

The flux of carbonyl sulfide and carbon disulfide between the atmosphere and a spruce forest

X. Xu, H. G. Bingemer, U. Schmidt

▶ To cite this version:

X. Xu, H. G. Bingemer, U. Schmidt. The flux of carbonyl sulfide and carbon disulfide between the atmosphere and a spruce forest. Atmospheric Chemistry and Physics Discussions, 2002, 2 (1), pp.181-212. hal-00300824

HAL Id: hal-00300824 https://hal.science/hal-00300824

Submitted on 18 Jun 2008

HAL is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers. L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés. Atmos. Chem. Phys. Discuss., 2, 181–212, 2002 www.atmos-chem-phys.org/acpd/2/181/ © European Geophysical Society 2002



ACPD

2, 181-212, 2002

The flux of carbonyl sulfide and carbon disulfide

Xu et al.

The flux of carbonyl sulfide and carbon disulfide between the atmosphere and a spruce forest

X. $Xu^{1,*}$, H. G. Bingemer¹, and U. Schmidt¹

¹Institute for Meteorology and Geophysics, University of Frankfurt, P.O. Box 111932, 60054 Frankfurt am Main, Germany ^{*}now at Max Planck Institute for Chemistry, Mainz, Germany

Received: 9 January 2002 - Accepted: 19 February 2002 - Published: 21 February 2002

Correspondence to: X. Xu (xu@mpch-mainz.mpg.de)

Title Page		
Abstract Introduction		
Conclusions	References	
Tables	Figures	
•	•	
Back	Close	
Full Screen / Esc		
Print Version		
Interactive Discussion		

Abstract

Turbulent fluxes of carbonyl sulfide (COS) and carbon disulfide (CS₂) were measured over a spruce forest in Central Germany using the relaxed eddy accumulation (REA) technique. A REA sampler was developed and validated using simultaneous measurements of CO₂ fluxes by REA and by eddy correlation. REA measurements were conducted during six campaigns covering spring, summer, and fall between 1997 and 1999. Both uptake and emission of COS and CS₂ by the forest were observed, with deposition occurring mainly during the sunlit period and emission mainly during the dark period. On the average, however, the forest acts as a sink for both gases. The average fluxes for COS and CS₂ are -93 ± 11.7 pmol m⁻² s⁻¹ and -18 ± 7.6 pmol m⁻² s⁻¹, respectively. The fluxes of both gases appear to be correlated to photosynthetically active radiation and to the CO₂ and H₂O fluxes, supporting the idea that the airvegetation exchange of both gases is controlled by stomata. An uptake ratio COS/CO₂ of 10 ± 1.7 pmol μ mol⁻¹ has been derived from the regression line for the correlation between the COS and CO₂ fluxes. This uptake ratio, if representative for the global ter-

¹⁵ between the COS and CO₂ fluxes. This uptake ratio, if representative for the global terrestrial net primary production, would correspond to a sink of $2.3 \pm 0.5 \text{ Tg COS yr}^{-1}$.

1. Introduction

20

25

The tropospheric abundance (~ 500 ppt) and long tropospheric lifetime (2–7 years) of carbonyl sulfide (COS) makes it the major source of the stratospheric background aerosol (Crutzen, 1976; Engel and Schmidt, 1994), which plays a role in the Earth's radiation budget and in heterogeneous reactions causing chemical ozone destruction (Lacis et al., 1992; Rodriguez et al., 1991; Grainier and Brasseur, 1992).

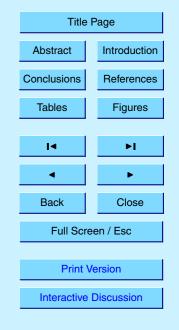
COS is released to the atmosphere by oceans, biomass burning, oxidation of CS_2 and DMS, and several anthropogenic sources (tire wear, aluminium production, coal combustion, sulfur recovery, etc.), and is removed by terrestrial vegetation, soils, photolysis, and reactions with OH and O radicals (Khalil and Rasmussen, 1984; Chin and

ACPD

2, 181–212, 2002

The flux of carbonyl sulfide and carbon disulfide

Xu et al.



Davis, 1993; Andreae and Crutzen, 1997; Watts, 2000). Terrestrial vegetation is recognized as a dominant sink of atmospheric COS, but the magnitude of this sink has not been satisfactorily quantified. The uptake of COS by vegetation was first observed by Taylor et al. (1983) and Kluczewski et al. (1983, 1985) during laboratory chamber experiments with *synthetic* air containing 120 ppb and 4 ppb of COS, respectively. Brown and Bell (1986) obtained a preliminary estimate of the global vegetation sink of 2–5 Tg COS yr⁻¹, based on data from Kluczewski et al. (1983, 1985). However, air-vegetation exchange of COS depends on the mixing ratio level of this gas (e.g. Kesselmeier and Merk, 1993). Therefore, estimates of the global COS uptake by vegetation that are derived from experiments employing largely enhanced COS levels will not be representative of the real natural conditions. Laboratory experiments with crops conducted by Goldan et al. (1988) showed that the uptake of COS and CO₂ for atmospheric levels and under daylight conditions has a common pathway, i.e., through the open stomata, and that the transport resistances for both gases are virtually the same.

¹⁵ This similarity led Goldan et al. (1988) to estimate the COS vegetation sink on the basis of the atmospheric mixing ratios of COS and CO₂ and the primary productivity of terrestrial plants. They reported an uptake rate ranging from 0.2 to 0.6 Tg COS yr⁻¹, one order of magnitude smaller than the estimate of Brown and Bell (1986).

Enzymological studies revealed that all enzymes involved in CO₂ assimilation can ²⁰ metabolize COS, with carbonic anhydrase (CA) being the key enzyme which catalyzes the hydrolysis of COS to CO₂ and H₂S (Protoschill-Krebs and Kesselmeier, 1992; Protoschill-Krebs et al., 1995, 1996). According to this finding, all living higher plants should be able to consume COS. This view is consistent with laboratory results (Kluczewski et al., 1983, 1985; Taylor et al., 1983; Goldan et al., 1988; Fried et al.,

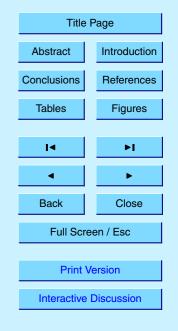
²⁵ 1993; Kesselmeier and Merk, 1993). However, it is not generally supported by field studies. While some in-situ experiments clearly showed uptake of COS by plants or soil/plant systems (Mihalopoulos et al., 1989; Hofmann et al., 1992; Bartell et al., 1993; Dippel and Jaeschke, 1996; Kuhn et al., 1999), other measurements found either no evidence of deposition (Berresheim and Vulcan, 1992) or the existence of both de-

ACPD

2, 181–212, 2002

The flux of carbonyl sulfide and carbon disulfide

Xu et al.



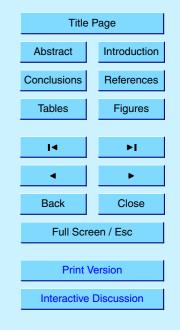
position and emission (Kesselmeier et al., 1993; Hofmann, 1993; Huber, 1994). This indicates that the role of vegetation under natural conditions is more complicated. The air-plant exchange of COS appears to be bidirectional and dependent on the ambient COS mixing ratio, the environmental conditions, the plant type, etc. A compensation

- ⁵ point, at which deposition equals emission, was observed during experiments of the air-plant exchange of COS (Rennenberg et al., 1991; Kesselmeier and Merk, 1993; Huber, 1994; Dippel and Jaeschke, 1996). It varies from 90 to 515 ppt and is usually lower than the tropospheric level of COS (~ 500 ppt), suggesting that vegetation tends to take up COS in most cases. The air-plant exchange flux of COS was often found to
- ¹⁰ be correlated to the CO₂ assimilation rate, to photosynthetically active radiation (PAR) and to the H₂O flux (Kesselmeier and Merk, 1993; Kesselmeier et al., 1993; Bartell et al., 1993; Hofmann, 1993; Huber, 1994), implying the importance of plant physiological processes in controlling the COS exchange rate. The close relationship between COS uptake and CO₂ fixation is an encouraging finding, as it allows an estimate of the
- ¹⁵ global vegetation sink of COS on the basis of the observed uptake ratio COS/CO₂ and the terrestrial plant productivity. Assuming that the ratio of the COS and CO₂ uptake equals the ratio of the atmospheric burden of both gases, Kesselmeier and Merk (1993) obtained a vegetation sink of 0.86–1.0 Tg COS yr⁻¹. This figure is at the high end of the estimated range of 0.16–0.91 Tg COS yr⁻¹ reported by Chin and Davis (1993). How-²⁰ ever, the actual value of this COS sink might be larger, since it was observed that the
- plants investigated prefer to take up COS over CO_2 , as indicated by Kesselmeier and Merk (1993).

Only a few studies have been made on the air-plant exchange of CS₂. Most of the earlier studies (Adams et al., 1981; Steudler and Peterson, 1985; Goldan et al., 1987;

²⁵ Lamb et al., 1987; Hines and Morrison, 1992) reported emission of CS_2 from vegetated soils. Castro and Galloway (1991) demonstrated that those chamber measurements employing sulfur-free sweep gases resulted in artificial release of CS_2 and COS. During the experiments of Taylor et al. (1983), vegetation uptake was observed at a high CS_2 level of 120 ppb. Fall et al. (1988) found indirect evidence for the uptake of CS_2 by 2, 181–212, 2002

The flux of carbonyl sulfide and carbon disulfide



wheat at lower ambient levels. Gradient measurements of sulfur gases in a loblolly pine forest did not reveal any evidence of CS_2 uptake by the trees (Berresheim and Vulcan, 1992). More recent field and laboratory studies (Hofmann, 1993; Kesselmeier et al., 1993; Huber, 1994; Dippel and Jaeschke, 1996) indicate that CS_2 is both deposited

⁵ to and emitted from soil/plant systems. It is presently not clear which physiological process influences and controls the exchange of CS₂ between the atmosphere and plants.

Considering the high variability of the exchange rates of COS and CS₂ between the atmosphere and terrestrial plants, present databases are obviously not adequate to allow a reliable estimate of global vegetation sinks for both gases. More field measurements on major ecosystems are required to improve our understanding of the role of terrestrial vegetation in the atmospheric cycle of COS and CS₂.

Dynamic enclosure and gradient methods are commonly used for measuring COS uptake by plants. In the former, the chamber disturbs the natural microclimate and the ¹⁵ results have only a limited spatial representativity. The latter method overcomes these problems and measures the surface fluxes on an ecosystem scale, but the assumptions of the gradient method are not always tenable. The method is not recommended for measuring the surface fluxes over tall vegetation, such as forests (den Hartog and Neumann, 1984). The widely adopted eddy correlation (EC) method needs fast-response

²⁰ sensors, which are unavailable for many species. A modified method, the relaxed eddy accumulation (REA), also known as conditional sampling (CS), overcomes the need for fast-response sensors and has proven a promising technique for measuring fluxes using slow-response gas sensors (Oncley et al., 1993; Pattey et al., 1993, 1999; Beverland et al., 1996; Moncrieff et al., 1998). According to Businger and Oncley (1990), the gas flux is expressed as

$$F_c = \beta \sigma_w (\overline{c^+} - \overline{c^-}),$$

where $\overline{c^+}$ and $\overline{c^-}$ are the average concentrations (pmol m⁻³) associated with updrafts and downdrafts, respectively, β is a dimensionless coefficient, and σ_w is the standard

2, 181-212, 2002 The flux of carbonyl sulfide and carbon disulfide Xu et al. **Title Page** Introduction Abstract Conclusions References Figures Tables ► Back Close Full Screen / Esc Print Version Interactive Discussion

© EGS 2002

(1)

ACPD

deviation of the vertical windspeed ($m s^{-1}$).

We have developed a REA sampler and used it to measure COS and CS₂ fluxes over a spruce forest. In the following sections we present some details of the REA sampler and its validation. The measurements of the COS and CS₂ fluxes by REA are discussed.

2. Experimental

2.1. Site

10

Measurements were made on a plateau in the Solling Mountains, Germany (51° 46′ N, 9° 35′ E, 505 m a.s.l). The plateau extends about 1300 m and has a slight downward slope of 1° 20′ towards east. Fig. 1 shows a scheme of the research site. REA flux measurements were taken from a 52 m meteorological tower of the Institute for Bioclimatology (IFB), University of Göttingen.

The site is covered mainly by Norway Spruce (*Picea abies*) planted in 1888, with a tree density of 461 trees ha⁻¹ (Ellenberg et al., 1986). The average canopy height of the spruce stand is about 30 m and the leaf area index is about 7. A beech stand (*Fagus sylvatica*) with a canopy height of about 29 m is located south and southwest of the tower. The small triangle-shaped area located about 200 m northwest of the tower represents younger spruces of about 20 m height. There are some beech patches to the southeast of the tower. If these beech patches are neglected, the spruce fetch ex-

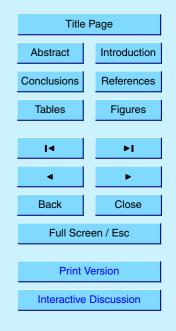
- tends 200–300 m in the south and southwest directions and up to 600–1500 m in other directions from the tower (lbrom et al., 1996). The largest source area extent (or the footprint) of the site, i.e., the maximum upwind distance from which the observed flux can originate, has been estimated by Laubach et al. (1994) and lbrom et al. (1996). For the measurement height of this work (39 m), Laubach et al. (1994) predicted that 80%
- ²⁵ of the flux originates from a distance between 0 and 600 m under neutral conditions. Taking the influence of stability on the source area into account, lbrom et al. (1996)

ACPD

2, 181–212, 2002

The flux of carbonyl sulfide and carbon disulfide

Xu et al.



obtained a maximum extent of the source area ranging from 200 m during daytime to 600 m during the night. Therefore, the fetch requirements for flux measurements are met during daytime even in the worst case, i.e., southwest wind. Inhomogeneous fetch may be encountered during nocturnal measurements, which only account for 2% of flux measurements in this work.

Soils at the site are characterized as cambisol with a pH value of 3.5 (Ellenberg et al., 1986). The soils are covered by a humus layer of 5 cm depth. Herbs, such as *Trientalis europea*, *Avenella flexuosa*, *Galium harcynium*, are the dominant ground plants. They cover 40% of the area. In addition, about 10% of the ground is covered by moss species, such as *Dicranella heteromalla*, *Polytrichum formosum*, etc.

The climate of the site is characterized as montane (sub-oceanic / sub-continental). The annual mean temperature is 6.6° C. The maximum and minimum monthly mean temperatures are approximately 18° C (July) and 4° C (February), respectively. The annual precipitation is about 1045 mm, with relatively large interannual variations. There is no clear dry or wet season at the site. The prevailing wind direction is west. More details about the site are described by Ellenberg et al. (1986), Laubach et al. (1994), and Ibrom et al. (1996).

2.2. Sampling

5

10

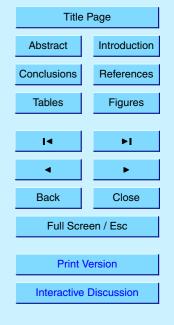
15

Samples were collected at a platform at 39 m altitude using a REA sampler depicted in

- Fig. 2. A 3-dimensional sonic anemometer-thermometer with a USA-1 sensor (Metek) was used for monitoring wind components and the virtual temperature. Wind and temperature data were recorded at a rate of 10 Hz. Three computer controlled Teflon solenoid valves (Cole Parmer) were employed to direct upstream and downstream air into two different sample reservoirs.
- ²⁵ Bags with a volume of 9 L, made of Tedlar PVF film (5 μ m, Du Pont), were used as sample reservoirs. They were protected by polystyrene boxes, which on one hand prevented photochemical reactions in the samples, and on the other hand could easily be transported. Before sampling, the bags were evacuated and then connected to the

2, 181–212, 2002

The flux of carbonyl sulfide and carbon disulfide



solenoid valves. During sampling, air from the height of the anemometer sensor was drawn at a rate of 2200 ml min^{-1} through a 3 m Teflon FEP tubing (1/8" O. D., 1/16" I. D.) by a Teflon diaphragm pump (N86 KTE, KNF Neuberger). This flow of compressed air was split in two sub-streams. A flow of 600 ml min⁻¹, set by a mass flow controller (MFC, 0–1000 ml min⁻¹, Bronk Horst), was directed to the selected reservoir, and the remainder was vented.

5

20

The sampling period was 30 min. The effective sampling time was, however, less than 30 min, because a deadband (w_d) was applied, i.e. an interval of the vertical wind speed in which air was not accumulated in the reservoir for up or downdraft, if $|w| \le w_d$.

- ¹⁰ The use of a deadband increases the concentration differences between the two sample reservoirs and prolongs the lifetime of the valves (Pattey et al., 1993; Oncley et al., 1993; Valentini et al., 1997; Xu, 2001). To establish a suitable deadband, a test run (3–5 min) was made before each REA sampling. The turbulence intensity (σ_w), as derived from the vertical wind data during the test run, was taken as a reference to select the value of w_d . w_d was usually set to less than $0.2\sigma_w$, corresponding to an effective
- sampling period of 25–30 min.

Although the sampling system can be operated at 10 Hz, the valves were switched at 5 Hz, with a few exceptions only. This prolongs the valve's lifetime without significant reduction in the accuracy of the flux measurements. This is confirmed by the good agreement between the sensible heat fluxes obtained by REA simulations at sampling frequencies 5 Hz and 10 Hz, respectively (Xu, 2001).

The lag time, i.e. the travel time of air from the intake point to the switching valve, is estimated to be 1.0 s, based on the flow rates and dead volume of the tubing. This lag time was taken into account in the control program.

Damping of concentration fluctuations in the tubing due to mixing may cause an underestimate of flux. However, an estimate of this effect using formulae of Leuning and Moncrieff (1990) showed that the reduction of flux is negligible even under extreme conditions. Therefore, no correction of the damping effect is necessary. More details about this REA sampler are discussed in Xu (2001) 2, 181–212, 2002

The flux of carbonyl sulfide and carbon disulfide

Title Page			
Abstract Introduction			
Conclusions Reference			
Tables Figures			
• •			
Back Close			
Full Screen / Esc			
Print Version			
Interactive Discussion			



2.3. Sample analysis

Air samples were analyzed for COS, CS_2 and CO_2 within 3 hour after sampling. The inertness of the Tedlar film ensures that loss and production of these trace gases in the bags are negligible during storage. Ambient air can be stored in the bags for more than 10 h without significant changes in the COS and CS_2 contents (Xu, 2001)

- COS and CS₂ were measured using a gas chromatograph (GC, HP 6890, Hewlett Packard) with a flame photometric detector (FPD, Tracor). Figure 3 shows the analyzing system. The mixing ratios of COS and CS₂ in the bags were measured by multiple analysis of 0.4–0.5 L aliquots from the bags. Each bag was analyzed at least 3 times. Samples were cryogenically focused (liquid Argon, –186°C) in a capillary glass trap (20 cm × 2 mm I. D.) filled with 2–3 cm silanized glasswool. Water vapor in the air samples was removed by passing the sample through a Nafion dryer (Perma Pure,
- samples was removed by passing the sample through a Nation dryer (Perma Pure, KNF Neuberger), operated with a counterflow of dry air (Drierite, Cole Parmer). This method has proved to be effective and economical.
- ¹⁵ The sulfur compounds were separated on a $5' \times 1/8''$ Teflon column packed with Chromosil 310 (Supelco). Nitrogen (99.999%, Messer Griesheim) was used as a carrier gas. For optimal separation within a short time (4–5 min), a pressure program was used instead of the conventional temperature program. The carrier gas pressure at the inlet of the column was controlled by the electronic pneumatic control (EPC) of the GC.
- ²⁰ The oven temperature was set to 30 ± 0.1 °C. A typical pressure program was: (1) start at 280 kPa and hold for 0.1 min, (2) decrease to 120 kPa at a rate of 240 kPa min⁻¹ and hold for 2 min, and (3) increase to 340 kPa at a rate of 1000 kPa min⁻¹ and holding for 1.7 min.

The GC/FPD system was calibrated daily by injecting standard samples with a gas tight Teflon/glass syringe (Precision sampling Corp.). The standard gas mixture (0.6– 1 ppm COS and 0.1–0.25 ppm CS₂) was produced by a permeation dilution device kept at $30 \pm 0.1^{\circ}$ C (Haunold). The permeation rates were determined by weighing the permeation tubes (VICI Metronics) every one or two months with an electronic balance

ACPD

2, 181–212, 2002

The flux of carbonyl sulfide and carbon disulfide

Xu et al.

Title Page			
Abstract	Introduction		
Conclusions	References		
Tables Figures			
•	•		
Back Close			
Full Scre	een / Esc		
Print Version			
Interactive Discussion			

(0.01 mg, Sartorius). The relative standard deviation of the permeation rates is less than 10%.

Most of the samples were also analyzed for CO₂, using a Li-6262 infrared CO₂ / H₂O analyzer (LI-COR). Pressure changes in the sample cell were measured by a pressure transducer (6262-03, LI-COR). The LI-6262 was programmed to automatically correct for the influence of pressure and H₂O fluctuations on CO₂ measurements due to dilution and pressure broadening. Slight zero and span drifts were corrected using working standards, which were calibrated against secondary standards at the Institute for Stratospheric Chemistry, Research Center Jülich, Germany, that can be traced to the Scripps CO₂ calibration scale.

2.4. Other measurements

Continuous eddy correlation measurements of the CO_2 , H_2O and heat fluxes as well as measurements of meteorological parameters were conducted by the IFB. The EC measurements of the CO_2 and H_2O fluxes were made at the same height as the REA sampling, at 39 m above the ground. The same sonic anemometer-thermometer was used for the EC and REA measurements. The sensor of the instrument was mounted at the end of a metal bar of 2.5 m length, which was pointed towards west. The CO_2 and H_2O mixing ratios at the height of the sonic sensor were continuously measured at 10 Hz, using a second LI-6262. An automatic daily two point calibration of the system was made for both CO_2 and H_2O , using CO_2 calibration gases and a stable capacitive humidity sensor, respectively. More details about the instrumentation and the EC

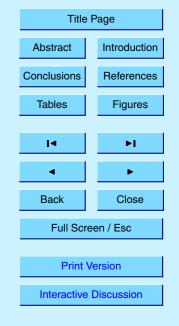
measurements were described by Laubach et al. (1994) and lbrom et al. (1996). The fluxes of COS and CS_2 between the soil and the atmosphere were measured at several positions within 30 m distance from the tower. Three dynamic chambers in

parallel were exposed to the ground. The chambers were flushed with ambient air. Air samples of the in- and outflow of the chambers were collected in Tedlar bags, and analyzed as described in Sect. 2.3. The results are reported in detail by Steinbacher **ACPD**

2, 181-212, 2002

The flux of carbonyl sulfide and carbon disulfide

Xu et al.



(2000) and Steinbacher et al. (2002)

2.5. Assessment of errors

15

20

The turbulent flux F_c of a trace gas c was calculated using Equation 1. σ_w and β are derived from wind and temperature data measured by the sonic anemometerthermometer. $\overline{c^+}$ and $\overline{c^-}$ are obtained as averages of at least three concentration measurements in samples of the respective bags.

According to the error propagation theory, the relative systematic error in F_c may be estimated by

$$\frac{\Delta F_c}{F_c} = \frac{\Delta \sigma_w}{\sigma_w} + \frac{\Delta \beta}{\beta} + \frac{\Delta (\overline{c^+} - \overline{c^-})}{\overline{c^+} - \overline{c^-}},$$
(2)

¹⁰ where ΔF_c , $\Delta \sigma_w$, $\Delta \beta$, and $\Delta(\overline{c^+} - \overline{c^-})$ are the absolute systematic errors in F_c , σ_w , β , and $(\overline{c^+} - \overline{c^-})$, respectively. The three terms at the right-hand side of Eq. (2) are estimated to be less than 2%, 5%, and 3%, respectively (Xu, 2001). Therefore, the accuracy for the flux measurements is estimated to be better than 10%.

The relative random error in F_c is given by

$$\frac{\delta_{F_c}}{F_c} = \sqrt{\left(\frac{\delta_{\sigma_w}}{\sigma_w}\right)^2 + \left(\frac{\delta_{\beta}}{\beta}\right)^2 + \left(\frac{\delta_{(\overline{c^+} - \overline{c^-})}}{\overline{c^+} - \overline{c^-}}\right)^2},\tag{3}$$

where δ_{F_c} , δ_{σ_w} , δ_{β} , and $\delta_{(\overline{c^+}-\overline{c^-})}$ are the absolute random errors in F_c , σ_w , β , and $(\overline{c^+}-\overline{c^-})$, respectively. Detailed considerations (Xu, 2001) show that $\delta_{(\overline{c^+}-\overline{c^-})}/(\overline{c^+}-\overline{c^-})$ is normally dominant among the three error sources at the right-hand side of Eq. (3). Because of the relatively small $|\overline{c^+}-\overline{c^-}|$ values, and the less perfect precision in the GC measurements of COS and CS₂, $\delta_{(\overline{c^+}-\overline{c^-})}/(\overline{c^+}-\overline{c^-})$ was as large as 10-100% in most REA runs. In a few extreme cases, when the fluxes (or the $|\overline{c^+}-\overline{c^-}|$ values) were

ACPD

2, 181-212, 2002

The flux of carbonyl sulfide and carbon disulfide

Xu et al.

Title Page		
Abstract Introduction		
Conclusions References		
Tables Figures		
•	•	
Back	Close	
Full Screen / Esc		
Print Version		
Interactive Discussion		

close to zero, it was even larger than 1000%. Consequently, with a few exceptions, random errors in σ_w and β contribute only negligibly to the overall precision of the COS and CS₂ fluxes. CO₂ could be measured at higher precision than sulfur gases. In spite of the small differences between $\overline{c^+}$ and $\overline{c^-}$ for CO₂, the relative random error in $(\overline{c^+} - \overline{c^-})$ for CO₂ was smaller than 10% in most cases. This is about the same order of magnitude as the relative random errors in σ_w and β .

3. Results and discussion

5

3.1. Validation of the REA sampler

To verify the reliability of the REA sampler, CO₂ flux measurements were intercompared with those from EC measurements by IFB. 94 pairs of CO₂ fluxes measured simultaneously by REA and EC techniques are plotted in Fig. 4. Although there are some outliers, the overall agreement is good. The regression line indicates a slight overestimate of the CO₂ deposition flux by REA. Considering the standard errors of the intercept and the slope, 0.55 and 0.06, respectively, this overestimate is not significant. The 95% confidence belt covers the 1:1 line, suggesting a reliable performance of the REA system under most conditions.

3.2. COS and CS₂ fluxes

Six campaigns of 10 to 15 days were conducted in August and September of 1997, in September of 1998, and in May, July and September–October of 1999, respectively. The observed fluxes ranged from -497 to 311 pmol m⁻² s⁻¹ for COS and from -305 to 236 pmol m⁻² s⁻¹ for CS₂, indicating that the spruce forest can both take up and emit COS as well as CS₂. Uptake of both gases occurred mainly during the sunlit period, whereas emission was observed mostly at night (see Sect. 3.4 for details). On the average, the forest is a sink for atmospheric COS and CS₂. The average fluxes are 2, 181–212, 2002

The flux of carbonyl sulfide and carbon disulfide

Xu et al.



 -93 ± 11.7 pmol m⁻² s⁻¹ for COS and -18 ± 7.6 pmol m⁻² s⁻¹ for CS₂.

Soil chamber measurements at the site showed that the soil always acted as a sink for atmospheric COS (Steinbacher, 2000). However, based on the average flux of $-0.81 \pm 0.24 \text{ pmol m}^{-2} \text{ s}^{-1}$, the soil sink of COS accounts for only less than 1% of the mean deposition of COS into the ecosystem, as observed in the REA measurements.

- The air-soil exchange of CS_2 , ranging from $-0.11 \text{ pmol m}^{-2} \text{ s}^{-1}$ to $0.23 \text{ pmol m}^{-2} \text{ s}^{-1}$, is also negligibly small compared to the exchange flux of CS_2 between the forest and the atmosphere. Therefore, the COS and CS_2 fluxes observed by the REA technique are mainly caused by the air-vegetation exchange.
- ¹⁰ For the purpose of comparison, the values of the observed deposition fluxes of COS and CS₂ were converted to deposition velocities relative to the leaf area. The deposition velocity for COS averaged 1.1 ± 0.7 mm s⁻¹. This agrees well with deposition velocities obtained in other laboratory and in situ studies (Taylor et al., 1983; Kluczewski et al., 1985; Goldan et al., 1988; Kesselmeier and Merk, 1993; Huber, 1994; Kuhn, 15 1997) although quite different plant species were investigated in most of these studies. The deposition velocity for CS₂ averaged 5.4 ± 5.9 mm s⁻¹. CS₂ deposition velocities reported by Huber (1994) and Taylor et al. (1983) range from 0 to 1.5 mm s⁻¹.

3.3. Seasonal variations

Field experiments covered the seasons of spring, summer and fall. Because of unsuitable weather conditions (cold, snow coverage, etc.), no field experiment was carried out in the winter months. Table 1 lists statistical data for the REA measurements of COS and CS₂ fluxes in different seasons and years. Data for CO₂ fluxes are also presented in this table.

During all seasons, COS on the average was taken up by the forest, although emission was observed sometimes during individual measurements. The largest COS deposition was observed in September of 1997, the smallest in September of 1998. There is no clear seasonal variation of the COS flux due to the large year-to-year fluctuation

ACPD 2, 181-212, 2002 The flux of carbonyl sulfide and carbon disulfide Xu et al. **Title Page** Introduction Abstract Conclusions References Figures Tables Back Close Full Screen / Esc **Print Version** Interactive Discussion © EGS 2002

in fall. CS_2 flux data show a net deposition in August, September and October and a minor, but insignificant net emission in May and July. The CO_2 flux data show a clear seasonal variation, with larger CO_2 deposition observed in later spring and in summer, and a smaller deposition in fall. This seasonal variability is consistent with the continuous measurements of CO_2 flux at the same site (lbrom et al., 1996).

3.4. Diurnal variations

5

Because of the high variability of the COS and CS₂ fluxes and the small number of measurements which could be performed on an individual day, it is difficult to identify any potential diurnal variation of the COS and CS₂ fluxes on the basis of data for individual observational days. Such data do not always show any common features. Some daily profiles are even contradictory. However, the average diurnal variations, presented in Fig. 5, clearly indicate systematic features of the fluxes of COS and CS₂, as well as of CO_2 . It is obvious that COS and CS_2 are taken up by the forest predominantly during the sunlit hours, and that the release of these gases occurs during the rest of the day, similarly to CO₂. The strongest deposition of both sulfur gases 15 was observed around local noon, i.e. during the period with most intense solar radiation and strongest turbulence. The similar diurnal features for COS and CO₂ suggest the existence of a common uptake pathway, i.e. the open stomata, as suggested by Goldan et al. (1988). The change of stomatal aperture, which is controlled by various environmental factors, such as the light intensity, water stress, etc., may cause the diurnal cycles of the COS and CS₂ fluxes, as it does for the CO₂ flux. This is supported by the correlations shown in the following sections.

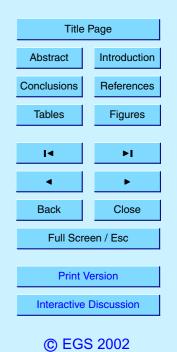
3.5. Correlations to PAR and to the H₂O flux

Figures 6a and 6b show the relationships of the COS flux to photosynthetically active radiation (PAR) and to the H₂O flux, respectively. The correlation coefficients in the figures indicate that the COS flux is significantly ($\alpha < 0.01$) correlated both to PAR and

ACPD

2, 181-212, 2002

The flux of carbonyl sulfide and carbon disulfide



to the H₂O flux.

As implied by the slopes of the regression lines in Figs. 6a and 6b, the uptake of COS increases with increasing PAR and H₂O flux. This may be a result of physiological regulation through stomatal aperture, which is related to both PAR and the H₂O flux.

- 5 At higher PAR, the trees tend to take up more COS and CS₂ and release more water vapor, because of decreased stomatal resistance. Although the relationships of Fig. 6 between the COS flux and physiological parameters are plausible, they are not suitable for parameterization purposes, since the correlations capture only a very small fraction (< 10%) of the variance.
- The CS₂ flux is also correlated to PAR and the H₂O flux. However, both correlations 10 are only significant at confidence levels of 75% and 87%, respectively. There is no known enzymatic mechanism which can explain the deposition of CS₂ to plants and the slight correlations of the CS₂ flux to PAR and the H₂O flux.

3.6. Correlation to the CO_2 flux

- Laboratory studies show that COS is consumed by plants in nearly the same way 15 as CO₂, after being split by the key enzyme, CA (Protoschill-Krebs and Kesselmeier, 1992; Protoschill-Krebs et al., 1995, 1996). This finding not only reveals the physiological background of the uptake of COS by higher plants, but also implies the possibility of extrapolating these measurements to obtain the global COS deposition to vegeta-
- tion, using the ratio of the COS uptake to CO_2 assimilation and the global CO_2 fixation, 20 which is better quantified. Measurements performed during this work support this idea. Figure 7 shows that the COS and CO₂ fluxes are positively correlated at the 99% confidence level. Although the data points are relatively scattered, the correlation line is well defined, as indicated by the dotted lines of the 95% confidence belt of the regression line. 25

The regression line in Fig. 7 contains a small intercept $(-19.6 \text{ pmol m}^{-2} \text{ s}^{-1})$. Both the intercept and the origin lie within the 95% confidence belt, suggesting that the deviation of this intercept from the origin is not significant. This is consistent with the

Tables

Back

ACPD

2, 181-212, 2002

Print Version Interactive Discussion

Full Screen / Esc

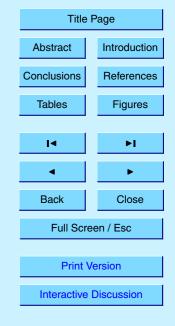
Close

idea that both COS and CO₂ are exchanged between the atmosphere and vegetation through a common pathway, i.e. the open stomata. On the other hand, since the soils at the site were observed to be a permanent source of CO_2 and a minor sink of COS, it is to be expected that a small negative intercept should exist.

- The slope (10.0 with a standard deviation of 1.7) of the regression line in Fig. 7 can be considered as a representative value for the ratio of the COS- and CO₂-uptake. Such uptake ratios were also investigated in several laboratory as well as in situ studies (Hofmann, 1993; Kesselmeier and Merk, 1993; Kesselmeier et al., 1993; Velmeke, 1993; Huber, 1994; Kuhn, 1997). Table 2 lists various values of uptake ratios which
- ¹⁰ were reported in these articles, along with the one obtained in this work. Considerable differences exist between the uptake ratios reported for different plant species, and even between various values reported for spruce. Hofmann (1993) and Huber (1994) measured exchanges of COS and CO₂ between the atmosphere and a spruce forest (80 years) in Schachtenau (Nationalpark Bayerischer Wald), Germany, using the gradi-
- ent and chamber methods, respectively. The uptake ratio obtained in the present work agrees reasonably well with the one derived from the measurements of Huber (1994), but it is about one order of magnitude larger than that derived by Hofmann (1993). If compared with the other uptake ratios listed in Table 2, our results fall roughly into the middle of all values reported so far.
- 20 3.7. Extrapolation to global vegetation sinks

One of the goals of the present work was to derive an estimate of the global vegetation sink of atmospheric COS by extrapolating the ratio between COS uptake and CO₂ fixation from the flux measurements. As shown in Sect. 3.6, a correlation of the COS flux to CO₂ flux was obtained for the spruce forest. The measurements of Hofmann (1993) and Huber (1994) on a 80 year spruce stand also showed a significant correlation between the COS and CO₂ fluxes. However, such a correlation was not found during cuvette experiments on young (10 years) spruce trees (Huber, 1994). A poor correlation between the COS exchange and the CO₂ assimila2, 181–212, 2002

The flux of carbonyl sulfide and carbon disulfide



tion was also found by Kesselmeier et al. (1993) during field experiments on tropical plants (Poterandia cladantha and Saccoglottis gabonensis), while a good correlation was observed by Kuhn (1997) during experiments on a temperate plant (Quercus agrifolia). Laboratory studies on agricultural plants (wheat, corn, rapeseed and pea) ⁵ showed a certain correlation of the COS uptake to the CO₂ assimilation (Hofmann, 1993; Kesselmeier and Merk, 1993), while no relationship between COS and CO₂ exchanges was observed over a wheat field (Hofmann, 1993). The inconsistency among these results indicates the uncertainty of any estimates of the global COS vegetation sink based on extrapolation. Nevertheless, it is worthwhile to use this extrapolation method until better methods become available. Based on the uptake ratio COS/CO₂ of 10 $10.0 \pm 1.7 \text{ pmol } \mu \text{mol}^{-1}$ obtained in this work and the recent estimate of the terrestrial net primary production (NPP) of about $45 \pm 5 \text{ Pg C yr}^{-1}$ (or 90 Pg dry matter yr⁻¹) from Matthews (1997), the global vegetation sink is estimated to be 2.3 ± 0.5 Tg COS yr⁻¹. This value lies within the ranges 2-5 Tg COS yr⁻¹ and 1-3.4 Tg COS yr⁻¹ estimated by Brown and Bell (1986) and Hofmann (1993), respectively, but is much larger than the 15 estimates of $0.24-0.59 \text{ Tg COS yr}^{-1}$, $0.16-1.0 \text{ Tg COS yr}^{-1}$ and $0.86-1.0 \text{ Tg COS yr}^{-1}$ by Goldan et al. (1988), Chin and Davis (1993), and Kesselmeier and Merk (1993), respectively. The COS vegetation uptake derived from our REA measurements should not be considered as the current best estimate of this sink. While its uncertainty of ± 0.5 Tq COS yr⁻¹ is due to the scatter of the REA data and the uncertainty in the NPP 20 estimate, the total error in the extrapolation is unknown and maybe large. A plausible range for the uptake ratio (see Table 2) seems to be 2–12 pmol μ mol⁻¹, corresponding to a sink of $0.5-2.8 \text{ Tg COS yr}^{-1}$. The figure obtained in this work falls within this range.

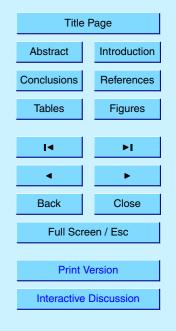
However, as it is derived from a large number of in situ data covering several years and seasons, it should be given special consideration.

The above method can also be applied to tentatively estimate the vegetation sink of CS₂. A linear fit shows $F_{CS_2} = (1.9 \pm 1.2)F_{CO_2} - 3.8$, with F_{CS_2} in pmol m⁻² s⁻¹ and F_{CO_2} in μ mol m⁻² s⁻¹. Assuming that the uptake ratio CS₂/CO₂ is equal to 1.9 ± 1.2 pmol μ mol⁻¹, the global vegetation sink is estimated to be 0.54 ± 0.35 Tg CS₂ yr⁻¹.

ACPD

2, 181-212, 2002

The flux of carbonyl sulfide and carbon disulfide



However, this preliminary estimate of the vegetation sink of atmospheric CS_2 should be viewed with caution since the correlation between the CS_2 and CO_2 fluxes is only significant at the 90% confidence level. Considering the large uncertainty in this estimate, further studies are necessary to quantify the air-vegetation exchange of CS_2 .

5 4. Conclusions

A REA sampler for measuring gas fluxes over forests has been developed and validated. Exchange fluxes of COS and CS₂ between a tall spruce forest and the atmosphere were measured using this system. Both deposition and emission of COS and CS₂ were observed. On the average, however, the forest acted as a net sink of both gases. The average fluxes for COS and CS₂ were -93 ± 11.7 pmol m⁻² s⁻¹ and 10 -18 ± 7.6 pmol m⁻² s⁻¹, respectively. The uptake of COS by the forest showed no clear seasonal pattern. A net deposition of CS₂ was observed in August, September and October and a minor net emission in May and July. On the average, a maximum deposition is found around noon, indicating the importance of stomata in controlling the air-plant exchange of COS and CS_2 . This is supported by the correlations of the fluxes of both gases to PAR and to the H₂O and CO₂ fluxes. Based on the uptake ratio COS/CO₂ (10.0 ± 1.7 pmol μ mol⁻¹) and a recent estimate of the NPP, the global COS vegetation sink is estimated to be 2.3 ± 0.5 Tg COS yr⁻¹. This estimate suggests that the vegetation sink of COS may have been significantly underestimated in earlier budget reviews of atmospheric COS. Since the tropospheric burden of COS is about 20 4.6 Tg, such a large vegetation sink of COS may limit the residence time of atmospheric COS to about 2 years and cause significant seasonal variations in the mixing ratio of tropospheric COS at the middle and high latitudes of the Northern Hemisphere.

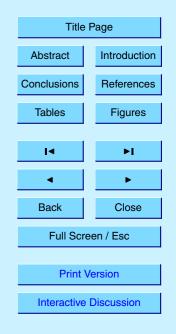
Acknowledgements. This work was funded by the Deutsche Forschungsgemeinschaft (DFG)
 through grant BI 462/1-3. We are very grateful to Drs. K. Morgenstern, A. Ibrom and others from the Institute for Bioclimatology, University of Göttigen for their kind supports. We thank

ACPD

2, 181–212, 2002

The flux of carbonyl sulfide and carbon disulfide

Xu et al.



References

Adams, D. F., Farwell, S. O., Pack, M. R., and Robinson, E.: Biogenic sulfur gas emissions from soils in eastern and southern United States, J. Air Pollut. Cont. Ass., 31, 1083–1089, 1981.

5 **184**

15

- Andreae, M. O. and Crutzen, P. J.: Atmospheric aerosols: Biogeochemical sources and role in atmospheric chemistry, Science, 276, 1052–1058, 1997. 183
- Bartell, U., Hofmann, U., Kreuzburg, B., Andreae, M.O., and Kesselmeier, J.: COS and H₂S fluxes over a wet meadow in relation to photosynthetic activity: An analysis of measurements
- made on 6 September 1990, Atmos. Environ., 27, 1851–1864, 1993. 183, 184
 Berresheim, H. and Vulcan, V.D.: Vertical distributions of COS, CS₂, DMS and other sulfur compounds in a loblolly pine forest, Atmos. Environ., 26 (11), 2031–2036, 1992. 183, 185
 - Beverland, I. J., Moncrieff, J. B., ÓNéill, D. H., Hargreaves, K. J., and Milne, R.: Measurement of methane and carbon dioxide fluxes from peatland ecosystems by the conditional-sampling technique, Q. J. R. Meteorol. Soc., 122, 819–838, 1996. 185
 - Brown, K. A. and Bell, J. N. B.: Vegetation the missing sink in the global cycle of COS, Atmos. Environ., 20, 537–540, 1986. 183, 197

Businger, J. A. and Oncley, S. P.: Flux measurement with conditional sampling, J. Atmos. Ocean. Technol., 2, 349–352, 1990. 185

- ²⁰ Castro, M. S. and Galloway, J. N.: A comparison of sulfur-free and ambient air enclosure techniques for measuring the exchange of reduced sulfur gases between soils and the atmosphere, J. Geophys. Res., 96, 15427–15437, 1991. 184
 - Chin, M. and Davis, D. D.: Global sources and sinks of OCS and CS₂ and their distributions, Global Biogeochem. Cycles, 7, 321–337, 1993. 182, 184, 197
- ²⁵ Crutzen, P.J.: The possible importance of CSO for the sulfate layer of the stratosphere, Geophys. Res. Lett., 3, 73–76, 1976. 182
 - den Hartog, G. D. and Neumann, H. H.: An empirical study on dry deposition of air pollutants to forests, Tech. Rep. AQRB-84-011-L, AES, Canada, 1984. 185

Dippel, J. and Jaeschke, W.: A comparison between natural and anthropogenic emissions of

ACPD

2, 181-212, 2002

The flux of carbonyl sulfide and carbon disulfide

Title Page		
Abstract	Introduction	
Conclusions	References	
Tables	Figures	
Back Close		
Full Screen / Esc		
Print Version		
Interactive Discussion		



the reduced sulfur compounds H_2S , COS, and CS_2 in a tropical industrialized region, J. Atmos. Chem., 25, 251–270, 1996. 183, 184, 185

- Ellenberg, H., Mayer, R., and Schauermann, J.: Ökosystemforschung Ergebnisse des Sollingprojektes 1966–1986, Verlag Eugen Ulmer, Stuttgart, 1986. 186, 187
- ⁵ Engel, A. and Schmidt, U.: Vertical profile measurements of carbonylsulfide in the stratosphere, Geophys. Res. Lett., 21, 2219–2222, 1994. 182
 - Fall, R., Albritton, D. L., Fehnsenfeld, F. C., Kuster, W. C., and Goldan, P. D.: Laboratory studies of some environmental variables controlling sulfur emissions from plants, J. Atmos. Chem., 6, 341–362, 1988. 184
- ¹⁰ Fried, A., Klinger, L. F., and III, D. J. E.: Atmospheric COS exchange in bog microcosms, Geophys. Res. Lett., 20 (2), 129–132, 1993. 183
 - Goldan, P. D., Kuster, W. C., Albritton, D. L., and Fehsenfeld, F. C.: The measurement of natural sulfur emissions from soil and vegetation: Three sites in the Eastern United States revisited, J. Atmos. Chem., 5, 439–467, 1987. 184
- Goldan, P. D., Fall, R., Kuster, W. C., and Fehnsenfeld, F. C.: Uptake of COS by growing vegetation: A major tropospheric sink, J. Geophys. Res., 93, 14186–14192, 1988. 183, 193, 194, 197
 - Grainier, C. and Brasseur, G.: Impact of heterogeneous chemistry on model predictions of ozone changes, J. Geophys. Res., 97 (D16), 18015–18033, 1992. 182
- Hines, M. E. and Morrison, M. C.: Emission of biogenic sulfur gases from Alaskan tundra, J. Geophys. Res., 97 (D15), 16703–16707, 1992. 184
 - Hofmann, U.: Vergleichende Untersuchungen zum Austausch von reduzierten Schwefelverbindungen zwischen Vegetation und Atmosphäre im Feldversuch und unter kontrollierten Bedingungen, Ph.D. thesis, Universität Mainz, 1993. 184, 185, 196, 197, 205
- ²⁵ Hofmann, U., Hofmann, R., and Kesselmeier, J.: Field measurements of reduced sulfur compounds over wheat during a growing season, in: Precipitation scavenging and atmosphere-surface exchange. Volume 2: Atmosphere-Surface Exchange Processes, (Eds) Schwartz, S. E. and Slinn, W. G. N., 967–977, Hemisphere, Washington, D. C., 1992. 183 Huber, B.: Austausch flüchtiger Schwefelverbindungen in land- und forstwirtschaftlichen
- Ökosystemen, Wissenschaftsverlag Maraun, Frankfurt, 1994. 184, 185, 193, 196, 205
 - Ibrom, A., Schuetz, C., Tworek, T., Morgenstern, K., Oltchev, A., Falk, M., Constaintin, J., and Gravenhorst, G.: Eddy-correlation measurements of the fluxes of CO₂ and H₂O above a spruce stand, Phys. and Chem. Earth, 21, 409–414, 1996. 186, 187, 190, 194

2, 181–212, 2002

The flux of carbonyl sulfide and carbon disulfide

Title Page		
Abstract Introductio		
Conclusions	References	
Tables	Figures	
•	►	
Back Close		
Full Screen / Esc		
Print Version		
Interactive Discussion		



Kesselmeier, J. and Merk, L.: Exchange of carbonyl sulfide (COS) between agricultural plants and the atmosphere: Studies on the deposition of COS to peas, corn and rapeseed, Biogeochem., 23, 47–59, 1993. 183, 184, 193, 196, 197, 205

Kesselmeier, J., Meixner, F.X., Hofmann, U., Ajavon, A.-L., Leimbach, S., and Andreae, M.O.:

- Reduced sulfur compounds exchange between the atmosphere and tropical tree species in Southern Cameroon, Biogeochem., 23, 23–45, 1993. 184, 185, 196, 197, 205
 - Khalil, M. A. K. and Rasmussen, R. A.: Global sources, lifetimes and mass balances of carbonyl sulfide (OCS) and carbon disulfide (CS₂) in the Earth's atmosphere, Atmos. Environ., 8 (9), 1805–1813, 1984. 182
- Kluczewski, S. M., Bell, J. N. B., Brown, K. A., and Minski, M. J.: The uptake of ³⁵S-carbonyl sulphide by plants and soils, in: Ecological aspects of radionuclide release, Special Publication of British Ecological Society III, (Ed) Coughtrey, P. J., pp. 91–104, Blackwell Scientific Press, Oxford, 1983. 183

Kluczewski, S. M., Brown, K. A., and Bell, J. N. B.: Deposition of [³⁵S]-carbonyl sulphide to vegetable crops, Radiat. Prot. Dosim, 11, 173–177, 1985. 183, 193

15

Kuhn, U.: Spurengasaustausch klimarelevanter reduzierter Schwefelverbindungen zwischen Biosphäre und Atmosphäre: COS Transfer der Flechten und anderer biotischer Kompartimente, Ph.D. thesis, Universität Mainz, 1997. 193, 196, 197, 205

Kuhn, U., Ammann, C., Wolf, A., Meixner, F.X., Andreae, M.O., and Kesselmeier, J.: Carbonyl

- ²⁰ sulfide exchange on an ecosystem scale: Soil represents a dominant sink for atmospheric COS, Atmos. Environ., 33 (6), 995–1008, 1999. 183
 - Lacis, A., Hansen, J., and Sato, M.: Climate forcing by stratospheric aerosols, Geophys. Res. Lett., 19, 1607–1610, 1992. 182

Lamb, B., Westberh, H., Allwine, G., Bamesberger, L., and Guenther, A.: Measurement of

- biogenic sulfur emission from soil and vegetation: Application of dynamic enclosure methods with Natusch filter and GC/FPD analysis, J. Atmos. Chem., 5, 469–491, 1987.
 - Laubach, J., Raschendorfer, M., Kreilein, H., and Gravenhorst, G.: Determination of heat and water vapour fluxes above a spruce forest by eddy correlation, Agric. For. Meteorol., 71, 373–401, 1994. 186, 187, 190
- ³⁰ Leuning, R. and Moncrieff, J.: Eddy-covariance CO₂ flux measurements using open- and closed-path CO₂ analysers: Corrections for analyser water vapour sensitivity and damping of fluctuations in air sampling tubes, Bound.-Layer Meteorol., 53, 63–76, 1990. 188 Matthews, E.: Global litter production, pools, and turnover times: estimates from measurement

ACPD

2, 181–212, 2002

The flux of carbonyl sulfide and carbon disulfide

Xu et al.

Title Page		
Abstract	Introduction	
Conclusions	References	
Tables Figures		
	►I	
•	►	
Back Close		
Full Screen / Esc		
Print Version		
	ersion	

data and regression models, J. Geophys. Res., 102, 18771-18800, 1997. 197

- Mihalopoulos, N., Bonsang, B., Nguyen, B. C., Kanakidou, M., and Belviso, S.: Field observations of COS deficit near the ground: Possible implication of vegetation, Atmos. Environ., 23 (10), 2159–2166, 1989. 183
- Moncrieff, J. B., Beverland, I. J., ÓNéill, D. H., and Cropley, F. D.: Controls on trace gas exchange observed by a conditional sampling method, Atmos. Environ., 32 (19), 3265–3274, 1998. 185
 - Oncley, S. P., Delany, A. C., Horst, T. W., and Tans, P. P.: Verification of flux measurement using relaxed eddy accumulation, Atmos. Environ., 27, 2417–2426, 1993. 185, 188
- Pattey, E., Desjardins, R. L., and Rochette, R.: Accuracy of the relaxed eddy-accumulation technique, evaluated using CO₂ flux measurements, Bound.-Layer Meteorol., 66, 341–355, 1993. 185, 188
 - Pattey, E., Desjardins, R. L., Westberg, H., Lamb, B., and Zhu, T.: Measurement of isoprene emissions over a black spruce stand using a tower-based relaxed eddy-accumulation sys-
- tem, J. Appl. Meteorol., 38 (7), 870–877, 1999. 185
- Protoschill-Krebs, G. and Kesselmeier, J.: Enzymatic pathways for the consumption of carbonyl sulphide (COS) by higher plants, Bot. Acta, 105, 206–212, 1992. 183, 195
- Protoschill-Krebs, G., Wilhelm, C., and Kesselmeier, J.: Consumption of carbonyl sulphide by chlamydomonas reinhardtii with different activities of carbonic anhydrase (CA) induced by different CO₂ growing regimes, Bot. Acta, 108, 445–448, 1995. 183, 195
- different CO₂ growing regimes, Bot. Acta, 108, 445–448, 1995. 183, 195
 Protoschill-Krebs, G., Wilhelm, C., and Kesselmeier, J.: Consumption of carbonyl sulphide (COS) by higher plants carbonic anhydrase (CA), Atmos. Environ., 30, 3151–3156, 1996. 183, 195

Rennenberg, H., Schröder, P., and Huber, B.: Emission of reduced sulfur compounds from

- agricultural and forest ecosystems, in: EUROTRAC Annual Report 1990, Part 4, BIATEX, pp. 156–162, Garmisch-Partenkirchen, 1991. 184
 - Rodriguez, J. M., Ko, M. K. W., and Sze, N. D.: Role of heterogeneous conversion of N₂O₅ on sulphate aerosols in global ozone loses, Nature, 352, 134–137, 1991. 182
- Steinbacher, M.: Untersuchungen zum Austausch von Carbonylsulfid (COS) über einem Waldökosystem, Master thesis, Universität Frankfurt, 2000. 190, 193
 - Steinbacher, M., Bingemer, H. G., and Schmidt, U.: Exchange of COS and CS₂ between forest soil and atmosphere at a site in Central Europe, in preparation, 2002. 191
 Steudler, P. A. and Peterson, B. J.: Annual cycle of gaseous sulfur emissions from a New Eng-

2, 181–212, 2002

The flux of carbonyl sulfide and carbon disulfide

Xu et al.

Title Page		
Abstract Introduction		
Conclusions	References	
Tables	Figures	
	►I	
•	Þ	
Back	Close	
Full Screen / Esc		
Print Version		
Interactive Discussion		

land spartina alterniflora marsh, Atmos. Environ., 19 (9), 1411–1416, 1985. 184

- Taylor, Jr., G. E., McLaughlin, Jr., S. B., Shriner, D. S., and Slevidge, W. J.: The flux sulfurcontaining gases to vegetation, Atmos. Environ., 17, 789–796, 1983. 183, 184, 193
- Valentini, R., Greco, S., Seufert, G., Bertin, N., Ciccioli, P., Cecinato, A., Brancaleoni, E., and Frattoni, M.: Fluxes of biogenic VOC's from Mediterranean vegetation by trap enriched re-
- laxed eddy accumulation, Atmos. Environ., 31(S1), 229–238, 1997. 188
- Velmeke, F.: Versuche zur Korrelation von Photosynthese mit dem Austausch flüchtiger Schwefel-Verbindungen zwischen Bäumen and Atmosphäre, Master thesis, Universität Mainz, 1993. 196, 205
- ¹⁰ Watts, S. F.: The mass budgets of carbonyl sulfide, dimethyl sulfide, carbon disulfide and hydrogen sulfide, Atmos. Environ., 34, 761–779, 2000. 183
 - Xu, X.: Investigations into the Tropospheric Cycle of COS: Atmospheric Distribution, Air-Sea and Air-Vegetaion Exchanges, Ph.D. thesis, Unversität Frankfurt, 2001.

188, 189, 191

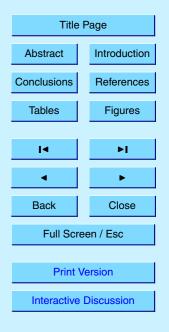
5

ACPD

2, 181–212, 2002

The flux of carbonyl sulfide and carbon disulfide

Xu et al.



Time F_{COS} F_{CS_2} F_{CO_2} μ mol m⁻² s⁻¹ $pm \tilde{o} l \, m^{-2} \, s^{-1}$ $\rm pmol\,m^{-2}\,s^{-1}$ Aug. 1997 -105 ± 47 -44±37 no data Sept. 1997 -173±51 -17±8 no data Sept. 1998 -43±27 -32 ± 20 -5.98±0.81 May 1999 -93±32 1±20 -10.2 ± 1.4 Jul. 1999 -88±19 5±9 -9.29±1.03 Sep./Oct. 1999 -105±23 -36 ± 19 -5.06 ± 1.26

Table 1. REA measurements of COS, CS ₂ and CO ₂ fluxes over a spruce forest in Solling,
Germany. Mean values and standard errors of the mean are listed for the individual campaigns

ACPD

2, 181-212, 2002

The flux of carbonyl sulfide and carbon disulfide

Xu et al.

Title Page			
Abstract	Abstract Introduction		
Conclusions	References		
Tables	Figures		
•	•		
Back	Close		
Full Screen / Esc			
Print Version			
Interactive Discussion			

2, 181–212, 2002

The flux of carbonyl sulfide and carbon disulfide

Xu et al.

Plant	$\text{COS/CO}_2 \text{ (pmol } \mu \text{mol}^{-1} \text{)}$	Method	Reference
Spruce	10.0±1.7	REA, in situ	This work
Spruce	11.7±4.8	Chamber, in situ	Huber (1994)
Quercus agrifolia	2.2	Chamber, in situ	Kuhn (1997)
Spruce Wheat Corn Pea	0.8±0.1 5.7±1.0 11.2±3.8 6.2±2.1	Gradient, in situ Chamber, in situ	Hofmann (1993)
Pea Rapeseed Corn	0.3–4.2 0.9–1.4 1.9–16.7	Chamber, laboratory	Kesselmeier and Merk (1993)
Sacoglottis gabonensis Porterandia cladantha	14.4(0.8–38.4) 1.7(0.14–6.0)	Chamber, in situ	Kesselmeier et al. (1993)
Querscus petrea	2.2	Chamber, laboratory	Velmeke (1993)

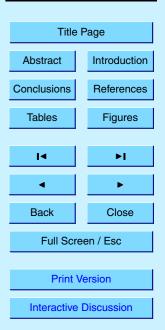
Table 2. Measurements of the uptake of COS by different plants in relation to CO2 assimilation

Title Page		
Abstract Introduction		
Conclusions	References	
Tables	Figures	
•	•	
Back	Close	
Full Screen / Esc		
Print Version		
Interactive Discussion		

2, 181–212, 2002

The flux of carbonyl sulfide and carbon disulfide

Xu et al.



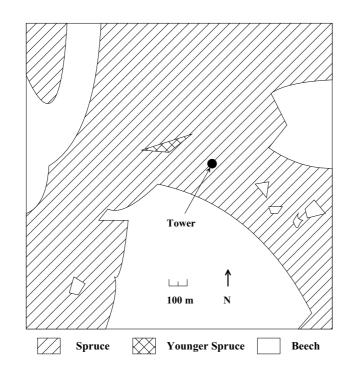
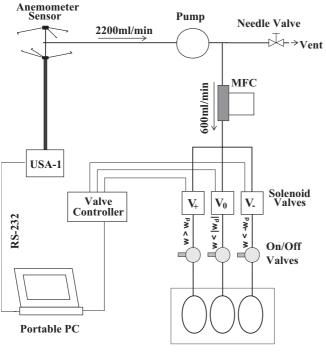


Fig. 1. Forest stands around the experimental site.

2, 181–212, 2002



Tedlar Sample Bags

Fig. 2. Schematic view of the REA sampler.

The flux of carbonyl sulfide and carbon disulfide

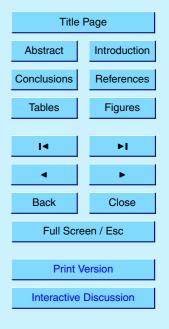
Xu et al.



2, 181–212, 2002

The flux of carbonyl sulfide and carbon disulfide





© EGS 2002

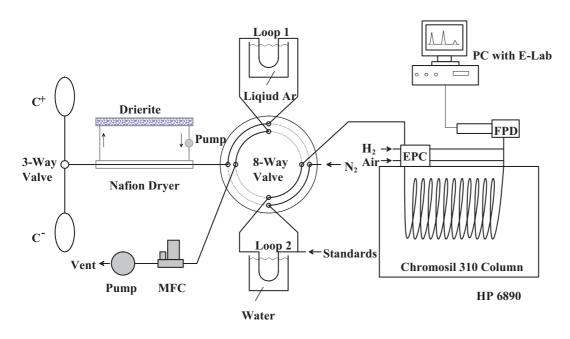


Fig. 3. System for the analysis of sulfur gases in air samples collected in Tedlar bags.

2, 181–212, 2002

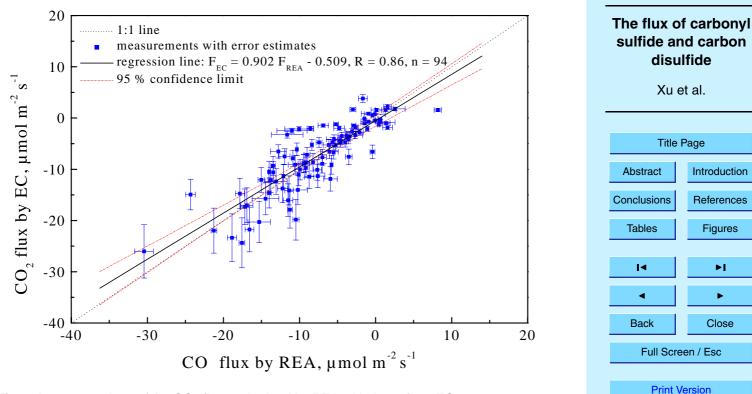


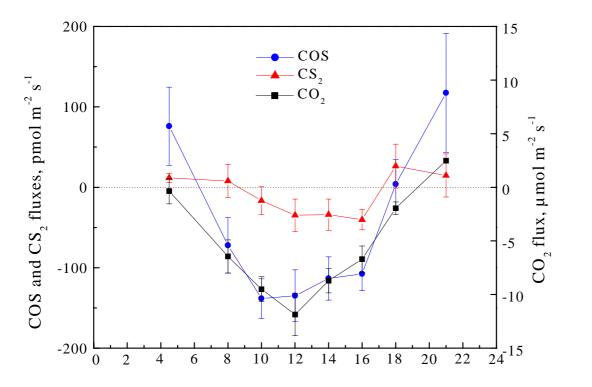
Fig. 4. Intercomparison of the CO₂ fluxes obtained by REA with those from EC measurements.

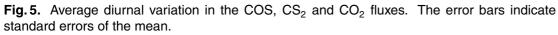
Interactive Discussion

2, 181–212, 2002

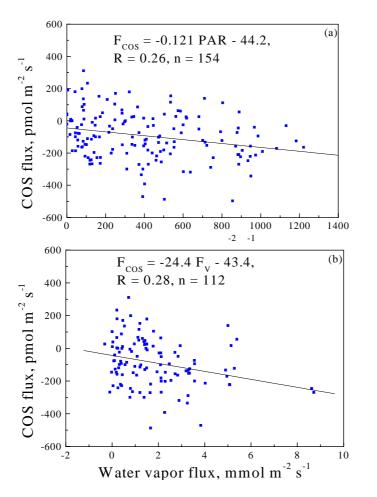
The flux of carbonyl sulfide and carbon

disulfide





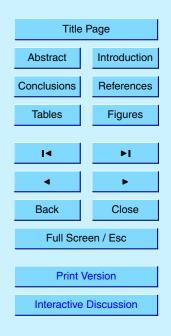




2, 181-212, 2002

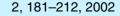
The flux of carbonyl sulfide and carbon disulfide

Xu et al.



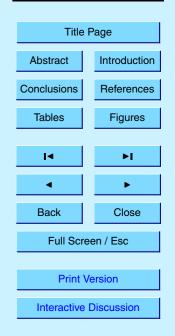
© EGS 2002

Fig. 6. Correlation of the COS flux to (a) photosynthetically active radiation (PAR) and (b) the H_2O flux (F_V).



The flux of carbonyl sulfide and carbon disulfide





© EGS 2002

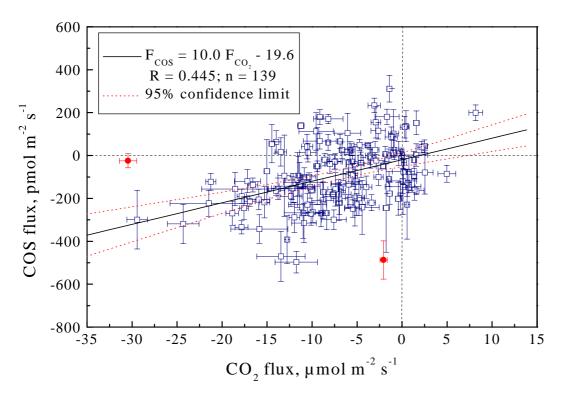


Fig. 7. Correlation between the COS and CO_2 fluxes. The vertical and horizontal bars present the estimated errors of the observed COS and CO_2 fluxes. The two outliers marked with filled circles were not included in the regression since the results are too sensitive to them.