liq = crd + qtz.$$
 (8c)

Late biotite forms from the breakdown of orthopyroxene and cordierite in the presence of melt according to the high variance reactions

$$opx + liq = bt (9a)$$

and

$$\operatorname{crd} + \operatorname{liq} = \operatorname{bt}$$
 (9b)

which are related to the KFMASH univariant dehydration melting reaction bt+qtz+grt+kfs=opx+crd+liq located under the [spinel, sapphirine] KFMASH invariant point. However, locally, the reverse of Eq. (9a) is also observed and expressed by tiny orthopyroxene rims around biotite which indicates a reaction

$$bt + liq = opx + liq. (9c)$$

Finally, tiny symplectitic intergrowths of ilmenite and cordierite formed at the expense of biotite (Fig. 3k).

6. P-T data and petrogenetic grids

6.1. Construction of the petrogenetic grids

In addition to the melt phase that is present under ultra high temperature conditions, eight major phases are recognized: garnet, orthopyroxene, cordierite, sapphirine, spinel, biotite, k-feldspar and quartz. At least 6 major components are necessary to describe the system: $K_2O-FeO-MgO-Al_2O_3-SiO_2-H_2O$ (KFMASH). The choice of phases and chemical system is described in detail in Cenki et al. (2002).

The petrogenetic grids are presented in Figs. 4 and 5. The P-T location of KFMASH invariant points [spinel, sapphirine], [biotite, spinel] is constrained by stability calculations (Spear et al., 1999; Kelsey et al., 2004) and experimental results (Carrington and Harley, 1995). Most of the corresponding univariant lines are taken from experimental and topological data (McDade and Harley, 2001). The KFMASH univariant reactions bt+ sil+qtz=grt+crd+liq and bt+qtz+grt=opx+crd+liqhave the same P-T location as in Cenki et al. (2002). The remaining univariant lines and the divariant bundles around each of these critical univariant lines have been balanced with matrix analysis in the KFMASH system (using the following $X_{\rm Mg}$: garnet — 0.5, cordierite — 0.85, orthopyroxene — 0.7, sapphirine — 0.77) and built according to the geometrical rules of Schreinemaker analysis (Zen, 1966). The reaction sequence is well described around the [biotite, spinel] and [spinel, sapphirine] KFMASH invariant points. Due to the way the grid has been built, the reactions cordierite-absent ((crd)) and liquid-absent ((liq)) usually appear degenerate but in reality the divariant reactions (crd) and (lig) would involve a slight proportion of melt or cordierite, respectively. In addition, k-feldspar singularities are not treated in detail and k-feldspar is therefore considered in excess in the petrogenetic grids.

The location of the solidus, the kyanite–sillimanite boundary and the three related reactions involving a vapor phase (V) as reactant are taken from Kelsey et al. (2004).

6.2. Description of the P–T path and thermobarometry

Based on the calculated KFMASH grids, the interpretation of the mineral assemblages and reaction textures yields a clockwise P-T path, segmented in three parts: heating, cooling, decompression (Figs. 4 and 5).

During prograde heating, melt-producing reactions ((1)–(4) and (5b)) produce one or more incongruent phases (garnet, cordierite, orthopyroxene, sillimanite)

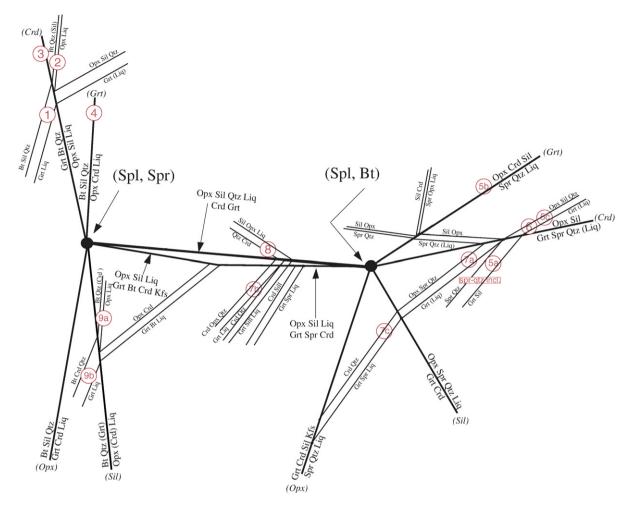


Fig. 5. Qualitative partial pseudosection in a PT space (P vertical axis, T horizontal axis) presenting the reactions observed in the sapphirine–quartz rocks from Mottamala. Invariant KFMASH points (points) and univariant KFMASH lines (thick lines) are taken from the literature (Carrington and Harley, 1995; McDade and Harley, 2001; Cenki et al., 2002; Kelsey et al., 2004). Divariant KFMASH fields (thin lines) built with Schreinemakers geometrical rules (Zen, 1966). Kfs is in excess. See text for further explanations.

along with melt above the [spl, spr] invariant point (i.e. within the stability field of orthopyroxene and sillimanite >8.8 kbar) and below the kyanite–sillimanite joint (Fig. 5). The early assemblage sapphirine–quartz formed at much higher T (in the biotite-out field) either from destabilization of sillimanite inclusions in garnet or the breakdown of orthopyroxene–sillimanite–cordierite assemblages.

The retrograde path (Fig. 5) is marked by a series of melt-consuming reactions indicating cooling related to the breakdown of garnet (reactions (6)–(7c)) or orthopyroxene–sillimanite assemblages (reactions (8a)–(8c)). These trends are located around the [spinel, biotite] invariant point and on both sides of the orthopyroxene–sillimanite joint. Reactions (6)–(7c) consume little melt as the crd- and liq-absent reactions are

almost degenerate. The shallow slope of this cooling path is further constrained by the absence of sapphirine—quartz retrograde assemblages and spinel in coronas or symplectites. The path has to be isobaric above the [spl, opx] and [bt, opx] invariant points and below the [spinel, sapphirine] and [spinel, biotite] invariant points (Fig. 4) until decompression takes place (i.e. below 6 kbar; 850 °C; Raith et al., 1997). This last stage is not visible in spinel-free assemblages but may be recorded in rocks that have a higher alumina content.

Breakdown of orthopyroxene and cordierite in the presence of melt (reactions (9a) and (9b)) occurs at the end of this cooling path and indicates the onset of decompression. Evidences for the shape of the decompression path are rare. It may follow an isothermal

Table 2
Results of Al-in-orthopyroxene coexisting with garnet thermobarometry (Harley and Green, 1982)

	Garnet analyse	es	
	2	7	14
X_{Ca}	0.01	0.02	0.02
$X_{\rm Fe}$	0.49	0.50	0.51
X_{Mg}	0.49	0.48	0.47
Fe/Mg	0.99	1.04	1.08
	Orthopyroxen	e analyses	
	3	8	13
$X_{Al,M1}$	0.20	0.21	0.21
$X_{\rm Fe}$	0.29	0.32	0.32
$X_{ m Mg}$	0.71	0.68	0.68
Fe/Mg	0.41	0.46	0.47
K	0.17	0.17	0.17
ΔVr	-212.10	-213.03	-212.80
T (°C)	1075	1075	1075
T(K)	1348	1348	1348
P (kbar)	11.32	10.74	10.77

decompression arrow parallel to the univariant reaction bt+sil+qtz=grt+crd+kfs+liq on its low temperature side.

Semi-quantitative estimations (Fig. 4) yield a heating-cooling P-T path culminating at 8–11 kbar and 1000–1100 °C. Thermobarometric calculations using aluminium in orthopyroxene coexisting with garnet (Harley and Green, 1982) gave estimations for the peak of metamorphism at ~11 kbar at 1075 °C (Table 2).

7. Geochronology

Electron-microprobe (EPMA) dating of monazite was carried out on the thin sections which were prepared for drilling of single monazite grains. The main purpose of this step was to identify grains that host a pre-Pan-African (>600 Ma) age population, which then could be chosen for more precise U-Pb TIMS dating. We therefore concentrated on samples which previously have been investigated carefully by EPMA dating (Braun and Appel, 2006). The results of these investigations confirmed the earlier studies and show that the host granite of the enclaves (sample I02-8-5a) exclusively contains monazite with Neoproterozoic ages (470–570 Ma). By contrast, monazites in the two granulite samples (I01-93-5, I02-46-5) display a large range of late to early Neoproterozoic ages (Braun unpubl. data; Braun and Appel, 2006).

According to this, U-Pb ID-TIMS (method after Paquette and Pin, 2001, all the errors quoted at the 2σ

level) dating of separated monazite grains from the granite sample I02-8-5a gave four individual but concordant Neoproterozoic III to Cambrian ages between 551 ± 1 and 493 ± 1 Ma (Fig. 6, Table 3). They are in perfect agreement with similar concordant U-Pb ID-TIMS ages $(551\pm1-481\pm5 \text{ Ma})$ from micro-drilled monazite inclusions (method described in Paquette et al., 2004) in orthopyroxene and quartz from the spr-qtz-bearing granulite sample I02-46-5 (Fig. 6). A monazite inclusion in garnet (Fig. 7a) was analysed in the granulite sample I01-93-5 for U-Th-Pb geochronology with laser ablation (LA)-ICPMS. The instrument couples a 193 nm ArF excimer laser (type GeoLas102-MicroLas) with a sector filed ICPMS (type Element I from ThermoFinnigan). The analytical method is basically that developed for zircon by Tiepolo (2003) and later modified for monazite by Paquette and Tiepolo (in press). Five spot analyses using a spot size of 5 µm, 12 J cm⁻² of energy density and 3 Hz of repetition rate of the laser were carried out on the monazite crystal. Time resolved signals were carefully inspected to evaluate the presence of common Pb, inclusions or perturbations related to fractures. Data reduction was carried out with the Glitter software (van Achterbergh et al., 2001) and all analytical errors were fully propagated. Each analysis is thus retained accurate within the quoted errors (Table 4). The 5 analyses give consistent Th–Pb and U–Pb ages. The mean 208 Pb/ 232 Th age is $501\pm12\,$ Ma ($2\sigma,\,$ MSWD=0.98), the mean 206 Pb/ 238 U age is 497 ± 10 Ma $(2\sigma, MSWD=1.5)$ and the three concordant U-Pb ages (Fig. 7b, Table 4) yield a mean concordia age of 497 ± 12 Ma $(2\sigma, MSWD=9.2)$. In contrast to this, U-Pb ID-TIMS dating of a micro-drilled

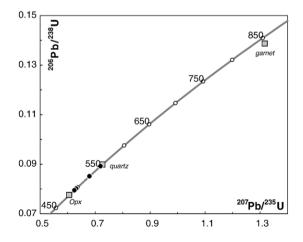


Fig. 6. Concordia diagram showing the results of U-Pb monazite dating. Black dots represent monazite grains from the granite (sample I02-8-5a) that hosts the metapelitic enclaves, squares are monazite inclusions in minerals from the enclave (sample I02-46-5).

Table 3
Results of U-Pb TIMS dating of monazite grains from the granite host (mnz 1-4) and the granulite (mnz 5-7)

No.	Wt.	U	Pb rad	²⁰⁶ Pb	²⁰⁸ Pb	²⁰⁶ Pb	²⁰⁷ Pb	²⁰⁷ Pb	²⁰⁶ Pb	²⁰⁷ Pb	²⁰⁷ Pb	Correlation
	(μg)	(ppm)	(ppm)	²⁰⁴ Pb	²⁰⁶ Pb	²³⁸ Pb	²³⁵ Pb	²⁰⁶ Pb	²³⁸ Pb	²³⁵ Pb	²⁰⁶ Pb	Coefficient
						Atomic ra	atios	Appare	ent ages (
mnz1 I02-8-5a	13	1064	1338	5382	15.898	0.08515 (0.16%)	0.6796 (0.27%)	0.05789 (0.20%)	527	527	526	0.66
mnz2 I02-8-5a	20	3365	1021	24833	3.8653	0.08919 (0.14%)	0.7199 (0.15%)	0.05855 (0.05%)	551	551	550	0.95
mnz3 I02-8-5a	10	576	622	534	13.912	0.08015 (0.21%)	0.6315 (0.37%)	0.05714 (0.30%)	497	497	497	0.60
mnz4 I02-8-5a	11	701	506	3537	9.3633	0.07948 (0.17%)	0.6246 (0.28%)	0.05700 (0.22%)	493	493	492	0.64
mnz5 I02-46-5 opx	1	757	1625	819	30.202	0.07756 (1.06%)	0.6055 (1.97%)	0.05662 (1.59%)	482	481	477	0.59
mnz6 I02-46-5 qtz	1	1960	1035	4565	5.6915	0.08987 (0.15%)	0.7266 (0.23%)	0.05864 (0.17%)	555	555	554	0.68
mnz7 I01-93-5 grt	5	3486	1699	21,291	2.9842	0.1387 (0.15%)	1.3168 (0.15%)	0.06884 (0.04%)	837	853	894	0.96

Monazite grains were not mechanically abraded. The isotopic ratios are corrected for mass discrimination $(0.1\pm0.015\%)$ per amu for Pb and U), isotopic tracer contribution and analytical blanks: 5.5 ± 3.5 pg for Pb and less than 1 pg for U. Initial common Pb is determined for each fraction in using the Stacey and Kramers (1975) two-step model.

monazite inclusion in garnet from this sample gave a slightly discordant early Neoproterozoic age, the corresponding $^{207}\text{Pb}/^{206}\text{Pb}$ minimum age of which is 894 ± 1 Ma (Fig. 6, Table 3).

8. Discussion

Since its first description in southern India by Grew (1982) sapphirine has been reported from an increasing number of rock types and occurrences in the eastern part of the Madurai Block (Santosh et al., 2004; Tateishi et al., 2004 and references therein). In particular, the recognition of sapphirine—spinel and sapphirine—quartz provided evidence for a stage of high- to ultra-high-temperature metamorphism in southern India. However, it is still unclear whether the reaction textures and phase assemblages observed in these rocks reflect a single stage metamorphic evolution or belong to independent orogenic events.

Tateishi et al. (2004) first presented sapphirine–quartz assemblages in rocks from the Mottamala quarry, which are reported here. From phase petrology and thermobarometric calculations they inferred minimum P-T conditions of 8 kbar and 1050 °C followed by an initial stage of isobaric cooling and the onset of decompression at temperatures exceeding 900 °C. The results presented in this study do not entirely support this view. Whereas the P-T conditions for the peak of metamorphism in our study are fairly similar to those of Tateishi et al. (2004), our results indicate protracted near-isobaric

cooling before decompression starts at temperatures of ca. 800 °C. The main reasons for this discrepancy are (i) the widespread occurrence of orthopyroxenesillimanite assemblages in our samples which requires a cooling path well above the [spinel, biotite] invariant point, (ii) the occurrence of back reactions of garnet and orthopyroxene with melt-producing biotite, cordierite and quartz symplectites that marks the end of the cooling path at ca. 850 °C and 8 kbar and (iii) the fact that we take a melt phase (e.g. quartz component of the melt) instead of a solid phase (e.g. quartz) into account for most of the retrograde reactions. Kriegsman and Hensen (1998) demonstrated that this would shift the invariant point to significantly lower pressure and steepen the slope of the corresponding univariant curves. Reactions which involve a melt phase have a subvertical slope whereas reactions that involve solid phases only have subhorizontal slopes. Therefore any P-T path that crosses melt-present reactions will have a shallower slope than a path in a grid whose reactions exclusively involve solid phases. As a result our retrograde path is dominated by cooling reactions compared to the one proposed by Tateishi et al. (2004), which is dominated by decompression reactions. Similar results were obtained by Moraes et al. (2002) for sapphirine-quartz rocks from Brazil who proposed a P-T path that also takes into account of melt-producing and melt-consuming reactions. They discuss in detail the role of melt in the evolution of the UHT granulites of central Brazil and located the melt-bearing invariant points [spinel,



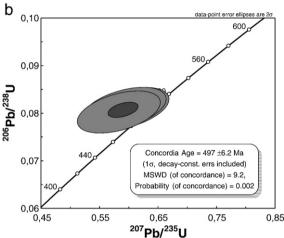


Fig. 7. a. Microphotograph of a monazite inclusion in garnet from sample I01-93-5 which has been dated by LA-ICP-MS at 501 ± 16 Ma. Width of individual spot sizes is 5 μ m. b. Concordia diagram with individual U–Pb ages from the analysed monazite inclusion.

sapphirine] and [spinel, biotite] at similar pressures as in this paper (*ca.* 8–9 kbar).

The geochronological data presented in this study are consistent with the results of EPMA monazite dating recently presented by Braun and Appel (2006). They provide additional support for the occurrence of two stages of high-grade metamorphism in the Madurai Block during the early and late Neoproterozoic (*ca.* 900 and 550 Ma ago). The major challenge in interpreting

these ages is to find a clear correlation between these age data and the metamorphic history that is recorded in the rocks which also host the monazites. In particular, a precise knowledge on the age of UHT metamorphism is crucial for a better understanding of the geodynamic evolution of the Madurai Block. Unfortunately, the results and observations which so far have been collected for the UHT granulites described in this study do not allow to unravel their P-T-t evolution unequivocally.

The slightly discordant Tonian age from the monazite inclusion in garnet could reflect a stage of UHT metamorphism that led to the development of the sprqtz symplectite inclusions. The occurrence of monazite inclusions with such old relic cores in garnet and in proximity to the sapphirine-quartz symplectites could favour such an interpretation although the evidence for this correlation is not unequivocal. The heatingcooling path described in this paper could then reflect an early Neoproterozoic orogeny during which the studied granulite samples experienced a complex melting and mineral-melt back reaction sequence. In such a case and depending on the time of formation of the host garnet (very early according to reaction (1) or significantly later through reaction (5c)) the appropriate monazite ages represent maximum or minimum values for the time of UHT metamorphism, respectively. The rocks subsequently may have stayed in the lower continental crust and cooled down from ultra-high (>1000 °C) to high temperatures (~800 °C) before they were invaded by granitic melts during Pan-African orogeny and brought to a higher crustal level along a steep isothermal decompression path. Monazite and particularly spr-qtz intergrowths must have been shielded by garnet during this stage which allowed them to preserve their early Neoproterozoic geochronological and petrological features. Alternatively, one could argue that UHT metamorphism was related to Pan-African orogeny and occurred 550-600 Ma ago. We consider this explanation to be unlikely since in this case it would be difficult to explain that monazite was able to preserve ages older than 600 Ma although temperatures were well above

Table 4
U-Th-Pb ages from in-situ LA-ICP-MS dating of a monazite inclusion in garnet from the granulite sample I01-93-5

Spot #	Atomic rati	Apparent ages												
	²⁰⁶ Pb/ ²³⁸ U	1σ	²⁰⁷ Pb/ ²³⁵ U	1σ	²⁰⁸ Pb/ ²³² U	1σ	²⁰⁶ Pb/ ²³⁸ U	1σ	²⁰⁷ Pb/ ²³⁵ U	1σ	²⁰⁸ Pb/ ²³² U	1σ	Concordia age	2σ
Ju29f04	0.082	0.002	0.599	0.030	0.026	0.001	510	11	477	24	514	14	_	_
Ju29f05	0.081	0.002	0.597	0.030	0.026	0.001	502	11	475	24	516	14	499	21
Ju29f06	0.080	0.002	0.583	0.029	0.024	0.001	498	11	467	23	487	13	495	21
Ju29f07	0.076	0.002	0.537	0.027	0.025	0.001	475	11	436	22	491	13	_	_
Ju29f08	0.081	0.002	0.592	0.029	0.025	0.001	502	11	472	23	499	13	499	21

common estimates for the closure temperature for Pb diffusion in monazite.

The late Neoproterozoic ages reported from monazite inclusions in quartz and orthopyroxene agree very well with available data from these rocks as well as from other places in the Southern Granulite Terrain (Santosh et al., 2003; Ghosh et al., 2004) and again confirm the widespread high-grade metamorphic event at the end of the Pan-African orogeny in this area.

Raith et al. (1997) provided a detailed petrologic study of sapphirine-bearing UHT granulites from the Palni Hills in southern India. Sapphirine-quartz assemblages were not observed but sillimanite pseudomorphs after kyanite point to significantly higher pressure (ca. 12 kbar) and slightly lower temperature (1000 °C) conditions than those reported in this study. Raith et al. (1997) proposed that the first stage of this path was dominated by high-P (ca. 12 kbar) isobaric heating and a high-T (ca. 1000 °C) isothermal decompression (down to ca. 8 kbar). Despite the lack of geochronological data they suggested that this stage took place during the Paleoproterozoic (ca. 2.5 Ga) and presumably was coeval with the tectono-metamorphic evolution of the Nilgiri, Biligiri Rangan and Shevaroy Hills.

Our geochronological data are in line with the possibility of two distinct orogenic events but do not show any evidence for a Paleoproterozoic stage of high-grade metamorphism as it is widespread in the granulite terrains further north in the SGT. This raises the question whether all sapphirine-bearing rocks in southern India have shared the same tectono-metamorphic evolution or not. Most of the rocks reflecting UHT conditions occur as enclaves or form small outcrops and yield a fairly wide range in P-T conditions and evolution. This as well as the fact, that with the exception of the data presented in this study, no texturally constrained geochronological information is available from these occurrences makes it difficult to judge if UHT metamorphism was a local or a regional phenomenon in the exposed crustal level of the SGT. Precise geochronology on the Palni Hills rocks could shed some light on the possible linkage between them and the Rajapalayam granulites. However, in both localities the last decompression stage (from 8 kbar), as recorded by the reaction of orthopyroxene and cordierite with melt-producing biotite, is agreed to take place at Pan-African times.

Magnesium- and aluminium-rich metapelites with similar sapphirine—quartz assemblages that indicate ultrahigh-temperature and medium to high pressure metamorphic conditions are reported from neighbouring Gondwana terrains in Sri Lanka and the Lützow—Holm

Complex in Antarctica (Yoshimura et al., 2003; Osanai et al., 2006). They also occur as lenses or enclaves in ordinary pelitic gneisses and/or two-pyroxene granulites and charnockites. Merging observations in the host mafic granulites and the enclaves, Osanai et al. (2006) presented a P-T path that is dominated by two isothermal decompression stages. The occurrence of kyanite in the metapelites indicates significantly higher pressures (ca. 17 kbar) than in the rocks presented in this study and isothermal decompression to ca. 11 kbar, which was followed by isobaric cooling. Taking into account that the grids of Osanai et al. (2006) do not involve a melt phase, the pressure at which isobaric cooling occurs might be overestimated. The difference between the last segments of the isothermal decompression paths in Rajapalayam and Sri Lanka concerns spinel that is widely reported in symplectites and retrograde products in Sri Lankan rocks but clearly absent in the rocks studied in the South Indian UHT granulites. This might be explained by a model according to which the South Indian rocks cooled at depth (ca. 6 kbar), preventing the growth of retrograde spinel, whereas the Sri Lankan rocks were cooled during decompression (from ca. 6 kbar) to allow the widespread formation of spinel-bearing symplectites without entering the spinel-quartz stability field. On the other hand, it is known that oxygen fugacity plays a key role in the appearance of spinel, and the rocks from South India and Sri Lanka might have reequilibrated at different oxygen fugacity conditions. Osanai et al. (2006) report a high oxygen fugacity indicated by the coexistence of magnetite and spinel as well as the occurrence of magnetite exsolutions in spinel. The rocks found in Rajapalayam do no present such assemblages and probably have not seen spinel-bearing reactions as the bulk oxygen fugacity was too low.

UHT granulites from the Lützow-Holm Bay are interpreted to have evolved along a similar P-T path as the Sri Lankan occurrences (Yoshimura et al., 2003), which is characterised by isothermal decompression from a kyanite-stabilized granulite field through sapphirine+quartz, orthopyroxene+sillimanite and garnet-cordierite stability fields. Even though no precise geochronology has been undertaken on these rocks, Osanai et al. (2006) speculate that the localized UHT/ HP metamorphism in the rocks from the Highland Complex and the Lützow-Holm Complex might have taken place before the Pan-African metamorphism. The correlation of the tectono-metamorphic evolution of these crustal fragments with that of UHT granulites in the Madurai Block in southern India has to be investigated by future studies.

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