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A normalised seawater strontium isotope curve and the Neoproterozoic-Cambrian chemical weathering event

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A normalised seawater strontium isotope curve

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Abstract

The strontium isotope composition of seawater has varied over geological time in response to changes in the rates of continental weathering relative to ocean crust alteration. However, the potential of the seawater ⁸⁷Sr/⁸⁶ Sr curve to trace globally integrated chemical weathering rates has not been fully realised because ocean ⁸⁷Sr/⁸⁶Sr is also strongly influenced by the isotopic evolution of Sr sources to the ocean. A first attempt is made here to normalise the seawater ⁸⁷Sr/⁸⁶Sr curve to plausible trends in the ⁸⁷Sr/⁸⁶Sr ratios of the three major Sr sources: carbonate dissolution, silicate weathering rates, which can be linked to increased nutrient availability, bioproductivity and oxygenation of Earth's surface environment. Use of normalised seawater ⁸⁷Sr/⁸⁶Sr curves will, it is hoped, help to improve future geochemical models of Earth System dynamics.

15 **1** Introduction

Strontium in modern seawater (⁸⁷Sr/⁸⁶Sr ~0.709) derives from two major sources: the submarine, chemical alteration of ocean crust (⁸⁷Sr/⁸⁶Sr ~0.703: Hofmann, 1997) and the subaerial, chemical weathering of the continental crust and its sedimentary cover (⁸⁷Sr/⁸⁶Sr ~0.712: Palmer and Edmond, 1989; Peucker-Ehrenbrink and Miller, 2006).
²⁰ The ⁸⁷Sr/⁸⁶Sr signatures of these two isotopic end members indicate that about twice as much ocean Sr derives from river runoff (RR) than from hydrothermal exchange (MI). Because seafloor spreading rates, which determine ocean crust production, are not considered to change erratically over geological time (Rowley, 2002), attempts have been made to use the marine carbonate-based seawater ⁸⁷Sr/⁸⁶Sr curve to trace
²⁵ changes in globally integrated chemical weathering rates (Francois and Walker, 1992; Kennedy et al., 2006). This approach is complicated, however, by the fact that seawater



⁸⁷Sr/⁸⁶Sr is affected not only by changes in the rates of continental weathering relative to sea-floor spreading but also by variations in the isotopic composition of river runoff and the upper mantle. Uncertainties in the ⁸⁷Sr/⁸⁶Sr of past river runoff, in particular, undermine the potential of the seawater ⁸⁷Sr/⁸⁶Sr curve to be an unambiguous tracer
 ⁵ of chemical weathering (Veizer and Mackenzie, 2003).

The ⁸⁷Sr/⁸⁶Sr signature of RR is a function of the ratio between carbonate and silicate weathering rates with respect to Sr. Although it is widely held that carbonate weathering is the dominant source of Sr in river runoff (75%: Brass, 1976; 67%: Berner and Rye, 1992; 78%: Veizer and Mackenzie, 2003), variations in the ⁸⁷Sr/⁸⁶Sr ratio of

- the carbonate component in river runoff have generally not been considered in interpretations of first order seawater ⁸⁷Sr/⁸⁶Sr evolution. Here I propose plausible, but not incontrovertible evolutionary trends for the three major Sr sources to the ocean: 1) submarine, elemental exchange with ocean crust (MI); 2) subaerial, chemical weathering of silicate minerals (RRS); and 3) subaerial dissolution of marine carbonate (and a sector of the sector of the
- ¹⁵ marine sulphate) rocks (RRC), in order to illustrate how the seawater ⁸⁷Sr/⁸⁶Sr curve can be normalised against the isotopic evolution of Sr sources to the ocean. The use of more sophisticated, normalised seawater ⁸⁷Sr/⁸⁶Sr curves will it is hoped lead to improved biogeochemical models of the exogenic Earth system.

2 Evolving ⁸⁷Sr/⁸⁶Sr signatures of Sr sources to the ocean

- 1) Exchange between seawater and ocean crust (MI in Fig. 1) occurs generally at midocean ridges and ridge flanks and supplies Sr to the ocean that inherits the isotopic composition of its magmatic precursor, the depleted mantle (Spooner, 1976), which is ~0.703 (Hofmann, 1997). Modern MI represents the accumulation of radiogenic ⁸⁷Sr in the depleted mantle, which deviated early in Earth history from the bulk earth ⁸⁷Sr/⁸⁶Sr
 trend as a result of the preferential incorporation of Rb over Sr into the crust. This
- process led to a lessening of the rate of ⁸⁷Sr/⁸⁶Sr increase in MI, either since the onset of modern tectonics close to the Archaean-Proterozoic boundary (O'Nions, 1979) or



according to some estimates already by c. 3.8 Ga (McCulloch, 1994).

2) Silicate weathering (RRS in Fig. 1) supplies relatively radiogenic Sr to the oceans by virtue of the high Rb/Sr ratios of most rock-forming silicate minerals and the high mean age of the Upper Continental Crust (UCC). However, it has been noted by several authors that the average isotopic composition of surface rocks exposed to weathering appears to be less radiogenic than would normally be expected simply from the radioactive decay of ⁸⁷Rb, e.g. ~0.716 (Goldstein and Jacobsen, 1988) instead of the expected value of >0.730 (Veizer and Mackenzie, 2003), meaning that radiogenic Sr has been lost from the silicate pile over time (Goldstein, 1988). The mean ⁸⁷Sr/⁸⁶Sr ra10 tio of the detrital silicate component in modern rivers is 0.7178 (Bickle, 1994), and this is accented here to be the best estimate for the isotopic signature of silicate-derived.

is accepted here to be the best estimate for the isotopic signature of silicate-derived, dissolved Sr in modern rivers (RRS). Tracing the ⁸⁷Sr/⁸⁶Sr ratio of RRS into the past is difficult because it depends not only on the isotopic evolution of the upper crust but also on the relative susceptibility to weathering, age and Sr content of the various silicate ¹⁵ minerals that make up the exposed parts of the crust.

The Sr isotopic evolution of crustal silicates would normally be expected to deviate from that of the depleted mantle as a mirror image, albeit with a steeper gradient, taking into account the relative size differences of these two Sr reservoirs (O'Nions et al., 1979). However, there is a tendency for Rb and Sr to partition into different mineral phases in the surface environment, which complicates any simplistic trend. Sr is 20 leached preferentially from silicates during chemical weathering and tends to accumulate conservatively in seawater, whereas Rb, although soluble, has a strong affinity to the clay silicate fractions of both soils and fine-grained marine sediments. Although some current thinking (e.g. Kemp et al., 2006) considers that any such surface effects were more than compensated for by the preferential partitioning of Rb into granitic 25 melts during partial melting and fractional crystallisation, there is no consensus on the evolution of crustal ⁸⁷Sr/⁸⁶Sr and it is not the aim of this article to contribute to this complex debate. Accordingly, the RRS trend in Fig. 1a is an idealised curve; however, this should not greatly affect the conclusions of the present article. For example, were



I to assume unchanging RRS ⁸⁷Sr/⁸⁶Sr through time (Goldstein, 1988), this would only dampen the major features of the normalised curve shown in Fig. 1b but not eliminate them.

3) The dissolution of carbonate rocks (limestones and dolomites) and evaporitic sulphates (gypsum, anhydrite) is considered to be the major source of Sr to the oceans 5 (Brass, 1976) by virtue of the high Sr contents and high solubility of those minerals. The isotopic signature of this largely carbonate source today (~0.708) is close to the range of Phanerozoic seawater (~0.707-0.709) and so changes in the rates and isotopic composition of the carbonate weathering flux (RRC) are widely disregarded in interpretations of ocean ⁸⁷Sr/⁸⁶Sr trends (Veizer and Mackenzie, 2003). However, the 10 ⁸⁷Sr/⁸⁶Sr of RRC is largely dependent on past seawater ⁸⁷Sr/⁸⁶Sr and so will have evolved considerably over time. According to recent estimates (Peucker-Ehrenbrink and Miller, 2006), about half of all sediments undergoing weathering at the Earth's surface are younger than 0.25 Ga, while the rest must be aged between 0.25 and 3.5 Ga with a strong bias towards younger Phanerozoic sedimentary rocks Assuming that this 15 skewed age distribution has remained the same since carbonate platforms first became abundant around 2.5 Ga, the ⁸⁷Sr/⁸⁶Sr of RRC can be traced back through time (Fig. 1a). Clearly, the changing seawater ⁸⁷Sr/⁸⁶Sr will have exerted considerable control over the isotopic evolution of river runoff independently of the age distribution of rocks undergoing weathering or crustal evolution. Significantly, seawater and predicted 20 RRC ⁸⁷Sr/⁸⁶Sr reached a maximum divergence of ~0.004 at around 0.5 Ga (Middle Cambrian-Early Ordovician).

3 River runoff versus mantle input through time

Modern-day river runoff ⁸⁷Sr/⁸⁶Sr, at 0.7124 (Palmer and Edmond, 1989; Peucker Ehrenbrink and Miller, 2006), reflects a mixture of Sr from the weathering of carbon ates (⁸⁷Sr/⁸⁶Sr = 0.7077) and silicates (⁸⁷Sr/⁸⁶Sr = 0.7178). Using these estimates, approximately 55% of the Sr in modern rivers derives from the weathering of carbonate



minerals (cf. 60%: Bickle, 1994). This appears to be a conservative estimate, being considerably lower than the initially proposed 75% ratio (Brass, 1976) and is also lower than the 67% ratio used in published models of the exogenic Sr isotope system (Francois and Walker, 1992; Berner and Rye, 1992). In this regard, it can be argued that the

- 5 estimated isotopic composition of RRW is only a minimum constraint as marine carbonate and evaporite minerals generally incorporate more radiogenic Sr during diagenetic alteration. The isotopic composition of the modern RRS flux is not well constrained. In particular, the assumption that the Sr isotopic composition of detrital silicates accurately represents that of the dissolved Sr component (Bickle, 1994) may overestimate the ⁸⁷Sr/⁸⁶Sr of RRS due to the generally greater susceptibility to weathering of less

radiogenic silicate minerals.

In order to use the seawater ⁸⁷Sr/⁸⁶Sr curve as a measure of changes in river runoff (RR) versus mantle Sr input (MI), I have first estimated the ⁸⁷Sr/⁸⁶Sr of RR by assuming that the modern ratio of 45:55 between the silicate and carbonate weathering fluxes has remained the same throughout the past 2.5 Ga. This assumption appears to be

- 15 somewhat valid as the resultant curve of river runoff ⁸⁷Sr/⁸⁶Sr (Fig. 1a) does not cross even the highest peaks of the seawater ⁸⁷Sr/⁸⁶Sr trend. Indeed, the relative influence of carbonate weathering on the isotopic composition of river runoff is likely to have been even greater in the geological past. This is because carbonate dissolution is
- not constrained directly by the endogenic (volcanic and metamorphic) CO₂ flux unlike 20 silicate weathering, and was probably favoured in the geological past due to the greater acidity of rainfall caused by higher atmospheric CO₂ (Kasting and Siefert, 2004), while the absence of land plants and eukaryotic soil biota meant that silicate weathering efficiency was likely depressed during the Precambrian relative to today (Lenton and
- Watson, 2004). The flux of Sr derived from carbonate dissolution seems therefore 25 likely to have dominated over that from silicate weathering for as long as carbonate platforms have been abundant on Earth, i.e. since the early Proterozoic or possibly the late Archean. The relative influence of river runoff versus mantle input (Fig. 1b) can now be estimated by considering seawater ⁸⁷Sr/⁸⁶Sr to be a binary mixture of RR and



MI.

The resultant normalised ⁸⁷Sr/⁸⁶Sr curve in figure 1b displays some significant deviations from the seawater ⁸⁷Sr/⁸⁶Sr curve from which it was derived, and therefore from previous estimates of river runoff versus mantle input. These differences are not particularly sensitive to the age model chosen for carbonate sediments undergoing 5 weathering or to the evolutionary model for the silicate weathering component in rivers. Most importantly, there is no longer any evidence for a stepwise change in the ⁸⁷Sr/⁸⁶Sr base-level (Fig. 1a) across the Proterozoic-Phanerozoic transition, which had been interpreted previously to indicate an irreversible shift in chemical weathering rates due to the evolution of soil biota on the continents (Kennedy et al., 2006). 10 The shift in the base-level noted by those authors can instead be explained by increases in the ⁸⁷Sr/⁸⁶Sr ratios of the Phanerozoic carbonate weathering and mantle fluxes. Another feature of note is the relative deflation of the Cenozoic ⁸⁷Sr/⁸⁶Sr peak compared with the previously equivalent Cambrian⁸⁷Sr/⁸⁶Sr peak. The significance of the Neoproterozoic-Cambrian rise in seawater ⁸⁷Sr/⁸⁶Sr can only be fully appreciated 15 when one considers that the ⁸⁷Sr/⁸⁶Sr compositions of both the MI and the RRC fluxes would have been considerably lower during the Neoproterozoic-Cambrian interval.

4 The Neoproterozoic-Cambrian chemical weathering event

The Neoproterozoic-Cambrian rise in seawater ⁸⁷Sr/⁸⁶Sr (normalised or otherwise) can
be explained by one or more of four different factors: 1) an increase in the ⁸⁷Sr/⁸⁶Sr ratio (and age) of the rocks undergoing weathering above that predicted by the idealised trend shown in figure 1a; 2) a decrease in the mantle Sr input and/or seafloor spreading rates; 3) an increase in overall continental weathering rates; and/or 4) an increase in the silicate/carbonate weathering ratio. As I attempt to show below, the most likely
explanation involves a combination of options 3 and 4.

The excellent correlation between the normalised seawater 87 Sr/ 86 Sr curve and seawater δ^{34} S (Fig. 1b) implies that absolute fluxes and not the isotopic evolution of those



fluxes led to first-order ⁸⁷Sr/⁸⁶Sr trends. This is because the seawater δ^{34} S record is thought to reflect changes in the biogeochemical cycling of sulphur (Strauss, 1999), and so ought to be independent of changes in the ⁸⁷Sr/⁸⁶Sr isotopic ratios of rocks exposed to weathering. In this regard, although tectonic upheavals can feasibly change the globally integrated age (Halverson et al., 2007), and therefore ⁸⁷Sr/⁸⁶Sr of rocks undergoing weathering, it seems that such changes only controlled second order variations in the seawater ⁸⁷Sr/⁸⁶Sr curve. This conclusion is consistent with the widely held interpretation that the Neoproterozoic-Cambrian interval was a time of CO₂ drawdown (and O₂ increase) caused by elevated weathering rates (Derry et al., 1992). A 50% de-

¹⁰ crease in the mantle input flux would have been sufficient to cause the Neoproterozoic-Cambrian rise. However, such a large, sustained decrease in seafloor spreading rates seems improbable during a period of exceptionally rapid continental reconfiguration, generally high sea-level and widespread rifting (Kirschvink et al., 1997). The most plausible explanation for the Neoproterozoic-Cambrian rise is therefore a sustained increase in overall continental and/or silicate weathering rates.

Times of low continental influence on ocean ⁸⁷Sr/⁸⁶Sr correlate with the existence of the supercontinents Rodinia and Pangaea, while their break-up is associated with sustained increases in continental influence on the ocean Sr isotope budget. One possible explanation for this connection is that increased rates of physical weathering due

- to tectonic uplift led to higher overall chemical weathering rates during times of supercontinent break-up and microcontinent collision (Jacobsen and Kaufman, 1999; Squire et al., 2006). In support of this interpretation, the normalised ⁸⁷Sr/⁸⁶Sr curve correlates well with sediment flux rates (Hay et al., 2001) and rifting events (D. S. Bradley, personal communication), while the rise to peak ⁸⁷Sr/⁸⁶Sr values correlates with the formation of huge mountain chains, such as the "Transgondwanan Supermountain" (Squire et al.,
- 25 Huge mountain chains, such as the mansgondwaran supermountain (Squire et al., 2006). However, sustained increases in silicate weathering rates are only thought to be possible so long as increased CO_2 fluxes accompany them (Walker et al., 1981), in which case increased rates of CO_2 degassing from carbonate metamorphism (Bickle, 1994) and granitisation during, may be relevant. Increases in the efficiency of sili-



cate weathering may also have occurred during the Neoproterozoic-Cambrian interval, caused by the evolution of soil biota (Kennedy et al., 2006; Lenton and Watson, 2004); however, any such irreversible change in the silicate versus carbonate weathering ratio must have been accompanied by a decline in overall continental weathering rates after the Cambrian Period.

It has long been noted that the long-term seawater 87 Sr/ 86 Sr record correlates positively with δ^{34} S during the Phanerozoic and negatively with δ^{13} C and this has been taken to indicate a tectonic control on global C- and S-biogeochemical cycles (Veizer et al., 1999). The normalised 87 Sr/ 86 Sr curve improves the correlation with δ^{34} S and extends it back to c. 800 Ma (Fig. 1b), which implies that increased chemical weathering rates led to higher rates of sulphate reduction (and pyrite burial), presumably linked to higher rates of organic carbon degradation. The alternative proposition that rates of weathering might control seawater δ^{34} S directly by changing the riverine sulphate flux can be excluded as this would lead to an inverse correlation between the two iso-

- topic parameters (Bottrell and Newton, 2006). Correlation between globally integrated organic degradation rates and enhanced chemical weathering rates is consistent with the notion that bioproductivity is limited over geological time scales by the availability of phosphorus (Lenton and Watson, 2000), which can only be derived from the weathering of rocks. The onset of the ⁸⁷Sr/⁸⁶Sr rise also marks the beginning of unusually high
- 20 δ^{13} C values that were set to become characteristic of the later Neoproterozoic (Halverson et al., 2007). Thus it would appear that this sustained, weathering-driven increase in productivity also led to higher rates of organic carbon burial, and so presumably was the major driver towards the further oxygenation of the surface environment at this time (Derry et al., 1992).

Recent studies have emphasised the role of terrestrial biota in increasing both P weathering rates (Lenton and Watson, 2004) and overall silicate weathering rates (Kennedy et al., 2006) during the Neoproterozoic-Cambrian interval. Although the evolution of terrestrial eukaryotes by 600 Ma (Heckman et al., 2001; Yuan et al., 2005) may be one cause for the increase in weathering rates, their influence was clearly reversible

2, 69-84, 2007 A normalised seawater strontium isotope curve G. A. Shields **Title Page** Introduction Abstract Conclusions References Tables Figures Back Close Full Screen / Esc Printer-friendly Version Interactive Discussion

as seawater 87 Sr/ 86 Sr decreased steadily after its Late Cambrian peak. This may be due to the establishment of a new equilibrium between more efficient drawdown of atmospheric CO₂ by silicate weathering and the endogenic CO₂ flux. A similar scenario has been suggested to take place after the evolution of vascular land plants during the later Palaeozoic, whereby the evolution of land plants is considered by some to have led to eutrophication, CO₂ disequilibrium and the contemporaneous Permo-Carboniferous glaciations, and higher atmospheric oxygen levels (Lenton, 2001).

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The existence of prolonged pulses of chemical weathering may explain the hitherto enigmatic age distribution of phosphorus-rich rocks on Earth. Sedimentary phospho-

- rite rocks of economic importance have only been deposited during four intervals of Earth history (Shields et al., 2000). The c. 2.0 Ga Paleoproterozoic, Neoproterozoic-late Cambrian (0.65–0.5 Ga) and mid-Cretaceous-Recent (0.1–0.0 Ga) intervals, were true P-giant episodes (Fig. 1b) with phosphorus-rich rocks and related organic-rich strata distributed worldwide. The other major deposit, the Phosphoria Formation, is
 limited in areal distribution and formed during the Permian icehouse. All three P-giant
- episodes stand out as times of enhanced chemical weathering rates according to the normalised ⁸⁷Sr/⁸⁶Sr record.

This new reading of the seawater 87 Sr/ 86 Sr record suggests that its first order trend records changes in globally integrated chemical weathering rates, while its second order features are more likely to be related to changes in the isotopic composition of source fluxes (Jones and Jenkyns, 2001). By association, long-term trends in seawater δ^{34} S reflect variations in ocean productivity driven by changes in the riverine nutrient flux. Four possible chemical weathering events can be recognised in Earth his-

tory, representing periods of disequilibrium and elevated organic productivity. Of these,
 the Neoproterozoic-Cambrian interval stands out as a time when increased chemical weathering rates pushed the Earth System to a new state characterised by higher O₂ and lower CO₂ and accompanied by the appearance of new forms of animal, plant and fungal life.



5 Conclusions

The carbonate dissolution component to river runoff has never been considered in interpretations of first order trends in the seawater ⁸⁷Sr/⁸⁶Sr curve despite the dominance of this component today and the certainty that it has must have changed greatly over

- ⁵ Earth history. When predictable trends in this parameter and other factors influencing seawater ⁸⁷Sr/⁸⁶Sr are considered, it appears that there was a reversible trend towards unusually high rates of chemical weathering during the late Neoproterozoic and Cambrian. These high rates may help to explain the evidence for eutrophication around this time in the form of massive phosphorite deposits, black shales, high organic carbar have have been been as the state of the state of
- ¹⁰ bon burial rates (high δ^{13} C) and high organic degradation rates (high δ^{34} S). Sustained high chemical weathering rates and related CO₂ drawdown are both consistent with the notion that Earth's exogenic system became oxygenated over this interval, thus paving the way for the further evolution of animals.

References

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¹⁵ Berner, R. A. and Rye, D. M.: Calculation of the Phanerozoic strontium isotope record of the oceans from a carbon cycle model, Am. J. Sci., 292, 136–148. 1992.

Bickle, M. J.: The role of metamorphic decarbonation reactions in returning strontium to the silicate sediment mass, Nature, 367, 699–704, 1994.

Bottrell, S. H. and Newton, R. J.: Reconstruction of changes in global sulphur cycling from marine sulphate isotopes, Earth Sci. Rev. 75, 59–83, 2006.

- Brass, G. W.: The variation of the marine ⁸⁷Sr/⁸⁶Sr ratio during Phanerozonic time: interpretation using a flux model, Geochim. Cosmochim. Acta, 40, 721–730, 1976.
- Derry, L. A., Kaufman, A. J., and Jacobsen, S. B.: Sedimentary cycling and environmental change in the Late Proterozoic: Evidence from stable and radiogenic isotopes, Geochim. Cosmochim. Acta, 56, 1317–1329, 1992.
- Francois, L. M. and Walker, J. C. G.: Modelling the Phanerozoic carbon cycle and climate: constraints from the ⁸⁷Sr/⁸⁶Sr isotopic ratio of seawater, Am. J. Sci., 292, 81–135, 1992.

2, 69-84, 2007 A normalised seawater strontium isotope curve G. A. Shields **Title Page** Introduction Abstract Conclusions References Tables Figures Back Close Full Screen / Esc Printer-friendly Version Interactive Discussion

- Goldstein, S. J. and Jacobsen, S. B.: Nd and Sr isotopic systematics of river water suspended material: Implications for crustal evolution, Earth Planet. Sci. Lett., 87, 249–265, 1988.
- Goldstein, S. L.: Decoupled evolution of Sr and Nd isotopes in the continental crust and mantle, Nature, 336, 733–738, 1988.
- ⁵ Halverson, G., Dudas, F.Ö., A. C. Maloof, and Bowring, S. A.: Evolution of the ⁸⁷Sr/⁸⁶Sr composition of Neoproterozoic seawater, Palaeogeog. Palaeoclimat., Palaeoecol, in press, 2007.
 Hay, W. W., Wold, C. N., Söding, E., and Flögel, S.: Evolution of sedimentary fluxes and ocean salinity, in: Geologic modelling and simulation: sedimentary systems, edited by: Merriam, D. F. and Davis, J. C., Kluwer Academic/Plenum Publishers, New York, pp. 153–167 (2001).
- Heckman, D. S., Heckmann, D. S., Geiser, D. M., Eidell, B. R., et al.: Molecular evidence for the early colonization of land by plants and fungi, Science, 291, 1129–1133, 2001.
 Hofmann, A. W.: Mantle geochemistry: the message from ocean volcanism, Nature, 385, 219– 229, 1997.

Hough, M. L., Shields, G. A., Evins, L. Z., Strauss, H., Henderson, R. A., and Mackenzie, S.:

A major sulphur isotope at c. 510 Ma: a possible connection anoxia-extinction-volcanism connection during the Early-Middle Cambrian transition?, Terra Nova, 18, 257–263, 2006.
 Jacobsen, S. B. and Kaufman, A. J.: The Sr, C and O isotopic evolution of Neoproterozoic seawater, Chem. Geol., 161, 37–57, 1999.

Jones, C. E. and Jenkyns, H. C.: Seawater Strontium Isotopes, Oceanic Anoxic Events, and

- ²⁰ Seafloor Hydrothermal Activity in the Jurassic and Cretaceous, Am. J. Sci., 312, 112–149, 2001.
 - Kamber, B. A. and Webb, G. M.: The geochemistry of late Archaean microbial carbonate: Implications for ocean chemistry and continental erosion history, Geochim. Cosmochim. Acta, 65, 2509–2525, 2001.
- Kampschulte, A. and Strauss, H.: The sulfur isotopic evolution of Phanerozoic seawater based on analysis of structurally substituted sulfate in carbonates, Chem. Geol. , 204, 255–286, 2004.

Kasting, J. F. and Siefert, J. L.: Life and the evolution of Earth's atmosphere, Science, 296, 1066–1068, 2004.

Kemp, A. I. S., Hawkesworth, C. J., Paterson, B. A., and Kinny, P.: Episodic growth of the Gondwana Supercontinent from hafnium and oxygen isotopes in zircon, Nature, 439, 580– 583, 2006.

Kennedy, M., Droser, M., Meyer, L. M., Pevear, D., and Mrofka, D.: Late Precambrian oxygena-

eED 2, 69–84, 2007		
A normalised seawater strontium isotope curve		
G. A. Shields		
Title Page		
Abstract	Introduction	
Conclusions	References	
Tables	Figures	
I	▶1	
•	· ·	
Back	Close	
Full Screen / Esc		
Printer-friendly Version		
Interactive Discussion		

tion: inception of the clay mineral factory, Science 311, 1446–1449, 2006.

- Kirschvink, J.L., Ripperdan, R. L., and Evans, D. A. D.: Evidence for a large-scale reorganization of Early Cambrian continental masses by inertial interchange true polar wander, Science, 277, 541–545, 1997.
- ⁵ Lenton, T. M.: The role of land plants, phosphorus weathering and fire in the rise and regulation of atmospheric oxygen, Global Change Biology, 7, 613–629, 2001.
 - Lenton, T. M. and Watson, A. J.: Redfield revisited 1: Regulation of nitrate, phosphate and oxygen in the ocean, Global Biogeochem. Cycles, 14, 225–248, 2000.

Lenton, T. M. and Watson, A. J.: Biotic enhancement of weathering, atmospheric oxy-

- ¹⁰ gen and carbon dioxide in the Neoproterozoic, Geophys. Res. Lett., 31, L05202, doi:10.1029/2003GL018802, 2004.
 - McCulloch, M. T.: Primitive ⁸⁷Sr/⁸⁶Sr from and Archean barite and conjecture on the Earth's age and origin, Earth Planet. Sci. Lett., 126, 1–13, 1994.

O'Nions, R. K., Evenson, N. M., and Hamilton, P. J.: Geochemical modelling of mantle differentiation and crustal growth, J. Geophys. Res., 84, 6091–6101, 1979.

Peucker-Ehrenbrink, B. and Miller, M. W.: Marine ⁸⁷Sr/⁸⁶Sr record mirrors the evolving upper continental crust. Geochim. Cosmochim Acta, 70, 18, A487, 2006.

15

- Palmer M. R. and Edmond, J. M.: Strontium isotope budget of the modern ocean, Earth Planet. Sci. Lett., 92, 11–26, 1989.
- Rowley, D.: Rate of plate creation and destruction, Geol. Soc. Am. Bull., 114, 927–933, 2002. Shields, G. A., Stille, P., and Brasier, M. D.: Isotopic records across two phosphorite giant episodes compared: the Precambrian-Cambrian and the late Cretaceous-recent, in: Marine authigenesis: From Global to Microbial, edited by: Glenn, C., Prévôt-Lucas, L., Lucas, J., SEPM special volume 66, p. 103–116, 2000.
- ²⁵ Shields, G. A. and Veizer, J.: Precambrian marine carbonate isotope database: Version 1.1, Geochem., Geophys., Geosyst., 6, 1–12, 2002.
 - Spooner, E. T. C.: The strontium isotopic composition of seawater, and seawater-oceanic crust interaction, Earth Planet. Sci. Lett., 31, 167–174, 1976.

Squire, R. J., Campbell, I. H., Allen, C. M., and Wilson, C. J. L.: Did the Transgondwanan

³⁰ Supermountain trigger the explosive radiation of animals on Earth?, Earth Planet. Sci. Lett., 250, 116–133, 2006.

Strauss, H.: Geological evolution from isotope proxy signals – sulphur, Chem. Geol., 161, 89– 101, 1999.

2, 69–84, 2007		
A normalised seawater strontium isotope curve G. A. Shields		
Title Page		
Abstract	Introduction	
Conclusions	References	
Tables	Figures	
I	۶I	
•		
Back	Close	
Full Screen / Esc		
Printer-friendly Version		
Interactive Discussion		

Veizer, J. and Mackenzie, F. T.: Evolution of sedimentary rocks, in: , Treatise on geochemistry, edited by: Mackenzie, F. T. (Volume editor), Volume 7, Sediments, diagenesis, and sedimentary rocks, 369–407, 2003.

Veizer, J., Ala, D., Azmy, K., et al.: 87 Sr/ 86 Sr, δ^{13} C and δ^{18} O evolution of Phanerozoic seawater, Chem. Geol., 161, 59–88, 1999.

5

Walker, J. C. G., Hays, P. B., and Kasting, J. F.: A negative feedback mechanism for the long-term stabilization of Earth's surface temperature, J. Geophys. Res., 86, 9776–9782, 1981.

Yuan, X., Xiao, S., and Taylor, T. N.: Lichen-like symbiosis 600 million years ago, Science, 308, 1017–1020, 2005.

2, 69-84, 2007 A normalised seawater strontium isotope curve G. A. Shields Title Page Introduction Abstract Conclusions References Tables Figures Back Close Full Screen / Esc Printer-friendly Version Interactive Discussion

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Fig. 1.

Fig. 1. Sr isotope evolution of seawater and ocean fluxes (A) and its implications for the relative influence of continental weathering on ocean composition (B). The seawater ⁸⁷Sr/⁸⁶Sr curve (Veizer et al., 1999; Shields and Veizer, 2002) in (A) has been constructed from the lowermost ratios for each time interval. The river runoff (RR) curve has been determined by assuming a modern-like 9:11 relationship between Sr input from marine carbonate weathering (RRC) and silicate weathering (RRS), respectively. The RRC curve assumes that sediments undergoing weathering have a skewed age distribution, with a mean age of 250 Ma (Peucker-Ehrenbrink and Miller, 2006), and so lags seawater 87 Sr/ 86 Sr by >250 Ma. The RRS curve is an idealistic representation based on predicted crustal evolution (O'Nions et al., 1979), other authors assume much earlier crustal Rb/Sr differentiation with minimal isotopic evolution (Kamber and Webb, 2001). Ocean crust alteration provides less radiogenic Sr to the oceans (MI). The curve in (B) assumes that seawater ⁸⁷Sr/⁸⁶Sr results from simple binary mixing between RR and MI, and shows that the influence of continental weathering was less prior to 2.5 Ga, and negligible prior to 3.0 Ga. The normalised 87 Sr/ 86 Sr and seawater δ^{34} S curves (Kampschulte and Strauss, 2004; Hough et al., 2006) have been smoothed by plotting values for every 50 million years only. P-giant episodes (Shields et al., 2000).

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